



**JSS COLLEGE OF ARTS, COMMERCE & SCIENCE
(AUTONOMOUS)**

OOTY ROAD, MYSURU-570 025

(Autonomous under University of Mysore: Re-accredited by NAAC with 'A' Grade)

MASTERS OF VOCATIONAL COURSE

M.Voc. (Software Development)

Department of M.Voc.

JSS College of Arts, Commerce and Science

Ooty Road, Mysore-25

2018

PROGRAMME
MASTER OF VOCATIONAL COURSE - Software Development
2 Years/4 Semesters

PROGRAMME OUTCOMES:

After Completing the M.Voc. Program Students will be able to:

- PO1. Contribute to the design of embedded software modules
- PO2. Develop software code to specification
- PO3. Create documents for knowledge sharing
- PO4. Manage your work to meet requirements
- PO5. Work effectively with colleagues
- PO6. Maintain a healthy, safe and secure working environment
- PO7. Provide data/information in standard formats
- PO8. Develop knowledge, skills and competence

Contribute to the design of embedded software modules

C++, Embedded C#

NOS Code : SSC/N6004

Course Outcomes: At the end of the course students will be able to:

- CO1. Establish your role and responsibilities in contributing to the design of embedded software modules.
- CO2. Establish clearly module requirements and constraints
- CO3. Identify any issues with module requirements or constraints and clarify these with appropriate people
- CO4. Document designs using standard templates and tools
- CO5. Update your organization's knowledge base with new designs.
- CO6. Obtain advice and guidance on designing embedded software modules from appropriate people, where required.

Amazon web services

Hr:-15

- Introduction to cloud computing
- What is AWS
- Success stories
- AWS services
 - EC2
 - ELB
 - S3
 - CLOUD FRONT
 - OTHER SERVICES

Develop Software code to specification

C, Java

NOS Code : SSC/N0502

Course Outcomes: At the end of the course students will be able to:

- CO1. Check your understanding of the Business Requirements specification (BRS), Software Requirements Specification (SRS), High Level Design (HLD) and Low Level Design (LLD) with appropriate people access reusable components, code generation tools and unit testing tools from your organization's knowledge base.
- CO2. Convert technical specifications into code to meet the requirements, leveraging reusable components, where available
- CO3. Create appropriate Unit test cases (UTCs)
- CO4. Review codes and UTCs with appropriate people
- CO5. Analyze inputs from appropriate people to inform future designs
- CO6. Submit tested code for approval by appropriate people

Mango DB on AWS

Hr:-15

- Introduction
- Creating, reading and updating data
- Schema design

- Performance
- Aggregation framework
- Application engineering
- Case studies

Create documents for knowledge sharing

TCL, Squirrel

NOS Code : SSC/N0703

Course Outcomes: At the end of the course students will be able to:

- CO1. Establish with appropriate people the purpose, scope, formats and target audience for the documents
- CO2. Access existing documents, language standards, templates and documentation tools from your organization's knowledge base.
- CO3. Liaise with appropriate people to obtain and verify the information required for the documents
- CO4. Create documents using standard templates and agreed language standards
- CO5. Review documents with appropriate people and incorporate their inputs
- CO6. Publish documents in agreed formats
- CO7. Update your organization's policies, procedures and guidelines when creating documents for knowledge sharing

R programming language

Hr:-15

- A first session in R
- Getting data into R
- Basic data manipulation
- Basic plotting 1
- Basic plotting 2
- Loops and functions
- Basic stats1
- Basic stats2
- Advance data manipulation

- Final project presentation

Manage your Work to meet requirements

NOS Code : SSC/N9001

Course Outcomes: At the end of the course students will be able to:

- CO1. establish and agree your work requirements with appropriate people
- CO2. keep your immediate work area clean and tidy
- CO3. utilize your time effectively
- CO4. use resources correctly and efficiently
- CO5. treat confidential information correctly
- CO6. work in line with your organization's policies and procedures
- CO7. work within the limits of your job role
- CO8. obtain guidance from appropriate people, where necessary
- CO9. ensure your work meets the agreed requirements

Business English

Hr:-30

- Understanding company structures
- Developing relationships in the work place
- Correspondence- E-mailing written Reports and telephoning
- Public speaking and presentations
- Meeting and Negotiations

Work effectively with colleagues

NOS Code : SSC/N9002

Course Outcomes: At the end of the course students will be able to:

- CO1. communicate with colleagues clearly, concisely and accurately
- CO2. work with colleagues to integrate your work effectively with theirs
- CO3. pass on essential information to colleagues in line with organizational requirements
- CO4. work in ways that show respect for colleagues
- CO5. carry out commitments you have made to colleagues
- CO6. let colleagues know in good time if you cannot carry out your commitments, explaining the reasons

CO7. identify any problems you have working with colleagues and take the initiative to solve these problems

CO8. follow the organization's policies and procedures for working with colleagues

Presentation skills

Hr:-30

- Understand basic patterns of an effective presentation
- Create and deliver an effective presentation
- Asking questions
- Answering questions
- Evaluate presentation
- Case study- Entrepreneurship development

Meeting management skills

- Understanding meeting management
- Types of meeting
- Setting meeting objectives
- Greetings
- Conduct of meeting
- Meeting closure
- Minutes of meeting
- Meeting follow up

Maintain a healthy safe and secure working environment

NOS Code : SSC/N9003

Course Outcomes: At the end of the course students will be able to:

- CO1. comply with your organization's current health, safety and security policies and procedures
- CO2. report any identified breaches in health, safety, and security policies and procedures to the designated person
- CO3. identify and correct any hazards that you can deal with safely, competently and within the limits of your authority
- CO4. report any hazards that you are not competent to deal with to the relevant person in line with organizational procedures and warn other people who may be affected
- CO5. identify and recommend opportunities for improving health, safety, and security to the designated person
- CO6. complete any health and safety records legibly and accurately

Time management skills

Hr:-30

- Beginnings
- Values and beliefs
- Goals and benchmarks – The ladders of success
- Managing projects and commitments
- Prioritizing your to DO's
- Designing the projects that matters
- Inspired action- getting results you need
- Tracking projects
- Managing for Accomplishment
- Tools that work

Provide data/information in standard formats

NOS Code : SSC/N9004

Course Outcomes: At the end of the course students will be able to:

- CO1. establish and agree with appropriate people the data/information you need to provide, the formats in which you need to provide it, and when you need to provide it
- CO2. obtain the data/information from reliable sources
- CO3. check that the data/information is accurate, complete and up-to-date
- CO4. obtain advice or guidance from appropriate people where there are problems with the data/information
- CO5. carry out rule-based analysis of the data/information, if required
- CO6. insert the data/information into the agreed formats

Term work skills

- Team structure
- Stages of team
- Traditional teams VS Collaborative teams
- Taking team action and problem solving
- Team communication
- Conflict resolution
- Mediation
- Role-specific issues
- Projects presentation

Develop your knowledge , skills and competence

NOS Code : SSC/N9005

Course Outcomes: At the end of the course students will be able to:

- CO1. obtain advice and guidance from appropriate people to develop your knowledge, skills and competence
- CO2. identify accurately the knowledge and skills you need for your job role
- CO3. identify accurately your current level of knowledge, skills and competence and any learning and development needs

CO4. agree with appropriate people a plan of learning and development activities to address your learning needs

CO5. undertake learning and development activities in line with your plan

CO6. apply your new knowledge and skills in the workplace, under supervision

Communication skills

- Communication in English
- Listening skills
- Reading skills
- Writing skills
- Speaking skills.

General Education Component3:

- Project estimation skills
- Understanding project estimation
- How to estimate time accurately
- Methods of estimating time
- Preparing your scheduling

Choice Based Credit System & Continuous Assessment Grading Pattern

SYLLABUS FOR MASTER OF ARTS IN ENGLISH



Programme Code: ENG

2020-21

JSS COLLEGE OF ARTS, COMMERCE & SCIENCE

(An Autonomous College of University of Mysore; Re-Accredited by NAAC with 'A' Grade)

**POSTGRADUATE DEPARTMENT OF STUDIES & RESEARCH IN
ENGLISH**

OOTY ROAD, MYSURU-570 025, KARNATAKA

**POSTGRADUATE DEPARTMENT OF STUDIES & RESEARCH IN
ENGLISH**

COURSES OFFERED

I SEMESTER

Course I – (Hard Core-I) English Literature from Chaucer to Milton	ENA 010
Course II – (Hard Core-II) Elizabethan Age	ENA 020
Course III – (Hard Core-III) 17th and 18th Century English Literature	ENA 030
Course IV – (Hard Core-IV) 19th Century English Literature	ENA 040
Course V – *(Soft Core-I) Realism and Fiction	ENA 250
*(Soft Core-II) Contemporary Indian Novels in English	ENA 220

II SEMESTER

Course I - (Hard Core – I) The Modern Age-I	ENB 040
Course II - (Hard Core-II) Literary Criticism-I	ENB 020
Course III – (Hard Core-III) Indian Writing in English – I	ENB 030
Course IV – (Hard Core-IV) 20th Century Women’s Writing: Theory & Practice	ENB 050
Course V – *(Soft Core-I) Dalit Literature	ENB 230
*(Soft Core-II) English Essayists	ENB 220

Note: * Soft Core Elective Courses

III SEMESTER

Course I – (Hard Core-I) The Modern Age-II	ENC 010
Course II – (Hard Core-II) Indian Writing in English-II	ENC 020
Course III – (Hard Core-III) New Literatures in English	ENC 030
Course IV (Open Electives) A Course in Written and Spoken English	OE
Course V –* (Soft Core-I) Indian English Poetry After Independence	ENC 230
*(Soft Core-II) Feminism	ENC 220

IV SEMESTER

Course I – (Hard Core-I) Literary Criticism-II	END 010
Course II – (Hard Core-II) American Literature	END 020
Course III – (Hard Core) Major Project Work leading to a Dissertation	END 030
Course IV – *(Soft Core I) Indian Diaspora Fiction	END 240
*(Soft Core II) African Fiction	END 220

Note: * Soft Core Elective Courses

**POSTGRADUATE DEPARTMENT OF STUDIES &
RESEARCH IN ENGLISH**

Choice Based Credit System & Continuous Assessment Grading Pattern Syllabus

MA PROGRAMME IN ENGLISH

2021-24

Semester	Type of Course	Course Title	L	T	P	Credits	Credits required
First Semester	Hard Core	1. English Literature from Chaucer to Milton	3	1	0	4	16
		2. Elizabethan Age	3	1	0	4	
		3. 17 th and 18 th Century English Literature	3	1	0	4	
		4. 19 th Century English Literature	3	1	0	4	
	Soft Core Electives	1. Realism and Fiction	3	1	0	4	4
		2. Contemporary Indian Novels in English	3	1	0	4	
Total Credits							20
Second Semester	Hard Core	1. The Modern Age-I	3	1	0	4	16
		2. Literary Criticism-I	3	1	0	4	
		3. Indian Writing in English – I	3	1	0	4	
		4. 20 th Century Women's Writing: Theory & Practice	3	1	0	4	
	Soft Core Electives	1. Dalit Literature	3	1	0	4	4
		2. English Essayists	3	1	0	4	
Total Credits							20

Semester	Type of Course	Course Title	L	T	P	Credits	Credits required
Third Semester	Hard Core	1. The Modern Age-II	3	1	0	4	12
		2. Indian Writing in English-II	3	1	0	4	
		3. New Literatures in English	3	1	0	4	
	Soft Core Electives	1. Indian English Poetry After Independence	3	1	0	4	4
		2. Feminism	3	1	0	4	
	Open Elective	1. A Course in Written and Spoken English	3	1	0	4	4
Total Credits							20
Fourth Semester	Hard Core	1. Literary Criticism-II	3	1	0	4	12
		2. American Literature	3	1	0	4	
		3. Major Project Work leading to Dissertation	0	0	4	4	
	Soft Core Electives	1. Indian Diaspora Fiction	3	1	0	4	4
		2. African Fiction	3	1	0	4	
Total Credits							16
Total Credits at the end of the Course							76

Programme Outcomes

On graduating from this programme student is able to

- PO1: Learn to write logically relating the real-life scenario with the issues depicted in literary texts.
- PO2: Demonstrate critical reading, writing, and thinking skills in writing analytical essays.
- PO3: Recognise and understand figurative language in literary works of various literatures.
- PO4: Equips the students to understand theoretical developments in literary studies.
- PO5: Acquire skills of criticism in reading literary works of different periods of various genres.
- PO6: Write focused, organised, well-developed text-based essays.

Programme Specific Outcomes

On completion, the graduate of this programme is able to

- PSO1: Acquire the competence to work as English Language teacher at Primary, Secondary, Higher secondary and Pre-University level.
- PSO2: Gain basic knowledge needed to enroll for M Phil or PhD programmes.
- PSO3: Demonstrate good communication skills.
- PSO4: Draft literary essays demonstrating the skills of critical thinking and creative writing.
- PSO5: Participate in discussions and debates demonstrating good communication skills.
- PSO6: Learn the skills to work as English language trainer.

Course I – (HC-I) ENGLISH LITERATURE FROM CHAUCER TO MILTON

Course Code: ENA 010

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

- CO1: Recognize and understand figurative language, such as allegory and metaphor, and literary techniques, such as irony, rhyme, and allusion.
- CO2: Identify the unique qualities of the authors studied, and compare and contrast them.
- CO3: Develop a well-written argument about one or more literary texts or authors, and accurately cite literary and other sources.
- CO4: Develop ability to read, summarize and critically analyse poems and sonnets of various themes.

UNIT – I

- 1. Background – 14th Century – 1658
- 2. Renaissance, Reformation, Puritan Upsurge
- 3. Jacobean, Metaphysical School

UNIT – II

- 1. Chaucer -- Prologue to the Canterbury Tales
- 2. John Milton – Paradise Lost: Book-I
- 3. Book of Job

UNIT – III

- 1. Sidney: Sonnets I, V & VI (Astrophel and Stella)
- 2. Spenser: Prothalamion, Epithalamion
- 3. Surrey: 1. Love that liveth and reigneth in my thought
2. Set me whereas the sun doth parch the green
- 4. Wyatt: 1. Who so list to hunt 2. They flee from me

UNIT – IV

- 1. John Donne – The Goodmorrow, Song: Go and Catch a Falling Star, Death be Not Proud, At the Round Earth's Imagined Corner, Sunne Rising, Resurrection
- 2. George Herbert – The Pulley, The Collar, Virtue, Discipline
- 3. Andrew Marvell – To His Coy Mistress, Thoughts in a Garden
- 4. Robert Herrick- To the Virgins, To Find God.

Books for Reference and Further Reading:

- 1. David Daiches – *A Critical History of English Literature*- Four volumes.
- 2. Boris Ford (Ed) - *Pelican Guide to English Literature*- Eight volumes.
- 3. Herbert Grierson - *Metaphysical Poets*.
- 4. Abrams M H et al. *The Norton Anthology of English Literature*. New York: Norton, 2006.
- 5. Al Poplaski: *English Literature in Context*.

Course II – (HARD CORE-II) ELIZABETHAN AGE

Course Code: ENA-020

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Understand the origin and growth of English Theatres and Renaissance plays.

CO2: Demonstrate knowledge of Elizabethan culture, society and politics.

CO3: Analyse Shakespearean Tragedies and Comedies in terms of language, character and Themes.

CO4: Develop ability to read, summarize and critically analyse Shakespearean sonnets on various themes.

UNIT – I

Background – Elizabethan Age

Elizabethan theatre and audience Shakespeare- Tragedy and Comedy

UNIT – II

Marlowe: *Dr. Faustus*

Ben Jonson: *Volpone*

Thomas Kyd: *Spanish Tragedy*

UNIT – III

William Shakespeare: 1) *Macbeth*, 2) *Julius Caesar*, 3) *As you Like It*

UNIT – IV

Shakespeare's Sonnets: **18, 29, 30, 33, 60, 71, 73, 114, 116.**

On Love: 29, 114, and 116, On Death: 18, 19, and 55 On Time: 33, 60, and 73

On Dark Lady: 130, 131, and 133

Books for Reference and Further Reading:

1. A.C. Bradley – *Shakespearean Tragedy*
2. F R Leavis – *The Common Pursuit*
3. Wilson Knight – *The Wheel of Fire*
4. Stewart Justman – *Shakespeare: The Drama of Generations*
5. S. Vishwanathan - *Exploring Shakespeare*
6. Cleanth Brooks - *Understanding Drama*
7. Toropov - *Shakespeare for Beginners*

Course III – (HARD CORE-III) 17th and 18th CENTURY ENGLISH LITERATURE

Course Code: ENA 030

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

- CO1: Apply knowledge of the historical and cultural contexts of the literature of this period to comprehend the works of major authors.
- CO2: Recognize and understand figurative language, such as allegory and metaphor, and literary techniques - irony, rhyme, and allusion.
- CO3: Reflect and write analytically about the literary texts and their contexts.
- CO4: Develop skills of literary critical analysis reading the prescribed plays, novels and essays.

UNIT –I

Background – Restoration, Neo-Classical, Augustan Satire, Comedy of Manners Spectator Essays

UNIT – II

Dryden – *Absalom and Achitophel*

Alexander Pope – *The Rape of the Lock*

Bunyan – *Pilgrim's Progress Book-1*

UNIT – III

William Congreve - *The Way of the World*

Sheridan – *The School for Scandal*

Aphra Behn- *Oroonoko (Royal Author)*

UNIT – IV

Daniel Defoe: *Robinson Crusoe*

Jonathan Swift – *Gulliver's Travels – Book IV* (Voyage to the Land of Houyhnhnms)

Addison & Steele: Spectator Essays

Addison: (1) Sir Roger at Church, (2) Sir Roger at Assizes

Steele: (1) The Gentleman 2) The Spectator Club

Books for Reference and Further Reading:

1. M.H. Abrams (Ed) *The Norton Anthology of English Literature* (Vol.1 & 2)
2. David Daiches – *A Critical History of English Literature* –Four volumes
3. Arnold Kettle- *The English Novel*- Two volumes
4. Ian Jack – *The Augustan Satire: Intention and Idiom in English poetry 1660-1750*
5. Pramod Nayyar (ed) – *English Poetry 1660-1780: An Anthology*

Course IV (Hard Core – IV) – 19th CENTURY ENGLISH LITERATURE

Course Code: ENA 40

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Understand the impact of French Revolution on Romantic and Victorian age.

CO2: Learn the issues related to Woman's Question during the period and contributions of Mary Wollstonecraft and J S Mill to this movement.

CO3: Recognize and understand the use of allegory, metaphor, irony, rhyme, rhythm, allusion in Romantic and Victorian poetry

CO4: Reflect analytical skill of understanding literary essays of Victorian philosophers.

CO5: Develop ability to read, summarize and critically analyse the novels of Jane Austen, Emily Bronte, Charles Dickens and Thomas Hardy.

Unit I: Background:

1. French Revolution;
2. The Romantic Movement in Literature with special reference to leading Romantic poets; Influence of German Philosophy on Romanticism (Schiller and Kant)
3. Introduction to 19th Century Prose and Victorian Poetry
4. Women Question with reference to J S Mill and Mary Wollstonecraft

Unit II: Poetry

1. William Blake: 1) Tyger; 2) London 3) The Chimney Sweeper
2. William Wordsworth: Tintern Abbey, Ode to Immortality, Lucy Poems: A Slumber Did My Spirit Seal, She Dwelt Among Untrodden Ways.
3. S T Coleridge: The Rime of the Ancient Mariner
4. P B Shelley: Ode to the West Wind; To A Skylark
5. John Keats: Ode on the Grecian Urn, Ode to Autumn
6. Robert Browning: My Last Duchess
- 4 Tennyson: Ulysses, Lotus Eaters

Unit III: Fiction

1. Jane Austen: Emma
2. Emily Bronte: Wuthering Heights
3. Charles Dickens: Hard Times
4. Thomas Hardy: Jude the Obscure

Unit IV: Prose

1. J. Ruskin: *Unto the Last* (Chs.1 and 2)
2. J.S. Mill: "On Liberty"
3. Mathew Arnold: Culture and Anarchy
(Ch 1: Sweetness & Light , Ch 2: Hellenism & Hebraism)

Books for Reference and Further Reading:

1. Russell Noyes (Ed.): *English Romantic Poetry and Prose*.
2. Harold Bloom and Lionel Trilling: *Romantic Poetry and Prose*.
3. M. Bowra: *The Romantic Imagination*.
4. William D. Templeman and Charles F. Harrold: *English Prose of the Victorian Era*.

Any one of the soft cores to be chosen for earning 4 credits

PAPER- V – (SOFT CORE ELECTIVE -I) INDIAN DRAMA

Course Code: ENA 230

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

On completion of the Course, the students are able to

CO1: Understand the important aspects and features of Indian Drama.

CO2: Learn to interpret and appreciate poetic devices in Indian Classical Dramas.

CO3: Compare and analyse the classical Indian dramas with the contemporary time.

CO4: Write down the characteristics of interpretation of Indian classical dramas.

1. Abhijnana Shakuntalam: Kalidasa
2. SvapnaVasavadattam: Bhasa
3. Mudrarakshasa: Vishakadatta
4. Mrucchakatika: Shudraka

Books/Texts for Reference and Further Reading

1. Translations made by Prof. M R Kale
2. Global Sanskrit Literature Series in English

SOFT CORE II: Contemporary Indian Novels in English

Course Code: ENA-240

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

- CO1. Learn to explore the thematic paradigm shift in the contemporary Indian writings.
- CO2. Appreciate the postmodern literary devices employed by the writers.
- CO3. Read and critically analyse the Indian novels published in 2000 and after.
- CO4. Develop ability to read, summarize and critically analyse various themes in the works of Anurag Mathur, Anita Desai, Vikram Seth, Shashi Deshpande, Amit Chaudhuri, Anita Nair, Upamanyu Chatterjee and Manju Kapur.

Unit I:

1. Anurag Mathur: *The Inscrutable Americans* (1991)
2. Anita Desai: *Fasting, Feasting* (1999)

Unit II:

1. Vikram Seth: *A Suitable Boy* (1993)
2. Shashi Deshpande: *A Matter of Time* (2000)

Unit III:

1. Amit Chaudhuri: *A New World* (2000)
2. Anita Nair: *Ladies Coupe* (2001)

Unit IV:

1. Upamanyu Chatterjee: *Way to Go* (2010)
2. Manju Kapur: *Custody* (2011)

***Note: Two lecture to introduce new themes and techniques of Contemporary Indian writing**

Books for Reference and Further Reading:

1. K.R. Srinivasa Iyengar: *Indian Writing in English*, 1994.
2. Mukherjee, Meenakshi, C. Vijayshree and Vijay Kumar, eds. *The Nation Across the World: Postcolonial Literary Representations*. New York: OUP, 2008.
3. P K Dhawan and Veena Noble Das, ed. *Fiction of the Nineties*. 1990.
4. Shukla, Sheo Bhushan and Anu Shukla, ed. *The Novels of Amit Chaudhuri: An Exploration in the Alternative Tradition*, Sarup and Sons, 2012.

II SEMESTER

Course I – (HARD CORE-I) THE MODERN AGE- I

Course Code: ENB 040

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Understand the social, political and cultural milieu of the age.

CO2: Learn the impact of World War I and II on 20th Century poetry.

CO3: Identify and analyse literary elements like rhyme, rhythm, tone, style, imagery and, symbols, etc.

CO4: Reflect the analytical skills of understanding war poetry.

UNIT – I

World Wars – I & II, Imagism, Surrealism, Symbolism, Free Verse, Existentialism, and Phenomenology

UNIT – II

G.M. Hopkins: The Wreck of Deutschland, Windhover

W.B Yeats: Sailing to Byzantium, Byzantium, Second Coming, Easter 1916

Thomas Hardy: The Darkling Thrush, The Man He Killed

Wilfred Owen: Strange Meeting, Insensibility

UNIT – III

T.S. Eliot – The Wasteland, The Journey of the Magi

W H Auden – In Memory of W. B. Yeats, Muse des Beaux Arts

Sylvia Plath: The Arrival of the Bee Box: Lady Lazarus

UNIT – IV

Dylan Thomas: After the Funeral, Fern Hill

Seamus Heaney: Tollund Man, Digging, Casualty

Philip Larkin: Church Going, Next Please

Ted Hughes: Thought Fox, Hawk Roosting

Books for Reference and Further Reading:

1. F. R. Leavis. *New Bearings in English Poetry*
2. Faber Book of Modern Verse
3. Norton Anthology of English Literature

Course II (Hard Core-II) - LITERARY CRITICISM-I

Course Code: ENB 020

Credits: 4

Course Outcomes

At the end of the Course, student able to

CO1: Learn the meaning, elements and characteristics of classical literary criticism.

CO2: Understand the basics of literary/ critical theories.

CO3: Learn the technique of early literary criticism.

CO4: Acquire the skills to interpret literary works using literary theories.

Unit I:

Origin, growth, and development of Literary Criticism, Various Posits and Literary Contexts
Principles of Literary and Practical Criticism

Unit II:

1. Longinus: *On the Sublime*
2. Plato on Mimesis in *A Short History of English Literary Criticism* by Wimsatt & Brooks
3. Aristotle: *Poetics*

Unit III:

1. Sidney: *Apology for Poetry*
2. Samuel Johnson: *Preface to Shakespeare*
3. Dryden: *On Dramatic Poesy*

Unit IV:

1. W. Wordsworth: *Preface to Lyrical Ballads*
2. Coleridge: *Biographia Literaria* (Chapters 13, 14, 17)
3. Arnold: *The Function of Criticism at the Present Time*

Books for Reference and Further Reading:

1. I.A. Richards. Principles of Literary Criticism
2. C.T. Indira et al. English Literary Criticism
3. M.S. Nagarajan. English Literary Criticism and Theory
4. Vernon Hall. A short history of literary criticism

Course III – (HC-II) INDIAN WRITING IN ENGLISH – I

Course Code: ENB 030

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Understand the Origin and Growth of Indian English Writing.

CO2: Study different phases of Indian writing and understand their features.

CO3: Understand the social, political and cultural milieu of Pre-independent India explored in literary works.

CO4: Explore different issues in Indian Writing in English depicted in poetry, plays, novels, and essays.

Unit I

Anglicists and Orientalists Debates, *Macaulay's Minute*, Rajaram Mohan Roy's *Letter to Lord Amherst*, The Rise of the Indian English Novel, Novel as a Social Act.

Unit II: Poetry

1. Toru Dutt: Prahlad; Our Casuarina Tree;
2. R. Tagore: Gitanjali: First five and last five poems
3. Sarojini Naidu: Coromandel Fishers, Indian Weavers
4. Aurobindo: *Savitri* Canto-I

Unit III: Drama

1. Girish Karnad: *Hayavadana*
2. Vijay Tendulkar: *Silence! The Court is in Session*
3. Mahesh Dattani: *The Final Solution*

Unit IV: Fiction

1. Mulk Raj Anand: *Untouchable*
2. Raja Rao: *Kanthapura*
3. R.K. Narayan: *A Tiger for Malgudi*

Unit IV: Criticism

1. Hiriyanna: *Art Experience*: Indian Aesthetics (Chs. 1 and 2)
2. Ananda Coomaraswamy: "The Dance of Shiva"
3. Aurobindo: The Poets of Dawn -1&2 (From Future Poetry)

Books/Texts for Reference and Further Reading:

K.R. Srinivasa Iyengar: *Indian Writing in English*

C.D. Narasimhaiah: "Towards an Understanding of the Species Called Indian Writing in English"

Meenakshi Mukherjee: *The Perishable Empire* Chapter on: "The Anxiety of Indianness"

Course IV – (HC-IV) 20th CENTURY WOMEN’S WRITING: THEORY AND PRACTICE

Course Code: ENB-040

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Learn history and growth of feminism as a movement, and the waves of feminism.

CO2: Understand the phrases such as Sex and Gender, women’s rights.

CO3: Explore feminist issues in the novels of Buchi Emecheta, Margaret Atwood and Mahasweta Devi.

CO4: Read and understand the feminist ideas in the works of Simone de Beauvoir, Virginia Woolf and Showalter.

CO5: Appreciate and analyse the poems of Kamala Das and Maya Angelou.

Unit I (Background Study)

Feminism, History of Feminism, Feminist Movements, Sex and Gender, Society and Gender, Women’s rights

Unit II (Novel)

Buchi Emecheta: *The Joys of Motherhood*

Mahasweta Devi: *Imaginary Maps* (A Collection of Short Stories)

Margaret Atwood: *Edible Woman*

Unit III (Prose)

Simone de Beauvoir: *The Second Sex*

Virginia Woolf: *A Room of One’s Own*

Showalter Elaine: “*Feminist Criticism in Wilderness*”

Unit IV (Poetry)

Kamala Das: *An Introduction, My Grandmother’s House, Words, Suicide*

Maya Angelou: *Phenomenal Woman, Caged Bird, Human Family, Women Work*

Books for Reference and Further Reading:

1. The Female Imagination: Patricia Mayor Spack
2. Gender Trouble: Feminism and Subversion of Identity: Judith Butler
3. The Feminine Mystique: Betty Friedan
4. Feminism and Recent Fiction in English: Sushila Singh
5. The New Feminist Criticism: Essays on Women’s Literature and Theory: Elaine Showalter
6. Sexual / Textual Politics: Toril Moi

Any one of the soft cores to be chosen for earning 4 credits

Course IV – (SOFT CORE-II) DALIT LITERATURE

Course Code: ENB-230

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Understand the origin and growth of Dalit literature in India.

CO2: Understand the sufferings of marginalised in Dalit writings.

CO3: Compare and analyse the life of oppressed in the works various languages translated into English like Kannada, Gujarathi, Punjabi, Tamil and Telugu.

CO4: Compare and analyse the different forms of Dalit Literature based on different experiences.

UNIT I: Introduction to Dalit Literature

Essays:

1. Movements of Dalit identity- K C Das
2. Indian Dalits (Introduction- Sharankumar Limbale. (Dalit Literature and Aesthetics)
3. Dalit Literature- B Krishnappa (The Exercise of Freedom)
4. We Too Are Human-B R Ambedkar (The Exercise of Freedom)

Unit II: Poetry

1. H Govindaiah. "In The Soil of Tears" and "A Letter to Father Searching For Me"
2. Challapalli Swaroopa Rani: "Water" and "Forbidden Theory"
3. Sukirtharani: "Paraia God" and "Untitled Poem"
4. Jyoti Lanjewar: "Cave" and "Mother"
5. Gurudas Alam: "For Freedom" and "Treatment of Untouchables" (From The *Core Of Untouchable's*)
6. Bipin Gohel: "To The Fading Man I Sing" and "To A Poet At Mushaira"

Unit III: Short Stories

1. Devanoora Mahadeva: "Odallala"
2. Pathik Parmar : "Naked Feet"
3. M.M. Vinodini: "The Parable of Lost Daughter –Luke15"

Unit IV: Autobiographies

1. Dr. Siddalingaiah – "Ooru Keri-I"
2. Urmila Pawar – "The Weave of My Life: A Dalit Woman's Memoirs"
3. Bama Faustina- "Karukku"

Books for Reference and Further Reading:

- D.R. Nagaraj *The Flaming Feet*
Eleanor Zelliott *From Untouchable To Dalit*
Mulk Raj Anand *Apology For Heroism*
Arjun Dangle *Poisoned Bread*
Encyclopedia of Dalits in India: Literature
Sathyanarayana: *The Exercise of Freedom*
K. Singh: *Dalitism and Feminism: Locating Woman in Dalit Literature*

(SOFT CORE-II) ENGLISH ESSAYISTS

Course Code: ENB-220

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Understand the genre of prose essays and appreciate the essayist's artistic statements.

CO2: Learn to appreciate the literary devices employed by the essayists.

CO3: Understand the importance of essays as a genre to bring social change based on close reading of the essayist's observations on society.

CO4: Appreciate and analyse the views in Bacon, Hazlitt, Charles Lamb, Bertrand Russell, Mathew Arnold and Orwell's essays.

Unit I

General Introduction to Essay as a genre of literary art, Notable Essayists of British Literature

Unit II

1. Francis Bacon: a) Of Studies
 b) Of Ambition

2. Hazlitt: a) On Going a Journey
 b) On the Ignorance of the Learned

Unit III

3. Charles Lamb: a) Poor Relations
 b) Chimney Sweepers

4. Bertrand Russell: a) Behaviorism and Values
 b) Freedom versus Authority in Education

Unit IV

5. Mathew Arnold: a) Wordsworth
 b) John Keats

6. George Orwell: a) Charles Dickens
 b) Politics and English Language

Books/Texts for Reference and Further Reading:

1. The English Critical Tradition: An Anthology of English Literary Criticism, Vol-2, Edited by S. Ramaswamy & V.S. Sethuraman
2. Wimsatt and Brook, Literary Criticism: A Short History
3. G. Tillostone, Criticism and the Nineteenth Century

III SEMESTER

Course I – (HARD CORE-I) THE MODERN AGE – II

Course Code: ENC 010

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Understand the social, political and cultural milieu of the age.

CO2: Learn and analyse the impact of World War I and II on 20th Century fiction.

CO3: Appreciate the use of various literary devices and postmodern techniques such as Stream of Consciousness, Dark Humour in modern writings.

CO4: Understand and write about the new theatres evolved in modern age.

UNIT – I

Post-World War Scenario, Twentieth Century Social Milieu, Twentieth Century Theatre, Twentieth Century Novel, Great Economic Depression, Stream of Consciousness, Postmodernism

UNIT – II

DH Lawrence: *The Rainbow*

Virginia Woolf: *To the Lighthouse*

E.M. Forster: *A Passage to India*

Doris Lessing: *The Golden Notebook*

UNIT – III

Samuel Beckett- *Waiting for Godot*

John Osborne: *Look Back in Anger*

UNIT – IV

Virginia Woolf: “On Modern Fiction”

George Lukacs: “The Meaning of Contemporary Realism” (chapters on Kafka & Modernist Fiction)

Raymond Williams: “When was Modernity”

Books for Reference and Further Reading:

1. Norton Anthology of English Literature
2. Vasudevan. *Perspectives: Selection from Modern English Prose and Fiction*

Course II – (HARD CORE-II) INDIAN WRITING IN ENGLISH - II

Course Code: ENC 020

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Understand characteristic features of post-independent Indian Writing in English.

CO2: Read, compare and critically analyse essays of Indian critics.

CO3: Describe the Indianness in Indian Writing in English.

CO4: Write about the use of various literary devices by Indian writers, such as Arundati Roy, Amitav Ghosh, Shanta Gokhale and RK Narayan.

UNIT – I

Colonization and its aftermath, Culture Vs Modernity: The Indian Context, The Modern Indian Psyche vis a vis Indian Writing in English and in Translation,

UNIT – II

1. Arundhati Roy: *God of Small Things*
2. Amitav Ghosh: *Shadow Lines*
3. Jahnvi Barua: *Next Door*

UNIT – III

1. M. K. Gandhi: *My Experiments with Truth*
2. Amrita Pritam: *Revenue Stamp*
3. R K Narayan: *My Days*

UNIT – IV

1. Gayathri Spivak: “Can the Subaltern Speak?”
2. Meenakshi Mukherjee: “The Anxiety of Indianness”
3. Aijaz Ahamed: *In Theory (Chapter 8 on Indian Literature)*

Books for Reference and Further Reading:

1. K R Srinivasa Iyengar – *Indian Writing in English*
2. M. K Naik – *Critical Essays in Indian Writing in English*
3. Ramakrishnan E V- *Locating Indian Literature*
4. A K Mehrotra (ed): *A Concise History of Indian Literature in English*,
5. Saleem Peeradina (ed): *Contemporary Indian Poetry in English*
6. Makarand Paranjape (ed): *Indian Poetry in English*
7. Girish Karnad – *Fire and Rain*

Course III – (HARD CORE-III) NEW LITERATURES IN ENGLISH

Course Code: ENC 030

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Learn the emergence of New Literatures from Commonwealth literature.

CO2: Learn the thematic concerns in New Literatures.

CO3 Read and analyse the cultural conflict in New literatures such as African, Australian, Canadian and Caribbean and the impact of colonisation on native cultures.

CO4: Write focused analytical essays on the novels of Chinua Achebe, Wole Soyinka, Alice Munro, Patrick White, and V S Naipaul.

CO5: Appreciate the use of various literary devices in the poetry of Dennis Brutus, David Diop, AJM Smith, Judith Wright, Derek Walcott, and Braithwaite.

CO6: Learn and write critical analysis on the essays of Ngugi, Northrop Frye and Wilson Harris.

UNIT 1

Philosophy and Aesthetics of Commonwealth Literature, paradigm shifts from commonwealth to New Literatures

UNIT II- African Literature

Chinua Achebe: *A Man of the People*

Ngugi Wa Thiong'o: *Homecoming* Part-II (a) The Writer and His Past (b) The Writer in a Changing Society

Wole Soyinka: *The Lion and the Jewel*

Denis Brutus: If This Life is All That We Have, I am the tree...

I must conjure from my Past

2) David Diop: *Africa, Vultures*

3) Gabriel Okara: *Piano and Drums, Once Upon A Time*

UNIT III- Canadian and Australian Literature

Alice Munro: *Lives of Girls and Women*

Northrop Frye: "Conclusion" to *Literary History of Canada*

AJM Smith: *The Lonely Land* E.J. Pratt: *The Dying Eagle*

Patrick White: *Solid Mandala*

Judith Wright: *Preoccupations in Australian Poetry* (Chapter 13 on A.D. Hope)

Judith Wright: *Woman to Man, A.D. Hope: Australia*

UNIT- IV- Caribbean Literature

V S Naipaul: *Half a Life*

Wilson Harris: *Tradition and the West Indian Novel*

Derek Walcott: *Almond Trees, A Far Cry from Africa*

Braithwaite: *Starvation, Caliban*

Books/Texts for Reference and Further Reading:

1. Anna Rutherford. Commonwealth
2. Oxford Companion to Canadian Literature
3. Macaulay. *A Map of Australian Verse*
4. Arnold. *Companion to Postcolonial Literature*
5. Helen Tiffin et al. *The Empire Writes Back*

Any one of the soft cores to be chosen for earning 4 credits

Course IV – (SOFT CORE-I) INDIAN ENGLISH POETRY AFTER INDEPENDENCE

Course Code: ENC 230

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Learn and appreciate the use of Indianness in the modern Indian poetry.

CO2: Analyse the themes, imagery, symbolism in the poems of Ezekiel, Ramanujan, Daruwalla, de Souza, Mahapatra, Parthasarathy, Anita Nair and Vikram Seth.

CO3: Understand and appreciate the human values and human predicament in modern Indian poetry.

CO4: Analyse the trend setting themes explored in contemporary Indian poetry.

Unit-I

1. **Nissim Ezekiel:** Night of the Scorpion; Enterprise
Urban; Poet, Lover, Birdwatcher
2. **A K Ramanujan:** Obituary, A River
Love Poem for a Wife II, Small Reflections on a Great House

Unit II

3. **Keki N Daruwalla:** Death of a Bird, The Mistress, The Ghaghra in Spate
4. **Eunice de Souza:** Bequest, Advice to Women, Women in Dutch Paintings,
Feeding the Poor on Christmas

Unit III

5. **Jayanta Mahapatra:** Hunger, Freedom, Grandfather, Dhauli
6. **R Parthasarathy:** Rough Passage: a. Exile b. Trial c. Homecoming

Unit IV

7. **Anita Nair:** Happenings On the London Underground, The Last Rites
Hello Lust How Men Eat
8. **Vikram Seth:** How rarely all these few years, From California, The Wind

***Note:** Two lectures to introduce new themes and techniques of recent Indian poetry in English

Books/Texts for Reference and Further Reading:

1. King Bruce (ed): *Modern Indian Poetry in English*
2. *The Oxford Anthology of Modern Indian Poetry*
3. Saleem Peeradina (ed): *Contemporary Indian Poetry*
4. Zenia Mitra (ed): *Indian Poetry in English: Critical Essays*
5. R. Parthasarathy (ed): *Ten Twentieth Century Indian Poets*

Course V – (SOFT CORE-II) FEMINISM

Course Code: ENC 220

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Analyse the texts critically and write the main ideas given in the prescribed texts.

CO2: Write logically the feelings expressed by the feminists.

CO3: Learn to appreciate the literary theories employed by the feminists to bring gender justice in the society and in literary representation.

CO4: Analyse the critical ideas expressed by Shoshona Felman, Elaine Showalter, Toril Moi and Susie Tharu in their prescribed essays or prose works.

1. Shoshona Felman: "Women and Madness: The Critical Fallacy"

2. Elaine Showalter: "The Female Tradition"

3. Toril Moi: Feminist, Female, Feminine

4. Susie Tharu: Problems for a Contemporary Theory of Gender

Books for Reference and Further Reading:

The Female Imagination: Patricia Mayor Spack

Gender Trouble: Feminism and Subversion of Identity: Judith Butler

The Feminine Mystique: Betty Friedan

Feminism and Recent Fiction in English: Sushila Singh

The New Feminist Criticism: Essays on Women's Literature and Theory: Elaine Showalter

Sexual / Textual Politics: Toril Moi

OPEN ELECTIVES COURSE
A COURSE ON WRITTEN AND SPOKEN ENGLISH

Dept. code 570

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Learn the correct use of parts of speech and English grammar.

CO2: Understand grammar rules and apply them in conversation and communication.

CO3: Write effectively describing impressions, feelings and experiences.

CO4: Talk about familiar topics and give explanations and reasons for opinions, past actions and future plans.

CO5: Understand comprehension passages and answer the implied questions rightly.

UNIT I

Basic Grammar

Definition of Parts of Speech and correct usage

1. Noun
2. Pronoun
3. Adjectives: Degrees of comparison and correct usage.
4. Verb, Tenses: Present, Past and Future and their correct usage.
5. Adverb: Kinds of Adverbs and their correct usage
6. Conjunction
7. Preposition
8. Articles
9. Active and Passive voice
10. Direct and Indirect Speech
11. Punctuation

UNIT II

Oral communication is the ability to explain and present one's ideas in clear English, to diverse audiences; speaking effectively, Effective **Listening** and **Reading skills** to be taught.

Short and long conversation involving two or more people: Greeting, introducing, making an enquiry, casual/formal telephone conversation, Conversation at the bank, at the department store, at the post office, at the doctor's, at the travel agent, at the railway station/bus stop etc.

Informal conversation between friends.

Reading Skills with a focus on Idioms, Phrases, Antonyms, Homophones, Homonyms and Figures of Speech

Unit-III

Written communication: The ability to write effectively in a range of contexts and for different audiences and purposes, with a good command of the English language is taught.

1. Letter Writing:

Personal letters

Leave note

Application for a job

Letter to the editor

- Letters of complaints
Placing orders
2. Precise Writing
3. Short Essay writing

Unit - IV

Writing skills to be taught through:

- A. Guided Composition
B. Expansion of an Idea/ Proverb
C. Comprehension

Comprehension of Poetry/ Prose/ Short Story/ Essay of selected texts

- Comprehension of Poetry:** 1. Robert Frost: **Stopping by Woods on a Snowy Evening**
2. Wole Soyinka: **Telephone Conversation**

Comprehension of Prose: Radhakrishnan: **Humanities V/S Science**

Comprehension of Short Story: Maxim Gorky: **One Autumn Night**

Books for Reference and Further Reading:

1. Sidney Green Baum, The Oxford English Grammar, Oxford University Press,
2. Cowie, A.P. & R. Macklin, Oxford Dictionary of Phrasal verbs, Oxford University Press,
3. Cowie, A.P. & R. Macklin & I.R. Mc Caig, Oxford Dictionary of Phrasal English Idioms, Oxford University Press,
4. Stuart Redman, English Vocabulary in Use pre-intermediate and intermediate; low price editions, Oxford University Press,
5. Rosemary T., Fruehling & Joan M. Lacombe, Communication for Results, A.I.T.B.S. Publishers and Distributors,
6. How to write and speak better, Reader's Digest
7. Modern Grammar with Practical Exercises, Vikas Publishing House Pvt. Ltd.

IV SEMESTER

Course I – (HARD CORE-I) LITERARY CRITICISM-II

Course Code: END 010

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Learn the meaning, elements and characteristics of contemporary literary criticism.

CO2: Writes the essays using the skills of literary critical analysis.

CO3: Learn to write analytical essays on the literary texts of the prescribed critics.

CO4: Articulate and discuss the latest developments in the specific field of practice of literary theories.

Unit I

Russian Formalism, Psychoanalysis, Structuralism, Deconstruction, Post colonialism, Phenomenology and certain other forms of Hermeneutics. The Archetypes of Literature“ Marx, Nietzsche, Freud, Gramsci

Unit II:

1. T.S. Eliot: Tradition and Individual Talent
2. F.R. Leavis: Literature and Society, Tragedy and the Medium (From The Common Pursuit)
3. Carl Jung: “Psychology and Literature”

Unit III:

1. Northrop Frye: Archetypes of Literature
2. G. Genette: “Structuralism and Literary Criticism
3. J. Derrida: “Structure, Sign and Play in the Discourse of Human Sciences”

Unit IV:

1. Elaine Showalter: Towards a Feminist Poetics
2. Helene Cixous: The Laugh of the Medusa
3. Sara Suleri: Woman Skin Deep: Feminism and Postcolonial Condition

Books for Reference and Further Reading:

Eagleton, Terry: *Theory – An Introduction*

Selden, Raman: *A Reader’s Guide to Contemporary Literary Theory* Belsey, Catherine.

Critical Practice: Culler, Jonathan.

Structuralist Poetics: Structuralism, Linguistics and the Study of Literature Newton,

K.M. *Twentieth Century Literary Theory: A Reader*

Course II – (HARD CORE-II) AMERICAN LITERATURE

Course Code: END 020

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Understand the significance of Renaissance, Transcendentalism and journey metaphor in American literature.

CO2: Appreciate and analyse the poems of Emily Dickinson, Wallace Stevens, Walt Whitman and Robert Frost.

CO3: Understand and write critically about the themes, narrative techniques, character analysis in the novels of Mark Twain, Douglas, Toni Morrison and Ray Bradbury.

CO4: Acquire the ability to understand and explore the human condition in the plays of Arthur Miller, Eugene O'Neill and Edward Albee.

UNIT –I

American Renaissance, Journey as Metaphor, Westward Movement, Transcendentalism

UNIT – II

Emerson: American Scholar

Thoreau: Walden (Chapters on Economy & Where I Lived and What I Lived For)

Emily Dickinson: Because I Could Not Stop for Death, The Soul Selects her Own Society
I Heard a Fly Buzz

Wallace Stevens: Emperor of Ice-Cream, Anecdote of The Jar

Walt Whitman: When Lilacs Last in the Dooryard Bloomed, A Noiseless Patient Spider,

Robert Frost: Mending Wall, The Road Not Taken, Birches

UNIT III

Mark Twain: Huckleberry Finn

Fredrick Douglas: Narrative of the Life of an American Slave

Toni Morrison: The Bluest Eye

Ray Bradbury: Fahrenheit 451

UNIT – IV

Arthur Miller: Death of a Salesman

Eugene O'Neill: The Hairy Ape

Edward Albee: The Zoo Story

Books for Reference and Further Reading:

Norton Anthology of American Literature

Richard J Gray. *A History of American Literature*

The Cambridge History of American Literature. Vol. 1 to 4

Any one of the soft cores to be chosen for earning 4 credits

COURSE - III: SOFTCORE Elective I: INDIAN DIASPORA FICTION

Course Code: END 240

Credits: 4

Course Outcomes:

At the end of the Course, student able to
Course Outcomes

CO1: Learn the background of Diaspora Literature & major themes of Diaspora Literature

CO2: Compare and Contrast authors' treatment of themes, characters, subject matter etc.

CO3: Identify and analyse literary elements like plot, setting, tone, point of view, style, image, symbols, etc.

CO4: Trace the role of partition, corruption, fantasy, migration, etc. and psychological aspects behind human behaviour in the novels prescribed.

Unit I:

1. Salman Rushdie: *Midnight Children*
2. Tanuja Desai Hidier: *Born Confused*

Unit II:

1. Jhumpa Lahiri: *The Namesake*
2. Chitra Banerjee Divakaruni: *The Mistress of Spices*

Unit III:

1. Kiran Desai: *Inheritance of Loss*
2. Rohinton Mistry: *A Fine Balance*

Unit IV:

1. Aravind Adiga: *White Tiger*
2. Hari Kunzru: *Gods without Men*

***Note: Two lectures to introduce India Diaspora**

Books for Reference and Further Reading:

1. Avtar Brah. *Cartographies of Diaspora: Contesting Identities*. London: Routledge, 1996.
2. Homi K. Bhabha. *The Location of Culture*, 1994.
3. Edward W. Said. *Orientalism*. New Delhi: Penguin, 2001.

COURSE - III (SOFTCORE Elective-II) AFRICAN FICTION

Course Code: END-230

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Learn the social, political and cultural milieu of the African society represented in fiction.

CO2: Understand and write critical essays on contemporary African novels such as *Anthills of the Savannah*, *Purple Hibiscus*, *The Bride Price* and *Changes: A Love Story*.

CO3: Articulate and discuss the latest developments in the specific fields of postcolonial African writings to bring gender justice in the society.

CO4: Write critically on the role of the characters in the novels of Achebe, Adichie, Emecheta and Aidoo.

Unit I: Chinua Achebe: *Anthills of the Savannah*

Unit II: Chimamanda Adichie Ngozi: *Purple Hibiscus*

Unit III: Buchi Emecheta: *The Bride Price*

Unit IV: Ama Ata Aidoo: *Changes: A Love Story*

Suggested Reading:

1. Butler, Judith. *Gender Trouble: Feminism and the Subversion of Identity*. New York: Routledge, 1990.
2. Ogundipe-Leslie, Omolara. *Re-Creating Ourselves African Women and Critical Transformations*. Trenton, NJ: Africa World P, 1994.
3. Palmer, Eustace (ed.). *An Introduction to the African Novel. A Critical Study of Twelve Books*. London: Heinemann, 1979.
4. Ashcroft, Bill. *Post-Colonial Transformation*. London and New York: Routledge, 2001.
5. Ashcroft, Bill, Gareth Griffiths, and Helen Tiffin. Eds. *The Post-Colonial Studies Reader*. London and New York: Routledge, 2002.

Course – IV – (HARDCORE - III) Major Project Work Leading to a Dissertation

Course Code: END 030

Credits: 4

Teaching Hours: 60 (4 Hours/Week)

Course Outcomes

At the end of the Course, student able to

CO1: Learn to investigate the area of topic chosen for project work in detail.

CO2: Learn research skills and demonstrate scholarly expertise in exploring the subject to prepare the dissertation for the project work.

CO3: Learn the skills of research analysis in writing thesis.

CO4: Able to think logically and relate the issues and findings to real life scenario.

Students will be encouraged to undertake a major project work in disciplines related to literature of contemporary interest.

Suggested Areas:

Folklore

Performing Arts

Popular Culture

Gender and Sexuality

Dalit and Subaltern Discourses

Media Studies

World Literatures in English

General Survey of Poetic Works of Classical Writers

General Survey of Poetic Works of Modern Writers

The distribution of marks for The Project Work will be as per the stipulations laid down by the university.

SOFTCORES OFFERED

1. Cross Cultural Women Writers
2. Caribbean Literature
3. Novels of Childhood
4. Indian Classics in Translation
5. Folklore and Literature-I
6. Folklore and Drama
7. Myth and Drama-I
8. Myth and Drama-II
9. Cultural Theory
10. Canada and the World
11. Feminist Theory-I
12. Australian Poetry
13. Folk Epics of Karnataka
14. Postcolonial Theory
15. Writers from African Diaspora
16. Canadian Science Fiction by Women
17. Theories on Culture: An Introduction
18. Postcolonial Criticism
19. Travel Literature
20. Popular Culture and Censorship
21. Adventure Novels
22. Novels and Metropolis
23. Short Fictions of Russia and USA
24. Contemporary Indian Novels in English
25. Translations
26. Dalit Literature-I
27. Recent Indian Poetry in English
28. Dalit Literature-II
29. Women Writings from the Margins
30. English Essayists
31. Post 1990 Indian Women Narratives
32. Indian Novels in English: 2000 and after
33. Twentieth Century Indian Poets in English
34. Contemporary Indian Regional Poetry in English Translation
35. Realism and Fiction
36. Indian Women Novelists
37. African Fiction in English
38. Jewish American Fiction
39. Introduction to Canadian Literature
40. South Asian Immigrant Literature in Canada
41. Introduction to Australian Literature.
42. Feminist Theory-II
43. Indian Classics in Translation
44. Shakespeare Criticism
45. Literature and Popular Culture
46. Postcolonial African Fiction
47. Arab Women Short Stories
48. Indian Diaspora Fiction

POSTGRADUATE DEPARTMENT OF ENGLISH
Question Paper Pattern

END SEMESTER EXAM (C 3)

Course Code Title of the Course (HC/ SC)

Max Marks: 70

Time: 3 Hours

Instruction: Answer all sections.

SECTION – I

I. Answer the following questions in **one or two** sentences

10X1= 10

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.
- 9.
- 10.

SECTION-II (from Unit-I)

II. Answer any one of the following.

1x15=15

- 1.
- 2.
- 3.

SECTION-III (from Unit-II)

III. Answer any one of the following.

1x15=15

- 1.
- 2.
- 3.

SECTION-IV (from Unit-III)

IV. Answer any one of the following.

1x15=15

- 1.
- 2.
- 3.

SECTION-V (from Unit-IV)

V. Answer any one of the following.

1x15=15

- 1.
- 2.
- 3.

JSS MAHAVIDYAPEETA
JSS COLLEGE OF ARTS, COMMERCE & SCIENCE
(Affiliated to University of Mysore)
Autonomous, NAAC 'A' grade and College with Potential for Excellence
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PROPOSED SCHEME of INSTRUCTION
and
SYLLABI
for
M.Sc. MATHEMATICS
under
CHOICE BASED CREDIT SYSTEM (CBCS)

PG DEPARTMENT OF MATHEMATICS
2017 – 18

1. **Scheme of Admission:**

- 50% seats of the total intake for M.Sc. Mathematics Programme of the College will be filled by the University of Mysore through Centralized Admission Cell as per University regulations.
- Remaining 50% seats will be filled by the College under College Quota.

2. **Eligibility:** B.Sc. degree with Mathematics as Major/Optional subject with 45% or B.Sc.Ed., degree of Regional Institute of Education with Mathematics as a special subject.

3. **Scheme of Examination:**

- (i) Theory paper of 03 hours duration (C_3 component) : **70** marks
(ii) Internal Assessment: **30** marks
(C_1 Component: 15 marks, C_2 Component: 15 marks)

4. **Pattern of Question Paper: Theory paper:** There are 5 questions. All questions must be answered. Each question carries 14 marks.

5. **Minimum Marks for Securing Credits:** 40% (with minimum of 30% in C_1 and C_2 and minimum of 30% in C_3).

6. **Minimum Credits for getting the M.Sc. Mathematics degree:** 76 credits.

7. **Scheme of study for Masters:**

Total No. of credits to be earned	76 credits
Minimum No. of credits to be earned from Hard Core papers	44 credits
Minimum No. of credits to be earned from Soft Core papers	28 credits
Minimum No. of credits to be earned from Open elective papers	4 credits



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

(Affiliated to University of Mysore)

Autonomous, NAAC 'A' grade and College with Potential for Excellence

B N Road, Mysuru – 570025

PG DEPARTMENT OF MATHEMATICS

CBCS - M.Sc. Mathematics

List of courses with credit pattern

Sl. No.	Code	Type of the paper	Title of the Paper	Credit pattern in L:T:P	Credit Value	No. of hrs
FIRST SEMESTER						
1	MAA 010	HC	Algebra I	3 : 1 : 0	4	5
2	MAA 020	HC	Real Analysis I	3 : 1 : 0	4	5
3	MAA 030	HC	Real Analysis II	3 : 1 : 0	4	5
4	MAA 040	HC	Complex Analysis I	3 : 1 : 0	4	5
5	MAA 210	SC	Linear Algebra	3 : 1 : 0	4	5
SECOND SEMESTER						
6	MAB 010	HC	Algebra II	3 : 1 : 0	4	5
7	MAB 020	HC	Real Analysis III	3 : 1 : 0	4	5
8	MAB 030	HC	Complex Analysis II	3 : 1 : 0	4	5
9	MAB 210	SC	Ordinary and Partial Differential Equations	3 : 1 : 0	4	5
10	MAB 230	SC	Graph Theory	3 : 1 : 0	4	5
THIRD SEMESTER						
11	MAC 010	HC	Elements of Functional Analysis	3 : 1 : 0	4	5
12	MAC 020	HC	Topology I	3 : 1 : 0	4	5
13	MAC 210	SC	Commutative Algebra	3 : 1 : 0	4	5
14	MAC 220	SC	Theory of Numbers	3 : 1 : 0	4	5
15	CSC/MCC/ZOC/ BTC/BOC/BCC 580	OE (For others)	Basic Mathematics	3 : 1 : 0	4	5
FORTH SEMESTER						
16	MAD 010	HC	Measure and Integration	3 : 1 : 0	4	5
17	MAD 020	HC	Topology-II	3 : 1 : 0	4	5
18	MAD 220	SC	Theory of Partitions	3 : 1 : 0	4	5
19	MAD 230	SC	Differential Geometry	3 : 1 : 0	4	5
Total Credits					76	

Program Outcomes

PO1 : Apply the underlying unifying structures of mathematics (i.e. sets, relations and functions, logical structure) and the relationships among them

PO2 : To include methods of facilitating learning such as projects, group work and participative learning.

PO3 : To Innovate, invent and solve complex mathematical problems using the knowledge of pure and applied mathematics.

PO4 : To impart knowledge of some basic concepts and principles of the discipline.

PO5 : To establish inter-disciplinarily between mathematics and other subjects from Humanities and the Social Sciences.

PO6 : To encourage collaborative learning through group activities and hands-on learning.

PO7 : To provide in-service training for school teachers. To learn to apply mathematics to real life situations and help in problem solving

Program Specific Outcomes

PSO1 : Explain the importance of mathematics and its techniques to solve real life problems and provide the limitations of such techniques and the validity of the results

PSO2 : Propose new mathematical and statistical questions and suggest possible software packages and/or computer programming to find solutions to these questions

PSO3 : Continue to acquire mathematical and statistical knowledge and skills appropriate to professional activities and demonstrate highest standards of ethical issues in mathematics

PSO4 : Ability to use computer calculations as a tool to carry out scientific investigations and develop new variants of the acquired methods, if required by the problem at hand.

PSO5 : Crack lectureship and fellowship exams approved by UGC like CSIR – NET and SLET.

PSO6 : Apply knowledge of Mathematics, in all the fields of learning including higher research and its extensions.

First Semester

Course Code MAA 010	Algebra I
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Course Outcome

Students are able to

- CO1 Define and interpret the concepts of divisibility, congruence, greatest common divisor, prime, and prime-factorization
- CO2 Apply the Law of Quadratic Reciprocity and other methods to classify numbers as primitive roots, quadratic residues, and quadratic non-residues
- CO3 Assess properties implied by the definitions of groups.
- CO4 Analyze Permutation groups
- CO5 Uses various canonical types of groups and canonical types of rings
- CO6 Analyze and demonstrate examples of subgroups, normal subgroups and quotient groups.
- CO7 Use the concepts of isomorphism and homomorphism for groups
- CO8 Demonstrate knowledge of conjugates, the Class Equation and Sylow theorems.

COURSE CONTENT

Unit I

Number theory - Congruences, residue classes, theorems of Fermat, Euler and Wilson, linear congruences, elementary arithmetical functions, primitive roots, quadratic residues and the law of quadratic reciprocity.

Unit II

Groups - Lagrange's Theorem, homomorphism and isomorphism, normal subgroups and factor groups.

Unit III

The fundamental theorem of homomorphism, two laws of isomorphism.

Unit IV

Permutation groups and Cayley's theorem, Sylow's theorems.

Books for Reference:

1. D. M. Burton – Elementary Number Theory, Tata McGraw-Hill, New Delhi, 6th Ed.,
2. I. Niven, H. S. Zuckerman and H. L. Montgomery – An Introduction to the Theory of Numbers, New York, John Wiley and Sons, Inc., 2004, 5th Ed.,
3. G. A. Jones and J. M. Jones – Elementary Number Theory, Springer, 1998.
4. Thomas W. Hungerford – Algebra, Springer International Edition, New York.
5. Michael Artin – Algebra, Prentice-Hall of India, New Delhi.
6. J. A. Gallian – Contemporary Abstract Algebra, Narosa Publishing House, 4th Ed.,
7. D. S. Dummit and R. M. Foote – Abstract Algebra, John Wiley and Sons, 1999.
8. I. N. Herstein – Topics in Algebra, Vikas Publishing House, New Delhi.
9. J. B. Fraleigh – A First course in Abstract Algebra, Addison-Wesley,
10. N. S. Gopalakrishnan – University Algebra, New Age International, 2nd Ed.

Course code MAA020

Real Analysis I

Course Outcome

Students are able to

- CO1 Describe fundamental properties of the real numbers that lead to the formal development of real analysis
- CO2 Describe the real line as a complete ordered field
- CO3 Acquire knowledge extended real number system, the n-dimensional Euclidean space
- CO4 Study the types of mean inequalities and application
- CO5 Use the definitions of convergence as they apply to sequences, series, and functions
- CO6 Use the definitions of convergence as they apply to re-arrangements. Double series, infinite products

COURSE CONTENT**Unit I**

The extended real number system, the n-dimensional Euclidean space, the binomial inequality, the inequality of the arithmetic and geometric means, the inequality of the power means, Cauchy's, Holder's inequality and Minkowski's inequality.

Unit II

Numerical sequences, convergent sequences, Cauchy sequences, upper and lower limits.

Unit III

Series of real numbers series of non-negative terms, the number 'e', tests of convergence.

Unit IV

Multiplications of series, re-arrangements. Double series, infinite products.

Books for Reference:

1. W. Rudin – Principles of Mathematical Analysis, International Student edition, McGraw Hill, 3rd Ed.
2. T. M. Apostol – Mathematical Analysis, Addison Wesley, Narosa, New Delhi, 2nd Ed.
3. R. R. Goldberg – Methods of real Analysis, Oxford and IBH, New Delhi.
4. Torence Tao – Analysis I, Hindustan Book Agency, India, 2006.
5. Torence Tao – Analysis II, Hindustan Book Agency, India, 2006.
6. Kenneth A. Ross – Elementary Analysis: The Theory of Calculus, Springer International Edition, 2004.

Course code MAA 030	Real Analysis II
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Course Outcome

Students are able to

- CO1 Determine the basic topological properties of subsets of the real numbers
- CO2 Analyze Sets and functions , distinguish between finite, countable and uncountable sets
- CO3 Determine the Riemann integrability and the Riemann-Stieltjes integrability of a bounded function and prove a related of theorems concerning integration
- CO4 Apply the Mean Value Theorem and the Fundamental Theorem of Calculus to

problems in the context of real analysis

CO5 Prove Taylor's theorem, determine maxima and minima, study the Functions of bounded variation

CO6 Produce rigorous proofs of results that arise in the context of real analysis.

COURSE CONTENT

Unit I

Finite, countable and uncountable sets, the topology of the real line.

Unit II

Continuity, uniform continuity, properties of continuous functions, discontinuities, monotonic functions.

Unit III

Differentiability, **mean value theorems**, L' Hospital rule, Taylor's theorem, maxima and minima, Functions of bounded variation.

Unit IV

The Riemann-Stieltje's integral, criterion for integrability. Properties of the integral, classes of integrable functions. The integral as the limit of a sum. First and second mean value theorems. Integration and differentiation.

Books for Reference:

1. W. Rudin – Principles of Mathematical Analysis, International Student edition, McGraw-Hill, 3rd Ed..
2. Torence Tao – Analysis I, Hindustan Book Agency, India, 2006.
3. Torence Tao – Analysis II, Hindustan Book Agency, India, 2006.
4. T. M. Apostol – Mathematical Analysis, Addison Wesley, Narosa, 2nd Ed.,
5. R. R. Goldberg – Methods of real Analysis, Oxford and IBH Publishing Company, New Delhi.
6. Kenneth A. Ross – Elementary Analysis: The Theory of Calculus, Springer International Edition, 2004.

Course Outcome

Students are able to

- CO1 Represent complex numbers algebraically and geometrically,
- CO2 Define and analyze limits and continuity for complex functions as well as consequences of continuity
- CO3 Apply the concept and consequences of analyticity and the Cauchy-Riemann equations and of results on harmonic and entire functions including the fundamental theorem of algebra
- CO4 Analyze sequences and series of analytic functions and types of convergence.
- CO5 Apply Cauchy's theorem for disk and the integral formula.
- CO6 Understand Local properties of Analytic functions

COURSE CONTENT

Unit I

Algebra of complex numbers, geometric representation of complex numbers. Riemann sphere and Stereographic projection, Lines, Circles. Limits and Continuity.

Unit II

Analytic functions, Cauchy-Riemann equations, Harmonic functions, Polynomials and Rational functions. Elementary theory of power series - sequences, series, uniform convergence of power series, Abel's limit theorem, The elementary functions.

Unit III

Topology of the complex plane. Linear fractional transformations, Cross-ratio, Symmetry, Elementary conformal mappings. Complex integration – Line integrals, Rectifiable arcs.

Unit IV

Cauchy's theorem for a rectangle. Cauchy's theorem in a Circular disk, Cauchy's integral formula. Local properties of analytic functions.

Books for Reference:

1. L. V. Ahlfors – Complex Analysis, McGraw-Hill, Kogakusha, 1979.
2. J. B. Conway – Functions of one complex variable, Narosa, New Delhi.
3. R. P. Boas – Invitation to Complex Analysis, The Random House, 1987
4. B. C. Palka – An Introduction to Complex Function Theory, Springer, 1991.
5. S. Ponnusamy – Foundations of Complex Analysis, Narosa, 1995.

Course code MAA 210	Linear Algebra
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Course Outcome

Students are able to

- CO1 Solve systems of linear equations,
- CO2 Analyze vectors in \mathbb{R}^n geometrically and algebraically,
- CO3 Recognize the concepts of the terms span, linear independence, basis, and dimension, and apply these concepts to various vector spaces and subspaces,
- CO4 Use matrix algebra and the related matrices to linear transformations,
- CO5 Compute and use determinants, and use eigenvectors and eigenvalues,
- CO6 Determine and use orthogonality, have thorough understanding of the Linear Transformations
- CO7 Study Inner Products and Norms, The Gram-Schmidt Orthogonalization Process and Orthogonal Complements.
- CO8 Analyze the elementary canonical forms, rational and Jordan forms, Apply the inner product space

COURSE CONTENT

Unit I

Vector Spaces, Subspaces, Linear Combinations and Systems of Linear Equations, Linear Dependence and Linear Independence, Bases and Dimension, Maximal Linearly Independent Subsets; Linear Transformations, Null Spaces, and Ranges, The Matrix Representation of a Linear Transformation, Composition of Linear Transformations and Matrix Multiplication, Invertibility and Isomorphisms, The Change of Coordinate Matrix, The Dual Space; Elementary Matrix Operations and Elementary Matrices, The Rank of a Matrix and Matrix Inverses, Systems of Linear Equations.

Unit II

Properties of Determinants, Cofactor Expansions, Elementary Operations and Cramer's Rule, Eigenvalues and Eigenvectors, Diagonalizability, Invariant Subspaces and the Cayley-Hamilton Theorem; Inner Products and Norms, The Gram-Schmidt Orthogonalization Process and Orthogonal Complements.

Unit III

The Adjoint of a Linear Operator, Normal and Self-Adjoint Operators, Unitary and Orthogonal Operators and Their Matrices, Orthogonal Projections and the Spectral Theorem; Bilinear and Quadratic Forms;

Unit IV

The Diagonal form, The Triangular form; The Jordan Canonical Form; The Minimal Polynomial; The Rational Canonical Form.

Books for Reference:

1. S. Friedberg, A. Insel, and L. Spence - Linear Algebra, Fourth Edition, PHI, 2009.
2. Jimmie Gilbert and Linda Gilbert – Linear Algebra and Matrix Theory, Academic Press, An imprint of Elsevier.
3. I. N. Herstein – Topics in Algebra, Vikas Publishing House, New Delhi.
4. Hoffman and Kunze – Linear Algebra, Prentice-Hall of India, 1978, 2nd Ed.,
5. P. R. Halmos – Finite Dimensional Vector Space, D. Van Nostrand, 1958.
6. S. Kumeresan – Linear Algebra, A Geometric approach, Prentice Hall India, 2000.

SECOND SEMESTER

Course code MAB 010	Algebra II
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Course Outcome

Students are able to

- CO1 Assess properties implied by the definitions of rings.
- CO2 Analyze and demonstrate examples and properties of ideals and quotient rings
- CO3 Demonstrate knowledge of polynomial rings and associated properties
- CO4 Derive and apply Gauss Lemma, Eisenstein criterion for irreducibility of rationals
- CO5 Demonstrate the characteristic of a field and the prime subfield.

COURSE CONTENT

Unit I

Rings, Integral domains and Fields, Homomorphisms, Ideals and Quotient Rings, Prime and Maximal ideals.

Unit II

Euclidean and principal ideal rings, Polynomials, Zeros of a polynomial, Factorization, Irreducibility criterion.

Unit III

Adjunction of roots, algebraic and transcendental extensions, Finite fields.

Unit IV

Separable and inseparable extensions, Perfect and imperfect fields. Theorem on the primitive element.

Books for Reference:

1. Thomas W. Hungerford – Algebra, Springer International Edition, New York.
2. Michael Artin – Algebra, Prentice-Hall of India, New Delhi.
3. Joseph A. Gallian – Contemporary Abstract Algebra, Narosa, 4th Ed.,
4. D. S. Dummit and R. M. Foote – Abstract Algebra, John Wiley and Sons, 1999, 2nd Ed.,
5. I. N. Herstein – Topics in Algebra, Vikas Publishing House, New Delhi.
6. J. B. Fraleigh – A First course in Abstract Algebra, Addison-Wesley,
7. N. S. Gopalakrishnan – University Algebra, New Age International, 2nd ed.,

Course Outcome

Students are able to

- CO1 Understand the concept of sequences and series with respect to the real numbers system and establish whether a given series/ sequences is convergent/ divergent at a specified point or interval
- CO2 Recognize the difference between pointwise and uniform convergence of a sequence/series of functions
- CO3 Illustrate the effect of uniform convergence on the limit function with respect to continuity, differentiability, and integrability
- CO4 Derive and apply properties of convergence of power series.
- CO5 Analyze Functions of several variables, partial derivatives, continuity and differentiability
- CO6 Understand the importance of the Implicit function theorem, Taylor's theorem, the Maxima and Minima, Lagrange's multipliers

COURSE CONTENT

Unit I

Sequences and series of functions, Uniform convergence, Uniform convergence and continuity, Uniform convergence and integration, Uniform convergence and differentiation.

Unit II

Power series, The exponential and logarithmic functions, The trigonometric functions. Improper integrals and their convergence.

Unit III

Functions of several variables, partial derivatives, continuity and differentiability, the chain rule, Jacobians.

Unit IV

The Implicit function theorem, Taylor's theorem, the Maxima and Minima, Lagrange's multipliers.

Books for Reference:

1. W. Rudin – Principles of Mathematical Analysis, International Student edition, McGraw-Hill, 3rd Ed.,
2. T.M. Apostol – Mathematical Analysis, Addison Wesley, Narosa, 2nd Ed.,
3. R.R. Goldberg – Methods of Real Analysis, Oxford and IBH, New Delhi.
4. D.V. Widder – Advanced Calculus, Prentice Hall of India, New Delhi, 2nd Ed.,
5. Torence Tao – Analysis I, Hindustan Book Agency, India, 2006.
6. Torence Tao – Analysis II, Hindustan Book Agency, India, 2006.
7. Kenneth A. Ross – Elementary Analysis: The Theory of Calculus, Springer International Edition, 2004.

Course code MAB 030	Complex Analysis II
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Course Outcome

Students are able to

- CO1 Classify singularities and poles, find residues and evaluate complex integrals using the residue theorem.
- CO2 Apply the Cauchy integral theorem in its various versions, and the Cauchy integral formulas
- CO3 Evaluate complex contour integrals directly and by the fundamental theorem
- CO4 Study Residue theorem and the argument principle
- CO5 State and prove the Cauchy Riemann Equation and use it to show that a function is Analytic
- CO6 Represent functions as Taylor, power and Laurent series
- CO7 Study properties of Partial fractions, Infinite products, Canonical products, The Gamma and Beta functions

COURSE CONTENT

Unit I

The Calculus of Residues – The residue theorem, argument principle, Evaluation of definite integrals.

Unit II

Harmonic functions – Definition and basic properties, mean value property, Poisson's formula, Schwarz's theorem, reflection principle.

Unit III

Power series expansions – The Weierstrass theorem, The Taylor series, The Laurent series.

Unit IV

Partial fractions and factorization – Partial fractions, Mittag - Leffer's theorem, Infinite products, Canonical products, The Gamma and Beta functions, Sterling's formula. Entire functions – Jensen's formula, Hadamard's theorem.

Books for Reference:

1. L. V. Ahlfors – Complex Analysis, McGraw-Hill, Kogakusha, 1979.
2. J. B. Conway – Functions of one complex variable, Narosa, New Delhi.
3. R. P. Boas – Invitation to Complex Analysis, The Random House, 1987.
4. B. C. Palka – An Introduction to the Complex Function Theory, Springer, 1991.
5. S. Ponnusamy – Foundations of Complex Analysis, Narosa, 1995.

Course code MAB 210	Ordinary and Partial Differential Equations
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Course Outcome

Students are able to

- CO1 Solve problems in ordinary differential equations, dynamical systems, stability theory, and a number of applications to scientific and engineering problems
- CO2 Study of Differential focuses on the existence and uniqueness of solutions also emphasizes the rigorous justification of methods for approximating solutions in pure and applied mathematics.

- CO3 Implement solution methods using appropriate technology
- CO4 Investigate the qualitative behavior of solutions of systems of differential equations
- CO5 Recognize the major classification of PDEs and the qualitative differences between the classes of equations
- CO6 Solve linear PDEs using classical solution methods.
- CO7 Solve differential equations widely used in formulating many fundamental laws of physics and chemistry.

COURSE CONTENT

Unit I

Linear Second Order Equations - Initial value problem, Existence and Uniqueness by Picard's Theorem, Wronskian, separation and comparison theorems, Poincare phase plane, variation of parameters.

Unit II

Power series solutions - Solution near ordinary and regular singular point. Convergence of the formal power series, applications to Legendre, Bessel, Hermite, Laguerre and hypergeometric differential equations with their properties.

Unit III

Partial differential equations - Cauchy problems and characteristics, Classification of Second order PDE's, reduction to canonical forms, derivation of the equations of mathematical physics and their solutions by separation of variables.

Unit IV

Boundary value problems - Transforming Boundary value problem of PDE and ODE, Sturm - Liouville system, eigen values and eigen functions, simple properties, expansion in eigen functions, Parseval's identity, Green's function method.

Books for Reference:

1. E. A. Coddington and N. Levinson – Theory of Ordinary Differential equations, Tata McGraw-Hill, New Delhi.
2. R. Courant and D. Hilbert – Methods of Mathematical Physics, Vol. I. & II, Tata McGraw-Hill, New Delhi, 1975.

3. G. F. Simmons – Differential Equations with applications and Historical Notes, Tata McGraw-Hill, New Delhi, 1991.
4. I. N. Sneddon – Theory of Partial differential equations, McGraw-Hill, International Student Edition.
5. S. G. Deo and V. Raghavendra – Ordinary Differential Equations and Stability Theory, Tata McGraw-Hill, New Delhi.

Course code MAB 230	Graph Theory
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Course Outcome

Students are able to

- CO1 Construct examples and proofs pertaining to the basic theorems.
- CO2 Understand the definitions of cut vertex, bridge, blocks and Automorphism group of a graph.
- CO3 Study the properties of trees and connectivity
- CO4 Discuss and understand the importance of the concepts Matchings and Menger's theorem

COURSE CONTENT

Unit I

Types of Graphs, Walk and connectedness, degrees, Extremal graphs, Intersection graph, Operations on graphs.

Unit II

Cutpoints, Bridges and Blocks, Block graphs and cutpoints.

Unit III

Characterization of trees, Centers and Centroids, Spanning Tree .

Unit IV

Connectivity and line connectivity, Menger's theorem, Coverings Independence, Critical points and lines.

Books for Reference:

1. F. Harary – Graph Theory, Addition Wesley Reading Mass, 1969.
2. N. Deo – Graph Theory With Applications to Engineering and Computer Science, Prentice Hall of India, 1987.
3. K. R. Parthasarathy – Basic Graph Theory, Tata McGraw-Hill, New Delhi, 1994.
4. G. Chartand and L. Lesniak – Graphs and Diagraphs, Qwadsworth and Brooks, 2nd Ed.,
5. Clark and D. A. Holton – A First Look at Graph Theory, Allied publishers.
6. D. B. West – Introduction to Graph Theory, Pearson Education Inc., 2001, 2nd Ed.,
7. J. A. Bondy and U. S. R. Murthy – Graph Theory with applications, Elsevier, 1976.

THIRD SEMESTER

Course code MAC 010	Elements of Functional Analysis
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Course Outcome

Students are able to

- CO1 Explain the fundamental concepts of functional analysis.
- CO2 Understand the approximation of continuous functions.
- CO3 Understand concepts of Hilbert and Banach spaces
- CO4 Understand the definitions of linear functional and prove the Hahn-Banach theorem, open mapping theorem, uniform boundedness theorem, etc.
- CO5 Define linear operators, self adjoint, isometric and unitary operators on Hilbert spaces.

COURSE CONTENT

Unit I

Metric completion. Banach's contraction mapping theorem and applications, Baire' category theorem, Ascoli - Arzela theorem.

Unit II

Linear spaces and linear operators, Norm of a bounded operator, The Hahn – Banach extension theorem, Stone - Weirstrass theorem.

Unit III

Open mapping and Closed Graph theorems. The Banach - Steinhaus Principle of Uniform Boundedness.

Unit IV

Hilbert spaces- The orthogonal projection, Nearly orthogonal elements, Riesz's lemma, Riesz's representation theorem.

Books for Reference:

1. G. F. Simmons – Introduction to Topology and Modern Analysis, Tata McGraw-Hill, New Delhi.
2. A. E. Taylor – Introduction to Functional Analysis, Wiley, New York, 1958.
3. A. Page and A. L. Brown – Elements of Functional Analysis.
4. George Bachman and Lawrence Narici – Functional Analysis, Dover Publications, Inc., Mineola, New York.
5. J. B. Conway – A Course in Functional Analysis, GTM, Vol. 96., Springer, 1985.

Course code MAC 020	Topology I
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Course Outcome

Students are able to

- CO1 Define and illustrate the concept of topological spaces and continuous functions
- CO2 Provide precise definitions and appropriate examples and counter examples of fundamental concepts in topology.
- CO3 Define and illustrate the concept of product topology, metric and quotient topology
- CO4 Prove a selection of theorems concerning topological spaces, continuous functions product topologies, and quotient topologies
- CO5 Define connectedness and compactness, and prove a selection of related theorems

COURSE CONTENT

Unit I

Set theoretic preliminaries. Topological spaces and continuous maps - topological spaces, basis for a topology, the order topology, the product topology on $X \times X$, the subspace topology.

Unit II

Closed sets and limit points, continuous functions, the product topology, the metric topology, the quotient topology.

Unit III

Connectedness - connected spaces, connected sets on the real line, path connectedness.

Unit IV

Compactness - compact spaces, compact sets on the line, limit point compactness, local compactness.

Books for Reference:

1. J. R. Munkres – A First Course in Topology, Prentice Hall India, 2000, 2nd Ed.,
2. G. F. Simmons – Introduction to Topology and Modern Analysis, McGraw-Hill, Kogakusha, 1968.
3. S. Willard – General Topology, Addison Wesley, New York, 1968.
4. J. Dugundji – Topology, Allyn and Bacon, Boston, 1966.
5. J. L. Kelley – General Topology, Van Nostrand and Reinhold Co., New York, 1955.

Course code MAC 210	Commutative Algebra
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Course Outcome

Students are able to

- CO1 Know the basic definition of commutative rings, local rings, prime and maximal Ideals over commutative rings
- CO2 Explain basic concepts of The prime spectrum of a ring, the nil radical and Jacobson radical
- CO3 Study the operation on ideals and extension and contraction

- CO4 Be Familiar with the notions of Noetherian and Artinian modules
- CO5 Discuss and understand the Free modules Finitely generated modules, Simple modules, Exact sequences of modules
- CO6 Know how to localize rings and modules, and familiar with important applications of localization
- CO7 Know basic theory of Noetherian rings and study the Hilbert basis theorem .

COURSE CONTENT

Unit I

Rings and ideals - Rings and ring homomorphisms, Ideals, Quotient rings, zero-divisors, nilpotent elements, units, prime ideals and maximal ideals.

Unit II

The prime spectrum of a ring, the nil radical and Jacobson radical, operation on ideals, extension and contraction.

Unit III

Modules - Modules and modules homomorphisms, submodules and quotient modules, Direct sums, Free modules Finitely generated modules, Nakayama Lemma, Simple modules, Exact sequences of modules.

Unit IV

Modules with chain conditions - Artinian and Noetherian modules, modules of finite length, Artinian rings, Noetherian rings, Hilbert basis theorem.

Books for Reference:

1. M. F. Atiyah and I. G. Macdonald – Introduction to Commutative Algebra, Addison-Wesley.
2. C. Musili – Introduction to Rings and Modules, Narosa Publishing House.
3. Miles Reid – Under-graduate Commutative Algebra, Cambridge University Press.
4. N. S. Gopalakrishnan, Commutative Algebra, Oxonian Press.

Course code MAC 220	Theory of Numbers
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Course Outcome

Students are able to

- CO1 Define and interpret the concepts of divisibility, congruence, greatest common

divisor, prime, and prime-factorization

- CO2 Apply the Law of Quadratic Reciprocity and other methods to classify numbers as primitive roots, quadratic residues, and quadratic non-residues
- CO3 Produce rigorous arguments (proofs) centered on the material of number theory, most notably in the use of Mathematical Induction and/or the Well Ordering Principle in the proof of theorems
- CO4 Know the diophantine equations, irrational numbers, continued fractions

COURSE CONTENT

Unit I

Prime numbers, The Fundamental theorem of Arithmetic, The series of Reciprocals of primes, The Euclidean Algorithm. Fermat and Mersenne numbers. Farey series, Farey dissection of the continuum, Irrational numbers-Irrationality of m^{th} root of N , e and π .

Unit II

Arithmetical Functions – The Mobius function, The Euler' function and Sigma function, The Dirichlet product of Arithmetical functions, Multiplicative functions. Averages of Arithmetical functions – Euler summation formula, Some elementary asymptotic formulas, The average orders of $d(n)$, $\sigma(n)$, $\varphi(n)$, $\mu(n)$. An application to the distribution of lattice points visible from the origin.

Unit III

Approximation Irrational numbers, Hurwitz's Theorem, Representation of a number by two or four squares, Definition $g(k)$ and $G(k)$, Proof of $g(4) < 50$, Perfect numbers. The series of Fibonacci and Lucas.

Unit IV

Continued fractions - Finite continued fractions, Convergent of a continued fraction, Continued fractions with positive quotients. Simple continued fractions, The representation of an irreducible rational fraction by a simple continued fraction. The continued fraction algorithm and Euclid's algorithm. The difference between the fraction and its convergents, Infinite simple continued fractions, the representation of an irrational number by an infinite continued fraction, Equivalent numbers and periodic continued fractions, some special quadratic surds.

Books for Reference:

1. G. H. Hardy and E. M. Wright – An Introduction to Theory of Numbers, Oxford University Press, 1979, 5th Ed.,

2. I. Niven, H. S. Zuckerman and H. L. Montgomery – An Introduction to the Theory of Numbers, New York, John Wiley and Sons, Inc., 2004, 5th Ed.,
3. Bruce C. Berndt – Ramanujan's Note Books Volume-1 to 5, Springer.
4. G. E. Andrews – Number Theory, Dover Books, 1995.
5. T. M. Apostol – Introduction to Analytic Number Theory, Narosa Publishing House, New Delhi.

OPEN ELECTIVE (For others)

Coursecode MCC/BCC/BTC/BOC/ZOC/CSC 580	Basic Mathematics
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Course Outcome

Students are able to

- CO1 Write an argument using logical notation and determine if the argument is or is not valid
- CO2 Identify sets as well defined collections, represents sets in roster and set builder form, conversion from set builder form to roster form and vice versa.
- CO3 Understand the meaning of equal and equivalent sets and differentiate between equal and equivalent sets.
- CO4 Identify the subsets of the given sets, find the complement of a subset of a given Set, within a given universe. Represent venn diagram using sets.
- CO5 Use the simple method to solve small linear programming models by hands, given a basic feasible point
- CO6 Solve a two dimensional linear programming problem graphically
- CO7 Understand the definitions of graphs, path, connectedness, cut vertex, bridge, blocks of a graph.
- CO8 Study the properties of trees and matrix of a graph

COURSE CONTENT

Unit I

Mathematical Logic: Connection – Normal Forms – Theory of Inferences –Predicate Calculus.

Unit II

Set Theory: Operations on Sets – Basic Set Identities – Relations and Orderings, Functions.

Unit III

Introduction: formulation of LP problems, Graphical solution of LP problems. Introduction to Simplex, revised simplex methods and Dual simplex, Game theory.

Unit IV

Graph Theory: Basic Concepts of Graph Theory- Paths – Connectedness – Matrix Representation of Graphs – Trees – List structures and Graphs

Books for Reference:

1. C. L. Liu – Elements of Discrete Mathematics, McGraw-Hill, 1986.
2. Kenneth H. Rosen – Discrete Mathematics and its Applications, McGraw-Hill, 2002.
3. F. Harary – Graph Theory, Addition Wesley Reading Mass, 1969.
4. N. Deo – Graph Theory With Applications to Engineering and Computer Science, Prentice Hall of India, 1987.
5. K. R. Parthasarathy – Basic Graph Theory, Tata McGraw-Hill, New Delhi, 1994.
6. G. Chartand and L. Lesniak – Graphs and Diagraphs, wadsworth and Brooks, 2nd Ed.,
7. Clark and D. A. Holton – A First Look at Graph Theory, Allied publishers.
8. D. B. West – Introduction to Graph Theory, Pearson Education Inc.,2001, 2nd Ed.,
9. J. A. Bondy and U. S. R. Murthy – Graph Theory with applications, Elsevier, 1976.
10. S. D Sharma- Operations Research.

Fourth Semester

Course code MAD 010	Measure and Integration
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Course Outcome

Students are able to

- CO1 Understand the fundamental concepts of Mathematical Analysis and the classical theorems in Advanced Real Analysis
- CO2 Be familiar with measurable sets and functions

- CO3 Discuss the Integration a measurable function
- CO4 Discuss theorems on monotonic functions, functions of bounded variation, differentiation of an integral, absolute continuity
- CO5 Understand the fundamentals of measure theory and be acquainted with the proofs of the fundamental theorems underlying the theory of integration.
- CO6 Learn about measure theory of Signed measures, product measures

COURSE CONTENT

Unit I

Lebesgue measure - outer measure, measurable sets and Lebesgue measure, a nonmeasurable set, measurable functions.

Unit II

The Lebesgue integral – the Lebesgue Integral of a bounded function over a set of finite measure, the integral of a non-negative function, the general Lebesgue integral.

Unit III

Differentiation and integration - Differentiation of monotonic functions, functions of bounded variation, differentiation of an integral, absolute continuity.

Unit IV

Measure and integration - Measure spaces, Measurable functions, integration, Signed measures, the Radon - Nikodym theorem, Measure and outer measure, outer measure and measurability, the extension theorem, product measures.

Books for Reference:

1. H. L. Royden – Real Analysis, Prentice Hall, 3rd Ed.,
2. G. de Barra – Measure Theory and Integration, Wiley Eastern Limited.
3. Inder K. Rana – An Introduction to Measure and Integration, Narosa, 1997.

Course code MAD 020	Topology II
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Course Outcome

Students are able to

- CO1 Define and illustrate the concepts of the countability and separation axioms.
- CO2 Appreciate the beauty of deep mathematical results like Uryzohn's lemma and Tietze's extension theorem and understand the dynamics of the proof techniques.
- CO3 Discuss Tychonoff's theorem, locally compact spaces, Compactness of metric spaces
- CO4 Study Stone-Weierstrass theorems
- CO5 Describe different examples distinguishing general, geometric, and algebraic topology.
- CO6 Construct of the fundamental group of a topological space and applications to covering spaces and homotopy theory.

COURSE CONTENT

Unit I

Countability and Separation axioms - the countability axioms, the separation axioms, normality of a compact Hausdorff space.

Unit II

Urysohn's lemma, Tietze's extension theorem, Urysohn's metrization theorem, Partitions of unity.

Unit III

Tychonoff's theorem on the product of compact spaces. Local finiteness, Paracompactness, Normality of a paracompact space.

Unit IV

The Fundamental group and the Fundamental group of a circle, The Fundamental group of the punctured plane, Essential and Inessential Maps, The Fundamental Theorem of Algebra.

Books for Reference:

1. James R. Munkres - A First Course in Topology , Prentice Hall India, 2000, 2nd Ed.,
2. G. F. Simmons – Introduction to Topology and Modern Analysis, McGraw-Hill, Kogakusha, 1968.
3. S. Willard – General Topology, Addison Wesley, New York, 1968.
4. J. Dugundji – Topology, Allyn and Bacon, Boston, 1966.

5. J. L. Kelley – General Topology, Van Nostrand and Reinhold Co., New York, 1955.

Course code MAD 230	Differential Geometry
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Course Outcome

Students are able to

- CO1 Define and analyze the basic concepts of space curves and plane curves
- CO2 Introduce the fundamentals of differential geometry primarily by focusing on the theory of curves and surfaces in three space.
- CO3 Compute quantities of geometric interest such as curvature, as well as develop a facility to compute in various specialized systems
- CO4 Understand the global properties of curves such as the four vertex theorem.
- CO5 Know the theory of surfaces introduces the fundamental quadratic forms of a surface, intrinsic and extrinsic geometry of surfaces, and the Gauss theorem.
- CO6 Introduce the method of the moving frame and over determined systems of differential equations as they arise in surface theory.

COURSE CONTENT

Unit I

Plane curves and Space curves – Frenet-Serret Formulae. Global properties of curves – Simple closed curves, The isoperimetric inequality, The Four Vertex theorem. Surfaces in three dimensions – Smooth surfaces, Tangents, Normals and Orientability, Quadric surfaces.

Unit II

The First Fundamental form – The lengths of curves on surfaces, Isometries of surfaces, Conformal mappings of surfaces, Surface area, Equiareal Maps and a theorem of Archimedes.

Unit III

Curvature of surfaces – The Second Fundamental form, The Curvature of curves on a surface, Normal and Principal Curvatures.

Unit IV

Gaussian Curvature and The Gauss' Map – The Gaussian and The mean Curvatures, The Pseudo sphere, Flat surfaces, Surfaces of Constant Mean Curvature, Gaussian Curvature of Compact surfaces, The Gauss' Map.

Books for Reference:

1. A. Pressley – Elementary Differential Geometry, Under-graduate Mathematics Series, Springer.
2. T. J. Willmore – An Introduction to Differential Geometry, Oxford University Press.
3. D. Somasundaram – Differential Geometry: A First Course, Narosa, 2005.

Course code MAD 220	Theory of Partitions
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Course Outcome

Students are able to

- CO1 Know the definitions of partitions
- CO2 Apply the q-binomial theorem and Ramanujan ${}_1\psi_1$ - summation formula
- CO3 Know the congruence of partition
- CO4 Apply the q-series

COURSE CONTENT

Unit I

Partitions - partitions of numbers, the generating function of $p(n)$, other generating functions, two theorems of Euler, Jacobi's triple product identity and its applications.

Unit II

${}_1\psi_1$ - summation formula and its applications, combinatorial proofs of Euler's identity, Euler's pentagonal number theorem, Franklin's combinatorial proof.

Unit III

Congruence properties of partition function, the Rogers - Ramanujan Identities.

Unit IV

Elementary series - product identities, Euler's, Gauss', Heine's, Jacobi's identities. Restricted Partitions – Gaussian, Frobenius partitions.

Books for Reference:

1. G. H. Hardy and E. M. Wright – An Introduction to Theory of Numbers, Oxford University Press, 1979, 5th Ed.,

2. I. Niven, H. S. Zuckerman and H. L. Montgomery – An Introduction to the Theory of Numbers, New York, John Wiley and Sons, Inc., 2004, 5th Ed.,
3. Bruce C. Berndt – Ramanujan's Note Books Volumes-1 to 5.
4. G. E. Andrews – The Theory of Partitions, Addison Wesley, 1976.
5. A. K. Agarwal, Padmavathamma, M. V. Subbarao – Partition Theory, Atma Ram & Sons, Chandigarh, 2005.

MASTER OF SOCIAL WORK

MSW

SYLLABUS

**CREDIT BASED, CHOICE BASED CONTINUOUS ASSESSMENT PATTERNED
EDUCATION SYSTEM**

(Regulations, Scheme of Examination and Course Content)

To be effective from the Academic Year 2020-21 onwards

**DEPARTMENT OF STUDIES IN SOCIAL WORK
JSS COLLEGE OF ARTS, COMMERCE AND
SCIENCE, OOTY ROAD, MYSORE**

JSS College of Arts, Commerce and Science

(Autonomous)

Ooty Road, Mysore

Master of Social Work Programme

DISTRIBUTION OF COURSE CONTENT AND CREDITS

MSW- I Semester

Compulsory additional papers for non BSW students

Sl. No.	Code No.	Paper Title	L	T	P	Credits
1	NSW-1	Social Science Perspectives for Social Work Practice	-	-	-	-
2	NSW-2	Term Project	-	-	-	-

1. A bridge course will be conducted for a period of 5 days covering the required course input. There are no credits allotted to this course and no written examination too. However, attendance of 75% is compulsory.

2. Paper code: NSW2 : **Term project:**

The term project is a team-exercise consisting 3 to 5 students. The team is expected to select a theme relevant to current social issues in consultation with the supervisor and make an exhaustive survey of literature on the chosen theme including empirical studies made on the same.

Further, the group shall also collect the experiences or opinions of people on the issues and make a detailed presentation. Flexibility is accorded in planning and executing the term project. Creative and analytical approaches are to be carried out under the direct supervision of a faculty supervisor.

The report of the term project has to be submitted before the end of theory examination of the first semester. The term project is offered only for the non-BSW students. However, students with BSW background are also encouraged to opt for the term project, as an innovative approach in social work, if they desire so.

DISTRIBUTION OF CREDITS

Sl. No.	Course Type	Credits
1	HARD CORE (HC)	52
2	SOFT CORE (SC)	20
3	OPEN ELECTIVE (OE)	04
	TOTAL	76

Semester- wise Distribution of Course Content and Credits

I Semester

Sl. No.	Course Code	Course Title	L:T:P	Credits
1.	SWA HC-1	Social Work - History and Ideologies	2:1:0	3
2.	SWA HC-2	Work with Individuals and Families	2:1:0	3
3.	SWA HC-3	Work with Groups	2:1:0	3
4.	SWA HC-4	Work with Communities	2:1:0	3
5.	SWA HC-5	Human Growth and Development	2:1:0	3
6.	SWA HC-6	Social Work Practicum – I	0:1:2	3
		Total		18

II Semester

Sl. No.	Course Code	Course Title	L:T:P	Credits
1.	SWB HC-7	Management of Developmental and Welfare Services	2:1:0	3
2.	SWB HC-8	Social Work Research and Statistics	2:1:0	3
3.	SWB HC-9	Social Work Practicum – II (Social Work Camp and Summer Placement)	0:0:3	3
4.	SWB HC-10	Social Work Practicum - III	0:1:2	3
5.	SWB SC-1	Communication and Counselling / Gandhian Approach to Welfare and Development	3:1:0	4
6.	SWB SC-2	Personal and Professional Growth/ Population and Environment	2:1:0	3
			Total	19

III Semester

Sl. No.	Course Code	Course Title	L:T:P	Credits
1.	SWC HC-11	Human Resource Management	2:1:0	3
2.	SWC HC-12	Social Work Practicum – IV	0:1:2	3
3.	SWC SC-3	Social Work with Tribal and Rural communities/Organizational Behavior and Organizational Development	2:1:0	3
4.	SWC SC-4	Preventive and Social Medicine and Medical Social Work /Rehabilitation and After Care Services	2:1:0	3
5.	SWC SC-5	Social Policy, Planning and Development/ Legal System in India	2:1:0	3
6.	SWC OE	Gerontological Social Work / Social Work Practice with Children	4:0:0	4
			Total	19

IV Semester

Sl. No.	Course Code	Course Title	L:T:P	Credits
1	SWD HC-13	Employee Relations and Legislation	2:1:0	3
2	SWD HC-14	Mental Health and Psychiatric Social Work	2:1:0	3
3	SWD HC-15	Major Project	0:2:4	6
4	SWD HC-16	Social Work Practicum – V	0:1:2	3
5	SWD HC-17	Social Work Practicum – VI (Block Placement)	0:0:2	2
6	SWD SC-6	Human Resource Development and Employee Wellness/Case Studies	3:1:0	4
			Total	21

Note: In a Semester for only one Soft Core Course, there can be two choices.

Objectives of the Course (Master of Social Work):

1. To provide education and training in social work to those desirous of making a career in social work practice.
2. To provide opportunities through intensive field practicum to work with variety of people in their development and provide service to those who are in need of it.
3. To provide inter-disciplinary collaboration for better understanding of human problems, services and issues related to human development.
4. To link theory with practice in every sphere of human service.
5. To develop requisite knowledge, skills and values in working with people.
6. To promote among learners a sense of responsibility and commitment to work with different sections of people and especially of the vulnerable sections of the society
7. To promote opportunities and to create awareness for personal growth
8. To acquire knowledge and skills in undertaking practice-based research and to administer human service organizations

Name of the Course:

The course shall be called ' **Master of Social Work**' (MSW).

Duration of the Course:

The Course of study **for MSW Degree** shall extend over a period of four semesters spreading over two academic years.

Regulations:

ELIGIBILITY FOR ADMISSION TO MSW COURSE

Candidates who have passed BSW/ BA/ B.Sc. / B.Com. / BBM/ B.C.A / LLB of the University of Mysore or any other university recognized as equivalent there to are eligible for admission to MSW course. Candidates will be selected for admission as per the general guidelines issued from the University of Mysore from time to time. The Department/University shall conduct entrance examination for admission to the course.

The examination is of two hour duration and the question paper comprises of 100 objective type questions - 20% questions from general knowledge and current social issues, 60% from science & social sciences, and another 20% questions will be from reasoning and numerical ability. Merit will be assessed on the basis of performance in the entrance examination and performance in the undergraduate examination on equal weightage.

PATTERN OF QUESTION PAPER

Pattern 3

(The Question paper comprising of 3 parts: A,B and C as follows)

PART – A

There are 8 questions and a candidate has to answer any 5 questions. Each question carries 2 marks. This part covers all units of the syllabus.

PART – B

There are 8 questions and a candidate has to answer any 5 questions. Each question carries 5 marks. This part covers all units of the syllabus.

PART –C

There is a single question such as case study (may contain sub questions) covering entire syllabus carrying 15 marks. No choice.

ASSESSMENT OF SOCIAL WORK PRACTICUM

A viva-voce examination shall be conducted for each candidate in all semesters. The performance of the candidate shall be assessed by a committee consisting of three members as follows.

1. Chairperson of the Department
2. One Senior Member of the Faculty
3. One External Examiner

SOCIAL WORK PRACTICUM

The practicum with different learning opportunities is designed to provide scope to develop and enhance professional practice skills. Learning is aided through observation, analysis of social realities and experience of participation in designing and providing social work intervention.

The tasks are organized to help the learner acquire beginning skills, practice those already acquired, and master them from simple to complex. The learner is gradually encouraged to become an independent worker during the course of study.

Objectives

The objectives are met by providing a variety of experiences to learners to:

1.
 - i. Develop the ability to observe and analyze social realities. Understand the characteristics of social systems and their dynamics. Appreciate society's response to people's needs, problems and social issues.
 - ii. Develop critical understanding of the application of legislation, legal process, and social policy.
2.
 - i. Develop the ability to examine the process of programme management and participate in the effort at various levels.
 - ii. Develop the ability to recognize the need for newer programs, initiate and Participate in them.
 - iii. Use Human Rights tools, understanding of gender justice, and need for equity in all intervention.
 - iv. Develop an understanding of organizational structure, resource management, and day-to-day administration for human service programmes - developmental and welfare programmes
 - v. Develop the capacity to integrate knowledge and practice-theory by participating in intervention.
3.
 - i. Clarify and imbibe values which sustain positive attitude and professional ethics.
 - ii. Develop the capacity for self-direction, growth and change through self awareness.
4.
 - i. Enhance writing skills to document practice appropriately. Recordings to be viewed as an expression of interest, motivation and involvement in practice and as evidence of enrichment in the process of professional growth.

To meet these outcomes, several opportunities with specific objectives are designed. The different sets of opportunities with details of content and related tasks are listed separately.

Paper code: NSW -1

Paper Title: SOCIAL SCIENCE PERSPECTIVES FOR SOCIAL WORK PRACTICE

INTRODUCTION

This course provides the learners basic understanding of relevant concepts from social sciences to help the learners to study and understand social phenomenon. Further, it helps the learner develop skills for social analysis and understand developmental processes.

OBJECTIVES

- a. Understand the concepts to examine social phenomenon.
- b. Develop skills to analyse Indian society and change.
- c. Understand change and conflict.
- d. Understand the system for economic order.
- e. Develop skills for social analysis.
- f. Understand the development and its impact.

Course Content

UNIT I

Sociology and its relationship to other disciplines: Meaning, scope and significance - Its relationship with other social sciences such as History, Economics, Politics, Psychology, Anthropology and Social work.

Society and Culture: Society as a system of relationship - Social Structure: Meaning, status and roles - Culture: Meaning and contents-Tradition, customs, values, norms, folklore and mores.

Indian Society: Composition of Indian Society: the concept of unity amidst diversity - Social classification in India: Tribal, rural and urban divisions - Social stratification in India: Meaning, caste, class divisions.

Socialization: Meaning, process of socialisation - The development of self - Agencies of socialisation.

UNIT II

Social Groups, Social Institutions and Social Control - Meaning and types: Primary and Secondary groups, in-groups and out-groups, reference groups - Types of social institutions: Marriage, Family, Religion, State and Law.

Meaning and Functions: Social Control exercised through the social institutions

Social Change: Meaning, characteristics and factors inducing change with reference to India.

Social Movements in India: Meaning, factors essential for a Movement - Dominant social movements in India - Social reform movement and contributions of social reforms - Peasant movement - Trade Union movement - Social movements and social change in India.

UNIT III

Development - A Human Right Perspective: Social Ideals of Indian Constitution - Fundamental Rights - Human Rights.

Socio-economic order and comparative economic system: Capitalism, Socialism and Mixed economy, their features, merits and demerits - Marxian political economy.

Social Analysis: Significance of social analysis: A brief analysis of socioeconomic, political and cultural systems - Inter-linkages in the Indian context.

UNIT IV

Under-development and its causes and Contemporary Development Dynamics: A historical overview with reference to developing countries of Asia, Africa and Latin America - North-south relations, world trades, Multinational corporations and their influences on Third World economics - Trends and counter trends (Paradoxes) in the global, political, economic, military, ecological and socio-cultural spheres.

Theories of Economic Development, Globalisation and its impact on Developing Countries: Stages of growth theory - Structural internationalist theory Privatization, liberalization and structural adjustment programmes - Role of international financial institutions.

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2. Agrawal, A. N. and Lal Kundan 1989 Economics and Development and Planning, Delhi, New Vikas Publishing House Private Limited
3. Augushine, John S. (Ed.) 1989 Strategies for Third World Development, New Delhi: Sage Publications.
4. Basu, Durga Das. 1983 Introduction to the Constitution of India, New Delhi, Prentice-Hall of India Private Ltd.
5. Bert N. Adams. 1975 A Sociological Interpretation, Chicago: Rang McNally College.

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Introduction to Social Sciences, Delhi: C.B.S. Publication.
23. Panday Rajendra. 1986
Reality, New Delhi: Criterion Publication.
24. Pandey, A. K. 1997
Tribal Society in India, New Delhi: Manak Publishing Ltd.
25. Rao, D. Bhaskar, 1998
World Summit for Social Development, New Delhi: Discovery Publications.
26. Rao, V. Lakshmana 1994
Ashish
Essays on Indian Economy, New Delhi: Publishing House.
27. Reddy, D. V. 1994
Development and New International Economic Order, New Delhi: Deep and Deep Publications.
28. Ross, Aileen D. 1961
The Hindu Family in its Urban Setting, Delhi: Oxford University Press.
29. Rugman, Alan 2000
The End of Globalization, London: Random House.
30. Sharma, S. L. (Ed.) 1986
Development: Socio-Cultural Dimensions, Jaipur: Rawat Publications.
31. Sharma, K. L. (Ed.) 1999
Caste and Race in India, Mumbai: Popular Prakashan.
32. Sharma, R. K. 1997
Indian Society - Institution and Change, New Delhi: Atlantic Publishers.
33. Singh, K. S. 1985
Tribal Society in India, New Delhi: Manohar Publishing House.
34. Srinivas, M. N. 1996
Village, Caste, Gender and Method (Essay in Indian Social Anthropology), Delhi: Oxford University Press.
35. Steveus Smart M., Laura S. Smart. 1976
Families Developing Relationships, New York: Macmillan Publishing Co.
36. Tandon, B. B., and Tandon, K. K. 1997.
Indian Economy, New Delhi: Tata McGraw-Hill Publishing Company Limited

Journals/ Magazines

Sociological Bulletin (Journal of the Indian Sociological Society).

Contribution to Indian Sociology.

Social Change, Issues and Perspectives (Journal of the Council for Social Development).

Economic and Political Weekly, EPW Research Foundations, Mumbai.

Paper code: NSW 2

TERM PROJECT

The term project is a team-exercise consisting 3 to 5 students. The team is expected to select a theme relevant to current social issues in consultation with the supervisor and make an exhaustive survey of literature on the chosen theme including empirical studies made on the same.

Further, the group shall also collect the experiences or opinions of people on the issues and make a detailed presentation.

Flexibility is accorded in planning and executing the term project. Creative and analytical approaches are to be carried out under the direct supervision of a faculty supervisor.

The report of the term project has to be submitted before the end of theory examination of that semester to the Department of Social Work, University of Mysore, Mysore through the supervisor and Chairman/Principal of the college.

The term project is offered only for the non-BSW students. However, students with BSW background are also encouraged to opt for the term project, as an innovative approach in social work, if they desire so.

Evaluation of the term project will be done along with the viva-voce examination by the viva-voce committee, constituted for the assessment of social work practicum or similar committee may be constituted, if required.

Odd semester

Paper code: SWAHC- 1

Paper Title: SOCIAL WORK - HISTORY AND IDEOLOGIES

INTRODUCTION

This course aims at introducing the learners to a critical inquiry into the history and ideologies of social change and professional social work.

OBJECTIVES

- a. Understand the history of evolution of social work profession, both in India and the West.
- b. Develop insights into the origin and development of ideologies, approaches to social change.
- c. Understand rationale, goals, ideals and ethics for social change.
- d. Understand the perceptions of people and social problems, the status of benefactors and their motives.
- e. Develop skills to understand contemporary reality in its historical context.
- f. Understand self as a part of own environment and explore own assumptions, ideals, values to develop sensitivity to marginalization of vulnerable groups.

Course Content:

UNIT I

Indian History of Social Work Profession: Introduction - Beginning of social work education - Welfare versus developmental orientation in social work - Professionalization of social work values, education, knowledge and professional associations - Goals, values, functions/roles and process of social work - Interface between professional and voluntary social work, social work ethics.

UNIT II

Indian History of Ideologies for Social Change -Ancient period: Vedic, Vedantic and non-Vedic Ideologies, Spirituality - Medieval period: Zoroastrianism and Islam in India - Mysticism of Bhakti and Sufi movements and Sikhism.

Modern period: Christianity in India - Hindu reform movements - Dalit movements - Gandhian ideology and Sarvodaya movement - Nationalism - Ideology of the Indian Constitution - Ideology of voluntary organisations and voluntary action.

UNIT III

Contemporary Ideologies for Social Change: Neoliberalism and Globalisation - Post modernism - Multiculturalism - Ideology of action groups and social movements - Ideology of non-governmental organisations.
Role of state in providing social welfare services.

UNIT IV

Western History of Ideologies for Social Change: Organized and scientific charity - Beginning of social work education - Clinical social work - Ecological social work - Attributes of a profession.

Western History of Social Work Profession - Medieval period: Judeo-Christian-ideologies - Secular humanism and Protestantism - Modern period: Rationalism and Welfarism - Liberalism and democracy - Utilitarianism and Social Darwinism - Socialism and human rights - Emerging ideologies of professional social work.

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Charles R. and Contributors
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and Political Thought, New Delhi: Sage.
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Bombay, Asia Publishing House.
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Social Work Education in India -- Retrospect
and Prospect Udaipur, Himansu Publications.
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Social Work: In the Third Millennium (Some
Concerns and Challenges), Sriniketan, Department
of Social Work, Visva-Bharati.
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Perspective, Bangalore: Visthar
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Journals/ Magazines

Economic and Political Weekly, The Indian Journal of Social Work, Lokayan Bulletin and Vikalp.

Odd semester

Paper code: SWAHC- 2

Paper Title: WORK WITH INDIVIDUALS AND FAMILIES

INTRODUCTION

This course aims to develop simple to complex skills of working with individuals and families in various situations like crisis, preventive, facilitative and developmental.

OBJECTIVES

- a. Understand casework as a method of social work, and appreciate its place in social work practice.
- b. Understand the values and principles of working with individuals and families.
- c. Develop the ability to critically analyse problems of individuals and families and factors affecting them.
- d. Enhance understanding of the basic concepts, tools and techniques in working with individuals and families, in problem-solving and in developmental work.
- e. Develop appropriate skills and attitudes to work with individuals and families.

Course Content

UNIT I

Social case work: Definitions, scope, historical development - Influence of psychoanalysis on casework - Introduction of casework as a method of social work - Concepts of adjustment and maladjustment - Philosophical assumptions and casework values.

Principles of casework: Individualization, acceptance, non-judgmental attitude, participation, relationship, effective communication of feeling, client self-determination, and confidentiality.

Components of social casework: The person, the problem, the place and the process.
Process in casework: Study, assessment, intervention, evaluation, follow-up, and termination.

UNIT II

Types of problems faced by Individuals and families; individual differences and needs - Family assessment in casework practice.

Theories and approaches: Psycho-social approach, Functional approach, Problem-solving approach, Crisis Theory, Family intervention, Behavioural modification, Transactional analysis and Holistic approach.

UNIT III

Tools for Help: Case work tools: Interview, home visit, observation, listening, communication skills, rapport building.

Records: Nature, purpose and principles of recording.

Techniques of casework: Supportive, resource enhancement and counseling.

Self as a professional: Professional self - Conflicts and dilemmas in working with individuals and families.

UNIT IV

Application of Method: Primary and secondary settings - Application of methods in family, women, and child welfare settings, marriage counselling centres, schools settings, medical and psychiatric settings, correctional institutions, and industry.

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Odd semester

Paper code: SWAHC -3

Paper Title: WORK WITH GROUPS

INTRODUCTION

This course aims at developing the understanding of Group Work as a method, developing skills for intervention, and gaining knowledge of the scope of this method in various settings.

OBJECTIVES

- a. Develop awareness about the specific characteristics of Group Work and its contributions as a method of social work intervention,
- b. Gain knowledge about group formation and the use of a variety of group approaches.
- c. Develop understanding of concepts, dynamics and small group theory in relation to all types of groups, e.g. family, staff, committee, long-term client groups.
- d. Identify the various situations and settings where the method could be used in the context of social realities of the country.

Course Content

UNIT I

Introduction and history of Group Work: Understanding of groups - Characteristics and significance of group - Definition of Social Group Work - Characteristics of Social Group Work - Purpose of Social Group Work; Historical evolution of group work with special emphasis on the Indian Context.

Type of Groups: Types and approaches based on objectives and purpose - Type of membership - Time duration - Social group work in different settings and analysis of group processes.

Values and Principles in group work and Characteristics of Group formation: Values in social group work- Principles in group work - Assumptions underlying social group work - Factors of group formation - Formulation of goals - Identification of problems for work.

Pre-group and Initial Phase: Planning model - Characteristics of pre group phase - Group structures - Facilitation skills and role of worker in pre-group and initial phase.

UNIT II

Group Processes and Group Dynamics: Importance of group processes - Typical patterns - Processes in different type of groups - Worker's skills in identifying and understanding processes - Bond, sub-groups, role.

Leadership - Isolation - Decision making - Conflict - Communication - Relationships.

UNIT III

Middle Phase and Use of Program: Characteristics of middle phase - Group structures - Group dynamics - Facilitation skills - Role of group workers - - Comparison across phases - Concept and principles - Program planning - Skills in program planning

Facilitation: Knowledge of skills and techniques for effective work with groups/problem solving.

Recordings in Group work: Importance of recording in social group work - Principles of recording - Recording structure - Types of recording.

UNIT IV

Evaluation in Groups and Termination Phase : Importance of evaluation - Types of evaluation - Methods of evaluation - Need for termination - Types of termination - Characteristics of termination phase - Worker's skills.

Application of Group Work: Application in health settings, school settings, family welfare settings, industrial settings, women and child welfare settings.

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Odd semester

Paper code:SWAHC-4

Paper Title: WORK WITH COMMUNITIES

INTRODUCTION

Community organization / development, as a method of social work practice, is seen as a means to facilitate communities towards self-directed change. It takes as its basis the inequalities in society manifested through processes of marginalization, discrimination or disempowerment of groups, which have resulted in the loss of control over resources, be they tangible or intangible. The strategies of Community Organisation practice being addressed as part of the course cover a range spanning different ideologies, from those being people-initiated, and those that are initiated by the elite. Community organization is seen as a means as well as an end, where collective processes are to sustain the community's capacity to bring about change.

OBJECTIVES

- a. Understand the critical elements of community organisation practice.
- b. Enhance critical understanding of the models and strategies for community organisation practice.
- c. Make the micro-macro connections between the range of complex issues in practice.
- d. Develop attitudes conducive to participatory activities for civil society.

Course Content

UNIT I

Community: Concept, characteristics, types and functions.

Understanding of community organisation practice: Definition of community organisation, values and principles of Community Organisations, ethics of community organisation practice.

Community Organisation Practice: Community work within social work, Understanding Human Rights in community organisation practice.

Historical development of community organisation practice.

Power: Concept of power - The range of perspectives - Dimensions of power relevant to community organization.

Empowerment: Concept of Empowerment - Barriers to, process and cycle of empowerment.

Gender and Empowerment: Gender sensitive community organization practice

UNIT II

Models and Strategies of Community Organization - Locality Development Model - Social Planning Model - Social Action Model - Select methods of public interest mobilization, litigation, protests and demonstrations, Dealing with authorities, Public Relations, Planning, Monitoring and Evaluation - Roles in different models attributes and attitude.

UNIT III

Community Organization as a Method: Relevance of community organisation as a method across different spheres of social work intervention and relook at own attitudes.

Skills of Community Organization Practitioner: Problem analysis, resource mobilization, conflict resolution, organizing meetings, writing and documentation, networking, training.

UNIT IV

Strategy and Roles: Unionization as a strategy - Advocacy in community organization.

Current debates in Community Organisation Practice: Emerging issues - Impact of macro policies.

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Paper code: SWAHC-5

Paper Title: HUMAN GROWTH AND DEVELOPMENT

INTRODUCTION

The course aims to introduce the learners to the development of the individual across the life span, in a system and an ecological perspective. It also provides an understanding of human development and behaviour, in contextual influences, including individuals in disadvantaged or special contexts. The theoretical inputs are to enhance the understanding of people's growth, health, and development at various stages as bio-psycho-socio-spiritual being over the life span.

OBJECTIVES

- a. Develop an overall understanding of the principles of growth; their relevance and application to behaviour at various phases in the life span.
- b. Understand the twin roles of individual's heritage and environmental influences in growth and development.
- c. Understand interactional nature of growth and behaviour at various stages in the life span: infancy, childhood, adolescence, youth, adulthood and old age, and impact of cultural aspects.
- d. Develop sensitivity towards needs, developmental tasks and health status along with need for developmental programmes for the same.
- e. Apply the information of growth, development and health in social work practice in general and individuals, groups and communities in particular.

Course Content

UNIT I

Life Span: Beginning of life - Human reproductive system; Fertilization and Foetal development - Delivery and pre-natal and post-natal care and their importance in development.

Principles of growth and development - Methods of studying human behaviour, - Role of heredity and environment - Social customs traditions, values in parenting and child rearing practices, deprivation and development during stages of life span. Understanding of the Indian concept of life span stages.

UNIT II

Developmental periods: Infancy, babyhood, childhood, puberty, adolescence -. Growth, hazards, lifestyle effects

Adulthood - Growth, personal and social adjustment, health, sexuality, vocational and marital adjustment.

Aging - Characteristics, hobbies, adjustment, physical and mental health, death, dying and bereavement.

Special focus is on psychosocial development, moral development, and personality development vis-a-vis the influence of the contexts of development., (The contexts here refers to gender, family, significant others, neighbourhood: peers, school, community, work place and other larger contexts like the society and culture. Emphasis is placed on the Indian context of development, variations from the

normal patterns of development and views on the stages)

UNIT III

Theories of Human Development: A critical look at the theories of human development - Freud's psychosexual theory, Erikson's psychosocial theory, learning theories.

UNIT IV

Basic human needs: Physical, psychological and intellectual needs, stress - Coping and social support.

Motivation, frustration and conflicts - Emotions and emotional behaviour.

Personality: Definition, nature, types and assessment of personality.

Intelligence: Concept, levels of intelligence, influence of heredity and environment, assessment of intelligence.

Relevance of Psychology to social work practice across the stages of development, period specific needs, tasks and challenges.

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Odd semester

Code: SWAHC-6

Title: SOCIAL WORK PRACTICUM - I

Orientation provides information regarding.

- i. the importance and place of the practicum in the educational programme.
- ii. the purpose, functions and ethics in professional practice.

In the first four weeks, the learners may make a local directory to include emergency numbers of Hospitals/ PHCs/ Police/ Panchayath Office and Network Agencies along with references to other developmental and welfare services in the location.

Visits - provide an exposure to and understanding of the services provided in responses to people's needs. (Agencies in health setting, education, community, institutional services, criminal justice system, civic administration, rehabilitation etc.)

Structured experience laboratory - is a classroom activity, which provides opportunities through the games/activities, to form the involvement of self in various practice skills. These laboratory experiences are designed in small groups to encourage participation, sharing of the experience and aid in examining learning and applications of skills. These sessions have a specific objective of experiencing self, and applying /using self in practice. (Relationship skills, Communication skills etc., will be focused)

Concurrent practice learning of two-days a week - on going learning of practice is an opportunity to develop intervention skills in reality situations. This entails learning social work practice for two, or two and a half days or its equivalent, each week of the semester.

The learners may be placed in agencies or in communities to initiate and participate in direct service delivery. Practice learning is a vital component of the educational opportunity to be provided to the learner. The teaching-learning process must be designed to help the learner to move on the mastering strategies, skills and techniques to practice social work.

Even semester (II Semester)

Code: SWBHC -7

Paper Title: MANAGEMENT OF DEVELOPMENTAL AND WELFARE SERVICES

INTRODUCTION

The course aims to develop management competencies to function in organizations, participate as a team member and understand the role of a social work programmes manager.

OBJECTIVES

- a. Understand the overall environment and its impact on the nature, structure and development of organizations in corporate, public and voluntary sectors in the context of social work profession.
- b. Understand policies and procedures involved in establishing and maintaining human service organizations.
- c. Acquire skills to network and participate in the management of resources - human, material and environmental.
- d. Develop skills to participate in management of programmes, as a part of the inter-disciplinary team and initiate as well as develop new programmes.
- e. Develop ability to analyse the practices applied in specific settings.

Course Content

UNIT I

Social Services: Need for welfare and developmental organisations, Factors determining social welfare programmes, Development and Welfare organizations' response to societal needs; role of state, voluntary and corporate sector.

Management services: Types of settings, organizational characteristics like origin, nature, size, structure, and design, organizational climate and impact of socio-political environment - Management process: Vision, Planning, Organizing, Directing, Staffing, Coordination, Reporting, Budgeting.

Establishment: Registration, different types of legislations, legal status, constitution, rules and procedure, goals - Financial resources: Organizational Budget, Sources of finance, Fund Raising, Records, Audit.

UNIT II

Physical: All activities related to acquiring, hiring and maintaining importable structure and infrastructure, maintenance of premises and daily upkeep.

Enhancing the involvement and the potential of people in organization's executive boards, committees; professionals and other staff-relationship, communication, team work, and facilitating team building, supervision, and participation in training.

UNIT III

Programme Development: Programme management: long term, short term, and Documentation.

Project proposals based on felt-needs, nature of resources, eligibility criteria, records, evaluation and research.

Impact analysis - Qualitative and quantitative.

UNIT IV

Public Relations: Public relations need and its promotion by all in the organisation. Representing the organization, networking, public, corporate and voluntary sector, resource building, accountability, transparency, use of media for publicity.

Change and its Management: Understand and manage change, innovation in a rapidly changing social environment: for policy programmes and structure.

Organizational understanding: Conflict, conflict resolution, creating positive climate.

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Paper Code: SWBHC-8

Paper Title: SOCIAL WORK RESEARCH AND STATISTICS

INTRODUCTION

This course is to equip learners to utilize, and conduct research as service managers to improve services, evaluate, and develop new services and intervention methods: strategies and techniques and also, be an effective consumer of other researches.

OBJECTIVES

- a. Develop an understanding of scientific approach to human inquiry in comparison to the native or common sense approach in various aspects, and its process.
- b. Understand major research strategies, meaning, scope and importance of social work research.
- c. Develop an ability to see the linkages between practice, research, theory and their role in enriching one another.
- d. Develop ability to conceptualize, formulate and conduct simple research projects/exercises (This would include a broad range of basic research skills such as conceptualization of a research strategy and problem; writing a research proposal; developing tools for collecting data; use of sampling, strategies; data collection, processing, presentation, analysis and interpretation; and writing research report etc).
- e. Make informed assessment and judicious use of research studies and findings.
- f. Develop skills for use of library and documentation services for research.

Course Content

UNIT I

Science - Meaning and assumptions, scientific approach in comparison to the native or common sense approach.

Scientific attitude; Scientific method; application of scientific method for the study of social phenomena.

Research: Definition and objectives, Social Work Research: Meaning, objectives, functions and limitations; Scope of social work research in India; Agencies sponsoring and conducting social work research, ethics in research.

Problem identification: Criteria for the selection of research problem; Problem formulation.

Concepts, constructs, variables, conceptual and operational definitions. Hypothesis: Meaning, importance, uses and requirements.

UNIT II

Design of research: Definition and importance; types of research design; exploratory, descriptive, experimental, evaluative design, participatory research and action research.

Source and Types of Data: Primary and secondary, objective and subjective, qualitative and quantitative.

Sampling: Sample and population: Rationale and Characteristics of sampling; methods of sampling, general considerations in the determination of sample size.

Methods of collection of primary data:

Observation: Structured and unstructured; participant and non-participant. Questionnaire, interview schedule and interview guide. Pilot study and Pre-testing.

Scales: Need for scales, some prominent scaling procedures.

Case study: Meaning, uses, steps.

Secondary data: Official data, personal documents, problem in the use of secondary data

UNIT III

Processing of data: Content, editing, coding data classification, manual and mechanical tabulation of data; frequency distribution, diagrammatic and graphic presentation - use of computers.

Issues related to Social Work Research: Interpretation of data, research reporting: contents of research report: foot-note, references, bibliography, preparation of abstract; the art of making book review.

UNIT IV

Statistics: Definition, functions and importance

Measures of Central Tendency; Measures of Dispersion.

Chi-square, Correlation Coefficient, 't' distribution; Analysis of Variance and 'F' distribution.

SPSS package.

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Code : SWBHC-9

Title: SOCIAL WORK PRACTICUM - II:

SOCIAL WORK CAMP:

Rural/ Tribal camps with a duration of 7 - 10 days - provide opportunities to experience rural life, analyze rural dynamics, and observe the functioning of local self government and voluntary organisations. This experience aids peer participation in planning for activities for own group and those for local people. It also helps develop skills to carry out, evaluate, and report the experience.

SUMMER PLACEMENT:

Summer Placement - provides an opportunity to experience day-to-day work in a setting. The learner gets involved with direct practice with the client system and with the ongoing management operations of the setting. The time frame recommended for this experience is about three weeks, after the first year of the post-graduate programme. The learner may use the same setting for data collection of research project, if such an arrangement is part of the plan.

Code: SWBHC -10

Title : SOCIAL WORK PRACTICUM - III

Concurrent practice learning of two-days a week - on going learning of practice is an opportunity to develop intervention skills in reality situations. This entails learning social work practice for two, or two and a half days or its equivalent, each week of the semester. The learners may be placed in agencies or in communities to initiate and participate in direct service delivery. Practice learning is a vital component of the educational opportunity to be provided to the learner. The teaching-learning process must be designed to help the learner to move on the mastering strategies, skills and techniques to practice social work.

Even semester

Paper Code: SWBSC-1

Paper title: COMMUNICATION AND COUNSELING

INTRODUCTION

This paper relates the relevance of components of communication and counseling in social work practice.

OBJECTIVES

- a. Understand the meaning and importance of communication in day-to-day life.
- b. Focus on interpersonal communication of interviewing and allied aspects.
- c. Develop holistic understanding of counseling as a tool for help.
- d. Acquire knowledge of various approaches: their theoretical under-pinnings for goals, values, processes and techniques,
- e. Develop skills of application to real life situations.

Course Content

UNIT I

Communication: Meaning and importance of communication.

Process of communication: Key elements in the communication process - Communication, message, audience; channel of communication. Verbal and non-verbal communication.

Basics of Communication.

Education and communication for national development.

Interpersonal communication: Interviewing - Objectives, principles of interviewing; listening, qualities of effective communicator.

Seminars, conferences, lectures, group discussion, panel discussion, symposium, workshop, role playing, simulation exercises, written communication, report writing, letter writing, article/essay writing, games, brain storming, street play, field work exposure.

UNIT II

Visual aids in communication: Poster making, use of notice boards, flip charts, charts, flash cards, photographs, pamphlets, slide shows.

Mass Communication: Television, exhibition, newspapers and magazines, advertisements, radio, film, VCD/ DVD, e-mail, internet.

Impact of mass communication on society, family, marriage and child development.

Communication Analysis and Planning: Planning and executing a communication campaign on an issue using various methods of communication.

UNIT III

Counseling: Definition, nature and goals, areas of counseling; Historical background and origins of counseling, ethical nature of counseling, qualities of an effective counselor.

Counseling Situations: Developmental, preventive, facilitative, and crisis.

Counseling and Psychotherapy - Skills in counseling - Establishing the relationship.

Process of Counseling.

Approaches to Counseling: Approaches; Theoretical base, thrust, goals, key concepts, techniques - Approaches like person-centered, rational-emotive, behavioural approaches, gestalt, existential approaches, Egans three stage model, eclectic model.

Indigenous Approach: Indigenous approaches of help and self-help like yoga, reflection. Act of Prayashchit.

UNIT IV

Couple and Family Counseling: Issues in such counseling, its process and stages.

Crisis Counseling

Group Counseling: Counseling for groups - Process, advantages and disadvantages of group counseling.

Practice of counseling in family counseling centres, family courts, counseling bureau - Premarital and marital counseling, vocational counseling centres, mental health centres, child guidance clinics, correctional institutions, deaddiction and rehabilitation centres, educational institutions.

REFERENCES

1. Brown, Leland 1970 Communicating Facts and Ideas in Business, New Jersey: Prentice-Hall Inc., Englewood Cliffs.
2. Chandrashekar, C. R. A Manual on Counseling for Lay- Counselors, (Ed.) 1999 Bangalore, Prasanna Counseling Centre.

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Paper code: SWBSC -1

Paper Title: GANDHIAN APPROACH TO WELFARE AND DEVELOPMENT

INTRODUCTION

The course aims at sensitizing the learner to the Gandhian approach and to utilize some of the skills in practice.

OBJECTIVES

- a. Develop an understanding of Gandhi's concept of society and his approach to social transformation.
- b. Develop knowledge of the specific programmes formulated by Gandhi for rural reconstruction and the development of the weaker sections of society, with the focus on strategies and skills.
- c. Develop the ability to identify similarities and differences between the Gandhian and professional social work approaches to social change, welfare and development.

Course Content

Unit I

Gandhian thought: Salient features of Gandhian thought; Gandhian values; Concepts and methods; Concept of a healthy society; Sarvodaya.

Unit II

Gandhian Approach: Economic and its organization: Ownership of property; Concept of trusteeship, distribution and economic equality; System of production, problems of mechanization, decentralization of production, rural- urban relationship

Unit III

Social Organisation: Marriage and family, position of women, social stratification, caste and untouchability, education and its role; Basic education.

Unit IV

Constructive programmes: Contents training of constructive workers, skills involved, nature of programmes; Bhoodan, Gramdan.

Gandhian and Vinobha's movements with special reference to Bhoodan and Gramdan

Gandhian and Professional Social Work Approach: Similarities and differences between Gandhian and professional approach to social development and welfare

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2. Bandopadhyaya, J 1969 Social and Political Thought to Gandhi, Bombay: Allied Publishers
3. Gandhi, M.K Social Service, Work and Reform; 3 Volumes, Ahmedabad: Navjivan Press
4. Ganguli, B.N 1972 Gandhi's Vission of Ideal Society, Hyderabad: Andhra Mahila Sabha
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9. Narayan, J 1965 From Socialism to Sarvodaya, Varanasi: Sarva Seva Sangh

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Even semester

Paper Code: SWSC-2

Paper title: PERSONAL AND PROFESSIONAL GROWTH.

INTRODUCTION

The course aims at enhancing personal and professional effectiveness by developing a continuous awareness and deeper insight into one's being. It encourages value clarification, upholding of professional ethics, and ability to make effective choices for integration. It provides opportunities to understand stress, stressors and methods to handle stress experienced.

OBJECTIVES

- a. Understand self as a being, as one in the process of becoming and experience self-awareness.
- b. Examine own values and attitudes and explore choices made to express self in own environment.
- c. Develop positive life skills and practice self-help methods for integration and for stress reduction.
- d. Understand and uphold professional values and ethics.

Course Content:

UNIT I

Self and Self Awareness: Understand self through a cognitive construct/paradigm (two/three models from among those available may be offered as workshops). Suggested approaches are: Rational Emotive Therapy, Gestalt Approach, Transactional Analysis, Reality Therapy, Yoga for Therapy, Meditation Techniques.

Explore self as being, and understand the process of becoming. (through observation)

Practice consciously measures to sustain and experience continuous awareness.

Observation and Reflection: Theory and techniques.

Communication Choices: Communication mode and patterns and effectiveness, Interpersonal communication, nature of choices made.

UNIT II

Emotions and their Expression: Emotions, nature of expression.

Understand own pattern of communication, choices made to express emotions, modes used, examine need for change.

Communication: Informal and knowledge and skills of rapid reading, writing, creative writing, report writing and public speaking.

UNIT III

Creativity and Self: Understand brain functions: Creativity, need and development

Life Style: Conscious life style - enhanced life skills: Communication, decision making, empathy, critical thinking, use of time and money, building and sustaining bonds-relational, collegial and personal.

Self defeating behaviour - nature and impact. Choices for change.

UNIT IV

Values, Attitude and Professional Ethics: Values and attitudes - their role in life, Value conflict - its impact, value clarification.

Integration: Through Eastern and Western approaches experience the processes of integration. Approaches recommended are: Yoga as a science, meditation (tool for meditation - own choice).

Stress / Burn out - Self help Methods: Stress, Stressors, nature and impact of stress, its expression, and burnout.

Spirituality and Growth.

REFERENCES:

1. Becavar, D. (Ed.) 1997 The Family, Spirituality and Social Work, Journal of Family Social Work, Vol.2, No.4,
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| 8. Hamilton, G. 1954 | Self-Awareness in Professional Education', Journal of Social Casework, Vol. 35, No.9, 371-379. |
| 9. Haskar, S. L. 1976 | Know Thyself, Triveni 45(2), 88. |
| 10. Paul, Brunton. 1975 | The Hidden Teaching Beyond Yoga, Triveni, 44 (3), 91. |
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| 13. Singh, N. P. 1970 | The Concept of Self in Modern Psychology, Indian Education Review, 5 (1), 84 -99. |

Paper code: SWBSC-2

Paper Title: POPULATION AND ENVIRONMENT

INTRODUCTION

The content has two aspects to it. Population dynamics and its relatedness to the environment, natural resources, utilization and their preservation.

OBJECTIVES:

- a. Understand characteristics, determinants of population growth.
- b. Examine population policy, plan and initiatives.
- c. Understand inter-relatedness of human life, living organisms and environment.
- d. Examine utilization and management of resources.
- e. Develop skills to participate in activities related to the two areas.

2. Family planning Association of India
Family planning Counseling Guide,
Population Reports Service Series J.N 35
and 36
3. Fisher, W.F 1997
Towards Sustainable Development
(Struggling over India's Narmada River),
New Delhi: Rawat Publications.
4. Gadgil, and Guha. 1997
This Fissured Land - An Ecological History
of India: Delhi: Oxford University Press.
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Hertford: IPPF
6. Krishna. M. 1995
Air Pollution and Control, Kakinada:
Kaushal and Co.
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Living in the Environment, California:
Wordsworth International Group.
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"Urbanization in India's Future",
Population and Development Review, Vol.
11(4)
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Our Common Future, Delhi: Oxford
University Press.
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Population Planning, Policy and
Programmes, New Delhi: Deep and Deep
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Population Education, New Delhi: Ashish
Publication.
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Environmental Management Handbook,
Ahmedabad: IOS Press.
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Environment Management in India, Vol. II,
New Delhi: Ashish Publishing House
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Sustainable Development (An Alternative
Paradigm), Ahmedabad: Karnavati
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Population Education, A Natural Source
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16. Sharma, P.D. 1995
Ecology and Environment, New Delhi:
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odd semester (III Semester)

Odd Semester

Paper code: SWCHC-11

Paper Title: HUMAN RESOURCE MANAGEMENT

INTRODUCTION

The main objective of this course is to prepare young graduates for management and administrative positions in various industrial, business, governmental/non-governmental organisations and service sector organisations.

OBJECTIVES

- a. Develop managerial skills in different functional areas of management with practical focus on HRM.
- b. Develop the competence to evolve the problem-solving approaches by applying conceptual and behavioural skills.
- c. Develop interpersonal skills/ competence and leadership qualities to work in a group with team building approach.
- d. Develop sound theoretical base in various concepts and theories to enable the student to develop a broad perspective of the management field.
- e. Distinguish the strategic approach to Human Resources from the traditional functional approach.
- f. Understand the relationship of HR strategy with overall corporate strategy.

Course Content

UNIT I

Human Resource Management: Concept, scope, philosophy and objectives; Evolution; Approaches, Structure and Functions; Line and staff relations of HRM; HRM Model. Hierarchy, formal and informal structure, Organization chart/reporting structure.

Human Resource Planning: Concept and objectives; Human resource inventory; Human resource planning process; job analysis; job description; job specification; job design; career planning and career paths; job rotation.

UNIT II

Talent Acquisition: Goals; policies, sources and methods. Selection: Concept, process. Talent Acquisition Tests, Theories and issues in psychological testing, Intelligence testing – theoretical background, Aptitude Testing, Personality Assessment, MBTI. Placement, Induction and socializing the new employee. Talent retention: Concept, importance and methods.

UNIT III

Compensation Management: Factors influencing compensation plans and policies; Job evaluation - Fixation of salary, components of salary. Pay for performance - Incentive Schemes, principles and types, Employee Stock Option Plan, compensation survey / review

UNIT IV

Strategic Human Resource Management (SHRM): Business strategy and organizational capability, SHRM: aligning HR with Corporate strategy, Strategic HR planning and Development, Change Management and restructuring and SHRM, Corporate Ethics, Values and SHRM, Competencies of HR professional in a SHRM scenario.

REFERENCES

1. Agarwal, R. D. (Ed.) 1973 Dynamics of Personnel Management in India, New Delhi: Tata McGraw-Hill Publishing Company.
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| 14. Rudrabasavaraj, M. N. 1986 | Cases in Human Resource Management, Bombay: Himalaya Publishing House. |
| 15. Subba Rao, P. 1996 | Essentials of Human Resource Management and Industrial Relations, Himalaya Publishing House. |

Odd semester

Code: SWCHC-12

Title: SOCIAL WORK PRACTICUM - IV

Workshops: Skills Development - help learners acquire specific skills for situations encountered during practice and acquire skills for intervention. These may be for problems/ concerns, issues or situations like work with alcoholics, HIV/AIDS affected persons, adolescents for life skills development, youth for leadership development and couples for marital relationship and enrichment work with elderly. These workshops are to enhance skills/ develop new skills for practice in specific situation, specific problems and issues.

Concurrent practice learning of two-days a week -on going learning of practice is an opportunity to develop intervention skills in reality situations. This entails learning social work practice for two, or two and a half days or its equivalent, each week of the semester. The learners may be placed in agencies or in communities to initiate and participate in direct service delivery. Practice learning is a vital component of

the educational opportunity to be provided to the learner. The teaching-learning process must be designed to help the learner to move on the mastering strategies, skills and techniques to practice social work

Odd semester

Paper code : SWCSC-3

Paper Title : SOCIAL WORK WITH TRIBAL AND RURAL COMMUNITIES.

INTRODUCTION

This course aims at introducing the learner the programmes of tribal and rural development, and the importance of social work practice with tribal and rural communities.

OBJECTIVES

- a. Develop an understanding of tribal and rural communities.
- b. Understand the characteristics and problems of tribal and rural communities.
- c. Acquire knowledge about the contribution of Governmental and Non-governmental Organisations to tribal and rural development.
- d. Develop an understanding of the functions of Panchayath Raj Institutions with particular reference to Karnataka.
- e. Gain knowledge about the application of social work in tribal and rural development programmes.

Course Content

UNIT I

Tribe in relation to caste and nation - Nature and Characteristics of Primitive Cultures- Tribes in India and their ecological distribution.

Emerging Trends in Tribal Social Institutions - Family and Kinship Systems, Jati Structure, Economic Structure, Political organisations.

Characteristics of Tribal Society - Economic, Social, Political and Cultural Problems of Tribal Life.

UNIT II

Government Programmes since Independence and their Impact on Tribal Societies - Programmes of Voluntary Agencies and their Impact on Tribal Societies.

Analysis and Assessment of Tribal Community Problems - Special Problems of the Tribals in a particular area.

Social Work Practice in Tribal Development: Community organisation as a method of intervention, Participatory Rural Appraisal (PRA), Logical Framework Approach/Analysis (LFA), techniques of intervention and its scope in tribal community development.

UNIT III

Rural Society and Poverty - Historical perspective - Dynamics in the village society - Caste/class relationships - Control and Power, Conflict and Integration. Poverty in the rural context - Its nature and manifestations. Analysis of Basic Problems - Issues faced by the rural poor such as indebtedness, Bonded labour, Low wages, Unemployment, Underemployment, and other forms of exploitations.

UNIT IV

Current Rural Development Programmes in India: Council for the Advancement of People's Action and Rural Technology (CAPART) and other Rural Development Statutory Bodies.

Panchayath Raj System in Karnataka and its role in rural and tribal development.

Role of social worker in tribal and rural development programmes. **REFERENCES:**

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20. Sodhi, J. S. 1990 Poverty Alleviation of Rural Development, New Delhi: Criterion Publications.
21. Swaminathan, M. S. 1982 Science and Integrated Rural. Development, New Delhi: Concept Publishing company.

Paper code SWCSC-3

Paper Title: ORGANIZATIONAL BEHAVIOUR AND ORGANIZATIONAL DEVELOPMENT

INTRODUCTION

The course aims to provide an understanding of human behavior at work so that the learner may acquire the skills required to analyze problems and develop a problem-solving approach.

OBJECTIVES:

- a. To impart knowledge about individual, group and organizational dynamics and their consequences,
- b. To make clear the concepts and approaches that help in developing models or systems that support human ingenuity.
- c. To acquaint the students with the knowledge of theories and practices that govern human behavior at work,
- d. To help the learner understand the value and worth of human resources in an organization.
- e. To enable the students to become aware of their communication skills and sensitize them to their potential to become successful managers.
- f. To gain self-confidence and healthy self-respect while retaining respect for other's rights.
- g. To understand the application of Transactional Analysis in several areas of employee management.

Course content

UNIT I

Conceptual Framework: Organization Behavior: Definition, concept, approaches and scope, historical background of Organization Behavior.

Introduction to Enneagram, personality types according to Enneagram. Emotional Intelligence; Attitude, Values, Personality; Job satisfaction, Employee Morale : Meaning, influences and outcomes - Measuring job satisfaction.

Assertiveness Training: Benefits of assertiveness - components of assertive behavior, measuring assertiveness, handling fear, handling anger, handling depression, developing assertive behavior skills, assertiveness on the job, assertiveness in interpersonal relations.

UNIT II

Transactional Analysis (TA), TA and self awareness, Winners and Losers, Structural analysis, Life positions, transactions, games and strokes, Life scripts, TA applications in motivation, Leadership and Teamwork, TA in counseling.

Motivation: Concept and theories, techniques of motivation, role of reinforcement and punishment, motivation and organization reward system, awards, employee empowerment and engagement.

UNIT III

Leadership: Meaning, roles, skills, and styles, leadership theories, types of leadership, powerful persuasion strategies.

Group dynamics: Concept, types of groups, dynamics of group formation, decision making in groups.

Organization Development: Concept, emerging approaches and techniques, Foundations of OD, Organizational Diagnosis, OD interventions - An overview, individual and interpersonal interventions, team/group interventions, comprehensive interventions, organizational transformation, success and failure of OD, Planned Organizational change, feedback and OD.

UNIT IV

Organizational Conflict: Concepts, causes and types, conflict-resolution strategies.

Organizational change: Concept, forces of change and resistance to change, managing organizational change and diversity, facilitating creative and divergent thinking, planned organizational change.

13. Ryan, Rosemary K C, 2008 Leadership Development- A Guide for HR and Training Professionals, New Delhi, Elsevier Publications.
14. Sadler, Philip, 2004 Leadership- Styles, Role Models, Qualities, Behaviours, Concepts, New Delhi, Kogan Page India Pvt., Ltd.
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Odd semester

Paper code: SWCSC-4

Paper Title: PREVENTIVE AND SOCIAL MEDICINE AND MEDICAL SOCIAL WORK

INTRODUCTION

This course introduces the basic health issues and the application of social work in health setting both in hospital and community.

OBJECTIVES

- a. Understand the concept and dimensions of health.
- b. Understand the issues related to the prevention, clinical features and treatment of major communicable and non-communicable diseases.
- c. Trace the historical development of medical social work in India and abroad.
- d. Understand the nature of medical social work services.
- e. Understand the tenets of National Health Policy of India and modernization of community based health care services. .
- f. Understand the health care services at different levels.

COURSE CONTENT

UNIT I

Concept of health : Physical, social, mental and spiritual dimensions of health - Positive health - Determinants of health - Health and development - Indicators of health. Concept of Prevention: Levels of prevention - Hygiene, public health, preventive medicine, community health, social medicine, community medicine. Health Care of the Community; Concept of health care - Levels and principles of health care.

UNIT II

Communicable and Non-communicable Diseases: Leprosy, Tuberculosis, Sexually Transmitted Diseases (STDs), HIV/AIDS. Cancer, Hypertension, Accidents, Diabetes, Blindness, Neurological problems, Mental illnesses.

Maternal and Child Health Services - Immunization - Integrated Child Development Services (ICDS) Scheme - School health programmes.

UNIT III

Medical Social Work: Meaning, Definition and Scope - Historical background and nature: Medical Social Work in India and Abroad - Team work and Multidisciplinary approach in health care; Organization and administration of medical social work departments in hospitals.

Patient as a person and Role of Social Worker: Understanding the patient as a person; Illness behaviour and treatment behaviour of the patient - Impact of illness on the patient and family.

Role of social worker with patients and their families - Rehabilitation.

UNIT IV

National Health Policy of India, Directorate General of Health Services, Indian Council of Medical Research (ICMR), Health as a concurrent subject.

Health System in India - at the Centre, at the State level, at the district level, and village level. Health Education and Communication.

Voluntary Health Agencies in India - International health - World Health Organisation (WHO), UNICEF, UNDP, FAO, ILO, World Bank.

Non - governmental and other Agencies - Ford Foundation, CARE, International Red Cross, Indian Red Cross.

REFERENCES:

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8. Jordan, William. 1972 The Social Worker in Family Situations, London: Routledge and Kegan Paul.
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11. Mechanic, David 1968 Medical Sociology- A Selective View, New York, Free Press.
12. Mishne, Judith (Ed.) 1980 Psychotherapy and Training in Clinical Social Work, New York: Gardner Press.
13. Nichols, P. J. R. (Ed.) 1980 Rehabilitation Medicine, London: Butterworths.
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16. Ramachandrudu, G. 1997 Health Planning in India,' New Delhi, A. P. H. Publishing Corporation.
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Paper code: SWCSC-4

Paper Title: REHABILITATION AND AFTER CARE SERVICES

INTRODUCTION

Rehabilitation of differently abled people is a noble and worthy endeavor, requiring the combined knowledge of the psycho-social theory and practical skills and techniques of social work. The current paper facilitates social work students to work with the specific group of clientele suffering from various types of disabilities and impart application of specific professional social work methods to cater to the needs of this population.

OBJECTIVES:

- a. To understand the concept of handicap, rehabilitation and the scope for practice.
- b. To identify the specific client categories requiring the rehabilitation services, problem specificity and rehabilitation service interventions.
- c. To acquaint oneself with different rehabilitation settings, different therapeutic approaches to rehabilitation process.
- d. To acquire the social work skills adapted to facilitate the process of rehabilitation, the rights and legal provisions provided for differently abled people and assimilate the knowledge of social work practice to disability specific client service.

Course Content

UNIT I

Rehabilitation: Definition and scope for social work interventions; definition of Impairment, Disability, Handicap; causes of Handicap - heredity, acquired, Major illnesses - physical, neurological and psychiatric Stress, vulnerability, coping and competence to deal with handicaps; Need for comprehensive rehabilitation - psycho-social rehabilitation

UNIT II

History, philosophy and principles of psycho-social rehabilitation; specific problem areas - physical handicap - vision, hearing, orthopedic, speech and language difficulties, mental retardation and others; neurological, psychiatric problems, disasters, alcohol and drug usage, terminal illnesses and any other.

Intervention in rehabilitation: Assessment, planning, intervention, evaluation, tools for assessment, follow-up services.

UNIT III

Rehabilitation Settings: Hospital based, day-care, night-care, quarter-way home, half-way-home, group home, hostels, long-stay homes, vocational guidance centre,

sheltered workshop, occupational therapy centre, community based rehabilitation centre, home care, inclusive education and others

Approaches: Therapeutic community, behavior modifications, transactional analysis and eclectic approach

UNIT IV

Practice of Social work methods in the process of rehabilitation: Case work, group work, community organisation, research, administration and social action.

Legal provisions for differently abled people - The Persons with Disabilities (Equal Opportunities, Protection of Rights and Full Participation) Act 1995, Rehabilitation Council of India: Formation, scope and functions, governmental policies and programmes, initiatives from the non-governmental sectors.

International trends and national initiatives in the rehabilitation scenario.

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Paper code: SWCSC-5

Paper Title: SOCIAL POLICY, PLANNING AND DEVELOPMENT

INTRODUCTION

The course introduces the learner as to how policy is a link between Constitutional Principles, Development Plans, Legislative and Executive Actions. The analysis of these processes is to enable utilization of the knowledge to improve social work practice.

Further, it provides a critical and analytical framework to understand key concepts, development processes and current issues, pertaining to different parts of the world, with specific reference to India. This course is expected to provide the social work students with a context for micro-level interventions.

OBJECTIVES

- a. Gain knowledge of policy analysis and the policy formulation process.
- b. Acquire skills in critical analysis of social policies and development plans.
- c. Develop an understanding of social policy in the perspective of national goals as stated in the Constitution, particularly with reference to Fundamental Rights and the Directive Principles of State Policy.
- d. Critically understand the concept, content and process of social development.
- e. Develop the capacity to identify linkages among social needs, problems, development issues and policies.
- f. Locate strategies and skills necessary for social development and reinforce values of social justice, gender justice and equality.

Course Content

UNIT I

Social Policy and Constitution: Concept of social policy, sectoral policies and social services - Relationship between social policy and social development-- Values underlying social policy and planning based on the Constitutional provisions (i.e. the Directive Principles of State Policy and Fundamental Rights) and the Human Rights - Different models of social policy and their applicability to the Indian situation.

UNIT II

Sectoral Social Policies in India: Evolution of social policy in India in a historical perspective-Different sectoral policies and their implementation, e.g. Policies concerning education, health, social welfare, women, children, welfare of backward classes, social security, housing, youth, population and family welfare, environment and ecology, urban and rural development, tribal development and poverty alleviation.

UNIT III

Social Planning: Concept of social planning - Scope of social planning - the popular restricted view as planning for social services and the wider view as inclusive of all sectoral planning to achieve the goals of social development - Indian planning in a historical perspective - The Constitutional position of planning in India. Niti Ayog - Coordination between Centre and State, need for decentralization - Panchayath Raj - people participation.

UNIT IV

Social Development: Concept of social development - Current debates of development - Approaches to development - Development indicators.

Social Development in India: The historical and social context of development in India - Demographic transitions - Rural development: Agrarian and land reforms; Green Revolution - Industrialization and urban development - Labour relations-Gender issues - Environmental issues (land, water, forest) - Education - Health.

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Recommended Journals / Periodicals

Alternatives; Development and Change; Economic and Political Weekly.

Paper code : SWCSC-5

Paper Title : LEGAL SYSTEM IN INDIA

INTRODUCTION

The course is to help learners understand the legal system and procedures in India. It supports understanding the processes in public interest litigation and develops skills for the same.

OBJECTIVES

- a. Acquire information on the legal rights of people.
- b. Develop an understanding of the legal system and get acquainted with the process of the legal system with emphasis on functioning in India.
- c. Understand the role of the police, prosecution, judiciary and correction. d. Gain insight into the problems faced by the people belonging to different strata of society, in interacting with this system.
- e. Develop an understanding of the processes and problems of public interest litigation and legal aid to marginalized.

Course Content

UNIT I

Social Justice: Meaning and Concept; Social legislation: Meaning, definitions and concept. Social justice as an essential basis of social legislations; Social legislations in a welfare state with special reference to India.

Rights: Concept and definitions of Rights; types of Rights; Rights of women and children; Rights of Scheduled Castes and Scheduled Tribes; Rights of accused and offender under Constitution of India, Indian Penal Code and Criminal Procedure Code.

UNIT II

Division of Law: Substantive Law and Procedural Law.

Legislations pertaining to Social Institutions: Marriage, divorce, maintenance of spouse, adoption.

Legislations for prevention of Crime and Deviance: Indian Penal Code (relevant chapters like of Offences against Public Tranquility, of Offences affecting the Public Health, Safety, Convenience, of Decency and Morals, of Offences relating to Religion, of Offences affecting the Human Body, of Offences relating to Marriage, of Cruelty by Husband or Relatives of Husband)

Legislations pertaining to women.

UNIT III

Criminal Justice System in India:

Police: Structure, powers and functions and their role in maintaining peace and order in the society.

Prosecution: Meaning, structure, its role in criminal justice, trial participation.

Judiciary: Supreme Court, High Court - Constitution of Supreme Court and High Court: Powers and functions.

Sub-ordinate Courts - District Sessions Court, Magistrate Courts, and other subordinate courts.

UNIT IV

Correction and Correctional Laws: Corrective measures as per Criminal Procedure Code, Probation of Offenders Act, Juvenile Justice (Care and Protection of Children) Act.

Legal Aid: Concept of legal-aid, history of legal-aid, persons needing legal-aid, legal-aid schemes.

Public Interest Litigation: Meaning, Concept, Process and Problems.

Right to Information Act- Provisions and implementation.

Role of Social Worker: Social Work intervention, need, methods.

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Open Elective**Paper Code: SWOE****Paper Title: GERONTOLOGICAL SOCIAL WORK****INTRODUCTION**

Changing demographic profile in India has led to rise in the number of elderly as never before. Along with the enhanced longevity, a number of issues related to care and management of elderly have come into focus. Social work as a profession concerned with providing professional service to the needy, has recognized the need to address the concerns of the senior citizens. The paper envisages training the learners in professional social work practice with the elderly.

The paper focuses on senior citizens as target client group for social work intervention; the paper deals with the issues, concerns, problems and social work methods in facilitating healthy adaptation of the client group in the current Indian context.

OBJECTIVES:

- a. To get an overview of the perspectives on aging and scope for practice.
- b. To understand the various challenges related to aging, healthy aging and problems of the elderly in difficult situations.
- c. To identify agencies working with elderly, the different care settings and issues in working with elderly in different settings. To gain an insight into process of working with elderly.
- d. To train the learners in applying specific social work intervention measures in working with senior citizens, care givers and to have an understanding of
- e. National Policy on Older Persons, and the role of International and NGOs in improving the quality of life of the elderly.

Course Content**UNIT I**

Gerontology - Definition and scope. Understanding the elderly - demographic, developmental, psychological, socio cultural, economic, and health perspectives. The issues pertaining to elderly- health, occupation, income, retirement planning, family support, gender issues, property Rights and any other

UNIT II

Developmental tasks in elderly: Issues in health care, changes in family structure, coping with aging process, challenges due to changing physiological, economic, safety, status in the family and other issues, Healthy aging, quality of life, coping with demise of the life partner, bereavement, resolving one's own death, and any other.

UNIT III

Care settings for elderly: General hospitals, geriatric wards/ hospitals, home-based care, homes for the aged, nursing homes, day-care-centers, hobby centers, and facilities for homeless elderly, elder helpline, and senior citizen forum.

Tools for assessment of the problems of elderly, intervention and follow up services and evaluation.

UNIT IV

Social work intervention measures for senior citizens through methods of social work: Case work, group work, community organisation, welfare administration, social work research, social action

Care giver issues - Needs, burden, coping and training; training for caregivers of institutions for the elderly

National Policy on Older Persons, Legal and governmental welfare benefits for senior citizens, Role of HelpAge India and other prominent Organisations working for elderly.

International scenario

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JOURNALS.

1. Indian Journal of Gerontology, C-207, Manu Marg, Tilak Nagar, Jaipur 302
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2. R & D Journal of Helpage India . C-14, Qutab Institutional Area, New Delhi,
110016.

Open Elective

Paper code SWOE

Paper Title: SOCIAL WORK PRACTICE WITH CHILDREN

INTRODUCTION

Children are the future of human society. Profession of social work has to work with children in difficult circumstances while rendering services in varied settings. There is a need for social workers specially trained in working with the children and adolescents. Such trained social workers can render valuable services to children in need of professional help.

The current paper focuses on children as a special group for focused social work intervention through facilitating acquisition of knowledge about children from different perspectives, types of settings where the children can be helped and application of social work methods to render social work intervention to children.

OBJECTIVES

- a. To understand children facing difficult circumstances and the impact of difficult circumstances on children's development.
- b. To gain an overview of agencies where children form the major client group, and appropriate evaluation of children's problems.
- c. To impart to the trainee, specific social work intervention methods in dealing with children as a client group; to understand the Rights of children in the legal, national and international context.

Course Content

UNIT I

Human reproductive system - beginning of life till beginning of adulthood. Understanding the children and adolescents from different perspectives - developmental, demographic, economic, psychological, sociological, environmental, familial, educational dimensions of child development. Issues in adolescence - self image, peer group, career choice, sexuality, education, vocation and other issues. Healthy child development, importance of supportive environment in upbringing of the children.

UNIT II

Children in difficult circumstances - developmental delay, physical and intellectual handicaps; chronic illnesses, nutritional deficiencies, accidents, poverty, child labour, abandoned and orphaned children, adoption issues, children in institutions, psychological problems in children, self harm and suicides in children, addiction related problems in children, children brought up by single parent due to death, divorce and other related issues, problems in formal schooling, children living in difficult situations - children in streets, slums, war zones, migration, children in conflict with law, truancy, drug abuse, running away from homes, neglected children, child abuse, child trafficking, child marriage and any other. Special focus on adolescent issues as applicable.

UNIT III

Children in difficulties - Helping agencies, Settings and issues - paediatric hospitals, nursing homes, child care centres, child guidance clinics, residential care services for children - residential schools, orphanages, homes for children in conflict with law, agencies dealing with differently abled children, any other. Assessment, intervention, follow up and evaluation of children and adolescents facing difficulties.

UNIT IV

Social Work Intervention Programmes - Case work, group work, community organisation methods in helping children, school mental health programmes, home visits, school visits, life skills training, family life education for adolescents, creative use of play therapy, art, dance, drama and other mediums for helping children, child

help lines, child care centres, adoption services, special rehabilitation services for rescued children and any other.

Legislations pertaining to children, legal protection, International, National and non-governmental organisations working with children, Rights of the children.

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Even semester (IV Semester)

Paper code SWDHC-13

Paper Title: EMPLOYEE RELATIONS AND LEGISLATION

INTRODUCTION

The purpose is to provide an in-depth knowledge about the relationship between employer, employee and the state, to bring out the importance of cordial employee relations for organizational productivity and gain an understanding of the mechanism of inter-personal relations, collective bargaining and productivity improvement functions in the organisation through involvement of all groups.

OBJECTIVES

- a. Develop the skills of interpersonal relationship as per organisational requirement.
- b. Understand the trends and dynamics between the partners in the organisation.
- c. Enhance the knowledge on organisational performance, role and responsibility.
- d. Develop the knowledge on various statutory / legal aspects influencing the organizations.
- e. To stimulate thinking on rationale behind the Laws and their enforcement.

Course Content

UNIT I

Employee relations, History of industrialization in India - Issues related to employees in organized and unorganized sector.

Concept, Definition, Philosophy and Principles of employee relations. Employee relations with special reference to Occupation - Safety - Health and Environment (OSHE) Education.

Analysis of the terms 'industry' and 'industrial dispute', industrial discipline - misconduct, disciplinary proceedings.

Domestic Enquiry: Contents and Process, Principles of Natural Justice, Tribunal; Discharge/Dismissal.

UNIT II

Trade Unions: Trade Unionism in India, emergence, history and growth, Trade Union as an organization - Various Trade Unions in India, Trade Union policies, Role of Trade Unions in India, Employers' Associations - Objectives, structure and activities. Contemporary issues in employee relations.

UNIT III

Employee Legislations: - The Payment of Bonus Act, 1965, Employees Provident Fund (and Misc. Provisions) Act 1952, Workmen's Compensation Act 1923, Employees State Insurance Act 1948, Payment of Gratuity Act, 1972, Child Labour (Prohibition and Regulation) Act, 1986.

Fundamentals of Labour laws, The Constitution of India: Preamble, Fundamental Rights including writs, Directive Principles of State Policy, The Factories Act 1948, The Contract Labour (Regulation and Abolition) Act 1970, The Minimum Wages Act 1948 and The Payment of Wages Act 1936; The Apprentices Act, 1961, The Maternity Benefit Act 1961.

UNIT IV

The Trade Union Act 1926, The Industrial Employment (Standing Orders) Act 1946, The Industrial Dispute Act 1947, The Employment Exchanges (Compulsory Notification of Vacancies) Act 1958. Introduction to Right to Information Act, Intellectual Property Rights, Patent Law, Copyrights, Trademark Law.

Collective Bargaining: Definitions, characteristics, critical issues in collective bargaining, theories of collective bargaining, Hick's Analysis of Wages setting under collective bargaining, conflict-choice model of negotiation, Behavioral Theory of Labor Negotiation, Collective Bargaining in India, Collective bargaining in practice, levels of bargaining, coverage and duration of agreements, administration of agreements, negotiating a contract, the negotiation process, effective negotiation, negotiation and collective bargaining, post negotiation - Administration of the agreement.

Employee relations in knowledge based industry - Concepts of self-managed teams (SMT) - Changing employee/ employer and trade union relationship. Current rules of Taxation of Salaries.

Labor Welfare Officer - Duties and functions; Social Work in Industry.

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23. Subramanian, H. N. 1967 Labour Management Relations in India, Bombay, Asia Publishing House.
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25. Tyagi, B. P. 1976 Labour Economics and Social Welfare, Meerut, Jai Prakash Nath & Co.
26. Vaid, K. N. 1970 Labour Welfare in India, New Delhi, Sri Ram Centre for Industrial Relations.
27. Yoder, D. 1972 Personnel Management an Industrial Relations, New York, Prentice-Hall India.

Paper code: SWDHC-14

Paper Title: MENTAL HEALTH AND PSYCHIATRIC SOCIAL WORK

INTRODUCTION

This course is to provide awareness about mental health and mental health problems and also application of social work in mental health settings.

OBJECTIVES

- a. Understand the concepts 'mental health' and 'mental illness'.
- b. Understand the signs and symptoms, etiology, diagnosis and treatment of mental health problems.
- c. Understand different services for the care of mentally ill.
- d. Understand historical background of psychiatric social work in India and abroad. Understand the nature of psychiatric social work services and relevance of team work.
- e. Understand the nature of collaboration with voluntary organisations for the welfare of mentally ill.
- f. Identify the issues related to psychiatric social work department in hospitals and community mental health settings.

Course Content

UNIT I

Concept of mental health and mental illness - Mental health as a part of general health - Misconceptions about mental illnesses. General approaches to the mentally ill - International Classification of Mental Disorders.

Signs, symptoms, etiology, diagnosis, prognosis and management of the following:

- Neuroses
- Psychoses
- Psycho physiologic disorders
- Personality disorders
- Psychiatric disturbances in children and adolescents
- Organic psychotic conditions
- Mental retardation.

UNIT II

Introduction to Psychiatric Social Work: Meaning and Scope - Historical background of psychiatric social work in India and abroad - Reasons for its development as a specialty. Application of social work methods and other related techniques used in the field - Multi-disciplinary approach and team work in mental health care - Problems of hospitalization - Impact of mental illness on the patient, family and community.

Practice of Social Work: Importance of home visit and visit to the place of work - Role of family in the treatment of mentally ill - Preparing the family and community for the return of the affected individual, follow-up.

UNIT III

Care of mentally ill: Day-care centre, night-care centre, half-way-home, sheltered workshop, Occupational therapy units - Role of social worker and role of voluntary organisations.

Role of voluntary organisations, governmental-agencies and paraprofessionals in the welfare of mentally ill.

Role of social worker in mental health centers, departments of psychiatry in general hospitals, child guidance clinics, community mental health units, correctional institutions, industries, and family welfare centres.

Role of social worker with head injured, paraplegics and epileptics.

Role of social worker in the management of substance abuse - Educational avenues in psychiatric social work - Research avenue in the field of mental health for social workers.

UNIT IV

Organisation of psychiatric social work department - Functions; and collaboration with other departments.

Community mental health and social work, NMHP, Innovations like Satellite clinics, district mental health programme etc.

Rehabilitation and Acts: Occupational therapy - Principles and practice - Psychosocial rehabilitation.

Mental Health Act, 1987.

The Persons with Disabilities (Equal Opportunities, Protection of Rights and Full Participation) Act, 1995.

12. Henderson, Sir David and Batchelor, I. R. C. 1962
Textbook of Psychiatry, New York
Oxford University Press.
13. Hudson, Barbara L. 1982
Social Work with Psychiatric Patients,
London: Macmillan.
14. Humble, Stephen and Unell, Judith (Ed.) 1989
Self Help in Health and Social Welfare,
London: Routledge.
15. Jones, Kathleen. 1972
A History of the Mental Health
Services, London: Routledge and
Kegan Paul.
16. Jordan, William. 1972
The Social Worker in Family
Situations, London: Routledge and
Kegan Paul.
17. Maller, Joshua-o. 1971
The Therapeutic Community with
Chronic Mental Patients, S. Karger.
18. Mishne, Judith (Ed.) 1980
Psychotherapy and Training in
Clinical Social Work, New York:
Gardner Press.
19. Page, J. D. 1983
Abnormal Psychology, New York,
McGraw-Hill.
20. Robbins, Arthur J. 1957
Mental Hospitals in India and Social
Work Service, Delhi School of Social
Work.
21. Strean, Herbert S. 1979
Psychoanalytic Theory and Social
Work Practice, New York: Free Press.
22. Stroup, H. H. 1960
Social Work - An Introduction to the
Field, (Chapter 9: Psychiatric Social
Work), New Delhi: Eurasia Publishing
House.
23. Todd, F. Joan. 1967
Social Work with the Mentally
Subnormal, New York: Routledge and
Kegan Paul.
24. Towle, Charlotte. 1941
Social Case Records from Psychiatric
Clinics with Discuss Notes, Chicago;

Illinois: University of Chicago Press.

25. Yelloly, Margaret. 1980

Social Work Theory and
Psychoanalysis, New York: Van
Nostrand Reinhold Company.

26. National Mental Health Programme for India

Code SWDHC -15

Title: MAJOR PROJECT

Students are given broad guidelines for undertaking empirical evidence-based project in the fourth semester. In case of group project work, the group will be formed by the college or the university department by adopting random method of selection. The project shall comprise of selection of the topic, methodological details, analysis, interpretation and deductions made. The department will prepare a set of guidelines for presenting the report.

Evaluation of the Project will be done along with the viva-voce examination by the viva-voce committee constituted for the assessment of social work practicum.

Even semester

Code: **SW DHC-6**

Title : **SOCIAL WORK PRACTICUM - V**

Concurrent practice learning of two-days a week - on going learning of practice is an opportunity to develop intervention skills in reality situations. This entails learning social work practice for two, or two and a half days or its equivalent, each week of the semester. The learners may be placed in agencies or in communities to initiate and participate in direct service delivery. Practice learning is a vital component of the educational opportunity to be provided to the learner. The teaching-learning process must be designed to help the learner to move on the mastering strategies, skills and techniques to practice social work.

Even semester

Code: **SWDHC-17**

Title: SOCIAL WORK PRACTICUM - VI: (BLOCK PLACEMENT)

Block Placement - enables learners to integrate learning and generate newer learning by participating in the intervention process over a period of 6 weeks continuously, in a specific agency. Usually, block field work is provided at the end of the two-year programme. There shall be a professionally qualified worker in the setting willing to plan orientation and provide consultation, when needed.

Paper code: SWDSC-6

Paper Title: HUMAN RESOURCE DEVELOPMENT AND EMPLOYEE WELLNESS

INTRODUCTION

The purpose of this course is to provide practical exposure and knowledge in behavioural science to develop skills not only to understand and analyse problems but also to develop a problem-solving approach to issues.

OBJECTIVES

- a. To develop multi facets of the personality and to build self confidence.
- b. To develop a spirit of continuous learning and innovation.
- c. To strengthen the competency base of individuals, teams and organization and also familiar with the organizational culture.
- d. Understand and further the organization culture.
- e. To appreciate the importance of bottom-line focus to the Human Resource function and trend toward HR Accountability.
- f. To understand the various approaches to and techniques of measuring HR issues.
- g. To create awareness of different types of information systems in an organization so as to enable the use of computer resources efficiently, for effective decision- making.

Course Content

UNIT I

Human Resource Development (HRD): Concept, origin and needs for HRD; Overview of HRD as a Total system; Approaches to HRD; human capital approach; social psychology approach and poverty alleviation approach; HRD and its dimensions, Competency Mapping.

UNIT II

HRD Interventions: Performance Measurement Systems - Fundamental issues. Feedback sessions. Organizational goal setting process, Key Result Area (KRA) and Key Performance Indicator (KPI), Coaching, Mentoring, career planning, career development, reward system, quality of work life. HRIS: - Computers and computer based Information Systems. Measuring HR : Changing role of HR, HR as a strategic partner, the need for measuring HR. Approaches to measuring HR: - Competitive Benchmarking, HR Accounting, HR Auditing, HR Effectiveness Index, HR Key Indicators, HR MBO (Management by Objectives).

Instructional Technology: Learning and HRD; Building Learning Organization: measuring learning - the intellectual capital, architecting a learning organization, Organizational Learning, models and curriculum; factors and principles of learning; group and individual learning; HRD trends; behavioural sciences; transactional analysis; Concepts of continuous learning, behavior modeling and self-directed learning; evaluating the HRD effort; data gathering; analysis and feedback; HRD experience in Indian organizations; future of HRD - Organization culture and development.

UNIT III

Talent Development: Concept and importance; Training Need Analysis, process of training, designing and evaluating training and development programs. Use of information technology, Types and Methods of Training: Training within industry (TWI), External; on the job and off the job; Training methods; lecture, incident process, role play, structured and unstructured discussion, in-basket exercise, simulation, vestibule, training, management games, case study, programmed instruction, team development, and sensitivity training; review of training programs.

UNIT IV

Employee Wellness: Concept, philosophy, principles and scope; Importance and relevance of wellness programs, Role of Welfare Officer as per the Factories Act 1948. Relevance - with reference to Accidents, Absenteeism, Alcoholism, Domestic Violence: Preventive and remedial measures.

Employee Counseling. Role of Counselor in Organizations. Corporate Social Responsibility (CSR): CSR as a business strategy.

Environmental management systems ISO 14001, ISO 26000: Social responsibility guidance standard, environmental impact assessment.

Development, New Delhi: Oxford and
IBH Publishing Co. Pvt. Ltd

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|---|--|
| 13. Rudrabasavaraj, M. N. 1984 | Human Factors in Administration,
Bombay: Himalaya Publishing House. |
| 14. Sahni, P. and Sharma, K. K. 1988 | Organisational Behaviour, New Delhi:
Deep and Deep Publications. |
| 15. Singh M. K. and Bhattacharya
(Eds.) 1990 | Personnel Management, New Delhi :
Discovery Publishing House. |
| 16. Vroom, V. H. and
Grant, L. 1969 | Organisational Behaviour and Human
Performance, New York. Wiley. |

Code: SWDSC-6

Title: CASE STUDIES

Every Candidate is expected to take up five cases, study them in depth and present the intervention, if any. Case refers to a unit of study – an individual, an institution, a community or an incident. The candidate has to work under the guidance of faculty member and submit the report on or before the date prescribed.

The university or the college concerned can develop guidelines for undertaking case studies. However, the students are encouraged to start his/her work on case studies from the beginning of the course.

Evaluation of the case study will be done along with the viva-voce examination by the viva-voce committee constituted for the assessment of social work practicum or similar committee may be constituted, if required.

JSS COLLEGE OF ARTS, COMMERCE & SCIENCE

(An Autonomous College of University of Mysore)

Re-accredited by NAAC with 'A' grade

OOTY ROAD, MYSORE-570 025, KARNATAKA



ESTD-1964

SYLLABUS

2017-2020

M.VOC. FOOD PROCESSING & ENGINEERING

Scheme of Instruction For M. Voc. (Food Processing & Engineering)

General Education Component

(L-Lecture; T-Tutorial; P-Practical/Practice) (1 Credit = 15 Hrs)

Semesters	Paper No.	Title	L:T:P	Theory Hours	Tutorial	Practical Hours	Total Hours	Total Credits
					Hours			
Sem I	MFA 510	Food Chemistry	2:0:1	30	0	15	45	3
	MFA 520	Food and Nutrition	2:0:1	30	0	15	45	3
SemII	MFB 510	Food Microbiology	2:0:1	30	0	15	45	3
	MFB 520	Biostatistics	2:0:1	30	0	15	45	3
SemIII	MFC 510	Information Communication Technology	2:0:1	30	0	15	45	3
	MFC 520	EDP	2:0:1	30	0	15	45	3
SemIV	MFD 510	Food Marketing	2:0:1	30	0	15	45	3
	MFD 520	Food Standards, Regulatory Affairs and IPR Issues	2:0:1	30	0	15	45	3

Sl. No.	Food Chemistry	Hrs
1.	Introduction to food chemistry, its role in processing and food formulations,	1
2.	Moisture in foods: Role and type of water in foods, Functional properties of water, role of water in food spoilage, Water activity and sorption isotherm, Molecular mobility and food stability.	2
3.	Dispersed systems of foods: Physicochemical aspects of food dispersion system: a) Sol b) gel c) foam d) emulsions.	1
4.	Carbohydrates: Functional characteristics of different carbohydrates. Maillard reaction, caramelization, methods to control non enzymatic reactions. Starch and Dietary fibres, Functional properties of polysaccharides, natural vegetable gums, carbohydrate composition of various natural foods.	5
5.	Proteins in foods: Protein content and composition in various foods- cereal grains, legumes and oilseed proteins, proteins of meat, milk, egg and fish. Functional properties of proteins in foods – water and oil binding, foaming, gelation, emulsification. Effect of processing on functional properties of proteins-heat processing alkali treatments, chilling, freezing, dehydration and radiations. Unconventional sources of proteins- SCP fish protein concentrates, leaf proteins	5
6.	Lipids in foods: Role and use of lipids /fat, occurrence, fat group classification, Physicochemical aspects of fatty acids in natural foods, hydrolysis, reversion,. Chemical aspects of lipolysis, autooxidation, antioxidants, Technology of fat and oil processing: Refining, Hydrogenations, Inter etherification, Safety use of oils and fats in food formulation.	5
7.	Vitamins and minerals, Dietary sources, requirements, Allowances, Enrichment, Restorations, Fortifications, Losses of vitamins and minerals, Optimization and retention of vitamins and minerals	2
8.	Enzymes in food industry, Carbohydrases (Amylases, cellulases, pectinases,) Proteases, Lipases and oxidases in food processing.	2
8.	Chemistry of food flavour: definitions of flavour, Flavourmatics / flavouring compounds, Sensory assessment of flavour, Technology for flavour retention.	2

9.	Food additives: Buffer systems/ salts / Acids, Chelating agents and sequestrants, Antioxidants, Antimicrobial agents, Non-nutritive and low calorie sweeteners, Stabilizer and thickeners,	2
10.	Food colours, natural and synthetic, Regulatory aspects –Natural and synthetic permitted food colours.	1
11.	Food toxicants – anti nutritional factors and their occurrence, effects and methods of elimination or inactivation- protease inhibitors, lectins, lathrogens, phytates and flatulence factors.	2
12.	Food Contaminants, Pesticidal residues – permitted limits. Toxicology and public health.	2

Sl. No.	Practical	Hrs
1.	Determination of moisture content of foods using different methods	1
2.	Determination of crude proteins by microkjeldahl method	2
3.	Determination of crude fat by soxlet method	2
4.	Determination of acid value, saponification value and iodine number of fat/ oil	2
5.	Determination of minerals and acid insoluble ash and estimation of Calcium and phosphorus	3
6.	Assay of amylases, papain and lipases	2
7.	Detection of common food adulterants	2
8.	Determination of food colors	1

CODE NO:

Semester - I

Food Chemistry

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

I. Write short notes for the following(any 5): (5x2=10)

1. -----
2. -----
3. -----
4. -----
5. -----
6. -----

PART-B

II. Answer any 4 of the following: (4x5=20)

1. -----
2. -----
3. -----
4. -----
5. -----

PART -C

III. Answer any 4 of the following: (4x10=40) IV.

1. -----
2. -----
3. -----
4. -----
5. -----

**INFORMATION AND COMMUNICATION TECHNOLOGY
PRACTICAL**

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

1. Determination of moisture content of foods using different methods
2. Determination of crude proteins by microkjeldahl method
3. Determination of crude fat by soxlet method
4. Determination of acid value, saponification value and iodine number of fat/ oil
5. Determination of minerals and acid insoluble ash and estimation of Calcium and phosphorus
6. Assay of amylases, papain and lipases
7. Detection of common food adulterants
8. Determination of food colors

Sl. no.	Food and Nutrition	Hrs
1.	Introduction to Food: Definition, classification and constituents of food : Carbohydrates, Fats , Proteins ,Fat soluble vitamins-A, D, E and K , Water soluble vitamins – Thiamin, Riboflavin, Niacin, Pyridoxine, Folate, Vitamin B12 and Vitamin C, Minerals – Calcium, Iron, Zinc, Iodine and Flourine.	3
2.	Food Design: Nutritive values of cereals, pulses, oil seeds, fruits, vegetables, fish, meat and eggs.	2
3.	Functions of food, Effect of deficiency & overconsumption of dietary sources on health, Basic Food Groups, Recommended dietary Allowance (RDA), Food guide pyramid, Dietary fibers, Functions of water in body. Balanced Diet: Concept of Balanced Diet: Definition, food groups used in planning balanced diets.	5
4.	Traditional and contemporary methods of food processing and quality evaluation of food products	3
5.	Nutrition: Basic terms used in Nutrition, relationship between food, health and nutrition, Bioavailability of nutrients. Basal Metabolic Rate (BMR). Protein quality, Dietary allowances and standards for different age groups: Adult man/woman, Preschool children, Adolescent children, pregnant woman. Geriatric nutrition, Nutrition for athletes	10
6.	Digestion and absorption of carbohydrates, proteins and fats. Factors influencing the sensory acceptability and digestion of foods	1
7.	Food Design: Nutritive values of cereals, pulses, oil seeds, fruits, vegetables, fish, meat and eggs. Nutrient composition of foods and Energy calculations	4
8	Antinutritional factors: Sources and harmful effects of anti vitamins (e.g.: avidin, dicoumarol), Natural toxicants, (e.g.: Lathyrus sativa).Food adultrants- structure and harmful effects of - Butter yellow, lead chromate and malachite green.	2

Sl. no.	Practical	Hrs
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1.	Sensory acceptability of food products: Physical Attributes (Appearance, color, texture, taste and overall acceptability). Texture measurement of food products by instrumental methods. Preparation of food labelling. Formulation for foods for target groups (weaning, pre-school children, geriatric, therapeutic foods etc.). Processing of spices for traditional products. Storage and shelf determination.	15
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MODEL QUESTION PAPER

CODE NO:

Semester - I

Food and Nutrition

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

V. Write short notes for the following(any 5):

(5x2=10)

- 7. -----
- 8. -----
- 9. -----
- 10. -----
- 11. -----
- 12. -----

PART-B

VI. Answer any 4 of the following:

(4x5=20)

- 6. -----
- 7. -----
- 8. -----
- 9. -----
- 10. -----

PART -C

VII. Answer any 4 of the following:

(4x10=40)

VIII.

- 6. -----
- 7. -----
- 8. -----
- 9. -----
- 10. -----

Food and Nutrition

PRACTICAL

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

1. Sensory acceptability of food products: Physical Attributes (Appearance, color, texture, taste and overall acceptability).
2. Texture measurement of food products by instrumental methods.
3. Preparation of food labelling.
4. Formulation for foods for target groups (weaning, pre-school children, geriatric, therapeutic foods etc.).
5. Processing of spices for traditional products.
6. Storage and shelf determination.

General Component MFB 510 Semester - II

Sl. no.	Food Microbiology	Hrs
1.	Microbiology: Introduction, historical developments in food microbiology; prokaryotes and eukaryotes; Microscope; classification & morphology of microbes; Techniques of pure culture; Bacteriology of air & water; classification of microorganisms-a brief account; sources of microorganisms in foods; microbial growth, growth curve; Thermal inactivation of microbes; Concept, determination & importance of TDT, F, Z & D values; Factors affecting heat resistance; Pasteurization and sterilization. factors affecting growth-intrinsic and extrinsic factors controlling growth of microorganisms. Microbiology of various food stuffs.-Cereals, legumes, oilseeds, fruits & vegetables, Milk and their processed products	8
2.	Disinfection & disinfectants; Energy metabolism of aerobic & anaerobic microbes	4
3.	Effect of food preservatives, heating process, irradiation, low temperature storage, chemical preservatives and high-pressure processing on the microbiology of foods; control of water activity and microbial growth	
4.	Foods microbiology and public health: food poisoning, types of food poisonings, important features etc; bacterial agents of food borne illness, food poisoning by <i>Clostridium</i> , <i>Salmonella</i> , <i>E. coli</i> , <i>Bacillus</i> , <i>Staphylococcus</i> etc.; non-bacterial agents of food borne illness: poisonous algae, and fungi-a brief account.	5
5.	Food spoilage and microbes of milk, fruits, vegetables and various plant products, spoilage of canned foods; methods of isolation and detection of microorganisms or their products in food; conventional methods; rapid methods, retention of microbes, (newer techniques)-immunological methods; fluorescent, anti body, radioimmunoassay, principles of ELISA, PCR (Polymerase chain reactions)	8
6.	Indicators microorganisms; microbiological criteria of foods and their significance; the HACCP system and food safety used in controlling microbiological hazards, applications of hurdle	4
	technology for controlling microbial growth.	

7.	Microbiology of Fermented foods:, Cereals, Vinegar, Oriental foods, Alcoholic beverages. Food poisoning and microbial toxins, standards for different foods. Food borne intoxicants and myco toxins	4
8.	Microbiology of milk & milk products like cheese, butter, ice cream, and milk powder etc	4
9.	Microbiology of fruits & vegetable and products like jam, jelly, sauce, juice; etc	4
10.	Microbiology of cereal & cereal products like bread, biscuits, confectionary etc	4

Sl. no.	Practical	Hrs
1.	Equipments used in microbiology laboratory, study of microscope, observation of microbial slides, preparation and sterilization of media, methods of sterilization, staining techniques, effects of environmental factors on growth of microorganisms, Assignment -microbiological analysis of market samples- milk & milk products, fresh & processed fruits and vegetables, Cereal & bakery products	45

CODE NO:

Semester - II

Food Microbiology

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

IX. Write short notes for the following(any 5):

(5x2=10)

- 13. -----
- 14. -----
- 15. -----
- 16. -----
- 17. -----
- 18. -----

PART-B

X. Answer any 4 of the following:

(4x5=20)

- 11. -----
- 12. -----
- 13. -----
- 14. -----
- 15. -----

PART -C

XI. Answer any 4 of the following:

(4x10=40) XII.

- 11. -----
- 12. -----
- 13. -----
- 14. -----
- 15. -----

Food Microbiology**PRACTICAL****SCHEME OF EXAMINATION**

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

1. Equipments used in microbiology laboratory, study of microscope, observation of microbial slides, preparation and sterilization of media, methods of sterilization, staining techniques, effects of environmental factors on growth of microorganisms,
2. Assignment-microbiological analysis of market samples- milk & milk products, fresh & processed fruits and vegetables, Cereal & bakery products

General Component Semester - IIMFB 520

Sl. no.	Biostatistics	Hrs
1.	Statistical concepts: Data structure, sampling methods, collection, classification and tabulation of data, graphical and diagrammatic representation, histogram, frequency polygon, frequency curve, bar graph, pie chart etc.	4
2.	Measure of Central Frequency: Mean, median, mode.	2
3.	Measure of dispersion of data: Range, semi-interquartile range, mean deviation, standard deviation, standard error, coefficient of variation, confidence limits.	5
4.	Types of distribution of data: Normal, Binomial, Poisson.	7
5.	Z-test, t-test, ANOVA, multiple comparisons, LSD and DMRT, Chi-square test.	4

6.	Regression estimate, correlation coefficient.	4
7.	Experimental designs, data transformation.	4

Sl. no.	Practical	Hrs
1.	Analytical Problems / calculations	15

MODEL QUESTION PAPER

CODE NO:

Semester - II

Biostatistics

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

XIII. Write short notes for the following(any 5):

(5x2=10)

- 19. -----
- 20. -----
- 21. -----
- 22. -----
- 23. -----
- 24. -----

PART-B

XIV. Answer any 4 of the following:

(4x5=20)

- 16. -----
- 17. -----
- 18. -----
- 19. -----
- 20. -----

PART -C

XV. Answer any 4 of the following:

(4x10=40)

XVI.

- 16. -----
- 17. -----
- 18. -----
- 19. -----
- 20. -----

II SEMESTER PRACTICAL EXAMINATION

Biostatistics

PRACTICAL

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

1. Analytical Problems / calculations

General Component Semester – IIIMFC 510

Sl. no.	Information Communication Technology	Hrs
1	The humanitarian supply chain – Definition, system of organizations, people, technology, activities, information and resources involved in moving a product or service from supplier to customer	5
2	Technology framework – Front-end services, Middleware services and Infrastructure services: Supporting the food assistance supply chain; Mapping technologies; Web portals	5
3	Mobile technologies - Combining hand-held and wireless communications technologies	5
4	Beneficiary identification - Challenges in food assistance to ensure that assistance goes to the right beneficiaries	5
5	ICT in emergencies – Requirement inputs of Food assistance interventions during emergencies	5
6	Linking the humanitarian supply chain - Ways in which WFP uses technology and technological techniques to fulfill its role as the provider of food assistance in development and emergencies	5

Sl. no.	Practical	Hrs
1.	Identification of software related to Food Processing and Engineering	3
2.	Practicing the use of software	6
3.	Requirement development for Food Processing Software	6

MODEL QUESTION PAPER

CODE NO:

Semester – III

Information Communication Technology

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

XVII. Write short notes for the following(any 5):

(5x2=10)

- 25. -----
- 26. -----
- 27. -----
- 28. -----
- 29. -----
- 30. -----

PART-B

XVIII. Answer any 4 of the following:

(4x5=20)

- 21. -----
- 22. -----
- 23. -----
- 24. -----
- 25. -----

PART –C

XIX. Answer any 4 of the following:

(4x10=40)

XX.

- 21. -----
- 22. -----
- 23. -----
- 24. -----
- 25. -----

II SEMESTER PRACTICAL EXAMINATION

Information Communication Technology

PRACTICAL

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

- I. Identification of software related to Food Processing and Engineering.
- II. Practicing the use of software.
- III. Requirement development for Food Processing Software

General Component Semester - III MFC 520

Sl. no.	EDP	Hrs
1	Need for EDP, Entrepreneurship and enterprise – Concept, definition and characteristics with special reference to Food and allied areas of the Indian scenario.	2
2	Entrepreneurial development – objectives, evaluation and the existing experience, soft skill for entrepreneurship	2
3	Functions and classification of Entrepreneur and supporting institution and schemes by the National and International agencies Factors influencing entrepreneurship groups	3
4	Gender equality in Entrepreneurship, Women Entrepreneurship, selection of enterprising men and women. The short comings for women entrepreneurship and remedial majors	3
5	Identifying products, services and enterprise establishmentSWAT Analysis	1
6	Institution working for promotion of entrepreneurship in the country such as NSIC, NIMSME, NIESBUD, KVIC/KVIB etc. And also National Financial Institutions such as banks, corporations and Agro industry projects	5
7	Identification of potential areas of food processing and regions for SMES, appraisal implementation, monitoring and evaluation, Globalization and the emerging business / entrepreneurial environment, business plan format for tiny and small enterprises, planning small scale units	5
8	Training the identified entrepreneurs, Investment analysis, Risk analysis and probable approach for successful entrepreneurship, cost benefit analysis, assessing financial viability of the project, market survey tools and market management	3
9	Network establishment for food chain, corporate and social responsibility	2
10	Communication skills, listing and noting down, project preparation and presentation skills, field dairy maintenance, upgradation of skills and knowledge on the contemporary food processing technology, public private partners	4

Sl. no.	Practical	Hrs
1	Different methods to identify potential entrepreneurs – men and women from both rural and urban areas	2
2	Selection of enterprise best suited for men and women, identification of business opportunities and financial processing sector	1
3	Selection and identification of enterprise based on local/regional – financial support, resources	3
4	Training on communication skills for development of enterprise by the entrepreneur	2
5	Market survey and identification of potential food processing entrepreneurships	3
6	Preparation of project reports, business plan and feasibility report	2
7	Presentation of the project proposed and documentation	2
1	Visit to Industries / Research Institutions	4
2	Project	8
3	Internship	12

MODEL QUESTION PAPER

CODE NO:

Semester – III

EDP

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

XXI. Write short notes for the following(any 5):

(5x2=10)

- 31. -----
- 32. -----
- 33. -----
- 34. -----
- 35. -----
- 36. -----

PART-B

XXII. Answer any 4 of the following:

(4x5=20)

- 26. -----
- 27. -----
- 28. -----
- 29. -----
- 30.-----

PART –C

XXIII. Answer any 4 of the following:

(4x10=40)

XXIV.

- 26. -----
- 27. -----
- 28. -----
- 29. -----
- 30. -----

III SEMESTER PRACTICAL EXAMINATION

EDP

PRACTICAL

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

- I. Different methods to identify potential entrepreneurs – men and women from both rural and urban areas
- II. Selection of enterprise best suited for men and women, identification of business opportunities and financial processing sector
- III. Selection and identification of enterprise based on local/regional – financial support, resources
- IV. Training on communication skills for development of enterprise by the entrepreneur
- V. Market survey and identification of potential food processing entrepreneurships
- VI. Preparation of project reports, business plan and feasibility report
- VII. Presentation of the project proposed and documentation
- VIII. Visit to Industries / Research Institutions
- IX. Project
- X. Internship

General Component Semester – IV MFD 510

Sl. No.	Food Marketing	Hrs
1.	Food Marketing: Definition, meaning, characteristics of rural and urban marketing	3
2.	Opportunities and challenges marketing food products by small scale entrepreneurs	2
3.	Rural marketing segmentation, rural consumer behavior, changing trends in rural consumer selection and decision, marketing process and influential factors, marketing needs for export products.	5
4.	Urban marketing segmentation, urban consumer behavior, changing trends in urban consumer selection and decision, marketing process and influential factors	5
5.	Product design, innovativeness presentation, services, prices, method of pricing, network for sourcing raw materials and distribution of products in both rural and urban area.	4
6.	Designing advertisement, campaign, sales promotion, choice of media, techniques, personal selling and publicity	4
7.	Online Marketing: Target population, product packing, distribution through courier and other mode of transportation.	3
8.	Food packaging, labelling for consumer acceptability	2
9.	Relevant of marketing information system, market research in accessing consumer behavior	2

Sl. No.	Practical	Hrs
1.	Regulatory aspects and food hygiene and safety for packing and marketing of food products. Costing of food products.	15
	Visit to marketing federation, cooperatives APMCs and other marketing organization and institution for familiarization of marketing strategy, handling and transportation of fresh package products, perishable goods and self stable and transport table. Financial management, securing financial support, advancing the products for marketing, bulk and retail sales, recalling the products recovery of advances.	

CODE NO:

Semester – IV

Food Marketing

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

XXV. Write short notes for the following(any 5):

(5x2=10)

- 37. -----
- 38. -----
- 39. -----
- 40. -----
- 41. -----
- 42. -----

PART-B

XXVI. Answer any 4 of the following:

(4x5=20)

- 31. -----
- 32. -----
- 33. -----
- 34. -----
- 35. -----

PART –C

XXVII. Answer any 4 of the following:

(4x10=40)

- 31. -----
- 32. -----
- 33. -----
- 34. -----
- 35. -----

IV SEMESTER PRACTICAL EXAMINATION

PRACTICAL

Food Marketing

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

1.Regulatory aspects and food hygiene and safety for packing and marketing of food products. Costing of food products

2.Visit to marketing federation, cooperatives APMCs and other marketing organization and institution for familiarization of marketing strategy, handling and transportation of fresh package products, perishable goods and self stable and transport table.

3.Financial management, securing financial support, advancing the products for marketing, bulk and retail sales, recalling the products recovery of advances.

General Component Semester – IV MFD 520

Sl. No.	Food Standards, Regulatory Affairs and IPR Issues	Hrs
1.	Introduction to concepts of food quality, food safety, food quality assurance and food quality management; objectives, importance and functions of quality control, Current challenges to food safety	3
2.	Principles of food quality assurance, total quality management (TQM) –good manufacturing/management practices, good hygienic practices, good lab practices, general awareness and role of management practices in quality control	3
3.	Microbial quality control: determination of microorganisms in foods by cultural, microscopic, physical, chemical methods. Statistical quality control in food industry Food adulteration, nature of adulterants, methods of evaluation of food adulterants and toxic constituents	3
4.	Food safety management, applications of HACCP in food safety, concept of food trace ability for food safety, Food safety and Standards Act 2006: salient provision and prospects	3
5.	Role of national and international regulatory agencies, Bureau of Indian Standards (BIS), AGMARK, Food Safety and Standards Authority of India (FSSAI)	3
6.	Introduction to WTO agreements: SPS and TBT agreements, Codex Alimentarius Commission, International organization for standards (ISO) and its standards for food quality and safety (ISO 9000 series, ISO 22000, ISO 15161, ISO 14000)	5
7.	Food safety in USA, USFDA, Legislation in Europe: Directives of the official journal of the EU, council regulations, food legislation in UK. Regulating methods for food analysis, case studies. Enforcers of Food Laws Approval Process for Food Additives, Nutritional Labeling	5
8.	Concept of property, rights, duties and their correlation; History and evaluation of IPR; Copyrights and related rights. Distinction among Various forms of IPR. Patent rights/protection and procedure; Infringement or violation; Remedies against infringement; Indian Patent Act 1970 and TRIPS; Geographical indication and Industrial design	5
Sl. No.	Practical	Hrs
1.	Study of food regulations in various countries ; study of nutritional labeling of packaged food items by visiting food market, Visit the websites of FSSAI, BIS, AGMARK, ISO, Codex Alimentarius Commission, USFDA Study of patent law in India and the procedure for grant of patent in India	15

MODEL QUESTION PAPER

CODE NO:

Semester – IV

Food Standards, Regulatory Affairs and IPR Issues

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

XXVIII. Write short notes for the following(any 5):

(5x2=10)

- 43. -----
- 44. -----
- 45. -----
- 46. -----
- 47. -----
- 48. -----

PART-B

XXIX. Answer any 4 of the following:

(4x5=20)

- 36. -----
- 37. -----
- 38. -----
- 39. -----
- 40.-----

PART –C

XXX. Answer any 4 of the following:

(4x10=40)

- 36. -----
- 37. -----
- 38. -----
- 39. -----
- 40. -----

IV SEMESTER PRACTICAL EXAMINATION

PRACTICAL

Food Standards, Regulatory Affairs and IPR Issues

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

1. Study of food regulations in various countries ;
- 2.study of nutritional labeling of packaged food items by visiting food market, Visit the websites of FSSAI, BIS, AGMARK, ISO, Codex Alimentarius Commission , USFDA
- 3.Study of patent law in India and the procedure for grant of patent in India

M.Voc (Food Processing and Engineering) Syllabus		
1st -M.Voc		
NSQF Level: 8 – Semester I & II		
Sub Sector: Fruits & Vegetables		
Job Role: Head of Production		
UNIT I		
Post Harvest Management of Fruits		
S.No	THEORY	Hrs
1.	General Introduction of fruits-citrus, tropical and subtropical, pome, stone, soft, and berry fruits, melons and watermelons	1
2.	Importance and scope of post harvest management of fruits, Morphology, structure and composition of fruits	2
3.	Maturity Indices and standards for standards for selected fruits, methods of maturity determination	2
4.	Post-harvest physiological and biochemical changes in fruits; ripening of climacteric and non-climacteric fruits	2
5.	Harvesting and handling of important fruits. Harvesting tools; field heat removal/precooling of fruits. Sorting and grading at farm and cluster level; factors affecting post harvest losses	2
6.	Nature of post harvest deterioration; physiological change- physical damage; chemical injury-pathological decay; identification of diseases and disorders in fruit-nutritional disorders, respiratory disorders, temperature disorders and miscellaneous disorders. Classification of diseases and diseases organisms, types of diseases and agents of diseases in fruits.	4
7.	Pre-cooling of fruits and cold storage, zero energy cool chamber	3
8.	Shelf life enhancement- permitted chemicals for ripening, wax coating	3
9.	Storage practices: Refrigerated storage, modified atmospheric storage-novel MAP gases and their role, novel MAP applications, Applying high oxygen MAP; MAP of minimally processed fruits; controlled atmosphere storage/ultra low oxygen storage of fruits, recent advances in CAP and MAP	9
TOTAL		28
S.No	PRACTICALS	Hrs
1.	Familiarization of various fruits available in India and categorization of fruits used for pulping	3
2.	Studies on morphological features of some of the fruits	3
3.	Studies on maturity indices; Studies on harvesting of fruits	3
4.	Studies on permitted chemicals for ripening and enhancing the shelf life of fruits	3
5.	Studies on regulations of ripening of banana and mango	3
6.	Studies on physiological disorders like chilling injury of certain fruits	3
7.	Studies on pre cooling and storage of fruits and vegetables	3
8.	Demonstration on wax coating on apples, citrus and Mango	3
9.	Studies on various storage systems and structures;	3
10.	Studies on pre packaging of whole and cut vegetables	3
11.	MAP of minimally processed fruits & vegetables	3
12.	Visit to commercial packaging houses for mango, banana, pomegranate, grapes	3
13.	Visit to Controlled Atmospheric packaging centres	4
14.	Visit to commercial storage structures for onion and potato	4
15.	Visit to multi chamber cold storages for fruits and vegetables	4
16.	Visit to Fruit Orchards -Observations on Pruning, orchard Hygiene, Irrigation, Manuring, Insect Pests, Pathological Spoilages, Pre-harvest spray schedules to	4

	control pathological spoilages and insect infestation	
17.	Visit to Fruit Orchards - Studies on Causes for pre and post harvest losses. Spoilage factors, post harvest field operations including methods to reduce the post harvest losses	4
	TOTAL	56
UNIT II		
Technology for processing of Fruit Pulp		
S.No	THEORY	Hrs
1.	Process of receiving, ripening, checking raw material quality, sorting, washing, cutting/slicing, deseeding/destining, pulping, precooking/pasteurization, sterilizing, aseptic packaging or canning, retort pouching, sampling for quality analysis and storing	2
2.	Machineries and tools used for the fruit pulping process such as fruit washer, peeler, slicer, fruit pulper, steam jacketed kettles, packaging machines etc	2
3.	Quality assessment of packaging materials	2
4.	Enzymes in quality and processing of tropical and sub tropical fruits	3
5.	Non thermal processing methods-ultra violet light, high pressure processing, ultrasound, ozone application, irradiation, pulsed electric field	7
6.	Introduction, canning machineries, various steps involved in canning of fruit pulp, syrup preparation, pretreatment for canning operation	7
7.	Canning of various fruits, process flow diagram for canning, filling, exhausting, sealing and processing operations	5
	TOTAL	28
S.No	PRACTICALS	Hrs
1.	Canning of mango pulp	5
2.	Canning of tomato pulp	5
3.	Preservation of tomato pulp by chemical preservation method	5
4.	Preservation of banana pulp by freezing method	5
5.	Canning of mango slices in syrup	5
6.	Canning of pineapple slices in syrup	5
7.	Canning of banana slices in syrup	5
8.	Visit to fruit processing units and collection of data on wastes and by products	5
9.	Visit to Aseptic packing units for fruit pulps & concentrates	8
10.	Visit to the pilot plants of CFTRI & DFRL Mysore	8
	TOTAL	56
UNIT III		
Food Quality and Food Microbiology		
S.No	THEORY	Hrs
1.	Introduction – definition, historical development and significance of food microbiology; Microscope; Classification & morphology of microbes; Techniques of pure culture; Bacteriology of air & water; Anti-microbial agents – physical & chemical – mechanism & action	3
2.	Sources of Contamination: Air, Water, Soil, Sewage, Post processing Contamination. Intrinsic & extrinsic factors influencing the growth of Microorganisms in foods	3
3.	Disinfection & disinfectants; Energy metabolism of aerobic & anaerobic microbes; Thermal inactivation of microbes; Concept, determination & importance of TDT, F, Z & D values; Factors affecting heat resistance; Pasteurization and sterilization	4
4.	Microbiology of Fruits and vegetables and their products like jam, jelly, sauce, juice/pulp	3
5.	Food Quality aspects of Fruits & vegetables; Introduction, Quality principles, Quality enhancement model. Application of quality enhancement model	3
6.	Food Waste Treatment : Liquid waste, Solid waste vessel containers & wrapping	3

	waste, Hazardous waste .Quality and Safety of Frozen Foods: Fruits, Vegetable	
7.	Measuring and Controlling Devices: Role of transducers measurements in food processing; Humidity, Turbidity and Color, Food & Process temperature controller and indicators. Statistical Quality Control for food Industry : Food Quality System, Fundamentals, Process control implementing quality control program, six sigma, RSM	4
8.	Food additives – preservatives, antioxidants, sequestrates, surface active agents, stabilizers and thickeners, bleaching and maturing agents, starch modifies, buffers, acids, alkalis, food colors, artificial sweeteners, nutritional additives, flavoring agents.	5
	TOTAL	28
	PRACTICALS	3
1.	Determination of firmness of fruits	3
2.	Determination of moisture content	3
3.	Titrate acidity estimation	3
4.	Estimation of SO ₂ in food sample	3
5.	Estimation of sodium benzoate in food sample	3
6.	Estimation of polyphenol and polyphenol oxidase	3
7.	Estimation of Reducing sugar, Non-reducing and total sugars	3
8.	Determination of organic acid content	3
9.	Ascorbic acid estimation	3
10.	Determination of pH in food products	3
11.	Determination of total Ash	3
12.	Determination of total soluble solids	3
13.	Estimation of ash content	3
14.	Estimation of crude fibre	3
15.	Estimation of pectin	3
16.	Flow process chart of food plant Waste utilization processes, various treatment for waste disposal analysis of cleaners & sanitizers, CIP Cleaning	11
	TOTAL	56
UNIT IV		
Food Safety, Hygiene and Sanitation for Processing of Fruit Pulp		
S.No	THEORY	Hrs
1.	Food safety, hygiene and sanitation for processing of fruit pulp: food safety standards and regulations for fruit pulp, definition of hygiene, hygiene practices and its importance at every stage of fruit pulp processing at industrial level; personal hygiene requirements; physical, chemical and biological hazards and methods for prevention of various hazards; CIP and COP methods and procedures, GHP, GMP and HACCP; waste management-pre and post production.	14
2.	Microbiological aspect of Food; types of food microbes, causes of food spoilage, types of food spoilage/deterioration, criteria to check the food spoilage, need for food preservation, different types of food preservation methods, method of assessing the quality of products based on physical parameters	14
	TOTAL	28
S.No	PRACTICALS	Hrs
1.	Clean and maintenance of work area using appropriate sanitizers, ensure the work area safe and hygienic for fruit processing, disposal of waste material as per SOPs and industrial requirement	8
2.	Check the working and performance of machineries and tools for fruit pulp process, clean the machineries and tools used with recommended sanitizer, to place the necessary tools required for the process, to attend minor repair, faults of all machineries if required.	8
3.	Disassembling and assembling of machineries used in fruit pulp industry (Fruit mill, crusher etc)	8
4.	Demonstration of CIP and COP methods of cleaning the machines with approved	8

	sanitizers	
5.	Visit to industry to learn about GHP, GMP, HACCP	12
6.	Visit to industry to learn about waste management pre and post production	12
	TOTAL	56
S.No	UNIT V	Hrs
Plant Design, Plant Economics and Plant Management		
1.	Food Industry management- location of plant land and building requirements, plant capacity, plant and machinery requirement, building and plant layout, utilities, byproducts, waste, energy and safety audit, manpower requirements	5
2.	Introduction to economics: Meaning, scope, and contribution to business decisions. Analysis of Demand: Law of demand, Utility function, Rate of commodity substitution, Maximization of utility, Demand functions, Indifference curve analysis, Substitution and income effects. Market demand and demand elasticities: concept of market demand, price and income elasticities of demand, importance of elasticity. Demand forecasting: causes and techniques of demand forecasting	6
3.	Analysis of supply and market equilibrium: Law of supply, price elasticity of supply, equilibrium of demand and supply. Theory of the Firm: Production function, returns to scale, Optimizing behavior, Input demands, Cost functions, Profit maximization, economics & diseconomies of scale, break even analysis. Market structures perfect competition: Profit maximization and equilibrium of firm and industry, Short run and long run supply curves; Price and output determination, practical applications	6
4.	Plant maintenance program; Role of maintenance staff and plant operators, Preventive maintenance; Guidelines for good maintenance & safety precautions; Lubrication & lubricants; Work place improvement through '5S'. Hygiene and sanitation requirement in food processing and fermentation industries; CIP methods, sanitizing & disinfestation, pest control in food processing; storage and service areas	6
5.	Supply chain management for fruits	5
	TOTAL	28
PRACTICALS		
1.	Visit to industry to learn the management system	16
2.	Visit to Fruit & Vegetable Processing Industries. Preparation of a Business Plan for setting up fruit & vegetable processing unit	40
	TOTAL	56
VI	Hands on Training in Fruit Pulp Processing Industry and submission of report	120
	TOTAL	540

Model Curriculum

Plant Manager

SECTOR: FOOD PROCESSING

**SUB-SECTOR: FRUIT & VEGETABLE, FOOD GRAIN MILLING
(INCLUDING OILSEEDS), DAIRY
PRODUCTS, MEAT & POULTRY, FISH &
SEAFOOD, BREAD & BAKERY,
ALCOHOLIC BEVERAGES, AERATED
WATER/ SOFT DRINKS, SOYA FOOD,
PACKAGED FOOD**

OCCUPATION:

PROCESSING REF

ID: FIC/Q9004,

V1.0

NSQF LEVEL: 9



Certificate

CURRICULUM COMPLIANCE TO QUALIFICATION PACK – NATIONAL OCCUPATIONAL STANDARDS

is hereby issued by the

FOOD INDUSTRY CAPACITY AND SKILL INITIATIVE (FICSI)

for the

MODEL CURRICULUM

Complying to National Occupational Standards of
Job Role/Qualification Pack: **'Plant Manager'**
QP No. **'FIC/Q0004, Version 1.0, NSQF Level 9'**

Date of issuance: **March 30, 2023**

Valid up to: **March 30, 2024**

* Valid up to the next review date of the Qualification Pack

Authorized Signatory
(Food Industry Capacity and Skill Initiative)

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Plant Manager

CURRICULUM / SYLLABUS

This program is aimed at training candidates for the job of a “Plant Manager”, in the “Food Processing” Sector/Industry and aims at building the following key competencies amongst the learner

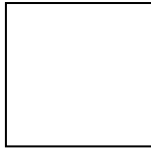
Program Name	Plant Manager		
Qualification Pack Name & Reference ID. ID	FIC/Q9004, v1.0		
Version No.	1.0	Version Update Date	30/03/2016
Pre-requisites to Training	Preferably Class 12 and 6-7 years' experience in a food processing unit		
Training Outcomes	After completing this programme, participants will be able to: Daily management of food processing unit Coordination of food processing unit operations including production planning, managing human resources, supply chain, production operation, maintenance, quality assurance, storage and distribution of finished products.		

This course encompasses 3 out of 3 National Occupational Standards (NOS) of “Plant Manager” Qualification Pack issued by “Food Industry Capacity and Skill Initiative”.

Sr. No.	Module	Key Learning Outcomes	Equipment Required
1	Introduction to the training program Theory Duration (hh:mm) 00:30 Practical Duration (hh:mm) 00:00 Corresponding NOS Code Bridge Module	Introduce each other and build rapport with fellow participants and the trainer.	White board/Chart papers, marker
2	Overview of the “Plant Manager” Role Theory Duration (hh:mm) 01:00 Practical Duration (hh:mm) 00:00 Corresponding NOS Code	Understanding the roles and responsibilities of plant manager Awareness of the nature and availability of job opportunities	Laptop/computer white board, marker, projector, chart papers
3	Introduction to the Food Processing Industry Theory Duration (hh:mm) 01:30 Practical Duration (hh:mm) 00:00 Corresponding NOS Code	Define food processing List the various sub sectors of food processing industry	Laptop, white/black board, marker, chart papers, projector, Trainer’s guide, Student manual
4	Introduction to food processing process Theory Duration (hh:mm) 02:00 Practical Duration (hh:mm)	List the common machineries used in food processing Explain the process of testing food for accepted quality standards Demonstrate the test for checking the quality of food Describe the procedure for processing various food	Laptop, white board, marker, chart papers, projector, trainer’s guide and student handbook

Sr. No.	Module	Key Learning Outcomes	Equipment Required
	04:00 Corresponding NOS Code	Identify different equipment used in food industry	
5	Organizational standards and norms Theory Duration (hh:mm) 04:00 Practical Duration (hh:mm) 02:00 Corresponding NOS Code	State the roles and responsibilities of a plant manager State how to conduct yourself at the workplace State the personal hygiene and sanitation guidelines State the food safety hygiene standards to follow in a work environment	Laptop, white board, marker, chart papers, projector, trainer's guide and student handbook, protective gloves, head caps, aprons, safety goggles, safety boots, mouth masks, sanitizer, safety manual
6	Lead operations of a food processing unit Theory Duration (hh:mm) 15:00 Practical Duration (hh:mm) 11:40 Corresponding NOS Code FIC/N9017	<ul style="list-style-type: none"> Develop operational plans for the operation of food processing unit that is consistent with the objectives and goals of organisation, and to produce quantity and quality products Develop operational plan that is flexible and complements supply chain, inventory, human resource, production, maintenance, quality and logistics management of production unit Develop operational plan to improve output in all areas of functions with the objective to reduce overall cost, and to produce quantity and quality products Develop operational plan considering national and international regulatory requirements, health and safety, food safety and hygiene requirements on process and product(s), and to maintain safe and environmental compliant workplace Develop objectives and set demanding but achievable targets for operation function managers, and assign clear responsibilities with expected targets/performance Provide direction and professional expertise to all function managers to achieve organisation goals Monitor and control the operational plan to achieve its overall objectives 	Laptop, white board, marker, chart papers, projector, trainer's guide and student handbook

Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<ul style="list-style-type: none"> • Evaluate the implemented operational plan periodically, analyze performance data, identify areas for improvement and recommend changes • Monitor performance of managers and employees to ensure that departmental and individual objectives are achieved within scheduled timelines and budget • Design new work processes, procedures, systems, structures and roles for any changes implemented in the organisation to achieve organizational goal and regulatory requirements • Review and ensure implemented changes are effective and meet the requirements of the organization • Maintain professionalism, tact, diplomacy, sensitivity, diversity and equality, and lead food processing unit to achieve organisation objectives and goals • Ensure that work arrangements, resources and business processes respond to different needs, abilities and values • Develop and implement new business strategies for improving processes and procedures to improve performance • Develop a leadership style and apply them appropriately for managers to follow the lead willingly to achieve organisation targets and goals • Communicate clearly the organisation vision, values and goals to employees, make managers understand and commit their expertise to achieve organisation goals • Lead managers of all operation functions, link operational plans and drive managers towards achieving organisation vision, objectives and goals • Lead managers through difficulties, challenges and conflicts • Conduct meetings with managers regularly and effectively, encourage them to share their views, provide guidance and support to overcome process issues and lead to achieve organisation goal 	



Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<ul style="list-style-type: none"> • Encourage managers to take lead in their own areas of expertise, take own decisions in their area of function, and provide recognitions when they are successful • Lead the managers and organisation successfully through difficulties and challenges • Design processes with achievable targets and realistic timeline, proper resource allocation, with defined process responsibilities to manage food processing operation based on organizational goals • Develop processes that are effective and sustainable, implement and ensure it is followed, review its effectiveness and make necessary changes if required • Develop process measures that are affordable, and provide enough information and required training for managers and employees to manage the process • Review and understand resource requirements for process and allocate necessary resources to all functional areas • Develop systems to link all function processes, and encourage function heads and employees to interact across the organisation to form a complete system • Establish effective methods to review the quality of work and product, and improve the process • Focus attention on issues that are critical to achieve results, provide solutions and guidance to overcome the issues that affect the process • Identify issues and trends and recognize their impact upon current and future work, work out solutions and implementation plan to overcome and utilize latest trends to achieve long term goals of the organisation • Develop policies and procedures for any change in organisation goal, organisation structure PC31. set responsibilities for managers, set and prioritize objectives for change, clearly communicate change and make the managers understand their responsibilities and commitment 	



Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<ul style="list-style-type: none"> • Implement change, identify and deal with obstacles to change, and support managers and employees through the change process • Brief managers on their responsibilities and make them understand their role, objectives for their area and the overall organisation, and expected performance • Monitor progress and performance quality of the managers on regular basis against the level of expected performance and provide prompt and constructive feedback PC35. support managers in identifying and dealing with problems and unforeseen events • Identify gaps and performance issues, discuss the causes and recommend solutions to improve performance of managers and their team • Monitor performance, analyze employee strength and weakness, and make changes in their tasks/responsibilities • Review performance and update work plans in their area, monitor and conduct review meetings on regular basis, recognize successful completion of work or work activities by function manager(s) and their teams • Motivate managers to complete expected target and any additional work allocated and provide additional support and resources to complete work 	
7	<p>Ensure proper production and operation management</p> <p>Theory Duration (hh:mm) 08:00</p> <p>Practical Duration (hh:mm) 12:00</p> <p>Corresponding NOS Code FIC/N9018</p>	<p>Update self with an understanding of the goals of the organisation and forecast/requirements of the sales and marketing manager, with the knowledge of production method and process, plant capacity, resource availability, plan products and quantity to be produced</p> <p>Monitor and regulate supply chain management which includes sourcing and procurement, conversion of raw materials to finished products, all logistics activities, coordination and collaboration with suppliers,</p>	Laptop, white board, marker, chart papers, projector, trainer's guide and student handbook ,



Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<p>intermediaries, third party service providers, and customers, to integrate supply and demand management within and across companies</p> <p>Monitor and regulate inventory process to meet the production requirement of the organisation, review current procurement procedures, analyze benefits and risks that may impact the procurement of supplies, implement plans and methods to improve and provide solutions to resolve any immediate problems</p> <p>Evaluate current storage methods and identify ways of improving the storage of supplies to provide better fit with supply chain strategy</p> <p>During production process, coordinate production activities with procurement, maintenance, and quality control function to obtain optimum production and efficient utilization of human resources, machines and equipment</p> <p>Make adjustments/revise/reschedule production schedules and priorities in case of breakdown down of equipment/issues with physical or human resource/ urgent orders/unforeseen issues or any operational problems</p> <p>Direct production activities and establish production priorities to produce quantity and quality products within the operation budget</p> <p>Review and analyze human resource, production, quality control, maintenance, and operational reports to identify reason for nonconformance/ non-compliance to organisation and regulatory standards for product and process, develop and implement operating methods and procedures to eliminate problems and improve product and process quality</p> <p>Monitor storage and distribution of products to and from the plant/processing unit warehouse, ensure storage and distribution norms and procedures like palletizing, stacking height, labeling, fefo etc are followed</p>	



Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<p>Establish systems to collect and assess information on performance of all functions, analyze data and evaluate performance of departments and organisation, through knowledge and understanding identify reasons for problems and low performance</p> <p>Establish and implement methods and procedures for improvement, ensure implemented methods deliver expected result, and identify opportunities to improve organization performance</p> <p>Read financial responsibilities, compile available financial information, evaluate the cost, benefits and risks of the current budget, and estimate financial requirements for operation of food processing unit</p> <p>Consult with department managers the objectives and associated plans, discuss and identify priorities and develop a realistic master budget for food processing operation, communicate the final proposed budget with all managers</p> <p>Submit the proposed master budget with clear proposals to the management for approval, assist them to evaluate the budget, negotiate with clarity and strong reasoning and get the budget approved</p> <p>Evaluate, analyze and allocate budget to departments of food processing operation, allocate budget to each department managers with expectations and targets, provide required ongoing support and resources</p> <p>Establish systems to monitor and evaluate performance against delegated budgets and the master budget and put contingency plans in place</p> <p>Identify reason for significant variances between budget and actual expenditure, discuss with managers, provide solutions and ensure immediate corrective action is taken</p> <p>In case of unforeseen situation/emergency/shortage,</p>	



Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<p>identify any additional financial needs, prepare provisional budget, negotiate and get it approved by the management, delegate provisional budget to respective managers, monitor and control expenditure</p> <p>Encourage managers to identify ways of reducing expenditure, analyze and pursue potential ideas, implement those in all areas of function</p> <p>Review the financial performance of managers regularly, and identify improvement opportunities and ensure it is implemented, provide information to the management on the financial performance of the operation management</p> <p>Determine human resource requirement including contingencies to achieve organisation goal, organize interview, hiring and training of new employees through human resources manager</p> <p>Ensure that all employees receive appropriate training on job duties, corporate policies and applicable regulations</p> <p>Oversee and direct the activities of subordinate managers, provide coaching and mentoring, and conduct evaluations of all</p> <p>Discuss with managers of operation function and identify resource requirement for their area, analyze, estimate and approve resources, monitor effective use of those resources</p> <p>Ensure compliance of all employees with organization policy, procedures and applicable regulations</p> <p>Conduct meetings to address grievances, to resolve or effect settlements within the scope of authority, and refer unresolved grievances for management-union negotiations</p> <p>Take personnel actions, such as promotions, transfers, discharges or disciplinary measures, within the scope of authority</p> <p>Update self with knowledge of quality management system, legal and regulatory requirements, environmental issues related to the</p>	



Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<p>organisation, process and products produced</p> <p>Ensure system, plan and resources are in place to assure food products produced in the organisation meet the organisation standards, national and international regulations</p> <p>Implement procedure, standards and specifications to meet quality goals of the organisation, co-ordinate departments and provide support to implement food safety system like HACCP in the organisation</p> <p>Evaluate records of quality of product and process to assess the effectiveness of quality system followed in the organisation, review and revise the quality system through quality assurance manager and implement changes</p> <p>Organize training for employees to update on latest developments/systems/ tools and techniques in quality management system and evaluate their competency to fulfill organisation goals</p> <p>Encourage employees of all functions to take personal responsibility for achieving quality standards of product and process and address or report/address any non-conformance</p> <p>Monitor process and product quality against target and plan, identify and assess risks of shortfalls in the quality of processes and products/services and take immediate corrective action to address risks</p> <p>Direct and coordinate implementation of quality system such as ISO, HACCP, etc. in the organisation through quality manager</p> <p>Ensure managers responsible for organizational processes understand the requirements of quality system, establish their roles in implementation of quality system in their functional areas, enhance their confidence and commitment to quality by providing continuous support</p> <p>Encourage and support department heads and employees for quality</p>	



Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<p>audit process to obtain accreditation, certifications to a standard or a mark of quality, monitor quality audit process, review results and take immediate corrective action through concerned managers</p> <p>Identify the environmental impact related to the resources, process and products produced in the organisation such as air/water/noise pollution, effluent treatment, waste disposal etc, identify risks to the environment, consult with experts and identify opportunities to improve environmental performance</p> <p>Set and implement policies and procedures through managers, monitor to ensure its efficiency and effectiveness and make changes as required to meet the regulatory requirements</p>	
8	<p>Manage new projects and ensure compliance to regulatory requirements</p> <p>Theory Duration (hh:mm) 07:00</p> <p>Practical Duration (hh:mm) 09:00</p> <p>Corresponding NOS Code FIC/N9019</p>	<p>Implement new project/business plans of the organisation for introducing new products or for improving processes, procedures and performance</p> <p>Map or perform comparative study of the project with the current project/product to understand the ways proposed project fits with the overall vision, objectives and plans of the organization</p> <p>Read the key objectives and scope of the proposed project, prepare resource requirement for implementation of new project, negotiate with clarity and strong reasoning and get approval from superiors/management</p> <p>Consult with experts and managers and prepare realistic and thorough plan to implement the project successfully, prepare project report considering all possibilities</p> <p>Submit the project report to the superiors/management, discuss plan, consider suggestions and recommendations and make necessary changes where necessary, take approval of final plan</p> <p>brief project team managers on the project plan and their roles and responsibilities, start implementation of project and provide ongoing</p>	Laptop, white board, marker, chart papers, projector, trainer's guide and student handbook, logbooks, internal audit register, food safety manual, quality policy etc.



Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<p>support, encouragement and information for successful completion</p> <p>Monitor, control and review project plan during each stage of implementation</p> <p>Provide sufficient resources to deal with contingencies and to manage any potential risks</p> <p>Inform the management/superiors of the developments in the project on regular basis, discuss progress and problems, take approval for any changes in project plan</p> <p>Complete project within agreed level of resources, meeting all legal and regulatory requirements, share the success with the project team members, recognize and reward their contribution</p> <p>Update self with understanding of national and international food safety regulations and standards related to the food processing units, process and products produced in the organisation</p> <p>Ensure effective policies and procedures are in place in the organization to meet to legal and regulatory requirements</p> <p>Ensure regulatory standards set by the organisation for products are stringent in context of the national and international legal requirements</p> <p>Ensure managers of all functional area have a clear understanding of the policies and procedures on food regulatory standards</p> <p>Organize training for all employees on policies and procedures on food regulatory standards and the importance of following regulations</p> <p>Monitor and ensure relevant legal and regulatory requirements pertaining to food processing units and products produced in the organisation are followed and met</p> <p>Identify reasons for non-compliance, review and revise the policies and procedures in consultation with quality and regulatory affairs manager to correct and overcome failures, provide support to all managers to implement corrective actions for the organisation and</p>	



Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<p>products to comply with regulatory standards</p> <p>Update self with understanding of health and safety requirements, and food safety, hygiene and sanitation requirements for the organisation and products produced</p> <p>Ensure that the organisation has written policy and procedures on health and safety, food safety, hygiene and sanitation, and those are clearly communicated to all employees of the organisation, and are put into practice and followed</p> <p>Implement a system for identifying hazards and assessing risk in food processing and products and set procedures to control and prevent them</p> <p>Implement system for GMP, HACCP, FIFO/FEFO, product recall, etc., organize training to the employees on health and safety, food safety, hygiene and sanitation for effective implementation of the systems, allocate required resources for implementation, and ensure those are followed by all employees</p> <p>Ensure systems are in place for effective monitoring, measuring and reporting on the performance of health and safety system</p> <p>Evaluate the existing systems and procedures, consult with managers and experts and identify methods to reduce risks/improve control measure</p> <p>Ensure health and safety policies are practiced across the organisation, effectively monitored, reviewed and revised at regular intervals to meet the changes in national and international regulations</p>	
9	<p>Professional and Core Skills</p> <p>Theory Duration (hh:mm) 03:00</p> <p>Practical Duration (hh:mm) 05:00</p>	<p>Undertake a self-assessment test</p> <p>Identify personal strengths and weaknesses</p> <p>Plan and schedule the work order and manage time effectively to complete the tasks assigned</p> <p>Prevent potential problems from occurring</p>	<p>Laptop, white/black board, marker, chart papers, projector, Trainer's guide, Student manual</p>



Sr. No.	Module	Key Learning Outcomes	Equipment Required
	Corresponding NOS Code	Resolve issues and problems using acquired knowledge and realize the importance of decision making Identify potential problems and make sound and timely decision Improve your reading skills State the importance of listening	
10	IT Skills Theory Duration (hh:mm) 05:00 Practical Duration (hh:mm) 07:00 Corresponding NOS Code	Identify parts of the computer Use the computer keyboard effectively to type Use computer applications effectively to record day-to-day activities Use the word processor effectively Use the spreadsheet application effectively Use the computer to document day-to-day activities	Laptop, white/black board, marker, chart papers, projector, Trainer's guide, Student manual
11	Field Visits Theory Duration (hh:mm) 04:00 Practical Duration (hh:mm) 30:00 Corresponding NOS Code	Observe the factory location, layout and safety aspects of food processing Observe the storage facilities for raw materials and finished products Observe the various machineries used in process Observe the various machineries used in process Observe the cleaning methods and processes followed to maintain the process machineries and tools Observe the raw materials used and their storage procedures Observe the packaging and storage processes of raw material and finished product Observe the post-production cleaning and maintenance process followed in the industry	All the tools and equipment listed above must be available at the site of field visit
12	Revision Theory Duration (hh:mm) 02:00 Practical Duration (hh:mm) 02:00 Corresponding NOS Code	Revised the knowledge gained so far	All the tools and equipment listed above must be available at the time of revision
13	Evaluation	Assess the knowledge and skills acquired by the participants	All the tools and equipment listed above



Sr. No.	Module	Key Learning Outcomes	Equipment Required
	Theory Duration (hh:mm) 08:00 Practical Duration (hh:mm) 20:00 Corresponding NOS Code		must be available for evaluation
14	On-the-job Training Theory Duration (hh:mm) 30:00 Practical Duration (hh:mm) 65:00 Corresponding NOS Code	Apply the skills and knowledge acquired in the training program in the field	All the tools and equipment listed above must be available on the site at the time of OJT
	Total Duration 240:00 Theory Duration 88:00 Practical Duration 152:00	Unique Equipment Required: Laptop, white board, marker, chart papers, projector, trainer's guide and student handbook, protective gloves, head caps, aprons, safety goggles, safety boots, mouth masks, sanitizer, safety manual	

Grand Total Course Duration: **240Hours, 0 Minutes**

(This syllabus/ curriculum has been approved by SSC: Food Industry Capacity and Skill Initiative)



Trainer Prerequisites for Job role: “Plant Manager” mapped to Qualification Pack: “FIC/Q9004, v1.0”

Sr. No.	Area	Details
1	Description	To deliver accredited training service, mapping to the curriculum detailed above, in accordance with the Qualification Pack “FIC/Q9004”, Version 1.0
2	Personal Attributes	An aptitude for conducting training, and pre/ post work to ensure competent, employable candidates at the end of the training, and pre/post work to ensure competent, employable candidates at the end of the training. Strong communication skills, ability to work as part of a team; a passion for quality and for developing others; well-organized and focused, eager to learn and keep oneself updated with the latest in the mentioned fields.
3	Minimum Educational Qualifications	M.Sc/M.Tech/ME in Food Technology or Food Engineering with 7-8 years of hands on experience in a food industry B.Sc (home Sc) /B.Tech/BE in Food Technology or Food Engineering with 9-10 years of hands on experience in a food industry
4a	Domain Certification	Certified for Job Role: “Plant Manager” mapped to QP: “FIC/Q9004, v1.0”. Minimum accepted score is 80%
4b	Platform Certification	Recommended that the Trainer is certified for the Job Role: “Trainer”, mapped to the Qualification Pack: “SSC/Q1402”. Minimum accepted SCORE IS 80 % as per FICSI guidelines.
5	Experience	M.Sc/M.Tech/ME in Food Technology or Food Engineering with 7-8 years of hands on experience in a food industry B.Sc (home Sc) /B.Tech/BE in Food Technology or Food Engineering with 9-10 years of hands on experience in a food industry

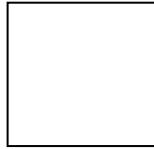


Annexure: Assessment Criteria

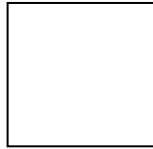
Assessment Criteria	
Job Role	Plant Manager
Qualification Pack	FIC/Q9004, v1.0
Sector Skill Council	Food Processing

Sr. No.	Guidelines for Assessment
1	Criteria for assessment for each Qualification Pack will be created by the Sector Skill Council. Each Performance Criteria (PC) will be assigned marks proportional to its importance in NOS. SSC will also lay down proportion of marks for Theory and Skills Practical for each PC.
2	The assessment for the theory part will be based on knowledge bank of questions created by the SSC.
3	Individual assessment agencies will create unique question papers for theory part for each candidate at each examination/training centre(as per assessment criteria below)
4	Individual assessment agencies will create unique evaluations for skill practical for every student at each examination/training center based on this criteria
5	To pass the Qualification Pack, every trainee should score a minimum of 70% (overall) in every QP
6	The marks are allocated PC wise; however, every NOS will carry a weight age in the total marks allocated to the specific QP

Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
1. FIC/N9017: Lead Operations of a food processing unit	PC1. Develop operational plans for the operation of food processing unit that is consistent with the objectives and goals of organisation, and to produce quantity and quality products	100	3	1	2
	PC2. Develop operational plan that is flexible and complements supply chain, inventory, human resource, production, maintenance, quality and logistics management of production unit		2.5	1	1.5
	PC3. Develop operational plan to improve output in all areas of functions with the objective to reduce overall cost, and to produce quantity and quality products		2.5	0.5	2
	PC4. Develop operational plan considering national and international regulatory requirements, health and safety, food safety and hygiene requirements on process and product(s), and to maintain safe and environmental compliant workplace		2.5	1	1.5



Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
	PC5. Develop objectives and set demanding but achievable targets for operation function managers, and assign clear responsibilities with expected targets/performance		3	1	2
	PC6. Provide direction and professional expertise to all function managers to achieve organisation goal		2.5	1	1.5
	PC7. Monitor and control the operational plan to achieve its overall objectives		2.5	0.5	2
	PC8. Evaluate the implemented operational plan periodically, analyze performance data, identify areas for improvement and recommend changes		2.5	1	1.5
	PC9. Monitor performance of managers and employees to ensure that departmental and individual objectives are achieved within scheduled timelines and budget		3	1	2
	PC10. Design new work processes, procedures, systems, structures and roles for any changes implemented in the organisation to achieve organizational goal and regulatory requirements		2.5	1	1.5
	PC11. Review and ensure implemented changes are effective and meets the requirements of the organisation		2.5	1	1.5
	PC12. Maintain professionalism, tact, diplomacy, sensitivity, diversity and equality, and lead food processing unit to achieve organisation objectives and goals		2.5	1	1.5
	PC13. Ensure that work arrangements, resources and business processes respond to different needs, abilities and values		2.5	0.5	2
	PC14. Develop and implement new business strategies for improving processes and procedures to improve performance		2.5	1	1.5
	PC15. Develop a leadership style and apply them appropriately for managers to follow the lead willingly to achieve organisation targets and goals		2.5	1	1.5
	PC16. Communicate clearly the organisation vision, values and goals to employees, make managers understand and commit their expertise to achieve organisation goals		2.5	0.5	2



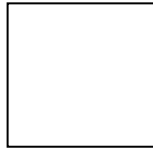
Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
	PC17. Lead managers of all operation functions, link operational plans and drive managers towards achieving organisation vision, objectives and goals		3	1	2
	PC18. Lead managers through difficulties, challenges and conflicts		2.5	1	1.5
	PC19. Conduct meetings with managers regularly and effectively, encourage them to share their views, provide guidance and support to overcome process issues and lead to achieve organisation goal		2.5	0.5	2
	PC20. Encourage managers to take lead in their own areas of expertise, take own decisions in their area of function, and provide recognitions when they are successful		2.5	1	1.5
	PC21. Lead the managers and organisation successfully through difficulties and challenges		3	1	2
	PC22. Design processes with achievable targets and realistic timeline, proper resource allocation, with defined process responsibilities to manage food processing operation based on organizational goals		2.5	1	1.5
	PC23. Develop processes that are effective and sustainable, implement and ensure it is followed, review its effectiveness and make necessary changes if required		2.5	1	1.5
	PC24. Develop process measures that are affordable, and provide enough information and required training for managers and employees to manage the process		2.5	0.5	2
	PC25. Review and understand resource requirements for process and allocate necessary resources to all functional areas		2.5	1	1.5
	PC26. Develop systems to link all function processes, and encourage function heads and employees to interact across the organisation to form a complete system		2.5	0.5	2
	PC27. Establish effective methods to review the quality of work and product, and improve the process		2.5	0.5	2



Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
	PC28. Focus attention on issues that are critical to achieve results, provide solutions and guidance to overcome the issues that affect the process		2.5	1	1.5
	PC29. Identify issues and trends and recognize their impact upon current and future work, work out solutions and implementation plan to overcome and utilize latest trends to achieve long term goals of the organisation		3	1	2
	PC30. Develop policies and procedures for any change in organisation goal, organisation structure		2.5	1	1.5
	PC31. Set responsibilities for managers, set and prioritize objectives for change, clearly communicate change and make the managers understand their responsibilities and commitment		2.5	1	1.5
	PC32. Implement change, identify and deal with obstacles to change, and support managers and employees through the change process		2.5	1	1.5
	PC33. Brief managers on their responsibilities and make them understand their role, objectives for their area and the overall organisation, and expected performance		2	1	1
	PC34. Monitor progress and performance quality of the managers on regular basis against the level of expected performance and provide prompt and constructive feedback		3	1	2
	PC35. Support managers in identifying and dealing with problems and unforeseen events		2.5	1	1.5
	PC36. Identify gaps and performance issues, discuss the causes and recommend solutions to improve performance of managers and their team		3	1	2
	PC37. Monitor performance, analyze employee strength and weakness, and make changes in their tasks/responsibilities		2.5	1	1.5
	PC38. Review performance and update work plans in their area, monitor and conduct review meetings on regular basis, recognize successful completion of work or work activities by function manager(s) and their teams		2.5	1	1.5



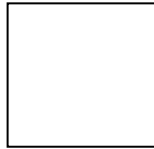
Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
	PC39. Motivate managers to complete expected target and any additional work allocated and provide additional support and resources to complete work		2	1	1
			100	35	65
2. FIC/N9018: Ensure proper production and proper management	PC1. Update self with an understanding of the goals of the organisation and forecast/requirements of the sales & marketing manager, with the knowledge on production method and process, plant capacity, resource availability, plan products and quantity to be produced	100	2	1	1
	PC2. Monitor and regulate supply chain management which include sourcing and procurement, conversion of raw materials to finished products, all logistics activities, coordination and collaboration with suppliers, intermediaries, third-party service providers, and customers, to integrates supply and demand management within and across companies		3	0.5	2.5
	PC3. Monitor and regulate inventory process to meet the production requirement of the organisation, review current procurement procedures, analyze benefits and risks that may impact the procurement of supplies, implement plans and methods to improve, provide solutions to resolve any immediate problems		3	1	2
	PC4. Evaluate current storage methods, identifying ways of improving the storage of supplies to provide better fit with supply chain strategy		3	1	2
	PC5. During production process, coordinate production activities with procurement, maintenance, and quality control function to obtain optimum production and efficient utilization of human resources, machines and equipment		1.5	0.5	1
	PC6. Make adjustments/revise/reschedule production schedules and priorities in case of breakdown down of equipment/issues with physical or human resource/ urgent		1.5	0.5	1



Assessable Outcome	Assessment Criteria	Total Mark (600)	Marks Allocation		Skills Practical
			Out Of	Theory	
	orders/unforeseen issues or any operational problems				
	PC7. Direct production activities and establish production priorities to produce quantity and quality products within the operation budget		3	1	2
	PC8. Review and analyze human resource, production, quality control, maintenance, and operational reports to identify reason for non-conformance/ noncompliance to organisation and regulatory standards for product and process, develop and implement operating methods and procedures to eliminate problems and improve product and process quality		3	1	2
	PC9. Monitor storage and distribution of products to and from the plant/processing unit warehouse, ensure storage and distribution norms and procedures like palletizing, stacking height, labeling, FEFO etc are followed		3	0.5	2.5
	PC10. Establish systems to collect and assess information on performance of all functions, analyze data and evaluate performance of departments and organisation, through knowledge and understanding identify reasons for problems and low performance		1.5	1	0.5
	PC11. Establish and implement methods and procedures for improvement, ensure implemented methods deliver expected result, and identify opportunities to improve organization performance		1.5	1	0.5
	PC12. Read financial responsibilities, compile available financial information, evaluate the cost, benefits and risks of the current budget, and estimate financial requirements for operation of food processing unit		3	1.5	1.5
	PC13. Consult with department managers the objectives and associated plans, discuss and identify priorities and develop a realistic master budget for food processing operation, communicate the final proposed budget with all managers		3	1	2



Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
	PC14. Submit the proposed master budget with clear proposals to the management for approval, assist them to evaluate the budget, negotiate with clarity and strong reasoning and get the budget approved		2.5	1	1.5
	PC15. Evaluate, analyze and allocate budget to departments of food processing operation, allocate budget to each department managers with expectations and targets, provide required ongoing support and resources		3	1	2
	PC16. Establish systems to monitor and evaluate performance against delegated budgets and the master budget and put contingency plans in place		2.5	1	1.5
	PC17. Identify reason for significant variances between budget and actual expenditure, discuss with managers, provide solutions and ensure immediate corrective action is taken		2.5	1	1.5
	PC18. In case of unforeseen situation/emergency/shortage, identify any additional financial needs, prepare provisional budget, negotiate and get it approved by the management, delegate provisional budget to respective managers, monitor and control expenditure		3	1	2
	PC19. Encourage managers to identify ways of reducing expenditure, analyze and pursue potential ideas, implement those in all areas of function		2.5	1	1.5
	PC20. Review the financial performance of managers regularly, and identify improvement opportunities and ensure it is implemented, provide information to the management on the financial performance of the operation management		2.5	1	1.5
	PC21. Determine human resource requirement including contingencies to achieve organisation goal, organize interview, hiring and training of new employees through human resources manager		3	1	2
	PC22. Ensure that all employees receives appropriate training on job duties,		2.5	1	1.5



Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
	corporate policies and applicable regulations				
	PC23 Oversee and direct the activities of subordinate managers, provide coaching and mentoring, and conduct evaluations of all		2.5	1	1.5
	PC24 Discuss with managers of operation function and identify resource requirement for their area, analyze, estimate and approve resources, monitor effective use of those resources		3	1	2
	PC25 Ensure compliance of all employees with organization policy, procedures and applicable regulations		2.5	1	1.5
	PC26 Conduct meetings to address grievances, to resolve or effect settlements within the scope of authority, and refer unresolved grievances for management-union negotiations		2.5	0.5	2
	PC27 Take personnel actions, such as promotions, transfers, discharges or disciplinary measures, within the scope of authority		3	1	2
	PC28 Update self with knowledge of quality management system, legal and regulatory requirements, environmental issues related to the organisation, process and products produced		2.5	0.5	2
	PC29 Ensure system, plan and resources are in place to assure food products produced in the organisation meet the organisation standards, national and international regulations		3	1	2
	PC30 Implement procedure, standards and specifications to meet quality goals of the organisation, coordinate departments and provide support to implement food safety system like HACCP in the organisation		3	1	2
	PC31 Evaluate records on quality of product and process to assess the effectiveness of quality system followed in the organisation, review and revise the quality system through quality assurance manager and implement changes		2.5	1	1.5
	PC32 Organize training for employees to update on latest		2.5	0.5	2



Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
	developments/systems/ tools and techniques in quality management system and evaluate their competency to fulfill organisation goals				
	PC33 Encourage employees of all functions to take personal responsibility for achieving quality standards of product and process and to address or report/address any non-conformance		2.5	1	1.5
	PC34 Monitor process and product quality against target and plan, identify and assess risks of shortfalls in the quality of processes and products/services and take immediate corrective action to address risks		3	1	2
	PC35 Direct and coordinate implementation of quality system like ISO, HACCP etc in the organisation through quality manager		3	1	2
	PC36 Ensure managers responsible for organizational processes understand the requirements of quality system, establish their roles in implementation of quality system in their functional areas, enhance their confidence and commitment to quality by providing continuous support		3	1	2
	PC37 Encourage and support department heads and employees for quality audit process to obtaining accreditation, certifications to a standard or a mark of quality, monitor quality audit process, review results and take immediate corrective action through concerned managers		2.5	1	1.5
	PC38 Identify the environmental impact related to the resources, process and products produced in the organisation like air/water/noise pollution, effluent treatment, waste disposal etc, identify risks to the environment, consult with experts and identify opportunities to improve environmental performance		1.5	0.5	1
	PC39 Set and implement policies and procedures through managers, monitor to ensure its efficiency and effectiveness and make changes as required to meet the regulatory requirements		1.5	0.5	1
			100	35	65



Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
3. FIC/N9019: Manage new projects and implement health and safety system in food processing unit	PC1. Implement new project/business plans of the organisation for introducing new products or for improving processes, procedures and performance	100	3	1	2
	PC2. Map or perform comparative study of the project with the current project/product to understand the ways proposed project fits with the overall vision, objectives and plans of the organisation		4	1	3
	PC3. Read the key objectives and scope of the proposed project, prepare resource requirement for implementation of new project, negotiate with clarity and strong reasoning and get approved from superiors/management		5	2	3
	PC4. Consult with experts and managers and prepare realistic and thorough plan to implement the project successfully, prepare project report considering all possibilities		4	1.5	2.5
	PC5. Submit the project report to the superiors/management, discuss plan, consider suggestions and recommendations and make necessary changes where necessary, take approval of final plan		4	1.5	2.5
	PC6. Brief project team managers on the project plan and their roles responsibilities, start implementation of project and provide ongoing support, encouragement and information for successful completion		5	2	3
	PC7. Monitor, control and review project plan during each stage of implementation		4	1.5	2.5
	PC8. Provide sufficient resources to deal with contingencies and to manage any potential risks		4	1.5	2.5
	PC9. Inform the management/superiors of the developments in the project on regular basis, discuss progress and problems, take approval for any changes in project plan		4	1.5	2.5
	PC10. Complete project within agreed level of resources meeting all legal and regulatory requirements, share the success with the project team		4	1.5	2.5



Assessable Outcome	Assessment Criteria	Total Mark (600)	Marks Allocation	
			Out Of	Skills Practical
	members, recognize and reward their contribution			
	PC11. Read national and international food safety regulations and standards related to the food processing units, process and products produced in the organisation	3	1.5	2.5
	PC12. Ensure effective policies and procedures are in place in the organization to meet to legal and regulatory requirements	5	2	3
	PC13. Ensure regulatory standards set by the organisation for products are stringent than the national and international legal requirements	4	1.5	2.5
	PC14. Ensure managers of all functional area have a clear understanding of the policies and procedures on food regulatory standards	4	1.5	2.5
	PC15. Organize training for all employees on policies and procedures on food regulatory standards and the importance of following regulations	4	1	3
	PC16. Monitor and ensure relevant legal and regulatory requirements pertaining to food processing units and products produced in the organisation are followed and met	5	2	3
	PC17. Identify reasons for noncompliance, review and revise the policies and procedures in consultation with quality and regulatory affairs manager to correct and overcome failures, provide support to all managers to implement corrective actions for the organisation and products to comply with regulatory standards	5	2	3
	PC18. Read the health and safety requirements, and food safety, hygiene and sanitation requirements for the organization and products produced	3	1	2
	PC19. Ensure that the organisation has written policy and procedures on health and safety, food safety, hygiene and sanitation, and those are clearly communicated to all employees of the organisation, and are put into practice and being followed	4	1	3



Assessable Outcome	Assessment Criteria	Total Mark (600)	Out Of	Marks Allocation	
				Theory	Skills Practical
	PC20. Implement system for identifying hazards and assessing risk in food processing and products, set procedures to control and prevent them		4	1	3
	PC21. implement system for GMP, HACCP, FIFO/FEFO, product recall etc, organize training to the employees on health and safety, food safety, hygiene and sanitation for effective implementation of the systems, allocate required resources for implementation, and ensure those are followed by all employees		5	2	3
	PC22. Ensure systems are in place for effective monitoring, measuring and reporting on the performance of health and safety system		4	1	3
	PC23. Evaluate the existing systems and procedures, consult with managers and experts and identify method to reduce risks/improve control measure		4	1	3
	PC24. Ensure health and safety policies are practiced across the organisation, effectively monitored, reviewed and revised at regular intervals to meet the changes in national and international regulations		4	1	3
	Total		100	35	65
	Grand Total	300	300	200	100
	Percentage Weightage		100	60%	40%
	Minimum Pass% to qualify (aggregate):			70%	

JSS MAHAVIDYAPEETHA



JSS College of Arts, Commerce & Science (Autonomous)
Ooty Road, Mysuru-25

PG Department of Physics

*(Autonomous under University of Mysore, Re-accredited by NAAC with 'A' Grade
Recognised by UGC as "College with Potential for Excellence")*

M.Sc. Physics

Course Structure and Syllabus

Under

Choice Based Credit Scheme (CBCS)

&

Continuous Assessment Grading Pattern (CAGP)

2019-20 (With PO.PSO and Cos)



JSS MAHAVIDYAPEETHA

JSS College of Arts, Commerce & Science (Autonomous), Ooty Road, Mysuru-25

PG Department of Physics

Details of Courses offered and associated credits

Paper Code	Paper	HC/SC/EL/OE	Credits			
			L	T	P	Total
I Semester						
PHA 120	Classical Mechanics	HC 1	3	-	-	03
PHA 130	Mathematical Methods of Physics 1	HC 2	3	-	-	03
PHA 140	Mathematical Methods of Physics 2	HC 3	3	-	-	03
PHA 150	Classical Electrodynamics & Plasma Physics	HC 4	3	-	-	03
PHA 160	Computer Lab CL-A	HC 5	-	-	2	02
PHA 220	Electronics Lab	SC 1	-	-	4	04
						18
II Semester						
PHB 070	Continuum Mechanics and Relativity	HC 6	3	-	-	03
PHB 080	Thermal Physics	HC 7	3	-	-	03
PHB 090	Quantum Mechanics 1	HC 8	3	-	-	03
PHB 100	Spectroscopy and Fourier Optics	HC 9	3	-	-	03
PHB 110	Computer Lab CL-B	HC 10	-	-	2	02
PHB 210	Optics Lab	SC 2	-	-	4	04
						18
III Semester						
PHC 050	Quantum Mechanics 2	HC 11	3	-	-	03
PHC 060	Condensed Matter Physics	HC 12	3	-	-	03
PHC 070	Nuclear and Particle Physics	HC 13	3	-	-	03
PHC 080/090	Condensed Matter Physics Lab / Nuclear and Particle Physics Lab	HC 14	-	-	4	04
Students are permitted to choose any one of the following (special paper) and corresponding practical coupled to the special paper						
PHC 240	Solid State Physics 1	SC 3	3	-	-	03
PHC 250	Solid State Physics Lab 1	SC 4	-	-	2	02
PHC 260	Nuclear Physics 1	SC 3	3	-	-	03
PHC 270	Nuclear Physics Lab 1	SC 4	-	-	2	02
PHY 306	Theoretical Physics 1	SC 3	3	-	-	03
PHY 315	Theoretical Physics Lab 1	SC 4	-	-	2	02
Students from other departments can register for any one of the following						
PHY-321/322	Modern Physics/Energy Science	OE	3	1	-	04
						22

IV Semester						
PHD 090/080	Nuclear and Particle Physics Lab/ Condensed Matter Physics Lab	HC 15	-	-	4	04
A student has to register for one particular discipline in confirmation with the corresponding SC (special paper) opted in III semester						
PHD 250	Solid State Physics 2	SC 5	3	-	-	03
PHD 260	Solid State Physics 3	SC 6	3	-	-	03
PHD 240	Solid State Physics Lab 2	SC 7	-	-	2	02
PHD 300	Nuclear Physics 2	SC 5	3	-	-	03
PHD 310	Nuclear Physics 3	SC 6	3	-	-	03
PHD 320	Nuclear Physics Lab 2	SC 7	-	-	2	02
PHY-405	Theoretical Physics 2	SC 5	3	-	-	03
PHY-406	Theoretical Physics 3	SC 6	3	-	-	03
PHY-425	Theoretical Physics Lab 2	SC 7	-	-	2	02
Students are permitted to choose any one of the following (Elective papers 1)						
PHD 270	Accelerator Physics	SC 8	2	-	-	02
PHY-408	Liquid Crystals		2	-	-	02
PHY-409	Atmospheric Physics		2	-	-	02
PHY-410	Numerical Methods		2	-	-	02
						04
Students are permitted to choose any one of the following (Elective papers 2)						
PHY-411	Nuclear Spectroscopy Methods	SC 9	3	1	-	04
PHY-412	Modern Optics					
PHD 280	Electronics					
PHY-414	Minor Project					
						04
18						
Semester		HC	SC	OE	Total	
<i>I Semester</i>		14	4		18	
<i>II Semester</i>		14	4		18	
<i>III Semester</i>		13	5	04	22	
<i>IV Semester</i>		04	14		18	
Total		45	27	04	76	

HC: Hard Core; SC: Soft Core; OE: Open Elective; EL: Elective; EC: Extra Credit;

Scheme of Assessment

1. First Theory and Practical Internal Assessment tests (for C1) will be conducted after 8th week of the semester comprising 50% of the syllabus.
2. Second Theory and Practical Internal Assessment tests (for C2) will be conducted in the 16th week of the semester comprising remaining 50% of the syllabus.
3. The Theory C1 and C2 Internal Assessment tests evaluation will be an aggregate of written test, assignments, seminar by the corresponding Course teacher/teachers.
4. Practical Internal Assessment tests C1 and C2 will be assessed based on the performance of the student in the given Lab experiment, Record writing and viva by the corresponding Course teacher/teachers.
5. A student will be eligible for C3 semester end examination, if he/she has scored a minimum of 30% which is the sum total of C1 test and C2 test.
6. C3 semester end examination for theory Courses (papers) will be conducted in 18th week of the Semester. Three sets of question papers will be set by both internal and external Examiners, who were chosen from the panel of examiners approved by the Board of Studies for each Course. The Board of Examination convened well before the C3 semester end examination will scrutinise and approve the question papers sets and submit the same to Controller of Examination.. From each of the sets, one question paper is randomly chosen by Chief Controller of Examination.
7. C3 semester end Laboratory examination for practical Courses will be conducted conveniently before or after C3 semester end examination for theory Courses. The Laboratory examination will be conducted by both internal and external examiners.
8. Overall grade will be based on marks scored in C1, C2 and C3 examination

The details of the allotment of marks for evaluation of the theory and practical question papers of C1, C2 and C3 examinations were also prepared on the recommendations and approval of Board of Studies. The pattern of evaluation is provided below:

Theory Examination : C1+C2+C3=100 Marks

Theory	Unit 1	Unit 2	Unit 3	Problems from the three units	Average Score	Assignment	Seminar	Total Marks
C1	10	10	10	-	10	5	-	15
C2	10	10	10	-	10	-	5	15
C3	18	18	18	16	--	-	-	70

Practical Examination: C1+C2+C3=100 Marks

Practical	Experiment	Record	Viva	Total Marks
C1	7	5	3	15
C2	7	5	3	15
C3	50	-	20	70

Programme Outcomes:

- PO1. Identify, formulate and analyze complex problems using first principles.*
- PO2. A research oriented learning to develop analytical problem-solving approaches.*
- PO3. Awareness of ethical issues and regulatory considerations.*
- PO4. Understand the basic concepts, fundamental principles and the scientific Theories.*
- PO5. Acquire skills in handling scientific instruments, planning and performing in laboratory experiments.*
- PO6. Think creatively in explaining solutions to the problems.*
- PO7. Realize developments in science subject and interdisciplinary approach.*
- PO8. Develop scientific outlook towards all aspects of life.*
- PO9. Effective influence, which inspires in new scientific theories and inventions.*
- PO10. To Imbibe ethical, moral and social values in personal and social life and develop Positive attitude that leads to successful life.*

PROGRAMME SPECIFIC OUTCOMES:

- PSO1. Acquire substantial knowledge in Physics, basic knowledge in Mathematics and computer science*
- PSO2. Specialisation in Solid State Physics/Nuclear Physics provides special expertise.*
- PSO4. Provide hands-on research experience in a specific field of Physics, through the supervised project.*
- PSO5. Familiarise with contemporary research within various fields of Physics.*
- PSO6. Provides the candidate knowledge with general competence and analytical skills needed in research, education, industry, consultancy and public administration.*
- PSO7. Develop interdisciplinary approach.*
- PSO8. Critically assess and evaluate research methods and results.*
- PSO9. Understands the role of Physics in society and ethical problems.*
- PSO10. To avail Global research opportunities.*

Syllabus for the 4-Semester M.Sc., (Physics) Choice Based Credit Scheme (CBCS)

PHA 120: Classical Mechanics

Course Outcome:

- Students who have completed this course should be able to*
- C01. Solve the Newton equations for simple configurations
 - C02. Use conservation to solve dynamics problems.
 - C03. Represent the equations of motion for complicated mechanical systems
 - C04. Derive and solve the equations of motions for systems subject to the Principle of Least Action
 - C05. Calculate conserved quantities from symmetries
 - C06. Acquire a deep knowledge of the Lagrangean and Hamiltonian formalism
 - C07. Understand theory of small oscillations

Mechanics of a system of particles: Conservation of linear and angular momenta in the absence of (net) external forces and torques using centre of mass. The energy equation and the total potential energy of a system of particles using scalar potential (**Goldstein H**).

The Lagrangean method: Constraints and their classifications. Generalized coordinates. Virtual displacement, D'Alembert's principle and Lagrangean equations of the second kind. Examples of (1) single particle in Cartesian, spherical polar and cylindrical polar coordinate systems, (2) Atwood's machine, (3) a bead sliding on a rotating wire in a force-free space and (4) Simple pendulum. Derivation of Lagrange equations from Hamilton principle (**Goldstein H**).

Central forces: Reduction of two particle equations of motion to the equivalent one-body problem, reduced mass of the system. Conservation theorems (First integrals of the motion). Equations of motion for the orbit, classification of orbits, conditions for closed orbits. The Kepler problem (inverse-square law of force) (**Aruldas G, Goldstein H, Srinivasa Rao K.N**). **[16 hours]**

Hamilton's equations: Generalised momenta. Hamilton's equations. Examples - simple harmonic oscillator, charged particle moving in an electromagnetic field. Hamiltonian for a free particle in different coordinates. Cyclic coordinates. Physical significance of the Hamiltonian function. Derivation of Hamilton's equations from a variational principle (**Goldstein H**).

Canonical transformations: Definition, Generating functions (Four basic types). Examples of Canonical transformations. The harmonic Oscillator. Infinitesimal contact transformation. Poisson brackets; properties of Poisson brackets, angular momentum and Poisson bracket relations. Equation of motion in the Poisson bracket notation. The Hamilton-Jacobi equation; the example of the harmonic oscillator treated by the Hamilton-Jacobi method (**Goldstein H**). **[16 hours]**

Mechanics of rigid bodies: Degrees of freedom of a free rigid body. Angular momentum and kinetic energy of rigid body. Moment of inertia tensor, principal moments of inertia, products of inertia, the inertia tensor. Euler equations of motion for a rigid body. Torque free motion of a rigid body. Precession of earth's axis of rotation, Euler angles, angular velocity of a rigid body (**Goldstein H**).

Small oscillations of mechanical system: Introduction, types of equilibria, Quadratic forms of kinetic and potential energies of a system in equilibrium. General theory of small oscillations, secular equation and eigenvalue equation. Small oscillations in normal coordinates and normal modes, examples of two coupled oscillators. Vibrations of a linear triatomic molecule (**Goldstein H**). **[16 hours]**

Total work load

48 hours

References:

1. Goldstein H., Poole C. and Safko J., Classical mechanics, 3rd Edn., Pearson Education, New Delhi. 2002
2. Upadhaya J.C., Classical mechanics, Himalaya Publishing House, Mumbai. 2006.
3. Srinivasa Rao K.N., Classical mechanics, Universities Press, Hyderabad. 2003.
4. Takwale R.G. and Puranik S., Introduction to classical mechanics, Tata McGraw, New Delhi, 1991.
5. Landau L.D. and Lifshitz E.M., Classical mechanics, 4th Edn., Pergamon Press, 1985.
6. Aruldas G., Classical Mechanics, PHI Learning Private Limited, New Delhi

PHA 130: Mathematical Methods of Physics 1

Course Outcome:

Upon completion of the course students should be

- CO1. Familiar with mathematical methods for solving advanced problems in physics.
- CO2. Familiar with Tensors, algebra of tensors and Tensor Calculus and its applications in applied sciences and engineering;
- CO3. Able to solve abstract mathematical problems, recognize real-world problems and to formulate mathematical models for such problems.
- CO4. Familiar with Hermite and Laguerre polynomials solutions
- CO5. Familiar with generating function of the polynomials
- CO6. Able to Use Legendre polynomials, associated Legendre polynomials in Physics
- CO7. Able to Use Bessel functions, Spherical harmonics in Physics

Curvilinear coordinates and Tensors: Curvilinear coordinates in the Euclidean 3-space, Orthogonal curvilinear coordinates. Differential vector operators; Grad, divergence, curl and Laplacian in arbitrary curvilinear coordinates. Circular cylindrical coordinates, spherical polar coordinates (**Arfken & Weber**).

Tensors: Tensors of rank r as a r -linear form in base vectors. Transformation rules for base vectors and tensor components. Tensor algebra, contraction, Raising and lowering of indices, Associated tensors, quotient rule. Mention of pseudo tensor, dual tensor and non-cartesian tensor. Metric tensor, Covariant and contravariant components of the metric tensor, Christoffel symbols. Tensor derivative operators, Covariant differentiation. The contracted Christoffel symbol (**Arfken & Weber**). [16 hours]

Differential equations, Hermite function and Laguerre functions: Differential equations: Partial differential equations, Separation of variables - Helmholtz equations in Cartesian, circular cylindrical coordinates Spherical polar coordinates. Regular and irregular singular points of a second order ordinary differential equation. Series solution-Frobenius power series method, Examples of Harmonic oscillator and Bessel's equation. Linear dependence and independence of solutions-Wronskian. Non-homogeneous equations-Green's function, examples (**Arfken & Weber**).

Hermite functions: Hermite's differential equation and its Solution, Hermite polynomials, Generating functions, Recurrence relations, Rodrigues representation, Orthogonality (**Arfken & Weber**).

Laguerre functions: Laguerre differential equation and its solution, Laguerre polynomials, Generating function, Recurrence relations, Rodrigues representation, Orthogonality. Associated Laguerre functions: Definition, Generating function, Recurrence relations and Orthogonality (**Arfken & Weber**). [16 hours]

Special functions: Sturm - Liouville theory - Self adjoint ODE's, Hermitian operators, completeness of eigenfunction, Green's function—eigenfunction expansion (**Arfken & Weber**).

Bessel functions: Bessel functions of the first kind $J_\nu(x)$, Bessel differential equation, generating function for $J_\nu(x)$, Integrals for $J_0(x)$ and $J_\nu(x)$, recurrence formulae for $J_\nu(x)$, orthogonal properties of Bessel polynomials (**Arfken & Weber**).

Legendre functions: Legendre differential equation, Legendre polynomials, generating functions, recurrence formulae, Rodrigues representation, Orthogonality. Associated Legendre polynomials; The differential equation, Orthogonality relation (**Arfken & Weber**).

Spherical harmonics: Definition and Orthogonality (**Arfken & Weber**).

[16 hours]

Total work load

48 hours

References:

1. Arfken G.B. and Weber H.J., *Mathematical methods for physicists*, 6th Edn., Academic Press, New York (Prism Books, Bangalore, India), 1995.
2. Harris E.G., *Introduction to modern theoretical physics*, Vol. 1, John Wiley, New York, 1975.
3. Srinivasa Rao K.N., *The rotation and Lorentz groups and their representations for physicists*, Wiley Eastern, New Delhi, 2003.
4. Gupta B.D., *Mathematical physics*, 4th Edn, 2011.
5. Bali N. P., *Engineering Mathematics*, Laxmi Publications, New Delhi
6. Dass H. K., *Higher Engineering Mathematics*, S. Chand, New Delhi
7. Chattopadhyay P. K., *Mathematical Physics*, New Age International.

PHA 140: Mathematical Methods of Physics 2

Course Outcome:

Upon completion of the course students should be able to

- CO1. Explain the concepts of Linear vector space.
- CO2. Express conditions for transformations.
- CO3. Explain matrix representation of a linear transformation, Matrix representations
- CO4. Explain concepts of eigenvalues and eigenvectors of a matrix.
- CO5. Use matrices and determinants to solve sets of simultaneous linear equations
- CO6. Understand the concepts of inner product, orthogonality and orthonormality
- CO7. Recognize real-world problems and formulation of mathematical models of such problems.
- CO8. Understand linear representations of groups and Rotation groups and applications in Physics
- CO9. Apply Fourier transforms and Integral equations in Physics

Linear vector space: Linear vector space - Definition. Linear dependence and independence of vectors. Dimension. Basis. Change of basis. Subspace. Isomorphism of vector spaces. Linear operators. Matrix representative of a linear operator in a given basis. Effect of change of basis. Invariant subspace. Eigenvalues and eigenvectors. Characteristic equation. The Schur canonical form. Diagonalization of a normal matrix. Schur's theorem (**Arfken & Weber**). **[16 hours]**

Linear representations of groups: Groups of regular matrices; the general linear groups $GL(n, C)$ and $GL(n, R)$. The special linear groups $SL(n, C)$ and $SL(n, R)$. The unitary groups $U(n)$ and $SU(n)$. The orthogonal groups $O(n, C)$, $O(n, R)$, $SO(n, C)$ and $SO(n, R)$. Homogeneous Lorentz group (**Arfken & Weber**).

Rotation group: The matrix exponential function-Definition and properties. Rotation matrix in terms of axis and angle. Eigenvalues of a rotation matrix. Euler resolution of a rotation. Definition of a representation. Equivalence. Reducible and irreducible representations. Schur's lemma. Construction of the $D^{1/2}$ and D^1 representation of $SO(3)$ by exponentiation. Mention of the D^1 irreps $SO(3)$ (**Srinivasa Rao K.N.**) **[16 hours]**

Fourier transforms and Integral equations: General properties, completeness, use of Fourier series. Applications of Fourier series (**Arfken & Weber**).

Integral transforms; Development of Fourier Integral, Fourier transform - inversion theorem, Fourier transform of derivatives, convolution theorem. Momentum representation (**Arfken & Weber**).

Integral equations: Definitions, transformation of a differential equation into an integral equation, Integral transforms, generating functions, Abel's equation, Neumann series, separable kernels, Numerical solution, non-homogeneous integral equations (**Arfken & Weber**). **[16 hours]**

Total work load

48 hours

References:

1. Shankar R., Principles of quantum mechanics, 2nd Edn., Plenum Press, New York, 1984.
2. Srinivasa Rao K.N., The rotation and Lorentz groups and their representations for Physicists, Wiley Eastern, New Delhi, 1988.
3. Arfken G.B. and Weber H.J., Mathematical methods for Physicists, 5th. Edn., Academic Press, New York, 2001.
4. Gupta B.D., Mathematical Physics, 4th Edn. (Page no. 8.48–8.83, 8.16-8.48) 2011
5. Bali N. P., Engineering Mathematics, Laxmi Publications, New Delhi
6. Dass H. K., Higher Engineering Mathematics, S. Chand Publications, New Delhi
7. Charlie Harper, Introduction to Mathematical Physics, PHI Publications, 2008.

PHA 150: Classical Electrodynamics, Plasma Physics and Optics

Course Outcome:

Students who have completed this course should

- CO1. Have theoretical foundations of electromagnetic phenomena.
- CO2. Be able to solve the Maxwell equations for simple configurations.
- CO3. Formulate and solve electromagnetic problems with the help of electrodynamic potentials and super potentials
- CO4. make a detailed account for gauge transformations and their use
- CO5. Learn the techniques of deriving and evaluating formulae for the electromagnetic fields from general charge and current distributions
- CO6. calculate the electromagnetic radiation from radiating systems, formulate and solve electrodynamic problems, covariant form in 4 dimensional space-time
- CO7. Formulate self-consistent models for the interaction between matter and em fields,
- CO8. Covariant formulation of electrodynamics, Lagrange formalism
- CO8. Apply the concept of Special theory of relativity for relativistic electrodynamics

Electric multipole moments: The electric dipole and multipole moments of a system of charges. Multipole expansion of the scalar potential of an arbitrary charge distribution **(Griffiths D.J.)**.

Potential formulation: Maxwell equations in terms of electromagnetic potentials. Gauge transformations. The Lorentz, Coulomb and radiation gauges **(Griffiths D.J.)**.

Fields of moving charges and radiation: The retarded potentials. The Lienard-Wiechert potentials. Fields due to an arbitrarily moving point charge; the special case of a charge moving with constant velocity **(Griffiths D.J.)**.

Radiating systems: Radiation from an oscillating dipole. Power radiated by a point charges - Larmor formula. Lienard's generalisation of Larmor formula. Energy loss in bremsstrahlung and linear accelerators. Radiation reaction - Abraham-Lorentz formula **(Griffiths D.J)** **[16 hours]**

Relativistic electrodynamics: Charge and fields as observed in different frames. Covariant formulation of electrodynamics; Electromagnetic field tensor, Transformation of fields, Field due to a point charge in uniform motion. Lagrangian formulation of the motion of charged particle in an electromagnetic field **(Griffiths D.J.)**.

Plasma Physics: Quasineutrality of a plasma, plasma behaviour in magnetic fields, Plasma as a conducting fluid. Magnetohydrodynamics; magnetic confinement, Pinch effect, instabilities, Plasma waves. **(Laud B. B.)**

[16 hours]

Electromagnetic waves: Monochromatic plane waves - velocity, phase and polarization. Propagation of plane electromagnetic waves in (1) conducting media and (2) ionised gases. Reflection and refraction of electromagnetic waves; Fresnel formulae for parallel and perpendicular components. Brewster's law. Normal and anomalous dispersion; Clausius-Mossotti relation **(Born M. and Wolf E)**.

Interference: General theory of interference of two monochromatic waves. Two beam and Multiple beam interference with a plane-parallel plate. Fabry-Perot interferometer; etalon construction, resolving power and its application. Interference filters **(Born M. and Wolf E)**.

Diffraction: Integral theorem of Helmholtz and Kirchoff. Fresnel-Kirchoff diffraction formula; conditions for Fraunhofer and Fresnel diffraction. Fraunhofer diffraction due to a circular aperture. **(Born M. and Wolf E)**

[16 hours]

Total work load

48 hours

References:

1. Griffiths D.J., Introduction to Electrodynamics, 5th Edn., Prentice-Hall of India, New Delhi, 2006.
2. Jackson J.D., Classical Electrodynamics, 2nd Edn., Wiley-Eastern Ltd, India, 1998.
3. Born M. and Wolf E., Principles of Optics, 6th Edn., Pergamon Press, Oxford, 1980.
4. Matveev A.N., Optics, Mir Publishers, Moscow, 1988.
5. Laud B.B., Electromagnetics, Wiley Eastern Limited, India, 2000.
6. Hecht E., Optics, Addison-Wesley, 2002.
7. Lipson S.G., Lipson H. & Tannhauser D.S., Optical physics, Cambridge University Press, USA, 1995.
8. Ajoy Ghatak, Optics, Tata McGraw - Hill, New Delhi
9. Gupta A. B. Modern Optics, Books and Allied (P) Ltd, Kolkata
10. Sen S.N., Plasma Physics, Pragathi Prakashan

PHA 160: Computer Lab CL-A

Course Outcome:

Students who have completed this course should

- CO1. Learn scientific typesetting with LaTeX program.*
- CO2. Plot the functions and data with Gnuplot program,*
- CO3. Solve problems with Octave program*

- *Linux operating system basics (4 sessions) :
Login procedure; creating, deleting directories; copy, delete, renaming files; absolute and relative paths; Permissions—setting, changing; Using text editor.*
- *Scientific text processing with LATEX.
Typeset text using text effects, special symbols, lists, table, mathematics and including figures in documents.*
- *Using the plotting program GNUPLOT (2 sessions) :
Plotting commands; To plot data from an experiment and applying least-squares fit to the data points. Including a plot in a LATEX file.*
- *Using the mathematics package OCTAVE (2 sessions), To compute functions, matrices, eigenvalues, inverse, roots.*

Total work load: 1 day(s) per week × 4 hours × 16 weeks = 64 hours

PHA 220: Electronics Lab

Course Outcome:

Students who have completed this course should

- CO1. Maintain digital and analog devices and circuits.*
- CO2. Analyze components associated with digital and analog electronic systems.*
- CO3. Demonstrate proficiency in the use of electronic equipment and devices.*
- CO4. Assist in the design, operation, and troubleshooting of electronic systems.*
- CO5. Analyse electronics devices and circuits using computer simulations.*
- CO6. Realise electronic circuits using electronic devices and mathematical concepts.*
- CO7. Accept professional and ethical responsibilities of the engineering technology*

Any ten of the following experiments:

1. *Regulated power supply.*
2. *Active filters : low pass (single pole).*
3. *Active filters : high pass (double pole).*
4. *Voltage follower.*
5. *Colpitts' oscillator.*
6. *Opamp as an integrator and differentiator.*
7. *Opamp as a summing and log amplifier.*
8. *Opamp as an inverting and non-inverting amplifier.*
9. *Coder and encoder.*
10. *Half adder and full adder.*
11. *Boolean algebra-Logic gates.*
12. *Opamp astable multivibrator.*

Total work load: 2 day(s) per week × 4 hours × 16 weeks = 128 hours

PHB 070: Continuum Mechanics and Relativity

Course Outcome:

After the completion of the course, student will be familiar with

- C01. Internal response of materials to external loading
- C02. Development of physical intuition for the behaviour of solids and fluids
- C03. Unique connections between solid and fluid mechanics.
- C04. The non-existence ether through Michelson-Morley experiments
- C05. Minkowski space-time
- C06. The true nature of Newtonian mechanics and Lorentz Transformations
- C07. The concept of constant relative motion of different bodies in different frames
- C08. The calculation of proper time, dilated time, proper length and contracted length
- C09. Principles of the special theory of relativity, Covariant form of Newton laws
- C010. Einstein field equations, their underlying mathematical structure and solutions
- C011. Lagrangean formulation of General Relativity.
- C012. Derivation of testable physical consequences of General Relativity

Continuum mechanics of solid media: Small deformations of an elastic solid; the strain tensor. The stress tensor. Equations of equilibrium. The symmetry of the stress tensor. The generalised Hooke's law for a homogeneous elastic medium; the elastic modulus tensor. Navier equations of motion for a homogeneous isotropic medium. **(Landau L.D. and Lifshitz)**

Fluid mechanics: Equation of continuity. Flow of a viscous fluid; Navier-Stokes equation and its solution for the case of flow through a cylindrical pipe. The Poiseuille formula **(Landau L.D. and Lifshitz)**. **[16 hours]**

Minkowski space-time: Real coordinates in Minkowski space-time. Definition of 4-tensors. The Minkowski scalar product and the Minkowski metric $\eta_{ij} = \text{diag}(1 -1 -1 -1)$. Orthogonality of 4-vectors. Raising and lowering of 4-tensor indices. Time like, null and space like vectors and world-lines. The light-cone at an event **(Griffiths)**.

Relativistic mechanics of a material particle: The proper-time interval $d\tau$ along the world - line of a material particle. The instantaneous (inertial) rest-frame of a material particle; Components of 4-velocity, 4-acceleration and 4-momentum vector, statement of second law of Newton. Determination of the fourth component F_4 of the 4-force along the world-line of the particle. Motion of a particle under the conservative 3-force field and the energy integral. The rest energy and the relativistic kinetic energy of a particle. **[16 hours]**

Einstein's equations: The Principle of Equivalence and general covariance. Inertial mass, gravitational mass, Eötvös experiment. Gravitation as space-time curvature. Einstein Gravitational field equations and its Newtonian limits.

The Schwarzschild metric: Heuristic derivation of the Schwarzschild line element. Motion of particles and light rays in the Schwarzschild field. Explanation of the (1) perihelion advance of planet Mercury, (2) gravitational red shift and (3) gravitational bending of light. A brief discussion of the Schwarzschild singularity and the Schwarzschild black hole. **[16 hours]**

Total work load

48 hours

References:

1. Landau L.D. and Lifshitz E.M., Fluid Mechanics, Pergamon Press, 1987.
2. Landau L.D. and Lifshitz E.M., Theory of Elasticity, Pergamon Press, 1987.
3. Synge J.L., Relativity: The Special Theory, North-Holland, 1972.
4. Landau L.D. and Lifshitz E.M., The Classical Theory of Fields, 4th Edn., (Sections 1 to 6, 16 to 18, 23 to 25, 26 to 35), Pergamon Press, Oxford, 1985.
5. Wald R.M., General relativity, The University of Chicago Press, Chicago, 1984.
6. Schutz B.F., A first course in general relativity, Cambridge University Press, Cambridge, 1985.
7. Bergman P., Introduction to theory of relativity, Prentice-Hall of India, 1969.
8. Rindler R., Relativity: Special, general and cosmological, Oxford University Press, 2006.
9. Narlikar J. V., An introduction to Cosmology, Cambridge Publications
10. Somnath Datta, Introduction to Special theory of Relativity, Allied Publishers, India, 1998
11. Griffiths D. J. Introduction to Electrodynamics, Pearson Publications, 2013.

PHB 080: Thermal Physics

Course Outcome:

After the completion of the course, Student will be familiar with

- CO1. Basic concepts of Thermodynamics
- CO2. Formulation of the first law of thermodynamics for a closed systems
- CO3. Mode of heat transfer, the amount of heat energy transferred and conservation of mass and energy equations.
- CO4. Second law of thermodynamics; thermal efficiency & coefficient of performance.
- CO5. concept of ensemble, phase space & conservation of phase-space density(Liouville's theorem)
- CO6. Solution to statistical mechanics problems for simple non-interacting systems,
- CO7. Phase transitions, Statistical equilibrium,. Entropy and probability
- CO10. Microcanonical ensemble, Partition function, Gibbs paradox, canonical ensemble
- CO11. Quantum statistical mechanics. Symmetry of wave functions.
- CO12. Quantum distribution functions (BE and FD), Applications of Quantum Statistics

Thermodynamics Preliminaries: Zeroth law of thermodynamics, vander Walls equation of state second law of thermodynamics (Huang K., Laud B.B, Satya Prakash).

Entropy: Change in entropy for reversible an irreversible process, entropy and second law of thermodynamics, thermodynamic functions and Maxwell's relations TdS equations, heat capacities equations, third law of thermodynamics. Irreversible thermodynamics Onsager's reciprocal relation (Huang K., Laud B.B, Satya Prakash).

Phase equilibria; Equilibrium conditions. Classification of phase transitions; phase diagrams; Clausius-Clapeyron equation, applications. Thermoelectric phenomenon, Peltier effect, Seebeck effect, Thompson effect. Systems far from equilibrium (Huang K., Laud B.B, Satya Prakash). [16 hours]

Classical Statistical Mechanics: Probability, phase space, division of phase space, ensembles, density distribution in phase space, ergodic hypotheses, Liouville theorem. Statistical equilibrium, postulate of equal a priori probability, general expression for probability, Stirlings formula, the most probable distribution, Maxwell Boltzmann distribution law, law of equipartition of energy. Entropy and probability.

Microcanonical ensemble, connection between statistical and thermodynamic quantities, Partition function of system of particles, Gibbs paradox, canonical ensemble, perfect monoatmic gas in canonical ensemble, grand canonical ensemble. Vibrational partition function of diatomic molecules (Einstein relations), Rotational partition function of diatomic molecule (Huang K., Laud B.B, Satya Prakash). [16 hours]

Quantum Statistical Mechanics: The postulates of quantum statistical mechanics. Symmetry of wave functions. The Liouville theorem in quantum statistical mechanics; condition for statistical equilibrium; Ensembles in quantum mechanics; the quantum distribution functions (BE and FD), the Boltzmann limit of Boson and Fermion gases, the derivation of the corresponding distribution functions.

Applications of Quantum Statistics: Equation of state of an ideal Fermi gas (derivation not expected), Application of Fermi-Dirac statistics to the theory of free electrons in metals, degeneracy. Application of Bose statistics to the photon gas, derivation of Planck's law, comments on the rest mass of photons. Thermodynamics of Black body radiation. Bose-Einstein condensation (Huang, Laud, Satya Prakash). [16 hours]

Total work load

48 hours

References:

1. Agarwal B.K. and Eisner M., Statistical mechanics, New Age International Publishers, 2000.
2. Roy S.K., Thermal physics and statistical mechanics, New Age International Pub., 2000.
3. Huang K., Statistical mechanics, Wiley-Eastern, 1975.
4. Laud B.B., Fundamentals of statistical mechanics, New Age International Pub.,2000.
5. Schroeder D.V., An introduction to thermal physics, Pearson Education New Delhi, 2008.
6. Salinas S.R.A., Introduction to statistical physics, Springer, 2004.
7. Mark W Zemansky Heat and Thermodynamics, McGraw – Hill
8. Gupta A. B and Roy H. B., Thermal Physics Books and Allied (P) Ltd, Kolkata
9. Satya Prakash, Statistical Mechanics, Kedarnath Ramnath, 2017.
10. Mike Glazer, J.S. Wark, Statistical Mechanics: A Survival Guide, Oxford Publications, 2001.

PHB 090: Quantum Mechanics 1

Course Outcome:

After the completion of the course Student will be familiar with

- CO1. The Basic concepts and mathematical foundations of quantum mechanics
- CO2. Solutions to the Schrödinger equation for simple potentials.
- CO3. The effect of symmetries in quantum mechanics
- CO4. The significance of wave function, normalization, uncertainty Principle
- CO5. The Physical significance of eigen functions and eigen vectors
- CO6. Formalism; Hilbert space,, Dirac notation, concepts of Dynamical variables and operators
- CO7. General Uncertainty relation. Matrix representation of wave functions & operators.
- CO8. The Schrödinger equation and time evolution of a system.
- CO9. The Schrödinger picture and Heisenberg picture
- CO10. Schrodinger equation in three dimensions, Hydrogen atom.

The wave function and uncertainty Principle: Wave particle duality, interpretation of the wave function, wave functions for particles having definite momentum, wave packet, Gaussian wave packet. Heisenberg uncertainty principle.

Time independent Schrodinger equation, conservation of probability, expectation values and operators, the Ehrenfest theorem, Time dependent Schrodinger equation, stationary states. Energy quantisation. Properties of energy eigenfunction, general solutions of time dependent Schrodinger equation for a time independent potential. Schrodinger equation in momentum space **(Bransden & Joachain).** [16 hours]

Formalism: Hilbert space. The state of a system, Dirac notation. Dynamical variables and operators – Hermitian operators, adjoint operator, projection operators. Inverse and unitary operators. Expansion in eigenfunctions - eigenvalue and eigenfunction of an operator. Commutator algebra. General Uncertainty relation. Unitary transformation, Representation in discrete basis; Matrix representation of wave functions and operators. Change of representation and Unitary transformations. Matrix representation of eigenvalue problem. Representation in continuous bases. The Schrödinger equation and time evolution of a system. The Schrödinger picture and Heisenberg picture.

Schrodinger equation in one dimension: The free particle, the potential step, potential barrier, infinite square well, finite square well, the linear harmonic oscillator (Algebraic and Analytic method), the periodic potential **[Bransden and Joachain, Nouredine Zettili].** [16 hours]

Angular Momentum: Orbital angular momentum; Orbital angular momentum and spatial rotations, eigenvalues and eigenfunctions of L^2 and L_z . Particle on a sphere and the rigid rotator. General angular momentum. The spectrum of J^2 and J_z . Matrix representation of angular momentum operators, spin angular momentum, spin one-half, total angular momentum. Addition of angular momenta - CG Coefficients.

Schrodinger equation in three dimensions: Separation of the Schrodinger equation in Cartesian coordinates - the free particle. Central potential. Separation of the Schrodinger equation in spherical polar coordinates; the Hydrogenic atom and its solutions **(Bransden & Joachain).** [16 hours]

Total work load

48 hours

References:

1. Nouredine Zettili, Quantum Mechanics, WILEY Publications, U K 2009
2. Griffiths D.J., Introduction to quantum mechanics, Prentice-Hall, USA, 1994.
3. Bransden & Joachain, 2004, II edition, Pearson Low Price Edition
4. Sakurai J.J. and Tuan S.F. (Editor), Modern quantum mechanics, AddisonWesley, India, 1999.
5. Shankar R., Principles of quantum mechanics, 2nd Edn., Plenum Press, New York, 1984.
6. Schiff L.I., Quantum mechanics, 3rd. Edn., McGraw-Hill, Kogakusha Ltd., New Delhi, 1968.
7. Aruldas G., Quantum Mechanics, PHI, New Delhi
8. Mathews P. M. and Venkatesan K., Quantum mechanics, Tata - McGraw-Hill, New Delhi
9. Verma H. C., Quantum Physics, Surya Publications, Ghaziabad
10. Merzbacher E., Quantum Mechanics, III edition, Wiley publication.

PHB 100: Spectroscopy and Fourier Optics

Course Outcome:

After completing this course, the student will be able to

- CO1. Explain the applications of spectroscopic methods for qualitative & quantitative analysis.
- CO2. Compare and contrast atomic and molecular spectra.
- CO3. Understand the molecular absorption & scatter from particulate matter in atomic absorption spectroscopy
- CO4. Recognise the effect of changing the slit width on qualitative & quantitative analysis.
- CO5. Explain the selection rule for infrared-active transitions
- CO6. design a non-dispersive infrared spectrophotometer
- CO7. Understand the Fourier Transform infrared spectroscopy
- CO8. understand Raman spectroscopy
- CO9. Elements of Nonlinear Optics

Atomic spectroscopy: vector model of atom- orbital magnetic moment, Larmor precession, electron spin, coupling of orbital and spin angular momenta. Spectroscopic terms and their notations, spin-orbit interaction, quantum mechanical relativistic correction. Fine structure of hydrogen, Lamb shift. L-S and J-J coupling. Lande interval rule, selection rules.

Zeeman effect, Examples $1 \quad {}^2_{32}D - {}^2_{12}P$ $2) \quad {}^2_{52}D - {}^2_{32}P$ $3) \quad {}^3_0P - {}^2_0S$.

Anomalous Zeeman effect, Lande-g factor, Paschen-Back effect – spin-orbit correction. Stark effect – weak field effects and strong field effects. Hyperfine structure of spectral lines. Nuclear spin and hyperfine splitting, intensity ratio and determination of nuclear spin. Breadth of spectral lines, natural breadth. Doppler Effect and external effect (**Rajkumar**). **[16 hours]**

Nuclear magnetic resonance: Quantum mechanical expression for the resonance condition. Relaxation Mechanisms; Expression for spin lattice relaxation. Chemical shift; spin-spin interaction, example of ethyl alcohol. Fourier transform technique in NMR. FTNMR spectrometer and experimental procedure. NMR in medicine.

Microwave spectroscopy: The classification of molecules. The rotational spectra of rigid diatomic rotator, the spectra of non-rigid diatomic rotator, example of HF. Microwave oven.

Infrared spectroscopy: The Born-Oppenheimer approximation. Vibrational energy of diatomic molecule. Anharmonic oscillator. Diatomic vibrating rotator, example of the CO molecule. The vibrations of polyatomic molecules; skeletal and group frequencies. Experimental technique in FTIR.

Raman spectroscopy: The quantum theory of Raman effect. Pure rotational Raman spectra of linear molecules and symmetric top molecules. Vibrational Raman spectra. Rotational fine structure. Instrumentation technique in Raman spectroscopy (**Banwell C.N. & McCash E.M and Aruldas**). **[16 hours]**

Fourier optics: Spatial frequency filter; effect of a thin lens on an incident field distribution. Lens as a Fourier transforming element. Application to phase contrast microscopy. (**Hecht**)

Propagation of light in an anisotropic medium: Structure of a plane electromagnetic wave in an anisotropic medium. Dielectric tensor. Fresnel's formulae for the light propagation in crystals. Ellipsoid of wave normals and ray normals. Normal surface and ray surface. Optical classification of crystals. Light propagation in uniaxial and biaxial crystals. Refraction in crystals. (**Born M. and Wolf E.**)

Elements of Nonlinear Optics: Second harmonic generation, optical rectification and phase matching; third harmonic generation (**Lipson, Srivatsava**). **[16 hours]**

Total work load

48 hours

References:

1. Tralli N. and Pomilla P.R., Atomic theory, McGraw-Hill, New York, 1999.
2. Banwell C.N. and McCash E.M., Fundamentals of Molecular Spectroscopy, 4th Edn., Tata McGraw-Hill, New Delhi, 1995.
3. Mahan B.H., University Chemistry, 3rd Edn. (Chapters 3, 10, 11 and 12), Narosa, New Delhi, 1975.
4. Hecht E., Optics, Addison-Wesley, 2002.
5. Lipson S.G., Lipson H. and Tannhauser D.S., Optical physics, Cambridge University Press, USA, 1995.
6. Rajkumar, Atomic and molecular spectra: Laser, Kedarnath Ramanath Publications, Meerut.
7. Born M. and Wolf E., Principles of optics, 6th Edn., Pergamon Press, Oxford, 1980
8. Srivatsava, P K Optics, CBS Publisher & Distributors I Edition, 2011

PHB 110: Computer Lab CL-B

Course Outcome:

After completing this course the student will

- CO1. *Learn Computer Programming,*
- CO2. *Have Hands on experience with C programming language and its syntax,*
- CO3. *Apply C techniques for physics problem solving and data analysis.*
- CO4. *Learn PERL scripting language*

Programming in C

- *Check whether given number is odd or even.*
- *Find the largest and smallest number in the input set.*
- *Compute the Fibonacci sequence.*
- *Check whether the input number is prime or not.*
- *Compute the roots of a quadratic equation.*
- *Generate Pascal's triangle.*
- *To add two m X n matrices.*
- *To find the sum and average of a data stored in a file.*
- *Linear least-squares fitting to data in a file.*
- *To find the trajectory of a projectile shot with an initial velocity at an angle. Also, find the maximum height travelled and distance travelled. Write the trajectory data to a file specified and plot using Gnuplot.*

Programming in Perl

- *Searching for a pattern in a string.*
- *Counting the number of characters, words and lines in a given file.*
- *Sorting strings.*
- *Check whether the input number is prime or not.*
- *Compute the roots of a quadratic equation.*
- *Linear least squares fitting to data in a file.*

Total work load : 1 day(s) per week × 4 hours × 16 weeks = **64 hours**

PHB 220: Optics Lab

Course Outcome:

After completing this course the student will

- CO1. *Have hands on experience of experiments using spectrometer*
- CO2. *Use the principles of superposition to explain interference, diffraction and polarisation*
- CO3. *Describe the operation of optical devices, including, polarisers and interferometers.*
- CO4. *Solve problems in optics by selecting the appropriate equations & performing numerical or analytical calculations.*
- CO5. *Follow instructions to perform experiments & document their results, using correct procedures and protocols.*
- CO6. *Analyse, interpret and communicate results from laboratory experiments, orally or in a written laboratory report.*

Any ten of the following experiments:

1. *Verification of the Brewster law of polarisation.*
2. *Verification of Fresnel laws of reflection from a plane dielectric surface.*
3. *Determination of the inversion temperature of the copper-iron thermocouple.*
4. *Birefringence of mica by using the Babinet compensator.*
5. *Birefringence of mica by using the quarter-wave plate.*
6. *Experiments with the Michelson interferometer.*
7. *Determination of the refractive index of air by Jamin interferometer.*
8. *Determination of the size of lycopodium spores by the method of diffraction haloes.*
9. *Determination of wavelength by using the Fabry-Perot etalon.*
10. *Dispersion of the birefringence of quartz.*
11. *The Franck-Hertz experiment.*
12. *Experiments with the laser.*
13. *Determination of the Stokes vector of a partially polarised light beam*
14. *Determination of the modes of vibration of a fixed-free bar.*

Total work load : 2 day(s) per week × 4 hours × 16 weeks = **128 hours**

PHC 050: Quantum Mechanics 2

Course Outcome:

- After completing this course the student will be able to
- CO1. Understand the effect of symmetries in quantum mechanics
 - CO2. Understand the significance of wave function, normalization, uncertainty principle, Physical significance of eigenfunctions and eigenvectors
 - CO3. Time-independent perturbation theory: Non degenerate Perturbation Theory
 - CO4. Degenerate Perturbation Theory; Fine Structure of Hydrogen, Zeeman Effect.
 - CO5. The Variational Principle: Theory, the Ground State of Helium.
 - CO6. WKB Approximation: The Classical Region, Tunneling; connection formulae, α - decay
 - CO7. The time-dependent perturbation theory
 - CO8. Adiabatic approximation and scattering phenomena
 - CO9. Have a basic understanding of relativistic effects in quantum mechanics.
 - CO10. Mathematical foundations of quantum mechanics & Schrödinger equation for simple cases.
 - CO11. Relativistic quantum mechanics using Klein-Gordon equation & Dirac equation

The time-independent perturbation theory: Non degenerate Perturbation Theory; first and second order perturbation, Perturbed Harmonic Oscillator. Degenerate Perturbation Theory; Fine Structure of Hydrogen, The Zeeman Effect.

The Variational Principle: Theory, the Ground State of Helium.

WKB Approximation: Classical Region, Tunneling; connection formulae, α -particle decay (Griffiths). [16 hours]

Time-dependent perturbation theory: Time dependent perturbation theory; general features, constant and periodic perturbations. Two-Level Systems; Emission and Absorption of Radiations, Spontaneous Emission, Fermi golden rule, Rabi Oscillations.

Adiabatic approximation - The Adiabatic Theorem, Berry's Phase. Sudden approximation.

Scattering: Introduction, scattering cross section, scattering by a spherically symmetric potential. Partial Wave Analysis, phase shifts. Optical theorem, Lippmann- Schwinger equation. Born Approximation, Rutherford scattering (Griffiths D J). [16 hours]

Relativistic quantum mechanics: Klein-Gordon equation: free particle, stationary state solutions, continuity equation. The Dirac equation; free-particle, stationary state solutions, continuity equation. Covariant formulation; Covariant form of Dirac equation, Lorentz invariance of the Dirac equation, Plane wave solutions of the Dirac equation -non-relativistic limit. Spin and helicity operators. Normalization of the solutions. Brief discussion of the hydrogen atom according to Dirac theory, Non-relativistic limit of Dirac equation. Negative energy states - Hole theory (Sakurai J J). [16 hours]

Total work load

48 hours

References:

1. Bransden and Joachain, II edition, Pearson Low Price Edition
2. Sakurai J.J. and Tuan S.F. (Editor), Modern Quantum Mechanics, Addison Wesley, India, 1999.
3. Shankar R., Principles of Quantum Mechanics, 2nd Edn., Plenum Press, New York, 1984.
4. Schiff L.I., Quantum mechanics, 3rd. Edn., McGraw-Hill, Kogakusha Ltd., New Delhi, 1968.
5. Griffiths D.J., Introduction to Quantum mechanics, Prentice-Hall, USA, 1994.
6. Sakurai J.J., Advanced quantum mechanics, Addison-Wesley, Harlow, England, 1999.
7. Griffiths D., Introduction to Elementary particles, John Wiley and Sons, New York, 1987.
8. Gasiorowicz S., Elementary Particle Physics, John-Wiley, New York, 1966.
9. Muirhead H., The Physics of Elementary Particles, Pergamon Press, London, 1965.

PHC 060: Condensed Matter Physics**Course Outcome:**

After completing this course, the student will be able to

- C01. Understand the principles of Condensed Matter Physics, Mathematical descriptions of physical phenomena.
- C02. Understand the principles of crystal structure of elements Instrumentation for crystal studies
- C03. Evaluation of crystals data and their suitability for single crystal structure analysis.
- C04. Crystal growth techniques
- C05. Liquid crystals; Classification and structures
- C06. Understand the Structural, Magnetic, Electrical and Semiconductor Properties
- C07. Theoretical and practical experience in the Condensed Matter Physics

X-ray crystallography: Crystalline state. Reference axes, equation of a plane, Miller indices. External symmetry of crystals; symmetry operations. Two and three dimensional point groups. Lattices; two dimensional lattices, choice of unit cell. **(Buerger, p12-20, 23-45).**

Three-dimensional lattices; crystal systems and Bravais lattices. Screw and glide operations. Space groups; Examples of space groups. Diffraction of X rays by crystals; Laue equations. Reciprocal lattice. **[Sherwood, p272-288].** Bragg equation. Equivalence of Laue and Bragg equations. Significance of structure of solid for applications **(Ladd and Palmer, p55-66, p114-121).**

Atomic scattering factor (qualitative).

Electron and neutron diffraction: Basic principles. Differences between electron, neutron and X-ray diffractions, applications (qualitative). **(Vainshtein, p 336 - 357).**

Crystal growth techniques: General methods of crystal growth. Czochralski, Kyropoulos, Stockbarger-Bridgman. Zone refining techniques **(Rose et al p 146 - 154). [16 hours]**

Disordered materials: Amorphous solids. Aperiodic materials.

Liquid crystals: Introduction, Classification and their applications. Morphology. The smectic (A-H), nematic and cholesteric phases. Birefringence, texture and X-ray studies. Orientational order and its determination for nematic liquid crystals **(DeGennes P.G. and Prost J, Gray and Goodby)**

Crystal lattice dynamics: Vibration of an infinite one-dimensional monoatomic lattice, First Brillouin Zone. Group velocity. Finite lattice and boundary conditions. Vibrations of a linear diatomic lattice; optical and acoustical branches, dispersion relations. **(Wahab, p288-305).**

Magnetic properties of solids: Diamagnetism and its origin. Expression for diamagnetic susceptibility. Paramagnetism; Quantum theory of paramagnetism, Brillouin function. Ferromagnetism; Curie-Weiss law, Spontaneous magnetisation and its variation with temperature. Ferromagnetic domains. Antiferromagnetism. Two sub-lattice model. Susceptibility below and above Neel's temperature. **(Dekker, p446-490). [16 hours]**

Superconductivity: Experimental facts. Type I and type II superconductors. Phenomenological theory. London equations. Meissner effect. High frequency behaviour. Thermodynamics of superconductors; Entropy and Specific heat. Qualitative ideas of the theory of superconductivity. **(Kittel, p333-364).**

Semiconductors: Elemental and compound Semiconductors **[Streetman, p61-95].** Crystal structure and bonding. Expressions for carrier concentrations. Fermi energy, electrical conductivity and energy gap in intrinsic semiconductors. Extrinsic Semiconductors; impurity states and ionization energy of donors. Carrier concentrations and their temperature variation **(Mckelvey, p256-277). [16 hours]**

Total work load

48 hours

References:

1. Stout G.H. and Jensen L.H., X-ray structure determination, MacMillan, USA, 1989.
2. Ladd M.F.C. and Palmer R.A., Structure determination by X-ray crystallography, Plenum Press, USA, 2003.
3. Buerger M.J., Elementary crystallography, Academic Press, London.
4. Dekker A.J., Solid state physics, Prentice Hall, 1985.
5. Kittel C., Introduction to solid state physics, 7th Edn., John Wiley, New York, 1996.
6. Mckelvey J.P., Solid state and semiconductor physics, 2nd Edn., Harper and Row, USA, 1966.
7. Streetman B.G., Solid state electronic devices, 2nd Edn., Prentice-Hall of India, New Delhi, 1983.
8. DeGennes P.G. and Prost J., The physics of liquid crystals, 2nd Edn., Clarendon Press, Oxford, 1998.
9. Wahab M.A., Solid state physics, Narosa Publishing House, New Delhi, 1999.
10. Azaroff L.V., Introduction to solids, McGraw-Hill Inc, USA, 1960.
11. Sherwood D., Crystals, X-rays and proteins, Longman, UK, 1976.
12. Rose R.M., Shepard L.A. and Wulff J., The structure and properties of materials Vol. 4, Electronic properties, Wiley Eastern, 1965.

13. Vainshtein B.K., *Modern crystallography, Vol. I, Springer-Verlag, Germany, 1981.*
 14. Pillai S.O., *Solid state physics, New Age International Publications, 2002.*

PHC 070: Nuclear and Particle Physics

Course Outcome:

After completing this course the student will be able to

- CO1. Understand the relation between the standard model and QCD.
 CO2. Understanding strong interactions with many-body physics.
 CO3. Quantitatively estimates for nuclear phenomena
 CO4. Understand the Standard Model
 CO5. Familiarise with theoretical and experiments used in particle physics.
 CO6. Stabilise the nucleus, nuclear forces, interactions and models.
 CO7. Critically assess a range of applications of nuclear technology.
 CO8. Develop Theoretical & practical experience in the scattering experiments & gamma-ray spectroscopy.

Properties of the Nucleus: Nuclear radius; determination by mirror nuclei, Mesic X-rays and electron scattering methods. Nuclear moments; spin, magnetic dipole moment. Relation between J and μ on the basis of single particle model. Determination of nuclear magnetic moment by Molecular beam experiment. Electric quadrupole moment – reduced Electric quadrupole moment .

Nuclear Models: Liquid drop model; Weissacker's formula and its application to (1) stability of isobars and (2) fission process. Shell model; Infinite square well potential, Magic numbers. Fermi gas model; well depth, level density and nuclear evaporation.

Nuclear reactions: Q -values, threshold energy. Reactions induced by proton, deuteron and particles. Photodisintegration (**Krane & Taya**). [16 hours]

Nuclear decay modes: Beta decay; Beta ray spectrum, Pauli neutrino hypothesis, mass of the neutrino from beta ray spectral shape, Fermi theory of beta decay, Kurie plot, f_t - values and forbidden transitions. Methods of excitation of nuclei; Nuclear isomerism, Mossbauer effect (qualitative only), Auger effect.

Interaction of nuclear radiation with matter: Energy loss due to ionization for proton -like charged particles, Bethe-Bloch formula, Range energy relations. Ionisation and Radiation loss of fast electrons (Bremsstrahlung - qualitative only). Interaction of gamma and X-rays with matter. Detectors; Brief description of NaI (TI) gamma ray spectrometer. Boron trifluoride counter.

Nuclear reactors: Condition for controlled chain reactions, slowing down of neutrons, logarithmic decrement in energy. Homogeneous spherical reactor; critical size, effect of reflectors. Breeder reactor(Qualitative discussion) (**Krane & Taya**). [16 hours]

Nuclear forces and elementary particles: General features of nuclear force; spin dependence, charge independence, exchange character, saturation other features. Meson theory of nuclear forces; Yukawa's theory. Properties of pi mesons; charge, mass, spin, isospin and parity, decay modes, meson resonances.

Particle interactions and families: Conservation laws; classification of fundamental forces and elementary particles. Associated particle production, Gellmann-Nishijima scheme, strange particles. CP violations in Kaon decay. Symmetries; Eight-fold way symmetry, quarks and gluons. Elementary ideas of the Standard model (**Griffiths D J**). [16 hours]

Total work load

48 hours

References

1. Taya D.C., *Nuclear Physics, Himalaya Publishing House, New Delhi, 2012 (Unit 1. Chapter Page 6-14. Page 30- 35, 40-49. Chapter 9. Page 355-369. Chapter 10. Page 401-411.)*
2. Krane K.S., *Introductory nuclear physics, Wiley, New York, 1987. (Unit 1. Chapter 16 page 605-610.)*
3. Ghoshal S.N., *Nuclear physics, S.Chand and Company, Delhi, 1994. (Unit 2: Chapter 5 page 137-155, Chapter 6 page 187-204, 222, 262, Chapter 13, page 647-651, chapter 15, page717-721.)*
4. Wong S.S.M., *Introductory nuclear physics, Prentice Hall of India, Delhi, 1998.*
5. Khanna M.P., *Introduction to particle physics, Prentice Hall of India, Delhi, 2008.*
6. Kapoor S.S. and Ramamoorthy V., *Nuclear radiation detectors, Wiley Eastern, Bangalore,2007*

PHC 240: Solid State Physics 1

Course Outcome:

After completing this course the student will

- CO1. Understand the Dielectrics properties of Solids;
- CO2. Understand Physics of Ferroelectric Properties of Solids
- CO3. Quantitatively estimate the Magnetic properties of solids
- CO4. Compute the magnetic relaxation time
- CO5. Understand BCS theory of Superconductors
- CO6. Understand the fundamentals of nanomaterials
- CO7. Understand the elastic properties of solids

Dielectric properties of solids: Macroscopic description of static dielectric constant, the static electronic and ionic polarisabilities of molecules, orientation polarization. Local electric field at an atom; Lorentz field, field of dipoles inside cavity. The static dielectric constant of solids; Clausius- Mossotti relation. Complex dielectric constant. Polarization catastrophe. Dielectric losses and Debye relaxation time. Classical theory of electronic polarization and optical absorption.

Ferroelectricity: Basic properties and classification of ferroelectric materials. The dipole theory of ferroelectricity, objections against the dipole theory. Ionic displacements and behavior of Barium titanate above the Curie temperature. Theory of spontaneous polarization of Barium titanate. Thermodynamics of ferroelectric transitions. Landau theory of phase transitions, Dielectric constant near the Curie point. Ferroelectric domain **(Dekker and Kittel)**. [16 hours]

Magnetic properties: Definition of magnetization and susceptibility. Hund's rule; calculation of L , S and J for $3d$ and $4f$ shells. Setting up of Hamiltonian for an atom in an external magnetic field; explanation of diamagnetism, Van Vleck Paramagnetism and quantum theory of paramagnetism **(Ashcroft & Mermin)**. Interpretation of the Weiss field in terms of exchange integral **(Dekker p473-474)**. Calculation of the singlet triplet splitting, spin Hamiltonian and Heisenberg model **(Ashcroft and Mermin)**.

Zero-temperature properties: Ground state of the Heisenberg ferromagnet. First excitation of one dimensional ferromagnetism at zero-temperature; spin waves, anti-ferromagnetism. Low-temperature behaviour of ferromagnets; Bloch's $T^{3/2}$ law **(Ashcroft and Mermin, Kittel)**.

Magnetic resonance: Phenomenological description, Relaxation mechanisms, Derivation of Casimir Dyrpe relation. Nuclear Magnetic moments, condition for resonance absorption, setting up of Bloch's equations, solutions for steady state and weak RF field. Expression for power absorption, change of inductance near resonance. Dipolar line width in a rigid lattice **(Dekker p498-512)**. [16 hours]

Band theory of solids: Statement and proof of Bloch theorem; periodic potentials in solids. Reciprocal lattice, periodic boundary conditions, density of states. Construction of Brillouin zones for a square lattice. Nearly free electron model and solution at the boundary. Energy gap using nearly free electron model. Tightly bound electron approximation, application to SC, BCC and FCC lattices **(Dekker)**.

Superconductivity: BCS theory; Cooper pairs, Energy gap, Meissner effect. Flux quantization. Theory for DC and AC bias; Josephson tunnelling, Josephson junction. High T_c superconductors **(Ibach and Luth)**.

Elastic constants of crystals: Elastic strains and stresses. Elastic compliance and stiffness constants, applications to cubic crystals and isotropic solids. Elastic waves and experimental determination of elastic constants **(Kittel, R S Krishnan, R J Asero)** [16 hours]

Total work load

48 hours

References:

1. Dekker A.J., Solid state physics, Prentice Hall, 1985.
2. Kittel C., Introduction to solid state physics, 7th Edn., John Wiley, New York, 1996.
3. Ashcroft N.W. and Mermin N.D., Solid State Physics, Saunders College Publishing, 1996.
4. Ibach H. and Luth H., Solid State Physics Narosa, New Delhi, 1996.
5. Pillai S.O., Solid state physics, New Age International Publications, 2002.
6. Wahab M.A., Solid state physics, Narosa Publishing House, New Delhi, 1999.
7. R S Krishnan, Progress in Crystal Physics, Vol 1, Central Art Presst, Chennai
8. R J Asero, Mechanics of Solids and Mateials, Uni. Of Caifornia,

PHC 260: Nuclear Physics 1

Course Outcome:

After completing this course the student will

- CO1. Conceptualise the Nuclear Detectors, Nuclear Pulse techniques and Nuclear models
- CO2. Develop Theoretical and practical experience in the scattering experiments and gamma-ray spectroscopy.
- CO3. Learn High-energy nuclear physics, the behaviour of nuclear matter under extreme conditions.
- CO4. Learn Low-energy nuclear physics, structure & dynamics of nuclei, nuclear reactions and their probabilities;
- CO5. Understand development of sensors, detectors, and larger complex Nuclear instruments

Nuclear detectors: Scintillation processes in inorganic crystals (NaI(Tl)). Semiconductor detector -Diffused junction, Surface barrier and Lithium drifted detectors. Relation between applied voltage and depletion layer thickness in junction detectors, Hyper pure germanium detectors, Cerenkov detectors.

Nuclear pulse techniques: Preamplifier circuits; charge sensitive and voltage sensitive preamplifiers.

Linear pulse amplifiers; Linearity, stability, pulse shaping, pulse stretching.

Operational amplifiers; analog to digital converters. Scalars, Schmidt trigger as a pulse discriminator, Single channel analyser; Integral and differential discriminators.

Multichannel Analysers, memory devices and online data processing.

[16 hours]

Shell model: Motion in a mean potential, Square well and simple harmonic oscillator potential well, spin orbit interaction and Magic numbers. Extreme single particle model, Ground state properties of nuclei based on shell model. Nordheim's Rules.

Collective model: Evidences for collective motion. Nuclear rotational motion; Rotational energy spectrum and nuclear wave functions for even-even nuclei. Odd- A nuclei energy spectrum and wave function.

Nilsson model: Nilsson diagrams.

Many body self-consistent models: Hartree-Fock model. (Hans H.S)

[16 hours]

Timing spectroscopy: Coincidence and anti-coincidence circuits. Delay circuits. Time to amplitude conversion; start-stop and overlap converters.

Gamma ray spectroscopy: Life time measurements. Gamma-gamma, beta-gamma angular correlation studies. Angular distribution of gamma rays from oriented nuclei. Polarization of gamma rays.

[16 hours]

Total work load

48 hours

References:

1. Mermier P. and Sheldon E., *Physics of the nuclei and particles*, Vol. 1 and 2, Academic Press, New York 1970.
2. Segre E., *Nuclei and particles*, Benjamin Inc, New York, 1977.
3. Arya A.P., *Fundamentals of nuclear physics*, Allyn and Bacon, USA, 1968.
4. Blatt J.M. and Weisskopf V.F., *Theoretical nuclear physics*, Wiley and Sons, New York, 1991.
5. Siegbahn K., *The alpha, beta and gamma ray spectroscopy: Vol. 1 and 2*, North Holland, Amsterdam, 1965.
6. Price J.W., *Nuclear radiation detectors*, McGraw Hill, New York, 1965.
7. Kapoor S.S. and Ramamoorthy V., *Nuclear radiation detectors*, Wiley Eastern, Bangalore, 1993.
8. Kowalski E., *Nuclear electronics*, Springer Verlag, Berlin, 1970.
9. Leo W.R., *Techniques for nuclear and particle physics experiments*, Springer Verlag, 1992.
10. Roy R.R. and Nigam B.P., *Nuclear physics*, New Age International, New Delhi, 1986.
11. Hans H.S., *Nuclear physics—Experimental and theoretical*, New Age International Publishers, 2001.
12. Tayal D.C., *Nuclear Physics*, Himalaya Publishing House, New Delhi, 2012

PHY-306: Theoretical Physics 1

Course Outcome:

After completing this course the student will

- CO1. Tackle a wide range of topics using powerful analytical tools including formal methods in

classical and quantum physics

- CO2. *Clearly communicate information & conclusions in written and verbal form of the ideas in Theoretical Physics*
- CO3. *Evaluate complex problems & formulate solutions, indentifying the role of theory, hypothesis and experiment in the scientific method*
- CO4. *Apply computation to solve the problems in theoretical physics*
- CO5. *Plan, carry out and report theoretical physics based investigation*
- CO6. *Apply classical and quantum theoretical techniques in research*

General theory of relativity: *Tensor Calculus and Riemannian geometry: Covariant Differentiation, Parallel Transport, Geodesies, The Curvature Tensor.*

Riemannian geometry: *Riemannian space, The determinant of $g_{\mu\nu}$, Metrical Densities, The Connection of a Riemannian Space: Christoffel Symbols, Geodesies in a Riemannian Space, The Curvature of a Riemannian Space: The Riemann Tensor.* **[16 hours]**

Gravitational field: *The Principle of Equivalence, The Field Equations of General Relativity, Metrics with Spherical Symmetry, The Schwarzschild Solution. Geodesies in the Schwarzschild Space, Advance of the Perihelion of a Planet, The Deflection of Light Rays, Red Shift of Spectral Lines, The Schwarzschild Sphere. Gravitational Collapse. Black Holes.* **[16 hours]**

Quantum field theory-1: *Classical and quantum fields: Particles and fields, Discrete and continuous mechanical systems, Classical scalar fields, Maxwell fields Quantum Theory of Radiation: Creation, annihilation, and number operators, Quantized radiation field, Fock states, Emission and absorption of photons by atoms, Rayleigh scattering, Thomson scattering, and the Raman effect.* **[16 hours]**

Total work load

48 hours

References:

1. Papapetrou A., *Lectures on general relativity*, D. Reidel Publishing Company, USA, 1974.
2. Dirac P.A.M., *The general theory of relativity*, John Wiley and Sons, New York, 1975.
3. Adler R., Bazin M. and Schiffer M., *Introduction to general relativity*, McGraw-Hill Kogakusha, Ltd. New Delhi, 1965.
4. Hartle J.B., *Gravity: An introduction to Einstein's general relativity*, Benjamin-Cummings Pub. Co., USA, 2002.
5. Sakurai J.J., *Advanced quantum mechanics*, Addison-Wesley, Harlow, England, First ISE Reprint, 1999.
6. Griffiths D., *Introduction to elementary particles*, John Wiley and Sons, New York, 1987.
7. Gasiorowicz S., *Elementary particle physics*, John-Wiley, New York, 1966.
8. Muirhead H., *The physics of elementary particles*, Pergamon Press, London, 1965.

Open Elective Papers

Paper to be offered to Non-Physics Postgraduate students

PHY-321: Modern Physics

Course Outcome:

After completing this course the student will

- CO1. Apply basic principles and laws of electricity and magnetism
- CO2. Solve Problems involving the fundamental principles of physics
- CO3. Apply Mathematical techniques for quantitative solutions to problems
- CO4. Understand the fundamentals of Condensed Matter, Nuclear and Quantum Physics

Nuclear physics: A brief overview of nuclear physics. Nuclear reactions, a brief description of nuclear models. Interactions of X-rays and γ -rays with matter, slowing down and absorption of neutrons. Fundamental particles, classification of fundamental particles, fundamental forces, conservation laws in particle physics, a brief outline of the quark model.

Nuclear power: Nuclear fission, fission chain reaction, self sustaining reaction, uncontrolled reaction, nuclear bomb. Nuclear reactors, different types of reactors and reactors in India. Nuclear waste management. Nuclear fusion, fusion reactions in the atmosphere. Radiation effects; dosage calculation. Nuclear energy; applications and disadvantages. **[16 hours]**

Condensed matter physics: Amorphous and crystalline state of matter. Crystal systems. Liquid crystals. X-ray diffraction; Bragg equation. Structure of NaCl. FTIR; Experiment analysis. NMR; Experiment and analysis. Electrical conductivity of metals and semiconductor. Magnetic materials; para,ferro, ferri and anti-magnetism. Dielectrics— para, ferro, pyro and piezo properties. Symmetry in physics. **[16 hours]**

Quantum physics: Qualitative discussion. Molecules, atoms, nucleus, nucleons, quarks and gluons. Particle physics (qualitative). Stern-Gerlach experiment and consequences. Uncertainty relation. Hydrogen atom. Positron annihilation. Laser trapping and cooling. Ion traps. Electromagnetic, strong, weak and Gravitational forces. Big Bang theory, String theory. LHC experiment, consequences. Higgs Boson. **[16 hours]**

Tutorial

[32 hours]

References:

1. Ghoshal S.N., Atomic and nuclear physics, Vol.2., S. Chand and Company, Delhi, 1994.
2. Evans R.D., Atomic nucleus, Tata Mc Grow Hill, New Delhi, 1976.
3. Penrose R., Road to Reality, Vintage Books, 2007.
4. Ladd M.F.C. and Palmer R.A., Structure determination by X-ray crystallography, Plenum Press, USA, 2003.
5. De Gennes P.G. and Prost J., The physics of liquid crystals, 2nd Edn., Clarendon Press, Oxford, 1998.
6. Myer R., Kennard E.H. and Lauritsern T., Introduction to modern physics, 5th Edn., McGraw- Hill, New York, 1955.
7. Halliday D., Resnick R. and Merrill J., Fundamentals of physics, Extended 3rd Edn., John Wiley, New York, 1988.

PHY-322: Energy Science

Course Outcome:

After completing this course the student will

- C01. Learn the need for renewable energy in the growing world
 C02. Understand the conservation of renewable energy resources
 C03. Understand the proper utilisation of renewable energy resources
 C04. Understand the physics of renewable energy resources
 C05. Learn the technology behind the biogas and biomass production

Renewable energy resources: Forms of Energy, Basics of Thermodynamics: Heat capacity, Heat transfer mechanism, entropy, First and second law of thermodynamics Carnot Cycle, Rankin cycle. Fossil fuels, time scale of fossil fuels. Solar energy: Sun as the source of energy and its energy transport to the earth, Extraterrestrial and terrestrial solar radiations, Measurement techniques of solar radiations using Pyranometer and Pyrliometer.

[16 hours]

Materials and solar cell technology : Single, poly and amorphous silicon, GaAs, CdS, fabrication of single and polycrystalline silicon solar cells, amorphous silicon solar cells, photovoltaic systems and technical problems. Wind Energy Origin and classification of winds, Aerodynamics of windmill: Maximum power and Forces on the Blades and thrust on turbines; Wind data collection and field estimation of wind energy, Site selection, Basic components of wind mill, Types of wind mill, Wind energy farm, Hybrid wind energy systems: The present Indian Scenario.

[16 hours]

Biomass energy and biogas technology: Nature of Biomass as a fuel, Biomass energy conversion processes, Direct combustion: heat of combustion, combustion with improved Chulha and cyclone furnace; Dry chemical conversion processes: pyrolysis, gasification, types of gasification. Importance of biogas technology, anaerobic decomposition of biodegradable materials, Factors affecting Bio-digestion, Types of biogas plants, Applications of biogas.

[16 hours]

Tutorial

[32 hours]

References:

1. Peter A., *Advances in energy systems and technology*, Academic Press, USA, 1986.
2. Neville C.R., *Solar energy conversion: The solar cell*, Elsevier North-Holland, 1978.
3. Dixon A.E. and Leslie J.D., *Solar energy conversion*, Pergamon Press, New York, 1979.
4. Ravindranath N.H., *Biomass, energy and environment*, Oxford University Press, 1995.
5. Cushion E., Whiteman A. and Dieterle G., *World Bank Report*, 2009.

PHC 080: Condensed Matter Physics Lab

Course Outcome:

After completing this course the student will

- C01. Independently collect single crystal XRD data and evaluate the crystal structure,
 C02. Analyse Diffraction data and their suitability for single crystal structure.
 C03. Independently work on raw diffraction data and Solve and refine crystal data
 C04. Understand the properties of Condensed Materials such as Structural, Magnetic, Electrical and Semiconducting Properties
 C05. Acquire theoretical and practical experience in the Condensed Matter Physics

Any eight of the following experiments:

1. Determination of the paramagnetic susceptibility of the given salt by Quincke's method
2. Study of mercury spectrum by superimposing it on brass spectrum
3. Sodium spectrum analysis by using Edser-Butler fringes
4. Temperature coefficient of resistance of a thermistor
5. Analysis of the powder X-ray photograph of a simple cubic crystal
6. Thermionic work function of a metal (Richardson-Dushman formula)
7. Energy gap of a semiconductor
8. Frank Hertz experiment
9. Measurement of magneto resistance of semiconductors
10. Stefan's Constant of Radiation
11. Thermal Conductivity of Poor Conductor
12. Di-electric constant of a Non polar liquid
13. Dipole moment of an organic Molecule
14. High Resistance by Leakage

Total work load : 2 day(s) per week \times 4 hours \times 16 weeks = **128 hours**

PHC 090: Nuclear and Particle Physics Lab

Course Outcome:

After completing this course the student will

- CO1. Be familiar with theoretical concepts & experimental techniques of Nuclear & particle physics,.
- CO2. Be able to make quantitative estimates of phenomena in elementary particles
- CO3. Understand stability of the nucleus, nuclear forces, interactions & models, Critically assess a range of applications of nuclear technology.
- CO4. Acquire Theoretical & practical experience in the scattering experiments and gamma-ray spectroscopy.
- CO5. Be able to realise the Nuclear Electronic circuits

Any eight of the following experiments:

1. Half-life of Indium-116 measurement.
2. Energy Resolution of a NaI(Tl) scintillation spectrometer.
3. Compton scattering—determination of the rest energy of an electron.
4. Beta absorption coefficient measurement.
5. Dekatron as a counter of signals.
6. Gamma-ray absorption coefficient measurement.
7. End-point energy of Beta particles by half thickness measurement.
8. Common Source amplifier.
9. Astable multivibrator using timer IC 555.
10. Dead time of the G.M. counter.

Total work load : 2 day(s) per week \times 4 hours \times 16 weeks = **128 hours**

Reference: 1. Varier K. M., Antony Joseph and Pradyumnan P. P., *Advanced experimental techniques in Modern Physics*, Pragati Prakashan, 2011

PHC 250: Solid State Physics Lab 1

For those who have opted for **Solid State Physics Specialisation**

Course Outcome:

After completing this course the student will

- CO1. Be able to make quantitative estimates for phenomena in solid state physics.
- CO2. Describe thermal & vibrational properties of solids,
- CO3. Compute band structures using the tight-binding approximation
- CO4. Compute trajectories in real and reciprocal space, Characterize magnetism
- CO5. Characterize intrinsic and doped semiconductors
- CO6. Describe superconductivity
- CO7. Understand the mechanism and working of semiconductor devices.

Any five of the following experiments:

1. Optical rotatory dispersion of a uniaxial crystal.
2. Birefringence of quartz using spectrometer.
3. Paramagnetic susceptibility by Gouy balance method.
4. Fermi energy of copper.
5. Cell parameter(s) from an X-ray powder diffractogram.
6. Verification of Langmuir-Child's law.
7. Curie temperature of a ferroelectric material.
8. Dielectric constant and its temperature variation.
9. Determination of the polarisabilities of the molecules of an uniaxial crystal using spectrometer.
10. Determination of Stefan's constant using Photo Cell
11. Calibration of Si Diode
12. Measurement of Electrical and Thermal Conductivity of Copper
13. Verification of Curie-Weiss law
14. BH Curve in a ferromagnetic Material

Total work load : 1 day(s) per week \times 4 hours \times 16 weeks = **64 hours**

PHC 270: Nuclear Physics Lab 1

For those who have opted for **Nuclear Physics Specialisation**

Course Outcome:

After completing this course the student will

- CO1. *Be familiar with Nuclear Detectors, Pulse techniques, nuclear models, timing spectroscopy and nuclear technology.*
- CO2. *Be familiar with Theoretical & practical experience in the scattering experiments and gamma-ray spectroscopy.*
- CO3. *Learn Instrumentation, development of sensors, detectors, & larger complex instruments.*
- CO4. *Familiarise in Fundamentals of reactor theory and nuclear decay mechanisms*
- CO5. *Learn Accelerator physics, research and development for the next generation of particle accelerators.*

Any five of the following experiments:

1. *Cockroft-Walton voltage multiplier.*
2. *Coincidence circuit.*
3. *Linear amplifier.*
4. *Transistorised binary circuit.*
5. *Pulse shaping circuits.*
6. *Linear Gate.*
7. *Randomicity of radioactive decay.*
8. *Nomogram method : Measurement of endpoint energy of beta rays.*
9. *Study of linearity of the NaI(Tl) gamma ray spectrometer.*
10. *Determination of the energy of an unknown gamma ray source.*

Total work load : 1 day(s) per week × 4 hours × 16 weeks = 64 hours

PHY-315: Theoretical Physics Lab 1

For those who have opted for **Theoretical Physics Specialisation**

Course Outcome:

After completing this course the student will

- CO1. *Tackle a wide range of topics using powerful analytical tools including formal methods in classical and quantum physics*
- CO2. *Clearly communicate information and conclusions in written and verbal formats on ideas in Theoretical Physics*
- CO3. *Evaluate complex problems and formulate solutions, indentifying the role of theory, hypothesis and experiment in the scientific method*
- CO4. *Apply computation to solve the problems in theoretical physics*
- CO5. *Plan, carry out and report theoretical physics based investigation*
- CO6. *Apply classical and quantum theoretical techniques in research*

Any five of the following experiments:

1. *Calculation of Christoffel symbols.*
2. *Geodesics and curvature calculations.*
3. *Exterior Schwarzschild metric calculations.*
4. *Robertson-Walker metric calculations.*
5. *Lagrangian and Hamiltonian, Euler Lagrange equations for Schroedinger field.*
6. *Lagrangian for Maxwell's field and The field equations.*
7. *Symmetries of the Lagrangian and Constants of motion.*
8. *Operator algebra-BCH formula.*
9. *Relativistic kinematics-1: Relations between center of momentum and laboratory frames.*
10. *Relativistic kinematics-2: Non-relativistic limit of relativistic kinematics.*

Total work load : 1 day(s) per week × 4 hours × 16 weeks = 64 hours

PHD 250: Solid State Physics 2

Course Outcome:

After completing this course the student will

- CO1. Gain an overall idea about X-ray diffraction method Solid State Physics
 CO2. Get the understanding about X-ray diffraction (XRD) by Crystals.
 CO3. Understand the Physical phenomena and significance of XRD
 CO4. Be able to make quantitative estimates for structural phenomena of solids.
 CO5. Describe thermal and vibrational properties of solids
 CO6. Understand the concepts of Dislocations, Imperfections and Defects in Solids
 CO7. Appreciate the Luminescent effects and colour centres in ionic crystals

X-ray diffraction by crystals: The reciprocal lattice. Ewald sphere and construction. Scattering by an electron and atom; Atomic scattering factor. Anomalous scattering. Fourier analysis and inversion of Fourier series; Physical significance. Geometrical structure factor of the unit cell. Absent reflections and space groups. (Sherwood, P290 – 358).

Experimental techniques: Brief introduction to Laue, Powder and single crystal methods. Use of Synchrotron radiation for structure studies. Weissenberg and precession methods. Cell parameter and space group determination. Molecular weight determination. (Stout and Jensen, p 90–211). [16hours]

Structure analysis: Low angle scattering. Reduction of intensities to structure amplitudes. Various corrections. Absolute scale factor and temperature factor from statistical methods. Statistical method for finding the presence of center of symmetry Fourier analysis of electron density. Patterson synthesis. Harker sections and lines. Heavy atom methods. Direct methods for phase determination. The inequality relations. Difference Patterson synthesis and error Fourier synthesis. Figure of merit. Cyclic Fourier refinement, Difference Fourier synthesis. Refinement of structures: The least squares method. Accuracy of the parameters. Bond lengths and angles. (Sherwood, Ladd and Palmer)

SAXS; Particle Size study of Fibre structure [16 hours]

Imperfections in solids: Different types of imperfections. Schottky and Frenkel defects; expression for energy for the formation of Frenkel and Schottky defects. Diffusion in metals; Kirkendall effect. Ionic conductivity in pure and doped halides. Photoconductivity (Kittel).

Dislocations: Burger's Vector. Expression for strain in edge and screw dislocations (Wahab and Kittel).

Synthesis and Device fabrication of Nanomaterials: Nanomaterials. Bottom-Up approach; Sol-gel synthesis, hydrothermal growth, thin-film growth, physical vapor deposition, chemical vapor deposition. Top-Down Approach; Ball milling, Microfabrication, Lithography, Ion-beam lithography (Ramachandra rao and Shubra singh, p129-142).

Luminescence: Excitation and Emission. Franck-Condon principle. Decay mechanisms; Temperature dependent and independent decays. Thermoluminescence and glow curve. Gudden-Pohl effect (Dekker). [16 hours]

Total work load 48 hours

References:

1. Stout G.H. and Jensen L.H., X-ray structure determination, MacMillan, USA, 1989.
2. Ladd M.F.C. and Palmer R.A., Structure determination by X-ray crystallography, Plenum Press, USA, 2003.
3. Sherwood D., Crystals, X-rays and proteins, Longman, London, 1976.
4. Wahab M.A., Solid state physics, Narosa Publishing House, New Delhi, 1999.
5. Azaroff L.V., Introduction to solids, McGraw-Hill Inc, USA, 1960.
6. Weertman J. and Weertmann J.R., Elementary dislocation theory, McMillan, USA, 1964.
7. Pillai S.O., Solid state physics, New Age International Publications, 2002.

PHD 260: Solid State Physics 3

Course Outcome:

After completing this course the student will

- CO1. Have overall idea about properties of Solids.

- CO2. Be provided with the understanding about free electron theory of metals
 CO3. Understand the properties of impurity semiconductors.
 CO4. Learn Semiconductor phenomena; Hall effect, Magneto-resistance phenomenon
 CO5. Be able to make quantitative estimates of semiconducting phenomena of solids.
 CO6. Describe the effect of excess carriers in semiconducting solids
 CO7. Understand the mechanism and working of semiconductor devices.

Free electron theory of metals: Boltzmann transport equation, Sommerfeld's theory of electrical conductivity, mean free path in metals, dependence of resistivity on temperature and impurities. Matthiessens rule. Electron-phonon collisions. Electrical conductivity of metals at high frequencies. Plasma frequency. Transparency of alkali metals to UV radiation. Anomalous skin effect. Plasmons. Field enhanced emission, Schottky effect. Hall effect and magnetoresistance in metals. Cyclotron frequency (**Kittel & Pillai**). Thermal conductivity of insulators; Umklapp processes (**Dekker, p275-292**). **[16 hours]**

Impurity semiconductors: A brief discussion on Elemental and Compound Semiconductors and their properties. Carrier concentrations; effect of temperature and impurity density. Electrical neutrality condition. Fermi energy; Variation with temperature and impurity density, when the Boltzmann approximation is valid,. Effect of impurity density at very low temperatures. Mobility of current carriers; effect of temperature and impurity. Electrical conductivity; effect of temperature, impurity density and the energy band gap.

Hall effect in semiconductors; Expression for Hall co-efficient,
Magneto-resistance phenomenon (qualitative) (M A Wahab).

Cyclotron resonance;Cyclotron resonance in Si and Ge semiconductors. Effective mass tensor. Variation of cyclotron resonance frequency with orientation of the crystal in the magnetic field (**Mckelvey, p270-300**). **[16 hours]**

Excess carriers in semiconductors: Generation and recombination rates. Continuity equations; Einstein equations, Expression for the diffusion length of electrons and holes (**Mckelvey, p320-335**). High field transport in semiconductors; electron temperature. Gunn effect, Expression for drift velocity. Superlattice Phenomenon (**Roy, p29-39**).

Semiconductor devices: The pn junction; space charge region, effect of the applied field on barrier potential, barrier thickness and contact field. Transition capacitance. Current density for excess carriers. Characteristics and applications of phototransistors, JFET, SCR and UJT(**Mckelvey, p390-441**). **[16 hours]**

Total work load

48 hours

References:

1. Dekker A.J., Solid state physics, Prentice Hall, 1985.
2. Mckelvey J.P., Solid state and semiconductor physics, 2nd Edn., Harper and Row, USA, 1966.
3. Roy D.K., Physics of semiconductor devices, University Press, Hyderabad, 1992.
4. Schur M., Physics of semiconductor devices, Prentice-Hall of India, New Delhi, 1999.
5. Wilson J. and Hawkes J.F.B., Optoelectronics—An introduction, 2nd Edn., Prentice-Hall of India, New Delhi, 1996.
6. Streetman B.G., Solid state electronic devices, 2nd Edn., Prentice-Hall of India, New Delhi, 1983.
7. Omar M.A., Elementary solid state physics, Addison Wesley, New Delhi, 2000.
8. Wahab M.A., Solid state physics, Narosa Publishing House, New Delhi, 1999.
9. Pillai S. O. Solid State Physics, ew Age International Publications, New Delhi.

PHD 300: Nuclear Physics 2

Course Outcome:

After completing this course the student will

- CO1. Understand the phenomenon of nuclear fission and its application in energy production.
 CO2. Gain an overview on the neutron physics and nuclear reactor theory.
 CO3. Develop the understanding of quantum theory beta decay.

CO4. Learn about the multipole transition involving gamma decay and internal conversion.

Nuclear fission: Nuclear fission, Mass-energy distribution of fission fragments. Statistical model of fission.

Reactor theory-1: Neutron and its interaction with matter-collision kinematics, differential elastic scattering cross sections, isotropic scattering, the criticality condition for a reactor. Neutron transport equation using elementary diffusion theory. One group critical equation, critical size on the basis of Fermi age theory.

[16 hours]

Reactor theory-2: Reactors; One group theory, spherical and cylindrical homogeneous reactor. Effective multiplication factor. Reflector reactors: effects of reflector. One group method of a homogeneous reactor with reflector. reflector savings. Infinite multiplication factor, critical size and critical mass.

Heterogeneous reactor system; calculation of thermal utilization factor. Fast Breeder reactor, Evaluation of Buckling using one group model.

[16 hours]

Beta decay: Classification of beta interactions. Matrix elements. Fermi and Gamow-Teller selection rules for allowed beta decay. The non conservation of parity in beta decay. Wu et al experiment. The universal Fermi interaction.

Gamma decay: Electromagnetic interactions with nuclei. Multipole transitions. Transition probabilities in nuclear matter. Weisskopf's estimates. Structure effects. Selection rules. Internal conversion Photo disintegration of deuteron and radiative capture of neutron by proton.

[16 hours]

Total work load

48 hours

References:

1. Glasstone S.& Edlund M.C., Elements of nuclear reactor theory, D. Van Nostrand Co., USA, 9th Print, 1963.
2. Garg S., Ahmed F. and Kothari I.S., Physics of nuclear reactors, Tata McGraw-Hill, New Delhi, 1986.
3. Roy R.R. and Nigam B.P., Nuclear physics, New Age International, New Delhi, 1986.
4. Hans H.S., Nuclear physics—Experimental and theoretical, New Age International Publishers, 2001.
5. Ghoshal S.N., Nuclear physics, Vol. 2., S.Chand and Company, Delhi, 1994. Chapter 15, page 714-730.

PHD 310: Nuclear Physics 3

Course Outcome:

After completing this course the student will

- CO1. Know about the two particle nuclear bound system like deuteron.
- CO2. Learn the theory of nucleon-nucleon scattering processes.
- CO3. Get an understanding of plane wave theory of nuclear reactions.
- CO4. Acquire knowledge on optical model and heavy ion physics.

Two particle systems: Deuteron; Schrodinger equation for a two nucleon system, Theory of the ground state of the deuteron under central and non central forces, Excited states of the deuteron. Rarita-Schwinger relations. Deuteron magnetic and Quadrupole moments.

Nucleon-nucleon scattering processes: Theory of s-wave scattering of neutrons by free protons and experimental results. Wigner's formula for n-p scattering. Theory of scattering of slow neutrons by bound protons (Ortho and Para hydrogen) and experimental results. Effective range theory for n-p scattering. S wave theory of proton-proton scattering. Mott's modification of Rutherford's formula. Pion-nucleon scattering experimental results, (3/2, 3/2) resonance. **[16 hours]**

Nuclear reactions-1: Plane wave theory of direct reactions. Born approximation (Plane wave); Butler's theory. Cross section for nuclear scattering and reactions. Shadow scattering, Breit-Wigner resonance formulae.

Nuclear reactions-2: Bohr's independence hypothesis. The compound nucleus (CN) reactions, decay rates of CN, Statistical theory of nuclear reactions. Evaporation probability and cross sections for specific reactions. **[16 hours]**

Optical model: Giant resonances, Kapur-Pearls' dispersion formula for potential scattering. Direct reactions: Kinematics of stripping and pickup reactions. Theory of stripping and pickup reactions. Inverse reactions.

Heavy ion physics: Special features of heavy ion Physics. Remote heavy ion electromagnetic interactions. Coulomb excitations. Close encounters.

Heavy ion scattering. Grazing interactions. Particle transfer. Direct and head on collisions, compound nucleus and quasi molecule formation. **[16 hours]**

Total work load

48 hours

References:

1. Roy R.R. and Nigam B.P., Nuclear physics—Theory and experiment, New Age International Ltd, New Delhi, 1986.
2. Hans H.S., Nuclear physics—Experimental and theoretical, New Age International Publishers 2001.
3. Sachtler G.R., Nuclear reactions, Addison Wesley, New York, 1983.
4. Mermier P. and Sheldon E., Physics of nuclei and particles, Vol. 2 Academic Press, USA, 1971.
5. Jackson D.F., Nuclear reactions, Chapman and Hall, London, 1975
6. Mermier P. and Sheldon E., Physics of nuclei and particles, Vol. 3 Academic Press, USA, 1971.

PHY-405: Theoretical Physics 2

Course Outcome:

After completing this course the student will

- CO1. Tackle a wide range of topics using powerful analytical tools including formal methods in classical and quantum physics
- CO2. Clearly communicate information and conclusions in written and verbal formats on ideas in Theoretical Physics

- CO3. Evaluate complex problems and formulate solutions, indentifying the role of theory, hypothesis and experiment in the scientific method
- CO4. Apply computation to solve the problems in theoretical physics
- CO5. Plan, carry out and report theoretical physics based investigation
- CO6. Apply classical and quantum theoretical techniques in research

Relativistic quantum mechanics: Probability conservation in relativistic quantum mechanics, The Dirac equation, Conserved current, Representation independence, large and small components, approximate Hamiltonian for an electrostatic problem, free particle solutions, Relativistic covariance, Space inversion, Bilinear covariants and their properties, Klein's paradox, Hole theory and charge conjugation. **[16 hours]**

Quantization of the Dirac field: Second quantization, positron operators and positron spinors, Electromagnetic and Yukawa couplings. Weak interactions and parity nonconservation: Classification of interactions, parity and hyperon decay, Fermi theory of beta decay, the two-component neutrino. Pion decay and the CPT theorem. **[16 hours]**

Covariant perturbation theory: Natural units and dimensions, S-matrix expansion in the Interaction representation. Unitarity, First order processes: Matrix element for electron scattering. Cross section for Mott scattering. Helicity change and spin projection operator. Pair annihilation, pair creation, hyperon decay. S -matrix for two photon annihilation, electron propagator, Matrix element for Compton scattering, Feynman rules. Cross section for two photon annihilation. **[16 hours]**

Total work load

48hours

References:

1. Sakurai J.J., *Advanced quantum mechanics*, Addison-Wesley, Harlow, England, First ISE Reprint, 1999.
2. Griffiths D., *Introduction to elementary particles*, John Wiley and Sons, New York, 1987.
3. Gasiorowicz S., *Elementary particle physics*, John-Wiley, New York, 1966.
4. Muirhead H., *The physics of elementary particles*, Pergamon Press, London, 1965.

PHY-406: Theoretical Physics 3**Course Outcome:**

After completing this course the student will

- CO1. *Tackle a wide range of topics using powerful analytical tools including formal methods in classical and quantum physics*
- CO2. *Clearly communicate information and conclusions in written and verbal formats on ideas in Theoretical Physics*
- CO3. *Evaluate complex problems and formulate solutions, indentifying the role of theory, hypothesis and experiment in the scientific method*
- CO4. *Apply computation to solve the problems in theoretical physics*
- CO5. *Plan, carry out and report theoretical physics based investigation*
- CO6. *Apply classical and quantum theoretical techniques in research*

Angular momentum theory and applications: *Angular momentum: Transformations under rotations. Coupling of three and four angular momenta. Racah coefficients, Wigner 9j symbols, applications. Wigner-Eckart theorem. Projection theorem. j-j and L-S coupling. Angular momentum in nuclear reactions, Spherical tensors. Evaluation of matrix elements between coupled angular momentum states. Vector spherical harmonics. Gradient theorem (without proof). Multipole radiation.* **[16 hours]**

Spin density matrix: *Spin and helicity in a relativistic process. Effect of Lorentz and discrete transformations on helicity states. Wick and Wigner rotations, pure rotation, pure boost, parity, time reversal and charge conjugation. The spin density matrix (ρ), general properties, multipole parameters, combined systems, Diagonalization of ρ . Oriented and non-oriented systems, Polarized and aligned systems, Spherical tensor basis and $SU(N)$ basis.* **[16 hours]**

Relativistic density matrix: *Helicity multipole parameters and their transformation laws. Helicity amplitudes for elastic reactions and their symmetry properties. Polarization in scattering of spin $\frac{1}{2}$ particles, Final state density matrix. Observables of a reaction, reactions involving polarized beam and polarized targets.* **[16 hours]**

Total work load

48 hours

References:

1. Sakurai J.J. and Tuan S.F. (Editor), *Modern quantum mechanics*, AddisonWesley, India, 1999.
2. Leader E., *Spin in particle physics*, Cambridge University Press, London, 2001.
3. Rose M.E., *Elementary theory of angular momentum*, John Wiley and Sons, USA, 1957.
4. Blum K., *Density matrix theory and applications*, Plenum Press, New York, 1981.

Elective Papers 1**PHD 270: Accelerator Physics****Course Outcome:**

After completing this course the student will

- CO1. Know about different types of ion sources and their working.
- CO2. Learn ion optics and focussing.
- CO3. Know about different particle accelerators.
- CO4. Get theory behind electron accelerator.

Ion sources: Brief introduction to ion sources for positive and negative ions. Ion production. Semi classical treatment of ionization, Townsend theory-comparison of theory and experiment for ion production. Examples of ion sources-properties of ion sources. Insulation at high voltages-Spark voltage. Paschen's law for gas breakdown.

Ion optics and focussing: Focussing properties of linear fields. Electrostatic and magnetic lenses. [16 hours]

Particle accelerators: Introduction, development of accelerators. Direct-voltage accelerators: Cockroft-Walton generator, Van de Graff generator, Tandem accelerators, Pelletron. Resonance accelerators: Cyclotron - fixed and variable energy, principles and longitudinal dynamics of the uniform field cyclotron. Linear accelerators.

[16 hours]

Electron accelerators: Betatron; Beam focusing and Betatron Oscillation. Microtron. Synchronous accelerators; Principle of phase stability, Mathematical theory for Principle of phase stability. Electron synchrotron. Proton synchrotron.

Alternating gradient machines; Alternating gradient principle, AG proton synchrotron.

[16 hours]

Total work load

48 hours

References:

1. Townsend P.D., Kelly J.C. and Hartley N.E.W., Ion implantation, sputtering and their applications, Academic Press, London, 1976.
2. Humphrey S. Jr., Principles of charged particle acceleration, John Wiley, 1986.
3. Arya A.P., Fundamentals of nuclear physics, Allyn and Bacon, USA, 1968.
4. Ghoshal S.N., Atomic and nuclear physics, Vol. 2, S.Chand and Company, Delhi, 1994.
5. Varier K.M., Joseph A. and Pradyumnan P.P., Advanced experimental techniques in modern physics, Pragathi Prakashan, Meerut, 2006.

PHY-408: Liquid Crystals**Course Outcome:**

After completing this course the student will

- CO1. Be familiar with the fundamentals of anisotropic fluids.
- CO2. Understand the principles of Long and short range order in nematics.
- CO3. Learn static distortion in nematics, defects and textures in nematics
- CO4. Know about Dynamical properties of liquid crystals and nematics.
- CO5. Study Optical properties of Cholesterics.

Anisotropic fluids: Main Types and properties: Introduction. The building blocks. Small organic molecules. Long helical rods. Associated structures. Nematics and Cholesterics. Nematics proper. Static pretransitional effects above T_{N-1}^i . The cholesterics. A distorted form of the nematic phase. Smectic. Smectic A. Smectic B. Smectic C. Other mesomorphic phases. Exotic smectics; long range order in a system of long rods. Lyotropic systems. Remarkable features of liquid crystals. Applications of liquid crystals. [De Gennes and Prost] [16 hours]

Long and short range order in nematics: Definition of an order parameter. Microscopic approach. Order parameter from optical method, from diamagnetic anisotropy. Mean field theory with S2 interaction (Maier-Saupe).

Static distortion in nematics: Long range distortions, distortion free energy. Magnetic field effects—Molecular diamagnetism, Magnetic coherence length.

Defects and textures in nematics: Observations. Black filaments. Schlieren structures. Types of defects (qualitative discussion only).

Smectics: Continuum description of smectics A and C, Mean field description of S_A -N transition.

[De Gennes and Prost]

[16 hours]

Dynamical properties of nematics: Experiments measuring the Leslie coefficients-Laminar flow under a strong orienting field, Attenuation of ultrasonic shear waves, Laminar flow in the absence of external fields. Convective instabilities under electric fields. Basic electrical parameters, Experimental observations at low frequencies, The Helfrich interpretation. Extension to higher frequencies (qualitative).

Cholesterics: Optical properties of an ideal helix—The planar texture, Bragg reflection, Transmission properties at arbitrary wavelengths (normal incidence), The Mauguin limit, Rotatory Power. Agents influencing the pitch—Physicochemical factors, External fields (qualitative). Textures in cholesterics. [De Gennes and Prost]

[16 hours]

Total work load

48 hours

References:

1. De Gennes P.G. and Prost J., *The physics of liquid crystals*, 2nd Edn., Clarendon Press, Oxford, 1998.
2. Chandrashekhar S., *Liquid crystals*, Cambridge University Press, 1977.
3. Gray G.W., *Molecular structure and the properties of liquid crystals*, Academic Press, 1962.
4. Maier G., Sackmann E. and Grabmanier I.G., *Applications of liquid crystals*, Springer Verlag, 1975.
5. Gray G.W. and Goodby J.W., *Smectic liquid crystals (Textures and structures)*, Leonard Hill, London, 1984.

PHY-409: Atmospheric Physics

Course Outcome:

After completing this course the student will

- CO1. Understand the composition of atmosphere and its different layers.
- CO2. Get on the thermodynamics of the atmosphere.
- CO3. Learn on terrestrial radiations and its effects on atmosphere.
- CO4. Understand on aerosols, clouds and atmospheric radioactivity.
- CO5. Understand working of atmospheric electricity.

Atmospheric composition: Energy in the atmosphere, heating of the atmosphere, motions in the atmosphere. Variations in atmospheric composition, Structure on the basis of composition. Thermal structure of the atmosphere.

Thermodynamics: Entropy of dry air, vertical motion of saturated air, tephigram, potential energy of an air column.

Dynamics: Escape of hydrogen, photodissociation of oxygen, photo chemical processes. Equations of motion, the geostrophic approximation, cyclostrophic motion. [16 hours]

Terrestrial and extra terrestrial radiation: General features of direct, diffuse and global radiation-attenuation of direct solar radiation-Rayleigh and Mie scattering. Angstrom turbidity formula for all aerosols. Direct transmittance due to continuum attenuation, diffuse spectral irradiance due to Rayleigh and aerosol scattering.

Aerosols: Production and properties of aerosols. Aerosol optical depth, Beer's law - Sun Photometer. Optical filters.

Clouds: Microphysics of clouds, Macro characterization of clouds. Radiative transfer in clouds and aerosols.

[16 hours]

Atmospheric radioactivity: Background Radiation, Radioactivity in Atmosphere, Radon, Properties of radon, Origin of radon, Radon entry into the atmosphere: Diffusion, Advection and Convection. Health Effects: Dose.

Atmospheric electricity: The generation of an ion, The mobility of ions, Ion size, recombination of ions. Ions in an electric field, Ionizing agencies, radioactivity. The conductivity of the atmosphere and its origin, Measurement of conductivity of the atmosphere near the ground. Relationship between ions and conductivity. The current voltage characteristics in a gas under conditions of volume ionization. [16 hours]

Total work load

48 hours

References:

1. Salby M.L., *Fundamentals of atmospheric physics*, Academic Press, USA, 2006.
2. Houghton J., *The physics of the atmosphere*, Cambridge University Press, 2002.
3. Siddhartha K., *Atmosphere, weather and climate*, Kisalaya Publications, 2000.
4. Lutgens F.K. and Tarbuk E.K., *The atmosphere: An introduction to meteorology*, Prentice Hall USA, 1986.

5. Holton, J.R., *Dynamic meteorology*, 3rd edition, Academic Press, USA, 1992.
6. Keshvamurthy R.N. and Shankar Rao M., *The physics of monsoons*, Allied Publishers, 1992.
7. Iqbal M., *An introduction to solar radiation*, Academic Press, USA, 1983.
8. Wilkening M., *Radon in the environment*, Elsevier Science Publishers, The Netherlands, 1990.
9. Israel H., *Atmospheric electricity-Vol II*, Israel Program for Scientific Translations, Jerusalem. 1973.

PHY-410: Numerical Methods

Course Outcome:

After completing this course the student will

- CO1. Understand Computer arithmetic in solving the problems.
- CO2. Learn Iterative methods like Bisection method, Newton-Raphson method, Secant method.
- CO3. Solve Linear algebraic equations by numerical techniques like the Gauss elimination method.
- CO4. Learn numerical Interpolations, Least-squares approximation of functions.
- CO5. Learn different techniques of Numerical integration
- CO6. Solve Numerical solution of differential equations

Computer arithmetic: *Integers; Floating point representation of numbers; Arithmetic operations with normalisation; Errors in representation; Commonly used number types and their limits like max. and min. integer, float, double precision, long, etc.*

Iterative methods: *Bisection method, Newton-Raphson method, Secant method, the method of successive approximations. Solution of a polynomial equation.* **[16 hours]**

Linear algebraic equations: *The Gauss elimination method, LU decomposition method, Gauss-Jordan method, An introduction to the solution of simultaneous non-linear equations.*

Interpolations: *Introduction, Newton interpolation formulae, extrapolation, Lagrange interpolation. spline interpolation.*

Least-squares approximation of functions: *Introduction, linear regression, algorithm for linear regression. Polynomial regression, fitting exponential and trigonometric functions.* **[16 hours]**

Numerical integration. *Trapezoidal method, Simpson rule. Errors in integration formulae (Romberg method). Algorithms for integration of a tabulated function. Algorithms for integrating a known function. Gaussian quadrature formulae.*

Numerical solution of differential equations: *Euler method, Runge - Kutta methods, Runge - Kutta 4th order formulae, predictor - corrector method. comparison of predictor-corrector and Runge- Kutta methods.* **[16 hours]**

Total work load

48 hours

References:

1. Atkinson K.E., *An introduction to numerical analysis*, John Wiley and Sons, USA, 1988.
2. Press W.H., Flannery B.P., Teukolsky S.A. and Vetterling W.T., *Numerical recipes in C*, Cambridge University Press, UK, 1989.
3. Krishnamurthy E.V. and Sen S.K, *Numerical algorithms*, Affiliated East West Press Pvt. Ltd., India, 1993.
4. Rajaraman V., *Computer oriented numerical methods*, Prentice Hall of India Pvt. Ltd., India, 2001.

Elective Papers 2

PHY-411: Nuclear Spectroscopy Methods

Course Outcome:

After completing this course the student will

- CO1. Understand Ion implantation and backscattering spectroscopy
- CO2. Study Compton scattering
- CO3. Know about Positron annihilation spectroscopy
- CO4. Understand Experimental methods of positron annihilation spectroscopy

Ion implantation and backscattering spectroscopy: *Ion implantation, Implantation technique, Ion beam diffusion, Thermal annealing and sputtering, Analysis techniques. Backscattering, Energy loss and straggling. Kinematics factor, differential scattering cross sections, depth scale, backscattering yield, instrumentation. Application to elemental and compound targets. Axial and planar half angles. Estimates of minimum yield. Lattice location of impurities, alignment procedures. Ion induced X-rays. Application of ion implantation. [16 hours]*

Compton scattering: *Compton scattering from free electrons. Effects of external potential. Klein-Nishina cross sections for polarized and unpolarized radiation. Compton profiles, momentum distributions and impulse Compton profiles. Calculation of Compton profiles for electron models. Relativistic profile corrections: experimentation. Discussion of methodology including sources, detectors and geometry. Data accumulation, analysis and multiple scattering corrections. Discussion of experimental results for some simple metals, ionic and covalent crystals. [16 hours]*

Positron annihilation spectroscopy: *The positron and its discovery, Positronium, its characteristics, formation. Spur model and Ore gap model of positronium formation. Quenching and enhancement. Theory of 2-gamma and 3-gamma annihilations. Positron and positronium states in solids: trapping of positrons. Two state trapping model.*

Experimental methods of positron annihilation spectroscopy: *Positron lifetime techniques (PLT), Angular Correlation of Annihilation Radiation (ACAR), Doppler broadening (DB) and Coincidence DB. Methods of data analysis: PLT and ACAR. Experimental results of some metals and defected materials. Interpretation of the experimental results. PAS in the study of polymers. Multiparameter techniques. A brief mention of slow positron beams. [16 hours]*

Tutorial

[32 hours]

References:

1. Townsend P.D., Kelly J.C. and Hartley N.E.W., *Ion implantation, sputtering and their applications*, Academic Press, London, 1976.
2. Chu W.K., Mayer J.W. and Nicholate Mar A.O., *Backscattering spectroscopy*, Academic Press, New York, 1978.
3. Mayer J.W. and Rimini B. (Eds.), *Ion beam handbook for material analysis*, Academic Press, 1977.
4. Williams B. (Ed.), *Compton scattering*, McGraw-Hill, New York, 1977.
5. Hautajarvi P. (Ed.), *Positrons in solids*, Springer Verlag, New York, 1979.
6. Fava R.A. (Ed.), *Methods of experimental physics*, Academic Press, New York, 1980.
7. Schradev D.M. and Jean Y.C., *Positron and positronium chemistry*, Elsevier Science Publication, Amsterdam, 1988.
8. Jayaram B., *Mass spectrometry—Theory and applications*, Plenum Press, New York, 1966.

PHY-412: Modern Optics

Course Outcome:

After completing this course the student will

- CO1. Understand the phenomenon of polarization of light using quantum mechanics

- CO2. Familiar with Non linear Optics
 CO3. Understand Pancharatnam phase in Quantum features of radiation field.
 CO4. Learn the concept of Radiation Field Quantization.
 CO5. Understand Squeezed states of light.

Polarization of light: Pure states and mixed states. Density operator, properties and equation of motion. Polarization of light, states of polarized light, Jones matrices, Jones formalism, Stokes parameters, Poincaré sphere, Mueller matrices and Mueller formalism, Mueller matrices and their characterization, Few illustrative examples; comparison of Jones and Mueller formalisms. Pancharatnam phase, dynamical phase, cyclic evolution of polarization state on Poincaré sphere; Applications of the concept of Pancharatnam phase. [16 hours]

Quantum features of radiation field: Planck's law of radiation and Einstein coefficients, Thermal equilibrium, Semi-classical theory of two level atoms, quantum theory of B coefficient, Optical resonance, damping, Theory of chaotic light, coherence, temporal, spatial, mutual coherence, line broadening, natural and Doppler width, collision broadening. [16 hours]

Quantized radiation field: Quantization of radiation field, States of radiation field; Fock states and phase eigenstates; Interaction of radiation with matter, theory of spontaneous emission; Coherent states and their properties, BCH formula, P, Q and Wigner distribution functions, Squeezed states of light and their properties; applications. Correlation functions, Brown-Twiss correlations. [16 hours]

Tutorial [32 hours]

References:

1. Loudon R., *The quantum theory of light*, Clarendon Press, Oxford, 1973.
2. Mandel L. and Wolf E., *Optical coherence and quantum optics*, Cambridge University Press, 1995.
3. Louisell W.H., *Quantum statistical properties of radiation*, John Wiley and Sons, New York, 1973.
4. Blum K., *Density matrix theory and applications*, Plenum Press, New York, 1981.
5. Pancharatnam S., *Collected works*, Oxford University Press, 1975.

PHD 280: Electronics

Course Outcome:

After completing this course the student will

- CO1. Learn analyzing digital and analog devices and circuits.
 CO2. Analyze components associated with digital and analog electronic systems.
 CO3. Demonstrate proficiency in the use of electronic equipment and devices.
 CO4. Assist in the design, operation, and troubleshooting of electronic systems.
 CO5. Analyze electronics devices and circuits using computer simulations.
 CO6. Solve electronic devices and systems using mathematical concepts.

BJT AC Analysis: Amplification in AC domain. BJT transistor modeling, common emitter voltage divider bias configuration. Emitter follower configuration. Darlington connection. Hybrid equivalent model, Approximate Hybrid equivalent circuit ; Voltage divider configuration, Complete hybrid equivalent model.

Feedback and Oscillator Circuit: Feedback concept, Feedback connections types, Practical feedback circuits. Feedback amplifier; Phase and frequency considerations. Oscillator operation, Phase - shift Oscillator, Wien-bridge Oscillator, Crystal Oscillator—BJT version.

FET amplifiers: JFET small signal model, Biasing of FET, Common drain, common gate configurations, FET amplifier and its frequency response. MOSFET – types and E – MOSFET Voltage divider configurations (Boylestad and Nashelsky) [16 hours]

Operational amplifiers: Concepts of differential amplifier, Ideal op-amp, op-amp parameters, ideal voltage transfer curve, open loop and closed op-amp configurations, inverting amplifier, non inverting amplifier, limitations of open loop op-amp configurations.

Operational amplifier applications: Summing, scaling and averaging amplifiers, voltage to current converter with grounded load, current to voltage converter, integrator, differentiator, V to I and I to V converters, Log and antilog amplifiers, Wave form generators, phase shift oscillator, Wein bridge oscillator. Non-linear circuit

applications: Crossing detectors, 555 timer as a mono-stable and astable multivibrators, Active Filters—First and second order Low pass and High pass filters, Butterworth filters (**Gaekwad R.A**) **[16 hours]**

Digital electronics: Boolean Laws and Theorems, addition and subtraction based on 1's and 2's complements, Families of gates, RS and JK flip-flops, The Master-Slave JK Flip-Flop, D and T flipflops. Karnaugh maps for 3 and 4 variables, Decoders-BCD decoders, Encoders.

Combinational logic circuits: Shift registers-series, series in-series out and parallel in parallel out. Half and full adders, Registers, Counters - Binary Ripple Counters, Synchronous Binary counters, Counters based on Shift Registers, Synchronous counters, Synchronous Mod-6 Counter using clocked JK Flip-Flops. Synchronous Mod-6 Counter using clocked D, T, or SR Flip-Flops. Memory cells, memory registers **16 hours]**

Tutorial

[32 hours]

References:

1. Boylestad R.L. and Nashelsky L., *Electronic devices and circuit theory*, 4th Edn., Pearson Education, 2006.
2. Bell D.A., *Operational amplifiers and linear circuits*, 2nd Edn., Pearson Education, 2004.
3. Gayakwad R.A., *Operational amplifiers and linear integrated circuits*, Prentice-Hall of India, New Delhi, 1993.
4. Malvino A.P. and Leach D.P., *Digital principles and applications*, 4th Edn., Tata McGraw Hill, 1988.
5. Arivazhagan S. and Salivahananan S., *Digital circuits and design*, Vikash Publishing House Pvt. Ltd. New Delhi, 2001.
6. *Op-amps and linear integrated circuits*, ramakanth A Gaekwad, 3rd edition, Pearson education Asia, 2002
7. *Linear ICs and applications* Uday A Bakshi & Atul P Godse, Technical Publications
8. *Linear integrated Circuits*, Roy & Choudary
9. *Digital fundamentals*, Thomas L Floyd

PHY-414: Minor Project

Course Outcome:

After completing this course the student will have

- CO1. *Hands on experience to various experimental Techniques*
- CO2. *Research exposure to Physics experiments.*
- CO3. *Knowledge on construction of electronic circuits for various application.*
- CO4. *Data analysis techniques and plotting of experimental results.*

PHD 090: Nuclear and Particle Physics Lab

For those who have completed **Condensed Matter Physics Lab PHY311**

Course Outcome:

After completing this course the student will

- CO1. *Be familiar with theoretical concepts & experimental techniques of Nuclear & particle Physics.*
- CO2. *Be able to make quantitative estimates of phenomena in elementary particles*
- CO3. *Understand stability of the nucleus, nuclear forces, interactions and nuclear models; critically assess applications of nuclear technology.*
- CO4. *Acquire Theoretical and practical experience in the scattering experiments and gamma-ray spectroscopy.*
- CO5. *Be able to realise the Nuclear Electronic circuits*

For those who have completed **Condensed Matter Physics Lab PHY311**

Any eight of the following experiments:

1. Half-life of Indium-116 measurement.
2. Energy Resolution of a NaI(Tl) scintillation spectrometer.
3. Compton scattering determination of the rest energy of an electron.
4. Beta absorption coefficient measurement.
5. Dekatron as a counter of signals.
6. Gamma-ray absorption coefficient measurement.

7. End-point energy of beta particles by half thickness measurement.
8. Common source amplifier.
9. Astable multivibrator using timer IC 555.
10. Dead time of the G.M. counter.

Total work load : 2 day(s) per week \times 4 hours \times 16 weeks = 128 hours

PHD 080: Condensed Matter Physics Lab

For those who have completed **Nuclear Physics Lab PHY 312**

Course Outcome:

After completing this course the student will

- CO1. Independently collect single crystal XRD data and evaluate the crystal structure,
- CO2. Analyse Diffraction data and their suitability for single crystal structure.
- CO3. Independently work on raw diffraction data and Solve and refine crystal data
- CO4. Understand the properties of Condensed Materials such as Structural, Magnetic, Electrical and Semiconducting Properties.
- CO5. Acquire theoretical and practical experience in the Condensed Matter Physics

Any eight of the following experiments :

1. Determination of the paramagnetic susceptibility of the given salt by Quincke's method.
2. Study of mercury spectrum by superimposing it on brass spectrum.
3. Sodium spectrum analysis by using Edser-Butler fringes.
4. Temperature coefficient of resistance of a thermistor.
5. Analysis of the powder X-ray photograph of a simple cubic crystal.
6. Thermionic work function of a metal (Richardson-Dushman formula).
7. Energy gap of semiconductor.
8. Determination of Stefan's constant.
9. Frank Hertz experiment
10. Magnetic hysteresis.
11. Measurement of magneto resistance of semiconductors.

Total work load : 2 day(s) per week \times 4 hours \times 16 weeks = 128 hours

PHD 240: Solid State Physics Lab 2

For those who opted for **Solid State Physics Specialisation**

Course Outcome:

After completing this course the student will

- CO1. Be able to make quantitative estimates for phenomena in solid state physics.
- CO2. Describe characteristics of thermal and vibrational properties of solids, Compute Band structures
- CO3. Compute trajectories in real and reciprocal space, Characterize magnetism
- CO4. Characterize intrinsic and doped semiconductors
- CO5. Describe superconductivity
- CO6. Understand the mechanism and working of semiconductor devices.

Any five of the following experiments:

1. Photovoltaic cell.
2. Photoconductive cell.
3. Hall effect in semiconductors.
4. Determination of the energy gap of semiconductors by four-probe method.
5. Temperature variation of the junction voltage of a p-n diode.

6. Temperature variation of the reverse saturation current in a p-n diode.
7. Depletion capacitance of a junction diode.
8. Determination of material constant of an intrinsic semiconductor.
9. Schottky effect.
10. Ionic conductivity of an alkali halide crystal.
11. Dielectric constant and its temperature variation.
12. Ultrasonic velocity and elastic constants of a solid.
13. Determination of Curie temperature of a magnetic material
14. Magnetic field variation along with axis of the solenoid
15. Magnetic Hysteresis
16. Thermal Diffusivity of Brass
17. Temperature co-efficient of resistance of copper

Total work load: 1 day(s) per week × 4 hours × 16 weeks = **64 hours**

PHD 250: Nuclear Physics Lab 2

For those who opted for **Nuclear Physics Lab Specialisation**

Course Outcome:

After completing this course the student will

- CO1. Be familiar with Concepts of Nuclear Detectors, Pulse techniques, nuclear models, timing spectroscopy and nuclear technology.
- CO2. Be familiar with Theoretical and practical experience in the scattering experiments and gamma-ray spectroscopy.
- CO3. Learn Instrumentation, development of sensors, detectors, and larger complex instruments.
- CO4. Familiarise in Fundamentals of reactor theory and nuclear decay mechanisms
- CO5. Learn Accelerator physics, research and development for the next generation of particle accelerators.

Any five of the following experiments:

1. Schmitt trigger.
2. Variable delay line.
3. Pulse recorder.
4. Display devices.
5. Feather analysis: End-point energy of beta rays measurement.
6. Z dependence of external Bremsstrahlung radiation.
7. Fermi-Kurie plot : Determination of the end-point energy of beta rays using a plastic scintillation detector.
8. Determination of the resolving time of a coincidence circuit.
9. Determination of source strength by gamma-gamma coincidence.
10. Determination of source strength by beta-gamma coincidence.
11. Multichannel analyser : Study of the variation of energy resolution as a function of gamma ray energies.
12. Verification of Mosley's law
13. Beta ray absorption studies – relation between $\frac{\mu}{\rho}$ and end point energy.
14. Absorption coefficient of Al using Sr-90 and Y-90 beta sources.

Total work load : 1 day(s) per week × 4 hours × 16 weeks = **64 hours**

PHY-425: Theoretical Physics Lab 2

For those who opted **Theoretical Physics Lab Specialisation**

Course Outcome:

After completing this course the student will

- CO1. Tackle a wide range of topics using powerful analytical tools including formal methods in classical and quantum physics

- CO2. Clearly communicate information and conclusions in written and verbal formats on ideas in Theoretical Physics
- CO3. Evaluate complex problems and formulate solutions, indentifying the role of theory, hypothesis and experiment in the scientific method
- CO4. Apply computation to solve the problems in theoretical physics
- CO5. Plan, carry out and report theoretical physics based investigation
- CO6. Apply classical and quantum theoretical techniques in research

Any five of the following experiments:

1. Density matrix description of polarization of light.
2. Double scattering of spin-1/2 particles on spin-zero targets.
3. Second order QED processes (Compton scattering).
4. Evolution of matrix elements between coupled angular momentum states.
5. Dirac matrix representations.
6. Algebra of Dirac matrices.
7. Electron-proton scattering, Rosenbluth formula.
8. Relativistic kinematics-3: Study of decay and production processes.
9. Feynman diagrams and calculations.
10. Energy matrix calculation.

Total work load : 1 day(s) per week \times 4 hours \times 16 weeks = **64 hours**

JSS MAHAVIDHYAPEETHA
JSS COLLEGE OF ARTS COMMERCE & SCIENCE
(Autonomous) Ooty road, Mysore – 25
II and IV Semester Examination May / June 2018

M.Sc PHYSICS Question Paper Pattern

Time:3 Hours

Answer all the questions

Max. Marks: 70

SECTION A

1. (a)
(b)
(c)

18 Marks

OR

2. (a)
(b)
(c)

18 Marks

SECTION B

3. (a)
(b)
(c)

18 Marks

OR

4. (a)
(b)
(c)

18 Marks

SECTION C

5. (a)
(b)
(c)

18 Marks

OR

6. (a)
(b)
(c)

18 Marks

SECTION D - (Problems only)

7.

5 Marks

OR

8.

5 Marks

9.

6 Marks

OR

10.

6 Marks

11.

5 Marks

OR

12.

5 Marks

**J.S.S. College of Arts, Commerce and Science
(Autonomous)
Ooty Road, Mysuru-570 025**

DEPARTMENT OF ZOOLOGY (PG)

**Programme outcome, Programme specific outcome, Course outcome and
curriculum for Postgraduate Zoology
(2018-2019 & onwards)**

Program Outcome

1. Imbibe the knowledge with facts and figures related Zoology.
2. Understand the basic concepts, fundamental principles, and the scientific theories related to various scientific phenomena and their relevancies in the day-to-day life.
3. Identify, formulate, research literature, and analyze complex problems reaching substantiated conclusions using first principles of mathematical, biological, physical and chemical sciences.
4. Will be able to think creatively to propose novel ideas in explaining facts and figures or providing new solution to the problems.
5. Develop scientific outlook not only with respect to Zoology but also in all aspects related to life.
6. Realize that interdisciplinary knowledge in other faculties can have greatly and effectively influence which inspires in evolving new scientific theories and inventions.
7. Imbibe ethical, moral and social values in personal and social life leading to highly cultured and civilized personality.
8. Develop various communication skills such as reading, listening, speaking, etc.
9. Realize that acquiring knowledge is a continuous process and in combination with untiring efforts and positive attitude and other necessary qualities leads towards a successful life.

Programme Specific outcome:

At the completion of M.Sc. in Zoology the students are able to:

1. Understand the classification and taxonomic aspects of the animal world (chordates and non-chordates). The students will be able to identify the taxonomic group of a given animal based on the external characteristics.
2. Understand the basic concepts of Animal physiology. The students will be able to identify and understand the important life processes which are essential for continuation of life on earth.
3. Understand the nature and structure of biomolecules and basic concepts of Biological chemistry.
4. Understand the concepts of Genetics, Cell Biology and Molecular Biology.
5. Understand the basic principles and concepts of environmental science, ecology and nature conservation.
6. Understand the importance of knowledge of wild life and animal behaviour for conservation and balancing the nature.
7. Understand the tools and techniques employed in Biological research and experiments.
8. Understand the process of evolution.
9. Understand the concept and applications of sericulture, apiculture, animal husbandry, Lac culture etc.

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
OOTY ROAD, MYSURU
PG DEPARTMENT OF ZOOLOGY
Syllabus Adopted from the academic year 2018-19

Semester	HC / SC	Paper title	CREDITS			Total Credits
			L	T	P	
I	HC - 1.1	Biosystematics & Non Chordata	2	0	2	4
	HC - 1.2	Biological Chemistry	2	0	2	4
	HC - 1.3	Cytogenetics	2	0	2	4
	SC - 1.4	Tools and techniques in Biology	3	1	0	4
	SC - 1.5	Chronobiology	3	1	0	4
	SC - 1.6	Histology and Histopathology	3	0	1	4
Any two of the Soft core paper may be opted						20
II	HC - 2.1	Chordata	2	0	2	4
	HC - 2.2	Animal Physiology	2	0	2	4
	HC - 2.3	Entomology	2	0	2	4
	SC - 2.4	Developmental Biology	3	0	1	4
	SC - 2.5	Immunology	3	1	0	4
	SC - 2.6	Evolutionary Biology	3	1	0	4
Any two of the Soft core paper may be opted						20
III	HC - 3.1	Molecular Biology & Biotechnology	2	0	2	4
	HC - 3.2	Reproductive Biology	2	0	2	4
	HC - 3.3	Ecology and Wildlife**	2	0	2	4
	SC - 3.4	Ethology *	3	1	0	4
	SC - 3.5	Pollution and Toxicology *	3	1	0	4
	OE - 3.6	Concepts of Zoology	3	1	0	4
*Any one of the Soft core paper may be opted ** Field visits are included in this paper						20
IV	HC - 4.1	Advanced Genetics and Computational Biology	2	0	2	4
	HC - 4.2	Applied Zoology*	2	0	2	4
	HC - 4.3	Project	0	2	6	8
* Field visits are included in this paper						16

Total credits

Hard Core - 52 Credits

Soft Core - 20 Credits

Open Elective - 04 Credits

Total credits required to complete M.Sc Course - 76 Credits

**M.Sc, I SEMESTER
HC 1.1 Non Chordata**

32hrs

Course Outcome:

After completing the course student will be able to

1. Understand the classification of major and minor invertebrate phyla
2. Give some examples and basic characteristics of some examples of each phylum
3. Understand the evolutionary pathway and its significance
4. Adaptive characters of animals coming under different invertebrate phyla

UNIT I Basic concepts of animal taxonomy:

8 hrs

- A. Introduction and history of taxonomy
- B. Species concept
- C. Zoological classification - theories of classification - taxonomic ranks – hierarchy
- D. Zoological nomenclature: Binomial nomenclature, trinomial nomenclature-ICZN
- E. taxonomical keys: key to the species
- F. Linnaean taxonomy and classical taxonomy - level of taxonomy.

Unit II : Classification, Locomotion and Nutrition:

8 hrs

- A. General Characteristics of Non chordata.
- B. **Locomotion:** Muscle filaments and myonemes,
Flagella and cilia. Amoeboid movement.
- C. **Nutrition in Protozoa:** Filter feeding in polychaetes, Filter feeding
and digestion in Deuterostomia and molluscs.
- D. **Respiration:**
Structure and function of respiratory organs- Skin, gills, book lungs and
Trachea. Respiratory pigments

Unit III:

8 hrs

- A. **Excretion and osmoregulation:**
Osmoregulation in fresh water and marine Invertebrates
Structure and function of excretory organs- Coelom, Coelomoducts, Nephridia,
Malpighian tubules and Coxal glands
- B. **Nervous system:**
Primitive nervous system: Coelenterata and Echinodermata
Advanced nervous system: Annelida, Arthropoda(Crustacea and insecta) and
Mollusca (Cephalopod)
- C. Sense organs and their importance

Unit IV:

8 hrs

- A. **Invertebrate paleontology and larval forms:**
Free living and parasitic Larval forms
- B. **Fossil:** types and importance of fossil study, overview of Geological Time Scale

NON CHORDATA –PRACTICALS

4x16=64 Hrs

1. PROTOZOA;

4x12=48 hrs

Slides : 1) *Trypanosoma cruzi* 2) Plasmodium – signet ring stage 3) Ceratium
4) *Leishmania donovani* 5) Vorticella 6) Noctiluca 7) Radiolaria 8) *Entamoeba histolytica*
9) Foraminifera 10) Opalina

2. PORIFERA;

a) Slides: 1)Sponge spicules 2)Sponge gemmules

b) Specimen: 1) Grantia 2) Euspongia 3) Clypeaster

3. CNIDARIA:

a) Slides: 1) Obelia polyp and Medusa 2) Pennaria 3) Aurelia-tentaculocyst

b) Specimens: 1) Physalia 2) Gorgonia 3) Spongodus 4) Zoanthus 5) Favia 6) Pennatula
7) Sea anemone 8) *Corallium rubrum*

4. HELMINTHES:

a) Slides: 1) *Fasciola hepatica* 2) *Ancylostoma*

b) Specimens: 1) Planaria 2) Male and female *Ascaris lumbricoides* 3) *Taenia solium* 4)

5. ANNELIDA:

a) Slides: 1) Leech 2) Earthworm setae

b) Specimens: 1) Neries 2) *Chloea flava* 3) *Pheretima postuma* 4) *Terebella* 5) *Eurythoe*

6. ARTHROPODA:

a) Slides: 1) Daphnia 2) Sacculina 3) T.S of Peripatus

b) Specimens: 1) Balanus 2) Lepas 3) Palinurus 4) Scolopendra 5) Rhinoceros beetle
6) Spider 7) Gongylus 8) Belostoma 9) Limulus 10) Squilla 11) Eupagarus 12) Julus

7. MOLLUSCA :

Specimens: 1) Aplysia 2) Glochidium 3) Loligo 4) Chiton 5) Cypreae 6) Octopus
7) Sanguinolaria 8) Chicoreus 9) Ficus 10) Lambis 11) Mytillus 12) Doris 13) Onchidium
14) Oliva 15) Murex 16) Turritella 17) Cardium

8. ECHINODERMATA:

Specimens: 1) Sea Urchin 2) Linckia 3) Echinodiscus 4) Holothuria 5) Antedon

9. MINOR PHYLA: —1) Lingula

10. LARVAL FORMS:

Slides: 1) Cercaria 2) Trochophore 3) Megalopa larva 4) Nauplius 5) Zoea 6) Mysis

7) Phyllosoma 8) Protozoa 9) Bipinnaria 10) Veliger 11) Tornaria

12) Glochidium 13) Pluteus

11. Field Study: Visit to different areas around the college campus, to observe and study

Non chordates in their natural habitat.

4x2=8 hrs

II. Study of Nervous system, Respiratory system, Reproductive system and Excretory system

in invertebrates by employing computer animation/charts:

4x2=8 hrs

REFERENCES :

1. Barnes, R.D. 1974. Invertebrate Zoology, III edition. W.B Saunders Co., Philadelphia
2. Barrington, E.J.W, 1976. Invertebrate Structure and Function. Thomas Nelson and Sons Ltd., London.
3. Hyman L.H. 1940. The invertebrates. Vol. 1. Protozoa through Ctenophora, McGraw hill Co., N.Y.
4. Hyman. L H. 1959. The Invertebrates smaller coelomate groups, Vol. V. McGraw Hill Co.,
5. Hyman. L. H. 1951. The Invertebrates. Vol. 2. McGraw Hill Co., N.Y.
6. Hyman. L H. 1968. The invertebrates Vol. 8. McGraw Hill Co., N.Y and London.
7. Simpson, G C. Principles of Taxonomy.

M.Sc, I SEMESTER
HC -1.2 BIOLOGICAL CHEMISTRY

32 hrs

Course Outcome:

After completing the course student will be able to

1. Identify the five classes of polymeric biomolecules and their monomeric building blocks.
2. Explain the specificity of enzymes (biochemical catalysts), and the chemistry involved in enzyme action.
3. Understand types, Structure, biochemical properties and functions of vitamins.
4. Explain how the metabolism of organic compounds leads ultimately to the generation of large quantities of ATP.

UNIT I Chemical Bonds and Carbohydrates: 8 Hrs

- A. Structure of an atom, orbitals, chemical bonds - covalent, co-ordinate, ionic and hydrogen; Vander-Waal's force; hydrophobic interactions; Normality and Molarity of solutions.
- B. Carbohydrates – Chemistry and biological properties

UNIT II Proteins and Lipids: 8 Hrs

- A. Proteins- Chemistry and biological properties, Christian Anfinsen's experiment, Biological values of proteins
- B. Lipids: Chemistry, triglycerides; prostaglandins and steroids –biosynthesis, Chemical importance of lipids.

UNIT III Enzymes: 8 Hrs

- A. Enzymes: Nomenclature – current status; factors influencing velocity of enzyme reaction, enzyme dynamics and enzyme inhibition.
Ribozymes and abzymes; co-enzymes, isozymes, clinical importance.

UNIT IV Nucleic acids & Vitamins: 8 Hrs

- A. Nucleic acids: Chemistry, alternative models of DNA,
- B. Vitamins and trace elements – chemical nature, vitamins as co-enzymes, Deficiency diseases, role of trace elements

Biological Chemistry practicals 4x16=64 Hrs

1. Qualitative analysis for identification of carbohydrates (Starch, Glycogen, Sucrose, Lactose, Maltose, Glucose, Fructose).
2. Qualitative analysis for identification of Proteins (Egg albumin, Casein, Gelatin, Peptone)
3. Precipitation reaction of proteins (Egg albumin, Peptone)
4. The absorbance curves for two dyes and demonstration of Beer-Lambert's law.
5. Estimation of amino acids by Sorenson's formal titration (Arginine, Alanine, Leucine, lysine)
6. Determination of concentration of Glucose and Maltose by calibration curve.
7. Determination of amylase activity.
8. Determination of effect of temperature, pH and incubation period on amylase activity.
9. Test for non-esterified fatty acid.
10. Demonstration of gel electrophoresis.

REFERENCES

1. Barrington, E. J. W (1976) An introduction to general and comparative endocrinology, Oxford University press, London.
2. Conn, E. E., Stumft, P. K., Bruencing, G. and Dol, R. G. 1995. Outlines of Biochemistry. Pub. John Wiley, Singapore.

3. Eckert, R and Randall, D. 2002, Animal physiology, 2nd Edn, W.H..Freman
4. Guyton. A.G. 1986, Text book of Medical Physiology, 7th Edn., Saunders Publication
5. Harper, H. A. 1993. A review of Physiological Chemistry, Lange Medical Publication, 2nd Edn.
6. Lehninger, A. L., Nelson, D. L. and Cox, M. M., 2nd Edn. 1993. Principles of Biochemistry, CBS Publishers and Distributors, New Delhi.
7. Oser, B. L. (Ed.) 1993. Hawk's Physiological Chemistry. Tata Graw Hill Publishing Co. Ltd. New Delhi.

**M.Sc., I SEMESTER
HC – 1.3 CYTOGENETICS**

32 Hrs

Course Outcome:

After completing the course student will be able to

1. Described the fundamental molecular principles of genetics
2. Understood the structure and function of DNA & RNA
3. Understand about the transmission, distribution, arrangement, and alteration of genetic information and how it functions and is maintained in populations
4. Described the basics of genetic mapping
5. Explain basic structure of animal cell and its organelles
6. Describe the functions and organization of cell organelles

Unit I: Introduction to the Cell & Cell Organelles

8 hrs

- A. The origin and evolution of the cell, From molecules to first cell, from Prokaryotes to eukaryotes, from single cell to multicellular organisms.
- B. Membrane Structure and Function,
- C. Structural organization and functions of intracellular organelles- The nucleus, Mitochondria, Lysosomes, Peroxisomes, Golgi apparatus, and endoplasmic reticulum.

Unit II: Cell Cycle and Cell signalling

8 Hrs

- A. Phases of cell cycle.
- B. Biochemical studies with oocytes, eggs and early embryos.
- C. Regulation of cell cycle: Molecular mechanisms regulating mitotic events.
Regulation of cell cycle progression.
Check points in cell cycle regulation.
Cell cycle control in polytene cells.
- D. Molecular basis of signal transduction
- E. Cellular aging and death: (a) Causes of aging
(b) Cellular changes due to aging
(c) Theories of aging
(d) Apoptosis
(e) Longevity genes

UNIT III Gene mutations

8 Hrs

- A. Types of mutations (Spontaneous, Induced, Base substitutions and frameshifts - Transitions, Transversions, gain in function, loss in function, Neutral mutations),
- B. Molecular mechanism of mutations (Base analogs, alkylating agents); Detection of mutations: Dominant lethal test, Sex-linked recessive lethal test, II-III translocations, Ames test, P-mediated mutagenesis

UNIT IV Chromosomal mutations

8 Hrs

- A. Structure and organization of eukaryotic chromosomes
- B. Structural and numerical variations of chromosomes, Chromosomal rearrangements and their cytogenetic consequences with examples from plants, Drosophila and Man,

Practical applications of chromosome rearrangements - Balancers and attached X-chromosome in *Drosophila*. Cytogenetic effects of ionizing and nonionizing radiations

CYTOGENETICS PRACTICALS

4X16 =64 Hrs

- 1) Life cycle of *Drosophila melanogaster* 1x4=04hrs
- 2) Preparation of culture media. Culture of *Drosophila* - Methods of maintenance. 1x4=04hrs
- 3) Study of morphology of *Drosophila melanogaster* 1x4=04hrs
- 4) Mounting of Sex comb of *Drosophila melanogaster* 1x4=04hrs
- 5) Mounting of Wing of *Drosophila melanogaster* 1x4=04hrs
- 6) Study of mutants of *D. melanogaster* 1x4=04hrs
- 7) Preparation of genital plate of *D. melanogaster* 2x4=08hrs
- 8) Chi square Analysis of F1, F2 and Test cross progeny in *Drosophila melanogaster* to understand pattern of inheritance of different characters and to demonstrate. 3x4=12hrs
 - a) Law of segregation
 - b) Law of Independent assortment
 - c) Sex-linked inheritance
- 9) Temporary squash preparation of Mitotic chromosomes from root tip meristem of *Allium cepa* 2x4=08hrs
- 10) Temporary squash preparation of Meiotic chromosomes from testis of *Poicelocerus pictus* 2x4=08hrs
- 11) Study of Barr body using buccal smear of volunteers 1x4=04hrs

REFERENCES:

1. Alberts, B., A. Jhonson, J. Lewis, M. Raff, K. Roberts and P. Walter 2008. Molecular Biology of the cell. V Ed. Garland Science, New York.
2. Brachet, J. 1985. Molecular Cytology, Academic Press, N. Y.
3. Furukawa, R., and M. Fechheimer. 1997. The structure, function and assembly of actin filament bundles. Int. Rev. Cytol. 175: 29-90.
4. Lewin B. (1997) Gene VI Oxford University Press, Oxford
5. Lodish, H., A. Berk, C.A Kaiser, M.P. Scott, A Bretscher, H. Ploegh, P. Matsudaira. 2008. Sixth Edition, Molecular Cell Biology. W. H. Freeman and Co., N. Y.
6. Pollard, T. D. and W. C. Earnshaw. 2002. Cell Biology. Saunders
7. Russel P.J (1998) Genetics. The Benjamin Cummings Publishing Co Inc.
8. Snustad D.P and M.J.Simons. (1997) Principles of Genetics. John Wiley and Sons Inc. N.Y.
9. Strickberger M.W. (1977) Genetics. MacMillan Collier Co. Pvt Ltd
10. Watson J.D, Hopkins, N.H, Roberts J.A, Steitz and A.M.Weiner. (1987) Molecular biology of gene. The Benjamin Cummings Publishing Co Inc.
11. Wolfe, A. 1995. Chromatin: Structure and function. Academic Press, N. Y.

M.Sc., I SEMESTER
SC – 1.4 TOOLS AND TECHNIQUES OF BIOLOGY

48 hrs

Course Outcome:

After completing the course student will be able to

1. Describe the methodology involved in biotechniques.
2. Describe the applications of bioinstruments
3. Demonstrate knowledge and practical skills of using instruments in biology and medical field.
4. Perform techniques involved in molecular biology and diagnosis of diseases
5. Update current knowledge regarding biomedical engineering involving new methods and the instrumentation.

UNIT I: MICROSCOPY:

12hrs

Basic principles of microscopy, Types of microscopes and their biological applications
Bright-field microscope, numerical aperture, limit of resolution, types of objectives, ocular & stage micrometers, Electron Microscope, SEM, Confocal microscope.

Dark-field microscope

Phase-contrast microscope

Differential interference contrast microscope

Fluorescence microscope

Photomicrography and image processing

UNIT II: SEPARATION TECHNIQUES:

12hrs

Centrifugation - Basic principles, Types of rotors, Clinical, high speed & ultracentrifuge

Electrophoresis – Agarose and polyacrylamide gel, Two-dimensional, Isoelectrofocussing

Chromatography - Paper and Thin layer chromatography, Column chromatography, Gel filtration, Ion-exchange, Affinity, Introduction to FPLC and HPLC

UNIT III:

12hrs

A. Radio-tracer techniques

Unit of radioactivity and half life, Measurement of radioactivity (β and γ emission), Applications of radioisotopes, Safety measures

B. Techniques in immunodetection: Immunoblotting and immunofluorescence

C. Immunological techniques: Immunodiffusion and Immunoelectrophoresis

UNIT IV:

12hrs

A. Cell culture techniques: Design and functioning of tissue culture laboratory; Culture media, essential components and preparation; Cell viability testing

B. Cytological techniques: Mitotic & Meiotic chromosome preparations from insects and vertebrates Chromosome banding techniques (G-, C-, Q-, R- banding etc.)

C. Molecular cytological techniques: In situ hybridization (radiolabelled & non-radiolabelled methods), FISH, and Restriction banding

D. Molecular biology techniques: Southern hybridization and Northern hybridization DNA sequencing Polymerase chain reaction (PCR)

TUTORIALS

2x16 = 32 Hrs

REFERENCES

1. Alberts et al: Molecular Biology of the Cell, Garland, 2002
2. Karp: Cell and Molecular Biology, John Wiley & Sons, 2002
3. Lodish et al: Molecular Cell Biology, Freeman, 2000
4. Pollard & Earnshaw: Cell Biology, Saunders, 2002
5. Ruthman: Methods in Cell Research, Bell & Sons, 1970.

6. Boyer: Modern Experimental Biochemistry and Molecular biology (2nd Ed.), Benjamin/Cumin, 1993
7. Freifelder: Physical Biochemistry (2nd Ed.), Freeman, 1982
8. Holme and Peck: Analytical Biochemistry (3rd Ed.), Tata McGraw Hill, 1998
9. Plumer: An Introduction to Practical Biochemistry (3rd Ed.), Tata-McGraw Hill, 1990
10. Switzer and Garrity: Experimental Biochemistry 92nd Ed.), Freeman, 1999
11. Wilson and Walker: Practical Biochemistry (3rd Ed.), Cambridge Univ. Press, 2000

**M.Sc., I SEMESTER
SC – 1.5 CHRONOBIOLOGY**

48 hrs

Course Outcome:

After completing the course student will be able to

1. Understand the concept of Chronobiology
2. Identify the way by which circadian rhythms affect life from the genome to the complex behaviour of the individual
3. Acknowledge the role of Chronobiology and chronodisruption on several physiopathological events
4. Acknowledge the input of the synchronizers on homeostasis
5. Characterize the biological relevance of several chronotypes
6. Acknowledge the relevance of circadian rhythms on therapeutic interventions
7. Acknowledge the importance of scientific research on Chronobiology
8. To interpret study designs and scientific parameters related to Chronobiology.

UNIT I: Introduction:

4 hrs

History, Biological rhythms, Biological clocks, Significance of biological timekeeping

UNIT II: Biological rhythms:

10 hrs

- A. Types of rhythms- Circadian, Circatidal, Circalunar, Circannual
- B. Methods of measurement
- C. Properties: Entrainment, Re-entrainment, Phase angle difference, Freerun, Phase shift, Phase response curve, Arrhythmia.

UNIT III: Factors influencing biological rhythms:

10 hrs

- A. Environmental: Photoperiod -Photoreception and photo-transduction;
The physiological clock and measurement of day length;
Role of photic and non-photic cues in seasonality, Other zeitgebers
Reversal of roles of principal and supplementary cues.
- B. Evolution of photoperiodism: comparative studies; Circannual rhythms and seasonality.

UNIT III: Circadian pacemaker system:

8 hrs

- A. Suprachiasmatic nuclei, B. Pineal gland, C. Optic lobes.

UNIT IV: Molecular basis of circadian rhythms

8 hrs

- A. Clock genes, B. Drosophila, C. Mouse

UNIT V: Applied Chronobiology:

8 hrs

- A. Human circadian rhythms: Melatonin: Input or output signal of the clock system, Clock function (dysfunction); Human health and diseases
- B. Applications of circadian rhythm principles: Jet-lag/shift work, Depression and

sleep disorders, Chronopharmacology and Chronotherapy

TUTORIALS

2X16=32 Hrs

References

1. Binkley, S. (1990): The clockwork sparrow: time, clocks, and calendars in biological organisms, Prentice-Hall, New Jersey.
2. Chandrashekar, M. K. (1985): Biological rhythms, Madras Science Foundation, Chennai.
3. Jay. C. Dunlap, Jennifer. J. Loros, Patricia J. DeCoursey (ed). 2004: Chronobiology Biological Timekeeping, Sinauer Associates, Inc. Publishers, Sunderland, MA, USA
4. Nelson, R. J. (2000) An Introduction to Behavioural Endocrinology, 2nd edition, Sunderland Publishers, Massachusetts.
5. Saunders D.S., C.G.H. Steel, X., afopoulou (ed.)R.D. Lewis. (3rd Ed) 2002: Insect Clocks, Baren and Noble Inc. New York, USA
6. Shapiro, C. M. and Heslegrave, R. J. (1996): Making the shift work, Joli Joco Publications, Inc. Toronto.
7. Vinod Kumar (ed 2002) : Biological Rhythms Narosa Publishing House, Delhi/ Springer-Verlag, Germany

**M.Sc., I SEMESTER
SC – 1.6 HISTOLOGY AND HISTOPATHOLOGY**

48 Hrs

Course Outcome:

After completing the course student will be able to

1. Understand the applications of dyes and its classification.
2. Know the functional morphology of various mammalian organs.
3. Imbibe the knowledge on histochemical techniques.
4. Describe the etiology and pathology of liver cirrhosis and atherosclerosis.
5. Explain histopathology of breast and prostate tumours.

UNIT – I Basics of Histology

8Hrs

- A. Objectives and applications
- B. Tissue fixation : Objectives, methods, chemical fixatives-types and chemistry of fixation; Physical methods:-freezing and microwave fixation; choice of fixatives, fixation artifacts.
- C. Dyes. –Natural and Synthetic, Classification

UNIT-II Functional Morphology (mammalian)

8 Hrs

- A. Histological organization of GI tract- stomach and intestine,
- B. Histological organization of lungs & kidney
- C. Histological organization of spleen & thymus,
- D. Bone and bone marrow.

Unit-III Histochemistry

8 Hrs

- A. Principles and methods of application
- B. Classical histochemical Techniques: for localization of glycoproteins (PAS), nucleic acids (Feulgen) and steroid dehydrogenase activity.

Unit-IV Immunohistochemistry

8 Hrs

- A. Principles, method of application
- B. Immunohistochemistry techniques for localization of proteins in endocrine cells (Pituitary cell types or islet of Langerhans)

C. Immunofluorescence: In situ hybridization of nucleic acids

UNIT-V Histopathology

8 Hrs

- A. Morphological alterations in cells due to disease,
- B. Types of degeneration: clouding, hyaline, hydrophic and fatty degeneration.
- C. Etiology, pathogenesis and histopathology of Liver cirrhosis and atherosclerosis
- D. Neuropathology of alcoholism and methanol poisoning.

Unit-VI Histopathology of tumors

8 Hrs

- A. Malignant and non-malignant
- B. Types of carcinoma
- C. histopathology of breast and prostate tumors

PRACTICALS

2x8= 16 Hrs

I. Histology:

- 1. Microtomy and staining: Hematoxylin-eosin - Demonstration 2x2=4 hrs
- 2. Histology: 2x2=4 hrs
Observations of permanent slides of mammalian organs – stomach, intestine, spleen, liver, kidney, lungs, testis, epididymis, vas deferens, ventral prostate, seminal vesicle, ovary, uterus and Fallopian tube.

II. Histometry:

2x3=6hrs

Histometrical measurements and statistical analysis of some tissues.

III. Histopathology:

2x1=2hrs

Study of histopathological changes (permanent slides) – gastric ulcers, cirrhosis of liver, breast tumors, cystic follicles of ovary, pancreas in diabetics, cryptorchid testis and leukemia.

REFERENCES:

- 1. Boyd, W. 1976: A text book of Pathology. Structure and function in disease, 4th edition. Lea and Fibiger, Philadelphia.
- 2. Pearse, A.G.E. (1980): Histochemistry, theoretical and Applied, J & A, Churchill Ltd., London.
- 3. Rogers, A.W. (1983): Cells and Tissues, An introduction to Histology and Cell Biology, Academic Press, NY.
- 4. Telford, I.R. and Bridgman, C.F. (1990). Introduction to Functional Histology, Harper and Row, NY.

**M.Sc., II SEMESTER
HC – 2.1 CHORDATA**

32 Hrs

Course Outcome:

After completing the course student will be able to

- 1. Understand the classification of chordates
- 2. Give some examples and basic characteristics of some examples of protochordates
- 3. Give some examples and basic characteristics of some examples of vertebrates
- 4. Understand the evolutionary pathway and its significance
- 5. Analyse adaptive characters of animals coming under different vertebrate classes

UNIT I General characters and outline classification of Chordata

8hrs

- A. General and Comparative study: Comparison of three Protochordates, Subphyla in terms of General comparison, Habits and habitats,

- B. Alimentary canals and associated glands, Pharynx, Food and feeding and excretory system in Protochordates.
- C. Adaptive radiation in vertebrates – fishes, amphibians, reptiles, aves and mammals

UNIT II

8hrs

- A. **Integument and its Derivatives:** Epidermal Integument or Skin Functions, Structure & its Derivatives (Glands, Scales and scutes, digital cornifications, horns, feathers, hairs), Integument in different classes of Chordates.
- B. **Nervous system-** Development of Brain, spinal cord, Peripheral nerves and sense organs

UNIT III

8hrs

- A. **Respiratory System:** Introduction Respiratory organs: Gills (Internal or true gills, External or Larval gills). Lungs and Ducts, Accessory Respiratory organs and Swim Bladders.
- B. **Circulatory system:** Evolution of heart and aortic arches

UNIT IV

8hrs

- A. **Digestive System:** Introduction Embryonic Digestive Tract Alimentary Canal: Divisions, Digestive Glands
- B. **Urinogenital System:** Vertebrate kidneys and ducts, Gonads and their ducts

CHORDATA PRACTICALS

4x16=64 Hrs

1. **Protochordates:** Specimens: 1) *Amphioxus*, *Herdmania*

Slides- *Salpa* (sexual), *Doliolum*

2. **Fishes :** 1) *Rhinobatus* 2) *Hippocampus* 3) Goldfish (aquarium fish) 4) *Clarius*

5) *Anabas* 6) *Coffe* fish 7) *Acipenser* 8) *Periophthalmus* 9) *Triacanthus*

10) *Notopterus* 11) *Exocoetus* 12) *Diodon hystrix* 13) *Echeneis neucrates*

3. **Amphibians :** 1) *Ichthyophis* 2) Axolotl Larva 3) *Rana tigrina* 4) *Amblystoma*

4. **Reptiles :** 1) *Calotes* 2) *Mabuya* 3) Chameleon 4) *Phrynosoma* 5) *Chelone mydas*

5) *Varanus* 6) *Naja naja* 7) Krait 8) *Hydrophis* 9) Viper

5. **Birds :** 1) Blue jay 2) Indian koel -male and female 3) Kite

6. **Mammals :** 1) Guinea pig 2) Domestic cat 3) Loris 4) *Megaloderma lyra* (bat)

5) Pangolin

7. **Integuments of vertebrates:** Scales of fish, Hoofs, nails, horns, claws,

plastron and carapace of tortoise, snout of saw fish

8. **Osteology :**

1) **Skull and lower jaw:-** a) Crocodile b) Bird c) Carnivore mammal (dog)
d) Herbivore mammal (horse)

2) **Types of vertebrae:-** a) Procoelous b) Ophisthocoelous c) Amphicoelous
d) Amphiplatian e) Heterocoelous f) Axis and atlas vertebrae.

II. Study of following systems in rat by employing computer animation/charts:

- a) Circulatory system b) Nervous system c) Reproductive system
- d) Digestive system e) Sense organs f) Urinary system

REFERENCES :

1. Alexander, R. M. 1975. The Chordata. Cambridge University Press, London.
2. Barrington, E.J.W. 1965. The Biology of Hemichordata and Protochordata, Oliver and Boyd, Edinburgh.
3. Colbert, E. H, 1969. Evolution of the vertebrates, John Wiley and Sons, Inc., N.Y.
4. Kent, C. G. 1954. Comparative anatomy of vertebrates
5. Kingsley, J.S. 1962. Outlines of Comparative anatomy of vertebrates. Central book depot Allahabad.

M.SC., II SEMESTER HC – 2.2 ANIMAL PHYSIOLOGY

32 Hrs

Course Outcome:

After completing the course student will be able to

1. Understand the mechanism of transport of molecules, stepwise release of energy , aerobic and anaerobic respiration
2. Describe the physiology of digestive and respiratory system of human beings.
3. Understand the blood composition, types, groups and circulatory system.
4. Describe the physiology of excretory system and nervous system of human beings.
5. Know the physiology of sense organs, muscles and reproductive system.

UNIT I: Membrane Transport, Bioenergetics & Circulation

8 Hrs

A. Membrane Transport:

Molecular mechanisms of passive and active transport.

B. Bioenergetics:

- a) Energy – Concept, laws of thermodynamics
- b) Redox potential
- c) Stepwise release of energy through cytochromes, production of ATP, uncoupling of oxidative phosphorylation, inhibitors.
- d) Anaerobic and aerobic breakdown of glucose, alternate pathway – HMP shunt and glucuronic acid pathway.
- e) Citric acid cycle as common metabolic pathway.

C. Circulation:

- a) Major types of body fluids and their composition.
- b) Neurogenic and myogenic hearts.
- c) Mammalian heart – cardiac cycle, ECG.

UNIT II: Physiology of excitation & Excretion

8 Hrs

A. Muscle Physiology:

- a) Molecular organization of sarcomere.
- b) Mechanism of contraction with emphasis on sliding filament and Davies models, regeneration of storage phosphate.
- c) Physiological adaptations of muscles for jumping, swimming and flight.

B. Neurophysiology:

- a) Axonal and synaptic transmission of nerve impulses.
- b) Synaptic integrity, synaptic plasticity.
- c) Molecular mechanism of sensory transduction and neural output in receptor cells.

C. Excretion:

- a) Comparative physiology of excretion in animals- Nitrogenous wastes and waste elimination.
- b) Mammalian kidney- Structure and physiology of urine formation.

Unit III: Basic Concepts of Endocrinology

8 hrs

A. Chemical messengers:

Autocrine, Paracrine and endocrine secretions,
Types of hormones, an overview of human endocrine system

B. Hormone synthesis: Peptide and steroid hormones.

Role of Hormones in homeostasis- Glucose and Water balance

C. Hypothalamus and pituitary gland:

Structure, function and control of hypothalamic hormones.
Pituitary hormones and their physiological actions
chemical structure and. Feedback regulation. Pathophysiology.
Hypothalamo - hypophysial portal system

D. Pineal gland–Structure and function.

Unit IV:

8 hrs

A. Thyroid gland: Structure, function and biosynthesis of thyroid hormone

B. Parathyroid : Structure and PTH – Calcitonin – Role of hormones in calcium and phosphate metabolism.

C. Adrenal gland hormones

Adrenal cortex hormones: Corticoids: role played in Stress management – Aldosterone and the rennin- angiotensin system

Adrenal medullary hormones: Catecholamines as emergency hormones

D. Gastrointestinal hormones: Secretion, control and function

E. Pancreatic Hormones: Insulin and glucagons, their role in the regulation of Carbohydrate, protein and lipid metabolisms.

ANIMAL PHYSIOLOGY PRACTICALS

4x16=64 Hrs

1. Estimation of Proteins by Lowry *et al* method. (in tissue sample from slaughter house)
2. Determination of serum cholesterol. (Clinical sample)
3. Determination of glucose content by Anthrone method. ((in tissue sample from slaughter house)
4. Estimation of liver and skeletal muscle glycogen. (in tissue sample from slaughter house)
5. Determination of serum/ blood urea by DAMO method. (Clinical sample)
6. Estimation of creatinine in the urine sample.
7. Total count of RBC and WBC.
8. Differential count of WBC
9. Response of RBC's to Hypertonic, hypotonic and isotonic solutions
10. Observation of permanent slides of T.S of endocrine glands
 - a. Pituitary gland
 - b. Thyroid gland
 - c. Adrenal gland
 - d. Pancreas
11. Identification of chemical structures of steroid hormones

REFERENCES:

1. Adler N. T (1981) Neuroendocrinology of Reproduction, Physiology and Behaviour. Austin, C. R and R. V. Short (eds) (1972) Reproduction in mammals. (1) Germ cells and Fertilization (2) Embryonic and Foetal development (3) Hormones in Reproduction (4) Reproduction pattern (5) Artificial control of reproduction, Cambridge University press, London.
2. Barrington, E. J. W (1976) An introduction to general and comparative endocrinology, Oxford University press, London
3. Raghavendra Puri (2003) Mammalian endocrinology Vol. I & II, Dominant Publishers and Distributors, New Delhi.
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**M.Sc., II SEMESTER
HC – 2.3 ENTOMOLOGY**

32hrs

Course Outcome:

After completing the course student will be able to

1. Understand insects encountered in agricultural fields.
2. Envisage an insight on economically important pests of various foods, fiber and household
3. Understand various insect pest management methods and its significance
4. Learn to apply various agricultural equipment and understand the effect of chemicals and its dosages in agricultural pest management
5. Learn to apply the pest control methods wisely to minimise ecological backlash
6. Discuss the evolutionary significance of insect plant interaction and insect animal interaction.

Unit I: General Entomology **10 hrs**

A. Classification of class Insecta up to orders with suitable examples; Integument appendages.

B. Insect Endocrinology

- I. Insect Hormones and their regulation: Chemistry and functions of hormones, Hormones in metamorphosis, Ecdysis and Diapause
- II. Semiochemicals:: Allelochemicals and Pheromones (Primer & releaser)

Unit II: Agricultural Entomology **10hrs**

A. Role of insects in plant pollination

B. Insects pests: Classification and categories of pests, origin and emergence of pests, pest out breaks and pest resurgence
Structure, life history, significance, nature of damage and control methods of major pests of sugarcane, Paddy and Coconut.

C. Structure, life history, significance, nature of damage and control measures of stored grain pests: (a) *Sitophilus* (b) *Trogoderma* (c) *Rhizopertha* (d) *Tribolium* (e) *Bruchus* (f) *Sitotruga* (g) *Ephestia*

Unit III: General and household insect pests **06hrs**

A. Structure, life history, significance, nature of damage and control measures of following general pests: (a) grasshoppers & locusts (c) termites (d) aphids (e) hairy caterpillars

B. Household pests: Cockroaches, Ants, Wasps, Silverfish, furniture beetle, and their control

Unit IV: Medical Entomology **06hrs**

A. Insect vectors: Role of insect as vectors of human diseases (Malaria, filariasis, Kala azar and their control)

Mosquitoes as pests and their control.

Housefly: A human health hazard and its control

B. Arboviral diseases: Dengue, chicken gunya, swine flu.

PRACTICALS: **4x16=64 Hrs**

1. Collection and preservation of dead insects for systematic studies & field report 4x4=16 hrs

2. Identification of different insects upto orders- House fly, Cockroach :

Mosquitoes, stored grain beetles, destructive insects, important crop and household pests

4x4=16 hrs

4. Fixing and preservation of dead insects by Plastination technique. 4x4=16 hrs
 5. Field studies of insects to understand their habit: Ants, Butterflies, termite, wasps, Moths. 4x2=08 hrs
 6. Study of insect mouth parts: Mosquito, Cockroach, House fly, Butterfly 4x2=08 hrs

REFERENCES:

1. Awasti V.B. 2009 Introduction to general entomology 3rd Ed. Scientific publication (India), Jodhpur
2. Awasti V.B. 2007, Agricultural Insect Pests and their control. Scientific publishers (India) Jodhpur
3. Trigunayat M.M. 2009, A Mannual of practical entomology, scientific publishers, Jodhpur, India.
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5. L . K Jha. Applied Agricultural Entomology. New central book agency. Culcutta

**M.Sc., II- SEMESTER
 SC – 2.4 DEVELOPMENTAL BIOLOGY**

48 Hrs

Course Outcome:

After completing the course student will be able to

1. Understand the molecular concepts of developmental biology during fertilization.
2. Know about Noble prize concepts during frog development viz., Nucleocytoplasmic interactions.
3. Explain on axis development in drosophila.
4. Describe endocrine and molecular control in metamorphosis of insects and amphibians.
5. Explain the various stages of chick embryonic development.

Unit I:

- A) Introduction : Descriptive V/s. Experimental Embryology **8hrs**
 B) Fertilization : a) An overview of structure and differentiation of egg and sperm
 b) General sequence and molecular events during fertilization

Unit II: Early development - I

8 hrs

- a) Nucleocytoplasmic interactions in early development: An overview of Nuclear transplanted experiments in Amphibians and mammals
- b) Creations of multicellularity: Cleavage-Regulatory mechanism
- c) Gastrulation: Morphogenetic movements and regulatory mechanisms in amphibian and mammalian embryo.

Unit III: Early development - II

8hrs

- a) Morphogenetic determinants and their role in development:
 Yellow cytoplasm in Ascidians, Polar body in Mollusca, Pole plasm in *Drosophila*
- b) Laying down the embryonic body plan :
 Determination of embryonic axes in *Drosophila* – Anterior-posterior (maternal effect genes) & Dorsoventral; Amphibians (cell-cell interaction) & Mammals (Hox Genes)
- c) Cell lineage studies and cell death genes in *Caenorhabditis elegans*.

Unit IV: Morphogenesis –I

8 hrs

- a) Early embryogenesis in *Drosophila* : Regional specification by. Segmentation genes: Gap genes, Pair rule genes, Segment polarity genes, and Homeotic genes.
- b) Cellular differentiation and morphogenesis:
 - i. Neuronal v/s epidermal fate specification in *Drosophila*.

ii. Vulval induction in *Caenorhabditis elegans*.

Unit V: Morphogenesis-II

8 hrs

- a) Role of Cell Adhesion molecules in morphogenesis : Cadherins and Fibronectins
- b) Genetics of imaginal discs and transdetermination
- c) Limb development-an over view :
 - i. Proximo-distal axis specification in developing limb.
 - ii. Cell death and formation of digits.

Unit VI: Post embryonic development

8 hrs

- a) Metamorphosis : Endocrine and molecular control of metamorphosis in insects and amphibians
- b) Types of growth
- c) Regeneration : Types, Blastema formation, Sources of cells for regeneration
- d) Abnormal development as seen in Teratogenesis.

PRACTICALS

16X2=32Hrs

- 1. Study of internal changes during early development of frog & chick (permanent slides) 3X2=06hrs
- 2. Development of chick-Embryo mounting-permanent preparation 2X2=04hrs
- 3. Study of early developmental stages of *Drosophila* (Live Observation of embryo) and dechoriation and observation of embryos 2X2=04hrs
- 4. Study of Imaginal discs – the precursors of adult structures in *Drosophila* 3X2=06hrs
- 5. Demonstration of window technique to observe chick embryo development 2X2=04hrs
- 6. Effect of thyroid hormone on development in frog 2X2=04hrs
- 7. Study of various developmental stages in frog up to tadpole stage 2X2=04hrs

REFERENCES:

- 1. Balinsky, B.I., 1965. An introduction to embryology, W.B.Saunders company.
- 2. Gilbert, S. F. 2006, Developmental Biology, 8th Ed. Sinauer Associates Inc.,
- 3. Kalthoff, 2000, Analysis of Biological Development, 2nd Ed., McGraw-Hill Science, New Delhi, INDIA. Massachusetts, USA.
- 4. Vasudeva Rao, 1994. Developmental Biology: A modern synthesis, Oxford & IBH, New Delhi.
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- 6. Wolpert, L, Beddington, R Jessell, T. Lawrence P, Meyerowitz, E, Smith J., 2001, Principles of Deveopment Oxford University Press Oxford.
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**M.Sc., II SEMESTER
SC – 2.5 IMMUNOLOGY**

48hrs

Course Outcome:

After completing the course student will be able to

- 1. Outline the key components of the innate and adaptive immune responses.
- 2. Describe about cell types and organs which are involved in an immune response—
- 3. Describe the Infectious diseases, hypersensitivity, autoimmune disorders,— immunodeficiency diseases

- Unit I: Introduction to immunity** **8hrs**
- A. History; types of immunity – Innate and acquired immunity.
 - B. Cells and Organs of immune system: Cells: Lymphocytes (T & B cells), monocytes, macrophage; eosinophills, basophills, neutrophils and mast cells.
 - C. Primary and secondary lymphoid organs: Bone marrow, Thymus, Spleen, Lymph nodes
- Unit II: Antigens and Immunoglobulins** **8hrs**
- A. Antigens: factors influencing immunogenicity, adjuvant, epitope, hapten
 - B. Immunoglobulins: Basic structure of the immunoglobulin;
Types and functions of immunoglobulins.
 - C. Monoclonal antibodies:Antigen-antibody reactions
- Unit III: Immune response** **8hrs**
- A. Humoral and cell mediated immune responses
 - B. Primary and secondary immune modulation; Cytokines; role of complement system in immune response (Classical pathway, Alternate pathway);
 - C. Immune response against bacterial (tuberculosis), parasitic (malaria) and viral (HIV) infections;
- Unit IV Immunotechniques** **8hrs**
- A. Agglutination; Precipitation;
 - B. Immunofluorescence; RIA, ELISA, Immuno-electrophoresis and Western blotting.
- Unit V Major histocompatibility complex and Hypersensitivity** **8hrs**
- A. Transplantation and graft rejection,
 - B. Genetic organization of H2 and HLA complexes, HLA typing;
 - C. Immediate and delayed hypersensitivity.
- Unit VI Vaccines and Vaccination** **8hrs**
- A. Types of Vaccines and their significance
 - B. Vaccine delivery systems.
 - C. Congenital and acquired immunodeficiencies

TUTORIALS **2X16=32 Hrs**

References:

1. Austyn, J.M. and Kathym, J. Wood. 1993. Principles of cellular and molecular Immunology. Oxford University Press. Oxford.
2. Benjamin, Elisunshine, Geoffrey Leskowitz.1996. Immunology: A short course. 3rd Edition. New York.
3. Kubey, J.M. 1990. Essential Immunology. 6th Edition. Blackwell Scientific Publication, New York.
4. Rao, C.V. 2002. An introduction to Immunology. Narona Publishing House, New Delhi.
5. Rotti, I. 1994. Essential Immunology. Blackwell, London.
6. Stibes, D.P. and Terr, A.I. 1991. Basic and Clinical Immunology. 7th Edition. Appleton and Large. California.

M.Sc., II SEMESTER
SC – 2.6 EVOLUTIONARY BIOLOGY

48 Hrs

Course Outcome:
After completing the course student will be able to

1. Understand that many of the organisms that inhabit the Earth today are different from those that inhabited it in the past
2. Understand that the propositions underlying Darwin's theory of evolution.
3. Explain adaptation, providing examples from several different fields of biology
4. Explain how the molecular record provides evidence for evolution
5. Understand the Human origin and evolution.

UNIT I Emergence of concept of evolution: 8 Hrs

- A. Pre Darwinian concepts, Darwinism and its impact in the development of synthetic theory.
- B. Neodarwinism: Birth of population genetics, Components of population genetics, Mendelian population, gene pool, allele frequencies and genotype frequencies,

UNIT II Speciation: 8 Hrs

- A. Concept of species,
- B. Types of species
- C. Models of speciation,
- D. Patterns and mechanisms of reproductive isolation,
- E. Hybridization, polyploidy and speciation.

UNIT III Molecular evolution 8 Hrs

- A. Phyletic gradualism and punctuated equilibrium.
- B. Micro and macroevolution.
- C. Molecular evolution: Selectionists theory of evolution, Neutral theory of evolution and Molecular clock and emergence of non-darwinism,

UNIT IV Phylogeny 8 Hrs

- A. Phylogenetic trees : Construction with nucleic acid and amino acid sequences,
- B. Types of trees and Techniques employed in construction of phylogenetic trees,
- C. Molecular phylogenetics of Homo sapiens.

UNIT V Population genetics and Evolution 8 Hrs

- A. Gene pool, gene frequency, Hardy-Weinberg Law.
- B. Destabilizing forces of evolutionary equilibrium (Mutation, Migration, Selection, Meiotic drive and genetic drift).
- C. Founder effect, Isolating mechanisms and speciation.
- D. Micro Macro and Mega evolution, Co-evolution.

UNIT VI Genome and Evolution 8 Hrs

- A. Genes and gene clusters
- B. Origin of new genes by gene duplication (Ohno's concept)
- C. Selfish DNA
- D. Karyotypic evolution (Drosophila).

TUTORIALS 2X16=32Hrs

REFERENCES:

1. Dobzhansky Th, (1951) Genetics and origin of species, 3rd Edn. Chapman and Hall, London.
2. Dobzhansky Th, Ayala F.J, Stebbins G.L and J.M. Valentine, (1976) Evolution, Surjeet Publication, New Delhi.
3. Futuyama D.J (1986) Evolutionary Biology, Sinuauer Associates Inc. USA
4. Hartl D.L (2000) A primer of population genetics, Sinuauer Associates Inc. USA
5. Jha A.P (1992) Genes and Evolution - John Wiley Publicaion, New Delhi

6. King M (1993) Species evolution - The role of chromosomal change. The Cambridge University Press, Cambridge

M.Sc., III SEMESTER
HC – 3.1 MOLECULAR BIOLOGY AND BIOTECHNOLOGY

32 hrs

Course Outcome:

After completing the course student will be able to

1. Know nucleic acids, DNA replication and its mechanism.
2. Understand transcription and its modifications.
3. Explain genetic code, enzymes, factor and the process of translation.
4. Analyse gene regulation, lytic and lysogenic cycles in prokaryotes.
5. Understand gene regulation in eukaryotes.
6. Explain molecular mechanism of DNA damage repair.

Part A: Molecular Biology

Unit I Introduction to nucleic acids

8hrs

- A. DNA Replication: i) Enzyme components of replication unit ii) Mechanism with emphasis on Dna A in initiation, Co-ordinated synthesis, End replication in eukaryotes iii) Fidelity.
- B. Transcription: i) Transcription apparatus and process (RNA polymerase, cisregulatory elements, terminators, transcription factors). ii) Post transcriptional modifications of mRNA in eukaryotes (G-cap, Poly tail, Splicing).
- C. Translation: i) Genetic code (major features, usage of different codons). ii) Enzymes, factors and the process (Aminoacyl t-RNA synthetase, Peptidyl transferase, IFs, EFs, RFs and Ribosome)

Unit II Gene regulation

8hrs

- A. Gene regulation in Prokaryotes: (i) Regulation at transcription initiation: Eg. lac operon (+ve and -ve control) (ii) Regulation beyond transcription initiation: trp attenuator (iii) Regulation in Lambda Phage - Lytic and lysogenic cycle induction.
- B. Gene regulation in Eukaryotes: (a) Transcriptional activators (b) Transcriptional repression: (i) direct repression, indirect repression (ii) Gene silencing by modification of histones and DNA (c) RNA interference
- C. Molecular basis of homologous recombination: Models and protein machinery
- D. Molecular mechanisms of DNA damage repair.

Part B: Biotechnology

Unit III:

8 hrs

A. Genetic engineering:

Definition, objectives and outline of recombinant DNA technology procedure.

Enzymes: Restriction Enzymes; DNA ligase, Klenow enzyme,

T4 DNA polymerase, Polynucleotide kinase, Alkaline phosphatase.

Cloning vectors: Plasmids, Phages, Cosmids, Phagemids, Artificial chromosomes (YAC, BAC, HAC),

B. Cloning:

Construction of Genomic and cDNA libraries.

Identification of Recombinants: Genetic selection, Use of chromogenic substrates, Insertional inactivation.

Analysis of recombinant DNA clones: Characterization of clones, Restriction mapping, Southern hybridization.

Polymerase chain reaction and DNA sequencing-Maxam and Gilbert's method, Sanger's method, Automated DNA sequencing

Unit IV:**8 hrs****C. Applications of Biotechnology:**

Production of medicinally important products – vaccines, Gene therapy, AIDS therapy, Biofertilizers, biopesticides, medicine and human health

D. Animal Biotechnology

Animal cell and Tissue culture: Principles of cell culture, cell and tissue types, cell lines, transformation.

Cell and tissue culture media: Natural and defined, role and components of serum in culture.

Applications of tissue culture: Tissue culture in biomedical research karyological studies, amniocentesis, mutagenesis, Cytotoxicity assays.

PRACTICALS**4x16=64 Hrs**

1. Extraction of DNA by rapid method.
2. Extraction of DNA by standard method.
3. Estimation of DNA concentration by Diphenylamine method.
4. Localization of DNA in prefixed paramecium slides by Feulgen staining
5. Localization of nucleic acids in prefixed paramecium slides by Toluidine blue staining
6. Estimation of RNA concentration by Orcinol method
7. PCR amplification of DNA and gel electrophoresis.
8. Restriction digestion and gel electrophoresis.
9. Isolation of plasmid DNA from bacteria.
10. Molecular biology problems

REFERENCES

1. Griffiths A J F, H. J. Muller, D. T. Suzuki, R. C. Lewontin and W. M. Gelbart 2000. An introduction to genetic analysis. W. H. Greeman. New York.
2. Lewin, B 2003 Genes VIII. Oxford University Press. Oxford
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4. Das, H.K. 2007. Text book of Biotechnology. Wiley India Pvt. Ltd. New Delhi
5. Freshney, Ian, R. 2006. Culture of Animal Cell (5th edn). Wiley- Liss publications
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**M.Sc., III SEMESTER
HC – 3.2 REPRODUCTIVE BIOLOGY**

32 hrs**Course Outcome:**

After completing the course student will be able to

1. Understand structure and function of reproductive organs
2. Explain the structure of reproductive cells
3. Describe the role of internal cues in reproduction
4. Describe the role of external factors in reproduction
5. Analyse the role of endocrine glands and their secretions in reproduction
6. Identify the factors affecting fertility
7. Know different types of assisted reproductive technologies.

UNIT I: Male reproduction: **8 hrs**

- A. Functional morphology of male reproductive system
- B. Kinetics of spermatogenesis – wave and cycle
- C. Hormonal control of mammalian testis and spermatogenesis
- D. Ultrastructure of spermatozoa
- E. Abnormalities of sperm
- F. Brief description of histomorphology and hormonal control of male accessory organs viz., epididymis, vas deferens, seminal vesicles, ventral prostate, bulbourethral gland and preputial gland
- G. Sperm maturation – morphological and biochemical events, influence of accessory organ secretions
- H. Biochemistry of semen and capacitation

UNIT – II Female reproduction : **8 hrs**

- A. Origin and migration of primordial germ cells; genetic and hormonal control of differentiation of gonads and gonadal ducts in mammals.
- B. Female Reproductive System-Functional morphology of mammalian ovary, Fallopian tube and uterus.
- C. Ovarian steroid hormones and their actions

UNIT III: Reproductive cycles in Mammals: **8 hrs**

- A. Comparison of estrous and menstrual cycles
- B. Menstrual cycle : Different phases, changes in the ovary and uterus and hormonal control
- C. Implantation – Process, Types and hormonal control
- D. Pregnancy – length of gestation, hormonal control
- E. Parturition – Process of birth and influence of hormones
- F. Lactation – Hormonal control of mammary gland, development and lactogenesis

UNIT – IV: Fertility and reproductive management **8 hrs**

- A. Fertility control – Need, principles of different male and female temporary and permanent contraceptive methods.
- B. Assisted Reproduction: Causes of infertility, Artificial insemination, different methods of assisted reproduction (*In-vitro* Fertilization, Gamete Intra Fallopian tube Transfer, Zygote Intra Fallopian tube Transfer).

PRACTICALS **16X4=64 hrs**

1. Demonstration of surgical technique by video clipping
2. Counting of spermatozoa in semen sample collected from volunteers
3. Staining of spermatozoa for abnormalities in semen samples collected from volunteers /clinical samples
4. Study of different contraceptive devices
5. Observation of permanent Histology slides
 - a. Comparative morphology of ovary
 - b. Comparative morphology of testis
 - c. Comparative study of male accessory organs
 - d. Comparative study of female accessory organs
6. Observation of permanent slides of T.S of endocrine glands
 - a. Pituitary gland
 - b. Thyroid gland
 - c. Adrenal gland
 - d. Pancreas

REFERENCES

1. Adler N. T (1981) Neuroendocrinology of Reproduction, Physiology and Behaviour.
2. Austin, C. R and R. V. Short (eds) (1972) Reproduction in mammals. (1) Germ cells and Fertilization (2) Embryonic and Foetal development (3) Hormones in Reproduction (4) Reproduction pattern (5) Artificial control of reproduction, Cambridge University press, London.
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4. Raghavendra Puri (2003) Mammalian endocrinology Vol. I & II, Dominant Publishers and Distributors, New Delhi.
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8. Jones, R. E (1991) Human Reproductive Biology press N.Y
9. Knobil, E and Neil J. D (1994) The physiology of reproduction, Vol. I & II. Raven press, New York.

**M.Sc., III SEMESTER
HC – 3.3 ECOLOGY AND WILDLIFE**

32 hrs

Course Outcome:

After completing the course student will be able to

1. Demonstrate and Understand ecological relationships between organisms and their environment.
2. Present an overview of diversity of life forms in an ecosystem.
3. Explain and identify the role of the organism in energy transfers
4. Describe the Habitat ecology and Resource ecology
5. Understand the types of environmental Pollution and their management
6. Scope, Values and Conservation strategies of wildlife.

Part-A Ecology

UNIT - I

8 hrs

A. Ecosystem: Historical account, Scope, Basic concepts and Approaches to the study of Environmental Biology. Components of Environment - An overview of abiotic factors and Biotic factors. Concepts of habitat and Ecological niche. Ecotone and Edge effect. Food chains, Food-webs and their structure in Ecological Pyramids in aquatic, terrestrial and parasitic Environments.

B. Population Ecology: Introduction. An overview of important population attributes – Density, Natality, Growth rates, Growth forms and concept of carrying capacity, Patterns in human population growth and its explosion -Remedial measures. Mortality - life tables and survivorship curve, sex ratio, age distribution, dispersal and dispersion, aggregation and Allee's principle, population fluctuation and cyclic oscillations and Population interactions.

UNIT - II

8 hrs

A. Community Ecology Concept of community - community structure and attributes, concept of climax Species diversity in community and it's measurement- Alpha diversity- Simpson's diversity index, Shannon index, Fisher's alpha, rarefaction. Beta diversity- Sorensen's similarity index, Whittaker's index, Evenness, Gamma diversity. Drivers of species diversity loss and conservation.

B. Bioecology of Freshwater Zooplankton: Definition, Types and adaptations of Zooplankton. Brief study of organizations, life cycles and Ecological importance of Rotifers, Cladocerans, Copepods-Calanoids, Harpacticoids and Cyclopoids, and Ostracods. Mass culturing of Zooplankton.

C. Microbial Ecology: Ecological role, beneficial and pathogenic Microorganisms. Indicator Microorganisms. Role of microorganisms in biodegrading and bioremediation of organic and metal pollution.

Part B Wildlife Biology

UNIT – III

8hrs

- A.** Scope and values of wildlife (Ecological, Aesthetic, Scientific, Recreational, Medicinal)
- B.** Causes of wildlife depletion: Degradation and destruction of natural habitats, Exploitation for commercial purposes, Deforestation, Agricultural expansion, Urbanization and Industrialization, forest fires and hunting.
- C.** Wildlife corridors, Human-wildlife conflicts
- D.** Wildlife awareness and education, Wildlife and tribal welfare

UNIT – IV

- A.** Conservation strategies: Red data book, protected area network, Role of NGOs in conservation.
- B.** Wildlife act and legislation: Wildlife Protection Act 1972; Biological Diversity Act 2002.
- C.** Wildlife conservation projects in India (with special reference to Project Tiger, Project Hungul and Gir Project)
- D.** In-situ conservation: Bioreserves, National parks, Wildlife sanctuaries and Safari's in India
- E.** Management of Bioreserves, National parks, Wildlife sanctuaries and Safari.
- F.** Ex-situ conservation: Zoo garden, Management of Zoos, Captive breeding, Artificial insemination, Cryopreservation (techniques and applications) Germplasm banks,

PRACTICALS:

4X16=64 Hrs

1. Qualitative and Quantitative study of freshwater planktons.
2. Determination of species diversity by Shannon-Weiner Index
3. Determination of species diversity by Simpson's index
4. Field visit to Sewage pond, Natural lake (and if possible river): Collection of water samples and study of physico-chemical parameters such as colour, pH, temperature, conductivity, total solids and turbidity
5. Estimation of Dissolved Oxygen in three natural (sewage, pond and Tap) water samples.
6. Estimation of free Carbon di-Oxide in three natural (sewage, pond and Tap) water samples.
7. To study the relationship between Dissolved Oxygen and free Carbon di-Oxide, if any, in three natural (sewage, pond and Tap) water samples.
8. Determination of BOD in three natural (sewage, pond and Tap) water samples
9. Determination of COD in three natural (sewage, pond and Tap) water samples
10. To study the relationship between BOD and COD, if any, in three natural (sewage, pond and Tap) water samples
11. Collection, observation of planktons (Phytoplankton and Zooplankton) from polluted and non-polluted water bodies.
12. Estimations of bacterial abundance in different water samples – using DMT.
13. Visit to RMNH, Mysore, to study models of freshwater, marine, estuarine and terrestrial habitats.
14. Survey of Animal Population - to visit different habitats/areas in and around Mysore and collect data on some population attributes, application of Bio-statistical tests to the collected data and its interpretation.
15. Visit to nearby Zoological garden, wildlife sanctuaries, Animal rehabilitation centres.

REFERENCES

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12. Saharia, V.B. 1982. Wildlife in India. Natraj Publishers. Dehara Dun.

M.Sc., III SEMESTER SC 3.4 ETHOLOGY

48 Hrs

Course Outcome:

After completing the course student will be able to

1. Evaluate the learning and instinct behavior.
2. Explain the mechanisms in instinct and behaviour
3. Explain how animals learn
4. Compare learning and instinct behaviour.
5. Analyse any problem about animal behaviour
6. Explain the importance of evolution for animal behaviour.
7. Explain evolution and behaviour.
8. Explain natural selection and behaviour.
9. Explain the relationship between predators and prey
10. Explain social behaviour.

UNIT - I

8 Hrs

- A. Descriptive versus experimental approaches
- B. Reflexes and complex behaviour- Latency, after discharge, summation, warm up, fatigue inhibition and feedback control
- C. Instinctive Behaviour - Fixed action pattern, Types of sign stimuli and releasers as triggers, Genetic basis of instinctive behaviour.

UNIT- II

8 Hrs

- A. Development and behaviour- Causes of behavioral changes during development, development of bird song.
- B. Learning- Classical conditioning experiment, latent and insight learning. Social learning, learning sets and play.
- C. Importance of early experience – Critical period- Filial imprinting, Sexual imprinting in birds, Imprinting like process in mammals.

UNIT- III Foraging and anti-predator behaviour

8 Hrs

- i. Anti predator behaviour – avoiding detection through colour and Markings (Mullarian mimicry)

- ii. Warning coloration
- iii. Batesian mimicry

UNIT-IV Biological communication **8 Hrs**

- i. Forms of signals,
- ii. Visual communication with suitable examples,
- iii. Auditory Communication
- iv. Tactile and Chemical communication

UNIT -V Sexual Behaviour **8 Hrs**

- i. Hormones and sexual behaviour – Selected examples of courtship and mating behaviour.
- ii. Pheromones in Insects and Mammals
- iii. Lee Boot, Whitten, Bruce, Collidge and Castro-Vandenberg effect/s
- iv. Selected examples of courtship and mating behaviour

UNIT-VI Social Behaviour **8 Hrs**

- i. Introduction
- ii. Advantages of grouping
- iii. Social organization in insects with special reference to ants and honeybees
- iv. Social organization in sub human primates
- v. Altruism, Kin selection and Genetic control of behaviour

TUTORIALS – On the basis of the proposed chapters. **2x16 = 32 Hrs.**

REFERENCES

- 1) Goodenough J.E., Mc Guire B. and Wallace R. A. (1993) Perspectives on Animal Behaviour. John Wiley and sons, New York.
- 2) Tinbergen (2006) Social behaviour in Animals. J.V. Publishing House Jodhpur India.
- 3) Vandenberg. J.E.(Ed) (1983). Pheromones and Reproduction in mammals. Academic Press. NewYork.
- 4) Agrawal, K.C. 2000. Biodiversity. Agrobios. India.

M.Sc., III SEMESTER
SC – 3.5 POLLUTION AND TOXICOLOGY

48hrs

Course Outcome:

After completing the course student will be able to

1. broader understanding of how science and the scientific method work to address environmental problems.
2. Earth's major systems (ecosystems and biogeochemical cycles), how they function and how they are affected by human activity (population growth, air, water and soil pollution, ozone depletion, global warming, and solid waste disposal).
3. the interaction of human society (urban sprawl, energy use/generation, resource consumption and economics) with the Earth's systems.

Part A - Pollution **24 hrs**

Unit I: **8 hrs**

A. Concept of Biosphere: Its components, hydrosphere, atmosphere, and lithosphere, Origin of life in the biosphere.

B. Water pollution: Definition, sources Types and classification of pollutants. Effects of Water Pollution, River Pollution, Oxygen sag curves and Eutrophication Drinking water: Collection, purification and distribution. Wastewater treatment: Primary, secondary and tertiary treatment.

Unit II:**8 hrs**

A. Atmospheric pollution: Primary and secondary air pollutants. Biological effects of Nox, SO_x, SPM, Hydrocarbons, Acid rain, Global warming, Photochemical smog and Ozone hole.

B. Solid waste and Biomedical waste: Sources, collection, transport, treatment and Disposal methods.. Noise Pollution: Sources, Biological effects, Control measures and OSHA standards.

Unit III:**8hrs**

A. Radiation & Thermal pollution: Sources, types, effects, Atmospheric fallout and abatement.

B. Environmental Impact Assessment: Basic elements, Methods Guideline for industrial EIA, Aquaculture related EIA, Transport related EIA and Water related EIA. Case studies: Konkan Railway, Silent valley, Bhopal Tragedy and Love canal tragedy, Mangalore Bojpe tragedy

Part B – Toxicology**24 hrs****Unit IV:****8hrs**

A. General Principles of Toxicology: Introduction, Definition of toxicology Importance of Dose and Dose-response, factors influencing toxicity, Bioassay-toxicity evaluation studies using fish as model.

B. Toxic compounds: Heavy metals-Lead and mercury, Hydrocarbons- Aromatic and Aliphatic, and cyanides, and toxic gases - Bhopal tragedy.

Unit V:**8hrs**

A. Biotransformation: Bioactivation, Biotoxification of organo phosphates and organo chorines in the bodies of animals.

B. Natural toxins, Venoms and poisons: Properties and their effects, Major Sites and mechanism of action, Toxins in lower and higher organisms, Toxin and Venom therapy.

Unit VI:**8hrs**

A. Smoking aids: Active and Passive smoking, Consumption of tobacco, Mariguana(Ganja), their effects and Prevention measures.

B. Cosmetics: Types of cosmetics, Chemical Characteristics, Applications, Exposure and risk assessment, Cosmetic safety regulations.

C. Risk assessment: Exposure assessment, Dose-Dosage, Risk characterization, Risk analysis and communications, Occupational health and illness.

TUTORIALS – On the basis of the proposed chapters**2x16 = 32 Hrs****REFERENCES:**

1. Nandini, .N. Sunitha N. and T. Sucharita 2010. Environmental Studies, Sapna Book House Bangalore
2. Frant C.L.V. 1991, Basic Toxicology II (Eds.), Hemisphere publishing corporation, Washington, London
3. Sambasiva Rao K.R.S. 1999. Pesticide impact on fish metabolism. (Eds.) Discovery Publishing House, New Delhi.
4. Bio-pesticides in Insect Pest Management 1999. S. Ignacimuthu and Alok Sen, Phoenix Publishing House Pvt., Ltd., New Delhi.
5. APHA, AWWA and WEF. 1992: Standard Methods for Examination of Water and Wastewater, XVIII Ed, American Public Health Association. NY, USA
6. Nebel, B.T. and Wrigly R.T. 1998. Environmental Science, VI Ed. Prentice Hall New Jersey, USA

7. Hosetti, B.B. 2001. A Text Book of Applied Aquatic Biology, Daya Publishing House, Delhi.
8. Hassall, K.A. 1990. The Biochemistry and uses Pesticides structure, metabolism and Mode of action and uses in crop protection, John Wiley & Sons. Inc.
9. Pandey, K. and J.P. Shukla, 1990. Elements of Toxicology. Radha publ. New Delhi.

**M.Sc., III Semester:
OPEN ELECTIVE-(For Science discipline students).
CONCEPTS OF ZOOLOGY.**

48 Hrs

Course Outcome:

After completing the course student will be able to

1. Broader understanding of Zoology and its concepts
2. Understand the concepts and basics of animals taxonomy
3. Understand the basics of histology
4. Describe the structure and basic functions of organ systems
5. Explain ecological concepts and effects of environmental pollution
6. Explain the mechanism of inheritance.

1. Introduction:

8 Hrs

a) Branches of animal science: Taxonomy, Animal Physiology, Genetics, Developmental Biology, Evolution, Ethology, Ecology, Applied Zoology, Entomology, Histology, c) Indian Wildlife- Status, Causes of wildlife depletion, Wildlife corridors, Conservation strategies- *In situ* and *Ex situ* d) e) Animals and human welfare.

2. Animal Taxonomy:

4 Hrs

a) Carl Linnaeus – Taxonomic hierarchy: Kingdom, Division, Phylum, Class, Order, Family and Binomial nomenclature

3. Animal cells and Tissues :

8 Hrs

a) Brief description of animal cell (light and ultra structure) b) Functions of cell organelles c) Structure and functional diversity in animal cell d) Cell division: Types and significance e) Structure and functions of basic tissues.

5. Structure and functions of organ systems:

16 Hrs

a) Human alimentary canal and outlines of digestion and absorption
 b) Respiration: Human respiration – exchange of gases.
 c) Circulation : Structure of human heart, Blood vessels and capillaries, composition of blood, blood coagulation.
 d) Excretion : Mammalian kidney and urine formation.
 e) Locomotion in vertebrates – Swimming, walking running, flying
 f) Nervous system and their functions, A brief account of human endocrine system
 g) Reproduction : Asexual and sexual reproduction, significance of sexual reproduction, outlines of human reproduction and fertility control

6. Ecology and Environmental Biology:

8 Hrs

a) Abiotic and Biotic factors b) Environmental Pollution – brief account of Air, Water and Noise pollution.

7. Heredity:

4 Hrs

a) Continuity of life – Mendel's laws b) Structure of chromosomes c) DNA and RNA

TUTORIALS

2x16=32 Hrs

REFERENCES :

1. Barnes, R. D. 1974. Invertebrate Zoology, III edition, W. B. Saunders Co., Philadelphia.
2. Barrington, E. J. W. 1976. Invertebrate structure and function. Thomas Nelson and Sons Ltd., London
3. Ltd., London

4. Hyman L. H. 1940. The invertebrates Vol.1 Protozoa through Ctenophora, McGraw hill co., N. Y.
5. co., N. Y.
6. Hyman. L. H. 1968. The Invertebrates Vol.8 McGraw Hill Co., N. Y and London.
7. Parker, T. J. Haswell, W. A. 1961. Text book of Zoology, Vol.I, Macmillon Co., London.
8. Russel – Hunter, W.D 1969. A. biology of higher invertebrates, Mac millon Co., Ltd.,
9. London.
10. Barrington, E. J. W. 1965. The Biology of Hemichordata and Protochordata – Oliver and Boyd, Edinborough.
11. Clark, W. E 1963. History of the Primates IV Edn., Univ. of Chicago Press, Chicago.
12. Malcom Jollie, 1962. Chordata morphology – East-West Press Pvt. Ltd., New Delhi.
13. Romer, A. S. 1966. Vertebrate Paleontolgy, 3rd Ed., Univ. of Chicago Press, Chicago.
14. Romer A. S., 1960. Vertebrate body, 3rd Ed., W. B. Saunders Co., Philadelphia.
15. Young. J. Z., 1950. Life of vertebrates The Oxford University Press, London
16. Young J Z 1957 Life of mammals, Oxford University Press, London.

M.Sc., IV SEMESTER

HC – 4.1 ADVANCED GENETICS AND COMPUTATIONAL BIOLOGY

32 hrs

Course Outcome:

After completing the course student will be able to

1. Understand the genomic organization of prokaryotes and eukaryotes.
2. Know the applications of various model organisms in genomic research.
3. Able to analyse the pedigree, psychosomatic disorders, prenatal diagnosis and genetic counselling.
4. Recognise few heritable diseases in man.
5. Understand the basic concepts of genomics
6. Understand the basic concepts of proteomics
7. Understand the nucleic acid and proteinr databases and tools.

Part A-Advanced Genetics

Unit I: Genome organization:

3 hrs

Prokaryotes, Eukaryotic nuclear genomes - C-value paradox, Eukaryotic organelle genomes Split Genes Mobile genetic elements in Prokaryotes (bacteria) and Eukaryotes (*Drosophila*, maize and humans), Genome Projects of model organisms (*C. elegans*, *Drosophila* and Mouse).

Unit II: Cancer Genetics:

5 hrs

Cancer incidence and mortality, types of cancer, causes of cancer, properties of cancer cells, Genetic basis of Carcinogenesis- Oncogenes: proto-oncogenes, oncogenes, retroviral oncogenes in human cancer. Tumor suppressor genes: Functions of tumor suppressor gene products. Cancer as a multistep process. Animal models of cancer research: Transgenic mouse and *Drosophila* models.

Unit III: Human genetics:

5 hrs

History of human genetics, pattern of inheritance, pedigree analysis. Human genome: Organization, distribution of genes, gene families. Genetic basis of syndromes and disorders: Cystic fibrosis, Neurofibromatosis, Schizophrenia, Anxiety disorder, Congenital heart diseases, Dyslexia.

Unit IV: Quantitative genetics:

3hrs

Introduction, types of quantitative trait, Nature of quantitative traits and their inheritance- Polygenic inheritance (Multifactorial hypothesis) – analysis of continuous variation; Variations associated with polygenic traits.

Part B-Computational Biology

Unit VII: Introduction and Scope of the Computational Biology **4 hrs**

Genomics: Definition and types of genomics Structural genomics: whole genome shotgun sequencing, gene annotation, gene families and clusters. Orthologs and paralogs. Functional genomics: Transcriptome, Microarray technology.

Unit VIII: Proteomics: **4 hrs**

Definition, Protein structure determination, protein domains, protein folding, Computer aided protein structure analysis, Protein-protein interactions, Protein microarrays.

Unit IX: Nucleic acid sequence and Protein analysis: **4 hrs**

Alignment, similarity searches including remote similarity searches, secondary structure element, motifs, Single nucleotide polymorphism (SNP), Two dimensional polyacrylamide gel electrophoresis, Mass Spectrometry.

Unit X: Genomics and proteomics databases and tools: **4 hrs**

Nucleic acid sequence databases and tools: Genbank, UCSC, ENSEMBL, EMBL, DDBJ, BLAST vs FASTA, file formats-FASTA, GCG, Genscan and ClustalW. Protein sequence databases and tools: Uni-prot, PDB, PIR, BLAST, PSI-BLAST (steps involved in use and interpretation of results).

PRACTICALS:

1. Study of mitotic chromosomes of *Drosophila* species- *Drosophila melanogaster*, *Drosophila nasuta*.
2. Preparation of metaphase chromosomes from bone marrow cells of mouse.
3. Karyotypic studies of normal human chromosomes and syndromes.
4. Creation of pedigrees and study of patterns of inheritance.
5. Studies on phenotypes of different diseases and syndromes.
6. Study of Quantitative characters: Sternopleurals, Acrosticals – mean, standard deviation.
7. Data mining for sequence analysis.
8. Web-based tools for sequence searches and homology screening-BLAST, FASTA
9. Nucleic acid sequence databases: GenBank retrieval, GeneScan.
10. Proteomics data bases: Uni-Prot, PROSITE, PDB, PIR, ProtParam.
11. Annotations: ORF finder, Use of ARTEMIS or any other suitable software

REFERENCES:

1. The Human Genome 2001, Nature Vol. 409.
2. The Drosophila Genome. 2000, Science Vol. 267.
3. The Caenorhabditis elegans genome 1998. Science Vol. 282.
4. Introduction to Genetic Analysis. Griffiths, Anthony J.F.; Miller, Jeffrey H.; Suzuki, David T.; Lewontin, Richard C.; Gelbart, William M. New York: W.H. Freeman & Co.; 1999
5. Fundamental Neuroscience. Larry R. Squire, Darwin Berg, Floyd Bloom, and Sascha du Lac. Third Edition, Academic Press; 3 edition (2008)
6. Principles of Neural Science. Eric R. Kandel, James H. Schwartz, and Thomas M. Jessell. McGraw-Hill Medical; 4 edition(2000)
7. Neurogenetics: Scientific and Clinical Advances (Neurological Disease and Therapy) David R. Lynch, Informa HealthCare; 1 edition (2005)
8. The Molecular and Genetic Basis of Neurologic and Psychiatric Disease. Roger N Rosenberg, Salvatore DiMauro, Henry L Paulson, and Louis Pt (2007) Lippincott Williams & Wilkins; Fourth edition

9. Bioinformatics for Dummies, Claverie J. M., Notredame C., (2nd Ed., 2007), Wiley Publishing, Inc., New York, USA
10. Brown T. A. 2007, Genomes 3. Garland Science Publishing, New York.
11. A.Malcolm Campbell and Laurie J.Heyer. Discovering Genomimcs, Proteomics and Bioinformatics. 2004. Low Price edition. Pearson Education, Inc.

**M.Sc., IV SEMESTER
HC – 4.2 APPLIED ZOOLOGY**

32 hrs

Course Outcome:

After completing the course student will be able to

1. Explain plant insect interaction, origin of pest and its control.
2. Understand vectors and its communicable diseases.
3. Explain races of silkworm their disease and its control.
4. Know about the importance of insects in forensic science and medicine.
5. Know about aquaculture and its practices in India.

UNIT I: Aquaculture

8hrs

Aquaculture in India: an overview – nutritional value and food security - Site selection and preparation of culture ponds - Fish culture: carps, marine fishes and ornamental fishes. Prawn culture: Freshwater prawns and marine shrimps. Fattening of crabs. Crayfish and Lobster - Molluscs: mussels, clams, chanks and oysters including pearl oyster. Live feeds: micro algae, micro-invertebrates (*Artemia* nauplii, Rotifers, Cladocerans, Copepods, Ostracodes) and worms as live baits – Water quality management and maintenance of sanitation - Plant and animal nutrients - Balanced diet (iso-nitrous and iso-caloric) - Artificial feed formulation – Low cost feed formulation - Aquatic weeds.

UNIT II: Sericulture

8hrs

Salient features of Saturnidae and Bombycidae. Mulberry and non mulberry silkworms, classification based on voltinism, moulting and geographic origin. Morphology and life cycle of *Bombyx mori*. Structure and functions of Silk glands. Silkworm rearing technology: Building, equipments, disinfection, environmental factors, Seed cocoons, preservation, grainage activity, LSPs, egg production, incubation, artificial hatching. **Pests and diseases:** Protozoan, Fungal, Viral and Bacterial diseases and their control measures. Silkworm pests and Predators

UNIT III: Apiculture

8hrs

Scope and its importance, Classification and morphology of honey bees, species and races of honey bees, tribal life and bee hunting. sex seperation, comb building, orientation of comb, communication, collection of propolis and water. Honey and its chemical composition, medicinal importance. Economic importance of honey, wax, bee pollination, pollen and Venom.

UNIT IV: Vermiculture

8hrs

A. Introduction to vermiculture. Definition, meaning, history, economic importance, their value in maintenance of soil structure. Useful species : Local species and Exotic species of earthworms. Role of four R's.
 B. Taxonomy Anatomy, Physiology and Reproduction of Lumbricidae and Eudrilidae.
 C. Earthworm Farming (Vermiculture) for home gardens, larger scale, Extraction (harvest), vermicomposting harvest and processing.
 D. Nutritional Composition of Vermicompost for plants, comparison with other fertilizers
 E. Enemies of Earthworms, Sickness

PRACTICALS:**16X4=64 hrs**

1. Study of morphometric characters of Indian major carps.
2. Diversity of fishes.
3. Collection of phytoplankton and zooplankton from natural resources and their identification.
4. Study of morphology of honey bee and cast system.
5. Mounting of mouth parts, stinging apparatus of honey bee.
6. Study of digestive system of honeybee.
7. Study of structure and types of honey comb.
8. Study of bee plants.
9. Study of morphology of lifecycle of *Bombyx mori*
10. Study of digestive and silk gland of *Bombyx mori*
11. Study of Non mulberry silkworms and their food plants.
12. Field trip- Collection of native earthworms & their identification
13. Study of systematic position& External characters of locally available earthworm species.
14. Mounting of setae and identification of earthworm species.
15. Study of equipments used in Vermiculture.

REFERENCES

1. Ashok Kumar (2009) Textbook of Animal Diseases
2. Edwards, C.A. and J.R. Lofty (1977) "Biology of Earthworms" Chapman and Hall Ltd., London.
3. G.S. Shukla, V.B. Upadhyay (2006) Economic Zoology.
4. Kevin, A and K.E.Lee (1989) " Earthworm for Gardeners and Fisherman" (CSIRO, Australia, Division of Soils)
5. Lee, K.E. (1985) "Earthworms: Their ecology and Relationship with Soils and Land Use" Academic Press, Sydney.
6. Pradip. V Jabde, (2005) Text Book of Applied Zoology.
7. R. L. Kotpal (2000) Modern Textbook of Zoology. Rastogi Publications
8. Satchel, J.E. (1983) "Earthworm Ecology" Chapman Hall, London.
9. Wallwork, J.A. (1983) "Earthworm Biology" Edward Arnold (Publishers) Ltd. London.

**M.Sc., IV SEMESTER
HC – 4.3 Project****Course Outcome:**

After completing the course student will be able to

1. understand the concepts of Project Management for planning to execution of projects
2. find importance of reference work Using tools of information such as periodical , journals, online resources
3. break work down the tasks of project and determine handover procedures
4. Interpret, analyse and presentation of the results obtained and compare with similar works and draw conclusion.

M.Sc., Examination
(Scheme CBCS)
M.Sc., ZOOLOGY
HARD CORE- Model question paper

Time: 3 hrs

Max Marks: 70

Instructions: *1. Answer all questions*

2. Illustrate your answer wherever necessary

I. Write short notes on the following:

[8×2=16]

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.

II. Write elaborate notes on any FIVE of the following:

[5×6=30]

- 9.
- 10.
- 11.
- 12.
- 13.
- 14.
- 15.
- 16.

Q3. Answer the following:

[2×12=24]

17. (i)
Or
(ii)
18. (i)
Or
(ii)

**M.Sc Examination
(Scheme CBCS)
M.Sc., ZOOLOGY
Softcore - Model question paper**

Time: 3 hrs

Max Marks: 70

Instructions: *1. Answer all questions*

2. Illustrate your answer wherever necessary

I. Write short notes on the following:

[8×2=16]

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.

II. Write elaborate notes on any FIVE of the following:

[5×6=30]

- 9.
- 10.
- 11.
- 12.
- 13.
- 14.
- 15.
- 16.

Q3. Answer the following:

[2×12=24]

17. (i)
Or
(ii)
18. (i)
Or
(ii)

**M.Sc Examination
(Scheme CBCS)
M.Sc., ZOOLOGY
Open Elective-Model question paper**

Time: 3 hrs

Max Marks: 70

Instructions: *1. Answer all questions*

2. Illustrate your answer wherever necessary

I. Write short notes on the following:

[8×2=16]

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.

II. Write elaborate notes on any FIVE of the following:

[5×6=30]

- 9.
- 10.
- 11.
- 12.
- 13.
- 14.
- 15.
- 16.

Q3. Answer the following:

[2×12=24]

17. (i)
Or
(ii)
18. (i)
Or
(ii)

Credit Matrix, Course of Study and Scheme of Examination for M.Sc. Degree Programme in Biochemistry

(With effect from 2019-20)

Programme Code: BIC

Course Type	Credits to be earned				Total Credits
	I Semester	II Semester	III Semester	IV Semester	
Hard Core Course	12	12	12	16	52
Soft Core Course	04	04	04	–	12
Elective Course	04	04	–	–	08
Open Elective Course*	–	–	04	–	04
Semester Total	20	20	20	16	76

*An Open Elective course offered by PG Dept. of Biochemistry to the students of other Depts.

Course Code	Course Type	Course Title	Credit Pattern (L:T:P)	Credits
Semester – I				
BCA040	HC	Analytical Biochemistry–I	3:1:0	4
BCA050	HC	Chemistry and Metabolism of Proteins and Nucleic Acids	3:1:0	4
BCA060	HC	Experiments in Biochemical Techniques and Enzymology** and Seminar	0:0:4	4
BCA230	SC	Enzymology	3:1:0	4
BCA250	Choose any ONE from the following		3:1:0	4
	SC	(i) Chemical Principles and Biochemical Reactions (ii) Plant Biochemistry (iii) Microbial Biochemistry		
Semester Total Credits				20

Course Code	Course Type	Course Title	Credit Pattern (L:T:P)	Credits
Semester – II				
BCB040	HC	Analytical Biochemistry–II	3:1:0	4
BCB050	HC	Chemistry and Metabolism of Carbohydrates and Lipids	3:1:0	4
BCB060	HC	Experiments in Immunology and Biochemical Estimations** and Seminar	0:0:4	4
BCB250	SC	Immunology and Microbiology	3:1:0	4
BCB260	Choose any ONE from the following		3:1:0	4
	SC	(i) Human Physiology and Nutrition (ii) Research Methodology and Biostatistics (iii) Clinical Research Methods and Industrial Biochemistry		
Semester Total Credits				20

Course Code	Course Type	Course Title	Credit Pattern (L:T:P)	Credits
Semester – III				
BCC070	HC	Cell Biology, Endocrinology and Cell Signaling	3:1:0	4
BCC050	HC	Clinical Biochemistry	3:1:0	4
BCC060	HC	Experiments in Clinical Biochemistry and Molecular Biology** and Research Paper Presentation	0:0:4	4
BCC220	Choose any ONE from the following		4:0:0	4
	SC	(i) Genomics, Proteomics and Bioinformatics (ii) Biotechnology (iii) Pharmaceutical Biochemistry		
BCC630	OE	Nutrition and Health	4:0:0	4
Semester Total Credits				20

Course Code	Course Type	Course Title	Credit Pattern (L:T:P)	Credits
Semester – IV				
BCD010	HC	Molecular Biology and Gene Regulation	3:1:0	4
BCD070	HC	Genetics and Genetic Engineering	3:1:0	4
BCD060	HC	Project Work OR Dissertation***	0:4:4	8*
Semester Total Credits				16
Total CREDITS to be earned for M.Sc. BIOCHEMISTRY				76

* Grade Point will be calculated with respect to the allotted credits

HC	Hard Core Course
SC	Soft Core Course
E	Elective Course
OE	Open Elective Course
C1	Component 1 of Internal Assessment (IA)
C2	Component 2 of Internal Assessment (IA)
C3	Component 3 (Semester-end Exam)
L	Lecture (1 Credit=1 hr)
T	Tutorial (1 Credit=2 hrs)
P	Practical (1 Credit=2 hrs)

** Weekly Four hrs of practical for Two days

*** Project work OR Dissertation should be in-house only and may be allotted to the students in the 2nd/3rd semester

Note: Two Practical examinations of four hrs duration each for C3 (component 3) of Hardcore Course with Practical Component Only.

SCHEME OF ASSESSMENT

Course Code	Course Type	Course Title	Exam Hrs	Max. Marks			
				IA		Exam C3	Total
				C1*	C2*		
Semester - I							
BCA040	HC	Analytical Biochemistry–I	3	15	15	70	100
BCA050	HC	Chemistry and Metabolism of Proteins and Nucleic Acids	3	15	15	70	100
BCA060	HC	Experiments in Biochemical Techniques and Enzymology and Seminar	–	15	15	70	100
BCA230	SC	Enzymology	3	15	15	70	100
BCA250	Choose any ONE from the following		3	15	15	70	100
	E	(i) Chemical Principles and Biochemical Reactions (ii) Plant Biochemistry (iii) Microbial Biochemistry					
Semester Total Marks							500

Course Code	Course Type	Course Title	Exam Hrs	Max. Marks			
				IA		Exam C3	Total
				C1*	C2*		
Semester - II							
BCB040	HC	Analytical Biochemistry–II	3	15	15	70	100
BCB050	HC	Chemistry and Metabolism of Carbohydrates and Lipids	3	15	15	70	100
BCB060	HC	Experiments in Immunology and Biochemical Estimations and Seminar	–	15	15	70	100
BCB250	SC	Immunology and Microbiology	3	15	15	70	100
BCB260	Choose any ONE from the following		3	15	15	70	100
	SC	(i) Human Physiology and Nutrition (ii) Research Methodology and Biostatistics (iii) Clinical Research Methods and Industrial Biochemistry					
Semester Total Marks							500

Course Code	Course Type	Course Title	Exam Hrs	Max. Marks			
				IA		Exam C3	Total
				C1*	C2*		
Semester - III							
BCC070	HC	Cell Biology, Endocrinology and Cell Signaling	3	15	15	70	100
BCC050	HC	Clinical Biochemistry	3	15	15	70	100
BCC060	HC	Experiments in Clinical Biochemistry and Molecular Biology and Research Paper Presentation	–	15	15	70	100
BCC220	Choose any ONE from the following		3	15	15	70	100
	SC	(i) Genomics, Proteomics and Bioinformatics (ii) Biotechnology (iii) Pharmaceutical Biochemistry					
BCC630	OE	Clinical Research and Drug Development	3	15	15	70	100
Semester Total Marks							500

Course Code	Course Type	Course Title	Exam Hrs	Max. Marks			
				IA		Exam C3	Total
				C1*	C2*		
Semester - IV							
BCD010	HC	Molecular Biology and Gene Regulation	3	15	15	70	100
BCD070	HC	Genetics and Genetic Engineering	3	15	15	70	100
BCD060	HC	Project Work OR Dissertation**	–	15	15	70	100
Semester Total Marks							300

- C1* & C2* Internal test will be conducted for 20 marks of one hour duration and it is reduced to 10 marks + 5 marks for continuous assessment
- Continuous assessment comprise of assignments, group discussions, seminars and tutorials
- ** The Project evaluation is as below
 - Component 1 (C1): Periodic Progress Report (15%)
 - Component 2 (C2): Periodic Progress Report (15%)
 - Component 3 (C3): Final Viva-Voce and Evaluation (70%)
 - (The report evaluation is for 40% and the Viva-Voce examination is for 30%)

Program Outcome(s):

PO	PO Code	Group
Provides with the necessary knowledge and skills to undertake a career in research, either in industry or in an academic setting	PO1	2019
Provides the breadth and depth of scientific knowledge in 'Biochemistry' and allied areas	PO2	2019
Equips to apply for a PhD or to gain employment in biochemistry and allied areas	PO3	2019
Provides a substantial element of hands-on research experience, with enhanced experimental skills	PO4	2019
Demonstrates detailed knowledge and understanding of the principles and theories of biochemistry	PO5	2019
Helps to understand the principal techniques of biomolecular structural characterization, including spectroscopy	PO6	2019

Program Specific Outcome(s): The Specific Outcome of this programme is to train and provide the candidate with knowledge related to

PSO	PSO Code	Group
Global level research opportunities to pursue PhD programme targeted approach of CSIR–NET examination	PSO1	2019
Enormous job opportunities at all level of chemical, pharmaceutical, food products, life oriented material industries	PSO2	2019
Specific placements in R & D and Quality Control or Analysis division of nutraceutical, pharmaceutical industries & Allied Division	PSO3	2019

Course Code	Title of the Course	Credits
BCA040	ANALYTICAL BIOCHEMISTRY-I	4

COURSE OUTCOME(S):

- CO1 Specify in depth cell fractionation techniques
- CO2 Write down in details with application, if applicable, chromatography and spectroscopy
- CO3 Write down in details with application, if applicable, principle and applications of electrophoresis
- CO4 Understand the classification and characteristics of centrifugation and microscopy

		No. of Lectures
Unit I:		08
1.1	Cell Fractionation	
1.1.1	<u>Cell fractionation techniques</u> : Preparation of extracts for biochemical investigations. Physico-chemical properties of solvents, solubility and miscibility, salting-in and salting-out.	
1.1.2	Choice of solvent for solvent extraction, mixed solvents, solid phase extraction. Cell lysis, dialysis, precipitation and ultra filtration.	

Unit II:		14
2.1	Chromatography and Spectroscopy	
2.1.1	Adsorption <u>vs.</u> Partition chromatography. Paper, TLC, Ion exchange, Reverse phase, Gel filtration, Affinity, HPLC, and Gas chromatographic techniques.	
2.1.2	Beer-Lamberts Law, Its verifications and Deviations, Concept of Absorptions, Transmission, Scattering, Phosphorescence, Fluorescence, Luminescence, Diffraction Spectra.	
2.1.3	Principle, instrumentation, working and applications of-UV and Visible Spectroscopy,	
2.1.4	Turbidometry and Nephelometry.	

Unit III:		12
3.1	Electrophoresis	
3.1.1	Theory of electrophoresis, continuous and discontinuous PAGE, SDS-PAGE.	
3.1.2	Other electrophoretic methods-Isoelectric focusing, 2-dimensional gel electrophoresis, Capillary electrophoresis and PFGE.	
3.1.3	Agarose gel electrophoresis of nucleic acids. Isotachopheresis.	
3.1.4	Separation of proteins, lipoproteins, visualizing separated	

	components–staining, fluorescence, PAS staining, zymogram and reverse zymogram,	
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Unit IV:		
4.1	Centrifugation and Microscopy	
4.1.1	Analytical and Preparative Ultracentrifuge–Principle, instrumentation and applications.	14
4.1.2	Analysis of subcellular fractions, marker enzymes and determination of relative molecular mass–Svedberg's constant, sedimentation velocity and sedimentation equilibrium.	
4.1.3	Theories of Tissue Fixation and Staining Techniques. Principles of Transmission and Scanning Electron Microscopy.	
4.1.4	Principles of Phase Contrast and Fluorescence Microscopy, Confocal Microscopy	

References	
[1]	Analytical techniques in Biochemistry and Molecular Biology; Katoch, Rajan. Springer (2011)
[2]	Basic Methods for the Biochemical Lab; Martin Holtzhauer, Springer, (2007).
[3]	Principles and Techniques of Biochemistry and Molecular Biology 7th Edn. Keith Wilson and John Walker, Cambridge University Press, (2010).
[4]	Biochemistry LabFax, Ed. J.A.A. Chambers and D. Rickwood,, Blackwell Science, (1993),
[5]	Protein Purification Applications, S.L.V. Harris and Angal IRL Press, (1990)
[6]	Laboratory Techniques in Biochemistry and Molecular Biology, Work and Work Vol. I & II, North Holland, (1969).
[7]	Basic Mathematics for Biochemists; Cornish Bowden, Oxford University Press (1998),.
[8]	Biophysical Tools for Biologists <i>In Vivo</i> Techniques; John Correia H. Detrich, III Elsevier (2008).
[9]	Practical Biochemistry by Keith Wilson and Walker 5th ed. Cambridge.
[10]	Biophysical chemistry, Upadhyaya, A., Upadhyaya, K. and Nath, N. Himalayan Publishing House.
[11]	Practical biochemistry- Principles and Techniques. Wilson and Walker. J.Cambridge Uni. Press.
[12]	Physical Biochemistry-David Freifelder, 2nd Edition.
[13]	Principles of Instrumental Analysis. 5th Ed. Douglas A Skoog, James Holler and Timothy A Nieman.
[14]	Introduction to Electron Microscopy for Biologists; Terry Allen, Academic Press (2008).

Course Code	Title of the Course	Credits
BCA050	CHEMISTRY AND METABOLISM OF PROTEINS AND NUCLEIC ACIDS	4

COURSE OUTCOME(S):

- CO1 Identify the details of amino acids and proteins
- CO2 Understand in details with application, if applicable, nitrogen metabolism and degradation
- CO3 Write down the classification and characteristics of synthesis of amino acids and proteins
- CO4 Write down in details with application, if applicable, metabolism of nucleic acids

		No. of Lectures
Unit I:		10
1.1	Chemistry of Amino acids and Proteins	
1.1.1	Classification and structure of 20 amino acids, newly discovered amino acids, essential, non-essential, unusual and non-protein	
1.1.2	General properties of aa, acid-base titrations, pKa Peptide bond-stability and formation, chemical synthesis of peptide. Primary structure and determination, GN Ramachandran plots	
1.1.3	Secondary structure and motifs, α helix, β sheet, Leucine zipper, Zinc finger	
1.1.4	Tertiary & Quaternary structure (myoglobin, hemoglobin) Protein-protein interactions (actin, tubulin) Small peptides (glutathione, peptide hormones), Cyclic peptides (Gramicidin)	
1.1.5	Classification of proteins-globular, fibrous, membrane, metallo-proteins, Denaturation (pH, temperature, chaotropic agents), refolding, Role of chaperones in folding	
Unit II:		14
2.1	Nitrogen Metabolism and Degradation of Amino Acids	
2.1.1	Nitrogen cycle, Nitrogen fixation – symbiotic and non-symbiotic, Nitrogenase complex. Assimilation of ammonia	
2.1.2	Metabolic fate of dietary proteins and amino acids Degradations to glucose and ketone bodies	
2.1.3	Amino acids degraded to Pyruvate, Oxaloacetate	
2.1.4	Amino acids degraded to Acetyl-CoA, Succinyl-CoA Metabolism of branched chain amino acids, urea cycle, regulation of urea cycle	
2.1.5	Genetic defects in metabolism of amino acids (albinism, Phenylketonuria, maple syrup urine disease, homocystinuria, alkaptonuria, methyl malonic Acidemia)	

Unit III:		
3.1	Biosynthesis of Amino Acids and Protein Degradation	
3.1.1	Biosynthesis of amino acids and regulation of amino acid metabolism	08
3.1.2	Biosynthesis and degradation of heme	
3.1.3	Biosynthesis of polyamines, creatine, gramicidine and glutathione	
3.1.4	Biosynthesis and degradation of glycoproteins and proteoglycans	
3.1.5	Protein degradation pathway–Ubiquitin–Proteosome pathway, lysosomal pathway	

Unit IV:		
4.1	Chemistry and Metabolism of Nucleic Acids	
4.1.1	Purines, pyrimidines, nucleosides, nucleotides, unusual bases. Structure of DNA – Watson Crick Model, A- and Z- forms.	16
4.1.2	Supercoiling of DNA – negative and positive, linking number	
4.1.3	Structure of RNA, tRNA, rRNA, siRNA / miRNA Denaturation and renaturation, T _m (factors affecting T _m) and Cot curves, Isolation and purification of nucleic acids from biological sources.	
4.1.4	Biosynthesis of purines and pyrimidines, Degradation of purines and pyrimidines, Regulation: de novo, salvation, nucleotide analogs, conversion of nucleotides to deoxynucleotides, mechanism of action of methotrexate, 5-fluorouridine, azathymidine.	
4.1.5	Gout and Lesch–Nyhan syndrome	
4.1.6	Biosynthesis of NAD, FAD and Co–enzyme A	

References	
[1]	Lehninger Principles of Biochemistry 4th Ed By David L. Nelson and Michael M. Cox, WH Freeman and Company.
[2]	Biochemistry by Lubert Stryer. WH Freeman and Co.
[3]	Biochemistry: The Molecular Basis of Life by Trudy McKee and James R McKee. Publisher: McGraw-Hill Higher education.
[4]	Biochemistry and Molecular biology By William H. Elliott and Daphne C. Elliott. Oxford University Press.
[5]	Biochemistry 3rd Ed. By Donald J. Voet and Judith G. Voet. John Wiley and Sons.
[6]	Biochemistry: Biomolecules, Mechanisms of Enzyme Action and Metabolism Vol 1 by D Voet. John Wiley and Sons.

Course Code	Title of the Course	Credits
BCA060	EXPERIMENTS IN BIOCHEMICAL TECHNIQUES AND ENZYMOLOGY AND SEMINAR	4

COURSE OUTCOME(S):

- CO1 Identify the details of spectrophotometer
 CO2 Identify the details of specific activity of enzymes
 CO3 Deliberate the characteristics of gel electrophoresis
 CO4 Deliberate the characteristics of use of pipettes

Group I:	1. Determination of Normality, Molarity and Molality of solutions 2. Preparation of buffers: Acetate, Phosphate and Tris buffer 3. Colorimetry–Beer's law and its applications 4. Determination of Molar Extinction Coefficient 5. Chromatography–Separation of amino acids by ascending, descending, circular paper chromatography 6. TLC of amino acids 7. Gel filtration, Ion exchange chromatography	
Group II:	8. Estimation of protein by Lowry's method. 9. Estimation of protein by Biuret reagent method. 10. Estimation of amino acids by Ninhydrin method 11. Isolation of casein from milk and its quantification 12. Electrophoresis–Separation of proteins by Native and SDS-PAGE 13. Determination of pK_a and pI of amino acid, formal titration. 14. Separation of nucleic acids by agarose gel electrophoresis	
Group III:	15. Isolation of microbes from air, soil and water 16. Gram's staining 17. Determination of growth curve of bacteria 18. Antibiotic sensitivity tests 19. Determination of specific activity of <ul style="list-style-type: none"> (i) Acid Phosphatase (ii) Alkaline Phosphatase (iii) Salivary Amylase (iv) Protease (v) Invertase (vi) Aminotransferase 	
Group Study	Extraction, Isolation, Purification and enzyme characterization. Determination of specific activity, optimum pH, temperature, time and energy of activation. Determination of K_m and V_{max} Enzyme inhibition studies	

References

- [1] Analytical techniques in Biochemistry and Molecular Biology; Katoch, Rajan. Springer (2011).
- [2] Basic Methods for the Biochemical Lab; Martin Holtzhauer, Springer, (2007).
- [3] Principles and Techniques of Biochemistry and Molecular Biology 7th Edn. Keith Wilson and John Walker, Cambridge University Press, (2010).
- [4] Biochemistry LabFax, Ed. J.A.A. Chambers and D. Rickwood,, Blackwell Science, (1993),
- [5] Protein Purification Applications, S.L.V. Harris and Angal IRL Press, (1990)
- [6] Laboratory Techniques in Biochemistry and Molecular Biology, Work and Work Vol. I & II, North Holland, (1969).
- [7] Physical Biochemistry, Kansal Edward Van Halde. Prentice Hall.
- [8] Modern Experimental Biochemistry R.F.Boyer [Ed.] (1986) Addition Wesley.
- [9] Analytical Biochemistry; D.J. Holme and H. Pick Longman (1983).
- [10] Principles and techniques of Biochemistry and Molecular Biology; Keith Wilson and John Walker; 6th Edn. (2005) Cambridge University Press.
- [11] Biochemical Calculations, Irwin H. Segel (1976) 2nd Ed. John Wiley and Sons.

Course Code	Title of the Course	Credits
BCA230	ENZYMOLGY	4

COURSE OUTCOME(S):

- CO1 Write down in details with examples enzyme kinetics
CO2 Identify in details with examples enzyme catalysed reactions
CO3 Identify the characteristics of cooperativity reactions
CO4 Learn the classification and characteristics of multienzyme complex reactions

		No. of Lectures
Unit I:		16
1.1	Enzyme Kinetics and Inhibition	
1.1.1	Nature of enzymes, Nomenclature and IUB classification of enzymes, Units of enzyme activity, IU and activity and specific activity. Localization, isolation, purification and characterization of enzymes. Criteria of purity of enzymes. Assay methods—coupled enzyme assays, continuous, end point and kinetic assay.	
1.1.2	<u>Enzyme Kinetics</u> : Rate of a reaction, order and molecularity. Michaelis Menten equation, initial velocity approach, steady state approach. Vmax, Km and their significance. Linear transformation of Michaelis Menten equation—Lineweaver Burk plot, Eadie Hofstee, Haynes–Wolf and Cornish–Bowden.	
1.1.3	<u>Inhibition</u> : Reversible inhibition—Competitive, Non competitive and Uncompetitive, product inhibition, irreversible inhibition—suicide inhibition. Determination of <i>K_i</i> . Fast reactions—Stopped flow, temperature jump method with examples of enzymes.	
Unit II:		08
2.1	Enzyme Catalyzed Reactions	
2.1.1	Bisubstrate enzyme catalysed reactions—Cleland's notation with examples for ordered, ping pong, and random.	
2.1.2	General rate equation. Primary and secondary plots. Mechanisms of enzyme catalysis—Active site structure and its investigation.	
2.1.3	Methods of determining active site structure—isolation of ES/EI complex, affinity labeling, chemical modification studies.	

Unit III:		
3.1	Enzyme Catalysis and Cooperativity	
3.1.1	<u>Nature of enzyme catalysis</u> : Transition state theory, proximity and orientation, orbital steering, acid base catalysis, covalent catalysis, metal ion catalysis, nucleophilic and electrophilic catalysis, intramolecular catalyses, entropy effects.	12
3.1.2	Effect of temperature and pH on enzyme catalyzed reactions.	
3.1.3	<u>Oligomeric proteins and Cooperativity</u> : Binding of ligands to macromolecules–Scatchard plot, Positive and Negative cooperativity. Oxygen binding to hemoglobin.	
3.1.4	Hill equation, homotropic and heterotropic effectors. Allosteric enzyme–Aspartyl transcarbamylase.	

Unit IV:		
4.1	Multienzyme Complex and Coenzymes	
4.1.1	<u>Mechanisms of action of specific enzyme</u> : Chymotrypsin zymogen activation, acid–base catalysis, charge relay net work. Lysozyme, alcohol dehydrogenase, ribonuclease, Carboxypeptidase–A, RNA as enzyme, coenzymic action of NAD+ FAD, TPP, PLP, biotin, CoA, folic acid and lipoic acid.	12
4.1.2	<u>Multienzyme complexes</u> : Isoenzymes, eg. LDH. Multifunctional enzyme (DNA polymerase) multi enzyme complex (PDC)	
4.1.3	Metabolic regulation of enzyme activity–Feedback regulation.	

References

- [1] Fundamentals of Enzymology; 3rd Edn. Nicholas C. Price and Lewis Stevens, Oxford University Press (2012).
- [2] Enzymes; Trevor Palmer, East - West Press Pvt. Ltd., Delhi (2004).
- [3] Enzymes: A Practical Introduction to Structure, Mechanism, and Data Analysis; Robert A. Copeland, Wiley-VCH Publishers (2000).
- [4] Enzyme Kinetics and Mechanism; Paul F. Cook, W. W. Cleland, Garland Science (2007).
- [5] Enzyme Kinetics; Roberts, D.V. (1977), Cambridge University Press.
- [6] The Enzymes; Boyer, Academic Press, (1982).
- [7] Principles of Enzymology for Food Sciences; Whitaker, Marcel Dekker (1972) Academic Press.
- [8] Introduction to Enzyme and Co-enzyme Chemistry. Ed. T. Bugg, (2000), Blackwell Science.

Course Code	Title of the Course	Credits
BCA250	CHEMICAL PRINCIPLES AND BIOCHEMICAL REACTIONS	4

COURSE OUTCOME(S):

- CO1 Specify in details with examples chemical principles and bonding
 CO2 Write down in depth thermodynamics
 CO3 Learn in details with application, if applicable, stereochemistry
 CO4 Deliberate in depth secondary metabolites

		No. of Lectures
Unit I:		16
1.1	Chemical Principles and Bonding	
1.1.1	<u>Chemical principles</u> : Acids and bases, Buffers. Buffering capacity. Ionic strength- Molarity, Normality, Mole concept, Avogadro number, structure and special properties of water.	
1.1.2	<u>Bonding</u> : Covalent bond, ionic bond, Coordinate bond. Coordinate bond formation by transition metals in biological complex structures.	
1.1.3	Crystal field theory, ligand field theory, valence bond theory.	
1.1.4	Bonding of iron in hemoglobin and cytochromes, cobalt in Vit B12, and Mg ²⁺ in chlorophyll. Chelates and complexes.	

Unit II:		08
2.1	Thermodynamics	
2.1.1	<u>Physiological importance of electrolytes</u> : Osmotic pressure, vapour pressure, vapour pressure osmometer, Donnan membrane equilibrium.	
2.1.2	<u>Introduction to thermodynamics</u> : I, II and III law. Enthalpy, entropy and free energy. Free energy and chemical equilibrium	
2.1.3	<u>Electrodes</u> : Hydrogen electrode, oxygen electrode, oxidation and reduction reactions, redox potential.	

Unit III:		12
3.1	Stereochemistry and Heterocyclic Compounds	
3.1.1	Importance of Stereochemistry, position and order of groups around carbon. Geometric and optical isomerism, absolute and relative configuration. Symmetry view of chirality, relation between chirality and optical activity, representation of chiral structures by Fischer.	
3.1.2	Structure and stereochemistry of glucose—anomers, epimers and stereoisomers, D and L, + and – R and S notations.	
3.1.3	Heterocyclic Compounds: Chemistry, biological	

	occurrence of furan, indole, thiazole, pterine, pteridine, isoalloxazine, pyrrole.	
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Unit IV:		
4.1	Organic Reactions and Secondary Metabolites	
4.1.1	Mechanism of Organic Reactions: Classification of organic reactions. Reaction intermediates, reaction energetics, rate, order and molecularity of reactions.	12
4.1.2	Mechanisms and stereochemistry of substitution, addition, and elimination. Rearrangements reactions. Mechanisms of ester hydrolysis. Aromaticity and resonance structure. Hydrogenation- homogenous and heterogenous hydrogenation	
4.1.3	Secondary metabolites: Phytochemicals, terpenes, polyphenols, procyanidins, flavonoids, xanthones, alkaloids and pigments.	

References
[1] Basic principles of organic chemistry- Robers and Caserio
[2] Organic chemistry, Hendrickson, Cram and Hammonal.
[3] Organic chemistry, I. L. Finar, Longman group Ltd.
[4] Organic chemistry, Morrison and Boyd, 4th edition Allyn and Bacon Inc.

Course Code	Title of the Course	Credits
BCA250	PLANT BIOCHEMISTRY	4

COURSE OUTCOME(S):

- CO1 Specify in details with examples Photosynthetic pathways and its regulations
CO2 Write down in depth plant growth hormones in the agriculture
CO3 Learn in details with application, if applicable, Medicinal plants
CO4 Deliberate in depth secondary metabolites of plants and its significance

		No. of Lectures
Unit I:		12
1.1	Plant Cell and Photosynthesis	
1.1.1	Plant cell–Structure and functions of subcellular organelles, plant cell wall, Mechanism of water absorption, Ascent of sap. Transpiration - types, stomatal opening, Mechanism and factors affecting transpiration.	
1.1.2	Photosynthesis–Photosynthetic pigments, Photo synthetic apparatus, Light reactions, cyclic and non cyclic Phosphorylation. Calvin cycle, Hatch–Slack cycle, CAM plants.	
1.1.3	Regulation of photosynthesis, Photorespiration.	

Unit II:		12
2.1	Cycles of elements	
2.1.1	Nitrogen cycle, Biochemistry of symbiotic and non symbiotic nitrogen fixation, Sulphur cycle, Phosphorus cycle.	
2.1.2	Plant nutrition–Biological functions of micro and macro nutrients in plants and their deficiency symptoms.	

Unit III:		16
3.1	Growth Regulators	
3.1.1	Plant growth regulators–chemistry, biosynthesis, mode of action, distribution and physiological effects of Auxins, Gibberellins, Cytokinins, ABA and Ethylene.	
3.1.2 3.1.3	Biochemistry of seed dormancy, Seed germination, Fruit ripening and Senescence.	

Unit IV:		08
4.1	Medicinal Importance	
4.1.1	Medicinal value of different parts of plants.	
4.1.2	Basic methods to identify the secondary metabolites. Role of secondary metabolites in Ayurvedha and Siddha treatment.	
4.1.3	Medicinal value of Amla, Stevia, Aswagandha, Turmeric and other Indian medicinal plants.	

References	
[1]	Plant physiology, Verma, 7th Revised edition, Emkay Publications 2001.
[2]	Plant Physiology, S. N. Pandey and B.K. Sinha, Vikas Publishing House Pvt. Ltd, 3rd edition, 1999.
[3]	Plant Biochemistry and Molecular Biology, Peter Jhea, Richard C. Leegood,
[4]	Introduction to plant physiology, William. G.Hopkins, Norman. P.A. Hunger, 3rd edition
[5]	A Handbook of Medicinal Plants –Prajapathi, Purohit,Sharma, Kumar
[6]	Medicinal Plants –a compendium of 500 species.

Course Code	Title of the Course	Credits
BCA250	MICROBIAL BIOCHEMISTRY	4

COURSE OUTCOME(S):

- CO1 Specify in details with examples staining techniques used for the identification of microbes
- CO2 Write down in depth Molecular biology of prokaryotes
- CO3 Learn in details with application, if applicable, Operon systems in gene regulation of bacteria
- CO4 Deliberate in depth antimicrobial drugs are used for the microbial infections

		No. of Lectures
Unit I:		10
1.1	Pure Culture, Staining Technique and Growth	
1.1.1	Principles of microbial nutrition: Nutritional requirements, different kinds of media, factors affecting growth.	
1.1.2	Enrichment culture techniques for isolation of chemoautotroph's, chemoheterotroph's and photosynthetic microorganisms. Modes of reproduction, Biosynthesis of cell wall components, enumeration, growth curve, generation time, synchronous growth, Chemostat. Adaptation to stationary phase, heat and cold shock, osmolarity and salinity, oxidative stress.	
1.1.4	Gram, Acid fast & flagellar staining. Mechanism of bacterial motility.	

Unit II:		14
2.1	Regulation of Genes in Bacteria	
2.1.1	Nucleic Acids as Carriers of Genetics Information, Arrangement and Organization of Gene in Prokaryotes:	
2.1.2	Operon Concept, Catabolite Repression, Instability of Bacterial RNA, Inducers and Co repressors E. coli Lac Operon: Negative Regulation and Positive Regulation, E. Coli Arabinose Operon: Regulation by Attenuation, His and Trp Operons: Anti-termination, Genetic Transfer: Conjugation, Transformation and Transduction.	

Unit III:		
3.1	Virology and Biological Nitrogen Fixation	
3.1.1	Introduction to Virus, Classification, Assay Methods, Properties and Characteristic of Bacterial, Plant and Animal Viruses	16
3.1.2	Virus Host Interaction, Acute Virus Infections, Persistent of Virus Infection, Influenza, Herpes, Hepatitis A and B.	
3.1.3	<u>Nitrogen Metabolism</u> : Mechanism and Regulation of Utilization of Ammonia, Nitrate and other Nitrogen Source	
3.1.4	<u>Nitrogen Fixation</u> : Mechanism and Regulation of Nitrogen Fixation, Symbiotic and Asymbiotic Nitrogen Fixation and Biochemistry of Nitrogenase.	

Unit IV:		
4.1	Antimicrobial Agents	
4.1.1	The Development of Antimicrobial Agents, Past, Present and Future, Selection of Antimicrobial Agents	08
4.1.2	Synthetic Organic Antimicrobials, β -Lactam Antibiotics, Amino glycoside Antibiotics, Antifungal Drugs, Antiviral Drugs	
4.1.3	Resistance to Antimicrobial Drugs	

<p>References</p> <p>[1] Microbial physiology, 2nd Edn. I.W. Dawes and I.W. Sutherland (1991) Blackwell Scientific.</p> <p>[2] Microbial physiology, 4th Edn. Albert G. Moat, John W. Foster and Michael P. Spector, Wiley-Liss (2002).</p> <p>[3] Biology of Microorganisms, Brock Prentice Hall (1996).</p> <p>[4] Microbiology: Lansing M. Prescott, Hartley and Klein, 5th Edn. McGraw Hill (2002).</p> <p>[5] General Microbiology, Stainer <i>et al.</i>, 4th Edn. McMillan (1975).</p> <p>[6] Microbiology, Pelczer, Reid and Kreig Tata McGraw Hill (1996).</p>
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Course Code	Title of the Course	Credits
BCB040	ANALYTICAL BIOCHEMISTRY–II	4

COURSE OUTCOME(S):

- CO1 Identify in details with application, if applicable, flow cytometry
CO2 Specify the characteristics of biosensor technology
CO3 Understand in details with examples spectroscopy
CO4 Write down the details of x-ray crystallography

		No. of Lectures
Unit I:		08
1.1	Flow Cytometry and Model Systems	
1.1.1	<u>Flow Cytometry</u> : Principle and design of flow cytometer, cell sorting.	
1.1.2	<u>Animal models</u> : Choice of animals, types of studies, mutant organisms, cultured cells, plant as models and tissue culture models.	

Unit II:		16
2.1	Biosensor Technology and Radioactivity	
2.1.1	Concept and design of biosensors, types and uses of biosensors.	
2.1.2	Principle and applications of biosensors for glucose, triglyceride, uric acid, cholesterol and oxalate.	
2.1.3	Units of radioactivity. Detection and measurement of radioactivity–solid and liquid scintillation counting, scintillation cocktails and sample preparation. Cerenkov counting. Applications of radioisotopes in biology. Radiation hazards.	
2.1.4	Principle and Applications of Autoradiography	

Unit III:		10
3.1	Spectroscopy	
3.1.1	Principle, instrumentation, working and application of–Spectrofluorimetry, Flame Spectrophotometry, Atomic Absorption Spectrometry.	
3.1.2	<u>IR spectroscopy</u> : Physical basis of IR spectroscopy. Instrumentation, use of IR in structure determination, Fourier Transfer–IR spectroscopy.	
3.1.3	<u>NMR</u> : Principle, effect of atomic, identity on NMR, chemical shift, spin coupling NMR, measurement of NMR spectra, biochemical application of NMR.	
3.1.4	<u>ESR</u> : Principle, measurement of ESR spectra, biochemical application of ESR.	
3.1.5	Principle, instrumentation and applications of ORD and CD	

Unit IV:		
4.1	Mass spectroscopy, X-ray Crystallography and Nanoparticles	
4.1.1	Theory and construction of mass spectrometer. Ionization, fragmentation, m/z , time of flight, MALDI and ESI.	
4.1.2	<u>Other methods</u> : MS/MS, LC/MS, GC/MS, Peptide mapping, post translation modification analysis, determination of disulfide bridges	
4.1.3	X-ray crystallography–Bragg's law, Unit cell, Isomorphous replacement, Fibre pattern of DNA.	14
4.1.4	<u>Introduction to Nanoscience</u> : Importance and fundamental science behind nanotechnology.	
4.1.5	<u>Applications of Nanoparticles</u> : Tools to make nanostructures, Nanoscale lithography, E–beam lithography, molecular synthesis, self assembly. Drug and Gene delivery for human health, Biosensors and sensors, cleaning environment (for heavy metal & Bioremediation).	

References

- [1] Analytical techniques in Biochemistry and Molecular Biology; Katoch, Rajan. Springer (2011)
- [2] Basic Methods for the Biochemical Lab; Martin Holtzhauer, Springer, (2007).
- [3] Principles and Techniques of Biochemistry and Molecular Biology 7th Edn. Keith Wilson and John Walker, Cambridge University Press, (2010).
- [4] Biochemistry LabFax, Ed. J.A.A. Chambers and D. Rickwood,, Blackwell Science, (1993),
- [5] Protein Purification Applications, S.L.V. Harris and Angal IRL Press, (1990)
- [6] Laboratory Techniques in Biochemistry and Molecular Biology, Work and Work Vol. I & II, North Holland, (1969).
- [7] Basic Mathematics for Biochemists; Cornish Bowden, Oxford University Press (1998),.
- [8] Biophysical Tools for Biologists *In Vivo* Techniques; John Correia H. Detrich, III Elsevier (2008).
- [9] Practical Biochemistry by Keith Wilson and Walker 5th ed. Cambridge.
- [10] Biophysical chemistry, Upadhyaya, A., Upadhyaya, K. and Nath, N. Himalayan Publishing House.
- [11] Practical biochemistry- Principles and Techniques. Wilson and Walker. J.Cambride Uni. Press.
- [12] Physical Biochemistry-David Freifelder, 2nd Edition.
- [13] Principles of Instrumental Analysis. 5th Ed. Douglas A Skoog, James Holler and Timothy A Nieman.

Course Code	Title of the Course	Credits
BCB050	CHEMISTRY AND METABOLISM OF CARBOHYDRATES AND LIPIDS	4

COURSE OUTCOME(S):

- CO1 Understand the classification and characteristics of chemistry of carbohydrates
CO2 Deliberate the classification and characteristics of bioenergetics
CO3 Write down the characteristics of chemistry of lipids
CO4 Learn in depth metabolism of lipids

		No. of Lectures
Unit I:		10
1.1	Chemistry of Carbohydrates	
1.1.1	Classification, monosaccharides (aldoses & ketoses) Configuration and conformation of monosaccharides (pyranose & furanose, chair & boat).	
1.1.2	Reducing and optical properties of sugars. Stability of glycosidic bond disaccharides, oligosaccharides.	
1.1.3	Structural polysaccharides—cellulose, hemicellulose, pectin, lignin, chitin, chitosan	
1.1.4	Storage polysaccharides: starch, glycogen, inulin Steric factors in polysaccharides folding, blood group polysaccharides and lectins. Glycosaminoglycans, mucopolysaccharides, hyaluronic acid Chondroitin sulfate, keratan sulfate, dermatan sulfate. Bacterial cell wall polysaccharides, proteoglycans (syndecan and agrecan)	

Unit II:		14
2.1	Metabolism of Carbohydrates and Bioenergetics	
2.1.1	Reactions and energy balance in Glycolysis, Gluconeogenesis, TCA cycle, HMP Shunt pathway, Pasteur and Crabtree effect, Anapleurotic reactions	
2.1.2	Glyoxylate cycle, Glucuronic acid cycle, Glycogen metabolism.	
2.1.3	Photosynthesis reactions for biosynthesis of glucose C3 and C4 cycle in plants	
2.1.4	Mitochondrial ETC—Organization of respiratory chain complexes, P/O ratio, ATP synthesis, Mitchell's hypothesis, uncouplers and inhibitors.	

Unit III:		12
3.1	Chemistry of Lipids	
3.1.1	Classification & biological significance of lipids, fatty acids and Steroids	
3.1.2	Bile acids and salts, Phospholipids, Oils, waxes, isoprene units, Lipoproteins, Glycolipids, Sphingolipids	

3.1.3	Cerebrosides, Gangliosides, Prostaglandins, Prostacyclins, Thromboxanes, Leukotrienes, cysteinyl leukotrienes	
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Unit IV:		
4.1	Metabolism of Lipids	
4.1.1	Fate of dietary lipids and Apo-lipoproteins Fatty acid biosynthesis, Desaturation of fatty acids Beta oxidation, breakdown of odd chain fatty acids, energy yields	12
4.1.2	Regulation of β -oxidation, ω -oxidation & α -oxidation Metabolism of phospholipids & Sphingolipids Regulation and Biosynthesis of cholesterol, action of statins	
4.1.3	Fate of acetyl CoA, formation of ketone bodies and ketosis	
4.1.4	Biosynthesis of prostaglandins, Prostacyclins, Thromboxanes, Leukotrienes, Action of aspirin	
4.1.5	Genetic defects in lipid metabolism, Medium chain acyl coenzyme A dehydrogenase deficiency MCAD, Long-chain 3-hydroxyacyl-CoA dehydrogenase (LCHAD) deficiency, Familial hypercholesterolemia	

References

- [1] Lehninger Principles of Biochemistry 4th Ed By David L. Nelson and Michael M. Cox, WH Freeman and Company.
- [2] Biochemistry by Lubert Stryer. WH Freeman and Co.
- [3] Biochemistry: The Molecular Basis of Life by Trudy McKee and James R McKee. Publisher: McGraw-Hill Higher education.
- [4] Biochemistry and Molecular biology By William H. Elliott and Daphne C. Elliott. Oxford University Press.
- [5] Biochemistry 3rd Ed. By Donald J. Voet and Judith G. Voet. John Wiley and Sons.
- [6] Biochemistry: Biomolecules, Mechanisms of Enzyme Action and Metabolism Vol 1 by D Voet. John Wiley and Sons.

Course Code	Title of the Course	Credits
BCB060	EXPERIMENTS IN IMMUNOLOGY AND BIOCHEMICAL ESTIMATIONS	4

COURSE OUTCOME(S):

- CO1 Understand in details with examples antigen antibody reactions
CO2 Specify in details with application, if applicable, oils and fats estimation
CO3 Understand in depth acid value principle and determination
CO4 Identify in details with examples mitosis and meiosis

Group I:	<ol style="list-style-type: none"> 1. Demonstration of Ag-Ab interaction: Radial immunodiffusion and ODD. 2. Demonstration of direct agglutination reaction using human blood group antigens. 3. Demonstration of indirect agglutination reaction– latex agglutination. 4. Fluorescence emission of proteins and vitamins 5. UV–Vis spectra of proteins, nucleic acids and other aromatic compounds 6. Extraction of neutral lipids, phospholipids 7. TLC of lipids and estimation of phospholipids 	
Group II:	<ol style="list-style-type: none"> 8. Iodine No. of Oils/Fats 9. Saponification Value of Oils/Fats 10. Acid Value/Peroxide Value of Oils/Fats 11. Estimation of α-Keto-acid 12. Estimation of ascorbic acid 13. Estimation of Iron 14. Estimation of Calcium 	
Group III:	<ol style="list-style-type: none"> 15. Isolation of Starch from potato and purity determination 16. Colorimetric estimation of reducing sugars (DNS reagent method) 17. Estimation of reducing sugar: Hegedorn and Jensen Method 18. Estimation of Phosphate 19. Mitosis in onion root tips 20. Meiosis in <i>tradescantia</i>/grasshopper testis 21. Total and Differential Cell Counting of blood 	
Group Study	Preparation of antigen adjuvant mixture, injection and raising antibodies in rat. Purification of antibodies Antibody titer and ELISA	

References

- [1] Analytical techniques in Biochemistry and Molecular Biology; Katoch, Rajan. Springer (2011).
- [2] Basic Methods for the Biochemical Lab; Martin Holtzhauer, Springer, (2007).
- [3] Principles and Techniques of Biochemistry and Molecular Biology 7th Edn. Keith Wilson and John Walker, Cambridge University Press, (2010).
- [4] Biochemistry LabFax, Ed. J.A.A. Chambers and D. Rickwood,, Blackwell Science, (1993),
- [5] Protein Purification Applications, S.L.V. Harris and Angal IRL Press, (1990)
- [6] Laboratory Techniques in Biochemistry and Molecular Biology, Work and Work Vol. I & II, North Holland, (1969).
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- [8] Modern Experimental Biochemistry R.F. Boyer [Ed.] (1986) Addition Wesley.
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- [10] Principles and techniques of Biochemistry and Molecular Biology; Keith Wilson and John Walker; 6th Edn. (2005) Cambridge University Press.
- [11] Biochemical Calculations, Irwin H. Segel (1976) 2nd Ed. John Wiley and Sons.
- [12] Methods in Immunology and Immunochemistry; Curtis Williams, Academic Press (1971).
- [13] Immuno Assay Hand Book; David Wild, Elsevier (2013).

Course Code	Title of the Course	Credits
BCB250	IMMUNOLOGY AND MICROBIOLOGY	4

COURSE OUTCOME(S):

- CO1 Identify in details with examples antigens and antibodies
- CO2 Understand the details of cellular basis of immunity
- CO4 Identify the classification and characteristics of MHC Complex
- CO4 Learn in depth basic concepts of microbiology

		No. of Lectures
Unit I:		
1.1	Antigens and Antibodies	
1.1.1	<u>Introduction</u> : Historical development and milestones in immunology. Barriers to infection–skin, mucous membrane, Definitions–Antigenicity, Immunogenicity, primary and secondary lymphoid organs, self and non self discrimination. Innate and acquired immunity.	12
1.1.2	<u>Antigens and Antibodies</u> : Haptens and determinants–Epitopes and paratopes. Antigenicity, carbohydrates, proteins, nucleic acids, and cells as antigens. Valency of antigen.	
1.1.3	Classes and subclasses of immunoglobulins, structure of immunoglobulins, hyper variable region, isotypic, allotypic and idiotypic variations.	
Unit II:		
2.1	Complement and Cellular Basis of Immunity	
2.1.1	<u>Complement</u> : Structure, components, properties and functions of complement pathways, biological consequences of complement activation.	12
2.1.2	Hyper sensitivity reactions (Type I, II, III and IV).	
2.1.3	<u>Cellular basis of immunity</u> : Primary and secondary immune response. Reticuloendothelial system. T, B and accessory cells. Subsets of T (T–helper cells, T–killer cells, T–suppressor cells) and B cells. Development of T and B cells. T and B cell receptors, antigen processing and presentation.	
2.1.4	Cytokines and co–stimulatory molecules–Lymphokines, interleukins structure and function of IL-2, TNF α . T and B interaction. Suppression of immune response, immunoglobulin, diversity of gene rearrangement, factors affecting diversity, class switching and clonal selection theory of Burnet.	

Unit III:		
3.1	MHC, Transplantation, Tumor Immunology and Vaccines	
3.1.1	<u>MHC</u> : MHC gene and its polymorphism, role of MHC in immune response.	16
3.1.2	<u>Transplantation</u> : Autograft, isograft, allograft and xenograft, Graft rejection, graft Vs host reaction, MHC in transplantation.	
3.1.3	<u>Immunochemical techniques</u> : Precipitation, agglutination, complement fixation, immunodiffusion, immunoelectrophoresis, immunofluorescence, RIA, ELISA.	
3.1.4	<u>Tumor immunology</u> : Tumor associated antigens, factors favoring tumor growth, immune surveillance. Tumor necrosis factors α and β Disorders of immunity: Immunological tolerance, auto immune disorders, AIDS, SCID, lupus erythematosus <u>Vaccines</u> : Adjuvants; vaccines and their preparations. Polyclonal and monoclonal antibodies–hybridoma technique.	
Unit IV:		
4.1	Microbiology	
4.1.1	Historical aspects - Discovery of microorganisms. Theory of spontaneous generation. Era of Louis Pasteur. Microbes and fermentation. Microbes and diseases-Koch's Postulates.	08
4.1.2	General characteristics: Morphology, nomenclature and classification of bacteria, virus, yeasts and fungi.	
4.1.3	Microbial nutrition-Factors influencing growth, growth curve of bacteria. Measurement of growth, continuous culture, synchronous culture and chemostat. Auxotrophs, autotrophs, heterotrophs. Methods of cultivations and preservation of microorganisms.	
4.1.4	Methods of control of microorganisms-Sterilization Techniques: Definitions of physical methods, heat (dry & moist) filtration, radiation; chemical agents-phenols, alcohols, halogens, heavy metals, aldehydes, quaternary ammonium compounds & gases.	

References

- [1] Antibodies—A Laboratory Manual; E. D. Harlow, David Lane, 2nd Edn. CSHL Press (2014).
- [2] Basic and Clinical Immunology; Stites *et al.*, [Ed] (1982) Lange.
- [3] Roitt's Essential Immunology; Ivan, M. Roitt & Peter J Delves (2001) Blackwell Science.
- [4] Immunology: Roitt *et al.*, Mosby (2001),
- [5] Kuby Immunology; Owen, Punt, Stranford, 7th Edn. W. H. Freeman (2013).
- [6] Immune System; M. C. Connel *et al.*, Eds. (1981) Blackwell Science.
- [7] Immunology at a Glance: J.H.L. Playfare [ed.] Blackwell Science, (1987).
- [8] Immunology; Jan Klein [Ed.], Blackwell Science (1990).
- [9] Introduction to Immunology; Kim Bell [Ed.,] 3rd Edn. McMillan (1990).

Course Code	Title of the Course	Credits
BCB260	HUMAN PHYSIOLOGY AND NUTRITION	4

COURSE OUTCOME(S):

- CO1 Specify the classification and characteristics of blood and respiratory systems
CO2 Identify in depth digestive and excretory systems
CO3 Learn in details with application, if applicable, concepts of nutrition
CO4 Specify the details of vitamins and minerals

		No. of Lectures
Unit I:		12
1.1	Blood and Respiratory System	
1.1.1	<u>Blood</u> –Composition, cells. Erythrocytes–structure and function, WBC–types and functions.	
1.1.2	Platelets and their function. Buffer systems; hemostasis–blood volume, blood pressure and its regulation. Blood clotting, Dissolution of clot; anticoagulants. CSF–composition and function.	
1.1.3	<u>Respiratory System</u> –Mechanism of gas exchange, oxygen binding by hemoglobin and factors affecting oxygenation. Acid–base balance and its regulation.	

Unit II:		12
2.1	Hepatobiliary, Digestive and Excretory System	
2.1.1	<u>Hepatobiliary system</u> –Anatomy of the liver, blood supply; cells–hepatocytes, endothelial cells, Kupffer cells and paranchymal cells.	
2.1.2	Secretory and excretory function; detoxification and formation of bile	
2.1.3	<u>Digestive system</u> –GI tract, digestion and absorption of carbohydrates, proteins and lipids. Mechanism of HCl production in the stomach. Gastrointestinal hormones.	
2.1.4	<u>Excretory System</u> –Ultra structure of the nephron, glomerular filtration, tubular reabsorption and tubular secretion, formation of urine.	

Unit III:		16
3.1	Nutrition, Carbohydrates, Proteins and Fats	
3.1.1	<u>Nutrition</u> –Concepts of macro and micro nutrients, essential nutrients and their classification. Food groups, proximate analysis of foods, chemical and biological analysis for nutrients.	
3.1.2	Food as source of energy, methods of determining energy value of foods, calorimetry, physiological fuel values and daily requirement of energy, high and low	

	calorie diets. Basal metabolic rate (BMR), factors affecting BMR, specific dynamic action of foods.	
3.1.3	<u>Carbohydrates</u> –dietary sources, dietary fiber essentiality of carbohydrates.	
3.1.4	<u>Proteins</u> –Evaluation of nutritive value of dietary protein PER, BV, essential amino acids, nutritional classification of proteins, supplementary value of proteins, protein calorie malnutrition–Kwashiorkor and Marasmus.	
3.1.4	<u>Fats</u> –Sources, invisible fat, essential fatty acids, PUFA.	

Unit IV:		
4.1	Vitamins and Minerals	
4.1.1	<u>Vitamins</u> –Fat soluble and water soluble vitamins, provitamins, antivitamins, dietary sources, structure, daily requirements and functions.	08
4.1.2	Deficiency symptoms of B complex members and fat soluble vitamins, hypervitaminosis, vitamin like compounds.	
4.1.3	<u>Minerals</u> –Macro and micronutrients, sources, requirements, functions and deficiency symptoms.	
4.1.4	Water metabolism–distribution in body, function and factors affecting water balance.	
4.1.5	Recommended daily allowances, special nutrition for infants, children, during pregnancy, lactation and old age.	

References

- [1] The Cell, Copper, Geoffery, M., Oxford University Press, (2001)
- [2] Text Book of Biochemistry with Clinical correlations; Thomas Devlin [Ed.] (1997), Wiley –Liss.
- [3] Lehninger- Principles of Biochemistry; DL Nelson and MM Cox [Eds], 6th Edn. Macmillan Publications (2012).
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- [9] Molecular Biology of Cell; Albertis *et. al.* (2002) Garland Science.
- [10] Biochemistry Ed. Donald Voet & Judith G. Voet, John Wiley & Sons, Inc. (2010).
- [11] Mammalian Biochemistry; White, Handler and Smith, McGraw-Hill, (1986).
- [12] Textbook of human Physiology by Guyton, 11th ed.Elesvier.
- [13] Introduction to Human Nutrition, 2nd Edn. Michael J. Gibney, Susan A. Lanham-New, Aedin Cassidy, Hester H. Vorster, Wiley-Blackwell (2009).
- [14] Nutrition: Everyday Choices, 1st Edition; Mary B. Grosvenor, Lori A. Smolin Wiley (2006).
- [15] Bioactive Food as Dietary Interventions for Liver and Gastrointestinal Disease; Watson Elsevier (2012).
- [16] Nutrition and Metabolism, 2nd Edn., Lanham S, Mac Donald I and Roche H. The Nutrition Society, London, UK, (2012).
- [17] Introduction to Human Nutrition, 2nd Edn., Gibney M, Lanham S, Cassidy A and Vorster H. The Nutrition Society, London, UK, (2012).

Course Code	Title of the Course	Credits
BCB260	RESEARCH METHODOLOGY AND BIOSTATISTICS	4

COURSE OUTCOME(S):

- CO1 Specify the classification and characteristics of research methodologies and the experimental designs
- CO2 Identify in depth statistical analysis
- CO3 Learn in details with application, if applicable, scientific write-up
- CO4 Specify the details of Interpretation of data

		No. of Lectures
Unit I:		12
1.1	Research Methodology and Design	
1.1.1	<u>Research Methodology</u> : Meaning of research, Objectives of research, Motivation in Research, Types of Research, and Research approaches.	
1.1.2	Research methods vs. Research Methodology, Research process–scientific method, Criteria for good research, Defining the research problem.	
1.1.3	<u>Research Design</u> : Meaning and need for research design, features of good design. Preparation of Scientific report, presentation of a review.	

Unit II:		12
2.1	Scientific Writing	
2.1.1	Mechanical and stylistic aspects of scientific writing–Precision and clarity of language, writing style, writing process, presentation of numerical data and scientific figures.	
2.1.2	Constraints on scientific writing–audience, format and mechanics (grammar, word choice, punctuation, tenses).	
2.1.3	Objectives and design of experiment–experimental unit, identifying variables, replications & controls, power analysis in planning experiments, treatment structure and design structure.	
2.1.4	Graphical analysis of data and presentation of results.	

Unit III:		16
3.1	Statistical Significance Analysis	
3.1.1	Significance and limitations of statistical calculations, Sampling techniques.	
3.1.2	Probability theory, random variables and distribution functions, Point and interval estimation, linear regression. Statistical evaluation of results–Hypothesis testing, interpretation of statistic for analysis of error.	
3.1.3	Measures of central tendency and dispersion	

Unit IV:		08
4.1	Testing Methods	
4.1.1	ANOVA, F-test, t-test, z-test, chi-square, correlation coefficient.	

References

- [1] Research Methodology: Methods & Techniques By CR Kothari. Publisher: New Age International
- [2] From Research to Manuscript: A Guide to Scientific Writing (Paperback) By Michael Jay Katz. Publisher: Springer
- [3] The Craft of Scientific Writing (3rd Edition) By Michael Alley. Publisher: Springer-Verlag.
- [4] Writing Scientific Research Articles: Strategy and Steps (Hardcover) By Margaret Cargill and Patrick O.Connor. Publisher: WileyBlackwell.
- [5] The Mayfield Handbook of Technical and Scientific Writing By Leslie Perelman and Edward Barrett. McGraw-Hill NY
- [6] Scientific Style and Format: The CBE Manual for Authors, Editors, and Publishers(Hardcover) 6th Ed By Edward J. Huth. Publisher: Cambridge University Press.
- [7] The Handbook of Technical Writing, Eighth Edition (Handbook of Technical Writing Practices) (Hardcover) By Gerald J. Alred, Charles T. Brusaw and Walter E. Oliu, St. Martin's Press.
- [8] Science and Technical Writing: A Manual of Style (2nd Ed.) By Philip Rubens. Publisher: Routledge, London.
- [9] The Elements of Technical Writing (Elements of Series) (Paperback) By Gary Blake and Robert W. Bly. Publisher: Longman.
- [10] Technical Writing: Principles, strategies and readings (7th Edition) By Diana C. Reep. Publisher: Longman.
- [11] Biostatistics By PN Arora and PK Malhan, Himalaya Publishing House.
- [12] Experimental Design and Data Analysis for Biologists By Gerry P. Quinn and Michael J. Keough. Publisher: Cambridge University Press.
- [13] Principles of Biostatistics (with CD-ROM) (Hardcover) By Marcello Pagano and Kimberlee Gauvreau. Publishers: Duxbury Press
- [14] Biostatistics: Experimental Design and Statistical Inference (Hardcover) By James F. Zolman. Oxford University Press.
- [15] Intuitive Biostatistics By Harvey Motulsky. Publisher: Oxford University Press

Course Code	Title of the Course	Credits
BCB260	CLINICAL RESEARCH METHODS AND INDUSTRIAL BIOCHEMISTRY	4

COURSE OUTCOME(S):

- CO1 Specify the classification and characteristics of clinical practice and clinical research
- CO2 Identify in depth fermentation technology and downstream processing
- CO3 Learn in details with application, if applicable, clinical research methods
- CO4 Specify the details of steps involved in drug discovery

		No. of Lectures
Unit I:		10
1.1	Introduction to Clinical Research	
1.1.1	Introduction to Clinical Research, Terminologies and definition in Clinical Research, Origin and History of Clinical Research	
1.1.2	Difference between Clinical Research and Clinical Practice, Types of Clinical Research, Phases of clinical research	
1.1.3	Clinical Trials in India–The National Perspective, Post marketing surveillance	
1.1.4	Pharmaceutical Industry–Global and Indian Perspective Clinical Trial market, Career in Clinical Research	

Unit II:		14
2.1	Clinical Research Methods	
2.1.1	Design of experiments, factorial experiments, randomization, interaction among factors.	
2.1.2	Types of studies: Cohort studies, double blind, placebo control, cross over and double dummy.	
2.1.3	Introduction to Good Clinical Practices, Clinical Trial Development: Protocol Design and Development, Case Report Form Design and Development, Principals of Data Management, Clinical Trial Management: Maintaining and Managing Essential Documents, Recording and Reporting Non–Serious and Serious Adverse Events.	

Unit III:		12
3.1	Drug Discovery Concepts and Biostatistics	
3.1.1	Proof of concept, target identification and validation. Identifying the lead compound, optimization of lead compound, mechanism of action, drug target and validation of target.	
3.1.2	Safety pharmacology, pharmaco–kinetics and	

	pharmaco–dynamics, acute and chronic toxicity Development of new drug/molecules and elucidation of their mechanisms of actions, formulations, factors affecting drug efficacy, drug resistance, traditional medicines; biotransformation.	
3.1.3	Statistical concept: Data structure, sampling methods, collection, classification and tabulation of data, graphical and diagrammatic representation, histogram, frequency polygon, frequency curve, bar graph, pie chart.	
3.1.4	Measure of central tendency: Mean, median, mode, mean deviation, standard deviation, standard error	
3.1.4	Types of distribution of data: Normal, binomial, Poisson, Z-test, t-test and ANOVA. Correlation and regression.	
Unit IV:		
4.1	Bioprocess Methods	
4.1.1	Basics of chemical engineering, mass transfer, heat generation and removal, fluid dynamics:	12
4.1.2	Bernoulli's principle, viscosity, hydraulic conductivity, capillary flow, control and applications of industrial processes, process evaluation and development, over production of metabolites and methods;	
4.1.3	Fermentation–Submerged and solid state fermentation Fermentor design, Industrial use of microbes. Strain improvement, Inocula preparation, Downstream processing–Recovery and purification of intracellular and extra cellular products. Methods to maximize the yield.	

References

- [1] Basic Test for Drugs, WHO-GENEVA 1998 edition
- [2] Who Expert Committee on Specification for Pharmaceutical Preparation WHO-GENEVA, 2005 edition
- [3] Who Expert Committee on Biological Standardization WHO-GENEVA 2003 edition
- [4] Clinical Research Fundamental and Practice –Vishal Bansal Parar Medical Publisher, 2010 edition
- [5] Introduction to Pharmacopoeia CBS Publishers and Distributors 1991 edition
- [6] Essential of Clinical Research –Dr. Ravindra B. Ghooi and Sachin C. Itkar Nirali Prakashan 2010 edition
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- [8] A Comprehensive Clinical Research Manual-Samir Malhotra, Nusrat Shafiq, Promila Pandhi Jaypee Brothers Medical Publishers (P) Ltd, 2008 edition
- [9] Industrial microbiology, A.H. Patel
- [10] Principles of Fermentation technology, Stanburry. P. Whitaker and S.J. Hall, 1995
- [11] Biotechnology–U. Sathyanarayana. YLL

Course Code	Title of the Course	Credits
BCC070	CELL BIOLOGY, ENDOCRINOLOGY AND CELL SIGNALING	4

COURSE OUTCOME(S):

- CO1 Specify in details with examples cellular organization
- CO2 Learn the characteristics of endocrinology
- CO3 Learn in depth cell signaling
- CO4 Write down the characteristics of membrane biology

		No. of Lectures
Unit I:		12
1.1	Cellular Organization, Division and Cytoskeletons	
1.1.1	Cell types–organization of prokaryotic and eukaryotic cells.	
1.1.2	Cell division–mitosis and meiosis, cell cycle–phases of cell cycle, cyclins and cdk. Regulation of cell growth and cell cycle.	
1.1.3	Cell motility–molecular motors, microtubules, structure and composition. Microtubular associated proteins–role in intracellular motility.	
1.1.4	Cellular organelles–Nucleus–internal organization, traffic between the nucleus the nucleolus, and cytoplasm. Endoplasmic reticulum–protein sorting and transport, golgi apparatus and lysosomes, morphology and function of mitochondria, chloroplasts and peroxisomes, glyoxysomes.	

Unit II:		12
2.1	Membrane Biology	
2.1.1	Organization of lipid monolayer, bilayer, Physicochemical properties of biological membranes - compositions, supra molecular organization - Singer and Nicolson's model.	
2.1.2	Membrane asymmetry-lipids proteins and carbohydrates, lateral diffusion, biogenesis of lipids and proteins. Polarized cells, membrane domains- caveolae, rafts, Membrane lipid and protein turnover, intracellular targeting of proteins. Factors influencing fluidity of membrane	
2.1.3	Membrane transport - Laws of diffusion across membranes, simple diffusion, facilitated diffusion and active transport - glucose transporter Na ⁺ K ⁺ ATPase (Structure and mechanism of action), bacterial phosphotransferase system. Endocytosis, receptor mediated endocytosis, exocytosis, ion channels, aquaporin channel, ionophores. Patch clamp technique.	

Unit III:		
3.1	ENDOCRINOLOGY	
3.1.1	<u>Endocrine System</u> –Endocrine organs in man. Hierarchy and regulation of hormone release.	12
3.1.2	Structure and control of hypothalamus, GRH, somatostatin, TRH, CRH, GnRH. Pituitary-anatomy and structure.	
3.1.3	Hormones of anterior, posterior and median lobes. Pro-opiomelanocortin. Thyroid, parathyroid, adrenals, gonads–Testes and ovaries. Menstrual cycle. Hypothalamus–pituitary target organ axis and regulation by feedback mechanism, Pineal gland, melatonin and circadian rhythm	
3.1.4	Classification of hormones based on solubility and structure, mechanism of action of water soluble and lipid soluble hormones.	
Unit IV:		
4.1	Cell Signaling	
4.1.1	Nerve transmission–Central and peripheral nervous systems. Structure of neuron, axon, dendrites, synapse neuromuscular junction. Neurotransmitters- mechanisms of nerve conduction. α and β adrenergic neurons, nicotinic and muscarinic neurons.	12
4.1.2	Muscle contraction–Skeletal muscle and smooth muscle contraction, muscle proteins–actin, myosin, tropomyosine, troponins, mechanisms of muscle contraction, role of calcium and calmodulin Biochemistry of vision	
4.1.3	Cellular signaling: Extra cellular signaling–G Protein linked receptors ,Role of cyclic AMP, IP3, DAG, Ca^{2+} as a second messenger, receptor tyrosine kinases , MAP kinase pathway, NF κ B pathway, apoptosis, Cell survival pathway, Jak/Stat pathway, TGF β Signaling. Multiple signaling path ways–Insulin receptor (regulation of blood glucose)	
4.1.4	Steroid hormone receptors, structural organization of receptor protein, hormone binding domain, antigenic domain and DNA binding domain.	

References

- [1] The World of the cell by Becker, Kleinsmith and Harden Academic Internet Publishers; 5th edition (2006)
- [2] The Cell: A Molecular Approach, Fourth Edition by Geoffrey M. Cooper and Robert E. Hausman.
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Course Code	Title of the Course	Credits
BCC050	CLINICAL BIOCHEMISTRY	4

COURSE OUTCOME(S):

- CO1 Identify in details with application, if applicable, specimen collection and analysis
CO2 Specify in details with application, if applicable, metabolic disorders
CO3 Write down the characteristics of hormonal disorders
CO4 Write down in details with application, if applicable, hematology

		No. of Lectures
Unit I:		10
1.1	Specimen Collection and Analysis	
1.1.1	Concepts of accuracy, precision, reproducibility, reliability, and other factors in quality control.	
1.1.2	Normal values. Specimen collection and Processing: Collection of blood–venipuncture, skin puncture, arterial puncture. Anticoagulants. Collection and analysis of normal and abnormal urine–timed urine specimens, preservatives.	
1.1.3	Clinical significance of sugars, proteins, ketone bodies, bilirubin and porphyrins. CSF–collection, composition and analysis. Amniotic fluid–Origin, collection, composition.	

Unit II:		14
2.1	Disorders	
2.1.1	Disorders of carbohydrate metabolism: Diabetes mellitus, glycohemoglobins, hypoglycemias, galactosemia and ketone bodies.	
2.1.2	Various types of glucose tolerance tests. Glycogen storage diseases.	
2.1.3	Lipid profile, lipidosis and multiple sclerosis. Causes and diagnosis of the disorders of HDL–cholesterol, LDL–cholesterol and triglycerides.	
2.1.4	Cancer: Etiology, diagnosis, treatment and prognosis. Carcinogens, oncogens, mechanism.	
2.1.5	Biochemistry of ageing: Cellular senescence, Role of Telomerase in aging, Alzheimer’s disease, Parkinson’s disease.	

Unit III:		12
3.1	Enzymes and Hormonal Disorders	
3.1.1	<u>Evaluation of organ function tests</u> : Clinical assessment of renal, hepatic, pancreatic, gastric, intestinal and thyroid functions. Clinical importance of bilirubin.	
3.1.2	<u>Diagnostic enzymes</u> : Principles of diagnostic enzymology. Clinical significance of aspartate aminotransferase, alanine aminotransferase, creatine	

3.1.3	kinase, aldolase and lactate dehydrogenase. Enzyme tests in determination of myocardial infarction. Enzymes of pancreatic origin and biliary tract.	
3.1.4	<u>Hormonal disorders</u> : Protein hormones (anterior pituitary hormones, posterior pituitary hormones), steroid hormones, adrenocorticosteroids, and reproductive endocrinology. Disorders of thyroid hormones.	

Unit IV:		
4.1	Hematology	
4.1.1	<u>Biochemical aspects of hematology</u> : Total cell count, differential count, hematocrit.	12
4.1.2	Disorders of erythrocyte metabolism, hemoglobinopathies, thalassemias, thrombosis, porphyrias and anemias. Laboratory tests to measure coagulation and thrombolysis.	
4.1.3	Doping	
4.1.4	<u>Detoxification in the body</u> : Enzymes of detoxification, polymorphism in drug metabolizing enzymes. Mechanism of drug action and channels of its excretion.	
4.1.5	Test for lung function: Chest X-ray, Spirometry. Test for Brain function: EEG, MRI, CT.	

References

- [1] Textbook of Medical Biochemistry by MN Chatterjea and Rana Shinde, Jaypee Brothers.
- [2] Lehninger Principles of Biochemistry 5th Ed by David L. Nelson and Michael M. Cox, WH Freeman and Company.
- [3] Davidson's Principles and Practice of Medicine: A Textbook for Students and Doctors (Hardcover) 15th Ed by LSP Davidson, J MacLeod and CRW Edwards. Publisher: Churchill Livingstone.
- [4] Medical Biochemistry (Paperback) by John W. Baynes and Marek Dominiczak. Publisher: Mosby.
- [5] Clinical Biochemistry: An Illustrated Colour Text (Paperback) 3rd Ed By Allan Gaw, Michael Murphy, Robert Cowan, Denis O'Reilly, Michael Stewart and James Shepherd. Publisher: Churchill Livingstone.
- [6] Review of Medical Physiology (Lange Basic Science) (Paperback) By William F. Ganong. Publisher: McGraw-Hill Medical
- [7] Harper's Biochemistry (Lange Medical Books) (Paperback) By Robert K. Murray, Daryl K. Granner, Peter A. Mayes and Victor W. Rodwell. Publisher: Appellton and Lange.
- [8] Clinical Biochemistry by Richard Luxton. Scion Publishing Ltd.
- [9] Principles of Medical Biochemistry: With STUDENT CONSULT Online Access (Paperback) by Gerhard Meisenberg and William H. Simmons. Publisher: Mosby.

Course Code	Title of the Course	Credits
BCC060	EXPERIMENTS IN CLINICAL BIOCHEMISTRY AND MOLECULAR BIOLOGY	4

COURSE OUTCOME(S):

- CO1 Specify the details of urine and blood analysis
- CO2 Specify the characteristics of determination of enzyme activity
- CO3 Identify the classification and characteristics of DNA quantification and analysis
- CO4 Deliberate the details of isolation of nucleic acids from plant, animal and microbial sources

Group I:	<p>Urine analysis</p> <ol style="list-style-type: none"> 1. Qualitative analysis of urine for normal organic and inorganic constituents 2. Qualitative analysis of urine for abnormal constituents- Glucose, albumin, Ketone bodies. 3. Quantitative estimation of Creatine and Creatinine, Urea, Uric acid, Sulphate, Chloride 4. Titrable acidity <p>Blood analysis</p> <ol style="list-style-type: none"> 5. Quantitative estimation of Urea, Uric acid, Creatine, Cholesterol HDL-C and LDL-C 6. Blood glucose and GTT 	
Group II:	<p>Determination of Enzyme activity of</p> <ol style="list-style-type: none"> 7. Alkaline phosphatase 8. SGOT 9. SGPT 10. LDH 11. Electrophoresis of lipoproteins: Serum proteins. 12. Albumin/Globulin Ratio. 13. Fractionation of serum proteins-Ammonium sulphate precipitation. 14. Isolation of DNA and RNA from biological sources. 15. Quantitative determination of DNA and RNA. 	
Group III:	<ol style="list-style-type: none"> 16. Determination of melting temperature of DNA (T_m) 17. Sub-cellular fractionation of rat liver by differential centrifugation and marker analysis 18. Determination of activities of marker enzymes 19. Preparation of erythrocyte ghosts 20. Kinetics of uptake of glucose by erythrocytes 21. Viability of cells by trypan blue dye exclusion 22. Study of morphology of <i>Drosophila melanogaster</i> 	

	23. Study of mutants of <i>Drosophila melanogaster</i> 24. Study of polytene chromosomes of <i>Drosophila melanogaster</i>	
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Group Study	Isolation of plasmid DNA, Restriction digestion of plasmid DNA, ligation of DNA fragment into a plasmid vector, preparation of competent cells, <i>E.Coli</i> transformation and amplification of DNA by PCR.	
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References

- [1] Practical Clinical Biochemistry, ed. Harold Varley, 4th edn. CBS Publishers (1988).
- [2] Practical Clinical Biochemistry: Methods and Interpretation, ed. Ranjna Chawla, Jaypee Brothers Medical Publishers (1996).
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- [4] Hawk' s Physiological Chemistry, ed. Oser, 14th Edn.(1976), Tata-McGrawHill.
- [5] Biochemistry, ed. Plummer Tata-McGraw Hill, (1971).
- [6] Molecular Biology Techniques; Sue Carson, Heather Miller and D. Scott Witherow, Academic Press (2011).
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- [8] Principles of Gene Manipulations; 6th Edn. S.B. Primrose, R.M. Twyman, and R.W. Old, Blackwell Science (2012).
- [9] Gene Cloning and DNA analysis- An Introduction; T. A. Brown, 5th Edition, Wiley-Blackwell (2006).
- [10] Laboratory methods in Enzymology; Part-A; Jon Lorsch, Academic Press (2014).
- [11] Gene Cloning Laboratory Manual 4th Edn. Michael R. Green and Joseph Sambrook, CSHL Press (2014).
- [12] Current Protocols in Molecular Biology; S Gallagher, Wiley Interscience (2008).

Course Code	Title of the Course	Credits
BCC220	GENOMICS, PROTEOMICS AND BIOINFORMATICS	4

COURSE OUTCOME(S):

- CO1 Specify the details of DNA sequencing methods
- CO2 Specify the characteristics of determination of Proteins
- CO3 Identify the classification and characteristics of microarray data
- CO4 Deliberate the details of bioinformatics in biological databases and sequencing analysis

		No. of Lectures
Unit I:		12
1.1	Structural Organization of Genome and Sequencing	
1.1.1	Structural organization of genome in Prokaryotes and Eukaryotes, Organelle DNA–mitochondrial, chloroplast,	
1.1.2	DNA sequencing–principles and translation to large scale projects, Recognition of coding and non–coding sequences and gene annotation. Tools for genome analysis–RFLP, DNA fingerprinting, RAPD, PCR, Linkage and Pedigree analysis–physical and genetic mapping.	
1.1.3	Microbes, plants and animals, Accessing and retrieving genome project information from web, Comparative genomics, ESTs and SNPs.	

Unit II:		12
2.1	Proteomics	
2.1.1	Protein analysis (includes measurement of concentration, amino-acid composition, N-terminal sequencing),	
2.1.2	2–D electrophoresis of proteins, Microscale solution isoelectricfocusing, Peptide fingerprinting,	
2.1.3	LC/MS-MS for identification of proteins and modified proteins, MALDI-TOF	
2.1.4	SAGE and Differential display proteomics, Protein-protein interactions, Yeast two hybrid systems.	

Unit III:		08
3.1	Functional Genomics, Proteomics and Metabolomics	
3.1.1	Analysis of microarray data, Protein and peptide microarray–based technology; PCR–directed protein <i>in situ</i> arrays	
3.1.2	Structural proteomics	
3.1.3	Metabolomics	

Unit IV:		
4.1	Biological Databases and Sequence Analysis	
4.1.1	<u>Introduction biological databases</u> : Types (relational & object-oriented). Primary, secondary & specialized databases.	16
4.1.2	Types of databases–Nucleotide sequence database, EMBL, Genebank, Unigene, Genome biology, Protein dBase (Swiss-prot & Trembl and Motif) and 3D structure databases (PDB, SCOP, Cath, Genecards, SRS & Entrez).	
4.1.3	Computational approaches for gene identification, ORF and Human Genome Project.	
4.1.4	<u>Basics of sequence analysis</u> : Alignments using BLAST and FASTA, Multiple Sequence Alignment (CLUSTAL-X and CLUSTAL-W), Application of multiple sequence alignment	
4.1.5	Protein Structure Prediction in Bioinformatics– <i>Ab initio</i> based methods, Homology based methods, secondary structure prediction.	
4.1.6	Protein structure comparison–intermolecular and intramolecular methods. Phylogenetic construction by distance based methods	

References

- [1] Voet D, Voet JG & Pratt CW, Fundamentals of Biochemistry, 2nd Edition. Wiley 2006
- [2] Brown TA, Genomes, 3rd Edition. Garland Science 2006
- [3] Campbell AM & Heyer LJ, Discovering Genomics, Proteomics and Bioinformatics, 2nd Edition. Benjamin Cummings 2007
- [4] Primrose S & Twyman R, Principles of Gene Manipulation and Genomics, 7th Edition, Blackwell, 2006.
- [5] Glick BR & Pasternak JJ, Molecular Biotechnology, 3rd Edition, ASM Press, 1998.
- [6] Essential Bioinformatics (Paperback) by Jin Xiong. Cambridge University Press.
- [7] Bioinformatics: Methods & Protocols by Stephen Misener and Stephen A. Krawetz, Humana Press.
- [8] Humana Press.
- [9] Essentials of Bioinformatics by Irfan Ali Khan and Atiya Khanum. Publisher: Ukaaz Publications.
- [10] Bioinformatics: Sequence and Genome Analysis (Hardcover) by David W. Mount. Cold Spring Harbor Laboratory Press
- [11] Introduction to Bioinformatics (Paperback) by Arthur M. Lesk. Oxford Univ Press.
- [12] Introduction to Bioinformatics: A Theoretical and Practical Approach (Paperback) by David Womble, Stephen A. Krawetz and David D. Womble. Humana Press Inc., U.S.
- [13] Applied Bioinformatics: An Introduction (Paperback) by Paul M. Selzer, Richard Marhofer and Andreas Rohwer. Publisher: Springer-Verlag Berlin and Heidelberg GmbH & Co. K.
- [14]

Course Code	Title of the Course	Credits
BCC220	BIOTECHNOLOGY	4

COURSE OUTCOME(S):

- CO1 Write down the characteristics of tools of genetic engineering
- CO2 Learn in depth DNA Cloning
- CO3 Write down the characteristics of Industrial biotechnology and gene therapy
- CO4 Identify in depth biosafety and bioethics

		No. of Lectures
Unit I:		12
1.1	Tools of Genetic Engineering	
1.1.1	Basic principles–mechanism of natural gene transfer by Agrobacterium, generation of foreign DNA molecules.	
1.1.2	Restriction enzymes, their types and target sites, cutting and joining DNA molecules, linkers, adapters, homopolymers, enzymes used in genetic engineering.	
1.1.3	Cloning vehicles and their properties, natural plasmids, in vitro vectors, cosmids and T-DNA based hybrid vectors.	
1.1.4	Cloning strategies–cloning with single strand DNA vectors, cDNA cloning and gene libraries, recombinant selection and screening methods, expression of cloned genes–problems and solutions, shuffle vectors.	
1.1.5	DNA sequencing strategies–Sanger’s and Maxam–Gilbert’s methods, applications of PCR and DNA hybridization, Southern, Northern and Western blotting.	

Unit II:		12
2.1	DNA Cloning and Sequencing	
2.1.1	Techniques of tissue culture–culturing explants and haploids, protoplasts fusion and embryoids.	
2.1.2	Methods of gene transfer to plants, animals and bacteria–Ca transfection, electroporation, shotgun and others.	
2.1.3	Transgenic plants, GM foods and Biopesticides, gene knockouts and transgenic animals.	
2.1.4	Biodegradation and its applications, bioleaching.	

Unit III:		
3.1	Industrial Biotechnology and Gene Therapy	
3.1.1	Applications of biotechnology–industrial biotechnology–Fermentors, principle, types product recovery and purification of ethanol, citric acid, vitamin B12, streptomycin.	16
3.1.2	Enzyme biotechnology–production and uses of industrially important enzymes such as protease, immobilization of enzymes and their applications	
3.1.3	Waste treatment, bioenergy and biogas production.	
3.1.4	Gene therapy (somatic)-the principle and approaches.	

Unit IV:		
4.1	Biosafety and Bioethics	
4.1.1	Biotechnology–potential hazards, biological weapons, biosafety of GM foods and GMOs–substantial equivalence and safety testing, gene drain, the tangled genes	08
4.1.2	Human genome research–the objectives and approaches, genomics and genome prospecting–the controversies, issues of biotechnology-social and scientific, technology protecting systems and the terminator.	
4.1.3	IPR, its concepts and conditions–patenting of genes, cells and life forms, evaluation of life patenting.	

<p>References</p> <p>[1] Fermentation Biotechnology O.P. Ward. 1989 Prentice Hall.</p> <p>[2] Biotechnology J.E. Smith Cambridge University Press 1996.</p> <p>[3] Introduction to Biotechnology Brown, Campbell and Priest Blackwell Science 1987.</p> <p>[4] A Textbook on Biotechnology H.D. Kumar 2nd edition East West Press 1998.</p> <p>[5] Molecular Biotechnology Glick and Pasternak, Panima Publ.</p> <p>[6] From Genes to clones Winnaecker VCH Publication.</p> <p>[7] Elements of Biotechnology P.K. Gupta, Rastogi Publication, 1998.</p> <p>[8] Molecular Biology and Biotechnology. Walker and Gingold. 3rd ed. Panima Publ. 1999.</p> <p>[9] Plant Biotechnology. Ignacimuthu, Oxford, IBH.</p> <p>[10] Recombinant DNA Technology, Watson, Scientific American Publ.</p> <p>[11] Principles of Genome analysis, Primrose, Oxford University Press, 1998.</p>
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Course Code	Title of the Course	Credits
BCC220	PHARMACEUTICAL BIOCHEMISTRY	4

COURSE OUTCOME(S):

- CO1 Identify the details of ADME mechanism of drugs
- CO2 Learn in details with application, if applicable, Drug receptor interactions
- CO3 Deliberate in details with application, if applicable, Mode of action of anti cancer drugs
- CO4 Write down in depth Drug tolerance and abuse

		No. of Lectures
Unit I:		10
1.1	Drugs	
1.1.1	Drugs: History of Drugs Classification of drugs, routes of drug administration, absorption and distribution of drugs.	
1.1.2	Factors influencing drug absorption and elimination of drugs.	

Unit II:		14
2.1	Drug Receptor and Metabolism	
2.1.1	Drug-Receptor interactions involvements of binding forces in drug receptor interaction, drug action not mediated by receptors.	
2.1.2	Drug metabolism: Mechanism of phase I and II enzyme reactions, biochemical importance of xenobiotic metabolism.	

Unit III:		12
3.1	Anticancer Drugs	
3.1.1	Cancer: Cancer and principles of cancer chemotherapy, mode of action of anti cancer drugs.	
3.1.2	Antimetabolites, antibiotics, alkylating agents and other agents,	

Unit IV:		12
4.1	Adverse Drug Reactions	
4.1.1	Adverse drug reactions and drug induced side effects.	
4.1.2	Biological effects of drug abuse and drug dependence.	
4.1.3	Drug tolerance and intolerance.	

References

- [1] The Pharmacology volume I and II –Goodman and Gillman
- [2] Basic Pharmacology –Foxter Cox
- [3] Oxford text book of Clinical Pharmacology and Drug Theraphy ,D.G Grahme Smith and J.K.Aronson
- [4] Pharmacology and Pharmatherapeutics – R.S.Satoskar,S.D.Bhandhakarand
- [5] Essentials of Pharmacotherapeutics ,Barav.F.S.K
- [6] Lippincotts illustrated review Pharmacology, Mary.J.Mycek,Richards ,Pamela

Course Code	Title of the Course (Open Elective)	Credits
BCC630	NUTRITION AND HEALTH	4

COURSE OUTCOME(S):

- CO1 Identify the details of basic concepts of nutrition
CO2 Learn in details with application, if applicable, nutrients
CO3 Deliberate in details with application, if applicable, nutrition associated problems
CO4 Write down in depth social health problems

		No. of Lectures
Unit I:		10
1.1	Basic Concepts in Nutrition	
1.1.1	Understanding relationship between food, nutrition, health and food pyramid.	
1.1.2	Functions of food: Physiological, psychological and social Basic food groups and concept of balanced diet	
1.1.3	Energy: Functions, sources and concept of energy balance.	
1.1.4	Nutritional requirements: Physiological considerations and nutritional concerns for the following life stages: Adult man / woman Preschool children Adolescent children Pregnant woman, Nursing woman and infant Geriatrics	

Unit II:		14
2.1	Nutrients	
2.1.1	Functions, Recommended Dietary Allowances, dietary sources, effects of deficiency and/ or excess consumption on health of the following nutrients: Carbohydrates and dietary fibre Lipids Proteins Fat soluble vitamins: A, D, E and K Water soluble vitamins: Thiamin, Riboflavin, Niacin, Pyridoxine, Folate, Vitamin B12 and Vitamin C Minerals: Calcium, Iron, Zinc and Iodine	
2.1.2	Gut Microbiome	

Unit III:		
3.1	Nutritional problems, their implications and related nutrition programmes	
3.1.1	Etiology, prevalence, clinical features and preventive strategies of <u>Undernutrition:</u> Protein energy malnutrition, nutritional anemia's, vitamin A deficiency and iodine deficiency disorders <u>Overnutrition:</u> Obesity, Coronary Heart Disease and Diabetes	14
3.1.2	<u>National Nutrition Policy and Programmes:</u> Integrated Child Development Services (ICDS) Scheme Mid day Meal Programme (MDMP) National programmes for prevention of Anemia Vitamin A deficiency and Iodine Deficiency Disorders	

Unit IV:		
4.1	Social health problems	
4.1.1	Smoking Alcoholism AIDS including AIDS Control Programme	10
4.1.2	<u>Nutrition for special conditions:</u> Nutrition for physical fitness and sport, BMI Feeding problems in children with special needs Considerations during natural and man-made disasters e.g. floods, war. Basic guidelines in disaster management	

References

- [1] Text Book of Biochemistry with Clinical correlations; Thomas Devlin [Ed.] (1997), Wiley –Liss.
- [2] Harper's Review of Biochemistry, Murray et. al., (1997) 24th Edn., Lange
- [3] Bryan Derrickson, Gerard J Tortora Principles of Anatomy and Physiology , twelfth Ed, 2011, Wiley & Sons Limited.
- [4] Bamji MS, Krishnaswamy K and Brahmam GNV (Eds) (2009). Textbook of Human Nutrition, 3rd edition. Oxford and IBH Publishing Co. Pvt. Ltd. New Delhi.
- [5] Mudambi, SR and Rajagopal, MV. Fundamentals of Foods, Nutrition and Diet Therapy; 2012; New Age International Publishers
- [6] Wardlaw GM, Hampl JS. Perspectives in Nutrition; Seventh Ed; 2007; McGraw Hill.
- [7] Swaminathan M. Handbook of Foods and Nutrition; Fifth Ed; 1986; BAPPCO.
- [8] Suri S. and Malhotra A. Food Science, Nutrition & Food Safety Pearson India Ltd. 2014.
- [9] Edelstein S, Sharlin J (ed). Life Cycle Nutrition- An Evidence Based Approach; 2009; Jones and Barlett Publishers.
- [10] ICMR (1989) Nutritive Value of Indian Foods. National Institute of Nutrition, Indian Council of Medical Research, Hyderabad.
- [11] ICMR (2011) Dietary Guidelines for Indians – A Manual. National Institute of Nutrition, Indian Council of Medical Research, Hyderabad
- [12] World Health Organization (2006). WHO Child Growth Standards: Methods and development: Length/height-for-age, weight-for-age, weight-for-length, weight-for-height and body mass index-for-age (d).
- [13] Lakra P, Singh MD. Textbook of Nutrition and Health; First Ed; 2008; Academic 14 Excellence

Course Code	Title of the Course	Credits
BCD010	MOLECULAR BIOLOGY AND GENE REGULATION	4

COURSE OUTCOME(S):

- CO1 Write down the characteristics of DNA characteristics and replication
- CO2 Write down in depth Transcription and regulation
- CO3 Learn in depth translation
- CO4 Identify in depth translational regulation

		No. of Lectures
Unit I:		14
1.1	DNA Replication and Gene Expression	
1.1.1	<u>Introduction</u> : Historical perspective, types of RNA, Central dogma of molecular biology.	
1.1.2	<u>DNA Replication</u> : Nearest neighbor base frequency analysis. Replication of DNA semiconservative model- Meselson and Stahl experiment. Direction of replication of <i>E.coli</i> , discontinuous replication-Okazaki fragments.	
1.1.3	Composition and properties of DNA polymerase I, II and III. of <i>E.coli</i> DNA ligase, fidelity of replication. DNA topoisomerases and gyrases.	
1.1.4	Replication in viruses single stranded DNA virus, ϕ X174, rolling circle model. Replication of mitochondrial DNA.	
1.1.5	Organization of prokaryotic and eukaryotic gene-promoters, introns, exons, other regulatory sequences, enhancers, silencers, function of introns.	
1.1.6	<u>Regulation of Gene expression in prokaryotes</u> : Operon model-Lac operon-structure and regulation; Galactose operon-role of two promoters; Arabinose operon-positive control; tryptophan operon-attenuation control.	
1.1.7	<u>Regulation of gene expression at the level of DNA structure</u> : Super coiling, DNA methylation, role of nucleosome structure of eukaryotic DNA in gene expression-eg. glucocorticoid gene, chromatin remodeling	

Unit II:		
2.1	Transcription and Regulation	
2.1.1	<u>Transcription</u> : RNA biosynthesis in prokaryotes and eukaryotes- initiation, elongation and termination. RNA polymerase I, II and III. RNA dependent RNA synthesis - RNA replicase of QB virus.	10
2.1.2	Processing of eukaryotic mRNA–cap addition, poly A tail addition, intron splicing, RNA editing. Processing of t–RNA.	
2.1.3	<u>Regulation at the level of transcription</u> : Transcription factors, TF II. Formation of initiation complex. Role of enhancers	
2.1.4	<u>Regulation at the level of RNA processing</u> : RNA export and RNA stability. Factors affecting RNA stability. RNA degradation.	
Unit III:		
3.1	Translation	
3.1.1	<u>Translation</u> : Genetic code, triplet codon, Universality features of the genetic code, assignment of codons studies of Khorana, Nirenberg, triplet binding techniques, degeneracy of codons, wobble hypothesis, evolution of genetic code and codon usage, variation in the codon usage.	12
3.1.2	<u>3D structure of prokaryotic and eukaryotic ribosomes</u> , <u>Translation</u> : initiation, elongation and termination. Role of m–RNA and t–RNA; aminoacyl t–RNA synthetase and its role in translation accuracy, signal sequence, translational proof-reading, translational inhibitors.	
3.1.3	<u>Post translational modification of proteins</u> –signal peptide cleavage, disulphide bond formation, O–and N–Glycosylation, folding of nascent protein, role of chaperones, attachment of glycosyl anchor, and other modifications.	
Unit IV:		
4.1	Translational Regulation	
4.1.1	<u>Regulation at the level of translation</u> : Secondary structure in the 5' and 3' untranslated region–eg. Regulation of Ferritin and Transformation of m-RNA. Role of upstream AUG codons. (eg. GCN 4 gene regulation), transplicing and translational introns, protein splicing introns.	12
4.1.2	<u>Role of ribosomes in the regulation of translation</u> : Proof–reading mechanism. Ribosomal optimization of translation. Regulation at the level of ribosome assembly. Regulation at the level of post-translational modification, protein stability, N–end rule, PEST and other sequences	

References

- [1] Molecular Biology of the Cell, Alberts et al., Garland Publications, (2012).
- [2] Molecular Biology, David Freifelder, Narosa Publishers, (1997).
- [3] Molecular Biology Robert F. Weaver, McGraw Hill (2012).
- [4] Molecular Biology of Gene; Watson, J.D. et al., 5th Edn. Pearson Education; (2004).
- [5] Principles of Virology; S.J. Flint et al., ASM Press (2000).
- [6] Biochemistry and Molecular Biology; 5th Edn. D.Papachristodoulou, A. Snape, W.H. Elliott, and D. C. Elliott Oxford University Press (2014)
- [7] Chromatin structure and Gene Expression; 2nd Edn. Sarah Elgin, Jerry Workman, Oxford University Press (2000)
- [8] Molecular Cell Biology; Harvey Lodish 5th Edn. (2010)
- [9] Biochemistry 5th Edn. Jeremy M. Berg, John L. Tymoczko, Lubert Stryer (2011).
- [10] Genome Stability: DNA Repair and Recombination; James Haber, Garland Science (2013)

Course Code	Title of the Course	Credits
BCD070	GENETICS AND GENETIC ENGINEERING	4

COURSE OUTCOME(S):

- CO1 Deliberate in details with examples gene organization
- CO2 Understand the classification and characteristics of population genetics
- CO3 Deliberate the details of cloning vectors
- CO4 Understand the details of applications of genetic engineering

		No. of Lectures
Unit I:		12
1.1	Mendelism and Gene Organization	
1.1.1	Basic principles of Mendelism–Laws of inheritance, dominance, codominance, epistasis, (eg. Comb shape in chickens). Pleiotropism. Cytoplasmic inheritances (eg. Shell Coiling)	
1.1.2	Organisation of genes in chromosomes–Single copy gene, gene families, tandemly repeating genes, pseudo genes	
1.1.3	Chromosome number–Ploidy, karyotyping, sex chromosome and dosage compensation. Mobile genetic elements,	
1.1.4	Chromosomal basis of human diseases–Extra or missing chromosome, abnormality in chromosome structure–deletion duplication, inversion and translocation.	
1.1.5	Gene and development–Model systems for studying development in Drosophila, genetic control of development in Drosophila, anteroposterior axis, specification role of maternal genes, segmentation of larval body, gap genes, pair rule genes, homeotic genes, complex gene interaction in development, sequential gene action.	

Unit II:		12
2.1	Population Genetics and Mutations	
2.1.1	Population Genetics–Genetic variation, Hardy–Weinberg Law, genetic frequency, migration, genetic equilibrium	
2.1.2	Mutations- nature of mutations–spontaneous and induced mutation, conditional lethal (eg. Temperature sensitive) mutation. Biochemical basis of mutation. Point mutation, base substitution mutation, missense, nonsense and silent mutations. Mutation rates. Chemical mutagens, radiation induced mutation, reverse mutations and suppressor mutations–intergenic and intragenic suppression, reversion as a means of detecting mutagens - Ames test	
2.1.3	Repair Mechanism–DNA repair mechanisms. Reciprocal	

	recombination, site specific recombination, <i>E. coli</i> rec system. Holliday model of recombination, SOS repair.	
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Unit III:		
3.1	Cloning Vectors, Gene Transfer and Sequencing Methods	
3.1.1	Introduction to genetic engineering: Basic steps of gene cloning, enzymes used in genetic engineering.	08
3.1.2	Cloning vectors: Plasmids, Phages, Cosmids, Phagemids, Yeast vectors, Shuttle vectors, Ti Plasmids, Ri plasmids.	
3.1.3	Methods of gene transfer. Isolation and purification of cellular and plasmid DNA.	
3.1.4	Methods for labeling nucleic acids and probes. Methods of DNA sequencing. DNA Microarray	

Unit IV:		
4.1	Amplification & Applications of Genetic Engineering	
4.1.1	Amplification of DNA by PCR technique and applications.	16
4.1.2	<i>In situ</i> hybridization, analysis of DNA, RNA and protein by blotting techniques.	
4.1.3	Marker and Reporter genes. Applications of genetic engineering: Transgenic plants and animals DNA vaccines Gene therapy	

References

- [1] Singh, J.S., Singh, S.P. and Gupta, S. (2006) Ecology Environment and Resource Conservation. Anamaya Publications, New Delhi
- [2] Wilkinson, D.M. (2007). Fundamental Processes in Ecology. An Earth System Approach. Oxford.
- [3] Daubenmier, R.F. (1970). Plants and Environment: A text book of Plant Autoecology, Wiley Eastern Private Limited
- [4] Daubenmier, R.F. (1970), Plant Communities, Wiley Eastern Private Limited
- [5] Odum, E. (2008) Ecology. Oxford and IBH Publisher.
- [6] Sharma, P.D. (2010) Ecology and Environment, (8th Ed.) Rastogi Publications, Meerut.
- [7] Tom Strachan & Andrew P.Read 1999. Human Molecular Genetics (2nd Edition), John Wiley & Sons.
- [8] Ricki Lewis, 1998. Human Genetics-Concepts & Applications (3rd Edition), McGraw-Hill.
- [9] T. A. Brown, 1999. Genomes, John Wiley & Sons (Asia) PTE Ltd.
- [10] Scott Freeman & Jon C. Herron, 2001. Evolutionary Analysis (2nd Edition), Prentice Hall.
- [11] Garner E.J, Simmons, M.J. & Snustad, D.P.1991. Principles of Genetics, John Wiley & Sons Inc, N.Y
- [12] Watson, J.D., Hopkins, N. H., Roberts, J. W. Steitz & Weiner, A. M., 1987. Molecular Biology of the Genes, The Benjamin/Cummings Publishing Company Inc., Tokyo.

Course Code	Title of the Course	Credits
BCD060	PROJECT WORK OR DISSERTATION	8

COURSE OUTCOME(S):

- CO1 Identify the classification and characteristics of literature survey
- CO3 Learn in depth define of objective of project work
- CO3 Write down the classification and characteristics of design of experimental methods
- CO4 Understand the details of result analysis and interpretation

BLUE PRINT OF QUESTION PAPER FOR C1 & C2 COMPONENT

JSS Mahavidyapeetha
JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(Autonomous) B N Road, Mysuru - 25

M.Sc. Biochemistry

I/II/III/IV Semester First/Second Internal Assessment Test (Component 1/2)

Title of the Course & Code

Duration: 1hr

Max Marks: 20

A) Answer any FOUR of the following

4X2=08

- 1.
- 2.
- 3.
- 4.
- 5.

B) Answer any ONE of the following

4X1=04

- 1.
- 2.

C) Answer any ONE of the following

8X1=08

- 1.
- 2.

BLUE PRINT OF QUESTION PAPER FOR C3 COMPONENT

JSS Mahavidyapeetha
JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(Autonomous), Ooty Road, Mysuru – 570025

M.Sc Degree

I/II/III/IV Semester Examination, _____

BIOCHEMISTRY

Course Title & QP Code

Time: 3 Hours

Max. Marks: 70

Instructions to Candidates:

Answer any Five questions from Part – A

Any Four questions from Part – B

Any Three questions from Part – C

Part – A

2X5=10

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.

Part – B

6X4=24

- 8.
- 9.
- 10.
- 11.
- 12.
- 13.

Part – C

12X3=36

- 14.
- 15.
- 16.
- 17.
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**CHOICE BASED CREDIT SYSTEM AND
CONTINUOUS ASSESSMENT AND GRADING PATTERN**

SYLLABUS FOR

**MASTER'S DEGREE PROGRAM
IN
BIO-TECHNOLOGY
(2018-19 ONWARDS)**



JSS COLLEGE OF ARTS COMMERCE AND SCIENCE
(An Autonomous College of University of Mysore; Re-Accredited by NAAC with 'A' Grade)

POSTGRADUATE DEPARTMENT OF STUDIES IN BIO-TECHNOLOGY
Ooty Road, Mysore – 570 025, India

POSTGRADUATE DEPARTMENT OF STUDIES IN BIOTECHNOLOGY

Choice Based Credit System and Continuous Assessment and Grading Pattern Syllabus

M.Sc., PROGRAM IN BIOTECHNOLOGY

Scheme of Study – 2018-19 onwards

Credit matrix for Master's Degree Program in Biotechnology

Credits to be earned	I Sem	II Sem	III Sem	IV Sem	Total Credits
Hard Core Course	12	12	12	16	52
Soft Core Course	08	08	04	–	20
Open Elective Course	–	–	04	–	04
Semester Total	20	20	20	16	76

I SEMESTER			
Course title	Hard Core(HC)/ Soft Core(SC)	Credit pattern (L:T:P)	Credits
Biomolecules and Bioenergetics	HC	3:1:0	4
Bioanalytical Techniques	HC	3:1:0	4
Lab – I	HC	0:0:4	4
Choose any TWO from the following	SC	3:1:0	4
1. Molecular Genetics			
2. Microbiology	SC	3:1:0	4
3. Cancer Biology			
4. Cell Biology			
NON CREDIT COURSES			
Communication Skills			
Total credits			20
II SEMESTER			
Course title	Hard Core(HC)/ Soft Core(SC)	Credit pattern (L:T:P)	Credits
Molecular Biology	HC	3:1:0	4
Immunology and Immunotechnology	HC	3:1:0	4
Lab – II	HC	0:0:4	4
Choose any TWO from the following	SC	3:1:0	4
1. Cell Signalling and communication			
2. Metabolomics	SC	3:1:0	4
3. Food and Environmental Biotechnology			
4. Pharmaceutical Biotechnology			
NON CREDIT COURSE			
Employability Skills			
Total credits			20

III SEMESTER			
Course title	Hard Core(HC)/ Soft Core(SC)/ Open Elective(OE)	Credit pattern (L:T:P)	Credits
Bioprocess Engineering and Technology	HC	3:1:0	4
Genetic Engineering	HC	3:1:0	4
Lab – III	HC	0:0:4	4
Choose any ONE from the following 1. Biostatistics, Bioinformatics and Bioentrepreneurship 2. Clinical and Advanced Techniques in Biotechnology	SC	4:0:0	4
Applied Biotechnology * (For other discipline students)	OE	4:0:0	4
Total credits			20
IV SEMESTER			
Course title	Hard Core(HC)	Credit pattern (L:T:P)	Credits
Plant Biotechnology	HC	3:1:0	4
Animal Biotechnology	HC	3:1:0	4
Project Work/Dissertation	HC	0:4:4	8
Total credits			16
Total credits to be earned for M.Sc. Biotechnology			76

*** Open Elective Course shall be from different discipline of study**

1. A student opting I, II and III semester has to appear for at least 12 credits. (Soft core course may be studied any time).
2. Minimum number of students per Soft core course is 15.

L – Lecture – 1 credit = 1 hour

T – Tutorial – 1 credit = 2 hours

P – Practical – 1 credit = 2 hours

ASSESSMENT:

Continous Assessment: C1 – 15% & C2 – 15% (at the end of 8th and 16th week respectively)

Assessment	IA Test (20 Marks)	Assignment (5M)		Total (25 Marks)	Total reduced to 15 Marks
		a - Collection of material - 2.5 Marks	b - Preparation of report - 2.5 Marks		
C1					
C2					

Semester End Assessment: C3 – 70% – By written exam.

Conversion of grades in to credits should be based on relative evaluation calculations.

Program: M.Sc. Biotechnology

Program outcomes (PO):

PO1: To make the students develop interpersonal skills, written and oral communication and also to improve their body language and eye contact during presentations.

PO2: To train the students in group discussions to develop leadership qualities and to respect the others idea and take the decisions for the welfare of society.

PO3: To teach the students not to demoralize the others ideas and not to differentiate the intelligent and the ignorant, poor and the rich and to uphold the moral values in the society.

PO4: Upon completion of course students will have the ability to design the experiments to solve the current problems in the society related to health, environment and industries.

PO5: To make the students competent enough to write the research papers, project proposals and application of mathematics in understanding biological science.

Program Specific Outcomes (PSO):

PSO1: To make the students understand the nature, bio-molecules, their analysis and application in day to day life, so that we are transforming knowledge from nature to lab and lab to beside.

PSO2: Higher studies like M.Phil and Ph.D can be pursued to attain research positions.

PSO3: Various examinations such as CSIR-NET, ARS-NET GATE, ICMR, DBT and many other opens channels for career development.

PSO4: Students have various opportunities in different industrial sector.

PSO5: Several career opportunities are available for students with biotechnology background abroad

PSO6: In practical we teach the students to follow the standard operating procedures of the equipment, troubleshooting the problems and analyse and interpretation of data.

PSO7: To train the students regarding bio-safety in handling corrosive, explosive and radioactive and bio-hazardous compounds.

I SEMESTER
BIOMOLECULES AND BIOENERGETICS (HARD CORE) - 48 Hrs

COURSE CODE: BTA040

Course Outcome

CO1-Study of different biomolecules

CO2-Metabolism and their regulation

CO3-Enzymes and their role in metabolism

CO4- Application of thermodynamics to understand the basic concepts of life.

Unit – I

12 Hrs

Chemical basis of life; Composition of living matter; Water – properties, pH, ionization and hydrophobicity; Emergent properties of biomolecules in water; Biomolecular hierarchy; Macromolecules; Molecular assemblies; Structure-function relationships

Amino acids – structure and functional group properties; Peptides and covalent structure of proteins; Ramchandran's plot; Elucidation of primary and higher order structures; Evolution of protein structure; Structure-function relationships in model proteins like ribonuclease A, myoglobin, hemoglobin and chymotrypsin.

Unit – II

12 Hrs

Enzyme catalysis – general principles of catalysis; Quantitation of enzyme activity and efficiency; Enzyme characterization and Michaelis-Menten kinetics; activation, inhibition (reversible & irreversible) and covalent modification; Single substrate enzymes; Bisubstrate reaction (ping-pong and sequential), Applications of enzymes (food& Pharmacy).

Unit – III

12 Hrs

Sugars - mono, di, and polysaccharides; Suitability in the context of their different functions-cellular structure, energy storage, signaling; Glycosylation of other biomolecules - glycoproteins and glycolipids; Lipids - structure and properties of important members of storage and membrane lipids; lipoproteins

Unit – IV

12 Hrs

Bioenergetics-basic principles; Equilibria and concept of free energy; Coupled processes; Glycolytic pathway; Krebs's cycle; Oxidative phosphorylation; Photophosphorylation; Elucidation of metabolic pathways; Logic and integration of central metabolism; entry/ exit of various biomolecules from central pathways; Principles of metabolic regulation; Regulatory steps; Nucleosides, nucleotides, nucleic acids - structure, diversity and function

Texts/References

1. V.Voet and J.G.Voet, Biochemistry, 3rd edition, John Wiley, New York, 2004.
2. A.L. Lehninger, Principles of Biochemistry, 4th edition, W.H Freeman and Company, 2004.
3. L. Stryer, Biochemistry, 5th edition, W.H. Freeman and Company, 2002.

BIOANALYTICAL TECHNIQUES (HARD CORE) - 48 HRS

COURSE CODE: BTA050

Course Outcome

CO1-To understand the separation of molecules by different chromatography, centrifugation and electrophoretic techniques

CO2-Analysis and characterization of molecules by spectroscopy techniques

CO3-Use of radioactive material in understanding metabolic pathways

Unit- I

12 Hrs

Basic Techniques

Buffers; Methods of cell disintegration; Enzyme assays and controls; Detergents and membrane proteins; Dialysis, Ultrafiltration and other membrane techniques

Spectroscopy Techniques

UV, Visible and Raman Spectroscopy; Theory and application of Circular Dichroism; Fluorescence; MS, MALDI-TOF; NMR and Plasma Emission spectroscopy; Protein crystallization; Theory and methods; API-electrospray; Peptide Synthesis.

Imaging techniques: Compound microscope, fluorescent, phase contrast, TEM, SEM, cryo-electron microscope

Unit-II

12 Hrs

Chromatography Techniques

TLC and Paper chromatography; Chromatographic methods for macromolecule separation - Gel permeation, Ion exchange, Hydrophobic, Reverse-phase and Affinity chromatography; HPLC and FPLC; Criteria of protein purity

Electrophoretic techniques

Theory and application of Polyacrylamide and Agarose gel electrophoresis; Capillary electrophoresis; 2DElectrophoresis; Gradient electrophoresis; Pulsed field gel electrophoresis

Unit- III

12 Hrs

Centrifugation

Basic principles; (RCF, Sedimentation coefficient etc); Types of centrifuge -Microcentrifuge, High speed & Ultracentrifuges; Preparative centrifugation; Differential & density gradient centrifugation; Applications (Isolation of cell components); Analytical centrifugation; Determination of molecular weight by sedimentation velocity & sedimentation equilibrium methods

Unit- IV

12 Hrs

Radioactivity

Radioactive & stable isotopes; Pattern and rate of radioactive decay; Units of radioactivity; Measurement of radioactivity; Geiger-Muller counter; Solid & Liquid scintillation counters (Basic principle, instrumentation & technique); Brief idea of radiation dosimetry; Cerenkov radiation; Autoradiography; Measurement of stable isotopes; Falling drop method; Applications of isotopes in biochemistry; Radiotracer techniques; Distribution studies; Isotope dilution technique; Metabolic studies; Clinical application; Radioimmunoassay

Texts/References

1. Freifelder D., Physical Biochemistry, Application to Biochemistry and Molecular Biology, 2nd Edition, W.H. Freeman & Company, San Fransisco, 1982.
2. Keith Wilson and John Walker, Principles and Techniques of Practical Biochemistry, 5th Edition, Cambridge University Press, 2000.

3. D. Holme & H. Peck, Analytical Biochemistry, 3rd Edition, Longman, 1998.
4. R. Scopes, Protein Purification - Principles & Practices, 3rd Edition, Springer Verlag, 1994.
5. Selected readings from Methods in Enzymology, Academic Press.

LAB – I (HARD CORE)

COURSE CODE: BTA060

Course Outcome

CO1 - Course objective is to introduce the students to the fundamental experiments in the field of Biochemistry, Microbiology and Genetics.

CO2 - Students get the insight to operate simple equipments like colorimeter and spectrophotometer.

CO3 - Identification of microorganisms by morphology and staining techniques. Study of growth kinetics.

CO4 - In genetics students are exposed to know about culture and maintenance of *Drosophila melanogaster* (model organism), Study of mutants, salivary gland chromosome and karyotyping techniques.

Practicals/ Experiments

1. Good laboratory practices
2. Measurement of pH
3. Preparation buffers and solutions
4. Determination of pKavalues of amino acids
5. Estimation of reducing sugar by DNS method
6. Estimation of proteins by Lowry's method
7. Ascending, descending and circular paper chromatography for separation of amino acids (1D & 2D)
8. TLC of amino acids/lipids (1D & 2D)
9. HPLC
10. Estimation of ascorbic acid by DNPH method
11. Estimation of urea
12. Estimation of Phosphate
13. Gel electrophoresis- native and SDS-PAGE and determination of molecular weight of proteins
14. Salivary amylase assay, time kinetics, specific activity, determination of optimum temperature and pH; Effect chloride ions on salivary amylase activity
15. Determination of Km and Vmax. and activation energy for an acid phosphatase (from potato)
16. Effect of inhibitors on enzyme activity
17. Purification of amylase from sweet potatoes: Extraction, ammonium sulphate fractionation, gel filtration. Monitoring of enzyme activity, % activity and % recovery during purification
18. Preparation of liquid and solid media for growth of microorganisms
19. Isolation and maintenance of organisms by plating, streaking and serial dilution methods, slants and stab cultures, storage of microorganisms
20. Isolation of pure cultures from soil and water
21. Growth, growth curve; measurement of bacterial population by turbidometry and serial dilution methods. Effect of temperature, pH, carbon and nitrogen sources on growth.
22. Microscopic examination of bacteria, yeast and molds and study of organisms by gram stain, acid fast stain and staining for spores.
23. Assay of antibiotics and demonstration of antibiotic resistance.
24. Culture of *Drosophila melanogaster* and Observation of drosophila mutants
25. Isolation of salivary gland chromosomes
26. Biotech Industry and/ or R & D institution visit/s

MOLECULAR GENETICS (SOFT CORE) – 48 Hrs

COURSE CODE: BTA230

Course Outcome

CO1- To understand the molecular mechanism of inheritance

CO2-Mutation and DNA repair mechanism

CO3-Gene mapping and study of chromosomal abnormalities

CO4-Phylogenetics and micro-evolution

CO4-Development of an organism

Unit- I

12 Hrs

Laws of inheritance in haploid organisms- *Chlamydomonas* and *Neurospora*, uniparental, maternal and cytoplasmic inheritance in yeast, *Neurospora*, paramecium and plants

Genomic organization: Prokaryotes, eukaryotes, viral genome, extrachromosomal genome-plasmids, mitochondria & chloroplast, repetitive elements- LINES and SINES, simple sequence repeats

Mobile genetic elements: discovery, insertion sequence in prokaryotes, complex transposons (Tn10, Tn5, Tn9 and Tn3 as examples), mechanisms, control, consequences and applications of transposition by simple and complex elements.

Unit – II

12 Hrs

Mutation: Types, causes and detection, mutant types – lethal, conditional, biochemical, loss of function, gain of function, germinal versus somatic mutants, Molecular basis of mutations, insertional mutagenesis

Recombination: Homologous and non-homologous recombination, Holliday model, site-specific recombination

DNA Repair: Mechanism of genetic repair- direct repair, photo reactivation, excision repair, mismatch repair, post-replicative recombination repair, SOS repair

Unit-III

12 Hrs

Gene mapping methods: Linkage maps, tetrad analysis, mapping with molecular markers, mapping by using somatic cell hybrids, development of mapping population in plants.

Microbial genetics: Methods of genetic transfers – transformation, conjugation, transduction and sex-duction, mapping genes by interrupted mating, fine structure analysis of genes.

Quantitative genetics: Polygenic inheritance, heritability and its measurements, QTL mapping.

Unit-IV

12 Hrs

Genes and development: Model systems for studying development- *Drosophila*, *Caenorhabditis*, *Arabidopsis*. Genetic control of development in *Drosophila*: anterioposterior axis specification, role of maternal genes, segmentation of larval body, gap genes, pair rule genes, homeotic genes, complex gene interaction in development, sequential gene action. Floral meristems and floral development in *Arabidopsis*.

Human Genetics: Human chromosomes, karyotype – construction, characteristics, staining techniques and nomenclature; chromosomal abnormalities – sex chromosomal and autosomal, inherited disorders, genetic counselling, gene therapy; Human Genome Project, Human Genome Map.

Population Genetics: Genetic variation, Hardy-Weinberg Law, random mating, genetic frequency, natural selection, genetic drift, migration, genetic equilibrium.

Evolution: Molecular basis of evolution, Molecular clock, Molecular phylogenetics

Reference Books

1. Genetics. Strickberger, M. W., Prentice Hall of India Pvt. Ltd.
2. Genetics – A Molecular Approach. Brown, T. A. Chapman and Hall.
3. Genes VII. Lewin, B. Pearson Education International. 2003.
4. Genetics- A Conceptual Approach. Benjamin A Pierce.

CANCER BIOLOGY (SOFT CORE) – 48 Hrs

COURSE CODE:

Course Outcome

CO1-Understanding the normal and cancerous cell

CO2-Protooncogenes, tumor suppressor genes and apoptotic genes – regulation

CO3-Diagnosis and treatment of cancer

Unit-I

Cancer Biology:

12 Hrs

Introduction, historical perspective, classification, Carcinogenesis, cancer initiation, promotion and progression, Cancer cell cycles, Genomic instability, Apoptosis, Genes and proteins as players in apoptosis, DNA viruses/ cell immortalization.

Unit-II

12 Hrs

Cancer Genes I: Oncogenes and signal transduction

Cellular proto-oncogenes, oncogene activation, Growth factors, growth factor receptors, signal transduction, Transcription, Transcription factors and cancer, Retroviral oncogenes, Tumor suppressor, Tumor suppressor gene pathways, DNA methylation, epigenetic silencing of suppressor genes.

Unit-III

12 Hrs

Understanding Cancer as a Disease: natural history of cancer development

Free radicals, antioxidants and metabolic oxidative stress and cancer, Epidemiology of selected cancers, Gene rearrangements, detecting oncogene abnormalities in clinical specimens, Cell: cell interactions, cell adhesion, angiogenesis, invasion and metastasis, Antiangiogenic therapy of cancer.

Unit-IV

12 Hrs

Current concepts in cancer therapy

Strategies of anticancer chemotherapy, Strategies of anticancer gene therapy/translating therapies from the laboratory to the clinic, Gene discovery in cancer research, cancer genome anatomy project, Cancer immunity and strategies of anticancer immunotherapy, stem cells and their applications in cancer therapy.

Reference Books

1. Molecular Biology of the Cell. Bruce Alberts

MICROBIOLOGY (SOFT CORE) – 48 Hrs

COURSE CODE: BTA240

Course Outcome

CO1- To understand the microbial taxonomy

CO2-Handling, preservation and sterilization of microbes

CO3-Microbial interactions with different hosts

CO4-Application of microorganisms in the field of agriculture, environment and health sciences

Unit-I

12 Hrs

Microbial Diversity & Systematics

The beginning of microbiology: The discovery of the microbial world – Hook, Anton van Leeuwenhoek and Cohn; Contribution of Pasteur and Koch. Development of pure culture methods; the enrichment culture methods. Methods in Microbiology: Pure culture techniques; the theory and practice of sterilization.

Classical and modern methods and concepts; Domain and Kingdom concepts in classification of microorganisms; Criteria for classification; Classification of Bacteria according to Bergey's manual; Molecular methods such as Denaturing Gradient Gel Electrophoresis (DGGE), Temperature Gradient Gel Electrophoresis (TGGE), Amplified rDNA Restriction Analysis and Terminal Restriction Fragment Length Polymorphism (T-RFLP) in assessing microbial diversity; 16S rDNA sequencing and Ribosomal Database Project.

Unit-II

12 Hrs

Microbial Growth & Physiology

Ultrastructure of Archaea (Methanococcus); Eubacteria (*E.coli*); Unicellular Eukaryotes (Yeast) and viruses (Bacterial, Plant, Animal and Tumor viruses); Microbial growth: Batch, fed-batch, continuous kinetics, synchronous growth, yield constants, methods of growth estimation, stringent response, death of a bacterial cell. Factors affecting growth like temperature, acidity, alkalinity, water availability and oxygen. Microbial physiology: Physiological adaptation and life style of Prokaryotes; Unicellular Eukaryotes and the Extremophiles (with classical example from each group)

Unit-III

12 Hrs

Microbial Interactions and Infection

Host-Pathogen interactions; Microbes infecting humans, veterinary animals and plants; Pathogenicity islands and their role in bacterial virulence. Chemotherapy/antibiotics: Types, mode of action, resistance to antibiotics.

Unit-IV

12 Hrs

Microbes and Environment

Role of microorganisms in natural system and artificial system; Influence of Microbes on the Earth's Environment and Inhabitants; Ecological impacts of microbes; Symbiosis (Nitrogen fixation and ruminant symbiosis); Microbes and Nutrient cycles; Microbial communication system; Quorum sensing; Microbial fuel cells; Prebiotics and Probiotics.

Texts/References

1. Pelczar MJ Jr., Chan ECS and Kreig NR., Microbiology, 5th Edition, Tata McGraw Hill, 1993.
2. Maloy SR, Cronan JE Jr., and Freifelder D, Microbial Genetics, Jones Bartlett Publishers, Sudbury, Massachusetts, 2006.

3. Crueger and A Crueger, (English Ed., TDW Brock); Biotechnology: A textbook of Industrial Microbiology, Sinaeur Associates, 1990.
4. G Reed, Prescott and Dunn's, Industrial Microbiology, 4th Edition, CBS Publishers, 1987.
5. M.T. Madigan and J.M. Martinko, Biology of Microorganisms, 11th Edition, Pearson Prentice Hall, USA, 2006.

CELL BIOLOGY (SOFT CORE) – 48 Hrs

COURSE CODE:

Course Outcome

CO1-Understanding the structure and function of bacterial, plant and animal cell

CO2-Cell signalling and communication

CO3-Study of growth factors and their function

CO4-Tumor biology of a cell

Unit-I

12 Hrs

Membrane and membrane phenomenon: Membrane structure and principles of organization, Membrane proteins, glycoproteins and glycolipids, specialization of plasma membrane, transport across cell membrane – types of transport, ion channels, active transport and ion pumps, symport, antiport, plant and prokaryotic membrane transport proteins. Cell organelle and membrane proteins: Mechanism and regulation of vesicular transport, Golgi and post golgi storing, receptor mediated endocytosis.

Unit-II

12 Hrs

Microfilament, cell motility and cell shape: actin, actin architecture and assembly, myosin, muscle contraction, microtubules structure and dynamics, microtubule associated protein, cilia, flagella, intermediate filaments.

Multicellularity: Extracellular matrix, hyaluronan and proteoglycan, matrix proteins and their receptors, adhesive proteins, cell junctions, structure and function of plant cell wall.

Cellular signaling: Extra cellular signaling, G-protein linked receptors, role of cAMP, receptor tyrosine kinases, Ca^{2+} as a second messenger, multiplex signaling pathways, insulin receptor and regulation of blood glucose, regulation of cell surface receptors and transcription factors in signaling pathways, Chemical messenger – peptide and steroid hormones, mechanism of hormone action.

Unit-III

12 Hrs

Growth factor: Growth factor structure (PDGF, VEGF), mechanism of action (PDGF, VEGF), receptors, signal transduction, plant growth factors and hormones – auxins, cytokinins and other

Cell Cycle: General strategy of cell cycle, discrete cell cycle events, cell cycle control, early embryonic cell cycle, yeast cell cycle, molecular genetics of cell cycle control, cyclins, cyclin dependent kinase, inhibitors, cell division control in multicellular organism, apoptosis.

Unit-IV

12 Hrs

Tumor biology: Retroviruses, retro viral transformation of host, development and causes of cancer, proto-oncogene, conversion from proto-oncogene to oncogene, tumor suppressor gene, role of p53 in cancer, cell culture uses in research, molecular medicine and cancer.

Nerve cells: Action potential, voltage gated ion channels, nicotinic acetylcholine receptor, other neurotransmitters and their transporters, sensory transduction – the visual and olfactory system.

Reference Books

1. Molecular Biology of the Cell. Alberts, B., *et al.*, 4th Edition. Garland Publ. Inc.
2. Molecular Cell Biology. 5th Edn. Lodish, H., *et al.*, W H Freeman.
3. Genes VII. Lewin, B. Pearson Education International.
4. Cell and Molecular Biology. Karp, J. John Wiley and Sons Inc.

NON CREDIT COURSE

COURSE CODE:

Course Outcome

CO1-Interpersonal skills (body language, eye contact)

CO2-Presentation skills

CO3-Writing emails, research papers and proposals and business reports

Communication Skills Module

Business Etiquette – Video Conferencing (VC): Introduction to Video Conferencing; Concept & uses of VC; VC Etiquette

Business Communication: Seven Cs of communication: Complete, Courteous, Considerate, Clear, Concise, Concrete, Correct; Verbal/ Nonverbal Communication

Writing Process: Identifying objective; Categorizing Information; Organizational Patterns; Designing document; Memo writing; Revision checklist; Releasing document

Business Proposal & Report Writing: Types of Proposals; Top-Down & Bottom-Up Approach; Study of Technical Bid & Cost Bid; Transmittal Letters; Formal Reports (Short and Long); Types of graphics & illustrations; Business Report Templates; Study of Sample Proposals;

Project Report Writing: Project Charter; Project Plan; Gant Chart; Activities List; Resources List; Risks List, Project Status Report; Project Closure Report; Types of graphics & illustrations; Study of Project Report Templates

Email Writing: Problems resulting out of emails; Contents of email, Importance of a good subject line; Dos and Don'ts; Using your email software to its maximum; Setting up signatures; Setting up accounts; Creating HTML stationary; Creating email templates for common emails; Using short mails for internal communication; Importance of acknowledging emails; Creating folder structure for easily accessing emails; Care to be taken while deleting emails, Archiving emails; Comparison of emails and letters; Writing typical emails, sending point-wise reply to emails

Cross-Cultural Training: Cross-Cultural Sensitivity; American, European, Australian, Middle-east and South-East Asian countries culture training; Cultural Foundations; Cross Cultural Communication; Communication Styles; Comparative Values (American – Indian); Regional Dialects; Cross-Cultural Customer Attitude & Expectations

Interpersonal Skills: Introduction & Importance of Interpersonal Skills; Cost of Poor interpersonal skills; Standing up for self assertiveness; Strategies to achieve self-assertiveness; Managing conflicts, disputes; Dealing with Diversity Interpersonal Relationship and influence

II SEMESTER

MOLECULAR BIOLOGY (HARD CORE) – 48 Hrs

COURSE CODE: BTB020

Course Outcome

CO1- The student will get an idea about the genomic organization of prokaryotes and eukaryotes.

CO2- Obtain in depth knowledge of genetic code, DNA replication and transcription.

CO3- Understand principles, concepts of translation, post translation mechanism

CO4- Regulation of gene expression in prokaryotes and eukaryotes

CO5- Gain the insight into molecular mechanism of antisense molecules, inhibition of splicing and application of antisense and ribozyme technologies.

Unit-I

12 Hrs

Genome organization: Organization of bacterial genome; Structure of eukaryotic chromosomes; Role of nuclear matrix in chromosome organization and function; Matrix binding proteins; Heterochromatin and Euchromatin; DNA reassociation kinetics (Cot curve analysis); Repetitive and unique sequences; Satellite DNA; DNA melting and buoyant density; Nucleosome phasing; DNase I hypersensitive regions

DNA topology: Closed and super coiled DNA, DNA topoisomerases,

DNA replication: Enzymes in DNA replication, DNA Pol I, II III, replication in single stranded DNA viruses, replication in prokaryotes, eukaryotic DNA replication, eukaryotic polymerases, role of other proteins and enzymes in replication, fidelity of replication, replication of mitochondrial DNA, inhibitors of replication.

Unit- II

12 Hrs

Genetic code: Elucidation, Contributions of Khorana and others, triple binding assay, Wobble hypothesis.

Transcription: Transcription unit, RNA polymerase in prokaryotes, mechanism of transcription- initiation, elongation and termination. Eukaryotic transcription - eukaryotic RNA polymerase, transcription factors, initiation, elongation and termination of transcription, inhibitors of transcription; post transcriptional modifications – capping, polyadenylation, splicing, introns and exons. Structural organization of mRNA, tRNA and rRNA, nuclear export of mRNA and mRNA stability

Unit-III

12 Hrs

Translation: Molecular anatomy and biogenesis of ribosome, partial reconstitution experiments; Amino acid activation- amino acylation of tRNA; prokaryotic and eukaryotic translation– mechanism of initiation, elongation and termination, inhibitors of translation, post translational modifications, protein glycosylation.

Protein localization: Synthesis of secretory proteins and membrane proteins; import into nucleus, mitochondria, chloroplast and peroxisomes.

Regulation of gene expression in Prokaryotes: Basic control circuits, positive and negative regulation; Operon concept – *lac*, *ara* and *trp* operons- catabolite repression, regulatory elements in prokaryotes, attenuation, antitermination, regulation of gene expression in Bacteriophage.

Unit-IV

12 Hrs

Regulation of gene expression in Eukaryotes: *cis* control elements – promoters, enhancers, *trans* acting factors, DNA binding motifs of transcription factors, mechanism of regulation by transcription factors, NFkB histone acetyl transferase and deacylase, hormonal regulation of gene expression, post transcriptional control.

Antisense RNA and ribozymes: Molecular mechanism of antisense molecules, inhibition of splicing, disruption of RNA structure, hammerhead, hairpin ribozymes, Application of antisense and ribozyme technologies. RNA interference, RNA induced gene silencing.

Reference Books

1. Molecular Biology. Freifelder, D. Narosa Pub House.
2. Advance Molecular Biology. Twyman, R. M. Viva Book Pvt. Ltd.
3. Molecular Biology. JD Watson
4. Molecular Biology of the Cell. Bruce Alberts.
5. Genes, Benjamin XII ,2017

IMMUNOLOGY AND IMMUNOTECHNOLOGY (HARD CORE) – 48 Hrs

COURSE CODE: BTB050

Course Outcome

- CO1- Study basic concepts of immunology
- CO2- MHC and their role in transplantation
- CO3-Cytokines and their role in immune system
- CO4-Tumor immunology
- CO5-Autoimmune diseases
- CO6-Hypersensitivity
- CO7-Vaccine production.

Unit-I

12 Hrs

Immune system: Structure, functions and organization of cells and organs involved in immune systems – T cells, B-cells, macrophages, Eosinophils, Neutrophils, Mast cells; bone marrow, spleen, thymus, lymph node, peyer's patch; Infections and immune responses – Innate immunity, acquired immunity; clonal nature of immune response; Immunohaematology – blood groups antigens, blood transfusion and Rh incompatibilities.

Antigens: Types, haptens, adjuvants, antigenic specificity.

Antibodies: Structure of immunoglobulins, heterogeneity, sub-types – iso-, allo- and idio- types and their properties

Unit-II

12 Hrs

Complements: Structure, components, properties and functions of complement pathways, biological consequences of complement activation; Immunological diversity;

Effector mechanism: T-cell cloning, mechanism of antigen recognition by T-cells and B-lymphocytes and their properties, receptors and related diseases.

Role of class II MHC molecules in T-cell cloning, antigen specific and alloreactive T-cell cloning, applications of T-cell cloning in understanding relevant antigens and T-cell subtypes; T-cell cloning in vaccine development

MHC and Tumor immunology: Structure and function of MHC and the HLA system; regulation of Ir-genes; Tumor immunology– Tumor specific antigens, Immune response to tumors, theory of surveillance, immune diagnosis of tumor; Tumor markers – Alpha fetofetal proteins, carcinoembryonic antigen

Unit-III

12 Hrs

Immune responses and Transplantation: HLA and tissue transplantation; Tissue typing methods for organ and tissue transplantation in humans; Graft versus host rejection, Host versus graft rejection; Xenotransplantation; Immunosuppression theory; Autoimmune diseases – Hashimoto's disease, Systemic lupus erythematosus, Multiple sclerosis, Myasthenia gravis, Rheumatoid arthritis and the remedies.

Allergy: Type I – Antibody mediated – Anaphylaxis, Type II – antibody dependent – Cytolytic and Cytotoxic, Type III – Immune complex mediated reactions– Arthus reaction, serum sickness, Type IV– Cell mediated hypersensitivity reaction– Tuberculin type.

Unit-IV

12 Hrs

Lymphokines and Cytokines– assay methods, related diseases; Immunological tolerance; production of interleukins and interferons– applications.

Immunizations: Conventional vaccines, sub-unit vaccines, DNA vaccines, toxoids, antisera; common immunization – small pox, DPT, hepatitis, polio, measles

Reference Books

1. Immunology. Roitt, Gower Medical Publisher.
2. Fundamental Immunology. Paul W E Raven Press.
3. Immunology. Kuby
4. Immunology, JanewasTraves, Walpart, SHlomehik. Churchill Livingstone.
5. An introduction to Immunology. Rao, C. V. Nasora pub house.
6. Immunology – A short course. Coico, R., Sunshine, G. and Benjamini, E. John Wiley and sons.
7. Cellular Interactions and Immunobiology. BIOTOL series. Butterworth-Heinemann.

LAB – II (HARD CORE)

COURSE CODE: BTB060

Course Outcome

CO1-Students are trained to get the skills in the field of Molecular biology and Genetic engineering

CO2- , Isolation and purification of nucleic acids and their quantification

CO3-Study of antigen and antibody interactions.

CO4 -Preparation of wine and analysis of food samples

Practicals/ Experiments

1. Identification of normal and abnormal human karyotype
2. Localization of Barr bodies
3. Estimation of free fatty acids by titrametric method
4. Saponification value for commercial oil samples
5. Determination of iodine value of an oil
6. Determination of total carbohydrates by phenol-sulphuric acid method
7. Estimation of cholesterol
8. *In vitro* transcription
9. Total RNA extraction
10. Estimation of DNA by Diphenylamine (DPA) method
11. Estimation of RNA by orcinol method
12. Isolation of DNA different samples: plant leaves, coconut endosperm, yeast, animal tissues
13. Determination of purity and concentration of isolated DNA using spectrophotometer
14. Agarose gel electrophoresis of DNA
15. Analysis of microbial quality of foods – Litmus test, catalase test and dye reductase test in milk, estimation of lactic acid in milk
16. Preparation of wine
17. Estimation of percentage of alcohol in wine
18. Chemical method to differentiate between ethanol from methanol
19. Estimation of total acids in wine
20. Conjugation
21. Phage titration
22. Preparation of antigen and antibody production
23. Purification of IgG/IgY
24. Slide agglutination test/blood grouping
25. Antibody labeling
26. Immunoprecipitation test- ODD
27. ELISA for quantification of an antigen
28. Lymphocyte preparation
29. Rossette assay
30. Rocket immunoelectrophoresis

Biotech Industry and/ or R & D institution visit

CELL SIGNALLING AND COMMUNICATION (SOFT CORE) – 48 Hrs

COURSE CODE: BTB220

Course Outcome

CO1- Understanding the multi-cellularity of organisms

CO2-role of extracellular matrix in signalling

CO3-various signalling pathways from the cell surface to the nucleus

CO4-cell signalling in plants

CO5-microbe-plant and insect-plant interaction.

Unit-I

12 Hrs

Multicellularity: Role of Extracellular matrix - hyaluronan and proteoglycan. Matrix proteins and their receptors, adhesive proteins and cell junctions in multicellularity. Structure and function of plant cell wall

The importance of the matrix in signal transduction: Cell surface receptors as reception of extracellular signals, Amplification of signal during transmission - a quantitative study, Tyrosine kinase and tyrosine phosphatase, Cell membrane components and adapter proteins required for signal transmission, Upstream and downstream signal transduction without cell surface receptor activation, G-protein coupled signaling; the secondary messengers in signal transduction pathways cAMP, Ca²⁺, Reactive Oxygen Species and Hypoxia Signalling, Apoptosis Signaling Transduction Pathway, PI3K/AKT Cell Survival Pathway.

Unit-II

12 Hrs

Various signal transduction pathways from cell surface to nucleus: MAP kinase pathway, SAP/JNK pathway, p38 pathway, ERK pathway, NFκB pathway, Cell survival pathway, Wnt signaling pathway, Jak/Stat pathway, Smad pathway, TGF β Signaling, EGFR, VEGF And their Signalling, Cytoskeleton And Cell Signalling, Carbohydrate Recognition Signaling, MMPs And Cell Signalling, Cross talk among cell surface receptors, Cross talks among cytoplasmic components, Translocation of signal components during signal transmission, From cytoplasm to cell membrane, NF-κB Signaling from cytoplasm to nucleus, Cell cycle and its Signalling.

The end point of signal transduction--- gene transcription: Nuclear receptors and transcription factors in signalling, Signalling from single gene expression to multiple gene expression: Super array as a tool for the study of multiple gene transcription, Practical application of the signal transduction research, RNA Interference And Cell Signalling, Senescence and Its Signaling Pathways.

Unit-III

12 Hrs

Signal transduction in plants: Cross-talk with the environment- wound and mechanical signalling - fatty acid signalling, peptide signalling, oligosaccharide signalling; protein kinases and signal transduction. Abiotic stresses - Dehydration-stress, salt-stress, cold acclimation, heat-stress

Role of active oxygen species (AOS) in plant signal transduction: AOS in plants, AOS as signal molecules, AOS-part of a signalling network.

Action of phytohormones: Multiple signals regulating growth and development of plant organs and their adaption to environmental stresses.

Unit-IV

12 Hrs

Symbiotic plant-microbe interaction: Rhizospheric signals (PGPR) and early molecular events in the ectomycorrhizal symbiosis; Lipo-chito-oligosaccharides (LCO) signalling in the interaction between rhizobia and legumes; endophytes.

Recognition and defencesignalling in plant-microbe interaction: Resistance genes - gene-for-gene resistance; co-evolution and specificity of R genes; the TIR domain, the NBS domain; genetic organization of resistance genes; quorum sensing.

Plant-insect interaction: Induction of direct and indirect defence

Reference Books

1. Animal Cell Biotechnology – Methods and Protocols. Nigel Tenkins.
2. Molecular biology of the Cell –Alberts et al.
3. Molecular Cell Biology. 5th Edn. Lodish, H, et al., W H Freeman
4. Cell and Molecular Biology. Karp, J.JohnWiley and Sons In.
5. The Cell-Molecular approach. 4th Ed. Geoffrey M Cooper and Robert E Hausman.
6. Cell Biology- A Laboratory Handbook. 3rd Ed, 4th Vol, Julio E Celis

METABOLOMICS (SOFT CORE) – 48 Hrs

COURSE CODE:

Course Outcome

CO1-Understanding the basic metabolism of plants

CO2-Different pathways involved in secondary metabolite production

CO3-Altering the metabolic pathways by changing the precursors

CO4-Purification of useful secondary metabolites and their kinetics and dynamics

CO5-Applications in food and pharmaceutical industries

Unit-I

12 Hrs

Plant Metabolomics: Developments and history of plant metabolomics, Nature and prospecting of metabolism-related secondary plant products, tools and techniques, production in culture: optimization; selection, hormonal kinetics for secondary metabolites, production, mechanism and control.

Unit-II

12 Hrs

Production of secondary metabolites: Induction, Alkaloids, antitumor compounds, food additives, steroids and saponins, detoxification of secondary metabolites, production of secondary metabolites by bioconversion, genetic transformation for production of secondary metabolites, large-scale production in bioreactors, Metabolomics-assisted breeding.

Unit-III

12 Hrs

Microbial metabolomics: Systems biology of microbial metabolism; microbe sensors, *In silico* metabolomes, Food and Applied metabolomics, Biomarker discovery. Experimental Approaches- Genome sequencing, Gene expression arrays, Nuclear Magnetic Resonance, Mass spectroscopy, Capillary electrophoresis, Two dimensional gel electrophoresis, Gene expression arrays, Pathway analysis, HPLC, Protein sequencing, Bench-scale fermentation, AFLP/RLFP analysis.

Unit-IV

12 Hrs

Pharmacometabolomics: personalized medicine and future of health system, Pathways discovery and disease pathophysiology, Bioinformatics analysis of targeted metabolomics; Environmental metabolomics, Bioactive compounds and Pharmacognosy, Clinical Applications of Metabolomics, Nutrigenomics and Metabolomics, Novel Technologies for Metabolomics, Data Handling for Metabolomics.

Reference Books

1. V.Voet and J.G.Voet, Biochemistry, 3rd edition, John Wiley, New York, 2004.
2. A.L. Lehninger, Principles of Biochemistry, 4th edition, W.H Freeman and Company, 2004.

FOOD AND ENVIRONMENTAL BIOTECHNOLOGY (SOFT CORE) – 48 Hrs

COURSE CODE: BTB210

Course Outcome

CO1-Comprehensive insight into the fermented foods and enzymes in food industry

CO2-Obtain knowledge of functional foods, genetically modified foods and nutraceuticals.

CO3-Students will be able to understand current status of biotechnology in environment protection.

CO4-Understand the principles of bioremediation and significance of GMO to the environment.

Unit-I

12 Hrs

Fermented foods, milk-based products, fermented vegetables, fermented meats, fish, beverages, vinegar, mould fermentation - tempeh, soysauce, rice wine.

Enzymes in dairy industry, cheese making and whey processing, impact of enzyme technology (protein hydrolysates, bioactive peptides), Enzymatic processing of fruit juices; role of enzymes in baking, meat and meat processing, phytase in animal feeds, DNA-based methods for food authentication, comparative methods of toxicity testing in (novel) foods, biological approach to tailor-made foods, application of generic technologies in food and nutritional sciences; anti-cancer components in foods.

Unit-II

12 Hrs

Functional foods and Biotechnology: applying molecular, biochemical, cellular and bioprocessing concepts, use of specific phenolic metabolites from botanical species. Pre- and Pro-biotics, single cell protein, single cell lipids. Manipulation of fruit ripening process.

Food processing, principles and practices, food ingredients and processing aids from biotechnological processes, corn sweeteners, bacterial starter cultures, Food spoilage, preservation, mycotoxins in food commodities. Genetically modified foods, designer foods, Nutraceuticals, detection of GM foods.

Unit-III

12 Hrs

Renewable and non-renewable resources, current status of biotechnology in environment protection. Characterization of waste. Waste water management: Bioreactors for waste-water treatment, Aerobic biological treatments, anaerobic biological treatments, treatment of industrial effluents-dairy, distillery, paper and sugar industries. Membrane-based waste water treatment.

Oil pollution – treatment with microorganisms.

Unit-IV

12 Hrs

Bioremediation: Concepts and principles, bioremediation using microbes, in situ and ex situ bioremediation, biosorption and bioaccumulation of heavy metals.

Xenobiotics: Degradation capabilities of microorganisms with reference to toxicology, pesticides, herbicides, polyaromatic hydrocarbons.

Renewable energy: Relevance of GMO to the environment.

Solid waste management: Waste as a source of energy, biotechnology in paper and pulp industry, production of oil and fuels from wood waste, anaerobic and aerobic composting, vermiculture, biofuels.

Reference Books

1. Food Microbiology. Frazier, W. C. and Westhoff, D. C. Tata McGraw Hill.
2. Agriculture Bio-technology. Purohit. Agrobios India.
3. Food Bio-technology. Knorr, D. Marcel Dekker Inc.
4. Environmental Bio-technology. Jogand, S. N. Himalaya Publishing House, New Delhi.

PHARMACEUTICAL BIOTECHNOLOGY (SOFT CORE) – 48 Hrs

COURSE CODE:

Course Outcome

CO1-Rules and regulation regarding development of drugs

CO2- Study of Pharmacodynamics and pharmacokinetics of drugs

CO3-Different phases of clinical trials and drug toxicity studies.

CO4-GMP and GLP in production management and quality control and assessment

Unit- 1

12 Hrs

Introduction to pharmaceuticals and Drug development process: Introduction to pharma industry, history of the pharmaceutical industry, traditional pharmaceuticals of biological origin (animal, plant and microbial) biopharmaceuticals and pharmaceutical biotechnology, age of biopharmaceuticals, biopharmaceuticals: current status and future prospects.

Steps involved in drug development process, drug delivery systems, preclinical studies and principles practices, phases of clinical trials. Regulatory authorities in India, USA and Europe and Japan, prescription, non- prescription drugs and orphan drugs.-

Unit-II

12 Hrs

Drug Receptors, Pharmacodynamics and pharmacokinetics.: Different types of drug receptors, second messengers (cAMP, Ca²⁺ and phosphoinositides) and their signalling mechanism, relation between drug concentration and response, concentration effect curves, concentration- effect curves, relation between drug dose and clinical responses. Volume of distribution of drug, clearance, drug accumulation, bioavailability, alternative routes of administration and the first pass effect, therapeutic drug monitoring

Unit-III

12 Hrs

Drug biotransformation and drug toxicity: The role of biotransformation in drug disposition, phase I metabolism (microsomal oxidation, hydroxylation, dealkylation), phase II metabolism (Drug conjugation pathway) CYP families, clinical relevance of drug metabolism, drug-drug interaction. Mechanisms of toxicity, production of toxic metabolites, harmful immune response, idiosyncratic toxicity, contexts of drug toxicity, drug overdose, drug- drug interactions, pathology of drug toxicity. Cellular toxicity, organ and tissue toxicity.

Unit-IV

12 Hrs

The drug manufacturing process and drugs of biopharmaceutical origin: Guides to good manufacturing practice, manufacturing facility. Clean rooms, cleaning, decontamination and sanitations (CDS), CDS of the general manufacturing area, CDS of the Process equipment, generation of purified water, water for injection, documentation, specifications, Concept and testing of pre- formulations & their parameters. Tablets: Compressed, granulation, coatings, pills and capsules, parenteral preparations, herbal extracts, oral liquids, Ointments. Processing and packing instructions.

Therapeutic enzymes: asparaginase, DNase, Glucocerebrosidase, galactosidase and urate oxidase, superoxide dismutase, Lactase.

Reference Books

1. Textbook of Pharmaceutical Biotechnology. Chandrakant Kokate, Pramod H.J, SS Jalalpure. Elsevier Health Sciences, 2012
2. Pharmaceutical Biotechnology: Concepts and Applications. Gary Walsh. John Wiley & Sons, 2013
3. Pharmaceutical Biotechnology, Second Edition. Michael J. Groves. Taylor & Francis, 2005

NON CREDIT COURSE

EMPLOYABILITY SKILLS MODULE

COURSE CODE:

Course Outcome

CO1- Concepts of corporate communication

CO2- English grammar skills

CO3- Develop strategies for negotiation and marketing

CO4- Personality development and interview skills

CO5- This course will enable students to learn about the project management, entrepreneurship.

Campus to Corporate: Transition from College to Corporate world; Perceptions v/s Real Corporate life; Working in Teams; Basics of corporate communication

Corporate & Office Etiquette: Elements of a good handshake; Visiting cards exchange & How to manage business cards; Small Talk & Networking; Basics dining etiquette

English Grammar: A quick round up: Nouns, Pronouns, Adjectives, Verbs, Adverbs, Tenses, Prepositions, Clauses, Subject and Predicate, Punctuations, Subject- verb agreement, Confusing prepositions, Missing Articles, Editing paragraphs

Negotiation Skills: Introduction to Bargaining and Negotiation; The Negotiation Process: Four Stages; An Analytical framework of Negotiation; Bargaining Approaches; Strategy for Value Added Negotiation

Selection & Interviewing Skills: Current market for talent & methods for attracting & sourcing; Best practices for different hiring situations - Campus, Market, Head hunter agencies; Selection process design & assessment centers; Effective interview

Personality Development: Self assessment: SWOT; Understanding Personality - Identifying different personalities; Levels of Human Learning; Change v/s Transformation; Sensitivity - Sharpen your senses; Creativity and Lateral thinking; Developing Positive Mental Attitude; Emotional Quotient; Handling Criticism; Positive Health; Food habits and Meditation; Goal setting - Creative Visualization - Law of Attraction; Living a created life - Personal Leadership

III Semester

BIOPROCESS ENGINEERING AND TECHNOLOGY (HARD CORE) – 48 Hrs

COURSE CODE:

Course Outcome

CO1-To have the comprehensive insight into the different type of fermenter

CO2-To obtain knowledge of media design and industrial culture

CO3-Students will be able to understand different type of fermenter and bioreactor.

CO4-Understand the principles of downstream processing

CO5- To understand the enzyme technology and their applications in industry.

Unit-I

12 Hrs

Basic principle of Biochemical engineering and Microbial Growth Kinetics:

General Introduction to metabolic pathways involved in microbial products, concepts of over production, primary and secondary metabolites, estimation of biomass. Isolation, screening and maintenance of industrially important microbes; Microbial growth kinetics, Strain improvement for increased yield and other desirable characteristics.

Batch culture, continuous culture, fed batch culture, the growth cycle, effect of nutrients, growth rate and cell cycle.

Unit II

10 Hrs

Media design and industrial cultures: Introduction, typical media, Oxygen requirement, antifoams, media formulation, energy sources, carbon and nitrogen source, other components, media optimization, Media sterilization, Batch process (thermal death kinetics), continuous sterilization process. Sterilization of fermenter and other ancillaries, filter sterilization of air and media. Rheological properties of medium. Screening for industrial useful metabolites, maintenance of stock cultures

Unit III

10 Hrs

Types of fermenters and bioreactors: design, control system, operation, optimization, control and monitoring of variables such as temperature, agitation, pressure, pH, online measurements and control, Scale up of bioreactors. Bubble column, airlift reactor, packed bed, fluidized bed, trickle bed, Membrane reactor, Photobioreactor, Solid state fermenter, Animal and plant cell bioreactors. Scale up and Scale down studies of bioreactors. Biosensor

Unit IV

16 Hrs

Downstream processing (Recovery and purification of products) of biologicals: Separation of cells, foam separation, disintegration of micro organism, mechanical and non mechanical methods, flocculation, filtration, plate filters, rotary vacuum filters, centrifugation, Stoke's law, continuous centrifugation, basket centrifuge, bowl centrifuge, membrane filtration, ultra filtration and reverse osmosis, chromatographic techniques, absorption, spray drying, drum drying, freeze drying.

Enzyme Technology: production, recovery, stability and formulation of bacterial and fungal enzymes-amylase, protease, penicillin acylase, glucose isomerase; Immobilised Enzyme and Cell based biotransformation steroids, antibiotics, alkaloids.

Texts/ References

1. Jackson AT., Bioprocess Engineering in Biotechnology, Prentice Hall, Engelwood Cliffs, 1991.
2. Shuler ML and Kargi F., Bioprocess Engineering: Basic concepts, 2nd Edition, Prentice Hall, Engelwood, Cliffs, 2002.

3. Stanbury RF and Whitaker A., Principles of Fermentation Technology, Pergamon press, Oxford, 1997.
4. Baily JE and Ollis DF., Biochemical Engineering fundamentals, 2nd Edition, McGraw-Hill Book Co., New York, 1986.
5. Aiba S, Humphrey AE and Millis NF, Biochemical Engineering, 2nd Edition, University of Tokyo press, Tokyo, 1973.
6. Comprehensive Biotechnology: The Principles, Applications and Regulations of Biotechnology in Industry, Agriculture and Medicine, Vol 1, 2, 3 and 4. Young M.M., Reed Elsevier India Private Ltd, India, 2004.
7. Mansi EMTEL, Bryle CFA. Fermentation Microbiology and Biotechnology, 2nd Edition, Taylor & Francis Ltd, UK, 2007.

GENETIC ENGINEERING (HARD CORE) - 48 Hrs

COURSE CODE:

Course Outcome

CO1-To have the comprehensive insight into the different enzymes used in Genetic engineering lab

CO2-To obtain knowledge of construction of vectors

CO3-Students will be able to understand different type of cloning methods.

CO4-Understand the principles of PCR & types

CO5- To know the different sequence methods

Unit I

10 Hrs

Basics Concepts: DNA Structure and properties; Restriction Enzymes; DNA ligase, Klenow enzyme, T4 DNA polymerase, Polynucleotide kinase, Alkaline phosphatase; CRISPR- cas9, Cohesive and blunt end ligation; Linkers; Adaptors; Homopolymeric tailing; Labeling of DNA: Nick translation, Random priming, Radioactive and non-radioactive probes, Hybridization techniques: Northern, Southern and Colony hybridization, Fluorescence in situ hybridization; Chromatin Immunoprecipitation; DNA-Protein Interactions-Electromobility shift assay; DNaseI footprinting; Methyl interference assay

Unit II

10 Hrs

Cloning Vectors: Plasmids; Bacteriophages; M13 mp vectors; PUC19 and Bluescript vectors, Phagemids; Lambda vectors; Insertion and Replacement vectors; Cosmids; Artificial chromosome vectors (YACs; BACs); Animal Virus derived vectors-SV-40; vaccinia/baculo & retroviral vectors; Expression vectors; pMal; GST; pET-based vectors; Protein purification; His-tag; GST-tag; MBP-tag etc.; Intein-based vectors; Inclusion bodies; Methodologies to reduce formation of inclusion bodies; Baculovirus and pichia vectors system, Plant based vectors, Ti and Ri as vectors, Yeast vectors, Shuttle vectors

Unit III

6 Hrs

Cloning Methodologies: Insertion of Foreign DNA into Host Cells; Transformation; Construction of libraries; Isolation of mRNA and total RNA; cDNA and genomic libraries; cDNA and genomic cloning; Expression cloning; Jumping and hopping libraries; Southwestern and Far-western cloning; Protein-protein interactive cloning and Yeast two hybrid system; Phage display; Principles in maximizing gene expression

Unit IV

22 Hrs

PCR and Its Applications: Primer design; Fidelity of thermostable enzymes; DNA polymerases; Types of PCR – multiplex, nested, reverse transcriptase, real time PCR, touchdown PCR, hot start PCR, colony PCR, cloning of PCR products; T-vectors; Proof reading enzymes; PCR in gene recombination; Deletion; addition; Overlap extension; and SOEing; Site specific mutagenesis; PCR in molecular diagnostics; Viral and bacterial detection; PCR based mutagenesis, Mutation detection: SSCP, DGGE, RFLP, Oligo Ligation Assay (OLA), MCC (Mismatch Chemical Cleavage, ASA (Allele-Specific Amplification), PTT (Protein Truncation Test)

Sequencing methods: Enzymatic DNA sequencing; Chemical sequencing of DNA; Automated DNA sequencing; RNA sequencing; Chemical Synthesis of oligonucleotides; Introduction of DNA into mammalian cells; Transfection techniques; Gene silencing techniques; Introduction to siRNA; siRNA technology; Micro RNA; Construction of siRNA vectors; Principle and application of gene silencing; Gene knockouts and Gene Therapy; Creation of knock out mice; Disease model; Somatic and germ-line therapy- in vivo and ex-vivo; Suicide gene therapy; Gene replacement; Gene targeting; Transgenics; cDNA and intragenic arrays; Differential gene expression and protein array.

Text/References

1. S.B. Primrose, R.M. Twyman and R.W.Old; Principles of Gene Manipulation. 6th Edition, S.B.University Press, 2001.
2. J. Sambrook and D.W. Russel; Molecular Cloning: A Laboratory Manual, Vols 1-3, CSHL, 2001.
3. Brown TA, Genomes, 3rd ed. Garland Science 2006
4. Selected papers from scientific journals.
5. Technical Literature from Stratagene, Promega, Novagen, New England Biolab etc.
6. Genes, Benjamin XII , 2017

LAB-III (HARD CORE)

COURSE CODE:

Course Outcome

- CO1 -
- CO2 -
- CO3 -
- CO4 -
- CO5 -

Practicals/ Experiments

1. Animal cell culture: preparation of media, culture and maintenance of cell lines, trypsinization
2. Culture of transformed cells
3. MTT assay for cytotoxicity
4. Western blotting and detection
5. Study of fermenter (demo)
6. Immobilization of yeast by calcium alginate, gel entrapment and assay for enzyme *invertase*
7. Assay of catalase.
8. Study of alcohol fermentation – alcohol from different substrates – estimation of alcohol content
9. Solid state fermentation
10. Determination of the activity of enzyme protease
11. Determination of the activity of enzyme amylase
12. Estimation of Vitamin E
13. Estimation of Vitamin C
14. Estimation of aminoacid by ninhydrin method
15. Preparation of MS media
16. Induction of callus
17. Micropropagation
18. Suspension culture- production of secondary metabolites
19. Preparation of synthetic seeds
20. Database search for nucleotide and aminoacid sequences using BLAST
21. Study of sequence alignment
22. Construction of trees/dendrogram using sequence analysis
23. Structure prediction using homology searches
24. RAPD
25. Transformation
26. Bacterial gene expression
27. RFLP mapping
28. Isolation of plasmid DNA from *E.coli*
29. Restriction digestion of DNA
30. DNA ligation
31. production of citric acid by *A.niger* by submerged fermentation.
32. Estimation of citric acid by titrametric method
33. PCR
34. Isolation of antibiotic producing actinomycetes from soil sample

CLINICAL AND ADVANCED TECHNIQUE IN BIOTECHNOLOGY (SOFT CORE) – 48 Hrs

COURSE CODE:

Course Outcome

CO1-Diagnosis of diseases using enzymes as markers

CO2-analysis of blood and urine sample to interpret the diseases

CO3-Study of metabolic disorders and their diagnosis

CO4- Clinical trails of designed drugs/ biomolecules

CO5-Tools of Histopathology, Immunotechnology, microarray and DNA chips in understanding the diseases

Unit I

16 Hrs

Diagnostic Enzymology: Mechanisms of elevated enzyme activities. Some important enzymes – alkaline phosphates, creatine kinase, LDH, AST, ALT – isozyme changes

Blood: Composition, cells, functions of plasma proteins and lipo proteins in diseases. Disorders of hemoglobin – Thalassemia, sickle cell anemia. Anemias – Microcytic, normocytic and macrocytic.

Advanced methods in clinical analysis: Blood, urine and quantitative determination of metal ions in body fluids

Liver: Biochemical indices of hepatobiliary diseases. Bile pigments – formation of bilirubin, urobilinogen bile acids, jaundice – pre-hepatic, hepatic and post-hepatic; liver function tests, diseases of the liver – hepatitis, cholestasis, cirrhosis, Gallstones.

Unit II

8 Hrs

Kidney: Assessment of renal function – creatine clearance, renal calculi, uremia, laboratory investigation of kidney disorders.

Cardiovascular Disorders: major cardio vascular system – Atherosclerosis – risk factors, pathogenesis. Diagnosis and prognosis

Disorders of Amino Acid and nucleotide metabolism: Gout Lesch – Nyhan syndrome, orotic acid urea phenyl ketonuria, alkaptonuria, maple-syrup urine.

Clinical trails of designed drugs/biomolecules.

Molecular detection of diseases, Amniocentesis

Unit III

12 Hrs

Microscopy: Phase Contrast Microscopy, Fluorescence Microscopy, Confocal and Inverted Microscopy), Electron Microscopy (Transmission Electron Microscopy, Scanning Electron Microscopy)

Diagnostics and immunological techniques: applications of immunological and molecular diagnostic methods (RIA, ELISA, PCR, DNA finger printing) in forensic science and disease diagnosis. *In vitro* antigen-antibody reactions, Coombs' test, complement titration test (Direct and indirect), Immunofluorescence, Immuno-enzymatic and ferritin technique, Immuno-electromicroscopy. Immuno-electrophoresis, Western blot analysis. Hybridoma technology – Monoclonal and polyclonal antibodies and their application

Unit IV

12 Hrs

Nanobio-technology: Introduction, types and synthesis of nanomaterial, protein – based nano structures, DNA-based nano structures. Applications of nanomaterials, nano biosensors, drug and gene diversity, disease diagnostics, cancer therapy, risk potential of nanomaterials.

DNA chip technology and micro arrays: Types of DNA chips and their production, hybridization, application of micro arrays on DNA chips.

Genomic research: Methods for whole genome sequencing, genome sequence data, e-PCR, genome sequence to annotation- methods for annotation of genome sequence.

Reference Books

1. Biochemistry – With Clinical Correlations. Devlin.
2. Clinical Biochemistry. Latner.
3. Principles of Instrumental Analysis. 5th Ed. Douglas A Skoog, James Holler and Timothy A Nieman.
4. Analytical and Preparative Separation Methods of Biomacromolecules. Hassan Y Aboul – Enein.
5. Microbiology – Principles and Explorations. 5th Ed. Jacquelyn G Black.
6. Genetic Engineering: Primose, S. B.
7. An introduction to molecular Bio-technology (Ed.) Wink.
8. Principles of gene manipulation and genomics. Primose, S. B. and Twyman, R. M.
9. Gene cloning and DNA analysis an Introduction. Brown, T. A. Blackwell Science Company.
10. Molecular Biology and Biotechnology. Walker, J. M. and Rapley, R. Panima Publishing Corporation.
11. Molecular Biotechnology – Principles and application of Recombinant DNA. Glicks, R. Bernard and Pasternak, J. Jack. Panima Publishing Corporation.
12. Molecular Biomethods Hand Book. Rapley, R and Walker, M. Jhon. Humana Press.
13. Genes (VIII edition) Benjamin Lewin, Pearson Education International

**BIOSTATISTICS, BIOINFORMATICS AND BIOENTREPRENEURSHIP (SOFT CORE)
– 48 HRS**

COURSE CODE:

Course Outcome

CO1-Application of statistics to understand and analyse the experimental results of biological sciences

CO2-retrieval of biological data

CO3-phylogenetic analysis

CO4-primer designing

CO5-drug discovery and molecular docking

Unit I

12 Hrs

Statistical concept: Data structure, sampling methods, collection, classification and tabulation of data, graphical and diagrammatic representation, histogram, frequency polygon, frequency curve, bar graph, pie chart.

Measure of central frequency: **Mean, median, mode, mean deviation, standard deviation, standard error**

Types of distribution of data: Normal, binomial, Poisson, Z-test, t-test and ANOVA.

Correlation and regression

Unit II

18 Hrs

Bioinformatics: Introduction, history, internet and bioinformatics, knowledge, discovery and data mining, problems faced in bioinformatics area, opportunities in bioinformatics, human genome project.

Biological databases and their management: database concept, introduction, history of databases, databases management systems, types of database, Codd rules, data normalization biological databases – introduction, application and its importance, biological database and their functioning, types of biological database, microbiological database, primary sequence database, carbohydrate database, RNA database, genome database, organism database, biodiversity.

Sequence database: Introduction, nucleotide sequence database, protein sequence database, the EMBL nucleotide sequence database, structure databases.

Bioinformatics software: Clustal V Multiple sequence alignment, Clustal W Version 1.7, Ras Mol, Oligo, Mol script, TREEVIEW, ALSCRIPT, genetic analysis software, Phylip.

Computational biology: Introduction, data mining and sequence analysis, database similarities searches, practical aspects of multiple sequence alignment, phylogenetic analysis, predictive methods using nucleic acid and protein sequences, submitting DNA sequences to the databases.

Unit III

10 Hrs

Innovation: Idea to enter into business, Designing and development of new products as per market demands and their future prospective. Needs of customer, branding, distribution, promotion and advertising.

Types of bio-industries and IPR: biopharma, bioagri and bioservices. IP protection & commercialization strategies- freedom to operate.

Accounting and Finance : Business plan preparation, contracts, partnerships, business feasibility analysis by SWOT, socio-economic costs benefit analysis; funds/support from Government agencies like MSME/banks and private agencies like venture capitalists:/angel investors for bio entrepreneurship; business plan proposal for virtual start up company. statutory and legal requirements for starting a company/venture; basics in accounting practices: concepts of balance

sheet, profit and loss statement, Valuation, Cash flow, double entry. Information technology for business administration and expansion. Technology transfer.

Incubation centres: Govt. (C-CAMP, KBITS, CFTRI) and Private incubation centres for start-ups.

Unit IV

8 Hrs

Marketing : Market conditions, segments, prediction of market changes; identifying needs of customers; Market linkages, branding issues; developing distribution channels - franchising; policies, promotion, advertising; branding and market linkages for virtual start-up company.

Business Strategy & HR: Entry and exit strategy; pricing strategy; negotiations with financiers, bankers, government and law enforcement authorities; dispute resolution skills; external environment/ changes; avoiding/managing crisis; broader vision–global thinking; mergers & acquisitions.

Regulatory understanding:- GLP, GMP, GCP, PCB, IBSC, ISO

Bioentrepreneurship and case study: Importance of entrepreneurship; advantages of being entrepreneur - freedom to operate; introduction to bioentrepreneurship – biotechnology in a global scale; Scope in bioentrepreneurship; innovation – types, out of box thinking; skills for successful entrepreneur – creativity, leadership, managerial, team building, decision making, Risk assessment, opportunities for bioentrepreneurship- development programs of public and private agencies (MSME, DBT, BIRAC, Start-up & Make in India).

References:

1. Singh Narendra, Project management and control, (Himalaya publishing house)
2. Prasanna Chandra, Projects: Planning, Analysis, selection, implementation& review (Tata McGraw Hill)
3. P. GopalaKrishna& V.E. Rama Moorthy, Project management (Mac Millan India)
4. Chandra prasanna, proect preparation, Appraisal and Implementation (Tata Mcgrow Hill)
5. A. N. Desai, The dynamics of Entrepreneurial development and management (Himalaya publishing house)
6. Biostatistical Analysis. Zar J. H. Printice-Hall International.
7. Methods in Biostatistics. Mahajan, B. K. Smt. Hindu Mahajan
8. Bioinformatics. David W. Mount.
9. Bioinformatics A Practical Guide to the Analysis of Genes and Proteins Andreas D. Baxevanis and B. F. Francis Ouellette. A John Wiley & Sons, Inc., Publication.
10. Biostatistics. Daniel.
11. Handbook of Biostatistics A Review and Text. Christopher and Carvounis.

APPLIED BIOTECHNOLOGY (OPEN ELECTIVE) – 48 Hrs

Unit I Scope of Biotechnology in India and Karnataka. Structure of plant, animal and bacterial cells. Biomolecules and their importance. Enzyme Biotechnology Introduction to application of enzymes in industry: Food & beverage, detergent, textile pharmaceutical and leather.	12 Hrs
Unit II Applications of Plant Cell and tissue culture technology Improvement of hybrids, encapsulated seeds, production of disease resistant, stress resistant plants, secondary metabolites from cell cultures Transgenic plants for crop improvement, molecular farming from transgenic plants, edible vaccines. Bioethics in plant genetic engineering.	14 Hrs
Unit III Animal Cell Culture Techniques Manipulation of reproduction in animals: Artificial insemination, embryo transfer, embryo splitting, embryo sexing In vitro fertilization technology (IVF): Embryo cloning, embryonic stem cells In vitro fertilization and embryo transfer in humans. Transgenic animals Valuable products from animal cell culture (Tissue plasminogen activator, Blood factor VIII, erythropoietin.) Hybridoma technology: Production of monoclonal and polyclonal antibodies and their applications. Bioethics in animal genetic engineering cryopreservation, quantitation of cells, cytotoxicity assays.	10 Hrs
Unit IV Industrial and microbial biotechnology Growth media, sources of nutrition, sterilization, design of fermenter, batch, fed batch and continuous culture. Production of primary metabolites (vitamins, organic acids, alcohols and amino acids). Production of secondary metabolites (antibiotics) Biopesticides (Biological control of plant pathogens, pests and weeds.) Biofertilizers (microbial inoculants) Food Biotechnology – Genetically modified foods, Nutraceuticals, detection of genetically modified foods. Production of single cell proteins and mycoproteins.	12 Hrs

Reference Books

1. Biotechnology. B. D. Singh
2. Biotechnology. R. C. Dubey

OPEN ELECTIVE –FUNDAMENTALS OF BIOINFORMATICS

Contact Hours/ Week	: 4	Credits	: 4
Total Lecture Hours	: 48		

Course Outcomes:

- Ability to use popular bioinformatics tools to generate biologically meaningful results
- Ability to interpret biological results generated by a bioinformatics tool
- Application of some basic models and algorithms
- The students will gain an understanding of the computational challenges (and their solutions) in the analysis of large biological data sets; they will understand how some of the commonly used bioinformatics tools work, how to use these tools effectively

Unit I

12 Hrs

Introduction to Bioinformatics and Biological Database:

Introduction to bioinformatics, Review of Central Dogma, Genome organization -Prokaryotic and Eukaryotic. Overview of Genome Projects – Human genome project. Introduction to DNA and protein databases and their classification, file formats, information retrieval tools – Entrez, SRS, ARSA. Nucleotide and Protein sequence and structure databases (NCBI, EMBL, DDBJ and PDB). Focus on GenBank, UniProt, and Gene Ontology.

Unit II

12 Hrs

Sequence Alignment and Database Similarity Searching:

Pairwise alignment: Alignment algorithm: Pairwise: Dot matrix method, Dynamic programming Method (Needleman-Wunsch & Smith Waterman), Scoring Matrices – PAM and BLOSUM, Database Similarity Searching: FASTA and BLAST. BLAST variants, Statistical parameters for BLAST output – e value, p value and Bit Score.

Unit III

12 Hrs

Multiple sequence alignment:

Iterative, Progressive alignment. Application of MSA – 1. Phylogenetics – Phylogenetics Basics, Terminologies, Gene versus species phylogeny, Forms of tree representation: Maximum Parsimony and Distance methods 2. Gene prediction: Gene prediction in prokaryotes and eukaryotic 3. Protein Motif and Domain Prediction: Identification of Motif and Domains in MSA – PSSM and Profile HMMs.

Protein sequence analysis:

Analysis of Scalar parameters: ProtParam and pepstats: Hydropathy analysis (Membrane protein prediction): Kyte-Doolittle plot, Helical Wheel representation. Secondary structure prediction, Protein structure building-Homology modelling (Comparative modelling only) – SWISS MODEL server and MODELLER, Protein Structure Visualization: Rasmol, Pymol, CN3D, Swiss PDB viewer, Chimera and Discovery studio visualizer

Applications of Bioinformatics: Bioinformatics in pharmacy: overview of drug discovery process, structure based and ligand-based drug design (CADD). Pharmacokinetics: absorption, distribution, metabolism, excretion and toxicity of drugs.

REFERENCE BOOKS:

1	David W Mount	“Bioinformatics sequence and Genome analysis”, Cold Spring Harbor Laboratory Press, 2 nd Edition, 2013, 9989332257358
2	Jin Xiong	Essentials Bioinformatics, Cambridge university press, 3 rd Edition, 2006, 9789335657325
3	Neil C. Jones and Pavel A. Pevzner	An Introduction to Bioinformatics Algorithms, MIT Press, 5 th Edition, 2005, 8789432449328
4	Steffen Schulze-Kremer	Molecular Bioinformatics: Algorithms and Applications, Walter de Gruyter, 4 th Edition, 1996, 9789432449327
5	Attwood T K, D J Parry-Smith	Introduction to Bioinformatics, Pearson Education, 3 rd Edition, 2005, 9789332447329
6	Michael R Barnes and Ian C grey	Bioinformatics for Geneticists, John Wiley & Sons Ltd, The Atrium, Southern Gate, Chichester, West Sussex PO19 8SQ, England
7	Rui Jiang, Xuegong Zhang. Michael Q. Zhang	Basics of Bioinformatics, Springer Heidelberg New York Dordrecht London,
8	Supratim Choudhuri	Bioinformatics for Beginners, Academic Press.
9	Peter Lake and Paul Crowther	Concise Guide to Databases. Springer London Heidelberg New York Dordrecht
10	Arthur M. Lesk	Introduction to Bioinformatics, Oxford University Press Inc., New York
11	Mahmood A. Mahdavi	Bioinformatics –Trends and Methodologies, InTech Janeza Trdine 9, 51000 Rijeka, Croatia
12	Catherine Hack and Gary Kendal	Bioinformatics: Current Practice and Future Challenges for Life Science Education: Biochemistry and Molecular Biology Education Vol. 33, No. 2, pp. 82–85, 2005
13	Teresa K. Attwood	The Babel of Bioinformatics, SCIENCE, Volume 290, Number 5491, Issue of 27 Oct 2000, pp. 471-473.

IV SEMESTER

PLANT BIOTECHNOLOGY (HARD CORE) – 48 Hrs

Unit I

10 Hrs

Plant tissue culture-General: Historical background: Requirements for in-vitro culture- Tissue culture laboratory, Preparation of media, sterilization. Conventional plant breeding and plant tissue culture.

Cell and Tissue Culture Technology: **Role of hormones in growth and development of plants, tissue-specific hormones. Callus Induction, Organogenesis, Somatic embryogenesis, cell suspension culture and synthetic seeds**

Somaclonal variations: Isolation of somoclonal variants, Factors affecting somoclonal variants – applications

Micropropagation: Propagation from pre-existing meristem, shoot apical meristem, shoot and node culture, micropropagation stages and applications

Unit II

15 Hrs

Germplasm preservation: cryopreservation, cryoprotectant, warming rate and recovery, gene banks, applications.

Seed Health Technology

Introduction: Importance of Seed health, important seed-borne diseases; Seed Health diagnostics; Management of seed-borne diseases.

Haploid Technology: Methods of haploid culture, Factors affecting anther and microspore cultures, applications. Cytoplasmic male sterility in Indian Mustard.

Protoplast Technology: Isolation, purification and culture of protoplasts, protoplast fusion and somatic hybridization, applications of somatic hybrids/ cybrids.

Secondary metabolite production: Induction of secondary metabolites by plant cell culture, technology of plant cell culture for production of chemicals, biotransformation using plant cell culture. Bioreactor systems and models for mass cultivation of plant cells.

Unit III

7 Hrs

Plant transformation techniques: Methods of gene transfer in plants, *Agrobacterium* mediated transfer- mechanism of DNA transfer, general features of Ti and Ri plasmids, role of *vir* genes, design of expression vectors, use of promoters and reporter genes; viral vectors, direct gene transfer methods- electroporation, microinjection, particle bombardment, selection of transformants, screening and field trials.

Unit IV

16 Hrs

Transgenic plants: Herbicide resistance, resistance against biotic stress- bacterial, viral, fungal and insect resistance, abiotic stress, improved crop productivity, improved nutritional quality, transgenic plants for floriculture, Qualitative trait loci and marker studies.

Growth- promoting bacteria in plants: Biological nitrogen fixation, genetic manipulation for nitrogen fixation. Biocontrol of phytopathogens.

Molecular farming: Transgenic plants as production systems-production of alkaloids, steroids, colouring agents, flavoring agents, biodegradable plastics, industrial enzymes, therapeutic proteins, biopharmaceuticals, edible vaccines, plantibodies.

Intellectual Property Rights (IPR): IPRs and agricultural technology- implications for India, WTO, WIPO, GATT, TRIPS. Plant Breeder's Rights, legal implications, commercial exploitation of traditional knowledge, protection. Ethical issues associated with consumption of GM food, labelling of GM crops and foods.

Reference Books

1. Plant Signal Transduction. Scheel D and Wasterpack C. Oxford University Press.
2. Introduction to Plant Pathology. Strange R N. John Wiley and Sons Ltd.
3. Applied plant virology. Walkey. Chapman and Hall London.
4. Molecular Plant Pathology by Agrios.
5. Plant Tissue Culture Concepts and Laboratory Exercise. Trigiano R. N. and Gray, D. L. CRC Press.
6. Plant Tissue culture – Supplement-7. Lindsey, K. Springer International Edition.
7. Introduction to Plant Tissue Culture. Razdon, M. K. Oxford and IBH Publishing Co. Pvt Ltd.
8. Introductory to plant physiology. Noggle, R., Fritz, J. G. Prentice Hall of India Pvt. Ltd.
9. Plant Molecular Biology – A Practical Approach. Shaw, C. H. Panima Publishing Corporation.
10. A Laboratory Manual of Plant Biotechnology. Purohit. Publisher Agrobios.
11. Introduction to Plant Biotechnology. Chawla, H. S.
12. Practical Application of Plant Molecular Biology. Henry, R. J. Chapman and Hall.
13. Plant Biotechnology – Laboratory manual. Chawla, H. S. Oxford and IBH publishing Co. Pvt. Ltd.
14. Biotechnology. Gupta, P. K. Rastogi Publications.
15. Biochemistry and Molecular Biology of Plants. Buchanan, Gmissem and Jones.
16. Genetic Engineering of Crop Plants. Lyrett, G. W., Grierson, D.
17. Plant Molecular Biology. Grierson and S. N. Covey.

ANIMAL BIOTECHNOLOGY (HARD CORE) – 48 Hrs

Unit I

10 Hrs

Culture of animal cells: Advantages and limitations of tissue culture, aseptic handling, facilities required, media and cell lines. Primary culture: Isolation of mouse and chick embryos, human biopsies, methods for primary culture, nomenclature of cell lines, sub culture and propagation, immortalization of cell lines, cell line designation, selection of cell line and routine maintenance.

Cloning and Selection: Cloning protocol, stimulation of plating efficiency, suspension cloning, isolation of clones, isolation of genetic variants, interaction with substrate, selective inhibitors.

Unit II

16 Hrs

Cell separation and characterization: Density based, antibody based, magnetic and fluorescence based cell sorting. Characterization of cells based in morphology, chromosome analysis, DNA content, RNA and protein, enzyme activity, antigenic markers, cytotoxicity assays, cell quantitation, cell culture contamination: monitoring and eradication, cryopreservation.

Culturing of specialized cells: Epithelial, mesenchymal, neuro ectodermal, hematopoietic gonad and tumor cells, Lymphocyte preparation, culture of amniocytes, fish cells, confocal microscopy. Stem cell culture and its applications

Organic and embryo culture: Choice of models, organ culture, histotypic culture, filter-well inserts, neuronal aggregates whole embryo culture eggs, chick and mammalian embryos.

Unit III

16 Hrs

Cell and Tissue engineering: Growth factors for *in situ* tissue regeneration, biomaterials in tissue engineering, approaches for tissue engineering of skin, bone grafts, nerve grafts. Hemoglobin based blood substitutes, bio artificial or biohybrid organs. Limitations and possibilities of tissue engineering.

***In vitro* fertilization and Embryo transfer:** *In vitro* fertilization in Humans, Embryo transfer in Humans, Super ovulation and embryo transfer in farm animals e.g: Cow.

Cloning of Animals: Methods and uses. Introduction, nuclear transfer for cloning, cloning from-embryonic cells, adult and fetal cells. Cloning from short term cultured cells: cloning of sheep, monkeys, mice, pets, goats and pigs. Cloning from long term cultured cells: Cloning of cows from aged animals. Cloning efficiency, Cloning for production of transgenic animals, gene targeting for cloned transgenic animals, cloning for conservation, human cloning: ethical issues and risks.

Unit IV

6 Hrs

Transfection methods and transgenic animals: Gene transfer or transfection, transfection of fertilized eggs or embryos, unfertilized eggs, cultured mammalian cells, targeted gene transfer. Transgenic animals and applications: mice and other animals, sheep, pigs, goats, cows and fish.

The legal and socio-economic impact of biotechnology at national and international levels, public awareness. Biosafety regulations- guidelines for research in transgenic animals, public awareness of the processes of producing transgenic organisms

Reference Books

1. Anthony Atala, Robert P Lanza. 2002, Methods of tissue engineering, Academic press
2. Ian Freshney R. 2005, Culture of animal cells–A manual of basic techniques, John Wiley and Sons Inc. Hoboken, New Jersey
3. Animal Cell Culture – A Laboratory Manual. Frushney.
4. Animal Biotechnology. Ballinic, C. A., Philip, J. P and Moo Young, M. Pergamon Press.
5. Genetic Engineering of Animals. Puhler, A. VCH Publisher.
6. Methods of Tissue Engineering. Anthony Atala, Robert P. Lanza.
7. Animal Cell Biotechnology – Methods and Protocols. Nigel Tenkins.

PROJECT WORK/DISSERTATION (HARD CORE)

COURSE CODE:

Course Outcome

CO1-Review of recent research articles published in high impact journals and presentation by students.

CO2-Students do conduct review of literature followed by hands on training to do piece of research work.

CO3-They would be skill full to understand the experiment and interpret the result.

CO4-They get an idea to compile the data and present in the form of dissertation.

- Includes exhaustive review of literature on the topic selected, design of work, standardization of techniques and execution of work
- Compiling of the data generated in the form of thesis. Interpretation of the result correlating with the advanced information available in the literature.
- Research Paper presentation.



JSS COLLEGE OF ARTS COMMERCE AND SCIENCE
(An Autonomous College of University of Mysore; Re-Accredited by
NAAC with 'A' Grade)

OOTY ROAD, MYSURU- 25

PG DEPARTMENT OF BOTANY

Choice - Based Credit System (CBCS)

BOTANY

M.Sc. DEGREE SYLLABUS

2018-19 ONWARDS
(MODIFIED ON 2022)

JSS MAHAVIDYAPEETHA
 JSS COLLEGE FOR ARTS, COMMERCE AND SCIENCE
 (AUTONOMOUS) OOTY ROAD, MYSURU- 25
POST GRADUATE DEPARTMENT OF BOTANY

**M.Sc., Botany Choice - Based Credit System (CBCS) Syllabus
 (CBCS-CGPA-Modified (2018-19))
 CORE SUBJECT: BOTANY – [POST GRADUATE]**

DEGREE: M.Sc., BOTANY

1st and 3rd semester Changes made at BOS meeting held on 13.01.2022 (in %)

HC 1.3 Systematics of Angiosperms (5.17%)

HC 3.3 Plant Biotechnology (40.22%)

SC 3.3 Plant Propagation and Plant Breeding (1.7%)

OE 3.1 Plant Propagation Techniques (1.35%)

(CHANGES MADE ARE HIGHLIGHTED IN THE TEXT)

FIRST SEMESTER				Credits: 22
No.	Course/Paper Code	Title of the Course/ Paper	Hrs/Week L:T:P	Credits
1	HARD CORE 1.1	Virology, Bacteriology, Mycology and Plant Pathology	2:2:2	2:1:1
2	HARD CORE 1.2	Phycology, Bryophytes, Pteridophytes and Gymnosperms	2:2:2	2:1:1
3	HARD CORE 1.3	Systematics of Angiosperms	2:2:2	2:1:1+ (2 credits for submission of tour report) 2:1:3
4	SOFT CORE 1.1**	Fungal Biology and Biotechnology	2:2:2	2:1:1
5	SOFT CORE 1.2**	Algal Biology and Biotechnology	2:2:2	2:1:1
6	SOFT CORE 1.3**	Lichenology and Mycorrhizal Technology	2:2:2	2:1:1
7	SOFT CORE 1.4**	Phytopathology	2:2:2	2:1:1
<p>*Field Study/Tour: The student shall undertake a field trip for a minimum of 2-3 days and shall submit the herbaria and tour report for evaluation-2 credits.</p> <p>**Any two soft core papers shall be studied.</p>				

SECOND SEMESTER			Credits: 18	
No.	Course/Paper Code	Title of the Course / Paper	Hrs/Week L:T:P	Credits
1	HARD CORE 2.1	Reproductive Biology of Angiosperms and Plant Morphogenesis	2:2:2	2:1:1
2	HARD CORE 2.2	Cell Biology and Genetics	2:2:2	2:1:1
3	HARD CORE 2.3	Plant Breeding and Evolutionary Biology	2:2:2	2:1:1
4	SOFT CORE 2.1*	Plant Anatomy and Histochemistry	2:0:2	2:0:1
5	SOFT CORE 2.2*	Ethno-Botany and Intellectual Property Rights (IPR)	2:0:2	2:0:1
6	SOFT CORE 2.3*	Economic Botany	2:0:2	2:0:1
7	OPEN ELECTIVE 2.1	Medicinal Plants	2:2:0	2:1:0
** Any two soft core papers shall be studied.				

THIRD SEMESTER			Credits: 16	
No.	Course/Paper Code	Title of the Course /Paper	Hrs/Week L:T:P	Credits
1	HARD CORE 3.1	Biochemistry and Plant Physiology	2:2:2	2:1:1
2	HARD CORE 3.2	Molecular Biology	2:2:2	2:1:1
3	HARD CORE 3.3	Plant Biotechnology	2:2:2	2:1:1
4	SOFT CORE 3.1*	Molecular Genetics of Plants	2:2:2	2:1:1
5	SOFT CORE 3.2*	Molecular Plant Pathology	2:2:2	2:1:1
6	SOFT CORE 3.3*	Plant Propagation and Plant Breeding	2:2:2	2:1:1
7	SOFT CORE 3.4*	Phyto-chemistry and Herbal Technology	2:2:2	2:1:1
8	OPEN ELECTIVE 3.1	Plant Propagation Techniques	2:2:0	2:1:0
* Any one soft core courses/papers shall be studied.				

FOURTH SEMESTER				Credits:
16				
No.	Course/Paper Code	Title of the Course /Paper	Hrs/Wk L:T:P	Credits
1	HARD CORE 4.1	Ecology, Conservation Biology and Phytogeography	2:2:2	2:1:1
2	HARD CORE 4.2	Project Work *	4:2:2	8
3	SOFT CORE 4.1*	Seed Technology	2:2:2	2:1:1
4	SOFT CORE 4.2*	Seed Pathology	2:2:2	2:1:1
5	SOFT CORE 4.3*	Bio -Analytical Techniques	2:2:2	2:1:1
6	OPEN ELECTIVE 4.1	Plant Diversity and Human Welfare	2:2:0	2:1:1
<p>*Project Work: The student shall undertake a Project Work in the Department or in any other University or Institute under the guidance of a Research Supervisor and shall submit a Project Report duly signed by Student and Research Supervisor for Evaluation.</p>				

Semester- Wise Credit Pattern:

I Semester= 22 [HC- 12+2=14 + 08 (SC)]

II Semester= 24 [HC- 12 + 08 (SC) + 04 (OE)]

III Semester= 18 (HC- 08 + 06 (SC) + 04 (OE)]

IV Semester= 20 (HC-12 +04 (SC) + 04 (OE)]

In total= 46 HC + 26 (SC) + 12 (OE)= The Department is offering 84 Credits of B.Sc. Honors/ M.Sc. Botany (CBCS) Course including three Open Elective Course to the outside Department Students/

Important Note:

Student is required to earn the credit for qualifying B.Sc. Honors/ M.Sc. Botany from Department of Botany as follows:

Hard Core offered by the Department= 46 (Against maximum of 56)

Soft Core offered by the Department = 26 (Against minimum of 16)

Minimum Open Elective to be earned by the Student (Outside the Department) = 04

A total of 76 Credit is required for qualifying B.Sc. Honors/ M.Sc. Botany Course.

**SCHEME OF EXAMINATION/ASSESSMENT
MODEL QUESTION PAPER (THEORY)
JSS COLLEGE FOR ARTS, COMMERCE AND SCIENCE
(AUTONOMOUS) OOTY ROAD, MYSURU- 25
POST GRADUATE DEPARTMENT OF BOTANY
M.Sc., Degree -----Semester Examination May/June-20--
BOTANY**

Course/Paper:
Course/Paper Code.....

Time: 3 Hrs

Max Marks: 70

**Instructions: 1) Answer all questions.
2) Draw neat and labelled diagrams wherever necessary.**

I. Answer the following; (10MCQs of 1 Marks each)

10 X 1 = 10

- 2 from Unit I
- 3 from Unit II
- 2 from Unit III
- 3 from Unit IV

II. Answer the following;

4 X 5 = 20

- 2 from Unit I with internal choice
- 2 from Unit II with internal choice
- 2 from Unit III with internal choice
- 2 from Unit IV with internal choice

III. Answer the following;

4 X10 = 40

- 2 from Unit I with internal choice
- 2 from Unit II with internal choice
- 2 from Unit III with internal choice
- 2 from Unit IV with internal choice

**SCHEME OF PRACTICAL EXAMINATION/ASSESSMENT
MODEL QUESTION PAPER (PRACTICALS)**

**JSS COLLEGE FOR ARTS, COMMERCE AND SCIENCE
(AUTONOMOUS) OOTY ROAD, MYSURU- 25
POST GRADUATE DEPARTMENT OF BOTANY
M.Sc., Degree I Semester Examination May/June-2018
BOTANY**

Course/Paper:
Course/Paper Code.....

Time: 3 Hrs

Max Marks: 70

Conducting Experiment/Micro-preparation /Plant identification	15	
Q II. Minor experiment/ Demonstrations/ Procedure Writing		10
Q III. Critically comments (3x5 Marks)	15	
Q IV. Identification 5x2 Marks)	10	
Q V. Viva-voce examination	10	
Q VI. Class Records/ Submissions	10	

Q I.

PO M.SC. BOTANY

Sl. No.	PO
1.	Conduct investigations of complex problems by the use of research-based knowledge on an independent term project.
2.	Transfer of appropriate knowledge and methods from one topic to another within the subject.
3.	Carry out practical work, in the field and in the laboratory, with minimal risk.
4.	Able to think logically and organize tasks into a structured form and assimilate knowledge and ideas based on wide reading of text books and through the internet.
5.	Apply the scientific knowledge of basic science, life sciences and fundamental process of plants to study and analyse any plant form.
6.	Knowledge and understanding of the range of plant biology in terms of structure, function and environmental relationships.
7.	Apply reasoning informed by the contextual knowledge to assess plant diversity, and the consequent responsibilities relevant to the biodiversity conservation practice.

PSO M.SC. BOTANY

Sl. No.	COURSE	PSO
1.	Algal Biology and Biotechnology	Phylogeny, thallus organisation, economic and ecological importance of algal community
2.	Biochemistry and Plant Physiology	Biomolecules, metabolic pathways and stress physiology in plants
3.	Cell Biology and Genetics	Cell originals and Mendelian principles
4.	Ecology, Conservation Biology and Phytogeography	Diversity of vegetation, distribution and its conservation
5.	Economic Botany	Economic values of different crop plants and their applications
6.	Major Project	Hands on experience in various fields of plant science
7.	Molecular Biology	Molecular level organisation in prokaryotes and eukaryotes with respect to various mechanisms involved
8.	Plant Anatomy and Histochemistry	Anatomical features and organisation of cells in plants
9.	Plant Breeding and Evolutionary Biology	Plant breeding methods, procedures and their application for crop improvement
10.	Plant Biotechnology	Tissue culture techniques and its application in development of resistant varieties
11.	Plant Propagation and Plant Breeding	Propagation methods and plant breeding procedures and their application in different fields
12.	Plant Propagation Techniques	Propagation methods and procedures and their application in different fields
13.	Phycology, Bryophytes, Pteridophytes and Gymnosperms	Distribution, classification and phylogeny of lower plant communities
14.	Phytopathology	Concepts of plant diseases defence mechanisms in plants and study of plant diseases
15.	Reproductive Biology of Angiosperms and Plant Morphogenesis	Embryological study of growth and development using plant models
16.	Seed Technology	Industrial scale processing of seeds up to marketing

17.	Systematics of Angiosperms	Angiospermic plant family study with their phylogeny
18.	Virology, Bacteriology, Mycology and Plant Pathology	Diversity, distribution of microorganism with respect to their economic aspects

CO M.SC. BOTANY

Sl. No.	COURSE	CO
1.	Algal Biology and Biotechnology	Specify in depth of thallus organization and phylogeny in algae
2.	Algal Biology and Biotechnology	Understand the details of toxins, blooms and distributions of algae
3.	Algal Biology and Biotechnology	Deliberate in depth about cultivation and marketing algae
4.	Algal Biology and Biotechnology	Specify the details of Algal products and uses
5.	Biochemistry and Plant Physiology	Learn in details with biomolecules and their function
6.	Biochemistry and Plant Physiology	Understand in depth about solute transport and photosynthesis in plants
7.	Biochemistry and Plant Physiology	Specify the details of metabolism of nitrogen, lipids and plant hormones
8.	Biochemistry and Plant Physiology	Understand in depth about Stress physiology
9.	Cell Biology and Genetics	Learn in detail about cell membranes transport and proteins
10.	Cell Biology and Genetics	Deliberate the Functions of cell organelles, programmed cell death
11.	Cell Biology and Genetics	Specify the extensions of Mendelian principles
12.	Cell Biology and Genetics	Learn about Sex determination and dosage compensation
13.	Ecology, Conservation Biology and Phytogeography	Understand the diversity of ecosystem and types of ecosystems
14.	Ecology, Conservation Biology and Phytogeography	Learn the in details of pollution and environmental biology
15.	Ecology, Conservation Biology and Phytogeography	Study the importance of biodiversity and conservation biology
16.	Ecology, Conservation Biology and Phytogeography	Detailed study of phytogeography and crop distribution
17.	Economic Botany	Specify the details of cereals, millets, pulses, oil yielding plants and study of horticultural plants and floriculture
18.	Economic Botany	Deliberate the characteristics of sugar yielding plants, spices and condiments
19.	Economic Botany	Understand the importance of fibre, timber and gum yielding plant
20.	Economic Botany	Deliberate on the medicinal plants and their applications
21.	Major Project	Learn the details of literature survey and methodology in research
22.	Molecular Biology	Identify the characteristics of genetic materials and its replication
23.	Molecular Biology	Learn the details of molecular basis of mutation, repair and recombination
24.	Molecular Biology	Deliberate the details of RNA formation, processing of RNA and post-RNA
25.	Molecular Biology	Understand in depth of gene regulation in prokaryotes and eukaryotes
26.	Plant Anatomy and Histochemistry	Learn in details of primary vegetative body of the plants
27.	Plant Anatomy and Histochemistry	Deliberate in details of differentiation in vascular tissues and study of apical meristems in shoot and root
28.	Plant Anatomy and Histochemistry	Deliberate the characteristics of secondary growth
29.	Plant Anatomy and	Understand the details of plant histochemistry

	Histochemistry	
30.	Plant Breeding and Evolutionary Biology	Learn in depth about plant breeding methods and techniques
31.	Plant Breeding and Evolutionary Biology	Understand the details of breeding for specific purposes
32.	Plant Breeding and Evolutionary Biology	Learn the details of Nature of evolution
33.	Plant Breeding and Evolutionary Biology	Identify the characteristics of variation and speciation
34.	Plant Biotechnology	Understand in depth about plant tissue culture and its techniques
35.	Plant Biotechnology	Specify the genetic engineering and tools used in it
36.	Plant Biotechnology	Understand the details of genetic manipulation, transgenic approaches to produce resistant plants
37.	Plant Biotechnology	Learn the details of engineering of crop plants for production of secondary metabolites
38.	Plant Propagation and Plant Breeding	Learn the details of importance of plant propagation, vegetative propagation and micro propagation
39.	Plant Propagation and Plant Breeding	Understanding of basic concepts of plant breeding and genetics
40.	Plant Propagation and Plant Breeding	Study types, purposes of plant breeding
41.	Plant Propagation and Plant Breeding	Deliberate study of advanced breeding aspects
42.	Plant Propagation Techniques	Learn the details of importance of plant propagation
43.	Plant Propagation Techniques	Understand in depth about types of vegetative propagation
44.	Plant Propagation Techniques	Learn the techniques of budding and layering
45.	Plant Propagation Techniques	Deliberate in details with examples of micro propagation in forestry and horticulture plants
46.	Phycology, Bryophytes, Pteridophytes and Gymnosperms	Understand the details of diversity, distribution, pigmentation and life cycle of algae
47.	Phycology, Bryophytes, Pteridophytes and Gymnosperms	Deliberate in depth of Bryophytes life cycle, classification, phylogeny and Economic importance
48.	Phycology, Bryophytes, Pteridophytes and Gymnosperms	Understand the details of Pteridophytes life cycle, phylogeny, classification, economic importance and anatomy
49.	Phycology, Bryophytes, Pteridophytes and Gymnosperms	Write down in details with examples Gymnosperms history, reproduction, edconomic importance and interrelationship
50.	Phytopathology	Learn the details of the concept, causative agents and disease cycle of plant pathogens
51.	Phytopathology	Deliberate the details of defense mechanisms in plants and its genetics
52.	Phytopathology	Study of Management of plant diseases
53.	Phytopathology	Identify in details with examples of diseases in crop plants
54.	Reproductive Biology of Angiosperms and Plant Morphogenesis	Understanding the microsporogenesis and historical overview
55.	Reproductive Biology of Angiosperms and Plant Morphogenesis	Specify in details with examples about megasporogenesis, fertilization, endosperm and embryo
56.	Reproductive Biology of Angiosperms and Plant Morphogenesis	Specify the details of models and concepts of plant morphogenesis
57.	Reproductive Biology of Angiosperms and Plant Morphogenesis	Understand in details with examples of plant growth and development, photomorphogenesis
58.	Seed Technology	Understand the seed science and concepts
59.	Seed Technology	Study the seed production and processing methods

60.	Seed Technology	Learn about seed quality parameters and tests
61.	Seed Technology	Deliberate the procedure of seed certification
62.	Systematics of Angiosperms	Understand the principles and applications of Taxonomy of angiosperms
63.	Systematics of Angiosperms	Specify the details of taxonomic literature
64.	Systematics of Angiosperms	Deliberate in details with examples Dicot and monocot family and features of classification systems
65.	Systematics of Angiosperms	Specify in details molecular systematics with examples of softwares and databases
66.	Virology, Bacteriology, Mycology and Plant Pathology	Learn the classification and characteristics of viruses, viroids, prions and diseases of it
67.	Virology, Bacteriology, Mycology and Plant Pathology	Deliberate in details with examples of Bacteria, archeabacteria, actinomycetes and mycoplasma and its economic importance
68.	Virology, Bacteriology, Mycology and Plant Pathology	Specify the Fungal diversity, life cycle and economic importance of fungi
69.	Virology, Bacteriology, Mycology and Plant Pathology	Understand in details of etiology, distribution and management of plant disease

BOTANY: I SEMESTER- HARD CORE 1.1
VIROLOGY, BACTERIOLOGY, MYCOLOGY AND PLANT PATHOLOGY

Theory-32 Hrs

Unit-1: Virology: Origin and evolution of viruses; Classification of viruses-ICTV and Baltimore Systems; Genome diversity in viruses; Methods of cultivation of viruses; Purification and detection of viruses; Transmission of viruses; Mechanism of replication of DNA and RNA viruses; Viroids - Structure and multiplication; Prions - structure and multiplication; Prion diseases.

Unit-2: Bacteriology: Introduction and classification of Bacteria by Bergey's Manual of Determinative and Systematic Bacteriology; C. R. Woese- Three domain classification of Bacteria; Archaeobacteria and Eubacteria - diversity and evolution; Nutritional types of bacteria; Bacterial growth; Recombination in bacteria (conjugation transformation, and transduction); Brief account on actinomycetes; Structure and multiplication of Mycoplasma and Phytoplasmas; Economic importance of bacteria.

Unit -3: Mycology: Present status of fungi; Outline classification of fungi (Ainsworth-1973). Vegetative organization in fungi; Nutrition in fungi (saprotrophs, biotrophs, necrotrophs; symbiotrophs); Methods of reproduction in fungi - Asexual and sexual methods; Spore liberation in fungi; Evolution of sex in fungi; Heterothallism and parasexuality; Life cycle pattern and phylogeny of Myxomycotina, Mastigomycotina, Zygomycotina, Ascomycotina, Basidiomycotina and Deuteromycotina; Fungi and their economic importance.

Unit-4: Plant Pathology: Concepts and scope of plant pathology; Plant diseases and crop losses; Classification of plant diseases; Parasitism and disease development; Effect on physiology of host; Host range of pathogens; Defence Mechanisms in Plants; Plant Disease epidemics and plant disease forecasting; Methods of plant disease management; Study of plant diseases- Sandal Spike, Citrus Canker, Bacterial Blight of Paddy, Late Blight of Potato, Downy Mildew of Bajra, Tikka Disease of Ground nut, Grain Smut of Sorghum. Phloem Necrosis of Coffee, Root Knot Disease of Mulberry.

Practicals-32 Hrs

- 1) Laboratory guidelines, design, tools, equipments and other requirements for studying microorganisms.
- 2) Measuring the dimensions of microorganisms using Micrometry.
- 3) Determining total count of microbes using Haemocytometer.
- 4) Gram and special staining of bacteria.
- 5) Preparation of NA, PDA, sterilization, pouring, inoculation and culturing of bacteria/fungi.
- 6) Staining of fungi including VAM fungi.
- 7) Identification of fungi.
- 8) Measurement of bacterial growth by Spectrophotometer.
- 9) Recording environmental factors (Temperature, RH, and Rainfall and wind velocity).
- 10) Splash liberation of spores from diseased tissue.
- 11) Estimation of total phenols in diseased and healthy plant tissues.
- 12) Study of the following diseases: Sandal Spike, Citrus canker, Bacterial Blight of paddy, Late Blight of Potato. Downy Mildew of Bajra, Tikka disease of ground nut, Grain smut of Sorghum, Phloem Necrosis of Coffee, Root Knot disease of Mulberry.

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- 16) Holt, J.G., Krige, N.R., Sneath, P.H.A. Stuley, J.T. and Williams, S.T. 2010. Bergey's Manual of Determinative Bacteriology, 9th edn. Williams and Wilkins, USA.

BOTANY: I SEMESTER - HARD CORE 1.2
PHYCOLOGY, BRYOPHYTES, PTERIDOPHYTES AND GYMNOSPERMS

Theory-32 Hrs

Unit-1: Phycology: Diversity and distribution of algae; Unicellular, colonial, filamentous, heterotrichous, parenchymatous, pseudoparenchymatous, siphonous forms; General characteristics, classification and phylogeny of algae; Pigmentation in algal groups; Role of photosynthetic and accessory pigments; Life cycles in algae - haplontic, diplontic, isomorphic, heteromorphic; Economic importance of algae.

Unit -2: Bryophytes: Introduction, general characteristics, classification and phylogeny of Bryophytes; Distribution, habitat, external and internal morphology and reproduction; Comparative account on gametophytes and sporophytes of bryophytes; Economic and ecological importance.

Unit -3: Pteridophytes: Introduction, classification and phylogeny; Morphology, anatomy reproductive biology and phylogeny; Psilophytes, Lycophytes, Sphenophytes, Filicophyta; Evolution of sorus; evolution of sporangium; Gemetophyte development - homosporous and heterosporous ferns; Heterospory and seed habit; Stellar evolution in Pteridophytes; Ecology of Pteridophytes; Economic importance.

Unit- 4: Gymnosperms: Distribution, general characteristics, classification and phylogeny of Gymnosperms; Range in morphology, anatomy, reproduction and interrelationships of - Cycadales, Ginkgoales, Coniferales, Gnetales; Pteridosperms; Economic importance of Gymnosperms.

Practicals-32 Hrs

1-4) Algae: Study of Cyanophyceae: *Anabaena*, *Oscillatoria*; Study of Chlorophyceae: *Oedogonium*, *Pediastrum*; Study of Phaeophyceae: *Turbinaria*, *Ectocarpus*; Study of Rhodophyceae: *Gracilaria*, *Batrachospermum*; Economic products of algae.

5-7) **Bryophytes:** Study of morphology, anatomy and reproductive morphology - Hepaticopsida- *Marchantia*, *Dumortiera*; Anthocerotopsida- *Anthoceros*, *Notothylas*; Bryopsida- *Bryum* and *Polytrichum*.

8-10) **Pteridophytes:** Study of vegetative habit, anatomy and reproductive morphology of *Psilotum*, *Lycopodium*, *Isoetes*, *Ophioglossum*, *Botrychium*, *Angiopteris*, *Pteris*, *Hymenophyllum*, *Marselia*, *Salvinia*, *Azolla*; **Paleobotany-** Study of Lepidodendrales, Calamitales, Sphenophyllales and Coenopteridales (Fossil Pteridophytes).

11-12) **Gymnosperms:** Study of morphology, anatomy and reproductive morphology of *Zamia*, *Pinus* and *Ephedra*, *Ginkgo*, *Auracaria*, *Podocarpus*, *Gnetum*, *Agathis*, *Cupressus*, *Thuja*; Economic importance of Gymnosperms.

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BOTANY: I SEMESTER - HARD CORE 1.3
SYSTEMATICS OF ANGIOSPERMS

Theory-32 Hrs

Unit-1: Introduction to plant systematics; Plant classification systems-artificial, natural and phylogenetic systems; Contributions of Carolus Linnaeus, Michel Adanson, de Jussieu, de Candolle to plant classification; Concepts of taxonomic hierarchy; Taxonomic Categories- Genus concept; Species concept; Intraspecific categories; subspecies; varieties and forms; History of botanical nomenclature; ICBN and ICN aims and principles; Rules and recommendations; Rule of priority; Typification; Author citation, Legitimate and illegitimate names; Name changes and synonyms; Effective and valid publication; Herbarium and its significance; Botanical gardens.

Unit-2: Taxonomic Literature: General taxonomic indices, world floras and manuals; Monographs and revisions; Bibliographies, catalogues and reviews; Periodicals, glossaries and dictionaries; Hortus Malabaricus; Taxonomic websites-IPNI, Plant List, Tropicos, Botanicum-Periodicum-Huntianum (BPH); Biodiversity Heritage Library (BHL); Botanicus, Index Herbariorum; Taxonomic Keys- bracketed keys, indented keys, numbered keys, edge punched and body punched keys.

Unit-3: Study of plant classification Systems; Broad outlines of Bentham and Hooker's system, Engler and Prantl's system, Hutchinson's system, Takhtajan's system, and Cronquist's system; Numerical Taxonomy-principles, selection of characters, merits and demerits; Angiosperm Phylogeny Group (APG) III & IV classification; Study of angiosperm families-Magnoliaceae, Nymphaeaceae, Urticaceae, Papaveraceae, Euphorbiaceae, Acanthaceae, Rubiaceae, Alismataceae, Cyperaceae, Commelinaceae, Zingiberaceae, Liliaceae, Dioscoreaceae and Orchidaceae.

Unit-4: Molecular Systematics: Nuclear, mitochondrial and chloroplast genes. Gene sequencing, analysis of molecular data, alignment of sequences; Phylogenetic tree construction-Maximum Likelihood and Neighbour Joining Methods; Phylogenetic analysis-rooted and unrooted trees; Data analysis- alignment, substitution, model building; Phylogenetic softwares-CLUSTAL W, MEGA, Mesquite, PAUP, PHYLIP, Treefinder, TreeBase.

Practicals-32 Hrs

1) Methods of preparation and maintenance of Herbaria.

2-4) A field trip of three days to a floristically rich area to study plants belonging to different families (Every student shall submit a report for evaluation for two credits).

5-10) Identification of the flowering plants in and around Mysore using keys, floras and monographs.

11-12) Construction of phylogenetic tree based on molecular data of plant species retrieved from GenBank.

References:

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BOTANY: I SEMESTER - SOFT CORE 1.1
FUNGAL BIOLOGY AND BIOTECHNOLOGY

Theory-32 Hrs

Unit-1: Introduction and historical overview of mycology; General characteristics and importance of fungi in human life; Fungi –Taxonomy and Systematics; Fungi in genetic and applied research; Estimation of Fungal diversity; Quantitative Indices- species richness, species evenness and species abundance; Molecular methods used for fungal diversity estimation-nuclear genome, messenger RNA transcripts, Ribosomal/DNA sequence comparisons and mitochondrial genome.

Unit-2: Macro fungi and micro fungi living on plant substrata; Lignicolous macrofungi; Lichenized fungi; Sequestrate fungi; Endophytic fungi; Saprobic soil fungi; Fungi in stressful environment; Mutualistic, arbuscular, and endomycorrhizal fungi; Yeasts; Fungicolous fungi; Fungi in fresh and marine water habitats; Fungi associated with aquatic animals; Fungi as parasites of humans and plants; Fungi associated with animals, insect, arthropod and nematodes; Coprophilous fungi.

Unit-3: Fungal Fermentation and Food Products: Food and Beverages; Single cell proteins- Myco-proteins; Food processing by fungi-bread, soybean products, cheese and fermented milk; Fungal secondary metabolites-antibiotics, immunosuppressive agents, anti-tumour agents, fungal toxins as medicines; Fungal pigments; Steroid transformation; Fungal enzymes; Bio-control agents; Application of molecular biology in fungal biotechnology.

Unit-4: Mushrooms and fungi in medicine; Toxic macromycetes; Mushroom cultivation; Model organisms- *Saccharomyces cerevisiae/Neurospora crassa*; Bio-deterioration of food grains and mycotoxins; Fungal communities of herbivore dung; The fungal communities of composts; Fungal interactions and practical exploitation; Heavy metals in fungi-accumulation and sorption; Biotechnology of wood rotting fungi.

Practicals-32 Hrs

- 1) Study of Myxomycetes and Chytridiomycetes
- 2) Study of Plasmodiophoromycetes and Oomycetes
- 3) Study of Zygomycetes
- 4) Study of Ascomycetes
- 5) Study of Basidiomycetes
- 6) Study of
- Deuteromycetes 7) Study of
- Lichens
- 8) Study of VAM fungi
- 9) Detection of aflatoxin B1
- 10) Cultivation of Oyster mushroom.
- 11) Alcoholic fermentation of grape juice by *Saccharomyces*.
- 12) Cultivation of *Penicillium* and testing antibiotic principle.
- 13) Study of edible and poisonous mushrooms.
- 14) Study of fungal model organisms - *Saccharomyces cerevisiae/Neurospora crassa*

References:

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- 2) Deacon, J. W. 1997. Modern Mycology 3rd edn. Blackwell Science publishers, London.

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BOTANY: I SEMESTER - SOFT CORE 1.2
ALGAL BIOLOGY AND BIOTECHNOLOGY

Theory-32 Hrs

Unit-1: Algal Biology: Historical development of Phycology and contributions of Phycologists; Thallus organization in algae-Cyanophyceae, Chlorophyceae, Charophyceae, Euglenophyceae, Xanthophyceae, Bacillariophyceae, Phaeophyceae and Rhodophyceae; General characteristics, algal classification, affinities and phylogeny- polyphasic approach; Molecular markers for phylogenetic study; Algal physiology- ultra-structure of cells; Photosynthesis and respiration.

Unit-2: Algal blooms and Toxins: Blooms produced by algal groups; Toxins produced by cyanobacteria, diatoms, dinoflagellates, prymnesiophytes and eugleoids; bioaccumulation and biomagnification; effects of toxins on aquatic life and humans; Scenario in coastal waters of India- monitoring and safety measures; Algal communities of extreme environments- Thermal hot springs, cold springs, snow and ice; **Fresh water algae-** Ecological classification of fresh water organisms; Lentic communities of algae (pond, lake, bog, swamp); Lotic communities (streams, rivers, rapids; **Marine algae-** Marine biota; zonation; quantitative study of phytoplanktons, marine communities of algae.

Unit-3: Algal Biotechnology: Algal culture techniques; general principles; physical parameters; culture media; strain improvement; **Algal cultivation methods-**conventional, advanced; **Cultivation of microalgae-***Spirulina* and *Dunaliella*; Media, seeding, cultivation systems, harvesting; processing, drying methods, packaging, marketing; Algal cultivation and production in India; **Cultivation of macroalgae- *Porphyra***; Nutritional value; importance of life cycle; methods of cultivation in advanced countries; Pillar, semi raft floating and open sea cultivation.

Unit-4: Applications of algae/products: Pollution indicators, treatment of waste water plants, heavy metal toxicity and phyco-remediation; Bio-fouling and biofuel production; Algal products as sources of nutraceuticals; Food colorants; Aquaculture feed; Therapeutics and cosmetics; Medicines; Dietary fibres from algae and uses; Biotechnological applications of algal silica and oils.

Practicals-32 Hrs

- 1) Study of fresh water planktonic forms in the lake samples.
- 2) Study of fresh water diatoms.
- 3) Chlorophyceae: *Ulva*, *Caulerpa*, *Halimeda*, *Acetabularia*.
- 4) Xanthophyceae: Mounting of *Botrydium* from soils.
- 5) Phaeophyceae: *Dictyota*, *Sargassum*, *Cystophyllum*.
- 6) Rhodophyceae: *Gracilaria*, *Gelidium*.
- 7) Cyanophyceae: *Microcystis*, *Nostoc*, *Spirulina*.
- 8) Estimation of carotene content in algal cells .
- 9) Culturing of microalgae: *Spirulina*/*Chlorella*/*Scenedesmus*/*Dunaliella*.
- 10) Applications of algal products: Agar, spirulina tablets/powder, beta-carotene, phycobiliproteins, triglycerides, Mycosporine like amino acids (MAA), diatom silica as nanoparticles.
- 11) Visit to National Institute of Oceanography, Goa.
- 12) Study of algal herbaria.

References

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- 11) Chu, W. 2012. Biotechnological Applications of Microalgae. *JeJSME* 6(1): S24-S37.

BOTANY: I SEMESTER - SOFT CORE 1.3
LICHENOLOGY AND MYCORRHIZAL TECHNOLOGY

Theory-32 Hrs

Unit-1: Introduction: Photobionts- identification, reproduction, and taxonomy of photobionts; Occurrence within lichens; Mycobionts- Lichenized versus nonlichenized fungi; Bryophilous and folicolous lichens; Thallus morphology and anatomy; Growth forms - crustose lichens, foliose lichens, fruticose lichens; Vegetative structures- Homoiomerous thallus, stratified thallus, cortex, epicortex, and epinecral layer, photobiont layer and medulla, lower cortex, Attachment organs and appendages; Cyphellae and pseudocyphellae; Cephalodia (Photosymbiodemes); Reproductive structures- sexual reproduction in lichen-forming ascomycetes; Mating systems, dikaryon formation, Ascomal ontogeny, Ascosporeogenesis; Ascus structure and function; Generative reproduction: ascoma, perithecia, apothecia, Thallinocarpia, Pycnoascocarpia, Hysterothecia, Asci, Basidioma; Vegetative reproduction- aposymbiotic propagules, symbiotic propagules; Systematics of lichenized fungi- History, classification and phylogeny.

Unit-2: Morphogenesis- Acquisition of a compatible photobiont; Recognition and specificity; Structural and functional aspects of the mycobiont–photobiont interface; Genotypes and phenotypes, growth patterns; Biochemistry and secondary metabolites- intracellular and extracellular products; The fungal origin of the secondary metabolites; Major categories of lichen products; Application to pharmacology and medicine; Harmful properties of lichen substances, lichens in perfume, lichens in dyeing; Stress physiology and the symbiosis- stress tolerance, limits to stress tolerance; harmful effects of stress, constitutive and inducible stress tolerance, evolution of stress tolerance in lichens; Modes of water uptake, light, temperature, carbon dioxide; The carbon economy of lichens.

Unit-3: Nitrogen, its metabolism and potential contribution to ecosystems, Methods of determination of nitrogen fixation; Nutrients- chemical and physical properties of nutrients and metals; Nutrient requirements, sources of nutrients, accumulation mechanisms, compartmentalization of elements within lichens; Metal toxicity, metal tolerance; Environmental role of lichens- dispersal, establishment, pedogenesis and biodeterioration; Community structure, succession, ecosystem dynamics; Animal and lichen interactions; Forest management, conservation, environmental monitoring; Lichen sensitivity to air pollution- lichens in relation to sulfur dioxide, oxidants and lichens, hydrogen fluoride and organopollutants.

Unit-IV: Mycorrhizal fungi: Introduction and classification; Types of mycorrhizas- Arbutoid mycorrhizas, ectomycorrhizas, vesicular arbuscular mycorrhizas or arbuscular mycorrhizas, ectendomycorrhizas, ericoid mycorrhizas, monotropoid mycorrhizas and orchid mycorrhizas; Phosphate solubilisation; Ecological significance of AM fungi; Importance of mycorrhiza in evolution of land plants; Role of mycorrhiza in agriculture, horticulture and forestry.

Practicals-32 Hrs

- 1-3) Survey of lichen vegetation in the study area: Frequency, density and abundance.
- 4) Determination of species richness and species diversity.
- 5) Isolation and maintenance of cyanobionts and phycobionts
- 6) Isolation and maintenance of mycobionts

- 7) Analysis of secondary metabolites of lichens.
- 8) Biological activity of secondary metabolites of the lichens.
- 9) Culture methods for lichens and lichen symbionts.
- 10) Root clearing and staining technique to study arbuscular mycorrhizal fungi.
- 11) Assessment of % root colonization of arbuscular mycorrhizal fungi.
- 12) Isolation and identification of arbuscular mycorrhizal fungi.

References:

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BOTANY: I- SEMESTER - SOFT CORE 1.4
PHYTOPATHOLOGY

Theory-32 Hrs

Unit-1: Concept of plant disease, Economic aspects of plant diseases; Types of plant diseases- Infectious diseases and non-infectious diseases; Causative agents of plant diseases; Angiospermic parasites; Development of plant pathology; Plant pathology in practice- Plant Clinic and Plant Doctor Concept; Parasitism and pathogenicity; Disease triangle; Infections and colonization; Weapons of plant pathogens; Effect of pathogen on physiology of host plant (photosynthesis, translocation and transpiration, respiration, permeability, transcription and translation).

Unit-2: Defence mechanisms in Plants- Pre-existing structural and chemical defences, induced structural and biochemical defences; Plant disease epidemiology- Elements of an epidemic and development of epidemics; Plant Disease forecasting; Genes and Diseases, Gene for gene concept, non-host resistance; Types of plant resistance to pathogens (Horizontal and Vertical Resistance); 'R' Genes and 'avr' genes; Genetics of virulence in pathogens and resistance in host plants; Breeding for disease resistance.

Unit-3: Management of Plant Diseases: Exclusion, eradication, cross protection, direct protection, integrated disease management, chemical methods of plant disease control; Biotechnological approaches to plant disease management; Gene silencing and disease control; Mechanism of gene silencing and control of viral diseases; Engineered resistance to viral, bacterial, fungal and insect diseases of crop plants.

Unit-4: Study of diseases of crop plants: Potato Spindle Tuber Disease, Tobacco Mosaic Disease, Sandal Spike Disease, Bacterial blight of Paddy, Citrus Canker, Late Blight of Potato, Downy Mildew of Maize, Blight of Paddy, Angular leaf spot of Cotton, Tikka disease of ground nut, Rust of coffee, Grain and Head smut of Sorghum. Leaf blight of Paddy, Blast of Paddy, Powdery mildew of cucurbits, Wilt of Tomato, Phloem Necrosis of Coffee, Root Knot of Disease of Mulberry and Vegetables; Non-parasitic diseases of plants; Seed-borne diseases.

Practicals-32 Hrs

- 1) Isolation of bacterial, fungal, and nematode plant pathogens of crop plants.
- 2) Study of mineral deficiency diseases of Tomato and French bean.
- 3) Estimation of foliar infection by Stover's method.
- 4) Study of spore germination.
- 5) Estimation of total phenols in diseased and healthy plant tissues.
- 6) Mycoflora analysis by Standard Blotter Method SBM/agar plating method.
- 7)-9) Study of Tobacco mosaic, Bacterial blight; Downy mildew of Maize; Powdery mildew of cucurbits; Grain smut of sorghum; Leaf rust of Coffee; Root Knot of Mulberry. Bunchy top of banana, Grassy shoot of sugar cane, Little leaf of Brinjal; Potato Spindle Tuber Disease (PSTVd)
- 10) Study of effect of pathogens on seed germination and vigour index.
- 11) Study of effect of fungicide on seed-borne pathogens.
- 12) Study of Fungal bio-control agents.

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BOTANY: II- SEMESTER- HARDCORE 2.1

REPRODUCTIVE BIOLOGY OF ANGIOSPERMS AND PLANT MORPHOGENESIS

Theory-32 Hrs

Unit-1: Reproductive Biology of Angiosperms: Historical overview; Contributions of P. Maheshwari; BM Johri; BGL Swamy to the development of embryology in India; Microsporogenesis and Microgametogenesis- wall layers and functions; Tapetum- types, concept of male germ unit and its significance; Pollen morphological features; Unusual features-pollen development in Cyperaceae, pollen embryo sac; Concept and scope of palynology.

Unit-2: Megasporogenesis and Megagametogenesis; Ovular structure and types; Development of monosporic, bisporic, tetrasporic and special types of embryo sacs; Ultra structure and nutrition of female gametophyte, concept of female germ unit and its significance; Fertilization- a general account, double fertilization, single fertilization, heterofertilization and polyspermy; Pollen recognition and rejection reactions - types, structures, methods to overcome incompatibility reactions; Endosperm- types, haustorial variations, ruminant and composite endosperm; Embryo- structure, development of monocot, dicot and grass embryo; Significance of embryonal suspensor; Experimental Embryology- scope and applications.

Unit-3: Plant Morphogenesis: Models of morphogenesis- comparison of plant v/s animal morphogenetic pathways: Embryo, *Arabidopsis thaliana*; Concepts- cell fate/ fate maps, gradients, stem cells in plants and their significance in development, polarity, symmetry, totipotency of cell types, pluripotency, plasticity, differentiation, redifferentiation, dedifferentiation and regeneration in *Acetabularia* and *Arabidopsis thaliana*.

Unit-4: Plant Growth and Development: Types, shoot apical meristems, root meristems; control of cell division in meristems; Quiescent center and meristeme de attente; *Arabidopsis*- vascular patterning and leaf development, abnormal growth; Cellular basis of growth- maintenance of cell shape; Cytoskeletal elements; Photomorphogenesis- definition, history, Hartmann's technique; Photoreceptors and photo morphogenesis, localization and properties; Effect of blue light-mediated photomorphogenesis with suitable examples.

Practicals-32 Hrs

Reproductive Biology of Angiosperms:

- 1) Study of microsporangium- slides: wall layers, tapetal types, two-celled and three-celled pollen; pollen tetrads.
- 2) Study of pollen germination: *Balsam*, *Delonix*, *Hibiscus* and *Peltaphorum*
- 3) Study of megasporangium-slides: female gametophyte development in *Penstemon*, *Xyris pauciflora*, 2, 4, 8-nucleate stages, mature embryo sac.
- 4) Endosperm mounting- *Cucumis sativus*, *Grevillia robusta* and *Croton sparsiflorus*
- 5) Embryo: Slides-monocot, dicot and grass embryo.
- 6) Embryo mounting : *Crotalaria*.

Plant Morphogenesis:

- 7) Study of stem cells in plants: SAM, RM.
- 8) Regeneration abilities of shoot apical meristems of dicots on media with combinations of growth regulators.
- 9) Study of totipotency in cell types: stomata, epidermal cells, stem and leaf explants on a tissue culture media.
- 10) Polarity in stem cuttings: *Pothos* spp.
- 11) Study of regeneration in succulents *Kalanchoe*, *Byrophyllum*.

- 12) Study of leaf galls of plants: *Pongamia pinnata* and *Achyranthes aspera*: Morphological observations and histology.
- 13) Study of *Arabidopsis thaliana* as a model plant.

References:

- 1) Johri, B. M. 1984. The embryology of Angiosperms. Springer Verlag.
- 2) Johri, B. M. 1982. The experimental embryology of vascular plants. Springer Verlag, New York.
- 3) Swamy, B.G.L. & Krishnamurthy, K. V. 1982. From flower to fruit: The embryology of angiosperms. Tata McGraw Hill Co. New Delhi.
- 4) Eames, 1961. Morphology of Angiosperms. McGraw Hill book Co., Inc., New York.
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- 12) Aloni, R. 1987. Differentiation of vascular tissues. Annu. Rev. Plant Physiol. 38:179- 219.
- 13) Raman, A. 2007. Insect induced plant galls of India; unresolved questions. Curr. Sci. 92 (6): 748-757.
- 14) Smith, H. 1975. Phytochrome and Photomorphogenesis- an introduction to the photocontrol of plant development. Mc Graw- Hill Book Co. (UK), Ltd.
- 15) Mohr, H. 1972. Lectures in photomorphogenesis. Springler- Vohrleg, Berlin, Germany.

BOTANY: II- SEMESTER - HARD CORE 2.2
CELL BIOLOGY AND GENETICS

Theory-32 Hrs

Unit-1: Bio Molecules and Membranes: Structure, composition of bio-molecules and their stabilizing interactions (carbohydrates, lipids, proteins and nucleic acids); Unit membrane structure and functions; Membrane proteins, membrane transport and the electrical properties; Intra-cellular compartments and protein sorting; Intracellular membrane traffic; Cytoskeletons.

Unit-2: Functions of Organelles: Cell wall, membranes, nucleus, mitochondria, Golgi bodies, lysosomes, spherosomes, peroxisomes, ribosomes, endoplasmic reticulum, Plastids, chloroplast, vacuoles and cytoskeleton; Cell cycle and mechanism of cell cycle regulations; A brief account of cell signalling, receptors, second messengers; General mechanism of signal transduction pathway; Programmed cell death in life cycles of plants.

Unit-3: Extensions of Mendelian Principles co-dominance, incomplete dominance, gene interactions, multiple alleles, lethal alleles, pleiotropy, penetrance and expressivity, polygenic inheritance, linkage and crossing over, sex linked inheritance, sex limited and influenced traits, genome imprinting, extra nuclear inheritance; **Concept of the gene-**classical-alleles, multiple alleles, pseudo-alleles, complementation test, experiments on rII locus and lozenge locus, modern- jumping genes, overlapping and genes within genes, split genes, nested genes, fusion genes; **Gene mapping methods-** linkage maps, tetrad analysis; Recombination in bacteria mapping genes in bacteria by interrupted mating technique, fine structure mapping, transduction and transformation mapping, mapping genes in Bacteriophages,

Unit-4: Sex Determination and Dosage Compensation: Chromosomal and genetic basis of sex determination; Mechanism of sex determination in *Melandrium*, *C. elegans*, *Drosophila* and humans, dosage compensation mechanisms in humans, *Drosophila* and *C. elegans*. **Transposable elements-** discovery in maize and bacteria, transposal elements in bacteria and bacteriophage, types and functions; Transposable elements in eukaryotes- Plants, *Drosophila* and Humans, mechanisms of transpositions; Transposable elements in research.

Practicals-32 Hrs

- 1) Determination of reducing sugars by Nelson-Somogyim's method.
- 2) Estimation of total soluble sugars by volumetric method.
- 3) Quantitative determination of free Amino acid content in germinating seeds.
- 4) Estimation of ascorbic acid in plant tissues.
- 5) Estimation of Phospholipids by TLC.
- 6) Slides/Charts/photos NP (Cytology Genetics and Embryology).
- 7) Study of mitosis in normal and induced root tips cells of Onion.
- 8) Study of meiosis in onion flower buds , translocation in *Rhoeo*.
- 9) Study of special chromosomes- B chromosomes, and sex chromosomes.
- 10) Determination of chiasma frequency in onion.
- 11) -12) To solve genetic problems on linkage, ordered and unordered tetrads.

References:

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- 2) Griffith, A.J.F. Gelbart, W.M. Muller, J.H. and Lewintin, R.C. 1999. Modern Genetic Analysis. W.H. Freeman and Co. New York.

- 3) Hartl. D. 1991. Basic Genetics. 2edn., Jones and Barlett Publishers Inc. Boston.
- 4) Fairbanks, D.J. and Anderson, W.R. 1999. Genetics the continuity of Life. Brooks's/Cole publishing Company, California.
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- 9) Strickberger, Monroe W. 2000. Evolution. 3rd Edn. Jones & Bartlett Publishers, Inc. 40 Tall Pine Drive Sudbury, MA 01776, USA.
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- 12) Kleinsmith, L.J. and Kish, V.M. 1995 .Principles of Cell and Molecular Biology 2nd edn. Harper Collins College Publishers, New York, USA.
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BOTANY: II SEMESTER HARD CORE 2.3
PLANT BREEDING AND EVOLUTIONARY BIOLOGY

Theory-32 Hrs

Unit-1: Introduction: Objective and role of plant breeding; Evolution of plant breeding, scope of plant breeding, sciences related to plant breeding, Vavilov's concept of origin of centers of origin of crop plants; Recent trends in plant breeding; **Breeding Methods**-plant introduction and acclimatization, domestication and agriculture, pure line, clonal, mass and progeny selections, recurrent selection, pedigree, bulk and back cross methods; Heterosis breeding synthetic and composite varieties; **Breeding Techniques**-Mutation breeding, polyploidy, hybridization, tissue culture techniques in crop improvement, protoplast fusion, electrophoration, electro-fusion, biolistics, somatic hybridization, transgenic plants (GMO's); The role of Gene technology in plant breeding.

Unit-2: Breeding for Specific Purposes: Breeding for disease resistance, insect resistance, drought and salinity, quality trait, multiple cropping systems, ideotype breeding, breeding for Adaptation; **Crop breeding and seed production**- Breeding field crops, seed production techniques, release of new varieties, intellectual property rights, computer application in plant breeding, crop breeding Institutes/Centers; Genetic resources and germplasm conservation; Scientific Plant breeding; Green revolution; The elite crop (Golden rice); Contributions of **Dr.**

M.S. Swaminathan, Dr. Norman E. Borlaug and N.I. Vavilov.

Unit-3: Nature of Evolution : The origin, theories of evolution of life, earth and the universe,; Conditions of the early earth, emergence of the first living cell, origin of prokaryotic and eukaryotic cells, life in the Palaeozoic, Mesozoic and Coenozoic era. **Development of Evolutionary thoughts;** Ecological context, before Darwin, Darwinism, Darwin's evolutionary theory, Neo – Darwinism, modern synthesis: **Fossil evidence of Ancient life,** fossilization,; Interpreting geological time scale and fossil records; Evidences from comparative, morphology, patterns of development, comparative physiology and biochemistry, biogeography, palaeontology, taxonomy, anatomy and embryology, plant and animal breeding; Evidence from changing earth and sea; Extinctions; Evolutionary ecology.

Unit-4: Natural Selection : Types of natural selection, selective forces, selection models, sexual selection, selection and non adaptive characters, Adaptive radiation, artificial selection, **Variation-** gene flow, genetic drift, gene mutation - Mendelian concept, chromosomal mutation, architectural changes in chromosomes; The Hardy – Weinberg law, polyploidy in plant evolution; Speciation and origin of higher categories -Types of speciation, models of speciation, pattern of speciation, isolating mechanism and species formation, signification of speciation; Molecular evolution.

Practicals-32 Hrs

- (1) Study of floral biology of crops - typical examples of self and cross pollinated plants.
- (2) Selfing and hybridization techniques - Bagging and emasculation.
- (3) Pollen viability: germination test and TTC test.
- (4) Studying of centre's of origin of cultivated crops - N.I. Vavilov Concept.
- (5) Mode of pollination study in different crops.
- (6) Identification of crop breeding institutes/ centers and logos.
- (7) Studying and identification of contributors of plant breeding - M.S. Swaminathan, N.I. Vavilov, Norman . E. Borlaug .
- (8) Study of contributions of scientists to evolutionary biology.
- (9)-12) Study of models and photographs related to evolution.

References

- 1) Atherly, A.G. Girton, J.R. Donald, J.R. 1999. The Science of Genetics. Saunders College Publishers. Fortworth.
- 2) Griffith, A.J.F., Gelbart, W.M. Muller, J.H. and Lewintin, R.C. 1999. Modern Genetic analysis. W.H. Freeman and co. New York.
- 3) Hartl. D. 1991. Basic Genetics. II edn. Jones and Barlett Publishers Inc. Boston.
- 4) Fairbanks, D.J. and W.R. Anderson. 1999. Genetics the continuity of life. Brooks's/Cole publishing company. California.
- 5) Brooker, R.J. 1999. Genetics –analysis and principles. Addison Wesley Longman Inc. California.
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- 11) Dodson E. O. and Dodson P. 1976. Evolution: Process and Product. 2nd Ed., D. Van Nostrand Company, 450 West 33rd Street, New York, N.Y. 10001
- 12) Chopra, V.L. 2000. Plant Breeding- theory and practices. Oxford and IBH Publishing Co. Pvt. Ltd., Oxford.
- 13) Chahal, G.S. and Gosal, S.S. 2002. Principles and procedures of Plant Breeding. Narosa Publishing House, New Delhi.

BOTANY: II- SEMESTER - SOFT CORE 2.1
PLANT ANATOMY AND HISTO-CHEMISTRY

Theory-32 Hrs

Unit-1: Plant Anatomy: Primary vegetative body of the plant; Anatomical features of leaf, stem and root (dicot and monocot); leaf of fern and gymnosperm; Structure of modified leaves- Kranz anatomy and C4 photosynthesis; Ultra-structure and chemistry of the cell wall; formation of the cell wall and its uses.

Unit-2: Anatomy of Vascular Tissue: Ultra structure and differentiation of xylem and phloem tissues; Apical meristems- shoot apex in Pteridophytes, Gymnosperms and Angiosperms, theories, root apical meristems.

Unit -3: Secondary Growth: Vascular cambium, secondary xylem of gymnosperms and dicots and secondary phloem of Gymnosperms and dicots; Periderm and bark; Anomalous secondary growth in monocots and climbers; Leaf ontogeny - Dicot- simple, compound, Monocot; Floral anatomy-flower parts, floral meristem, vascular system.

Unit-4: Plant Histochemistry: Tests for minerals, carbohydrates, lignins, polyphenols, proteins, lipids and nucleic acids; Study of instruments: (a) Camera lucida (b) Micrometry (c) Microtome. Principles of histo-chemical stains; Killing, fixing and staining of plant tissues; Double staining- TBA method.

Practicals-32 Hrs

- 1) Staining of xylem and phloem elements.
- 2) Study of anatomy of roots in: *Ficus, Musa, Dieffenbachia, Vanda*.
- 3) Study of anomalous secondary growth in the following examples: Stem of *Aristolochia, Nyctanthes, Pyrostegia, Peperomia, Tinospora, Achyranthes*.
- 4) Study of Ecological anatomy.
- 5) Study of Vasculature in floral organs.
- 6) Studying double staining technique.
- 7-11) Embedding: TBA method, embedding for electron microscope, Sectioning, Microtomes, whole mounts maceration.
- 12) Histochemical- PAS Test, Sudan black- lipids, Feulgen reaction – Nucleic acids.

References:

- 1) Abraham, F. 1982. Plant Anatomy. 3rd edn. Pergaon Press. Oxford.
- 2) Cariquist, S. 1967. Comparative Plant Anatomy- Holt Reinert and Winston, New York.
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- 9) Krishnamurthy, K. V. 1988. Methods in Plant Histochemistry. S. Viswanathan (Printers and Publishers) Pvt. Ltd. Madras.

BOTANY: II- SEMESTER - SOFT CORE 2.2
ETHNO-BOTANY AND INTELLECTUAL PROPERTY RIGHTS (IPR)

Theory - 32 Hrs

Unit-1: Ethno-botany: Introduction, concept, scope and objectives; Ethno-botany as an interdisciplinary science; The relevance of ethno-botany in the present context; Ethnic groups; Ethno-botany- Major and minor ethnic groups of India and their life styles; Forest Vs. ethnic groups; Plants in tribal life with reference to Magico-religious rituals and social customs; Sacred groves.

Unit-2: Methodology used in the study of Ethnobotany and Ethno pharmacology: Field work, Herbarium, Ancient Literature, Archaeological findings, temples and sacred places, protocols. Preliminary phyto-chemical analysis of ethno-botanical important medicinal plants.

Unit-3: Role of ethno-botany in modern Medicine with special examples; Medico-ethno-botanical Sources in India with special reference to Karnataka; Tribals Vs. Agriculture: Shifting, Podu and Jhum cultivation; Role of ethnic groups on surrounding environment; Crop genetic sources; Endangered taxa and forest management (participatory forest management); Ethno- botany as a tool to protect interests of ethnic groups; Sharing of wealth concept with few examples from India.

Unit-4: Study of Intellectual Property Rights – patents, trademark, geographical indication, copyright; IPR and Traditional Knowledge; Bio-piracy of traditional knowledge; Ethno botany and legal aspects; National and international organizations and treaty related to traditional knowledge – WIPO, TKDL, TRIPS, CBD, Nagoya protocol etc., Ethno botany as a source (recent) of already known drugs: a) *Withania* as an antioxidant and relaxant b) *Sarpagandha* in brain ailments c) *Becopa* and *Centella* in epilepsy and memory development in children d) *Phyllanthus fraternus* in diabetic and viral jaundice e) *Artemisia* as a powerful cerebral anti malarial agent and its possible use in tuberculosis.

Practicals-32 Hrs

- 1) Survey and collection important ethno botanical plants by using questionnaire and interview.
- 2) Preliminary phyto- chemical analysis of medicinal plants.
- 3) Study of biological functional properties of crude drugs – Anti microbial activity.
- 4) Study of methods of *in-situ* or *ex-situ* conservation of important medicinal plants.
- 5) Study of techniques used in Pharmacognosy – organoleptic, anatomy and chemical methods.
- 6) A visit to a Tribal area to conduct field work and collect ethno botanical information / data.
- 7) Listing of Crude drugs in Pansali shops (local crude drugs shops) and their identification (little known drugs only).
- 8) -12) Visit to nearby Western Ghats and Sacred Groves.

References:

- 1) Jain, S.K. 1995. Manual of Ethno-botany, Scientific Publishers, Jodhpur.
- 2) Jain, S.K. 1981. Glimpses of Indian. Ethno-botany, Oxford and I B H, New Delhi
- 3) S.K. Jain 1989. Methods and approaches in ethno-botany. (ed.) Society of ethno botanists, Lucknow, India.
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- 5) Colton C.M. 1997. Ethno botany – Principles and applications. John Wiley and sons –
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- 8) Faulks, P.J. 1958. An introduction to Ethno-botany, Moredale pub. Ltd. London

BOTANY: II- SEMESTER - SOFT CORE 2.3
ECONOMIC BOTANY

Theory -32 Hrs

Unit- 1: Economic Botany: The origin of cultivated plants and Agriculture; The future role of plants in relation to mankind; Introduction to Green revolution; Study of origin, distribution, cultivation and utility of the useful parts of the following- - rice, wheat, maize, barley, sorghum and millets; Red gram, green gram, black gram, horse gram, pea, cow pea, bengal gram; Oil Yielding plants- sunflower, safflower, groundnut, linseed, rape seed; A brief account of economically important horticultural and floricultural plants.

Unit- 2: Economic Botany: Study and utility of the useful parts of the following- Sugar yielding plants- sugar cane and sweet potato, sugar beet and *Stevia*; Spices and condiments - ginger, turmeric, cardamom, cinnamon, clove, saffron, all spice, black pepper, nutmeg, red pepper, coriander, cumin, fennel and *Vanilla*.

Unit -3: Economic Botany Study and utility of the useful parts of the following- fibre- cotton, jute, flax, hemp, Sunn hemp, China grass, coconut and Kapok; Timber yielding plants- *Tectona* and *Dalbergia*; Dyes- indigo, henna; Masticatories and fumitories-areca nut, betel leaf, tobacco; rubber- Para rubber and other substitutes; Gums- Gum Arabic, Karaya gum.

Unit-4: Medicinal Botany: Scope and importance of medicinal plants; Indigenous medicinal Sciences; Important medicinal plants and their uses; Major exporters and importers of traditional medicinal plants and plant products; Application of natural products to certain diseases- jaundice, cardiac, infertility, diabetics, blood pressure and skin diseases; Poisonous plants.

Practicals-32 Hrs

- 1) Utility, uses and economic importance of cereals and millets.
- 2) Utility, uses and economic importance of horticultural and floricultural plants
- 3) Utility, uses and economic importance of pulses and oil yielding crops.
- 4) Utility, uses and economic importance of sugar yielding crops.
- 5) Utility, uses and economic importance of spice and condiments.
- 6) Utility, uses and economic importance of fiber and timber yielding plants.
- 7) Utility, uses and economic importance of dye, rubber and gum yielding plants
- 8) Utility, uses and economic importance of masticatories and fumitories
- 9) -12) Study of medicinal and poisonous plants.

References:

- 1) Hill, A.F. 1952. Economic Botany, TataMcGraw Hill, New Delhi.
- 2) Kochhar, S.L. 1998. Economic Botany of Tropics, Macmillan India Publishers, New Delhi.
- 4) Pandey, B.P. 2000. Economic Botany. S. Chand & Company, New Delhi.
- 5) Pandey, S.N. and Chandha, A. 1999. Economic Botany. Vikas Publishing House Pvt. Ltd. New Delhi.

BOTANY: II SEMESTER- OPEN ELECTIVE 2.1
MEDICINAL PLANTS

Theory-32 Hrs

Unit-1: Medicinal Plants: History, scope and importance of medicinal plants; Indigenous medicinal sciences; History, origin, panchamahabhutas, saptadhatu and tridosha concept, Rasayana, plants used in ayurvedic treatments, Siddha: Origin of Siddha medicinal systems, Basis of Siddha system, plants used in Siddha medicine. Unani: History, concept: Umoor-e-tabiya, tumors treatments/ therapy, polyherbal formulations.

Unit-2: Medicinal Plants Conservation: Conservation of endangered and endemic medicinal plants; Endemic and endangered medicinal plants; Red list criteria; *In-situ* conservation- biosphere reserves, sacred groves, national parks; *Ex situ* conservation- botanic gardens, ethno medicinal plant gardens; Propagation of medicinal plants - objectives of the nursery, its classification, important components of a nursery, sowing, pricking, use of green house for nursery production, propagation through cuttings, layering, grafting and budding.

Unit - 3: Funding for Cultivation of Medicinal Plants: Sources of financial aids for medicinal plant cultivation: Aims and objectives, Functions and activities of the board, Schemes and Projects for Financial assistance, Funding of projects; Procedure for processing project proposal for approval, Implementation and monitoring.

Unit- 4: Ethno botany and Folk medicines: Definition; Ethno botany in India: Methods to study ethno botany; Applications of Ethno botany: National interacts. Ethno medicine. Application of natural products to certain diseases- Jaundice, cardiac, infertility, diabetics, Blood pressure and skin diseases. Brief introduction to poisonous plants.

References:

- 1) Trivedi, P. C. 2006. Medicinal Plants: Ethnobotanical Approach, Agrobios, India.
- 2) Purohit and Vyas, 2008. Medicinal Plant Cultivation: A Scientific Approach, 2nd edn.
- 3) Agrobios, India.
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BOTANY: III- SEMESTER - HARD CORE 3.1
BIOCHEMISTRY AND PLANT PHYSIOLOGY

Theory -32 Hrs

Unit-1: Biochemistry- Brief account of plant structural and functional molecules- carbohydrates, proteins, lipids and nucleic acids; classification, structural and functional properties of bio molecules; Biochemistry of cell membranes; **Lipids**-building and storage molecules, classification and significance; **Proteins**- classification, structure- primary, secondary, tertiary and quaternary structure; properties of proteins; **Enzymes**- Nomenclature, nature and properties of enzymes, active sites, co-enzymes, kinetics of enzyme action, catalysis, specificity and inhibition, allosteric enzymes, ribozyme and abzyme.

Unit-2:Solute transport: Transport of solutes across the membranes Transmembrane proteins, Transport of ions, solutes and macro-molecules, Mechanism of translocations in phloem; Role played in signal transduction pathway stomatal physiology; **Phytosynthesis in higher plants** (i) Photophosphorylation - Calvin cycle; **Photorespiration** - C4 – Pathway, CAM in plants; Oxidative Phosphorylations; Glycolysis -TCA – Cycle and terminal oxidation.

Unit-3: Plant Hormones- plant hormones-discovery, biosynthesis, metabolism, transport and physiological effects of plant hormones and their applications; **Nitrogen metabolism** -(i) Molecular mechanism of N₂ fixation (ii) Biosynthesis of amino acids (iii) Assimilation of nitrate and ammonium; **Lipid metabolism**- fats and oils biosynthesis and oxidation of lipids; Physiology of seed germination and flowering.

Unit -4: Stress Physiology: Water deficit and its physiological consequences; Drought tolerance mechanisms, Salinity stress and plant responses. Heat stress and heat shock proteins; Metal toxicity in plants. Biotic stress, HR and SAR mechanisms; **Mineral nutrition**- in plants and deficiency diseases; **Plant development**- physiology of flowering; **Phytochrome**- photochemical and biochemical properties of phytochrome; Concept of photoperiodism and vernalization and its influence on flowering;

Practicals-32 Hrs

- 1) Estimation of protein by Lowry's method
- 2) Determination of water potential of tissue by plasmolytic method
- 3) Determination of water potential by Gravimetric method
- 4) Quantitative estimation of chlorophyll a, chlorophyll b and total chlorophyll in plant tissue
- 5) Determination of diurnal fluctuation of acid content of CAM plants (TAN)
- 6) Determination of temperature quotient (Q₁₀) of water uptake
- 7) Separation of chlorophyll pigments/Anthocyanin by TLC
- 8) Protein analysis by SDS PAGE method.
- 9) Estimation of Alpha-amylase activity in germinating seedling.
- 10) Silver staining of proteins.
- 11-12) Visit to Molecular Biology Laboratories.

References:

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- 13) Zimmermann, M.H., and Milburn, J.A. Transport in Plants. 1. Phloem transport (Encyclopedia of Plant Physiology. New Series Vol. 1), Springer, New York.
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- 15) Hopkins, W.G. 1995. Introduction to Plant Physiology, John Wiley & Sons. Inc., NewYork, USA.
- 16) Moore, T.C. 1989. Biochemistry and Physiology of Plant Hormones. Springer Verlag, New York, USA.
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BOTANY: III- SEMESTER - HARD CORE 3.2
MOLECULAR BIOLOGY

Theory-32 Hrs

Unit-1: Organization of chromosomes and genes in prokaryotes and eukaryotes - Operon, interrupted genes, gene families, unique and repetitive DNA, heterochromatin, euchromatin, transposons, mitochondrial and chloroplast genome organization, Transposable elements in prokaryotes and eukaryotes, genetic and evolutionary significance, **DNA replication**- patterns, Messelson and Stahl's and Taylor's experiment, enzymes of replication, mechanism of DNA replication in prokaryotes and Eukaryotes, proof reading and error correction mechanisms.

Unit-2: Molecular mechanism of mutation, repair and recombination:- Mutation-DNA damage by spontaneous mutations, physical and chemical mutagens and their molecular mechanisms, **Repair mechanisms**- direct reversal of damage, base and excision repair, recombinational repair, SOS repair, translation repair synthesis, transcription coupled repair, **Recombination**- homologous recombination, models of recombination, mechanisms, protein machinery of homologous recombination, genetic consequence of homologous recombination, gene conversion, site specific recombination, mechanism and biological significance, non homologous recombination- transposition, molecular mechanisms of transposition- conservative, replicative and retro-transposition.

Unit-3: RNA synthesis, processing and translation: transcription activators and repressors, promoters, RNA polymerases and transcription factors, mechanism of transcription in prokaryotes and eukaryotes, **RNA processing**- capping, polyadenylation, splicing, alternative splicing, RNA editing, exon shuffling and RNA transport, **Translation and processing**- ribosomes, tRNA aminoacylation, aminoacyl tRNA synthetase, genetic code, wobble hypothesis, deciphering of the code, translation mechanism , translation proof reading, translation inhibitors and post translational modifications.

Unit-4: Regulation of gene expression in Prokaryotes: Operon concept, regulation at transcription initiation- lac and trp operon control, regulation of lytic and lysogenic cycles in lambda phage, regulation beyond transcription initiation-premature termination- trp operon, ribosomal proteins as translational repressors, riboswitches, **Regulation of gene expression in eukaryotes**-transcription activators and repressors, regulation after transcription initiation- alternative splicing, translational control in ferritin and transferrin mRNA, RNA interference, role of chromatin in regulation of gene expression and gene silencing.

Practicals-32 Hrs

- 1) Isolation of DNA from CTAB method.
- 2) Isolation of DNA from Onion.
- 3) Isolation of DNA from mulberry leaves.
- 4) Estimation of DNA by DPA method.
- 5) Extraction of RNA by trizol/ phenol-chloroform methods.
- 6) Estimation of proteins by Biuret method.
- 7) Estimation of protein by Bradford method.
- 8) Determination of T_m value of DNA.
- 9-12) Photo graphs/ charts related to molecular biology/Molecular Biologists.

References:

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- 6) Gunning.B.E.S. and Steer, M.W.1996. Plant Cell Biology; Structure and Function. Jones and Bartlett Publishers, Boston, Massachusetts.
- 7) Harris,Nand Oparka, K.J. 1994. Plant Cell Biology A Practical Approach. IRL Press, Oxford University Press, U.K.
- 8) F.M. Ausubel, R.Brent, R.E. Kingston, D.D. Moore, J.G. Seidman, J.A. Smith, K. Struhl, (Current Edition) (2005). Current Protocols in Molecular Biology.
- 9) B.B. Buchanan, W.Gruissem and R.L. Jones . USA (2000) .Biochemistry and Molecular Biology of Plants. Ed. ASPP Press.
- 10) T.A. Brown, 2000. Essential of Molecular Biology, Vol-I & 2 Oxford University Press.
- 11) James D. Watson, Tania, A. Baker, Stephen, P. Bell, Alexander ,Gannm, Michael Levine.2004. Molecular Biology of the gene. 5th Edition, Pearson Education.Philip M Gilmartin and Chris.
- 12) Bowle.2002. Molecular Biology of Plants. Vol 1 & 2 Oxford University Press.

BOTANY: III-SEMESTER - HARD CORE 3.3
PLANT BIOTECHNOLOGY

Theory-32 Hrs

Unit-1: Plant Tissue Culture: Scope and importance of plant tissue culture - Media composition and types, hormones and growth regulators, explants for organogenesis; Micro propagation, embryo and endosperm culture, somatic embryogenesis, variation and cell line selection, androgenesis and microspore culture, significance of haploids, diploidization and bulbosum technique; Cryopreservation, germplasm collection; Somatic Hybrids- Isolation and protoplast culture and somatic hybridization and its significance, Synthetic seed production and somaclonal variations.

Unit-2: Genetic Engineering: Milestones in plant recombinant DNA technology; Importance of gene manipulation in future perspectives; **Tools in Genetic Engineering-** Enzymes in genetic engineering - restriction endonucleases, types and their actions, other DNA modifying enzymes; Cloning vectors- plasmids isolation and purification - Ti Plasmid, pBR322, pUC-series. Phage vectors-M13 phage vectors, Cosmids -types, phasmids or phagemids, shuttle vectors-types; YAC and BAC vectors, Lambda phage vectors, Lambda phage DNA as a vectors; Cloning vectors and expression vectors; Vectors for plant cells; Vectors for animal cells, baculovirus vectors- adenoviruses, retroviruses, transposons as vectors, Synthetic construction of vectors.

Unit 3: Applications of Genetic Engineering for pest, disease and stress tolerance: The genetic manipulation of herbicide resistance with suitable examples; The genetic manipulation of pest and disease resistance with suitable examples; Transgenic approaches to viral and bacterial disease resistance. Engineering for stress tolerance and Metabolic Engineering of Plants; Future prospects for GM crops.

Unit 4: Biofertilizers: Preparation and applications of biofertilizers such as Rhizobium, Azotobacter, Blue Green Algae and VAM. Single Cell proteins (SCP): Health benefits and advantages of single cell proteins- *Spirulina*. Biofuels: Ethanol and Biofuel production from plants. Mushroom cultivation and its advantages. Bioremediation: Phytoremediation; Biodegradation, Xenobiotics. Biotechnology of medicinal and aromatic plants for human welfare.

Practicals-32 Hrs

- 1) Preparation of plant tissue culture media and types.
- 2) Organ culture (Shoot tip, nodal and leaf culture) for callus Initiation and regeneration.
- 3) Anther culture for the production of haploids.
- 4) Suspension culture and production, separation and estimation of secondary metabolites.
- 5) Encapsulation of somatic embryos and production of Synthetic seed.
- 6) Extraction of secondary metabolites using Soxhlet extractor and Identification of In vitro secondary metabolites-alkaloids, steroids and flavonoids.
- 7) Restriction digestion of plasmid and genomic DNA and gel electrophoresis.

- 8) Isolation of genomic DNA from bacteria/plants and purification by agarose gel electrophoresis.
- 9) Restriction analysis of plasmids, gel purification of DNA, small and large scale purification of plasmids.
- 10) Preparation of competent *E. coli* cells. Bacterial transformation and recovery of plasmid clones.
- 11) Gene cloning in plasmids, analysis of recombinant plasmids.
- 12) DNA amplification by PCR, RT-PCR, Real Time PCR.
- 13) Analysis of DNA and RNA and Protein by Southern, Northern and Western blotting.
- 14) Primer design for PCR.

References:

- 1) Slater, N. Scott and M. Fowler. Plant Biotechnology 2003: The genetic manipulation of plants. Oxford University Press, Oxford.
- 2) Plant Biotechnology. 2000. J.H. Hammond, P. Mcgarvey, and V. Yusibov (eds). Springer Verlag, Heidelberg.
- 3) Text Book of Biotechnology. 2004. H.K. Das (ed). Wiley India Pvt. Ltd., New Delhi.
- 4) Plant Biotechnology -The Genetic Manipulation of Plants, Adrian Slater, Nigel Scott and Mark Flower, Oxford University Press, (2000).
- 5) Plant Genetic Transformation and Gene Expression by (eds) J.Draper *et.al.* Blackwell Scientific Publications, Oxford (1988).
- 6) Reinert, J. 1982. Plant Cell and Tissue Culture: A Laboratory Manual. Narosa Publishing House, New Delhi.
- 7) Chawla H.S., 2009, Plant Biotechnology. Oxford and IBH Publishing Co. Pvt. Ltd, New Delhi.
- 8) Bhojwani, S.S. and Razdan, M.K. 2004. Plant Tissue Culture: Theory and practice. Elsevier Science Publishers, New York, USA.
- 9) PUROHIT S. D., 2012. Introduction To Plant Cell Tissue And Organ Culture PHI Learning Pvt. Ltd., New Delhi
- 10) Roberta, H. Smith, 2012. Plant Tissue Culture: Techniques and Experiments 3 edition. Academic Press; US.

BOTANY: III- SEMESTER- SOFT CORE 3.1
MOLECULAR GENETICS OF PLANTS

Theory-32 Hrs

Unit-1: Plants as genetic tools in Biology: *Arabidopsis*, *Rice*, *Maize*, *Saccharomyces*; Genome organization in plants; *Arabidopsis thaliana*- an experimental model for understanding plant development and functions; Plant genes and regulation; nucleus and chromatin organization; Histones and histone modifications; DNA packaging, organization and types of DNA sequences; functional and non- functional sequences, organization of plant nuclear genes, plastid genes and mitochondrial genes.

Unit-2: Genes responding to hormones, phytochrome, responses to abiotic stresses; Genes induced by water stress and freezing stress; Genes involved in photosynthesis and nitrogen fixation and their regulation; Molecular development of leaf and flower - ABC and revised model of flower development; Genes involved in fertilization, seed development, embryo development.

Unit-3: Genetics of *Agrobacterium*: Biology and genetics of *Agrobacterium tumefaciens*; The Ti- plasmid, *Vir* genes and expression, Mechanism of T-DNA transfer and integration; Basic features of vectors for plant transformation; Proteomics, genomics and bioinformatics; Structural and functional genomics, comparative genomics - biochemical, evolutionary, physiological and phylogenomics; Tools to study functional genomics.

Unit-4: Proteomics- functional and comparative proteomics; Protein distribution, characterization and identification, differential display proteomics, detection of functional linkages; Pharmacogenomics; Bioinformatics- tools of bioinformatics, data bases and data base management, bioinformatics in taxonomy, biodiversity, agriculture; Bioinformatics in drug design and drug discovery.

Practicals-32 Hrs

- 1) *Arabidopsis thaliana*- study of plant system and its biology.
- 2) *Arabidopsis* RNA extraction (total and polysomal) for Northern blotting.
- 3) Expression of foreign genes in plant cells through *Agrobacterium tumefaciens* (Chart)
- 4) Production of tobacco transgenic plants and assay for the introduced transgenic (Chart)
- 5) Co-cultivation of tobacco *Agrobacterium tumefaciens*
- 6) -12) Learning gene bank formats- EMBL format, FASTA format, Swiss- PROT, Ex PASy

References:

- 1) Buchmann, B.B., Gruissem, W., and Jones, R.L. 2000. Biochemistry and Molecular Biology of Plants. ASPP Press, USA.
- 2) Ausubel, F.M., Brent, R., Kingston, R.E., Moore, D.D., Seidman, J.G., Smith, J.A., and Struhl, K. 2005. Current protocols in molecular biology. Current Edition.
- 3) Brown, T.A. 2000. Essentials of Molecular Biology. Vol. I & II, Oxford University Press.
- 4) Potrykus, I., and Spangenberg, G. 1995. Gene transfer to plants. Springer, Berlin, Heidelberg.
- 5) Watson, J.D., and Baker, T.A., Bell, S.P. Gannm, A. and Levine, M. 2004. Molecular Biology of Genes. 5th edn., Pearson Education.
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- 11) Draper, J. 1988. Plant Genetic Transformation and Gene Expression. Blackwell Scientific Publications, Oxford.
- 12) Old, R.W., and Primrose, S.B. 2004. Principles of Gene Manipulation. An introduction to Genetic Engineering. 5th Edition, Blackwell Science Publications.

BOTANY: IV- SEMESTER- SOFT CORE 3.2
MOLECULAR PLANT PATHOLOGY

Theory-32 Hrs

Unit-1: Concepts and scope of physiological and molecular plant pathology; Molecular approaches to plant disease diagnosis; Nucleic acid based probes for detection of plant pathogens including non-culturable organisms; **Pathogenicity and Disease Development-**factors; induced resistance, virulence and pathogenicity factors; Plant-pathogen interactions with emphasis on incompatible interactions and induced resistance.

Unit -2: Pathogenesis: Necrogenic plant pathogenic bacteria with emphasis on hrp and avr genes and virulence factors; Fungal plant pathogens with emphasis on virulence and pathogenicity factors; Plant viruses with emphasis on virus replication, virus transport in plants and control of plant viruses with transgenic plants; **Signal Transduction-** recognition of the pathogen by the host, transmission of the alarm signal to the host defense providers; Necrotic defense reaction, defense through hypersensitive response; Molecular basis of induced biochemical reaction; Local and systemic acquired resistance (SAR).

Unit-3:Genetics of Plant Diseases and Resistance: Genes and diseases; physiological specialization among plant pathogens; Variability in viruses, bacteria and fungi; Levels of variability in pathogens and loss of virulence in plant pathogens; Genetics of virulence in pathogens and of resistance in host plants; Molecular plant breeding for disease resistance.

Unit-4: Genetics and molecular basis of host-pathogen interaction: Evolution of parasitism; genetics on host-pathogen interaction; Gene for gene relationship; Criteria for gene for gene type relationship; Molecular basis of host pathogen interaction; Host-parasite-interaction. **Biotechnological methods of plant disease management;** Genetic engineering and crop protection; Cross protection; Gene silencing and disease control- mechanism of gene silencing and control of viral diseases; Engineered resistance to viral, bacterial, fungal and insect diseases of crop plants.

Practicals-32 Hrs

- 1-2) Testing hypersensitivity reaction on *Nicotiana and Bajra*.
 - 3) Estimation of lipoxygenase in diseased and healthy plants.
 - 4) Estimation of polyphenols in diseased and healthy plants.
 - 5-7) Studying systemic acquired resistance in crop plants.
 - 8) Genetic testing of disease resistance in plants.
 - 9-11) Molecular detection of viruses, Mycoplasma, fungi and bacteria from infected plants.
 - 12) In-vitro testing of pathogen virulence.
- Visit to agricultural research station to study diseases on different crop plants.

References:

- 1) Singh, R. S. (1973). Plant Disease. Oxford and IBH Pub.Co. New Delhi.
- 2) Agrios, G. N. (1994). Plant Pathology 2nd Edn. Academic Press NY.
- 3) Johnston A and Both, C. 1983-Plant Pathologists Pocket-book. 2nd Edn. Commonwealth Mycological Institute, Oxford and IBH Pub. Co. Calcutta.
- 5) Rangaswamy G and Mahadevan A 2002. Diseases of crop plants in India, Prentice Hall of India Pvt. Ltd. New Delhi.
- 6) Mehrotra, R. S.1983-Plant Pathology Tata Mc. Graw Hill Pub. Co. Ltd., New Delhi.
- 7) Vidhyasekaran, P. 2004. Encyclopedia of Plant Pathology.Viva Books Pvt.Ltd. New Delhi.

BOTANY: III SEMESTER- SOFT CORE 3.3
PLANT PROPAGATION AND PLANT BREEDING

Theory-32 Hrs

Unit-1: Plant Propagation: History, scope and importance of plant propagation; Propagation structures with reference to green house equipment and media; Seed propagation and vegetative propagation; Propagation by cuttings; Biology and techniques of grafting; Techniques of budding; Layering and its natural modifications; Propagation by specialized stems and roots; Micro propagation – techniques and applications in forestry and horticulture; Limitations and applications of vegetative propagation; Propagation methods of some selected plants – Citrus, Grape, Mango, Mulberry, Hibiscus, Rose, Croton, Eucalyptus.

Unit-2: Plant Breeding: History of plant breeding, objectives of plant breeding, salient achievements of plant breeding; Centres of origin of crop plants, Exploration and collection of plant genetic resources, evaluation of germplasm collection, documentation, conservation of plant genetic resources, utilization of genetic resources; The theory of pure line selection – Genetic basis, sources of genetic variation in pure lines, the land variety (races); **Mendelian experiments of plant hybridization;** Quantitative Inheritance; Applications of biometrical genetics in plant breeding.

Unit-3: Plant Breeding: Types of plant breeding; Fertility regulating mechanisms - manual or mechanical control, genetic control, incompatibility, male sterility, genetic engineering for male sterility, chemical control, genetic basis of heterosis; Synthetic and composite varieties -genetic basis, procedure for developing synthetic and composite varieties - genetic basis, procedure for developing synthetic varieties; Breeding for resistance to disease and insect pests.

Unit - 4 :Mutation Breeding: Significance of induced mutations in plant breeding; Polyploidy in plant breeding- types of polyploids, induction of polyploidy, phenotypic effects of polyploidy, significance of polyploids; Tissue culture in crop improvement; Molecular approaches to crop improvement- probes, gel electrophoration, electrofusion, biolistics, gene cloning, transgenic plants (GMO's), molecular markers, construction of genetic maps, application of DNA makers in plant breeding, the role of gene technology in plant breeding; Crop breeding Institutes/Centers, Molecular biology in relation to intellectual property rights.

Practicals-32 Hrs

- 1) Study of types of vegetative propagation: Cutting, Grafting, budding, layering.
- 2) Study of propagation by modified stems and modified roots.
- 3) Preparation of media, explants, culture, initiation of shoot multiplication.
- 4) Pot and green house implants (demonstration) (5) Studying of floral biology.
- 6) Hybridization techniques - bagging and emasculation.
- 7) Pollen viability test : Seed germination test, TTC test.
- 8) Mode of pollination study in different crops.
- 9) Visit to crop breeding stations/institutes / centres.
- 10) Estimation of protein quality, Amino acid Analysis and determination of oil and fatty acids.
- 11) Observation of colour and conditions of mature anthers in different crops.
- 12) Identification of and studying of important plant breeders.

References:

- 1) Abbottt, A.J. and Atkin, R.K. eds. 1987. Improving vegetatively propagated crops.

Academic press, New York.

- 2) Bose, T.K., Sadhu, M.K., & Das, P., 1986. Propagation of Tropical and Subtropical Horticultural crops, Nowya Prakash, Calcutta.
- 4) Hartmann, H.T., Kester E.D., Davis, F.T., and Geneve, R.L. 1997. Plant propagation. Principles and practices. Prentice Hall of India Private Limited, New Delhi.
- 5) Krishnamurthy. H.M. 1981. Plant Growth substances including application in Agriculture.
- 6) Pierik, L.M. 1987. In vitro culture of Higher plants Murtinus Nijhoff pub. Dordrecht.
- 7) Razdan, M.K. 1994. An Introduction to Plant tissue culture, Oxford and IBH Pub. Co., PVT. Ltd., Bombay and Calcutta.
8. Mac Donald, B. 1987. Practical woody plant propagation for nursery growers. Portland, OR: Timber press.
9. Sadhu, M.K. 1989. Plant propagation Wiley eastern Ltd. N. Delhi.

BOTANY: III SEMESTER SOFT CORE 3.4
PHYTOCHEMISTRY AND HERBAL TECHNOLOGY

Theory-32 Hrs

Unit-1: Phytochemistry: Scope of phytochemistry, plants as source of chemical compounds, primary and secondary metabolites and its applications; Definition, source of herbal raw materials, identification, authentication, standardization of medicinal plants as per WHO guidelines and different herbal pharmacopoeias; Natural pigments, natural products as markers for new drug discovery.

Unit-2: Extraction, isolation and purification of phytochemicals: Selection of plant samples, processing and storage of samples for extraction; Factors influencing the choice of extraction, principles of extraction methods, infusion, decoction, digestion, maceration, percolation, solvent extraction, fluid extraction, ultrasound, microwave assisted extraction, advantage and disadvantage involved in each method; Isolation of selected primary and secondary metabolites – amino acids, proteins and carbohydrate; Phenolics, flavonoids, alkaloids, lipids, oils, terpenes and saponins; Purification techniques for primary and secondary metabolites – solvent-solvent fractionation and chromatography techniques.

Unit-3: Characterisation of Phytochemicals: Preliminary, qualitative and quantitative techniques – paper chromatography, thin layer chromatography, column chromatography-HPLC, GC (qualitative and quantitative), colour reactions for amino acids, sugars, phenolics, flavonoids, alkaloids, terpenes, saponins, oils, lipids; Spectroscopic estimations/gravimetric determination of total sugars, amino acids, proteins, phenolics, flavonoids, alkaloids, terpenes, saponins, oils, lipids; Characterisation using spectroscopic techniques - UV/VIS, FTIR, DSC (differential scanning calorimeter), NMR, MS, MALDI. XRD – single crystal and powder.

Unit-4: Standardisation and Validation of Photochemical: Quality determination of herbal drugs; Role of processing methods and storage conditions on quality of drugs; Standardisation parameters- impurity limit, ash content, extractable matter, moisture content, other phytochemicals, microbial contaminants, pesticides; Validation of drug – guidelines, limit of detection and quantification of impurities, organoleptic properties, physical, chemical, biological characteristics, stability testing, storage conditions and packing system/unit.

Practicals-32 Hrs

- 1) Survey and collection of medicinal plants for analysis.
- 2) Selection of plant part, processing and storage of samples for further analysis.
- 3) Extraction methods - aqueous and sequential solvent extraction of compounds.
- 4) Preliminary phytochemical analysis of active principles from the extracts.
- 5) Antibacterial/antifungal activity of crude /active principles
- 6) Identification of secondary metabolites using TLC- phenolics, flavonoids, alkaloids, terpenes, saponins etc.
- 7) Column chromatographic separation of active principles.
- 8) Characterisation of active principle using spectroscopy, HPLC, GCMS, LCMS, FTIR, and MALDI TOF.
- 9) -12) Submission of report on TEN important curative principles of Indian medicinal plants.

References:

- 1) Braithwaite, A. and Smith, F.J. 1996. Chromatographic Methods. 5th edn., Blackie Academic & Professional, London.
- 2) Bourne, U.K. Kokate, Purohit, C.K. and Gokhale S.B. 1983. Pharmacognosy. Nivali Prakashan Publication.
- 3) Braithwaite, A. and Smith, F. J. 1996. Chromatographic Methods. 5th edn. Blackie Academic & Professional, London.
- 4) Sadasivam. S. and A. Manickam, 0000. Bio Chemical methods 2ndedn. New Age International Pvt Ltd. New Delhi.
- 5) Harborne, J.B. 1984. Phytochemical Methods, 2ndedn. Chapman and Hall, London. Harborne J.B., 1973. Phytochemical methods a guide to modern techniques of plants analysis. Chapman and Hall Ltd. London.

BOTANY: III SEMESTER- OPEN ELECTIVE 3.1
PLANT PROPAGATION TECHNIQUES

Theory-32 Hrs

Unit-1: History, scope and importance of plant propagation; Propagation structures with reference to green house equipment and media; Seed propagation – the development of seeds, techniques of seed production and handling principles and media.

Unit-2: Vegetative propagation: Techniques of propagation by cuttings; stem cuttings – hard wood, semi hard wood, soft wood and herbaceous, leaf cuttings, leaf bud cuttings, root cuttings; Biology and techniques of grafting: Whip and tongue, wedge and cleft, bark, side grafting, approach.

Unit-3: Techniques of budding: T- budding patch budding, chip budding, ring budding; Layering and its natural modifications- simple layering, tip layering, mound or stool layering, air layering, compound or serpentine layering and trench layering; Propagation by specialized stems and roots.

Unit- 4: Micro propagation – techniques and applications in forestry and horticulture; Advantage, limitations and applications of vegetative propagation, **Somaclonal variations;** Propagation methods of some selected plants – Citrus, gape, mango, mulberry, hibiscus, rose, Croton, Eucalyptus.

References:

- 1) Abbott, A.J. and Atkin, R.K. (eds.) 1987. Improving vegetatively propagated crops. Academic press, New York.
- 2) Bose, T.K., Sadhu, M.K., and Das, P., 1986. Propagation of Tropical and Subtropical Horticultural crops, Nowya Prakash, Calcutta.
- 3) Hartmann and Kester, 1983. Plant propagation
- 4) Hartmann, H.T., Kester E.D., Davis, F.T. and Geneve, R.L. 1997. Plant propagation. Principles and practices. Prentice Hall of India Private Limited, New Delhi.
- 5) Krishnamurthy. H.M. 1981. Plant Growth substances including application in Agriculture.
- 6) L.M. Pierik 1987. In vitro culture of Higher plants Murtinus Nijhoff pub. Dordrecht.
- 7) M.K. Razdan 1994. An Introduction to Plant tissue culture, Oxford and IBH Pub. Co., PVT. Ltd., Bombay and Calcutta.
- 8) Mac Donald, B. 1987. Practical woody plant propagation for nursery growers. Portland, OR: Timber press.
- 9) Sadhu, M.K. 1989. Plant propagation Wiley eastern Ltd. N. Delhi.

BOTANY: IV- SEMESTER- HARD CORE 4.1
ECOLOGY, CONSERVATION BIOLOGY AND PHYTOGEOGRAPHY

Theory-32 Hrs

Unit-1: Introduction and scope of Ecology: Plants and the environment- plant adaptation, ecotypes, habitat ecology- fresh water and marine water ecology (ecosystems), wetlands and their characteristics; Ecosystem function; The distribution of biomes; Major Terrestrial Biomes; Forests-Tropical Forests-Temperate Forests, Taiga, Grasslands, Savanna, Temperate Grasslands/Prairies, Tundra, Deser and Chaparral.

Unit-2: Environmental Biology: Global warming: Greenhouse gases - causes and consequences; Ozone depletion- causes and consequences; Air, water and soil pollution - major pollutants, their source, permissible limits - and control methods; Radioactive pollution- Ionising radiation, disposal of radioactive waste, nuclear accidents; Environmental Education Programmes - WWF, UNEP, MAB; Role of plants in solving energy crisis and ameliorating global warming.

Unit-3: Biodiversity and Conservation Biology: Science in the service of Biodiversity, biodiversity and its value, biodiversity issues, concerns, management; Biodiversity hot spots; Biodiversity- threats and current status of biodiversity; IUCN categories, Red Data book and Red lists, invasive alien species as threat to biodiversity; Conservation strategies- past, present, and future; Attitudes about conservation; conservation movements; CITES (Convention on international trade in endangered species), WCU (World Conservation Union); Endangered species Act. 2002 (GOI); Protected areas, Network of India- history, size, scale and management; Heritage trees.

Unit-4: Phytogeography: Biogeography of the world, India and Karnataka; Climatic zones, tectonics, continental movements; Types of plant distribution – discontinuous distribution - land bridge theory, continental drift; continuous distribution-cosmopolitan, circumpolar, circumboreal, circumaustral, pantropical; Distribution of plants - islands; Phytochorea of the world, India; Plant dispersal, migrations and isolation; Eendemic plants of Western Ghats and Eastern Himalayas; Origin, distribution and acclimatization of coffee, cardamom, sugarcane, cashew, ragi, maize, wheat, rice and cotton; Remote sensing and GPS, study of vegetation by GIS (Geographical Information system).

Practicals-32 Hrs

- 1) Study of local vegetation by quadrat method.
- 2) Water analysis for pollution studies.(Bio-monitoring: TDS, Hardness, Chlorides, CO₂ COD, DO, BOD)
- 3) Rapid detection of bacteriological quality of water with special reference to faecal coliforms.
- 4) Morphology and anatomy of plants in relation to habitats - Xerophytes, Mesophytes, Hydrophytes.
- 5) *In situ* and *Ex situ* method of conservation.
- 6) Eminent phytogeographers of the world (photos).
- 7) Continental drift (charts).
- 8) Application of Remote Sensing, GIS and GPS in Forestry and Wild life management.
- 9) Biogeography of the world – Oceans, deserts, islands, mountains.

- 10) Biogeography of India –rivers, mountains, islands.
- 11) Floristic regions of world – India and Karnataka.
- 12) Study of endemic plants of India.
- 13) Origin, acclimatization and distribution of Coffee, Cardamom, Sugarcane, Cashew, Ragi, Maize, Wheat, Rice and Cotton.

References:

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- 2) Good R.D. 1974. Geography of the flowering plants.
- 3) James H. B. 1998. Biogeography.
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- 5) Croiat, 1952. Manual of Phytogeography.
- 6) Edgar A. 1972. Plants, Man and Life.
- 7) Valentine, D. H. 1972. Taxonomy, Phytogeography & Evolution.
- 8) Phil Gibson J. and Gibson Terri, R. 2006. Plant ecology.
- 9) Primack, R. B. 2006. Essentials of conservation biology.

- 10) Ricklefs, R. E. 2001. The Economy of Nature.
- 11) Narasaiah M. L., 2005. Biodiversity and Sustainable Development.
- 12) Tondon P, Abrol Y. P, Kumaria S., 2007. Biodiversity and its significance.
- 14) Krishnamurthy K. V. 2007. An Advanced Textbook on Biodiversity: Principles and Practice.
- 15) Christian Leveque and Jean-Claude Mounolou (2003). Biodiversity.
- 16) Jeffries Michael J. 2006. Biodiversity and conservation.

**BOTANY: IV- SEMESTER- SOFT CORE 4.2
PROJECT WORK**

BOTANY: IV- SEMESTER- SOFT CORE 4.1
SEED TECHNOLOGY

Theory-32 Hrs

Unit-1: Seed Technology: Introduction to seed science and technology and its goals; Development of seed technology industry in India; Seed as basic input in agriculture; Seed Biology - Seed development, morphology and anatomy of dicot and monocot seeds; Seed structure and functions; Seed programmes and organizations; Seed village concept, seed production agencies, seed industry and custom seed production in India; International Seed Science and Technology Organizations.

Unit-2:Seed Production: General principles of seed production in self and cross pollinated and vegetatively propagated crops; Hybrid seed production; Maintenance of inbred lines and breeders seeds; Synthetic and composite seeds; Improved seed and their identification; Germplasm banks; **Seed Processing**-Harvesting, seed drying, seed cleaning and grading; Equipments; Seed Storage- types of storage structure; seed factors affecting storage life, effect of storage on relative humidity, temperature and moisture; Seed deterioration; Seed treatment.

Unit-3: Seed Quality Testing: Devices and tools used in seed testing; ISTA and its role in seed testing; Seed sampling- physical purity and heterogeneity test; Seed moisture content-importance and determination and methods; Viability and vigour testing; Genetic purity testing -objective and criteria for genetic purity testing, seed health testing, field and seed standards, designated diseases, objectionable weeds; Significance of seed borne diseases, seed health testing and detection methods for seed borne fungi, bacteria, viruses and nematodes; Preparation and dispatch of seed testing reports, storage of guard samples, application and use of seed standards and tolerances.

Unit- 4: Seed Certification: Principles and philosophy of seed certification, purpose and procedures, national seed programme; National Seed Corporation (NSC) - agencies responsible for achieving self-reliance in seed production and supply of quality of seeds (State Seeds Corporation; National Seed Development Council (NSDC); Central Seed Committee(CSC) ; Seed market surveys, seed industry in relation to global market; Concept of WTO, GATT, IPR, Plant Variety Protection and its significance seed technology; UPOV and its role.

Practicals-32 Hrs

- 1) Determination of physical purity of seed samples.
- 2) Determination of density or weight per thousand seeds.
- 3) Determination of seed Heterogeneity.
- 4) Visual examination of dry seeds for disease symptoms.
- 5) Determination of moisture content by hot air oven method.
- 6) Seed viability test- TTC method.
- 7) Determination of seed germination by TP/BP/Sand method.
- 8) Evaluation of seedlings vigour by BP/Sand methods.
- 9) Seed vigour evaluation by (a) conductivity test (b) Hiltner's test (c) Performance test(d) Accelerated ageing test (e) Cold test.
- 10) Examination of suspensions obtained from washings of seed.
- 11) Infection sites studied by planting seed components.
- 12) Detection of seed-borne fungi and their characters of five seed borne pathogens. Vist: Visit to seed industries/seed companies/ seed research stations.

References:

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- 7) Michael, B. and Bewley, D. 2000. Seed technology and its biological basis. Wiley- Blackwell.
- 8) Neergaard, P. 2005. Seed Pathology, Palgrave, Macmillan, Denmark. Science, Technology and Uses. CABI, UK.
- 9) Vanangamudi, K., Natarajan, K., Saravanan, T., Natarajan, N., Umarani, R., Bharathi, A. and Srimathi, P. 2006. Advances in Seed Science and Technology: Vol: III: Forest Tree Seed Technology and Management, Agrobios, New Delhi.

BOTANY: IV- SEMESTER- SOFT CORE 4.2

SEED PATHOLOGY

Theory - 32 Hrs

Unit-1: Seed Pathology: Introduction, historical development, development of seed health testing; Reduction in crop yields loss in due to seed-borne diseases; Seed-borne pathogens (Fungi, Bacteria, Mycoplasma-like Organisms, fastidious Vascular Bacteria, Spiroplasmas, Viruses, Viroids, Nematodes); Location of seed-borne inoculums, histopathology of some seed-borne pathogens; Seed infection, mechanism of seed infection, seed infestation or contamination; Factors affecting seed infection; Longevity of seed-borne pathogens.

Unit-2: Seed transmission and inoculation, factors affecting seed transmission; Cultural practices, epidemiology and inoculum thresholds of seed-borne pathogens; Classification of seed-borne; Role of Seed-borne inoculum in disease development; Economic loss due to seed borne pathogens; Certification program; Seed health tests, Nonparasitic seed disorders; Deterioration of grains; Storage fungi, field and storage fungi; Invasion by storage fungi; effects of seed deterioration.

Unit-3: Detection of Seed-borne Diseases: Examination of dry seeds; Isolation of fungi, Bright-field microscopic examination, observation under UV light, measurement of gases, Determination of FAV, Moldy smell, collection of seed exudates; Immunoassays, ergosterol estimation; Avoiding damage to seeds during harvesting; Processing, threshing, storage conditions, reducing seed moisture to safe limits, seed treatment, resistance.

Unit-4: Mycotoxins - Fungi known to produce mycotoxins, Factors affecting mycotoxin production the effects and control of mycotoxins, storage conditions, sorting of grains, cultural operations, chemical treatment, biological control, detoxification, regulatory measures, use of resistant cultivars; Control of seed-borne pathogens; Selection of seed production areas; Crop management, crop rotation, isolation distances, rouging, biological control, chemical method, mechanical method, physical methods; Certification- certification standards, plant quarantine, national and international regulations.

Practicals-32 Hrs

- 1-5) Detection of seed-borne fungi and their identification.
- 6) Detection of Seed-borne bacteria.
- 6) Detection of seed-borne viruses.
- 7) Detection of seed-borne insects by egg-plug staining.
- 8) Detection seed-borne nematodes.
- 9) Effect of deterioration of grains by Storage Fungi.
- 10) Detection of seed-borne fungi by PCR.
- 11) Estimation of ergosterol by UV-visible Spectrophotometer.
- 12) Detection of mycotoxins by thin Layer chromatography.

References

- 1) Agarwal, V. K. and Sinclair, J. B. 1996. Principles of Seed Pathology, 2nd edn. CRC Press, Taylor and Francis, USA.
- 2) Neergaard, P. 1977. Seed Pathology. Vol. I..Macmillan Press, Cornell University, USA.
- 3) Agrios, G. N. 1994 -Plant Pathology 2nd edn. Academic Press, New York.
- 4) Mehrotra, R. S. 1983-Plant Pathology Tata Mc. Graw Hill Pub. Co. Ltd., New Delhi.

- 5) Rangaswamy, G. and Mahadevan, K. 2002. Diseases of Crop plants in India. Prentice Hall of India Private Limited New Delhi.
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- 7) Basra, A. S. 2006. Handbook of Seed Science and Technology, The Haworth Press, USA.
- 8) Copeland, L.A. 1995. Principles of Seed Science and Technology- Kluwer Academic Publishers, The Netherlands.
- 9) Vanangamudi, K., Natarajan, K., Saravanan, T., Natarajan, N., Umarani, R., Bharathi, A. and Srimathi, P. 2006. Advances in Seed Science and Technology: Vol: III: Forest Tree Seed Technology and Management, Agrobios, New Delhi.

BOTANY: IV- SEMESTER- SOFT CORE 4.3
BIO- ANALYTICAL TECHNIQUES

Theory-32 Hrs

Unit- 1: Spectroscopy: Principles of UV-Visible spectroscopy, chromophores and their interaction with UV-visible radiation and their utilization in structural, qualitative and quantitative analysis of drug molecules; Infrared Spectroscopy, Infrared radiation and its interaction with organic molecules, vibrational mode of bonds, instrumentation and applications, interpretation of IR spectra; FTIR and ATR, X-ray diffraction methods.

Unit-2: Nuclear Magnetic Resonance Spectroscopy: Magnetic properties of nuclei, field and precession, instrumentation and applications of NMR; Chromatographic techniques- Principles and applications- types- column, paper, thin layer and gas chromatography, HPLC, HPTLC, size exclusion chromatography, Affinity chromatography, Mass spectrometry, MALDI-TOF.

Unit-3: Electrophoresis: Principle and application of PAGE, SDS PAGE, immunostaining, immuno-electrophoresis, Iso-electric focusing, 2D electrophoresis Centrifugation- Principles, techniques of preparative and analytical centrifugation. Ultracentrifuges, molecular weight determination, sedimentation analysis, RCF. Microscopy- principles and applications of electron microscope (SEM and TEM), CryoEM, Preparations of specimen for electron microscopy- freeze drying, freeze etching, fixing, staining; confocal, fluorescent, flow cytometry - principles and applications.

Unit-4: Molecular Biology Techniques: Primer designing; Principles and applications of PCR; Blotting techniques; Hybridization techniques; Micro-array; Next Generation Sequencing- Nucleic acid sequencing.

Practicals-32 Hrs

- 1) Calibration of bio-analytical instruments.
- 2) Principles and instrumentation and applications of imaging techniques:
- 3) Separation of fatty acids/lipids by TLC/HPTLC.
- 4) Separation of proteins by PAGE, SDS- PAGE.
- 5) Agarose gel electrophoresis of DNA/RNA.
- 6) Immunoelectrophoresis
- 7) Agar gel diffusion, counter immuno electrophoresis.
- 8) Verification of Beer Lambert law with the U.V. spectrophotometer.
- 9) Demonstration of blotting techniques.
- 10) Performing PCR for amplification of ITS regions of fungi/ bacteria.

References

- 1) Braithwaite, A. and Smith, F.J. 1996. Chromatographic Methods. 5th edn. Blackie Academic & Professional London.
- 2) Budzikiewicz, H., Djerassi, C. and Williams, D.H. 1968. Mass Spectrometry of Organic Compounds. Holden-Day, San Francisco, CA
- 3) Harborne, J.B. 1984. Phytochemical Methods. 2nd edn. Chapman and Hall, London.
- 4) Harborne J.B. (1973) Phytochemical methods a guide to modern techniques of plants analysis. Chapman and Hall, London Ltd.

BOTANY: II SEMESTER - OPEN ELECTIVE 4.1
PLANT DIVERSITY AND HUMAN WELFARE

Theory-32 Hrs

Unit -1: Plant Diversity and Significance: Role of plant diversity in ameliorating energy crisis and global warming; Types of biodiversity-genetic diversity, species diversity, plant diversity at the ecosystem level; Agro-biodiversity and cultivated plant taxa, wild taxa; **Values and uses of Biodiversity-** Ethical and aesthetic values, precautionary principle, methodologies for valuation, uses of plants and microbes.

Unit -2: Loss of Biodiversity: Major causes of for biodiversity loss; Loss of genetic diversity, Loss of species diversity; Loss of ecosystem diversity; Loss of agro-biodiversity; Projected scenario for biodiversity loss; Management of Plant Biodiversity- Organizations associated with biodiversity management; Methodology for execution-IUCN, UNEP, UNESCO, WWF, NBPGR; Biodiversity legislation and conservations; Biodiversity information management and communication.

Unit -3: Conservation of Biodiversity: Conservation of genetic diversity, species diversity and ecosystem diversity, *In situ* and *ex situ* conservation, Social approaches to conservation, Biodiversity awareness programmes, Conservation of Heritage Trees.

Unit-4: Role of plants in relation to Human Welfare: Importance of forestry their utilization and commercial aspects, Avenue trees, Ornamental plants of India, Alcoholic beverages through ages, Fruits and nuts- Fruit crops of Karnataka and their commercial importance; Wood and its uses.

References:

- 1) Krishnamurthy K. V. 2007. An Advanced Textbook on Biodiversity: Principles and Practice. Oxford & IHB Publishing Co. Pvt. Ltd. New Delhi.
- 2) Christian Leveque and Jean-Claude Mounolou, 2003. Biodiversity. John Wiley & Sons Ltd, The Atrium, Southern Gate, Chichester, West Sussex PO19 8SQ, England.
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JSS Mahavidyapeetha



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

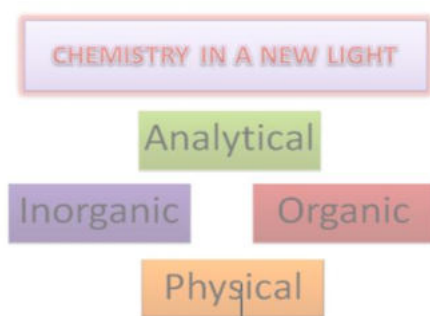
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POSTGRADUATE DEPARTMENT OF CHEMISTRY

SYLLABUS FOR M.Sc. DEGREE PROGRAMME IN CHEMISTRY



PROGRAMME CODE: CHE



under
Choice Based Credit System (CBCS) and
Continuous Assessment Grading Pattern (CAGP)
(With effect from 2018-19)

GENERAL REQUIREMENTS

Scheme of Instructions:

- A. A Masters Degree program is of 4 semesters-two years duration. A candidate can avail a maximum of 8 semesters – 4 years (in one stretch) to complete Masters Degree (including blank semesters, if any). Whenever a candidate opts for blank semesters, he/ she has to study the prevailing courses offered by the department when he/ she continues his/ her studies.
- B. A candidate has to earn a minimum of 76 Credits, for successful completion of a Master Degree. The 76 Credits shall be earned by the candidate by studying Hard Core, Soft Core and Open Elective.
- C. **Minimum for Pass:** In case a candidate secures less than Thirty percent in C₁ and C₂ put together, the candidate is said to have DROPPED the course, and such a candidate is not allowed to appear for C₃.
- D. In case a candidate secures less than Thirty percent in C₃, or secures more than Thirty percent in C₃ but less than Thirty percent in C₁, C₂ and C₃ put together, the candidate is said to have not completed the course and he/ she may either opt to DROP the course or to utilize PENDING option.
- E. **Credits (Minimum) Matrix:** A candidate has to study a minimum of 16 Credits in Soft Core (sum total of 4 semesters) and 04 Credits in Open Elective (in III Semester) for the successful completion of the Masters Degree course. A minimum of 15 students should register for every Soft Core or Open Elective course.
- F. All other rules and regulations hold good which are governed by the College/ University.

GENERAL SCHEME WITH RESPECT TO THE ASSESSMENT OF CREDITS

Semester	Hard Core (HC)		Total	Soft Core (SC)		Total	Open Elective (OE)
	Theory	Practicals		Theory	Practicals		
I	3+3+3+3=12	(4+4)+(4+4)=08 ^a	20	2+2+2+2=08 ^b	NIL	08^b	NIL
II	3+3+3+3=12	(4+4)+(4+4)=08 ^a	20	2+2+2+2=08 ^b	NIL	08^b	NIL
III	3+0+3+0=06	NIL	06	0+(2x2)+0+(2x2)=08	(4+4)+(4+4)=08 ^{a,c}	16	04
IV	0+3+0+3=06	NIL	06	(2x2)+0+(2x2)+0=08	4 ^d +(4+4)+(4+4)=08 ^{a,c}	18	NIL
Grand Total	36	16	52	26	16	42	04

^aFifty percent of the students will attend Analytical/ Inorganic Practical and remaining Fifty percent students will attend Organic/ Physical Practical in I or III Semester and *vice-versa* during II or IV Semester.

^bCourses are common for both I and II Semesters and the candidate can opt any course of his/ her choice in aforesaid semesters and should ascertain that the course/ s already studied in I Semester are not repeated in the II Semester.

^cPracticals are only for chemistry students which are compulsory courses.

^dDissertation/ Project work

SCHEME OF STUDY AND EXAMINATION

FIRST SEMESTER HARD CORE

THEORY

Course Code	Course Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHA 130	Fundamentals of Chemical Analysis	03	03	100	15	15	03	70
CHA 140	Inorganic Chemistry-I	03	03	100	15	15	03	70
CHA 150	Organic Chemistry-I	03	03	100	15	15	03	70
CHA 160	Physical Chemistry-I	03	03	100	15	15	03	70

PRACTICALS

Course Code	Course Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHA 050	Analytical Chemistry Practicals	08	04	100	15	15	06	70
CHA 060	Inorganic Chemistry Practicals	08	04	100	15	15	06	70
CHA 070	Organic Chemistry Practicals	08	04	100	15	15	06	70
CHA 080	Physical Chemistry Practicals	08	04	100	15	15	06	70

Note: Fifty percent of the students will attend Analytical and Inorganic practicals and remaining Fifty percent of the students will attend Organic and Physical practicals in I semester and *vice-versa* during II semester.

SOFT CORE

THEORY

Course Code	Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHA 500	Applied Analysis I	02	02	100	15	15	03	70
CHA 510	Frontiers in Inorganic Chemistry	02	02	100	15	15	03	70
CHA 520	Reaction Mechanisms	02	02	100	15	15	03	70
CHA 530	Solid State Chemistry and Chemistry of Nano Materials	02	02	100	15	15	03	70

SECOND SEMESTER HARD CORE

THEORY

Course Code	Course Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHB 090	Separation Techniques	03	03	100	15	15	03	70
CHB 100	Advanced Coordination Chemistry	03	03	100	15	15	03	70
CHB 170	Organic Chemistry-II	03	03	100	15	15	03	70
CHB 120	Physical Chemistry - II	03	03	100	15	15	03	70

PRACTICALS

Course Code	Course Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHB 130	Analytical Chemistry Practicals	08	04	100	15	15	06	70
CHB 140	Inorganic Chemistry Practicals	08	04	100	15	15	06	70
CHB 150	Organic Chemistry Practicals	08	04	100	15	15	06	70
CHB 160	Physical Chemistry Practicals	08	04	100	15	15	06	70

Note: It is same as that of I Semester. Students who have studied Analytical/ Inorganic or Organic/ Physical Practicals will get interchanged during II Semester.

SOFT CORE

All the courses are common for both I and II Semesters and the candidate can opt any course of his/ her choice in aforesaid semesters and should ascertain that the course/ s already studied in I Semester are not repeated in the II Semester.

THEORY

Course Code	Title	Contact Hours/ Week	Credits	Max. Marks/ Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHB 500	Applied Analysis I	02	02	100	15	15	03	70
CHB 510	Frontiers in Inorganic Chemistry	02	02	100	15	15	03	70
CHB 520	Reaction Mechanisms	02	02	100	15	15	03	70
CHB 530	Solid State Chemistry and Chemistry of Nano Materials	02	02	100	15	15	03	70

THIRD SEMESTER

HARD CORE

THEORY

Course Code	Course Title	Contact Hours/ Week	Credits	Max. Marks/ Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHC 030	Instrumental Methods of Analysis	03	03	100	15	15	03	70
CHC 040	Spectroscopy	03	03	100	15	15	03	70

OPEN ELECTIVE (for Non-Chemistry Students only)

Course Code	Course Title	Contact Hours/ Week	Credits	Max. Marks/ Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHC 600	Selected Topics in Chemistry	04	04	100	15	15	03	70

SOFT CORE

THEORY

Course Code	Course Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHC 500	Inorganic Chemistry-II	02	02	100	15	15	03	70
CHC 510	Structural Methods in Inorganic Chemistry	02	02	100	15	15	03	70
CHC 520	Biophysical Chemistry and polymers	02	02	100	15	15	03	70
CHC 530	Applications of Chemical Kinetics and Quantum Chemistry	02	02	100	15	15	03	70

PRACTICALS

Course Code	Course Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHC 210	Analytical Chemistry Practicals	08	04	100	15	15	06	70
CHC 220	Inorganic Chemistry Practicals	08	04	100	15	15	06	70
CHC 230	Organic Chemistry Practicals	08	04	100	15	15	06	70
CHC 240	Physical Chemistry Practicals	08	04	100	15	15	06	70

- Note:** 1. Fifty percent of the students will attend Analytical and Inorganic practicals and remaining Fifty percent of the students will attend Organic and Physical practicals in III semester and *vice-versa* during IV semester.
2. Practicals papers are only for chemistry students which are compulsory

FOURTH SEMESTER HARD CORE

THEORY

Course Code	Course Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
CHD 010	Bioinorganic Chemistry	03	03	100	15	15	03	70

CHD 020	Advanced Physical Chemistry	03	03	100	15	15	03	70
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SOFT CORE

THEORY

Course Code	Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
					CHD 500	Applied Analysis II	02	02
CHD 510	Applied Analysis III	02	02	100	15	15	03	70
CHD 520	Retrosynthesis and Organometallic Chemistry	02	02	100	15	15	03	70
CHD 530	Biomolecules and Natural Products	02	02	100	15	15	03	70

PRACTICALS

Course Code	Course Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
					CHD 210	Analytical Chemistry Practicals	08	04
CHD 220	Inorganic Chemistry Practicals	08	04	100	15	15	06	70
CHD 230	Organic Chemistry Practicals	08	04	100	15	15	06	70
CHD 240	Physical Chemistry Practicals	08	04	100	15	15	06	70

Note: It is same as that of III Semester. Students who have studied Analytical/ Inorganic or Organic/ Physical Practicals will get interchanged during IV Semester.

PROJECT /DISSERTATION WORK

Course Code	Course Title	Contact Hours/Week	Credits	Max. Marks/Course	Internal Assessment Marks		Semester End Exams (C ₃)	
					C ₁	C ₂	Duration	Max. Marks
					CHD 250	Project /Dissertation Work	08	04

SCHEME OF EXAMINATION FOR C₁, C₂ AND C₃ COMPONENTS

Preamble: In view of the CBCS syllabus, following is the model distribution of marks for C₁, C₂ and C₃ Components. At a glance, the model includes both theory (HC/ SC/ OE) as well as practicals (HC/ SC) assessment of marks.

The following is the scheme which will be followed for the assessment of marks for both theory (HC/ SC/ OE) as well as practicals (HC/ SC) irrespective of the Credits associated with each Course. Thirty percent of the marks will be assessed for the internals (C₁ and C₂) and remaining seventy percent will be for the semester end examinations (C₃). Each Course carries 100 marks and hence thirty marks for internal assessment and remaining seventy marks will be for Semester End Examinations. Out of thirty marks for internals, fifteen marks will be allotted to each C₁ and C₂ components. The distribution of marks for C₁ and C₂ varies with HC and SC papers.

Each theory Course (HC/ SC/ OE) consists of three components namely C₁, C₂ and C₃. C₁ and C₂ are designated as Internal Assessment (IA) and C₃ as Semester End Examination. Each Course (HC/ SC/ OE) carries **100 Marks** and hence the allotment of marks to C₁, C₂ and C₃ Components will be fifteen, fifteen and seventy marks respectively. i.e.,

C ₁ Component	: 15 Marks	} Internal Assessment Marks
C ₂ Component	: 15 Marks	
C ₃ Component	: 70 Marks	Semester End Examination
Total	: 100 Marks	

The above will be followed in common for all the theory (HC/ SC/ OE) and practical (HC/ SC) Courses in all the four semesters.

1. THEORY:

1.1. HARD CORE (03 CREDITS COURSES)

1.1.1 Distribution of Marks for C₁ and C₂ Components (I/ II Semesters):

IA consists of fifteen marks for each component it will be divided into three parts viz., **Internal Test, Home Assignment and Seminar**. Internal tests will be conducted during the 8th Week of the semester for C₁ and 16th Week of the semester for C₂. Home Assignment will be

concerned for C₁ Component and Seminar for C₂ Component only. Hence, a teacher may give only one assignment (or in their personal interest one more may be given). Since each Course has three units, the marks shall be divided equally. Allotment of marks for C₁ and C₂ is as follows: Out of fifteen Marks for IA for C₁, Internal test will be conducted for Thirty Marks (reduced to 10 Marks) and Home Assignment will be given for 05 Marks (Each Home Assignment from every unit will be assessed for 05 Marks and finally reduced to 05 Marks). IA for C₂ will be distributed as follows: Internal test will be conducted for Thirty Marks (reduced to 10 Marks) and Seminar will be assigned for 05 Marks for the favor of IA. Please note that actual Seminar will be assessed for 20 Marks and finally 05 Marks will be distributed to each theory HC Course. i.e.,

C ₁		C ₂	
Internal Test	: 30 Marks (10+10+10) Reduced to 10 Marks	Internal Test	: 30 Marks (10+10+10) Reduced to 10 Marks
Home Assignment	: 15 Marks (05+05+05) Reduced to 05 Marks	Seminar	: 20 Marks (05+05+05+05) Distributed 05 Marks to each HC Course
Total	: 15 Marks	Total	: 15 Marks

1.1.1a Distribution of Marks for C₁ and C₂ Components (III/ IV Semesters):

The modalities discussed above in 1.1.1 holds good for this also except for Seminar component. Seminar will be assigned for 05 Marks for the favor of IA. Please note that actual Seminar will be assessed for 10 Marks and finally 05 Marks will be distributed to each theory HC Course. i.e.,

C ₁		C ₂	
Internal Test	: 30 Marks (10+10+10) Reduced to 10 Marks	Internal Test	: 30 Marks (10+10+10) Reduced to 10 Marks
Home Assignment	: 15 Marks (05+05+05) Reduced to 05 Marks	Seminar	: 10 Marks (05+05) Distributed 05 Marks to each HC Course
Total	: 15 Marks	Total	: 15 Marks

1.1.2 Distribution of Marks for C₃ Component (Semester End Examination):

The question paper is of 3 hr duration with Max. Marks 70. The following question paper pattern will be followed for all the theory Courses (HC/ SC/ OE). Question paper will have 2 parts both parts will cover all units of the course with equal proportional of distribution. Part A is of Short Answer Type questions which will have ten questions and each question carries two

Marks. Part B carries fifty Marks and comprises of seven questions where in a student has to answer any five. Each question carries ten marks with sub question i.e.,

Model Question Paper Pattern:

Max. Duration: 3 Hr

Max. Marks: 70

Note: *Question paper has two parts, answer both the parts.*

PART A

Ten questions will be given and all ten should be answered. Each question carries two marks.

10 x 2 = 20

PART B

Seven questions will be given and any five should be answered. Each question carries Ten marks. An examiner may distribute marks as (3+3+4), (5+5), (7+3) & (4+6). Two marks question shall be avoided to maximum extent.

5 x 10 = 50

1.2 SOFT CORE (02 CREDITS COURSES):

1.2.1 Distribution of Marks for C₁ and C₂ Components:

IA consists of 15 marks for each components; it will be divided into two parts viz., ***Internal Test and Home Assignment***. Internal tests will be conducted during the 8th Week of the semester for C₁ and 16th Week of the semester for C₂. As far as Home Assignment is concerned, the concerned teacher will assign one or two Home Assignments to each student. Since each Course has two units, the marks will be divided equally. Allotment of marks for C₁ and C₂ is as follows: Out of 15 Marks for IA, Internal tests will be conducted for 10 marks and Home Assignment for 05 Marks. i.e.,

C₁		C₂	
Internal Test	: 20 Reduced to 10	Internal Test	: 20 Marks 10
Home Assignment	: 10 Marks Reduced to 05	Home Assignment	: 10 Marks Reduced to 05
Total	: 15 Marks	Total	: 15 Marks

1.2.2 Distribution of Marks for C₃ Component (Semester End Examination):

The above discussed pattern (1.1.2) holds good in this case also.

1.3 OPEN ELECTIVE (04 CREDITS COURSE):

1.3.1 Distribution of Marks for C₁ and C₂ Components:

IA consists of 15 marks for each components; it will be divided into two parts viz., *Internal Test and Home Assignment*. Internal tests will be conducted during the 8th Week of the semester for C₁ and 16th Week of the semester for C₂. As far as Home Assignment is concerned, the concerned teacher will assign one or two Home Assignments to each student. Allotment of marks for C₁ and C₂ is as follows: Out of 15 Marks for IA, Internal tests will be conducted for 20 marks and reduced to 10 Home Assignment for 05 Marks. i.e.,

C₁		C₂	
Internal Test	: 20 Marks Reduced to 10	Internal Test	: 20 Marks Reduced to 10
Home Assignment	: 20 Marks Reduced to 05	Home Assignment	: 20 Marks Reduced to 05
Total	: 15 Marks	Total	: 15 Marks

1.3.2 Distribution of Marks for C₃ Component (Semester End Examination):

The above discussed pattern (1.1.2) holds good in this case also.

2. PRACTICALS (04 CREDITS COURSES):

The following scheme will be applicable for both HC and SC in all the four semesters (SC courses are for chemistry students only which are compulsory Courses).

Each practical (HC/ SC) consists of three components namely C₁, C₂ and C₃. C₁ and C₂ are designated as Internal Assessment (IA) and C₃ as Semester End Examination. Each practical (HC/ SC) carries **100 Marks** and hence the allotment of marks to C₁, C₂ and C₃ Components will be fifteen, fifteen and seventy marks respectively. i.e.

C ₁ Component	: 15 Marks	} Internal Assessment Marks
C ₂ Component	: 15 Marks	
C ₃ Component	: 70 Marks	Semester End Examination
Total	: 100 Marks	

2.1 Distribution of Marks for C₁ and C₂ Components:

IA consists of **15 Marks**; it will be divided into three parts viz., *Internal Test, Continuous Assessment and Record*. Continuous assessment refers to the daily assessment of

each student based on his/ her attendance, skill, results obtained etc. Thus, three marks are allotted for Continuous Assessment. Internal tests will be conducted for ten marks during the 8th Week of the semester for C₁ and 16th Week of the semester for C₂. Finally, remaining two Marks will be for the record. i.e.,

C ₁		C ₂	
Internal Test	: 10 Marks	Internal Test	: 10 Marks
Continuous Assessment	: 03 Marks	Continuous Assessment	: 03 Marks
Record	: 02 Marks	Record	: 02 Marks
Total	: 15 Marks	Total	: 15 Marks

5.1.2 Distribution of Marks for C₃ Component (Semester End Examination):

The end examination will be conducted for **seventy Marks/ Course** with a maximum duration of six hours. Two experiments will be given to each student which carries thirty Marks each. Each student will be subjected to Viva-Voce Examination for which ten Marks is allotted. i.e.,

Two Experiments	: 60 Marks
Viva-Voce	: 10 Marks
Total	: 70 Marks

2.3 Evaluation of Project Work/ Dissertation (Minor):

Each student can take up Project Work/ Dissertation under the guidance of the faculty of the department during the IV Semester as a Soft Core course.

2.3.1 Distribution of Marks for C₁ and C₂ Components:

IA consists of **fifteen Marks** for each components; it will be divided into three parts viz., *Attendance, Continuous Assessment and Work Progress*. Continuous assessment refers to the daily assessment of each student based on his or her skill, results obtained, literature survey etc. C₁ will be assessed during the 8th Week of the semester and C₂ during the 16th Week of the semester. Hence, the concerned guide will prepare the marks list based on the above said parameters for both C₁ and C₂ Components.

2.3.2 Distribution of Marks for C₃ Component (Semester End Examination):

The semester end examination will be conducted for **seventy Marks**. Every student is suppose to prepare a hard copy of the findings of the work in the form of report and submitted for

evaluation. This part will be assessed for fourth Marks. Each student will be subjected to Viva-Voce Examination for which thirty Marks is allotted. i.e.,

Evaluation of Report	: 40 Marks
Viva-Voce	: 30 Marks
Total	: 70 Marks

Programme Outcomes (POs)

At the end of the programme the student able to:

POID	PO
39167	Work in the pure, interdisciplinary and multidisciplinary areas of chemical sciences and its applications.
39159	Learn about the potential uses of analytical, inorganic, organic and physical chemistry.
39165	Acquire knowledge, abilities and insight in well-defined area of research within Chemistry.
39168	Plan and execute research in frontier areas of chemical sciences.
39162	Develop knowledge of scientific theories and methods, gain experience in working independently with scientific questions and clearly express opinion on academic issues.
39163	Acquire the skills of planning and conducting advanced experiments by applying suitable simple and sophisticated analytical techniques.
39169	Learn professionalism, including the ability to work in teams and apply basic ethical principles.
39166	Adopt the skills and knowledge required to the professional life, and to qualify for training as scientific researcher.
39160	Develop scientific communication skills for differently specialized and non-specialized audiences.

POID	PO
39161	Gather attention about the physical aspects of chemistry.
39164	Examine specific phenomena theoretically and/or experimentally, contribute to the generation of new scientific insights or to the innovation of new applications of research in Chemistry.

Programme Specific Outcome (PSOs)

After completion of this programme the candidate able to

PSOID	PSOs
PSO1	Think and teach aspects of chemistry to the different levels of students in a futuristic manner.
PSO2	Reach the positions by employment in chemical, pharmaceutical, food and material industries.
PSO3	Reach a level to think about the scientific situations existing around him/her.
PSO4	Take up Global level research opportunities to pursue Ph.D. programme and will be more resourceful and will have targeted approach to qualify CSIR- NET and other competitive examinations.
PSO5	Analyse data obtained from sophisticated instruments for the structure determination and chemical analysis.
PSO6	Understand the background of organic mechanism and instrumental methods of chemical analysis.
PSO7	Apply modern methods of analysis to chemical systems in a laboratory setting.
PSO8	Find placements in R & D and synthetic division of polymer industries & allied division.
PSO9	Explore new areas of research in both chemistry and allied fields of science and technology.

FIRST SEMESTER
THEORY – HARD CORE

Course Title: FUNDAMENTALS OF CHEMICAL ANALYSIS

Course Code: CHA 130

Course Outcomes

After studying this course the student able to:

COID	CO
47457	Learn in depth Language of analytical chemistry, Errors and treatment of analytical data.
47470	Specify in depth Titrimetric analysis, Acid-base titrations in non-aqueous media, Precipitation titrations.
47494	Learn in depth Complexometric titrations, Redox titrations, Obtaining and preparing samples for analysis.

UNIT – I

Analytical Chemistry–Meaning, role, central location of analytical chemistry. Quantitative and qualitative analysis.Steps in quantitative analysis.

Language of analytical chemistry - Analysis, determination and measurement.Techniques, methods, procedures and protocols.Classifying analytical techniques.

Errors and treatment of analytical data: Limitations of analytical methods – Error: determinate and indeterminate errors, minimization of errors. Accuracy and precision,distribution of random errors, the normal error curve. Statistical treatment of finite samples-measures of central tendency and variability: mean, median, range, standard deviation and variance. Student’s t-test,confidence interval of mean.Testing for significance-comparison of two means and two standard deviations.Comparison of an experimental mean and a true mean.Criteria for the rejection of an observation- Q-test. Propagation of errors: determinate errors and indeterminate errors.

Standardization and calibration: Comparison with standards-direct comparison and titrations. External standard calibration-the least squares methods, regression equation, regression coefficient. Internal standard methods and standard-addition methods.

Selecting an analytical method: Accuracy, precision, sensitivity, selectivity, robustness and ruggedness,scale of operation, equipment, time and cost.Making the final choice.Figures of merit of analytical methods–sensitivity, detection and quantitation limit, linear dynamic range.

[16 HOURS]

UNIT – II

Titrimetric analysis: An overview of titrimetry. Principles of titrimetric analysis. Titration curves. Titrations based on acid-base reactions-titration curves for strong acid and strong base, weak acid and strong base and weak base and strong acid titrations. Selecting and evaluating the end point. Finding the end point by visual indicators, monitoring *pH* and temperature. Quantitative applications – selecting and standardizing a titrant, inorganic analysis-alkalinity, acidity and free CO₂ in water and waste waters, nitrogen, sulphur ammonium salts, nitrates and nitrites, carbonates and bicarbonates. Organic analysis-functional groups like carboxylic acid, sulphonic acid, amine, ester, hydroxyl, carbonyl. Air pollutants like SO₂. Quantitative calculations. Characterization applications-equivalent weights and equilibrium constants.

Acid-base titrations in non-aqueous media: Role of solvent in acid-base titrations, solvent systems, differentiating ability of a solvent, some selected solvents, titrants and standards, titration curves, effect of water, determining the equivalence point, typical applications-determination of carboxylic acids, phenols and amines.

Precipitation titrations: Titration curves, feasibility of precipitation titrations, factors affecting shape - titrant and analyte concentration, completeness of the reaction, titrants and standards, indicators for precipitation titrations involving silver nitrate, the Volhard, the Mohr and the Fajan's methods, typical applications.

[16 HOURS]

UNIT – III

Complexometric titrations: Complex formation reactions, stability of complexes, stepwise formation constants, chelating agents, EDTA-acidic properties, complexes with metal ions, equilibrium calculations involving EDTA, conditional formation constants, derivation of EDTA titration curves, effect of other complexing agents, factors affecting the shape of titration curves-completeness of reaction, indicators for EDTA titrations-theory of common indicators, titration methods employing EDTA-direct, back and displacement titrations, indirect determinations, titration of mixtures.

Redox titrations: Balancing redox equations, calculation of the equilibrium constant of redox reactions, calculating titration curves, detection of end point, visual indicators and potentiometric end point detection. Quantitative applications-adjusting the analyte's oxidation state, selecting and standardizing a titrant. Inorganic analysis-chlorine residuals, dissolved oxygen in water, water in non-aqueous solvents. Organic analysis-chemical oxygen demand (COD) in natural and waste waters, titrations of mercaptans and ascorbic acid with I₃⁻ and titration of organic compounds using periodate.

Obtaining and preparing samples for analysis: Importance of sampling, designing a sample plan-random, judgement, systematic-judgement, stratified and convenience sampling. Type of sample to collect - grab and composite samples. *In situ* sampling. Size of sample and number of samples. Implementing the sampling plan - solutions, gases and solids. Bringing solid samples into solution - digestion and decomposing.

[16 HOURS]

References

1. Fundamental of Analytical Chemistry, D.A. Skoog, D.M. West, Holler and Crouch, 8th edition, 2005, Saunders College Publishing, New York.
2. Analytical Chemistry, G.D. Christian, 5th edition, 2001, John Wiley & Sons, Inc, India.
3. Quantitative Analysis, R.A. Day and A.L. Underwood, 6th edition, 1993, Prentice Hall, Inc. New Delhi.
4. Vogel's Textbook of Quantitative Chemical Analysis, J. Mendham, R.C. Denney, J.D. Barnes and M.J.K. Thomas, 6th edition, Third Indian Reprint, 2003, Pearson Education Pvt. Ltd., New Delhi.
5. Analytical Chemistry Principles, John H. Kennedy, 2nd edition, Saunders College Publishing, California, 1990.
6. Principles and Practice of Analytical Chemistry, F.W. Fifield and Kealey, 3rd edition, 2000, Blackwell Sci., Ltd. Malden, USA.
7. Modern Analytical Chemistry, David Harvey, McGraw Hill, New Delhi, 2000.

Course Title: INORGANIC CHEMISTRY-I

Course Code: CHA 140

Course Outcomes

After studying this course the student able to:

COID	CO
48345	Understand the details of Molecular symmetry and group theory and applications, Representation of groups.
48356	Learn in details with examples VSEPR model , Non-aqueous solvents, Electron deficient compounds, Lanthanides & Actinides.
48360	Understand the classification and characteristics of Organometallics of transition metals.
48367	Specify in depth Ferrocene and ruthenocene, Complexes containing alkene, alkyne, arene and allyl ligands.

UNIT – I

Molecular symmetry and group theory: Symmetry elements and symmetry operations. Concept of a group, definition of a point group. Classification of molecules into point groups. Subgroups. Schoenflies and Hermann-Mauguin symbols for point groups. Multiplication tables (C_n , C_{2v} and C_{3v}). Matrix notation for the symmetry elements. Classes and similarity transformation.

Representation of groups: The Great Orthogonality theorem and its consequences. Character tables (C_s , C_i , C_2 , C_{2v} , C_{2h} and C_{3v}).

Applications of group theory: Group theory and hybrid orbital. Group theory to Crystal field theory and Molecular orbital theory (octahedral and tetrahedral complexes). Determining the symmetry groups of normal modes (both linear and non-linear molecules).

[16 HOURS]

UNIT – II

Structures and energetics of inorganic molecules: Introduction, Energetics of hybridization. VSEPR model for explaining structure of AB, AB₂E, AB₃E, AB₂E₂, ABE₃, AB₂E₃, AB₄E₂, AB₅E and AB₆ molecules. M.O. treatment of homonuclear and heteronuclear diatomic molecules. M.O. treatment involving delocalized π -bonding (CO_3^{2-} , NO_3^- , NO_2^- , CO_2 and N_3^-).

Non-aqueous solvents: Classification of solvents, Properties of solvents (dielectric constant, donor and acceptor properties) protic solvents (anhydrous H₂SO₄, HF and glacial acetic acid) aprotic solvents (liquid SO₂, BrF₃ and N₂O₄). Solutions of metals in liquid ammonia. Super acids.

Electron deficient compounds: Higher boranes, polyhedral boranes (preparations, properties, structure and bonding). Wade's rules, carboranes and metallocarboranes.

Lanthanides & Actinides: Spectral & magnetic properties. Use of lanthanide compounds as shift reagents.

[16 HOURS]

UNIT – III

Fundamental concepts: Introduction, Classification of organometallic compounds by bond type, nomenclature, the effective atomic number rule, complexes that disobey the EAN rule, common reactions used in complex formation.

Organometallics of transition metals: Preparation, bonding and structures of nickel, cobalt, iron and manganese carbonyls. Preparation and structures of metal nitrosyls in organometallics.

Ferrocene and ruthenocene: Preparation, structure and bonding.

Complexes containing alkene, alkyne, arene and allyl ligands: preparation, structure and bonding. The isolobal principles.

[16 HOURS]

References:

1. Symmetry and spectroscopy of molecules, 2nd Ed. Veera Reddy, New Age International Publication (2009).
2. Group Theory and its Chemical Applications, P.K. Bhattacharya, Himalaya Publishing House (1986).
3. Chemical Applications of Group Theory, 3rd Ed., F.A. Cotton, Wiley, New York (1990).
4. Inorganic Chemistry, 3rd edition. James E. Huheey, Harper and Row Publishers (1983).
5. Inorganic Chemistry, 3rd edition. G.L. Miessler and D.A. Tarr, Pearson Education (2004).
6. Inorganic Chemistry, 4th edition. P. Atkins, T. Overton, J. Rourke, M. Weller and F. Armstrong, Oxford University Press (2004).
7. Inorganic Chemistry, 2nd edition. C.E. Housecroft and A.G. Sharpe, Pearson Education Ltd. (2005).
8. Basic Organometallic Chemistry - B.D. Gupta and A.J. Elias, Universities Press (2010).

Course Title: ORGANIC CHEMISTRY-I**Course Code: CHA 150****Course Outcomes**

After studying this course the student to:

COID	CO
49763	Learn in details with examples Stereoisomerism, Stereoselectivity, Optical, Geometrical, isomerism and Conformational isomerism.
49791	Understand in details with examples Molecular rearrangements, Carbon to carbon migration, Carbon to nitrogen migration.
49796	Learn in details with examples Miscellaneous rearrangements.
49802	Learn the classification and characteristics of Heterocyclic chemistry.

UNIT – I

Stereoisomerism: Projection formulae [Fly wedge, Fischer, Newman and Saw horse], enantiomers, diastereoisomers, configurational notations of simple molecules, *DL* and *RS* configurational notations.

Stereoselectivity: Stereoselective reactions, diastereoselective reactions, stereospecific reactions, regioselective and regiospecific reactions, enantioselective and enantiospecific reactions.

Optical isomerism: Conditions for optical isomerism, Elements of symmetry – plane of symmetry, centre of symmetry, alternating axis of symmetry (rotation-reflection symmetry);

optical isomerism due to chiral centers and molecular dissymmetry, allenes and biphenyls, criteria for optical purity.

Geometrical isomerism: Due to C=C, C=N and N=N bonds, E, Z conventions, determination of configuration by physical and chemical methods.

Conformational isomerism: Elementary account of conformational equilibria of ethane, butane and cyclohexane. Conformation of cyclic compounds such as cyclopentane, cyclohexane, cyclohexanone derivatives and decalins. Conformational analysis of 1,2, 1,3, and 1,4-disubstituted cyclohexane derivatives and *D*-Glucose, Effect of conformation on the course/ rate of reactions.

[16 HOURS]

UNIT – II

Molecular rearrangements: Introduction

Carbon to carbon migration: Pinacol-pinacolone, Wagner-Meerwein, Benzidine, Demjanov, benzylic acid, Favorskii, Arndt-Eistert synthesis, Fries rearrangement, Steven's rearrangement, dienophile rearrangement.

Carbon to nitrogen migration: Hofmann, Curtius, Lossen, Schmidt and Beckmann rearrangements.

Miscellaneous rearrangements: Sommelet-Hauser, Wittig, Smiles, Neber, Japp-Klingemann rearrangement, Meisenheimer rearrangements, Bayer-Villegier rearrangement, Allylic rearrangements.

[16 HOURS]

UNIT – III

Heterocyclic chemistry: Nomenclature of heterocyclic systems

Structure, reactivity, synthesis and reactions of indole, pyridine, benzofuran, quinoline, isoquinoline, pyrazole, imidazole, pyrone, coumarin, chromones, pyrimidines and purines. Synthesis and synthetic applications of azirines and aziridines, isoxazole, oxazole and azepine.

[16 HOURS]

References:

1. H. Pine, Hendrickson, Cram and Hammond, Organic Chemistry, Mc Graw Hill, New York, 1987.
2. Organic Chemistry by Morrison & Boyd.
3. I.L. Finar, Organic Chemistry, ELBS Longmann, Vol. I & II, 1984.
4. E.L. Eliel and S.H. Wilen, Stereochemistry of Organic Compounds, John Wiley and Sons, New York. 1994.

5. Introduction to Stereochemistry by K. Mislow.
6. Basic Principles of Organic Chemistry by Roberts & Caserio
7. N.S. Issacs, Reactive Intermediates in Organic Chemistry, John Wiley and Sons, New York.1974.
8. R.K. Bansal, Organic Reaction Mechanism, Wiley Eastern Limited, New Delhi, 1993.
9. J. March, Advanced Organic Chemistry, Wiley Interscience, 1994.
10. E.S. Gould, Mechanism and Structure in Organic Chemistry, Halt, Rinhart & Winston, New York, 964.
11. A Guide Book to Mechanism in Organic Chemistry by Petersykes
12. Stereochemistry and Mechanism through Solved Problems by P.S. Kalsi.
13. Text book of Organic Chemistry by P.S. Kalsi.
14. F.A. Carey and Sundberg, Advanced Organic Chemistry – Part A & B, 3rd edition, Plenum Press, New York, 1990.
15. D. Nasipuri, Stereochemistry of Organic Compounds, 2nd edition, Wiley Eastern Limited, New Delhi, 1991.
16. S.K. Ghosh, Advanced General Organic Chemistry, Book and Alleied (P) Ltd, 1998.
17. Heterocyclic Chemistry – Joule & Smith
18. Heterocyclic Chemistry – Achaeson
19. Basic Principles of Heterocyclic Chemistry – L.A. Pacquette
20. Comprehensive Heterocyclic Chemistry – Kartritzky series, Pergamon Press, New York, 1984.

Course Title: PHYSICAL CHEMISTRY-I

Course Code: CHA 160

Course Outcomes

After studying this course the student to:

COID	CO
51245	Learn in depth Concepts of entropy and free energy, Partial molar properties.
51250	Learn the details of Fugacity, Statistical thermodynamics.
51272	Learn the details of Chemical Kinetics, Kinetics of reactions in solution, Linear free energy, Enzyme kinetics.
51303	Learn the characteristics of Electrochemistry, Energetics of cell reactions, Corrosion.

UNIT – I

Concepts of entropy and free energy: Second law of thermodynamics, definition of entropy, entropy of phase transition, entropy change during spontaneous process. Helmholtz and Gibbs free energies, Maxwell relations, Variation of free energy with temperature and pressure. Third law of thermodynamics, Nernst heat theorem & its applications, numericals based on entropy and free energy changes.

Partial molar properties: Partial molar quantities, Partial molar Gibbs function, Partial molar volume and its determination by intercept method and density measurements. Chemical potential and its significance. Variation of chemical potential with temperature and pressure. Formulation of the Gibbs Duhem equation. Derivation of Duhem-Margules equation.

Fugacity: Determination of fugacity of gases. Variation of fugacity with temperature and pressure. Activity and activity coefficients. Variation of activity with temperature and pressure. Determination of activity coefficients by vapour pressure, depression in freezing points and solubility measurements by electrical methods.

Statistical thermodynamics: Different types of ensembles, ensemble averaging, distribution law (Boltzmann statistics), partition function and thermodynamic parameters; relation between molecular and molar partition functions, translational partition function, rotational partition function for linear and non-linear molecules.

[16 HOURS]

UNIT – II

Chemical Kinetics: Basic concepts of chemical kinetics. Complex reactions: measurement of kinetics. Chain, parallel, consecutive and reversible reactions. Arrhenius equation, energy of activation and its experimental determination. Simple collision theory-mechanism of bimolecular reaction. Lindemann's theory, Hinshelwood's theory for unimolecular reaction. Activated complex theory of reaction rate.

Kinetics of reactions in solution-salt effects, effect of dielectric constant (single sphere and double sphere model), effect of pressure, volume and entropy change on reaction rates. Cage effect with an example. Oscillatory reactions: oxidation of malonic acid. Kinetics of heterogeneous reactions - Langmuir's theory, unimolecular and bimolecular surface reactions.

Linear free energy relationship: Hammett equation, Taft equation. Isokinetic relationship and significance of isokinetic temperature.

Enzyme kinetics: Effect of substrate concentration (Michaelis Menton equation), Effect of pH, effect of catalyts and inhibitors, effect of temperature.

[16 HOURS]

UNIT – III

Electrochemistry: Arrhenius theory of strong and weak electrolytes and its limitations. Factor effecting conductance, Debye-Huckel-Onsager equation of conductivity and its validity. Walden's rule. Debye-Huckel theory - concept of Ionic strength, Debye-Huckel limiting law (DHL), its modification for appreciable concentrations. Determination of transference number by moving boundary and Hittorf's methods. True and apparent transference numbers (TrN). Abnormal TrN, effect of temperature on TrN. Liquid junction potential-determination and minimization.

Energetics of cell reactions: Effect of temperature, pressure and concentration on energetics of cell reactions (calculation of ΔG , ΔH and ΔS). Electrochemical energy sources – batteries, classification, primary & secondary.

Corrosion: Manifestations of corrosion, types of corrosion, basis of electrochemical corrosion, theories of corrosion. Local cell theory (Wagner and Traud theory), Corrosion inhibition and prevention.

[16 HOURS]

References:

1. Thermodynamics for Chemists by S. Glasstone, Affiliated East-West Press, New Delhi, (1965).
2. Chemical Thermodynamics by I.M. Klotz, W.A. Benzamin Inc. New York, Amsterdam (1964).
3. Basic Physical Chemistry by W.J. Moore, Prentice Hall of India Pvt. Ltd., New Delhi (1986).
4. Text Book of Physical Chemistry by Samuel Glasstone, MacMillan Indian Ltd., 2nd edition (1974).
5. Theoretical Chemistry by S. Glasstone.
6. Elementary Statistical Thermodynamics by N.D. Smith Plenum Press, NY (1982).
7. Elements of Physical Chemistry by Lewis and Glasstone.
8. Physical Chemistry by P.W. Atkins, ELBS, 4th edition, Oxford University Press (1990)

9. Chemical Kinetics by K.J. Laidler.
10. Chemical Kinetics by Frost and Pearson.
11. Kinetics and Mechanism of Chemical Transformation by J. Rajaram and J.C. Kuriacose.
12. Chemical Kinetics by L.K. Jain.
13. Chemical Kinetics by Benson.

PRACTICALS – HARD CORE

Course Title: ANALYTICAL CHEMISTRY PRACTICALS

Course Code: CHA 050

Course Outcomes

After studying this course the student to:

COID	CO
CO1	Analyze various samples with different classical and simple instrumental skills.
CO2	Obtain knowledge for selection of analytical methods with suitable technique being adopted for the analysis different samples like, water, laboratory chemicals and reagents, body fluids such as urine etc.
CO3	Distinguish classical and instrumental methods.
CO4	Propose and conduct experiment for quantification of individual analytes.

[128 HOURS]

PART – I

1. Determination of total acidity of vinegar and wines by acid-base titration.
2. Determination of purity of a commercial boric acid sample, and Na_2CO_3 content of washing soda.
3. Determination of relative equivalent weight of a weak organic acid by titration with NaOH.
4. Determination of ephedrine and aspirin in their tablet preparations by residual acid-basetitrimetry.
5. Determination of carbonate and bicarbonate in a mixture by *pH*-metric titration and comparison with visual acid-base titration.
6. Determination of carbonate and hydroxide-analysis of a commercial washing soda by visual and *pH*-titrimetry.
7. Determination of purity of a commercial sample of mercuric oxide by acid-base titration.

8. Determination of benzoic acid in food products by titration with methanolic KOH in chloroform medium using thymol blue as indicator.
9. Determination of the pH of hair shampoos and pH determination of an unknown soda ash.
10. Analysis of water/ waste water for acidity by visual, pH metric and conductometric titrations.
11. Analysis of water/ waste water for alkalinity by visual, pH metric and conductometric titrations.
12. Determination of ammonia in house-hold cleaners by visual and conductometric titration.
13. Determination of chromate and dichromate in mixture by acid-base titration: visual and pH metric methods.
14. Potentiometric determination of the equivalent weight and K_a for a pure unknown weak acid.
15. Determination of purity of aniline by non-aqueous acid-base titration by visual and potentiometric methods.
16. Determination of purity of ethylene glycol and glycerol by oxidimetric method using periodate (Malprade reaction).
17. Spectrophotometric determination of creatinine and phosphorus in urine.
18. Flame emission spectrometric determination of sodium, potassium and calcium in river/ lake water.

PART – II

1. Determination of percentage of chloride in a sample by precipitation titration- Mohr, Volhard and Fajan's methods.
2. Determination of silver in an alloy and Na_2CO_3 in soda ash by Volhard method.
3. Mercurimetric determination of chloride in blood or urine.
4. Determination of total hardness, calcium and magnesium hardness and carbonate and bicarbonate hardness of water by complexation titration using EDTA.
5. Determination of calcium in calcium gluconate/ calcium carbonate tablets/ injections and of calcium in milk powder by EDTA titration.
6. Determination of zinc in a sample of foot powder and thallium in a sample of rodenticide by EDTA titration.
7. Analysis of commercial hypochlorite and peroxide solution by iodometric titration.
8. Determination of copper in an ore/ an alloy by iodometry and tin in stibnite by iodimetry.
9. Determination of ascorbic acid in vitamin C tablets by titrations with $KBrO_3$ and of vitamin C in citrus fruit juice by iodimetric titration.
10. Determination of iron in razor blade by visual and potentiometric titration using sodium metavanadate.

11. Determination of iron in pharmaceuticals by visual and potentiometric titration using cerium(IV) sulphate.
12. Determination of nickel in steel by synergic extraction and boron in river water/ sewage using ferroin.
13. Determination of total cation concentration of tap water by ion-exchange chromatography.
14. Determination of magnesium in milk of magnesium tablets by ion-exchange chromatography.
15. Cation exchange chromatographic separation of cadmium and zinc and their estimation by EDTA titration.
16. Gas chromatographic determination of ethanol in beverages.
17. Solvent extraction of zinc and its spectrophotometric determination.
18. Anion exchange chromatographic separation of zinc and magnesium followed by EDTA titration of the metals.
19. Separation and determination of chloride and bromide on an anion exchanger.
20. Separation of *o*- and *p*-nitroaniline and analysis by thin layer chromatography.

References

1. Fundamental of Analytical Chemistry, D.A. Skoog, D.M. West, Holler and Crouch 8th edition, 2005, Saunders College Publishing, New York.
2. Analytical Chemistry, G.D. Christian, 5th edition, 2001 John Wiley & Sons, Inc, India.
3. Quantitative Analysis, R.A. Day and A.L. Underwood, 6th edition, 1993, Prentice Hall, Inc. New Delhi.
4. Vogel's Textbook of Quantitative Chemical Analysis, J. Mendham, R.C. Denney, J.D. Barnes and M.J.K. Thomas, 6th edition, Third Indian Reprint, 2003, Pearson Education Pvt. Ltd., New Delhi.
5. Analytical Chemistry Principles, John H. Kennedy, 2nd edition, Saunders College Publishing, California, 1990.
6. Practical Clinical biochemistry methods and interpretations, R. Chawla, J.P. Bothers Medical Publishers (P) Ltd., 1995.
7. Laboratory manual in biochemistry, J. Jayaraman, New Age International Publishers, New Delhi, 1981.
8. Practical Clinical Biochemistry by Harold Varley and Arnold.Heinmann, 4th edition.

Course Title: INORGANIC CHEMISTRY PRACTICALS**Course Code: CHA 060****Course Outcomes**

After studying this course the student to:

COID	CO
CO1	Prepare reagents required for analysis.
CO2	Propose and conduct experiment for quantitative analysis of inorganic samples such as ore, metals, complexes mixture of metals and complexes etc.
CO3	Propose schemes for semi-micro qualitative analysis.
CO4	Develop skills for the scientific and relevant documentation and risk and security assessment.

[128 HOURS]**PART – I**

1. Determination of iron in haematite using cerium(IV) solution (0.02M) as the titrant, and gravimetric estimation of insoluble residue.
2. Estimation of calcium and magnesium carbonates in dolomite using EDTA titration, and gravimetric analysis of insoluble residue.
3. Determination of manganese dioxide in pyrolusite using permanganate titration.
4. Quantitative analysis of copper-nickel in alloy/ mixture:
 - i. Copper volumetrically using KIO_3 .
 - ii. Nickel gravimetrically using DMG
5. Determination of lead and tin in a mixture: Analysis of solder using EDTA titration.
6. Quantitative analysis of chloride and iodide in a mixture:
 - i. Iodide volumetrically using KIO_3
 - ii. Total halide gravimetrically
7. Gravimetric analysis of molybdenum with 8-hydroxyquinoline.
8. Micro-titrimetric estimation of :
 - a) Iron using cerium(IV)
 - b) Calcium and magnesium using EDTA
9. Quantitative estimation of copper(II), calcium(II) and chloride in a mixture.
10. Circular paper chromatographic separation of: (Demonstration)
 - a. Iron and nickel
 - b. Copper and nickel

PART – II

Semimicro qualitative analysis of mixtures containing **TWO** anions and **TWO** cations (excluding sodium, potassium and ammonium cations) and **ONE** of the following less common cations: W, Mo, Ce, Th, Ti, Zr, V, U and Li.

References

1. Vogel's Text Book of Quantitative Chemical Analysis – 5th edition, J. Basset, R.C. Denney, G.H. Jeffery and J. Mendhom.
2. A Text Book of Quantitative Inorganic Analysis by A.I. Vogel, 3rd edition.
3. Spectrophotometric Determination of Elements by Z. Marczenko.
4. Vogel's Qualitative Inorganic Analysis – Svelha.
5. Macro and Semimicro Inorganic Qualitative Analysis by A.I. Vogel.
6. Semimicro Qualitative Analysis by F.J. Welcher and R.B. Halin.
7. Quantitative Chemical Analysis by Daniel C. Harris, 7th edition, (2006).

Course Title: ORGANIC CHEMISTRY PRACTICALS

Course Code: CHA 070

Course Outcomes

After studying this course, the student to:

COID	CO
CO1	Prepare several simple organic compounds and also propose suitable mechanisms.
CO2	Acquire knowledge of different reactions, conditions to be maintained, precautions to be exercised before/during/after the reaction.
CO3	Learn qualitative analysis and to separate a mixture of two components.
CO4	Gain confidence to set up reactions individually either in the pharma industry or for the Research.

PART – I

1. Preparation of *p*-nitro aniline from acetanilide.
2. Preparation of *p*-bromo aniline from acetanilide.
3. Preparation of benzoic acid from benzaldehyde
4. Preparation of n-butyl bromide from n-butanol.
5. Preparation of *p*-nitroiodobenzene from paranitroaniline.
6. Preparation of aniline from nitrobenzene.
7. Preparation of β-D-Glucose penta acetate.

8. Preparation of phenoxy acetic acid.
9. Preparation of cyclohexanone from cyclohexanol.
10. Preparation of chalcone.
11. Preparation of *S*-benzylthiuronium chloride.
12. Condensation of anthracene and maleic anhydride (Diels-Alder reaction).
13. Preparation of *m*-nitrobenzoic acid from methyl benzoate.

PART – II

Qualitative analysis: Separation of binary mixtures, identification of functional groups and preparation of suitable solid derivatives.

References

1. Manual of Organic Chemistry -Dey and Seetharaman.
2. Modern Experimental Organic Chemistry by John H. Miller and E.F. Neugil, p 289.
3. An Introduction to Practical Organic Chemistry -Robert, Wingrove etc.
4. A Text Book of Practical Organic Chemistry–A.I. Vogel, Vol.III.
5. Practical Organic Chemistry, Mann & Saunders.
6. Semimicro Qualitative Organic Analysis by Cheronis, Entrikin and Hodnet.
7. J. N. Guthru & R. Kapoor, Advance experimental Chemistry, New Delhi-1991.
8. R. K. Bansal, Laboratory Manual of Organic Chemistry, New PGE International (P) LTd. London, 3rd edition. 1996.18
9. N. K. Visno, Practical Organic Chemistry, New PGE International (P) Ltd. London, 3rd edition, 1996.

Course Title: PHYSICAL CHEMISTRY PRACTICALS

Course Code: CHA 080

Course Outcomes

After studying this course, the student to:

COID	CO
CO1	An idea about handling of instruments like UV-Visible Spectrophotometer, Potentiometer, pH meter, etc.
CO2	Determine the concentration of the species in given solutions using kinetic methods.

CO3	Distinguish between different physical properties of substances or compounds.
CO4	Acquire knowledge of different thermodynamic parameters.

[128 HOURS]

PART – I (Non-instrumental)

1. Study of kinetics of hydrolysis of an ester using HCl/ H₂SO₄ at two different temperature, determination of rate constants and energy of activation.
2. Study of kinetics of the iodine-hydrogen peroxide clock reaction.
3. Determination of activation energy for the bromide-bromate reaction.
4. Determination of heat of solution of benzoic acid by variable temperature method (graphical method).
5. Determination of partial molar volume of NaCl-H₂O system.
6. Determination of critical solution temperature of phenol-water system.
7. Binary analysis of two miscible liquids by viscometric method (Ethanol & Water).
8. To study oscillating or periodic or rhythmic reactions of malonic acid.
9. Thermometric titration of hydrochloric acid with NaOH.
10. Kinetics of photodegradation of indigocarmine(IC) using ZnO as photocatalyst and study the effect of [ZnO] and [IC] on the rate of photodegradation.

PART – II (Instrumental)

1. Conductometric titration of a mixture of HCl and CH₃COOH against NaOH.
2. Conductometric titration of orthophosphoric acid /formic acid/ oxalic acid against NaOH and NH₄OH.
3. Determination of PI of glycine by by potentiometric method.
4. Potentiometric titration of KI vs KMnO₄ solution.
5. pH Titration of (a) polybasic acid(H₃PO₄), (b) (CH₃COOH+HCl) and (c) CuSO₄vsNaOH and determination of K_a.
6. To obtain the absorption spectra of colored complexes, verification of Beer's law and estimation of Ni ⁺² ions from [Ni(NH₃)₆]²⁺ by spectrophotometry.
7. Analysis of binary mixture (Glycerol and Water) by the measurement of refractive index.
8. Study the kinetics of reaction between CAT and indigo carmine spectrophotometrically and determination of rate constant.
9. Spectrophotometric titration of FeSO₄ against KMnO₄.
10. Determination of the molecular weight of a polymer material by viscosity measurements (cellulose acetate/methyl acrylate).

References

1. Practical Physical Chemistry – A.J. Findlay.
2. Experimental Physical Chemistry – F. Daniels *et al.*
3. Selected Experiments in Physical Chemistry – Latham.
4. Experiments in Physical Chemistry – James and Prichard.
5. Experiments in Physical Chemistry – Shoemaker.
6. Advanced Physico-Chemical Experiments – J. Rose.
7. Practical Physical Chemistry – S.R. Palit.
8. Experiments in Physical Chemistry – Yadav, Geol Publishing House.
9. Experiments in Physical Chemistry – Palmer.
10. Experiments in Chemistry – D.V. Jahagirdar, Himalaya Publishing House, Bombay, (1994).
11. Experimental Physical Chemistry – R.C. Das and B. Behera, Tata Mc Graw Hill.

THEORY – SOFT CORE

Course Title: APPLIED ANALYSIS-I

Course Code: CHA 500

Course Outcomes

After studying this course, the student to:

CO1: Describe the meaning of applied analysis.

CO2: Make out the causes for air pollution and water pollution, and knowledge an control devices or techniques or processes of such pollutions.

CO3: Understand the importance of food and drug analysis.

CO4: Acquire the knowledge to choose methodologies for the preliminary and complete analysis of air, water, food and drugs.

CO5: Adopt suitable analytical technique for sampling and analysis of air, water, food and drug samples for analysis.

CO6: Describe suitable analytical method for the determination of required analytes/components of the sample provided.

UNIT – I

Air pollution, analysis and control: Historical overview-global implications of air pollution, sources of pollutants, classification of pollutants. Sources and effects of particulates,

carbonmonoxide, sulphur oxides, nitrogen oxides, hydrocarbons and photochemical oxidants on human health, vegetation and materials. Standards for air pollutants.

Air quality monitoring: Sampling methods and devices for particulates and gaseous pollutants. SO₂: ambient air measurements and stack gas measurements- Turbidimetric, colorimetric, conductometric and coulometric methods, NO_x: Griess-Ilosvay and Jacobs-Hockheiser colorimetric methods, Hydrocarbons: total and individual hydrocarbons by gas chromatography. Oxidants and ozone: colorimetric, titrimetric and chemiluminescence methods.

Control devices for particulates: Gravitational settlers, centrifugal collectors, wet collectors, electrostatic precipitation and fabric filtration.

Control devices for gaseous pollutants: adsorption, absorption, condensation and combustion processes. Automotive emission control-catalytic converters.

Water pollution and analysis: Water resources, origin of wastewater, types of water pollutants; their sources and effects, chemical analysis for water pollution control-objectives of analysis, parameters of analysis, sample collection and preservation. Environmental and public health significance and measurement of colour, turbidity, total solids, acidity, alkalinity, hardness, chloride, residual chlorine, chlorine demand, sulphate, fluoride, phosphates, total nitrogen, NO₃⁻ and NO₂⁻ nitrogesn in natural and waste/ polluted waters, heavy metal pollution-public health significance of Pb, Cd, , Hg, and As, general survey of the instrumental techniques for the analysis of heavy metals in aquatic systems, organic loadings-significance and measurement of DO, BOD, COD, TOD, and TOC.

[16 HOURS]

UNIT – II

Food analysis: Objectives of food analysis. Sampling procedures. Detection and determination of sugars and starch. Methods for protein determination. Oils and fats and their analysis-iodine value, saponification value and acid value. Rancidity-detection and determination (peroxide number). Tests for common edible oils. Analysis of foods for minerals-phosphorus, sodium, potassium and calcium. General methods for the determination of moisture, crude fibre and ash contents of food. Analysis of milk for fat and added water. Non-alcoholic beverages-determination of chicory and caffeine in coffee; caffeine and tannin in tea. Alcoholic beverages-methanol in alcoholic drinks and chloral hydrate in toddy. Food additives-chemical, preservatives-inorganic preservatives-sulphur dioxide and sulphites, their detection and determination. Organic preservatives-benzoic acid and benzoates, their detection and determination. Flavouring agents-detection and determination of vanilla and vanillin. Coloring matters in foods-classification, certified colors, detection of water soluble dyes, color in citrus fruits, beet dye in tomato

products, mineral color. Pesticide residues in foods-determination of chlorinated organic pesticides. Control food quality-codex alimentarius, Indian standards.

Drugs and pharmaceutical analysis: Importance of quality control; drugs and pharmaceuticals. Sources of impurities in pharmaceutical chemicals. Analytical quality control in finished/ final products. Common methods of assay. Analysis of common drugs; Analgesics-aspirin, paracetamol; Anthelmintics-mebendazole; Antiallergies-chlorpheniramine maleate; Antibiotics-penicillin, chloramphenicol; Anti-inflammatory agents-oxycodone; Antimalarials-primaquine phosphate; Antituberculosists-INH; Narcotics-nicotine, morphine; Expectorants-Benadryl; Sedative-diazepam; Vitamins-A, C, B1, B2, B6, niacin and folic acid.

[16 HOURS]

References

1. Standard Methods of Chemical Analysis, A.J. Weleher (Part B), Robert E. Krieger Publishing Co. USA, 1975.
2. Environmental Chemistry, S.E. Manahan Willard grant press, London, 1983.
3. Environmental Chemical Analysis, Iain L Marr and Malcolm S. Cresser, Blackie and Son Ltd., London, 1983.
4. Chemistry for Environmental Engineering, Chair N. Sawyer and Perry L.M Canty, McGraw Hill Book, Co., New York, 1975.
5. The Air Pollution Hand Book, Richard Mabey, Penguin, 1978.
6. The Pollution Hand Book, Richard Mabey, Ponguin 1978.
7. Soil Chemical Analysis, M.L. Jackson, Prentice Hall of India Pvt, Ltd., New Delhi, 1973.
8. Experiments in Environmental Chemistry, P.D. Vowler and D.W. Counel, Pergamon press, Oxford 1980.
9. Manual Soil Laboratory Testing, vol I, K.H. Head, Pentech Press, London 1980.
10. A Text Book of Environmental Chemistry and Pollution Control, S.S. Dara, S.Chand and co. Ltd. New Delhi 2004.
11. Air pollution Vol II edition by A.C. Stern, Academic Press New York, 1968.
12. Instrumental Methods for Automatic Air Monitoring Systems in Air Pollution Control, Part-III edition by W. Strass, John-Wiley and Sons, New York, 1978.
13. Analysis of Air pollutants, P.O. Warner, John Wiley and Sons, New York, 1976.
14. The Chemical Analysis Air pollutants, Interscience, New York, 1960.
15. The Analysis of Air Pollutants, W. Liethe, Ann Arbor Science Pub. Inc. Michigan 1970.
16. Environmental Chemistry, A. K. De.
17. Food Analysis, A.G. Woodman, McGraw Hill. 1971.

18. Chemical Analysis of Foods, H.E. Cox and Pearson.
19. Analysis of Foods and Food Products, J.B. Jacob.
20. A First Course in Food Analysis, A.Y. Sathe, New Age Internationals (P) Ltd., Publishers, Bangalore, 1999.
21. Analytical Agricultural Chemistry, S.L. Chopra and J.S. Kanwar, Kalyani Publishers, New Delhi, 1999.
22. Pharmaceutical Analysis, (Ed). T. Higuchi and E.B. Hanssen, John Wiley and Sons, New York, 1997.
23. Pharmaceutical Analysis-Modern Methods, Part A and B, (Ed). James W. Hunson.
24. Quantitative Analysis of Drugs in Pharmaceutical Formulations, P. D. Sethi, 3rd edition. CBS Publishers and Distributors, New Delhi, 1997.

Course Title: FRONTIERS IN INORGANIC CHEMISTRY

Course Code: CHA 510

Course Outcomes

After studying this course, the student to:

CO1: Know the significance of materials chemistry

CO2: Acquire knowledge of various characterization techniques

CO3: Obtain the skills about the inorganic pigments.

CO4: Obtain the skills about the nanomaterials, nanoscience and nanotechnology.

UNIT – I

Materials chemistry

General principles-Defects, nonstoichiometric compounds and solid solutions, atom and ion diffusion, solid electrolytes. Synthesis of materials-The formation of extended structures, chemical deposition.

Metal oxides, nitrides and fluorides: Monoxides of the 3d metals, higher oxides and complex oxides, oxide glasses, nitrides and fluorides.

Chalcogenides, intercalation chemistry and metal rich phases: Layered MS₂ compounds and intercalation, Chevrel phases.

Framework structures: Structures based on tetrahedral oxoanions, structures based on octahedral and tetrahedral.

Inorganic pigments: Coloured pigments, white and black inorganic materials.

Molecular materials and fullerenes: Fullerenes, Molecular material chemistry.

Silicates: Structure, classification - silicates with discrete anions, silicates containing chain anion, silicates with layer structure, silicones with three dimensional net work and applications.

[16 HOURS]

UNIT – II

Nanomaterials, nanoscience and nanotechnology

Fundamentals-Terminology and history, novel optical properties of nanomaterials.

Characterization and fabrication: Characterization methods. Top-down and bottom-up fabrication. Solution based synthesis of nanoparticles. Vapour-phase synthesis of nanoparticles. Synthesis using frameworks, supports and substrates.

Artificially layered materials: Quantum wells and multiple quantum wells. Solid state superlattices. Artificially layered crystal structures.

Self-assembled nanostructures: Self-assembly and bottom-up fabrication. Supramolecular chemistry and morphosynthesis. Dimensional control in nanostructures.

Bioinorganic nanomaterials: DNA and nanomaterials. Natural and artificial nanomaterials-Biomimetics. Bionanocomposites.

Inorganic-organic nanocomposites: Uses and design strategies. Polymer nanocomposites.

[16 HOURS]

References:

1. Inorganic Chemistry, 4th edition. P. Atkins, T. Overton, J. Rourke, M. Weller and F. Armstrong, Oxford University Press (2006).
2. Inorganic Chemistry Principles of Structure and Reactivity: James E. Huheey, Ellen A. Keiter, Richard L. Keiter, Okhil K. Medhi, Delhi University, New Delhi (2006)
3. Chemistry of the Elements – N.N. Greenwood and A. Earnshaw, Pergamon Press (1985).
4. Industrial Inorganic Chemistry – 2nd edition. K.H. Buchel, H.H. Moretto and P. Woditsh, Wiley - VCH (2000).
5. Basic Inorganic Chemistry – 3rd edition. F.A. Cotton, G. Wilkinson and P.L. Gaus, John Wiley and Sons (2002).
6. Inorganic Chemistry, 3rd edition. James E. Huheey, Harper and Row Publishers (1983).
7. Inorganic Chemistry, 3rd edition. G.L. Miessler and D.A. Tarr, Pearson Education (2004).
8. Inorganic Chemistry, 2nd edition. C.E. Housecroft and A.G. Sharpe, Pearson Education.

Course Title: REACTION MECHANISMS

Course Code: CHA 520

Course Outcomes

CO1: Identify the reactivity of the molecules

CO2: Fate of the reaction by knowing the thermodynamic and kinetic requirements.

CO3: Identify the products, structure, and stability through mechanistic approach.

CO4: Mechanistic pathway of different reactions.

CO5: Know the nucleophilic, electrophilic and elimination reactions. Each of these will have different mechanistic route.

UNIT – I

Structure and reactivity: Brief discussion on effects of hydrogen bonding, resonance, inductive and hyperconjugation on strengths of acids and bases.

Methods of determining organic reaction mechanism: Thermodynamic and kinetic requirements for reactions, kinetic and thermodynamic control. Hammonds postulates and Curtin-Hammett principle.

Identification of products. Formation, structure, stability, detection and reactions of carbocations (classical and non-classical), carbanions, free radicals, carbenes, nitrenes, nitrile oxides, nitrile imines, nitrile ylides and arynes. Determination of reaction intermediates, isotope labeling and effects of cross over experiments. Kinetic and stereochemical evidence, solvent effect. Linear free energy relationship-Hammet equation and Taft treatment.

[16 HOURS]

UNIT – II

Basics of organic reactions: Meaning and importance of reaction mechanism, classification and examples for each class.

Aliphatic substitution reactions:

Nucleophilic substitution reactions: Kinetics, mechanism and stereochemical factors affecting the rate of S_N^1 , S_N^2 , S_N^i , S_N^1 , S_N^2 and S_N^i reactions, Neighbouring group participation.

Electrophilic substitution reactions: S_E^1 and S_E^2 reactions

Aromatic substitution reactions:

Nucleophilic substitution reactions: S_N^1 , S_N^2 and benzyne mechanism, Bucherer reaction.

Electrophilic substitution reactions: Mechanism of Friedel-Crafts alkylation and acylation, Mannich reaction, chloromethylation, Vilsmeier-Haack reaction.

Mechanism of hydrolysis of carboxylic acid derivatives: Hydrolysis of esters, amides and acid chlorides.

Elimination reactions: Mechanism and stereochemistry of eliminations - E₁, E₂, E_{1cB}. *cis* elimination, Hofmann and Saytzeff eliminations, competition between elimination and substitution, decarboxylation reactions. Chugaev reaction.

[16 HOURS]

References:

1. Organic Chemistry by Morrison and Boyd.
2. H. Pine, Hendrickson, Cram and Hammond, Organic Chemistry, Mc Graw Hill, New York, 1987.
3. I.L. Finar, Organic Chemistry, ELBS Longmann, Vol. I & II, 1984.
4. E.L. Eliel and S.H. Wilen, Stereochemistry of Organic Compounds, John Wiley and Sons, New York. 1994.
5. Basic Principles of Organic Chemistry by Roberts & Caserio
6. N.S. Issacs, Reactive Intermediates in Organic Chemistry, John Wiley and Sons, New York. 1974.
7. R.K. Bansal, Organic Reaction Mechanism, Wiley Eastern Limited, New Delhi, 1993.
8. J. March, Advanced Organic Chemistry, Wiley Interscience, 1994.
9. A Guide Book to Mechanism in Organic Chemistry by Petersykes
10. Stereochemistry and Mechanism through Solved Problems by P.S. Kalsi.
11. Text book of Organic Chemistry by P.S. Kalsi.
12. F.A. Carey and Sundberg, Advanced Organic Chemistry – Part A & B, 3rd edition, Plenum Press, New York, 1990.
13. S.K. Ghosh, Advanced General Organic Chemistry, Book and Alleied (P) Ltd, 1998.
14. Organic chemistry, Gram Solomons.

Course Title: SOLID STATE CHEMISTRY AND CHEMISTRY OF NANOMATERIALS

Course Code: CHA 530

Course Outcomes

After studying this course, the student to:

CO1: Learn solid state chemistry, X-ray crystallography, etc.

CO2: Learn the fundamentals of semiconductors, superconductors, nanomaterials and the methods by which nanoparticle is synthesized.

UNIT – I

Solid state chemistry: Types of imperfections, classification of imperfections, point defects, Schottky defects, Frenkel defects, disordered crystals, line defects, dislocation types, plane defects, small-angle and large-angle boundaries, stacking faults, crystal growth and twinning.

X-ray crystallography: law of interfacial angles, laws of symmetry, Miller indices, Bragg equation (no derivation), Experimental methods – powder and rotating crystal methods, indexing of powder and rotating crystal photographs. Atomic scattering factor, structure factor, Fourier synthesis and electron density diagrams. Electron diffraction of gases, experimental technique, Scattering-Intensity curves, Wierl equation (no derivation), Radial distribution method determination of bond lengths and bond angles. Heat capacity of solids: Einstein and Debye equations (with derivation).

[16 HOURS]

UNIT – II

Semiconductors: Band theory, energy bands, intrinsic and extrinsic semiconductors. Conductivity: electrons and holes, temperature dependence on conductivity, Optical properties: absorption spectrum, photoconductivity, photovoltaic effect and luminescence. Junction properties: metal-metal junctions, metal-semiconductor junctions, p-n junctions, transistors, industrial applications of semiconductors: Mixed oxides, spinels and other magnetic materials.

Superconductors: Meissner effect, type I and II super conductors, isotope effect, basic concepts of BCS theory, manifestations of the energy gap, Josephson devices.

Chemistry of nanomaterials: Nano particles. Synthesis - Laser ablation, chemical vapour transportor (CVT) and sol-gel methods. Metal oxides nanoparticles with supercritical water and precursor method. Synthesis of metal oxides and its composite nanoparticles by solvothermal and hydrothermal methods. Carbon nanotube, carbon nanowires and its composites. Applications of nanomaterials in renewable energy. Inorganic and organic nanoporous aerogels.

[16 HOURS]

References:

1. Solid State Chemistry and its Applications, Anthony R. West.
2. Solid State Chemistry: An Introduction, 3rd edition, Lesley E. Smart and Elaine A. Moore.
3. Introduction to Solid State Physics - C. Kittel, 5th edition, Wiley Eastern Ltd.

4. Advances in Technologically Important Crystals - Binay Kumar, R.P. Tandon, Mcmillan.
5. Hand Book of Nanotechnology, Bharat Bhushan, Springer Publisher.
6. Nanotechnology - Importance and Applications, M. H. Fulekar, Ink International publisher.

SECOND SEMESTER

THEORY – HARD CORE

Course Title: SEPARATION TECHNIQUES

Course Code: CHB 090

Course Outcomes:

After completion of this course, a student will be able to

COID	CO
51645	Learn the details of Solvent extraction, Extraction systems, Solid Phase Extraction, Chromatography.
51652	Learn the classification and characteristics of Gas chromatography, HPLC, Ion exchange, Size-exclusion chromatography.
51670	Learn in details with examples Thin layer, Affinity chromatography.
51682	Understand the details of Supercritical fluid extraction and Electrophoretic methods.

UNIT – I

Solvent extraction: Theory-Nernst partition law, efficiency and selectivity of extraction.

Extraction systems: Extraction of covalent neutral molecules, extraction of uncharged metal chelates and synergic extraction, extraction of ion-association complexes-non chelated complexes, chelated complexes and oxonium systems. Use of salting out agents. Methods of extraction-batch and continuous extractions. applications.

Solid Phase Extraction (SPE): Principles, apparatus and instrumentation. Solid phase sorbents, extraction formats - Automated solid phase extraction. Solid phase micro extraction (SPME). Applications of SPE and SPME.

Chromatography: Definition, principles and mechanism of separation, classification of chromatographic techniques. General descriptions of column chromatography-frontal analysis, displacement analysis and elution analysis. General theory of column chromatography: characterizing a chromatogram-retention time, retention volume and baseline width. Chromatographic resolution, capacity factor, column selectivity. Column efficiency-band

broadening-rate theory and plate theory. Peak capacity, non ideal behavior. Optimizing chromatographic separations using capacity factor, column selectivity and column efficiency-van Deemter equation, and its modern versions, Golay equation and Huber-Knox equations.

[16 HOURS]

UNIT – II

Gas chromatography (GC): Principles, instrumentation-mobile phase, chromatographic columns, stationary phases, sample introduction, temperature control, and detectors for gas chromatography. Quantitative and qualitative applications.

Highperformance liquid chromatography (HPLC): Principles, instrumentation- columns (analytical and guard columns), stationary phases, mobile phases, choosing a mobile phase, isocratic vs gradient elution, HPLC plumbing, sample introduction. Detectors for HPLC- spectroscopic, electrochemical and others, quantitative applications.

Ion exchange chromatography (IEC): Definitions, requirements for ion-exchange resin, synthesis and types of ion-exchange resins, principle, basic features of ion-exchange reactions, resin-properties-ion-exchange capacity, resin selectivity and factors affecting the selectivity, applications of IEC in preparative, purification and recovery processes. Ion chromatography (IC) : Double column IC and single column IC.

Size-exclusion chromatography: Theory and principle of size-exclusion chromatography, experimental techniques of gel-filtration chromatography (GFC) and gel-permeation chromatography (GPC), materials for packing-factors governing column efficiency, methodology and applications.

[16 HOURS]

UNIT – III

Thin layer chromatography: Principle, apparatus and methodology, applications, HPTLC

Affinity chromatography: Definitions, separation-mechanism-matrices, matrix activation, role of spacer arms and applications.

Supercritical fluid chromatography (SFC): Properties of supercritical fluids, instrumentation and operating variables, comparison of SFC with other types of chromatography, applications.

Supercritical fluid extraction: Advantages of supercritical fluid extraction, instrumentation, supercritical fluid choice, off-line and on-line extractions, typical applications of supercritical fluid extraction.

Electrophoretic methods - Electrophoresis & Capillary Electrophoresis: Theory-electrophoretic mobility, electroosmotic mobility, electroosmotic flow velocity, total mobility, migration time, efficiency, selectivity and resolution. Instrumentation-capillary tubes, hydrodynamic and electrokinetic methods of sample injection, applying electric field and detectors. Capillary

electrophoresis methods-capillary zone electrophoresis, micellarelectrokinetic capillary chromatography, capillary gel electrophoresis and capillary electrochromatography.

[16 HOURS]

References

1. Fundamental of Analytical Chemistry, D.A. Skoog, D.M. West, Holler and Crouch 8th edition, 2005, Saunders College Publishing, New York.
2. Analytical Chemistry, G.D. Christian, 5th edition, 2001 John Wiley & Sons, Inc. India.
3. Quantitative Analysis, R.A. Day and A.L. Underwood, 6th edition, 1993 Prentice Hall, Inc. New Delhi.
4. Vogel's Textbook of Quantitative Chemical Analysis, J. Mendham, R.C. Denney, J.D. Barnes and M.J.K. Thomas, 6th edition, Third Indian Reprint, 2003 Pearson Education Pvt. Ltd., New Delhi.
5. Analytical Chemistry Principles, John H. Kennedy, 2nd edition, Saunders College Publishing, California, 1990.
6. Introduction to Instrumental Analysis, Robert. D. Braun, Pharm. Med. Prem. India, 1987.
7. Instrumental Method of Analysis, W.M. Dean and Settle, 7th edition, 1986, CBS Publishers, New Delhi.
8. Instant Notes of Analytical Chemistry, Kealey and Haines, Viva Books Pvt. Ltd., 2002.
9. Modern Analytical Chemistry, David Harvey, McGraw Hill, New Delhi, 2000.
10. Principles and Practice of Analytical Chemistry, F. W. Fifield and Kealey, 5th edition, 2000, Blackwell Sci., Ltd. Malden, USA.

Course Title: ADVANCED COORDINATION CHEMISTRY

Course Code: CHB 100

Course Outcomes:

After studying this course, the student to:

COID	CO
47143	Learn in depth Preparation of coordination compounds, Stability of coordination compounds, Geometries of metal complexes, Determination of stability constants, Crystal field theory.
47150	Understand in details with examples Molecular Orbital Theory, Electronic spectra and Magnetic properties.

47159	Learn in details with examples Reaction and Mechanisms, Substitution reactions.
47165	Identify in details with examples Inner-sphere mechanism and outer-sphere mechanism.

UNIT – I

Preparation of coordination compounds: Introduction, Preparative methods - simple addition reactions, substitution reactions, oxidation-reduction reactions, thermal dissociation reactions, reactions of coordinated ligands, the trans-effect & other methods.

Stability of coordination compounds: Introduction, trends in stepwise stability constants, factors influencing the stability of metal complexes with reference to the nature of metal ion and ligands, the Irving-William series, chelate effect.

Geometries of metal complexes: Coordination numbers 2-8.

Determination of stability constants: Theoretical aspects of determination of stability constants of metal complexes by spectrophotometric, *pH* metric and polarographic methods.

Crystal field theory: Salient features of CFT, d-orbital splitting in octahedral, tetrahedral, square planar and tetragonal complexes, Jahn-Teller distortions, measurement of $10 Dq$ and factors affecting it. Evidences for metal-ligand covalency.

[16 HOURS]

UNIT – II

Molecular Orbital Theory: Introduction, Principles of Molecular orbital theory, sigma and pi-bonds in MOT. Applications to Ligand field theory. MOT to octahedral, tetrahedral and square planar complexes with and without pi-bonding.

Electronic spectra: Introduction, selection rules and intensities, electronic spectra of octahedral and tetrahedral complexes, Term symbols for d^n ions, Orgel and Tanabe-Sugano diagrams, charge-transfer spectra. Ligand-field transition, Optical rotatory dispersion and Circular dichroism.

Magnetic properties: Origin of magnetism, types of magnetism, Curie law, Curie Weiss law, magnetic susceptibility and its measurements. Spin and orbital contributions to the magnetic moment, the effects of temperature on μ_{eff} , spin-cross over, ferromagnetism, antiferromagnetism and ferrimagnetism.

[16 HOURS]

UNIT - III

Reaction and Mechanisms: Introduction

Substitution reactions - Inert and labile compounds, mechanisms of substitution.

Kinetic consequences of Reaction pathways - Dissociation, interchange and association.

Experimental evidence in octahedral substitution - Dissociation, associative mechanisms, the conjugate base mechanism, the kinetic chelate effect.

Stereochemistry of reactions- Substitution in trans and its complexes, isomerization of chelate rings.

Substitution reactions of square-planar complexes - kinetics and stereochemistry of square-planar substitutions, evidence for associative reactions, explanations of the trans effect.

Electron-transfer processes: Inner-sphere mechanism and outer-sphere mechanism, conditions for high and low oxidation numbers.

[16 HOURS]

References

1. Physical Inorganic Chemistry - A Coordination Chemistry Approach- S.F.A. Kettle, Spektrum, Oxford, (1996).
2. Inorganic Chemistry - 2nd edition, C.E. Housecroft and A.G. Sharpe, Pearson Education Ltd., (2005).
3. Inorganic Chemistry - 3rd edition, G.L. Miessler and D.A. Tarr, Pearson Education, (2004).
4. Inorganic Chemistry - 2nd edition, D.F. Shriver, P.W. Atkins and C.H. Langford, Oxford University Press, (1994).
5. Inorganic Chemistry- 3rd edition, James E. Huheey, Harper and Row Publishers, (1983).
6. Basic Inorganic Chemistry- 3rd edition, F.A. Cotton, G. Wilkinson and P.L. Gaus, John Wiley and Sons, (2002).

Course Title: ORGANIC CHEMISTRY-II

Course Code: CHB 170

Course Outcomes:

After studying this course, the student to:

COID	CO
50609	Understand in depth Reductions and Oxidations
50614	Learn in depth Reagents in organic synthesis, Green Synthesis
50628	Understand in details with examples Photochemistry and concerted reactions, Electrocyclic reactions

50648

Learn the details of Cycloaddition reactions, dipolar cycloadditions, cycloaddition reactions, Sigmatropic reactions

UNIT – I

Reductions: Catalytic hydrogenations (homogeneous and heterogeneous) - catalysts, reduction of functional groups, catalytic hydrogen transfer reactions. Wilkinson catalyst. Baker's yeast, LiAlH_4 , NaBH_4 , metal dissolving reactions (Birch reduction). Leukart reaction (reductive amination), diborane, Meerwein-Ponndorf-Verley reduction, Wolf-Kishner reduction, Clemensen reduction, tributyl tinhydride, stannous chloride.

Oxidations: Oxidation with chromium and manganese compounds (CrO_3 , $\text{K}_2\text{Cr}_2\text{O}_7$, PCC, PDC, Sarret reagent, Jones reagent, MnO_2 , KMnO_4), ozone, peroxides and peracids, lead tetra acetate, periodic acid, OsO_4 , SeO_2 , NBS, chloramine-T, Sommelet oxidation, Oppenauer oxidation, Sharpless epoxidation, Woodward and Prevost hydroxylation. Electrochemical Oxidation and reduction of organic compounds, green oxidation agents.

[16 HOURS]

UNIT – II

Reagents in organic synthesis: Use of following reagents in organic synthesis and functional group transformations: Lithium diisopropylamide (LDA), Gilmann reagent, dicyclohexyl carbodimide (DCC), dichloro dicyano quinone (DDQ), trialkyl silyl halides, phase transfer catalyst, crown ethers, Fenton's reagent, Ziegler-Natta catalyst, diazomethane, Stark enamine reaction, Phosphorus ylides – Wittig and related reactions, 1,3-dithiane anions - Umpolung reaction, sulphur ylides – reactions with aldehydes and ketones, Peterson reactions - synthesis of alkenes.

Green Synthesis: Designing of green synthesis, choice of reagents and catalysis. Microwave induced organic synthesis, ionic liquids in organic synthesis, polymer supported reagents and synthesis and the use of ultra sound in organic synthesis.

[16 HOURS]

UNIT – III

Photochemistry and concerted reactions: Introduction, light absorption and electronic transitions, Jablonski diagram, intersystem crossing, energy transfer, sensitizers, quenchers.

Photochemistry of olefins, conjugated dienes, aromatic compounds, ketones, enones, photooxidations, photoreductions, Norrish type I and II reactions, Paterno-Buchi reaction, Barton reaction, Di-pi-rearrangements.

Electrocyclic reactions: Stereochemistry, symmetry and Woodward-Hofmann rules for electrocyclic reactions, FMO theory of electrocyclic reactions, correlation diagram for cyclobutadiene and cyclohexadiene systems.

Cycloaddition reactions: Classification, analysis by FMO and correlation diagram method. **1,3-dipolar cycloadditions:** involving nitrile oxide, nitrile imine, nitrile ylide cycloaddition. Intra and intermolecular 3+2 cycloaddition and their application in organic synthesis.

[4+2] cycloaddition reactions: Deils-Alder reaction, hetero Diels-Alder reaction and their applications.

Sigmatropic reactions: Classification, stereochemistry and mechanisms. suprafacial and antarafacial shifts of H, sigmatropic shifts involving carbon moieties. [3,3] and [5,5]- sigmatropic rearrangement, Claisen, Cope and aza-Cope rearrangement

[16 HOURS]

References

1. H. Pine, Hendrickson, Cram and Hammond, Organic Chemistry, Mac Grow Hill, New York, 1987.
2. Organic Chemistry - Morrison and Boyd
3. I.L. Finar, Organic Chemistry, ELBS Longmann, Vol. 1 & II, 1984.
4. J. March, Advanced Organic Chemistry, Wiley Interscience, 1994.
5. E.S. Gould, Mechanism and Structure in Organic Chemistry, Halt, Rinhart & Winston, New York, 1964.
6. F.A. Carey and Sundberg. Advanced Organic Chemistry – Part A & B, 3rd edition, Plenum Press, New York. 1990.
7. Principles of Organic Synthesis - ROC Norman and Coxon
8. S.K. Ghosh, Advanced General Organic Chemistry, Book and Allied (P) Ltd. 1998.
9. R.K. Kar, Frontier orbital and symmetry controlled Pericyclic reaction.

Course Title: PHYSICAL CHEMISTRY-II

Course Code: CHB 120

Course Outcomes:

After studying this course, the student to:

COID	CO
50689	Learn in depth Quantum Chemistry

50698	Learn in details with examples Microwave and Vibration spectroscopy
50709	Understand in depth Raman and UV-Visible spectroscopy
50728	Learn the classification and characteristics of NQR, Mössbauer, ESR spectroscopy

UNIT – I

Quantum Chemistry: A brief resume of black body radiation, and atomic spectra-Bohr's theory of hydrogen atom. Photoelectric and Compton effects, de-Broglie concept, uncertainty principle, operators (algebra of operators, commutative and non-commutative operators, linear operator, Laplacian operator, Hermitian operator-Hamiltonian operator, turn over rule. Schrodinger wave equation for particles, Eigen values and Eigen functions, postulates of quantum mechanics. Application of Schrodinger equation to a free particle and to a particle trapped in a potential field (one dimension and three dimensions). Degeneracy, Wave equation for H-atom, separation and solution of R, ϕ and θ equations. Application of Schrodinger equation to rigid rotator and harmonic oscillator. Quantum numbers and their characteristics, orbital diagrams.

Approximate methods – Necessity of approximate methods, perturbation method, and the theory of perturbation method – first order and second order correction, application to He-atom (first order correction only).

[16 HOURS]

UNIT-II

Microwave spectroscopy: Rotation spectra of diatomic Molecules - rigid and non rigid rotator model. Rotational quantum number and the selection rule. Effect of isotopic substitution on rotation spectra. Relative intensities of the spectral lines. Classification of polyatomic molecules based on moment of inertia - Linear, symmetric top, asymmetric top and spherical molecules. Rotation spectra of polyatomic molecules (OCS, CH₃F and BCl₃). Moment of inertia expression for linear tri-atomic molecules. Applications - Principles of determination of Bond length and moment of inertia from rotational spectra. Stark effect in rotation spectra and determination of dipole moments.

Vibration spectroscopy: Vibration of diatomic molecules, vibrational energy curves for simple harmonic oscillator. Effects of anharmonic oscillation. Vibration - rotation spectra of carbon monoxide. Expressions for fundamental and overtone frequencies. Vibration of polyatomic molecules – The number of degrees of freedom of vibration. Parallel and perpendicular vibrations (CO₂ and H₂O). fundamental, overtone, combination and difference bands. Fermi resonance. Force constant and its significance. Theory of infrared absorption and theoretical group frequency. Intensity of absorption band and types of absorptions. Correlation chart. Important spectral regions - hydrogen stretching region, double and triple bonds regions, fingerprint region. Factors affecting the group frequency – Physical state, vibrational coupling,

electrical effect, hydrogen bonding, steric effect and ring strain. Applications: Structures of small molecules: XY_2 – linear or bent, XY_3 – planar or pyramidal.

[16 HOURS]

UNIT- III

Raman spectroscopy: Introduction, Raman and Rayleigh scattering, Stokes and anti-Stokes lines, polarization of Raman lines, depolarization factor, polarizability ellipsoid. Theories of Raman spectra - classical and quantum theory. Rotation-Raman and vibration-Raman spectra. Comparison of Raman and IR spectra, rule of mutual exclusion principle. Vibration modes of some simple molecules and their activity in Raman.

UV Visible spectroscopy: Quantitative aspects of absorption – Beer's law, Technology associated with absorption measurements. Limitations of the law – real, chemical, instrumental and personal.

NQR Spectroscopy: Quadrupolar nuclei, electric field gradient, nuclear quadrupole coupling constants, energies of quadrupolar transitions, effect of magnetic field. Applications.

Mössbauer spectroscopy: The Mössbauer effect, chemical isomer shifts, quadrupole interactions, measurement techniques and spectrum display, application to the study of Fe^{2+} and Fe^{3+} compounds, Sn^{2+} and Sn^{4+} compounds, nature of M-L bond, coordination number and structure), detection of oxidation states and inequivalent Mössbauer atoms.

Electron Spin Resonance Spectroscopy: Basic principles, hyperfine couplings, the 'g' values, factors affecting 'g' values, isotropic and anisotropic hyperfine coupling constants, Zero Field splitting and Kramer's degeneracy. Measurement techniques and Applications to simple inorganic and organic free radicals and to inorganic complexes.

[16 HOURS]

References

1. Vibrational Spectroscopy - Theory and Applications- D.N. Sathyanarayana, New Age International Publications, New Delhi (1996).
2. Spectroscopy, B.P. Straughan and S. Walker, John Wiley & Sons Inc., New York, Vol. 1 and 2, 1976.
3. Vibration Spectroscopy Theory and Applications, D.N. Satyanarayana, New Age International, New Delhi.
4. Spectroscopy, B.P. Straughan and S. Salker, John Wiley and Sons Inc., New York, Vol.2, 1976.

5. Organic Spectroscopy, William Kemp, English Language Book society, Macmillan, 1987.
6. Quantum Chemistry – A.K. Chandra. 2nd edition, Tata McGraw Hill Publishing Co. Ltd., (1983).
7. Quantum Chemistry – Eyring, Walter and Kimball. John Wiley and Sons, Inc., New York.
8. Quantum Chemistry – I.N. Levine. Pearson Education, New Delhi, (2000).
9. Theoretical Chemistry – S. Glasstone. East West Press, New Delhi, (1973).
10. Quantum Chemistry – R.K. Prasad, New Age International Publishers, (1996).
11. Valence Theory – Tedder, Murel and Kettle.
12. Quantum Chemistry – D.A. McQuarrie.
13. Theoretical Inorganic Chemistry – Day and Selbin.
14. Fundamentals of Molecular Spectroscopy, C.N. Banwell and E.M. McCash. 4th edition, Tata McGraw Hill, New Delhi.
15. Introduction to Spectroscopy - Pavia, Lampman and Kriz, 3rd edition, Thomson.
16. Spectroscopy, B.P. Straughan and S. Walker, John Wiley & Sons Inc., New York, Vol. 1 and 2, 1976.
17. Vibration Spectroscopy Theory and Applications, D.N. Satyanarayana, New Age International, New Delhi.
18. D. A. McQuarrie and J.D. Simon –Physical Chemistry, VIVA Students Ed. (2003).
19. J. D. Graybeat. Molecular Spectroscopy, McGraw-Hill International Edition (1988). Spectroscopy of Organic Compounds-3rd Ed.-P.S. Kalsi (New Age, New Delhi) 2000.
20. E.A.V. Ebsworth, D.W.H. Ranklin and S. Cradock: Structural Methods in Inorganic Chemistry, Blackwell Scientific, 1991.
21. J. A. Iggo: NMR Spectroscopy in Inorganic Chemistry, Oxford University Press, 1999.
22. C. N. R. Rao and J. R. Ferraro: Spectroscopy in Inorganic Chemistry, Vol I & II (Academic) 1970.
23. Spectroscopy, B. P. Straughan and S. Salker, John Wiley and Sons Inc., New Yourk, Vol.2, 1976.

PRACTICALS – HARD CORE

Course Title: ANALYTICAL CHEMISTRY PRACTICALS

Course Code: CHB 130

Course Outcomes

After studying this course the student to:

COID	CO
CO1	Analyze various samples with different classical and simple instrumental skills.

CO2	Obtain knowledge for selection of analytical methods with suitable technique being adopted for the analysis different samples like, water, laboratory chemicals and reagents, body fluids such as urine etc.
CO3	Distinguish classical and instrumental methods.
CO4	Propose and conduct experiment for quantification of individual analytes.

[128 HOURS]

PART – I

1. Determination of total acidity of vinegar and wines by acid-base titration.
2. Determination of purity of a commercial boric acid sample, and Na_2CO_3 content of washing soda.
3. Determination of relative equivalent weight of a weak organic acid by titration with NaOH .
4. Determination of ephedrine and aspirin in their tablet preparations by residual acid-basetitrimetry.
5. Determination of carbonate and bicarbonate in a mixture by $p\text{H}$ -metric titration and comparison with visual acid-base titration.
6. Determination of carbonate and hydroxide-analysis of a commercial washing soda by visual and $p\text{H}$ -titrimetry.
7. Determination of purity of a commercial sample of mercuric oxide by acid-base titration.
8. Determination of benzoic acid in food products by titration with methanolic KOH in chloroform medium using thymol blue as indicator.
9. Determination of the $p\text{H}$ of hair shampoos and $p\text{H}$ determination of an unknown soda ash.
10. Analysis of water/ waste water for acidity by visual, $p\text{H}$ metric and conductometric titrations.
11. Analysis of water/ waste water for alkalinity by visual, $p\text{H}$ metric and conductometric titrations.
12. Determination of ammonia in house-hold cleaners by visual and conductometric titration.
13. Determination of chromate and dichromate in mixture by acid-base titration: visual and $p\text{H}$ metric methods.
14. Potentiometric determination of the equivalent weight and K_a for a pure unknown weak acid.
15. Determination of purity of aniline by non-aqueous acid-base titration by visual and potentiometric methods.
16. Determination of purity of ethylene glycol and glycerol by oxidimetric method using periodate (Malprade reaction).
17. Spectrophotometric determination of creatinine and phosphorus in urine.

18. Flame emission spectrometric determination of sodium, potassium and calcium in river/ lake water.

PART – II

1. Determination of percentage of chloride in a sample by precipitation titration- Mohr, Volhard and Fajan's methods.
2. Determination of silver in an alloy and Na_2CO_3 in soda ash by Volhard method.
3. Mercurimetric determination of chloride in blood or urine.
4. Determination of total hardness, calcium and magnesium hardness and carbonate and bicarbonate hardness of water by complexation titration using EDTA.
5. Determination of calcium in calcium gluconate/ calcium carbonate tablets/ injections and of calcium in milk powder by EDTA titration.
6. Determination of zinc in a sample of foot powder and thallium in a sample of rodenticide by EDTA titration.
7. Analysis of commercial hypochlorite and peroxide solution by iodometric titration.
8. Determination of copper in an ore/ an alloy by iodometry and tin in stibnite by iodimetry.
9. Determination of ascorbic acid in vitamin C tablets by titrations with KBrO_3 and of vitamin C in citrus fruit juice by iodimetric titration.
10. Determination of iron in razor blade by visual and potentiometric titration using sodium metavanadate.
11. Determination of iron in pharmaceuticals by visual and potentiometric titration using cerium(IV) sulphate.
12. Determination of nickel in steel by synergic extraction and boron in river water/ sewage using ferroin.
13. Determination of total cation concentration of tap water by ion-exchange chromatography.
14. Determination of magnesium in milk of magnesium tablets by ion-exchange chromatography.
15. Cation exchange chromatographic separation of cadmium and zinc and their estimation by EDTA titration.
16. Gas chromatographic determination of ethanol in beverages.
17. Solvent extraction of zinc and its spectrophotometric determination.
18. Anion exchange chromatographic separation of zinc and magnesium followed by EDTA titration of the metals.
19. Separation and determination of chloride and bromide on an anion exchanger.
20. Separation of *o*- and *p*-nitroaniline and analysis by thin layer chromatography.

References

1. Fundamental of Analytical Chemistry, D.A. Skoog, D.M. West, Holler and Crouch 8th edition, 2005, Saunders College Publishing, New York.
2. Analytical Chemistry, G.D. Christian, 5th edition, 2001 John Wiley & Sons, Inc, India.
3. Quantitative Analysis, R.A. Day and A.L. Underwood, 6th edition, 1993, Prentice Hall, Inc. New Delhi.
4. Vogel's Textbook of Quantitative Chemical Analysis, J. Mendham, R.C. Denney, J.D. Barnes and M.J.K. Thomas, 6th edition, Third Indian Reprint, 2003, Pearson Education Pvt. Ltd., New Delhi.
5. Analytical Chemistry Principles, John H. Kennedy, 2nd edition, Saunders College Publishing, California, 1990.
6. Practical Clinical biochemistry methods and interpretations, R. Chawla, J.P. Bothers Medical Publishers (P) Ltd., 1995.
7. Laboratory manual in biochemistry, J. Jayaraman, New Age International Publishers, New Delhi, 1981.
8. Practical Clinical Biochemistry by Harold Varley and Arnold.Heinmann, 4th edition.

Course Title: INORGANIC CHEMISTRY PRACTICALS

Course Code: CHB 140

Course Outcomes

After studying this course the student to:

COID	CO
CO1	Prepare reagents required for analysis.
CO2	Propose and conduct experiment for quantitative analysis of inorganic samples such as ore, metals, complexes mixture of metals and complexes etc.
CO3	Propose schemes for semi-micro qualitative analysis.
CO4	Develop skills for the scientific and relevant documentation and risk and security assessment.

[128 HOURS]

PART – I

1. Determination of iron in haematite using cerium(IV) solution (0.02M) as the titrant, and gravimetric estimation of insoluble residue.
2. Estimation of calcium and magnesium carbonates in dolomite using EDTA titration, and gravimetric analysis of insoluble residue.

3. Determination of manganese dioxide in pyrolusite using permanganate titration.
4. Quantitative analysis of copper-nickel in alloy/ mixture:
 - a) Copper volumetrically using KIO_3 .
 - b) Nickel gravimetrically using DMG
5. Determination of lead and tin in a mixture: Analysis of solder using EDTA titration.
6. Quantitative analysis of chloride and iodide in a mixture:
 - a) Iodide volumetrically using KIO_3
 - b) Total halide gravimetrically
7. Gravimetric analysis of molybdenum with 8-hydroxyquinoline.
8. Micro-titrimetric estimation of :
 - a) Iron using cerium(IV)
 - b) Calcium and magnesium using EDTA
9. Quantitative estimation of copper(II), calcium(II) and chloride in a mixture.
10. Circular paper chromatographic separation of: (Demonstration)
 - a) Iron and nickel
 - b) Copper and nickel

PART – II

Semimicro qualitative analysis of mixtures containing **TWO** anions and **TWO** cations (excluding sodium, potassium and ammonium cations) and **ONE** of the following less common cations: W, Mo, Ce, Th, Ti, Zr, V, U and Li.

References

1. Vogel's Text Book of Quantitative Chemical Analysis – 5th edition, J. Basset, R.C. Denney, G.H. Jeffery and J. Mendhom.
2. A Text Book of Quantitative Inorganic Analysis by A.I. Vogel, 3rd edition.
3. Spectrophotometric Determination of Elements by Z. Marczenko.
4. Vogel's Qualitative Inorganic Analysis – Svelha.
5. Macro and Semimicro Inorganic Qualitative Analysis by A.I. Vogel.
6. Semimicro Qualitative Analysis by F.J. Welcher and R.B. Halin.
7. Quantitative Chemical Analysis by Daniel C. Harris, 7th edition, (2006).

Course Title: ORGANIC CHEMISTRY PRACTICALS**Course Code: CHB 150****Course Outcomes**

After studying this course, the student to:

COID	CO
CO1	Prepare several simple organic compounds and also propose suitable mechanisms.
CO2	Acquire knowledge of different reactions, conditions to be maintained, precautions to be exercised before/during/after the reaction.
CO3	Learn qualitative analysis and to separate a mixture of two components.
CO4	Gain confidence to set up reactions individually either in the pharma industry or for the Research.

[128 HOURS]**PART – I**

1. Preparation of *p*-nitro aniline from acetanilide.
2. Preparation of *p*-bromo aniline from acetanilide.
3. Preparation of benzoic acid from benzaldehyde
4. Preparation of n-butyl bromide from n-butanol.
5. Preparation of *p*-nitroiodobenzene from paranitroaniline.
6. Preparation of aniline from nitrobenzene.
7. Preparation of β -D-Glucose penta acetate.
8. Preparation of phenoxy acetic acid.
9. Preparation of cyclohexanone from cyclohexanol.
10. Preparation of chalcone.
11. Preparation of *S*-benzylthiuronium chloride.
12. Condensation of anthracene and maleic anhydride (Diels-Alder reaction).
13. Preparation of *m*-nitrobenzoic acid from methyl benzoate.

PART – II

Qualitative analysis: Separation of binary mixtures, identification of functional groups and preparation of suitable solid derivatives.

References

1. Manual of Organic Chemistry -Dey and Seetharaman.
2. Modern Experimental Organic Chemistry by John H. Miller and E.F. Neugil, p 289.
3. An Introduction to Practical Organic Chemistry -Robert, Wingrove etc.

4. A Text Book of Practical Organic Chemistry–A.I. Vogel, Vol.III.
5. Practical Organic Chemistry, Mann & Saunders.
6. Semimicro Qualitative Organic Analysis by Cheronis, Entrikin and Hodnet.
7. J. N. Guthru & R. Kapoor, Advance experimental Chemistry, New Delhi-1991.
8. R. K. Bansal, Laboratory Manual of Organic Chemistry, New PGE International (P) LTd. London, 3rd edition. 1996.18
10. N. K. Visno, Practical Organic Chemistry, New PGE International (P) Ltd. London, 3rd edition, 1996.

Course Title: PHYSICAL CHEMISTRY PRACTICALS

Course Code: CHB 160

Course Outcomes

After studying this course, the student to:

COID	CO
CO1	An idea about handling of instruments like UV-Visible Spectrophotometer, Potentiometer, pH meter, etc.
CO2	Determine the concentration of the species in given solutions using kinetic methods.
CO3	Distinguish between different physical properties of substances or compounds.
CO4	Acquire knowledge of different thermodynamic parameters.

[128 HOURS]

PART – I (Non-instrumental)

1. Study of kinetics of hydrolysis of an ester using HCl/ H₂SO₄ at two different temperature, determination of rate constants and energy of activation.
2. Study of kinetics of the iodine-hydrogen peroxide clock reaction.
3. Determination of activation energy for the bromide-bromate reaction.
4. Determination of heat of solution of benzoic acid by variable temperature method (graphical method).
5. Determination of partial molar volume of NaCl-H₂O system.
6. Determination of critical solution temperature of phenol-water system.
7. Binary analysis of two miscible liquids by viscometric method (Ethanol & Water).
8. To study oscillating or periodic or rhythmic reactions of malonic acid.
9. Thermometric titration of hydrochloric acid with NaOH.

10. Kinetics of photodegradation of indigocarmine(IC) using ZnO as photocatalyst and study the effect of [ZnO] and [IC] on the rate of photodegradation.

PART – II (Instrumental)

1. Conductometric titration of a mixture of HCl and CH₃COOH against NaOH.
2. Conductometric titration of orthophosphoric acid /formic acid/ oxalic acid against NaOH and NH₄OH.
3. Determination of PI of glycine by potentiometric method.
4. Potentiometric titration of KI vs KMnO₄ solution.
5. pH Titration of (a) polybasic acid(H₃PO₄), (b) (CH₃COOH+HCl) and (c) CuSO₄vsNaOH and determination of K_a.
6. To obtain the absorption spectra of colored complexes, verification of Beer's law and estimation of Ni⁺² ions from [Ni(NH₃)₆]²⁺ by spectrophotometry.
7. Analysis of binary mixture (Glycerol and Water) by the measurement of refractive index.
8. Study the kinetics of reaction between CAT and indigo carmine spectrophotometrically and determination of rate constant.
9. Spectrophotometric titration of FeSO₄ against KMnO₄.
10. Determination of the molecular weight of a polymer material by viscosity measurements (cellulose acetate/methyl acrylate).

References

1. Practical Physical Chemistry – A.J. Findlay.
2. Experimental Physical Chemistry – F. Daniels *et al.*
3. Selected Experiments in Physical Chemistry – Latham.
4. Experiments in Physical Chemistry – James and Prichard.
5. Experiments in Physical Chemistry – Shoemaker.
6. Advanced Physico-Chemical Experiments – J. Rose.
7. Practical Physical Chemistry – S.R. Palit.
8. Experiments in Physical Chemistry – Yadav, Geol Publishing House.
9. Experiments in Physical Chemistry – Palmer.
10. Experiments in Chemistry – D.V. Jahagirdar, Himalaya Publishing House, Bombay, (1994).
11. Experimental Physical Chemistry – R.C. Das and B. Behera, Tata Mc Graw Hill.

THEORY – SOFT CORE

All the courses are same as that of I Semester and a student can chose any course of his/ her choice provided that the same course has not been studied in the I Semester.

THIRD SEMESTER

THEORY – HARD CORE

Course Title: INSTRUMENTAL METHODS OF ANALYSIS

Course Code: CHC 030

Course Outcomes:

After studying this course, the student to:

COID	CO
48402	Learn the details of Electro analytical methods, Electrogravimetric analysis, Coulometric and Amperometric, Voltammetry.
48415	Understand in details with examples Thermal method of analysis, Thermogravimetric analysis, Differential thermal analysis and Differential scanning calorimetry.
48422	Learn in details with examples Enthalpimetric analysis, Microscopic analysis.

UNIT – I

Flame photometry and Atomic absorption spectrometry: Energy level diagrams-atomic absorption spectra. Flame characteristics. Flame atomizers and electrothermal atomization. Comparison of spectral interferences, chemical and physical interferences in FP and AAS. Background correction methods in AAS. Use of organic solvents. Quantitative techniques-calibration curve procedure and the standard addition technique. Typical commercial instruments for FP and AAS (Single and double beam atomic absorption spectrophotometers), applications of FES and AAS. Qualitative analysis and quantitative evaluations. Relative detectabilities of atomic absorption and flame emission spectrometry.

Molecular luminescence spectrometry: Theoretical basis for fluorescence and phosphorescence. Singlet and triplet excited states. Variables affecting luminescence-quantum efficiency, transition types, structure and structural rigidity, temperature and solvent effects, effect of pH, dissolved oxygen and concentration effect. Excitation spectra vs emission spectra. Origin of fluorescence, relationship between fluorescence and concentration. Fluorescence instrumentation-fluorometers and spectrofluorometers. Sensitivity and selectivity. Modification necessary to measure phosphorescence. Applications of fluorometry: inorganic and organic analyses.

Nephelometry and turbidometry: Principles, instrumentation and applications.

[16 HOURS]

UNIT – II

Electroanalytical methods: Classification. Potentiometers, galvanostats and potentiostats.

Potentiometric methods of analysis. Potentiometric electrochemical cells. The Nernst equation. Liquid junction potentials. Reference electrodes-SHE, calomel electrode and silver/silver chloride electrode. Metallic indicator electrodes-electrodes of first kind and second kind. Redox electrodes. Membrane electrodes –membrane potential, selectivity of membranes. Glass ion selective electrodes. Crystalline solid state ion selective electrodes. Liquid-based ion selective electrodes. Gas sensing electrodes. Potentiometric biosensors. Quantitative applications. Activity vs concentration. Quantitative analysis using external standards and the method of standard additions. Measurement of *pH*. Clinical and environmental applications.

Electrogravimetric analysis: Theory, apparatus, cell processes, deposition and separation, electrolytic separation of metals, applications.

Coulometric methods of analysis: General discussion, coulometry at controlled potential, apparatus and general technique, applications, coulometric titrations (amperometric/coulometric)-principles, apparatus, comparison of coulometric titrations with conventional titrations, automatic coulometric titrations, applications.

Amperometric titrations: Principle, titration curve, apparatus and techniques, applications.

Voltammetry: Fundamentals of voltammetry. Cyclic voltammetry: Principles and applications. Stripping analysis: Stripping voltammetry-basic principles, electrodes used for stripping analysis, apparatus for stripping analysis, applications, determination of lead in water voltammetry with micro electrodes.

[16 HOURS]

UNIT – III

Thermal method of analysis: Introduction,

Thermogravimetric analysis (TGA): Types of thermogravimetric analysis, principles and general thermal decomposition curve. Factors affecting the results-heating rate, furnace, instrument control/ data handling. Applications-purity and thermal stability, evaluation of correct drying temperature, analysis of complex mixture and determination of kinetic parameters of thermal degradation.

Differential thermal analysis (DTA): General principles. Theory-variables affecting the DTA curves. Instrumentation. Applications-analysis of the physical mixtures and thermal behaviour study. Determination of melting point, boiling point and decomposition point.

Differential scanning calorimetry (DSC): Basic principle. Instrumentation-power compensated DSC, Heat flux DSC. Applications- studies of thermal transitions and isothermal crystallization. Testing the purity of the pharmaceutical samples.

Thermomechanical analysis. Dynamic mechanical analysis.

Enthalpimetric analysis: Thermometric titrations and direct injection enthalpimetry: Principles, apparatus and applications.

Microscopic analysis: Principle and mechanism in characterization of compounds by scanning electron and transmission electron microscopic (SEM & TEM) techniques. Components of instruments of SEM and TEM.

[16 HOURS]

References

1. Fundamental of Analytical Chemistry, D.A. Skoog, D.M. West, Holler and Crouch 8th edition, 2005, Saunders College Publishing, New York.
2. Analytical Chemistry, G.D. Christian, 5th edition, 2001 John Wiley & Sons, Inc. India.
3. Quantitative Analysis, R.A. Day and A.L. Underwood, 6th edition, 1993 Prentice Hall, Inc. New Delhi.
4. Vogel's Textbook of Quantitative Chemical Analysis, J. Mendham, R.C. Denney, J.D. Barnes and M.J.K. Thomas, 6th edition, Third Indian Reprint, 2003 Pearson Education Pvt. Ltd., New Delhi.
5. Analytical Chemistry Principles, John H. Kennedy, 2nd edition, Saunders College Publishing, California, 1990.
6. Instrumental Methods of Analysis by H.H. Willard, L.L. Merritt and J.A. Dean, 7th Edition, CBS Publishers, New Delhi, 1988.
7. Principles and Practice of Analytical Chemistry, F.W. Fifield and Kealey, 3rd edition, 2000, Blackwell Sci., Ltd. Malden, USA.
8. Modern Analytical Chemistry, David Harvey, McGraw Hill, New Delhi, 2000.
9. Introduction to Instrumental Analysis, Braun, Pharm. Med. Press. India.
10. Instant Notes of Analytical Chemistry, Kealey and Haines, Viva Books Pvt. Ltd., New Delhi, 2002.
11. Analytical Transmission Electron Microscopy, An Introduction for Operators Thomas, Jürgen, Gemming, Thomas., Springer, 2014.
12. Scanning Transmission Electron Microscopy, Imaging and Analysis. Pennycook, Stephen J., Nellist, Peter D. (Eds.), Springer, 2011.

Course Title: SPECTROSCOPY

Course Code: CHC 040

Course Outcomes:

After studying this course, the student to:

COID	CO
51691	Understand in details with examples UV-Visible and IR spectroscopy.
51705	Understand in depth Nuclear magnetic resonance spectroscopy, Chemical shift.
51710	Learn the characteristics of ¹³ C-NMR spectroscopy.
51714	Understand the details of Mass spectroscopy and problems.

UNIT – I

UV Visible spectroscopy: Introduction, electronic transitions, simple chromophoric groups - systems of extended conjugation - aromatic systems - types of auxochromes - Functions of auxochromes - absorption and intensity shift - types of transitions - transition probability - types of absorption bands - solvent effects and choice of solvent - effect of polarity on various type of bonds Woodward's empirical rules for predicting the wavelength of maximum absorption for conjugated dienes, cyclic trienes and polyenes, α,β -unsaturated aldehydes and ketones, benzene and substituted benzene rings.

IR spectroscopy: Introduction, instrumentation, sample handling, modes of vibrations, Hooke's law, Characteristic group frequencies and skeletal frequencies. Finger print region, Identification of functional groups - alkenes, aromatics, carbonyl compounds (aldehydes and ketones, esters and lactones), halogen compounds, sulphur and phosphorus compounds, amides, lactams, amino acids and amines. Factors affecting group frequencies and band shapes, conjugation, resonance and inductance, hydrogen bonding and ring strain. Tautomerism, *Cis-trans* isomerism. Applications of IR spectroscopy.

[16 HOURS]

UNIT – II

Nuclear magnetic resonance spectroscopy: General introduction and definition, magnetic properties of nuclei (magnetic moment, g factor) and theory of nuclear resonance. Larmor precession frequency, resonance condition and relaxation processes.

Chemical shift: Standards employed in NMR, factors affecting chemical shift, electronegativity, shielding and deshielding mechanism, van der Waals deshielding, H-bonding, diamagnetic and paramagnetic anisotropies. Spin-spin coupling, chemical shift values and correlation for protons bonded to carbon and other nuclei. Instrumentation and sample handling.

Equivalence and magnetic equivalence proton exchange reactions, effects of chiral center, complex spin-spin interaction, stereochemistry, hindered rotation, Karplus curve - variation of coupling constants with dihedral angles. Simplification of complex spectra: isotopic substitution, increasing magnetic field strength, double resonance, spin decoupling, contact shift reagents, FT-NMR: Principle and applications, variable temperature profile, Nuclear Overhauser Effect (NOE).

[16 HOURS]

UNIT – III

¹³C-NMR spectroscopy: Comparison of ¹H-NMR and ¹³C-NMR. Multiplicity - proton decoupling, noise decoupling, off resonance decoupling, selective proton decoupling, noise decoupling by FT mode, chemical shift, application of ¹³C-NMR. ³¹P & ¹⁹F, Two dimensional NMR.

Mass spectroscopy: Principles, instrumentation, different methods of ionization. EI, CI, FD and FAB, Ion separators - single focusing separator with magnetic deflection, double focusing analyzer, time-of-flight separator and quadrupole analyzer, Mass spectra – molecular ion, base peak, meta-stable peak. General rules for fragmentation pattern. Nitrogen rule, ortho effect, Hydrogen transfer rearrangement and McLafferty rearrangement. Mass spectral fragmentation of organic compounds (acids, esters, hydrocarbons, halogenated hydrocarbons, alcohols, carbonyl compounds, amines, ethers and heterocyclic compounds).

Composite problems: Problems involving the application of the above spectroscopic techniques for structural elucidation of organic molecules.

[16 HOURS]

References:

1. Spectroscopy, B.P. Straughan and S. Salker, John Wiley and Sons Inc., New York, Vol.2, 1976.
2. Organic Spectroscopy, William Kemp, English Language Book society, Macmillan, 1987.
3. Application of Absorption Spectroscopy of Organic Compounds, John R. Dyer, Prentice Hall of India Private Ltd., New Delhi, 1974.
4. Spectrometric Identification of Organic Compounds, 4th edition, Robert M. Silverstein, G. Clayton Bassler and Terence C. Morrill, John Wiley & Sons, New York, 1981.
5. Organic Spectroscopy, V.R. Dani, Tata McGraw-Hall Publishing Company Limited, New Delhi. 1995.
6. Interpretation of Carbon-13 NMR Spectra, F.W. Wehrli and T. Wirthin, Heyden, London, 1976.
7. NMR spectroscopy – Powai.
8. Introduction to spectroscopy 3^{ed}, Pavia, Lampman, Kriz.

THEORY – OPEN ELECTIVE

Course Title: SELECTED TOPICS IN CHEMISTRY

Course Code: CHC 600

Course Outcomes:

After studying this course, the student to:

COID	CO
CO1	Gain knowledge on importance of chemistry or different branches of chemistry.
CO2	Acquire knowledge about the use of chemistry in everyday life.
CO3	Get knowledge about the extraction and purification techniques.
CO4	Learn importance of metals in biology; natural products and also various physical aspects.

UNIT – I

Solvent extraction: Basics of solvent extraction principal and application of solvent extraction.

Purification techniques: Crystallization, fractional crystallization, distillation techniques (simple distillation, steam distillation, distillation under reduced pressure, fractional distillation).

Chromatography: Definition, terms, classification of chromatographic techniques, principles of column and planar chromatography

Column chromatography: gas chromatography, high performance liquid chromatography, ion exchange chromatographic method.

Planar chromatography: Paper chromatography and TLC principles, mechanism of separation and application.

Electrophoretic methods: principles, definition, terms, types and applications.

[16 HOURS]

UNIT – II

An overview of metals in Biology: Introduction, the element content of living systems, biological chemistry of hydrogen, the economical use of resources- abundance and availability. Biological need for and the behaviors of inorganic elements.

Basic coordination chemistry for biologists: Introduction, ionic bonding, covalent bonding, coordination geometry, crystal field and ligand field theory.

Metal assimilation pathways: Introduction, metal assimilation in bacteria, plants, fungi and in mammals (iron, copper and zinc).

Metals in medicine: Introduction, *cis*-platin, radioactive pharmaceuticals, lithium compounds in therapy.

[16 HOURS]

UNIT – III

Chemistry of natural products: Carbohydrates (classification and structure of glucose, fructose, galactose, sucrose, maltose and lactose, carbohydrates as source of energy and breakdown process.

Proteins: amino acids classification and structure of α -amino acid, zwitter ion, isoelectric point and its determination by electrophoretic method, Elementary aspect of primary and secondary structures.

Vitamins: Classification, importance of vitamin A, D, E, K, B & C.

[16 HOURS]

UNIT – IV

Thermodynamics: First and second laws of thermodynamics. Concept of entropy and free energy, entropy as a measure of unavailable energy. Entropy and free energy changes and spontaneity of process. Variation of free energy with temperature and pressure. **Chemical kinetics:** Factor affecting the rate of reaction. Order of reaction and its determination. Energy of activation and its determination. Assumption of activated complex theory. **Electrochemistry:** Arrhenius theory of strong and weak electrolytes. Assumptions of Debye-Huckel theory of strong electrolytes. Electrode potential and construction of electrochemical cells. Corrosion and its prevention. **Photochemistry:** Laws of photochemistry, quantum yield and its determination, photodegradation.

[16 HOURS]

References:

1. Arthur I Vogel, Elementary Practical Organic Chemistry, Part I, II and III, CBS Publishers and Distributors, New Delhi, India.
2. I.L. Finar, Organic Chemistry, ELBS Longmann, Vol. I and II, 1984.
3. S K. Ghosh, Advanced General Organic Chemistry, Book and Allied (P) Ltd, 1998.
4. Organic Spectroscopy, William Kemp, English Language Book society, Macmillan, 1987.
5. Application of Absorption Spectroscopy of Organic Compounds, John R. Dyer, Prentice Hall of India Private Ltd., New Delhi, 1974.

6. Spectrometric Identification of Organic Compounds, 4th edition, Robert M. Silverstein, G. Clayton Bassler and Terence C. Morrill, John Wiley & Sons, New York, 1981.
7. Basic Inorganic Chemistry- 3rd edition, F.A. Cotton, G. Wilkinson and P.L. Gaus, John Wiley and Sons, (2002).
8. Inorganic Chemistry Principles of Structure and Reactivity: James E. Huheey, Ellen A. Keiter, Richard L. Keiter, Okhil K. Medhi, Delhi University, New Delhi (2006)
9. Elements of Physical Chemistry – Lewis and Glasstone.
10. Physical Chemistry by P.W. Atkins, ELBS, 4th edition, Oxford University Press (1990).
11. Basic Physical Chemistry by W.J. Moore, Prentice Hall, New Delhi, (1986).
12. Physical Chemistry – G.M. Barrow, McGraw Hill International Service (1988).

THEORY – SOFT CORE

Course Title: INORGANIC CHEMISTRY-II

Course Code: CHC 500

Course Outcomes:

After studying this course, the student to:

CO1: Gain knowledge from basic concepts of ionic solids, modern concept of acids and bases.

CO2: Demonstrate and understand the basic principles of acid-base chemistry and nonaqueous solvents.

CO3: Acquire knowledge to handle homogeneous and heterogeneous catalysis.

UNIT – I

Ionic solids: Introduction, Characteristic structures of ionic solids (NaCl, CsCl, ZnS, fluorite, rutile, β -cristobalite and cadmium iodide). The rationalization of structures.

The energetics of ionic bonding: Lattice enthalpy and Born-Haber cycle. Calculation of lattice enthalpies. Comparison of experimental and theoretical values of lattice enthalpy. The Kapustinskii equation. Consequences of lattice enthalpies.

Modern concept of acids and bases: Lux-Flood and Usanovich concepts, solvent system and leveling effect. Hard-Soft Acids and Bases, Classification and Theoretical backgrounds.

Supercritical fluids: Properties of supercritical fluids and their uses as solvents. Supercritical fluids as media for inorganic chemistry.

[16 HOURS]

UNIT – II

Biological and Medicinal Applications: Organomercury, boron, silicon and arsenic compounds.

Catalysis: General principles-The language of catalysis. Homogeneous and heterogeneous catalysts.

Homogeneous catalysis: Alkene hydrogenation, hydroformylation, The Wacker's process, Monsanto acetic acid process and L-DOPA synthesis, alkene oligomerizations, water-gas shift reactions. Palladium catalysed C-C bond forming reactions.

Heterogeneous catalysis: Alkene polymerization: Ziegler-Natta catalysis, Fischer-Tropsch carbon chain growth.

Zeolites as catalysts for organic transformation: Uses of ZSM - 5

Alkene metathesis, hydroboration, arylation or vinylation of olefins (Heck reaction).

Hybrid catalysts: Tethered catalysis. Biphasic systems.

Hydrosilylation: Platinum catalyst, Asymmetric palladium catalyst, Rhodium catalysts for asymmetric ketone reduction.

Asymmetric catalysis: General features of chiral ligands and complexes; mechanisms and catalytic cycles in hydrogenation, isomerization, epoxidation and catalytic reactions of C-C bond formation.

[16 HOURS]

References:

1. Basic Organometallic Chemistry - B.D. Gupta and A.J. Elias, Universities Press (2010).
2. Organometallics - A Concise Introduction, 2nd edition, Christoph Elschenbroich and Albert Salzer VCH, (1992).
3. Inorganic Chemistry, 2nd edition, C.E. Housecroft and A.G. Sharpe, Pearson Education Ltd., (2005).
4. Inorganic Chemistry- 3rd edition, G.L. Miessler and D.A. Tarr, Pearson Education, (2004).
5. Inorganic Chemistry, 4th edition. P. Atkins, T. Overton, J. Rourke, M. Weller and F. Armstrong, Oxford University Press (2004).

Course Title: STRUCTURAL METHODS IN INORGANIC CHEMISTRY

Course Code: CHC 510

Course Outcomes:

After studying this course, the student to:

CO1: Gain knowledge from basic concepts of instrumentation.

CO2: Demonstrate and understand basic principles of spectroscopic techniques.

CO3: Acquire knowledge about applications of spectroscopic techniques.

UNIT – I

NMR spectroscopy: Basic principles, chemical shift and factors affecting it, coupling constants. ^{19}F , ^{31}P , ^{11}B - NMR and NMR of paramagnetic complexes. Double resonance technique, The Nuclear Overhauser Effect, Magnetic susceptibility measurements by Evan's method. NMR to solids.

ESR spectroscopy: Theory, presentation of the spectrum, hyperfine coupling, the g value and factors affecting the magnitude of the g value. Zero-field splitting and Kramers' degeneracy. Application to simple inorganic and organic free radicals and to metal complexes.

NQR spectroscopy: Theory, energies of the quadrupole transitions, instrumentation, effect of magnetic field on the spectra, relationship between electric field gradient and molecular structures. Applications - interpretation of $e^2\text{Qq}$ data, structural information from NQR data.

[16 HOURS]

UNIT – II

Vibrational spectroscopy: Introduction, theory of infrared absorption, theoretical group frequencies, correlation chart. Applications to coordination compounds - aquo, amine, urea, DMSO, *cis* and *trans* metal complexes. Change in spectra accompanying change in symmetry upon coordination (nitrite, sulphate, nitrate, perchlorate and carbonate)

Mossbauer spectroscopy: Theoretical basis, interpretation of Mossbauer spectra - isomer shift, quadrupole splitting and magnetic hyperfine structures. Application: $\text{I}_2\text{Br}_2\text{Cl}_4$, $\text{Fe}_3(\text{CO})_{12}$, Prussian blue, nitroprusside, hexacyanoferrate.

Photoelectron spectroscopy: Introduction, principles, chemical shifts, photoelectron spectra of simple molecules, X-ray photoelectron and Auger electron spectroscopy. Applications.

Mass spectrometry: Theory, experimental techniques, molecular ions, fragmentation and ion reaction, Applications to coordination compounds.

[16 HOURS]

References:

1. Electronic Absorption Spectroscopy and Related Techniques – D.N. Sathyanarayana, Universities Press (2001).
2. Structural Methods in Inorganic Chemistry – E.A.V. Ebsworth, D.W.H. Ranklin and Cradock, Blackwell Scientific Publications (1988).
3. Physical Methods in Inorganic Chemistry – R.S. Drago, Saunders Publishers (1966).

Course Title: BIOPHYSICAL CHEMISTRY AND POLYMERS

Course Code: CHC 520

Course Outcomes:

After studying this course, the student to:

CO1: Learn electrophoresis, kinetics of polymerization, phase transition in polymer, polymers in solutions.

CO2: Distinguish different types of electrophoresis like free electrophoresis, zone electrophoresis, gel electrophoresis.

UNIT – I

Electrokinetic phenomena: Electrophoresis - principles of free electrophoresis, zone electrophoresis, gel electrophoresis and its applications in qualitative and quantitative study of proteins. Determination of isoelectric point of a protein. Electroosmosis and streaming potential and its biological significance. Biological significance of Donnan membrane phenomenon. Micelles and its involvement during digestion and absorption of dietary lipids. Diffusion of solutes across biomembranes and its application in the mechanism of respiratory exchange. “Salting In” and “Salting Out” of proteins. Osmotic behaviour of cells and osmo-regulation and its application in the evolution of excretory systems of organisms. Effect of temperature and *pH* on the viscosity of biomolecules (albumin solution). Significance of viscosity in biological systems - mechanism of muscle contraction, detection of intrastrand disulfide bonds in proteins, polymerization of DNA and nature of blood flow through different vessels. Effect of temperature, solute concentration (amino acids) on surface tension. Biological significance of surface tension - stability of Alveoli in lungs, interfacial tension in living cells (Danielli and Davson model). Application of sedimentation velocity and sedimentation equilibrium method for molecular weight determination of proteins.

[16 HOURS]

UNIT – II

Polymers: Fundamentals of polymers - monomers, repeat units, degree of polymerization. Linear, branched and network polymers. Classification of polymers. Polymerization - condensation, addition, free radical, ionic, co-ordination polymerization and ring opening polymerization. Molecular weight and size. Polydispersion. Average molecular weight concepts – number, weight and viscosity average molecular weight. Determination of molecular weights - viscosity method, osmotic pressure method, sedimentation and light scattering method.

Kinetics of Polymerization - condensation, addition, free radical, ionic, co-ordination polymerization. Kinetics of copolymerisation and polymer degradation.

Phase transitions in polymers and thermal characterization: Glass transition, crystallinity and melting- correlation with the polymer structure.

Polymers in solution: Criteria of polymer solubility. Thermodynamics of polymer solutions.

[16 HOURS]

References

1. Introduction to Physical Organic Chemistry, R.D. Gilliom, Madison – Wesley, USA (1970).
2. Physical Organic Chemistry, Reaction Rate and Equilibrium Mechanism – L.P. Hammett, McGraw HillBook, Co., (1970).
3. Biophysical Chemistry, Principle and Technique – A. Upadhyay, K. Upadhyay and N. Nath, Himalaya Publishing House, Bombay, (1998).
4. Essentials of Physical Chemistry and Pharmacy – H. J. Arnikaar, S. S. Kadam, K.N. Gujan, Orient Longman, Bombay, (1992).
5. Text book of polymer Science. F.W. Billmeyer, Jr., John Wiley. London (1994).
6. Polymer Science. V. R. Gowrikar, N. V. Vishwanathan and J. Sreedhar, Wiley Eastern, New Delhi (1990).
7. Fundamentals of Polymer Science and Engineering. A. Kumar and S.K. Gupta, Tata – McGraw Hill New Delhi (1978).
8. Polymer Characterization, D. Campbell and J. R. White, Chapman and Hall, New York.

Course Title: APPLICATIONS OF CHEMICAL KINETICS AND QUANTUM CHEMISTRY

Course Code: CHC 530

Course Outcomes:

After studying this course, the student to:

CO1: Understand the fundamentals of polymers, degree of polymerization and classification of polymers.

CO2: Acquire knowledge on different methods for the classification of compounds based on their molecular weights.

CO3: Develop knowledge on different methods for the classification of homogeneous catalysis.

UNIT - I

Homogenous catalysis: Acid-base catalysis, specific acid and base catalysis. General acid and base catalysis. Oxidation of amino acids and carbohydrates in presence of acid and base catalysis. Acidity functions - Bronstead, Hückel, Hammett and Bunnett hypothesis.

Chain reactions: Rice-Herzfeld mechanism for the thermal decomposition of acetaldehyde, Kinetics of explosive reactions, explosion limits (H_2 and O_2 reaction). Kinetics of autocatalytic and oscillatory chemical reactions, oscillatory chemical reaction of oxidation of malic acid by bromate ion catalyzed by Ce(III). Catalyzed and uncatalyzed reaction: Ru(III) catalyzed oxidation reaction of primary amines by chloramine – T in HCl medium.

[16 HOURS]

UNIT – II

Applications of quantum chemistry: Variation theorem- statement and proof. Application of variation method to He atom, the structure of many electron systems/ atoms (secular equations & determinants), Spin-orbit interaction, antisymmetry and Pauli exclusion principle. Angular momenta (commutations, relations, operators), Term symbols, Russell-Saunders terms and coupling schemes, Slater orbitals and SCF method for many electron systems.

Molecular wave functions: Born-Oppenheimer approximations. Covalent bond –valence bond and molecular orbital approaches with comparisons. MO theory applied to homonuclear and heteronuclear diatomics by LCAO methods, correlation diagrams, non-crossing rule.

Theory of directed valence-hybridization and geometry of molecules in terms of molecular orbitals (bond angle, dihedral angle), localised and delocalised molecular orbitals.

Conjugated and aromatic molecules: Huckel molecular orbital (HMO) theory of linear conjugated systems (ethane & allyl systems) and aromatic molecules (benzene as an example). Calculation of delocalization energies, bond order & charge density.

[16 HOURS]

References:

1. Statistical Thermodynamics by B.C. McLelland, Chapman and Hall, London (1973).
2. Elementary Statistical Thermodynamics by N.D. Smith, Plenum Press, NY (1982).
3. Elements of Classical and Statistical Thermodynamics by L.K. Nash, Addison-Wesley (1970).
4. Statistical Thermodynamics by I.M. Klotz.
5. Introduction to Statistical Thermodynamics by M. Dole, Prantice Hall, (1962).
6. Text Book of Physical Chemistry by Samuel Glasstone, McMillan Indian Ltd., 2nd edition (1974).

7. Elements of Physical Chemistry by Lewis and Glasstone.
8. Physical Chemistry by P.W. Atkins, ELBS, 4th edition, Oxford University Press (1990).
9. Kinetics and Mechanism of Chemical Transformation by J. Rajaram and J.C. Kuriacose.
10. Chemical Kinetics – L.K. Jain.
11. Chemical Kinetics – Benson.
12. Physical Organic Chemistry, Reaction Rate and Equilibrium Mechanism – L.P. Hammett, McGraw HillBook, Co., (1970).
13. Fundamentals of Photochemistry – Rohatgi and Mukherje (New Age Bangalore) 2000.

PRACTICALS – SOFT CORE

Course Title: ANALYTICAL CHEMISTRY PRACTICALS

Course Code: CHC 210

Course Outcomes:

After studying this course, the student to:

COID	CO
CO1	Get experience on analysis of various complex mixtures by following multistep reactions.
CO2	Acquire the knowledge on handling instruments and to overcome the general problems arises during the analysis.
CO3	Acquire industrial skills required for sampling, analytical and interpretation and presentation of results.
CO4	Possess adequate knowledge on literature search for developed analytical methods.

[128 HOURS]

PART – III

1. Determination of calcium in limestone by redox, acid-base and complexation titrations.
2. Determination of vitamin C in orange juice by titration with cerium(IV) and with 2,6-dichlorophenol indophenol.
3. Determination of mercury in an algacide by EDTA titration; and arsenic in ant control preparation by redox titration.
4. Determination of aluminium and magnesium in antacids by EDTA titration.
5. Analysis of a copper-nickel alloy sample for copper and nickel by EDTA titration using masking and selective demasking reactions.
6. Determination of saccharin in tablets by precipitation titration.
7. Determination of iodine value and saponification value of edible oils.

- Determination of ascorbic acid in goose berry/bitter gourd by titrimetry and spectrophotometry using *N*-bromosuccinimide (NBS).
- Analysis of a mixture of iron(II) and iron(III) by EDTA titration using *pH* control.
- Determination of sulpham drugs by potentiometry using NaNO_2 and iodometric assay of penicillin.
- Solvent extraction method for determination of silver as ion-associate with 1,10-phenanthroline and bromopyragallol red.
- Electrolytic determination of copper and lead in brass.
- Polarographic determination of copper and zinc in brass.
- Determination of sodium, potassium and calcium in mineral waters by atomic emission spectrometry.
- Determination of iron in mustard seeds and phosphorus in peas by spectrophotometry.
- Analysis of waste water for anionic detergents and phenol by spectrophotometry.
- Fluorimetric determination of riboflavin (vit. B₂) in tablets.
- Colorimetric analysis of procaine by diazotization and coupling reaction.
- Determination of manganese in steel by extraction-free spectrophotometry and molybdenum in steel by extractive spectrophotometry.
- Determination of ethanol in wine by titrimetric and spectrophotometric dichromate methods

PART – IV

- Analysis of waste waters for DO and COD by titrimetry.
- Analysis of a ground water sample for sulphate by titrimetry (EDTA) and turbidimetry.
- Potentiometric determination of formula and stability constant of a silver-ammonia complex ion.
- Determination of aspirin, phenacetin and caffeine in mixture and APC tablets by solvent extraction and UV spectrophotometry.
- Kinetic determination of urinary creatinine and purity of a commercial H_2O_2 sample.
- Determination of chromium(III) and iron(III) in a mixture by kinetic masking methods.
- Catalytic determination of traces of selenium in biological materials and iodide in blood serum.
- Photometric and potentiometric titration of iron(III) with EDTA.
- Photometric and potentiometric titration of copper with EDTA.
- Determination of copper(II) and iron(III) in mixture by photometric titration with EDTA.

11. Analysis of brackish water for chloride content by a) spectrophotometry (mercuric thiocyanate method), b) conductometry (silver nitrate) and c) potentiometry (silver nitrate).
12. Conductometric titration of sodium acetate with HCl and NH_4Cl with NaOH.
13. Ascorbic acid determination in natural orange juice by coulometry.
14. Spectrophotometric determination of iron in natural waters using thiocyanate and 1,10-phenanthroline as reagents.
15. Determination of fluoride in drinking water/ground water by spectrophotometry(alizarin red lake method).
16. Analysis of waste water for
 - a) phosphate by molybdenum blue method
 - b) ammonia-nitrogen by Nessler's method
 - c) nitrite-nitrogen by NEDA method
15. Analysis of a soil sample for
 - a) calcium carbonate and organic carbon by titrimetry.
 - b) calcium and magnesium by EDTA titration.
16. Analysis of a soil sample for
 - a) Nitrogen content by Kjeldahl method
 - b) Available phosphorus by spectrophotometry.
 - c) Nitrate-nitrogen/nitrite nitrogen/ammonia nitrogen by spectrophotometry.
 - d) sodium and potassium by flame photometry.
17. Analysis of urine for
 - a) urea and uric acid by titrimetry and spectrophotometry.
 - b) Sulphate by precipitation titration after ion-exchange separation.
 - c) Sugar by Benedict's reagent.
18. Analysis of blood for
 - a) cholesterol by spectrophotometry
 - b) bicarbonate by acid-base titration.
19. Fluorimetric determination of quinine in an antimalarial tablet.

References

1. Fundamental of Analytical Chemistry, D.A. Skoog, D.M. West, Holler and Crouch 8th edition, 2005, Saunders College Publishing, New York.
2. Analytical Chemistry, G.D. Christian, 5th edition, 2001 John Wiley & Sons, Inc. India.

3. Quantitative Analysis, R.A. Day and A.L. Underwood, 6th edition, 1993, prentice Hall, Inc. New Delhi.
4. Vogel's Textbook of Quantitative Chemical Analysis, J. Mendham, R.C. Denney, J.D. Barnes and M.J.K. Thomas, 6th edition, Third Indian Reprint, 2003 Pearson Education Pvt. Ltd., New Delhi.
5. Analytical Chemistry Principles, John H. Kennedy, 2nd edition, Saunders College Publishing, California, 1990.
6. Quantitative Analysis of Drugs in Pharmaceutical Formulations, P. D. Sethi, 3rd edition, CBS Publishers & Distributors, New Delhi, 1997.
7. Practical Clinical biochemistry methods and interpretations, R. Chawla, J.P. Bothers Medical Publishers (P) Ltd., 1995.
8. Laboratory Manual in Biochemistry, J. Jayaraman, New Age International Publishers, New Delhi, 1981.
9. Experiments on Water Pollution, D.I. Williams and D. Anglesia, Wayland Publishers Ltd., England, 1978.
10. Experiments on Land Pollution, D.I. Williams and D. Anglesia, Wayland Publishers Ltd., England, 1978.
15. Experiments in Environmental Chemistry, P.D. Vowler and D.W. Counel, Pergamon Press, Oxford 1980.
16. Manual Soil Laboratory Testing, vol.I, K.H. Head, Pentech Press, London 1980.

Course Title: INORGANIC CHEMISTRY PRACTICALS

Course Code: CHC 220

Course Outcomes:

After studying this course, the student to:

COID	CO
CO1	Get experience on analysis of various complex mixtures by multistep reactions.
CO2	Acquire knowledge on handling instruments.
CO3	Acquire skills required for sampling, analytical and interpretation and presentation of results.
CO4	Possess adequate knowledge on literature search for developed preparative methods.
CO5	Synthesize and characterization of complexes.

PART – III

1. Determination of aluminium and bismuth by complexometric titration.
2. Determination of lead and tin in a mixture.
3. Determination of calcium and lead in a mixture by pH control and complexation method.
4. Determination of zinc, manganese and magnesium in a mixture using fluoride as a demasking agent.
5. Quantitative analysis of copper(II) and iron(II) in a mixture:
 - i. Copper gravimetrically as CuSCN and
 - ii. Iron volumetrically using cerium(IV) solution
6. Determination of iron as the 8-hydroxyquinolate by solvent extraction method.
7. Determination of the composition of iron-phenanthroline complex by:
 - (a) Job's method
 - (b) Mole-ratio method and
 - (c) Slope-ratio method.
8. Polarographic estimation of cadmium and zinc.
9. Spectrophotometric determinations of:
 - a. Titanium using hydrogen peroxide
 - b. Chromium using diphenyl carbazide in industrial effluents
 - c. Nickel using dimethylglyoxime in steel solution
10. Solvent extraction of ferric thiocyanate complex and determination by colorimetry.

PART – IV

1. Preparation of hexaamminecobalt(III) chloride and estimate cobalt ion.
2. Preparation and characterization of Chloropentaamminecobalt(III) chloride and estimate cobalt ion.
3. Using chloropentaamminecobalt(III) chloride, prepare nitro and nitritopentamine cobalt(III) chloride. Record the IR spectra of the isomers and interpret.
4. Preparation of potassium tris-oxalatochromate(III) trihydrate.
5. Preparation of mercurytetrathiocyanatocobaltate(II) and estimation of mercury by gravimetry.
6. Preparation of tetraamminecopper(II) sulphate tetrahydrate

7. Preparation and characterization of manganese dioxide nano-particles
8. Preparation of bis-dichlorotriphenyl phosphine nickel (II).
9. Preparation and characterization of hexaamminenickel(II) chloride.
10. Demonstration Experiments:
 - (a) Recording and interpretation of IR and NMR spectra of complexes.
 - (b) Spectrochemical series - Evaluation of Dq value.
 - (c) DNA interaction with metal complexes by UV-visible absorption and viscosity methods.

References

1. Basic principles of Practical Chemistry – V. Venkateswaran, R. Veeraswamy and A.R. Kulandraivelu
2. Instrumental Analysis Manual - Modern Experiments for Laboratory – G.G. Guilbault and L.G. Hargis.
3. A Text Book of Quantitative Inorganic Analysis – A.I. Vogel, 5th edition.
4. Experimental Inorganic Chemistry – G. Palmer.
5. Inorganic Synthesis – O. Glemser.
6. Experimental Inorganic/ Physical Chemistry- Mounir A. Malati.
7. Quantitative Chemical Analysis – Daniel C. Harris, (2006) 7th edition.
8. Spectrophotometric Determination of Elements – Z. Marczenko

Course Title: ORGANIC CHEMISTRY PRACTICALS

Course Code: CHC 230

Course Outcomes:

After studying this course, the student to:

COID	CO
CO1	Develop experience in multistep synthesis and also mechanisms.
CO2	Learn different kinds of reactions under multistep synthesis.
CO3	Learn isolation experiments, preliminary identification and separation.
CO4	Acquire knowledge of various estimations like sugars, enol content, ketones, nitro, protein etc.

PART – III**Multi step synthesis**

1. Oxidation of cyclohexanol to adipic acid via cyclohexanone
2. Preparation of benzocaine from *p*-nitrotoluene
3. Preparation of *p*-chlorobenzoic acid from *p*-toluidine (Sandmeyer's reaction)
4. Molecular rearrangement:
 - i. Preparation of *o*-chlorobenzoic acid from phthalic anhydride
 - ii. Preparation benzilic acid from benzaldehyde
 - iii. Preparation of *o*-hydroxy benzophenone from phenyl benzoate via Fries rearrangement
 - iv. Preparation of benzanilide from benzophenone (Beckmann rearrangement).
5. Grignard reaction: Preparation of triphenyl carbinol
6. Preparation of luminol from phthalic anhydride
7. Synthesis of isoxazolines and pyrazolines via 1,3-dipolar cycloaddition.
8. Synthesis of tetralones from aryl aldehydes.
9. Synthesis of *m*-chloriodobenzene from *m*-dinitrobenzene
10. Synthesis of Schiff base from nitro compound.

PART – IV**Isolation of natural products**

1. Fractional crystallization: separation of mixture of naphthalene and biphenyl
2. Fractional distillation: Separation of mixture of hexane and toluene.
3. Thin layer chromatography: Separation of plant pigments
4. Column chromatography: Separation of mixture of *o* and *p*-nitro anilines
5. Isolation of piperine from pepper
6. Isolation of caffeine from tea
7. Isolation of azeleic acid from castor oil
8. Isolation of clove oil from clove
9. Estimation of sugars by Fehlings method
10. Determination of enol content by Meyer's method
11. Estimation of ketones by haloform reaction
12. Estimation of sugars by Bertrand's method
13. Estimation of nitro groups
14. Estimation of protein by biuret method

Spectral analysis: Structural elucidation of some simple organic compounds by UV, IR, NMR and mass. The spectra have to be provided by the teachers.

References

1. Manual of Organic Chemistry - Dey and Seetharaman.
2. Modern Experimental Organic Chemistry by John H. Miller and E.F. Neugil, p 289.
3. An Introduction to Practical Organic Chemistry - Robert, Wingrove etc.
4. A Text Book of Practical Organic Chemistry – A.I. Vogel, Vol.III
5. Practical Organic Chemistry - Mann & Saunders
6. Semimicro Qualitative Organic Analysis by Cheronis, Entrikin and Hodnet .
7. R.K. Bansal, Laboratory Manual of Organic Chemistry, New Age International (P) Ltd. London, 3rd edition, 1996.

Course Title: PHYSICAL CHEMISTRY PRACTICALS

Course Code: CHC 240

Course Outcomes:

After studying this course, the student to:

COID	CO
CO1	Acquire knowledge on handling instruments and to overcome the general problems arises during the analysis.
CO2	Learn concepts of rate constants, energy of activation, order of the reaction and also thermodynamics parameters.
CO3	Learn concepts of kinetics experiments.

[128 HOURS]

PART – III (NON-INSTRUMENTAL)

1. Determination of energy of activation for reaction between sodium formate and iodine.
2. To study the kinetics of reaction between acetone and iodine-determination of order of reaction w.r.t. iodine and acetone.
3. Determination of rate of decomposition of hydrogen peroxide with manganese dioxide.
4. Determination of order and rate constant of hydrolysis of ethyl acetate in acid medium.
5. Kinetics of decomposition of benzene diazonium chloride, determination of energy of activation and thermodynamic parameters.
6. Kinetics of decomposition of diacetone alcohol by NaOH-determination of energy of activation.

7. To determine the eutectic point of a two component system (Naphthalene-*m*-dinitrobenzene system).
8. Study of phase diagram of a three component system (e.g. acetic acid-chloroform water and system). Construction of binodal curve and indicating tie line.
9. Determination of heat of solution and lattice energy of calcium chloride.
10. Determination of partition co-efficient of acetic acid in water and butanol.
11. Study of kinetics of reaction between $K_2S_2O_8$ and KI, first order, determination of rate constants at two different temperatures and E_a .
12. To determine the rate constant for the reaction glycine and CAT.

PART IV (INSTRUMENTAL)

1. Kinetics of saponification of ethyl acetate by conductivity method and study the effect of dielectric constant of the medium (using CH_3OH).
2. Simultaneous spectrophotometric determination of manganese and chromium in $KMnO_4$ and $K_2Cr_2O_7$ mixture.
3. Determination of ionic product of water and study the effect of temperature.
4. Coulometric titration I_2 vs $Na_2S_2O_3$.
5. Conductometric study of charge transfer complex of *p*-phenylenediamine with phthalic acid.
6. Determination of mean ionic activity coefficient of a weak electrolyte (acetic acid) by conductometric measurements.
7. Conductometric determination of the degree of hydrolysis and hydrolysis constant of aniline hydrochloride.
8. Conductometric titration of potassium iodide with mercuric perchlorate.
9. Determination of pK value of an indicator (methyl orange).
10. Potentiometric titration of mixture of $KCl+KBr+KI$ vs $AgNO_3$.
11. Conductometric titration of a mixture of HCl , CH_3COOH and $CuSO_4$ against $NaOH$.
12. Thermometric titration of HCl and H_3BO_3 with $NaOH$.
13. Determination of quantum yield for the photolysis of Chloramine-T.
14. Determination of quantum yield for the photolysis of Chloramine-B.

References

1. Practical Physical Chemistry – A.J. Findlay.
2. Experimental Physical Chemistry – F. Daniels *et al.*

3. Selected Experiments in Physical Chemistry – Latham.
4. Experiments in Physical Chemistry – James and Prichard.
5. Experiments in Physical Chemistry – Shoemaker.
6. Advanced Physico-Chemical Experiments – J. Rose.
7. Practical Physical Chemistry – S.R. Palit.
8. Experiments in Physical Chemistry – Yadav, Geol Publishing House.
9. Experiments in Physical Chemistry – Palmer.
10. Experiments in Chemistry – D.V. Jahagirdar, Himalaya Publishing House, Bombay, (1994).
11. Experimental Physical Chemistry – Das. R.C. and Behera B, Tata Mc Graw Hill.

FOURTH SEMESTER

THEORY – HARD CORE

Course Title: BIOINORGANIC CHEMISTRY

Course Code: CHD 010

Course Outcomes:

After studying this course, the student to:

COID	CO
47414	Understand in details with examples Structural and molecular biology, Bioenergetics, Sodium and potassium-channels and pumps, Biochemistry of calcium, Vitamin B12 and Coenzymes.
47426	Understand the characteristics of Electron transport proteins and redox enzymes, Non-redox metalloenzymes.
47438	Specify the classification and characteristics of Identify the details of Metal ion transport and storage, Oxygen transport and oxygen uptake proteins.
47447	Learn the details of Metals in medicine, Disease due to metal deficiency and treatment, Metal complexes as drugs and therapeutic agents, Treatment of toxicity.

UNIT – I

Structural and molecular biology: Introduction, The structural building blocks of proteins, the structural building block of nucleic acids. Metal ion interactions with nucleosides and nucleotides.

General features of DNA - metal complex interaction.

Bioenergetics: Introduction, Redox reactions in metabolism, the central role of ATP in metabolism. Kinetic stability of ATP, Mitochondrial flow of electrons from NADH to O₂. Oxidative phosphorylation and respiratory chain.

Sodium and potassium-channels and pumps: Introduction, transport across membranes. Potassium and sodium channels, The sodium-potassium ATPase, Macro cyclic crown ether compounds, cryptands and ionophores.

Biochemistry of calcium: Introduction - comparison of Ca²⁺ and Mg²⁺. Biological roles of calcium, binding sites of calcium and proteins, storage of calcium, calcium in muscle contraction, calcium in blood clotting process.

Vitamin B₁₂ and Coenzymes: Structural feature, names of different forms, chemistry of cobalamin, biochemical functions of cobalamins, model compounds. Special characteristics of B₁₂ co-enzyme.

[16 HOURS]

UNIT – II

Metal ion transport and storage:

Iron storage and transport: Transferrin, ferritin, phosvitin and gastroferrin.

Iron transport in microbes: siderophores, *in vivo* microbial transport of iron

Oxygen transport and oxygen uptake proteins: Properties of dioxygen (O₂): Thermodynamic and kinetic aspects of dioxygen as an oxidant, activation of dioxygen through complexation with metal ions.

Haemoglobin (Hb) and Myoglobin (Mb) in oxygen transport mechanism: Introduction to porphyrin system, substituent effects on porphyrin rings, functions of Hb and Mb. Characteristics of O₂⁻ binding interaction with Hb and Mb. Model compounds for oxygen carriers (Vaska's complex and cobalt(III) – Schiff base complexes). Hemerythrin and hemocyanin.

Electron transport proteins and redox enzymes: Iron – sulfur proteins (rubredoxins and ferredoxins) and cytochromes including cytochrome P450. Catalase and peroxidase: Structure and reactivity.

Superoxide dismutase: Structure and reactivity.

Molybdenum containing enzymes: Aspects of molybdenum chemistry, Xanthine oxidase, aldehyde oxidase, sulfite oxidase, nitrogenase and nitrite reductase.

Non-redox metalloenzymes - Structure and reactivity: Carboxypeptidase-A, alcohol dehydrogenase, leucine aminopeptidase and carbonic anhydrase.

[16 HOURS]

UNIT - III

Therapeutic uses of Metals - Metals in medicine: Introduction, metals and human biochemistry, general requirements.

Disease due to metal deficiency and treatment: Iron, zinc, copper, sodium, potassium, magnesium, calcium and selenium.

Metal complexes as drugs and therapeutic agents: Introduction, antibacterial agents, antiviral agents, metal complexes in cancer therapy, metal complexes for the treatment of rheumatoid arthritis, vanadium diabetes, metal complexes as radio diagnostic agents.

Treatment of toxicity due to inorganics: General aspects of mechanism of metal ion toxicity,

(i) Mechanism of antidote complex with poison, rendering it inert: arsenic, lead, mercury, iron, copper.

(ii) Antidote accelerated metabolic conversion of poison to non-toxic product: cyanide and carbon monoxide.

[16 HOURS]

References:

1. The Inorganic Chemistry of Biological Process- 2nd edition, M. N. Hughes, John Wiley and Sons, (1988).
2. Bioinorganic Chemistry - R.W. Hay, Ellis Horwood Ltd., (1984).
3. Biological Inorganic Chemistry – An Introduction, R.R. Crichton, Elsevier, (2008).
4. Bioinorganic Chemistry - A.K. Das, Books and Allied (P) Ltd, (2007).
5. Bioinorganic Chemistry - K. Hussain Reddy, New Age International Ltd. (2003).
6. Bioinorganic Chemistry: A Survey - Eiichiro Ochiai, Academic Press, (2008).
7. Bioinorganic Chemistry: A Short Course - 2nd edition, R.M. Roat-Malone, Wiley Interscience, (2007).
8. Medicinal Applications of Coordination Chemistry - Chris Jones and John Thornback, RSC Publishing, (2007).
9. Transition Metal Complexes as Drugs and Chemotherapeutic Agents - N. Farrell, Kluwer Academic Publishers (1989).
10. The Biological Chemistry of the Elements: The Inorganic Chemistry of Life - 2nd edition, J.J.R. Frausto da Silva and R.J.P. Williams, Oxford University Press,(2001).

Course Title: ADVANCED PHYSICAL CHEMISTRY

Course Code: CHD 020

Course Outcomes:

After studying this course, the student to:

COID	CO
47184	Learn in depth Photochemistry, Mechanism of absorption and emission of radiation, Photophysical kinetics.

UNIT – I

Kinetics and Thermodynamics of Polymerization: Kinetics of addition, condensation and radiation induced polymerization. Thermodynamics of polymer solutions, The Flory-Huggins Theory, Flory Krigbaum and modified Flory-Huggins Theory.

Copolymerization: Kinetics of copolymerization, Copolymer equation, Monomer reactive ratios. Properties of Polymers: Crystalline melting point and the glass transition temperature.

Polymer molecular weights: Molecular weight distribution, Number average and Weight average molecular weight. Methods for determination of molecular weight – Osmometry-membrane osmometry, vapour pressure osmometry, light scattering, viscosity, ultracentrifugation.

Conducting Polymers: Structure, properties, characterization and applications.

Polymer Degradation, Stability and Environmental Issues: Types of degradation, Thermal degradation, Mechanical degradation, Photodegradation, Degradation by high energy radiation. Chemical, hydrolytic and UV stability. Recycling and biodegradation.

[16HOURS]

UNIT – II

Photochemistry: Introduction to photochemistry, quantum yield and its determination, factors affecting quantum yield, Actinometry - Uranyl oxalate and potassium ferrioxalate actinometers, acetone and diethylketone actinometers. Photosensitization: by mercury, dissociation of H₂. Photochemical kinetics of: Decomposition of CH₃CHO, formation of HCl. Photodegradation: Photocatalyst – ZnO, TiO₂, principle, application of ZnO/ TiO₂ in the photo degradation of dyes (IC), pesticides (DDT) and in industrial effluents. Effect of photo degradation on COD value.

Mechanism of absorption and emission of radiation: Einstein's treatment, selection rules, Life times of excited electronic states of atoms and molecules Types of electronic transitions in organic

molecules photochemical pathways, Jablonski diagram, Fluorescence, Phosphorescence. Fluorescence emission, factors affecting fluorescence, viz. structure, solvent, pH, temperature etc. Triplet state and phosphorescence.

Photophysical kinetics: kinetics of unimolecular processes, delayed fluorescence mechanisms, kinetics of collisional quenching, Stern-Volmer equation, quenching by added substances charge transfer mechanism, energy transfer mechanism.

[16 HOURS]

UNIT – III

Nuclear Chemistry: Radioactive decay – General characteristics, decay kinetics, parent – daughter decay growth relationships, determination of half-lives, Nuclear stability – packing fraction, binding energy, Brief survey of alpha, beta and gamma decays. Nuclear reactions – Bethe's notation, types of nuclear reactions – specific nuclear reactions, photonuclear reactions, Oppenheimer – Phillips process, spallation reactions. Definition of Curie and related calculations. Szilard-Chalmers process. Geiger-Muller counters – G.M. Plateau, dead time, coincidence loss, determination of dead time.

Radiation Chemistry: Introduction, units, interaction of electromagnetic radiation with matter, G-value, LET of radiation, dosimetry, Fricke dosimeter. Radiolysis - cysteine, and biphenyl. Radioisotopes as tracers, use of isotopic tracers in the elucidation of reaction mechanism, structure determination and solubility of sparingly soluble substances. ¹⁴C dating, medical applications of isotopic tracers. Hazards in radiochemical work and radiation protection.

[16 HOURS]

References

1. Polymer Science and Technology by Joel R. Fried Third edition, 2002.
2. Polymer Science, V.R.Gowarikar , N.V.Vishwanathan and Jaydev Sreedhar Reprint edition, 2002.
3. Text book of Polymer Science, Fred W. Billmeyer Jr. Third edition, 2000.
4. Principles of Polymerization, George Odian, Third edition 2002.
5. Handbook on Conducting Polymers – T.A.Skotheim, Ed Marcel Dekker Inc, New York, 1 and 2, 1986.
6. Essentials of Nuclear Chemistry, H. J Arnikar, Wiley Eastern Limited, 4th Edition, (1995).

7. Nuclear and Radiochemistry, G. Friedlander, J. W. Kennedy and J. M. Miller, John Wiley (1981).
8. Introduction to Radiation Chemistry, J. W. T. Spinks and R. J. Woods, John Wiley (1990)
9. Introduction to Nuclear Physics and Chemistry, B.G. Harvey, Prentice hall (1963).
10. Sourcebook on Atomic Energy-S. Glasstone, Van Nostrand Company (1967).
11. Radiochemistry and Nuclear methods of analysis-W.D.Ehman and D.E. Vance, John Wiley (1991).
12. Fundamentals of photochemistry by K.K.Rohatgi-Mukherjee, New Age International Publishers Revised Edition (Reprint 2003).
13. Chemistry and light by Paul Suppan, The Royal Society of Chemistry.
14. Nuclear Chemistry by Friedlander and Kennedy, John Wiley and Sons (1987).
15. Nuclear Physics and Chemistry by G. Harvey.
16. Essentials of Nuclear Chemistry by H.J. Arnikar, Eastern Wiley (1990).
17. Nuclear Chemistry by U.N. Dash, Sultan Chand and Sons (1991).
18. Source Book on Atomic Energy by S. Glasstone, 3rd edition Van Nonstrand (1967).
19. Nuclear Chemistry by Friedlander and Kennedy, John Wiley and Sons (1987).
20. Essentials of Nuclear Chemistry by H.J. Arnikar, Eastern Wiley (1990).
21. Fundamentals of Radiochemistry by D.D. Sood, A.V.R. Reddy and N. Ramamoorthy

THEORY – SOFT CORE

Course Title: APPLIED ANALYSIS II

Course Code: CHD 500

Course Outcomes:

After studying this course, the student to:

CO1: Acquire knowledge on fertility and essential minor and major nutrients of soil for better growth of plants

CO2: Learn meaning, classification, characteristic features and components of fuels.

CO3: Describe feasible analytical methods for the quantitative analysis of fuels

CO4: Learn importance of analysis of different components of body fluids with adequate knowledge and skills to employ a suitable analytical method

CO5: Know background on forensic analysis with reference to its importance and analytical Methods.

UNIT – I

Soil Analysis: Inorganic and organic components of soil, collection and preparation of soil samples for analysis. Measurement of soil pH and conductivity. Determination of organic carbon, total nitrogen, 53 available nitrogen, ammonia nitrogen, nitrate nitrogen and nitrite nitrogen. Available phosphorus and sulphur-their determination. Analysis of soil for sodium, potassium and calcium and magnesium. Micronutrient elements and their analysis. Pesticide residues in soil, their separation and determination.

Fuel analysis- Fuels and their classification. Solid fuels and their classes - natural, artificial and industrial solid fuels. Coal and its analysis - proximate analysis and ultimate analysis. Liquid fuels and their types. Aniline point, flash point and fire point and their determination, octane number of liquid fuels. Gaseous fuels and their classes, advantages. Combustion of a carbonaceous fuel – flue gas. Analysis of flue gas or automobile exhaust for CO₂, CO, O₂ and N₂ by Orsat's apparatus. Calorific value of fuel - net and gross calorific values. Determination of calorific value of solid and liquid fuels by bomb calorimeter method.

[16 HOURS]

UNIT - II

Biomedical and forensic analysis: Composition of body fluids and detection of abnormal levels of certain constituents leading to diagnosis of disease. Sample collection and preservation of physiological fluids. Analytical methods for the constituents of physiological fluids (blood, serum, urine).

Blood - estimation of glucose, cholesterol, urea, haemoglobin and bilirubin.

Urine - urea, uric acid, creatinine, calcium phosphate, sodium, potassium and chloride.

Biological significance, analysis and assay of enzymes (pepsin, monoaminoxidase, tyrosinase); and hormones (progesterone, oxytocin, insulin). Chemical, instrumental and biological assays to be discussed wherever necessary.

Forensic analysis: General discussion of poisons with special reference to mode of action of cyanide, organophosphates and snake venom. Estimation of poisonous materials such as lead, mercury and arsenic in biological materials.

[16 HOURS]

Course Title: APPLIED ANALYSIS III

Course Code: CHD 510

Course Outcomes:

After studying this course, the student to:

CO1: Learn meaning, laws and techniques of chemical kinetics.

CO2: Know importance of chemical kinetics in enzyme catalysed and non-enzyme

catalysed reactions.

CO3: Acquire knowledge on automated and automatic methods of analysis with choice on instrumental methods

CO4: Distinguish between conventional and radio-chemical methods

CO5: Know about Type of samples subjected to radio-chemical analysis and radioimmunoassay.

UNIT – I

Kinetic methods of analysis: Introduction, basis of kinetic methods, rate law expressions. Classifying chemical kinetic methods – direct-computation integral methods, direct-computation rate methods, curve-fitting methods. Instrumentation. Quantitative applications - enzyme catalyzed reactions, non-enzyme catalyzed reactions, non-catalytic reactions. Determining V_{max} , K_m for enzyme catalyzed reactions. Elucidating mechanism for the inhibition of enzyme catalysis. Determination of enzymes, LDH, GOT and GPT. Determination of substrates – urea, uric acid, blood glucose and blood alcohol. Analysis of closely related compounds - neglect of reaction of slow reacting component method and logarithmic extrapolation method.

Automated methods of analysis: An overview. Principles of automation. Automated instruments: process control. Continuous analyzers. Discrete autoanalyzers. Instruments used in automated process control. Automatic instruments - discrete and continuous flow sampling instruments. Flow injection analysis – principles - dispersion co-efficient. Factors affecting peak height, sample volume, channel length and flow rate, and channel geometry. Applications - limited dispersion applications, medium dispersion applications, stopped flow methods and flow injection titrations. Discrete automatic systems - centrifugal fast scan analyzer, automatic organic elemental analyzers.

Analysis based on multilayer films-general principles, film structures, instrumentation, performance and applications – blood urea nitrogen, blood glucose and potassium.

[16 HOURS]

UNIT – II

Radiometric methods: Radioactive isotopes. Nuclear emissions - α and β -particles, neutrons, gamma rays and miscellaneous nuclear particles. Nuclear reactions, radiochemical decay and activity. Instrumentation and measurement of radioactivity. Radiation detectors - gas ionization, scintillation and semiconductor detectors. Pulse height analysis. Autoradiography. Statistics of radioactive measurements.

Radiochemical analysis: Neutron activation methods - neutrons and their sources. Interaction of neutrons with matter. Theory, experimental considerations and applications. **Isotope dilution methods** - direct isotope dilution and inverse isotope dilution methods and their applications. Radiometric titrations. Radiorelease methods. Radioactive tracers.

Radio immunoassay: Principles of immunoassay. Specificity of immuno assays. Preparation of the antibody, incubation period for the assay, separation of the bound and free antigen. Fluorescence immunoassay. Enzyme immunoassay.

[16 HOURS]

References:

1. Fundamental of Analytical Chemistry, D.A. Skoog, D.M. West, Holler and Crouch 8th edition, 2005, Saunders College Publishing, New York.
2. Analytical Chemistry, G.D. Christian, 5th edition, 2001 John Wiley & Sons, Inc. India.
3. Quantitative Analysis, R.A. Day and A.L. Underwood, 6th edition, 1993 Prentice Hall, Inc. New Delhi.
4. Vogel's Textbook of Quantitative Chemical Analysis, J. Mendham, R.C. Denney, J.D. Barnes and M.J.K. Thomas, 6th edition, Third Indian Reprint, 2003 Pearson Education Pvt. Ltd., New Delhi.
5. Analytical Chemistry Principles, John H. Kennedy, 2nd edition, Saunders College Publishing, California, 1990.
6. Instrumental Methods of Analysis by H.H. Willard, L.L. Merritt and J.A. Dean, 7th edition, (1988).
7. Principles and Practice of Analytical Chemistry, F.W. Fifield and Kealey, 3rd edition, 2000, Blackwell Sci., Ltd. Malden, USA.
8. Modern Analytical Chemistry, David Harvey, McGraw Hill, New Delhi, 2000.
9. Introduction to Instrumental Analysis, Braun, Pharm. Med. Press. India.
10. Instrumental Method of Analysis, W. M. Dean and Settle, 7th edition, 1986, CBS Publishers, New Delhi.
11. Instant Notes of Analytical Chemistry, Kealey and Haines, Viva books Pvt. Ltd., 2002.
12. Soil Chemical Analysis, M.L. Jackson, Prentice Hall of India Pvt. Ltd., New Delhi, 1973.
13. Clinical Chemistry, Principles and Procedures, J.S. Annino, 2nd edition, Boston: Little, Brown, 1960.
14. Methods of Geochemical Analysis, D. Click, Ed., A Multi volume series, NewYork, Inter science.
15. Clinical Chemistry, Principles and Techniques, R.J. Henry, D.C. Cannon and J.W. Winkleman, Eds., 2nd edition, Hagerstorm, M.D: Harper and Row, 1974.

16. Fundamentals of Clinical Chemistry, N.W. Tietz, Ed., 2nd edition, Philadelphia: W.B. Saunders, 1976.

Course Title: RETROSYNTHESIS AND ORGANOMETALLIC CHEMISTRY

Course Code: CHD 520

Course Outcomes:

After studying this course, the student to:

CO1: Acquire knowledge of protection and deprotection in organic synthesis.

CO2: Learn about different named reactions which are highly useful for competitive exams and interviews.

CO3: Learn disconnection approach, their principles and terminologies.

CO4: Learn retrosynthesis of different complex organic molecules.

UNIT – I

Protecting groups: Protection of hydroxyl, carboxyl, carbonyl, thiol and amino groups. Illustration of protection and deprotection in synthesis.

Named reactions: Keto-enol tautomerism, mechanism and synthetic applications of aldol condensations, Claisen reaction, Schmidt reaction, Perkin reaction, Knoevenagel, benzoin and Stobbe condensation, Darzens glycidic ester condensation, Cannizaros reaction, Tischenko reaction. Michael addition, Robinson's annulation reaction.

Retrosynthesis: Introduction to disconnection approach: Basic principles and terminologies used in disconnection approach. One group C-X and two group C-X disconnections. Synthons and synthetic equivalents.

Retrosynthesis and synthesis of benzofurans, *p*-methoxy acetophenone, saccharine, α -bisabolene, nuciferal, penicillin-V.

[16 HOURS]

UNIT - II

Chemistry of organometallic compounds: Synthesis and reactions of organolithium (n-BuLi, PhLi) and organomagnesium (Grignard reagent) compounds.

Organoaluminium reagents: Preparation, site selective and stereoselective additions of nucleophiles mediated by organoaluminum reagents, reaction with acid chlorides, allyl vinyl ethers, 1,2-addition to imines and application in the synthesis of natural products.

Organopalladium compounds: Suzuki coupling, Heck reaction.

Organotin reagents: Barton decarboxylation reaction, Barton deoxygenation reaction, Stille coupling, Stille-Kelley coupling reactions, Barton McCombie reaction, Keck stereoselective allylation and other applications.

Organozinc reagents: Preparation - oxidative addition and transmetallation, addition reactions of alkyl, aryl, allylic and propargylic zinc reagents, diastereoselective and enantioselective addition reaction with aldehydes, Reformatsky reaction.

[16 HOURS]

References:

1. H. Pine, Hendrickson, Cram and Hammond, Organic Chemistry, Mac Grow Hill, New York, 1987.
2. Organic Chemistry - Morrison and Boyd
3. Organic Chemistry- Crabtree
4. Organic Chemistry- Clayden
5. I.L. Finar, Organic Chemistry, ELBS Longmann, Vol. 1 & II, 1984.
6. J. March, Advanced Organic Chemistry, Wiley Interscience, 1994.
7. E.S. Gould, Mechanism and Structure in Organic Chemistry, Halt, Rinhart & Winston, New York, 1964.
8. F.A. Carey and Sundberg. Advanced Organic Chemistry – Part A & B, 3rd edition, Plenum Press, New York. 1990.
9. Principles of Organic Synthesis - ROC Norman and Coxon.
10. S.K. Ghosh, Advanced General Organic Chemistry, Book and Allied (P) Ltd. 1998.

Course Title: BIOMOLECULES AND NATURAL PRODUCTS

Course Code: CHD 530

Course Outcomes:

After studying this course, the student to:

CO1: Know about amino acids, peptides, proteins: their structure, function and properties.

CO2: Learn structural determination of the proteins which are called as energy of the body.

CO3: About the chemistry lying behind the heredity.

CO4: Learn nomenclature, classification and biological importance of other natural Products.

UNIT - I

Amino Acids: General structure, Physiological properties

Peptides: Structure and conformation of peptide bond, peptide synthesis: Solution phase and Merrifield's solid phase synthesis, Racemization and use of HOBT, Synthesis of oxytocin and

vasopressin, biological importance of insulin, selective cleavage of polypeptide bonds (chemical and enzymatic).

Proteins: Structure determination: *C* and *N* terminal residue determination, primary, secondary, tertiary and quaternary structure determination, denaturing and renaturing of proteins.

Nucleic acids: Introduction, structure and synthesis of nucleosides and nucleotides, Solid phase synthesis of oligonucleotides, Structure of RNA and DNA, Crick-Watson model, role of nucleic acids in the biosynthesis of proteins.

[16 HOURS]

UNIT – II

Carbohydrates: Synthesis, industrial and biological importance of glycosides, amino sugars, sucrose, maltose and lactose. General methods of structure elucidation. Industrial importance and biological importance of cellulose, starch, glycogen, dextran, hemicellulose, pectin, agar-agar. Photosynthesis and biosynthesis of carbohydrates.

Carbohydrates Metabolism: Glycolysis and Krebs cycle.

Lipids: Nomenclature, classification, purification, synthesis of lipids, phospholipids, sphingolipids, biological importance of lipids: Lecithin, sphingolipids, oils and fats.

Terpenoids: Introduction, classification (natural and essential oils), isoprene rule and biological importance of terpenoids.

Steroids: Introduction, classification and biological significance of Testosterone, Progesterone, Estrogen

[16 HOURS]

References:

1. I. L. Finar, Organic Chemistry, ELBS Longmann, Vol. I & II, 1984.
2. Essentials of physiological chemistry – Anderson, John Wiley & Sons, New York, 1953.
3. K. Albert, L. Lehninger, D.L. Nelson, M.M. Cox, Principles of Biochemistry, CBZ publishers, 1st edition, New Delhi, 1993.
4. Harper's Biochemistry, Ed. R. Harper, 22nd edition, Prentice Hall Press, New York, 1990
5. Carbohydrates – Chemistry and Biochemistry by Pigmann and Harton.
6. An introduction to carbohydrate chemistry by Guthrie and Honeyman.
7. Protein chemistry by Neurath, Vol. I, II and III.
8. Peptide chemistry by Bodanski, Vol. I, II and III.
9. Introduction to the chemistry of fats and fatty acids by F. D. Gunstone.

PRACTICALS – SOFT CORE

Course Title: ANALYTICAL CHEMISTRY PRACTICALS

Course Code: CHD 210

Course Outcomes:

After studying this course, the student to:

COID	CO
CO1	Get experience on analysis of various complex mixtures by following multistep reactions.
CO2	Acquire the knowledge on handling instruments and to overcome the general problems arises during the analysis.
CO3	Acquire industrial skills required for sampling, analytical and interpretation and presentation of results.
CO4	Possess adequate knowledge on literature search for developed analytical methods.

[128 HOURS]

PART – III

1. Determination of calcium in limestone by redox, acid-base and complexation titrations.
2. Determination of vitamin C in orange juice by titration with cerium(IV) and with 2,6-dichlorophenol indophenol.
3. Determination of mercury in an algacide by EDTA titration; and arsenic in ant control preparation by redox titration.
4. Determination of aluminium and magnesium in antacids by EDTA titration.
5. Analysis of a copper-nickel alloy sample for copper and nickel by EDTA titration using masking and selective demasking reactions.
6. Determination of saccharin in tablets by precipitation titration.
7. Determination of iodine value and saponification value of edible oils.
8. Determination of ascorbic acid in goose berry/bitter gourd by titrimetry and spectrophotometry using *N*-bromosuccinimide (NBS).
9. Analysis of a mixture of iron(II) and iron(III) by EDTA titration using *pH* control.
10. Determination of sulpha drugs by potentiometry using NaNO_2 and iodometric assay of penicillin.
11. Solvent extraction method for determination of silver as ion-associate with 1,10-phenanthroline and bromopyragallol red.
12. Electrolytic determination of copper and lead in brass.
13. Polarographic determination of copper and zinc in brass.

14. Determination of sodium, potassium and calcium in mineral waters by atomic emission spectrometry.
15. Determination of iron in mustard seeds and phosphorus in peas by spectrophotometry.
16. Analysis of waste water for anionic detergents and phenol by spectrophotometry.
17. Fluorimetric determination of riboflavin (vit.B₂) in tablets.
18. Colorimetric analysis of procaine by diazotization and coupling reaction.
19. Determination of manganese in steel by extraction-free spectrophotometry and molybdenum in steel by extractive spectrophotometry.
20. Determination of ethanol in wine by titrimetric and spectrophotometric dichromate methods

PART – IV

1. Analysis of waste waters for DO and COD by titrimetry.
2. Analysis of a ground water sample for sulphate by titrimetry (EDTA) and turbidimetry.
3. Potentiometric determination of formula and stability constant of a silver-ammonia complex ion.
4. Determination of aspirin, phenacetin and caffeine in mixture and APC tablets by solvent extraction and UV spectrophotometry.
5. Kinetic determination of urinary creatinine and purity of a commercial H₂O₂ sample.
6. Determination of chromium(III) and iron(III) in a mixture by kinetic masking methods.
7. Catalytic determination of traces of selenium in biological materials and iodide in blood serum.
8. Photometric and potentiometric titration of iron(III) with EDTA.
9. Photometric and potentiometric titration of copper with EDTA.
10. Determination of copper(II) and iron(III) in mixture by photometric titration with EDTA.
11. Analysis of brackish water for chloride content by a) spectrophotometry (mercuric thiocyanate method), b) conductometry (silver nitrate) and c) potentiometry (silver nitrate).
12. Conductometric titration of sodium acetate with HCl and NH₄Cl with NaOH.
13. Ascorbic acid determination in natural orange juice by coulometry.
14. Spectrophotometric determination of iron in natural waters using thiocyanate and 1,10-phenanthroline as reagents.
15. Determination of fluoride in drinking water/ground water by spectrophotometry(alizarin red lake method).
16. Analysis of waste water for

- a) phosphate by molybdenum blue method
 - b) ammonia-nitrogen by Nessler's method
 - c) nitrite-nitrogen by NEDA method
15. Analysis of a soil sample for
- a) calcium carbonate and organic carbon by titrimetry.
 - b) calcium and magnesium by EDTA titration.
16. Analysis of a soil sample for
- a) Nitrogen content by Kjeldahl method
 - b) Available phosphorus by spectrophotometry.
 - c) Nitrate-nitrogen/nitrite nitrogen/ammonia nitrogen by spectrophotometry.
 - d) sodium and potassium by flame photometry.
17. Analysis of urine for
- a) urea and uric acid by titrimetry and spectrophotometry.
 - b) Sulphate by precipitation titration after ion-exchange separation.
 - c) Sugar by Benedict's reagent.
18. Analysis of blood for
- a) cholesterol by spectrophotometry
 - b) bicarbonate by acid-base titration.
19. Fluorimetric determination of quinine in an antimalarial tablet.

References

1. Fundamental of Analytical Chemistry, D.A. Skoog, D.M. West, Holler and Crouch 8th edition, 2005, Saunders College Publishing, New York.
2. Analytical Chemistry, G.D. Christian, 5th edition, 2001 John Wiley & Sons, Inc. India.
3. Quantitative Analysis, R.A. Day and A.L. Underwood, 6th edition, 1993, prentice Hall, Inc. New Delhi.
4. Vogel's Textbook of Quantitative Chemical Analysis, J. Mendham, R.C. Denney, J.D. Barnes and M.J.K. Thomas, 6th edition, Third Indian Reprint, 2003 Pearson Education Pvt. Ltd., New Delhi.
5. Analytical Chemistry Principles, John H. Kennedy, 2nd edition, Saunders College Publishing, California, 1990.
6. Quantitative Analysis of Drugs in Pharmaceutical Formulations, P. D. Sethi, 3rd edition, CBS Publishers & Distributors, New Delhi, 1997.
7. Practical Clinical biochemistry methods and interpretations, R. Chawla, J.P. Bothers Medical Publishers (P) Ltd., 1995.

8. Laboratory Manual in Biochemistry, J. Jayaraman, New Age International Publishers, New Delhi, 1981.
9. Experiments on Water Pollution, D.I. Williams and D. Anglesia, Wayland Publishers Ltd., England, 1978.
10. Experiments on Land Pollution, D.I. Williams and D. Anglesia, Wayland Publishers Ltd., England, 1978.
15. Experiments in Environmental Chemistry, P.D. Vowler and D.W. Counel, Pergamon Press, Oxford 1980.
16. Manual Soil Laboratory Testing, vol.I, K.H. Head, Pentech Press, London 1980.

Course Title: INORGANIC CHEMISTRY PRACTICALS

Course Code: CHD 220

Course Outcomes:

After studying this course, the student to:

COID	CO
CO1	Get experience on analysis of various complex mixtures by multistep reactions.
CO2	Acquire knowledge on handling instruments.
CO3	Acquire skills required for sampling, analytical and interpretation and presentation of results.
CO4	Possess adequate knowledge on literature search for developed preparative methods.
CO5	Synthesize and characterization of complexes.

[128 HOURS]

PART – III

1. Determination of aluminium and bismuth by complexometric titration.
2. Determination of lead and tin in a mixture.
3. Determination of calcium and lead in a mixture by pH control and complexation method.
4. Determination of zinc, manganese and magnesium in a mixture using fluoride as a demasking agent.
5. Quantitative analysis of copper(II) and iron(II) in a mixture:
 - a) Copper gravimetrically as CuSCN and
 - b) Iron volumetrically using cerium(IV) solution

6. Determination of iron as the 8-hydroxyquinolate by solvent extraction method.
7. Determination of the composition of iron-phenanthroline complex by:
 - a) Job's method
 - b) Mole-ratio method and
 - c) Slope-ratio method.
8. Polarographic estimation of cadmium and zinc.
9. Spectrophotometric determinations of:
 - a) Titanium using hydrogen peroxide
 - b) Chromium using diphenyl carbazide in industrial effluents
 - c) Nickel using dimethylglyoxime in steel solution
10. Solvent extraction of ferric thiocyanate complex and determination by colorimetry.

PART – IV

1. Preparation of hexaamminecobalt(III) chloride and estimate cobalt ion.
2. Preparation and characterization of Chloropentaamminecobalt(III) chloride and estimate cobalt ion.
3. Using chloropentaamminecobalt(III) chloride, prepare nitro and nitritopentamine cobalt(III) chloride. Record the IR spectra of the isomers and interpret.
4. Preparation of potassium tris-oxalatochromate(III) trihydrate.
5. Preparation of mercurytetrathiocyanatocobaltate(II) and estimation of mercury by gravimetry.
6. Preparation of tetraamminecopper(II) sulphate tetrahydrate
7. Preparation and characterization of manganese dioxide nano-particles
8. Preparation of bis-dichlorotriphenyl phosphine nickel (II).
9. Preparation and characterization of hexaamminenickel(II) chloride.
10. Demonstration Experiments:
 - a) Recording and interpretation of IR and NMR spectra of complexes.
 - b) Spectrochemical series - Evaluation of Dq value.
 - c) DNA interaction with metal complexes by UV-visible absorption and viscosity methods.

References

1. Basic principles of Practical Chemistry – V. Venkateswaran, R. Veeraswamy and A.R. Kulandraivelu

- Instrumental Analysis Manual - Modern Experiments for Laboratory – G.G. Guilbault and L.G. Hargis.
- A Text Book of Quantitative Inorganic Analysis – A.I. Vogel, 5th edition.
- Experimental Inorganic Chemistry – G. Palmer.
- Inorganic Synthesis – O. Glemser.
- Experimental Inorganic/ Physical Chemistry- Mounir A. Malati.
- Quantitative Chemical Analysis – Daniel C. Harris, (2006) 7th edition.
- Spectrophotometric Determination of Elements – Z. Marczenko

Course Title: ORGANIC CHEMISTRY PRACTICALS

Course Code: CHD 230

Course Outcomes:

After studying this course, the student to:

COID	CO
CO1	Develop experience in multistep synthesis and also mechanisms.
CO2	Learn different kinds of reactions under multistep synthesis.
CO3	Learn isolation experiments, preliminary identification and separation.
CO4	Acquire knowledge of various estimations like sugars, enol content, ketones, nitro, protein etc.

[128 HOURS]

PART – III

Multi step synthesis

- Oxidation of cyclohexanol to adipic acid via cyclohexanone
- Preparation of benzocaine from *p*-nitrotoluene
- Preparation of *p*-chlorobenzoic acid from *p*-toluidine (Sandmeyer's reaction)
- Molecular rearrangement:
 - Preparation of *o*-chlorobenzoic acid from phthalic anhydride
 - Preparation benzilic acid from benzaldehyde
 - Preparation of *o*-hydroxy benzophenone from phenyl benzoate via Fries rearrangement
 - Preparation of benzanilide from benzophenone (Beckmann rearrangement).
- Grignard reaction: Preparation of triphenyl carbinol
- Preparation of luminol from phthalic anhydride
- Synthesis of isoxazolines and pyrazolines via 1,3-dipolar cycloaddition.

8. Synthesis of tetralones from aryl aldehydes.
9. Synthesis of *m*-chloriodobenzene from *m*-dinitrobenzene
10. Synthesis of Schiff base from nitro compound.

PART – IV

Isolation of natural products

1. Fractional crystallization: separation of mixture of naphthalene and biphenyl
2. Fractional distillation: Separation of mixture of hexane and toluene.
3. Thin layer chromatography: Separation of plant pigments
4. Column chromatography: Separation of mixture of *o* and *p*-nitro anilines
5. Isolation of piperine from pepper
6. Isolation of caffeine from tea
7. Isolation of azeleic acid from castor oil
8. Isolation of clove oil from clove
9. Estimation of sugars by Fehlings method
10. Determination of enol content by Meyer's method
11. Estimation of ketones by haloform reaction
12. Estimation of sugars by Bertrand's method
13. Estimation of nitro groups
14. Estimation of protein by biuret method

Spectral analysis: Structural elucidation of some simple organic compounds by UV, IR, NMR and mass. The spectra have to be provided by the teachers.

References

1. Manual of Organic Chemistry - Dey and Seetharaman.
2. Modern Experimental Organic Chemistry by John H. Miller and E.F. Neugil, p 289.
3. An Introduction to Practical Organic Chemistry - Robert, Wingrove etc.
4. A Text Book of Practical Organic Chemistry – A.I. Vogel, Vol.III
5. Practical Organic Chemistry - Mann & Saunders
6. Semimicro Qualitative Organic Analysis by Cheronis, Entrikin and Hodnet .
7. R.K. Bansal, Laboratory Manual of Organic Chemistry, New Age International (P) Ltd. London, 3rd edition, 1996.

Course Title: PHYSICAL CHEMISTRY PRACTICALS**Course Code: CHC 240****Course Outcomes:**

After studying this course, the student to:

COID	CO
CO1	Acquire knowledge on handling instruments and to overcome the general problems arises during the analysis.
CO2	Learn concepts of rate constants, energy of activation, order of the reaction and also thermodynamics parameters.
CO3	Learn concepts of kinetics experiments.

[128 HOURS]**PART – III (NON-INSTRUMENTAL)**

1. Determination of energy of activation for reaction between sodium formate and iodine.
2. To study the kinetics of reaction between acetone and iodine-determination of order of reaction w.r.t. iodine and acetone.
3. Determination of rate of decomposition of hydrogen peroxide with manganese dioxide.
4. Determination of order and rate constant of hydrolysis of ethyl acetate in acid medium.
5. Kinetics of decomposition of benzene diazonium chloride, determination of energy of activation and thermodynamic parameters.
6. Kinetics of decomposition of diacetone alcohol by NaOH-determination of energy of activation.
7. To determine the eutectic point of a two component system (Naphthalene-*m*-dinitrobenzene system).
8. Study of phase diagram of a three component system (e.g. acetic acid-chloroform water and system). Construction of binodal curve and indicating tie line.
9. Determination of heat of solution and lattice energy of calcium chloride.
10. Determination of partition co-efficient of acetic acid in water and butanol.
11. Study of kinetics of reaction between $K_2S_2O_8$ and KI, first order, determination of rate constants at two different temperatures and E_a .
12. To determine the rate constant for the reaction glycine and CAT.

PART IV (INSTRUMENTAL)

1. Kinetics of saponification of ethyl acetate by conductivity method and study the effect of dielectric constant of the medium (using CH_3OH).

2. Simultaneous spectrophotometric determination of manganese and chromium in KMnO_4 and $\text{K}_2\text{Cr}_2\text{O}_7$ mixture.
3. Determination of ionic product of water and study the effect of temperature.
4. Coulometric titration I_2 vs $\text{Na}_2\text{S}_2\text{O}_3$.
5. Conductometric study of charge transfer complex of p-phenylenediamine with phthalic acid.
6. Determination of mean ionic activity coefficient of a weak electrolyte (acetic acid) by conductometric measurements.
7. Conductometric determination of the degree of hydrolysis and hydrolysis constant of aniline hydrochloride.
8. Conductometric titration of potassium iodide with mercuric perchlorate.
9. Determination of pK value of an indicator (methyl orange).
10. Potentiometric titration of mixture of $\text{KCl} + \text{KBr} + \text{KI}$ vs AgNO_3 .
11. Conductometric titration of a mixture of HCl , CH_3COOH and CuSO_4 against NaOH .
12. Thermometric titration of HCl and H_3BO_3 with NaOH .
13. Determination of quantum yield for the photolysis of Chloramine-T.
14. Determination of quantum yield for the photolysis of Chloramine-B.

References

1. Practical Physical Chemistry – A.J. Findlay.
2. Experimental Physical Chemistry – F. Daniels *et al.*
3. Selected Experiments in Physical Chemistry – Latham.
4. Experiments in Physical Chemistry – James and Prichard.
5. Experiments in Physical Chemistry – Shoemaker.
6. Advanced Physico-Chemical Experiments – J. Rose.
7. Practical Physical Chemistry – S.R. Palit.
8. Experiments in Physical Chemistry – Yadav, Geol Publishing House.
9. Experiments in Physical Chemistry – Palmer.
10. Experiments in Chemistry – D.V. Jahagirdar, Himalaya Publishing House, Bombay, (1994).
11. Experimental Physical Chemistry – Das. R.C. and Behera B, Tata Mc Graw Hill.

PROJECT WORK/ DISSERTATION–SOFT CORE

Course Title: PROJECT WORK/ DISSERTATION–SOFT CORE

Course Code: CHD 250

Course Outcomes

After studying this course, the student to:

COID	CO
CO1	Carry out literature survey on the problem/s to be solved.
CO2	Learn and follow suitable research methodologies to propose and to perform Experiments.
CO3	Attain the state of ability to take up research work.
CO4	Better understanding about research articles, patents, book chapters or books on relevant research problem
CO5	Acquire skills of writing research reports in the form of articles or thesis.

Each student can take up Project Work/ Dissertation under the guidance of the faculty of the department during the IV Semester as a Soft Core course.



JSS COLLEGE OF ARTS, COMMERCE & SCIENCE
(AUTONOMOUS)
OOTY ROAD, MYSORE-570 025

Postgraduate Department of Commerce

Syllabus

2016– 17

ONWARDS

COURSE STRUCTURE AND SYLLABUS

MINIMUM CREDITS REQUIRED FOR M.COM. DEGREE

I to IV Semester s	HARD CORE COURSE		SOFT CORE COURSE		OPEN ELECTIVE COURSE		TOTAL	
	Number	Credits	Number	Credits	Number	Credits	Number	Credits
Total	12	48	6	24	1	4	19	76

Semester wise requirements of course and credits

Semester	HARD CORE COURSE		SOFT CORE COURSE		OPEN ELECTIVE COURSE		TOTAL	
	Number	Credits	Number	Credits	Number	Credits	Number	Credits
I	4	16	1	4	-	-	5	20
II	3	12	2	8	-	-	5	20
III	3	12	1	4	1	4	5	20
IV	2	8	2	8	-	-	4	16
Total	12	48	6	24	1	4	19	76

First Semester

Sl. No.	Course Code	Title of the course	Hard core/ Soft core/	Number of Credits			
				L	T	P	TOTAL
HC01	MCA010	Accounting Theory	HC	3	1	0	4
HC02	MCA080	Corporate Governance and Business ethics	HC	3	1	0	4
HC03	MCA090	Advanced Financial Management	HC	3	1	0	4
HC04	MCA100	Strategic Management	HC	3	1	0	4
SC01	MCA210	Business Policy and Environment	SC	3	1	0	4
SC02	MCA220	Statistics for Business Decisions	SC	3	1	0	4

Note: students can register for any one of the above soft cores.

Second Semester

Sl. No.	Course Code	Title of the course	Hard core/ Soft core	Number of Credits			
				L	T	P	TOTAL
HC05	MCB030	Organizational Behavior	HC	3	1	0	4
HC06	MCB050	Entrepreneurial Development	HC	3	1	0	4
HC07	MCB010	Capital Market Instruments	HC	3	1	0	4
SC05	MCB240	Human Resource Management	SC	3	1	0	4
SC06	MCB270	Management of Financial Services	SC	3	1	0	4
SC07	MCB250	Banking Technology	SC	3	1	0	4

Note: Students can register for any two of the above soft cores.

Third Semester

Sl. No.	Course Code	Title of the course	Hard core/ Soft core/ Open elective	Number of Credits			
				L	T	P	TOTAL
HC08	MCC010	International Business	HC	3	1	0	4
HC09	MCC030	Business Research Methods	HC	3	1	0	4
HC10	MCC040	Security Analysis and Portfolio Management	HC	3	1	0	4
SC15	MCC230	Elective- Paper- 1	SC	3	1	0	4
SC17	MCC250	Elective – Paper -1	SC	3	1	0	4
OE	540/580	Retail Banking/ Personal Financial Management/Financial Accounting/Management of non-profit organization	OE	3	1	0	4

Note

1. Students can register for any one of the elective groups as soft core (first paper in the group)
2. Non-commerce students can register any one of the following open-electives.

Fourth Semester

Sl. No.	Course Code	Title of the course	Hard core/ Soft core/ Open elective	Number of Credits			
				L	T	P	TOTAL
HC11	MCD010	International Accounting	HC	3	1	0	4
HC12	MCD020	Current Trends in Business and Commerce	HC	3	1	0	4
SC09	MCD260	Advanced Cost Accounting	SC	3	1	0	4
SC10	MCD210	Supply chain Management	SC	3	1	0	4
SC12	MCD220	Major Project Work	SC	0	1	3	4
SC16	MCD230	Elective – Paper 2	SC	3	1	0	4
SC18	MCD250	Elective- Paper 2	SC	3	1	0	4

Note: 1 Major project work is a soft core course. The students can register for the Project work or for Advanced Cost Accounting or Supply Chain Management carrying four credits each.

Note: 2 Guiding 4 students for Major Project work is considered to be equal to 01 hour of class room teaching

Note: 3 Students must register for 2nd paper in the elective group selected by them in the III semester as soft core.

Elective Group

Any one group from the available electives shall be selected by a student at the commencement of the III Semester. Once a group has been selected, no change in the selected group will be allowed later. While the first paper of the selected group will be taught in Semester III, second paper of the selected group will be taught in Semester IV as Soft Core Courses. The Department will announce in the beginning of the third semester, any one or more elective groups which will be offered during III and IV semesters depending upon the availability of faculty members and the demand for electives.

Group A: Financial Accounting

1. Contemporary Areas of Financial Accounting
2. International Financial Reporting Standards (IFRS)

Group B: Financial Management

Futures, Options & Swaps
Strategic Financial Decisions

Group C: Business Taxation

1. Indirect Tax Laws and Practice
2. Corporate Tax Laws and Planning

Group D: Cost Management

1. Marginal Costing & Decision Making
2. Tools and Techniques of control.

PROGRAMME OUTCOMES:

After studying this programme the students are able to

- PO1 Understand role of accounting and finance in the present business scenario.
- PO2 Identify the latest trends in banking and finance
- PO3 Use wide varieties of tools and techniques to meet the emerging opportunities and challenges
- PO4 Become an entrepreneur based on the knowledge gained.
- PO5 Strengthen the knowledge base to take up CA/ICWA/ICS and other competitive examination
- PO6 Acquire the ability to engage in independent & lifelong learning in the broader context of social and technical changes.
- PO7 Accept the challenges of business world.
- PO8 Enhance logical thinking and decision making ability.

PROGRAMME SPECIFIC OUTCOMES:

After studying this programme the students are able to

- PSO1 Inculcate the knowledge of business and the techniques of managing the Business with special focus on Accounting, finance, and financial services.
- PSO2 Identify knowledge based accounting principles and the latest application oriented corporate accounting methods.
- PSO3 Develop decision-making skill through costing methods and practical application of management accounting principles.
- PSO4 Enhance taxation skills through a thorough understanding of tax laws.

First Semester

HARD CORE SUBJECTS

HC01: MCA010 ACCOUNTING THEORY

1. Course Description: The course provides the coverage of the meaning of accounting theory, its types, approaches to formulate accounting theory; the IASB's conceptual framework; recognition, measurement and disclosure of elements of financial statements.

2. Course Outcome:

- CO1 Deliberate the characteristics of structure of accounting theory
- CO2 Deliberate the classification and characteristics of relationship between research and accounting theory
- CO3 Specify the classification and characteristics of types of accounting theories
- CO4 Identify in detail with examples conceptual frame work for financial reporting
- CO5 Identify the characteristics of recognition, measurement and disclosure of elements of financial statements

3. Pedagogy: Course activities consist of lectures, case study analysis, group discussions, seminar presentation, assignment writing and tests. Reading and analysis of annual reports of companies will be the integral part of instruction.

4. Course Contents:

Module 1: The Meaning of Accounting Theory: The meaning of accounting theory; relationship with accounting research and practice. Types of accounting theories- syntactical, semantical and behavioral theories. Approaches to formulation of Accounting Theory. Ownership theories - proprietary, entity and fund theories.

Module 2: The Conceptual Framework of Accounting: The IASB Framework for the Preparation and Presentation of Financial Statements-preface, purpose and scope; Users and their information needs; Objectives of financial statements; Underlying concepts and assumptions; Qualitative characteristics of financial statements; Constraints on relevant and reliable information; True and fair view; the elements of financial statements; Structure and format of financial statements. The accounting cycle.

Module 3: Recognition, Measurement and Disclosure of Elements of Income Statement: The concept of revenue and expense; their recognition principles and processes, measurement methods and disclosure issues. Recognition, measurement and disclosure of gains and losses from extraordinary items, unusual items, discontinued operations, effects of changes in accounting principles, and prior period items. Problems. Case studies on annual reports of companies.

Module 4: Recognition, Measurement and Disclosure of Elements of Balance Sheet: Definition of asset, liability and owners' equity; their types and characteristics. Recognition, measurement and disclosure of long-term and short-terms assets and liabilities, owners' equity and retained earnings. Problems. Case studies of annual reports of companies.

References:

Anthony R.N., D.F. Hawkins and K.A. Merchant, *Accounting: Text and Cases*, McGraw Hill, 1999

Richard G. Schroeder, Myrtle W. Clark and Jack M. Cathey, *Financial Accounting Theory and Analysis: Text Readings and Cases*, John Wiley and Sons, 2005.

Ahmed Riahi Belkaoui, *Accounting Theory*, Quorm Books, 2000.

Jawahar Lal, *Accounting Theory and Practice*, Himalaya Publishing House, 2008.

L.S. Porwal, *Accounting Theory*, TMH, 2000.

Thomas R. Dyckman, Charles J. Davis, Roland E. Dukes, *Intermediate Accounting*, Irwin McGraw-Hill.

Eldon S. Hendriksen, *Accounting Theory*; www.iasb.org. www.icai.org. www.dca.gov.in

HC02:MCA080 CORPORATE GOVERNANCE AND BUSINESS ETHICS

Course Outcome:

- CO1 Understand in depth concept of corporate governance
- CO2 Deliberate in depth corporate ethics and cultural influences
- CO3 Understand the detail of Corporate social responsibility and accountability
- CO4 Learn in depth E-governance
- CO5 Write down the detail of Committees in corporate governance

Pedagogy:

The subject matter will be presented through lecture, class discussion, student presentation, guest lectures and laboratory experiences.

Module 1: Business ethics: Ethics; Doctrine of Trusteeship: Unethical practices; Good ethics and Good Business. Corporate Ethics: Concept and Importance – Benefits of corporate Ethics- Corporate Philosophy and Culture – Managing Ethics and Legal Compliance – Case Analysis.

Module 2: Social Responsibility of Business; Doctrine of Social Responsibilities of Business, Dimensions of Social Responsibility, and Assumptions of S.R. Corporate Social Responsibility: Corporate Crimes – Company and Society Relations – corporate Social Challenges – Corporate Accountability – Business and Ecology –Case Analysis.

Module 3: Concept of Corporate Governance: It's Importance –Agency Theory – Benefits of Good Corporate Governance – Present Scenario – Case Studies.

Module 4: Corporate Governance In India: Reforming BOD –Birla Committee – Naresh Chandra Committee – Narayana Murthy Committee – Audit Committee – Corporate Governance Code – The Future

References:

Business ethics by L.P. Hartman, (Tata Mc Graw Hill)

Business ethics by W.H. Shaw, (Thomson)

Corporate management and Accountability by L.C. Gupta (Mc Millan Institute for FM and Research, Chennai-1974)

Strategic Management by Hill, Ireland and Horkisson (Thomson)

Business and society by Keith Davis (Mc Graw Hill)

Corporate Governance – Fernando.

Business ethics and corporate Governance – Bansal Sandeep, Bansal Sanjiv, Bansal Rama – Kalyani Publishers.

HC03: MCA090 ADVANCED FINANCIAL MANAGEMENT

1. Course Description:

Financial management is a functional area in general management. This subject is focusing on introduction, scope and importance of financial management, investment decisions, capital structure decisions, dividend decisions and working capital management.

2. Course Outcome:

- CO1 Understand in depth financial management concepts and its important functions
- CO2 Understand in depth financial management concepts and its important functions
- CO3 Identify the detail of Capital structure theories
- CO4 Identify in detail with examples Risk analysis in capital budgeting
- CO5 Learn in depth Dynamics of financial markets

3. Pedagogy:

Students must work out assigned individual topics, present seminars and participate in case studies or group discussions.

Course Contents:

Module 1: Introduction : Meaning, Scope, and functions of Financial Management. Corporate Objectives-Profit Maximization and Wealth Maximization, their social implications.

Module 2: Capital Budgeting Decision – Importance – Challenges – NCF estimation DCF Techniques- NPV vs IRR – Multiple IRR Problem- Modified IRR (MIRR) – Capital Rationing.

Module 3: Risk Analysis in Capital Budgeting- Approaches to risk absorption- Expected Net Present Value (ENPV) - Payback method - Risk-Adjusted Discount rate - Use of Normal Distributions - Sensitivity analysis - Measurement of Project risk- Risk Analysis of Project Portfolios.

Module 4: Capital Structures Decision – Traditional View – MM’s theory- Factors impacting leverage decision. Cost of capital – Cost of equity – Cost of preferred capital- Cost of debt- Cost of retained earnings – WACC- Marginal cost of capital

References:

- Pandey, I.M. financial Management, Vikas Publishing House, New Delhi.
- Khan M.Y. and Jain P.K. Financial Management, Tata McGraw Hill, New Delhi.
- Kishore, R., Financial Management, Taxmans Publishing House, New Delhi.
- Chandra, Prasanna; Financial Management TMH, New Delhi.
- Horn, Van; Financial management and Policy, Prentice Hall of India.
- Brigham & Houston, Fundamentals of Financial Management, Thomson Learning, Bombay.
- Richard Brealey and Stewart Myers, Principles of Corporate Finance, Tata McGraw Hill, 2000.
- V K Bhalla, Financial Management and Policy: Text and Cases, Annual Publishers, 2002.

HC04: MCA100 STRATEGIC MANAGEMENT

1. Course Outcome:

- CO1 Deliberate the detail of strategy formulation
- CO2 Deliberate in detail with examples implementation of strategy
- CO3 Identify the detail of strategic analysis and choice
- CO4 Specify in depth strategy evaluation
- CO5 Understand in depth E-commerce business model

2. Pedagogy: Teaching method comprises of lecture sessions and tutorials. Lecture sessions focus on providing conceptual understanding and analytical setting for select aspects of the course content.

3. Course Contents:

Module 1: Strategic Management - An Introduction Concept of strategic management - Characteristics of strategic management - Defining strategy, Strategy formulation - Stakeholders in business - Vision, mission and purpose - Business definition, objectives and goals - Environmental appraisal - Types of strategies - Guidelines for crafting successful business strategies, Tailoring strategy to fit specific industry

Module 2: Strategic analysis and choice - Environmental Threat and Opportunity Profile(ETOP) – Organizational Capability Profile - Strategic Advantage Profile - Corporate Portfolio Analysis - SWOT Analysis - Synergy and Dysergy - GAP Analysis - Porter's Five Forces Model of competition - Mc Kinsey's 7s- Framework - GE 9 Cell Model - Distinctive competitiveness - Selection of matrix. Case study.

Module 3: Strategy implementation - Issues in implementation - Project implementation – Procedural implementation – Resource Allocation - Budgets - Organization Structure – Matching structure and strategy - Behavioral issues - Leadership style - Corporate culture – Values - Power - Social responsibilities – Ethics. Case study.

Module 4: Strategy Evaluation - Importance - Symptoms of malfunctioning of strategy - Organization anarchies - Operations Control and Strategic Control - Measurement of performance - Analyzing variances - Role of organizational systems in evaluation,. New Business Models and strategies for Internet Economy - Shaping characteristics of E-Commerce environment - E-Commerce Business Model and Strategies - Internet Strategies for Traditional Business - Key success factors in E-Commerce

References:

- A concept of corporate planning-, Russel Ackoff, Newyork wiley
- Business policy and strategic management- Tokyo, McGraw hill
- Strategic Management-Text and Cases- V.S.P. Rao and V. Harikrishna
- Strategic Management-Azar Kazmi
- Strategic Management-Francis Cherunillam
- Strategic Management-Subba Rao
- Strategic Planning Formulation of Corporate Strategy - Ramaswamy
- Strategic Management, 12th Ed. - Concepts and Cases - Arthur A. Thompson Jr. and A.J.Strickland
- Management Policy and Strategic Management (Concepts, Skills and Practices) - R.M.Shrivastava
- Strategic Management – Pearce
- Strategy & Business Landscape - Pankaj Ghemawat

Second Semester

HC05: MCB030 ORGANISATIONAL BEHAVIOUR

1. Course Outcome:

- CO1 Learn in depth individual behaviour in the organization
- CO2 Understand in detail with examples foundation of individual behaviour
- CO3 Deliberate the detail of skills in motivation
- CO4 Deliberate in detail with examples individual behaviour in group
- CO5 Identify in detail with examples resolve the conflicts

2. Pedagogy:

Teaching method comprises of lecture sessions and tutorials. Lecture sessions focus on providing conceptual understanding and analytical setting for select aspects of the course content.

3. Course Contents

Module1: Introduction: Meaning-Definitions and scope of organizational behavior-people-Organizational structure-technology and environment-OB as a Behavioral science-Contributing Discipline to OB-Psychology-Sociology-social psychology-Anthropology-Political science-OB and Management-Comparative roles in organization-Case studies.

Module 2: Foundations of Individual Behaviour: Biological Characteristics-Age-Sex-Marital Status-Number of Dependents-Tenure-Ability-Intellectual Abilities- Physical Abilities-The Ability-Job fit personality-personality determinants-Personality Traits-Major Personality Attributes influencing OB-Matching personality and Jobs-Values, attitudes, and Job satisfaction: Importance of Values-Sources of Value system-Sources and types of Attitudes-learning –Theories of learning shaping values-Case Studies.

Module 3: Motivation: The concept of Motivation-Early Theories of Motivation-Hierarchy of Needs theory-theory X and Theory Y-Hygiene theory-contemporary theories of motivation-ERG Theory-three needs theory-cognitive evaluation theory and others –case studies.

Module 4: Foundation of group behaviour: Defining and classifying groups-group process-group tasks-cohesive groups-group dynamics-leadership-nature and importance-functions-styles-communications-nature and types-effective communication-Roles of Formal and informal communication-Conflict management-The process of conflict-types of conflict-functional and dysfunctional conflict-resolution of conflict-case studies.

References:

Organisational Behaviour: Concept, Theory and Practice-Nirmal Singh

Organisational Behaviour - Fred Luthans

Organisation Theory and Behaviour - V S P Rao and PS Narayana

Organisational Behaviour - Niraj Kumar

Organisational Behaviour – K. Aswathappa

Management of organisational change –Harigopal

Course Outcome:

- CO1 Understand the characteristics of foundations and different dimensions of entrepreneurial development
- CO2 Write down in detail with examples skills of an young entrepreneurs
- CO3 Write down in detail with examples techniques of project planning, implementation and execution
- CO4 Identify in detail with examples institutional support to entrepreneurs
- CO5 Learn the characteristics of MIS in project

Pedagogy: A combination of Lectures, Case Analysis, Group Discussion, Seminars, Assignments, etc.

Module – 1: Entrepreneurship: Need, Scope, Entrepreneurial Competencies and Traits, Factors affecting Entrepreneurial Development, Entrepreneurial Motivation, Conceptual Model of Entrepreneurship, Entrepreneur Vs Intrapreneur, and Classification of Entrepreneurs. Micro, Small and Medium Enterprises (MSMEs): Meaning and Definitions of MSMEs, Features, Scope, Objectives, Relationship between Small and Large Units; Indian MSME Sector - Nature, Contribution to Economy, Problems and Government Schemes; and MSMEs Act, 2006.

Module – 2: Entrepreneurial Development Programs and Small Business: Relevance and Achievements of EDPs, Role of Government in Organizing such Programs, Women and Rural Entrepreneurs - Present Status in India. Small Business: Concept and Definition, Role of Small Business in Modern Indian Economy, Small Entrepreneur in International Business, Steps for starting a Small Industry, Registration as SSI, Role of SIDBI, Advantages and Problems of SSIs, Institutional Support Mechanism in India, EDI, Incubation Centers, Incentives and Facilities, and Government Policies for SSIs.

Module – 3: Project: Definition, Characteristics, Types, Steps in identification of Projects, Project Life Cycle. Project Management - Meaning, Scope and Importance, Role of Project Manager. Project Appraisal - Preparation of a Real Time Project, Feasibility Report containing Technical Appraisal, Environment Appraisal, Market Appraisal and Managerial Appraisal. Project Identification - Environment for Business Opportunities, Idea Generation, Short Listing and Selection of Product/Service, Stages in Venture Appraisal, Factory Design and Layout, and Feasibility Report Preparation.

Module – 4: Project Planning: Functions, Project Objectives and Policies, Identifying Strategic Project Variables; Statement of Work; Mile Stone Schedules Tools for Planning Hierarchy of Plans. Project Financing: Project Cost Estimation and Working Capital Requirement, Sources of Fund, Preparation of Projected Income Statement, etc; Implementation of Projects - Graphic representation of Project Activities, Management and Control of Projects, Project Scheduling, MIS in Project, Problems of Project Implementation.

Reference

1. Singh Narendra, Project Management and Control, Himalaya Publishing House.
2. Prasanna Chandra, Projects: Planning, Analysis, Selection, Implementation and Review, Tata McGraw Hill.
3. P. Gopala Krishnan and V. E Rama Moorthy, Project Management, MacMillan India.
4. Chandra Prasanna, Project Preparation, Appraisal and Implementation, Tata McGraw Hill.
5. A. N Desai, Entrepreneurship Management, Ashish Publishing House.
6. Vasanth Desai, The Dynamics of Entrepreneurial Development and Management, Himalaya Publishing House.
7. Nicholas, Project Management for Business and Technology: Principles and Practice, Prentice Hall of India.
8. Hall B. L, Pickle and Yance, Small Business Management, John Wiley & Sons, USA.

9. Kenneth R. Van Vloorthis, Entrepreneurship and Small Business Management, Allyn and Bacon. Kuvempu University, Master of Commerce (Regular), 2017-18 51
10. C. M. Bamback and J. R. Manscusu, Entrepreneurship and Venture Management, Prentice Hall of India.
11. Yound, Trevour L, Planning and Implementing Project, Sterling Publishing Ltd.
12. C. A Dailey, Entrepreneurship Management, McGraw Hill.

HC07: MCB010 CAPITAL MARKET INSTRUMENTS

1. Course Description:

Capital markets in recent times are flooded with new and innovative instruments enhancing vibrancy and volume of capital markets. Every advanced programme in commerce should consist of a course in analysis and evaluation of various instruments traded in capital markets today.

2. Course Outcome:

- CO1 Learn the characteristics of capital market in India
- CO2 Deliberate the classification and characteristics of various capital market instruments
- CO3 Specify the characteristics of dynamics of global capital markets
- CO4 Understand the classification and characteristics of derivatives in risk management
- CO5 Understand the characteristics of option and swaps

3. Pedagogy:

Teaching method comprises of lecture sessions and tutorials. Lecture sessions focus on providing conceptual understanding and analytical setting for select aspects of the course content.

4. Course Contents:

Module 1: Origin, Nature and Role of Capital Markets -Globalization of Capital markets- Capital Markets in India - Stock exchanges.

Module 2: Stocks, Bonds, Debentures – Types of Debentures, Private Placement, Guidelines for IPO – ADRs - GDRs – ETFs –Units of Mutual funds

Module 3: Derivatives- Basic features – Role of Derivatives markets - Forwards and Futures - Commodity Futures – stock Futures and Index futures.

Module 4: Options - Stock options and Index options-Swaps – Currency Swaps and Interest Rate Swaps.

References:

Financial institutions and markets- by Bhole (TMH).

Financial markets- by M.Y.Khan (Vivek).

Financial Derivatives – Dr.G.Kotreshwar

1. Course Outcome:

- CO1 Write down in depth scope of international business along with drivers of globalisation
- CO2 Specify the characteristics of policy and practice skills related to international business
- CO3 Understand the classification and characteristics of international business environment and the issues associated with them
- CO4 Identify in detail with examples modes of entry in international business
- CO5 Learn the detail of functions of WTO and MNCs

2. Pedagogy:

The course would be taught under LTP method. The lecture sessions are designed to be interactive with the student expected to come prepared with basic reading suggested before every session. The tutorial sessions are basically group exercises with each designated group handling a prescribed module for presentation and interaction, in a three-way interactive process. It basically involves preparing field reports and presenting them for plenary discussions.

3. Course Contents:

Module 1: Introduction: International Marketing-Trends in International Trade-Reasons for Going International-Global Sourcing and Production Sharing-International Orientations-Internationalization Stages and Orientations-Growing Economic Power of Developing Countries-International Business Decision-Case Studies.

Module 2: International Business Environment: Trading Environment-Commodity Agreements-Cartels-State Trading-Trading Blocks and Growing Intra-Regional Trade-Other Regional Groupings-SAARC-GATT/WTO and Trade Liberalization-The Uruguay Round-Evaluation-UNCTAD and WTO negotiation

Module 3: Multinational Corporations: Definition-Organizational Structures-Dominance of MNC's-Recent Trends-Code of Conduct-Multinationals in India-Case Studies.

Module 4: India in the Global Setting: India an Emerging Market-India in the Global Trade-Liberalization and Integration with Global Economy-Obstacles in Globalization-Factors Favoring Globalization-Globalization Strategies. Trade Policy and Regulation in India: Trade Strategies-Trade Strategy of India-Export-Import Policy-Regulation and Promotion of Foreign Trade in India-Case studies.

References:

- Chadha.G.K : WTO and Indian Economy
- G.S.Batra & R.C.Dangwal : International Business : New Trends
- Jean Pierre & H.David Hennessay : Global Marketing Strategies
- Justine Palu : International Business
- Francis Cheruniulam : International Business

1. Course Outcome:

- CO1 Learn in detail with examples Objectives and role of business research
CO2 Understand in depth Developing research proposal
CO3 Learn in detail with examples Questionnaire design
CO4 Write down in detail with application, if applicable, Measurement and scaling concepts
CO5 Deliberate the characteristics of Measurement and scaling concepts

2. Pedagogy:

The lecture sessions focus on providing conceptual understanding and analytical setting for select aspects of the course content. This session focuses on student involved and student driven content study. Identified groups of students make presentations and interact with both the faculty and the other students. The aspects reinforced through lecture and tutorial is taken up for practical study. Here the students would undertake field exercises related to different aspects of the course content.

3. Course Content:

Module 1: Introduction: Objectives and Role of Business Research–Distinct Features of Business Research-Theoretical Setting for Business Research–Ethical Issues in Business Research.

Module 2: Research Process: Developing a Research Proposal–Exploratory Research and Qualitative Analysis–Sources of Data- Methods of data collection–Techniques of Communicating with Respondents.

Module 3: Managing Research Assignment: Questionnaire Design-Sampling and Fieldwork Techniques-Measurement and Scaling Concepts-Attitude Measurement.

Module 4: Analysis and Presentation: Application of Univariate, Bivariate and Multivariate methods of Statistical Analysis-Methods of Business Research Report Writing–Language-Referencing-Bibliography.

References:

Business Research Methods, William G. Zikmund, The Dryden Press

Research for Development: A Practical Guide, Sophie Laws, VISTAAR Publications

Methodology in Social Research, Partha Nath Mukherjee, Sage Publications

1. Course Outcome:

CO1	Understand in depth practical aspects of investment analysis
CO2	Specify in depth functions of SEBI
CO3	Deliberate the detail of various investment alternatives
CO4	Deliberate in depth skills to construct investment portfolio
CO5	Identify in depth technical analysis of portfolio

2. Pedagogy:

In the LTP framework the Lecture sessions focus on building conceptual clarity and providing basic information on the nature and role of nonprofit sector. The reinforcement through Tutorial sessions focus on group exercises related to Indian nonprofit sector. Practical exercises involve field reports by students, both at individual and group levels.

3. Course Contents:

Module 1: Investment Setting - Securities – Sources of investment information – Security market indications – Security Contract regulation Act. Investor Protection. Over view of capital market, Institutional structure in capital market, Reforms and state of capital market, New issue market and problems, Securities and Exchange Board of India (SEBI), Debt Market.

Module 2: Fundamental Analysis - Economic Analysis – Economic forecasting and stock Investment Decisions – Forecasting techniques. Industry Analysis – Industry classification. Economy and Industry Analysis. Industry life cycle – Company Analysis Measuring Earnings – Forecasting Earnings – Applied Valuation Techniques – Graham and Dodds investor ratios.

Module 3: Technical Analysis - Fundamental Analysis Vs Technical Analysis – Charting methods – Market Indicators. Trend – Trend reversals – Patterns - Moving Average – Exponential moving Average – Oscillators – ROC Momentum – MACD – RSI – Stoastics.

Module 4: Portfolio Theory – Portfolio Construction – Diagnostics Management – Performance Evaluation – Portfolio revision- Mutual Funds.

References:

1. Donald E.Fischer & Ronald J.Jordan, 'Security Analysis & Portfolio Management', Prentice Hall of India Private Ltd., New Delhi 2000.
2. V.A.Avadhani – 'Securities Analysis and Portfolio Management', Himalaya Publishing House, 1997.
3. V.K.Bhalla, 'Investment Management', S.Chand & Company Ltd., Seventh Edition, 2000.
4. Punithavathy Pandian, 'Security Analysis & Portfolio Management' – Vikas Publishing House Pvt., Ltd., 2001.

1. Course Description: This course is designed to provide a deeper understanding of international accounting issues related to global financial reporting. It focuses on major diversities and challenges of financial reporting in the global arena, harmonization and international financial reporting standards. It also covers accounting for foreign currency transactions and major translation methods. It focuses on main issues in international financial statement analysis.

2. Course Outcome:

- CO1 Understand in depth international accounting issues related to global financial reporting
- CO2 Understand the detail of harmonisation of accounting
- CO3 Identify in detail with examples diversities and challenges of financial reporting
- CO4 Write down the detail of techniques of international financial statement analysis
- CO5 Learn the characteristics of accounting for foreign exchange rate fluctuations

3. Pedagogy: Method of instruction consists of lectures, analysis of international financial statements, group discussions, seminar presentations, writing assignments and tests. Reading and analysis of annual reports of multi-national organizations will be integral part of instruction.

4. Course Contents:

Module 1: International Financial Reporting: Definition and scope of international accounting. Main causes of diversity in international financial reporting. Harmonization vs. standardization. Rationale for and obstacles to harmonization. Role of regional and international organizations engaged in accounting harmonization – EU, ASEAN, IASB, IFAC, IOSCO, OECD, U.N., etc. Case studies on current evidence on accounting harmonization at regional level and global level.

Module 2: International Financial Reporting Standards (IFRS):The structure of IFRS. Process of IFRS Standard Setting. Constraints. Conceptual Framework for Financial Reporting under IFRS. Hierarchy of Standards. IFRS adoption or convergence. Case studies on compliance with IFRS by Indian Companies and MNCs.

Module 3: Accounting for Foreign Exchange Rate Fluctuations: An overview of foreign currency markets and exchange rates. Foreign exchange exposures. Accounting for spot and forward foreign currency transactions. Foreign currency translation methods- current rate method, current/non-current method, monetary/non-monetary method and temporal method. IFRS on foreign currency exchange accounting. Case studies on the annual reports of companies.

Module 4: International Financial Statement Analysis: Main issues in international financial statement analysis –understandability, availability, reliability, comparability and timeliness of financial information. Language, terminology and format. Techniques of financial statement analysis-horizontal, vertical, trend and ratio analysis. Analysis of income statement, balance sheet and cash flow statements of multinational corporations.

References:

- International Accounting by Shirin Rathore, Prentice-Hall of India, New Delhi
- Comparative International Accounting by Christopher Nubs and Robert Parker, Pearson Education Asia, New Delhi.
- International Accounting: A User Perspective by Shahrokh M. Saudagaran, South-Western Thomson Learning, Australia.
- International Accounting by A.K.Das Mohapatra, Prentice-Hall of India, New Delhi
- The Analysis and use of Financial Statements by Gerlad I.White, Ashwinipaul C.Sondhi and Dov Fried, John Wiley, New York.

The Economic Times, The Business Line and Financial Express daily papers.
Research Journals on International Accounting.
Internet Sources; www.iasb.org. www.worldbank.org. www.unctad.org. etc.,.

HC12: MCD020 CURRENT TRENDS IN BUSINESS AND COMMERCE

Course Outcome:

- CO1 Identify in detail with examples changing business and financial environment
- CO2 Specify the detail of reforms in the areas of banking and insurance
- CO3 Learn in depth trends in inflation
- CO4 Understand in depth trends in international business and collaboration
- CO5 Deliberate in depth social, economics and political event in the country

Pedagogy: LTP Model of facilitating learning process, working out MCQ's and group discussions.

Course contents:

1. All the compulsory subjects of M.Com programme.
2. The study of trends in the area of business and commerce – Trading, manufacturing, Finance, Capital Market, Foreign Trade, Foreign Exchange, Industrial relations, International Business and Collaborations, Taxation, Internal Marketing Scenario.
3. Trends in Inflation, Economics Development, Employment, Industry and Agriculture, World Economics and Trade Problems; Social, Economics and Political event in the Country and the world.

Reference:

1. Reserve bank of Indian Bulletin
2. Yojana
3. The Economic Times Daily
4. Business Line Daily
5. Business Standards
6. Business India

First Semester

SC01: MCA210 BUSINESS POLICY AND ENVIRONMENT

1. Course Outcome:

- CO1 Specify in detail with examples Policy formation
- CO2 Understand the detail of Environmental factors that influence business
- CO3 Understand the detail of Corporate social responsibility
- CO4 Deliberate the detail of Principles of business ethics
- CO5 Learn in detail with examples Control of MRTP

2. Pedagogy:

Teaching method comprises of lecture sessions and tutorials. Lecture sessions focus on providing conceptual understanding and analytical setting for select aspects of the course content.

3. Course Contents:

Module 1: Introduction: Business Policy: Importance of business policy-essentials of business policy-classification of business policy-Vision, Mission, Objectives–Types of Policies: Production policy-personnel policy- Financial policy-Marketing Policy-case studies.

Module 2: Business in a Social System, Business and Economic system –Business Objectives-Internal and External Business Environment.

Module 3: Business Environment: Internal Environment ofBusiness-External Environment-Economic-Political-Socio-Cultural-Technological Environment –case studies

Module 4: Business Ethics: Principles of Business Ethics; Doctrine of trusteeship; unethical practices; good ethics and good business. Social responsibility of business; Doctrine of social responsibility: Rationale of social responsibility; control of monopoly and restrictive and unfair trade practices.

References:

A concept of corporate planning-, Russel Ackoff, Newyork wiley

Business policy and strategic management- Tokyo, McGraw hill

Strategic Management-Text and Cases- V.S.P. Rao and V. Harikrishna

Strategic Management-Azar Kazmi

Strategic Management-Francis Cherunillam

Strategic Management-Subba Rao.

SC02: MCA220 STATISTICS FOR BUSINESS DECISIONS

1. Course Description:

The course comprises of probability theories, sampling techniques, time series analysis and multivariate analysis.

2. Course Outcome:

CO1	Understand the detail of probability theory and sampling
CO2	Learn in detail with examples various methods of theoretical probability distribution
CO3	Specify the detail of different tools in taking business decision
CO4	Understand the detail of advanced application oriented test
CO5	Understand in detail with examples time series analysis

3. Pedagogy:

Class room teaching of basic statistical models shall be followed by solving problems involving business applications. Assigned problems are to be worked on an individual basis, followed by group discussion of case problems.

4. Course Contents:

Module 1:Probability: Meaning, terminology, types and rules. Random variables and use of expected value in decision making. Binomial, Poisson and Normal probability distributions- their characteristics and applications in business decisions.

Module 2:Sampling: Meaning of sample and population. Probability and non-probability methods of sampling. Use of random digits to choose random samples. Sampling from normal and non-normal populations. The Central limit theorem. Use of sampling in business decisions

Module 3:Time Series Analysis : Variations in time series. Cyclical, seasonal and irregular variations. Trend analysis. Application of time series analysis in forecasting.

Module 4:Multivariate Analysis: Multiple regression and correlation analysis. Analysis of Variance. Application of multivariate analysis in business decisions. Using Statistical Package for Social Sciences (SPSS) to solve problems.

References:

Wonnacott and Wonnacott: "Statistics for Business and Economics" Wiley Publications

Wonnacott and Wonnacott: "Econometrics" Wiley Publications

Sanchetti and Kapoor: "Statistics"

Morris Hamber: "Statistical Analysis for Decision Making"

Richard Livin and David Robin: "Statistics for Management"

- 1. Course Description:** The course essentially covers all aspects of disaster management.
- 2. Course Outcome:**
 - CO1: Understand the concept of disaster management.
 - CO2: Learn the management of natural and man-made disaster.
 - CO3: Knowledge of National policies on Disaster Management.
- 3. Pedagogy:** Method of instruction consists of lectures, case study design and analysis, group discussions, seminar presentations, writing assignments and tests.

4. Course content:

Module 1: Basics of Disaster: Definition of hazard and disaster; typology of disaster; natural and human; made disasters. Geological disaster; earth quakes and seismology, volcanic eruptions, tsunami and landslides. Hydrological disasters- floods, droughts and famines, cyclones and hurricanes.

Module 2: Human made disaster: Human instigated disaster; communal forces and violence, caste conflicts, ethnic conflicts, refugees. Industrial and technological accidents, system failure, explosion and chemical leakers/spillage, biological weapons.

Module 3: National policy on disaster management, contingency action plans, financial assistance, from center to states under calamity relief fund. Crisis management groups; their formations and functions

Module 4: Management of disaster: Risk assessment and disaster response, quantification techniques, NGO management, SWOT Analysis based on design and formulation strategies, insurance and risk management, role of financial institutions in mitigation effort.

References:

- Encyclopedia of disaster management By Goel, S. L. Deep & Deep, Publication Pvt, Ltd.
Disaster management by G. K. Ghosh, A. P.H Publishing corporation.
Disaster management by R. B. Singh, Rawat Publications
Disaster management : Through the new millennium by Ayaz Ahmed, Anmol Publications.
Emergency medical services and disaster management: A Holistic approach by P.K.Dave, Gaytee Brothers medical Publishers(P) ltd.
Disaster management by D. Narayan, A.P.H. Publishing Corporations.
Model Encyclopedia of Disaster and Hazard management by B.C. Bose, Rajath Publications.
Disaster management by Nikuj Kumar, Alpha Publications.
Disaster management: recent approaches by Aravind Kumar. Anmol Publications.

SC05: MCB240 HUMAN RESOURCE MANAGEMENT

1. Course Outcome:

- CO1 Learn in depth significance and management of human resources in organisations
- CO2 Specify in depth human resource planning
- CO3 Understand in detail with examples human resource development
- CO4 Specify the characteristics of reward system
- CO5 Deliberate in detail with examples 360 degree appraisal

2. Pedagogy:

Teaching method comprises of lecture sessions and tutorials. Lecture sessions focus on providing conceptual understanding and analytical setting for select aspects of the course content.

3. Course Contents:

Module 1: Environmental context: New economic policy and changing business-technological –socio-economic and political and legal environment, structural reforms and their implications for HRM in India-Response of the management-worker and unions to structural reforms and their implications for HRM in India-Response of the management –Worker and unions to structural adjustment. Concepts of human resource management-Meaning-Objectives-Scope and functions-Perspective of HRM: linking corporate strategies and policies with HRM Organization of HRM department.

Module 2: Human Resources planning and Procurement; Job analysis and evaluation-job description-job specification -job rotation and job enrichment. Human resource planning- importance-objectives and problems. Recruitment-meaning-recruitment policy, sources –factors affecting selection decision-selection procedure. Human resource information system.

Module 3: Human resource development: Meaning-concepts of HRD-objectives of training-organization of training programmers-methods-advantages and limitations of training. Evaluation of training program HRD for total quality management. Transfer policy Promotion policy-Demotion and Discipline-consequences of indiscipline –disciplinary procedure.

Module 4: Compensation/Rewards system: Significance of reward system in business organisation. Compensation system in practice-systems of promoting -factors determining employee compensation and rewards-dearness allowance, employee benefits-bonus-laws on wages, bonus and social security-managerial compensation. Performance Appraisal: concepts, objectives philosophy and process of performance appraisal system- counseling-career planning and management.

References:

1. Human Resource Management: Strategies and Action -Armstrong
2. Human Resource Management -Dr.Ashwathappa
3. Personnel and Human Resource Management -D.A. Deonz and F.P. Robins
4. Personnel Management - Edwin Phillip
5. Human Resources Management—L.M. Prasad

SC06: MCB270 MANAGEMENT OF FINANCIAL SERVICES

1. Course Description:

This course is all about Financial Services industry analysis, trends, globalization and government policy. It also includes an analysis of future of Financial Services, particularly in respect of investment banking, micro-finance/insurance and angel investing.

2. Course Outcome:

- CO1 Specify in detail with examples role of financial service industry in India
- CO2 Understand the detail of globalisation of financial service
- CO3 Deliberate the detail of trends in financial services in India
- CO4 Understand the detail of Factoring service and securitisation system
- CO5 Identify in detail with examples futures of financial services industries

3. Pedagogy:

Students must work out assigned individual topics, present seminars and participate in case studies or group discussions.

4. Course Contents:

Module 1: Concept and Scope of Financial Services – Marketing of Financial Services Problems and Prospects of Financial Services Industry in India.

Module 2: Globalization of Financial Services – GATS – Functions and the Role of GATS Evaluation of Government’s policy towards globalization.

Module 3: An Overview of Trends in Financial Services in India – Lease Financing; Types, History and Development of Lease Financing, Structure of Leasing Industries, Factoring and International Factoring, Securitization- meaning, features, process, IPO Management – Venture Capital; Features, Methods of Venture Financing, Initiatives – Credit Rating; functions –Credit Rating Agencies. Portfolio Management Services.

Module 4: Future of Financial Services Industry – Micro Finance/Insurance – Angel investing – Investment Banking.

References:

1. Financial Markets and Institutions – Bhole L. M. (Tata McGraw Hill)
2. Indian Financial System – Theory and Practice – Khan M.V (Vikas Publishing)
3. Financial Management and Control – Chakraborty S.K (Mc Milan)
4. New Issues Markets - Khan M.Y (Allied Publishers)
5. Emerging Scenario of Financial Services – Gordon & Natarajan (Himalaya Publishing House)
6. Financial Services in India - M.A. Kohok (Himalaya Publishing House)
7. Management of Financial Services – Avadhani (Himalaya Publishing House)

1. Course description: The course deals with study of technology adaption in banking, besides providing scope of practical side of banking

2. Course Outcome:

- CO1 Identify the characteristics of Recent developments in banking technology
- CO2 Identify in depth Impact of technology on banks
- CO3 Specify in depth Payment channels and their delivery system
- CO4 Deliberate the detail of Global developments in banking technology
- CO5 Understand in depth Confidentiality information system audit

3. Pedagogy: LTP model, field survey, summary and group discussions.

4. Course Contents:

Module 1: Branch Operation and Core Banking - Introduction and Evolution of Bank Management – Technological Impact in Banking Operations – Total Branch Computerization – Concept of Opportunities – Centralized Banking – Concept, Opportunities, Challenges & Implementation . – Case study: Analysis of Recent Core Banking Software.

Module 2: Delivery Channels - Overview of delivery channels – E – Lounge establishment - Phone Banking – Call centers – Internet Banking – Mobile Banking – Payment Gateways – Card technologies – MICR electronic clearing

Module 3: Interbank Payment System - Interface with Payment system Network – Structured Financial Messaging system – Electronic Fund transfer – RTGSS – Negotiated Dealing Systems & Securities Settlement Systems – Electronic Money – E Cheques

Module 4: Global developments in banking technology - Information technology in finance and service delivery. Impact of technology on banks – protecting the confidentiality information system audit. Audit measures in computerized environment information system security and disaster management.

References:

1. Financial Services Information Systems - Jessica Keyes Auerbach publication; 2nd edition (March 24, 2000)(Text Book)
2. Kaptan S S & Choubey N S., “E-Indian Banking in Electronic Era”, Sarup & Sons, New Delhi, 2003
3. Vasudeva, “E – Banking”, Common Wealth Publishers, New Delhi, 2005
4. Turban Rainer Potter, Information Technology, John Wiely & Sons Inc
5. Banking Technology - Indian Institute of Bankers Publication

Fourth Semester

SC09: MCD260 **ADVANCED COST ACCOUNTING**

1. **Course Description:** This course is intended to expose students on advanced aspects of cost accounting

2. **Course Outcome:**

CO1	Specify the characteristics of reconciliation between cost and cost financial accounts
CO2	Deliberate in depth Technique of product life cycle costing
CO3	Learn in depth Value chain analysis
CO4	Deliberate the detail of Productivity and concept of measurement
CO5	Deliberate in depth Advanced cost accounting practices

3. **Pedagogy:** Method of instruction consists of lectures, case study design and analysis, group discussions, seminar presentations, writing assignments and tests.

4. **Course content:**

Module 1: Cost Book Keeping and Reconciliation between Cost and Cost financial Accounts – Book - keeping, Cost Ledgers, interlocking and integral Accounts. Reconciliation of Cost and Financial Accounts, Reasons, Needs, Methods.

Module 2 : Product Life Cycle Costing: Introduction, Product Life cycle, Phases and Characteristics of Product Life Cycle, Stages of Product Life Cycle, Product Life Cycle Costing Features and benefits of Product Life Cycle Costing.

Module 3 : Value Chain Analysis Introduction - Definition – Role of Management Accountant – Value Chain Analysis – approach for assessing competitive advantages – value chain analysis v/s conventional management accounting.

Module 4 : Productivity & Concept and Measurement i) Productivity Meaning, Measurement of Material, Labour, Capital and Management Productivity. Productivity V/s Efficiency. Capacity - Theoretical, Practical and idle capacity, Capacity utilization and effect of same on cost. ii) Concept and Measurement Measures to improve productivity - Technical, Financial, Operational Measures. Restructuring of activities - Business Process Re-engineering elementary knowledge. Human aspect of productivity.

References:

Cost accounting – S P Jain and Narang
Cost accounting – NK Prasad

SC 10: MCD210

SUPPLY CHAIN MANAGEMENT

1. Course Outcome:

CO1	Understand the detail of fundamentals of supply chain management
CO2	Deliberate the detail of foundational role of logistics management
CO3	Understand in detail with examples evaluation of marketing channels
CO4	Learn in depth channel management
CO5	Identify the detail of staffing the sales team

2. Pedagogy:

Method of instruction consists of lectures, group discussions, seminar presentations, writing assignments and tests. Reading and analysis of supply chain strategies of national and multi-national organisations will be integral part of instruction.

3. Course Contents:

Module 1 Introduction to Sales Management: Sales Management: Its Nature, Rewards, and Responsibilities, Social, Ethical, and Legal Responsibilities of Sales Personnel. Building Relationships through Strategic Planning, The Market-Driven Sales Organization, Forecasting Market Demand and Sales Budgets Design and Size of Sales Territories, Sales Objectives and Quotas, staffing the Sales Team - Planning for and Recruiting Successful Salespeople, Selection, Placement, and Socialization of Successful Salespeople, Training the Sales Team - The Management of Sales Training and Development, Contents of the Sales Training Program: Sales Knowledge and the Selling Process, Directing The Sales Team -Motivating Salespeople toward High Performance, Compensation for High Performance, Leading the Sales Team

Module 2 Marketing Logistics : Logistics and its importance, Functions of Logistics management - Procurement /Purchasing, Inward Transport, Receiving, Warehousing, Stock Control, Order Picking, Materials Handling, Outward Transport, Physical Distribution Management, Recycling, Returns, and Waste Disposal, Importance of Communication in Logistics, Technology in Logistics- Electronic Data interchange (EDI), Artificial Intelligence, Expert Systems, Communication Technology, Bar Coding and Scanning, Streamlining the Logistics Process, Strategic Issues in Logistics Management

Module 3 Marketing Channels: Evolution of Marketing Channels- The Production Era, The Sales Era, The Marketing Era, Relationship Marketing Era, Channel member and their roles, Roles of Channel Members, Channel Functions, Designing marketing channels - Channel Structure, Channel Intensity, Types of Channel Intermediaries at Each Level, Channel Flows and Cost.Importance of Channel Integration, Vertical Marketing Systems, Types of vertical marketing systems - Corporate VMS, Administered VMS, Contractual VMS, Horizontal Marketing Systems, Hybrid channel system, Designing and Managing Hybrid Channel Systems

Module 4 Channel Management: Recruiting Channel, Members - Recruiting as a Continuous Process, Recruiting Manufacturers, Screening, Criteria for Selecting Channel Members - Sales Factors, Product Factors, Experience Factors, Administrative Factors, Risk Factors, Motivating Channel Members, Distributor Advisory Councils, Modifying Channel Arrangements - PLC Changes, Customer-Driven Refinement of Existing Channels, Growth of Multi-Channel Marketing Systems, Managing Channel Relationships - Cooperation and coordination, Conflict, Power

References:

Sales And Distribution Management, Tapan Panda And Sunil Sahadev, Oxford Publications
Sales Management, Still And Cundiff, PHI
S.L. Gupta, Sales & Distribution Management, Excel Books
Marketing Channels, Coughian, Anderson, PHI
Sales And Distribution Management; Krishna Havaldar And Cavale; Tata Mcgraw-Hill
Sales Management, Dasgupta, PHI

ELECTIVE GROUPS

GROUP C: BUSINESS TAXATION

Paper-1

SC11: MCC230 INDIRECT TAX LAW AND PRACTICE

Goods and Services Tax and Customs Duty

1. **Course Description:** This paper is to educate the taxation students about Indian Tax System, its background, and its operation in the global competitive market. The importance and administration of the indirect taxes in the Indian market oriented economy and its role in achieving the objectives of modern welfare government. To understand the relevance of GST in the present Indian Tax Scenario.

2. Course Outcome:

- CO1 Understand in depth significance and contribution of indirect taxes in India
- CO2 Understand in detail with examples principles of taxation and incidence process of indirect taxes in market oriented economy
- CO3 Understand the implications of indirect taxes on the taxable capacity of consumers, dealers and society at large.
- CO4 Learn in depth tax planning, tax management, payment of tax and filling of tax returns
- CO5 Specify in depth impact of GST on Domestic, National and International trade

3. Pedagogy:

- 1) Lecture:
- 2) Tutorial and bridge class*(Non tax students)
- 3) Live leading cases pending and deciding in the high court and supreme courts.
- 4) Practical works: Tax planning, Tax management, filing of various tax returns and working as consultants and tax adviser for small companies nearby dealers and companies relating to GST and Customs

2. Course Contents:

Module 1: Constitutional Provision for Indian tax system, structure of Indian tax system. Public finance, public expenditure and public revenue. Principle and objectives of taxations in the modern welfare governments. Indian tax structure, different types of taxes- taxes under indirect tax family in India, methods of taxations, tax reforms and recent tax reforms committees and its recommendations and fiscal discipline.

Module 2: GST-Genesis, History, Constitutional Background of GST, GST Bills, GST- Central and State Financial relations, Finance commissions, Salient features of GST, Tax Reforms and GST in India, Tax compliance, GST administrative structure both central and state level, Advantages of GST- Economy, Governments, dealers and consumers, E-Commerce and GST GST-Issues, challenges and problems.- **Legal case study.**

Module-3: Taxes under GST- levy and incidence of GST, Value of Tax event, Rates and Schedules, CGST, SGST, IGST, Criteria for GST, GST on Exports, Imports and SEZ supplies input credit, Payment of Taxes, Returns Filing, Assessment and Audit of Accounts, under GST, GST refund, Appeals and Revision, Prosecution and Appellate Tribunals. -**Case studies.**

Module: 4: Customs Duty: Customs Act, 1962 and Rules, regulations Circulars and Notifications; Customs Tariff Acts, and the related Rules. Principles governing levy of customs duty, types of duties at global scenario. Basic principles of classification of goods and valuation of goods, special provisions regarding baggage, goods imported or exported by post, and stores. Duty drawback schemes, GATT and WTO objectives principles in customs duty.- **Case studies.**

References:

- 1) Indirect Taxes- Law and Practice: V.S.Datey
- 2) Karnataka GST Manuals
- 3) GST Ready Reckoner
- 4) Bare Acts of Customs, Customs Tariff Acts, GST
- 5) Public Fiancé- Mahaveer Thagy
- 6) Public Finance Prof.H.Doltan
- 7) GST Bill/Act 2016
- 8) CST Law and Practice-SS Gupta
- 9) Basic of GST-Nitya Tax association Taxman
- 10) GST Manual- Taxman publication
- 11) Indian GST for Beigneers –Jayaram Hiregange and Deepak Rao
- 12) CA Practical Manuals
- 13) www.gstindia.com

Paper-2: SC12: MCD230 CORPORATE TAX LAW AND PLANNING

1. Course Description: This course is focused on different heads of income, taxable in the hands of companies, computation of gross total income, deduction, exemptions, set off and carry forward of loss. Tax planning relating to various managerial decisions for reducing the tax burden, allocation of investments, and maximize the company wealth. As a tax consultant of the corporate tax laws of the company to give advice to the drawing officers regarding TDS, advance payment of tax and remittances of tax, for his employees.

2. Course Outcome:

CO1 Understand in detail with examples incidence of tax on residential status of the

	company
CO2	Deliberate in detail with examples different types of companies under corporate income tax
CO3	Understand the detail of different sources of income for corporate assesses
CO4	Specify in depth impudence of tax planning
CO5	Understand in depth procedure for assessment

3. Pedagogy: The course content is covered class room lecture, remedial class for non tax students, student's seminar, case discussion, and work out the problem on the company problems as student, as consultant and as a tax authority and also visiting company and tax office for practical exposure.

4. Course Content:

Module 1: Definition of company-Indian company, Domestic Company, Foreign Company, Widely Held Company, Closely held company, Residential Status of a company and incidence of Tax.

Module 2: Computation of Taxable incomeof companies- Computation of table income under different heads of income-House property, Profit and gain from business or profession, Capital gain and income other sources, carry forward and set off of losses in case of companies. Deduction from Gross Total income. Minimum Alternative Tax.

Module 3: Tax Planning- Tax avoidance and tax evasion. Tax planning with corporate dividend, Dividend policy-bonus shares. Tax planning with reference to specific managerial decisions- Make or Buy, Own or Lease, Purchase by installment or by Hire, Repair, Replace, Renewal or Renovation, shut down or continue.

Module 4: Procedure for assessment- Deduction of Tax at Source, Advance payment of Tax, Tax returns, refunds appeals and revision.

References:

Direct Taxes-Dr.H.C. Mehrotra and Dr.S.P.Goyal Sahitya Bhavn New Delhi.

Direct Taxes law and practice-Bhagavathi Prasad, Vishva Prakashana, New Delhi.

Direct Taxes Aggarval P.K "Tax Planning for Companies" Hind Law Publishers, New Delhi.

Corporate Tax Planning and Management, Lakhota, Vision Publishers.

Taxman's Direct Tax Laws and Practice, Dr.Vinod K Singhania and Kapil Singania Taxman's Publications(p)Ltd., New Delhi.

GROUP D: MANAGEMENT ACCOUNTING

Elective Paper-1

SC13: MCC250: MARGINAL COSTING AND DECISION MAKING

1. Course Outcome:

CO1	Deliberate in depth skill relating to marginal costing
CO2	Identify the detail of wide range of managerial decisions
CO3	Deliberate the detail of techniques of controlling cost through standard costing
CO4	Understand the detail of managerial cost control decisions
CO5	Learn the detail of direct costing

2. Pedagogy:

Course activities consist of lectures, case study analysis, group discussions, seminar presentation, assignment writing and tests. Solving problems and evaluating decisions involving the financial and cost data of selected firms will be integral part of instruction

3. Course Contents:

Module 1: Introduction: Meaning- terminology- Scope & Concepts- Cost Behavior Analysis- Break Even Analysis- Approaches of Break Even Analysis in relation to cost & revenue. Factors- Multi-product Break Even Analysis- Assumptions Underlying Break Even Analysis- Limitations of Break Even Analysis- Case Studies.

Module 2: Contribution Concepts & Short term Profitability Analysis: Profitability Analysis Under Constrained Conditions- Profit- Volume Ratio & its Uses- Profit Volume Graphs – Case Studies.

Module 3: Marginal Costing & Managerial Decisions: Profit Planning- Pricing Decision – Production Decision – Make and Buy Decision Joint & By-product Decision – Distribution Cost Analysis- Case Studies.

Module 4: Direct Costing: Meaning- Importance & Preparation of income statements- Comparison with Absorption Costing- Arguments in Favour of Direct Costing- Criticisms of Direct Costing. Value Analysis & Value Engineering: Basic Concept of Value- Constitution of a Value Analysis Team- Procedures Underlying Value Analysis Study- Benefits From & Resistance to Value Analysis Study- Reporting to Management- Objectives of Reporting- Reporting Needs of Different Management Levels- Types of Reports- General Principles of Reporting- Modes of Reporting- Reports to the Board of Directors- Reports to Top Management- Reporting to top Divisional Management- Reports to Junior Management Level- Preparation of Reports- use of Reports by Management- Case Studies.

References:

- J. Batty : Management Accountancy, ELBS
C.T.Horngel : Cost Accounting- A Managerial Emphasis
M.R.S. Murthy : cost Analysis for Management Decisions, Tata Mc Graw Hill

Elective Paper-2

SC14: MCD250: TOOLS AND TECHNIQUES OF CONTROL

1. Course Outcome:

- CO1 Specify the detail of cost control and management
- CO2 Deliberate the detail of cost control and management tools
- CO3 Learn in detail with examples costing system for job and process oriented manufacturing environments
- CO4 Identify the classification and characteristics of uniform costing and inter-firm comparison
- CO5 Learn in depth objectives and criticism of management audit

2. Pedagogy:

Course activities consist of lectures, case study analysis, group discussions, seminar presentation, assignment writing and tests. Analysis and evaluation of scope for cost reduction involving the value engineering and ABC activities of selected firms will be integral part of instruction.

3. Course Contents:

Module 1: Budgetary Control: Objectives of Budgetary Control-Preparation of the Budget-Functional Budgets-Sales Budgets-Production Budget-Cost Budget-Plant Utilization Budget Capital Expenditure Budget-Selling & Distribution Cost Budget-Purchasing Budget & Cost Budget-The Master Budget-Operation of Budgetary Control-Flexible Budgetary Control-Zero-Base Budgeting-Case Studies.

Module 2: Standard Costing: Objectives-Principles-Determination of Standards for Material-Labor-Direct Expenses & Overhead Costs-Variable and Fixed Costs-Case Studies.

Module 3: Variance analyses: Material, Labor, and Overhead Variances-sales & Profit Variances-Disposition of Variances-Assessing the Significance of Standard Cost Variance-Standard Cost Accounting-Case Studies.

Module 4: Uniform costing & Interfirm Comparisons: Objectives and Purposes Underlying Uniform Costing-Development of Uniform Costing-Cost Audit-Meaning & Definition-Inclusion of Clause B to Sec.208 to Sub Sec. (d) to Sec. 209-Indian Companies Act 1956-Appointment of Cost-Cost Audit Programme-Records Relating to Materials-Labor Overhead-Depreciation-Stores & Spare Parts-Work-in-progress and Incomplete Contracts-Cost Auditor's Report-Application of Cost Audit Report Rules, 1963-Sachar Committee's Report. Management Audit: Meaning & Definition-objectives & Criticisms-Types of Audits-Arguments for & Against Management Audit-Social Audit-Steps Underlying Social Audit Programme-Social Audit Report-Limitations of Social Audits-Case Studies.

References:

- Welsh, Glenn A. : Profit, Planning and Control(prentice Hall)
- J. Batty : Standard Costing
- M.R.S. Murthy : Cost Analysis for Management Decisions, Tata Mc Graw Hill

Open Elective

OE01: 540

RETAIL BANKING

1. Course Description:

This course is designed to provide a basic understanding of Personal Banking. It focuses on functions of banks, banker and customer relationship, Opening of bank accounts and their operations, bank deposits and loans and banking technology.

2. Course Outcome:

CO1	Understand the detail of functions associated with banking principles and practices
CO2	Deliberate in detail with examples services provided by banks to individual customers
CO3	Learn in depth retail banking scenario
CO4	Understand in detail with examples different models and strategies adopted by the banks
CO5	Understand the detail of banking technologies

3. Pedagogy:

Method of instruction consists of lectures, case study design and analysis, group discussions, seminar presentation, writing assignments and tests. Interaction with customers of banks, managers and employees of banks.

4. Course Contents:

Module 1: Functions of Banks: Introduction, Types of Banks, Payment Bankers-Recent Developments. Functions of Reserve Bank of India, Commercial Banks, Private Sector Banks, and Grameena Banks.

Module 2: Banker and Customer Relationship: Different Customers-Special type of Customers, Know your Customer [KYC] guidelines- Opening of different bank accounts and procedures for their operations.

Module 3: Bank Deposits and Loans: Pledge, Hypothecation, Mortgage-Utility. Different Deposit and Loan Products of Banks, Rate of Interest- Fixed and Floating, Documentation Procedures.

Module 4: Banking Technology: Net Banking-Procedure-Safety, Digital Banking-Pro's and Con's. Electronic Banking, Core Banking Technology, Debit and Credit Cards, ATMs.

References:

Machiraju, H.R., Indian Financial System, Vikas, New Delhi.
Verma, J.C. Merchant Banking, TMH, New Delhi.
Mithani and Gordeon, Banking Theory and Practice, Himalaya, Bombay.
Bhole, L.B., Financial Institutions and Markets, TMH, New Delhi.

OE02:580 PERSONAL FINANCIAL MANAGEMENT

1. Course Description: This course is designed to provide a deeper understanding of Personal Financial Management It focuses on basics of personal financial management, personal savings and investment plans, computation of return and risk factor of personal savings and investments, retirement savings plans.

2. Course Outcome:

- CO1 Understand in depth personal financial management process
- CO2 Learn the detail of plan personal budget
- CO3 Specify the detail of financial planning in the changing personal and financial environment
- CO4 Identify in depth various investment alternatives of personal financial planning
- CO5 Deliberate the detail of retirement savings plans

3. Pedagogy: Method of instruction consists of lectures, case study design and analysis, group discussions, seminar presentation, writing assignments and tests. Interaction individual investors with different profiles by age, income, sex, occupation, and region.

4. Course Contents:

Module 1: Basics of Personal Financial Management: Basics of Personal Financial Management: Budget, The Personal Financial Planning Process, Preparation of Personal Budget, Personal Financial Statements, Personal Income Tax Planning. Case studies on personal financial planning of individuals.

Module 2: Personal Savings & Investment: Investment Criteria- liquidity, safety and profitability. Savings instruments of Post Office and Banks. Chit Funds. Investment in Shares, Debentures, Corporate and Government Bonds, Mutual Fund. Investment in Physical Assets – Real Estate, Gold and Silver, Gold certificates. Risk and Return associated with these investments. Case studies on risk and return perception of retail investors on various investments.

Module 3: Computation of Return and Risk of Personal Investment: Present Value and Future Value of a Single Amount and an Annuity. Computation of interest, dividend and capital gains on personal investments. Impact of leverage on return. Personal tax planning,

Module 4: Retirement Savings Plans: Retirement Savings Plans:, Gratuity plans from the government. Life Insurance types. Health insurance. Pension Plans- Defined Contribution Plan and Defined Benefit Plan. Provident Fund, Gratuity. Life Insurance Plans. General Insurance Plans. Reverse Mortgage Plans.

References:

Personal Finance by Jack R. Kapoor, Les R. Dlabay and Robert J. Hughes, Tat McGraw-Hill Publishing Company Ltd. New Delhi.

Financial Education by Reserve Bank of India – rbi.org.

Personal Finance columns in The Economic Times, The Business Line and Financial Express Daily News Papers.

Information Broachers of Post Offices, Banks, Mutual Funds, Insurance Companies

Internet Sources- BSE, NSE, SEBI, RBI, IRDA, AMFI etc.

Question paper pattern (M.Com. 2016-17 Scheme) All courses except CTBC

Time: 3 Hours

Max Marks: 70

Section A

Answer any five questions: $5 \times 5 = 25$

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.

Section B

Answer any three questions: $10 \times 3 = 30$

- 9.
- 10.
- 11.
- 12.
- 13.

Section C

14. Case Study (Compulsory)

Note: In case problem subjects / working papers section A may comprise of two problems and section B may comprise of three problems. Section C shall be a problem only.

**Question Paper pattern M.Com. 2016-17 Scheme for Current Trends in Business and
Commerce**

Time : 3 Hours

Max Marks: 70

Section A

Answer any five questions: $5 \times 5 = 25$

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.

Section B

Answer any three questions: $10 \times 2 = 20$

- 9.
- 10.
- 11.

Section C: $1 \times 25 = 25$

Multiple choice questions (25 numbers) , each question carrying one mark

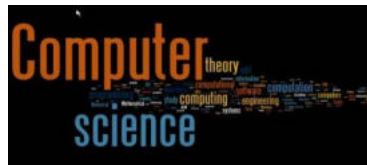


**JSS COLLEGE OF ARTS, COMMERCE & SCIENCE
(AUTONOMOUS)**

OOTY ROAD, MYSURU-570 025

(Autonomous under University of Mysore: Re-accredited by NAAC with 'A' Grade)

**Choice Based Credit System and CAGP
MASTERS DEGREE**



Syllabus

2020-22

Postgraduate Department of Computer Science

JSS College of Arts, Commerce and Science

Ooty Road, Mysore-25

2020-2022

PROGRAMME: MASTER OF SCIENCE IN COMPUTER SCIENCE

2 Years /4 Semesters (under CBCS-CAGP)

ADMISSIONS:

- i) 50% seats of the total intake for M.Sc., Computer Science Programme of the College will be filled-up by University of Mysore through Centralized Admission Cell as per University regulations.
- ii) Remaining 50% seats will be filled-up by the College under College Quota.

ADMISSION REQUIREMENT

Eligibility - All students who have 45% (40% for SC/ST) in their B.Sc degree (from Recognised University/ Open University) with Mathematics as one subject and any one-two among Electronics, Instrumentation, Computer Science or Computer Maintenance and Physics as other subject. OR Student with B.Sc (IT) are permitted provided they have not studied Mathematics in their 2 years PUC. OR Diploma (in Electronics, Computer Science and Information Technology) holders with B.Sc (IT) are permitted if they have SSLC level Mathematics. OR BCA degree from recognized University OR equivalent to this from any other recognized University in India or abroad is also permitted.

PROGRAMME OUTCOMES:

After Completing the M.Sc Program Students will be able to:

- PO1. Identify, formulate, and solve computer science problems
- PO2. Design, implement, test, and evaluate a computer system, component, or algorithm to meet desired needs
- PO3. Receive the broad education necessary to understand the impact of computer science solutions in a global and societal context
- PO4. Communicate effectively
- PO5. Success in research or industry related to computer science
- PO6. Have solid knowledge in computer science and engineering, including programming and languages, algorithms, theory, databases, etc.
- PO7. Integrate well into and contribute to the local society and the global community related to computer science
- PO8. Practice high standard of professional ethics
- PO9. Draw on and integrate knowledge from many related areas

PROGRAMME SPECIFIC OUTCOMES:

PSO1. Programmers or the Software Engineers with the sound knowledge of practical and theoretical concepts for developing software.

PSO2. Serve as the Computer Engineers with enhanced knowledge of computers And its building blocks. Work as the Hardware Designers/Engineers with the knowledge of Networking Concepts.

PSO3. Work as the System Engineers and System integrators Serve as the System Administrators with thorough knowledge of DBMS.

PSO4. Work as the Support Engineers and the Technical Writers

PSO5. Work as IT Sales and Marketing person.

PSO6. Serve as the IT Officers in Banks and cooperative societies.

PSO7. Computer Scientist in research and R & D laboratories.

PSO8. Faculty for Graduate and Under graduate Colleges.

MSc. in Computer Science – 2018**I Semester**

Course Code	Course	L:T:P	Credit Value
CSC101	HC1 (Data Structures and Algorithms)	3:0:1	4
CSC102	HC2 (System Software)	2:1:1	4
CSC103	HC3 (Computer Networks)	2:1:1	4
CSC104	SC1		4
CSC105	SC2		4
	TOTAL		20

II Semester

Course Code	Course	L:T:P	Credit Value
CSC201	HC4 (Analysis and Design of Algorithms)	2:1:1	4
CSC202	HC5 (Operating System and Unix)	2:0:2	4
CSC203	HC6 (Computer Graphics)	3:0:1	4
CSC204	SC3		4
CSC205	SC4		4
	TOTAL		20

III Semester

Course Code	Course	L:T:P	Credit Value
CSC301	HC7 (Software Engineering)	3:1:0	4
CSC302	HC8 (Theory of Languages)	3:1:0	4
CSC303	HC9 (Database Management System)	2:1:1	4
CSC304	SC5 / Term Work		4
CSC305	SC6 (Open Elective *****)		4
	TOTAL		20

IV Semester

Course Code	Course	L:T:P	Credit Value
CSC401	HC10 (Major Project)	0:1:7	8
CSC402	SC7		4
CSC403	SC8		4
	TOTAL		16

****** Open Elective Course: III Semester**

Course Code	Course	L:T:P	Credit Value
CSC305	OE- Computer Fundamentals / Programming with C	2:0:2	4

HARD CORE:

Sl. No.	Course	L:T:P	Credit Value
1	Computer Networks	2:1:1	4
2	Computer Graphics	3:0:1	4
3	Data Structures and Algorithms	3:0:1	4
4	Analysis and Design of Algorithms	2:1:1	4
5	DBMS	2:1:1	4
6	Software Engineering	3:1:0	4
7	Theory of Languages	3:0:1	4
8	Operating System and Unix	2:0:2	4
9	System Software	3:1:0	4

SOFT CORE:

Sl. No.	Course	L:T:P	Credit Value
1	Principles of Programming Language and C	2:1:1	4
2	Internet Technology	2:0:2	4
3	Java Programming	2:0:2	4
4	Multimedia	3:1:0	4
5	Microcontroller	3:1:0	4
6	Discrete Mathematics	3:1:0	4
7	Simulation and Modeling	3:1:0	4
8	Operations Research	3:1:0	4
9	Mobile Communication	3:1:0	4
10	C++	2:0:2	4
11	Pattern Recognition	3:0:1	4
12	Image Processing	2:1:1	4
13	Software Testing	3:0:1	4
14	Computational Techniques	3:0:1	4
15	Graph Theory	3:1:0	4
16	OOAD	2:1:1	4

17	Probability and Statistics	3:1:0	4
18	Data Mining	2:1:1	4
19	Artificial Intelligence	3:1:0	4
20	.NET Technologies	2:0:2	4
21	Object Oriented Modeling and Design with UML	2:1:1	4
22	Android Application Development	2:0:2	4
23	Advanced Database Management Systems	2:1:1	4
24	Compiler Design	3:0:1	4

SCHEME OF EXAMINATION AND ASSESMENT:

In view of the CBCS syllabus, Each Course is Assess with Components . Component 1 (C1), Component 2 (C2), and Component 3 (C3),

The following is the scheme which will be followed for the assessment of marks for both theory (HC/ SC/ OE) as well as practicals (HC/ SC) irrespective of the Credits associated with each Course. Thirty percent of the marks will be assessed for the internals (C1 and C2) and remaining seventy percent will be for the semester end examinations (C3). Each Course carries 100 marks and hence thirty marks for internal assessment and remaining seventy marks will be for Semester End Examinations. Out of thirty marks for internals, fifteen marks will be allotted to each C1 and C2 components.

Each theory Course (HC/ SC/ OE) consists of three components namely C1, C2 and C3. C1 and C2 are designated as Internal Assessment (IA) and C3 as Semester End Examination. Each Course (HC/ SC/ OE) carries 100 Marks and hence the allotment of marks to C1, C2 and C3 Components will be fifteen, fifteen and seventy marks respectively. i.e.,

C1 Component : 15 Marks	Internal Assessment Marks
C2 Component : 15 Marks	
C3 Component : 70 Marks	Semester End Examination
Total :	100 Marks

The above will be followed in common for all the theory (HC/ SC/ OE) Courses in all the four semesters.

DATA STRUCTURES & ALGORITHMS

3:0:1

Course Code:CSA100

Course Outcomes: At the end of the course students will be able to:

- CO1. Select appropriate data structures as applied to specified problem definition.
- CO2. Implement operations like searching, insertion, and deletion, traversing mechanism etc. on various data structures.
- CO3. Implement Linear and Non-Linear data structures.
- CO4. Implement appropriate sorting/searching technique for given problem.
- CO5. Design advance data structure using Non Linear data structure.

UNIT I: Introduction

Introduction to Stages in Problem Solving, Difference between Data Type, Data Structure and Storage Structure, Formal Definition of Data Structure, Classification of Data Structure
Analysis of Algorithms: Algorithm, Time Complexity and Space Complexity, O-Notation, Omega Notation and Theta Notation.
Primitive Data Structure: Integer, Real, Character and Boolean and Its Storage Representation

UNIT II: Non-Primitive Data Structures

Arrays: Introduction to Array Data Structure and Its Representation, Sparse Matrix Representation. Introduction to Structures
Stacks: Introduction, Implementation Using Arrays, Applications - Tower Of Hanoi, Expression Evaluation, Conversion of Expressions

UNIT III: Stacks and Queues

Queue: Introduction, Types – Ordinary, Circular, Doubly Ended, Priority, Implementation Using Arrays

Linked List: Introduction, Types, Operations, Implementation of Stacks and Queues Using Linked List

UNIT IV: Non-Linear Data Structures

Graphs: Introduction, Basic Terminologies, Graph Representation – Adjacency and Incidence Matrix Representation

Trees: Introduction, Binary Tree Representation – Array and Linked List Representations, Traversals – Preorder, In-order, Post order, Binary Search Tree, Introduction to B-Trees

Hash Tables: Direct Address Tables, Hash Tables, Hash Functions, Open and Closed Addressing

References:

1. Data Structures and Algorithms, 2nd Edition, 2006 , Andrew Tanenbaum
2. “An Introduction to Data Structures, with Applications” McGraw Hill, Kongakusha 1976, Trembley and Sorenson
3. “Data Structures” SBCS Publication, 1980, Horowitz and Sahni
4. Data Structures using c, A K Sharma
5. Data Structures and program design in C, Kruse Robert L
6. Data Structures and analysis in C, Mark Allen Weiss
7. Data Structures and Algorithms, Alfred V AHO and Jeffrey D Ullman

SYSTEM SOFTWARE

2:1:1

Course Code:CSA110

Course Outcomes:

At the end of the course students will be able to:

CO1. Understand fundamentals of language processing and grammar

CO2. Apply knowledge of compilation and code optimization steps to mimic a simple compiler

CO3. Demonstrate the working of various system software like assembler, loader, linker, editor and device driver

UNIT 1

Background: Machine Structure, Evolution of the components of a programming system, evolution of operating system, operating system user view point functions, facilities

General Machine Structure, General Approach to a new machine, Machine Structure – 360 and 370, Assembly Language

UNIT 2

Assemblers: General design procedure, design of an assembler.

Macro language and macro processor, macro instructions, features of macro, implementation

UNIT 3

Loaders, different types of loaders, loader schemes, design of an absolute loader, design of direct linking loader.

Compilers: Structure and phases

UNIT 4

Lex and yacc: The Simplest lex Program, Recognizing Words with Lex, Parser-Lexer

Communication, Regular Expressions, Grammars, Shift/Reduce Parsing, Structure of lex and yacc Programs, Programs in lex and yacc

References:

1. Systems Programming by Donovan
2. Principles of Compiler design by Ullman
3. System programming by Dhamdhare
4. Lex and yacc by John R Levine, Tony Mason and Doug Brown
5. System Software- Prof. Liland L Beck.

6. System Software- Prof. John R Levine

COMPUTER NETWORKS

2:1:1

Course code:CSA120

Course Outcomes:

At the end of the course students will be able to:

CO1. Master the terminology and concepts of the OSI reference model and the TCP-IP reference model.

CO2. Study the basic taxonomy and terminology of the computer networking and enumerate the layers of OSI model and TCP/IP model.

CO3. Master the concepts of protocols, network interfaces, and design/performance issues in local area networks and wide area networks

CO4. Acquire knowledge of Application layer and Presentation layer paradigms and protocols.

CO5. Study Session layer design issues, Transport layer services, and protocols.

CO6. Gain core knowledge of Network layer routing protocols and IP addressing.

CO7. Study data link layer concepts, design issues, and protocols.

CO8. Read the fundamentals and basics of Physical layer, and will apply them in real time applications.

CO9. Familiar with wireless networking concepts

CO10. Familiar with contemporary issues in networking technologies

CO11. Familiar with network tools and network programming

UNIT 1

USES OF COMPUTER NETWORK: Business Applications, Home Applications, Mobile Users, Social Issues

NETWORK HARDWARE: Local Area Networks, Metropolitan Area Networks, Wide Area Networks, Wireless Networks, Home Networks, Internetworks

NETWORK SOFTWARE: Protocol Hierarchies, Design Issues for the Layers, Connection-Oriented and Connectionless Service, Service Primitives, the Relationship of Services to Protocols

REFERENCE MODELS: The OSI Reference Model, The TCP/IP Reference Model, A comparison of OSI and TCP/IP Reference Model

UNIT 2

THE PHYSICAL LAYER

THE THEORETICAL BASIS FOR DATA COMMUNICATION: Fourier Analysis, Bandwidth-Limited Signals, The Maximum Data Rate of a Channel

GUIDED TRANSMISSION MEDIA: Magnetic Media, Twisted Pairs, Coaxial Cable, Fiber Optics

WIRELESS TRANSMISSION: The Electromagnetic Spectrum, Radio Transmission, Microwave Transmission, Infrared and Millimeter Waves, Light wave Transmission

COMMUNICATION SATELLITES: Geostationary Satellites, Medium-Earth Orbit Satellites, Low-Earth Orbit Satellites, Satellites versus Fiber

THE DATA LINK LAYER

DATA LINK LAYER DESIGN ISSUES: Services Provided to the Network Layer, Framing, Error Control, Flow Control

ERROR DETECTION AND CORRECTION: Error-Correcting Codes, Error-Detecting Codes

ELEMENTARY DATA LINK PROTOCOLS: A Utopian Simplex Protocol, A Simplex Stop-and-Wait Protocol

SLIDING WINDOW PROTOCOLS: A One-Bit Sliding Window Protocol, A Protocol Using Go-Back-N

UNIT 3

THE MEDIUM ACCESS CONTROL SUBLAYER

THE CHANNEL ALLOCATION PROBLEM: Static Channel Allocation, Dynamic Channel Allocation

MULTIPLE ACCESS PROTOCOLS: ALOHA, Carrier Sense Multiple Access Protocols, Collision-Free Protocols, Limited-Contention Protocols

ETHERNET: Ethernet Cabling, Manchester Encoding, The Ethernet MAC sub layer protocol, the binary exponential back off algorithm, Ethernet Performance, Switched Ethernet, IEEE 802.2: Logical Link Control

THE NETWORK LAYER

NETWORK LAYER DESIGN ISSUES: Store-and-Forward Packet Switching, Services Provided to the Transport Layer, Implementation of Connectionless Service, Implementation of Connection-Oriented Service, Comparison of Virtual-Circuit and Datagram Networks

ROUTING ALGORITHMS: The Optimality Principle, Shortest Path Algorithm, Flooding, Distance Vector Routing, Link State Routing, Hierarchical Routing

CONGESTION CONTROL ALGORITHMS: Approaches to Congestion Control, Traffic-Aware Routing, Admission Control, Traffic Throttling, Traffic Shaping

INTERNETWORKING: How Networks Differ, How Networks Can Be Connected, Tunneling, Internetwork Routing, Fragmentation

THE NETWORK LAYER IN THE INTERNET: The IP Protocol, IP Addresses, IPv6

UNIT 4

THE TRANSPORT LAYER

THE TRANSPORT SERVICE: Services Provided to the Upper Layers, Transport Service Primitives

ELEMENTS OF TRANSPORT PROTOCOLS: Addressing, Connection Establishment, Connection Release, Error Control and Flow Control, Multiplexing, Crash Recovery

THE INTERNET TRANSPORT PROTOCOLS: UDP: Introduction to UDP, Remote Procedure Call, Real-Time Transport Protocols

THE INTERNET TRANSPORT PROTOCOLS: TCP: Introduction to TCP, The TCP Service Model, The TCP Protocol, The TCP Segment Header, TCP Connection Establishment, TCP Connection Release, TCP Connection Management Modeling, TCP Sliding Window, TCP Timer Management

THE APPLICATION LAYER

DNS—THE DOMAIN NAME SYSTEM: The DNS Name Space, Resource Records, Name Servers

ELECTRONIC MAIL: Architecture and Services, The User Agent, Message Formats, Message Transfer, Final Delivery

NETWORK SECURITY:

CRYPTOGRAPHY: Introduction to Cryptography, Substitution Ciphers, Transposition Ciphers, Two Fundamental Cryptographic Principles

SYMMETRIC-KEY ALGORITHMS: DES—The Data Encryption Standard, Cipher Modes

PUBLIC-KEY ALGORITHMS: RSA, Other Public-Key Algorithms

DIGITAL SIGNATURES: Symmetric-Key Signatures, Public-Key Signatures

MANAGEMENT OF PUBLIC KEYS: Certificates

Reference Books:

1. Computer Networks, 5th Edition, Prentice Hall, 2006, Andrew S. Tanenbaum & David J. Wetherall
2. Data & Computer Communications, 6th Edition, Pearson Education, 2002, William Stallings
3. Computer Networks: 3rd Edition, Elsevier, 2003, Larry L. Peterson & Bruce S. Davie
4. Data Communication & Networking, 4th Edition, Mc Graw Hill, 2006, Behrouza Forouzan
5. Computer & Networks with Internet Applications, 4th Edition, Pearson Education, 2004, Douglas E. Comer

ANALYSIS AND DESIGN OF ALGORITHMS

2:1:1

Course Code:CSB060

Course Outcomes:

At the end of the course students will be able to:

- CO1. Analyze different scenarios for running time of algorithms using asymptotic notations and Design using Recursion.
- CO2. Apply divide and conquer strategy for design of various algorithms.
- CO3. Develop algorithms for well known problems using greedy methods.
- CO4. Describe and apply dynamic-programming approach for designing graph and matrix based algorithms.
- CO5. Understand the concept of backtracking for traversal and search algorithms.
- CO6. Apply the knowledge earned to determine the efficiency of algorithms considering time and space tradeoffs.

UNIT 1

INTRODUCTION: Algorithm specification, pseudo code conventions

PERFORMANCE ANALYSIS: Space Complexity, Time Complexity, Asymptotic Notation, Mathematical Analysis: Recursive and Non recursive algorithms

BRUTE FORCE – Bubble Sort, Selection Sort, Sequential Search, String Matching

UNIT 2

DIVIDE- AND – CONQUER: General Method, Binary Search, Finding the Maximum and Minimum, Merge Sort, Quick Sort, Strassen's Matrix Multiplication

THE GREEDY METHOD: The General Method, Knapsack Problem, Tree Vertex Splitting, Job Sequencing with Deadlines, Minimum-Cost Spanning Trees - Prim's Algorithm, Kruskal's Algorithm, Optimal Storage on Tapes, Optimal Merge Patterns, Single-Source Shortest Paths.

UNIT 3

DYNAMIC PROGRAMMING: The General Method, Binomial Coefficient, Multistage Graphs, All Pairs Shortest Paths Single-Source Shortest Paths: General Weights, String Editing, 0/1 Knapsack, the Traveling Salesperson Problem

BACKTRACKING: The General Method, the 8-Queens Problem, Sum of Subsets, Graph Coloring, Hamiltonian Cycles

UNIT 4

Elementary Graph Problems: Depth First Search, Breadth First Search, Topological Sort

NP-Hard and NP-Complete Problems: Basic Concepts, Nondeterministic Algorithms, The Classes NP-Hard And NP-Complete

NP-Hard Graph Problems: Clique Decision Problem (CDP), Node Cover Decision Problem, Chromatic Number Decision Problem (CNDP), Traveling Salesperson Decision Problem (TSP), AND/OR Graph Decision Problem (AOG)

References:

1. Analysis and Design of Algorithms: Horowitz Sahani
2. Analysis and Design of algorithms: Trembly
3. Introduction to Algorithms: Thomas H. Cormen
4. Analysis and Design of Algorithm: Padma Reddy
4. Introduction to the design and analysis of algorithm: Anany Levitin.
5. Design and analysis of algorithm: S Shridhar.

OPERATING SYSTEM and UNIX

2:0:2

Course Code:CSB070

Course Outcomes:

At the end of the course student will be able to:

- CO1. Understand device drivers
- CO2. Write applications with improved performance and stability
- CO3. Write set of small commands and utilities that do specific tasks well
- CO4. Run multiple programs each at the same time without interfering with each other or crashing the system.
- CO5. Implement Commands of UNIX.
- CO6. Implement various file processing commands and shell Programming.

UNIT 1

Introduction to Operating System, Operating System Structures

Process Management: Processes, CPU Scheduling

UNIT 2

Deadlocks, Storage Management: Memory management, Virtual Memory, File-System Interface

UNIT 3

Introduction: Why UNIX? The Unix Environment, UNIX Structure, accessing UNIX, UNIX commands

File Systems: File Names, File Types, Regular Files, Directories, File System Implementation, Operations unique to directories, Operations unique to regular files, Operations common to both.

Vi editor, local commands, range commands in vi, global commands in vi, rearrange text in vi, ex editor.

UNIT 4

Introduction to shells: Unix Session, Standard Streams, Redirection, pipes, tee command, command execution, command line editing, quotes, command substitution, job control, aliases, variables, predefined variables, options, shell/environment customization.

Security and file permission: User and groups, Security levels, changing permissions, User masks, Changing ownership and group, Regular expressions: Atoms, Operators, grep: operation, grep family, Searching for file content, sed and awk.

C Shell Programming: Basic script concepts, Expressions, Decision making selections,, Built in commands, Scripting techniques, Shell environment & Script, Script examples

References:

1. Unix and Shell Programming, Behrouz A Forouzan and Richard F.Gilberg, 2nd Edition, 2003, Thomson.
2. System Programming and Operating Systems, Dhamdhare. D.M., 4th Edition, TataMcGraw Hill, 2006
3. A Practical Guide to Linux, Mark G. Sobell, 1st Edition, 2002, Pearson Education (Chapters:1 to 5, 8, 10, 11, 15)
4. UNIX: The Complete Reference, Kenneth Rosen and others, 2nd Edition, 2002,Obsborne/McGraw Hill
5. Design of the UNIX Operating System, Maurice J Bach.
6. Operating System: A Modern Perspective , Gary J Nutt.

COMPUTER GRAPHICS

3:0:1

Course code:CSB080

Course Outcomes:

At the end of the course students will be able to:

CO1. Utilize the components of a graphics system and become familiar with building approach of graphics system components and algorithms related with them.

CO 2. Learn the basic principles of 3- dimensional computer graphics.

CO3. Provide an understanding of how to scan convert the basic geometrical primitives, how to transform the shapes to fit them as per the picture definition.

CO4. Provide an understanding of mapping from a world coordinates to device coordinates, clipping, and projections.

CO 5. Implement the applications of computer graphics concepts in the development of computer games, information visualization, and business applications.

CO6. Comprehend and analyze the fundamentals of animation, virtual reality, underlying technologies, principles

UNIT 1

Introduction, Video Display Devices, Refresh Cathode-Ray Tubes, Raster-Scan Display, Random-Scan Displays, Color CRT Monitors, Flat-Panel Displays, Raster Scan Systems, Input devices.

Output primitives: Points & Lines, Line Drawing Algorithms, Loading the Frame Buffer, Circle Generating Algorithms, Pixel Addressing and Object Geometry

UNIT 2

Two dimensional transformations Basic & other transformations, Matrix representations, Homogeneous coordinates Composite transformations, General-pivot-point transformations. Three Dimensional Transformations: Introduction to 3D Translation, Rotation & Scaling, Other Transformations, Modeling and Co-ordinate Transformations.

UNIT 3

Three Dimensional Viewing :Viewing Pipeling, window to viewport transformations, Projections, Types of projections.

Graphical User Interface & Interactive Input Methods : The User Dialogue, Windows & Icons, feedback, Input of Graphical Data, Interactive Picture Construction Techniques, Basic Positioning Methods, Constraints, Grids, Gravity Field, Rubber-Band Methods, Dragging, Painting & Drawing

UNIT 4

Curves & Surfaces: Properties, Bezier curves properties, Design techniques, Bezier surfaces, Displaying curves & surfaces

Hidden line removal algorithms

Introduction to fractals, Serpinsky's triangle, Construction, Koch curves.

Windowing & Clipping: Clipping operations, Line clipping algorithms, point clipping, text clipping, polygon clipping algorithms, Exterior clipping

Reference:

1. "Computer Graphics", Pearson Education, Donald D. Hern and M. Pauline Baker
2. "Principles of Interactive Computer Graphics" McGraw Hill 1989, W. M. Newman and Robert Sproull
3. "Computer Graphics a Programming Approach" McGraw Hill 1987, Steven Harrington
4. "Schaums outline of theory and problems of Computer Graphics" 2nd printing 1987, 1986 Edition, Roy A Plastock and Gardon Kelley
5. "Procedural Elements of Computer Graphics" McGraw Hill 2nd edition 1990, David F Frogers and J Alan Adams
6. Computer Graphics, James.D.Foley, A Vandam etal

SOFTWARE ENGINEERING

3:1:0

Course Code:CSC040

Course Outcomes:

At the end of the course students will be able to:

CO1. Understand the nature of software development and software life cycle process models, agile software development, SCRUM and other agile practices.

CO2 Learn methods of capturing, specifying, visualizing and analyzing software requirements.

CO3. Understand concepts and principles of software design and user-centric approach and principles of effective user interfaces.

CO 4. Basics of testing and understanding concept of software quality assurance and software configuration management process.

CO 5. Understand need of project management and project management life cycle.

CO 6. Understand project scheduling concept and risk management associated to various types of projects.

UNIT 1

INTRODUCTION TO SOFTWARE ENGINEERING: Software and Software Engineering, phases in Software Development, Software Development Process models, Agile modeling, Introduction to metrics.

UNIT 2

SOFTWARE REQUIREMENT SPECIFICATION: Role of SRS, Problem Analysis, Requirement specification, validation, metrics, Monitoring and control

PLANNING A SOFTWARE PROJECT: Cost Estimation, Project Scheduling, Staffing personal planning, Team Structure, Software Configuration Management, Quality Assurance Plans, Project Monitoring plans, Risk Management.

UNIT 3

SYSTEM DESIGN: Design Objectives, design principles, Module level Concepts, Design methodology – object oriented approach Design Specification, Verification, Metrics, monitoring and control.

DETAILED DESIGN: Module Specification, Detailed design and process design Language, Verification

UNIT 4

CODING: Programming practice, Verification, Metrics

TESTING: Testing Fundamentals, Fundamental testing, Structural Testing. Testing process
Clean Room approach

References:

1. “An Integrated approach to the Software Engineering” 2ed. Narosa Publishing House, New Delhi, 2002, Pankaj Jalote
2. Software Engineering Principles & Practice - 3rd Edition, Tata Mc Graw Hill Companies – 2006, Waman S Jawadekar
3. Software Engineering A Practitioner’s Approach - 6th Edition McGraw Hill – 2005, Roger S Pressman
4. Software Engineering - 7th Edition : Pearson Education Ltd- 2006, Sommerville

THEORY OF LANGUAGES

3:1:0

Course Code:CSC050

Course Outcomes:

At the end of the course students will be able to:

CO1. Design different types of Finite Automata and Machines as Acceptor, Verifier and Translator.

CO2. Understand, design, analyze and interpret Context Free languages, Expression and Grammars.

CO3. Design different types of Push down Automata as Simple Parser.

CO4. Design different types of Turing Machines as Acceptor, Verifier, Translator and Basic computing machine.

UNIT - 1

INTRODUCTION TO FINITE AUTOMATA: Introduction to Finite Automata; The central concepts of Automata theory; Deterministic finite automata; Nondeterministic finite automata.

FINITE AUTOMATA, REGULAR EXPRESSIONS: An application of finite automata; Finite automata with Epsilon-transitions; Regular expressions; Finite Automata and Regular Expressions

UNIT - 2

REGULAR LANGUAGES, PROPERTIES OF REGULAR LANGUAGES: Regular languages; Proving languages not to be regular languages; Closure properties of regular languages; Equivalence and minimization of automata.

CONTEXT-FREE GRAMMARS AND LANGUAGES: Context –free grammars; Parse trees; Ambiguity in grammars and Languages.

UNIT – 3

PUSHDOWN AUTOMATA: Definition of the Pushdown automata; the languages of a PDA; Equivalence of PDA's and CFG's; Deterministic Pushdown Automata.

PROPERTIES OF CONTEXT-FREE LANGUAGES: Normal forms for CFGs; The pumping lemma for CFGs; Closure properties of CFL

UNIT - 4

TURING MACHINE: The turing machine; Extensions to the basic Turing Machines;

UNDECIDABILITY: A Language that is not recursively enumerable; An Undecidable problem that is RE; Post's Correspondence problem

REFERENCES:

1. Introduction to Automata Theory, Languages and Computation – John E.. Hopcroft, Rajeev Motwani, Jeffrey D.Ullman:, 3rd Edition, Pearson education, 2007.
2. Fundamentals of the Theory of Computation: Principles and Practice – Raymond Greenlaw, H.James Hoove, Morgan Kaufmann, 1998.
3. Introduction to Languages and Automata Theory – John C Martin, 3rd Edition, Tata McGraw-Hill, 2007.

DATABASE MANAGEMENT SYSTEM

2:1:1

Course Code:CSC060

Course Outcomes:

At the end of the course students will be able to:

- CO1. Explain the features of database management systems and Relational database.
- CO2. Design conceptual models of a database using ER modelling for real life applications and also construct queries in Relational Algebra.
- CO3. Create and populate a RDBMS for a real life application, with constraints and keys, using SQL.
- CO4. Retrieve any type of information from a data base by formulating complex queries in SQL.
- CO5. Analyze the existing design of a database schema and apply concepts of normalization to design an optimal database.
- CO6. Build indexing mechanisms for efficient retrieval of information from a database

UNIT 1

Introduction and conceptual modeling databases and Database users, Data modeling using the entity relationship (ER) model, the enhanced entity – relationship (EER) model.

UNIT 2

Relational model: Concepts constraints, Languages, Design and programming.

The relational data model and relational database constraints, Relational algebra and relational calculus, Introduction to SQL Programming technique

UNIT 3

Database design theory and methodology functional dependencies and Normalization for relational database, Relational database design algorithms and further dependencies, practical database design methodology and use of UML diagrams.

UNIT 4

Introduction to transaction processing concepts and theory recovery

REFERENCES

1. Fundamentals of database system – 5th Edition – Ramez elmasri, Navathe – Person edition
- 2 .An introduction to database system – 8th Edition – C. J. Date, Kannan – Person Education
- 3.Database system concepts – 5th Edition – Korth, Sudarshan – McGraw Bill Edition

4. Database Management System- Raghuramakrishnan.
5. An Introduction to Database System- Bipin Desai
6. Principles of Database System- J D Ullman

Softcore:

PRINCIPLES OF PROGRAMMING LANGUAGES AND 'C'

2:1:1

Course Code:CSA020

Course Outcomes:

At the end of the course students will be able to:

- CO1. Analyzing semantic issues associated with function implementations, including variable binding, scoping rules, parameter passing, and exception handling.
- CO2. Implementation techniques for interpreted functional languages.
- CO3. Using object-oriented languages.
- CO4. Familiar with design issues of object-oriented and functional languages.
- CO5. Familiar with language abstraction constructs of classes, interfaces, packages, and procedures.
- CO6. Familiar with implementation of object-oriented languages.
- CO7. Familiar with using functional languages

UNIT 1: Preliminaries

Reasons for studying concepts of programming languages, Programming domains, Language evaluation criteria, Implementation methods Names, Bindings, Type Checking, and Scopes Introduction, Names, Variables, The concept of Binding, Type Checking, String Checking, Type Compatibility, Scope, Scope and Lifetime, Referencing Environments, Named Constants, Variable Initialization.

UNIT 2: Data Types

Introduction, Primitive Data types, Character String Types, User-Defined Ordinal Types, Array Types and Associative Arrays, Record Types, Union Types, Set Types, Pointes Types.

Expression, Assignment Statements and Statement Level Control Structures

Introduction, Arithmetic Expressions, Overloaded Operators, Type Conversions, Relational And Boolean Expressions, Short-Circuit Evaluation, Assignment Statements, Mixed-Mode assignment.

UNIT 3

Compound Statements, Selection Statements, Iterative Statements, Unconditional Branching, Guarded Commands, Conclusion.

Subprograms: Introduction, fundamentals, design issues, local referencing environments, parameter passing methods, overloaded programs, generic subprograms, coroutines, user defined overloaded operators

UNIT 4

C LANGUAGE: C Fundamentals, Operation data input and output, Control statements, Function Storage classes, Arrays, Pointers, structures and unions, Enumeration, Command line parameters, Macros, 'C' processor.

Reference:

1. Concepts of Programming Languages, Eight Edition, Pearson, Robert W. Sebesta
2. Foundation for Programming Languages, John C Mitchell
3. Principles of Programming Language, Chopra Rajiv
4. Principles of Programming Language, Dowek
5. Types and Programming Languages, Benjamin C Pierce
6. Programming Languages: Principle and Practices ,3rd Edition, Kenneth C Louden

INTERNET TECHNOLOGY

2:0:2

Course Code:CSD220

Course Outcomes:

At the end of the course students will be able to:

- CO1. Develop analytical ability in network technology.
- CO2. Create quality websites
- CO3. Work individually as a web designer and set up their own business
- CO4. Get the job opportunities in most companies for professional web designers and build websites more visually elegant and interactive
- CO5. Implement interactive web page(s) using HTML, CSS and JavaScript.
- CO6. Design a responsive web site using HTML5 and CSS3.

UNIT 1

Fundamentals: introduction to the Internet, WWW, Web Browsers, Web Servers, URL, Multipurpose Internet Mail Extensions (MIME), HTTP, Security, Introduction to HTML: Origins and Evolution, Basic Syntax, Document Structure, Basic tags, Images, Links, Lists, Tables, Forms, Frames.

UNIT 2

Introduction to XML: Syntax of XML, XML Document Structure, Document Type Definition.

Introduction to XHTML: Origins and Evolution, Basic Syntax, Document Structure, Basic tags, Images, Links, Lists, Tables, Forms, Frames, Syntactic difference between HTML and XHTML.

Cascading Style Sheets (CSS): Introduction, levels of style sheets, Selector Forms, Property value forms, Font properties, Color, Alignment of Text, Box model, Background Images, and <div> tags.

UNIT 3

The basics of JavaScript: Overview, Object Orientation and JavaScript, General syntactic characteristics, Primitives, Operations and Expressions, Screen Output and Keyboard Input, Control Statements, Objects, Arrays, Functions, Constructors, Errors.

JavaScript and XHTML Documents: Element access, Events and Event Handling, Handling Events from Body elements, Handling Events from Button elements, Handling Events from Text Box and password elements.

Dynamic Documents with JavaScript: Introduction, Positioning Elements, Moving Elements, Element Visibility, Changing Colors and Fonts, Dynamic Content, Stacking Elements, Locating Mouse Cursor, Reacting to Mouse Click, Slow Movement of Elements, Dragging and Dropping Elements.

UNIT 4

Introduction to PHP: Origins and Uses, Overview, General Syntactic Characteristics, Primitive, Operations and Expressions, Output, Control Statements, Arrays, Functions, Form Handling, Cookies.

REFERENCES:

1. Programming the World Wide Web – by Robert W. Sabesta 4th Edition Pearson Publications
2. HTML and XHTML the Complete Reference.
3. How to program the World Wide Web – by Deitel and Deitel
4. Mastering in HTML – by Ray and Ray.
5. Web programming and Internet Technologies: An E Commerce approach- By Porter ` Seobey and Pawan Lingras.
6. Internet Technology and Information services by Joseph Miller

JAVA Programming

2:0:2

Course Code:CSA270

Course Outcomes:

At the end of the course students will be able to:

- C01. Understand concept of Object Oriented Programming & Java Programming Constructs.
- C02. Understand basic concepts of Java such as operators, classes, objects, inheritance, packages ,Enumeration and various keywords.
- C03. Understand the concept of exception handling and Input/Output operations.
- C04. Design the applications of Java & Java applet.
- C05. Analyze & Design the concept of Event Handling and Abstract Window Toolkit.

UNIT 1

History and evolution of Java, An overview of Java, Data types, variables and arrays, Operators, Control statements- Introducing classes ,A closer look at methods and classes, Inheritance, Packages and interfaces.

UNIT 2

Exception handling, Multithreaded Programming, Enumeration, Autotoxins, I/O, Applets

UNIT 3

Networking, Event handling, Swings.

UNIT 4

String handling, Collection framework, Introduction to J2EE, Java servlet, Java server pages (JSP) and HTML, JDBC objects.

REFERENCES:

1. The complete reference Java – 7th Edition – Herbert Schildt – Tata Mcgraw hill Edition.
2. The complete reference J2EE – Jem Keogh – Tata Mcgraw hill Edition.
3. Object Oriented Programming with Java- M T Somashekara, D S Guru and K S Manjunatha.
4. The Complete Reference 7th Edition Herbert Schiidt
5. Introduction to Java Programming – E Balaguruswamy
6. Head First Java – 2nd Edition
7. Core Java- Horst Mann, C S –8th Edition-Cornell.

8. Core Servlet and Java Server pages- Hall, M-Brown L

MULTIMEDIA

3:1:0

Course Code:CSD210

Course Outcomes:

At the end of the course students will be able to:

- CO1. Understand various file formats for audio, video and text media.
- CO2. Develop various Multimedia Systems applicable in real time.
- CO3. Design interactive multimedia software.
- CO4. Apply various networking protocols for multimedia applications.
- CO5. Develop understanding of technical aspect of Multimedia Systems

UNIT 1

Introduction to Multimedia (MM) Communication, Scope, Range, Feasibility and Challenges of MM Communication Key aspects of MM: Compression, Coding, Transmission and Replay.

UNIT 2

Types of Compression: Quantization, Coding as PCM, DPCM, ADPCM. Simple Encoder and Decoders based on PCM Samples. Introduction to Transform domain Compression. Introduction to Audio part of MPEG, Psychoacoustics

UNIT 3

Compression in Spatial Domain Algorithms for Data Compression in Transform Domain: DCT. Variable Length Coding, Huffman code. Variable Length Coding: Arithmetic Coding. Introduction to JPEG 2000 Standard, Encoders-Decoders based on this. Audio Compression and MPEG Audio

UNIT 4

Fundamental concepts of Video. MPEG Architecture Details: Audio-Video- Systems. Video Coding standard related to H.263 and H. 264. MPEG- 1, 2 Video. MPEG- 4 : Video. Streaming and Transport Issues: Multiplexing, Synchronization and File formats. Errors in MPEG and Error handling, Concealment. Buffer structures and Buffer Management

Introduction to MPEG-7 and MPEG-21., HDTV. Content based Image Retrievals and Digital Libraries.

References:

1. Fundamentals of Multimedia, 2nd Ed, Pearson, 2005, Ze-Nian Li and Mark Drew
2. Multimedia Communications., Pearson, 2005, Fred Halsall
3. Introduction to Data Compression, 3rd Ed, Morgan Kaufman (India Ed), 2005, Khalid Sayood
4. The DATA compression; The Complete Reference, 3rd Ed, Springer (India Ed), 2006, David Solomon
5. Multimedia foundations: A Core Concepts of for Digital Design, Vic Cost Ello
6. Multimedia: Making it work, 9th edition, Tay Vaughan

MICROCONTROLLERS

3:1:0

Course Code:CSC210

Course Outcomes:

At the end of the course students will be able to:

- CO1. Describe the architecture of 8051 microcontroller and write embedded program for 8051 microcontroller.
- CO2. Design the interfacing for 8051 microcontroller.
- CO3. Understand the concepts of ARM architecture.
- CO4. Demonstrate the open source RTOS and solve the design issues for the same.
- CO5. Select elements for an embedded systems tool.
- CO6. Understand the concept and architecture of embedded systems

UNIT 1

Introduction, Numbering system and binary arithmetic,

UNIT 2

The 8051 Architecture, Basic assembly language programming concepts, Moving data,

UNIT 3

Logical operations, Arithmetic operations Jump and call instructions

UNIT 4

An 8051 Microcontroller design, Applications

References

1. “The 8051 Microcontroller”, 3rd Edition, Thomson India edition, 2007, Kenneth Ayala
2. “Programming and customizing the 8051 microcontroller”, Tata McGraw-Hill edition, 2006, Myke prick
3. “The 8051 Microcontroller and embedded systems”, Pearson India, 2006, Muhammad Ali Mazidi & Janice Gillispie Mazidi,
4. Microcontroller and Embedded System, Mazidi, M A- Mazidi
5. Microcontroller: Architecture, Programming and application, Ayala, Kenneth

DISCRETE MATHEMATICS

3:1:0

Course Code:CSA260

Course Outcomes:

At the end of the course students will be able to:

- CO1. Construct simple mathematical proofs and possess the ability to verify them.
- CO2. Have substantial experience to comprehend formal logical arguments .
- CO3. skillfull in expressing mathematical properties formally via the formal language of propositional logic and predicate logic.
- CO4. Specify and manipulate basic mathematical objects such as sets, functions, and relations and will also be able to verify simple mathematical properties that these objects possess.
- CO5. Apply basic counting techniques to solve combinatorial problems .
- CO6. Use various techniques of mathematical induction (weak, strong and structural induction) to prove simple mathematical properties of a variety of discrete structures

UNIT 1

Set Theory: Sets and Subsets. Operations on sets, Countable and uncountable sets, The addition principal, the concept of probability.

Mathematical Logic: Propositions, Logical Connectives, Tautologies; Contradictions, Logical equivalence, Application to switching networks, Duality, Commentates NAND and NOR, Converse, Inverse and Contrapositive, Rules of inference.

UNIT 2

Open statements; Quantifiers, Logical Implication involving Quantifiers, Statements with more than one variable, Methods of proof and disproof, Mathematical Induction.

UNIT 3

Relations and Ordering: Cartesian products of sets, Relations, Paths in relations and digraphs, Operations on relations, Composition of relations, Properties of relations, Equivalence relations, Partial orders, Total Orders, External elements in posets, Lattices.

Functions: Functions, Types of functions, Composition of function, Invertible functions, Permutation Function.

UNIT 4

Fundamental principles of counting: Principles of inclusion and exclusion: The rule of sum and product, Permutations, combinations: The binomial theorem, combinations with repetition,

Ramsey number, the Catalan numbers, sterling number and bell numbers, Generalizations of principles, the pigeonhole principle, Derangements-Nothing is in its Right place, Rook polynomials, Arrangements with Forbidden positions.

References:

1. Discrete Mathematics by Dr. Chandrashekar S .
2. Discrete and combinational Mathematics by Ralph P. Grimaldi, 5th edition, Addison Wesley, 2004
3. Discrete mathematical structures by Kolman, Robert C Busby and Sharon., 6th Edition, Prentice Hall, 2008
4. Discrete Mathematics and Application by Kenneth H Rosen.
5. Discrete Mathematics by Norman L Biggs.

SIMULATION & MODELLING

3:1:0

Course Code:CS/A/B/C/D300

Course Outcomes:

At the end of the course students will be able to:

- CO1. Understand the definition of simulation and how to develop and analyze a simulation model
- CO2. Understand the fundamental logic, structure, components and management of simulation modeling
- CO3. Demonstrate knowledge of how to use Arena
- CO4. Build a simulation model with basic operations and inputs
- CO5. Build a simulation model with detailed operations
- CO6. Perform statistical analysis of output from terminating simulation

UNIT 1

Introduction, Simulation of prepursuit problem, A system & its model, Simulation of an inventory problem, The basic nature of simulation

Simulation of continuous systems: A chemical reactor, Numerical integration vs continuous system simulation, Selection of an integration formula, Runge Kutta integration formulas, simulation of a servo system, Simulation of a water reservoir system, Analog vs digital simulation

UNIT 2

Discrete system simulation

Fixed time-step vs event to event model, On simulating randomness, Generation of random numbers, Generation of non uniformly distributed random numbers, Monte Carlo computation vs stochastic simulation

UNIT 3

Simulation of queuing systems

Rudiments of queuing theory, simulation of single server queue, Simulation of two server queue, Simulation more general queues.

Simulation of PERT network

UNIT 4

Network model of a project, Analysis of an activity network, Critical path computation, Uncertainties in activity duration, Simulation of an activity network, Computer program for simulation, Resource allocation and cost considerations, Inventory control & forecasting Elements of inventory theory, More complex inventory models, simulation examples, Generation of Poisson & erlang variates, Forecasting & regression analysis
Design and Evaluation of Simulation Experiments
Length of simulation runs, Variance reduction techniques, Experimental layout, validation

References:

1. System Simulation with Digital Computer Narsingh Deo
2. System Simulation and Modeling - Sengutta
3. Computer Methods for solving Dynamic Separation problems- C D Holland and A I Liapis.
4. Fundamentals of Modeling separation Process- C D Holand.
5. Process Modeling-M M Denn

OPERATIONS RESEARCH

3:1:0

Course Code: CS A/B/C/D 310

Course Outcomes:

At the end of the course students will be able to:

CO1: Formulate and solve mathematical model (linear programming problem) for a physical situations like production, distribution of goods and economics.

CO2: Apply the concept of simplex method and its extensions to dual simplex algorithm.

CO3: Solve the problem of transporting the products from origins to destinations with least transportation cost.

CO4: Convert and solve the practical situations into non-linear programming problem.

CO5: Identify the resources required for a project and generate a plan and work schedule

UNIT 1

Introduction: formulation of LP problems, graphical solution of LP problems, General formulation of L P problems, Slack & Surplus variables, Standard form, Matrix form, Simplex method, Revised Simplex method, Dual simplex

UNIT 2

Assignment model, Transportation model, Game theory

Probability: Introduction, Basic terms of probability, The Addition law of probability, discrete & continuous, variables, random variables, probability distribution of random variables, Mean variance& standard deviation, Mathematical expectation of a random variable.

UNIT 3

Queuing theory

Introduction, queuing system, distribution, Kendall's Notation, Classification, model I (m/m/1).

UNIT 4

Project management by PERT CPM

Introduction, history, Applications, Basic steps, Network diagram representation, rules of drawing network diagram, labeling Fulkerson's I-J rule, Time estimates & Critical path, PERT, Resource allocation, Uses of PERT/CPM.

References:

1. Operations Research - S D Sharma
2. Operations Research - R K Gupta & D S Hira
3. Introduction to Operation research – Frederick S Hillier ,Gerald J and Liberman.
4. Operation research: An Introduction by Hamdy A Taha.
5. Operation research: Application and algorithm by Wayne L Winston.

MOBILE COMMUNICATION

3:1:0

Course Code:CS A/B/C/D 320

Course Outcomes:

At the end of the course students will be able to:

- CO1. Design a mobile cellular network
- CO2. Optimize a radio channel system
- CO3. Select the apt diversity scheme for a given wireless system to improve the performance.
- CO4. Perform efficient spectral allocation using multiple access techniques such as CDMA, and OFDM.
- CO5. Select the correct MAC protocol and routing algorithm for mobile ad-hoc networks.
- CO6. Optimize the mobile ad-hoc network, MAC protocols and routing algorithms as per application.

UNIT 1

Introduction, Applications, History of wireless communication, reference model, Wireless transmission, Frequencies for radio transmission, signals, Antennas, Signal propagation Multiplexing, Modulation, Spread spectrum

UNIT 2

Cellular Systems, Medium Access control, Motivation for a specialized MAC, SDMA, FDMA, TDMA, CDMA, Comparison.

UNIT 3

Telecommunications Systems, GSM, DECT, TETRA, UMS & IMT 2000
Satellite Systems, history, Applications, Basics, Classical TCP improvements, TCP over 2.5/3G wireless networks, performance enhancing proxies

UNIT 4

Support for mobility, File Systems World Wide Web, Wireless Application protocol

References:

1. Cellular and Mobile Communication- Krishna.
2. Cellular Mobile Communication – V S Bhagat
3. Cellular and Mobile Communication- V Jeyasri Arokiamary.
4. Wireless Communication and Networks – William Stallings.

5. Cellular Communication; A Comprehensive and practical guide- Nishith Tripathi and Jeffreyreed.

C++

2:0:2

Course Code:CSA250

Course Outcomes:

At the end of the course students will be able to:

CO1 . Understand the features of C++

CO2. Understand the relative merits of C++ as an object oriented programming language

CO3. Understand how to produce object-oriented software using C++

CO4. Understand how to apply the major object-oriented concepts to implement object oriented programs in C++, encapsulation, inheritance and polymorphism

CO5 Understand advanced features of C++ specifically stream I/O, templates and operator overloading

Unit I

Quick overview of C : Expressions - Statements - Arrays and Null-Terminated Strings – Pointers - Functions – Structures, Unions, Enumerations and User-Defined Types – C Style Console I/O – File I/O -.

Unit II

An Overview of C++ - Classes and Objects – Arrays, Pointers, References, and the Dynamic Allocation Operators

Unit III

Function Overriding, Copy Constructors and Default Arguments – Operator Overloading - Inheritance – Virtual Functions and Polymorphism

Unit IV

Templates – Exception Handling - The C++ I/O System Basics

References :

JSSCACS

1. The Complete Reference C++, 4th Edition, Tata-McGraw-Hill, 2003, Herbert Schildt
2. Object Oriented Programming with C++ , M T Somashekara, D S Guru, H S Nagendraswamy and K S Manjunatha
3. C++ Premier, 5th Edition. Stanley B Lippman
4. C++ Programming language, E Balaguruswamy
5. The C++ programming language, 4th Edition, Bjarne Stroustrup

PATTERN RECOGNITION

3:0:1

Course Code: CAC230

Course Outcomes:

At the end of the course students will be able to:

- C01. Explain and compare a variety of pattern classification, structural pattern recognition, and pattern classifier combination techniques.
- C02. Summarize, analyze, and relate research in the pattern recognition area verbally and in writing.
- C03. Apply performance evaluation methods for pattern recognition, and critique comparisons of techniques made in the research literature.
- C04. Apply pattern recognition techniques to real-world problems such as document analysis and recognition.
- C05. Implement simple pattern classifiers, classifier combinations, and structural pattern recognizers.

UNIT 1

Machine perception, pattern recognition systems, Design Cycle, Learning and adaption, models of Pattern recognition

UNIT 2

Bayesian Decision Theory

Introduction, Bayesian, Decision theory- Two category classification, classifiers-Two category case and multi category case, missing and noisy features.

UNIT 3

Nonparametric Techniques

Introduction, Density estimation, Parzen window, KN neighbor estimation, The nearest neighbor rule, Metrics and Nearest Neighbor Classification, Error analysis of nearest decision rule

UNIT 4

Introduction, Heirarchical clustering techniques, partitional clustering techniques Dimensionality reduction techniques Introduction, principle component analysis, Fisher Linear Dicriminant Analysis, Spectral clustering based dimensionality reduction

References:

1. Pattern Classification, 2nd edition, Wiley publications, R. O Duda, P.E. Hart and D G Stork,
2. Pattern Recognition and Image Analysis, Prentice Hall of India, Pvt Ltd, Earl Gose, Richard, Johnsonbaugh, Steve Jost
3. Pattern Recognition and machine Learning, Cristopher M Bishop,
4. Pattern Recognition (Blue Ant) – Willian Gibsom.
5. Pattern Recognition, 4th Edition- Sergios Theodoridis and Konstantios Koutroumbas

IMAGE PROCESSING

2:1:1

Course Code:CS A/B/C/D 330

Course Outcomes:

At the end of the course students will be able to:

- C01. Analyze general terminology of digital image processing.
- C02. Examine various types of images, intensity transformations and spatial filtering.
- C03. Develop Fourier transform for image processing in frequency domain.
- C04. Evaluate the methodologies for image segmentation, restoration etc.
- C05. Implement image process and analysis algorithms.
- C06. Apply image processing algorithms in practical applications.

UNIT I

Introduction, digital image fundamentals

UNIT II

Image enhancement in the spatial domain, Image enhancement in the frequency domain

UNIT III

Image restoration, color image processing

UNIT IV

Wavelets and multi-resolution processing image compression

References:

1. Digital Image Processing-Rafel C.Gonzalez and Richard E Words.
2. The Image Processing hand Book- John C Cruss.
3. Fundamentals of Digital Image Processing- Anil K Jain.
4. Digital Image Processing –Jayaraman S.
5. Digital Image Processing- Sanjay M Shah Munesh Chandra Trivedi

SOFTWARE TESTING

3:0:1

Course Code:CSC440

Course Outcomes:

At the end of the course students will be able to:

CO1.Check Various test processes and continuous quality improvement

CO2.Verify Types of errors and fault models

CO3.Check Methods of test generation from requirements

CO4.Check Behavior modeling using UML: Finite state machines (FSM)

CO5.Test generation from FSM models

CO6.Input space modeling using combinatorial designs

UNIT 1

Assessing Testing Capabilities and Competencies, Building a software Testing Environment: Building a software Testing Strategy, Establishing a Software Testing Methodology, Determining your Software Testing Techniques, Selecting and Installing Software Testing Tools.

UNIT 2

The Eleven-Step Testing Process: Eleven-Step Testing Process Overview, Step1: Access Project Management Development Estimate and status, Step2: Develop Test Plan, Step3: Requirement Phase Testing,

UNIT 3

Step4: Design Phase Testing, step 5: Requirement Phase Testing, Step6: Execute Test and Record Results,

UNIT 4

Step7: Acceptance Test Results Step8: Report Test Results, Step9: Testing Software Installing, Step10: Test Software Changes, Step11: Evaluate Test Effectiveness.

REFERENCES:

1. Effective Methods for Software Testing, William E. Perry, 2nd Edition 2003, Wiley
2. *Surviving the Top Challenges of Software Testing*, New York: Dorset House, 1997.,
Rice, Randall and Peery, William E.,
3. A practitioner's Guide to Software Test Design, By Lee Copelane.
4. The Art of Software Testing By Glenford Myers.
5. Testing Object System: Models, Patterns and Tools by Robert V Binder.

GRAPH THEORY

3:1:0

Course Code:CSB270

Course Outcomes:

At the end of the course students will be able to:

CO1. Explain basic concepts in combinatorial graph theory

CO2. Define how graphs serve as models for many standard problems

CO3. Discuss the concept of graph, tree, Euler graph, cut set and Combinatorics.

CO4. See the applications of graphs in science, business and industry.

UNIT 1

Introduction to Graph theory: Basic terminologies—direct & undirect graphs, walks, paths & circuits, sub-graphs and complements, Graph Isomorphism, vertex degree and regular graphs,

UNIT 2

Konigsberg bridge problem & Euler graphs. Hamilton graphs & traveling salesman problem, planar graphs- definition & examples, Bipartite & Kuratowskis graphs, Euler's formula & detection of planarity, Dual of Planar graphs,

UNIT 3

Graph Coloring: Proper coloring & chromatic number of graphs, Chromatic polynomial, four color problems, Trees: Optimization & Matching: Trees; Definition & Properties, Rooted & binary rooted trees, ordered trees & trees sorting. Weighted trees & prefix codes

UNIT 4

Spanning trees, optimization, Networks, Cutset, Edge & Vertex connectivity of a graph, Max-flow Min-cut theorem and its applications, Matching theory and its applications

References:

1. Graph Theory, V.K Balakrishnan, Schaum Series, McGrawHill, 1997
2. Graph Theory, by Frank Harary, Westview Press, 1994.
3. Introduction to Graph Theory, Douglas B west.
4. Hand Book of Graph Theory, Jonathan L Gross and Jay Yellen.
5. Graph Theory with application to Engineering and Computer science, Narsingh Deo.

OOAD

2:1:1

Course Code:CSB210

Course Outcomes

At the end of the course students will be able to:

CO1. Analyse, design, document the requirements through use case driven approach.

CO2. Identify, analyse, and model structural and behavioural concepts of the system.

CO3. Develop, and explore the conceptual model into various scenarios and applications.

CO4. Apply the concepts of architectural design for deploying the code for software
Implementation of Object Oriented concepts using C++

UNIT II

Introduction, Object orientation, OO development, OO themes, OO modeling ,History. Modeling, Abstraction, Models .Class Modeling Object & class, Link & Association concepts, Generalization & Inheritance, sample Class Model, Navigation of class models. Advanced class modeling: advanced object & class concepts, Association end N ary Association, Aggregation, Abstract Classes, Multiple Inheritance, Meta Data, Reification, Constraints, Derived Data.

UNIT III

State modeling: Events, States, Transitions & Conditions, State Diagrams, State Diagram behavior. Advanced state modeling, interaction modeling.

UNIT IV

System design: Overview of System design, Estimating performance, making a Reuse plan, Breaking a System into Subsystems, Identifying concurrency, Allocation of subsystems, Management of Data Storage.

Reference:

1. Object Oriented Analysis and Design – Blaha, Rambaugh.
2. Object Oriented Analysis and Design with the Unified Process- W Satzinger, Robert B Jackson and Stephen D Burd.
3. Object Oriented Analysis and Design with application, 3rd edition- Grady Booch, Robert A Maksimchuk, Michael W Engel
4. Object Oriented Analysis and Design with application- Grady Booch.

PROBABILITY AND STATISTICS

3:1:0

Course Code: CS A/B/C/D 340

Course Outcomes

At the end of the course students will be able to:

- CO1. Apply probability theory to set up tree diagrams
- CO2. Apply probability theory via Bayes' Rule
- CO3. Describe the properties of discrete and continuous distribution functions
- CO4. Use method of moments and moment generating functions
- CO5. Assess the consistency, efficiency and unbiasedness of estimators
- CO6. Apply method of maximum likelihood estimation
- CO7. Apply the Central Limit Theorem
- CO8. Use statistical tests in testing hypotheses on data

UNIT 1

Introduction, basic terminology, Interpretation of probability: Axioms of probability, Some elementary theorems, Conditional probability, Mathematical Expectation

UNIT 2

Probability Distributions: Introduction, Discrete probability distributions continuous probability distributions, The expected value of a random variable Chebyshev's Theorem

UNIT 3

Sampling distributions, Populations and samples, Sampling distribution, The sampling distribution of the mean, sampling distributions of proportions, sampling distributions of mean, chi squared distribution, F distribution.

UNIT 4

Estimation and inference theory, introduction, point estimation, interval distribution, bayesian estimation, test of hypot, Introduction to ANOVA.

References:

1. Probability and Statistics: Bheeshma Rao
2. Probability and Statistics, 4th edition, Degroot, Schervish.
3. Probability and Statistics for Engineering and Science, 8th edition, Jay L Devore.
4. Probability and Statistics, Michael Akritas.

5. An Introduction to Probability and Statistics, 3rd edition, Vijay K Rohatgi and A K MD Ehsanes Saleh.

DATA MINING

2:1:1

Course Code:CSD230

Course Outcomes:

At the end of the course students will be able to:

- CO1. Demonstrate an understanding of the importance of data mining and the principles of business intelligence
- CO2. Organize and Prepare the data needed for data mining using pre preprocessing techniques
- CO3. Perform exploratory analysis of the data to be used for mining.
- CO4. Implement the appropriate data mining methods like classification, clustering or Frequent Pattern mining on large data sets.
- CO5. Define and apply metrics to measure the performance of various data mining algorithms.

UNIT 1

Introduction, what kind of patterns can be mined? which technologies are used? which kind of applications are targeted?, major issues in Data mining.

Getting to know your data: Data objects and attribute types, basic statistical description of data, Data Visualisation, Measuring Data similarity and dissimilarity.

UNIT 2

Data Preprocessing: data cleaning, Data integration, Data Reduction, Data transformation and Data Discretization

UNIT 3

Mining frequent patterns, Associations and correlations: Basic concepts, Frequent mining methods, pattern Evaluation methods, Pattern mining in Multilevel multi dimensional space, Decision tree induction.

UNIT 4

Cluster analysis, partitioning methods, heirarchical methods, density based methods, grid based methods, evaluation of clustering.

References:

1. Data Mining: Concepts and Techniques, Third Edition, Jaiwei Han, Micheline Kamber, Jian Pei.
2. Learning Data Mining with Python, 2nd edition, Robert Layton.
3. Data Mining; The Text book, Charu C Aggarwal.
4. Data Mining, 4th edition: Practical Machine learning Tools and Techniques by Lan H Witten and Fibe Frank.
5. Introduction to Data Mining – Pang- Ning Tan and Micheal Steinbach

ARTIFICIAL INTELLIGENCE

3:1:0

Course Code:CS A/B/C/D 350

Course Outcomes

CO 1. At the end of the course students will be able to:

CO 1. Create appreciation and understanding of both the achievements of AI and the theory underlying those achievements.

CO 2. Know concepts of a Rational Intelligent Agent and the different types of Agents that can be designed to solve problems

CO 3. Review the different stages of development of the AI field from human like behavior to Rational Agents.

CO4. Impart basic proficiency in representing difficult real life problems in a state space representation so as to solve them using AI techniques like searching and game playing.

CO5. The basic issues of knowledge representation and Logic and blind and heuristic search, as well as an understanding of other topics such as minimal, resolution, etc. that play an important role in AI programs.

CO6. Introduce advanced topics of AI such as planning, Bayes networks, natural language processing and Cognitive Computing.

UNIT 1

AI problems, AI techniques, defining the problem as state space search, production systems, problem characteristics

Heuristic Search: Generate and test, hill climbing, BFS, Problem Reduction, Constraint Satisfaction, Means-End Analysis

UNIT 2

Knowledge Representation: Representations and mappings, approaches to knowledge representation

procedural v/s declarative knowledge, normal forms in predicate logic and clausal forms, non-monotonic reasoning

Declarative Representations: semantic nets, conceptual dependency, frames, scripts

UNIT 3

Game playing: minimax search procedure, adding alpha-beta cut offs

Planning: An Example Domain – the blocks world, Components, goal stack planning

UNIT 4

Expert systems: expert systems v/s conventional computers, expert system shells, explanation based learning.

Learning: Learning from observation - Inductive learning – Decision trees – Explanation based learning – Reinforcement Learning, Neural Networks, Introduction to Natural Language Processing.

References:

1. Artificial Intelligence, Third Edition, Elaine Rich, Kevin Knight, Shivashankar B Nair, Tata McGraw-Hill.
2. Introduction to Artificial Intelligence, Wolfgang Ertl.
3. Artificial Intelligence, 2nd edition, Stuart Russel, peter Norvig.
4. Artificial Intelligence, Jeorge F Luger
5. Artificial Intelligence, Saroj kaushik

.NET TECHNOLOGIES

2:0:2

Course Code:CSB280

Course Outcomes

At the end of the course students will be able to:

- CO1. Design web applications using .NET
- CO2. Use .NET controls in web applications.
- CO3. Debug and deploy .NET web applications
- CO4. Create database driven .NET web applications and web services

Unit 1

Benefits of .NET Framework, Architecture of .NET Framework 4.0, Components of .NET Framework 4.0: CLR, CTS, Metadata and Assemblies, .NET Framework Class Library, Windows Forms, ASP .NET.

Need of C#, C# Pre-processor Directives, Creating a Simple C# Console Application, Identifiers and Keywords. Data Types, Variables and Constants: Value Types, Reference Types, Type Conversions, Boxing and Unboxing, Variables and Constants. Expression and Operators: Operator Precedence, Using the :: (Scope Resolution) Operator and Using the *is* and *as* Operators. Control Flow statements: Selection Statements, Iteration Statements and Jump Statements.

Unit 2

Arrays and Strings: One Dimensional and Multidimensional Arrays, Jagged Arrays
Classes and Objects: Creating a Class, Creating an Object, Using this Keyword, Creating an Array of Objects, Using the Nested Classes, Defining Partial Classes and Method, Returning a Value from a Method and Describing Access Modifiers. Static Classes and Static Class Members. Properties: Read-only Property, Static Property, Accessibility of accessors and Anonymous types. Indexers, Structs: Syntax of a struct and Access Modifiers for structs. Strings: Constructing Strings, Operating on Strings, Arrays of Strings, The String Class

Unit 3

Encapsulation: Encapsulation using accessors and mutators, Encapsulation using Properties.
Inheritance: Inheritance and Constructors, Sealed Classes and Sealed Methods, Extension methods. Polymorphism: Compile time Polymorphism/Overloading, Runtime Polymorphism/

Overriding. Abstraction: Abstract classes, Abstract methods. Interfaces: Syntax of Interfaces, Implementation of Interfaces and Inheritance.

Delegates: Creating and using Delegates, Multicasting with Delegates. Events: Event Sources, Event Handlers, Events and Delegates, Multiple Event Handlers. Exception Handling: The try/catch/finally statement, Checked and Unchecked Statements.

Unit 4

Introduction, Windows Forms, Life Cycle, Event Handling: A Simple Event- Driven GUI, Visual Studio Generated GUI Code, Delegates and Event- Handling Mechanism, Another Way to Create Event Handlers, Locating Event Information. Control Properties and Layout, Labels, TextBoxes and Buttons, GroupBoxes and Panels, CheckBoxes and RadioButtons, ToolTips, Mouse-Event Handling, Keyboard-Event Handling. Menus, MonthCalendar Control, ListBox Control, CheckedListBox Control, ComboBox Control, TreeView Control, ListView Control, TabControl Control ; Building an Multiple Document Interface (MDI) Application. Introduction to ADO.NET

References:

1. Programming in C# 4.0, Tata McGraw Hill, Hebert Schildt
2. C# with .net 4.0 by Andrew Troelsen
3. Programming in C# , 3rd Edition, E Balaguruswamy
4. The Complete Reference C#, Herbert Schildt.
5. The Complete Reference ASP.NET, Robert Standefer III

OBJECT ORIENTED MODELING AND DESIGN WITH UML

2:1:1

Course Code:CSA/B/C/D 360

Course Outcomes

At the end of the course students will be able to:

- CO1. Design & Programming course is a unique course that teaches students how to use object-oriented techniques to build software.
- CO2. Gathering requirements & end with implementation.
- CO3. Analyze and design classes, their relationships to each other in order to build a model of the problem domain.
- CO4. Use common UML diagrams throughout this process, such as use-case, class, activity & other diagrams.
- CO5. Create The diagrams through a free tool.
- CO6. Capture and manage requirements.

Unit I

INTRODUCTION : Object-Oriented Analysis and Design - Iterative, Evolutionary, and Agile. -
Case Studies : The NextGen POS System - INCEPTION : Inception is Not the Requirements
Phase – Evolutionary Requirements - Use Cases – Other Requirements:NextGen Example

Unit II

ELABORATION ITERATION 1: Basics – Domain Models – System Sequence Diagrams –
Operation Contracts – Requirements to Design-Iteratively – Logical Architecture of UML
Package Diagrams

Unit III

On to Object Design – UML Interaction Diagrams – UML class Diagrams - GRASP : Designing
Objects with Responsibilities – Object Design Example with GRASP:NextGen POS system –
Designing for Visibility – Mapping Designs to Code

Unit IV

ELABORATION ITERATION 2: UML Tools and UML as Blueprint – Quick Analysis Update:
NextGen POS – Iteration 2: More Patterns – GRASP: More Objects with Responsibilities –
applying GoF Design Patterns

Reference:

1. "Applying UML and Patterns: An Introduction to Object-Oriented Analysis and Design and Iterative Development", Third Edition, Pearson Education, 2005, Craig Larman,
2. Advanced Object Oriented Analysis and Design using UML , James J Odell.
3. Object Oriented Analysis and Design , Mike O-Docherty.
4. Object Oriented Modeling and Design with UML, 2nd edition, Michael R Blaha, James R Rumbaugh.
5. Object Oriented Systems and Analysis and Design using UML, Simon Benneth, Steve McRovv and Ray Farmer.

ANDROID APPLICATION DEVELOPMENT

2:0:2

Course Code:CSA/B/C/D 370

Course Outcomes

At the end of the course students will be able to:

- CO1. Describe and compare different mobile application models/architectures and patterns.
- CO2. Apply mobile application models/architectures and patterns to the development of a mobile software application.
- CO3. Describe the components and structure of a mobile development framework (Google's Android Studio).
- CO4. Apply a mobile development framework to the development of a mobile application.
- CO5. Demonstrate advanced Java programming competency by developing a maintainable and efficient cloud based mobile application

Unit 1

Introduction to Android & Open Handset Alliance

Installation of Android Studio and other Development Environments like Gradle.

Understanding Android File System.

Creating First Android Application

Understanding Intent, Activity, Service, Content Providers, Broadcast Receivers.

Understanding Android Application, Manifests, Layouts, Drawables, Styles, Android Activity, View

Understanding Android XML based layout (Linear Layout, Relative Layout, Frame Layout).

Introduction to Android Life Cycle Events

initialization and Button Click Listeners.

Unit 2

Development of Simple app containing Dialog Box, Intents, Toast, Spinners, Listeners examples.

Android Listview / GridView and Adapters

Android Date Picker Dialog, Time Picker Dialog

Launching sub Activity

Building Custom ListView and Understanding Adapters in detail
Understanding SQLite database. Populating database.
Developing simple app by using SQLite database (insert, delete, update)

Unit 3

Working with web server basics
Background loading, AsyncTask , Using Threads
Developing simple app by downloading image from web and showing it in ImageView
Understanding Importance of External Libraries and demonstration of simple external library
Image lazy loading, Image loading in list view, grid view

Unit 4

Working with Google Maps
ViewPager
Introduction to fragment, add, remove, replace fragment
ViewPager
Side Navigation Drawer
Action bar/ Toolbar
ViewPager Adapter / Swipe View

References:

1. Android Programming for Beginners by John Horton.
2. Professional Android 4 application development by Reto Meir.
3. Android Book by Lan F Darwin.
4. Learning Android Building application for The Android Market by Marko Gargenta.
5. Programming Android Java programming for the new generation of Mobile Devises by Zigurd Mellieks

ADVANCED DATABASE MANAGEMENT SYSTEM

2:1:1

Course Code:CS A/B/C/D 380

Course Outcomes

At the end of the course students will be able to:

- CO1. Evaluate and Apply Advanced Database Development Techniques.
- CO2. Evaluate Database Systems.
- CO3. Administer Database Systems.
- CO4. Design & Implement Advanced Database Systems.

Unit 1

Disk storage, Basic file Structures and hashing, indexing structures for files.
Algorithms for query optimization.

Unit 2

Physical database design and tuning, Introduction to transaction, Concurrency control techniques.
Concept for object databases, Object databases standard and design, database security.

Unit 3

Enhanced datamodels for advanced applications, distributed databases and client server architectures, Emerging database technologies and applications.

Unit 4

Definition of NoSQL, History of NoSQL and Different NoSQL products, NoSQL Basics. Exploring one among MongoDB/CouchDB/Cassandra along with Java/Ruby/Python interface : Interfacing and Interacting with NoSQL, NoSQL Storage Architecture, CRUD operations, Querying, Modifying and Managing NoSQL Datastores, Indexing and ordering datasets

References:

1. Fundamentals of Database System :5th Edition ,Navathe

2. Database Management System, Panneerselvam R.
3. Database Management System, Raghu Ramakrishnan and Johannes Gehrke.
4. Data Schemes: Models and algorithms (Advances in Database Systems), Charu C Aggarwal.
5. Multilevel secure Transaction Processing (Advances in Database system), Vijay Atluri and Sushin Jajodia.

COMPILER DESIGN

3:0:1

Course Code:CS A/B/C/D 390

Course Outcomes

At the end of the course students will be able to:

- CO1. Construct a parse tree, or explain why no parse tree exists, given a BNF grammar and a string over the appropriate alphabet.
- CO2. Implement a lexical analyzer from a specification of a language's lexical rules.
- CO3. Compute the FIRST set for a BNF grammar.
- CO4. Compute the FOLLOW set for a BNF grammar.
- CO5. Determine FIRST intersect FIRST constraint satisfaction - determine if a BNF grammar satisfies the constraint on intersection of FIRST sets required for single-symbol-lookahead, top-down, lookahead parsing ()
- CO6. Determine FIRST intersect FOLLOW constraint satisfaction - determine if a BNF grammar satisfies the constraint on intersection of FIRST and FOLLOW sets required for single-symbol-lookahead, top-down, lookahead parsing ()

Unit-1

Introduction, Classification of grammars. Context free grammars. Deterministic finite state automata (DFA) Non-DFA.

Lexical analysis :Language processors; The structure of a Compiler; The evolution Of programming languages; The science of building a Compiler; Applications of compiler technology; Programming language basics. Lexical analysis: The Role of Lexical Analyzer; Input Buffering; Specifications of Tokens; Recognition of Tokens.

Unit-2

SyntaxAnalysis

Introduction; Context-free Grammars; Writing a Grammar. Top-down Parsing; Bottom-up Parsing.

Introduction to LR Parsing: Simple LR; More powerful LR parsers (excluding Efficient construction and compaction of parsing tables) ; Using ambiguous grammars; Parser Generators.

Unit-3

Intermediate Code Generation

Variants of syntax trees; Three-address code; Translation of expressions; Control flow; Back patching; Switch-statements; Procedure calls.

Run-Time Environments

Storage Organization; Stack allocation of space; Access to non-local data on the stack; Heap management; Introduction to garbage collection.

Unit-4

Code Generation

Issues in the design of Code Generator; The Target Language; Addresses in the target code; Basic blocks and Flow graphs; Optimization of basic blocks; A Simple Code Generator
Code optimization. Folding, redundant sub-expression evaluation. Optimization within iterative loops.

References:

1. Compilers Principles, Techniques and Tools, 2nd Edition, Addison-Wesley, 2007, Alfred V Aho, Monica S. Lam, Ravi Sethi, Jeffrey D Ullman
- 2."The Theory and Practice of Compiler Writing". McGraw Hill, New York, 1985, Tremblay, et. al
3. Principles, Techniques and Tools of Compilers.- Allen I Holob.
4. Elements of Compiler Design.- Meduna
5. Compiler Design - K Muneeswaran.

OPEN ELECTIVES

2:0:2

OP1: COMPUTER FUNDAMENTALS

Course Code: PHC/CHC/BCC/BTC/BOC/MAC/SWC/KAC/ENC/MCC/ZOC 640

Course Outcomes:

At the end of the course students will be able to:

- CO1. Use technology ethically, safely, securely, and legally.
- CO2. Identify and analyze computer hardware, software, and network components.
- CO3. Design basic business web pages using current HTML/CSS coding standards.
- CO4. Install, configure, and remove software and hardware.
- CO5. Use systems development, word-processing, spreadsheet, and presentation software to solve basic information systems problems.

UNIT 1

INTRODUCTION

Computer, Characteristic of Computer, History of Computer, Components of Computer
Key Factors of Computers: Hardware, Software - types of Software (Application and system), forms of software (firmware, shareware, freeware), Translator - Assembler, Compiler and Interpreters. Computer Application – Business, Scientific, Entertainment and educational.

CLASSIFICATION OF COMPUTERS

Mode of operations – Analog, Digital and hybrid Computers
Size and capabilities – Micro, Mini, Main frame and Super computer

UNIT 2

MEMORY UNITS

Primary memory - RAM, ROM, PROM, EPROM, EEPROM, Flash memory
Secondary memory – Magnetic disk (Hard disk, Zip disk, Jaz disk, Super disk)
Optical disk (CD, CD – R, CD – RW, DVD), Magneto-optical disk device

COMPUTER PERIPHERALS DEVICES AND INTERFACES

Input devices – Working principle of Keyboard and mouse, Functional capabilities of Scanner, Digital Camera, OMR, OCR, touch pad, touch screen. Output Devices – Monitor, Printer, Plotter and projector.

Processors, Classification of Processors on speed, Motherboard, Power Supply, I/O Ports and its Maintenance

PROGRAMMING LANGUAGES

Machine, Assembly language and High Level Language

UNIT 3

Introduction to Windows, Elements of Word Processing, Spread Sheet, Presentations .

Nudi/Baraha.

UNIT 4: INTERNET

Basics of Internet: www, HTTP, DNS, IP address, Email, Web browsers, Search Engines

HTML: Introduction to HTML, CSS

E-Commerce: Introduction, Types, Advantages of e-commerce, Applications, survey on popular e-commerce sites

E-governance, Introduction to Cyber Ethics

References:

1. Computer Fundamentals (6th Edition) – Rajaraman.
2. Computer's Today – Suresh K Basandra.
3. Computer Fundamentals-P K Sinha
4. Computer System Architecture (3rd Edition) PHI-2002. Chapters 3.3 & 3.4- Morris Mano,
5. Digital Principles and application (4th Edition) – Malvino Leach, Tata Mc Graw-Hill Edition
6. Computer System Architecture (3rd Edition) – Morris Mano, PHI
7. Microsoft office 365-Katherine Murray.
8. Microsoft office 2016- Nita Rutkosky, Denise Seguin, Audrey Rutkosky Roggenkamp
9. The Complete reference HTML by Herbert Schildt
10. Learn to program HTML and CSS for beginners
11. HTML black book –Steven Holzner.

Programe Outcomes

- PO1: Demonstrate critical reading, writing, and thinking skills. Write well-developed, focussed and effective paragraphs, which support a clear thesis statement, and demonstrate competence in Standard Kannada usage.
- PO2: Get the opportunity to opt for career in the field of social media
- PO3: Helps to pursue reserach work at M.phil and Doctoral level
- PO4: Help to communicate effectively and fluently at various occassions
- PO5: Analyse and interpret text written in Dravidian Language.
- PO6: Learn to write logical and informative papers
- PO7: Imbibe good ethics explored in the works of great writers.
- PO8: Learn to participate effectively in debates, group discussions, seminars.

Programe Specific Outcomes

- PSO 1: Equipped to work efficiently in various positions of Social Media.
- PSO 2: Good communicative leads to huge career opportunitis
- PSO 3: Able enough to work as researcher in the field of Kannada Literature and other Dravidian Languages
- PSO 4: Capable to work as teachers, trainers and Faculthy of Kannada.
- PSO 5: Learn to interpret text with attention to ambiguity, complexity and aesthetic value

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YA AO A AA AY AAPAE AE

YAE AS E	YAEVA A OPE	PEm « EA A	PEm UA AS I	GA A A	AA A A A I	YAE CA AU UA AE	YAE CA MA	DAVAPA HEEA AY SA CAPEA	M AACAPMA
<p align="center">ADAN VESAY UAA (Hard core)</p>									
1	VA PA A VA: YA A A PA A a AA CA PA (KNC 050)	2:1	3	2	2	3	70	30	100
2	Dz PA PA A A VA A E E (KNC 020)	2:1	3	2	2	3	70	30	100
3	AAVAPA A AE MA E (KNC 030)	2:1	3	2	2	3	70	30	100
4	AAZ SA Z CA V/UA PA E (KNC 040)	2:1	3	2	2	3	70	30	100
<p align="center">G YA A A A A (Soft core)</p> <p align="center">PA A A A A « z A y O A A » Az EDA A A A PA A A A F Z AA MO A A e Cz A A A A A A</p>									
5.1	BAASA VJANADAYAN LUKASASA VJANADAYAN (PA A) (KNC 210)	3:1	4	3	2	3	70	30	100
5.2	e A A A Cz A A PA A P d A A A A A A (KNC 210)	3:1	4	3	2	3	70	30	100
5.3	Z A W C z A A A A A A A DA A A A A A A E (KNC 210)	3:1	4	3	2	3	70	30	100
<p align="center">a AA LA O (Open Elective)</p>									
6.1	KANNA BAASA SAHITHA (YA O CA A A A A A Dz A PA) (KNC 530)	3:1	4	3	2	3	70	30	100

ಅನುಷ್ಠಾನ ಕಾರ್ಯಕ್ರಮದ ವಿವರಣೆ

ಕ್ರ.ಸಂ.	ಪ್ರಾ.ಬಂ.ಸಂ.	ಪ್ರಾ.ಬಂ.ನಾಮ	ಕ್ರ.ಸಂ.	ಕ್ರ.ಸಂ.	ಕ್ರ.ಸಂ.	ಕ್ರ.ಸಂ.	ಗಂ.ಸಂ.	ಗಂ.ಸಂ.	ಗಂ.ಸಂ.	
										ಪ್ರಾ.ಬಂ.ಸಂ.
ಪ್ರಧಾನ ವಿಷಯ UAA(Hard core)										
1	2018-19	ಬಹುಮುಖ ಅಭಿವೃದ್ಧಿ ಯೋಜನೆ (KND 050)	2:1	3	2	2	3	70	30	100
2	2018-19	ನೃತ್ಯ ಮತ್ತು ಸಾಹಿತ್ಯ (KND 020)	2:1	3	2	2	3	70	30	100
3	2018-19	ಕಲಾ ಮತ್ತು ಸಾಹಿತ್ಯ (KND 030)	2:1	3	2	2	3	70	30	100
4	2018-19	ಕಲಾ ಮತ್ತು ಸಾಹಿತ್ಯ (KND 040)	1:2	3	1	4	3	70	30	100
GYAA ವಿಷಯಗಳು (Soft core)										
5	2018-19	ಬಹುಮುಖ ಅಭಿವೃದ್ಧಿ ಯೋಜನೆ (KND 210)	3:1	4	3	2	3	70	30	100
5.1	2018-19	ಭಾಷಾವಿಜ್ಞಾನ ಅಧ್ಯಯನ : ಕನ್ನಡ ಭಾಷಾಸ್ವರೂಪ : ಆಯ್ದ ಪಠ್ಯಗಳು (KND 210)	3:1	4	3	2	3	70	30	100
5.2	2018-19	ಭಾಷಾವಿಜ್ಞಾನ ಅಧ್ಯಯನ : ಕನ್ನಡ ಭಾಷಾಸ್ವರೂಪ : ಆಯ್ದ ಪಠ್ಯಗಳು (KND 210)	3:1	4	3	2	3	70	30	100
5.3	2018-19	ಭಾಷಾವಿಜ್ಞಾನ ಅಧ್ಯಯನ : ಕನ್ನಡ ಭಾಷಾಸ್ವರೂಪ : ಆಯ್ದ ಪಠ್ಯಗಳು (KND 210)	3:1	4	3	2	3	70	30	100

ಪ್ರಧಾನ ವಿಷಯ : 52 ಘಂಟೆ
 ಉಪಪ್ರಧಾನ ವಿಷಯ : 16 ಘಂಟೆ
 ವಿಶೇಷ ಉಪಪ್ರಧಾನ ವಿಷಯ : 04 ಘಂಟೆ
 ಉಪವಿಷಯ : 04 ಘಂಟೆ
 ಒಟ್ಟು : 76 ಘಂಟೆ

ಇದರಲ್ಲಿ ಉಪವಿಷಯ : 16 ಘಂಟೆ
 ಉಪವಿಷಯ : 16 ಘಂಟೆ
 ಉಪವಿಷಯ : 16 ಘಂಟೆ

1. ಪದ್ಯ ಓದುವ ಕ್ರಮ, ಅರ್ಥೈಸುವ ಕ್ರಮ, ವಿಶ್ಲೇಷಣೆ ಮತ್ತು ವಿಮರ್ಶೆ
2. ಭಾವಸ್ವಾರಸ್ಯ, ಸಂಭಾಷಣಾ ಕೌಶಲ, ಸನ್ನಿವೇಶಗಳ ಪರಿಚಯ
3. ವಸ್ತು, ಪಾತ್ರ, ಭಾಷೆ ಬಳಕೆ, ರಚನೆ, ತಂತ್ರಗಾರಿಕೆ

12. 1
13. Literary Criticism – A Short History
14. Contemporary Criticism

: W.K. Wimsatt & Cleanth Brooks
: (Ed) Sethuraman V S

Gyanakshara (Soft Core)

KNA 210: 5.1 bhasha vijñānaḥ mahatvāḥ (3:1=4)

Course Outcome

- CO1: Identify the details of Language
CO2: Understand the characteristics of Phonetics
CO3: Learn in details with examples Bashe mattu Baraha
CO4: Write down in details with examples Morphology

1. bhāṣā: śāstrāṅgaḥ; bhāṣā vijñānaḥ svarūpa- śāstrāṅgaḥ, bhāṣāyāḥ GUṇāḥ zāṅgāḥ, bhāṣāḥ varṇanāḥ zāṅgāḥ, bhāṣāḥ vācāḥ zāṅgāḥ
2. zāṅgāḥ : śāstrāṅgaḥ; zāṅgāḥ śāstrāṅgaḥ zāṅgāḥ : śāstrāṅgaḥ zāṅgāḥ
3. Dharmaḥ : śāstrāṅgaḥ śāstrāṅgaḥ Dharmaḥ zāṅgāḥ zāṅgāḥ
4. bhāṣā baddhāḥ; bhāṣā śāstrāṅgaḥ, bhāṣā mahatvāḥ

5.2

1. śāstrāṅgaḥ : J.Zi.J. i. 0/2j
2. bhāṣā vijñānaḥ : JA az śāstrāṅgaḥ
3. bhāṣā mahatvāḥ : śāstrāṅgaḥ
4. śāstrāṅgaḥ bhāṣā vijñānaḥ : śāstrāṅgaḥ
5. śāstrāṅgaḥ bhāṣā vijñānaḥ : śāstrāṅgaḥ
6. bhāṣā : (śāstrāṅgaḥ) śāstrāṅgaḥ
7. Language : Leonard Bloomfield
8. Language : Edward Sapir
9. A Course in Modern Linguistics : C.F. Hackett
10. Modern Linguistics : S Potter
11. Introduction to theoretical Linguistics : J. Lyons
12. Historical Linguistics An Introduction : W.P. Lehmann

KNA 210: 5.2 e śāstrāṅgaḥ śāstrāṅgaḥ śāstrāṅgaḥ śāstrāṅgaḥ

Course Outcome

- CO1: understood the knowledge of Kannada Folklore
CO2: Students must learn the Ancient and Present styles of Kannada Folklore
CO3: Helps the students to undertake Research in the field of Kannada Folklore
CO4: understood the importance of Kannada Folklore

3. ಉಪಲಕ್ಷಣಗಳು: ಪದ್ಯದ ಉದಾಹರಣೆಗಳನ್ನು ನೀಡಿ ವಿವರಿಸಿ
4. ಪದ್ಯದ ಉದಾಹರಣೆಗಳನ್ನು ನೀಡಿ ವಿವರಿಸಿ

ಉಪಲಕ್ಷಣಗಳು

1. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
2. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
3. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
4. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
5. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
6. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
7. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
8. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
9. Impact of Marxism on Indian life and Literature : H.M. Nayak (Ed)
10. Heritage of Karnataka : R.S. Mugali

KNC 030: ಪದ್ಯದ ಉದಾಹರಣೆಗಳು (2:1=3 ಪದ್ಯ)

Course Outcome

- CO1: Write down the characteristics of Sanskrit Poetics
- CO2: Learn in details with examples Kannada Poetics
- CO3: Learn in details with examples Tamil Poetics
- CO4: Identify the characteristics of Folk Poetics

1. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
2. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
3. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
4. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು

ಉಪಲಕ್ಷಣಗಳು

1. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
2. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
3. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
4. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
5. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
6. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
7. ವಿಮರ್ಶೆಯ ಪರಿಭಾಷೆ : ಓ.ಎಲ್. ನಾಗಭೂಷಣಸ್ವಾಮಿ
8. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
9. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು
10. ಪದ್ಯದ ಉದಾಹರಣೆಗಳು : ಪದ್ಯದ ಉದಾಹರಣೆಗಳು

f .J , i. 2ª gA 2018 – a 2018
 (D) Dz 2018: PÄÄ « 2018 – z 2018 É
 , Á Á CŞ ÁŞ PÄÄ – Z 2018

Y 2018 2018

1. PÄÄ Á vÄZ 2018 gA² 2018
2. 2018 « 2018 2018 Á vÄZ 2018, 2018 UÄÄ
3. 2018 2018 Á vÄZ 2018, 2018 UÄÄ: " 2018 «
4. PÄÄ Á vÄZ 2018 2018 2018 Á
5. PÄÄ Á vÄZ 2018 vÄÄ 2018 2018
6. PÄÄ Á vÄZ 2018 Á 2018 gA² 2018

ಇಂಡಿಆರ್ನಿಟಿ

ಪ್ರಧಾನ ವಿಷಯಗಳು (Hard core)

KND 050: ಯುನೈಟೆಡ್ ಸ್ಟೇಟ್ಸ್ ಆಫ್ ಅಮೆರಿಕಾ: ಕವಿ: ಪಾಪ್ ಸಂಗೀತ, ಕವಿ

Course Outcome

- CO1: Learn the details of Selected Poems
 CO2: Learn in depth Purnachandra Tejaswi
 CO3: Understand in depth Modern Kannada Poetry
 CO4: Identify in depth Kannada Novel

1-2. ಪಾಪ್ ಸಂಗೀತ: 12 ಕವಿಗಳು

1. ಒಂಟಿ ಒಂಟಿ : ಒಂಟಿ ಒಂಟಿ
2. ಜಿ. ಎಸ್. ಶೇಷಾದ್ರಿ : ಜಿ. ಎಸ್. ಶೇಷಾದ್ರಿ
3. ಪಾಪ್ ಸಂಗೀತ : ಪಾಪ್ ಸಂಗೀತ
4. "ಒಂಟಿ ಒಂಟಿ : ಜಿ. ಎಸ್. ಶೇಷಾದ್ರಿ
5. ಕೆ. ವಿ. ಗೋಕುಲಕೃಷ್ಣ ಅಡಿಗ : ಕೆ. ವಿ. ಗೋಕುಲಕೃಷ್ಣ ಅಡಿಗ
6. f J, i J, i : f J, i J, i
7. ಗೋಕುಲಕೃಷ್ಣ ಅಡಿಗ : ಗೋಕುಲಕೃಷ್ಣ ಅಡಿಗ
8. gAe AÉi z AÁ : gAe AÉi z AÁ
9. 1 z AÁ : 1 z AÁ
10. 2 AÁ : 2 AÁ
11. 3 AÁ : 3 AÁ
12. ಸ. ಉಷಾ : ಸ. ಉಷಾ

3.4. ಪಾಪ್ ಸಂಗೀತ

ಕವಿ: ಪಾಪ್ ಸಂಗೀತ : ಕವಿ. ಯುನೈಟೆಡ್ ಸ್ಟೇಟ್ಸ್ ಆಫ್ ಅಮೆರಿಕಾ

ಕವಿಗಳು: (ಕವಿಗಳು)

1. 2 + AÁ : 2 + AÁ
2. 3 + AÁ : 3 + AÁ
3. ಪಾಪ್ ಸಂಗೀತ : ಪಾಪ್ ಸಂಗೀತ
4. "ಒಂಟಿ ಒಂಟಿ : ಜಿ. ಎಸ್. ಶೇಷಾದ್ರಿ
5. 2 AÁ : 2 AÁ
6. 3 AÁ : 3 AÁ
7. Cz É : Cz É
8. 3 AÁ : 3 AÁ
9. 3 AÁ : 3 AÁ

3. Cf vAÀÁ : CwUÀ É ÉÀVÀ
(D±À1, ¥ÀÀ20 j Az À75)
 4. §, ÀÀÀÀÀ-UÀVÀ 2 ã ÀÀÀÀ :
 1. ÀÀÀÀ" ÉÀÀÀ ÀÀ
 2. - ÈÀÀÀq ÈÀÀÀ À ÈÉ
 3. ° ßÈVÀz À ÈÀÀÀÀ
 4. ° À ÀwAz ÀÀÀÀ, ° À À
 5. z È ÀÈÀÀ ÀÈ - ÈÀÀÀS À À
 6. ã ÀÀ - ÈÀÀÀS À ÀPÀÀgÀÀÀ ã À ÀÀÀ
 7. ã ÀÀ ÈÀz À ã ÈÀÀVÀz ÈÉ
 8. ÈÈÀÀÀ
 9. ã ÈÈMgÀÀÀ ÈÉ
 10. C¥ÀÀq ÈÀ ÀÀÀÀÀ
 11. GvÀÀÀz À ° ÀÀÀÀS
 12. CgÀÀÀ ÀÈ À èCgÀÀÀÀ¥À Àz À
 5. ÈÈ gÁ z ÀÀÀÀÀÀÈ ° ÀÀÀÀ
 6. 1z ÀÀ ÀZÀ vÀ ÈÀÀÀÀ, Àçü
 7. §, ÀÀÀÀ : ÀÀz È ° ÀVÈz À ã ÀÀÀÀÀÀÈ (Àçü51)
 8. gÀ À ÀÀÀÀVÉ PÀÀÀ À
 9. gÀ ÀÀÀÀVÉ ±ÀÀÀÀ (D±À6, Àçü2, ¥ÀÀ1 j Az À104)
 10. gÀ À ÀÀÀ ÀÀ ÀÀÀÀÀÀ
 11. | ã gÀ ÀVÈ z Àz ÀLwÀ ÀÀ À ÀÀÀ: ° ÀÀÀÀ" ÈgÀ
 12. ÈÈ ÀÀÀ G ¥À À §t f UÀÈÈ ÀÀÀ
-

α Άζ Ά δέ ύ Άέ

2018-19 £Έ , Ά εΆ Ά β ΠΈϊΆ ΆJ AJ ΠΆθ ΆΖΆΆ άό , Άϑ Ά ΓέΆ

(¹ © J , ῑ - ¹ J f | α ΆΖ Ά)

ΥΆέ Ά² Ά ῑ δέ

3 UάmέΆ

CAP: 70

ΆΖ έ J- Ά β ΠΈϊΆ ΆJ AJ ΠΆθ

(J) Cx ΆΆ (©)

CAP: 14

(¹) Cx ΆΆ (r)

CAP: 14

(E) Cx ΆΆ (Jϑ)

CAP: 14

(f) Cx ΆΆ (JZῑ)

CAP: 14

ΆΆ Ά ζ έ K¹/ Ά β ΠΈϊΆ

CAP: 7X 2=14

(L) (M)

(eÉ) (ῑ)

(Πέ) (ΠΆ)

(J ῑ) (Dgῑ)

(J^a Ά) (J , ῑ)

(J £ῑ) (n)

ΥΆέ Ά² Ά ῑ δέ Ά - Ά β ΠΈϊΆ ΆJ AJ ΠΆθ ΆΖΆΆ άό , Άϑ Ά ΓέΆ



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

(Autonomous)

B N ROAD, MYSURU- 570 025

UG Syllabus

CHOICE BASED CREDIT SYSTEM

2019-20 & 2020-21

DEPARTMENT OF HISTORY

TEMPLATE SYLLABUS IN HISTORY

Programme: BA

Subject: History

Sem	Core			DSE			GE		
	No. of Courses	Credits	Total Hours	No. of Courses	Credits	Total Hours	No. of Courses	Credits	Total Hours
I	1	6	75+15	-	-	-	-	-	90
II	1	6	75+15	-	-	-	-	-	90
III	1	6	75+15	-	-	-	-	-	90
IV	1	6	75+15	-	-	-	-	-	90
V	-	-	-	2	6	75+15	1	2	15
VI	-	-	-	2	6	75+15	1	2	15

CBCS Syllabus for BA HEG

SEM	CORE COURSE	PAPER	COURSE CODE	TEACHING HOURS PER WEEK	L: T: P	TOTAL CREDIT	THEORY EXAM	I.A. MARKS	TOTAL MARKS
I	DSC-I	History of Ancient India (upto 1100 AD)	ELA24021	5+1	5:1:0	6	70	30	100
II	DSC-II	History of Medieval India (upto 1700)	ELB24021	5+1	5:1:0	6	70	30	100
III	DSC-III	History of Karnataka (540-1565)	ELC24021	5+1	5:1:0	6	70	30	100
IV	DSC-IV	History of Modern Karnataka (1700-1956)	ELD24021	5+1	5:1:0	6	70	30	100
V	DSE –I	History of Modern India (1757-1947)	ELE24021	5+1	5:1:0	6	70	30	100
	GE-I	History of Modern India 1947	ELE24221	2	2:0:0	2	35	15	50
VI	DSE – II	History of Modern Asia(1900-1995)	ELF24021	5+1	5:1:0	6	70	30	100
	GE -II	History of Modern India 1707	ELF24221	2	2:0:0	2	35	15	50

CBCS Syllabus for BA HP

SEM	CORE COURSE	PAPER	COURSE CODE	TEACHING HOURS PER WEEK	L: T: P	TOTAL CREDIT	THEORY EXAM	I.A. MARKS	TOTAL MARKS
I	DSC-I	History of Ancient India (upto 1100 AD)	DLA24022	5+1	5:1:0	6	70	30	100
II	DSC-II	History of Medieval India (upto 1700)	DLB24022	5+1	5:1:0	6	70	30	100
III	DSC-III	History of Karnataka (540-1565)	DLC24022	5+1	5:1:0	6	70	30	100
IV	DSC-IV	History of Modern Karnataka (1700-1956)	DLD24022	5+1	5:1:0	6	70	30	100
V	DSE –I	History of Modern India (1757-1947)	DLE24022	5+1	5:1:0	6	70	30	100
	GE-I	History of Modern India 1947	DLE24222	2	2:0:0	2	35	15	50
VI	DSE – II	History of Modern Asia(1900-1995)	DLF24022	5+1	5:1:0	6	100	30	100
	GE -II	History of Modern India 1707	DLF24222	2	2:0:0	2	35	15	50

CBCS Syllabus for BA HE

SEM	CORE COURSE	PAPER	COURSE CODE	TEACHING HOURS PER WEEK	L: T :P	TOTAL CREDIT	THEORY EXAM	I.A. MARKS	TOTAL MARKS
I	DSC-I	History of Ancient India (upto 1100 AD)	DLA24024	5+1	5:1:0	6	70	30	100
II	DSC-II	History of Medieval India (upto 1700)	DLB24024	5+1	5:1:0	6	70	30	100
III	DSC-III	History of Karnataka (540-1565)	DLC24024	5+1	5:1:0	6	70	30	100
IV	DSC-IV	History of Modern Karnataka (1700-1956)	DLD24024	5+1	5:1:0	6	70	30	100
V	DSE –I	History of Modern India (1757-1947)	DLE24024	5+1	5:1:0	6	70	30	100
	GE-I	History of Modern India 1947	DLE24224	2	2:0:0	2	35	15	50
VI	DSE – II	History of Modern Asia(1900-1995)	DLF24024	5+1	5:1:0	6	100	30	100
	GE -II	History of Modern India 1707	DLF24224	2	2:0:0	2	35	15	50

Programme Specific Out Comes for Bachelor of Arts

After completing the graduation in the Bachelor of Arts the students are able to:

- PO1. Critically recognize the social, political, economic and cultural aspects of History.
- PO2. Demonstrate thinking skills by analysing, synthesizing, and evaluating historical information from multiple sources.
- PO3. Correctly extract evidence from primary sources by analysing and evaluating them in relation to their cultural and historical context.
- PO4. Develop an informed familiarity with multiple cultures.
- PO5. Emerge as a multifaceted personality who is self-dependent.
- PO6. Spread the messages of equality, nationality, social harmony and other human values.
- PO7. Comprehend the basic structures and processes of government systems and/or theoretical underpinnings.
- PO8. Analyse political problems, arguments, information, and/or theories.
- PO9. Apply methods appropriate for accumulating and interpreting data applicable to the Discipline of political science.
- PO10. Acquire the ability to engage in independent and life-long learning in the broadest context socio-technological changes

Programme Specific Outcomes of History& Political Science

After completing the graduation in the Bachelor of Arts the students are able to:

- PSO1 This Programme exerts it's Influence on life and destiny of Human beings.
- PSO2 It is a stepping stone for one's success in competitive examinations.
- PSO3 Understand the background of our religion, customs, institutions and so on.
- PSO4 Understand the present Social, political, religious and economic conditions of the people.
- PSO5 Analyze the relationship between the past and the present, is lively presented in the History
- PSO6 The study of History helps to impart moral education and the feeling of patriotism in the hearts of the pupils

Programme Outcomes for Bachelor of Arts in History and English:

After completing the graduation in the Bachelor of Arts the students are able to:

- PO1. Critically recognize the social, political, economic and cultural aspects of History.
- PO2. Demonstrate thinking skills by analyzing, synthesizing, and evaluating historical information from multiple sources.
- PO3. Correctly extract evidence from primary sources by analyzing and evaluating them in relation to their cultural and historical context.
- PO4. Develop an informed familiarity with multiple cultures.
- PO5. Demonstrate critical reading, writing, and thinking skills.
- PO6. Write focused, organized, well-developed, and text-based essays using effective paragraphs.

- PO7. Support a clear thesis statement, and demonstrate competence in Standard English Language and usage.
- PO8. Develop and carry out research projects, and locate, evaluate, organize, and incorporate information effectively.
- PO9. Write articles, novels, stories to spread the messages of equality, nationality, social harmony and other human values.
- PO10. Acquire the ability to engage in independent and life-long learning in a broader context about socio-technological and demographic changes.

Programme Specific Outcomes of History & English

PSO1 .Identify cultural and literary synthesis

PSO2 .Critically recognize the social, political, economic and cultural aspects of History.

PSO3.Support a clear thesis statement, and demonstrate competence in Standard English Language and usage.

PSO4 .Write articles, novels, stories to spread the messages of equality, nationality, social harmony and other human values.

PSO5.Correctly extract evidence from primary sources by analyzing and evaluating them in relation to their cultural and historical context.

PSO6 .Develop an informed familiarity with multiple cultures.

SEMESTER-I
DSC1: HISTORY OF ANCIENT INDIA Up to 1100
Credit (L: T: P = 5: 1: 0)

Course Outcomes:

- CO1. Familiarise the students of early civilizations. The birth of new religions. Jainism and Buddhism and the teachings of Mahaveera and Buddha
- CO2. Discuss ancient republics, establishment of great Empires political land military Adventures of our great rulers
- CO3. Gain knowledge of Economic, Social and religious conditions and education system of Ancient period.
- CO4. Inspire the students through the great literary books and contributions to the growth of Art & Architectures
- CO5. Understanding the administration of our great kingdoms and foreign trade And commercial activities are of great values in the development of the state.

Unit – I

Sources – Harappan culture – Vedic civilization - Rise of new Religions – Jainism; Life and Teachings of Mahaveera – Buddhism – Life and Teachings of Buddha, Ancient Indian Republics

Unit – II

Historical background the Mauryans – Chandragupta Maurya and Ashoka – Administration – Art and Architecture – The Shathavahanas – Cultural Contributions - The Kushanas – Kanishka – achievements – contribution to culture

Unit – III

The Guptas – Samudragupta and Chadragupta II – Golden Age Contributions, Vardhanas – Harshavardhana – Achievements – Nalanda University

Unit-IV

The Chalukyas - Pulikeshi II - Contributions to culture-The Rastrakutas-AmoghaVarsha Nrupatunga I-Contributions to culture-The Hoysalas-Vishnuvardhana-Art and Architecture

Unit – V

The Pallavas of Kanchi ; Mahendravarma I & Narasimha Varman I, Growth of Art and architecture – The Cholas – Rajaraja Chola I and Rajendra Chola I – Administration with reference to local – self government Art & Architecture

MAP

Places of Historical Importance:

- | | |
|----------------|-----------------|
| 1. Paithan | 6. Malkhed |
| 2. Lothal | 7. Nalanda |
| 3. Pataliputra | 8. Maski |
| 4. Badami | 9. Dwarasamudra |
| 5. Thanjur | 10. Kanchi |

Books for Study and Reference:

- 1) Romila Thapar – Recent perspective of India History – Oxford University Press Delhi – 1988
- 2) Romila Thapar – From Lineage to State – Oxford University Press – 2000
- 3) Romila Thaper – Ashoka and the decline of the Maurayas – Oxford University Press, Delhi – 1999
- 4) D. Kosambi – The Culture and Civilization of Ancient India – An historical outline – Vikas publications, New Delhi – 1997
- 5) A.L. Basham – The wonder that was India Vol – 1, Rupa & Co.,
- 6) R.S. Sharma – Indian Feudalism, Macmillan Publications, Madras – 1998
- 7) R.S. Sharma – Material Culture and Social Formation in Ancient India
- 8) R.S. Sharma – Aspects of political ideas and institutions in ancient India Motilal Banarasidas publication, New Delhi – 1999
- 9) B.N. Luniya – Evolution of Indian culture – Laxmi Narayan Agarwal, Agra – 1997
- 10) R.C. Manumdar – Ancient India – Motilal Banarsidas publications, New Delhi – 1982
- 11) A.S. Altekar – State & Government in Ancient India (Third Edition)
- 12) Vijaya Kumar – Urbanization in Ancient India (Thakur Abhinava Publication, New Delhi – 1981)
- 13) Irfan Habib and Vijaya Kumar Thakur – The Vedic Aryans
- 14) Burton Stein-Peasant and society in Medieval South India, Oxford, Delhi 1999
- 15) T.V. Mahalingam: South Indian politics
- 16) Dr. N. Chinnaswamy (Ed), Badami Chalukyas, Kannada Uni. Hampi

II – SEMESTER
PAPER- II
DSC2: HISTORY OF MEDIEVAL INDIA Upto 1700
Credit (L: T: P = 5: 1: 0)

Course outcomes:

- CO 1. Assimilate knowledge to the students to learn the developments in India during the Muslim rule.
- CO 2. Conceptualize the Persian literary sources, military adventures and administration of Delhi Sulthanate and the great Moghuals
- CO 3. Gain knowledge about the cultural contributions of sulthans in the medieval period.
- CO 4. Inspire the students through the great literary books and contributions to the growth of Art and Architectures.
- CO 5. Understanding the administration of our great kingdoms and foreign trade and commercial activities are of great values in the development of the state.

Unit-I

The Rajputs-Origin-Early Turkish invasions-Mahammed of Ghor-Battles of Tarain-Rajput Architecture

Unit – II

Delhi Sultanate – Early Turkish Rule; The Slave Dynasty – The Khalijis : Alauddin Khaji, expansion of the Delhi Sultanate and Administrative reforms – The Tughluqs : Muhammad – bin – Tughluq – Administrative Experiments – Tughluq

Unit – III

The Great Mughals – Babur – foundation of the Mughal empire – Revival of Afghan rule – Shershah and his administration – Akbar – expansion of the empire, administrative reforms – religious policy – Aurangzeb – religious policy and Deccan policy-Contribution to Culture

Unit – IV

Rise of the Marathas – Shivaji – Military achievements – administration – Rise of peshwas – Balaji Vishwanath, Bajirao 1 and expansion of the Maratha power and Balaji Bajirao – The third battle of Panipat – 1761 causes of results

Unit – V

Society and Economy – Bhakti Movement – features – Kabir and Guru Nanak – Sufi Movement – features – Muinuddin Chisti and Nizamuddin Aulia – Economy : Growth of new towns and cities, trade and commerce

MAP

Places of Historical Importance:

- | | |
|-------------|-------------------|
| 1. Tarain | 6. Kabul |
| 2. Delhi | 7. Golkonda |
| 3. Panipat | 8. Fathepur Sikri |
| 4. Devagiri | 9. Agra |
| 5. Poona | 10. Ahmadnagar |

Books for Study and Reference:

- 1) Irfan Habib – Medieval India (1200 – 1750) Oxford University 1997
- 2) Sathish Chandra – Medieval India from Sultante to Mughals (Har Anand Publication – New Delhi – 1998)
- 3) Irfan Habib – Akbar & His India. Oxford Press – New Delhi – 1998
- 4) Muzafar Alam & Sanjay Subramanya – The Mughal State 1526-1750 (Oxford University – 1998)
- 5) Tarachand – Influences of Islam on Indian Culture (The Indian Press Publication Ltd, Allahabad)
- 6) L.P. Sharma – Sultanate of Delhi (Konark publication, New Delhi – 1998)
- 7) Moreland W H – From Akbar to Arurangazeb
- 8) Tapan Roy Choudhari & Irfan Habib – The CAMBRIDGE Economic History of India (2 Vol) orient and Logman in association with vambridge uni press)
- 9) L.P. Sharma History of Medival India – (1000-1707) Konark publications – New Delhi)
- 10)R.C. Majumdar – Ancient India
- 11)R.C. Majumdar and others – Advanced History of India, Macmillan India Ltd., Delhi 2000
- 12)SAA Razvi – Wonder that was India II
- 13)A C Banerjee – Medieval India
- 14)M Habibi and KA Nizami – A Comprehensive History of India – vol – Delhi Sultanate

III – SEMESTER
DSC3: HISTORY OF KARNATAKA
(540-1565)

Credit (L: T: P = 5: 1: 0)

Course outcome:

- CO1. Understand the historical growth of Karnataka, sources-Geographical feature and Early kingdom
- CO2. Enable the students to learn the contributions of Chalukyas, Rastrakutas and Hoysalas development of Art and Architecture.
- CO3. Understand the glorious days of Vijayanagara Empire. The developments of Economy, Social and religious life style, contribution, Administration and culture.
- CO4. Gain knowledge about Bahamani and Adilshahi's Kingdom, their contribution to Education and Culture.
- CO 5. Helpful for the students to understand the policy of Wodeyar and their contributions to the of growth and development of Mysore

Unit – I

Sources: Literary sources and Archeological sources

Unit – II

Socio-Religious movement-Shankaracharya-Ramanujacharya-Madhvacharya and Basaveshwara

Unit – III

Vijayanagar Empire – Origin,Sangam Dynasty: Deveraya II – Krishnadevaraya – Battle of Talikote

Unit – IV

Administration Cultural Contributions during Vijayanagar, Vijayanagara as seen by Foreign Travelers

Unit – V

The Bahamani Kingdom – Mohammad Gawan – Bahamani and Adilshahi's Art and Architecture Wodeyars of Mysore, Early Wodeyars – Raja Wodeyar -Chikkadevaraja Wodeyar – Administration and literature

MAP

Places of Historical Importance:

- | | |
|-------------------|--------------------|
| 1. Udupi | 6. Srirangapattana |
| 2. Srirangam | 7. Talikote |
| 3. Kudala Sangama | 8. Mysore |
| 4. Gulbarga | 9. Raichur |
| 5. Hampi | 10. Bidar |

Books for Reference:

1. Burton Stein – Peasant State and Society in Medieval South India, Oxford, Delhi 1999
2. A.R. Kulkarni : Medieval Deccan History, Popular Pub House, Bombay 1996
3. K.A. Neelakanta Sastri : A History of South India, Oxford University Press, Madras
4. Heras : South India under Vijayanagar Empire (2 Vol)
5. Robert Sewl : A Forgotten Empire – Asian Educational Services, New Delhi
6. S. Rajashekara : Karnataka Architecture, Sujatha Pub 1985
7. T.V. Mahalingum : South Indian Polity
8. B.A. Salatore : Social Political Life in Vijayanagar Empire (II Vol)
9. Gordon Johnson : Architecture and Art of Southern India, Cambridge University Press 1998
10. Appadorai A : Economic Condition in South India, 1000-1500 AD (2 Vols) Madras University, Madras, 1936
11. K.R. Basavaraja : History and Culture of Karnataka, Chalukya Pub. Dharwad, 1984
12. G.S. Dixit – Krishnadevaraya – Life and Achievements, Pub. Directorate of Archaeology and Museums, Mysore
13. Shamarao : History of Mysore
14. Burton Stein : Vijayanagar, Cambridge Uni. Press
15. Dr. N. Chinnaswamy – (Ed) Badami Chalukyaru, Kannada Uni. Hampi

IV – SEMESTER
DSC5: HISTORY OF MODERN KARNATAKA
(1700-1956)

Credit (L: T: P = 5: 1: 0)

Course Outcomes:

- CO1. Acquire knowledge about Dalavoy's, Hyder Ali and Tippu-Anglo-Mysore war.
- CO2. Evaluate the Restoration of Mysore and understand Resistance against the British.
- CO3. Gain knowledge to students about the Commissioner's rule in Mysore and Development of Mysore during Diwans.
- CO4. Providing basic information on Backward Class movement in Mysore.
- CO5. Stimulate interest among students for Freedom Movement in Karnataka and Unification movement and Birth of United Karnataka in 1956-Historical places of Freedom struggle and unification movement

Unit – I

Mysore Between 1704 – 1750 Dalavoy's – rise of Hyder Ali – Relations with the British and the Marathas – Tippu Sultan – His relations with the British

Unit – II

Restoration of Mysore – Krishnaraja Wodeyar III – Dewan Purnaiah – Direct Rule of Krishnaraja Wodeyar – Nagar insurrection – Cultural achievements of Krishnaraja Wodeyar III

Unit – III

Commissioner's Rule – Mark Cubbon – administration – LB Bowring – Administration

Unit – IV

Rendition of Mysore – Dewan Rangacharlu – Sheshadri Ayer – Vishweshwaraiah – Sir Mirza Ismail

Unit – V

Backward class movement in Mysore – Freedom Movement in Karnataka – Belgaum Congress Session – Unification Movement – Birth of United Karnataka in 1956

MAP

Places of Historical Importance:

- | | |
|------------------|---------------|
| 1. Ankola | 6. Bangalore |
| 2. Mysore | 7. Belagum |
| 3. Vidhuraswatha | 8. Mangalore |
| 4. Easuru | 9. Bidanoor |
| 5. Kittur | 10. Shivapura |

Books for Reference:

1. Mohibul Hassan – History of Tippu Sultan the World Press Pvt. Ltd., 1971
2. P.B. Desai – History of Karnataka
3. K.R. Basavaraj – History and Culture of Karnataka, Chalukya Pub, Dharwad
4. Sheik Ali – Tippu Sultan – Geetha Book House, Mysore – 1982
5. James Manor – Political Change in an Indian State (Mysore 1831 – 1920) Rawal Pub. Jaipur 1996
6. Veerathappa K – Readings in Modern History of Mysore S Chand & Co. Delhi 1985
7. I.M. Muthanna – History of Modern Karnataka – Sterling Pub Delhi 1986
8. Naidu B.M – Intellectual History of Colonial India (Mysore 1831 – 1920) Rawal Pub. Jaipur 1996
9. Chandrashekar S – Dimenstions of Socio Political Change in Mysore 1918 – 1940 Delhi 1965
10. K.G. Made Gowda – Modern Mysore State, (A Study of Elite Polity and Society) Mysore Uni. Pub. 1997
11. Hayavadana Rao – Mysore Gazetteers Vol. 1 to 5, Bangalore 1929
12. R. Ramakrishna – Press and Politics in Modern Mysore 1997
13. G.S. Halappa – History of Freedom Movement in Karnataka
14. Shastri K.N.V. - Administration of Mysore under Mark Cubbon – London 1932
15. Mark Wiks – History of Mysore (3 Vol)
16. R.R. Diwakar – Karnataka through the Ages
17. Dr. R. Gopal and Dr. Narendra Prasad – Mummadi Krishnaraja Wodeyar Pub, Directorate of Archeology and Museums, Mysore
18. Dr. B.V. Sudhamani – Dalavoy Nanjaraja
19. Dr. K.S. Shivanna – Karnataka Charitre, Vol 3, Kannada University Hampi

DSE – 1 : HISTORY OF MODERN INDIA (1498-1947)

Credit (L: T: P = 5: 1: 0)

Course outcomes:

CO1. Understand the detailed picture of the heroic resistance Indian to the company's rule, the battle of Plassey, Buxar and Carnatic wars and their effects

CO2. Develop the knowledge of Consolidation of the British rule regulating Act 1773, subsidiary alliance, doctrine of lapse and land revenue policies.

CO3 Indian renaissance and change of administration, the great revolt of 1857. It will inspire students to appreciate and respect national leaders and values of patriotism and nationalism.

CO 4. Gain knowledge about foundation of Indian National congress. Role of moderates, extremists and Gandhian era., to the students

Unit 1

Advent of Europeans –Portuguese-English-French-Anglo-French Rivalry- Conquest of Bengal by the British –Battles of Plassey and Buxar –Robert Clive –Dual Government in Bengal –Anglo – Maratha Wars Rise of Sikhs – Ranjit Singh –Anglo Sikh Wars

Unit-II

Consolidation of the British rule- The regulating Act-1773 Pitts's India Act –lord Wellesly – Subsidiary Alliance _features and significance –doctrine of Lapse-Land Revenue Policies – Zamindari System –Roytwari System –its features

Unit III

Rebellion of 1857 –cause and consequences –Queen's proclamation act of 1858 – socio and Religious movements –Causes of the Indian Renaissance – the Brahma Samaj, Arya samaj, Ramakrishna Mission – The Aligarh Movement

Unit –IV

Rise and Growth of Nationalism – Foundation of Indian National Congress Moderates and their Achievements – Rise of Extrimism – Growth of Revolutionaries Nationalism till 1930 Gadar Movement, Chandrasjekar, Azad, Bhagat Sign

Unit –V

Gandhi Era – Round Table Conference – Government India Act of 1935 – Quit India Movement – Partition and Independence and Dr. B. R Ambedkar

Books For Reference:

1. Percival Spear: Oxford History of Modern India (1740-1975(new Delhi
2. Arthar D .Innes: History of British in India, Lal Pub. New Delhi 1999
3. Ranjit Guha :subaltern studies(10 Vols)
4. A.R Desai Social Background to Indian Nationalism
5. Tharachand History of Freedom Movement

V SEMESTER
GE – 1 :HISTORY OF MODERN INDIA (1498 – 1858)
Credit (L: T: P = 2: 0: 0)

Course outcome:

Unit 1

Advent of Europeans –Portuguese-English-French-Anglo-French Rivalry- Conquest of Bengal by the British –Battles of Plassey and Buxar –Robert Clive –Dual Government in Bengal – Anglo – Maratha Wars Rise of Sikhs – Ranjit Singh –Anglo Sikh Wars

Unit-II

Consolidation of the British rule- The regulating Act-1773 Pitts's India Act –lord Wellesly –Subsidiary Alliance _features and significance –doctrine of Lapse-Land Revenue Policies Rebellion of 1857 –causes and consequences –Queen's proclamation act of 1858

Books For Reference:

1. Percival Spear: Oxford History of Modern India (1740-1975(new Delhi
2. Arthar D .Innes: History of British in India, Lal Pub. New Delhi 1999
3. Ranjit Guha :subaltern studies(10 Vols)
4. A.R Desai Social Background to Indian Nationalism
5. Tharachand History of Freedom Movement

VI SEMESTER
DSE – 2: HISTORY OF MODERN ASIA (1900 – 1995)
Credit (L: T: P = 5: 1: 0)

Course outcome

CO1. Analyze the progress of Asian countries like China and Japan from insular nations to their present Dynamic position

CO2. Understand to trace their role in world affairs in the last 3 decades of the 20th Century

CO3. Develop the knowledge about diverse countries of the region and provide an insight into the historical background

CO4. Evaluate the basics of colonization and decolonization and analyse the areas of conflict in this vital region. Historical background of Iran, Arabs and Jews. Rise and growth of Arab nationalism, Zionist movement,

Unit I – China: Historical background – Opium Wars

- 1) The Boxer Rebellion : Causes and results – Boxer Protocol
- 2) Revolution of 1911 – Causes and Consequences – Role of Dr Sun Yat – Sen
- 3) Conflicts between the Kuomintang Nationalists and the Communists

Unit II – Japan

- 1) Historical background with reference to Meiji Restoration
- 2) The Anglo – Japanese Alliance – 1902 – Its significance
- 3) Japan's relations with China between 1914 to 1939

Unit III – Turkey

- 1) Disintegration of Ottoman Empire
- 2) The Young Turk Movement – achievements of the Young Turk government
- 3) Mustafa Kemal Pasha – Modernization of Turkey

Unit IV – Iran

- 1) Historical background
- 2) Pahlavi Regime – Reza Shah Pahlavi
- 3) Mossadeq – The Oil Crisis

Unit V – Arabs and Jews

- 1) Rise and Growth of Arab Nationalism
- 2) Zionist Movement: Genesis and growth – Balfour Declaration, Birth of Israel

Books for Reference:

- 1) H.M. Vinacke: A History of the far East in Modern time
- 2) W. David M.D. Rise and Growth of Modern China. Himalayan Pub. Bombay 1987
- 3) D.G.E. Hall History of South East Asia
- 4) S.L. Roy : History of Far East and South East Asia (2 Vol)
- 5) K.M. Panikar : Asia and Western Dominance
- 6) Luke: Modern Turkey
- 7) Karan Armstrong : A History of Jerusalem
- 8) Immanuel C.Y. HSU : The rise of Modern China, N.Y. 1995

- 9) George Ienczowsky : Middle East in the World Affairs
- 10) J.C. Michael – The Shaping of Arabs Clyde and Beers: The Far East
- 11) Shaik Ali thers : A short History of Modern Asia (1900 – 19600) Chandramouli Prakashana
– 1987
- 12) B.K. Gokhale – History of Modern World, Himalaya Pub. House, New Delhi

IV SEMESTER

GE-2: HISTORY OF MODERN INDIA (1828-1947)

Credit (L: T: P = 2: 0: 0)

Unit-I

Socio and Religious movements –Causes of the Indian Renaissance – the Brahma Samaj, Arya Samaj, Ramakrishna Mission – The Aligarh Movement

Rise and Growth of Nationalism – Foundation of Indian National Congress Moderates and their Achievements – Rise of Extremists

Unit –II

Gandhi Era – Round Table Conference – Government India Act of 1935 – Quit India Movement – Partition and Independence

Books For Reference:

1. Percival Spear: Oxford History of Modern India (1740-1975) (new Delhi)
2. Arthar D .Innes: History of British in India, Lal Pub. New Delhi 1999
3. Ranjit Guha :subaltern studies(10 Vols)
4. A.R Desai Social Background to Indian Nationalism
5. Tharachand History of Freedom Movement

Pattern of Question Paper

1. Internal Assessment

Each paper will have an internal assessment component for 30 marks out of the maximum 100 Marks

2. **Theory Examination**

Max Marks 70

Theory question paper pattern

Theory question paper for I – IV semester will have Four Parts

**Question Paper Pattern
Theory (6 Credits)**

Time: 3 Hours

Max. Marks: 70

PART – A

This part consists of **two** questions of 15 marks each and the student will have to answer **one** question **15 x 1 = 15**

PART – B

Part – B consists of five questions of **ten** marks each and the students will have to answer **three** questions **10 x 3 = 30**

PART – C

This part will have **five** questions of **five** marks and the students will have to answer **three** questions **5 x 3 = 15**

PART – D

Map question will have **Ten** marks out of which Five marks will be awarded to locating Five places. **Five** marks for writing explanatory note on the places **5 + 5 = 10**

The V & VI semester will have **Three parts** **Part – A** consists **01** question for **15** Marks. **Part – B** consists of **04** questions for **40** Marks and **Part – C** consists of **15** Marks for **03** questions.

GENERIC ELECTIVES

Pattern of Question Paper

1. Internal Assessment

Internal assessment component for 15 marks out of the maximum 50 Marks for C1&C2

2. **Theory Examination**

Max Marks:35

Theory question paper pattern

Theory question paper for all the V – VI semester will have 3 Parts

Question Paper Pattern

Theory (2 Credits)

Time: 2.00 Hours

Max. Marks: 35

PART – A

This part consists of **five compulsory** questions of 2 marks each and the student will have to answer **one** question **5 x 2 = 10**

PART – B

Part – B consists of two questions of **Ten** marks each and the students will have to answer **one** question **10 x 1 = 10**

PART – C

This part will have **Five** questions of Five marks and the students will have to answer **Three** questions **5 x 3 = 15**

ಭಾಷೆ ಕನ್ನಡ

ಸಿಬಿಸಿಎಸ್ ಪದ್ಧತಿಯ ಪಠ್ಯವಿನ್ಯಾಸ

2019-20ರಿಂದ ಆರಂಭವಾಗುವ ಶೈಕ್ಷಣಿಕ ವರ್ಷಕ್ಕೆ ಸಿಬಿಸಿಎಸ್ ಪದ್ಧತಿಯ ಪಾಠಕ್ರಮ ಮತ್ತು ಪರೀಕ್ಷಾ ಯೋಜನೆ
(ಎಲ್ಲಾ ನಿಕಾಯಗಳಿಗೂ ಅನ್ವಯವಾಗುವಂತೆ)

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್) ನಿರಂತರ
ಮೌಲ್ಯಮಾಪನ,

ಸಿ. - 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ - 2 (15 ಅಂಕಗಳು) ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

BA(AECC)

ಸೆಮಿಸ್ಟರ್	ಪತ್ರಿಕೆ	ಪತ್ರಿಕೆ ಶೀರ್ಷಿಕೆ					ಪರೀಕ್ಷಾ ಅವಧಿ (ಗಂಟೆಗಳು)	ಅಂಕಗಳು	ಆಂತರಿಕ ಮೌಲ್ಯಮಾಪನ ಅಂಕಗಳು	ಒಟ್ಟು ಅಂಕಗಳು
			ಕೋರ್ಸ್ ಕೋಡ್	ಕ್ರೆಡಿಟ್ ವಿನ್ಯಾಸ LTP	ಒಟ್ಟು ಕ್ರೆಡಿಟ್‌ಗಳು	ವಾರಕ್ಕೆ ಬೋಧನಾ ಗಂಟೆಗಳು				
ಮೊದಲನೆಯ ಚತುರ್ಮಾಸ										
1	1	ಘಟಕ 1 : ಪದ್ಯಭಾಗ ಘಟಕ 2 : ಗದ್ಯಭಾಗ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	ELA06021 ELA06022 ELA06023 ELA06024 ELA06025	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100
ಎರಡನೆಯ ಚತುರ್ಮಾಸ										
2	2	ಘಟಕ 1 : ನಾಟಕ ಘಟಕ 2 : ಕಾದಂಬರಿ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	ELB06021 ELB06022 ELB06023 ELB06024 ELB06025	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100
ಮೂರನೆಯ ಚತುರ್ಮಾಸ										
3	3	ಘಟಕ 1 : ಕಾವ್ಯ ಭಾಗ - ೧ ಘಟಕ 2 : ಕಥನ ಭಾಗ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	ELC06021 ELC06022 ELC06023 ELC06024 ELC06025	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100
ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸ										
4	4	ಘಟಕ 1 : ಕಾವ್ಯ ಭಾಗ - ೨ ಘಟಕ 2 : ಗದ್ಯ ಭಾಗ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	ELD06021 ELD06022 ELD06023 ELD06024 ELD06025	01 01 01 2:1:0	03	1 1 2 04		40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100

ಭಾಷೆ ಕನ್ನಡ

ಸಿಬಿಎಸ್ ಪದ್ಧತಿಯ ಪಠ್ಯವಿನ್ಯಾಸ

2019-20ರಿಂದ ಆರಂಭವಾಗುವ ಶೈಕ್ಷಣಿಕ ವರ್ಷಕ್ಕೆ ಸಿಬಿಎಸ್ ಪದ್ಧತಿಯ ಪಾಠಕ್ರಮ ಮತ್ತು ಪರೀಕ್ಷಾ ಯೋಜನೆ
(ಎಲ್ಲಾ ನಿಕಾಯಗಳಿಗೂ ಅನ್ವಯವಾಗುವಂತೆ)

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್) ನಿರಂತರ
ಮೌಲ್ಯಮಾಪನ,

ಸಿ. - 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ - 2 (15 ಅಂಕಗಳು) ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

BCOM /BBA (AECC)

ಸೆಮಿಸ್ಟರ್	ಪತ್ರಿಕೆ	ಪತ್ರಿಕೆ ಶೀರ್ಷಿಕೆ					ಪರೀಕ್ಷಾ ಅವಧಿ (ಗಂಟೆಗಳು)	ಅಂಕಗಳು	ಆಂತರಿಕ ಮೌಲ್ಯಮಾಪನ ಅಂಕಗಳು	ಒಟ್ಟು ಅಂಕಗಳು
			ಕೋರ್ಸ್ ಕೋಡ್	ಕ್ರೆಡಿಟ್ ವಿನ್ಯಾಸ LTP	ಒಟ್ಟು ಕ್ರೆಡಿಟ್‌ಗಳು	ವಾರಕ್ಕೆ ಬೋಧನಾ ಗಂಟೆಗಳು				
ಮೊದಲನೆಯ ಚತುರ್ಮಾಸ										
1	1	ಘಟಕ 1 : ಪದ್ಯಭಾಗ ಘಟಕ 2 : ಗದ್ಯಭಾಗ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	ENA06001 CDA06001	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100
ಎರಡನೆಯ ಚತುರ್ಮಾಸ										
2	2	ಘಟಕ 1 : ನಾಟಕ ಘಟಕ 2 : ಕಾದಂಬರಿ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	ENB06001 CDB06001	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100
ಮೂರನೆಯ ಚತುರ್ಮಾಸ										
3	3	ಘಟಕ 1 : ಕಾವ್ಯ ಭಾಗ - ೧ ಘಟಕ 2 : ಕಥನ ಭಾಗ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	ENC06001 CDC06001	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100
ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸ										
4	4	ಘಟಕ 1 : ಕಾವ್ಯ ಭಾಗ - ೨ ಘಟಕ 2 : ಗದ್ಯ ಭಾಗ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	END06001 CDD06001	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100

ಭಾಷೆ ಕನ್ನಡ

ಸಿಬಿಸಿಎಸ್ ಪದ್ಧತಿಯ ಪಠ್ಯವಿನ್ಯಾಸ

2019-20ರಿಂದ ಆರಂಭವಾಗುವ ಶೈಕ್ಷಣಿಕ ವರ್ಷಕ್ಕೆ ಸಿಬಿಸಿಎಸ್ ಪದ್ಧತಿಯ ಪಾಠಕ್ರಮ ಮತ್ತು ಪರೀಕ್ಷಾ ಯೋಜನೆ
(ಎಲ್ಲಾ ನಿಕಾಯಗಳಿಗೂ ಅನ್ವಯವಾಗುವಂತೆ)

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್) ನಿರಂತರ
ಮೌಲ್ಯಮಾಪನ,

ಸಿ. - 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ - 2 (15 ಅಂಕಗಳು) ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

BSC /BCA (AECC)

ಸೆಮಿಸ್ಟರ್	ಪತ್ರಿಕೆ	ಪತ್ರಿಕೆ ಶೀರ್ಷಿಕೆ	ಪರೀಕ್ಷಾ ಅವಧಿ (ಗಂಟೆಗಳು)				ಅಂಕಗಳು	ಆಂತರಿಕ ಮೌಲ್ಯಮಾಪನ ಅಂಕಗಳು	ಒಟ್ಟು ಅಂಕಗಳು	
			ಕೋರ್ಸ್ ಕೋಡ್	ಕ್ರೆಡಿಟ್ ವಿನ್ಯಾಸ LTP	ಒಟ್ಟು ಕ್ರೆಡಿಟ್‌ಗಳು	ವಾರಕ್ಕೆ ಬೋಧನಾ ಗಂಟೆಗಳು				
ಮೊದಲನೆಯ ಚತುರ್ಮಾಸ										
1	1	ಘಟಕ 1 : ಪದ್ಯಭಾಗ ಘಟಕ 2 : ಗದ್ಯಭಾಗ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	DMA06001 DMA06002 DMA06003 DMA06004 DMA06005 DMA06006 DMA06007 DMA06008 ECA06001	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 - 15)	100
ಎರಡನೆಯ ಚತುರ್ಮಾಸ										
2	2	ಘಟಕ 1 : ನಾಟಕ ಘಟಕ 2 : ಕಾದಂಬರಿ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ	DMB06001 DMB06002 DMB06003 DMB06004 DMB06005 DMB06006 DMB06007 DMB06008 ECB06001	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 - 15)	100

ಮೂರನೆಯ ಚತುರ್ಮಾಸ

3	3	ಘಟಕ 1 : ಕಾವ್ಯ ಭಾಗ - ಂ ಘಟಕ 2 : ಕಥನ ಭಾಗ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ } ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ }	DMC06001 DMC06002 DMC06003 DMC06004 DMC06005 DMC06006 DMC06007 DMC06008 ECC06001	01 01 01 2:1:0	03	1 1 2 04	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 - 15)	100
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ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸ

4	4	ಘಟಕ 1 : ಕಾವ್ಯ ಭಾಗ - ಂ ಘಟಕ 2 : ಗದ್ಯ ಭಾಗ ಘಟಕ 3 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ } ಘಟಕ 4 : ಪಠ್ಯಧಾರಿತಚಟುವಟಿಕೆ }	DMD06001 DMD06002 DMD06003 DMD06004 DMD06005 DMD06006 DMD06007 DMD06008 ECD06001	01 01 01 2:1:0	03	1 1 2	03	40 30 - 70	30 (ಸಿ-1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
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ಭಾಷೆ ಕನ್ನಡ

ಸಿಬಿಸಿಎಸ್ ಪದ್ಧತಿಯ ಪಠ್ಯವಿನ್ಯಾಸ

2019-2020ರ ಅರಂಭವಾಗುವ ಶೈಕ್ಷಣಿಕ ವರ್ಷಕ್ಕೆ

ಸಿಬಿಸಿಎಸ್ ಪದ್ಧತಿಯ ಪಾಠಕ್ರಮ ಮತ್ತು ಪರೀಕ್ಷಾ ಯೋಜನೆ

(ಎಲ್ಲಾ ನಿಕಾಯಗಳಿಗೂ ಅನ್ವಯವಾಗುವಂತೆ)

ಎಸ್.ಇ.ಸಿ (ಕೌಶಲ್ಯಾಭಿವೃದ್ಧಿ ಪತ್ರಿಕೆ) ಎಸ್ ಇ ಸಿ - (S.E.C - Skill Enhancement Course)

ಕನ್ನಡ ಸಂವಹನ ಕೌಶಲ : (Kannada Communication Skill)

ಮೂರನೆಯ ಚತುರ್ಮಾಸ(BA)

ಸೆ ಮಿಸ್ಟರ್	ಪತ್ರಿಕೆ	ಪತ್ರಿಕೆ ಶೀರ್ಷಿಕೆ					ಪರೀಕ್ಷಾ ಅವಧಿ (ಗಂಟೆಗಳು)	ಅಂಕ ಗಳು	ಆಂತರಿಕ ಮೌಲ್ಯಮಾಪ ನ ಅಂಕಗಳು	ಒಟ್ಟು ಅಂಕಗಳು
			ಕೋರ್ಸ್ ಕೋಡ್	ಕ್ರೆಡಿಟ್ ವಿನ್ಯಾಸ LTP	ಒಟ್ಟು ಕ್ರೆಡಿಟ್	ಬೋಧನಾ ಗಂಟೆಗಳು ಸೆಮಿಸ್ಟರ್ಗೆ				
3	1	ಘಟಕ 1 : ಮಾತಿನ ಕೌಶಲ ಕೇಳುವಿಕೆ ಕೌಶಲ ಘಟಕ 2 : ಓದಿನ ಕೌಶಲ ಬರವಣಿಗೆ ಕೌಶಲ ಘಟಕ 3 : ಪತ್ರ ವ್ಯವಹಾರ	ELC290	01	02	16	02	10		
				01		16		10		
				2:0:0				15		
						32		35	15	50

ಭಾಷೆ ಕನ್ನಡ
ಮೊದಲನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)

Course Code: ELA06025/ ELA06021/ELA06022/ELA06023/ELA06024 /ELA06026

ಕಲಾನಿಕಾಯ

ಬಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 1,

2 +1 = 3 ಕ್ರೆಡಿಟ್

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ - ಕಲಾ ಸಿರಿ - 1

ಪತ್ರಿಕೆ : 1 ಪದ್ಯಭಾಗ ಮತ್ತು ಕಥಾಭಾಗ

ಘಟಕ 1 : ಪದ್ಯಭಾಗ

ಅಂಕಗಳು :

40

1. ಇಂದಿನ ದೇವರು : ಕುವೆಂಪು
2. ಬೆಳಗು : ದ.ರಾ. ಬೇಂದ್ರೆ
3. ಏನಾದರೂ ಮಾಡುತ್ತಿರು ತಮ್ಮ : ಎಂ. ಗೋಪಾಲಕೃಷ್ಣ ಅಡಿಗ
4. ಮುಂಬೈ ಜಾತಕ : ಜಿ. ಎಸ್. ಶಿವರುದ್ರಪ್ಪ
5. ವ೦ದನೆ : ಸಿ.ಪಿ.ಕೆ
5. ಕುಂತಿಯ ಬೆಳಗು : ವಿಜಯಾದಿಪ್ಪೆ
6. ಅಕ್ಕಾರಾ ಅಕ್ಕಾರವೇ : ಎಲ್. ಹನುಮಂತಯ್ಯ
7. ಈ ರಾತ್ರಿ ಬುದ್ಧನಿಗಾಗಿ : ಅರಿಘ್‌ರಾಜ

ಘಟಕ 2 : ಕಥಾಭಾಗ

ಅಂಕಗಳು :

30

1. ವೆಂಕಟಶಾಮಿಯ ಪ್ರಣಯ : ಮಾಸ್ತಿ ವೆಂಕಟೇಶ ಅಯ್ಯಂಗಾರ್
2. ಸುಶ್ರುತ ಮತ್ತು ಹಕ್ಕಿ ಮರಿ : ಕೆ.ಪಿ. ಪೂರ್ಣಚಂದ್ರ ತೇಜಸ್ವಿ
3. ಗಿರಿಜವ್ವನ ರೊಟ್ಟಿ : ಅನಕೃ
4. ಮಾರಿಕೊಂಡವರು : ದೇವನೂರ ಮಹದೇವ
5. ನಿರಾಕರಣೆ : ವೀಣಾ ಶಾಂತೇಶ್ವರ

ಘಟಕ : 3 ಪದ್ಯಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಕವಿತೆಗಳನ್ನು ಭಾವ, ಅರ್ಥ ತಿಳಿಯುವಂತೆ ಓದಿಸುವುದು
- ಕವಿತೆಗಳಲ್ಲಿರುವ ವಿಶೇಷ ಪದ, ನುಡಿಗಟ್ಟು, ಗಾದೆ - ಇವುಗಳನ್ನು ಗುರುತಿಸಿ ಸ್ವಂತ ವಾಕ್ಯಗಳಲ್ಲಿ ಬಳಸಲು ಸೂಚಿಸುವುದು.
- ಕವಿತೆಗಳ ವಸ್ತು ಆಶಯಗಳನ್ನು ಪ್ರಸ್ತುತಕ್ಕೆ ಅನ್ವಯಿಸಿ ವಿಚಾರಮಂಡನೆ, ಗುಂಪು ಚರ್ಚೆ ಏರ್ಪಡಿಸುವುದು.
- ಕವಿತೆಗಳಿಗೆ ರಾಗ ಸಂಯೋಜಿಸಿ ಹಾಡಲು ಉತ್ತೇಜಿಸುವುದು, ಬಹುಮಾಧ್ಯಮಗಳನ್ನು ಬಳಸಿಕೊಳ್ಳುವುದು.
- ಕವಿತೆಗಳ ವಸ್ತು ಮತ್ತು ಆಶಯಗಳನ್ನು ಸೃಜನಾತ್ಮಕವಾಗಿ (ಕವಿತೆ ರಚನೆ, ಕಥನರೂಪ. ನಾಟಕ ರೂಪ, ಸ್ವಾನುಭವ ಕಥನ, ಚಿತ್ರಕಲೆ, ಮಣ್ಣಿನ ಮಾದರಿ, ಛಾಯಾಚಿತ್ರ ಇತ್ಯಾದಿ..) ಪರಿವರ್ತಿಸುವಂತೆ ಸೂಚಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳನ್ನು ಗುಂಪುಗಳನ್ನಾಗಿ ವಿಂಗಡಿಸಿ ಸಂಬಂಧಿಸಿದ ಚಟುವಟಿಕೆಗಳನ್ನು ನಡೆಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳಿಗೆ ಸಂಬಂಧಿಸಿದ ದಾಖಲೆಗಳನ್ನು ಸಂಗ್ರಹಿಸಿಡುವಂತೆ ಸೂಚಿಸುವುದು.

ಘಟಕ : 4 ಕಥಾಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಕಥಾಭಾಗಗಳ ಭಾವ, ಅರ್ಥ ತಿಳಿಯುವಂತೆ ಓದಿಸುವುದು
- ವಿಶೇಷ ಪದ, ನುಡಿಗಟ್ಟು, ಗಾದೆ - ಇವುಗಳನ್ನು ಗುರುತಿಸಿ ಸ್ವಂತ ವಾಕ್ಯಗಳಲ್ಲಿ ಬಳಸಲು ಸೂಚಿಸುವುದು.
- ವಸ್ತು ಆಶಯಗಳನ್ನು ಪ್ರಸ್ತುತಕ್ಕೆ ಅನ್ವಯಿಸಿ ವಿಚಾರಮಂಡನೆ, ಗುಂಪುಚರ್ಚೆ ಏರ್ಪಡಿಸುವುದು.
- ವಸ್ತು ಮತ್ತು ಆಶಯಗಳನ್ನು ಸೃಜನಾತ್ಮಕವಾಗಿ (ಕವಿತೆ ರಚನೆ, ಕಥನರೂಪ. ನಾಟಕ ರೂಪ, ಸ್ವಾನುಭವ ಕಥನ, ಚಿತ್ರಕಲೆ, ಮಣ್ಣಿನ ಮಾದರಿ, ಛಾಯಾಚಿತ್ರ ಇತ್ಯಾದಿ..) ಪರಿವರ್ತಿಸುವಂತೆ ಸೂಚಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳನ್ನು ಗುಂಪುಗಳನ್ನಾಗಿ ವಿಂಗಡಿಸಿ ಸಂಬಂಧಿಸಿದ ಚಟುವಟಿಕೆಗಳನ್ನು ನಡೆಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳಿಗೆ ಸಂಬಂಧಿಸಿದ ದಾಖಲೆಗಳನ್ನು ಸಂಗ್ರಹಿಸಿಡುವಂತೆ ಸೂಚಿಸುವುದು

ಭಾಷೆ ಕನ್ನಡ
ಎರಡನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)
Course Code: ELB06021/ELB06025 /ELB06022/ELB06023/ELB06024/ ELB06026

ಕಲಾನಿಕಾಯ

ಬಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 2,

2 +1 = 3

ಕ್ರೆಡಿಟ್

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ - ಕಲಾ ಸಿರಿ - 2

ಪತ್ರಿಕೆ : 2 ನಾಟಕ ಮತ್ತು ಕಾದಂಬರಿ

ಘಟಕ - 1

ಅಂಕಗಳು : 40

ಪಠ್ಯ - ನಾಟಕ : ಅಶ್ವತ್ಥಾಮನ್

ಲೇಖಕರು :ಬಿ. ಎಂ. ಶ್ರೀ

ಘಟಕ - 2

ಅಂಕಗಳು : 30

ಪಠ್ಯ - ಕಾದಂಬರಿ : ಚೋಮನ ದುಡಿ

ಲೇಖಕರು : ಶಿವರಾಮ ಕಾರಂತ

ಘಟಕ - 3 : ನಾಟಕ ಭಾಗದ ಪಠ್ಯ ಪೂರಕ ಚಟುವಟಿಕೆಗಳು

- ನಾಟಕದ ಆಯ್ದ ಭಾಗವನ್ನು ಧ್ವನಿ ಏರಿಳಿತಗಳೊಂದಿಗೆ ಓದಿಸುವುದು
- ಭಾಷಾ ವೈಶಿಷ್ಟ್ಯವನ್ನು ಗುರುತಿಸುವುದು.
- ನಾಟಕದ ಆಶಯವನ್ನು ಪ್ರಸ್ತುತಕ್ಕೆ ಅನ್ವಯಿಸಿ - ವಿಚಾರ ಮಂಡನೆ, ಗುಂಪು ಚರ್ಚೆ
- ಪಠ್ಯವನ್ನು ರಂಗರೂಪಕ್ಕೆ ತರುವ ಸಾಧ್ಯತೆಗಳ ಕುರಿತ ಚರ್ಚೆ
- ನಾಟಕದ ಸಂವಿಧಾನ, ತಂತ್ರ ಕುರಿತ ವಿಶ್ಲೇಷಣೆ
- ನಾಟಕದ ಭಾಗಗಳನ್ನು ಮತ್ತು ಪಾತ್ರಗಳನ್ನು ಅಭಿನಯಿಸುವಂತೆ ವಿದ್ಯಾರ್ಥಿಗಳನ್ನು ಉತ್ತೇಜಿಸುವುದು.
- ನಾಟಕಕಾರರ ಬದುಕು, ಸಾಹಿತ್ಯಕ ಸಾಧನೆ ಕುರಿತ ಮಾಹಿತಿ ಸಂಗ್ರಹಿಸುವುದು.
- ನಾಟಕದ ವಸ್ತು ಮತ್ತು ಆಶಯವನ್ನು ಸೃಜನಾತ್ಮಕವಾಗಿ ಮರು ರೂಪಿಸಲು (ಕವಿತೆ ರಚನೆ, ಪ್ರಹಸನ, ಸ್ವಾನುಭವ ಕಥನ, ಚಿತ್ರಕಲೆ, ಛಾಯಾಚಿತ್ರ ಇತ್ಯಾದಿ) ಸೂಚಿಸುವುದು
- ಇದಕ್ಕೆ ಸಂಬಂಧಿಸಿದ ವಿಡಿಯೋ ಪ್ರದರ್ಶನ

ಘಟಕ - 4 : ಕಾದಂಬರಿ ಭಾಗದ ಪಠ್ಯಪೂರಕ ಚಟುವಟಿಕೆಗಳು

- ಕಾದಂಬರಿಯ ಮುಖ್ಯ ಭಾಗಗಳ ವಾಚನ
- ಪ್ರಾದೇಶಿಕ ಭಾಷಾ ಪ್ರಭೇದ ನುಡಿಗಟ್ಟು, ಗಾದೆ ಇತ್ಯಾದಿ ಭಾಷಾ ವೈಶಿಷ್ಟ್ಯಗಳನ್ನು ಗುರುತಿಸುವುದು.
- ಆಶಯಕ್ಕೆ ಸಂಬಂಧಿಸಿದಂತೆ ಚರ್ಚೆ ಏರ್ಪಡಿಸುವುದು.
- ಕಾದಂಬರಿಗೆ ಸಂಬಂಧಿಸಿದಂತೆ ವಿಚಾರ ಮಂಡನೆಗೆ ಅವಕಾಶ ನೀಡುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಓದಿದ ಇನ್ನಿತರ ಕಾದಂಬರಿಗಳ ಕುರಿತು ಅಭಿಪ್ರಾಯ ಮಂಡನೆಗೆ ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.
- ಕಾದಂಬರಿ ಆಧರಿಸಿದ ಸಿನಿಮಾಗಳಿದ್ದರೆ ಅವುಗಳನ್ನು ಪ್ರದರ್ಶಿಸುವುದು.

ಭಾಷೆ ಕನ್ನಡ
ಮೂರನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)

Course Code: ELC06025/ ELC06021/ELC0602/2ELC06023/ELC06024/ ELC06026

ಕಲಾನಿಕಾಯ

ಬಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 3,

2 +1 = 3

ಕ್ರೆಡಿಟ್ ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್‌ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ : ಕಲಾ ಸಿರಿ - 3

ಪತ್ರಿಕೆ : 3 ಕಾವ್ಯ ಭಾಗ -1

ಘಟಕ 1 : ಕಾವ್ಯಭಾಗ

ಅಂಕಗಳು :

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1. ಅಣ್ಣ ತಂಗಿ : ಜನಪದ
2. ವಚನಗಳು : ಬಸವಣ್ಣ, ಆಯ್ದಕ್ಕೆ ಲಕ್ಕಮ್ಮ
ಮಾದಾರ ಚೆನ್ನಯ್ಯ, ಅಕ್ಕಮ್ಮ
3. ಚಂದ್ರಮತಿಯ ಪ್ರಲಾಪ : ರಾಘವಾಂಕ
4. ಮುಸುಡನಮರಾದ್ರಿಯಲಿ ತೇವೆನು
ದೇವಸಂತತಿಯ : ಕುಮಾರವ್ಯಾಸ
5. ನೃಪಾಲನ ಗರ್ವ ಸೋರಿದುದು : ರತ್ನಾಕರವರ್ಣಿ
6. ಎಳೆಗರುಂ ಎತ್ತಾಗದೆ : ಸೋಮೇಶ್ವರ

ಘಟಕ 2 : ಲಕ್ಷ್ಮೀಶನ ಜೈಮಿನಿ ಭಾರತ(ಆಯ್ದ ಭಾಗ)

ಅಂಕಗಳು : 30

ಘಟಕ : 3 ಕಾವ್ಯ ಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಜನಪದ ಗೀತೆಗಳ ಮೂಲಮಟ್ಟಗಳನ್ನು ಪರಿಚಯಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳ ಪರಿಸರದಲ್ಲಿರುವ ಜನಪದ ಸಾಹಿತ್ಯವನ್ನು ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು
- ಷಟ್ಪದಿಕಾವ್ಯಗಳನ್ನು ಗಮಕ ರೂಪದಲ್ಲಿ ಪ್ರಸ್ತುತ ಪಡಿಸುವುದು.
- ಆಯ್ದ ಕಾವ್ಯಭಾಗದ ನಾಟಕೀಯ ಸನ್ನಿವೇಶಗಳನ್ನು ಅಭಿನಯಿಸಲು ಉತ್ತೇಜಿಸುವುದು.
- ಕಾವ್ಯದ ಭಾಗವನ್ನು ನಾಟಕ ಅಥವಾ ಕಥೆಯ ರೂಪಕ್ಕೆ ಪರಿವರ್ತಿಸಲು ಸೂಚಿಸುವುದು.
- ಆಯ್ದ ವಚನಕಾರರ ಬದುಕು ಮತ್ತು ಸಾಹಿತ್ಯ ಕುರಿತಂತೆ ವಿಚಾರಗೋಷ್ಠಿ ನಡೆಸುವುದು.
- ಕಾವ್ಯಭಾಗದ ಕಥೆಗೆ ಸಂಬಂಧಿಸಿದ ಸಿನಿಮಾ ಪ್ರದರ್ಶನ ಹಾಗೂ ಗಾಯನಗಳನ್ನು ಕೇಳಿಸುವುದು.

ಭಾಷೆ ಕನ್ನಡ
ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)

Course Code: ELD06025/ELD06021/ELD06022/ELD06023/ELD06024/ ELD06026

ಕಲಾನಿಕಾಯ

ಬಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 4

2 +1 = 3

ಕ್ರೆಡಿಟ್ ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ - ಕಲಾ ಸಿರಿ - 4

ಪತ್ರಿಕೆ : 4 ಕಾವ್ಯ ಭಾಗ - 2 ಮತ್ತು ಗದ್ಯಭಾಗ

ಘಟಕ 1 : ಕಾವ್ಯಭಾಗ - 2

ಅಂಕಗಳು :

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1. ನೀಲಾಂಜನೆಯ ನೃತ್ಯ0 : ಪಂಪ
2. ದುರ್ಯೋಧನ ವಿಲಾಪಂ : ರನ್ನ
3. ನಳಕೂಬರನ ಪ್ರಸಂಗ : ನಾಗಚಂದ್ರ
4. ಕೊಲ್ವೊಡೆ ಕಾವರಾರೇಲವೊ : ರುದ್ರಭಟ್ಟ
5. ತುಪ್ಪೇರಿದ ದರ್ಪಣದೊಳ್ ಪಜ್ಜಳಿಸಲಾರ್ಪದೆ ಪ್ರತಿಬಿಂಬಂ : ಜನ್ನ
6. ತುರುಕಾಣ್ಯದಯ್ಯನ ವಚನ (ಮಂಟೇಸ್ವಾಮಿ ಕಾವ್ಯ) : ಸಂ. ಪಿ. ಕೆ. ರಾಜಶೇಖರ

ಘಟಕ 2 : ಗದ್ಯಭಾಗ

ಅಂಕಗಳು :

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1. ತ್ಸುನಾಮಿ : ಪೂರ್ಣಚಂದ್ರ ತೇಜಸ್ವಿ
2. ಆಸ್ಪತ್ರೆ : ಎಚ್.ಎಸ್.ಕೆ
3. ಉಪಘಾತ : ಅ.ರಾ. ಮಿತ್ರ
4. ಬಾಗಿಲಿಗೆ ಬಂದ ನವಿಲು ; ಕೃಪಾಕರ ಸೇನಾನಿ
5. ಬದುಕೆಂಬ ಅಗ್ನಿದಿವ್ಯ : ಕೆ. ಷರೀಫಾ

ಘಟಕ : 3 ಕಾವ್ಯ ಭಾಗ - 2 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಹಳಗನ್ನಡ ಪದ್ಯಗಳನ್ನು ಅರ್ಥಕ್ಕೆ ಅನುಗುಣವಾಗಿ ಬಿಡಿಸಿ ಓದುವುದು
- ಹಳಗನ್ನಡ ಪದ್ಯಗಳಲ್ಲಿ ಬರುವ ಕಠಿಣ/ವಿಶೇಷ ಪದಗಳಿಗೆ ಅರ್ಥಬರೆಯಲು ಸೂಚಿಸುವುದು ಮತ್ತು ನುಡಿಗಟ್ಟುಗಳನ್ನು ವಿವರಿಸಲು ತಿಳಿಸುವುದು.
- ವಸ್ತು ಮತ್ತು ಆಶಯಗಳನ್ನು ಪ್ರಸ್ತುತ ಸಂದರ್ಭಕ್ಕೆ ತಕ್ಕಂತೆ ಅನ್ವಯಿಸಿ ವಿಚಾರ ಮಂಡಿಸಲು ಗುಂಪು ಚರ್ಚೆ ನಡೆಸಲು ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.
- ಹಳಗನ್ನಡ ಕವಿತೆಗೆ ಗಮಕ ವ್ಯಾಖ್ಯಾನ ಮಾಡಿಸುವುದು..
- ಹಸ್ತಪ್ರತಿಗಳನ್ನು ತೋರಿಸುವುದು.

ಘಟಕ : 4 ಗದ್ಯಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಪರಿಸರ ಸಂರಕ್ಷಣೆಯಲ್ಲಿ ವಿದ್ಯಾರ್ಥಿಯು ತನ್ನ ಪಾತ್ರ ಕುರಿತು ತಿಳಿಯ ಪಡಿಸುವುದು.
- ಪರಿಸರ ಸಂರಕ್ಷಣೆ ಕುರಿತು ಚರ್ಚೆ ಮತ್ತು ಅಭಿಪ್ರಾಯ ಮಂಡನೆ
- ವಿದ್ಯಾರ್ಥಿಗಳಿಂದ ಅವರ ಜೀವನದ ಅನುಭವ ಕುರಿತು ಲೇಖನಗಳನ್ನು ಬರೆಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಯು ತಾನು ಕೈಗೊಂಡ ಪ್ರವಾಸದ ಅನುಭವಗಳನ್ನು ಅಭಿವ್ಯಕ್ತಪಡಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ತಾವು ತಿಳಿದಿರುವ ಹಾಸ್ಯ ಸನ್ನಿವೇಶಗಳನ್ನು ವರ್ಣಿಸುವುದು.
- ಕಲಾಸಾಧಕರನ್ನು ಗುರುತಿಸಿ ಪರಿಚಯಿಸಲು ಸೂಚಿಸುವುದು.
- ತಮ್ಮ ಪರಿಸರದಲ್ಲಿರುವ ವಿಶೇಷ ವ್ಯಕ್ತಿಗಳನ್ನು (ಸಾಹಿತಿ, ಕಲಾವಿದ, ವಿಜ್ಞಾನಿ, ಪರಿಸರ ಪ್ರೇಮಿ ಇತ್ಯಾದಿ..) ಸಂದರ್ಶಿಸುವುದು.

ಭಾಷೆ ಕನ್ನಡ
ಮೊದಲನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)
Course Code: CDA06001/ ENA06001

ವಾಣಿಜ್ಯ ನಿಕಾಯ
ಬಿ.ಕಾಂ/ ಬಿ.ಬಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 1,

2 +1 = 3

ಕ್ರೆಡಿಟ್ ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪಠ್ಯಪೂರಕ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ : ವಾಣಿಜ್ಯ ಸಿರಿ - 1

ಪತ್ರಿಕೆ : 1 ಪದ್ಯಭಾಗ ಮತ್ತು ಕಥಾಭಾಗ

ಘಟಕ 1 : ಪದ್ಯಭಾಗ

ಅಂಕಗಳು :

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1. ಪ್ರಾರ್ಥನೆ : ಬಿ.ಎಂ. ಶ್ರೀ
2. ನೀಂ ಮಹಚ್ಛಿಲ್ಪಿ ದಿಟಂ : ಕುವೆಂಪು
3. ಕುರುಡು ಕಾಂಚಾಣ : ದ.ರಾ. ಬೇಂದ್ರೆ
4. ರೈಲ್ವೇ ನಿಲ್ದಾಣದಲ್ಲಿ : ಕೆ. ಎಸ್. ನರಸಿಂಹಸ್ವಾಮಿ
5. ಖಾಲಿ ಸೈಟುಗಳು : ಕೆ. ಎಸ್. ನಿಸಾರ್ ಅಹಮದ್
6. ಅಡುಗೆ ಮನೆಯ ಹುಡುಗಿ : ವೈದೇಹಿ
7. ಅಪೂರ್ವ : ಹೇಮಾ ಪಟ್ಟಣಶೆಟ್ಟಿ
8. ನಾನು ಮತ್ತು ಚಪ್ಪಲಿಗಳು : ಮೂಡ್ಲಾಕೂಡು ಚನ್ನಸ್ವಾಮಿ

ಘಟಕ 2 : ಕಥಾಭಾಗ

ಅಂಕಗಳು :

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1. ಸುಗ್ಗಿ : ಬೆನಗರಹಳ್ಳಿ ರಾಮಣ್ಣ
2. ಕಪ್ಪುಹುಡುಗನ ಹಾಡು : ಅಬ್ದುಲ್ ರಶೀದ್
3. ರೊಟ್ಟಿ : ಪಿ. ಲಂಕೇಶ್
4. ದೇವರ ಹೆಣ : ಕುಂ. ವೀರಭದ್ರಪ್ಪ
5. ಮುನ್ನಾದಿನ : ಕೊಡಗಿನ ಗೌರಮ್ಮ

ಘಟಕ : 3 ಪದ್ಯಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಕವಿತೆಗಳನ್ನು ಭಾವ, ಅರ್ಥ ತಿಳಿಯುವಂತೆ ಓದಿಸುವುದು
- ಕವಿತೆಗಳಲ್ಲಿರುವ ವಿಶೇಷ ಪದ, ನುಡಿಗಟ್ಟು, ಗಾದೆ - ಇವುಗಳನ್ನು ಗುರುತಿಸಿ ಸ್ವಂತ ವಾಕ್ಯಗಳಲ್ಲಿ ಬಳಸಲು ಸೂಚಿಸುವುದು.
- ಕವಿತೆಗಳ ವಸ್ತು ಆಶಯಗಳನ್ನು ಪ್ರಸ್ತುತಕ್ಕೆ ಅನ್ವಯಿಸಿ ವಿಚಾರಮಂಡನೆ, ಗುಂಪುಚರ್ಚೆ ಏರ್ಪಡಿಸುವುದು.
- ಕವಿತೆಗಳಿಗೆ ರಾಗ ಸಂಯೋಜಿಸಿ ಹಾಡಲು ಉತ್ತೇಜಿಸುವುದು, ಬಹುಮಾಧ್ಯಮಗಳನ್ನು ಬಳಸಿಕೊಳ್ಳುವುದು.
- ಕವಿತೆಗಳ ವಸ್ತು ಮತ್ತು ಆಶಯಗಳನ್ನು ಸೃಜನಾತ್ಮಕವಾಗಿ (ಕವಿತೆ ರಚನೆ, ಕಥನರೂಪ. ನಾಟಕ ರೂಪ, ಸ್ವಾನುಭವ ಕಥನ, ಚಿತ್ರಕಲೆ, ಮಣ್ಣಿನ ಮಾದರಿ, ಛಾಯಾಚಿತ್ರ ಇತ್ಯಾದಿ..) ಪರಿವರ್ತಿಸುವಂತೆ ಸೂಚಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳನ್ನು ಗುಂಪುಗಳನ್ನಾಗಿ ವಿಂಗಡಿಸಿ ಸಂಬಂಧಿಸಿದ ಚಟುವಟಿಕೆಗಳನ್ನು ನಡೆಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳಿಗೆ ಸಂಬಂಧಿಸಿದ ದಾಖಲೆಗಳನ್ನು ಸಂಗ್ರಹಿಸಿಡುವಂತೆ ಸೂಚಿಸುವುದು.

ಘಟಕ : 4 ಕಥಾಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಕಥಾಭಾಗಗಳ ಭಾವ, ಅರ್ಥ ತಿಳಿಯುವಂತೆ ಓದಿಸುವುದು
- ವಿಶೇಷ ಪದ, ನುಡಿಗಟ್ಟು, ಗಾದೆ - ಇವುಗಳನ್ನು ಗುರುತಿಸಿ ಸ್ವಂತ ವಾಕ್ಯಗಳಲ್ಲಿ ಬಳಸಲು ಸೂಚಿಸುವುದು.
- ವಸ್ತು ಆಶಯಗಳನ್ನು ಪ್ರಸ್ತುತಕ್ಕೆ ಅನ್ವಯಿಸಿ ವಿಚಾರಮಂಡನೆ, ಗುಂಪುಚರ್ಚೆ ಏರ್ಪಡಿಸುವುದು.
- ವಸ್ತು ಮತ್ತು ಆಶಯಗಳನ್ನು ಸೃಜನಾತ್ಮಕವಾಗಿ (ಕವಿತೆ ರಚನೆ, ಕಥನರೂಪ. ನಾಟಕ ರೂಪ, ಸ್ವಾನುಭವ ಕಥನ, ಚಿತ್ರಕಲೆ, ಮಣ್ಣಿನ ಮಾದರಿ, ಛಾಯಾಚಿತ್ರ ಇತ್ಯಾದಿ..) ಪರಿವರ್ತಿಸುವಂತೆ ಸೂಚಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳನ್ನು ಗುಂಪುಗಳನ್ನಾಗಿ ವಿಂಗಡಿಸಿ ಸಂಬಂಧಿಸಿದ ಚಟುವಟಿಕೆಗಳನ್ನು ನಡೆಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳಿಗೆ ಸಂಬಂಧಿಸಿದ ದಾಖಲೆಗಳನ್ನು ಸಂಗ್ರಹಿಸಿಡುವಂತೆ ಸೂಚಿಸುವುದು

ಭಾಷೆ ಕನ್ನಡ

Course Code: CDB06001/ ENB06001

ಎರಡನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)

ವಾಣಿಜ್ಯ ನಿಕಾಯ

ಬಿ.ಕಾಂ/ಬಿ.ಬಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 2,

2 +1 = 3

ಕ್ರೆಡಿಟ್

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ : ವಾಣಿಜ್ಯ ಸಿರಿ - 2

ಪತ್ರಿಕೆ : 2 ನಾಟಕ ಮತ್ತು ಕಾದಂಬರಿ

ಘಟಕ - 1

ಅಂಕಗಳು : 40

ಪಠ್ಯ - ನಾಟಕ : ಅಂತಿಗೊನೆ

ಲೇಖಕರು : ಪಿ. ಲಂಕೇಶ್

ಘಟಕ - 2

ಅಂಕಗಳು : 30

ಪಠ್ಯ - ಕಾದಂಬರಿ : ಕರ್ವಾಲೋ

ಲೇಖಕರು : ಪೂರ್ಣಚಂದ್ರ ತೇಜಸ್ವಿ

ಘಟಕ - 3 : ನಾಟಕ ಭಾಗದ ಪಠ್ಯ ಪೂರಕ ಚಟುವಟಿಕೆಗಳು

- ನಾಟಕದ ಆಯ್ದ ಭಾಗವನ್ನು ಧ್ವನಿ ಏರಿಳಿತಗಳೊಂದಿಗೆ ಓದಿಸುವುದು
- ಭಾಷಾ ವೈಶಿಷ್ಟ್ಯವನ್ನು ಗುರುತಿಸುವುದು.
- ನಾಟಕದ ಆಶಯವನ್ನು ಪ್ರಸ್ತುತಕ್ಕೆ ಅನ್ವಯಿಸಿ - ವಿಚಾರ ಮಂಡನೆ, ಗುಂಪು ಚರ್ಚೆ
- ಪಠ್ಯವನ್ನು ರಂಗರೂಪಕ್ಕೆ ತರುವ ಸಾಧ್ಯತೆಗಳ ಕುರಿತ ಚರ್ಚೆ
- ನಾಟಕದ ಸಂವಿಧಾನ, ತಂತ್ರ ಕುರಿತ ವಿಶ್ಲೇಷಣೆ
- ನಾಟಕದ ಭಾಗಗಳನ್ನು ಮತ್ತು ಪಾತ್ರಗಳನ್ನು ಅಭಿನಯಿಸುವಂತೆ ವಿದ್ಯಾರ್ಥಿಗಳನ್ನು ಉತ್ತೇಜಿಸುವುದು.
- ನಾಟಕಕಾರರ ಬದುಕು, ಸಾಹಿತ್ಯಕ ಸಾಧನೆ ಕುರಿತ ಮಾಹಿತಿ ಸಂಗ್ರಹಿಸುವುದು.
- ನಾಟಕದ ವಸ್ತು ಮತ್ತು ಆಶಯವನ್ನು ಸೃಜನಾತ್ಮಕವಾಗಿ ಮರು ರೂಪಿಸಲು (ಕವಿತೆ ರಚನೆ, ಪ್ರಹಸನ, ಸ್ವಾನುಭವ ಕಥನ, ಚಿತ್ರಕಲೆ, ಛಾಯಾಚಿತ್ರ ಇತ್ಯಾದಿ) ಸೂಚಿಸುವುದು
- ಇದಕ್ಕೆ ಸಂಬಂಧಿಸಿದ ವಿಡಿಯೋ ಪ್ರದರ್ಶನ

ಘಟಕ - 4 : ಕಾದಂಬರಿ ಭಾಗದ ಪಠ್ಯಪೂರಕ ಚಟುವಟಿಕೆಗಳು

- ಕಾದಂಬರಿಯ ಮುಖ್ಯ ಭಾಗಗಳ ವಾಚನ
- ಭಾಷಾ ವೈಶಿಷ್ಟ್ಯಗಳನ್ನು ಗುರುತಿಸುವುದು (ಪ್ರಾದೇಶಿಕ ಭಾಷಾ ಪ್ರಭೇದ ನುಡಿಗಟ್ಟು, ಗಾದೆ ಇತ್ಯಾದಿ.)
- ಆಶಯಕ್ಕೆ ಸಂಬಂಧಿಸಿದಂತೆ ಚರ್ಚೆ ಏರ್ಪಡಿಸುವುದು.
- ಕಾದಂಬರಿಗೆ ಸಂಬಂಧಿಸಿದಂತೆ ವಿಚಾರ ಮಂಡನೆಗೆ ಅವಕಾಶ ನೀಡುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಓದಿದ ಇನ್ನಿತರ ಕಾದಂಬರಿಗಳ ಕುರಿತು ಅಭಿಪ್ರಾಯ ಮಂಡನೆಗೆ ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.
- ಕಾದಂಬರಿ ಆಧರಿಸಿದ ಸಿನಿಮಾಗಳಿದ್ದರೆ ಅವುಗಳನ್ನು ಪ್ರದರ್ಶಿಸುವುದು.

ಭಾಷೆ ಕನ್ನಡ
ಮೂರನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)
Course Code: CDC06001/ ENCO6001
ವಾಣಿಜ್ಯ ನಿಕಾಯ
ಬಿ.ಕಾಂ / ಬಿ.ಬಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 3,

2 +1 = 3

ಕ್ರೆಡಿಟ್

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ : ವಾಣಿಜ್ಯ ಸಿರಿ - 3

ಪತ್ರಿಕೆ : 3 ಕಾವ್ಯ ಭಾಗ -1

ಘಟಕ 1 : ಕಾವ್ಯಭಾಗ

ಅಂಕಗಳು :

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1. ವಚನಗಳು : ಜೇಡರ ದಾಸಿಮಯ್ಯ, ಅಲ್ಲಮ ಪ್ರಭು
ಅಮುಗೆ ರಾಯಮ್ಮ, ಲಿಂಗಮ್ಮ
2. ಸರ್ವಜ್ಞನ ತ್ರಿಪದಿಗಳು : ಸರ್ವಜ್ಞ
3. ಹರನಂಬುದೇ ಸತ್ಯ ಸತ್ಯವೆಂಬುದೇ ಹರನು : ರಾಘವಾಂಕ
4. ಮಾನವ ಜನ್ಮ : ಪುರಂದರ ದಾಸರು
5. ಧರ್ಮಯುದ್ಧ : ರತ್ನಾಕರವರ್ಣಿ
6. ಕುವರಿಯಾದೊಡೆ ಕುಂದೇನು : ಸಂಚಿಯ ಹೊನ್ನಮ್ಮ

ಘಟಕ 2 : ವಿರಾಟ ಪರ್ವ ಸಂಗ್ರಹ (ಆಯ್ದ ಭಾಗ)

ಅಂಕಗಳು :

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(ಕೀಚಕ ವಧಾ ಪ್ರಸಂಗ)

ಸಂ. ತ.ಸು. ಶಾಮರಾಯ

ಘಟಕ : 3 ಕಾವ್ಯ ಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಜನಪದ ಗೀತೆಗಳ ಮೂಲಮಟ್ಟಗಳನ್ನು ಪರಿಚಯಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳ ಪರಿಸರದಲ್ಲಿರುವ ಜನಪದ ಸಾಹಿತ್ಯವನ್ನು ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು
- ಷಟ್ಪದಿ ಕಾವ್ಯಗಳನ್ನು ಗಮಕ ರೂಪದಲ್ಲಿ ಪ್ರಸ್ತುತ ಪಡಿಸುವುದು.
- ಆಯ್ದ ಕಾವ್ಯಭಾಗದ ನಾಟಕೀಯ ಸನ್ನಿವೇಶಗಳನ್ನು ಅಭಿನಯಿಸಲು ಉತ್ತೇಜಿಸುವುದು.
- ಕಾವ್ಯದ ಭಾಗವನ್ನು ನಾಟಕ ಅಥವಾ ಕಥೆಯ ರೂಪಕ್ಕೆ ಪರಿವರ್ತಿಸಲು ಸೂಚಿಸುವುದು.
- ಆಯ್ದ ವಚನಕಾರರ ಬದುಕು ಮತ್ತು ಸಾಹಿತ್ಯ ಕುರಿತಂತೆ ವಿಚಾರಗೋಷ್ಠಿ ನಡೆಸುವುದು.
- ಕಾವ್ಯಭಾಗದ ಕಥೆಗೆ ಸಂಬಂಧಿಸಿದ ಸಿನಿಮಾ ಪ್ರದರ್ಶನ ಹಾಗೂ ಗಾಯನಗಳನ್ನು ಕೇಳಿಸುವುದು.

ಭಾಷೆ ಕನ್ನಡ
ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)
Course Code: END06001/CDD06001

ವಾಣಿಜ್ಯ ನಿಕಾಯ
ಬಿ.ಕಾಂ/ಬಿ.ಬಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 4,

2 +1 = 3

ಕ್ರೆಡಿಟ್

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ : ವಾಣಿಜ್ಯ ಸಿರಿ - 4

ಪತ್ರಿಕೆ : 4 ಕಾವ್ಯ ಭಾಗ - 2 ಮತ್ತು ಗದ್ಯಭಾಗ

ಘಟಕ 1 : ಕಾವ್ಯಭಾಗ - 2

ಅಂಕಗಳು :

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1. ಗಂಭೀರೆಯೆಂಬ ವೃದ್ಧಸ್ತ್ರೀಯ ಕಥೆ : ಭ್ರಾಜಿಷ್ಠಾ
2. ನೆತ್ತಮನಾಡಿ ಭಾನುಮತಿ ಸೋಲೊಡೆ : ಪಂಪ
3. ನೀರೊಳಗಿದುಂ ಬೆಮರ್ತನುರಗಪತಾಕಂ : ರನ್ನ
4. ರಾಮನಂ-----ಕೊಟ್ಟೆಪೆಂ ಸೀತೆಯಂ : ನಾಗಚಂದ್ರ
5. ಭಕ್ತಿಯೆ ಭವ ಸನ್ನಿಧಿಯೊಳ್ ರೂಪಗೊಂಡಂತಿರ್ದಳ್ : ಒಂದನೆ ನಾಗವರ್ಮ
6. ವಿಧಿವಿಳಸನದ ನೆರಂಬಡೆಯೆ ಕೊಂದು ಕೂಗದೆ ನರರಂ : ಜನ್ನ

ಘಟಕ 2 : ಗದ್ಯಭಾಗ

ಅಂಕಗಳು :

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1. ಹೆನ್ನಿ ಷುಲ್ಕಾನ್ ದಂಪತಿಗಳ
ಜಗತ್ತನ್ನು ಬೆಚ್ಚಿಸಿದ ಸಂಶೋಧನೆ : ಕೆ.ಪಿ. ಪೂರ್ಣಚಂದ್ರ ತೇಜಸ್ವಿ
2. ನಮ್ಮೂರ ಬೆಳದಿಂಗಳ ಕೆರೆಯಂತಹ ಬಾಲ್ಯದ ನೆನಪು : ಸಂಜೀವರಾಯ
3. ಮಸಾಲ್ ದೋಸೆ : ಪು.ತಿ. ನರಸಿಂಹಚಾರ್
4. ಸೊಸೆಯರ ಸಂಸಾರ : ಟಿ. ಸುನಂದಮ್ಮ
5. ಕುಣಿಯುವ ಕುರುಡು ಕಾಂಚಾಣ : ಜಿ. ಚಂದ್ರಶೇಖರ

ಘಟಕ : 3 ಕಾವ್ಯ ಭಾಗ - 2 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಹಳೆಗನ್ನಡ ಪದ್ಯಗಳನ್ನು ಅರ್ಥಕ್ಕೆ ಅರ್ಥೈಸುವುದು, ಅನ್ವಯಿಸುವುದು, ವ್ಯಾಖ್ಯಾನ ಮಾಡಲು ಹೇಳುವುದು.
- ಹಳೆಗನ್ನಡ ಪದ್ಯಗಳಲ್ಲಿ ಬರುವ ಪದಗಳಿಗೆ ಅರ್ಥ ಮತ್ತು ನುಡಿಗಟ್ಟುಗಳನ್ನು ವಿವರಿಸುವುದು.
- ಪೌರಾಣಿಕ ಸನ್ನಿವೇಶಗಳ ಹಿನ್ನೆಲೆಯನ್ನು ವಿವರಿಸುವುದು.
- ವಸ್ತು ಮತ್ತು ಆಶಯಗಳನ್ನು ಪ್ರಸ್ತುತ ಸಂದರ್ಭಕ್ಕೆ ತಕ್ಕಕಂತೆ ಅನ್ವಯಿಸಿ ವಿಚಾರ ಮಂಡಿಸಲು ಗುಂಪು ಚರ್ಚೆ ನಡೆಸಲು ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.
- ಹಳೆಗನ್ನಡ ಕವಿತೆಗೆ ಗಮಕ ವ್ಯಾಖ್ಯಾನ ಮಾಡಿಸುವುದು..
- ಹಸ್ತಪ್ರತಿಗಳನ್ನು ತೋರಿಸುವುದು.

ಘಟಕ : 4 ಗದ್ಯಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಪರಿಸರ ಸಂರಕ್ಷಣೆಯಲ್ಲಿ ವಿದ್ಯಾರ್ಥಿಯು ತನ್ನ ಪಾತ್ರ ಕುರಿತು ತಿಳಿಯ ಪಡಿಸುವುದು.
- ಪರಿಸರ ಸಂರಕ್ಷಣೆ ಕುರಿತು ಚರ್ಚೆ ಮತ್ತು ಅಭಿಪ್ರಾಯ ಮಂಡನೆ
- ವಿದ್ಯಾರ್ಥಿಗಳಿಂದ ಅವರ ಜೀವನದ ಅನುಭವ ಕುರಿತು ಲೇಖನಗಳನ್ನು ಬರೆಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಯು ತಾನು ಕೈಗೊಂಡ ಪ್ರವಾಸದ ಅನುಭವಗಳನ್ನು ಅಭಿವ್ಯಕ್ತಪಡಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ತಾವು ತಿಳಿದಿರುವ ಹಾಸ್ಯ ಸನ್ನಿವೇಶಗಳನ್ನು ವರ್ಣಿಸುವುದು.
- ವಾಣಿಜ್ಯ ಕ್ಷೇತ್ರದ ಸಾಧಕರನ್ನು ಗುರುತಿಸಿ ಪರಿಚಯಿಸಲು ಸೂಚಿಸುವುದು.

ಭಾಷೆ ಕನ್ನಡ

ಮೊದಲನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)

Course Code: DMA06001/DMA 06002/DMA 06003/DMA 06004/DMA 06005/DMA 06006/DMA 06007/

DMA 06008/ECA06001

ವಿಜ್ಞಾನ ನಿಕಾಯ

ಬಿ.ಎಸ್ಸಿ / ಬಿ.ಸಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 1

2 +1 = 3 ಕ್ರೆಡಿಟ್

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪಠ್ಯಪೂರಕ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ: ವಿಜ್ಞಾನ ಸಿರಿ - 1

ಪತ್ರಿಕೆ : 1 ಪದ್ಯಭಾಗ ಮತ್ತು ಕಥಾಭಾಗ

ಘಟಕ 1 : ಪದ್ಯಭಾಗ

ಅಂಕಗಳು :

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1. ಕಾಣಿಕೆ : ಬಿ. ಎಂ. ಶ್ರೀ
2. ಮಂಕುತಿಮ್ಮನ ಕಗ್ಗ (ಆಯ್ದ ಎಂಟು ಪದ್ಯಗಳು) : ಡಿ.ವಿ.ಜಿ
3. ನಲ್ಲಿ ತಾವ್ ನಂ ಮಲ್ಲಿ : ಜಿ. ಪಿ. ರಾಜರತ್ನಂ
4. ತಾಯಿ : ಗಂಗಾಧರ ಚಿತ್ತಾಲ
5. ಹಕ್ಕು : ಕೆ.ಎಸ್. ನಿಸಾರ್ ಅಹಮದ್
6. ಆ ಮರ ಈ ಮರ : ಡಾ. ಚಂದ್ರಶೇಖರ ಕಂಬಾರ
7. ನನ್ನ ಜನಗಳು : ಡಾ. ಸಿದ್ದಲಿಂಗಯ್ಯ
8. ದೇವದಾಸಿ : ಸುಕನ್ಯಾ ಮಾರುತಿ

ಘಟಕ 2 : ಕಥಾಭಾಗ

ಅಂಕಗಳು :

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1. ಬೈಚೇಗೌಡ : ಡಾ. ಮಾಸ್ತಿ ವೆಂಕಟೇಶ ಅಯ್ಯಂಗಾರ್
2. ಶಾಲಸಾಬಿ, ಕನ್ನಡ ಮೌಲ್ವಿ : ಗೋರೂರು ರಾಮಸ್ವಾಮಿ ಅಯ್ಯಂಗಾರ್
3. ಗಾಂಧಿ : ಬೆಸಗರಹಳ್ಳಿ ರಾಮಣ್ಣ
4. ಚಪ್ಪಲಿಗಳು : ಸಾ.ರಾ. ಅಬೂಬಕ್ಕರ್
5. ಹೊಸಹುಟ್ಟು : ನೇಮಿಚಂದ್ರ

ಘಟಕ : 3 ಪದ್ಯಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಕವಿತೆಗಳನ್ನು ಭಾವ, ಅರ್ಥ ತಿಳಿಯುವಂತೆ ಓದಿಸುವುದು
- ಕವಿತೆಗಳಲ್ಲಿರುವ ವಿಶೇಷ ಪದ, ನುಡಿಗಟ್ಟು, ಗಾದೆ - ಇವುಗಳನ್ನು ಗುರುತಿಸಿ ಸ್ವಂತ ವಾಕ್ಯಗಳಲ್ಲಿ ಬಳಸಲು ಸೂಚಿಸುವುದು.
- ಕವಿತೆಗಳ ವಸ್ತು ಆಶಯಗಳನ್ನು ಪ್ರಸ್ತುತಕ್ಕೆ ಅನ್ವಯಿಸಿ ವಿಚಾರಮಂಡನೆ, ಗುಂಪುಚರ್ಚೆ ಏರ್ಪಡಿಸುವುದು.
- ಕವಿತೆಗಳಿಗೆ ರಾಗ ಸಂಯೋಜಿಸಿ ಹಾಡಲು ಉತ್ತೇಜಿಸುವುದು, ಬಹುಮಾಧ್ಯಮಗಳನ್ನು ಬಳಸಿಕೊಳ್ಳುವುದು.
- ಕವಿತೆಗಳ ವಸ್ತು ಮತ್ತು ಆಶಯಗಳನ್ನು ಸೃಜನಾತ್ಮಕವಾಗಿ (ಕವಿತೆ ರಚನೆ, ಕಥನರೂಪ . ನಾಟಕ ರೂಪ, ಸ್ವಾನುಭವ ಕಥನ, ಚಿತ್ರಕಲೆ, ಮಣ್ಣಿನ ಮಾದರಿ, ಛಾಯಾಚಿತ್ರ ಇತ್ಯಾದಿ..) ಪರಿವರ್ತಿಸುವಂತೆ ಸೂಚಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳನ್ನು ಗುಂಪುಗಳನ್ನಾಗಿ ವಿಂಗಡಿಸಿ ಸಂಬಂಧಿಸಿದ ಚಟುವಟಿಕೆಗಳನ್ನು ನಡೆಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳಿಗೆ ಸಂಬಂಧಿಸಿದ ದಾಖಲೆಗಳನ್ನು ಸಂಗ್ರಹಿಸಿಡುವಂತೆ ಸೂಚಿಸುವುದು.

ಘಟಕ : 4 ಕಥಾಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಕಥಾಭಾಗಗಳ ಭಾವ, ಅರ್ಥ ತಿಳಿಯುವಂತೆ ಓದಿಸುವುದು
- ವಿಶೇಷ ಪದ, ನುಡಿಗಟ್ಟು, ಗಾದೆ - ಇವುಗಳನ್ನು ಗುರುತಿಸಿ ಸ್ವಂತ ವಾಕ್ಯಗಳಲ್ಲಿ ಬಳಸಲು ಸೂಚಿಸುವುದು.
- ವಸ್ತು ಆಶಯಗಳನ್ನು ಪ್ರಸ್ತುತಕ್ಕೆ ಅನ್ವಯಿಸಿ ವಿಚಾರಮಂಡನೆ, ಗುಂಪುಚರ್ಚೆ ಏರ್ಪಡಿಸುವುದು.
- ವಸ್ತು ಮತ್ತು ಆಶಯಗಳನ್ನು ಸೃಜನಾತ್ಮಕವಾಗಿ (ಕವಿತೆ ರಚನೆ, ಕಥನರೂಪ . ನಾಟಕ ರೂಪ, ಸ್ವಾನುಭವ ಕಥನ, ಚಿತ್ರಕಲೆ, ಮಣ್ಣಿನ ಮಾದರಿ, ಛಾಯಾಚಿತ್ರ ಇತ್ಯಾದಿ..) ಪರಿವರ್ತಿಸುವಂತೆ ಸೂಚಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳನ್ನು ಗುಂಪುಗಳನ್ನಾಗಿ ವಿಂಗಡಿಸಿ ಸಂಬಂಧಿಸಿದ ಚಟುವಟಿಕೆಗಳನ್ನು ನಡೆಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳಿಗೆ ಸಂಬಂಧಿಸಿದ ದಾಖಲೆಗಳನ್ನು ಸಂಗ್ರಹಿಸಿಡುವಂತೆ ಸೂಚಿಸುವುದು

ಭಾಷೆ ಕನ್ನಡ
ಎರಡನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)
Course Code: ECB06001 DMB06001DMB 06002DMB 06003DMB 06004DMB
06005DMB06006DMB06007DMB06008

ವಿಜ್ಞಾನ ನಿಕಾಯ

ಬಿ.ಎಸ್ಸಿ /ಬಿ.ಸಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 2,

2 +1 = 3

ಕ್ರೆಡಿಟ್

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯೂಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪಠ್ಯ : ವಿಜ್ಞಾನ ಸಿರಿ - 2

ಪತ್ರಿಕೆ : 2 ನಾಟಕ ಮತ್ತು ಕಾದಂಬರಿ

ಘಟಕ - 1

ಅಂಕಗಳು : 40

ಪಠ್ಯ - ನಾಟಕ : ಶೂದ್ರ ತಪಸ್ವಿ

ಲೇಖಕರು : ಕುವೆಂಪು

ಘಟಕ - 2

ಅಂಕಗಳು: 30

ಪಠ್ಯ - ಕಾದಂಬರಿ : ಕಾಡು

ಲೇಖಕರು : ಶ್ರೀ ಕೃಷ್ಣ ಅಲನಹಳ್ಳಿ

ಘಟಕ - 3 : ನಾಟಕ ಭಾಗದ ಪಠ್ಯ ಪೂರಕ ಚಟುವಟಿಕೆಗಳು

- ನಾಟಕದ ಆಯ್ದ ಭಾಗವನ್ನು ಧ್ವನಿ ಏರಿಳಿತಗಳೊಂದಿಗೆ ಓದಿಸುವುದು
- ಭಾಷಾ ವೈಶಿಷ್ಟ್ಯವನ್ನು ಗುರುತಿಸುವುದು.
- ನಾಟಕದ ಆಶಯವನ್ನು ಪ್ರಸ್ತುತಕ್ಕೆ ಅನ್ವಯಿಸಿ - ವಿಚಾರ ಮಂಡನೆ, ಗುಂಪು ಚರ್ಚೆ
- ಪಠ್ಯವನ್ನು ರಂಗರೂಪಕ್ಕೆ ತರುವ ಸಾಧ್ಯತೆಗಳ ಕುರಿತ ಚರ್ಚೆ
- ನಾಟಕದ ಸಂವಿಧಾನ, ತಂತ್ರ ಕುರಿತ ವಿಶ್ಲೇಷಣೆ
- ನಾಟಕದ ಭಾಗಗಳನ್ನು ಮತ್ತು ಪಾತ್ರಗಳನ್ನು ಅಭಿನಯಿಸುವಂತೆ ವಿದ್ಯಾರ್ಥಿಗಳನ್ನು ಉತ್ತೇಜಿಸುವುದು.
- ನಾಟಕಕಾರರ ಬದುಕು, ಸಾಹಿತ್ಯಕ ಸಾಧನೆ ಕುರಿತ ಮಾಹಿತಿ ಸಂಗ್ರಹಿಸುವುದು.

- ನಾಟಕದ ವಸ್ತು ಮತ್ತು ಆಶಯವನ್ನು ಸೃಜನಾತ್ಮಕವಾಗಿ ಮರು ರೂಪಿಸಲು (ಕವಿತೆ ರಚನೆ, ಪ್ರಹಸನ, ಸ್ವಾನುಭವ ಕಥನ, ಚಿತ್ರಕಲೆ, ಛಾಯಾಚಿತ್ರ ಇತ್ಯಾದಿ) ಸೂಚಿಸುವುದು
- ಇದಕ್ಕೆ ಸಂಬಂಧಿಸಿದ ವಿಡಿಯೋ ಪ್ರದರ್ಶನ

ಘಟಕ - 4 : ಕಾದಂಬರಿ ಭಾಗದ ಪಠ್ಯಪೂರಕ ಚಟುವಟಿಕೆಗಳು

- ಕಾದಂಬರಿಯ ಮುಖ್ಯ ಭಾಗಗಳ ವಾಚನ
- ಭಾಷಾ ವೈಶಿಷ್ಟ್ಯಗಳನ್ನು ಗುರುತಿಸುವುದು (ಪ್ರಾದೇಶಿಕ ಭಾಷಾ ಪ್ರಭೇದ ನುಡಿಗಟ್ಟು, ಗಾದೆ ಇತ್ಯಾದಿ.)
- ಆಶಯಕ್ಕೆ ಸಂಬಂಧಿಸಿದಂತೆ ಚರ್ಚೆ ಏರ್ಪಡಿಸುವುದು.
- ಕಾದಂಬರಿಗೆ ಸಂಬಂಧಿಸಿದಂತೆ ವಿಚಾರ ಮಂಡನೆಗೆ ಅವಕಾಶ ನೀಡುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಓದಿದ ಇನ್ನಿತರ ಕಾದಂಬರಿಗಳ ಕುರಿತು ಅಭಿಪ್ರಾಯ ಮಂಡನೆಗೆ ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.
- ಕಾದಂಬರಿ ಆಧರಿಸಿದ ಸಿನಿಮಾಗಳಿದ್ದರೆ ಅವುಗಳನ್ನು ಪ್ರದರ್ಶಿಸುವುದು.

ಭಾಷೆ ಕನ್ನಡ

ಮೂರನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)

Course Code:DMC06001/DMC06002/DMC06003/DMC06004/DMC06005/DMC06006/DMC06007/DMC06008

ವಿಜ್ಞಾನ ನಿಕಾಯ

ಬಿ.ಎಸ್ಸಿ/ಬಿ.ಸಿ.ಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 1,

2 +1 = 3

ಕ್ರೆಡಿಟ್

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪತ್ರಿಕೆ : 3 ನಡುಗನ್ನಡ ಕಾವ್ಯಗಳು (ಆಯ್ದ ಭಾಗ)

ಪಠ್ಯ : ವಿಜ್ಞಾನ ಸಿರಿ - 3

ಘಟಕ 1 : ಕಾವ್ಯಭಾಗ

ಅಂಕಗಳು :

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1. ಭಿನ್ನ ಭೇದವ ಮಾಡಬ್ಯಾಡಿರೋ : ಜನಪದ
2. ವಚನಗಳು : ಅಕ್ಕಮಹಾದೇವಿ
ಅರಿವಿನ ಮಾರಿತಂದೆ
ಮೋಳಿಗೆ ಮಹಾದೇವಿ
ಉರಿಲಿಂಗಪೆದ್ದಿ
3. ಇನ್ನು ಹುಟ್ಟಿದಿರಲಿ ನಾರಿಯರೆನ್ನವೊಲು : ಕುಮಾರವ್ಯಾಸ
4. ಕೋಪ ಪೈಸರಿಸಿತು ಹೃದಯ ತಂಪಾಯ್ತು : ರತ್ನಾಕರವರ್ಣಿ
5. ಮಳೆಯ ದಯಮಾಡೋ ರಂಗಯ್ಯ : ಹೆಳವನ ಕಟ್ಟೆ ಗಿರಿಯಮ್ಮ
6. ಶ್ರೀ ಗುರು ವಚನೋಪದೇಶ : ನಿಜಗುಣ ಶಿವಯೋಗಿ

ಘಟಕ - 2 ರಾಮಧಾನ್ಯ ಚರಿತೆ

ಅಂಕಗಳು: 30

ಸಂ. ಸಣ್ಣಯ್ಯ

ಘಟಕ : 3 ಕಾವ್ಯ ಭಾಗ - 2 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಹಳೆಗನ್ನಡ ಪದ್ಯಗಳನ್ನು ಅರ್ಥಕ್ಕೆ ಅನುಗುಣವಾಗಿ ಬಿಡಿಸಿ ಓದುವುದು.
- ಹಳೆಗನ್ನಡ ಪದ್ಯಗಳಲ್ಲಿ ಬರುವ ಪದಗಳಿಗೆ ಅರ್ಥ ಮತ್ತು ನುಡಿಗಟ್ಟು, ಪೌರಾಣಿಕ ಸನ್ನಿವೇಶಗಳ ಹಿನ್ನೆಲೆಯನ್ನು ವಿವರಿಸುವುದು.
- ವಸ್ತು ಮತ್ತು ಆಶಯಗಳನ್ನು ಪ್ರಸ್ತುತ ಸಂದರ್ಭಕ್ಕೆ ತಕ್ಕಂತೆ ಅನ್ವಯಿಸಿ ವಿಚಾರ ಮಂಡಿಸಲು ಗುಂಪು ಚರ್ಚೆ ನಡೆಸಲು ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.
- ಹಳೆಗನ್ನಡ ಕವಿತೆಗೆ ಗಮಕ ವ್ಯಾಖ್ಯಾನ ಮಾಡಿಸುವುದು.
- ಹಸ್ತಪ್ರತಿಗಳನ್ನು ತೋರಿಸುವುದು.

ಭಾಷೆ ಕನ್ನಡ
ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್)
Course Code: ECD06001/ DMD06001/DMD 06002/DMD 06003/DMD 06004/DMD
06005/DMD06006/DMD06007/DMD06008

ವಿಜ್ಞಾನ ನಿಕಾಯ
ಬಿ.ಎಸ್ಸಿ / ಬಿ.ಸಿಎ

ಭಾಷಿಕ ಕನ್ನಡ ಎ.ಇ.ಸಿ.ಸಿ - 4,
ಕ್ರೆಡಿಟ್

2 +1 = 3

ವಾರಕ್ಕೆ ಎರಡು ಕ್ರೆಡಿಟ್‌ಗಳ ಪಾಠ - ಒಂದು ಕ್ರೆಡಿಟ್ ಪೂರಕ ಪಠ್ಯ ಚಟುವಟಿಕೆ (ಟ್ಯುಟೋರಿಯಲ್)

ನಿರಂತರ ಮೌಲ್ಯಮಾಪನ, ಸಿ- 1, (15 ಅಂಕಗಳು) ಮತ್ತು ಸಿ- 2 (15 ಅಂಕಗಳು)

ಮುಖ್ಯ ಪರೀಕ್ಷೆ ಸಿ - 3 (70 ಅಂಕಗಳು)

ಪತ್ರಿಕೆ : 4 ಕಾವ್ಯ ಭಾಗ ಮತ್ತು ಗದ್ಯಭಾಗ

ಪಠ್ಯ : ವಿಜ್ಞಾನ ಸೌರಭ - 4

ಘಟಕ 1 : ಕಾವ್ಯಭಾಗ - 2

ಅಂಕಗಳು :

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1. ಕುರಿತೋದದೆಯುಂ ಕಾವ್ಯ ಪ್ರಯೋಗ ಪರಿಣತ ಮತಿಗಳ್ : ಶ್ರೀವಿಜಯ
2. ನಲೈಗೆ ದಿಬ್ಬಂಬಿಡಿವ೦ತವೊಲ್ : ಪಂಪ
3. ಭೀಮಸೇನ ಪ್ರತಿಜ್ಞಾ ವರ್ಣನಂ : ರನ್ನ
4. ಕಾವರೆ ಕಣೆಗೊಳ್ಳೊಡಬ್ಬೆ ಬಾರಿಪರೊಳರೇ : ಜನ್ನ
5. ಅಬ್ಬಿಯುಮೊರ್ಮೆ ಕಾಲವಶದಿಂ
ಮರ್ಯಾದೆಯಂ ದಾಂಟದೇ : ನಾಗಚಂದ್ರ
6. ಧರೆಗೆ ಒಬ್ಬನೆ ದಾನಶೂರ ಕರ್ಣ
(ಜನಪದ ಮಹಾಭಾರತ) : ಸಂ. ಪಿ.ಕೆ. ರಾಜಶೇಖರ

ಘಟಕ 2 : ಗದ್ಯಭಾಗ

ಅಂಕಗಳು:

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1. ರಾಮರಾವಣರ ಯುದ್ಧ : ಕುವೆಂಪು
2. ಚಿಗುರೊಡೆಯುತ್ತಿರುವ ಗಿಡದ ತನ್ಮಯತೆ : ಕ್ಷೀರಸಾಗರ
3. ಮಾಧವೀಲತೆ, ಸಾರ್ಕೇಂಡ್ರ : ಬಿ.ಜಿ.ಎಲ್ ಸ್ವಾಮಿ
4. ರೋಬಾಟ್‌ಗಳಿಗೆ ನಾಗರಿಕತೆಯ ಮೊದಲ ದೀಕ್ಷೆ : ನಾಗೇಶ್ ಹೆಗಡೆ
5. ಕಡಲ ಗರ್ಭದ ಅಪರಂಪಾರ : ರಹಮತ್ ತರಿಕೆರೆ

ಘಟಕ : 3 ಕಾವ್ಯ ಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಜನಪದ ಗೀತೆಗಳ ಮೂಲಮಟ್ಟಗಳನ್ನು ಪರಿಚಯಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳ ಪರಿಸರದಲ್ಲಿರುವ ಜನಪದ ಸಾಹಿತ್ಯವನ್ನು ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು
- ಷಟ್ಪದಿಕಾವ್ಯಗಳನ್ನು ಗಮಕ ರೂಪದಲ್ಲಿ ಪ್ರಸ್ತುತ ಪಡಿಸುವುದು.
- ಆಯ್ದ ಕಾವ್ಯಭಾಗದ ನಾಟಕೀಯ ಸನ್ನಿವೇಶಗಳನ್ನು ಅಭಿನಯಿಸಲು ಉತ್ತೇಜಿಸುವುದು.
- ಕಾವ್ಯದ ಭಾಗವನ್ನು ನಾಟಕ ಅಥವಾ ಕಥೆಯ ರೂಪಕ್ಕೆ ಪರಿವರ್ತಿಸಲು ಸೂಚಿಸುವುದು.
- ಆಯ್ದ ವಚನಕಾರರ ಬದುಕು ಮತ್ತು ಸಾಹಿತ್ಯ ಕುರಿತಂತೆ ವಿಚಾರಗೋಷ್ಠಿ ನಡೆಸುವುದು.

ಘಟಕ : 4 ಗದ್ಯಭಾಗದ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು

- ಪರಿಸರ ಸಂರಕ್ಷಣೆಯಲ್ಲಿ ವಿದ್ಯಾರ್ಥಿಯು ತನ್ನ ಪಾತ್ರ ಕುರಿತು ತಿಳಿಯ ಪಡಿಸುವುದು.
- ಪರಿಸರ ಸಂರಕ್ಷಣೆ ಕುರಿತು ಚರ್ಚೆ ಮತ್ತು ಅಭಿಪ್ರಾಯ ಮಂಡನೆ
- ವಿದ್ಯಾರ್ಥಿಗಳಿಂದ ಅವರ ಜೀವನದ ಅನುಭವ ಕುರಿತು ಲೇಖನಗಳನ್ನು ಬರೆಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಯು ತಾನು ಕೈಗೊಂಡ ಪ್ರವಾಸದ ಅನುಭವಗಳನ್ನು ಅಭಿವ್ಯಕ್ತಪಡಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ತಾವು ತಿಳಿದಿರುವ ಹಾಸ್ಯ ಸನ್ನಿವೇಶಗಳನ್ನು ವರ್ಣಿಸುವುದು.
- ವಿಜ್ಞಾನ ಕ್ಷೇತ್ರದ ಸಾಧಕರನ್ನು ಗುರುತಿಸಿ ಪರಿಚಯಿಸಲು ಸೂಚಿಸುವುದು.

ಮೂರನೆಯ ಚತುರ್ಮಾಸ (SEC)

ಪತ್ರಿಕೆ - 1 ಕನ್ನಡ ಸಂವಹನ ಕೌಶಲ - Kannada Communication Skill

Course Code : BA- ELC06021ELC06022ELC06023ELC06024 ELC06025ELC06026

ಘಟಕ 1 :

ಅ) ಮಾತಿನ ಕೌಶಲ

ನಿರೂಪಣೆ, ಭಾಷಣ, ಸಂಭಾಷಣೆ, ಚರ್ಚೆ, ಸಂದರ್ಶನ, ದೋಷಗಳು ಮತ್ತು ನಿವಾರಣಾ ಉಪಾಯಗಳು

ಆ) ಕೇಳುವಿಕೆ ಕೌಶಲ

ಮಹತ್ವ, ಲಕ್ಷಣಗಳು, ದೋಷಗಳು ಮತ್ತು ನಿವಾರಣಾ ಉಪಾಯಗಳು

ಘಟಕ 2 - ಅ) ಓದುವ ಕೌಶಲ

ಮಹತ್ವ, ಲಕ್ಷಣಗಳು, ವಿಧಗಳು, ಉತ್ತಮ ಓದುಗಾರಿಕೆಯ ಲಕ್ಷಣಗಳು

ಆ) ಬರವಣಿಗೆ ಕೌಶಲ

ಮಹತ್ವ, ಲಕ್ಷಣಗಳು, ಸೃಜನಶೀಲ ಬರವಣಿಗೆ, ಕನ್ನಡ ಬರವಣಿಗೆಯ ದೋಷಗಳು ಮತ್ತು ನಿವಾರಣಾ ಉಪಾಯಗಳು, ಬರವಣಿಗೆ ಉತ್ತಮಪಡಿಸುವ ವಿಧಾನಗಳು

ಘಟಕ 3 - ಪತ್ರ ವ್ಯವಹಾರ

1. ಪತ್ರಗಳು : ಸ್ವರೂಪ, ಲಕ್ಷಣ ಮತ್ತು ವಿವಿಧ ಅಂಗಗಳು - ಸಾಮಾನ್ಯ ಪತ್ರ, ಓದುಗರ ಪತ್ರ, ಮನವಿ ಪತ್ರ
2. ಸರ್ಕಾರಿ ಪತ್ರ : ಅಧಿಕೃತ, ಜ್ಞಾಪನ, ಸುತ್ತೋಲೆ, ಅಭ್ಯರ್ಥನ (ಸ್ವ ವಿವರ) ಸ್ವರೂಪ, ಲಕ್ಷಣ
3. ಕನ್ನಡ ಮತ್ತು ಗಣಕ ಜ್ಞಾನ
 - ಕಂಪ್ಯೂಟರ್ ನಲ್ಲಿ ಕನ್ನಡದ ಬಳಕೆ
 - ಕನ್ನಡ ಮತ್ತು ಡಿಟಿಪಿ
 - ಕನ್ನಡ ಮತ್ತು ಅಂತರ್ಜಾಲ
 - ಯುನಿಕೋಡ್

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಮಾತೃಭಾಷೆ ಕನ್ನಡ ತತ್ವ ಮತ್ತು ಬೋಧನಾ ಮಾರ್ಗ : ಅನಸೂಯಾ ವಿ. ಪರಗಿ
2. ಕುಶಲ ಕನ್ನಡ : ಎಚ್. ಎಸ್. ಕೃಷ್ಣಮೂರ್ತಿ, ಎಂ.ಎಸ್.
ಆಶಾದೇವಿ
3. ಆಡಳಿತ ಕನ್ನಡ : ಎಚ್.ಎಸ್.ಕೆ
4. ನುಡಿಹೆಜ್ಜೆ : ಕುವೆಂಪು ವಿಶ್ವವಿದ್ಯಾಲಯ
5. ತ್ರಿಯಾತ್ಮಕ ಕನ್ನಡ : ವಸಂತ ಕುಮಾರ
6. ಕನ್ನಡ ಬರಹ ಸರಿಪಡಿಸೋಣ : ಡಿ.ಎನ್. ಶಂಕರಭಟ್ಟ
7. ಕನ್ನಡ ಶಬ್ದ ರಚನೆ : ಡಿ.ಎನ್. ಶಂಕರಭಟ್ಟ
8. ಕನ್ನಡ ಭಾಷಾ ಪ್ರಯೋಗ : ಕಡಂಗೋಡ್ಲ ಶಂಕರಭಟ್ಟ
9. ಕನ್ನಡ ಭಾಷಾಲೋಕ : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ

ಪ್ರಶ್ನೆ ಪತ್ರಿಕೆ ಮಾದರಿ

ಮೂರನೆಯ ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್) - ಸಿಬಿಸಿಎಸ್

ಪತ್ರಿಕೆ - 1 ಕನ್ನಡ ಸಂವಹನ ಕೌಶಲ - Kannada Communication Skill

ಸಮಯ: 2 ಗಂಟೆಗಳು

ಗರಿಷ್ಠ

ಅಂಕಗಳು: 35

ಘಟಕ -1

(ಅಂಕಗಳು: 10)

1. ಎರಡು ಪ್ರಶ್ನೆಗಳಿಗೆ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

2 X 5 =

10

ಘಟಕ - 2

10)

(ಅಂಕಗಳು:

2. ಎರಡು ಪ್ರಶ್ನೆಗಳಿಗೆ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

2 X 5 =

10

ಘಟಕ - 3

15)

(ಅಂಕಗಳು:

3. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳಬೇಕು)

1 X 10 =

10

4. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

1 X 5 =

05

ಪ್ರಶ್ನೆ ಪತ್ರಿಕೆ ಮಾದರಿ

ಮೊದಲನೆಯ, ಎರಡನೆಯ, ಮೂರನೆಯ ಮತ್ತು ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸಗಳ (ಸೆಮಿಸ್ಟರ್)

ಸಿಬಿಸಿಎಸ್ - ಪಠ್ಯಕ್ರಮದ ಭಾಷಾ ಪತ್ರಿಕೆಗಳ ಪ್ರಶ್ನೆ ಪತ್ರಿಕೆ ಮಾದರಿ

ಕಲಾ, ವಾಣಿಜ್ಯ ಮತ್ತು ವಿಜ್ಞಾನ - ಮೂರು ನಿಕಾಯಗಳಿಗೂ ಅನ್ವಯವಾಗುವಂತೆ

ಸಮಯ: 3 ಗಂಟೆಗಳು

ಗರಿಷ್ಠ

ಅಂಕಗಳು: 70

ಪದ್ಯ / ನಾಟಕ / ಕಾವ್ಯ ಭಾಗ -1 / ಕಾವ್ಯ ಭಾಗ -2

ಅಂಕಗಳು: 40

೧. ಅ) ಒಂದು ಪದ್ಯ ಭಾಗದ ಭಾವವನ್ನು ಸಂದರ್ಭ ಸಹಿತ ವಿವರಿಸಿ

1 X 5 = 5

(ಎರಡು ಪದ್ಯ ಭಾಗಗಳನ್ನು ಕೇಳುವುದು)

ಆ) ಒಂದು ವಾಕ್ಯದಲ್ಲಿ ಉತ್ತರಿಸಿ (ಐದು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

5 X 1 = 5

2. ನಾಲ್ಕು ವಾಕ್ಯಗಳನ್ನು ಸಂದರ್ಭ ಸಹಿತ ವಿವರಿಸಿ (ಆರು ವಾಕ್ಯಗಳನ್ನು ಕೇಳುವುದು)

2. 1/2 X

4= 10

3. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

1 X 10 =

10

1. ಎರಡು ಪ್ರಶ್ನೆಗಳಿಗೆ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

2 X 5 = 10

ಕಥೆ / ಕಾದಂಬರಿ / ಗದ್ಯವೈವಿಧ್ಯ

30 ಅಂಕಗಳು

2. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

1 X 10 =

10

3. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

1 X 10 =

10

4. ಎರಡನ್ನು ಕುರಿತು ಸಂಕ್ಷಿಪ್ತವಾಗಿ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

2 X 5 = 10



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

(Autonomous)

B N ROAD, MYSURU- 570 025

DEPARTMENT OF MATHEMATICS

Syllabus

CHOICE BASED CREDIT SYSTEM

For B.Sc programmes

- **Physics, Mathematics and Chemistry**
- **Physics, Mathematics and Computer Science**
- **Physics, Mathematics and Computer Maintenance**
- **Physics, Mathematics and Electronics**

2019-2020

Scheme of Assessment:

Credits L:T:P	Percentage			Maximum marks in the Exam /Assessment			Exam Duration	
	Th	Pr	IA	Th	Pr	IA	Th	Pr
4:0:2	50	20	30	70	70	30	3h	3h
5:1:0	70	-	30	70	-	30	3h	-
4:0:0	70	-	30	70	-	30	3h	-
2:0:2	35	35	30	50	70	30	2h	3h
3:0:0	70	-	30	70	-	30	3h	-
2:1:0	70	-	30	70	-	30	3h	-
2:0:0	70	-	30	50	-	30	2h	-
0:0:1	-	70	30	-	70	30	-	2h

Note: L-Lecture, T-Tutorial, P-Practical; Th- Theory, Pr-Practical,

I A- Internal Assessment

SEMESTER III

Credits: L: T: P = 4:0:0

Teaching hours: 4 hours per week

DSC III : Real Analysis

Unit I: Finite and infinite sets, examples of countable and uncountable sets. Real line, bounded sets, supremum and infimum, completeness property of \mathbb{R} , Archimedean property of \mathbb{R} , intervals. Concept of cluster points and statement of Bolzano-Weierstrass theorem.

Unit II: Real Sequence, Bounded sequence, Cauchy convergence criterion for sequences. Cauchy's theorem on limits, order preservation and squeeze theorem, monotone sequences and their convergence .

Unit III: Infinite series. Cauchy convergence criterion for series, positive term series, geometric series, comparison test, convergence of p-series, Root test, Ratio test, alternating series, Leibnitz's test , Definition and examples of absolute and conditional convergence.

Unit IV: Sequences and series of functions, Point wise and uniform convergence. Mn-test, M-test, Statements of the results about uniform convergence, Power series and radius of convergence.

Reference Books:

1. S.C Malik –Real Analysis
2. S.C.Malik and Savita Arora, *Mathematical Analysis*, 2nd ed. New Delhi, India: New Age international (P) Ltd., 1992
3. Richard R Goldberg, *Methods of Real Analysis*, Indian ed.
4. Asha Rani Singhal and M .K Singhal, *A first course in Real Analysis*
5. E.Kreyszig- *Advanced Engineering Mathematics*, Wiley India Pvt. Ltd.
6. Raisinghania M. D., *Laplace and Fourier Transforms* S. Chand publications.
7. *Principles of Mathematical Analysis*- Walter Rudin.
8. *Mathematical Analysis*- Tom M Apostol

PRACTICAL COMPONENTS –III

Credits: L: T: P = 0:0:2

1. Illustration of convergent, divergent and oscillatory sequences.
2. Plotting of recursive sequences.
3. Study of convergence of sequences through plotting
4. Illustration of convergent, divergent and oscillatory series.
5. To study the convergence and divergence of infinite series by plotting their sequences of partial sums.
6. Using Cauchy's criterion on the sequence of partial sums of the series to determine convergence of series.
7. Cauchy's root test by plotting n^{th} roots.
8. Ratio test by plotting the ratio of n^{th} and $(n+1)^{\text{th}}$ terms.
9. Testing the convergence of binomial, exponential and logarithmic series and finding the sum.
10. To find the sum of the series and its radius of convergence.

SEMESTER IV

Credits: L: T: P = 4:0:0

Teaching hours: 4 hours per week

DSC IV: Algebra

Unit I: Definition and examples of groups, examples of abelian and non-abelian groups, the group Z_n of integers under addition modulo n and the group $U(n)$ of units under multiplication modulo n . Cyclic groups from number systems, complex roots of unity, cyclic group, groups of symmetries, the permutation group, Group of quaternion's.

Unit II: Subgroups, cyclic subgroups, the concept of a subgroup generated by a subset and the commutator subgroup of group, examples of subgroups including the center of a group. Cosets, Index of subgroup, Lagrange's theorem, order of an element, Normal subgroups: their definition, examples, and characterizations, Quotient groups. Homomorphism, Kernel and Image, Isomorphism, Fundamental Theorem of Homomorphism.

Unit III: Definition and examples of rings, examples of commutative and non-commutative rings: rings from number systems, Z_n the ring of integers modulo n , ring of real quaternion's, rings of matrices, polynomial rings, and rings of continuous functions. Sub rings and ideals.

Unit IV: Integral domains and fields, examples of fields: Z_p , Q , R , and C . Field of rational functions. Homomorphisms', Isomorphism'.

Reference Books :

1. I. N. Herstein – Topics in Algebra.
2. Joseph Gallian – Contemporary Abstract Algebra, Narosa Publishing House, New Delhi, Fourth Edition.
3. G. D. Birkhoff and S Maclane – A brief Survey of Modern Algebra.
4. J B Fraleigh – A first course in Abstract Algebra.
5. Michael Artin – Algebra, 2nd ed. New Delhi, India: PHI Learning Pvt. Ltd., 2011.
6. Vashista, A First Course in Modern Algebra, 11th ed.: Krishna Prakasan Mandir, 1980.
7. R Balakrishnan and N.Ramabadran, A Textbook of Modern Algebra, 1st ed. New Delhi, India: Vikas publishing house pvt. Ltd., 1991.
8. University algebra by N.S.Gopalakrishnan

PRACTICAL COMPONENTS-IV

Credits: L: T: P = 0:0:2

1. Verifying whether a given operator is binary or not.
2. To find identity element of a group.
3. To find inverse element of a group.
4. Finding all possible subgroups of a finite group.
5. Examples to verify Lagrange's theorem.
6. Illustrating homomorphism and isomorphism of groups.
7. Verification of normality of a given subgroup.
8. Verifying Cayley's theorem and isomorphism theorems.
9. Examples for finding left and right coset and finding the index of a group.
10. Examples on different types of rings.
11. Examples on integral domains and fields.
12. Examples on subrings, ideals and subrings which are not ideals.
13. Homomorphism and isomorphism of rings – illustrative examples.
14. Solving polynomial equations.
15. Finding the G.C.D of polynomials.
16. Finding units and associates.
17. Test for rational roots.

SEMESTER V

Credits: L: T: P = 4:0:0

Teaching hours: 4 hours per week

DSE I: Linear Algebra

Unit I: Vector spaces, subspaces, algebra of subspaces, quotient spaces, linear combination of vectors, linear span, linear independence, basis and dimension, dimension of subspaces, Direct sum of two subspaces, Quotient space.

Unit II: Inner Product, Inner Product of any two vectors in $V(\mathbb{R})$, Euclidean Vectorspace, Orthogonal Vectors, Ortho normal Basis, Orthogonal Projection, Orthogonal Compliment.

Unit III: Linear transformations, algebra of linear transformations , matrix representation of a linear transformation, null space, range, rank and nullity of a linear transformation,

Unit IV: Eigen values and Eigen vectors, Characteristic Polynomial. Isomorphism, Auto morphism, theorems, invertibility of non singular linear transformation, change of Basis and similar matrices.

Reference Books:

1. I. N. Herstien – Topics in Algebra.
2. Stewart – Introduction to Linear Algebra
3. S. Kumaresan – Linear Algebra
4. G. D. Birkhoff and S Maclane – A brief Survey of Modern Algebra.
5. N.S.Gopalakrishna – University Algebra
6. Saymour Lipschitz – Theory and Problems of Linear Algebra.
7. B.S Grewal – Higher engineering mathematics.
8. E.Kreyszig – Advanced Engineering Mathematics, Wiely India Pvt. Ltd.
9. J B Fraleigh – A first course in Abstract Algebra.

PRACTICAL COMPONENTS–V

Credits: L: T: P = 0:0:2

1. Vector space, subspace – illustrative examples.
2. Expressing a vector as a linear combination of given set of vectors.
3. Examples on linear dependence and independence of vectors.
4. Basis and Dimension – illustrative examples.
5. Verifying whether a given transformation is linear.
6. Finding matrix of a linear transformation.
7. Problems on rank and nullity.
8. Find characteristics polynomials.
9. To find Eigen values and their multiplicity.
10. Calculation of Eigen vector.
11. Change of basis.
12. Linear transformations to matrices and vice versa.
13. Matrix with respect to change of basis.
14. Orthogonal and orthonormal sets.
15. Gram- Schmidt orthogonalisation of the columns.

SEMESTER V

Credits: L: T: P = 4:0:0

Teaching hours: 4 hours per week

DSE I A: Matrices

Unit I: Types of matrices. Rank of a matrix. Invariance of rank under elementary transformations. Reduction to normal form, Solutions of linear homogeneous and non-homogeneous equations with number of equations and unknowns upto four.

Unit II: Matrices in diagonal form. Reduction to diagonal form upto matrices of order 3. Computation of matrix inverses using elementary row operations. Rank of matrix. Solutions of a system of linear equations using matrices. Illustrative examples of above concepts from Geometry, Physics, Chemistry, Combinatorics and Statistics.

Unit III: R_1, R_2, R_3 as vector spaces over R . Standard basis for each of them. Concept of Linear Independence and examples of different bases. Subspaces of R_2, R_3 .

Unit IV: Translation, Dilation, Rotation, Reflection in a point, line and plane. Matrix form of basic geometric transformations. Interpretation of eigen values and eigen vectors for such transformations and eigen spaces as invariant subspaces.

Reference Books:

1. A.I. Kostrikin, Introduction to Algebra, Springer Verlag, 1984.
2. S. H. Friedberg, A. L. Insel and L. E. Spence, Linear Algebra, Prentice Hall of India Pvt. Ltd., New Delhi, 2004.
3. Richard Bronson, Theory and Problems of Matrix Operations, Tata McGraw Hill, 1989.
4. E.Kreyszig – Advanced Engineering Mathematics, Wiley India Pvt. Ltd
5. B.S Grewal – Higher engineering mathematics

PRACTICAL COMPONENTS-V

Credits: L:T:P = 0:0:2

1. Introduction to matrices and commands connected to the matrices.
2. Addition and subtraction of matrices.
3. Multiplication and transpose of matrices.
4. Power of a matrix.
5. Row reduced echelon form.
6. Rank of a matrix.
7. Adjoint of a matrix.
8. Inverse of a non-singular matrix.
9. Systems of linear equations.
10. Trace of a matrix.

SEMESTER VI

Credits: L:T: P = 4:0:0

Teaching hours: 4 hours per week

DSE II: Complex Analysis

Unit I: Complex numbers, Polar and exponential form of complex numbers, Triangular inequality, Geometry of complex numbers, Equations of lines and circles in complex form, Functions of complex variables, Limits, Limits involving the point at infinity, continuity. Properties of complex numbers, regions in the complex plane, Derivatives, differentiation formulas, Cauchy-Riemann equations, sufficient conditions for differentiability.

Unit II: Analytic functions, examples of analytic functions, exponential function, Logarithmic function, trigonometric function, Harmonic functions, Construction of Analytic functions.

Unit III: Definite integrals of functions. Contours, Contour integrals and its examples, upper bounds for moduli of contour integrals. Cauchy-Goursat theorem, Cauchy integral formula, Cauchy's inequality, Liouville's theorem and the fundamental theorem of algebra. Convergence of sequences and series.

Unit IV: Linear and Bilinear Transformations, Cross ratio of four points, Conformal mapping, Transformations of the form z^2 , $\frac{1}{z}$, $\sin z$, $\cos z$, e^z , $\sinh z$, $\cosh z$ etc, Laurent series and its examples, Poles and residues.

Reference Books:

1. L. V. Ahlfors – Complex Analysis
2. Bruce P. Palica – Introduction to the Theory of Function of a Complex Variable
3. Serge Lang – Complex Analysis
4. Shanthinarayan – Theory of Functions of a Complex Variable
5. S. Ponnuswamy – Foundations of Complex Analysis
6. R. P. Boas – Invitation to Complex Analysis.
7. R V Churchill & J W Brown, Complex Variables and Applications, 5th ed.: McGraw Hill Companies., 1989.
8. A R Vashista, Complex Analysis, Krishna Prakashana Mandir, 2012.

PRACTICAL COMPONENTS-VI

Credits: L:T:P = 0:0:2

1. Declaring a complex number and graphical representation.
2. Complex numbers and their representations, operations like addition, multiplication, division, modulus, graphical representations of polar form.
3. To plot the complex functions and analyze the graph
(i) $f(z) = z$, (ii) $f(z) = z^3$, (iii) $f(z) = (z^4 - 1)^{1/4}$
4. Some problems on Cauchy – Riemann equations (polar forms).
5. Implementation of Milne – Thomson method of constructing analytic functions (simple examples).
6. Illustrating orthogonality of the surfaces obtained from the real and imaginary parts of an analytic function.
7. Verifying real and imaginary parts of an analytic function being harmonic (in polar coordinates)
8. Examples connected with Cauchy's integral theorem.
9. To compute the poles and corresponding residues of complex functions.
10. Illustrating the angle preserving property in a transformation.
11. Illustrating the circles are transformed to circles by a bilinear transformation.
12. To perform conformal mapping and bilinear transformations.

SEMESTER VI

Credits: L:T: P = 4:0:0

Teaching hours: 4 hours per week

DSE II B: Numerical Methods

Unit I: Algorithms, Convergence, Bisection method, False position method, Fixed point iteration method, Newton's method, Secant method, LU decomposition, Gauss-Jacobi, Gauss-Siedel and SOR iterative methods.

Unit II: Lagrange and Newton interpolation: linear and higher order, finite difference operators. Numerical differentiation: forward difference, backward difference and central Difference. Integration: trapezoidal rule, Simpson's rule, Euler's method.

Reference Books:

1. B. Bradie, A Friendly Introduction to Numerical Analysis, Pearson Education, India, 2007.
2. M.K. Jain, S.R.K. Iyengar and R.K. Jain, Numerical Methods for Scientific and Engineering Computation, 5th Ed., New age International Publisher, India, 2007.
3. Introduction to Numerical analysis by S.S.Shasthri
4. B.S Grewal – Higher engineering mathematics
5. E.Kreyszig – Advanced Engineering Mathematics, Wiely India Pvt. Ltd

PRACTICAL COMPONENTS-VI A

Credits: L:T:P = 0:0:2

1. Newton Gregory forward interpolation.
2. Lagrange interpolation.
3. Simpson's one-third method.
4. Simpson's three-eighth method.
5. Bisection method.
6. Regula-Falsi method.
7. Newton-Raphson method.
8. Modified Euler's method.
9. Runge Kutta second order method.
10. Runge Kutta fourth order method.

Vector Calculus

Credits: L:T: P = 2:0:0

Teaching hours: 2 hours per week

Skill Enhancement Course (SEC - II)

Unit I: Differentiation and partial differentiation of a vector function. Derivative of sum, dot product and cross product of two vectors.

Unit II: Gradient, divergence and curl, Standard derivations and Exercise ,

Reference Books:

1. Murray R Spiegel – Theory and problems of vector calculus.
2. Shanthinarayan and J N Kapur – A text book of Vector calculus.
3. B.S Grewal – Higher engineering mathematics.
4. Shanthi Narayan and P K Mittal, Differential Calculus, Reprint. New Delhi: S Chand and Co. Pvt. Ltd., 2014.

Question Paper Pattern

Mathematics

Time: 3 Hours

Max. Marks: 70

Section – A

I. Answer any five questions.

5 x 2 = 10

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.

Section – B

II. Answer any three questions.

3 x 5 = 15

- 1.
- 2.
- 3.
- 4.
- 5.

III. Answer any three questions.

3 x 5 = 15

- 1.
- 2.
- 3.
- 4.
- 5.

IV. Answer any three questions.

3 x 5 = 15

- 1.
- 2.
- 3.
- 4.
- 5.

V. Answer any three questions.

3 x 5 = 15

- 1.
- 2.
- 3.



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

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OOTY ROAD, MYSURU- 570 025

NATIONAL EDUCATION POLICY 2020 INITIATIVES

B.Sc. (Hons) Mathematics,

B.Sc. with Mathematics as a Major/Minor Subject

Syllabus for B.Sc (Hons) programmes

- Physics and Mathematics**
- Computer Science and Mathematics**

W.E.F. THE ACADEMIC YEAR 2021-22

DEPARTMENT OF MATHEMATICS

List of Courses with Credit Pattern B.Sc/B.Sc(Hons)

Sem	Programme	Programme Code	Course code	Title of the course	Course type	Credit Pattern L:T:P	Total credit	Teaching hrs/week	Evaluation pattern			
									C1	C2	C3	Total Marks
I	PM	BScPhMa32	FSA43032	Algebra-I and Calculus-I	DSC-Theory	4:0:0	4	4	20	20	60	100
				Theory based practical's on Algebra-I and Calculus-I	DSC-Practical	0:0:2	2	4	10	15	25	50
	CsM	BScCsMa34	FSA43034	Algebra-I and Calculus-I	DSC-Theory	4:0:0	4	4	20	20	60	100
				Theory based practical's on Algebra-I and Calculus-I	DSC- Practical	0:0:2	2	4	10	15	25	50
			FSA840	Mathematical Aptitude-I	OE- Theory	3:0:0	3	3	20	20	60	100
			FSA850	Business Mathematics-I	OE- Theory	3:0:0	3	3	20	20	60	100
				Optional Mathematics-I	OE- Theory	3:0:0	3	3	20	20	60	100
II	PM	BScPhMa32	FSB43032	Algebra-II (Number Theory)and Calculus-II	DSC- Theory	4:0:0	4	4	20	20	60	100
				Theory based practical's on Algebra-II (Number Theory)and Calculus-II	DSC- Practical	0:0:2	2	4	10	15	25	50
	CsM	BScCsMa34	FSB43034	Algebra-II (Number Theory)and Calculus-II	DSC- Theory	4:0:0	4	4	20	20	60	100
				Theory based practical's on Algebra-II (Number Theory)and Calculus-II	DSC- Practical	0:0:2	2	4	10	15	25	50
			FSB840	Mathematical Aptitude-II	OE- Theory	3:0:0	3	3	20	20	60	100
			FSB850	Business Mathematics-II	OE- Theory	3:0:0	3	3	20	20	60	100
				Optional Mathematics-II	OE- Theory	3:0:0	3	3	20	20	60	100

Preamble

The subject wise expert committee to draft model curriculum contents in Mathematics constituted by the Department of Higher Education, Government of Karnataka, Bangalore vide GO No. ED 260 UNE 2019 (PART-1) DATED 13.08.2021 is pleased to submit its partial report on the syllabus for the First Year (First & Second Semesters) B.Sc.(Basic/Honors) Mathematics and detailed Course Structure for B.Sc.(Honors) Mathematics and M.Sc. (One Year) Mathematics.

The committee discussed various models suggested by the Karnataka State Higher Education Council in its joint meetings with the Chairpersons of Board of Studies of all state universities in Karnataka and resolved to adopt Model IIA Bachelor of Science (Basic/Hons.) for the subjects with practical's with Mathematics as Major/Minor.

To achieve the core objectives of the National Education Policy 2020 it is unanimously resolved to introduce computer based practical's for the Discipline Core (DSC) courses by using Free and Open Source Software's (FOSS) tools for implementation of theory based on DSC courses as it is also suggested by the LOCF committee that the papers may be taught using various Computer Algebra System (CAS) software's such as Mathematica, MATLAB, Maxima and R to strengthen the conceptual understanding and widen up the horizon of students' self-experience. In view of these observations the subject expert committee suggested the software's Python /R / Maxima/ Scilab/ Maple/MatLab/Mathematica for hands on experience of implementation of mathematical concepts in computer based lab.

The expert committee suggests the implementation this curriculum structure in all the Departments of Mathematics in Universities/Colleges in Karnataka.

The subject expert committee designed the Course Learning Outcome (CO) to help the learners to understand the main objectives of studying the courses by keeping in mind of the Programme outcomes (PO) of the graduate degree with honors in Mathematics or a graduate degree with Mathematics as a major subject.

As the Mathematics subject is a vast with several branches of specializations, it is difficult for every student to learn each branch of Mathematics, even though each paper has its own importance. Hence the subject expert committee suggests number of elective papers (for both Discipline electives and Open Electives) along with Discipline Core Courses. The BoS in Mathematics of universities may include additional electives based on the expertise of their staff and needs of the students'. A student can select elective paper as per her/his needs and interest.

The subject expert committee in Mathematics suggests that the concerned Department/Autonomous Colleges/Universities to encourage their faculty members to include necessary topics in addition to courses suggested by the expert committee.

UG-Mathematics- CBCS

Name of the Degree Program	: B.Sc.
Discipline Course	: Mathematics
Year of Implementation	: 2021-22

Programme Outcomes (PO): By the end of the program the students will be able to :

PO 1	Disciplinary Knowledge : Bachelor degree in Mathematics is the culmination of in-depth knowledge of Algebra, Calculus, Geometry, differential equations and several other branches of pure and applied mathematics. This also leads to study the related areas such as computer science and other allied subjects
PO 2	Communication Skills: Ability to communicate various mathematical concepts effectively using examples and their geometrical visualization. The skills and knowledge gained in this program will lead to the proficiency in analytical reasoning which can be used for modeling and solving of real life problems.
PO 3	Critical thinking and analytical reasoning: The students undergoing this programme acquire ability of critical thinking and logical reasoning and capability of recognizing and distinguishing the various aspects of real life problems.
PO 4	Problem Solving : The Mathematical knowledge gained by the students through this programme develop an ability to analyze the problems, identify and define appropriate computing requirements for its solutions. This programme enhances students overall development and also equip them with mathematical modelling ability, problem solving skills.
PO 5	Research related skills: The completing this programme develop the capability of inquiring about appropriate questions relating to the Mathematical concepts in different areas of Mathematics.
PO 6	Information/digital Literacy: The completion of this programme will enable the learner to use appropriate softwares to solve system of algebraic equation and differential equations.
PO 7	Self – directed learning: The student completing this program will develop an ability of working independently and to make an in-depth study of various notions of Mathematics.
PO 8	Moral and ethical awareness/reasoning : : The student completing this program will develop an ability to identify unethical behavior such as fabrication, falsification or misinterpretation of data and adopting objectives, unbiased and truthful actions in all aspects of life in general and Mathematical studies in particular.
PO 9	Lifelong learning: This programme provides self directed learning and lifelong learning skills. This programme helps the learner to think independently and develop algorithms and computational skills for solving real word problems.
PO 10	Ability to peruse advanced studies and research in pure and applied Mathematical sciences.

PROGRAMME SPECIFIC OUTCOME:

**Bachelor of Science in (i) Physics and Mathematics and
(ii) Computer Science and Mathematics**

After completing the graduation in the above discipline, they are able to :

PSO1.Find carrier opportunities in organization , industries/civil service exam/teaching where a high level competence in mathematics and physics is demanded

PSO2.Develop proficiency in the analysis of complex physical problems and the use mathematical or other appropriate technique to solve them

PSO3.Apply for research in mathematics or in physics

PSO4.Apply for research in Mathematics or computer science

PSO5.Find jobs at all levels in corporate field

PSO6. Demonstrate skills in the use of computers for control, data acquisition and data analysis in experimental investigation.

Assessment

UG-Mathematics- CBCS

Weightage for the Assessments (in percentage) for first two semesters

Type of Course	Formative Assessment/ Internal Assessment	Summative Assessment (S.A.)
Theory	40%	60%
Practical	50%	50%
Projects	--	--
Experiential Learning (Internship etc.)	--	--

THEORY EXAMINATION (For Discipline Specific Courses):

UG-Mathematics- CBCS

Internal Assessment

(i) C1 Component: 20 Marks. This will be based on test for 10 marks and seminar for 10 marks. This should be completed by the 8th week of the semester.

C2 Component: 20 Marks. This will be based on test for 10 marks and assignment for 10 marks. This should be completed by the 15th week of the semester.

(ii) C3 component (Main Examination of 2 hours duration): 60 Marks. The pattern of the question paper will be as follows:

There will be 04 questions. All questions must be answered. All questions carry 15 marks.

Question 1. This question covers unit I of the syllabus. There will be 5 sub- questions each carrying 5 marks. The student has to answer any three of the 5 sub-questions.

Question 2. This question covers unit II of the syllabus. There will be 5 sub- questions each carrying 5 marks. The student has to answer any three of the 5 sub-question

Question 3. This question covers unit III of the syllabus. There will be 5 sub- questions each carrying 5 marks. The student has to answer any three of the 5 sub-questions.

Question 4. This question covers unit IV of the syllabus. There will be 5 sub- questions each carrying 5 marks. The student has to answer any three of the 5 sub-questions.

PRACTICAL EXAMINATION (For Discipline Specific Courses):

UG-Mathematics- CBCS

(i) Internal Assessment: 25 (10 +10+5)

This will be based on C1-test (10 marks), C2-test (10 marks), Seminar/practical record maintenance (5 marks)). This should be completed by the 15th week of the semester.

(ii) Main Examination (2 hours duration):25 (20 + 5)

There will be 3 questions each carrying equal marks. The student has to answer any two of the 3 questions. Each student will be subjected to viva-voce examination, based on practical syllabus, for 5 marks.

THEORY EXAMINATION (For Open Elective (OE) papers):

(i) Internal Assessment

C1 Component: 20 Marks. This will be based on test for 10 marks and seminar for 10 marks. This should be completed by the 8th week of the semester.

C2 Component: 20 Marks. This will be based on test for 10 marks and assignment for 10 marks. This should be completed by the 15th week of the semester.

(ii) C3 component (Main Examination of 2 hours duration): 60 Marks. The pattern of the question paper will be as follows:

There will be 03 questions. All questions must be answered. All questions carry 20 marks.

Question 1. This question covers unit I of the syllabus. There will be 6 sub- questions each carrying 5 marks. The student has to answer any four of the 6 sub-questions.

Question 2. This question covers unit II of the syllabus. There will be 6 sub- questions each carrying 5 marks. The student has to answer any four of the 6 sub-questions.

Question 3. This question covers unit III of the syllabus. There will be 6 sub- questions each carrying 5 marks. The student has to answer any four of the 6 sub-questions.

1. Minimum marks for Securing Credits: 30% in Theory Examination and 40% overall.
2. Minimum credits for getting B.Sc. Degree: As per NEP regulations.
3. Award of certificate/diploma/degree: As per NEP regulations.

Syllabus for B.Sc. with Mathematics as Major Subject & B.Sc. (Hons)

Mathematics

UG-Mathematics- CBCS

SEMESTER – I

Algebra - I and Calculus – I	
Teaching Hours: 4 Hours/Week	Credits: 4
Total Teaching Hours: 56 Hours	Max. Marks: 100 (S.A.-60 + I.A. – 40)
Course Code	FSA43032/FSA43034

Course Learning Outcomes: This course will enable the students to

- Learn to solve system of linear equations.
- Solve the system of homogeneous and non homogeneous linear of m equations in n variables by using concept of rank of matrix.
- Students will be familiar with the techniques of integration and differentiation of function with real variables.
- Students learn to solve polynomial equations.
- Learn to apply Reduction formulae.

Unit-I: Matrix: Recapitulation of Symmetric and Skew Symmetric matrices, Algebra of Matrices; Row and column reduction to Echelon form. Rank of a matrix; Inverse of a matrix by elementary operations; Solution of system of linear equations; Criteria for existence of non-trivial solutions of homogeneous system of linear equations. Solution of non-homogeneous system of linear equations. Cayley- Hamilton theorem, inverse of matrices by Cayley-Hamilton theorem (Without Proof). **14 Hours**

Unit-II: Theory of equations: Euclid's algorithm, Polynomials with integral coefficients, Remainder theorem, Factor theorem, Fundamental theorem of algebra(statement only), Irrational and complex roots occurring in conjugate pairs, Relation between roots and coefficients of a polynomial equation, Symmetric functions, Transformation, Reciprocal equations, Descartes' rule of signs, Multiple roots, Solving cubic equations by Cardon's method, Solving quartic equations by Descarte's Method. **14 Hours**

Unit-III: Polar Co-ordinates: Polar coordinates, angle between the radius vector and tangent. Angle of intersection of two curves (polar forms), length of perpendicular from pole to the tangent, pedal equations. Derivative of an arc in Cartesian, parametric and polar forms, curvature of plane curve-radius of curvature formula in Cartesian, parametric and polar and pedal forms- center of

curvature, circle of curvature.

14 Hours

UG Mathematics - CBCS
Unit IV: Successive Differentiation and Integral Calculus-I: nth Derivatives of Standard functions e^{as+b} , a^s , $(ax + b)^n$, $\log(ax + b)$, $\sin(ax + b)$, $\cos(ax + b)$, $e^{as} \sin(bx + c)$, $e^{as} \cos(bx + c)$, Leibnitz theorem and its applications.

Recapitulation of definite integrals and its properties. Reduction formulae for $\int \sin^n x \, dx$, $\int \cos^n x \, dx$, $\int \sin^n x \cos^N x \, dx$, $\int \tan^n x \, dx$, $\int \cot^n x \, dx$, $\int \sec^n x \, dx$, $\int \operatorname{cosec}^n x \, dx$, $\int x^n \sin x \, dx$, $\int x^n \cos x \, dx$, $\int x^n e^{as} \, dx$, $\int x^n (\log x)^N \, dx$ with definite limits.

14 Hour

Reference Books:

1. University Algebra - N.S. Gopala Krishnan, New Age International (P) Limited.
2. Algebra – Natarajan, Manicavasagam Pillay and Ganapathy.
3. Theory of Matrices - B S Vatsa, New Age International Publishers.
4. Matrices - A R Vasista, Krishna Prakashana Mandir.
5. Differential Calculus - Shanti Narayan, S. Chand & Company, New Delhi.
6. Applications of Calculus, Debasish Sengupta, Books and Allied (P) Ltd., 2019.
7. Calculus – Lipman Bers, Holt, Rinehart & Winston.
8. Calculus - S Narayanan & T. K. Manicavachogam Pillay, S. Viswanathan Pvt. Ltd., vol. I & II.
9. Schaum's Outline of Calculus - Frank Ayres and Elliott Mendelson, 5th ed. USA: Mc. Graw.
10. Shanthinarayan – Integral Calculus, New Delhi: S. Chand and Co. Pvt. Ltd.
11. Shanthinarayan and P K Mittal, Integral Calculus, Reprint. New Delhi: S. Chand and Co. Pvt. Ltd., 2013

UG

Theory based Practical's on Algebra - I and Calculus-I

Practical Hours : 4 Hours/Week	Credits: 2
Total Practical Hours: 56 Hours	Max. Marks: 50 (S.A.-25 + I.A.-25)
Course Code	FSA43332/FSA43334

Course Learning Outcomes: This course will enable the students to

- Learn *Free and Open Source Software (FOSS)* tools for computer programming
- Solve problem on algebra and calculus theory studied in **FSA43032/FSA43034** by using FOSS software's.
- Acquire knowledge of applications of algebra and calculus

through FOSS **Practical/Lab Work to be performed in Computer**

Lab (FOSS) Suggested Software's: Maxima/Scilab /Python/R.

Introduction to the software and commands related to the topic.

1. Computation of addition and subtraction of matrices,
2. Computation of Multiplication of matrices.
3. Computation of Trace and Transpose of Matrix
4. Computation of Rank of matrix and Row reduced Echelon form.
5. Computation of Inverse of a Matrix using Cayley-Hamilton theorem.
6. Solving the system of homogeneous and non-homogeneous linear algebraic equations.
7. Finding the nth Derivative of e^{ax} , trigonometric and hyperbolic functions
8. Finding the nth Derivative of algebraic and logarithmic functions.
9. Finding the nth Derivative of $e^{ax} \sin(bx + c)$, $e^{ax} \cos(bx + c)$.
10. Finding the roots of the equation, factoring.
11. Finding the angle between the radius vector and tangent.
12. Finding the curvatures of the given curves.

Open Elective Course

UG-Mathematics- CBCS
(For students of Science stream who have not chosen Mathematics as one of Core subjects)

Optional Mathematics – I	
Teaching Hours : 3 Hours/Week	Credits: 3
Total Teaching Hours: 42 Hours	Max. Marks: 100 (S.A.-60 + I.A. – 40)
Course Code	

Course Learning Outcomes: This course will enable the students to

- Learn to solve system of linear equations.
- Solve the system of homogeneous and non homogeneous m linear equations by using the concept of rank of matrix.
- Students will be familiar with the techniques of differentiation of function with real variables.
- Identify and apply the intermediate value theorems and L'Hospital rule.
- Learn to apply Reduction formulae.

Unit-I: Matrices: Recapitulation of Symmetric and Skew Symmetric matrices, Algebra of Matrices; Row and column reduction, Echelon form. Rank of a matrix; Inverse of a matrix by elementary operations; Solution of system of linear equations; Criteria for existence of non-trivial solutions of homogeneous system of linear equations. Solution of non-homogeneous system of linear equations. Cayley- Hamilton theorem, inverse of matrices by Cayley-Hamilton theorem (Without Proof). **14 Hours**

Unit-II: Theory of equations: Euclid's algorithm, Polynomials with integral coefficients, Remainder theorem, Factor theorem, Fundamental theorem of algebra(statement only), Irrational and complex roots occurring in conjugate pairs, Relation between roots and coefficients of a polynomial equation, Symmetric functions, Transformation, Reciprocal equations, Descartes' rule of signs, Multiple roots, Solving cubic equations by Cardon's method, Solving quartic equations by Descarte's Method. **14 Hours**

Unit-III: Polar Co-ordinates: Polar coordinates, angle between the radius vector and tangent. Angle of intersection of two curves (polar forms), length of perpendicular from pole to the tangent, pedal equations. Derivative of an arc in Cartesian, parametric and polar forms, curvature of plane curve-radius of curvature formula in Cartesian, parametric and polar and pedal forms- center of curvature, circle of curvature. **14 Hours**

UG-Mathematics- CBCS

Reference Books:

1. University Algebra - N.S. Gopala Krishnan, New Age International (P) Limited.
2. Algebra – Natarajan, Manicavasagam Pillay and Ganapathy.
3. Theory of Matrices - B S Vatsa, New Age International Publishers.
4. Matrices – A. R. Vasista, Krishna Prakashana Mandir.
5. Applications of Calculus, Debasish Sengupta, Books and Allied (P) Ltd., 2019.
6. Differential Calculus - Shanti Narayan, S. Chand & Company, New Delhi.
7. Calculus – Lipman Bers, Holt, Rinehart & Winston.
8. Calculus – S. Narayanan & T. K. Manicavachogam Pillay, S. Viswanathan Pvt. Ltd., vol. I & II.
9. Schaum's Outline of Calculus - Frank Ayres and Elliott Mendelson, 5th ed. USA: Mc. Graw.
10. Shanthinarayan – Integral Calculus, New Delhi: S. Chand and Co. Pvt. Ltd.
11. Shanthinarayan and P K Mittal, Integral Calculus, Reprint. New Delhi: S. Chand and Co. Pvt. Ltd., 2013.

Open Elective

UG-Mathematics- CBCS (For Students of all Streams)

Business Mathematics-I	
Teaching Hours : 3 Hours/Week	Credits: 3
Total Teaching Hours: 42 Hours	Max. Marks: 100 (S.A.- 60 + I.A. – 40)
Course Code	FSA850

Course Learning Outcomes: This course will enable the students to

- Translate the real word problems through appropriate mathematical modelling.
- Explain the concepts and use equations, formulae and mathematical expression and relationship in a variety of context.
- Finding the extreme values of functions.
- Analyze and demonstrate the mathematical skill require in mathematically intensive areas in economics and business.

Unit-I: Algebra – Set theory and simple applications of Venn Diagram, relations, functions, indices, logarithms, permutations and combinations. Examples on commercial mathematics. **14 Hours**

Unit - II: Matrices: Definition of a matrix; types of matrices; algebra of matrices. Properties of determinants; calculations of values of determinants upto third order; Adjoint of a matrix, elementary row and column operations; solution of a system of linear equations having unique solution and involving not more than three variables. Examples on commercial mathematics. **14 Hours**

Unit - III: Differential Calculus: Constant and variables, functions, Limits & continuity. Differentiability and Differentiation, partial differentiation, rates as a measure, maxima, minima, Partial Derivatives up to second order; Homogeneity of functions and Euler's Theorem; Total Differentials; Differentiation of implicit function with the help of total differentials, Maxima and Minima; cases of one variable involving second or higher order derivatives; Cases of two variables involving not more than one constraint. **14 Hours**

Reference Books:

1. Basic Mathematics, Allel R.G.A, Macmillan, New Delhi.
2. Mathematics for Economics, Dowling, E.T. , Schaum's Series, McGraw Hill London.
3. Quantitative Techniques in Management, Vohra, N.D., Tata McGraw Hill, New Delhi.
4. Business Mathematics, Soni R.S., Pitamber Publishing House, Delhi.

Open Elective

UG-Mathematics- CBCS

(For Students of all Streams)

Mathematical Aptitude-I	
Teaching Hours : 3 Hours/Week	Credits: 3
Total Teaching Hours: 42 Hours	Max. Marks: 100 (S.A.- 60 + I.A. – 40)
Course Code	FSA840

Course Learning Outcomes: This course will enable the students to

- have a strong base in the fundamental mathematical concepts.
- grasp the approaches and strategies to solve problems with speed and accuracy
- gain appropriate skills to succeed in preliminary selection process for recruitment

Unit-I: Number System, Types of Numbers, series (AP and GP), Algebraic operations BODMAS, Divisibility, LCM and HCF, Fraction, Simplification. **14 Hours**

Unit-II: Time and Distance, Problems based on Trains, Boats and Streams. **14 Hours**

Unit-III: Time, work and wages, Pipes and Cistern, Problems on Clock, Problems on Calendar.

14 Hours

Reference Books:

1. R.S. Aggarwal, "Quantitative Aptitude for Competitive Examinations", Revised Edition, S. Chand and Co. Ltd, New Delhi, 2018.
2. Quantitative Aptitude and Reasoning by R V Praveen, PHI publishers.
3. Quantitative Aptitude : Numerical Ability (Fully Solved) Objective Questions, Kiran Prakashan, Pratogitaprakasan, Kic X, Kiran Prakasan publishers.
4. Quantitative Aptitude for Competitive Examination by Abhijit Guha, Tata Mc Graw hill publications.

Algebra – II {Number Theory} and Calculus – II	
Teaching Hours : 4 Hours/Week	Credits: 4
Total Teaching Hours: 56 Hours	Max. Marks: 100 (S.A.-60 + I.A. – 40)
Course Code	FSB43032/FSB43034

Course Learning Outcomes: This course will enable the students to

- Learn the concept of Divisibility.
- Learn about prime and composite numbers.
- Learn the concept of congruences and its applications.
- Identify and apply the intermediate value theorems and L’Hospital rule.
- Understand the concept of differentiation and fundamental theorems in differentiation and various rules.
- Find the extreme values of functions of two variables.
- Students learn to find areas and volumes using integration.

Unit-I: Number Theory: Division Algorithm, Divisibility, Prime and composite numbers, Euclidean algorithm, Fundamental theorem of Arithmetic, The greatest common divisor and least common multiple. Congruences, Linear congruences, Simultaneous congruences, Euler’s Phi-function, Wilson’s, Euler’s and Fermat’s Theorems and their applications. **14 Hours**

Unit-II: Differential Calculus-I: Limits, Continuity, Differentiability and properties. Properties of continuous functions. Intermediate value theorem, Rolle’s Theorem, Lagrange’s Mean Value theorem, Cauchy’s Mean value theorem and examples. Taylor’s theorem, Maclaurin’s series, Indeterminate forms and evaluation of limits using L’Hospital rule. **14 Hours**

Unit-III: Partial Derivatives: Functions of two or more variables-explicit and implicit functions, partial derivatives. Homogeneous functions- Euler’s theorem and extension of Euler’s theorem, total derivatives, differentiation of implicit and composite functions, Jacobians and standard properties and illustrative examples. Taylor’s and Maclaurin’s series for functions of two variables, Maxima-Minima of functions of two variables.

Unit-IV: Integral Calculus-II: *Line integral:* Definition of line integral and basic properties, examples on evaluation of line integrals. *Double integral:* Definition of Double integrals and its conversion to iterated integrals. Evaluation of double integrals by changing the order of integration and change of variables. Computation of plane surface areas using double integrals. *Triple integral:* Definition of triple integrals and evaluation- change of variables, volume as triple integral. **14 Hours**

Reference Books:

1. Differential Calculus, Shantinayakan, S. Chand & Company, New Delhi.
2. Applications of Calculus, Debasish Sengupta, Books and Allied (P) Ltd., 2019.
3. Calculus – Lipman Bers, Holt, Rinehart & Winston.
4. Calculus - Shanthinarayanan & T. K. Manicavachogam Pillay, S. Viswanathan Pvt. Ltd., vol. I & II.
5. Schaum's Outline of Calculus - Frank Ayres and Elliott Mendelson, 5th ed. USA:Mc. Graw Hill, 2008.
6. Integral Calculus, Shanthinarayan, New Delhi: S. Chand and Co. Pvt. Ltd.
7. Integral Calculus, Shantinayakan and P K Mittal, S. Chand and Co. Pvt. Ltd.
8. Text Book of B.Sc. Mathematics, G K Ranganath, S Chand & Company.
9. David M Burton, Elementary Number Theory, 6th edition, McCraw Hill, 2007.
10. Emil Grosswald, Topics from the Theory of Numbers, Modern Birhauser, 1984.
11. Ivan Niven, Herbert S. Zuckerman and Hugh L. Montgomery, An Introduction to the Theory of Numbers, John Willey (New York), 1991

Theory based practical's On Algebra – II (Number Theory) and Calculus – II	
Practical Hours : 4 Hours/Week	Credits: 2
Total Practical Hours: 56 Hours	Max. Marks: 50 (S.A.-25 + I.A. -25)
Course Code	FSB43332/FSB43334

Course Learning Outcomes: This course will enable the students to

- Learn *Free and Open Source Software (FOSS)* tools for computer programming
- Solve problem on algebra and calculus by using FOSS software's.
- Acquire knowledge of applications of algebra and calculus through FOSS

Practical/Lab Work to be performed in Computer Lab

Suggested Software's: Maxima/Scilab /Python/R.

1. Programs related to Number Theory.
2. *Program to verify Mean value theorems.*
3. Program for finding the Taylor's and Maclaurin's expansions of the given functions.
4. Program to verify the Euler's theorem and its extension.
5. Programs to construct series using Maclaurin's expansion for functions of two variables.
6. Program to evaluate the line integrals with constant and variable limits.
7. Program to evaluate the Double integrals with constant and variable limits.
8. Program to evaluate the Triple integrals with constant and variable limits.

Open Elective

UG-Mathematics- CBCS

(For students of Science stream who have not chosen Mathematics as one of the Core subjects)

Optional Mathematics – II	
Teaching Hours : 3 Hours/Week	Credits:3
Total Teaching Hours: 42 Hours	Max. Marks: 100 (S.A.- 60 + I.A. – 40)

Course Learning Outcomes: This course will enable the students to

- Learn the concept of Divisibility.
- Learn about prime and composite numbers.
- Learn the concept of congruences and its applications.
- Understand the concept of differentiation and fundamental theorems in differentiation and various rules.
- Find the extreme values of functions of two variables.
- To understand the concepts of multiple integrals and their applications.

Unit-I: Number Theory: Division Algorithm, Divisibility, Prime and composite numbers, Euclidean algorithm, Fundamental theorem of Arithmetic, The greatest common divisor and least common multiple. Congruences, Linear congruences, Simultaneous congruences, Euler's Phi-function, Wilson's, Euler's and Fermat's Theorems and their applications. **14 Hours**

Unit-II: Partial Derivatives: Functions of two or more variables-explicit and implicit functions, partial derivatives. Homogeneous functions- Euler's theorem and extension of Euler's theorem, total derivatives, differentiation of implicit and composite functions, Jacobians and standard properties and illustrative examples. Taylor's and Maclaurin's series for functions of two variables, Maxima-Minima of functions of two variables.

14 Hours

Unit-III: Integral Calculus: *Line integral:* Definition of line integral and basic properties, examples on evaluation of line integrals. *Double integral:* Definition of Double integrals and its conversion to iterated integrals. Evaluation of double integrals by changing the order of integration and change of variables. Computation of plane surface areas, *Triple integral:* Definition of triple integrals and evaluation-change of variables, volume as triple integral.

14 Hours

Reference Books:

UG-Mathematics- CBCS

1. Differential Calculus, Shanti Narayan, S. Chand & Company, New Delhi.
2. Applications of Calculus, Debasish Sengupta, Books and Allied (P) Ltd., 2019.
3. Calculus – Lipman Bers, Holt, Rinehart & Winston.
4. Calculus - Shanthinarayanan & T. K. Manicavachogam Pillay, S. Viswanathan Pvt. Ltd., vol. I & II.
5. Schaum's Outline of Calculus - Frank Ayres and Elliott Mendelson, 5th ed. USA:Mc. Graw Hill, 2008.
6. Integral Calculus, Shanthinarayan, S. Chand and Co. Pvt. Ltd.
7. Integral Calculus, Shantinarayan and P K Mittal, S. Chand and Co. Pvt. Ltd.
8. Text Book of B.Sc. Mathematics, G K Ranganath, S Chand & Company.
9. David M Burton, Elementary Number Theory, 6th edition, McCraw Hill, 2007.
10. Emil Grosswald, Topics from the Theory of Numbers, Modern Birhauser, 1984.
11. Ivan Niven, Herbert S. Zuckerman and Hugh L. Montgomery, An Introduction to the Theory of Numbers, John Willey (New York), 1991.

Open Elective

(For Students of all streams)

Business Mathematics-II	
Teaching Hours : 3 Hours/Week	Credits: 3
Total Teaching Hours: 42 Hours	Max. Marks: 100 (S.A.- 60 + I.A. – 40)
Course Code	FSB850

Course Learning Outcomes: This course will enable the students to

- Integrate concept in international business concept with functioning of global trade.
- Evaluate the legal, social and economic environment of business.
- Apply decision-support tools to business decision making.
- Will be able to apply knowledge of business concepts and functions in an integrated manner.

Unit - I: Commercial Arithmetic: Interest: Concept of Present value and Future value, Simple interest, Compound interest, Nominal and Effective rate of interest, Examples and Problems Annuity: Ordinary Annuity, Sinking Fund, Annuity due, Present Value and Future Value of Annuity, Equated Monthly Installments (EMI) by Interest of Reducing Balance and Flat Interest methods, Examples and Problems. **14 Hours**

Unit - II: Measures of central Tendency and Dispersion: Frequency distribution: Raw data, attributes and variables, Classification of data, frequency distribution, cumulative frequency distribution, Histogram and give curves. Requisites of ideal measures of central tendency, Arithmetic Mean, Median and Mode for ungrouped and grouped data. Combined mean, Merits and demerits of measures of central tendency, Geometric mean: definition, merits and demerits, Harmonic mean: definition, merits and demerits, Choice of A.M., G.M.and H.M. Concept of dispersion, Measures of dispersion: Range, Variance, Standard deviation (SD) for grouped and ungrouped data, combined SD, Measures of relative dispersion: Coefficient of range, coefficient of variation. Examples and problems. **14 Hours**

Unit - III: Correlation and regression: Concept and types of correlation, Scatter diagram, Interpretation with respect to magnitude and direction of relationship. Karl Pearson's coefficient of correlation for ungrouped data. ~~UG-Mathematics-OBES~~ correlation coefficient. (with tie and without tie) Concept of regression, Lines of regression for ungrouped data, predictions using lines of regression. Regression coefficients and their properties (without proof). Examples and problems. **14 Hours**

**Reference Books:
UG-Mathematics- CBCS**

1. Practical Business Mathematics, S. A. Bari New Literature Publishing Company New Delhi
2. Mathematics for Commerce, K. Selvakumar Notion Press Chennai
3. Business Mathematics with Applications, Dinesh Khattar & S. R. Arora S. Chand Publishing New Delhi
4. Business Mathematics and Statistics, N.G. Das & Dr. J.K. Das McGraw Hill New Delhi
5. Fundamentals of Business Mathematics, M. K. Bhowal, Asian Books Pvt. Ltd New Delhi
6. Mathematics for Economics and Finance: Methods and Modelling, Martin Anthony and Norman, Biggs Cambridge University Press Cambridge
7. Financial Mathematics and its Applications, Ahmad Nazri Wahidudin Ventus Publishing APS Denmark
8. Fundamentals of Mathematical Statistics, Gupta S. C. and Kapoor V. K., Sultan Chand and Sons, New Delhi.
9. Statistical Methods, Gupta S. P.: Sultan Chand and Sons, New Delhi.
10. Applied Statistics, Mukhopadhyaya Parimal New Central Book Agency Pvt. Ltd. Calcutta.
11. Fundamentals of Statistics, Goon A. M., Gupta, M. K. and Dasgupta, B. World Press Calcutta.
12. Fundamentals of Applied Statistics, Gupta S. C. and Kapoor V. K., Sultan Chand and Sons, New Delhi.

Open Elective

(For Students of all Streams)

Mathematical Aptitude-II	
Teaching Hours : 3 Hours/Week	Credits: 3
Total Teaching Hours: 42 Hours	Max. Marks: 100 (S.A.- 60 + I.A. – 40)
Course Code	FSB840

Course Learning Outcomes: This course will enable the students to

- have a strong base in the fundamental mathematical concepts.
- grasp the approaches and strategies to solve problems with speed and accuracy
- gain appropriate skills to succeed in preliminary selection process for recruitment

Unit-I: Percentage, Average, Problems based on Ages, Ratio and Proportion, Partnership and share, Mixtures. **14 Hours**

Unit-II: Profit, Loss and Discount, Simple Interest, Compound Interest, Shares and Debentures. **14 Hours**

Unit-III: Permutations and Combinations, Probability, True discount and Banker's discount.

14 Hours

Reference Books:

1. R.S. Aggarwal, "Quantitative Aptitude for Competitive Examinations", Revised Edition, S. Chand and Co. Ltd, New Delhi, 2018.
2. Quantitative Aptitude and Reasoning by R V Praveen, PHI publishers.
3. Quantitative Aptitude : Numerical Ability (Fully Solved) Objective Questions, Kiran Prakashan, Pratogitaprakasan, Kic X, Kiran Prakasan publishers.
4. Quantitative Aptitude for Competitive Examination by Abhijit Guha, Tata Mc Graw hill publications.

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(An autonomous College of University of Mysuru)
Re-accredited by NAAC with 'A' grade
Ooty road, Mysuru-570 025, Karnataka



ESTD-1964

**CHOICE BASED CREDIT SYSTEM
CONTINUOUS ASSESSMENT GRADING PATTERN
(CBCS-CGPA)**

B. Sc., DEGREE SYLLABUS (CBCS)

MICROBIOLOGY

(Amended syllabus to be implemented W. E. F. (2020 – 2021))

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE, OOTY ROAD, MYSURU-25
DEPARTMENT OF MICROBIOLOGY
PROFORMA OF INSTRUCTIONS AND EXAMINATION FOR B.Sc. PROGRAMME IN MICROBIOLOGY (CBCS)
DURATION OF THE COURSE: 3YEARS (6SEMESTER)
PROGRAMME:BScBMBt, PROGRAMME CODE:BSc06 (2017-18)

Year	Semester	Course code & Core course	Title of the paper	Lecture + Practicals hours per week	No. of credits			Total credits	Total hours		Percentage			Maximum Marks in exam/Assessment			Exam Duration	
					L	T	P		Th	Pr	Th	Pr	IA	Th	Pr	IA	Th	Pr
I B.Sc	I	CMA28006 DSC-I :Theory	Introduction to Microbiology and Microbial diversity	04	4	-	2	06	60	60	50	20	30	70	70	30	3h	3h
		DSC-I: Pract-I	Introduction to Microbiology and Microbial diversity: Based on theory	04														
	II	CMB28006 DSC-II: Theory	Bacteriology	04	4	-	2	06	60	60	50	20	30	70	70	30	3h	3h
		DSC-II: Pract-II	Bacteriology: Based on theory	04														
II B.Sc	III	CMC28006 DSC-III:Theory	Microbial Physiology and Metabolism	04	4	-	2	06	60	60	50	20	30	70	70	30	3h	3h
		DSC-III: PractIII	Microbial Physiology and Metabolism Based on theory	04														
	IV	CMD28006 DSC-IV: Theory	Microbial Genetics and Genetic Engineering	04	4	-	2	06	60	60	50	20	30	70	70	30	3h	3h
		DSC-IV: Pract-IV	Microbial Genetics and Genetic Engineering Based on theory	04														
III B.Sc.	V	CME28006 / CME28206 DSE-V: Theory	No. of courses:1 DSE- A: Environmental Microbiology DSE-B: Agricultural Microbiology	04	4	-	1	5	60	45	50	20	30	70	70	30	3h	3h
		DSE- V:Pract-V	Based on theory	02														
		CME28406/ CME28606 SEC	No. of courses:1 SEC-A : Microbial diagnosis in health clinics SEC-B: Microbial analysis of Air and water	02														
	VI	CMF28006/ CMF28206 DSE-VI: Theory	No. of courses:1 DSE-A:Industrial and Food Microbiology DSE -B : Medical Microbiology and immunology	04	4	-	1	5	60	45	50	20	30	70	70	30	3h	3h
		DSE-VI: Pract-VI	Based on theory	02														

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE, OOTY ROAD, MYSURU-25
DEPARTMENT OF MICROBIOLOGY
PROFORMA OF INSTRUCTIONS AND EXAMINATION FOR B.Sc. PROGRAMME IN MICROBIOLOGY (CBCS)
DURATION OF THE COURSE: 3YEARS (6SEMESTER)
PROGRAMME:BScBBM, PROGRAMME CODE:BSc07 (2017-18)

Year	Semester	Course code & Core course	Title of the paper	Lecture + Practicals hours per week	No. of credits			Total credits	Total hours		Percentage			Maximum Marks in exam/Assessment			Exam Duration	
					L	T	P		Th	Pr	Th	Pr	IA	Th	Pr	IA	Th	Pr
I B.Sc	I	CMA28007 DSC-I :Theory	Introduction to Microbiology and Microbial diversity	04	4	-	2	06	60	60	50	20	30	70	70	30	3h	3h
		DSC-I: Pract-I	Introduction to Microbiology and Microbial diversity: Based on theory	04														
	II	CMB28007 DSC-II: Theory	Bacteriology	04	4	-	2	06	60	60	50	20	30	70	70	30	3h	3h
		DSC-II: Pract-II	Bacteriology: Based on theory	04														
II B.Sc	III	CMC28007 DSC-III:Theory	Microbial Physiology and Metabolism	04	4	-	2	06	60	60	50	20	30	70	70	30	3h	3h
		DSC-III: PractIII	Microbial Physiology and Metabolism Based on theory	04														
	IV	CMD28007 DSC-IV: Theory	Microbial Genetics and Genetic Engineering	04	4	-	2	06	60	60	50	20	30	70	70	30	3h	3h
		DSC-IV: Pract-IV	Microbial Genetics and Genetic Engineering Based on theory	04														
III B.Sc.	V	CME28007 / CME28207 DSE: Theory	No. of courses:1 DSE- A: Environmental Microbiology DSE-B: Agricultural Microbiology	04	4	-	1	5	60	45	50	20	30	70	70	30	3h	3h
		DSE- V:Pract-V	Based on theory	02														
		CME28407/ CME28607 SEC	No. of courses:1 SEC-A : Microbial diagnosis in health clinics SEC-B: Microbial analysis of Air and water	02														
	VI	CMF28007/ CMF28207 DSE: Theory	No. of courses:1 DSE-A:Industrial and Food Microbiology DSE -B : Medical Microbiology and immunology	04	4	-	1	5	60	45	50	20	30	70	70	30	3h	3h
		DSE	Based on theory	02														
		Pract-VI																
				Total credits				36										

B.Sc., Microbiology

DEPARTMENT OF MICROBIOLOGY

PROGRAMME: BSc BMBT

PROGRAMME CODE: BSC06

PROGRAMME OUTCOMES: B.Sc., BMBT

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate the ability to justify and explain their thinking and/or approach, both written and oral. Demonstrate the ability to present clear, logical and succinct arguments, including prose and mathematical language. Write and speak using professional norms, and demonstrate an ability to collaborate effectively.
- PO2. Develop state-of-the-art laboratory skills and professional communication skills.
- PO3. Apply the scientific method to design, execute, and analyze an experiment and also to explain their scientific procedures as well as their experimental observations.
- PO4. Demonstrate an understanding of fundamental biochemical principles, structure and biological function of biomolecules, metabolic pathways and their regulation.
- PO5. Work as a laboratory technician, biochemists or medical scientist.
- PO6. Possess knowledge of ethical practices in science.
- PO7. Describe/ explain the processes used by microorganisms for their replication, survival, and interaction with their environment and host populations.
- PO8. Explain the theoretical basis of the tools, technologies and methods common to microbiology.
- PO9. Apply the scientific method as a demonstration that they understand its application furthering our knowledge of the microbial world.
- PO10. Design and develop solution to Biotechnology problems by applying appropriate tools while keeping in mind safety factor for environmental & society.
- PO11. Create, select, and apply appropriate techniques, resources, and modern tools including prediction and modelling to different activities with an understanding of the limitations.
- PO12. Support biotechnology research activity with strong technical background knowledge.

B.Sc., Microbiology

PROGRAMME: BSc BBM

PROGRAMME CODE: BSC07

PROGRAMME OUTCOMES: B.Sc., BBM

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Identify the taxonomic position of plants using principles and methods of nomenclature and classification in Botany.
- PO2. Understand the impact of the plant diversity in societal and environmental contexts, and demonstrate the knowledge of, and need for sustainable development.
- PO3. Use interdisciplinary approaches with quantitative skills to work on biological problems.
- PO4. Demonstrate the ability to justify and explain their thinking and/or approach, both written and oral. Demonstrate the ability to present clear, logical and succinct arguments, including prose and mathematical language. Write and speak using professional norms, and demonstrate an ability to collaborate effectively.
- PO5. Develop state-of-the-art laboratory skills and professional communication skills.
- PO6. Apply the scientific method to design, execute, and analyze an experiment and also to explain their scientific procedures as well as their experimental observations.
- PO7. Demonstrate an understanding of fundamental biochemical principles, structure and biological function of biomolecules, metabolic pathways and their regulation.
- PO8. Work as a laboratory technician, biochemists or medical scientist.
- PO9. Possess knowledge of ethical practices in science.
- PO10. Describe/ explain the processes used by microorganisms for their replication, survival, and interaction with their environment, hosts, and host populations.
- PO11. Explain the theoretical basis of the tools, technologies and methods common to microbiology.
- PO12. Apply the scientific method as a demonstration that they understand its application furthering our knowledge of the microbial world.

B.Sc., Microbiology

DEPARTMENT OF MICROBIOLOGY

PROGRAMME: BSc BMBt

PROGRAMME CODE: BSC06

PROGRAMME SPECIFIC OUTCOME: B.Sc., BMBt

After completing the graduation in the Bachelor of Science the students are able to;

PSO 1: Gain and understanding of biochemical and molecular processes that occur in and between cells to expand understanding of biology

PSO2: Be knowledgeable in proper procedures and regulations in handling and disposal of chemicals.

PSO3: Communicate scientific information effectively, especially relating to microbes and their role in ecosystem and health related issues.

PSO4: Acquire, articulate, retain and demonstrate laboratory safety skills applicable to microbiological research or clinical methods, including accurately reporting observations and analysis.

PSO5: Demonstrate effectively the applications of biochemical and biological sciences

PSO6: Decide and apply appropriate tools and techniques in biotechnological manipulation.

PSO7: Justify societal, health, safety and legal issues and understand his or her responsibilities in biotechnological practices.

PROGRAMME: BSc BBM

PROGRAMME CODE: BSC07

PROGRAMME SPECIFIC OUTCOME: B.Sc., BBM

After completing the graduation in the Bachelor of Science the students are able to;

PSO 1: Demonstrate effectively the applications of biochemical and biological sciences.

PSO2: Inculcating proficiency in all experimental techniques and methods of analysis.

PSO3: Acquire, articulate, retain and demonstrate laboratory safety skills applicable to microbiological research or clinical methods, including accurately reporting observations and analysis.

PSO4: Communicate scientific information effectively, especially relating to microbes and their role in ecosystem and health related issues.

PSO5: Be knowledgeable in proper procedures and regulations in handling and disposal of chemicals.

PSO6: Gain and understanding of biochemical and molecular processes that occur in and between cells to expand understanding of biology

B.Sc., Microbiology

DSC-I

I B.Sc., I SEMESTER

TITLE: INTRODUCTION TO MICROBIOLOGY AND MICROBIAL DIVERSITY

60hrs (4hrs/week)

(THEORY)

credits: 4

COURSE CODE-CMA28006 / CMA28007

Course outcome:

After successful completion of this course students are able to:

- CO1:** Gain basic knowledge about Microbiology starting from history to Microorganisms.
- CO2:** An entire picture about the taxonomical classification of Microbes.
- CO3:** Understand the basic microbial structure, function and study of the comparative characteristics of prokaryotes and eukaryotes.
- CO4:** Understand the structural similarities and differences among various physiological groups of fungi, protozoa and algae.
- CO5:** Know how viruses are classified and understand the structure of viruses.
- CO6:** Know the replication strategies of representative viruses.

UNIT: I

No. of Hours: 15

HISTORY OF DEVELOPMENT OF MICROBIOLOGY

- A.** Milestones in the historical development of Microbiology. Germ theory of disease, Development of various microbiological techniques. Golden era of microbiology: Contributions of Anton von Leeuwenhoek, Louis Pasteur, Robert Koch, Joseph Lister, Alexander Fleming.
- B.** Development in the field of Soil Microbiology: Contributions of Martinus W. Beijerinck, Sergei N. Winogradsky and Selman A. Waksman.
- C.** Development in the fields of Medical Microbiology and Immunology: Contributions of Paul Ehrlich, Elie Metchnikoff and Edward Jenner.
- D.** Recent developments in the field of Microbiology.
- E.** Branches of Microbiology.
- F.** Scope of Microbiology

UNIT: II

No. of Hours: 15

MICROBIAL DIVERSITY

A. Systems of classification

Definition of taxonomy and systematics. Taxonomic ranks. Classification systems - artificial and phylogenetic. Numerical taxonomy. System of classification: Haeckel's three- kingdom, Whittaker's five-kingdom classification and Cavalier-Smith's eight kingdom classification.

General characteristics of different groups – a. Acellular microorganisms: Virus, Viroids, Prions. b. Cellular microorganisms: Bacteria, Algae, Fungi and Protozoa
Difference between prokaryotic and eukaryotic microorganisms.

B. Algae

- a.** History of phycology with emphasis on contributions of Indian scientists; Ghosh, M.O.P. Iyengar, T.V. Desikachary, Y. Bhardwaja, M. S. Randhawa and R. N. Singh (in brief).

B.Sc., Microbiology

- b.** Structure of typical algal cell (E.g: *Chlamydomonas*) - occurrence, thallus organization, Pigments, flagella, eyespot, food reserves and vegetative, asexual and sexual reproduction.
- c.** Outline classification (Fritsch, 1935).
- d.** Study of thallus structure, reproduction and economic importance of the following:
Chlorella, Spirogyra, Diatoms and Gracilaria

UNIT: III

No. of Hours: 15

FUNGI AND PROTOZOA

A. Fungi

- a.** Historical development of Mycology including significant contributions of eminent Mycologists: E J Butler, J F Dastur and C.V.Subramanian.
- b.** General characteristics of fungi including habitat, distribution, nutritional requirements, fungal cell ultra- structure, thallus organization and aggregation, fungal wall structure, asexual and sexual reproduction.
Definition- Heterokaryosis, Heterothallism and Parasexuality.
- c.** Outline classification as per Alexopoulos and Mims (1979)
- d.** Study of thallus structure, reproduction, life cycle and economic importance of the following: *Pythium, Saccharomyces, Penicillium, Agaricus and Fusarium.*

B. Protozoa

Outline classification, Morphology, reproduction and life cycle of: *Euglena, Paramecium, Entamoeba* and *Plasmodium.*

UNIT: IV

No. of Hours: 15

VIRUSES

- A.** Definition, early developments in Virology. General properties of viruses – size, shape and chemical composition, viral classification.
- B.** Study of structure of the following viruses:
 - 1 Bacteriophages – T-4 phage (replication in brief)
 - 2 Cyanophages
 - 3 Phytoviruses – TMV
 - 4 Zooviruses – Influenza virus and HIV
- C.** Significance of Viruses
- D.** Viroids and Prions-a brief account.

Total marks 100: 50(Theory) + 30 (C1+C2)+ 20 (Practicals)

B.Sc., Microbiology

DSC-I

I B.Sc., I SEMESTER

**TITLE: INTRODUCTION TO MICROBIOLOGY AND MICROBIAL DIVERSITY
(PRACTICALS)**

TOTAL HOURS: 60

CREDITS: 2

1. Microbiology Good Laboratory Practices and Biosafety.
2. Study of contributions of microbiologists
3. Study of typical prokaryotic and eukaryotic cell
4. Demonstration of the presence of microflora in the environment by exposing nutrient agar Plates to air.
5. Staining and mounting of Algae (Eg. *Spirogyra*) and Fungi (Eg. *Rhizopus*).
- 6-7. Study of the following Algae – *Chlamydomonas*, *Chlorella*, *Spirogyra*, *Diatoms* and *Gracilaria*
- 8-10. Study of the following Fungi – *Pythium*, *Rhizopus*, *Saccharomyces*, *Penicillium*, *Agaricus* and *Fusarium*
11. Microscopic examination of free-living Protozoa of a pond.
- 12-13. Study of the following Protozoans – *Euglena*, *Paramecium*, *Entamoeba*. And *Plasmodium*
14. Demonstration of plaque assay for coliphages.
15. Study of photographs of the following: Bacteriophages, TMV and HIV

DSC-II

B.Sc., Microbiology

**I B.Sc., II SEMESTER
TITLE: BACTERIOLOGY**

TOTAL HOURS: 60hrs (4hrs/week) (THEORY)

CREDITS: 4

COURSE CODE-CMB28006 / CMB28007

Course Outcome:

Enable the students to have sound knowledge about:

CO1: Bacteria, microscopes and basic laboratory techniques.

CO2: Demonstrate theory and practical skills in microscopy, their handling techniques and staining procedures.

CO3: Various Culture media and their applications and also understand various physical and chemical means of sterilization.

CO4: Know about microbial techniques for isolation of pure cultures of bacteria.

CO5: To identify the bacteria based on staining and cultural characteristics.

CO6: Able to perform routine culture handling tasks safely and effectively.

CO7: The maintenance and preservation of cultures.

UNIT I

No. of Hours: 15

BACTERIAL CELL ORGANIZATION

- A. Outline classification of bacteria as per Bergey's manual of Systematic Bacteriology. Occurrence, shape and arrangement of bacterial cell. Structure of eubacteria- cell wall (Gram positive, Gram negative, L-forms), Glycocalyx, capsule, cell membranes, periplasmic space, flagella, fimbriae, cilia and pili. Cell Membrane. Cytoplasm: Ribosomes, mesosomes, inclusion bodies, nucleoid, chromosome and plasmids. Endospore: Structure, formation and stages of sporulation. Reproduction in Bacteria. General characteristics of Archaea.
- B. Bacterial diversity:
- Methanogens, Rickettsiae, Chlamydiae, Mollicutes (Mycoplasmas), Spirochaetes and Actinomycetes
 - Cyanobacteria: Occurrence, structure, reproduction and economic importance of the following: *Microcystis*, *Spirulina* & *Anabaena*

Unit: II

No. of Hours: 15

BACTERIOLOGICAL TECHNIQUES

A. Cultivation of bacteria

- Culture media – Types, Cultivation of aerobic and anaerobic bacteria.
- Pure culture and Cultural characteristics: Pure culture techniques- Serial dilution, Pour plate, Spread plate, Streak plate and Micromanipulator technique. Cultural characteristics of bacteria – plate cultures/solid media and broth cultures/liquid media.
- Maintenance and Preservation of pure cultures – Sub culturing, overlaying with mineral oil, refrigeration (4°C), lyophilization and cryopreservation.

B. Microbiological stains and staining techniques

- Types of stains:** Acidic (Nigrosin), Basic (Crystal violet, Methylene blue); Stains for bacteria (Methylene blue, Nigrosin), Mechanisms of staining (in brief).

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- b. Preparation of bacterial smears for light microscopy** – fixation, simple staining, Negative staining, Differential staining – Gram's staining and Acid fast staining; Structural staining – capsule, flagella, cell wall, endospore and nuclear staining.
- c. Hanging drop method for bacterial motility.**

UNIT: III

No. of Hours: 15

MICROSCOPY

- A. Light Microscope:**
 - a.** Different types of microscopes, their construction and working principles. Simple microscope (dissection microscope), Compound microscope - bright field, dark field, phase contrast, stereomicroscope and fluorescence microscope.
 - b.** Micrometry.
- B. Electron Microscope:** Principle, construction and applications of Scanning and Transmission electron microscopes. Preparation of specimens for electron microscopic studies: TEM – Dehydration and fixation, ultra sectioning, Negative staining, shadow casting and freeze etching (in brief) and SEM – Dehydration, shadow casting and surface replica (in brief)

UNIT: IV

No. of Hours: 15

PHYSICAL AND CHEMICAL METHODS OF MICROBIAL CONTROL

Methods of sterilization

A. Physical methods:

- a) Heat –
 - i) Dry heat – Hot air oven
 - ii) Incineration –Incinerator, direct flaming.
 - iii) Moist heat method – Autoclave and Pressure cooker
 - iv) Tyndallization (fractional steam sterilization)
- b) Filtration–Types of filters: Membrane filter, HEPA filter (e.g., Laminar air flow) and Berkefeld filter (Diatomaceous earth)
- c) Radiation methods – UV rays, Gamma rays and Cathode rays

- B. Chemical method:** Definition of terms - Disinfectants, antiseptics, sanitizers, Microbicides: virucide, algicide, fungicide and sporicide. Microbistatic: bacteriostatic and fungistatic.
Use and mode of action - Alcohols, Aldehydes, Halogens, Phenols, Heavy metals, Quaternary Ammonium compounds and Sterilizing gases (ethylene oxide).

Total marks 100: 50 (Theory) + 30(C1+C2) + 20 (Practicals)

B.Sc., Microbiology

**I B.Sc., II SEMESTER
TITLE: BACTERIOLOGY
PRACTICAL**

TOTAL HOURS: 60hrs (4hrs/week)

CREDITS: 2

1. Study of photographs of microscopes mentioned in the theory syllabus
2. Study of simple and compound microscopes, including oil immersion objectives
3. Microscopic measurements of microorganisms or spores using Stage and Ocular micrometer.
4. Preparation of stains and mordant– Methylene blue, Crystal Violet, Safranin, Nigrosin, Carbol fuchsin, Malachite green and Gram's iodine.
5. Simple staining and Negative staining.
6. Differential staining (Gram's staining).
7. Structural staining- (cellwall and endospore of bacteria).
8. Demonstration of laboratory equipments – Autoclave, Pressure cooker, Hot air oven, Incubator, Refrigerator, Inoculation hood or chamber, Membrane filter, Colony counter. BOD incubator, pH meter & Biosafety cabinet.
9. Preparation of Chromic acid and its use.
10. Cleaning and Sterilization of glasswares. Preparation of culture media – Nutrient broth, Nutrient agar, Potato dextrose agar, Czapeck dox agar and Mac Conkey's agar.
11. Cultivation of microorganisms on Agar plate (Point inoculation), Broth, Anaerobic cultivation (Candle jar or Gas pack method).
12. Preparation of Physiological saline and Serial dilution.
13. Method of obtaining pure cultures of Microorganisms – Streak plate, Pour plate and Spread plate method.
14. Maintenance of pure culture – Sub culturing, Slope culture and refrigeration, Mineral oil overlay method and Stab culture
15. Demonstration of bacterial motility by Hanging drop technique

B.Sc., Microbiology

**II B.Sc., III SEMESTER
TITLE: MICROBIAL PHYSIOLOGY AND METABOLISM
THEORY**

TOTAL HOURS: 60hrs (4hrs/week)

CREDITS: 4

COURSE CODE-CMC28006 / CMC28007

After successful completion of this course students are able to:

- CO1.** Inculcate the knowledge regarding microbial growth, functions, physiology and metabolism.
- CO2.** Understand the microbial transport systems and microbial metabolism
- CO3.** Know the microbial growth in response to environmental factors.
- CO4.** Get equipped with various methods of bacterial growth measurement.
- CO5.** Know about the biological nitrogen fixation.
- CO6.** Knowledge of properties, structure, function of enzymes, enzyme kinetics and their regulation.

**UNIT I
MICROBIAL NUTRITION**

No. of Hours: 15

- A.** Classification of microorganisms based on energy- Phototroph and Chemotroph, Electron-Lithotroph and Organotroph and Carbon source- Autotroph and Heterotroph
Major nutritional type of Microorganisms: Chemolithoautotroph, Chemolithoheterotroph, Chemoheterotroph, Chemolithotroph, Photolithoautotroph and Photoorganoheterotroph. .
- B.** Nutritional requirements of Microorganisms. Elementary nutrients: Carbon, nitrogen, phosphorous, sulphur, oxygen and energy sources. Trace elements: Vitamins and Growth factors.
- C.** Uptake of nutrients: Diffusion- Simple and Facilitated, Active transport (use of Proton Motive force, ATP: ABC transporter), Group translocation, Iron uptake.

MICROBIAL GROWTH

- A.** Definition, Growth rate and generation time. The growth curve in batch culture - Phases of growth and their significance. Diauxic growth.
- B.** Microbial growth in response to environment -Temperature (psychrophiles, mesophiles, thermophiles, extremophiles, thermodurics, psychrotrophs), pH (acidophiles, alkaliphiles, neutrophiles), solute and water activity (halophiles, xerophiles, osmophilic), Oxygen (aerobic, anaerobic, microaerophilic, facultative aerobe, facultative anaerobe) and barophilic.
- D.** Measurement of growth by cell number (Haemocytometer) and cell mass (Turbidometer).
- E.** Batch culture and continuous culture of microorganisms – Chemostat, Turbidostat. Synchronization of cell division.

UNIT II

No. of Hours: 15

METABOLISM

- A. Microbial Enzymes:** Definition, Nomenclature, Classification, Properties, Mode and Mechanism of enzyme action, Factors effecting enzyme action, Enzyme regulation, Inhibition: Competitive and Noncompetitive and Allosteric enzymes, their importance.

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Cofactors and Coenzymes.

B. Nitrogen metabolism: Biological N₂ Fixation-Symbiotic and asymbiotic N₂ Fixation, nodule formation, bacteroids, Leg haemoglobin in Nitrogen fixation, Mechanism and Biochemistry of Nitrogen fixation, Role of Nitrogenase and Hydrogenase in Nitrogen fixation. Nitrogen assimilation.

C. Lipid metabolism: Breakdown of lipids by microorganisms, beta-oxidation of fatty acids.

UNIT III

No. of Hours: 15

CHEMOHETEROTROPHIC METABOLISM

A. Aerobic respiration: Concept of respiration: aerobic, anaerobic respiration and Fermentation. Ultra structure of Mitochondrion, Sugar degradation pathways i.e. EMP, ED, Pentose phosphate pathway, Formation of acetyl CoA from pyruvate, TCA cycle, Electron transport system and Oxidative phosphorylation.

B. Anaerobic respiration and Fermentation

Anaerobic respiration with special reference to dissimilatory nitrate reduction (Denitrification; nitrate/nitrite and nitrate/ammonia respiration; fermentative nitrate reduction). Fermentation - Alcohol fermentation and Pasteur effect. Lactate fermentation (homofermentative and heterofermentative pathways).

UNIT IV

No. of Hours: 15

CHEMOLITHOTROPHIC AND PHOTOTROPHIC METABOLISM

A. Introduction to aerobic and anaerobic chemolithotrophy with an example each. Hydrogen oxidation (definition and reaction) and methanogenesis (definition and reaction)

B. Photosynthesis: Definition, Photosynthetic microorganisms, Anoxygenic and Oxygenic photosynthesis, Light as a source of energy, Pigments of photosynthetic bacteria and photosynthetic apparatus in prokaryotes and eukaryotes. Mechanism of photosynthesis in bacteria. Comparison of photosynthesis in bacteria and eukaryotes.

Total marks 100: 50(Theory)+30(C1+C2)+ 20 (Practicals)

B.Sc., Microbiology

**II B.Sc., III SEMESTER
TITLE: MICROBIAL PHYSIOLOGY AND METABOLISM
PRACTICAL**

TOTAL HOURS: : 60hrs (4hrs/week)

CREDITS: 2

1. Effect of temperature on growth of microorganisms.
2. Effect of pH on growth of microorganisms.
3. Effect of carbon and nitrogen sources on growth of *E.coli*
4. Effect of salt on growth of *E. coli*
5. Study and plot the growth curve of *E. coli* by turbidometric method
6. Measurement of growth by cell number using Haemocytometer.
7. Study of bacteroids from root nodules.
8. Production of ammonia from organic compounds- Ammonification.
9. Acid and gas production from carbohydrates- Demonstration of fermentation of lactose
10. Starch hydrolysis.
11. Gelatin hydrolysis.
12. Detection of Catalase production by microorganisms.
13. Urease test
14. Isolation and culturing of photosynthetic bacteria
15. Demonstration of fermentation of glucose using Kuhne's fermentation vessel.

DSC-IV

B.Sc., Microbiology

II B.Sc., IV SEMESTER

**TITLE: MICROBIAL GENETICS AND GENETIC ENGINEERING
THEORY**

TOTAL HOURS: 60hrs (4hrs/week)

CREDITS: 4

COURSE CODE-CMD28006 / CMD28007

Course outcome

Enable the students to have sound knowledge about:

- CO1.** Genetics of microorganisms and also about recombinant DNA technology used in microbiological research.
- CO2.** Know the terms and terminologies related to molecular biology and microbial genetics.
- CO3.** Understand the properties, structure and function of genes in microorganisms at the molecular level.
- CO4.** Conceptual knowledge about DNA and RNA as a genetic material, enzymology and replication strategies.
- CO5.** Understand the molecular mechanisms involved in transcription and translation.
- CO6.** The importance of genetic code and wobble hypothesis.
- CO7.** The molecular mechanisms underlying mutations, DNA damage and repair mechanisms.
- CO8.** The concept of recombination and elucidate the gene transfer mechanisms in prokaryotes and eukaryotic microorganisms.
- CO9.** Understand about techniques in genetic engineering
- CO10.** Social and ethical issues concerning genetic engineering
- CO11.** Applications of genetic engineering in various fields

UNIT: I

No. of Hours: 15

MICROBIAL GENETICS

- A.** History and development of genetics. Chromosomes: Chromosome number, Morphology, Karyotype and Idiogram. Chemical composition. Prokaryotic and Eukaryotic chromosomal organization
Cell division: Mitosis, Meiosis and Cell cycle in brief.
- B. a.** Recombination in bacteria: Transformation, Transduction (types) and Conjugation process.
- b.** Extra-chromosomal genetic elements and their importance. Types of plasmids – F plasmid, R Plasmids, colicinogenic plasmids, Ti plasmids, linear plasmids, yeast- 2 μ plasmid.
- c.** Prokaryotic and Eukaryotic transposable elements. Transposition
- d.** Chemical basis of heredity: Evidence for DNA (Griffith experiment and Hershey and chase experiment) and RNA as genetic material (Fraenkel-Conrat's experiment).
- e.** DNA Structure: Miescher to Watson and Crick- historic perspective, Chemistry of nucleic acids. Watson and Crick model of DNA, Types of DNA, denaturation and renaturation . Organization of DNA: Prokaryotes, Eukaryotes and Viruses.
RNA Structure and function. Organelle DNA -- mitochondria and chloroplast DNA.

UNIT-II

No. of Hours: 15

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MOLECULAR GENETICS

- A. DNA Replication –Types, Modes and mechanism of DNA replication by semiconservative method, Replication in Prokaryotes (Cairn’s model). Mechanism of DNA replication: Enzymes and proteins involved in DNA replication –DNA polymerases, DNA ligase, primase, telomerase – for replication of linear ends .
- B. Genetic code – features, Wobble hypothesis and evolution of genetic code. Protein synthesis – Transcription and Translation in prokaryotes. Regulation of gene expression in prokaryotes (Lac operon concept).
- C. Gene mutation: Types of mutations. Mutagenic agents: Physical and chemical mutagens. Significance of mutations. DNA damage and repair: Photo reactivation and SOS repair

UNIT -III

No. of Hours: 15

GENETIC ENGINEERING

- A. a. Genetic engineering: Milestones in genetic engineering and biotechnology. Cloning tools; restriction modification systems: types I,II and III. mode of action, nomenclature, applications of type II restriction enzymes in genetic engineering
- b. DNA modifying enzymes and their applications: DNA polymerases, terminal deoxynucleotidyl transferase, kinases and phosphatases, and DNA ligases
- c. Cloning vectors –1. Cloning plasmids (pBR 322 and pUC 18). 2. Viruses as cloning vehicles (Lambda DNA, M13). 3. Hybrid vectors (Cosmid, YAC).
- d. Cloning host (*E. coli*) .
- B. Methods in Molecular cloning: Transformation of DNA-Calcium chloride method. Gene delivery-Microinjection, Electroporation, Biolistic method (gene gun), *Agrobacterium*- mediated delivery.
- C. Screening and detection of transformants: Blue white selection, replica plate technique and antibiotic resistance.

UNIT -IV

No. of Hours: 15

TECHNIQUES IN GENETIC ENGINEERING

- A. a. Gene cloning: DNA isolation (Phenol-Chloroform method). DNA separation by Gel electrophoresis: Agarose gel – principle and method, Transformation methods.
- b. DNA libraries: Brief account of genomic library -application
- c. Blotting – Southern and Western.
- d. Gene screening and Isolation – Nucleic acid hybridization method (DNA) – Colony and Plaque hybridization.
- e. DNA sequencing: Brief account of Sanger’s dideoxynucleotide synthetic method.
- f. DNA amplification – Principle of PCR.
- g. DNA fingerprinting- Restriction Fragment Length Polymorphism (RFLP)
- B. Applications of Genetic Engineering:
 - a. Medical Application.

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- b. Industrial Application.
- c. Agricultural Application.
- d. Environmental Application.

C. Social and ethical issues concerning Genetic Engineering.

Total marks 100: 50(Theory)+30(C1+C2)+ 20 (Practicals)

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DSC-IV

II B.SC., IV SEMESTER

**TITLE: MICROBIAL GENETICS AND GENETIC ENGINEERING
PRACTICALS**

TOTAL HOURS: 60hrs (4hrs/week)

CREDITS: 2

1. Study of mitosis in onion root.
2. Demonstration of meiosis from flower buds of onion / *Chlorophytum* / *Tradescantia*.
3. Demonstration of Bacterial Conjugation
4. Demonstration of bacterial transformation and transduction
- 5-6. Preparation of Master and Replica Plates
7. Isolation of streptomycin resistant strain of *E.coli* by gradient plate method.
8. Isolation and Quantification of Nucleic acids (DNA) from *E.coli* or Yeast.
9. Demonstration of AMES test
10. Demonstration of Amplification of DNA by PCR
11. Demonstration of Southern blotting
12. Study survival curve of bacteria after exposure to ultraviolet (UV) light
13. Isolation of Plasmid DNA from *E.coli*
- 14-15. Demonstration of the following models or photographs of – DNA, t-RNA, mRNA, Transformation, Conjugation and Transduction, Transcription, Translation and DNA replication.

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**DSE- A
V SEMESTER**

**TITLE: ENVIRONMENTAL MICROBIOLOGY
(THEORY)**

TOTAL HOURS: 60hrs (4hrs/week)

CREDITS: 4

COURSE CODE-CME28006 / CME28007

Course outcome

Enable the students to have sound knowledge about:

- CO1.** The role of microorganisms in soil, air, water, waste water and bioremediation.
- CO2.** Know about the diversity of microorganism and microbial communities inhabiting a wide range of ecological habitats.
- CO3.** Learn the occurrence, abundance and distribution of microorganisms in the environment and their role in the environment
- CO4.** Understand various biogeochemical cycles – Carbon, Nitrogen, Phosphorus cycles etc. and microbes involved in these cycles.
- CO5.** Understand various plant microbes interactions especially rhizosphere, phyllosphere and mycorrhizae and their applications especially the biofertilizers and their mass production.
- CO6.** Understand the basic principles of bioremediation.
- CO7.** The various methods to determine the Sanitary quality of water and sewage Treatment methods employed in waste water treatment

UNIT 1

No. of Hours: 15

SOIL MICROBIOLOGY

- A.** Introduction: Definition, Soil types, Soil profile and Physical characteristics of soil- Mineral particles, Organic residues, Water and Gases. Soil fertility. Role of microorganisms in soil formation (in brief).
- B.** Microbial flora of Soil: A brief account of Bacteria, Fungi, Algae, Actinomycetes, Protozoa and Viruses.
- C.** Biogeochemical cycles: Carbon cycle: Microbes involved in carbon cycle
Nitrogen cycle: Nitrogen fixation, ammonification, nitrification, denitrification and nitrate reduction
Phosphorus cycle: Phosphate immobilization and solubilisation.
Sulphur cycle: Microbes involved in sulphur cycle
- D.** Associated soil microorganisms with plants- the Rhizosphere and Rhizoplane microflora, Actinorrhizae, and Mycorrhizae (AM), Tripartite and Tetra partite association.
- E.** Interaction among soil microorganisms – Neutralism, Mutualism, Commensalism, Antagonism and Parasitism. (In brief).
Microbe-Plant interaction: Symbiotic and non symbiotic interactions
Microbe-animal interaction: Microbes in ruminants, nematophagus fungi and symbiotic luminescent bacteria

AEROBIOLOGY

- A. Introduction: Definition, history and development, aim and scope of aerobiology.
- B. Microbes and atmosphere: Atmospheric layers, sources of microorganisms, Air spora of indoor and outdoor environment. Factors affecting air spora. Significance of air borne microbes. Management of air-borne microbes. Human air borne diseases (Tuberculosis, Rhinitis and Aspergillosis).
- C. Techniques of trapping air-borne microorganisms: Impactors- The slit sampler, Hirst spore trap, Andersen sampler, Rotorod sampler, Vertical cylinder spore trap, Burkard spore traps. Impingers and Filtration. Advantages and disadvantages of the techniques.

UNIT-III

No. of Hours: 15

AQUATIC MICROBIOLOGY

- A. Introduction: Natural waters- atmospheric water, surface water and ground water. Distribution of microorganisms in aquatic environment-Neuston, plankton (Phytoplankton, Zooplankton). Aquatic microorganisms-lakes, ponds, streams, rivers estuaries, and marine plankton. Lotic and benthic population.
- B. Water pollution: Sources, water borne diseases- Viral (jaundice), Bacterial (cholera) and Protozoan (amoebic dysentery). Biological indicator of water pollution.
- C. Determination of sanitary quality of water: SPC, Tests for coliforms, MPN, IMViC reactions and membrane filter.
- D. Water purification in Municipal water supply, Parameters of potable water (According to WHO).

SEWAGE MICROBIOLOGY

- A. Introduction: Sources of waste water- Domestic, Agricultural and Industrial. Physical, chemical and microbiological characteristics of waste water
- B. Waste water treatment: Single dwelling unit-Septic tank. Municipal waste treatment – Primary (screening, coagulation and sedimentation), Secondary (trickling filter, activated sludge process, oxidation pond), Tertiary (reverse osmosis, ion exchange method and electro-dialysis in brief).
- C. Solid waste recycling- Anaerobic digestion process, Biogas and Composting.

MICROBIAL BIOREMEDIATION

In situ –Intrinsic, engineered and *Ex situ* bioremediation- Solid phase system (composting, composting process), Slurry phase system (aerated lagoons, low shear air lift reactor). Bioremediation of hydrocarbons- use of genetically engineered bacterial strains. Bioremediation of xenobiotics, Microbial leaching.

AGRICULTURAL MICROBIOLOGY

- A.** Introduction – Classification of plant diseases on the basis of spread and severity of infection
- B.** Microbes and Plant diseases - Entry of pathogens into host-prepenetration, penetration, post penetration.
- C.** Microbes in Agriculture: Biofertilizers: Definition and Types. Mass production of Bacterial inoculants (*Rhizobium* , *Azospirillum* & *Cyanobacteria*). Biopesticides: Definition, Types – Bacterial, Viral, Fungal and Protozoan, Mode of action, Microbial herbicides.
- D.** Plant diseases: Study of Symptoms, Etiology, Epidemiology, Management of the following diseases – Bean Mosaic, Sandal spike, Citrus canker, Downy mildew of Bajra, Powdery mildew of mulberry, Rust of sorghum, Blast of paddy, Red rot of sugarcane, Tikka disease of groundnut.

1. a. Isolation and identification of fungi from soil by serial dilution method.
b. Isolation and enumeration of bacteria from soil by serial dilution method.
2. Study of AM fungi
3. Isolation of Nitrogen fixing bacteria- *Rhizobium*
4. Study of antagonism between microorganisms
- 5a. Gram's staining of citrus canker specimen
b. Observation of specimens - Bean mosaic, Sandal spike, Citrus canker, Downy mildew of Bajra, Powdery mildew of mulberry, Rust of sorghum, Blast of paddy, Red rot of Sugarcane, Tikka disease of groundnut.
6. Isolation of airborne microorganisms (Bacteria and Fungi) by Petriplate exposure method.
7. Demonstration of air samplers: equipments / photographs of vertical cylindrical spore trap, Rotorod sampler, Hirst's spore trap, Andersen's sampler, Liquid impingement method (bead bubbler device) and Membrane filter.
8. Microscopic observation of different water samples for biological indicators of water pollution.
9. a. Standard analysis of water sample
b. Determination of MPN.
10. IMViC reactions.
11. Water quality test by Hydrogen sulphide strip test.
12. Display of photographs of water purification process (Baffles, Flocculator, Clarifier, Sand filter, Back wash, Chlorinometer and Chloroscope).
13. Determination of biological oxygen demand (BOD) of water.
14. a. Estimation of total solids in sewage.
b. Display of photographs - Septic tank, Trickling filter, Activated sludge process, Oxidation ponds, Sedimentation tank, and anaerobic digester.
15. a. Demonstration of composting
b. Display of photographs: composting, composting process, aerated lagoons, low shear air lift reactor and microbial leaching.

NOTE: Visit to water treatment plant/ sewage treatment plant/ industrial effluent treatment plant/Agricultural research institute. Each student shall submit an independent report on the visit along with the practical record for the internal assessment.

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V SEMESTER

**TITLE: AGRICULTURAL MICROBIOLOGY
THEORY**

TOTAL HOURS: 60hrs (4hrs/week)

CREDITS: 4

COURSE CODE-CME28206 / CME28207

TOTAL HOURS: 60hrs (4hrs/week)

CREDITS: 4

Course outcome

On successful completion of this course the student will gain knowledge about:

- CO1.** Microorganisms in agriculture, plant pathology and control of plant diseases and their significance
- CO2.** Understand the land mark in the field of Agricultural microbiology.
- CO3.** Gain knowledge about biofertilizers and biopesticide in agriculture.
- CO4.** Know about the stages in disease development, epidemiology and host pathogen interaction.
- CO5.** Know about principles and practices involved in the management of plant diseases by different methods.
- CO6.** Understand the important plant diseases caused by phytoplasma, viruses and viroids. Bacteria and fungi

UNIT I

No. of Hours: 15

INTRODUCTION AND HISTORY OF PLANT PATHOLOGY

A. Concept of plant disease- definitions of disease, disease cycle & pathogenicity, symptoms associated with microbial plant diseases, types of plant pathogens, Koch's postulates, economic losses and social impact of plant diseases.

B. Significant landmarks in the field of plant pathology- Contributions of Anton DeBary, Millardet, T J Burrill, E. Smith, Adolph Mayer, Dmitri Ivanowski, Diener, Stakman, H.H. Flor, Van Der Plank. Contributions of eminent Indian plant pathologists- E J Butler, B B Mundkar, K V Subbarao and M J Thirumalachar.

MICROORGANISMS IN AGRICULTURE

A. Biofertilizers: Definition, Types- Nitrogen fixing, Phosphate solubilizing and cellulolytic microbes. Mass production of Bacterial inoculants (*Rhizobium*, *Azospirillum*, *Azotobacter*, *Cyanobacteria*). Mode of application, Advantages and limitations.

B. Biopesticides: Definition, Types – Bacterial, Viral, Fungal and Protozoan, Mode of action, Microbial herbicides.

UNIT:II

No. of Hours: 15

PHYTOPATHOLOGY

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A. Stages in development of a disease : Introduction – Classification of plant diseases on the basis of spread and severity of infection.

Microbes and Plant diseases: Entry of pathogens into host- prepenetration (Infection) penetration, post penetration (invasion, colonization, dissemination of pathogens and perennation).

B. Plant disease epidemiology: Concepts of monocyclic, polycyclic and polyetic diseases, disease triangle & disease pyramid, forecasting of plant diseases.

C. Host Pathogen Interaction

a. Microbial Pathogenicity

Virulence factors of pathogen: Role of Enzymes-pectic enzymes, Toxins: Host specific (Tabtoxin) and host non-specific (Victorin and T toxin) and growth regulating substance in disease development- Auxins and Gibberellins.

b. Defense Mechanisms in Plants

Defence mechanism in plants: Preexisting (fungitoxic exudates and phenolic compounds) Structural (formation of cork layers, abscission layer and tyloses) and Biochemical defense mechanism (simple phenolic compounds), Hypersensitivity (in brief).

UNIT: III

No. of Hours: 15

CONTROL OF PLANT DISEASES

- A.** Principles & practices involved in the management of plant diseases by different methods, viz. regulatory - quarantine, crop certification, avoidance of pathogen, use of pathogen free propagative material : a. Cultural-Host eradication, crop rotation, sanitization, polythene traps and mulches(in brief).
- B.** Chemical- Inorganic chemicals: Copper compounds-Bordeaux mixture and Bordeaux paste, Organic chemicals- Organic sulfur compounds (Dithiocarbamates), Systemic fungicide, Heterocyclic compounds (Benomyl), antibiotics (Agrimycin).
- C.** Physical method-Soil sterilization by heat, soil solarization, hot water treatment of propagative organs and hot air treatment of storage organs (in brief)
- D.** Biological methods- suppressive soils, antagonism, antagonistic plants and trap plants (in brief).
- E.** IDM-Perennial Crop and annual crop (in brief).

UNIT:IV

No. of Hours: 15

SPECIFIC PLANT DISEASES

Study of some important plant diseases giving emphasis on its etiological agent, symptoms, epidemiology and control

A. Important diseases caused by fungi

Late blight of potato - *Phytophthora infestans*

Powdery mildew of wheat - *Erysiphe graminis*

Ergot of rye - *Claviceps purpurea*

Loose smut of wheat - *Ustilago nuda*

Wilt of tomato - *Fusarium oxysporum* f.sp. *lycopersici*

Red rot of sugarcane - *Colletotrichum falcatum*

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Blast of rice-*Magnaporthe grisea*

B. Important diseases caused by phytopathogenic bacteria: Bacterial leaf blight of rice, Bacterial cankers of citrus

C. Important diseases caused by phytoplasmas: Sandal spike

D. Important diseases caused by viruses: Papaya ring spot, Bunchy top of banana, Bean mosaic.

E. Important diseases caused by viroids: Potato spindle tuber.

Total marks 100: 50(Theory)+30(C1+C2)+ 20 (Practicals)

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**DSE- B
V SEMESTER
TITLE: AGRICULTURAL MICROBIOLOGY
PRACTICAL**

TOTAL HOURS: 30hrs (2hrs/week)

CREDITS: 01

1. Demonstration of Koch's postulates in fungal disease.
- 2-5. Study of important diseases of crop plants by cutting sections of infected plant material -
Late blight of potato, Powdery mildew of wheat, Ergot of rye, Loose smut of wheat, Wilt of tomato, Red rot of sugarcane, Blast of rice
6. Gram's staining of citrus canker specimen
- 7-8. Mounting of fungal pathogen- *Phytophthora infestans*, *Fusarium*, *Colletotrichum* and *Magnaporthe grisea*.
9. Observation of specimens-Bean mosaic and sandal spike
10. Observation of root nodule formation in plants (*Trigonella/Crotolaria*)
11. Demonstration of Indole acetic acid (IAA) production by soil fungi
12. Plant disease control by fungicides
13. Chemical determination of IAA produced by soil fungi *in vitro*
14. Isolation of fungal pathogens from soil
15. Isolation of fungal pathogens from diseased parts of plant

NOTE: Visit to Agricultural research station. Each student shall submit an independent report on the visit along with the practical record for the internal assessment.

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SEC-A

TITLE: MICROBIAL DIAGNOSIS IN HEALTH CLINICS

SEMESTER – V

TOTAL HOURS: 30hrs (2hrs/week)

CREDITS: 2

COURSE CODE-CME28406 / CME28407

Course outcome

CO1.Student will gain experience in health clinics such as examination, collection of clinical samples and diagnosis.

CO2.Demonstrate scientific quantitative skills, such as the ability to evaluate experimental design, read graphs, and understand and use information from scientific papers.

UNIT: I

No of Hours: 5

IMPORTANCE OF DIAGNOSIS OF DISEASES

Bacterial, viral, fungal and protozoan diseases of various human body systems. Disease associated clinical samples for diagnosis.

UNIT:II

No of Hours: 5

COLLECTION OF CLINICAL SAMPLES

Collection of clinical samples (oral cavity/sputum, throat, skin, blood, CSF, urine and faeces) and handling clinical specimens. Method of transport of clinical samples to laboratory and storage.

UNIT :III

N o of Hours: 15

DIRECT MICROSCOPIC EXAMINATION AND CULTURE

Examination of sample by staining - Gram stain, Ziehl-Neelson staining for tuberculosis, Giemsa stained, Thin blood film for malaria, Preparation and use of culture media – Blood agar, Chocolate agar, Lowenstein-Jensen medium, MacConkey agar, Distinct colony properties of various bacterial pathogens.

Serological and Molecular Methods

Serological Methods – Agglutination, Precipitation, ELISA and PCR.

Test for Typhoid, Dengue ,HIV and Swine flu

Laboratory exposure to students: demonstration of staining.

UNIT: IV

No of Hours: 5

TESTING FOR ANTIBIOTIC SENSITIVITY IN BACTERIA

Importance, Determination of resistance/sensitivity of bacteria using disc diffusion method, Determination of minimal inhibitory concentration (MIC) of an antibiotic by serial dilution method

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SEC-II

TITLE: MICROBIOLOGICAL ANALYSIS OF AIR AND WATER

SEMESTER – V

TOTAL HOURS: 30hrs (2hrs/week)

CREDITS: 2

COURSE CODE-CME28606 / CME28607

Course outcome

CO1. Know about bioaerosols, airsample collection and analysis.

CO2.Control measures of air microbes.

CO3. know about the water borne diseases and their management.

CO4.To identify water borne pathogens.

UNIT: I

No of Hours: 10

AIR MICROBIOLOGY

Bioaerosols, Air borne microorganisms (bacteria, Viruses, fungi) and their impact on human health and environment, significance in food and pharma industries and operation theatres, allergens

Air Sample Collection and Analysis

Bioaerosol sampling, air samplers, methods of analysis, CFU, culture media for bacteria and fungi.

Control Measures

Fate of bioaerosols, inactivation mechanisms – UV light, HEPA filters, desiccation and Incineration

UNIT:II

No of Hours: 5

WATER MICROBIOLOGY

Water borne diseases and their management: Cholera, Typhoid, Gastroenteritis and Traveller's diarrhoea.

UNIT: III

No of Hours: 5

MICROBIOLOGICAL ANALYSIS OF WATER

Sample Collection, Treatment and safety of drinking (potable) water, methods to detect potability of water samples: (a) standard qualitative procedure: presumptive/MPN tests, confirmed and completed tests for faecal coliforms (b) Membrane filter technique.

UNIT: IV

No of Hours: 5

LABORATORY SAFETY MEASURES

Precipitation, chemical disinfection, filtration, high temperature, UV light

Laboratory exposure to students: demonstration of air borne and water borne microbes.

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**DSE- A
VI SEMESTER
TITLE: INDUSTRIAL AND FOOD MICROBIOLOGY
THEORY**

TOTAL HOURS: 60hrs (4hrs/week)

CREDITS: 4

COURSE CODE-CMF28006 / CMF28007

Course outcome

Enable the students to get sufficient knowledge about:

- CO1.** Food related microorganisms, their contamination, spoilage and preservation
- CO2.** Understand the beneficial role of microorganisms in fermented dairy products.
- CO3.** Know the significance and activities of microorganisms in food
- CO4.** Understand the food borne intoxication and infections.
- CO5.** Learn about food safety and quality control.
- CO6.** Know the principles involving various methods of food preservation.
- CO7.** Understand how microbiology is applied in manufacture of industrial products
- CO8.** Know about design of bioreactors, medium formulation & design for microbial fermentation, the different types of fermentation processes
- CO9.** Identify techniques applicable for strain Improvement of microorganism
- CO10.** The underlying principles in downstream processing

UNIT: I

No of Hours: 15

INDUSTRIAL MICROBIOLOGY

- A.** Brief history and developments in industrial microbiology
- B.** Microorganisms of industrial importance; Isolation, Screening and Preservation of industrial important microbes..
- C.** Strain improvement of Microorganisms for industrial purposes.
- D.** A brief account of production medium, inoculum medium, raw materials-Molasses, corn steep liquor, sulphite waste liquor, yeast extract and whey. Buffers, Precursors, Inhibitors and Antifoam agents.
- E.** Fermenters and fermentation process: Design, types and basic function of fermenters, sterilization, devices for aeration and agitation (in brief).
Types of fermenters – laboratory, pilot-scale and production fermenters
Components of a typical continuously stirred tank bioreactor
Fermentation process – Surface, Submerged and Solid state fermentation. Types- Batch and Continuous fermentation.
Downstream processing: Steps in recovery and purification of fermented products – Precipitation, Filtration, Centrifugation, Distillation, Cell disruption, Solvent recovery, chromatography, Drying and crystallization.

UNIT: II

No of Hours: 15

INDUSTRIAL PRODUCTION

- A. a.** Organic acids – Citric acid.

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- b. Antibiotics – Penicillin.
 - c. Enzymes – Pectinase and amylase.
 - d. Alcohol – Ethanol.
 - e. Amino acid – Glutamic acid.
- B.** Mushroom cultivation – Oyster mushroom (bag method). Nutritional value.
- C.** Role of microorganisms in the production and recovery of minerals and petroleum.
- D.** Single cell protein: *Spirulina*.

Unit: III

No of Hours: 15

FOOD MICROBIOLOGY

- A.** Introduction to Food Microbiology: Definition, Concept and Scope. Food as a substrate for microorganisms, Factors influencing microbial growth in foods (intrinsic and extrinsic factors).
- B.** Sources of contamination, Microbial spoilage of foods – fruits, vegetables, meat, poultry, canned foods, cereals and cereal products.
- C.** Methods of food preservation: Physical method – high temperature, low temperature, canning. Drying – solar drying, drum drying, spray drying and Radiation. Chemical methods – chemical preservatives – (propionates, benzoate, sorbates, nitrates and nitrites, sugar and salt)
- D.** Food borne intoxication and infection:
Bacterial intoxication- Staphylococcal intoxication and Botulism.
Bacterial infection- Salmonellosis.
Mycotoxin –Types and importance of toxins with special reference to Aflatoxins.
- E.** Food safety and quality control. –A brief account on FPO, HACCP, Food laws and Food standards (in brief)

UNIT:IV

No of Hours: 15

DAIRY MICROBIOLOGY

- A.** Introduction to Dairy Microbiology: Source of milk contamination. Types of microorganisms in milk.
- B.** Methods to detect microbial spoilage by SPC, Reductase test.
- C.** Biochemical changes of milk - Souring, Gassy fermentation, Proteolysis, Lipolysis, and Ropiness.
- D.** Fermented dairy products (a brief account of characteristic and therapeutic value). Acidophilus milk, Yoghurt, Butter milk, Srikhand. Types of cheese. Probiotics and their benefits.
- E.** Preservation of milk and milk products – Pasteurization and Sterilization. Microbiological standard for milk and milk products (in brief).

Total marks 100: 50(Theory)+30(C1+C2)+ 20 (Practicals)

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**DSE- A
VI SEMESTER
TITLE: INDUSTRIAL AND FOOD MICROBIOLOGY
PRACTICAL**

TOTAL HOURS: 30hrs (2hrs/week)

CREDITS: 01

- 1-2. Isolation and enumeration of bacteria from utensils
Isolation and identification of fungi from food utensils
- 3-4. Isolation and enumeration of bacteria from spoiled vegetables
Isolation and identification fungi from spoiled vegetables.
- 5-6. Isolation and enumeration of bacteria from spoiled fruits.
Isolation and identification of fungi from spoiled fruits.
- 7-8. Isolation and identification of *Aspergillus* on groundnut by standard blotters
Method (ISTA,1982).
9. Estimation of lactic acid in milk.
10. Determination of phosphatase activity of milk
11. Turbidity test to detect boiled and unboiled milk.
12. Methylene blue reductase test to determine the quality of milk.
13. Preparation of wine from grapes.
- 14 a. Preparation of alcohol using jaggery or molasses.
b. Estimation of percentage alcohol in a given sample by specific gravity method.
15. Production of citric acid using *Aspergillus niger*

NOTE: Visit to food industries or food research laboratories, dairy industries and distilleries. Each student shall submit an independent report on the visit along with the practical record for the internal assessment.

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**DSE- B
VI SEMESTER**

**TITLE: MEDICAL MICROBIOLOGY AND IMMUNOLOGY
THEORY**

TOTAL HOURS: 60hrs (4hrs/week)

CREDITS: 4

COURSE CODE-CMF28206 / CMF28207

Course outcome

The course provides a solid foundation to understand:

- CO1.**The human immune response towards microbes in medical microbiology, knowledge is gained about the relationship between microorganism and human disease, pathogenicity, Laboratory diagnosis, treatment and prophylaxis.
- CO2.** Demonstrate an understanding of key concepts in immunology.
- CO3.** Understand the overall organization of the immune system.
- CO4.**To make them understand the salient features of antigen antibody reaction & its uses in diagnostics and various other studies.
- CO5 .**Learn about immunization and their preparation and its importance.

UNIT :I

No of Hours: 15

MEDICAL MICROBIOLOGY

- A.** Introduction – History and development of medical microbiology. Normal microflora of the human body: Importance of normal microflora, normal microflora of skin, throat, gastrointestinal tract, urogenital tract
- B.** Infection and disease transmission – Signs, symptoms, syndrome. Types of Infection: opportunistic infection and Nosocomial infection, mode of transmission.
- C.** Host pathogen interaction – Infection, Invasion, Pathogen, Pathogenicity, microbial virulence, microbial toxins, opportunistic and true pathogens.
- D.** Antimicrobial chemotherapy – General characteristics and types of antibiotics. Mode of action of -Penicillin, Aminoglycosides, Erythromycin, Chloramphenicol, Antifungal drugs- Griseofulvin, Nystatin Antiviral drugs-Acyclovir, Amantadine and Azidothymidine .Multiple Drug Resistance (in brief).

UNIT:II

No. of Hours:15

HUMAN DISEASES

- A.** Collection, transportation, culturing and identification of clinically important pathogens.
- B.** Pathogen –Cultural and Biochemical characteristics,clinical symptoms, laboratory diagnosis, prophylaxis and treatment of the following diseases:
 - a. Air borne: Influenza, Diphtheria, Blastomycosis
 - b. Direct contact: Warts, Syphilis, Sporotrichosis
 - c. Vector borne: Dengue, Malaria
 - d. Water borne: Typhoid, Amoebic dysentery

IMMUNOLOGY: IMMUNE CELLS AND ORGANS

- A.** Historical account and introduction to immune system – Blood and Plasma system.
- B.** Types of immunity – Innate (non specific) and Adaptive immunity (specific).
Humoral and cell mediated immunity.
- C.** Structure, Functions and Properties of: Immune Cells –T cell, B cell, Macrophage, Neutrophil, Eosinophil, Basophil, Mast cell, Dendritic cell, Phagocytes and NK cells.
Cells and tissues of immune systems-Structure and role of primary lymphoid organs (bone marrow,thymus),secondary lymphoid organs (spleen, lymph nodes and tonsils).

UNIT-IV

No. of Hours:15

IMMUNOLOGY: ANTIGENS AND ANTIBODIES

- A.** Antigens – Characteristics of an antigen (Foreignness, Molecular size and Heterogeneity); Haptens; Epitopes (T & B cell epitopes), Adjuvants.
- B.** Antibodies – Basic structure of immunoglobulin (Ig G). Biological properties of Immunoglobulin classes, monoclonal antibodies, antigen antibody reactions – salient features. precipitation reaction, neutralization test, opsonisation , agglutination reaction, compliment fixation. Immunotechniques – RIA, ELISA and ELISPOT.
Hypersensitivity (Type I to V - in brief).
Immunoprophylaxis – Vaccine – Types – killed, Live and Attenuated (Bacterial and Viral) and Toxoid with an example each.
National Immunization program (Tabular form).

Total marks 100: 50(Theory)+30(C1+C2)+ 20 (Practicals)

1. Determination of blood group and Rh factor.
2. Enumerate RBC in given blood sample
3. Enumerate WBC in given blood sample
4. Demonstration of precipitation reaction-Double diffusion in two dimensions (Ouchterlony procedure).
5. Antibiotic sensitivity test.
6. Estimation of urine bacteria by calibrated loop- direct streak method.
7. Determination of susceptibility to dental caries-Snydal test
8. Identification of dermatophytes from human skin.
9. Detection of typhoid by Widal test
10. Rapid plasma reagin (RPR) card test for syphilis
11. Identify bacteria on the basis of cultural, morphological and biochemical characteristics: IMViC, TSI, nitrate reduction, urease production and catalase tests
- 12-15. Material/ microscopic observation/ display of photographs of human pathogens as per theory syllabus: Influenza virus, *Corynebacterium diphtheriae*, *Blastomyces dermatitidis*, Human papilloma virus, *Trypanema pallidum*, *Sporothrix schenckii*, *Plasmodium*, Dengue viruses (DENV), *Salmonella typhi* and *Entamoeba histolytica*)

NOTE: Visit to pharmaceuticals and pathological laboratories. Each student shall submit an independent report on the visit along with the practical record for the internal assessment.

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B.Sc., Microbiology

**PATTERN OF QUESTION PAPER (CBCS)
SUBJECT: MICROBIOLOGY
DSCI-DSCIV
(THEORY: I SEMESTER TO IV SEMESTER)**

Time: 3hours

Max marks: 70

I. Answer the following

1X5=05Marks

- 1
- 2
- 3
- 4
- 5

II Answer any five of the following:

3X5=15 Marks

(Seven questions to be given and four to be answered)-short answer type

- 6
- 7
- 8
- 9
- 10
- 11
- 12

III Answer any four of the following:

5X4=20

(Six questions to be given and four to be answered)-short answer type

- 13
- 14
- 15
- 16
- 17
- 18

III Answer any three of the following

10X3=30

(Five questions to be given and four to be answered- essay type questions)

- 19
- 20
- 21
- 22
- 23

C1+C2=30(15+15) Continuous assessment

B.Sc., Microbiology

**PATTERN OF QUESTION PAPER (CBCS)
SUBJECT: MICROBIOLOGY
DSE (A & B) - DSE(A & B)
(THEORY: V SEMESTER TO VI SEMESTER)**

Time: 3hours

Max marks: 70

I. Answer the following

1X5=05Marks

- 1
- 2
- 3
- 4
- 5

II Answer any five of the following:

3X5=15 Marks

(Seven questions to be given and four to be answered)-short answer type

- 6
- 7
- 8
- 9
- 10
- 11
- 12

III Answer any four of the following:

5X4=20

(Six questions to be given and four to be answered)-short answer type

- 13
- 14
- 15
- 16
- 17
- 18

III Answer any three of the following

10X3=30

(Five questions to be given and four to be answered- essay type questions)

- 19
- 20
- 21
- 22
- 23

C1+C2=30(15+15) Continuous assessment

B.Sc., Microbiology

PATTERN OF QUESTION PAPER (CBCS)

SUBJECT: MICROBIOLOGY (SEI-SEII)

SEC(A) – SEC(B)

(THEORY: V SEMESTER)

Time: 2 hours

Max marks: 50

I. Answer the following

1X3=03

- 1
- 2
- 3

II Answer any four of the following:

3X4=12

(Six questions to be given and four to be answered)-short answer type

- 6
- 7
- 8
- 9
- 10
- 11

III Answer any three of the following:

5X3=15

(Five questions to be given and three to be answered)-short answer type

- 12
- 13
- 14
- 15
- 16

III Answer any two of the following

10X2=20

(Four questions to be given and two to be answered- essay type questions)

- 17
- 18
- 19
- 20

C1+C2=30(15+15) Continuous assessment

B.Sc., Microbiology

DSC-I
SCHEME OF THEORY EXAMINATION
I B.Sc., I SEMESTER

TITLE: INTRODUCTION TO MICROBIOLOGY AND MICROBIAL DIVERSITY

Times:3hrs

Max Marks:70

Question Paper to be set for total of 106marks including choices

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 History of development of microbiology	2	3	1	1	26
UNIT:2 Microbial Diversity	1	2	2	1	27
UNIT:3 Fungi and Protozoa	1	-----	1	2	26
UNIT:4 Viruses	1	2	2	1	27

I Main: 1x5=05Marks

II Main: 3x7=21Marks

III Main: 5x6=30Marks

IVMain: 10x5=50Marks

DSC-1

SCHEME OF PRACTICAL EXAMINATION

I B.Sc., I SEMESTER: PRACTICAL-I

TITLE: INTRODUCTION TO MICROBIOLOGY AND MICROBIAL DIVERSITY

Time: 3hours

Max marks: 70

- I.** Identify the materials **A, B** and **C** with labelled diagrams and reasons 5X2=10
 (1 material each from Algae and Fungi as per syllabus)
 (Identification -1mark; diagram and reasons-4mark)
- II.** Write critical notes on **D, E** and **F.** 5X3=15
 (Photographs/materials of Bacteriophages /TMV/HIV/ Plaque assay/ prokaryotic and Eukaryotic cell/Microbiologists/Exposed plates to air)
- III.** Identify the slides **G, H** and **I** with labelled diagrams and reasons 5X3=15
 (One slide each from Algae, Fungi and Protozoa as per the theory syllabus)
 (Identification –1mark; labelled diagram with reasons-4mark)
- IV.** Stain the given material **J** by.....method. Write the principle, procedure and leave the preparation for evaluation 10
 (Wet mounting of Algae/Fungi)
 (Preparation-4 marks; Principle and Procedure-4 marks)
- V.** Record 10
- VI.** Viva 10
- Total marks: 70: [50 (Practical Exam) + 20 (10 -record+ 10- viva)]**

DSC-II

B.Sc., Microbiology**SCHEME OF THEORY EXAMINATION****I B.Sc., II SEMESTER****TITLE: BACTERIOLOGY****Times:3hrs****Max Marks:70****Question Paper to be set for total of 106 marks including choices**

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 Bacterial cell organization	2	-	1	2	27
UNIT:2 Bacteriological techniques	1	2	2	1	27
UNIT:3 Microscopy	2	3	1	1	26
UNIT:4 Physical and chemical methods of Microbial control	-	2	2	1	26

I Main: 1x5=05Marks

II Main: 3x7=21Marks

III Main: 5x6=30Marks

IVMain: 10x5=50Marks

DSC-II**SCHEME OF PRACTICAL EXAMINATION****I B.Sc., II SEMESTER: PRACTICAL-II****TITLE: BACTERIOLOGY****Time: 3hours****Max marks: 70**

- I. Write critical notes on A, B, C and D** 3X4=12
(Microscopes-Charts/Photographs/Instruments/Oil immersion objective/ Stains / Laboratory equipments/Chromicacid/Detergents/Microbiologists/Media/cultivation of microorganisms/pure cultures/maintenance of culture) as per the theory syllabus.
- II. Measure the length/breadth/diameter of the given material E using Stage and Ocular Micrometer. Write the procedure and result.** 15
(Procedure-6marks; calibration -4marks; Results-5marks)
- III. Stain the given material F by.....method. Write the principle, procedure and leave the preparation for evaluation.** 08
(Simple staining/Negative staining/Gram-staining/Cell wall/ Endospore)
(Preparation-4marks; Principle and Procedure-4 marks)
- IV. Demonstrate/ Perform the experiment G giving the principle and procedure. Record the result.** 15
(Demonstration- 5marks; principle-5mark; procedure-3marks; results-2marks)
(Serial dilution/ measurement of growth by cell number using Haemocytometer/ Pour plate/Spread plate/Streak plate/Point inoculation)
- V. Record.** 10
- VI. Viva** 10
- Total marks: 70: [50 (Practical Exam) + 20 (10 -record+ 10- viva)]**

DSC-III**SCHEME OF THEORY EXAMINATION**

B.Sc., Microbiology**II B.Sc.,III SEMESTER****TITLE: MICROBIAL PHYSIOLOGY AND METABOLISM****Times:3hrs****Max Marks:70****Question Paper to be set for total of 106marks including choices**

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 Microbial Growth & Microbial nutrition	2	-	1	2	27
UNIT:2 Metabolism	1	2	2	1	27
UNIT:3 Chemoheterotrophic metabolism	2	3	1	1	26
UNIT:4 Chemolithotrophic & phototrophic metabolism	-	2	2	1	26

I Main: 1x5=05Marks

II Main: 3x7=21Marks

III Main: 5x6=30Marks

IVMain: 10x5=50Marks

DSC-3**II B.Sc.-IIISEMESTER****SCHEME OF PRACTICAL EXAMINATION****PRACTICAL III: MICROBIAL PHYSIOLOGY AND METABOLISM****Time: 3hours****Max. marks :70**

- I.** Demonstrate the experiment **A**, giving principle and procedure. Record the results. 15
(Demonstration-5marks; principle -5mark; procedure -3marks; result-2mark)
(Ammonification /Effect of temperature on growth of microorganisms/Effect of pH on the growth of microorganisms /Effect of salt concentration on growth of microorganism/ Effect of carbon and nitrogen on growth of microorganism).
- II.** Perform/conduct the experiment **B**, giving principle and procedure. Record the results. 10
(Demonstration-5marks; principle -2mark; procedure -2marks; result-1mark)
(Fermentation of lactose / starch hydrolysis/gelatin hydrolysis / catalase activity/urease test)
- III.** Prepare a temporary slide of **C** and identify the microorganisms giving reasons. 10
Leave the preparation for evaluation.
(Preparation of slide-5marks, identification- 1mark, reason-4mark, Material to be given is root nodules)
- IV.** Write critical notes on **D, E & F** 5X3=15
(Fermentation of lactose / glucose/Starch hydrolysis/Gelatin hydrolysis / Catalase Activity/Urease test/Haemocytometer/Turbidometer/fermentation of glucose by Kuhne's fermentation vessel)
- V.** Record 10
- VI.** Viva 10

Total marks: 70: [50 (Practical Exam) + 20 (10 -record+10- viva)]**DSC-IV****SCHEME OF THEORY EXAMINATION****II B.Sc.,IV SEMESTER**

B.Sc., Microbiology**TITLE: MICROBIAL GENETICS AND GENETIC ENGINEERING****Times: 3hrs****Max Marks:70****Question Paper to be set for total of 106marks including choices**

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 Microbial Genetics	2	-	1	2	27
UNIT:2 Molecular Genetics	1	2	2	1	27
UNIT:3 Genetic Engineering	2	3	1	1	26
UNIT:4 Tools of Genetic Engineering	-	2	2	1	26

I Main: 1x5=05Marks

II Main: 3x7=21Marks

III Main: 5x6=30Marks

IVMain: 10x5=50Marks

DSC-IV**II B.Sc.-IV SEMESTER****SCHEME OF PRACTICAL EXAMINATION****PRACTICAL IV: MICROBIAL GENETICS AND GENETIC ENGINEERING****Time: 3hours****Max. marks :70**

- I.** Identify the materials **A, B** and **C** with labelled diagrams and reasons 5X3=15
(conjugation/transduction/ AMES test/Amplification of PCR/Southern blotting/Plasmid DNA/Streptomycin resistant mutant)
(Identification -1mark; diagram and reasons-4mark)
- II.** Write critical notes on **D, E** and **F.** 5 X3=15
(DNA model /Transcription and Translation model/DNA replication model/t-RNA/Plasmids /Episomes/ mRNA, transformation, conjugation and transduction)
- III.** Demonstrate the experiment **G**, giving principle and procedure. Record the results. 10
(Replica plating /Quantification of DNA/Conjugation/transformation/transduction, Isolation of streptomycin resistant strain of *E.coli* by gradient plate method)
(Demonstration-5marks; principle -5mark; procedure -3marks; result-2mark)
- IV.** Prepare the slide **H** giving the procedure and results. 10
(Preparation of slide-5marks, Procedure-2 reason-2mark, Diagram-1)
(onion root tip or flower buds mentioned in the practical syllabus)
- V.** Record 10
- VI.** Viva 10
- Total marks: 70: [50 (Practical Exam) + 20 (10 -record+ 10- viva)]**

DSE-A**SCHEME OF THEORY EXAMINATION**

B.Sc., Microbiology**III B.Sc.,V SEMESTER
TITLE: ENVIRONMENTAL MICROBIOLOGY****Times: 3hrs****Max Marks:70****Question Paper to be set for total of 106marks including choices**

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 Soil Microbiology	2	-	1	2	27
UNIT:2 Aerobiology	1	2	2	1	27
UNIT:3 Aquatic, sewage & bioremediation	2	3	1	1	26
UNIT:4 Agricultural Microbiology	-	2	2	1	26

I Main: 1x5=05Marks

II Main: 3x7=21Marks

III Main: 5x6=30Marks

IVMain: 10x5=50Marks

**DSE-B
SCHEME OF THEORY EXAMINATION
III B.Sc.,V SEMESTER
TITLE: AGRICULTURAL MICROBIOLOGY****Times: 3hrs****Max Marks:70****Question Paper to be set for total of 106marks including choices**

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 Introduction & History of Plant pathology & Microorganism in Agriculture	-	2	2	1	26
UNIT:2 Phytopathology	1	2	2	1	27
UNIT:3 Control of Plant diseases	2	3	1	1	26
UNIT:4 Specific Plant disease	2	-	1	2	27

I Main: 1x5=05Marks

II Main: 3x7=21Marks

III Main: 5x6=30Marks

IVMain: 10x5=50Marks

**DSE-A
III B.Sc.-VSEMESTER
SCHEME OF PRACTICAL EXAMINATION**

B.Sc., Microbiology

PRACTICAL V: ENVIRONMENTAL MICROBIOLOGY

Time: 3hours

Max. marks :70

- I.** Demonstrate /perform the experiment **A**, giving principle and procedure. Record and interpret the result. 10
(Demonstration- 5marks; principle-2marks; procedure-2marks; results-1marks)
(Petriplate exposure method/standard analysis of water/ determination of MPN/ Isolation of Bacteria /Fungi from soil by serial dilution method/Antagonism between microorganisms).
- II.** Demonstrate /perform the experiment **B**, giving principle and procedure. Record and interpret the result. (Demonstration-3marks; principle-3mark; procedure-2mark; results-2marks) 10
(Demonstration of BOD of sewage/Estimation of total solids in sewage/IMViC/Hydrogen sulphide strip test).
- III.** Record the source and importance of microorganisms in the material **C** with Identification and label the diagrams. 10
(Source of the microorganisms and identification-5marks; labelled diagram- 3marks; importance- 2marks).
(Pond water, agar plates exposed to air, biological indicators of water pollution).
- IV.** Write critical notes on **D, E** and **F** 3x4=12
(Identification -1mark; critical comments-4marks)
(Air samplers, Results of standatrd analysis of water, MPN, IMViC reactions,Hydrogen sulphide stip test, photograghs of baffles ,floculator, clarifier,sand filter,back wash, chlorinometer,chloroscope,septic tank, Trickling filter, activated sludge process, oxidation pond, sedimentation tank, anaerobic digester, biogas plant, composting, composting process, aerated lagoons, low shear air lift reactor and microbial leaching/ Azolla/ VAM/Rhizosphere microflora/Plant diseases as per theory syllabus).
- V.** Prepare a temporary stained slide of **G**.Identify with labelled sketch and reasons. 08
Leave the preparation for evaluation.
(Identification -1mark; preparation-4marks; labeled diagram and reasons-3marks).
(Anabaena from Azolla/VAM/Rhizobium/Citrus canker)
- VI.** Record 10
- VII.** Viva 10

DSE-B

III B.Sc.-V SEMESTER

SCHEME OF PRACTICAL EXAMINATION

PRACTICAL V: AGRICULTURAL MICROBIOLOGY

Time: 3hours

Max. marks :70

- I.** Demonstrate /perform the experiment **A**, giving principle and procedure. Record and Interpret the result. 15
(Demonstration-5marks; principle-4marks; procedure-4marks; results- 2marks). (Isolation of Fungi from soil by serial dilution method/ from diseased parts of plants, chemical determination of IAA/plant disease control by fungicide).
- II.** Prepare a temporary stained slide of **B**. Identify with labeled sketch and reasons. 15
Leave the preparation for evaluation.
(Identification -2mark; preparation-5marks; labeled diagram-4 and reasons-4marks).
(Plant diseases as per theory syllabus)
- III.** Identify the slides/materials **C, D, E** and **F** with labelled diagrams and reasons 4X5=20
(Identification-1mark; reasons-2marks; labeled sketch-1mark).
(Plant diseases as per theory syllabus/ Koch postulates)
- IV.** Record +Report 10
- V.** Viva 10
- Total marks: 70: [50 (Practical Exam) + 10 (record+ report) + 5-(viva)]**

DSE-A

B.Sc., Microbiology**SCHEME OF THEORY EXAMINATION****III B.Sc.,VI SEMESTER****TITLE: FOOD MICROBIOLOGY AND INDUSTRIAL MICROBIOLOGY****Times: 3hrs****Max Marks:70****Question Paper to be set for total of 106marks including choices**

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 Industrial microbiology	2	-	1	2	27
UNIT:2 Industrial production	2	3	1	1	26
UNIT:3 Food Microbiology	1	2	2	1	27
UNIT:4 Dairy Microbiology	-	2	2	1	26

I Main: 1x5=05Marks

II Main: 3x7=21Marks

III Main: 5x6=30Marks

IVMain: 10x5=50Marks

DSE-B**SCHEME OF THEORY EXAMINATION****III B.Sc.,VI SEMESTER****TITLE: MEDICAL MICROBIOLOGY AND IMMUNOLOGY****Times: 3hrs****Max Marks:70****Question Paper to be set for total of 106marks including choices**

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 Medical microbiology	2	-	1	2	27
UNIT:2 Human diseases	1	2	2	1	27
UNIT:3 Immune cells& organs	2	3	1	1	26
UNIT:4 Antigens and antibody	-	2	2	1	26

I Main: 1x5=05Marks

II Main: 3x7=21Marks

III Main: 5x6=30Marks

IVMain: 10x5=50Marks

DSE-A

B.Sc., Microbiology

SCHEME OF PRACTICAL EXAMINATION
III B.Sc. – VI SEMESTER
TITLE: FOOD MICROBIOLOGY AND INDUSTRIAL MICROBIOLOGY

Time:3hours.

Max.marks:70

- I.** Demonstrate / Perform the experiment **A**, giving principle and procedure. Record and interpret the result. 15
(Demonstration -5marks; principle-4marks; procedure-4marks; results and interpretation-2marks).
(Isolation of microorganisms from utensils/spoiled vegetables/spoiled fruits).
- II.** Conduct the test for **B**. Write the principle and procedure. Record and interpret the results. 15
(Demonstration -5 marks; principle-4 marks; procedure-4 marks; results and interpretation- 2marks).
(Turbidity test, Phosphatase test, MBRT test, Estimation of % of alcohol in a given sample by specific gravity bottle method).
- III.** Write critical notes on **C, D** and **E**. (Identification -1mark; critical comments-1marks). 4X3=12
(Cheese, Yoghurt, Srikhand, Bread, Molasses, Wine, Alcohol, *Aspergillus* on groundnut, Citric acid production/alcohol from jaggery).
- IV.** Prepare temporary stained slide of **F**. Identify with labelled sketch and reasons. 08
Leave the preparation for evaluation.
(Identification -1mark; preparation-5marks; reasons- 4marks).
(*Spirullina, Chlorella, Aspergillus niger* and Yeast).
- V.** Record +Report 10
- VI.** Viva 10
-

DSE-B

SCHEME OF PRACTICAL EXAMINATION
III B.Sc. – VI SEMESTER
TITLE: IMMUNOLOGY AND MEDICAL MICROBIOLOGY

Time:3hours

Max.marks:70

- I.** Demonstrate / Perform the experiment **A**, giving principle and procedure. Record and interpret the result. 15
(Demonstration -5marks; principle-4marks; procedure-4marks;results and interpretation- 2).
(Antibiotic sensitivity test/Determination of blood group and Rh factor/Demonstration of precipitation reaction-ODD).
- II.** Demonstrate the experiment **B**. write the principle and procedure. Record and interpret the results. 15
(Demonstration -5marks; principle-4marks; procedure-4marks; results and interpretation-2m).
(RPR/Urine bacteria by calibrated loop /Enumerate RBC in given blood sample/ Enumerate WBC in given blood sample/ Snyder test).
- III.** Write critical notes on **C, D**, and **E**. 4x3=12
(Identification -1mark; critical comments-1marks)
(Antibiotic sensitivity test, Estimation of urine bacteria by calibrated loop/IMViC/TSI/Nitrate reduction/urease production/catalase/ Ouchterlony procedure, RPR, Widal test, Slides/ Photographs of human pathogens as per theory syllabus).
- IV.** Prepare temporary stained slide of **F**. Identify with labeled sketch and reasons. 08
Leave the preparation for evaluation.
(Identification -1mark; preparation-5marks; reasons- 4marks).
(Petri plates with Fungal colonies/Bacterial colonies).
- V.** Record +Report 10
- VI.** Viva 10

Total marks: 70: [50 (Practical Exam) + 10 – (record+ report) + 5-(viva)]

SEC- A

B.Sc., Microbiology**SCHEME OF THEORY EXAMINATION
III B.Sc.,V SEMESTER
TITLE: MICROBIAL DIAGNOSIS IN HEALTH CLINICS****Times: 3hrs****Max Marks:50****Question Paper to be set for total of 86 marks including choices**

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 Importance of diagnosis of diseases	1	1	1	1	19
UNIT:2 Collection of clinical samples	1	2	1	1	22
UNIT:3 Deirect microscopic examination and culture	--	1	2	1	23
UNIT:4 Testing for antibiotic sensitivity in bacteria	1	2	1	1	22

I Main: 1x3=03Marks

II Main: 3x6=18Marks

III Main: 5x5=25Marks

IVMain: 10x4=40Marks

**SEC-B
SCHEME OF THEORY EXAMINATION
III B.Sc.,V SEMESTER
TITLE: MICROBIOLOGICAL ANALYSIS OF AIR AND WATER****Times: 3hrs****Max Marks:70****Question Paper to be set for total of 86marks including choices**

UNITS	1 mark questions	3 mark questions	5 mark questions	10 mark questions	Total Marks
UNITS: 1 Air microbiology	1	1	1	1	19
UNIT:2 Water microbiology	1	2	1	1	22
UNIT:3 Microbial analysis of water	--	1	2	1	23
UNIT:4 Control Measures	1	2	1	1	22

I Main: 1x3=03Marks

II Main: 3x6=18Marks

III Main: 5x5=25Marks

IVMain: 10x4=40Marks

DEPARTMENT OF MICROBIOLOGY

B.Sc., Microbiology**LIST OF BOS MEMBERS (2018-19)**

SL.NO.	NAME
1	Dr. M. Seema –Chairman Lecturer and Head Department of Microbiology JSS College, Ooty road,Mysuru-25
2	Dr.B.Kavitha Associate Professor Department of Microbiology Government Science College Nrupathunga road Bengaluru- 560 001
3	M.Girish Assistant Professor and HOD Department of Microbiology JSS College for women, Saraswathipuram Mysuru
4	Dr. D.C. Mohana Associate Professor DOS in Microbiology and Biotechnology Bangalore University, Bengaluru

Approved list of Paper setters and Valuers

B.Sc., Microbiology

Sl No.	Name	College address
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B.Sc., Microbiology

1	Dr..M .Seema	Chairperson, Dept. of Microbiology JSS College, Ooty road, Mysore
2	H.P.Spoorthy	Assistant prof. Dept. of Microbiology JSS College, Ooty road, Mysore
3	Dr.S.Mahadevamurthy	Associate Prof & HOD Dept. of Microbiology Yuvaraja's college Mysore.
4	Dr.Syeda Kauser Fathima	Associate Prof. of Microbiology Maharani's Science College for women JLB road Mysore.
5	Dr. H.S. Jayanth.	Asso.Prof.of Microbiology Dept. of Microbiology Yuvaraja's college Mysore.
6	Dr.Nagarathnamma	Asso. Prof. of Microbiology Government women college Mandya
7	Sri. M. Girish	Assistant prof. Dept. of Microbiology JSS College for Women Saraswathipuram,Mysore
8	Dr. P.K.Maheshwar	Assistant Prof. Dept. of Microbiology Yuvaraja's college,Mysore.
9	Smt. M.S.Shobha	Assistant Prof, Dept. of Microbiology Maharani's Science College Mysore
10	Sri. R.A. Manjunath	Assistant Prof. Dept. of Microbiology Saradavilas College,Mysore
11	Dr.M.P. Ragavendra	Assistant Prof. Dept. of Microbiology Maharani's Science College,Mysore
12	Dr.K.Girish	Assistant Prof. Dept. of Microbiology Maharani's Science College, Mysore

B.Sc., Microbiology

13	Sri. G.S. Siddegowda	Assistant Prof. Dept. of Microbiology Maharani's Science College Mysore
14	Dr.N.S.Devaki	Assistant Prof. Dept. of Molecular Biology Yuvaraja's College , Mysore
15	Syeda Farahna Parveen	Assistant Prof. Dept. of Microbiology St.Philomina's College, Mysore
16	Smt. Vanitha	Assistant Prof. Dept. of Microbiology Maharani's Science College,Mysore
17	Smt. Revanamba	Assistant Prof. Dept. of Microbiology Maharani's Science College,Mysore
18	Mahadevaprasad	Assistant prof. Dept. of Microbiology JSS College for Women Saraswathipuram,Mysore
19	Dr.Nagalambika	JSS University,Mysuru
20	RajaRajeshwari.R	Assistant prof. Dept. of Microbiology SDM College JLB Road,Mysuru
21	Uzma Bathool	Assistant Prof. Dept. of Microbiology St.Philomina's College, Mysore
22	Shruthi Prakash	Assistant Prof. Dept. of Microbiology Mahajana's Science College,Mysuru
23	Athiya sultan	Assistant prof. Dept. of Microbiology SDM College JLB Road,Mysuru
24	C.Poornima Devi	Assistant Prof. Dept. of Microbiology Yuvaraja's college,Mysore
25	Dr.Rakshith	Assistant Prof. Dept. of Microbiology Yuvaraja's college,Mysore
26	Vasundara Devi R	Assistant Prof. Dept. of Microbiology Maharani's Science College,Mysore
	Niveditha Prakash	Assistant prof.

B.Sc., Microbiology

27		Dept. of Microbiology JSS College for Women Saraswathipuram, Mysore
28	Dr.K.Sumana	JSS University, Mysuru



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

(Autonomous)

OOTY ROAD, MYSURU- 570 025

DEPARTMENT OF PHYSICS

Syllabus

CHOICE BASED CREDIT SYSTEM

For B.Sc. Programmes

- **Physics, Mathematics, Chemistry**
- **Physics, Mathematics, Computer Science**
- **Physics, Mathematics, Electronics**
- **Physics, Mathematics, Computer Maintenance**

2017-18

Department of Physics

PROGRAMME: BSc PCM, PROGRAMME CODE: BSc-01 (2017-18)

Semester	Course Code	Course Title	Course type	No. of credits			
				L	T	P	Total
I	CMA29001	Mechanics	DSC-1	04	-	02	06
II	CMB29001	Electricity and Magnetism	DSC-2	04	-	02	06
III	CMC29001	Thermal physics and Statistical mechanics	DSC-3	04	-	02	06
IV	CMD29001	Waves and Optics	DSC-4	04	-	02	06
Discipline Specific Elective papers (DSE 1): Choose any 1							
V	CME29001	Digital, Analog circuits and Instrumentation	DSE-1A	04	-	01	05
	CME29201	Solid State Physics	DSE-1B	04	-	01	05
	CME29601	Mathematical Physics	DSE-1D	04	-	01	05
Discipline Specific Elective papers (DSE 2): Choose any 1							
VI	CMF29001	Elements of Modern Physics	DSE-1A	04	-	01	05
	CMF29201	Nuclear and particle physics	DSE-1B	05	-	-	05
	CMF29401	Quantum Mechanics	DSE-1C	04	-	01	05
Skill Enhancement Course (SEC)							
V	CME69001	Renewable Energy and Energy harvesting	SEC-1A	2	-	-	2
	CME69201	Radiation Safety	SEC-1B	2	-	-	2
	CME69401	Basic Instrumentation Skills	SEC-1C	2	-	-	2
	CME70601	Applied Optics	SEC-1I	2	-	-	2

PROGRAMME: BSc PMCS, PROGRAMME CODE: BSc-02 (2017-18)

Semester	Course Code	Course Title	Course type	No. of credits			
				L	T	P	Total
I	CMA29002	Mechanics	DSC-1	04	-	02	06
II	CMB29002	Electricity and Magnetism	DSC-2	04	-	02	06
III	CMC29002	Thermal physics and Statistical mechanics	DSC-3	04	-	02	06
IV	CMD29002	Waves and Optics	DSC-4	04	-	02	06
Discipline Specific Elective papers (DSE 1): Choose any 1							
V	CME29002	Digital, Analog circuits and Instrumentation	DSE-1A	04	-	01	05
	CME29202	Solid State Physics	DSE-1B	04	-	01	05
	CME29602	Mathematical Physics	DSE-1D	04	-	01	05
Discipline Specific Elective papers (DSE 2): Choose any 1							
VI	CMF29002	Elements of Modern Physics	DSE-1A	04	-	01	05
	CMF29202	Nuclear and particle physics	DSE-1B	05	-	-	05
	CMF29402	Quantum Mechanics	DSE-1C	04	-	01	05
Skill Enhancement Course (SEC)							
V	CME69002	Renewable Energy and Energy harvesting	SEC-1A	2	-	-	2
	CME69202	Radiation Safety	SEC-1B	2	-	-	2
	CME69402	Basic Instrumentation Skills	SEC-1C	2	-	-	2
	CME70602	Applied Optics	SEC-1I	2	-	-	2

PROGRAMME: BSc PMCM, PROGRAMME CODE: BSc-03 (2017-18)

Semester	Course Code	Course Title	Course type	No. of credits			
				L	T	P	Total
I	CMA29003	Mechanics	DSC-1	04	-	02	06
II	CMB29003	Electricity and Magnetism	DSC-2	04	-	02	06
III	CMC29003	Thermal physics and Statistical mechanics	DSC-3	04	-	02	06
IV	CMD29003	Waves and Optics	DSC-4	04	-	02	06
Discipline Specific Elective papers (DSE 1): Choose any 1							
V	CME29003	Digital, Analog circuits and Instrumentation	DSE-1A	04	-	01	05
	CME29203	Solid State Physics	DSE-1B	04	-	01	05
	CME29603	Mathematical Physics	DSE-1D	04	-	01	05
Discipline Specific Elective papers (DSE 2): Choose any 1							
VI	CMF29003	Elements of Modern Physics	DSE-1A	04	-	01	05
	CMF29203	Nuclear and particle physics	DSE-1B	05	-	-	05
	CMF29403	Quantum Mechanics	DSE-1C	04	-	01	05
Skill Enhancement Course (SEC)							
V	CME69003	Renewable Energy and Energy harvesting	SEC-1A	2	-	-	2
	CME69203	Radiation Safety	SEC-1B	2	-	-	2
	CME69403	Basic Instrumentation Skills	SEC-1C	2	-	-	2
	CME70603	Applied Optics	SEC-1I	2	-	-	2

PROGRAMME: BSc PME, PROGRAMME CODE: BSc-04 (2017-18)

Semester	Course Code	Course Title	Course type	No. of credits			
				L	T	P	Total
I	CMA29004	Mechanics	DSC-1	04	-	02	06
II	CMB29004	Electricity and Magnetism	DSC-2	04	-	02	06
III	CMC29004	Thermal physics and Statistical mechanics	DSC-3	04	-	02	06
IV	CMD29004	Waves and Optics	DSC-4	04	-	02	06
Discipline Specific Elective papers (DSE 1): Choose any 1							
V	CME29004	Digital, Analog circuits and Instrumentation	DSE-1A	04	-	01	05
	CME29204	Solid State Physics	DSE-1B	04	-	01	05
	CME29604	Mathematical Physics	DSE-1D	04	-	01	05
Discipline Specific Elective papers (DSE 2): Choose any 1							
VI	CMF29004	Elements of Modern Physics	DSE-1A	04	-	01	05
	CMF29204	Nuclear and particle physics	DSE-1B	05	-	-	05
	CMF29404	Quantum Mechanics	DSE-1C	04	-	01	05
Skill Enhancement Course (SEC)							
V	CME69004	Renewable Energy and Energy harvesting	SEC-1A	2	-	-	2
	CME69204	Radiation Safety	SEC-1B	2	-	-	2
	CME69404	Basic Instrumentation Skills	SEC-1C	2	-	-	2
	CME70604	Applied Optics	SEC-1I	2	-	-	2

Assessment Maximum marks - 100

Course type	C1		C2		C3 Exam Marks		Assigned Marks (Percentage)			Total
	Theory	Lab	Theory	Lab	Theory	Lab	Theory	Lab	IA	
DSC	10	05	10	05	70	70	50	20	30	100
DSE	10	05	10	05	70	70	50	20	30	100
DSE (non practical)	15	-	15	-	70	-	70	--	30	100
SEC	15	-	15	-	50	-	70	--	30	100

Note:

1. C1 will be conducted for 20 Marks (Theory) with one hour duration - 10 Marks (Lab) with continuous assessment and it will be reduced to assigned marks.
2. C2 will be conducted for 20 Marks (Theory) with one hour duration - 10 Marks (Lab) with continuous assessment and it will be reduced to assigned marks.
3. C3 will be conducted for 70 Marks (Theory) with three hours duration - 70 Marks (Lab) with 3 hours duration and to be reduced to assigned marks.
4. For non-practical course C3 will be conducted for 70 Marks (Theory) with three hours duration.
5. In case of SEC, C1 and C2 will be conducted for 15 Marks each with one hour duration and C3 will be conducted for 50 Marks with 2 hours duration.

Programme Outcome for Bachelor of Science in Physics, Chemistry, Mathematics:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate proficiency in Mathematics and the Mathematical concepts needed for a proper understanding of Physics.
- PO2. Demonstrate the ability to justify and explain their thinking and/or approach
- PO3. Develop state-of-the-art laboratory and professional communication skills
- PO4. Apply the scientific method to design, execute, and analyze an experiment
- PO5. Explain scientific procedures and experimental observations
- PO6. Appreciate the role of chemistry in the society
- PO7. Use this as a basis for ethical behaviour in issues facing chemists /drugs
- PO8. Understand chemistry as an integral part for addressing social, economic, and environmental problems
- PO9. Understand the value of Mathematical proof
- PO10. Demonstrate proficiency in writing and understanding proofs
- PO11. Apply mathematical problems and solution in aspects of science and technology
- PO12. Gain experience to investigate the real world problems
- PO13. Apply mathematical ideas and models to those problems

Programme Specific Outcome

Bachelor of Science in Physics, Chemistry, Mathematics

After completing the graduation in the Bachelor of Science the students are able to:

PSO1. Find career opportunities

PSO2. Develop competence to write competitive examinations

PSO3. Develop proficiency in the analysis of complex physical problems

PSO4. Use of mathematical or other appropriate techniques to solve problems

PSO5. Apply appropriate techniques for the qualitative and quantitative analysis of chemicals in laboratories and in industries

PSO6. Create a hypothesis and appreciate how it relates to broader theories

PSO7. Demonstrate skills in the use of computers

Programme Outcome for Bachelor of Science in Physics, Mathematics, Computer Science:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate proficiency in Mathematics and the Mathematical concepts needed for a proper understanding of Physics
- PO2. Demonstrate the ability to justify and explain their thinking and/or approach
- PO3. Develop state-of-the-art laboratory and professional communication skills
- PO4. Apply the scientific method to design, execute, and analyze an experiment
- PO5. Explain scientific procedure and experimental observations
- PO6. Understand the value of Mathematical proof
- PO7. Demonstrate proficiency in writing an understanding proofs
- PO8. Apply mathematical problems and solutions in aspects of science and technology.
- PO9. Gain experience to investigate the real world problems
- PO10. Apply mathematical ideas and models to those problems.
- PO11. Apply Mathematical principles for computing and logical design.
- PO12. Design, implements, and evaluates a computational system to meet desired needs within realistic constraints
- PO13. Use the System principles in the design and development of software for systems of varying complexity.

Programme Specific Outcome

Bachelor of Science in Physics, Mathematics, Computer Science

After completing the graduation in the Bachelor of Science the students are able to:

PSO1. Find career opportunities

PSO2. Develop competence to write competitive examinations.

PSO3. Develop proficiency in the analysis of complex physical problems

PSO4. Use of mathematical or other appropriate techniques to solve problems

PSO5. Create a hypothesis and appreciate how it relates to broader theories.

PSO6. Demonstrate skills in the use of Computers

PSO7. Join as Entry level Technical job role for an IT Industry

PSO8. Build small database ERP software/ web applications.

Programme Outcome for Bachelor of Science in Physics, Mathematics, Computer Maintenance:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate proficiency in Mathematics and the Mathematical concepts needed for a proper understanding of Physics
- PO2. Demonstrate the ability to justify and explain their thinking and/or approach
- PO3. Develop state-of-the-art laboratory and professional communication skills
- PO4. Apply the scientific method to design, execute, and analyze an experiment
- PO5. Explain scientific procedure and experimental observations
- PO6. Understand the value of Mathematical proof
- PO7. Demonstrate proficiency in writing and understanding proofs
- PO8. Apply mathematical problems and solutions in aspects of science and technology.
- PO9. Gain experience to investigate the real world problems
- PO10. Apply mathematical ideas and models to those problems
- PO11. Design, implement and evaluate a computational system to meet desired needs within realistic constraints
- PO12. Apply system design and development principals in the construction of software systems of varying complexity
- PO13. Apply the knowledge techniques , skills and modern tools in computer maintenance
- PO14. Understand networking applications to include basic electronics, programming, operation and computer network

Programme Specific Outcome

Bachelor of Science in Physics, Mathematics, Computer Maintenance

After completing the graduation in the Bachelor of Science the students are able to:

PSO1. Find career opportunities

PSO2. Develop competence to write competitive examinations

PSO3. Develop proficiency in the analysis of complex physical problems

PSO4. Use of mathematical or other appropriate techniques to solve them

PSO5. Create a hypothesis and appreciate how it relates to broader theories.

PSO6. Demonstrate skills in the use of Computers

PSO7. Start small enterprise in PC Maintenance/ Networking field.

PSO8. Join organizations related to Computer Hardware and Networking Maintenance.

Programme Outcome for Bachelor of Science in Physics, Mathematics, Electronics:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate proficiency in mathematics and the mathematical concepts needed for a proper understanding of physics.
- PO2. Demonstrate the ability to justify and explain their thinking and/or approach
- PO3. Develop state-of-the-art laboratory skills and professional communication skills.
- PO4. Apply the scientific method to design, execute, and analyze an experiment
- PO5. Understand the value of Mathematical proof
- PO6. Gain experience to investigate the real world problems
- PO7. Explain scientific procedures and their experimental observations
- PO8. Demonstrate proficiency in writing and understanding proofs.
- PO9. Apply mathematical problems and solutions in aspects of science and technology
- PO10. Apply mathematical ideas and models to problems.
- PO11. Apply appropriate troubleshooting techniques to electronic circuits / systems and perform test procedures
- PO12. Assist, Assemble, modify and test electronic circuits in accordance with job requirements.
- PO13. Communicate effectively in technical and non-technical environments

Programme Specific Outcome

Bachelor of Science in Physics, Mathematics, Electronics

After completing the graduation in the Bachelor of Science the students are able to:

PSO1. Find career opportunities

PSO2. Develop competence to write competitive examinations.

PSO3. Develop proficiency in the analysis of complex physical problems.

PSO4. Use mathematical or other appropriate techniques to solve complex physical problems.

PSO5. Create a hypothesis and appreciate how it relates to broader theories.

PSO6. Demonstrate skills in the use of Computers for control, data acquisition, and data analysis in experimental investigations

PSO7. Apply knowledge of Physics, Mathematics and Electronics fundamentals to the solve problems in electronic circuits & communication systems

PSO8. Apply appropriate troubleshooting techniques to Electronic circuits / systems and perform test procedures

SEMESTER I

Course code: CMA29001/ CMA29002/ CMA29003/ CMA29004

Credits: Theory – 04, Practical – 02

Theories: 60 Lectures

COURSE OUTCOME:

After successful completion of the course, the student is able to

CO1.Learn the details of Elasticity

CO2.Understand the classification and characteristics of motion of a point particle

CO3.Understand in details with examples Frames of reference and relative motion

CO4.Deliberate the classification and characteristics of Dynamics of particle in conservative field

CO5.Specify the classification and characteristics of Special theory of relativity and gravitation

CO6.Write down the characteristics of Surface tension and viscosity

MECHANICS: DSC1

Unit-1

Vectors: Vector algebra (with special reference to the rules of addition and multiplication), Scalar and vector products with specific examples.

Motion of a point particle: The position vector $r(t)$ of a moving point particle and its Cartesian components. Velocity and acceleration as the vector derivatives. Derivatives of a vector with respect to a parameter; Derivation of planar vector of a constant magnitude. Radial and transverse components of velocity and acceleration for arbitrary planar motion, deduction of results for uniform circular motion-centripetal force. **(05 Lectures)**

Frames of references and relative motion:

Newton's laws of motion and inertial mass. Galilean transformation; Galilean principle of relativity, Plumb line accelerometer and a freely falling elevator, Non-inertial frames and fictitious force, uniformly rotating frame of reference and coriolis force. Effect of rotation of earth on acceleration due to gravity. **(07Lectures)**

Dynamics of a particle in conservative fields:

Work done by force acting on a particle, work-energy theorem. Conservative and non conservative force field. Conservation of energy. Conservative force as a negative gradient of potential, central force as an example of conservative force field. **(05 Lectures)**

Conservation of momentum: Conservation of linear momentum, centre of mass, rocket equation. Angular momentum and torque, law of conservation of angular momentum, angular momentum of a system taking centre of mass of the system. **(06 Lectures)**

Dynamics of rigid bodies: Moment of inertia, radius of gyration, calculation of momentum of inertia of rectangular plate, circular plate and solid sphere, kinetic energy of rotation. **(04 Lectures)**

Oscillations: Simple harmonic motion. Differential equation of SHM and its solutions. Kinetic and Potential Energy, Total Energy and their time averages. Damped oscillations. **(03 Lectures)**

Unit-2

Gravitation: Newton's Law of gravitation. Motion of a particle in a central force field (motion is in a plane, angular momentum is conserved, areal velocity is constant). Kepler's Laws, derivations of Kepler's law, satellite in circular orbit and applications, geosynchronous orbits, weightlessness, basic idea of global positioning system (GPS). **(08 Lectures)**

Special Theory of Relativity: Postulates of Special Theory of Relativity. Lorentz transformation equations, Length contraction, time dilation, relativistic addition of velocities. Mass-Energy relation, energy-momentum relation. **(06 Lectures)**

Elasticity:

Review of concepts of moduli of elasticity, Hooke's Law and Poisson's ratio(σ). Relation between the elastic constants q , k , n and σ , limiting values for σ . Work done in stretching. Elastic potential energy. Bending moment. Theory of light single cantilever. I-section girders. Torsion; calculation of couple per unit twist. The Torsional pendulum, Static torsion, Searle's double bar experiment.

Surface Tension: Review of basic concepts. Pressure inside curved liquid surface. Surface tension

UG-Physics- CBCS Scheme

and interfacial tension by drop-weight method. Surface tension and angle of contact of mercury by Quincke's method.

Viscosity: Review of basic concepts; Variation of Viscosity of liquids with temperature and pressure.

(16 Lectures)

Reference Books:

- Halliday, Resnick, Jearl Walker, "Principles of Physics" 9th edition, Wiley, 2013.
- Berkeley Physics Course, Vol-1 "Mechanics", 2nd edition, Charles Kittel, Walter D Knight, Malvin A
- D S Mathur, "Elements of properties of matter", S Chand and company, New Delhi, Reprint-2007.
- D S Mathur, "Mechanics", S Chand and company, New Delhi, Reprint-2001.
- Brij Lal and N Subrahmanyam, "Properties of matter", 6th edition, Eurasia publishing house Ltd. New Delhi, Reprint-1993.
- Mechanics by Shankara Narayana & Chopra.
- Mechanics by Bhargava and Sharma.

PHYSICS LAB: DSC 1A LAB: MECHANICS

(Minimum of eight is to be conducted)

1. To determine the Moment of Inertia of a Flywheel.
2. Young's modulus by the single cantilever method.
3. Determination of the Young's modulus by Dynamic method (graphical and calculation method).
4. Drop weight method; Determination of surface tension of liquid and the interfacial tension between two liquids
5. Torsional pendulum; Determination of the rigidity modulus
6. To determine the Elastic Constants of a Wire by Searle's method.
7. Oscillations of a spiral Spring and calculate a) Spring Constant b) Value of g
8. Bar pendulum: Determination of the acceleration due to gravity and radius of gyration (Both graphical and calculation methods).
9. Determination of young's modulus by the method of uniform bending.
10. Determination of rigidity modulus by the static torsion method.

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11. To determine the Height of a Building using a Sextant.
12. To determine g by Kater's Pendulum
13. To determine the Modulus of Rigidity of a Wire by Maxwell's needle
14. To determine g and velocity for a freely falling body using Digital Timing Technique

Reference Books:

- Advanced Practical Physics for students, B.L.Flint and H.T.Worsnop, 1971, Asia Publishing House.
- Advanced level Physics Practicals, Michael Nelson and Jon M. Ogborn, 4th Edition, reprinted 1985, Heinemann Educational Publishers.
- Engineering Practical Physics, S.Panigrahi & B.Mallick, 2015, Cengage Learning India Pvt. Ltd.
- A Text Book of Practical Physics, Indu Prakash and Ramakrishna, 11th Edition, 2011, Kitab Mahal, New Delhi.

SEMESTER II

Course code: CMB29001/ CMB29002/ CMB29003/ CMB29004

Credits: Theory – 04, Practical – 02

Theories: 60 Lectures

COURSE OUTCOME:

After successful completion of the course, the student is able to

CO1.Deliberate in detail with examples vector analysis

CO2.Write down in detail with application, electrostatics and magnetostatic

CO3.Write down the classification and characteristics of AC Circuits

CO4.Specify in details with application, if applicable, properties of magnet material

CO5.Understand the characteristics of electromagnetic theory

CO6.Write down the characteristic of galvanometer

ELECTRICITY AND MAGNETISM: DSC 2

Unit-1

Vector Analysis: Review of vector algebra (Scalar and Vector product), Scalar and vector fields, gradient, divergence, Curl and their significance, Vector Integration, Line, surface and volume integrals of Vector fields, Gauss-divergence theorem and Stoke's theorem of vectors (statement only). **(07 Lectures)**

Electrostatics: Electrostatic Field, electric flux, Gauss's theorem of electrostatics and applications; 1) infinite line of charge and 2) plane charged sheet. Electric potential as line integral of electric field, potential due to a point charge, electric dipole, uniformly charged spherical shell and solid sphere. Calculation of electric field from potential.

Capacitance of an isolated spherical conductor. Parallel plate, spherical and cylindrical condenser. Energy per unit volume in electrostatic field.

Dielectric medium, Polarization, Displacement vector. Gauss's theorem in dielectrics. Parallel plate capacitor completely filled with dielectric.

Galvanometers: Construction, theory and working of Helmholtz galvanometer. **(15Lectures)**

Alternating current: R M S values, Response of LR, CR and LCR circuits to sinusoidal voltages (discussion using the j symbol), Series and parallel resonance, Half-power frequencies, bandwidth and Q-factor, Power in electrical circuits, power factor and Maximum power transfer theorem.

(08 Lectures)

Unit-2

Applications of ac circuits - ac bridges; Anderson's bridge and De-Sauty's bridge

(02 Lectures)

Magneto statics: Biot-Savart's law & its applications; long straight conductor, circular coil and solenoid carrying current. Divergence and curl of magnetic field. Magnetic vector potential. Ampere's circuital law.

Magnetic properties of materials: Magnetic intensity, magnetic induction, permeability, magnetic susceptibility. Brief introduction of dia-magnetic, para-magnetic and ferro-magnetic materials.

(10 Lectures)

Electromagnetic Induction: Faraday's laws of electromagnetic induction, Lenz's law, self Inductance (L) and mutual inductance (M), L of single coil, M of two coils. Energy stored in magnetic field.

(06 Lectures)

Electromagnetic Theory: Equation of continuity, Displacement current, Setting up of Maxwell's equations, wave equation in free space, Poynting vector, energy density in electromagnetic field, electromagnetic wave propagation through free space and isotropic dielectric medium, Transverse nature of electromagnetic waves, polarization.

(12 Lectures)

Reference Books:

- D. C. Tayal, Electricity and Magnetism, 1988, Himalaya Publishing House.
- K. K. Tewari: Electricity and magnetism, S. Chand Co. Ltd., New Delhi, Reprint 2007.
- B. B. Laud: Electrodynamics, Wiley Eastern Limited, New Delhi.
- David. J. Griffiths: Introduction to Electrodynamics, 3rd edition, Prentice-Hall of India Private limited, New Delhi.
- BrijLal and N. Subramanian: Electricity and Magnetism, 19th edition-Ratan Prakashan Mandir, Educational and University Publishers, Agra.

UG-Physics- CBCS Scheme

- D.N. Vasudeva: Fundamentals of Magnetism and Electricity, 12th edition-S.Chand and Co. Ltd., New Delhi

PHYSICS LAB- DSC 2A LAB: ELECTRICITY AND MAGNETISM (Minimum of eight is to be conducted)

1. To verify the Thevenin's theorem
2. Maximum Power Transfer Theorem
3. Black box – Identification of L,C & R
4. LCR series circuits – Determination of L & Q factor
5. LCR parallel circuits – Determination of L & Q factor
6. Anderson's Bridge – Determination of the self-inductance of the coil.
7. De-Sauty's bridge – Verification of laws of combination of capacitances, unknown capacitance.
8. B_H using Helmholtz double coil galvanometer
9. Maxwell's bridge-determination of mutual inductance
10. Low resistance-determination of the resistivity of the material.
11. Determination of capacitance by measuring impedance of RC circuit.
12. Determination of inductance by measuring impedance of RL circuit.
13. Measurement of Magnetic field strength B and its gradient in a Solenoid (Determine dB/dx).
14. To determine a Low Resistance by Carey Foster's Bridge.

Reference Books

- Advanced Practical Physics for students, B.L.Flint & H.T.Worsnop, 1971, Asia Publishing House.
- A Text Book of Practical Physics, Indu Prakash and Ramakrishna, 11th Edition, 2011, Kitab Mahal, New Delhi.
- Edition, reprinted 1985, Heinemann Educational Publishers

SEMESTER III

Course code: CMC29001/ CMC29002/ CMC29003/ CMC29004

Credits: Theory – 04, Practical – 02

Theories: 60 Lectures

COURSE OUTCOME:

After successful completion of the course, the student is able to

CO1.Specify in details with examples kinetic theory of gases

CO2.Specify in depth low temperature physics

CO3.Identify in detail with application thermal conductivity and theory of radiation

CO4.write down the classification and characteristics of laws of thermodynamics

CO5.Have a clear understanding about reversible and irreversible process

CO6.Understand the classification and characteristics of entropy and thermodynamic potential

THERMAL PHYSICS AND STATISTICAL MECHANICS: DSC 3

Unit-1

Laws of Thermodynamics: Thermodynamic description of system, Zeroth Law of thermodynamics and temperature. Applications of First Law; General Relation between C_P & C_V . Work Done during Isothermal and Adiabatic Processes. Compressibility & Expansion Coefficient. Reversible & irreversible processes; Carnot's theorem. Thermodynamic scale of temperature and its identity with perfect gas scale.

Entropy: The concept of entropy. Change of entropy in reversible and irreversible cycles. Entropy and non-available energy. Principle of increase of entropy; Clausius inequality. Second law of thermodynamics in terms of Entropy. Entropy of ideal gas, T-S diagram. Probability and entropy, Boltzmann relation. Concept of absolute zero and the third law of thermodynamics.

(15 Lectures)

Thermodynamic Potentials: Internal energy, Enthalpy, Helmholtz and Gibbs functions, Maxwell's thermodynamic relations & applications; Joule-Thompson Effect. Clausius-Clapeyron first Latent heat equation, effect of pressure on melting point of a solid, effect of pressure on boiling

UG-Physics- CBCS Scheme

point of a liquid, Expression for ($C_P - C_V$), C_P/C_V , TdS equations. **(08 Lectures)**

Kinetic Theory of Gases: Derivation of Maxwell's law of distribution of velocities and its experimental verification, Mean free path (Zeroth Order). Transport Phenomena; Viscosity, Conduction and Diffusion (for vertical case), Law of equipartition of energy (no derivation) and its applications to specific heat of gases; mono-atomic and diatomic gases. **(07 Lectures)**

Unit-2

Thermal conductivity: Equation of flow of heat through solid bar, determination of thermal conductivity of a bad conductor by Lee and Charlton method. **(03 Lectures)**

Theory of Radiation: Induced and spontaneous emission of radiation. Derivation of Planck's law of radiation using Einstein's A and B coefficients. Deduction of Rayleigh-Jeans law, Stefan's law and Wien's displacement law from Planck's law, Wien's formula. **(08 Lectures)**

Low temperature Physics: Ideal gas and real gas. Van-der Waals equation of state. Porous-plug experiment and its theory. Joule-Thomson expansion - expression for the temperature of inversion, inversion curve. Relation between Boyle temperature, temperature of inversion and critical temperature of a gas. Principle of regenerative cooling. Liquefaction of air by Linde's methods. Adiabatic demagnetization. **(07 Lectures)**

Statistical Mechanics: Probability concept, Phase space, Macrostate and Microstate, Entropy and Thermodynamic probability, Maxwell-Boltzmann law- distribution of velocity. Quantum statistics: Bose-Einstein, Maxwell-Boltzmann and Fermi-Dirac distribution law – electron gas-Bose-Einstein distribution law - photon gas - comparison of three statistics. **(12 Lectures)**

Reference Books:

- BrijLal, N. Subramanyam P.S. Hemne: Heat Thermodynamics and Statistical Physics, 1st edition. S Chand Publishing, 2007.
- S C Gupta: Thermodynamics, 1st edition, Pearson, 2005.
- C. L. Arora: Refresher Course in Physics Vol I, S Chand publishing, 2011.

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- S. R. Shankara Narayana: Heat and Thermodynamics, 2nd edition, Sulthan Chand and Sons, 1990.
- A Treatise on Heat, Meghnad Saha, and B.N. Srivastava, 1969, Indian Press.
- Heat and Thermodynamics, M.W.Zemasky and R. Dittman, 1981, McGraw Hill

PHYSICS LAB-DSC 3A LAB: THERMAL PHYSICS AND STATISTICAL MECHANICS (Minimum of eight is to be conducted)

1. Measurement of Planck's constant using black body radiation.
2. Verification of Stefan's-Boltzmann law.
3. To determine the coefficient of thermal conductivity of a bad conductor by Lee and Charlton's disc method.
4. Verification of distribution law using Monte-Carlo Method
5. To study the variation of thermo emf across two junctions of a thermocouple with temperature.
6. Specific heat of a liquid by cooling – graphical method
7. To determine Stefan's Constant
8. To determine Mechanical Equivalent of Heat, J, by Callender and Berne's constant flow method.
9. To record and analyze the cooling temperature of an hot object as a function of time using a thermocouple and suitable data acquisition system
10. To determine the temperature co-efficient of resistance by Platinum resistance thermometer.
11. To determine the coefficient of thermal conductivity of copper by Searle's Apparatus
12. To determine the Coefficient of Thermal Conductivity of Cu by Angstrom's Method.
13. To study the variation of thermo emf across two junctions of a thermocouple with temperature
14. To calibrate Resistance Temperature Device (RTD) using Null Method/Off-Balance Bridge

Reference Books:

- Advanced Practical Physics for students, B.L.Flint & H.T.Worsnop, 1971, Asia Publishing House.
- Advanced level Physics Practicals, Michael Nelson and Jon M. Ogborn, 4th Edition, reprinted 1985, Heinemann Educational Publishers
- A Text Book of Practical Physics, Indu Prakash and Ramakrishna, 11th Edition, 2011, Kitab Mahal, New Delhi.
- A Laboratory Manual of Physics for Undergraduate Classes, D.P. Khandelwal, 1985, Vani Publication.

SEMESTER IV

Course code: CMD29001/ CMD29002/ CMD29003/ CMD29004

Credits: Theory – 04, Practical – 02

Theories: 60 Lectures

COURSE OUTCOME:

After the completion of the course, Students will be able to

CO1.Specify the classification and characteristics of Fourier theorem

CO2.Learn in detail with application, superposition of simple harmonic motion

CO3.Understand in detail with application of wave motion

CO4.Deliberate in detail with examples Sound, wave optics and transducers

CO5.Learn the details of Interference, diffraction and polarization

CO6.Learn in detail with application of acoustics

WAVES AND OPTICS: DSC 4

Unit-1

Analysis of Complex Waves: Fourier's Theorem - Application to saw tooth wave and square wave. **(03 Lectures)**

Superposition of simple harmonic motion:

Superposition of two simple harmonic motion; Lissajous' figures. Damped vibration; Equation for damped vibrations. Forced vibration; solution in exponential form, Resonance, Expression for amplitude and phase at resonance.

Superposition of two collinear harmonic oscillations: Linearity and Superposition Principle. (1) Oscillations having equal frequencies and (2) Oscillations having different frequencies (Beats). **(11Lectures)**

Wave Motion:

Progressive waves; Differential equation of wave motion; Relation between amplitude and intensity. Expression for velocity of progressive waves in a medium; Newton's formula, Laplace's correction.

Transverse waves on a string. Travelling and standing waves on a string. Normal Modes of a string.

UG-Physics- CBCS Scheme

Expression for frequency of vibration of a stretched string, harmonics. Group velocity, Phase velocity. Longitudinal vibrations in a rod; Kundt's tube experiment. **(10 Lectures)**

Sound: Intensity and loudness of sound - Decibels - Intensity levels - musical notes - musical scale. Acoustics of buildings: Reverberation and time of reverberation, Absorption coefficient, Sabine's formula, measurement of reverberation time. Acoustic aspects of halls and auditoria. **(06 Lectures)**

Unit-2

Transducers: Types of transducers, dynamic microphone and loudspeaker - construction, working and their characteristics, Piezo electrical transducer. **(03 Lectures)**

Wave Optics: Electromagnetic nature of light. Definition and Properties of wave front. Huygens Principle. **(03 Lectures)**

Interference: Interference: Review of basic concepts, Coherent sources, conditions for constructive and destructive interference.

Coherent source by division of amplitude: Interference in Thin Films: – reflected and transmitted light, color of thin films, theory of air wedge, theory of Newton's rings; measurement of wavelength and refractive index.

Coherent source by division of division of wave front: Lloyd's Mirror and Fresnel's Biprism. Michelson's interferometer; Measurement of λ and $d\lambda$. **(11 Lectures)**

Diffraction: Fresnel and Fraunhofer diffraction. Explanation of rectilinear propagation of light. Theory of the zone plate and comparison with convex lens. Fresnel diffraction at a straight edge. Fraunhofer diffraction at a single slit. Transmission grating: theory for the case of normal incidence. **(06 Lectures)**

Polarization: Double refraction in uniaxial crystals. Huygens's theory. Positive and negative crystal. Principal refractive indices. Huygens's constructions of ordinary and extraordinary wave fronts in a uniaxial crystal, retarding plates. Production and analysis of linearly, circularly and elliptically polarized light. Optical activity, Fresnel's theory, Lorentz half shade polarimeter. **(07 Lectures)**

Reference Books:

- Fundamentals of Optics, F A Jenkins and H E White, 1976, McGraw-Hill. Principles of Optics, B.K. Mathur, 1995, Gopal Printing
- Fundamentals of Optics, H.R. Gulati and D.R. Khanna, 1991, R. Chand Publication
- Brij lal and N Subramanyam: Waves and Oscillations, 2nd edition, Vikas publishing house Pvt. Ltd., New Delhi.
- S K Gupta, O P Varma: Waves and Oscillations, 3rd edition, R.Chand & Co., New Delhi.
- R.L. Saihgal, A Text Book of Sound, S.Chand & Company Ltd. New Delhi, Reprint 1990.

PHYSICS LAB-DSC 4A LAB: WAVES AND OPTICS
(Minimum of eight is to be conducted)

1. To study Lissajous Figures
2. To determine the Refractive Index of the Material of a given Prism using Sodium Light.
3. To determine wavelength of sodium light using Fresnel Biprism.
4. To determine wavelength of sodium light using Newton's Rings.
5. To determine wavelength of Sodium light using plane diffraction Grating
6. Air wedge – Determination of thickness of a thin paper/diameter of a thin wire.
7. Helmholtz resonator-determination of frequency of tuning fork.
8. Cauchy's constants using spectrometer
9. Polarization – Determination of unknown concentration of sugar solution by graphical method using a polarimeter.
10. Diffraction at a Straight wire -To determine the diameter of the Straight wire
11. To investigate the motion of coupled oscillators
12. To determine the Frequency of an Electrically Maintained Tuning Fork by Melde's Experiment and to verify $\lambda^2 - T$ Law
13. Familiarization with Schuster's focusing; determination of angle of prism
14. To measure the intensity using photo sensor and laser in diffraction patterns of single and double slits.

Reference Books:

- Advanced Practical Physics for students, B.L. Flint & H.T. Worsnop, 1971, Asia Publishing House.
- Advanced level Physics Practicals, Michael Nelson and Jon M. Ogborn, 4th Edition, reprinted 1985, Heinemann Educational Publishers
- A Text Book of Practical Physics, Indu Prakash and Ramakrishna, 11th Edition, 2011, Kitab Mahal, New Delhi.

Discipline Specific Elective papers: Choose 1

SEMESTER V

Course code: CME29001/ CME29001/ CME29003/ CME29004

Credits: Theory – 04, Practical – 01

Theories: 60 Lectures

COURSE OUTCOME:

After the completion of the course, Students will be able to

CO1.Deliberate in detail with examples Digital Circuits

CO2.Specify the details of Operational amplifier

CO3.Learn the characteristics of sinusoidal oscillator

CO4.Identify in detail with application Instrumentation

CO5.Deliberate the classification and characteristics of gates

CO6.Identify in detail with application of semiconductor devices

**DIGITAL AND ANALOG CIRCUITS AND INSTRUMENTATION:
DSE 1A**

Unit-1

Digital Circuits

Difference between Analog and Digital Circuits. Binary Numbers. Decimal to Binary and Binary to Decimal Conversion, AND, OR and NOT Gates (Realization using Diodes and Transistor). NAND and NOR Gates as Universal Gates. XOR and XNOR Gates. **(04 Lectures)**

De Morgan's Theorems. Boolean Laws. Simplification of Logic Circuit using Boolean algebra. Fundamental Products. Minterms and Maxterms. Conversion of a Truth Table into an Equivalent Logic Circuit by (1) Sum of Products Method and (2) Karnaugh Map. **(05 Lectures)**

Binary Addition. Binary Subtraction using 2's Complement Method).Half Adders and Full Adders and Subtractors, 4-bit binary Adder-Subtractor. **(04 Lectures)**

Semiconductor Devices and Amplifiers:

Semiconductor Diodes: p and n type semiconductors. Barrier Formation in PN Junction Diode. Qualitative Idea of Current Flow Mechanism in Forward and Reverse Biased Diode. PN junction and its characteristics. Static and Dynamic Resistance. Principle and structure of (1) LEDs (2) Photodiode (3) Solar Cell. **(05 Lectures)**

Bipolar Junction transistors: n-p-n and p-n-p Transistors. Characteristics of CB, CE and CC Configurations. Active, Cutoff, and Saturation Regions. Current gains α and β . Relations between α and β . Load Line analysis of Transistors. DC Load line and Q-point. Voltage Divider Bias Circuit for CE Amplifier. h-parameter Equivalent Circuit. Analysis of a single-stage CE amplifier using Hybrid Model. Input and Output Impedance. Current, Voltage and Power Gains. Class A, B, and C Amplifiers. Two stage RC coupled amplifier. **(12 Lectures)**

Unit-2

Operational Amplifiers (Black Box approach):

Characteristics of an Ideal and Practical Op-Amp (IC 741), Open-loop & Closed-loop Gain. CMRR, concept of Virtual ground. Applications of Op-Amps: (1) Inverting and Non-inverting Amplifiers (2) Adder (3) Subtractor (4) Differentiator (5) Integrator (6) Zero Crossing Detector **(13 Lectures)**

Sinusoidal Oscillators: Barkhausen's Criterion for Self-sustained Oscillations. Determination of Frequency of RC Oscillator **(05 Lectures)**

Instrumentations:

Introduction to CRO: Construction and Working of CRO Applications of CRO: (1) Study of Waveform, (2) Measurement of Voltage, Current, Frequency, and Phase Difference. **(03 Lectures)**

Power Supply: Half-wave Rectifiers. Centre-tapped and Bridge Full-wave Rectifiers Calculation of Ripple Factor and Rectification Efficiency, Basic idea about capacitor filter, Zener Diode and Voltage Regulation. **(06 Lectures)**

Timer IC: IC 555 Pin diagram and its application as Astable & Monostable Multivibrator.

(03 Lectures)

Reference Books:

- Integrated Electronics, J. Millman and C.C. Halkias, 1991, Tata Mc-Graw Hill.
- Electronic devices and circuits, S. Salivahanan and N. Suresh Kumar, 2012, Tata Mc-Graw Hill.
- Microelectronic Circuits, M.H. Rashid, 2ndEdn. 2011, Cengage Learning.
- Modern Electronic Instrumentation & Measurement Tech., Helfrick&Cooper,1990, PHI Learning
- Digital Principles & Applications, A.P. Malvino, D.P. Leach & Saha, 7th Ed.,2011, Tata McGraw Hill
- Microelectronic circuits, A.S. Sedra, K.C. Smith, A.N. Chandorkar, 2014, 6th Edn., Oxford University Press.
- Fundamentals of Digital Circuits, A. Anand Kumar, 2nd Edition, 2009, PHI Learning Pvt. Ltd.
- OP-AMP and Linear Digital Circuits, R.A. Gayakwad, 2000, PHI Learning Pvt. Ltd.

**PRACTICALS - DSE LAB: DIGITAL AND ANALOG CIRCUITS AND INSTRUMENTS
(Minimum of eight is to be conducted)**

1. To verify and design AND, OR, NOT and XOR gates using NAND gates.
2. Zener diode as a voltage regulator.
3. Half adder and Full adder
4. FET-characteristics
5. To study the characteristics of a Transistor in CE configuration.
6. To study the characteristics of a Transistor in CB configuration.
7. Phase shift oscillator
8. Op-amp has inverting and non-inverting amplifier
9. Wien Bridge Oscillator
10. Hartley oscillator
11. CE-amplifier and negative feedback amplifier
12. To minimize a given logic circuit.
13. To design an astable multivibrator of given specifications using 555 Timer.

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14. To design a monostable multivibrator of given specifications using 555 Timer

Reference Books:

- Basic Electronics: A text lab manual, P.B. Zbar, A.P. Malvino, M.A. Miller, 1994, Mc-Graw Hill.
- Electronics: Fundamentals and Applications, J.D. Ryder, 2004, Prentice Hall.
- OP-Amps and Linear Integrated Circuit, R. A. Gayakwad, 4th edition, 2000, Prentice Hall.
- Electronic Principle, Albert Malvino, 2008, Tata Mc-Graw Hill

SEMESTER V

Course code: CME29201/ CME29202/ CME29203/ CME29204

Credits: Theory – 04, Practical – 01

Theories: 60 Lectures

COURSE OUTCOME:

After the completion of the course, Students will be able to

CO1. Write down in detail with application of crystal structure

CO2. Write down the details of Elementary lattice dynamics

CO3. Deliberate in detail with examples Magnetic properties of matter

CO4. Identify the characteristics of elementary band theory

CO5. Learn the classification and characteristics of superconductivity

CO6. Understand the elastic properties of solids and lattice vibrations

SOLID STATE PHYSICS: DSE 1B

Unit-1

Crystal Structure: Solids: Amorphous and Crystalline Materials. Lattice Translation Vectors. Lattice with a Basis – Central and Non-Central Elements. Unit Cell. Miller Indices. Reciprocal Lattice. Types of Lattices. Brillouin Zones. Diffraction of X-rays by Crystals. Bragg's Law. Bragg spectrometer. Continuous x-ray spectra, Duane and Hunt limit. Characteristic x-ray spectra. Mosley law and its significance. **(10 Lectures)**

Elementary Lattice Dynamics: Lattice Vibrations and Phonons: Linear Monatomic and Diatomic Chains. Acoustical and Optical Phonons. Qualitative Description of the Phonon Spectrum in Solids. Dulong and Petit's Law, Einstein and Debye theories of specific heat of solids. **(09 Lectures)**

Magnetic Properties of Matter: Dia, Para, Ferri and Ferromagnetic Materials. Classical Langevin Theory of dia – and Paramagnetic Domains. Quantum Mechanical Treatment of Paramagnetism. Curie's law, Weiss's Theory of Ferromagnetism and Ferromagnetic Domains. Discussion of B-H Curve. Hysteresis and Energy Loss. **(11 Lectures)**

Unit-2

Electrical Properties of metals: Free electron theory, Quantum theory, Conductivity of metals on the basis of free electron theory- Ohm's law. **(04 Lectures)**

Dielectric Properties of Materials: Polarization. Local Electric Field at an Atom. Depolarization Field. Electric Susceptibility. Polarizability. Clausius Mosotti Equation. Classical Theory of Electric Polarizability. Normal and Anomalous Dispersion. Cauchy and Sellmeier relations. Langevin-Debye equation. Complex Dielectric Constant. Optical Phenomena. Application: Plasma Oscillations, Plasma Frequency, Plasmon's. **(10 Lectures)**

Elementary band theory: Kronig Penny model. Band Gaps. Conductors, Semiconductors and insulators. P and N type Semiconductors, carrier concentration in intrinsic semiconductor, Fermi energy, effect of temperature and concentration on Fermi energy, Conductivity of Semiconductors, mobility, Hall Effect, Hall coefficient. **(10 Lectures)**

Superconductivity: Experimental Results. Critical Temperature. Critical magnetic field. Meissner effect. Type I and type II Superconductors, London's Equation and Penetration Depth. Isotope effect. BCS theory, application of Superconductors. **(06 Lectures)**

Reference Books:

- Introduction to Solid State Physics, Charles Kittel, 8th Ed., 2004, Wiley India Pvt. Ltd.
- Elements of Solid State Physics, J.P. Srivastava, 2nd Ed., 2006, Prentice-Hall of India
- Introduction to Solids, Leonid V. Azaroff, 2004, Tata Mc-Graw Hill
- Solid State Physics, Neil W. Ashcroft and N. David Mermin, 1976, Cengage Learning
- Solid State Physics, Rita John, 2014, McGraw Hill
- Solid-state Physics, H. Ibach and H Luth, 2009, Springer
- Elementary Solid State Physics, 1/e M. Ali Omar, 1999, Pearson India
- Solid State Physics, M.A. Wahab, 2011, Narosa Publications

PRACTICALS-DSE LAB: SOLID STATE PHYSICS

(Minimum of eight is to be conducted)

1. Determination of energy gap of a semiconductor using Meter Bridge.
2. Determination Fermi energy of metal(copper)
3. Study of powder X-ray photograph-the determination of inter planar distances.
4. Study of hall effect
5. To measure the Dielectric Constant of a given solid Materials
6. To determine value of Boltzmann constant using V-I characteristic of PN diode.
7. To determine value of Planck's constant using LEDs of at least 4 different colors.
8. Study of LDR characteristics.
9. Verification of Inverse square law for light Intensity using a Photo-diode.
10. To determine the refractive index of a dielectric layer using SPR
11. To study the PE Hysteresis loop of a Ferroelectric Crystal.
12. To measure the resistivity of a semiconductor (Ge) crystal with temperature by four-probe method (from room temperature to 150 °C) and to determine its band gap
13. To draw the BH curve of iron using a Solenoid and determine the energy loss from Hysteresis.
14. To determine the complex dielectric constant and plasma frequency of metal using Surface Plasmon resonance (SPR)

Reference Books

- Advanced Practical Physics for students, B.L. Flint and H.T. Worsnop, 1971, Asia Publishing House.
- Advanced level Physics Practicals, Michael Nelson and Jon M. Ogborn, 4th Edition, reprinted 1985, Heinemann Educational Publishers
- A Text Book of Practical Physics, Indu Prakash and Ramakrishna, 11th Ed., 2011, Kitab Mahal, New Delhi
- Elements of Solid State Physics, J.P. Srivastava, 2nd Ed., 2006, Prentice-Hall of India

SEMESTER V

Course code: CME29601/CME29602/CME29603/CME29604

Credits: Theory – 04, Practical – 01

Theories: 60 Lectures

COURSE OUTCOME:

After successful completion of the course, the student is able to:

CO1.Learn in detail with examples calculus of functions of more than one variable

CO2.Specify the details of fourier series

CO3.Specify in depth integrals

CO4.Deliberate in detail with application of partial differential equations

CO5.Understand in depth complex analysis

CO6.Deliberate in detail with examples of beta and gamma function

MATHEMATICAL PHYSICS: DSE 1D

Unit-1

Calculus of functions of more than one variable: Partial derivatives, exact and inexact differentials. Integrating factor, with simple illustration. Constrained Maximization using Lagrange Multipliers. **(06 Lectures)**

Fourier series: Periodic functions. Orthogonality of sine and cosine functions, Dirichlet Conditions (Statement only). Expansion of periodic functions in a series of sine and cosine functions and determination of Fourier coefficients. Complex representation of Fourier series. Expansion of functions with arbitrary period. Expansion of non-periodic functions over an interval. Even and odd functions and their Fourier expansions. Application. Summing of Infinite Series. **(10 Lectures)**

Frobenius Method and Special Functions: Singular Points of Second Order Linear Differential Equations and their importance. Frobenius method and its applications to differential equations. Legendre, Bessel, Hermite and Laguerre Differential Equations. Properties of Legendre Polynomials: Rodrigues Formula, Orthogonality. Simple recurrence relations. **(14 Lectures)**

Unit-2

Some Special Integrals: Beta and Gamma Functions and Relation between them. Expression of Integrals in terms of Gamma Functions. Error Function (Probability Integral). **(06 Lectures)**

Partial Differential Equations: Solutions to partial differential equations, using separation of variables: Laplace's Equation in problems of rectangular, cylindrical and spherical symmetry. **(10 Lectures)**

Complex Analysis: Brief Revision of Complex Numbers and their Graphical Representation. Euler's formula, De Moivre's theorem, Roots of Complex Numbers. Functions of Complex Variables. Analyticity and Cauchy-Riemann Conditions. Examples of analytic functions. Singular functions: poles and branch points, order of singularity, branch cuts. Integration of a function of a complex variable. Cauchy's Inequality. Cauchy's Integral formula.

(14 Lectures)

Reference Books:

- Mathematical Methods for Physicists: Arfken, Weber, 2005, Harris, Elsevier.
- Fourier analysis by M.R. Spiegel, 2004, Tata McGraw-Hill.
- Mathematics for Physicists, Susan M. Lea, 2004, Thomson Brooks/Cole.
- An Introduction to Ordinary Differential Equations, Earl A Coddington, 1961, PHI Learning.
- Differential Equations, George F. Simmons, 2006, Tata McGraw-Hill.
- Essential Mathematical Methods, K.F. Riley and M.P. Hobson, 2011, Cambridge University Press
- Partial Differential Equations for Scientists and Engineers, S.J. Farlow, 1993, Dover Publications.
- Mathematical methods for Scientists and Engineers, D.A. McQuarrie, 2003, Viva Books.

PRACTICALS -DSE LAB: MATHEMATICAL PHYSICS

The aim of this course is not just to teach computer programming and numerical analysis but to emphasize its role in solving problems in Physics.

- Highlights the use of computational methods to solve physical problems
- Use of computer language as a tool in solving physics problems (applications)
- The course will consist of lectures (both theory and practical) in the ComputerLab
- Evaluation done not on the programming but on the basis of formulating the problem
- Aim at teaching students to construct the computational problem to be solved
- Students can use anyone operating system Linux or Microsoft Windows

Topics Description with Applications	Introduction and Overview
Introduction and Overview	Computer architecture and organization, memory and Input/output devices
Basics of scientific computing	Binary and decimal arithmetic, Floating point numbers, algorithms, Sequence, Selection and Repetition, single and double precision arithmetic, underflow & overflow-emphasize the importance of making equations in terms of dimensionless variables, Iterative methods
Errors and error Analysis	Truncation and round off errors, Absolute and relative errors, Floating point computations.
Review of C & C++ Programming fundamentals	Introduction to Programming, constants, variables and data types, operators and Expressions, I/O statements, scanf and printf, c in and c out, Manipulators for data formatting, Control statements (decision making and looping statements) (<i>If-statement. If-else Statement. Nested if Structure. Else-if Statement. Ternary Operator. Goto Statement. Switch Statement. Unconditional and Conditional Looping. While-Loop. Do-While Loop. FOR Loop. Break and Continue Statements. Nested Loops</i>), Arrays (<i>1D&2D</i>) and strings, user defined functions, Structures and Unions, Idea of classes and objects
Programs: using C/C++ language	Sum & average of a list of numbers, largest of a given list of numbers and its location in the list, sorting of numbers in ascending-descending order, Binary search
Random number generation	Area of circle, area of square, volume of sphere, value of pi (π)

Solution of Algebraic and Transcendental equations by Bisection, Newton Raphson and Secant methods	Solution of linear and quadratic equation, solving $\alpha = \tan \alpha; I = I_0 \left(\frac{\sin \alpha}{\alpha} \right)^2$ in optics
Interpolation by Newton Gregory Forward and Backward difference formula, Error estimation of linear interpolation	Evaluation of trigonometric functions e.g. $\sin \theta, \cos \theta, \tan \theta, \text{etc.}$
Numerical differentiation (Forward and Backward difference formula) and Integration (Trapezoidal and Simpson rules), Monte Carlo method	Given Position with equidistant time data to calculate velocity and acceleration and vice-versa. Find the area of B-H Hysteresis loop
Solution of Ordinary Differential Equations (ODE) First order Differential equation Euler, modified Euler and Runge-Kutta (RK) second and fourth order methods	First order differential equation <ul style="list-style-type: none"> • Radioactive decay • Current in RC, LC circuits with DC source • Newton's law of cooling • Classical equations of motion Attempt following problems using RK 4 order method: <ul style="list-style-type: none"> • Solve the coupled differential equations $\frac{dx}{dt} = y + x - \frac{x^3}{3}; \frac{dy}{dx} = -x$ for four initial conditions $x(0) = 0, y(0) = -1, -2, -3, -4$. Plot x vs y for each of the four initial conditions on the same screen for $0 \leq t \leq 15$ The differential equation describing the motion of a Pendulum is $\frac{d^2\theta}{dt^2} = -\sin(\theta)$. The pendulum is released from rest at an angular displacement α , i.e. $\theta(0) = \alpha, \text{ and } \theta'(0) = 0$. Solve the equation for $\alpha = 0.1, 0.5$ and 1.0 and plot θ as a function of time in the range $0 \leq t \leq 8\pi$. Also plot the analytic solution valid for small $\theta \sin(\theta) = \theta$

Reference Books:

- Introduction to Numerical Analysis, S.S. Sastry, 5thEdn., 2012, PHI Learning Pvt. Ltd.
- Schaum's Outline of Programming with C++. J.Hubbard, 2000, McGraw-Hill Publications.
- Numerical Recipes in C++: The Art of Scientific Computing, W.H. Press et al., 3rdEdn., 2007, Cambridge University Press.
- A first course in Numerical Methods, Uri M. Ascher and Chen Greif, 2012, PHI Learning
- Elementary Numerical Analysis, K.E. Atkinson, 3rdEdn. , 2007, Wiley India Edition.
- Numerical Methods for Scientists and Engineers, R.W. Hamming, 1973, Courier Dover Pub.
- An Introduction to Computational Physics, T. Pang, 2ndEdn., 2006, Cambridge Univ. Press

Discipline Specific Elective papers: Choose 1

SEMESTER VI

Course code: CMF29001/ CMF29002/ CMF29003/ CMF29004

Credits: Theory – 04, Practical – 01

Theories: 60 Lectures

COURSE OUTCOME:

After successful completion of the course, the student is able to:

CO1.Understand in detail with examples planks quantum

CO2.Identify the classification and characteristics of Planck's quantum

CO3.Learn the classification and characteristics of fission and fusion reaction

CO4.Write down the characteristics of matter waves

CO5.Understand the classification and characteristics of radioactivity

CO6.Specify in depth Eigen values

ELEMENTS OF MODERN PHYSICS: DSE 1A

Unit-1

Planck's quantum, Planck's constant and light as a collection of photons; Photo-electric effect and Compton scattering. De Broglie wavelength and matter waves; Davisson-Germer experiment.

(08 Lectures)

Problems with Rutherford model- instability of atoms and observation of discrete atomic spectra; Bohr's quantization rule and atomic stability; calculation of energy levels for hydrogen like atoms and their spectra.

(04 Lectures)

Position measurement-gamma ray microscope thought experiment; wave-particle duality, Heisenberg uncertainty principle-impossibility of a particle following a trajectory; Estimating minimum energy of a confined particle using uncertainty principle; Energy-time uncertainty principle

(04 lectures)

UG-Physics- CBCS Scheme

Two slit interference experiment with photons, atoms and particles; linear superposition principle as a consequence; Matter waves and wave amplitude; Schrodinger equation for non-relativistic particles; Momentum and Energy operators; stationary states; physical interpretation of wave function, probabilities and normalization; Probability and probability current densities in one dimension. **(10 Lectures)**

Fission and fusion - mass deficit, relativity and generation of energy; Fission - nature of fragments and emission of neutrons. Nuclear reactor: slow neutrons interacting with Uranium 235; Fusion and thermonuclear reactions. **(04 Lectures)**

Unit-2

One dimensional infinitely rigid box- energy eigenvalues and Eigen functions, normalization; Quantum dot as an example; Quantum mechanical scattering and tunnelling in one dimension - across a step potential and across a rectangular potential barrier. **(12 Lectures)**

Size and structure of atomic nucleus and its relation with atomic weight; Impossibility of an electron being in the nucleus as a consequence of the uncertainty principle. Nature of nuclear force, NZ graph, semi-empirical mass formula and binding energy. **(07 Lectures)**

Radioactivity: stability of nucleus; Law of radioactive decay; Mean life & half-life; Alpha decay; Beta decay - energy released, spectrum and Pauli's prediction of neutrino; gamma-ray emission. **(11 Lectures)**

Reference Books:

- Concepts of Modern Physics, Arthur Beiser, 2009, McGraw-Hill
- Modern Physics, John R. Taylor, Chris D. Zafiratos, Michael A. Dubson, 2009, PHI Learning
- Six Ideas that Shaped Physics: Particle Behave like Waves, Thomas A. Moore, 2003, McGraw Hill
- Quantum Physics, Berkeley Physics Course Vol.4. E.H. Wichman, 2008, Tata McGraw-Hill

Co.

- Modern Physics, R.A. Serway, C.J. Moses, and C.A.Moyer, 2005, Cengage Learning
- Modern Physics, G. Kaur and G.R. Pickrell, 2014, McGraw Hill

**PRACTICALS –DSE LAB: ELEMENTS OF MODERN PHYSICS
(Minimum of eight is to be conducted)**

1. To determine value of Boltzmann constant using V-I characteristic of PN diode.
2. To determine work function of material of filament of directly heated vacuum diode.
3. To determine value of Planck's constant using LEDs of at least 4 different colors.
4. To determine the ionization potential of mercury.
5. To determine the wavelength of H-alpha emission line of Hydrogen atom.
6. To determine the absorption lines in the rotational spectrum of Iodine vapor. To study the diffraction patterns of single and double slits using laser source and measure its intensity variation using Photo sensor and compare with incoherent source – Na light.
7. Photo-electric effect: photo current versus intensity and wavelength of light; maximum energy of photo-electrons versus frequency of light
8. To determine the value of e/m by magnetic focusing.
9. To setup the Millikan oil drop apparatus and determine the charge of an electron.
10. To study the diffraction patterns of single and double slits using laser source and measure its intensity variation using Photo sensor and compare with incoherent source – Na light.
11. To determine work function of material of filament of directly heated vacuum diode.
12. To study the Divergence of laser
13. Spectral response of solar cell
14. Bridge rectifier with and without filters
15. Clipping and clamping circuits using diodes
16. Study of characteristics of GM tube and determination of operating voltage and plateau length using background radiation as source (without commercial source).
17. Study of counting statistics using background radiation using GM counter.
18. Study of radiation in various materials (e.g. KSO₄ etc.). Investigation of possible radiation in different routine materials by operating GM at operating voltage.
19. Study of absorption of beta particles in Aluminum using GM counter.

UG-Physics- CBCS Scheme

20. Half life of k-40
21. To determine the ionization potential of xenon.
22. Cockcroft-Walton Voltage multiplier.
23. Determination of range of electron in aluminum using GM counter

References:

- Physics of Radiation Therapy: F M Khan Williams and Wilkins, Third edition (2003)
- Advanced Practical Physics for students, B.L. Flint & H.T. Worsnop, 1971, Asia Publishing House.
- Practical Applications of Radioactivity and Nuclear Radiations, G.C. Lowental and P.L. Airey, Cambridge University Press, U.K., 2001
- A. Martin and S.A. Harbisor, An Introduction to Radiation Protection, John Willey & Sons, Inc. New York, 1981.
- Advanced Practical Physics for students, B.L. Flint & H.T. Worsnop, 1971, Asia Publishing House.
- Advanced level Physics Practicals, Michael Nelson and Jon M. Ogborn, 4th Edition, reprinted 1985, Heinemann Educational Publishers
- A Text Book of Practical Physics, Indu Prakash and Ramakrishna, 11th Edition, 2011, Kitab Mahal, New Delhi.

SEMESTER VI

Course code: CMF29201/ CMF29202/ CMF29203/ CMF29204

Credits: Theory – 05

Theories: 75 Lectures

COURSE OUTCOME:

After the completion of the course, Students will be able to

CO1. Write down in detail with application and properties of nuclei

CO2. Learn in detail with application and nuclear models

CO3. Understand in detail with examples radioactivity

CO4. Identify the details of particle physics

CO5. Write down the details of particle accelerators

CO6. Write down the details of detector for nuclear radiator

Nuclear & Particle Physics: DSE 1B

Unit-1

General Properties of Nuclei: Constituents of nucleus and their Intrinsic properties, quantitative facts about size, mass, charge density (matter energy), binding energy, average binding energy and its variation with mass number, main features of binding energy versus mass number curve, N/A plot, angular momentum, parity, magnetic moment, electric moments, nuclear excited states.

(10 Lectures)

Nuclear Models: Liquid drop model approach, semi empirical mass formula and binding energy, significance of various terms, condition of nuclear stability. Two nucleon separation energies, Fermi gas model (degenerate fermion gas, nuclear symmetry potential in Fermi gas), evidence for nuclear shell structure, nuclear magic numbers, basic assumption of shell model, concept of mean field, residual interaction, concept of nuclear force.

(12 Lectures)

Nuclear Reactions: Types of Reactions, Conservation Laws, kinematics of reactions, Q-value, reaction rate, reaction cross section, Concept of compound and direct reaction, resonance reaction, Coulomb scattering (Rutherford scattering).

(08 Lectures)

Unit-2

Radioactivity decay : (a) Alpha decay: basics of α -decay processes, theory of α -emission, Gamow factor, Geiger Nuttall law, α -decay spectroscopy. (b) beta-decay: energy kinematics for beta-decay, positron emission, electron capture, neutrino hypothesis. (c) Gamma decay: Gamma rays emission & kinematics, internal conversion.

(10 Lectures)

Interaction of Nuclear Radiation with matter: Energy loss due to ionization (Bethe-Block formula), energy loss of electrons, Cerenkov radiation, Gamma ray interaction through matter, photoelectric effect, Compton scattering, pair production, neutron interaction with matter.

(08 Lectures)

Detector for Nuclear Radiations: Gas detectors: estimation of electric field, mobility of particle, for ionization chamber and GM Counter. Basic principle of Scintillation. Detectors and construction of photo-multiplier tube (PMT). Semiconductor Detectors (Si & Ge) for charge particle and photon detection (concept of charge carrier and mobility)

(08 Lectures)

Particle Accelerators: Accelerator facility available in India: Van-de Graaff generator (Tandem accelerator), Linear accelerator, Cyclotron, Synchrotrons.

(04 Lectures)

Unit-3

Particle physics: Particle interactions; basic features, types of particles and its families. Symmetries and Conservation Laws: energy and momentum, angular momentum, parity, baryon number, Lepton number, Isospin, Strangeness and charm, concept of quark model, color quantum number and gluons.

(15 Lectures)

Reference Books:

- Introductory nuclear Physics by Kenneth S. Krane (Wiley India Pvt. Ltd., 2008).
- Concepts of nuclear physics by Bernard L. Cohen. (Tata Mcgraw Hill, 1998).
- Introduction to the physics of nuclei & particles, R.A. Dunlap. (Thomson Asia, 2004)
- Introduction to Elementary Particles, D. Griffith, John Wiley & Sons
- Quarks and Leptons, F. Halzen and A.D. Martin, Wiley India, New Delhi
- Basic ideas and concepts in Nuclear Physics - An Introductory Approach by K. Heyde (IOP-

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Institute of Physics Publishing, 2004).

- Radiation detection and measurement, G.F. Knoll (John Wiley & Sons, 2000).
- Theoretical Nuclear Physics, J.M. Blatt & V.F. Weisskopf (Dover Pub.Inc., 1991)

SEMESTER VI

Course code: CMF29401/CMF29402/CMF29403/CMF29404

Credits: Theory – 04, Practical – 01

Theories: 60 Lectures

COURSE OUTCOME:

After successful completion of the course, the student is able to:

CO1.Understand the classification and characteristics of time dependent Schrodinger equation

CO2.Understand in depth time independent Schrodinger equation

CO3.Understand the classification and characteristics of quantum theory of hydrogen like atoms

CO4.Specify the details of atoms in external magnetic field

CO5.Write down the characteristics of many electron atoms

QUANTUM MECHANICS: DSE 1C

Unit-1

Time dependent Schrodinger equation: Time dependent Schrodinger equation and dynamical evolution of a quantum state; Properties of Wave Function. Interpretation of Wave Function Probability and probability current densities in three dimensions; Conditions for Physical Acceptability of Wave Functions. Normalization. Linearity and Superposition Principles. Eigenvalues and Eigenfunctions. Position, momentum & Energy operators; commutator of position and momentum operators; Expectation values of position and momentum. Wave Function of a Free Particle. **(08 Lectures)**

Time independent Schrodinger equation-Hamiltonian, stationary states and energy eigenvalues; expansion of an arbitrary wavefunction as a linear combination of energy eigenfunctions; General solution of the time dependent Schrodinger equation in terms of linear combinations of stationary states; Application to the spread of Gaussian wavepacket for a free particle in one dimension; wave packets, Fourier transforms and momentum space wavefunction; Position-momentum uncertainty principle. **(10 Lectures)**

UG-Physics- CBCS Scheme

General discussion of bound states in an arbitrary potential- continuity of wave function, boundary condition and emergence of discrete energy levels; application to one-dimensional problem- square well potential; Quantum mechanics of simple harmonic oscillator-energy levels and energy eigenfunctions using Frobenius method. **(12 Lectures)**

Unit-2

Quantum theory of hydrogen-like atoms: time independent Schrodinger equation in spherical polar coordinates; separation of variables for the second order partial differential equation; angular momentum operator and quantum numbers; Radial wave functions from Frobenius method; Orbital angular momentum quantum numbers l and m ; s, p, d,.. Shells (idea only) **(10 Lectures)**

Atoms in Electric and Magnetic Fields:- Electron Angular Momentum. Space Quantization. Electron Spin and Spin Angular Momentum. Larmor's Theorem. Spin Magnetic Moment. Stern-Gerlach Experiment. Zeeman Effect: Electron Magnetic Moment and Magnetic Energy, Gyromagnetic Ratio and Bohr Magnetron. **(08 Lectures)**

Atoms in External Magnetic Fields:- Normal and Anomalous Zeeman Effect. **(02 Lectures)**

Many electron atoms:- Pauli's Exclusion Principle. Symmetric and Antisymmetric Wave Functions. Periodic table. Fine structure. Spin orbit coupling. Spectral Notations for Atomic States. Total Angular Momentum. Vector Model. Spin-orbit coupling in atoms-L-S and J-J couplings. **(10 Lectures)**

Reference Books:

- A Text book of Quantum Mechanics, P.M. Mathews & K. Venkatesan, 2nd Ed., 2010, McGraw Hill
- Quantum Mechanics, Robert Eisberg and Robert Resnick, 2ndEdn. 2002, Wiley.
- Quantum Mechanics, Leonard I. Schiff, 3rdEdn. 2010, Tata McGraw Hill.
- Quantum Mechanics, G. Aruldas, 2ndEdn. 2002, PHI Learning of India.
- Quantum Mechanics, Bruce Cameron Reed, 2008, Jones and Bartlett Learning.
- Quantum Mechanics for Scientists & Engineers, D.A.B. Miller, 2008, Cambridge University Press
- Quantum Mechanics, Eugen Merzbacher, 2004, John Wiley and Sons, Inc.

- Introduction to Quantum Mechanics, David J. Griffith, 2nd Ed. 2005, Pearson Education
- Quantum Mechanics, Walter Greiner, 4thEdn., 2001, Springer

PRACTICAL-DSE LAB: QUANTUM MECHANICS

Use C/C++/Scilab for solving the following problems based on Quantum Mechanics Like

1. Solve the s-wave Schrodinger equation for the ground state and the first excited state of the hydrogen atom:

$$\frac{d^2 y}{dr^2} = A(r)u(r), A(r) = \frac{2m}{\hbar^2} [V(r) - E] \text{ where } V(r) = \frac{-e^2}{r}$$

Here, m is the reduced mass of the electron. Obtain the energy Eigen values and plot the corresponding wave functions. Remember that the ground state energy of the hydrogen atom is ≈ -13.6 eV. Take $e=3.795$ (eVÅ)^{1/2}, $\hbar c = 1973$ (eVÅ) and $m = 0.511 \times 10^6$ eV/c²

2. Solve the s-wave radial Schrodinger equation for an atom

$$\frac{d^2 y}{dr^2} = A(r)u(r), A(r) = \frac{2m}{\hbar^2} [V(r) - E]$$

Where m is the reduced mass of the system (which can be chosen to be the mass of an electron), for the screened coulomb potential

$$V(r) = \frac{-e^2}{r} e^{-\frac{r}{a}}$$

Find the energy (in eV) of the ground state of the atom to an accuracy of three Significant digits. Also, plot the corresponding wave function. Take $e = 3.795$ (eVÅ)^{1/2}, $m = 0.511 \times 10^6$ eV/c², and $a = 3$ Å, 5 Å, 7 Å. In these units $\hbar c = 1973$ (eVÅ). The ground state energy is expected to be above -12 eV in all three cases.

3. Solve the s-wave radial Schrodinger equation for a particle of mass m:

$$\frac{d^2 y}{dr^2} = A(r)u(r), A(r) = \frac{2m}{\hbar^2} [V(r) - E]$$

For the anharmonic oscillator potential

$$V(r) = \frac{1}{2}kr^2 + \frac{1}{3}br^3$$

for the ground state energy (in MeV) of the particle to an accuracy of three significant digits. Also, plot the corresponding wave function. Choose $m = 940 \text{ MeV}/c^2$, $k = 100 \text{ MeV fm}^{-2}$, $b = 0, 10, 30 \text{ MeV fm}^{-3}$. In these units, $\hbar c = 197.3 \text{ MeV fm}$. The ground state energy I expected to lie between 90 and 110 MeV for all three cases.

4. Solve the s-wave radial Schrodinger equation for the vibrations of hydrogen molecule:

$$\frac{d^2 y}{dr^2} = A(r)u(r), A(r) = \frac{2\mu}{\hbar^2} [V(r) - E]$$

where μ is the reduced mass of the two-atom system for the Morse potential

$$V(r) = D \left(e^{-2\alpha r^1} - e^{-\alpha r^1} \right), r^1 = \frac{r - r_0}{r}$$

Find the lowest vibrational energy (in MeV) of the molecule to an accuracy of three significant digits. Also plot the corresponding wave function. Take: $m = 940 \times 10^6 \text{ eV}/c^2$, $D = 0.755501 \text{ eV}$, $\alpha = 1.44$, $r_0 = 0.131349 \text{ \AA}$

Laboratory based experiments:

5. Study of Electron spin resonance- determine magnetic field as a function of the resonance frequency
6. Study of Zeeman effect: with external magnetic field; Hyperfine splitting
7. To study the quantum tunnelling effect with solid state device, e.g. tunnelling current in backward diode or tunnel diode.

Reference Books:

- Schaum's Outline of Programming with C++. J.Hubbard, 2000, McGraw-Hill Publications.
- Numerical Recipes in C: The Art of Scientific Computing, W.H.Press et al., 3rdEdn., 2007, Cambridge University Press.
- Elementary Numerical Analysis, K.E.Atkinson, 3 r dEdn. , 2007, Wiley India Edition.

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- A Guide to MATLAB, B.R. Hunt, R.L. Lipsman, J.M. Rosenberg, 2014, 3rd Edn., Cambridge University Press
- Simulation of ODE/PDE Models with MATLAB®, OCTAVE and SCILAB:
- Scientific and Engineering Applications: A. Vande Wouwer, P. Saucez, C. V. Fernández. 2014 Springer ISBN: 978-3319067896
- Scilab by example: M. Affouf 2012 ISBN: 978-1479203444
- Scilab (A Free Software to Matlab): H. Ramchandran, A.S. Nair. 2011 S. Chand and Company, New Delhi ISBN: 978-8121939706
- Scilab Image Processing: Lambert M. Surhone. 2010 Betascript Publishing ISBN: 978-6133459274A
- Quantum Mechanics, Leonard I. Schiff, 3rd Edn. 2010, Tata McGraw Hill.
- Quantum Mechanics, Bruce Cameron Reed, 2008, Jones and Bartlett Learning.

Skill Enhancement Course (SEC) (Credit: 02 each)

SEMESTER V

Course code: CME69001/ CME69002/ CME69003/ CME69004

Credits: Theory – 02

Theories: 30 Lectures

COURSE OUTCOME:

After the completion of the course, Students will be able to

CO1.Understand the characteristics of fossil fuel

CO2.Learn in detail with application of wind energy

CO3.Specify in detail with application of ocean energy and hydro energy

CO4.Identify the characteristics of geothermal energy

CO5.Deliberate the characteristics of electromagnetic energy

CO6.Deliberate the characteristics of piezoelectric energy harvesting

RENEWABLE ENERGY AND ENERGY HARVESTING: SEC 1A

Fossil fuels and Alternate Sources of energy: Fossil fuels and Nuclear Energy, their limitation, need of renewable energy, non-conventional energy sources. An overview of developments in Offshore Wind Energy, Tidal Energy, Wave energy systems, Ocean Thermal Energy Conversion, solar energy, biomass, biochemical conversion, biogas generation, geothermal energy tidal energy, Hydroelectricity. **(03 Lectures)**

Solar energy: Solar energy, its importance, storage of solar energy, solar pond, non convective solar pond, applications of solar pond and solar energy, solar water heater, flat plate collector, solar distillation, solar cooker, solar green houses, solar cell, absorption air conditioning. Need and characteristics of photovoltaic (PV) systems, PV models and equivalent circuits, and sun tracking systems. **(06 Lectures)**

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Wind Energy harvesting: Fundamentals of Wind energy, Wind Turbines and different electrical machines in wind turbines, Power electronic interfaces, and grid interconnection topologies.

(03 Lectures)

Ocean Energy: Ocean Energy Potential against Wind and Solar, Wave Characteristics and Statistics, Wave Energy Devices.

(03 Lectures)

Tide characteristics and Statistics, Tide Energy Technologies, Ocean Thermal Energy, Osmotic Power, Ocean Bio-mass.

(02 Lectures)

Geothermal Energy: Geothermal Resources, Geothermal Technologies.

(02 Lectures)

Hydro Energy: Hydropower resources, hydropower technologies, environmental impact of hydro power sources.

(02 Lectures)

Piezoelectric Energy harvesting: Introduction, Physics and characteristics of piezoelectric effect, materials and mathematical description of piezoelectricity, Piezoelectric parameters and modeling piezoelectric generators, Piezoelectric energy harvesting applications, Human power

(04 Lectures)

Electromagnetic Energy Harvesting: Linear generators, physics mathematical models, recent applications

(02 Lectures)

Carbon captured technologies, cell, batteries, power consumption

(02 Lectures)

Environmental issues and Renewable sources of energy, sustainability.

(01 Lecture)

Demonstrations and Experiments

1. Demonstration of Training modules on solar energy, wind energy, etc.
2. Conversion of vibration to voltage using piezoelectric materials
3. Conversion of thermal energy into voltage using thermoelectric modules.

Reference Books:

- Non-conventional energy sources - G.D Rai - Khanna Publishers, New Delhi
- Solar energy - M P Agarwal - S Chand and Co. Ltd.
- Solar energy - Suhas P Sukhative Tata McGraw - Hill Publishing Company Ltd.
- Godfrey Boyle, “Renewable Energy, Power for a sustainable future”, 2004, Oxford University Press, in association with The Open University.
- Dr. P Jayakumar, Solar Energy: Resource Assesment Handbook, 2009
- J.Balfour, M.Shaw and S. Jarosek, Photovoltaics, Lawrence J Goodrich (USA).
- http://en.wikipedia.org/wiki/Renewable_energy

SEMESTER V

Course code: CME69201/ CME69202/ CME69203/ CME69204

Credits: Theory – 02

Theories: 30 Lectures

COURSE OUTCOME:

After the completion of the course, Students will be able to

CO1.Deliberate the details of photons

CO2.Understand the classification and characteristics of Atomic and nuclear physics

CO3.Write down the classification and characteristics of nuclear techniques

CO4.Write down the characteristics of radiation safety management

CO5.Identify in detail with application of radiation detection

RADIATION SAFETY: SEC 2A

Basics of Atomic and Nuclear Physics: Basic concept of atomic structure; X rays characteristic and production; concept of bremsstrahlung and auger electron, The composition of nucleus and its properties, mass number, isotopes of element, spin, binding energy, stable and unstable isotopes, law of radioactive decay, Mean life and half life, basic concept of alpha, beta and gamma decay, concept of cross section and kinematics of nuclear reactions, types of nuclear reaction, Fusion, fission. **(06 Lectures)**

Interaction of Radiation with matter: Types of Radiation: Alpha, Beta, Gamma and Neutron and their sources, sealed and unsealed sources, **Interaction of Photons** – Photoelectric effect, Compton Scattering, Pair Production, Linear and Mass Attenuation Coefficients, **Interaction of Charged Particles:** Heavy charged particles - Beth-Bloch Formula, Scaling laws, Mass Stopping Power, Range, Straggling, Channeling and Cherenkov radiation. Beta Particles- Collision and Radiation loss (Bremsstrahlung), **Interaction of Neutrons-** Collision, slowing down and Moderation. **(07 Lectures)**

Radiation detection and monitoring devices: Radiation Quantities and Units: Basic idea of different units of activity, KERMA, exposure, absorbed dose, equivalent dose, effective dose,

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collective equivalent dose, Annual Limit of Intake (ALI) and derived Air Concentration (DAC).

Radiation detection: Basic concept and working principle of gas detectors (Ionization Chambers, Proportional Counter, Multi-Wire Proportional Counters (MWPC) and Geiger Muller Counter), Scintillation Detectors (Inorganic and Organic Scintillators), Solid States Detectors and Neutron Detectors, Thermo luminescent Dosimetry. **(07 Lectures)**

Radiation safety management: Biological effects of ionizing radiation, Operational limits and basics of radiation hazards evaluation and control: radiation protection standards, International Commission on Radiological Protection (ICRP) principles, justification, optimization, limitation, introduction of safety and risk management of radiation. Nuclear waste and disposal management. Brief idea about Accelerator driven Sub-critical system (ADS) for waste management. **(05 Lectures)**

Application of nuclear techniques: Application in medical science (e.g., MRI, PET, Projection Imaging Gamma Camera, radiation therapy), Archaeology, Art, Crime detection, Mining and oil. *Industrial Uses:* Tracing, Gauging, Material Modification, Sterization, Food preservation. **(05 Lectures)**

Experiments:

1. Study the background radiation levels using Radiation meter

Characteristics of Geiger Muller (GM) Counter:

2) Study of characteristics of GM tube and determination of operating voltage and plateau length using background radiation as source (without commercial source).

3) Study of counting statistics using background radiation using GM counter.

4) Study of radiation in various materials (e.g. K₂SO₄ etc.). Investigation of possible radiation in different routine materials by operating GM at operating voltage.

5) Study of absorption of beta particles in Aluminum using GM counter.

6) Detection of α particles using reference source & determining its half life using spark counter

7) Gamma spectrum of Gas Light mantle (Source of Thorium)

Reference Books:

1. W.E. Burcham and M. Jobes – Nuclear and Particle Physics – Longman (1995)
 2. G.F.Knoll, Radiation detection and measurements
 3. Thermoluminescence Dosimetry, Mcknlly, A.F., Bristol, Adam Hilger (Medical Physics Handbook 5)
 4. W.J. Meredith and J.B. Massey, “Fundamental Physics of Radiology”. John Wright and Sons, UK, 1989.
 5. J.R. Greening, “Fundamentals of Radiation Dosimetry”, Medical Physics Hand Book Series, No.6, Adam Hilger Ltd., Bristol 1981.
 6. Practical Applications of Radioactivity and Nuclear Radiations, G.C. Lowental and P.L. Airey, Cambridge University Press, U.K., 2001
 7. W.R. Hendee, “Medical Radiation Physics”, Year Book – Medical Publishers Inc. London, 1981
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SEMESTER V

Course code: CME69401/ CME69402/ CME69403/ CME69404

Credits: Theory – 02

Theories: 30 Lectures

COURSE OUTCOME:

After completion of the course the student is able to

CO1.Deliberate in details with examples basic of measurement

CO2.Specify in depth electronic voltmeter

CO3.Understand in depth Cathode ray oscilloscope

CO4.Specify the characteristics of Impedance Bridge

CO5.Specify the classification and characteristics of digital multimeter

CO6.Identify the characteristics of signal generator

BASIC INSTRUMENTATION SKILLS: SEC 1C

Basic of Measurement: Instruments accuracy, precision, sensitivity, resolution range etc. Errors in measurements and loading effects. **Multimeter:** Principles of measurement of dc voltage and dc current, ac voltage, ac current and resistance. Specifications of a multimeter and their significance.

(04 Lectures)

Electronic Voltmeter: Advantage over conventional multimeter for voltage measurement with respect to input impedance and sensitivity. Principles of voltage, measurement (block diagram only). Specifications of an electronic Voltmeter/Multimeter and their significance. **AC millivoltmeter:** Type of AC millivoltmeters: Amplifier- rectifier, and rectifier- amplifier. Block diagram ac millivoltmeter, specifications and their significance.

(04 Lectures)

Cathode Ray Oscilloscope: Block diagram of basic CRO. Construction of CRT, Electron gun, electrostatic focusing and acceleration (Explanation only– nonmathematical treatment), brief discussion on screen phosphor, visual persistence & chemical composition. Time base operation, synchronization. Front panel controls. Specifications of a CRO and their significance.

(06 Lectures)

Use of CRO for the measurement of voltage (dc and ac frequency, time period. Special features of dual trace, introduction to digital oscilloscope, probes. Digital storage Oscilloscope: Block diagram and principle of working.

(03 Lectures)

Signal Generators and Analysis Instruments: Block diagram, explanation and specifications of low frequency signal generators. pulse generator, and function generator. Brief idea for testing, specifications. Distortion factor meter, wave analysis.

(04 Lectures)

Impedance Bridges & Q-Meters: Block diagram of bridge. working principles of basic (balancing type) RLC bridge. Specifications of RLC bridge. Block diagram & working principles of a Q-Meter. Digital LCR bridges.

(03 Lectures)

Digital Instruments: Principle and working of digital meters. Comparison of analog & digital instruments. Characteristics of a digital meter. Working principles of digital voltmeter.

(03 Lectures)

Digital Multimeter: Block diagram and working of a digital multimeter. Working principle of time interval, frequency and period measurement using universal counter/frequency counter, time- base stability, accuracy and resolution.

(03 Lectures)

The test of lab skills will be of the following test items:

1. Use of an oscilloscope.
2. CRO as a versatile measuring device.
3. Circuit tracing of Laboratory electronic equipment,
4. Use of Digital multimeter/VTVM for measuring voltages
5. Circuit tracing of Laboratory electronic equipment,
6. Winding a coil / transformer.
7. Study the layout of receiver circuit.
8. Trouble shooting a circuit
9. Balancing of bridges

Laboratory Exercises:

1. To observe the loading effect of a multimeter while measuring voltage across a low resistance and high resistance.
2. To observe the limitations of a multimeter for measuring high frequency voltage and currents.
3. To measure Q of a coil and its dependence on frequency, using a Q- meter.
4. Measurement of voltage, frequency, time period and phase angle using CRO.
5. Measurement of time period, frequency, average period using universal counter/frequency counter.
6. Measurement of rise, fall and delay times using a CRO.
7. Measurement of distortion of a RF signal generator using distortion factor meter.
8. Measurement of R, L and C using a LCR bridge/ universal bridge.

Open Ended Experiments:

1. Using a Dual Trace Oscilloscope
2. Converting the range of a given measuring instrument (voltmeter, ammeter)

Reference Books:

- A text book in Electrical Technology - B L Theraja - S Chand and Co.
- Performance and design of AC machines - M G Say ELBS Edn.
- Digital Circuits and systems, Venugopal, 2011, Tata McGraw Hill.
- Logic circuit design, Shimon P. Vingron, 2012, Springer.
- Digital Electronics, Subrata Ghoshal, 2012, Cengage Learning.
- Electronic Devices and circuits, S. Salivahanan & N. S.Kumar, 3rd Ed., 2012, Tata McGraw Hill
- Electronic circuits: Handbook of design and applications, U.Tietze, Ch.Schenk, 2008, Springer Electronic Devices, 7/e Thomas L. Floyd, 2008, Pearson India

SEMESTER V

Course code: CME70601/ CME70602/ CME70603/ CME70604

Credits: Theory – 02

Theories: 30 Lectures

COURSE OUTCOME:

After the completion of the course, Students will be able to

CO1.Specify the details of sources and detector

CO2.Deliberate the classification and characteristics of experiments on lasers

CO3.Specify the details of experiments on semiconductor sources

CO4.Write down in details with examples Fourier optics

CO5.Specify the classification and characteristics of photonic and holography

CO6.Specify the characteristics of photonic

APPLIED OPTICS: SEC 1I

(i) Sources and Detectors

(09 Lectures)

Lasers, Spontaneous and stimulated emissions, Theory of laser action, Einstein's coefficients, Light amplification, Characterization of laser beam, He-Ne laser, Semiconductor lasers.

Experiments on Lasers:

- a. Determination of the grating radial spacing of the Compact Disc (CD) by reflection using He-Ne or solid state laser.
- b. To find the width of the wire or width of the slit using diffraction pattern Obtained by a He-Ne or solid state laser.
- c. To find the polarization angle of laser light using polarizer and analyzer
- d. Thermal expansion of quartz using laser

Experiments on Semiconductor Sources and Detectors:

- a. V-I characteristics of LED
- b. Study the characteristics of solid state laser

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- c. Study the characteristics of LDR
- d. Photovoltaic Cell
- e. Characteristics of IR sensor

(ii) Fourier Optics

(06 Lectures)

Concept of Spatial frequency filtering, Fourier transforming property of a thin lens

Experiments on Fourier Optics:

a. Fourier optic and image processing

1. Optical image addition/subtraction
2. Optical image differentiation
3. Fourier optical filtering
4. Construction of an optical 4f system

b. Fourier Transform Spectroscopy

Fourier Transform Spectroscopy (FTS) is a powerful method for measuring emission and absorption spectra, with wide application in atmospheric remote sensing, NMR spectrometry and forensic science.

Experiment:

To study the interference pattern from a Michelson interferometer as a function of mirror separation in the interferometer. The resulting interferogram is the Fourier transform of the power spectrum of the source. Analysis of experimental interferograms allows one to determine the transmission characteristics of several interference filters. Computer simulation can also be done.

(iii) Holography

(06 Lectures)

Basic principle and theory: coherence, resolution, Types of holograms, white light reflection hologram, application of holography in microscopy, interferometry, and character recognition

Experiments on Holography and interferometry:

1. Recording and reconstructing holograms
2. Constructing a Michelson interferometer or a Fabry Perot interferometer
3. Measuring the refractive index of air

UG-Physics- CBCS Scheme

4. Constructing a Sagnac interferometer
5. Constructing a Mach-Zehnder interferometer
6. White light Hologram

(iv) Photonics: Fibre Optics

(09 Lectures)

Optical fibers and their properties, Principal of light propagation through a fibre, The numerical aperture, Attenuation in optical fibre and attenuation limit, Single mode and multimode fibres, Fibre optic sensors: Fibre Bragg Grating

Experiments on Photonics: Fibre Optics

- a. To measure the numerical aperture of an optical fibre
- b. To study the variation of the bending loss in a multimode fibre
- c. To determine the mode field diameter (MFD) of fundamental mode in a Single-mode fibre by measurements of its far field Gaussian pattern
- d. To measure the near field intensity profile of a fibre and study its refractive Index profile
- e. To determine the power loss at a splice between two multimode fibre

Reference Books:

- Fundamental of optics, F. A. Jenkins & H. E. White, 1981, Tata McGraw hill.
- LASERS: Fundamentals & applications, K.Thyagrajan & A.K.Ghatak, 2010, Tata McGraw Hill
- Fibre optics through experiments, M.R. Shenoy, S.K. Khijwania, et.al. 2009, Viva Books
- Nonlinear Optics, Robert W. Boyd, (Chapter-I), 2008, Elsevier.
- Optics, Karl Dieter Moller, Learning by computing with model examples, 2007, Springer.
- Optical Systems and Processes, Joseph Shamir, 2009, PHI Learning Pvt. Ltd.
- Optoelectronic Devices and Systems, S.C. Gupta, 2005, PHI Learning Pvt. Ltd.
- Optical Physics, A.Lipson, S.G.Lipson, H.Lipson, 4th Edn., 1996, Cambridge Univ. Press

The marks distribution for the final practical examination is as follows:

Formula/Formulae with explanation of symbols	05 marks
Diagram/Circuit diagram and tabular column	08 marks
Experimental setup+ taking readings	18 marks
Accuracy of readings	05 marks
Graph and Calculations	10 marks
Final result and units	04 marks
Viva	10 marks
Record	10 marks
Total for the practical examination	70 marks

Question Paper Pattern for DSC and DSE theory papers of 04 credits

From the academic year 2017-2018 onwards

Time: 3 hours

Max. Marks: 70

Part A:

- I. Answer all the questions (Multiple choice questions) 1 x 4 = 4 marks**
(One question should be selected from 15 lectures)

Part B:

- II. Answer any seven questions. 2 x 7 =14 marks**

Ten questions should be set

(At least two questions should be selected from 15 lectures)

Part C:

- III. Answer all the questions. 4 x 10 = 40 marks**
(Two questions should be set out of fifteen lectures of internal choice of ten marks each)

Part D:

- IV. Answer any three out of four questions 4 x 3 = 12 marks**
(One problem should be selected from 15 lecturers)

Question Paper Pattern for DSE theory papers of 05 credits

From the academic year 2017-2018 onwards

Time: 3 hours

Max. Marks: 70

Part A:

I. Answer all the questions (Multiple choice questions) 1 x 5 = 5 marks

(One question should be selected from 15 lectures)

Part B:

II. Answer any five questions. 2 x 5 =10 marks

Eight questions should be set

(At least two questions should be selected from 15 lectures)

Part C:

III. Answer the following questions. 8 x 5 = 40 marks

(Two questions should be set out of fifteen lectures of internal choice of eight marks each)

Part D:

IV. Answer any three out of five questions 5 x 3 = 15 marks

(One problem should be selected from 15 lecturers)

Question Paper Pattern for SEC theory paper of 02 credits

Time: 2 hours

Max. Marks: 50

Part A:

I. Ten Questions out of twelve. 2 x 10 = 20 marks

(Five questions from 15 lectures.)

Part B:

II. Answer all the questions 10 x 03 = 30 marks

(Two questions should be set out of ten lectures of internal choice of ten marks each)



**JSS COLLEGE OF ARTS, COMMERCE AND
SCIENCE**

(Autonomous)

B N ROAD, MYSURU- 570 025

DEPARTMENT OF SANSKRIT

Syllabus

CHOICE BASED CREDIT SYSTEM

For B.A/B.Sc PROGRAMME

2017-18

Scheme of study and Assessment pattern for BSc

		Course	Title	Hours / Week		Credits	Maximum Marks			Exam Duration	Total
				L	T/ P		L:T:P	IA			
								C1	C2		Marks
I Year	I SEM	CMA09001 CMA 09002 CMA 09003 CMA 09004 CMA 09005 CMA 09006 CMA 09007 CMA 09008	Selection of anthology Sanskrit prose	3	1	3:1:0	15	15	70	3 Hours	100
	II SEM	CMB09001 CMB 09002 CMB 09003 CMB 09004 CMB 09005 CMB 09006 CMB 09007 CMB 09008	Sanskrit poetry	3	1	3:1:0	15	15	70	3 Hours	100
II Year	III SEM	CMC09001 CMC 09002 CMC 09003 CMC 09004 CMC 09005 CMC 09006 CMC 09007 CMC 09008	Sanskrit drama l. Abhijnana shakunthale of kalidasa	3	1	3:1:0	15	15	70	3 Hours	100
	IV SEM	CMD09001 CMD09002 CMD09003 CMD09004 CMD09005 CMD09006 CMD09007 CMD09008	Champu kavyam Champu ramayane Ayodhyakanda	3	1	3:1:0	15	15	70	3 Hours	100

Programme: B.Sc (PCM/PMCs/PMCm/PME/CZBt/BMBt/BBM/CBZ)

Scheme of study and Assessment pattern for B.A

Programme: BA (EG/HP/KG/HE/JE)

		Course	Title	Hours / Week		Credits	Maximum Marks			Exam Duration	Total
				L	T/P		IA		Exam		
						C1	C2	Marks			
I Year	I SEM	DLA09001 DLA09002 DLA09003 DLA09004 DLA09005	Selection of anthology Sanskrit prose	3	1	3:1:0	15	15	70	3 Hours	100
	II SEM	DLB09001 DLB09002 DLB09003 DLB09004 DLB09005	Sanskrit poetry	3	1	3:1:0	15	15	70	3 Hours	100
II Year	III SEM	DLC09001 DLC09002 DLC09003 DLC09004 DLC09005	Sanskrit drama I. Abhijnana shakunthale of kalidasa	3	1	3:1:0	15	15	70	3 Hours	100
		DLC310	Bhasha samvardhane	2	0	2:0:0	15	15	70	3 Hours	100
	IV SEM	DLD09001 DLD09002 DLD09003 DLD09004 DLD09005	Champu kavyam Champu ramayane Ayodhyakanda	3	1	3:1:0	15	15	70	3 Hours	100

Programme Outcome for BA & B.Sc

PO1. To realize the human values.

PO2. To create the sense of social service.

PO3. Responsible and dutiful citizen.

PO4. To enhance the critical temper and creative ability.

Programme Specific Outcome for BA & B.Sc

PSO1. To develop the speaking skill.

PSO2. Clarity in Language.

PSO3. Usage of appropriate words.

BA/BSC

I SEMESTER

Selection of anthology Sanskrit prose

Credit (L: T: P = 3: 1: 0)

Course Outcome

- CO1. Identify the characteristics of tathavamasi
- CO2. Deliberate the classification and characteristics of tathavamasi
- CO3. Deliberate in details with examples chandrapedavidyabyasa
- CO4. Learn the classification and characteristics of chandrapeedavidyabyasa
- CO5. Deliberate in details with application, if applicable, mahatmabudha
- CO6. Specify in depth varthamanayugsyamahapuroshomahathmagandhigi

- 1. Tathvamsi
- 2. Chandrapeeda vidhyabhayyasa
- 3. Vyapadeshnamahatam
- 4. Mahatma Buddha
- 5. Varthamana yugasya mahapurusho mahatma Gandhi

GRAMMAR

- 1. Sandhi prakarana – Swara sandhi
- 2. Shabdha – Ajanta pullinga, Stree linga, Napumsaka linga
- 3. Samasaha – Tathpurusha, Karmadhareyaha
- 4. Karakani
- 5. Bhashantaram – Unseen Passage

Reference Books

- 1 Banas Kadambari – Editor by M.R. Kale
- 2 Raghavamsha Translation by Vidvan H.V. Nagaraj Rao
- 3 Samskrita Gadya Padya Sangrah by Dr. M. shivkumarswamy, Bangalore
- 4 Kavya Taranga 18 (Anthology of Sanskrit Prose) Published by Prasaranga University of Mysore

BA/BSC

II SEMESTER

Sanskrit poetry

Credit (L: T: P = 3: 1: 0)

Course Outcome

- CO1. Deliberate the classification and characteristics of selected poems of raguvamsha of kalidasa
- CO2. Write down in details with examples selected poems of raguvamsha of kalidasa
- CO3. Deliberate the characteristics of selected poems of raguvamsha of kalidasa
- CO4. Deliberate in depth selected poems of raguvamsha of kalidasa
- CO5. Write down the classification and characteristics of selected slokas of sidaratha manparivarthanm
- CO6. Specify the classification and characteristics of selected slokas of sidaratha manparivarthanm
- CO7. Specify the characteristics of selected slokas of sidaratha manparivarthanm
- CO8. Understand in details with application, if applicable, selected vachanas of basaveshwara
- CO9. Learn in depth selected vachanas of basaveshwara
- CO10. Specify the classification and characteristics of selected sloks of katopanisht
- CO11. Understand the classification and characteristics of selected sloks of katopanisht
- CO12. Understand the characteristics of selected sloks of katopanisht
- CO13. Understand in details with application, if applicable, selected vachanas of basaveshwara
- CO14. Deliberate the classification and characteristics of selected vachanas of basaveshwara

1. Raguvamsha of kalidasa – Ist Canto 1-30 sloakas
2. Siddarthamanaparivarthanam – Ashwaghosha – IIIrd Canto 1-20 sloakas
3. Basaveshwara vachanani – 1 – 15 Vachanas
4. Yamanachiketayoh – Samvada in katopanishath – Prathama valli- 28 sloakas

GRAMMAR

1. Sandhi prakarana – Vyanjana sandhi
2. Shabdaha – Sarvanama
3. Samasaha – Bahurvrihi, Avyaebhava samasa
4. Kriyapadani – Lat, Lung, Lrut, Lot, Vidhiling(Parsmipadi)
5. Krudannta avyyanee – Tavanta, Lebannta, Tumunannta
6. Translation – Unseen passage

Reference Books

- 1 Samskruta Kavya by K. Krishnamurthy, Mysore University, Mysore
- 2 Raghu vamsha Dr Nagarajarao
- 3 Samskrutha kavyatharanga 15 Prasaranga Mysuru

BA/B.Sc

III SEMESTER

Sanskrit drama1.Abhijnana shakunthale of kalidasa

Credit (L: T: P = 3: 1: 0)

Course Outcome

- CO1 Write down the characteristics of Abignanna shakuntalm of kalidasa
1-4acts
- CO2 Identify the details of Abignanna shakuntalm of kalidasa 1-4acts

Part - 1

Drama –Abhijnana shakuntalam of Kalidasa 1 – 4 Acts

Date, Place and works of Mahakavi Kalidasa

Part – II

Grammar

1. Halantha shabdah – Rajan, Athma, Maruth, Bhavath
2. Kriya padani - Lat, Lung, Lrut, Lot, Vidhiling(Atamanepadi)
3. Samasaha – Davndva, Divigu
4. Alankaraha – Upama, Swabhvokthi, Artantaranyyasa, Drustantha
5. Vakayya doshannpariharata
6. Translation – Seen passage

Reference Books

- 1 Abijnanna Sankunthala Edited by M.R. Kale
- 2 Sanskrita Drama by A.B. kith – published by Motilal Banarasidass, New Delhi.
- 3 Abijanna sakunthala Prof.K.T. Pandurangi
- 4 Samskritha Nataka Prof.A.R. Krishna Shastri
- 5 Kannada Abijnanna Sakunthala by Ganapathi Moleyara

DLC310

III SEMESTER

AECC(Only for III Sem B.com)

Bhasha samvardhana

Credit (L: T: P = 2:0: 0)

Course OutCome

CO1 Learn in details with examples padagalu

CO2 Deliberate the details of Vakayarachane

CO3 Understand in details with application, if applicable, Varadirachane

CO4 Deliberate the details of Sankshepakaranam

CO5 Deliberate the classification and characteristics of Sambhashanakaushalam

CO6 Learn in details with examples Bhashantharam

CO7 Write down in depth Patralekanam prabhedani

CO8 Identify the characteristics of Prathamika vyakarnam

CO9 Identify the characteristics of Geethagayanam

1. Lekhana kowshala
2. Bhasha prayoga, Shudha-Ashudha vyyathyasa
3. Vakhyya rachana
4. Varadhi rachana
5. Samkshepa karanam
6. Bhashantaram
7. Pathralekhanam, Types of letter
8. Viramatham, Avedana pathram
9. Adesha pathram, Anugata pathram, Sambhashana kowshalam, Geeta gayanam

Reference

1. Samskrutha kavya –Dr. K Krishna Murthy
2. Samskrutha vanijayam- Prasaranga Mysuru University

BA/BSc

IV SEMESTER

Champu kavayam Champu ramayane Ayodhyakanda

Credit (L: T: P = 3: 1: 0)

Course Outcome

CO1 Specify the characteristics of Ayodyakanada of champu ramayanam of bhoja

CO2 Write down the classification and characteristics of Ayodyakanada of champu ramayanam of bhoja

CO3 Write down in depth Ayodyakanada of champu ramayanam of bhoja

CO4 Understand in depth Ayodyakanada of champu ramayanam of bhoja

PART - 1

1. Champukavyam – Ayodhyakanda of champu Ramayana
2. Kavikaya parichyaha

PART - 2

GRAMMAR

1. Alankaraha – Roopaka, Anannvyaha, Slesha, Uthpreksha
2. Vakyya rachana –Sentence formation
3. Channdassu – Indravajra, Upendravajra, Anustup, Vasanta tilaka
4. Prayoga parivarthanam
5. Bhashantaram – Seen/ Un seen passage

Reference Books

- 1 Champu Ramayanam Edited by pandith S. Rangachar Sanskrita Sahithya Sadhana, mysore-4
- 2 Samskrutha Vanijyam Edited by Dr. G. Marulasiddaiah published by Prasaranga University of Mysore

- 3 Chandraloka Vrutharanthnakara Translation by
Prof.B.Rajashekaraiah, Published by Prof.M.Mallanna Mallikachala
Prakashana, Mysore
- 4 Arthashastra of Kautilya

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE (Autonomous),

OOTY ROAD, Mysore

SANSKRIT (CBCS) 2017-18

B.Sc (PCM/PMCs/PMCm/PME/CZBt/BMBt/BBM/CBZ)

Pattern of question paper

1. Essay type questions	1×8= 08
2. Essay type question 1 Out of 2	1×8= 08
3. Short notes 2 out of 4	2×4= 08
4. Translate and explain- 2 out of 4	2×5= 10
5. Annotations -3 out of 5	3×3= 09
6. Sanskrit questions -2 out of 4	2×4= 08
7. Sanskrit questions -5 out of 7	1×5= 05
8. Grammar - 7	12
9. Translation _ Seen passage	07

—

Total = 70

C1 – 15 Marks

C2 – 15 Marks

C3 - 70 Marks

Total =100 marks

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE (Autonomous),

OOTY ROAD, Mysore

SANSKRIT (CBCS) 2017-18

B.A (EG/HP/KG/HE/JE)

Pattern of question paper

1. Essay type questions	1×8= 08
2. Essay type question 1 Out of 2	1×8= 08
3. Short notes 2 out of 4	2×4= 08
4. Translate and explain- 2 out of 4	2×5= 10
5. Annotations -3 out of 5	3×3= 09
6. Sanskrit questions -2 out of 4	2×4= 08
7. Sanskrit questions -5 out of 7	1×5= 05
8. Grammar - 7	12
9. Translation _ Seen passage	07

— —

Total = 70

C1 – 15 Marks

C2 – 15 Marks

C3 - 70 Marks

Total =100 marks

SANSKRIT (CBCS) 2017-18
AECC-B.A (EG/HP/KG/HE/JE) III SEM

Pattern of question paper

2hrs

1.one out of two questions	1X10=10
2.two out of four	2X5=10
3.two out of three	2X5=10
4.translate kannada to Sanskrit	1X5=5
5.One third of its length suggest suitable title	1X10=10
6 correct the sentence	1X5=5
Total	50

C1 – 10 Marks

C2 – 10 Marks

C3 - 50 Marks

Total =70 marks

		Course	Title	Hours / Week		Credits	Maximum Marks			Exam Duration	Total
				L	T/ P		L:T:P	IA			
							C1	C2			
I Year	I SEM	DNA09001 BDA09001 DCA09001	Selection of anthology Sanskrit prose	3	1	3:1:0	15	15	70	3 Hours	100
	II SEM	DNB09001 BDB09001 DCB09001	Sanskrit poetry	3	1	3:1:0	15	15	70	3 Hours	100
II Year	III	DNC09001 BDC09001 DCC09001	Sanskrit drama1. Swapnavasavadatham	3	1	3:1:0	15	15	70	3 Hours	100
	BCOM	DNC310	Bhasha Samvardhana	2	0	2:0:0	15	15	70	3 Hours	100
	IV SEM	DND09001 BDD09001 DCD09001	Samskrutha vaanijyam	3	1	3:1:0	15	15	70	3 Hours	100

Scheme of study and Assessment pattern for BCOM/BBA/BCA

Programme Outcome for B.COM/BBA/BCA

PO1. inherit our ancient culture ,social ,political and economical knowledge

PO2. to understand trade and commerce as reflected in koutilyas Arthashastra and shukraneethi

PO3. write letter ,writing essays and regional language

PO4. understand moral values

PO5. understand life in a better way

PO6. have a good relation get a good job

PO7. serve society

Program Specific Outcome for B.COM/BBA/BCA

PSO1:Learn concentrate on studies by chanting Vedic shlokas and manthras

PSO2:Understand Vedic economics and policies

PSO3:Sanskrit is best suitable computer and is the mother of all languages

B.COM/BBA/BCA

I SEMESTER

Selection of anthology Sanskrit prose

Credit (L: T: P = 3: 1: 0)

Course Outcome

- CO1 Specify in details with application, if applicable, Thatvamasi
- CO2 Specify in details with application, if applicable, Learn in depth tathavamasi
- CO3 Deliberate in details with application, if applicable, Deliberate in details with examples tathavamasi
- CO4 Specify in details with application, if applicable, Understand in details with application, if applicable, tathavamasi
- CO5 Identify the classification and characteristics of Deliberate in details with examples chandrapedavidyabyasa
- CO6 Identify in details with examples Specify in depth chandrapedavidyabyasa
- CO7 Understand in details with examples Deliberate in depth chandrapedavidyabyasa
- CO8 Identify the details of Deliberate the details of vyapadwshenamahatha
- CO9 Specify the classification and characteristics of Deliberate the details of vyapadwshenamahath
- CO10 Learn the characteristics of Specify the classification and characteristics of mahatmabudha
- CO11 Understand the details of Deliberate in details with application, if applicable, mahatmabudha

- CO12 Identify in details with examples Identify in details with examples
varthamanayugsyamahapuroshomahathmagandhigi
- CO13 Identify in details with examples Deliberate the characteristics of
grammer(sandi prakaranna-swara sandhi)
- CO14 Specify in depth Write down the details of grammer(sandi prakaranna-swara
sandhi)
- CO15 Identify in details with examples Understand in details with examples
shabdasthrishulingesh

1. Rajaarshi vrutham in Kowtilya's Arthashastram
2. Tyriyopanishath – Shikshavalli
3. Sukanassupadeshasaraha in Bana's Kadambari
4. Kashi teerthayatravarmanam – In anthology of Prose
5. Digarthaha kasttasamshreyaha in Panchatantram

GRAMMER

1. Sandhi prakarana – Swara sandhi (Achh sandi)
2. Shabdhaha –Ajanta pullinga, Stree linga, Napumsaka linga
3. Samasaha – Tathpurusha, Karmadhareyaha
4. Krudanta avyyahani – Tavanta, Tumulnanta, Lybantha
5. Bhashantaram – Unseen Passage

Reference Books

- 3 Banas Kadambari – Editor by M.R. Kale
- 4 Raghavamsha Translation by Vidvan H.V. Nagaraj Rao
- 3 Samskrita Gadya Padya Sangrah by Dr. M. shivkumarswamy, Bangalore
- 4 Kavya Taranga 18 (Anthology of Sanskrit Prose) Published by Prasaranga
University of Mysore

B.COM/BBA/BCA

II SEMESTER

Sanskrit poetry

Credit (L: T: P = 3: 1: 0)

Course Outcome

- CO1 Understand the details of rajarashivrutham in koutilya's arthashatram
- CO2 Specify the characteristics of rajarashivrutham in koutilya's arthashatram
- CO3 Identify the classification and characteristics of rajarashivrutham in koutilya's arthashatram
- CO4 Learn the classification and characteristics of Iya's arthashatram
- CO5 Specify in depth Shishkhavalli in Thithriya upnishathu
- CO6 Learn the details of Shukunasopadesha in Banas kadmabri
- CO7 Understand in depth Shukunasopadesha in Banas kadmabri
- CO8 Understand the details of khashi theerthta yathra varnanum
- CO9 Identify the details of khashi theerthta yathra varnanum
- CO10 Learn the classification and characteristics of Degrthaha katha samshraah in panchathanthram
- CO13 Learn the details of selected sloks of Basvandhalahri of Gowrisankarawamy
- CO15 Learn the classification and characteristics of selected vachana of Sarvanjana
- CO14 Write down the characteristics of Selected slokas of Meghadutha by Kalidasa
- CO12 Write down the details of Selected slokas of Meghadutha by Kalidasa
- CO16 Deliberate the classification and characteristics of selected vachana of Sarvanjana
- CO15 Write down in depth selected sloks of Basvandhalahri of Gowrisankarawamy
- CO17 Deliberate the characteristics of selected vachana of Shubshithani
- CO18 Specify the characteristics of selected vachana of Shubshithani

PART – I

1. Meghaduta of kalidasa – poorva megha **1-30 sloakas**
2. Basavanandalahari of his holiness Sri Gowrishankara swamiji –**1-15 sloakas**
3. Sarvagnya vachanani – **1 – 15 stanzas selected from Purushothama Sarvagnya vachanani**
4. Subhashitani - **1 – 10 stanzas (Anthology of Poetry)**

GRAMMER

1. Vyanjana sandhi – Schuthva, Stuthva, Jasthva, Anunasika
2. Shabdhaha –Rajan, Maruth, Athman, Bhavath
3. Kriyapadani - Lat, Lung, Lrut, Lot, Vidhiling(Parsmipadi)
4. Prayoga parivartanam
5. Samasaha – Avyayi bhava, Bahurvihi
6. Translation – Unseen passage

Reference Books

- 1 Samskruta Kavya by K. Krishnamurthy, Mysore University, Mysore

B.COM/BBA/BCA

III SEMESTER

Sanskrit drama1. Swapnavasavadatham

Credit (L: T: P = 3: 1: 0)

Course Outcome

- CO1 Write down in details with examples swapnavasavdattam of bhasa
- CO2 Learn in details with examples swapnavasavdattam of bhasa
- CO3 Specify in details with examples swapnavasavdattam of bhasa
- CO4 Write down in details with examples swapnavasavdattam of bhasa

Part - 1

Drama –Swapnavasavadattam of Mahakavi Bhasa

Date, Place and works of Mahakavi Bhasa

Part – II

Grammar

1. Sarvanama shabdah – Tadh, Atadh, Yath, Kim, Trishulingeshu, Asmadh, Yushmadh
2. Kriya padani - Lat, Lung, Lrut, Lot, Vidhiling(Atamanepadi)
3. Samasaha – Davndva, Divigu
4. Alankaraha – Upama, Swabhvokthi, Artantaranyyasa, Drustantha

Reference Books

1. Swapnavasavadhatham
2. Sanskrita Drama by A.B. Kith – published by Motilal Banarasidass, New Delhi.
3. Abijanna sakunthala Prof.K.T. Pandurangi
4. Samskritha Nataka Prof.A.R. Krishna Shastri
5. Kannada Abijnanna Sakunthala by Ganapathi Moleyara

DNC310

III SEMESTER

AECC(Only for III Sem B.com)

Bhasha samvardhana

Credit (L: T: P = 2:0: 0)

Course OutCome

CO1 Learn in details with examples padagalu

CO2 Deliberate the details of Vakayarachane

CO3 Understand in details with application, if applicable, Varadirachane

CO4 Deliberate the details of Sankshepakaranam

CO5 Deliberate the classification and characteristics of Sambhashanakaushalam

CO6 Learn in details with examples Bhashantharam

CO7 Write down in depth Patralekanam prabhedani

CO8 Identify the characteristics of Prathamika vyakarnam

CO9 Identify the characteristics of Geethagayanam

10.Lekhana kowshala

11.Bhasha prayoga, Shudha-Ashudha vyyathyyasa

12.Vakhyya rachana

13.Varadhi rachana

- 14.Samkshepa karanam
- 15.Bhashantaram
- 16.Pathralekhanam, Types of letter
- 17.Viramatham, Avedana pathram
18. Adesha pathram, Anugata pathram, Sambhashana kowshalam, Geeta gayanam

Reference

3. Samskrutha kavya –Dr. K Krishna Murthy
4. Samskrutha vanijayam- Prasaranga Mysuru University

B.COM/BBA/BCA

IV SEMESTER

Samskrutha vaanijyam

Credit (L: T: P = 3: 1: 0)

Course Outcome

- CO1 Understand the characteristics of bussiness sanskrit
- CO2 Learn the details of bussiness Sanskrit
- CO3 Learn in depth bussiness Sanskrit
- CO4 Specify in depth bussiness Sanskrit
- CO5 Learn in details with application, if applicable, arthashatra of kutilya
- CO6 Write down the characteristics of arthashatra of kutilya
- CO7 Identify the characteristics of arthashatra of kutilya
- CO8 Deliberate the details of arthashatra of kutilya
- CO9 Specify in details with examples arthashatra of kutilya vinayadikarna
- CO10 Understand the details of arthashatra of kutilya vinayadikarna
- CO11 Identify in depth arthashatra of kutilya vinayadikarna
- CO12 Specify the classification and characteristics of arthashatra of kutilya vinayadikarna

PART – 1

1. Strayisthapana, Vartha dandaneeti, Indriya jaya in vinayadhikaranam – Arthashastra of Kowtilya
2. Samskruta vanijayam – Importance of Commerical education
3. Patralekhanam, Personal, official, Public and Business letters
4. Prabhanda lekhanam in Sanskrit
5. Samshkeshepa karanam
6. General essays in regional language
7. Translation from Kannada to Sanskrit

Reference

1. Samskrutha kavya –Dr. K Krishna Murthy
2. Samskrutha vanijayam- Prasaranga Mysuru University
3. Prabhanda pallava Dr s Jaganath
4. Koutilya’s arthashastram shyama shatri ORI Mysur

**SANSKRIT (CBCS) 2017-18
B.COM/BBA/BCA(I SEM - III SEM)**

Pattern of question paper

1. Sanskrit one marks questions	1×10=10
2. Essay type questions – 2 out of 4	2×7= 14
3. Short notes– 2 out of 4	2×5= 10
4. Annotations -3 out of 5	3×3= 09
5. Sanskrit questions -2 out of 4	2×5= 10
6. Grammar	12
7. Seen passage	05
Total	= 70

C1 – 15 Marks

C2 – 15 Marks

C3 - 70 Marks

Total =100 marks

SANSKRIT (CBCS) 2017-18
B.COM/BBA/BCA(IV SEM)

Pattern of question paper

- | | |
|--|----------|
| 1. One Essay type question out of 2 in vinayadhikaranam in Kowtilya of Arthashastram | 1×10=10 |
| 2. Short notes 2 out of 4 | 2×5= 10 |
| 3. Drafting of personal letter | 1×8= 08 |
| 4. Applications for suitable job | 1×8= 08 |
| 5. Sanskrit Essays - | 1×10=10 |
| 6. Essays in Regional language | 1×8= 08 |
| 7. Essay writing | 1×10= 10 |
| 8. Translation _ Seen passage | 06 |

Total = 70

C1 – 15 Marks

C2 – 15 Marks

C3 - 70 Marks

Total =100 marks

SANSKRIT (CBCS) 2017-18
AECC-B.COM III SEM

Pattern of question paper

2hrs

1.one out of two questions	1X10=10
2.two out of four	2X5=10
3.two out of three	2X5=10
4.translate kannada to Sanskrit	1X5=5
5.One third of its length suggest suitable title	1X10=10
6 correct the sentence	1X5=5
Total	50

C1 – 10 Marks

C2 – 10 Marks

C3 - 50 Marks

Total =70 marks



JSS COLLEGE OF ARTS, COMMERCE & SCIENCE

(An Autonomous College of University of Mysore)

Re-accredited by NAAC with 'A' grade

OOTY ROAD, MYSORE-570 025, KARNATAKA

SYLLABUS

Programme: B. Voc. (Software Development)

Scheme of Assessment:

Semester-I:

Sl. no	Course Code	Course Name	Credits		Marks					Hours Per Week		
			L:T:P	Total	Th	C1	C2	Pr	Total	T	P	Total
1	SDA 020	Communication Language Kannada	3:0:0	03	70	15	15	-	100	03	-	03
2	SDA 510	Basic Mathematics	3:0:0	03	70	15	15	-	100	03	-	03
3	SDA 520	C Programming	2:0:1	03	70	15	15	70	170	02	01	03
4	SDA 530	Digital Electronics	2:0:1	03	70	15	15	70	170	02	01	03

Semester-II:

Sl. no	Course Code	Course Name	Credits		Marks					Hours Per Week		
			L:T:P	Total	Th	C1	C2	Pr	Total	T	P	Total
1	SDB 540	Communication Language English	3:0:0	03	70	15	15	-	100	03	-	03
2	SDB 510	Discrete Mathematics	3:0:0	03	70	15	15	-	100	03	-	03
3	SDB 520	Algorithms & Data Structure	2:0:1	03	70	15	15	70	170	02	01	03
4	SDB 550	Microcontroller & Embedded System	2:0:1	03	70	15	15	70	170	02	01	03

Semester-III:

Sl. no	Course Code	Course Name	Credits		Marks					Hours Per Week		
			L:T:P	Total	Th	C1	C2	P	Total	T	P	Total
1	SDC 510	Advanced Calculus	4:0:0	04	70	15	15	-	100	04	-	04
2	SDC 520	Differential Equation	2:0:0	02	70	15	15	-	100	02	-	02
3	SDC 530	Software Architecture and SDLC & Process	2:0:1	03	70	15	15	70	170	02	01	03
4	SDC 550	Indian Constitution	3:0:0	03	70	15	15	-	100	03	-	03

Semester-IV:

Sl. no	Course Code	Course Name	Credits		Marks					Hours Per Week		
			L:T:P	Total	Th	C1	C2	P	Total	T	P	Total
1	SDD 510	Numerical Analysis & Statistics	4:0:0	04	70	15	15	-	100	04	-	04
2	SDD 520	Operation Research	2:0:0	02	70	15	15	-	100	02	-	02
3	SDD 530	Software Modelling & QA	2:0:1	03	70	15	15	70	170	02	01	03
4	SDD560	Environmental Science	3:0:0	03	70	15	15	-	100	03	-	03

Semester-V:

Sl. no	Course Code	Course Name	Credits		Marks					Hours Per Week		
			L:T:P	Total	Th	C1	C2	P	Total	T	P	Total
1	SDE 510	Project Management	2:0:2	04	70	15	15	70	170	02	02	04
2	SDE 520	Configuration Management	3:0:1	04	70	15	15	70	170	03	01	04
3	SDE 530	Human Computer Interaction	3:0:1	04	70	15	15	70	170	03	01	04

Semester-VI:

Sl. no	Course Code	Course Name	Credits		Marks					Hours Per Week		
			L:T:P	Total	Th	C1	C2	P	Total	T	P	Total
1	SDF 510	Operating System	3:0:1	04	70	15	15	70	170	03	01	04
2	SDF 550	Database Design	3:0:1	04	70	15	15	70	170	03	01	04
3	SDF 520	Computer Networks & Security Fundamentals	3:0:1	04	70	15	15	70	170	03	01	04

Program Outcome:

After completing the graduation in Software Development, the students are able to:

- PO1. Contribute to the design of software products and applications
- PO2. Develop software code to specification
- PO3. Provide data/information in standard formats
- PO4. Develop their knowledge, skills and competence
- PO5. Create documents for Knowledge Sharing
- PO6. Develop media content and graphic designs for software products and applications
- PO7. Appreciate the engineering nature of software development
- PO8. Describe key activities in software development and the role of modelling
- PO9. Explain key concepts in software development such as risk and quality
- PO10. Explain the basics of an object-oriented approach to software development

Program Specific Outcome:

After completing the graduation in Software Development, the students are able to:

- PSO1. Apply standard Software Engineering practices and strategies in real-time software project development
- PSO2. Acquaint with the contemporary trends in industrial/research settings and thereby innovate novel solutions to existing problems
- PSO3. Understand, analyze and develop computer programs of varying complexity
- PSO4. Apply standard Software Engineering practices and strategies in software project development
- PSO5. Acquaint with the contemporary issues, latest trends in technological development
- PSO6. Develop software code to specification
- PSO7. Manage their work to meet requirements
- PSO8. Develop their skills and competence

General Component Syllabus

SEMESTER I

Communication Language Kannada

Credits: 3 (45 hours)

(3 Hours of Theory per week)

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Course Outcome:

After completion of the course, the students are able to:

- CO1. Deliberate in depth Putta Vidhava
- CO2. Learn the details of Rangoli
- CO3. Specify in depth Mumbai Jathaka
- CO4. Learn in details with examples Mahiti Tantradnyan
- CO5. Understand in details with examples Mudanambikegala Bedinali

1. ಪಾಠ

12 ಉಪಲೇಖ

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|--------------------------------------|----------------|
| 1. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ | - ಪಾಠಗಳ |
| 2. ಏಕಾಕ್ಷರ ಮತ್ತು ದ್ವಿಾಕ್ಷರ | - " ಏಕಾಕ್ಷರ |
| 3. ಗುಣಗಳ | - ಏಕಾಕ್ಷರ |
| 4. ಅಕ್ಷರಗಳ ಸಂಯೋಗ | - f J, j, i, j |
| 5. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ | - ಅಕ್ಷರಗಳ |
| 6. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ | - 1 z ಪಾಠಗಳ |

2. ಉಪಲೇಖ

12 ಉಪಲೇಖ

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|--------------------------------------|---------------------------|
| 1. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ | - ಅಕ್ಷರಗಳ ಮತ್ತು ಅಕ್ಷರಗಳ |
| 2. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ | - f J, j, i. |
| 3. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ | ಅಕ್ಷರಗಳ ಮತ್ತು ಅಕ್ಷರಗಳ |
| 4. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ | - " ಅಕ್ಷರಗಳ ಮತ್ತು ಅಕ್ಷರಗಳ |
| | - J J, j, i |
| | ಅಕ್ಷರಗಳ ಮತ್ತು ಅಕ್ಷರಗಳ |

ಅಕ್ಷರಗಳ ಸಂಯೋಗ

3. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ

21 ಉಪಲೇಖ

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| 1. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ |
| 2. ಅಕ್ಷರಗಳ ಸಂಯೋಗ ಮತ್ತು ಅಕ್ಷರಗಳ ಸಂಯೋಗ |

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SDA 510

BASIC MATHEMATICS

Credits: 3 (45 Hours)

(3 Hours of Theory per week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Learn in depth Polynomials
- CO2. Write down the details of Cardon's Method
- CO3. Identify the details of Pair of straight lines
- CO4. Deliberate in details with examples Circle
- CO5. Specify in details with examples Radian Measure
- CO6. Learn in depth Complex Numbers

Unit 1: Algebra1

15 Hours

- Theory of Equations: Polynomials – Relations between the roots and coefficients – Symmetric functions –Synthetic division-Descartes' rule of signs –Cubic equations- Cardon's method.

Unit 2: Analytical Geometry –I

15 Hours

- Two dimensional coordinate geometry – straight line (Revision), Pair of straight lines – standard results and simple problems.
- Circle: Equations of circles, Tangent and normal, radical axis and radical centers.
- Conic: Parabola – Ellipse- Hyperbola (Equations in standard form and problems)

Unit 3: Trigonometry

15 Hours

- Radian measure-Trigonometric ratios –Trigonometric functions of compound angle, multiple angles and half angles-Inverse trigonometric functions-complex numbers.

Reference:

1. Algebra –Natarajan
2. Algebra – Hardy and Wright
3. Algebra –Shanthi Narayan
4. Algebra –Manicavachagam Pillay
5. Elements of Analytical Solid geometry – Shanti Narayan
6. Elements of Analytical Solid geometry –S.L.Loney
7. Differential Calculus –Shanthi Narayan
8. Trigonometry –S.L.Loney

C PROGRAMMING

Credits: 3 (45 Hours)

(2 Hours of Theory and 1 Hour of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Specify in details with examples Problem Design
- CO2. Learn in details with examples Algorithms
- CO3. Deliberate the details of Data Definition Structure
- CO4. Learn the details of Control Structures
- CO5. Learn the details of Functions
- CO6. Identify in details with examples Abstract Data Types

Unit 1

15 Hours

Problem Solving Technique: Problem definition, Problem analysis, Problem Design, Algorithms, Flow charts, Coding, Debugging, Program documentation, Program maintenance and Basic programming construct

Data Definition Structure: Types, constants, variables, keywords and identifiers.

Operators and Expressions: Arithmetic, Relational, Logical, operator precedence rules; input and output statement and Assignment statement

Unit 2

15 Hours

Control Structures: Sequential, Selection (one way, two way), looping (while, do while, for), combinations

Functions: Definition and passing (function depth look), Prototypes: parameter definition and passing (scope: local and global variables)

Data Structures: One and Two dimensional arrays

Abstract data types: Records (Structure definition statement); Strings: Use of main operations, string functions (concatenates string copy and compare etc).

Reference:

- Programming with ANSI C by: E. Balaguruswamy
- Let us C - Yashwanth kanetkar
- Computer concepts and C programming by - P. B. Kotur

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDA 530

DIGITAL ELECTRONICS

Credits: 3 (45 Hours)

(2 Hours of Theory and 1 Hour of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

CO1. Learn in detail with application, logic gates

CO2. Learn the classification and characteristics of combinational digital circuits

CO3. Deliberate the classification and characteristics of sequential digital circuits

CO4. Learn in details with application, flip flop

Unit 1

15 Hours

Representation of Information: Number system, integer and floating point representation, character codes (ASCII, EBCDIC)

Number Systems: Introduction to decimal, binary and hexadecimal number systems. Inter-conversion of decimal, binary and hex numbers

Binary Arithmetic and codes: Addition, multiplication and division in binary systems. Subtraction in binary systems –one’s and two’s complement methods. Subtraction of binary numbers by one’s and two’s complement methods. Concept of signed and unsigned numbers

Alphanumeric codes- ASCII and EBCDIC, concept of parity, error detection and correction

Logic Gates: Logic values and variables, positive and negative logic, AND, OR, NOT, NAND, NOR, AND, XOR gates, symbols and truth table. Definition of universal gates, NAND & NOR gates as universal gates

Boolean algebra: Laws of Boolean algebra. Principle of duality. DeMorgan’s theorems. Simplification of Boolean expressions. Boolean expression for logic circuits and vice versa SOP and POS notations. Canonical Expressions. Conversion from SOP to POS form and vice versa. Reduction of Boolean expressions (three/ four variables with don’t care conditions) using Karnaugh maps

Unit 2

15 Hours

Combinational Circuits: Half Adder, Full Adder, Half subtractor, Full subtractor, Encoders (Decimal to BCD) and decoders (BCD to Decimal), 4 X 1 Multiplexer and 1 X 4 demultiplexer - symbol and truth table

Sequential Circuits: RS flip flop, D flip flop. JK flip flop. Race around condition & T flip-flops. Shift registers –SISO, SIPO, PISO, PIPO registers. Brief explanation with Block diagrams. Counter - Synchronous and Asynchronous - Binary ripple counter and modulo counter

Semiconductor Memories: Idea of different types of Semiconductor memories (RAM, ROM, PROM, EPROM, EEPROM), process of data storage and retrieval, organization of memory, concept of PLA and PAL

Basic Building Blocks: ALU: arithmetic and logic unit operations, organization of control units, memory: types and organization, peripheral devices: I/O devices (video terminals and printers) and controllers, storage devices (tapes and disks), Programmed and interrupt control mechanism, I/O controllers, and bus bandwidths

Reference:

- Digital Electronic – Introduction to Theory & Practice by Gothmann
- Modern Digital Electronic (3rd Ed.) by Jain.
- Digital Principles & applications (6th Ed.) by Leech, Malvino and Saha.
- Digital Electronic by Thomas Floyd.
- The 8086 / 8088 Family Design, Programming & Interfacing by John Uffenbeck.
- 8086 Microprocessors Programming & Interfacing by Duglos V Hall.
- Intel Microprocessors Architecture, Programming & Interfacing (6th Ed.) by Barry B Bery.

Practicals

(1Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SEMESTER II

COMMUNICATIVE LANGUAGE ENGLISH

Credits: 3 (45 Hours)

(3 Hours of Theory per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Deliberate in details with examples Voice
- CO2. Specify in details with examples Articles
- CO3. Learn in depth Speech
- CO4. Deliberate the details of Writing Skills
- CO5. Learn in depth Speaking Skills

Module – 1 Grammar

	Marks	Hrs/ Week
1. Subject and Verb Agreement	5	6
2. Voice	5	5
3. Articles	5	3
4. Speech	5	6
5. Question tag	5	5
6. Framing of Questions	5	3+2=05

Module – 2 Writing Skills

1. Letter Writing Letter of Application/Letter of Grievances/Resume Preparation	10	4
2. Comprehension	10	3
3. Essay Writing	10	3

Module – 3 Speaking Skills

1. Greeting		
2. Requesting		
3. Enquiring		
4. Explaining	10	03+2=05
5. Reporting		
6. Permission		
7. Thanking		
	70	45

SDB 510

DISCRETE MATHEMATICS

Credits: 3 (45 Hours)

(3 Hours of Theory per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Understand the details of Matrices and Determinants
- CO2. Learn the details of Hamilton Theorem
- CO3. Understand in details with examples Graph Theory
- CO4. Understand the details of Calculus
- CO5. Identify in details with examples Definite and Indefinite Integrals

Unit 1: Matrices and Determinants

15 Hours

Algebra of Matrices and determinants –Elementary row operations- Rank of a matrix – Linear dependence of row and column vectors- System of Homogeneous linear equations-System of non homogeneous linear equations-Characteristic equations –Eigen values and Eigen vectors-Cayley –Hamilton theorem-Inverse of a matrix

Unit 2: Basics of graph theory

15 Hours

Definition-paths-matrix representation of graphs –planar graphs-non planar graphs- Coloring of graphs-chromatic number of graphs-Independent number

Unit 3: Calculus

15 Hours

Limits–Derivatives-Rules of differentiation-problems-differentiation of implicit Parametric and inverse functions-logarithmic differentiation and derivatives of second order Indefinite and definite integrals-simple problems

Reference:

- 1) Discrete Mathematics – Hari Kishan and Shiv Rajpundir
- 2) Matrices – Frank Ayers, Schaum publishing company
- 3) Text book of Matrices – Shanti Narayan and P k Mittal
- 4) Differential Calculus – Shanti Narayan
- 5) Differential Calculus and Integral Calculus – Piskunov
- 6) F.Harary – Graph Theory, Addition Wesley Reading Mass, 1969
- 7) Calculus – Volumes I & II.

SDB 520

ALGORITHMS AND DATA STRUCTURES

Credits: 3 (45 Hours)

(2 Hours of Theory and 1 Hour of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Deliberate in details with examples Algorithms and Data Structures
- CO2. Learn in depth Arrays
- CO3. Learn in details with examples Binary Search Trees
- CO4. Deliberate the characteristics of Heaps
- CO5. Specify the characteristics of Sorting Algorithms
- CO6. Learn the details of Shortest Path

Unit 1:

15 Hours

Algorithms and Data Structures: Asymptotic and Algorithm Analysis, Properties of data, Asymptotic Analysis, Algorithm Analysis.

Abstract Lists and Implementations: Linked lists and arrays, Stacks, Queues, De-queues.

Abstract Sorted Lists and Implementations: General trees, binary (including binary and complete trees), N-array trees and tree traversals, Abstract Sorted Lists, Binary search trees, Balanced search trees, AVL trees, B-Trees.

Unit 2:

15 Hours

Abstract Priority Queues: Heaps.

Abstract Sets/Maps: Chained Hash Tables, Linear Probing, Double Hashing.

Sorting Algorithms: Insertion and bubble sort, Heap, merge and quick sort, Bucket and radix sort

Graph and Direct Acyclic Graph Algorithms: Topological sort, Minimum spanning trees and shortest path.

Reference:

1. T.H. Cormen, C.E. Leiserson, R.L. Rivest and C. Stein, Introduction to Algorithms, 2nd Ed., Prentice-Hall of India, 2006.
2. Robert L. Kruse and A.J. Ryba, Data Structures and Program Design in C++, Prentice Hall, Inc., NJ, 1998.

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDB 550

INTRODUCTION TO MICROCONTROLLERS AND EMBEDDED SYSTEMS

Credits: 3 (45 Hours)

(2 Hours of Theory and 1 Hour of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Understand the architecture of 8051 microcontroller
- CO2. Write down the instruction set and simple programs of 8051 microcontroller
- CO3. Learn the details of 8051 microcontrollers
- CO4. Specify the characteristics of embedded system

Unit 1: Microcontrollers

15 Hours

Microcontroller 8051 - Introduction, block diagram of microprocessor, block diagram of microcontroller, comparison between microprocessor & microcontroller, Architecture of 8051 and pin out diagram of 8051.

Addressing modes - Data moves, Types of addressing modes - register addressing, immediate addressing, direct addressing, indirect addressing mode.

Instructions set - Data transfer instructions, arithmetic instructions, jump and call instructions.

PIC microcontroller - Core feature and over view of series.

Unit 2: Embedded Systems

15 Hours

Introduction to Embedded Systems - Definition of Embedded System, Embedded Systems Vs General Computing Systems, History of Embedded Systems, Classification, Major Application Areas, Purpose of Embedded Systems, Characteristics and Quality Attributes of Embedded Systems.

Hardware Side - introduction, The Core Level, Representing Information, Understanding Numbers, Addresses, Instructions, Registers. An Instruction Set View, Embedded Systems- A Register View, Register View of a Microprocessor

The Hardware Side: Storage Elements and Finite-State Machines - Theoretical model.

Reference:

- Microcontroller – K J Ayala.
- Introduction to Embedded Systems - Shibu K.V, McGraw Hill.
- Microcontroller – Mazadi.

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDC 510

SEMESTER III
ADVANCED CALCULUS

Credits: 3 (45 Hours)
(3 Hours of Theory per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Learn in depth Sequences
- CO2. Specify the details of Series
- CO3. Understand in details with examples Rolle's Theorem
- CO4. Deliberate in details with examples Taylor's Theorem
- CO5. Identify the details of Partial derivatives

Unit 1: Sequences and Series:

15 Hours

Sequences-Bounded and monotonic sequences-convergent, divergent and oscillatory sequences- standard results and simple problems

Infinite series-nth partial sum- geometric series-convergence of $\sum \frac{1}{n^p}$ - comparison test and ratio test-simple problems-alternating series

Unit 2: Calculus

15 Hours

Mean value theorems-Rolle's theorem, Lagrange's mean value theorem, Cauchy's mean value theorems (Statements and Geometrical interpretations)-Taylor's theorem – Maclaurin's expansion (Statement) and problems.

Unit 3: Partial derivatives

15 Hours

Limit and continuity of functions of two and three variables, Partial differentiation, Change of variables, Partial derivation and differentiability of real-valued functions of two and three variables, Euler's theorem on homogeneous functions. Taylor's theorem for functions of two and three variables - Jacobians

Reference:

1. A First Course in Real Analysis – Asharani Singhal.
2. Real Analysis – S.C. Malik.
3. Principles of Mathematical Analysis – Shanthinarayan
4. Calculus ,Volume -1 and Volume -2

SDC 520

DIFFERENTIAL EQUATIONS

Credits: 3 (45 Hours)

(3 Hours of Theory per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Specify in details with examples linear differential equations
- CO2. Write down in details with examples nonlinear differential equations
- CO3. Understand in details with examples Homogeneous linear differential equations
- CO4. Learn the details of Non homogeneous linear differential equations
- CO5. Specify the details of Partial differential equations

Unit 1: Linear and nonlinear differential equations

15 Hours

Elimination of arbitrary constant-solutions of linear differential equations - separation of variables –Homogeneous equations-exact equations- equations of the form $\frac{dy}{dx} + Py = Q$ - Integrating factor, Equations solvable for x, y, p. Clairaut's form and singular solutions

Unit 2: Homogeneous and non homogeneous linear differential equations

15 Hours

Homogeneous Linear differential equations with constant coefficients.-non homogeneous linear differential equations –inverse differential operators-Cauchy's homogeneous linear differential equations- Second order linear differential equations-variation of parameters and exact equations.

Unit 3: Partial differential equations

15 Hours

Total differential equations-simultaneous equations- partial differential equations-Lagranges form of linear partial differential equations-charpit's method.

Reference:

1. A short course in differential equations – Rainville and Bedient
2. Advanced Engineering Mathematics – Kreyszig
3. Higher Engineering Mathematics – Grewal
4. Laplace Transform –Murry R Spiegel
5. Applications of Differential equations –Martin Brown

SDC 530

SOFTWARE ARCHITECTURE AND SDLC& PROCESSES

Credits: 3 (45 Hours)

(2 Hours of Theory and 1 Hour of Practical per week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Specify in depth Software Process
- CO2. Deliberate in details with examples SDLC
- CO3. Learn in depth Data Modeling
- CO4. Understand the details of UML and ER Models
- CO5. Specify the details of Loose Coupling

Unit 1:

15 Hours

SDLC & Processes: Software Process, Software Development Life Cycle, Object-Oriented Concepts: connections between design and implementation, Software Testing, Object-Oriented Architecture and Design, Requirements analysis, Safety Critical Software.

Unit 2:

15 Hours

Software Architecture: Introduction to enterprise software architecture, the role of middleware, Cloud computing = SaaS + Utility Computing, Data Modeling, UML and E-R models. XML, Schemas, XML Schemas, Data Processing. Strategies for data processing, introduction to XQuery. JSON and JAXB, Domain-Driven Architecture. Domain-driven, design (DDD), Object-relational mapping (ORM), Service-oriented Architecture (SOA), Standardized service contract, Loose coupling, Service abstraction, Service-oriented Architecture (SOA),

Reference:

1. Designing Software Architectures: A Practical Approach (SEI Series in Software Engineering) 1st Edition, Kindle Edition by Humberto Cervantes
2. Just Enough Requirements and Sdlc: Requirements Documentation, Waterfall, and Agile Paperback – Import, 10 Mar 2017 by Ed Crook shanks

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDC 550

INDIAN CONSTITUTION

Credits: 3 (45 Hours)

(3 Hours of Theory per week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Learn the details of Features of Indian Constitution
- CO2. Understand the details of Fundamentals Rights
- CO3. Identify the details of Role of Prime Minister
- CO4. Learn the details of Power and Functions of Lok Sabha
- CO5. Specify the details of Power and Functions of Chief Minister

Unit 1 08 Hours

- a) Preamble of the Indian Constitution
- b) Salient features of Indian Constitution

Unit 2 10 Hours

- a) Fundamental Rights
- b) Fundamental Duties
- c) Directive principles of State Policy

Unit 3 14 Hours

- a) President – Election Method, Powers and Functions
- b) The Role of the Prime Minister
- c) The Parliament – Structure, Power and Functions(Lok Sabha and Rajya Sabha)
- d) Supreme Court – Organization and Jurisdiction

Unit 4 13 Hours

- a) The Role of Governor in the Administration of State
- b) Powers and Functions of the Chief Minister
- c) Composition , Powers and Functions of both the Houses of State Legislature
- d) High Court – Organization and Jurisdiction

Reference:

1. M.V. Pylee, “An Introduction to Constitution of India”, Vikas Publishing, 2002.
2. Brij Kishore Sharma, “Introduction to the Constitution of India”, PHI Learning Pvt. Ltd., New Delhi, 2011.
3. Latest Publications of Indian Institute of Human Rights, New Delhi.

SDD 510

**SEMESTER IV
NUMERICAL ANALYSIS AND STATISTICS**

Credits: 3 (45 Hours)
(3 Hours of Theory per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Understand the details of Bisection Method
- CO2. Identify in details with examples Range Kutta IV Order Method
- CO3. Learn the details of Finite differences
- CO4. Understand in depth Numerical Integration
- CO5. Identify in details with examples Linear programming

Unit 1: Numerical techniques

15 Hours

Numerical solutions of algebraic equations-Bisection method -Newton Raphson method, Regula Falsi method -iteration method-Euler method, Range kutta IV order methods

Unit 2: Finite differences and Numerical integration

15 Hours

Finite differences-Interpolation-Newton Gregory forward interpolation formula-Lagranges interpolation formula-Numerical integration: Trapezoidal rule, Simpson's $1/3^{\text{rd}}$ and $3/8^{\text{th}}$ rule Weddle's rule.

Unit 3: Linear programming

15 Hours

Linear programming –Requirements for a linear programming problem, Examples on the Applications of linear programming, Formulation of a linear programming, Standardization, Solving LPP by Graphical Method, Simplex Method (up to two variables)

Reference:

1. Numerical methods: S.S. Sastry.
2. Probability and statistics for engineers and Scientists – Ronald E .Walpole and Raymond H Mayers.
3. Mathematical Statistics - John Freund (Prentice Hall India PVT .Ltd)

OPERATION RESEARCH

Credits: 3 (45 Hours)

(3 Hours of Theory per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Understand the details of Different phases of operation research
- CO2. Deliberate in depth Simplex method
- CO3. Identify in depth Duality theorems
- CO4. Understand the details of Sequencing problems
- CO5. Learn in depth Transportation model problems
- CO6. Understand the details of Assignment problems

Unit 1:

15 Hours

Definition of the term Operation Research -Different phases of operation research
Advantages and limitations of O.R. Linear programming –Requirements for a linear programming problem , Examples on the Applications of linear programming ,Formulation of a linear programming , Standardization , Solving LPP by Graphical Method ,Simplex Method (up to two variables)

Unit 2:

15 Hours

Big M method revised simplex method, Dual simplex method, Duality theorems.
Sequencing problems: Processing ‘n’ jobs through two machines –Travelling salesman problems as an application of sequencing

Unit 3:

15 Hours

Transportation Model problems – Formulating, Solution –North West Corner Rule, Least Cost method, Row Minima method, Column minima method and Vogel’s approximation.
Assignment problem: formulating, method of finding initial basic feasible solution to Assignment problem using Hungarian method.

Reference:

1. ‘Operation Research’ by Kanthiswarup, Guptha, Manmohan – Sultan chand and sons Educational publishers, New Delhi,1996
2. ‘Operation Research’ by H. A.Taha Prentice Hall of India Ltd 1998
3. ‘Operation Research’ by S .D Sharma Kedarnath Ramnath and co (publishers)1997

SDD 530

SOFTWARE MODELING AND SOFTWARE QUALITY ASSURANCE

Credits: 3 (45 Hours)

(2 Hours of Theory and 1 Hour of Practical per week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Deliberate in details with examples Classes and Relationships
- CO2. Specify the details of State diagrams
- CO3. Identify in details with examples Events
- CO4. Deliberate in details with examples Software quality assurance
- CO5. Understand the details of Software quality assurance

Unit 1:

15 Hours

Software Modeling: What is Modeling?, Classes, Relationships, Common Mechanisms, Diagrams, Class Diagrams, Advanced Relationships, Instances, Object Diagrams, Use Cases, Interaction/Activity Diagrams, Events, State Machines, Time, Space, State Diagrams and Events

Unit 2:

15 Hours

Software Quality Assurance: Software Quality, Quality Assurance, Testing Concepts and Issues, Testing Activities, Testing Techniques. Other Techniques, Defect Prevention/Process Improvement, Inspection, Refactoring, CRC, Software Reliability Engineering, Quality Models and Measurements

Reference:

1. Modeling for Software Quality Assurance (English, Paperback, Srivyshnavi Pagadala)
2. Modeling for Software Quality Assurance Paperback – Import, 16 Nov 2013 by Srivyshnavi Pagadala

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDD560

ENVIRONMENTAL STUDIES

Credits: 3 (45 Hours)

(3 Hours of Theory per week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Deliberate the details of Components of environment
- CO2. Specify the details of Ecology and Ecosystems
- CO3. Identify in details with examples Natural resources
- CO4. Learn the details of Biodiversity
- CO5. Specify in details with examples Environmental pollution
- CO6. Identify the details of Environmental issues and policies

Unit 1: Environment and natural systems

4 Hours

- Introduction to Environment and Environmental Studies
- Definition and Components of Environment, Relationship between the different components of Environment
- Man and Environment relationship
- Impact of technology on Environment, Environmental Degradation
- Multidisciplinary nature of the Environment studies
- Its scope and importance in the present day Education System

Unit 2: Ecology and Ecosystems

5 Hours

- Introduction: Ecology- Objectives and Classification
- Concept of an ecosystem- structure and functions of ecosystem
- Components of ecosystem- Producers, Consumers, Decomposers
- Bio-Geo- Chemical Cycles- Hydrologic Cycle, Carbon cycle, Energy Flow in Ecosystem, Food Chains, Food webs, Ecological Pyramids
- Major Ecosystems: Forest Ecosystem, Grassland Ecosystem, Desert Ecosystem, Aquatic Ecosystem, Estuarine Ecosystem.

Unit 3: Natural Resources

6 Hours

Renewable and Non-renewable resources, exploitation and conservation,

- Water resources: Surface and Ground water sources, Indian and Global scenario.
- Land as a resource, land use change and land degradation
- Forest resources: Definition and Classification of Forests
Ecological and Economic importance and benefits of forest, Indian scenario
Deforestation: causes and effects, case studies remedial measures
- Food resources: Sources of food, Global and Indian food demand scenario
Limits of food production, Environmental effects of Agriculture
- Energy resources: Renewable and non renewable energy sources, use of alternate energy sources, growing energy needs, case studies
- Mineral resources: Definition and Classification of minerals, mining issues
Case studies
- Role of individual in conservation of natural resources.

Unit 4: Biodiversity and its Conservation

7 Hours

- Biodiversity: Definition, Levels of biological diversity: genetic, species and ecosystem diversity
- Bio geographic zones of India
- Hot spots of biodiversity
- Ecosystem and biodiversity services: Ecological, economic, social, ethical, aesthetic and Informational values
- Biodiversity patterns
- India as a mega-biodiversity nation
- Threats to biodiversity: Habitat loss, poaching of wildlife, man-wildlife conflicts, biological invasions.
- Endangered and endemic species of India
- Conservation of biodiversity: In-situ and Ex-situ conservation of biodiversity.

Unit 5: Environmental pollution

6 Hours

- Types of Environmental Pollution:
- Water Pollution: Introduction – Water Quality Standards, Sources of Water Pollution: Industrial Agricultural, Municipal; Classification of water pollutants, Effects of water pollutants, Eutrophication.
- Marine pollution: Causes, effects and control.
- Air Pollution: Composition of air, Structure of atmosphere, Ambient Air Quality Standards, Classification of air pollutants, Sources of common air pollutants like PM, SO₂, NO_X, Natural & Anthropogenic Sources, Effects of common air pollutants
- Soil Pollution: causes, effects and control.
- Noise Pollution: Introduction, Sound and Noise, Noise measurements, Causes and Effects
- Thermal Pollution: Causes, effects and control.
- Nuclear hazards and human health risks.
- Solid waste management: Control measures of urban and industrial waste.
- Role of individual in the prevention of pollution, Pollution case studies.

Unit 6: Sustainable development and Environmental issues and Policies 7 Hours

- Sustainable development: Meaning, changes in resource utilization.
- Water conservation: watershed management and Rain water harvesting.
- Environmental issues: Climate change, global warming, acid rain, ozone layer depletion.
- Disaster management: floods, drought, earthquake, cyclones and landslides.
- Wasteland reclamation.
- Environment Protection Act: Air, Water, Wildlife (Prevention and Control of Pollution)
- Forest Conservation Act.
- Issues involved in enforcement of environmental legislation.
- Environment: rights and duties.

Unit 7: Human Population and the Environment**5 Hours**

- Population growth, Explosion, demographic variation among nations.
- Family welfare Program.
- Environment, human health and welfare; infectious and lifestyle diseases in contemporary world.
- Value Education: Environmental ethics.
- HIV/AIDS
- Women and Child welfare.
- Role of information technology in Environment and human health

Unit 8: Field visit**5 Hours**

- Field work Visit to an area to document environmental assets river/ forest/ grassland/ hill/ mountain
- Visit to a local polluted site-Urban/Rural/Industrial/Agricultural.
- Visit to the solid waste treatment plant and water treatment plant.
- Video: The one degree (Equal to 5 lectures)

Reference:

1. Textbook of Environmental Studies for Undergraduate Courses by Erach Bharucha Second edition, 2013 Publisher: Universities Press (India) Private Ltd, Hyderabad.
2. Basics of Environmental Studies by Prof Dr N S Varandani, 2013 Publisher: LAP – Lambert Academic Publishing, Germany.
3. Environmental Studies by Anindita Basak, 2009 Publisher: Drling Kindersley (India) Pvt. Ltd Pearson
4. Textbook of Environmental Studies by Deeksha Dave & S S Kateva, Cengage Publishers.
5. Environmental Sciences by Daniel B Botkin & Edward A Keller Publisher: John Wiley & Sons.
6. Environmental Studies by R. Rajagopalan, Oxford University Press
7. Environmental Studies by Benny Joseph, TMH publishers
8. Environmental Studies by Dr. Suresh K Dhameja, 2007 Published by: S K Kataria & Sons New Delhi
9. Basics of Environmental Studies by U K Khare, 2011 Published by Tata McGraw Hill.
10. Environmental Studies by N. Arumugam & V.Kumaresan, Sara's publication.

SDE 510

SEMESTER V
PROJECT MANAGEMENT

Credits: 4 (60 Hours)

(2 Hour of Theory and 2 Hours of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Understand in depth Project plan
- CO2. Identify the details of User controlled scheduling
- CO3. Deliberate the details of Adding tasks
- CO4. Understand in details with examples Adding work resources
- CO5. Learn the details of Calendar

Unit 1:

15 Hours

- Creating a Project Plan
- User-Controlled Scheduling
- Adding Tasks
- Resolving Common Scheduling Issues

Unit 2:

15 Hours

- Adding Work Resources
- Adding Material and Cost Resources
- Defining a Calendar
- Viewing and Tracking Project Information Gantt Charts

Reference:

1. Project Management for Dummies, 5ed Paperback – 2017 by Stanley E. Portny
2. Project Management 3rd Edition (English, Paperback, Maylor)

Practicals

(2 Hours per week X 15 Weeks = 30 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDE 520

CONFIGURATION MANAGEMENT

Credits: 4 (60 Hours)

(3 Hour of Theory and 1 Hour of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Specify in depth Configuration management
- CO2. Understand the details of Configuration management planning
- CO3. Deliberate the details of Configuration control
- CO4. Learn in details with examples Configuration status accounting
- CO5. Identify the details of Configuration audits

Unit 1:

15 Hours

DEFINING CONFIGURATION MANAGEMENT: CM principles and standards, the recent growth of CM

CM PLANNING: Program phasing and milestones, Creating the CM organization, Defining CM system requirements, CM job classifications

CONFIGURATION IDENTIFICATION: First tasks of CM, Configurations & Baselines, Requirements traceability, Item identification and numbering

Unit 2:

15 Hours

ENGINEERING RELEASE: Control of technical data, the document control process, Development vs. formal release

CONFIGURATION CONTROL: Defining a closed-loop process, Change classifications, Review boards and CCBs, Processing changes and RDWs

INTRODUCTION TO SOFTWARE CONFIGURATION MANAGEMENT: Specific software CM tasks, SEI evaluation criteria

Unit 3:

15 Hours

CONFIGURATION STATUS ACCOUNTING: Defining CSA tasks and tailoring, Status accounting elements, Understanding the impact

CM PLANS: CMP preparation techniques, Software CMPs, Assessments and Plans
General procedures and work flow

CONFIGURATION AUDITS: Internal and informal audits, developing the audit plan, The Functional and Physical Audits

Reference:

1. Jessica Keyes, Software Configuration Management, Auerbach Publications, 2004
2. Anne Hass , Configuration Management Principles and Practice, Addison Wesley, 2002

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDE 530

HUMAN COMPUTER INTERACTION

Credits: 4 (60 Hours)

(3 Hours of Theory and 1 Hour of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Learn the details of Introduction of human computer interface
- CO2. Understand the details of Human consideration in screen design
- CO3. Identify in details with examples Windows
- CO4. Deliberate in depth Multimedia and coloring
- CO5. Specify in details with examples Hypermedia

Unit 1:

20 Hours

Introduction-Importance-Human-Computer interface-characteristics of graphics interface-Direct manipulation graphical system - web user interface-popularity-characteristic & principles. User interface design process- obstacles-usability-human characteristics in design - Human interaction speed-business functions-requirement analysis-Direct-Indirect methods-basic business functions-Design standards-system timings - Human consideration in screen design - structures of menus - functions of menus-contents of menu-formatting -phrasing the menu - selecting menu choice-navigating menus-graphical menus.

Unit 2:

13 Hours

Windows-Characteristics-components-presentation-styles-types-managements-organizations-operations-web systems-device-based controls: characteristics-Screen -based controls: operate control - text boxes-selection control-combination control-custom control-presentation control.

Unit 3:

12 Hours

Text for web pages - effective feedback-guidance & assistance-Internationalization-accessibility-Icons-Image-Multimedia –coloring, Windows layout-test: prototypes - kinds of tests - retest - Information search - visualization - Hypermedia - www - Software tools.

Reference:

1. Human-Computer Interaction, 3e Paperback – 2004 by Dix
2. Human Computer Interaction Paperback – 2014 by Meena K
3. Human-Computer Interaction Third Edition Paperback – Import, 24 Aug 2018 by Gerardus Blokdyk

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDF 510

SEMESTER VI

OPERATING SYSTEM

Credits: 4 (60 Hours)

(3 Hours of Theory and 1 Hour of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Identify the Characteristics of operating system
- CO2. Deliberate in depth Scheduling algorithms
- CO3. Learn in depth Semaphores
- CO4. Specify the details of Message passing
- CO5. Understand the details of Deadlock
- CO6. Identify in details with examples File organisation

Unit 1:

15 Hours

Overview of operating systems, functionalities and characteristics of OS

Hardware concepts related to OS, CPU states, I/O channels, memory hierarchy and microprogramming

The concept of a process, operations on processes, process states, concurrent processes, process control block, process context

UNIX process control and management, PCB, signals, forks and pipes

Interrupt processing, operating system organisation, OS kernel FLIH and dispatcher.

Job and processor scheduling, scheduling algorithms, process hierarchies.

Problems of concurrent processes, critical sections, mutual exclusion, synchronisation, deadlock.

Unit 2:

15 Hours

Mutual exclusion, process co-operation, producer and consumer processes.

Semaphores: definition, init, wait, signal operations.

Use of semaphores to implement mutex, process synchronisation etc., implementation of semaphores, Critical regions, Conditional Critical Regions, Monitors, Ada Tasks

Interprocess Communication (IPC), Message Passing, Direct and Indirect

Unit 3:

15 Hours

Deadlock: prevention, detection, avoidance, banker's algorithm.

Memory organisation and management, storage allocation

Virtual memory concepts, paging and segmentation, address mapping.

Virtual storage management, page replacement strategies

File organisation: blocking and buffering, file descriptor, directory structure, File and Directory structures, blocks and fragments, directory tree, inodes, file descriptors, UNIX file structure.

Reference:

1. D.M Dhamdhere: Operating systems - A concept based Approach, 3rd Edition, Tata McGraw- Hill, 2012.
2. P.C.P. Bhatt: Introduction to Operating Systems Concepts and Practice, 3rd Edition, PHI, 2010.
3. Harvey M Deital: Operating systems, 3rd Edition, Pearson Education, 2011.

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDF 550

DATABASE DESIGN

Credits: 4 (60 Hours)

(3 Hours of Theory and 1 Hour of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Understand the details of Database terminology and information types
- CO2. Specify in depth Database planning and designing
- CO3. Learn the details of Cloud computing
- CO4. Specify the details of Sets and normalization
- CO5. Specify in depth Report writing
- CO6. Understand in details with examples Customer service management

Unit 1

15 Hours

Database Basics
Introduction to Devise Digital Storage
Database Terminology & Database Information Types
Microsoft Excel versus Microsoft Access
Database Planning
Database Objects – Creating Fields and Tables
Planning and Designing a Database

Unit 2

15 Hours

Introduction to Cloud Computing
Database Relationships
Designing for the Business Case
Introduction to Data Security & Data Archives
Managing the Database
Database Relationship Development
Introduction to Visual Data Analytics
Introduction to Sets & Normalization
Database Extractions

Unit 3

15 Hours

Database Queries and Basic SQL
Emergence of Social Media Databases
Database Distribution
Report Writing
Introduction to Customer Service Management
Computing Databases
Database Inputs
Form Development

Reference:

1. Korth, Silberchatz, Sudarshan: "Database System Concepts", 6th Edition, McGraw – Hill
2. Elmasri and Navathe: "Fundamentals of Database Systems", 5th Edition, PEARSON Education.
3. Peter Rob and Carlos Coronel: "Database Systems Design, Implementation and Management", Thomson Learning, 5th Edition.

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

SDF 520

COMPUTER NETWORKS & SECURITY FUNDAMENTALS

Credits: 4 (60 Hours)

(3 Hours of Theory and 1 Hour of Practical per Week)

Course Outcome:

After completion of the course, the students are able to:

- CO1. Deliberate the details of OSI model
- CO2. Learn the details of Switches
- CO3. Understand the details of Protocols and services
- CO4. Identify in depth Security
- CO5. Learn in depth Encryption
- CO6. Specify the details of E-mail and server protection

Unit 1

15 Hours

COMPUTER NETWORKS

Network Infrastructure - Internet, intranet, and extranet.

Understand the OSI model.

TCP/IP

Local area networks (LANs), Wide area networks (WANs).

Network topologies and access methods

Network Hardware - switches.

Unit 2

15 Hours

Routers, media types

Protocols and Services

Understand IPv4 & IPv6.

Addressing names resolution & networking services.

SECURITY

Understanding Security Layers Principles, Physical, Wireless, Internet & Operating System Security

Unit 3

15 Hours

User authentication, permissions, password, audit policies, encryption, malware,

Network Security & dedicated firewalls.

Network Access Protection (NAP), network isolation, Protocol security, client, e-mail & server protection

Reference:

1. Fundamentals of Computer Networks Kindle Edition by SUDAKSHINA KUNDU
2. Introduction to Computer and Network Security Hardcover – 21 Aug 2013
by Richard R. Brooks

Practicals

(1 Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

QUESTION PAPER SCHEME

**JSS MAHAVIDYAPEETHA
JSS College of Arts, Commerce & Science
(Autonomous)**

Ooty Road, Mysore - 570 025

Title of the paper with code

Time: 3 Hours

Max Marks: 70

Part – A

I Answer any five Questions (out of six) 2x5=10

a)

b)

c)

d)

e)

f)

Part – B

Answer any two questions from each main

II a) 2x5=10

b)

c)

III a) 2x5=10

b)

c)

IV a) 2x5=10

b)

c)

Part – C

Answer any one question from each main

V a) 1x10=10

Or

b)

VI a) 1x10=10

Or

b)

VII a) 1x10=10

Or

b)

Question numbers V, VI and VII internal split-up (5+5, 7+3, 2+8 etc.)

QUESTION PAPER SCHEME

TIME: 3 Hrs

MAX MARKS: 70

Instructions to Students: All sections are compulsory.

SECTION A

- I Fill in the blanks with correct form of verbs 5 X 1 = 05
a)
b)
c)
d)
e)
- II Change the voice of the following sentences 5 X 1 = 05
a)
b)
c)
d)
e)
- III Fill in the blanks with suitable article 5 X 1 = 05
a)
b)
c)
d)
e)
- IV Change the speech of the following 5 X 1 = 05
a)
b)
c)
d)
e)
- V Add question tag to the following 5 X 1 = 05
a)
b)
c)
d)
e)
- VI Frame questions so as to get the underlined words as answers 5 X 1 = 05
a)
b)
c)
d)
e)

SECTION B

- VII 10 X 1 = 10
1. Write a letter of application
- Or
2. Prepare a Resume.
- VIII 10 X 1 = 10
- Read the following essay carefully and answer the questions.
- a)
 - b)
 - c)
 - d)
 - e)
- IX 10 X 1 = 10
- Write an essay on one of the following:
- a)
 - b)
 - c)
- X 5 X 2 = 10
- Answer the following (Dialogue writing)
- 1)
 - 2)

QUESTION PAPER PATTERN

Section A

Q1: Six questions of two marks each .Five questions to be answered $5 \times 2 = 10$

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.

Section B

Answer any four questions from each main. Each main carries 20 marks $3 \times 20 = 60$

Q2: Five sub questions of 5 marks each from unit 1

Q3: Five sub questions of 5 marks each from unit 2

Q4: Five sub questions of 5 marks each from unit 3

Skill Component Syllabus

Model Curriculum

JUNIOR SOFTWARE DEVELOPER

JUNIOR SOFTWARE DEVELOPER

SECTOR: **IT-ITeS**
SUB-SECTOR: **IT Services**
OCCUPATION: **Application Development**
REFERENCE ID: **SSC/Q0508, version 1.0**
NSQF LEVEL: **4**

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Junior Software Developer

CURRICULUM / SYLLABUS

This program is aimed at training candidates for the job of **Junior Software Developer** in the **IT-ITeS Sector/Industry** and aims at building the following key competencies in the learner.

Program Name	Junior Software Developer		
Qualification Pack Name & Reference ID.	Junior Software Developer SSC/Q0508, version 1.0		
Version No.	1.0	Version Update Date	31/12/2015
Pre-requisites to Training	10 th Standard		
Training Outcomes	<p>After completing this programme, participants will be able to:</p> <ul style="list-style-type: none"> assist in performing software construction and software testing entry-level tasks in the IT Services industry manage work to meet requirements maintain a healthy, safe and secure working environment 		

The Course encompasses all six National Occupational Standards (NOS) of **Junior Software Developer SSC/Q0508** Qualification Pack issued by **IT-ITeS Sector Skills Council NASSCOM**.

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
1	Basics of IT	05:00	15:00	Candidates will be able to: <ul style="list-style-type: none"> Demonstrate basic computer and internet literacy including operating a computer, describing its major components and how they work, using Windows and Linux OS, operating a browser, searching the internet, managing mails and using social internet media. 	SSC/N0506	Refer to Unique Equipment Required section
2	Problem Solving and Program Design	30:00	60:00	Candidates will be able to: <ul style="list-style-type: none"> Demonstrate aptitude for analysing information and making logical conclusions. 	SSC/N0506	Refer to Unique Equipment Required section



Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<ul style="list-style-type: none"> Demonstrate knowledge of the foundational mathematical concepts in computing. 		
3	Basic Algorithms and Application Development	30:00	60:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> Design algorithms to solve problems and convert them into code using the appropriate programming language constructs. Read and execute a test case and record the outcome in the appropriate template. Communicate effectively with appropriate people w.r.t. assigned roles in simple English – both oral and written. 	SSC/N0506	Refer to Unique Equipment Required section
4	Self and work Management	30:00	70:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> Establish and agree work requirements with appropriate people Keep immediate work area clean and tidy Utilize time effectively Use resources correctly and efficiently Treat confidential information correctly Work in line with organization’s policies and procedures Work within the limits of job role Obtain guidance from appropriate people, where necessary Ensure work meets the agreed requirements 	SSC/N9001	Refer to Unique Equipment Required section
5	Team Work and Communication	12:00	38:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> Obtain guidance from appropriate people to agree the analysis to be performed on the data Obtain advice and guidance from appropriate people on issues with data analysis 	SSC/N9002	Refer to Unique Equipment Required Section

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<p>outside their area of competence or</p> <ul style="list-style-type: none"> • Review the results of their analysis with appropriate people • Undertake modifications to your analysis based on inputs from appropriate people • Communicate with colleagues clearly, concisely and accurately • Work with colleagues to integrate their work effectively with them • Pass on essential information to colleagues in line with organizational requirements • Work in ways that show respect for colleagues • Carry out commitments they have made to colleagues • Let colleagues know in good time if they cannot carry out your commitments, explaining the reasons • Identify any problems they have working with colleagues and take the initiative to solve these problems • Follow the organization's policies and procedures for working with colleagues 		
6	Managing Health and Safety	12:00	38:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> • Comply with organization's current health, safety and security policies and procedures • Report any identified breaches in health, safety, and security policies and procedures to the designated person • Identify and correct any hazards that can deal with 	SSC/N9003	Refer to Unique Equipment Required section



Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<p>safely, competently and within the limits of authority</p> <ul style="list-style-type: none"> • Report any hazards that one is not competent to deal with to the relevant person in line with organizational procedures and warn other people who may be affected • Follow their organization’s emergency procedures promptly, calmly, and efficiently • Identify and recommend opportunities for improving health, safety, and security to the designated person • Complete any health and safety records legibly and accurately 		
7	Data and Information Management	15:00	35:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> • Establish and agree with appropriate people the data/information they need to provide, the formats in which you need to provide it, and when they need to provide it • Obtain the data/information from reliable sources • Check that the data/information is accurate, complete and up-to-date • Obtain advice or guidance from appropriate people where there are problems with the data/information • Carry out rule-based analysis of the data/information, if required • Insert the data/information into the agreed formats • Check the accuracy of work, involving colleagues where required • Report any unresolved anomalies in the 	SSC/N9004	Refer to Unique Equipment Required Section

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<p>data/information to appropriate people</p> <ul style="list-style-type: none"> • Provide complete, accurate and up-to-date data/information to the appropriate people in the required formats on time 		
8	Learning and Self Development	05:00	20:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> • Obtain advice and guidance from appropriate people to develop your knowledge, skills and competence • Identify accurately the knowledge and skills they need for your job role • Identify accurately their current level of knowledge, skills and competence and any learning and development needs • Agree with appropriate people a plan of learning and development activities to address their learning needs • Undertake learning and development activities in line with their plan • Apply new knowledge and skills in the workplace, under supervision • Obtain feedback from appropriate people on their knowledge and skills and how effectively you apply them • Review their knowledge, skills and competence regularly and take appropriate action 	SSC/N9005	Refer to Unique Equipment Required Section
	Total Duration:	<u>114:00</u>	<u>286:00</u>	<p>Unique Equipment Required: Training room should be fully furnished with the following equipment / tools / accessories. Additional / specific resources, wherever applicable (e.g. Hardware, software) are indicated in the main text corresponding to relevant learning outcome.</p>		



Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<p>For Domain NOS, For NOS SSC/N0506 – HTML, C++ / Java, IDE</p> <p>General:</p> <ul style="list-style-type: none"> • Comfortable seats with adequate lighting, controlled temperature and acoustics for training and learning • White Board, Markers and Eraser • Projector with screen • Flip chart with markers • Faculty’s PC/Laptop with latest configuration and internet connection • Supporting software / applications for projecting audio, video, recording, • Presentation Tools to support learning activities: • Intranet • Email • IMs • Learning management system e.g. Moodle, Blackboard to enable blended learning • Microphone / voice system for lecture and class activities • Handy Camera • Stationery kit – Staples, Glue, Chart Paper, Sketch Pens, Paint Box, Scale, A4 Sheets • For IT Lab sessions: Computer Lab with 1:1 PC : trainee ratio and having internet connection, MS Office / Open office, Browser, Outlook / Any other Email Client and chat tools. • Assessment and Test Tools for day to day online Tests and Assessments • For team discussions: Adequate seating arrangement in full / half circle format for one or more teams as per planned team composition. • Reading Resources: Access to relevant sample documents and learning forums to enable self-study before and after each training session. 		

Grand Total Course Duration: **400 Hours 0 Minutes**

(This Syllabus/Curriculum has been approved by IT-ITes Sector Skills Council NASSCOM.)

Notes from IT-ITeS Sector Skills Council NASSCOM

1. This document outlines the broad scope of coverage. This should be linked with OBF and training delivery plan. OBF (Outcome based framework) reflects the pedagogy used to ensure an expected outcome. Training delivery plan focuses on the sequence of delivery.
2. Though many NOSs have some seemingly common outcomes, notably core/generic, professional and technical skills, it is imperative to understand the contextual difference between them. For example, writing skills required to document program structure and code (in SSC/N0506) are different from the writing skills required to prepare a time plan (in SSC/N9001). Training providers are advised to,
 - a. Embed such skills development in the learning pedagogy for each expected outcome
 - b. Prepare a detailed session plan for training delivery with focus on sequence and duration of training
 - c. Run a diagnostic test to assess prior learning of students and help trainers / students identify the need for gap training, optimal duration and suitable training methodology. Accordingly, more introductory level sessions may be included in guided or self-paced mode of learning. E.g. adding some sessions on Functional English or Use of Internet and MS Office.



Annexure1: Assessment Criteria

Assessment Criteria for Junior Software Developer	
Job Role	Junior Software Developer
Qualification Pack	SSC/Q0508
Sector Skill Council	IT-ITeS

Sr. No.	Guidelines for Assessment
1	Criteria for assessment for each Qualification Pack (QP) will be created by the Sector Skill Council (SSC). Each performance criteria (PC) will be assigned Theory and Skill/Practical marks proportional to its importance in NOS.
2	The assessment will be conducted online through assessment providers authorised by SSC.
3	Format of questions will include a variety of styles suitable to the PC being tested such as multiple choice questions, fill in the blanks, situational judgment test, simulation and programming test.
4	To pass a QP, a trainee should pass each individual NOS. Standard passing criteria for each NOS is 70%.
5	For latest details on the assessment criteria, please visit www.sscnasscom.com .

ASSESSMENT OUTCOME (NOS CODE AND DESCRIPTION)	Assessment criteria (PC)	Total Marks	Out Of	MARKS ALLOCATION	
				Theory	Skills Practical
1.SSC/N0506 (Deal remotely with customer queries - Domestic)	PC1. greet customers and verify details, following your organization’s procedures	120	12.5	2.5	10
	PC2. read carefully, summarize, and obtain customer confirmation of, your understanding of queries		12.5	2.5	10
	PC3. express your concern for any difficulties caused and your commitment to resolving queries		15	0	15
	PC4. record and categorize queries accurately using your organization’s query management tool		5	0	5
	PC5. refer queries outside your area of competence or authority promptly to appropriate people		2.5	0	2.5
	PC6. access your organization’s knowledge base for solutions to queries, where available		2.5	0	2.5
	PC7. resolve queries within your area of competence or authority in line with organizational guidelines and service level agreements (SLAs)		15	0	15
	PC8. obtain advice and guidance from appropriate people, where necessary		2.5	0	2.5
	PC9. obtain confirmation from customers that queries have been resolved to satisfaction		10	0	10

	PC10. record the resolution of queries accurately using your organization's query management tool		35	15	20
	PC11. comply with relevant standards, policies, procedures and guidelines when dealing remotely with customer queries		7.5	0	7.5
		NOS Total	120	20	100
2.SSC/N9001 (Manage your work to meet requirements)	PC1. establish and agree your work requirements with appropriate people	40	10	5	5
	PC2. keep your immediate work area clean and tidy		5	0	5
	PC3. utilize your time effectively		5	5	0
	PC4. use resources correctly and efficiently		5	2.5	2.5
	PC5. treat confidential information correctly		5	0	5
	PC6. work in line with your organization's policies and procedures		2.5	0	2.5
	PC7. work within the limits of your job role		2.5	0	2.5
	PC8. obtain guidance from appropriate people, where necessary		2.5	0	2.5
	PC9. ensure your work meets the agreed requirements		2.5	0	2.5
			NOS Total	40	12.5
3.SSC/N9003 (Maintain a healthy, safe and secure working environment)	PC1. comply with your organization's current health, safety and security policies and procedures	40	10	5	5
	PC2. report any identified breaches in health, safety, and security policies and procedures to the designated person		5	0	5
	PC3. identify and correct any hazards that you can deal with safely, competently and within the limits of your authority		10	5	5
	PC4. report any hazards that you are not competent to deal with to the relevant person in line with organizational procedures and warn other people who may be affected		5	0	5
	PC5. follow your organization's emergency procedures promptly, calmly, and efficiently		5	0	5
	PC6. identify and recommend opportunities for improving health, safety, and security to the designated person		2.5	0	2.5
	PC7. complete any health and safety records legibly and accurately		2.5	0	2.5
		NOS Total	40	10	30



Annexure2: Trainer Prerequisites for Job role: Junior Software Developer mapped to Qualification Pack: SSC/Q0508

Sr. No.	Area	Details
1	Job Description	To deliver accredited training service, mapping to the curriculum detailed above, in accordance with the Qualification Pack SSC/Q0508.
2	Personal Attributes	Aptitude for conducting training, and pre/ post work to ensure competent, employable candidates at the end of the training. Strong communication skills, interpersonal skills, ability to work as part of a team; a passion for quality and for developing others; well-organised and focused, eager to learn and keep oneself updated with the latest in this field.
3	Minimum Educational Qualifications	Minimum 12 th Standard; Preferred Master’s degree in any discipline
4a	Domain Certification	Minimum accepted score in SSC Assessment is 90% per NOS being taught in QP SSC/Q0508. Additional certification in customer orientation, dealing with difficult customers, written communication etc. will be an added advantage.
4b	Platform Certification	Recommended that the Trainer is certified for the Job Role: “Trainer” mapped to the Qualification Pack: “SSC/Q1402”. Minimum accepted score is 70% per NOS.
5	Experience	Field experience: Minimum 2 years’ experience in the same domain Training experience: 1 year preferred



Certificate

CURRICULUM COMPLIANCE TO QUALIFICATION PACK – NATIONAL OCCUPATIONAL STANDARDS

is hereby issued by the

IT-ITES SECTOR SKILLS COUNCIL NASSCOM

for the

MODEL CURRICULUM

Complying to National Occupational Standards of
Job Role/ Qualification Pack: **'Junior Software Developer'** QP No. **'SSC/Q0508NSQF Level 4'**

Date of Issuance: December 31st, 2015

Valid up to*: December 31st, 2016

* Valid up to the next review date of the Qualification Pack

Authorised Signatory
Lakshmi Narayan
(Chairman, IT-ITeS Sector Skills Council NASSCOM)



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Model Curriculum

WEB DEVELOPER

WEB DEVELOPER

SECTOR: **IT-ITeS**
SUB-SECTOR: **IT Services**
OCCUPATION: **Application Development**
REFERENCE ID: **SSC/Q0503, version 1.0**
NSQF LEVEL: **5**

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Web Developer

CURRICULUM / SYLLABUS

This program is aimed at training candidates for the job of **Web Developer** in the **IT-ITeS Sector/Industry** and aims at building the following key competencies in the learner.

Program Name	Web Developer		
Qualification Pack Name & Reference ID.	Web Developer SSC/Q0503, version 1.0		
Version No.	1.0	Version Update Date	31/12/2015
Pre-requisites to Training	Graduate degree/ diploma in web design/ media design or any other related field		
Training Outcomes	<p>After completing this programme, participants will be able to:</p> <ul style="list-style-type: none"> • Contribute to the design of software products and applications • Develop media content and graphic designs for software products and Applications • Manage their work to meet requirements • Work effectively with colleagues • Maintain a healthy, safe and secure working environment • Provide data/information in standard formats • Develop their knowledge, skills and competence 		

The Course encompasses all seven National Occupational Standards (NOS) of **Web Developer SSC/Q0503** Qualification Pack issued by **IT-ITeS Sector Skills Council NASSCOM**.

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
1	Programming for the Web	20:00	30:00	Candidates will be able to: <ul style="list-style-type: none"> • Design basic programming structures to implement functionality in line with requirements defined in BRS/URS, SRS and HLD 	SSC/N0501	Refer to Unique Equipment Required Section
2	Analysis and Design of Web based Applications	20:00	30:00	Candidates will be able to: <ul style="list-style-type: none"> • Check their understanding of the Business Requirements Specification (BRS)/User 	SSC/N0501	Refer to Unique Equipment Required Section



Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				Requirements Specification (URS) with appropriate people <ul style="list-style-type: none"> • Check their understanding of the Software Requirements Specification (SRS) with appropriate people • Check their understanding of High Level Design (HLD) with appropriate people • Review their designs with appropriate people • Analyse inputs from appropriate people to identify, resolve and record design defects and inform future designs • Document their designs using standard templates and tools • Comply with their organization’s policies, procedures and guidelines when contributing to the design of software products and applications 		
3	Media Content and Graphics Design	20:00	80:00	Candidates will be able to: <ul style="list-style-type: none"> • Check their understanding of the Business Requirements Specification (BRS), Software Requirements Specification (SRS), High Level Design (HLD) and Low Level Design (LLD) with appropriate people • Access reusable components, media and graphical packages and tools from their organization’s knowledge base 	SSC/N0503	Refer to Unique Equipment Required Section

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<ul style="list-style-type: none"> Convert requirements into media content and graphic designs, leveraging reusable components where available Review media content and graphic designs with appropriate people and analyze their feedback Record any defects and corrective actions taken to inform future work Rework media content and graphic designs, incorporating feedback Submit media content timely and graphic designs for approval by appropriate people Update their organization's knowledge base with their experiences of the media content and graphic designs developed Comply with their organization's policies, procedures and guidelines when developing media content and graphic designs for software products and applications 		
4	Self and work Management	12:00	38:00	Candidates will be able to: <ul style="list-style-type: none"> Establish and agree their work requirements with appropriate people Keep their immediate work area clean and tidy utilize their time effectively 	SSC/N9001	Refer to Unique Equipment Required Section



Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<ul style="list-style-type: none"> • Use resources correctly and efficiently • Treat confidential information correctly • Work in line with organization’s policies and procedures • Work within the limits of their job role • Obtain guidance from appropriate people, where necessary • Ensure their work meets the agreed requirements 		
5	Team Work and Communication	12:00	38:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> • Communicate with colleagues clearly, concisely and accurately • Work with colleagues to integrate their work effectively with them • Pass on essential information to colleagues in line with organizational requirements • Work in ways that show respect for colleagues • carry out commitments they have made to colleagues • Let colleagues know in good time if they cannot carry out their commitments, explaining the reasons • Identify any problems they have working with colleagues and take the initiative to solve these problems • Follow the organization’s policies and procedures for working with colleagues 	SSC/N9002	Refer to Unique Equipment Required Section

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
6	Managing Health and Safety	05:00	20:00	Candidates will be able to: <ul style="list-style-type: none"> • Comply with their organization's current health, safety and security policies and procedures • Report any identified breaches in health, safety, and security policies and procedures to the designated person • Identify and correct any hazards that they can deal with safely, competently and within the limits of their authority • Report any hazards that they are not competent to deal with to the relevant person in line with organizational procedures and warn other people who may be affected • Follow their organization's emergency procedures promptly, calmly, and efficiently • Identify and recommend opportunities for improving health, safety, and security to the designated person • Complete any health and safety 	SSC/ N 9003	
7	Data and Information Management	15:00	35:00	Candidates will be able to: <ul style="list-style-type: none"> • Establish and agree with appropriate people the data/information they need to provide, the formats in which they need to provide it, and when they need to provide it 	SSC/N9004	Refer to Unique Equipment Required Section

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<ul style="list-style-type: none"> • Obtain the data/information from reliable sources • Check that the data/information is accurate, complete and up-to-date • Obtain advice or guidance from appropriate people where there are problems with the data/information • Carry out rule-based analysis of the data/information, if required • Insert the data/information into the agreed formats • Check the accuracy of their work, involving colleagues where required • Report any unresolved anomalies in the data/information to appropriate people • Provide complete, accurate and up-to-date data/information to the appropriate people in the required formats on time 		
8	Learning and Self Development	5:00	20:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> • Obtain advice and guidance from appropriate people to develop their knowledge, skills and competence • Identify accurately the knowledge and skills they need for their job role • Identify accurately their current level of knowledge, skills and 	SSC/N9005	Refer to Unique Equipment Required Section

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				competence and any learning and development needs <ul style="list-style-type: none"> • Agree with appropriate people a plan of learning and development activities to address their learning needs • Undertake learning and development activities in line with their plan • Apply their new knowledge and skills in the workplace, under supervision • Obtain feedback from appropriate people on their knowledge and skills and how effectively they apply them • Review their knowledge, skills and competence regularly and take appropriate action 		
	Total Duration:	<u>109:00</u>	<u>291:00</u>	Unique Equipment Required: Training room should be fully furnished with the following equipment / tools / accessories. Additional / specific resources, wherever applicable (e.g. Hardware, software) are indicated in the main text corresponding to relevant learning outcome. For Domain NOSs: <ul style="list-style-type: none"> • NOS SSC/N0501: HTML5, Javascript, CSS, SQL, Web Builder, Word Press, Joomla and modelling tools such as Visio, UML • NOS SSC/N0503: HTML5, CSS, Flash, Photoshop, Windows media player, Eclipse, XAMPP General: <ul style="list-style-type: none"> • Comfortable seats with adequate lighting, controlled temperature and acoustics for training and learning 		



Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<ul style="list-style-type: none"> • White Board, Markers and Eraser • Projector with screen • Flip chart with markers • Faculty’s PC/Laptop with latest configuration and internet connection • Supporting software / applications for projecting audio, video, recording, • Presentation Tools to support learning activities: <ul style="list-style-type: none"> ○ Intranet ○ Email ○ IMs ○ Learning management system e.g. Moodle, Blackboard to enable blended learning • Microphone / voice system for lecture and class activities • Handy Camera • Stationery kit – Staples, Glue, Chart Paper, Sketch Pens, Paint Box, Scale, A4 Sheets • For IT Lab sessions: Computer Lab with 1:1 PC:trainee ratio and having internet connection, MS Office / Open office, Browser, Outlook/ other Email Clients • Assessment and Test Tools for day to day online Tests and Assessments • For team discussions: Adequate seating arrangement in full / half circle format for one or more teams as per planned team composition. • Reading Resources: Access to relevant sample documents and learning forums to enable self-study before and after each training session. 		

Grand Total Course Duration: **400 Hours 0 Minutes**

(This Syllabus/Curriculum has been approved by IT-ITes Sector Skills Council NASSCOM.)



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1. This document outlines the broad scope of coverage. This should be linked with OBF and training delivery plan. OBF (Outcome based framework) reflects the pedagogy used to ensure an expected outcome. Training delivery plan focuses on the sequence of delivery.
2. Though many NOSs have some seemingly common outcomes, notably core/generic, professional and technical skills, it is imperative to understand the contextual difference between them. For example, writing skills required write design specifications (in SSC/N0501) are different from the writing skills required to prepare a time plan (in SSC/N9001). Training providers are advised to,
 - a. Embed such skills development in the learning pedagogy for each expected outcome
 - b. Prepare a detailed session plan for training delivery with focus on sequence and duration of training
 - c. Run a diagnostic test to assess prior learning of students and help trainers / students identify the need for gap training, optimal duration, and suitable training methodology. Accordingly, more introductory level sessions may be included in guided or self-paced mode of learning. E.g. adding some sessions on Functional English or Use of Internet and MS Office.



Annexure 1: Assessment Criteria

Assessment Criteria for Web Developer	
Job Role	Web Developer
Qualification Pack	SSC/Q0503
Sector Skill Council	IT-ITeS

Sr. No.	Guidelines for Assessment
1	Criteria for assessment for each Qualification Pack (QP) will be created by the Sector Skill Council (SSC). Each performance criteria (PC) will be assigned Theory and Skill/Practical marks proportional to its importance in NOS.
2	The assessment will be conducted online through assessment providers authorised by SSC.
3	Format of questions will include a variety of styles suitable to the PC being tested such as multiple choice questions, fill in the blanks, situational judgment test, simulation and programming test.
4	To pass a QP, a trainee should pass each individual NOS. Standard passing criteria for each NOS is 70%.
5	For latest details on the assessment criteria, please visit www.sscnasscom.com .

ASSESSMENT OUTCOME (NOS CODE AND DESCRIPTION)	ASSESSMENT CRITERIA (PC)	TOTAL MARKS	OUT OF	MARKS ALLOCATION	
				THEORY	SKILLS PRACTICAL
1. SSC/N0501 (Contribute to the design of software products and applications)	PC1. check their understanding of the Business Requirements Specification (BRS)/User Requirements Specification (URS) with appropriate people	100	10	10	0
	PC2. check their understanding of the Software Requirements Specification (SRS) with appropriate people		10	10	0
	PC3. check their understanding of High Level Design (HLD) with appropriate people		10	10	0
	PC4. design basic programming structures to implement functionality in line with requirements defined in BRS/URS, SRS and HLD		30	0	30
	PC5. review their designs with appropriate people		5	5	0
	PC6. analyze inputs from appropriate people to identify, resolve and record design defects and inform future designs		15	5	10
	PC7. document their designs using standard templates and tools		10	0	10
	PC8. comply with their organization’s policies, procedures and guidelines when contributing to the design of software products and applications		10	0	10
	Total		100	40	60
2. SSC/N0503 (Develop media)	PC1. check their understanding of the Business Requirements Specification (BRS), Software	100	10	10	0



Transforming the skill landscape

content and graphic designs for software products and Applications)	Requirements Specification (SRS), High Level Design (HLD) and Low Level Design (LLD) with appropriate people			
	PC2. access reusable components, media and graphical packages and tools from their organization's knowledge base	10	0	10
	PC3. convert requirements into media content and graphic designs, leveraging reusable components where available	20	0	20
	PC4. review media content and graphic designs with appropriate people and analyze their feedback	10	5	5
	PC5. record any defects and corrective actions taken to inform future work	10	0	10
	PC6. rework media content and graphic designs, incorporating feedback	10	5	5
	PC7. submit media content and graphic designs for approval by appropriate people	10	0	10
	PC8. update their organization's knowledge base with their experiences of the media content and graphic designs developed	10	0	10
	PC9. comply with their organization's policies, procedures and guidelines when developing media content and graphic designs for software products and applications	10	0	10
	Total	100	20	80
3.SSC/N9001 (Manage their work to meet requirements)	PC1. establish and agree their work requirements with appropriate people	7.5	0	7.5
	PC2. keep their immediate work area clean and tidy	15	7.5	7.5
	PC3. utilize their time effectively	15	7.5	7.5
	PC4. use resources correctly and efficiently	15	7.5	7.5
	PC5. treat confidential information correctly	7.5	0	7.5
	PC6. work in line with their organization's policies and procedures	15	0	15
	PC7. work within the limits of their job role	7.5	0	7.5
	PC8. obtain guidance from appropriate people, where necessary	7.5	0	7.5
	PC9. ensure their work meets the agreed requirements	10	0	10
	Total	100	22.5	77.5
4.SSC/N9002 (Work effectively with colleagues)	PC1. communicate with colleagues clearly, concisely and accurately	20	0	20
	PC2. work with colleagues to integrate their work effectively with theirs	10	0	10



	PC3. pass on essential information to colleagues in line with organizational requirements		10	10	0
	PC4. work in ways that show respect for colleagues		20	0	20
	PC5. carry out commitments you have made to colleagues		10	0	10
	PC6. let colleagues know in good time if you cannot carry out their commitments, explaining the reasons		10	10	0
	PC7. identify any problems you have working with colleagues and take the initiative to solve these problems		10	0	10
	PC8. follow the organization's policies and procedures for working with colleagues		10	0	10
		Total	100	20	80
5.SSC/N9003 (Maintain a healthy, safe and secure working environment)	PC1. comply with their organization's current health, safety and security policies and procedures		20	10	10
	PC2. report any identified breaches in health, safety, and security policies and procedures to the designated person		10	0	10
	PC3. identify and correct any hazards that you can deal with safely, competently and within the limits of their authority		20	10	10
	PC4. report any hazards that you are not competent to deal with to the relevant person in line with organizational procedures and warn other people who may be affected	100	10	0	10
	PC5. follow their organization's emergency procedures promptly, calmly, and efficiently		20	10	10
	PC6. identify and recommend opportunities for improving health, safety, and security to the designated person		10	0	10
	PC7. complete any health and safety records legibly and accurately		10	0	10
		Total	100	30	70
6.SSC/N9004 (Provide data/information in standard formats)	PC1. establish and agree with appropriate people the data/information you need to provide, the formats in which you need to provide it, and when you need to provide it		15	15	0
	PC2. obtain the data/information from reliable sources		15	0	15
	PC3. check that the data/information is accurate, complete and up-to-date	100	15	5	10
	PC4. obtain advice or guidance from appropriate people where there are problems with the data/information		5	5	0
	PC5. carry out rule-based analysis of the data/information, if required		20	0	20



	PC6. insert the data/information into the agreed formats		10	0	10
	PC7. check the accuracy of their work, involving colleagues where required		10	0	10
	PC8. report any unresolved anomalies in the data/information to appropriate people		5	5	0
	PC9. provide complete, accurate and up-to-date data/information to the appropriate people in the required formats on time		5	0	5
		Total	100	30	70
7.SSC/N9005 (Develop their knowledge, skills and competence)	PC1. obtain advice and guidance from appropriate people to develop their knowledge, skills and competence		20	7	13
	PC2. identify accurately the knowledge and skills you need for their job role		14	7	7
	PC3. identify accurately their current level of knowledge, skills and competence and any learning and development needs		14	0	14
	PC4. agree with appropriate people a plan of learning and development activities to address their learning needs	100	7	0	7
	PC5. undertake learning and development activities in line with their plan		12	0	12
	PC6. apply their new knowledge and skills in the workplace, under supervision		12	0	12
	PC7. obtain feedback from appropriate people on their knowledge and skills and how effectively you apply them		7	0	7
	PC8. review their knowledge, skills and competence regularly and take appropriate action		14	7	7
		Total	100	21	79



Annexure 2: Trainer Prerequisites for Job role: Web Developer mapped to Qualification Pack: SSC/Q0503

Sr. No.	Area	Details
1	Job Description	To deliver accredited training service, mapping to the curriculum detailed above, in accordance with the Qualification Pack SSC/Q0503.
2	Personal Attributes	Aptitude for conducting training, and pre/ post work to ensure competent, employable candidates at the end of the training. Strong communication skills, interpersonal skills, ability to work as part of a team; a passion for quality and for developing others; well-organised and focused, eager to learn and keep oneself updated with the latest in this field.
3	Minimum Educational Qualifications	Minimum Graduate degree/ diploma in web design/ media design or any other related field; Preferred Master’s Degree in Media Design
4a	Domain Certification	Minimum accepted score in SSC Assessment is 90% per NOS being taught in QP SSC/Q0503. Certification in relevant software competencies: Software Development Certifications in C++, Embedded, C#, C, Java etc., is an added advantage.
4b	Platform Certification	Recommended that the Trainer is certified for the Job Role: “Trainer” mapped to the Qualification Pack: “SSC/1402” . Minimum accepted score is 70%.
5	Experience	Field experience: Minimum 2 years’ experience in the same domain Training experience: 1 year preferred



Certificate

CURRICULUM COMPLIANCE TO QUALIFICATION PACK – NATIONAL OCCUPATIONAL STANDARDS

Is hereby issued by the

IT-ITES SECTOR SKILLS COUNCIL NASSCOM

for the

MODEL CURRICULUM

Complying to National Occupational Standards of
Job Role/ Qualification Pack: **'Web Developer'** QP No. **'SSC/Q0503 NSQF Level 5'**

Date of Issuance: December 31st, 2015

Valid up to*: December 31st, 2016

* Valid up to the next review date of the Qualification Pack

Authorised Signatory
Lalchimi Narayan
(Chairman, IT-ITeS Sector Skills Council NASSCOM)



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Model Curriculum

Software Developer

SECTOR: IT-ITES

SUB-SECTOR: IT SERVICES

OCCUPATION: DATA SCIENTISTS

REF. ID: SSC/Q0401, VERSION 1.0

NSQF LEVEL: 7



Certificate

CURRICULUM COMPLIANCE TO QUALIFICATION PACK – NATIONAL OCCUPATIONAL STANDARDS

is hereby issued by the

IT-ITES SECTOR SKILLS COUNCIL NASSCOM

for the

MODEL CURRICULUM

Complying to National Occupational Standards of
Job Role/Qualification Pack: **Software Developer** QP No. **SSC/Q0501 NSQF Level 7**

Date of Issuance: December 31st, 2015

Valid up to*: December 31st, 2016

* Valid up to the next review date of the Qualification Pack


Authorised Signatory
Lakshmi Narayan
(Chairman, IT-ITeS Sector Skills Council NASSCOM)

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Software Developer

CURRICULUM / SYLLABUS

This program is aimed at training candidates for the job of **Software Developer** in the **IT-ITeS** Sector/Industry and aims at building the following key competencies in the learner.

Program Name	Software Developer		
Qualification Pack Name & Reference ID.	Software Developer SSC/Q0501, version 1.0		
Version No.	1.0	Version Update Date	31/12/2015
Pre-requisites to Training	BSc (Stat, Math, Physics, Chemistry, Geology) or BE/ BTech		
Training Outcomes	<p>After completing this programme, participants will be able to:</p> <ul style="list-style-type: none"> • Contribute to the design of software products and applications • Develop software code to specification • Manage their work to meet requirements • Work effectively with colleagues • Maintain a healthy, safe and secure working environment • Provide data/information in standard formats • Develop their knowledge, skills and competence 		

The Course encompasses all seven National Occupational Standards (NOS) of Software Developer SSC/Q0501 Qualification Pack issued by IT-ITeS Sector Skills Council NASSCOM.

Sr. No.	Module	Key Learning Outcomes	Equipment Required
1	Programming and Algorithms Theory Duration (hh:mm) 20:00 Practical Duration (hh:mm) 30:00 Corresponding NOS Code SSC/N0501	Candidates will be able to: <ul style="list-style-type: none"> • Design basic programming structures to implement functionality in line with requirements defined in BRS/URS, SRS and HLD 	Refer to Unique Equipment Required Section
2	Analysis and Design of Software Applications Theory Duration (hh:mm) 20:00 Practical Duration (hh:mm) 30:00 Corresponding NOS Code SSC/N0501	Candidates will be able to: <ul style="list-style-type: none"> • Check their understanding of the Business Requirements Specification (BRS)/User Requirements Specification (URS) with appropriate people • Check their understanding of the Software Requirements Specification (SRS) with appropriate people • Check their understanding of High Level Design (HLD) with appropriate people • Review their designs with appropriate people • Analyse inputs from appropriate people to identify, resolve and record design defects and inform future designs • Document designs using standard templates and tools • Comply with organization's policies, procedures and guidelines when contributing to the design of software products and applications 	Refer to Unique Equipment Required Section
3	Application Development Theory Duration (hh:mm) 20:00 Practical Duration (hh:mm) 80:00 Corresponding NOS Code SSC/N0502	Candidates will be able to: <ul style="list-style-type: none"> • Check their understanding of the Business Requirements Specification (BRS), Software Requirements Specification (SRS), High Level Design (HLD) and Low Level Design (LLD) with appropriate people • Access reusable components, code generation tools and unit testing tools from their organization's knowledge base • Convert technical specifications into code to meet the requirements, leveraging reusable components, where available • Create appropriate unit test cases (UTCs) • Review codes and UTCs with appropriate people • Execute UTCs and document results • Rework the code and UTCs to fix identified defects 	Refer to Unique Equipment Required Section

Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<ul style="list-style-type: none"> Analyse inputs from appropriate people to inform future designs Record corrective actions for identified defects to inform future designs Submit tested code timely for approval by appropriate people Update their organization's knowledge base with their experiences of the code developed Comply with their organization's policies, procedures and guidelines when developing software code to specification 	
4	<p>Self and work Management</p> <p>Theory Duration (hh:mm) 12:00</p> <p>Practical Duration (hh:mm) 38:00</p> <p>Corresponding NOS Code SSC/N9001</p>	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> Establish and agree their work requirements with appropriate people Keep their immediate work area clean and tidy utilize their time effectively Use resources correctly and efficiently Treat confidential information correctly Work in line with organization's policies and procedures Work within the limits of their job role Obtain guidance from appropriate people, where necessary Ensure their work meets the agreed requirements 	Refer to Unique Equipment Required Section
5	<p>Team Work and Communication</p> <p>Theory Duration (hh:mm) 12:00</p> <p>Practical Duration (hh:mm) 38:00</p> <p>Corresponding NOS Code SSC/N9002</p>	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> Communicate with colleagues clearly, concisely and accurately Work with colleagues to integrate their work effectively with them Pass on essential information to colleagues in line with organizational requirements Work in ways that show respect for colleagues carry out commitments they have made to colleagues Let colleagues know in good time if they cannot carry out their commitments, explaining the reasons Identify any problems they have working with colleagues and take the initiative to solve these problems Follow the organization's policies and procedures for working with colleagues 	Refer to Unique Equipment Required Section
6	<p>Managing Health and Safety</p> <p>Theory Duration (hh:mm) 05:00</p>	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> Comply with their organization's current health, safety and security policies and procedures Report any identified breaches in health, safety, and security policies and procedures to the designated person Identify and correct any hazards that they can deal 	

Sr. No.	Module	Key Learning Outcomes	Equipment Required
	<p>Practical Duration (hh:mm) 20:00</p> <p>Corresponding NOS Code SSC/ N 9003</p>	<p>with safely, competently and within the limits of their authority</p> <ul style="list-style-type: none"> Report any hazards that they are not competent to deal with to the relevant person in line with organizational procedures and warn other people who may be affected Follow their organization's emergency procedures promptly, calmly, and efficiently Identify and recommend opportunities for improving health, safety, and security to the designated person Complete any health and safety 	
7	<p>Data and Information Management</p> <p>Theory Duration (hh:mm) 15:00</p> <p>Practical Duration (hh:mm) 35:00</p> <p>Corresponding NOS Code SSC/N9004</p>	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> Establish and agree with appropriate people the data/information they need to provide, the formats in which they need to provide it, and when they need to provide it Obtain the data/information from reliable sources Check that the data/information is accurate, complete and up-to-date Obtain advice or guidance from appropriate people where there are problems with the data/information Carry out rule-based analysis of the data/information, if required Insert the data/information into the agreed formats Check the accuracy of their work, involving colleagues where required Report any unresolved anomalies in the data/information to appropriate people Provide complete, accurate and up-to-date data/information to the appropriate people in the required formats on time 	Refer to Unique Equipment Required Section
8	<p>Learning and Self Development</p> <p>Theory Duration (hh:mm) 05:00</p> <p>Practical Duration (hh:mm) 20:00</p> <p>Corresponding NOS Code SSC/N9005</p>	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> Obtain advice and guidance from appropriate people to develop their knowledge, skills and competence Identify accurately the knowledge and skills they need for their job role Identify accurately their current level of knowledge, skills and competence and any learning and development needs Agree with appropriate people a plan of learning and development activities to address their learning needs Undertake learning and development activities in line with their plan Apply their new knowledge and skills in the workplace, under supervision Obtain feedback from appropriate people on their 	Refer to Unique Equipment Required Section

Sr. No.	Module	Key Learning Outcomes	Equipment Required
		<p>knowledge and skills and how effectively they apply them</p> <ul style="list-style-type: none"> Review their knowledge, skills and competence regularly and take appropriate action 	
	<p>Total Duration</p> <p>Theory Duration 109:00</p> <p>Practical Duration 291:00</p>	<p>Unique Equipment Required: Training room should be fully furnished with the following equipment / tools / accessories. Additional / specific resources, wherever applicable (e.g. Hardware, software) are indicated in the main text corresponding to relevant learning outcome.</p> <p>For Domain NOSs:</p> <ul style="list-style-type: none"> For NOS SSC/N0501: C/C++, UML tools such as Rational suite For NOS SSC/N0502: JDK / Eclipse <p>General:</p> <ul style="list-style-type: none"> Comfortable seats with adequate lighting, controlled temperature and acoustics for training and learning White Board, Markers and Eraser Projector with screen Flip chart with markers Faculty's PC/Laptop with latest configuration and internet connection Supporting software / applications for projecting audio, video, recording, Presentation Tools to support learning activities: <ul style="list-style-type: none"> Intranet Email IMs Learning management system e.g. Moodle, Blackboard to enable blended learning Microphone / voice system for lecture and class activities Handy Camera Stationery kit – Staples, Glue, Chart Paper, Sketch Pens, Paint Box, Scale, A4 Sheets For IT Lab sessions: Computer Lab with 1:1 PC: trainee ratio and having internet connection, MS Office / Open office, Browser, Outlook/ other Email Clients Assessment and Test Tools for day to day online Tests and Assessments For team discussions: Adequate seating arrangement in full / half circle format for one or more teams as per planned team composition. Reading Resources: Access to relevant sample documents and learning forums to enable self-study before and after each training session. 	

Grand Total Course Duration: 400 Hours 0 Minutes
(This Syllabus/Curriculum has been approved by IT-ITeS Sector Skills Council NASSCOM.)

Notes from IT-ITeS Sector Skills Council

1. This document outlines the broad scope of coverage. This should be linked with OBF and training delivery plan. OBF (Outcome based framework) reflects the pedagogy used to ensure an expected outcome. Training delivery plan focuses on the sequence of delivery.
2. Though many NOSs have some seemingly common outcomes, notably core/generic, professional and technical skills, it is imperative to understand the contextual difference between them. For example, writing skills required to communicate results of testing (in SSC/N0501) are different from the writing skills required to prepare a time plan (in SSC/N9001). Training providers are advised to,
 - a. Embed such skills development in the learning pedagogy for each expected outcome
 - b. Prepare a detailed session plan for training delivery with focus on sequence and duration of training
 - c. Run a diagnostic test to assess prior learning of students and help trainers / students identify the need for gap training, optimal duration, and suitable training methodology. Accordingly, more introductory level sessions may be included in guided or self-paced mode of learning. E.g. adding some sessions on Functional English or Use of Internet and MS Office.

Trainer Prerequisites for Job role: Software Developer mapped to Qualification Pack: SSC/Q0501

Sr. No.	Area	Details
1	Job Description	To deliver accredited training service, mapping to the curriculum detailed above, in accordance with the Qualification Pack SSC/Q0501.
2	Personal Attributes	Aptitude for conducting training, and pre/ post work to ensure competent, employable candidates at the end of the training. Strong communication skills, interpersonal skills, ability to work as part of a team; a passion for quality and for developing others; well-organised and focused, eager to learn and keep oneself updated with the latest in this field.
3	Minimum Educational Qualifications	Minimum Bachelor's Degree in Computer Science or any related field; Preferred Master's Degree in Computer Science
4a	Domain Certification	Minimum accepted score in SSC Assessment is 90% per NOS being taught in QP SSC/Q0501. Certification in relevant software competencies: Software Development Certifications in C++, Embedded, C#, C, Java etc., is an added advantage.
4b	Platform Certification	Recommended that the Trainer is certified for the Job Role: "Trainer" mapped to the Qualification Pack: "SSC/Q1402". Minimum accepted score is 70%.
5	Experience	Field experience: Minimum 2 years' experience in the same domain Training experience: 1 year preferred

Annexure: Assessment Criteria

Assessment Criteria for Software Developer	
Job Role	Software Developer
Qualification Pack	SSC/Q0501
Sector Skill Council	IT-ITeS

Sr. No.	Guidelines for Assessment
1	Criteria for assessment for each Qualification Pack (QP) will be created by the Sector Skill Council (SSC). Each performance criteria (PC) will be assigned Theory and Skill/Practical marks proportional to its importance in NOS.
2	The assessment will be conducted online through assessment providers authorised by SSC.
3	Format of questions will include a variety of styles suitable to the PC being tested such as multiple choice questions, fill in the blanks, situational judgment test, simulation and programming test.
4	To pass a QP, a trainee should pass each individual NOS. Standard passing criteria for each NOS is 70%.
5	For latest details on the assessment criteria, please visit www.sscnasscom.com .

				MARKS ALLOCATION	
ASSESSMENT OUTCOME (NOS CODE AND DESCRIPTION)	ASSESSMENT CRITERIA (PC)	TOTAL MARKS	OUT OF	THEORY	SKILLS PRACTICAL
1.SSC/N0501 (CONTRIBUTE TO THE DESIGN OF SOFTWARE PRODUCTS AND APPLICATIONS)	PC1. check their understanding of the Business Requirements Specification (BRS)/User Requirements Specification (URS) with appropriate people	100	10	10	0
	PC2. check their understanding of the Software Requirements Specification (SRS) with appropriate people		10	10	0
	PC3. check their understanding of High Level Design (HLD) with appropriate people		10	10	0
	PC4. design basic programming structures to implement functionality in line with requirements defined in BRS/URS, SRS and HLD		30	0	30
	PC5. review their designs with appropriate people		5	5	0
	PC6. analyze inputs from appropriate people to identify, resolve and record design defects and inform future designs		15	0	15
	PC7. document their designs using standard templates and tools		10	0	10
	PC8. comply with their organization's policies, procedures and guidelines when contributing to the design of software products and applications		10	0	10
			Total	100	35
2.SSC/N0502 (DEVELOP SOFTWARE CODE TO SPECIFICATION)	PC1. check their understanding of the Business Requirements Specification (BRS), Software Requirements Specification (SRS), High Level Design (HLD) and Low Level Design (LLD) with appropriate people	100	5	5	0
	PC2. access reusable components, code generation tools and unit testing tools from their organization's knowledge base		5	0	5
	PC3. convert technical specifications into code to meet the requirements, leveraging reusable components, where available		30	0	30
	PC4. create appropriate unit test cases (UTCs)		10	0	10
	PC5. review codes and UTCs with appropriate people		5	5	0
	PC6. execute UTCs and document results		5	0	5
	PC7. rework the code and UTCs to fix identified defects		10	0	10
	PC8. analyze inputs from appropriate people to inform future designs		5	5	0
	PC9. record corrective actions for identified defects to inform future designs		10	0	10
			5	5	0

				MARKS ALLOCATION	
ASSESSMENT OUTCOME (NOS CODE AND DESCRIPTION)	ASSESSMENT CRITERIA (PC)	TOTAL MARKS	OUT OF	THEORY	SKILLS PRACTICAL
	PC10. submit tested code for approval by appropriate people				
	PC11. update their organization's knowledge base with their experiences of the code developed		5	0	5
	PC12. comply with their organization's policies, procedures and guidelines when developing software code to specification		5	0	5
	Total		100	20	80
3.NOS/N9001 (MANAGE THEIR WORK TO MEET REQUIREMENTS)	PC1. establish and agree their work requirements with appropriate people	100	6.25	0	6.25
	PC2. keep their immediate work area clean and tidy		12.5	6.25	6.25
	PC3. utilize their time effectively		12.5	6.25	6.25
	PC4. use resources correctly and efficiently		18.75	6.25	12.5
	PC5. treat confidential information correctly		6.25	0	6.25
	PC6. work in line with their organization's policies and procedures		12.5	0	12.5
	PC7. work within the limits of their job role		6.25	0	6.25
	PC8. obtain guidance from appropriate people, where necessary		6.25	0	6.25
	PC9. ensure their work meets the agreed requirements		18.75	6.25	12.5
	Total	100	100	25	75
4.SSC/N9002 (WORK EFFECTIVELY WITH COLLEAGUES)	PC1. communicate with colleagues clearly, concisely and accurately	100	20	0	20
	PC2. work with colleagues to integrate their work effectively with theirs		10	0	10
	PC3. pass on essential information to colleagues in line with organizational requirements		10	10	0
	PC4. work in ways that show respect for colleagues		20	0	20
	PC5. carry out commitments you have made to colleagues		10	0	10
	PC6. let colleagues know in good time if you cannot carry out their commitments, explaining the reasons		10	10	0
	PC7. identify any problems you have working with colleagues and take the initiative to solve these problems		10	0	10
	PC8. follow the organization's policies and procedures for working with colleagues		10	0	10
		Total	100	100	20
5.SSC/N9003 (MAINTAIN A	PC1. comply with their organization's current health, safety and security policies and procedures	100	20	10	10

				MARKS ALLOCATION	
ASSESSMENT OUTCOME (NOS CODE AND DESCRIPTION)	ASSESSMENT CRITERIA (PC)	TOTAL MARKS	OUT OF	THEORY	SKILLS PRACTICAL
HEALTHY, SAFE AND SECURE WORKING ENVIRONMENT)	PC2. report any identified breaches in health, safety, and security policies and procedures to the designated person		10	0	10
	PC3. identify and correct any hazards that you can deal with safely, competently and within the limits of their authority		20	10	10
	PC4. report any hazards that you are not competent to deal with to the relevant person in line with organizational procedures and warn other people who may be affected		10	0	10
	PC5. follow their organization's emergency procedures promptly, calmly, and efficiently		20	10	10
	PC6. identify and recommend opportunities for improving health, safety, and security to the designated person		10	0	10
	PC7. complete any health and safety records legibly and accurately		10	0	10
		Total	100	30	70
6.SSC/N9004 (PROVIDE DATA/INFORMATION IN STANDARD FORMATS)	PC1. establish and agree with appropriate people the data/information you need to provide, the formats in which you need to provide it, and when you need to provide it	100	12.5	12.5	0
	PC2. obtain the data/information from reliable sources		12.5	0	12.5
	PC3. check that the data/information is accurate, complete and up-to-date		12.5	6.25	6.25
	PC4. obtain advice or guidance from appropriate people where there are problems with the data/information		6.25	0	6.25
	PC5. carry out rule-based analysis of the data/information, if required		25	0	25
	PC6. insert the data/information into the agreed formats		12.5	0	12.5
	PC7. check the accuracy of their work, involving colleagues where required		6.25	0	6.25
	PC8. report any unresolved anomalies in the data/information to appropriate people		6.25	6.25	0
	PC9. provide complete, accurate and up-to-date data/information to the appropriate people in the required formats on time		6.25	0	6.25
		Total	100	25	75
7.SSC/N9005 (DEVELOP THEIR KNOWLEDGE, SKILLS AND COMPETENCE)	PC1. obtain advice and guidance from appropriate people to develop their knowledge, skills and competence	100	10	0	10
	PC2. identify accurately the knowledge and skills you need for their job role		10	0	10
	PC3. identify accurately their current level of		20	10	10

				MARKS ALLOCATION	
ASSESSMENT OUTCOME (NOS CODE AND DESCRIPTION)	ASSESSMENT CRITERIA (PC)	TOTAL MARKS	OUT OF	THEORY	SKILLS PRACTICAL
	knowledge, skills and competence and any learning and development needs				
	PC4. agree with appropriate people a plan of learning and development activities to address their learning needs		10	0	10
	PC5. undertake learning and development activities in line with their plan		20	10	10
	PC6. apply their new knowledge and skills in the workplace, under supervision		10	0	10
	PC7. obtain feedback from appropriate people on their knowledge and skills and how effectively you apply them		10	0	10
	PC8. review their knowledge, skills and competence regularly and take appropriate action		10	0	10
		Total	100	20	80



IT-ITeS Sector Skill Council

4E-Vandana Building (4th Floor) 11, Tolstoy Marg, New Delhi-110001

Model Curriculum

User Interface (UI) Developer

User Interface (UI) Developer

SECTOR: IT-ITeS
SUB-SECTOR: IT Services
OCCUPATION: **Application Development**
REFERENCE ID: **SSC/Q0502**
NSQF LEVEL: **7**

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User Interface (UI) Developer

Curriculum / Syllabus

This program is aimed at training candidates for the job of a **User Interface (UI) Developer** in the **IT-ITeS Sector/Industry** and aims at building the following key competencies amongst the learner.

Program Name	User Interface (UI) Developer		
Qualification Pack Name & Reference ID.	User Interface (UI) Developer SSC/Q0502		
Version No.	1.0	Version Update Date	31/01/2015
Pre-requisites to Training	Bachelor's Degree in Science/Technology/Computers or any graduate course		
Training Outcomes	<p>After completing this programme, participants will be able to:</p> <ul style="list-style-type: none"> • Contribute to the design of software products and applications • Develop software code to specification • Develop media content and graphic designs for software products and applications • Manage your work to meet requirements • Work effectively with colleagues • Maintain a healthy, safe and secure working environment • Provide data/information in standard formats • Develop your knowledge, skills and competence 		

This course encompasses all Eight National Occupational Standards (NOS) of **User Interface (UI) Developer** Qualification Pack issued by **IT-ITeS Sector Skills Council NASSCOM**.

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
1.	Contribute to the design of software products and applications	17:00	33:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> • check your understanding of the Business Requirements Specification (BRS)/User Requirements Specification (URS) with appropriate people • check your understanding of the Software Requirements Specification (SRS) with appropriate people • check your understanding of High Level Design (HLD) with appropriate people • design basic programming structures to implement functionality in line with requirements defined in BRS/URS, SRS and HLD • review your designs with appropriate people 	SSC/N0501	Refer to Unique Equipment Required

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<ul style="list-style-type: none"> analyze inputs from appropriate people to identify, resolve and record design defects and inform future designs document your designs using standard templates and tools comply with your organization's policies, procedures and guidelines when contributing to the design of software products and applications 		
2.	Develop software code to specification	20:00	80:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> check your understanding of the Business Requirements Specification (BRS), Software Requirements Specification (SRS), High Level Design (HLD) and Low Level Design (LLD) with appropriate people access reusable components, code generation tools and unit testing tools from your organization's knowledge base convert technical specifications into code to meet the requirements, leveraging reusable components, where available create appropriate unit test cases (UTCs) review codes and UTCs with appropriate people execute UTCs and document results rework the code and UTCs to fix identified defects analyze inputs from appropriate people to inform future designs record corrective actions for identified defects to inform future designs submit tested code for approval by appropriate people update your organization's knowledge base with your experiences of the code developed comply with your organization's policies, procedures and guidelines when developing software code to specification 	SSC/N0502	

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
3.	Develop media content and graphic designs for software products and applications	12:00	38:00	Candidates will be able to: <ul style="list-style-type: none"> • check your understanding of the Business Requirements Specification (BRS), Software Requirements Specification (SRS), High Level Design (HLD) and Low Level Design (LLD) with appropriate people • access reusable components, media and graphical packages and tools from your organization’s knowledge base • convert requirements into media content and graphic designs, leveraging reusable components where available • review media content and graphic designs with appropriate people and analyze their feedback • record any defects and corrective actions taken to inform future work • rework media content and graphic designs, incorporating feedback • submit media content and graphic designs for approval by appropriate people • update your organization’s knowledge base with your experiences of the media content and graphic designs developed • comply with your organization’s policies, procedures and guidelines when developing media content and graphic designs for software products and applications 	SSC/N0503	
4.	Manage your work to meet requirements	12:00	38:00	Candidates will be able to: <ul style="list-style-type: none"> • establish and agree your work requirements with appropriate people • keep your immediate work area clean and tidy • utilize your time effectively • use resources correctly and efficiently • treat confidential information correctly • work in line with your organization’s policies and procedures • work within the limits of your job role • obtain guidance from appropriate people, where necessary • ensure your work meets the agreed requirements 	SSC/N9001	

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
5.	Work effectively with colleagues	10:00	40:00	Candidates will be able to: <ul style="list-style-type: none"> • communicate with colleagues clearly, concisely and accurately • work with colleagues to integrate your work effectively with theirs • pass on essential information to colleagues in line with organizational requirements • work in ways that show respect for colleagues • carry out commitments you have made to colleagues • let colleagues know in good time if you cannot carry out your commitments, explaining the reasons • identify any problems you have working with colleagues and take the initiative to solve these problems • follow the organization's policies and procedures for working with colleagues 	SSC/N9002	
6.	Maintain a healthy, safe and secure working environment	7:00	18:00	Candidates will be able to: <ul style="list-style-type: none"> • comply with your organization's current health, safety and security policies and procedures • report any identified breaches in health, safety, and security policies and procedures to the designated person • identify and correct any hazards that you can deal with safely, competently and within the limits of your authority • report any hazards that you are not competent to deal with to the relevant person in line with organizational procedures and warn other people who may be affected • follow your organization's emergency procedures promptly, calmly, and efficiently • identify and recommend opportunities for improving health, safety, and security to the designated person 	SSC/N9003	

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<ul style="list-style-type: none"> complete any health and safety records legibly and accurately 		
7.	Provide data/information in standard formats	12:00	38:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> establish and agree with appropriate people the data/information you need to provide, the formats in which you need to provide it, and when you need to provide it obtain the data/information from reliable sources check that the data/information is accurate, complete and up-to-date obtain advice or guidance from appropriate people where there are problems with the data/information carry out rule-based analysis of the data/information, if required insert the data/information into the agreed formats check the accuracy of your work, involving colleagues where required report any unresolved anomalies in the data/information to appropriate people provide complete, accurate and up-to-date data/information to the appropriate people in the required formats on time 	SSC/N9004	
8.	Develop your knowledge, skills and competence	5:00	20:00	<p>Candidates will be able to:</p> <ul style="list-style-type: none"> obtain advice and guidance from appropriate people to develop your knowledge, skills and competence identify accurately the knowledge and skills you need for your job role identify accurately your current level of knowledge, skills and competence and any learning and development needs agree with appropriate people a plan of learning and development activities to address your learning needs undertake learning and development activities in line with your plan apply your new knowledge and skills in the workplace, under supervision obtain feedback from appropriate people on your knowledge and skills and how effectively you apply them 	SSC/N9005	

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<ul style="list-style-type: none"> review your knowledge, skills and competence regularly and take appropriate action 		
	Total Duration:	<u>95:00</u>	<u>305:00</u>	<p>Unique Equipment Required: Training room should be fully furnished with the following equipment / tools / accessories. Additional / specific resources, wherever applicable (e.g. Hardware, software) are indicated in the main text corresponding to relevant learning outcome.</p> <p>Domain NOS requirements</p> <ul style="list-style-type: none"> Visio, UML, freeminds, mockingbird HTML 5, CSS, Java Script and SQL IDEs such as Web Builder, Word Press, Joomla Wordpress, psdGraphics etc. HTML, CSS, Flash, Photoshop, Windows media player, Eclipse, XAMPP <p>Common requirements</p> <ul style="list-style-type: none"> Comfortable seats with adequate lighting, controlled temperature and acoustics for training and learning White Board, Markers and Eraser Projector with screen Flip chart with markers Faculty's PC/Laptop with latest configuration and internet connection Supporting software / applications for projecting audio, video, recording, Presentation Tools to support learning activities: <ul style="list-style-type: none"> Intranet Email IMs Learning management system e.g. Moodle, Blackboard to enable blended learning Microphone / voice system for lecture and class activities Handy Camera Stationery kit – Staples, Glue, Chart Paper, Sketch Pens, Paint Box, Scale, A4 Sheets For IT Lab sessions: Computer Lab with 1:1 PC:trainee ratio and having internet connection, MS Office / Open office, Browser, Outlook / Any other Email Client and chat tools. Assessment and Test Tools for day to day online Tests and Assessments For team discussions: Adequate seating arrangement in full / half circle format for one or more teams as per planned team composition. 		

Sr. No.	Module	Theory Duration (hh:mm)	Practical Duration (hh:mm)	Key Learning Outcomes	Corresponding NOS Code	Equipment Required
				<ul style="list-style-type: none"> Reading Resources: Access to relevant sample documents and learning forums to enable self-study before and after each training session. 		

Grand Total Course Duration: **400 Hours 0 Minutes**

(This syllabus/ curriculum has been approved IT-ITeS Sector Skills Council NASSCOM.)

Notes from IT-ITeS Sector Skills Council NASSCOM

1. This document outlines the broad scope of coverage. This should be linked with OBF and training delivery plan. OBF (Outcome based framework) reflects the pedagogy used to ensure an expected outcome. Training delivery plan focuses on the sequence of delivery.
2. Though many NOSs have some seemingly common outcomes, notably core/generic, professional and technical skills, it is imperative to understand the contextual difference between them. Training providers are advised to,
 - a. Embed such skills development in the learning pedagogy for each expected outcome
 - b. Prepare a detailed session plan for training delivery with focus on sequence and duration of training
3. Run a diagnostic test to assess prior learning of students and help trainers / students identify the need for gap training and suitable training methodology. Accordingly, more introductory level sessions may be included in guided or self-paced mode of learning. E.g. adding some sessions on Functional English or Use of Internet and MS Office.

**Annexure1: Assessment Criteria**

Assessment Criteria for <QP Name>	
Job Role	User Interface (UI) Developer
Qualification Pack	SSC/Q0502
Sector Skill Council	IT-ITeS

Sr. No.	Guidelines for Assessment
1	Criteria for assessment for each Qualification Pack (QP) will be created by the Sector Skill Council (SSC). Each performance criteria (PC) will be assigned Theory and Skill/Practical marks proportional to its importance in NOS.
2	The assessment will be conducted online through assessment providers authorised by SSC.
3	Format of questions will include a variety of styles suitable to the PC being tested such as multiple choice questions, fill in the blanks, situational judgment test, simulation and programming test.
4	To pass a QP, a trainee should pass each individual NOS. Standard passing criteria for each NOS is 70%.
5	For latest details on the assessment criteria, please visit www.sscnasscom.com .

Assessable Outcomes	Assessment criteria for the outcome	Total Mark	Out of	Theory	Skills Practical
1. SSC/N0501 (Contribute to the design of software products and applications)	PC1. check your understanding of the Business Requirements Specification (BRS)/User Requirements Specification (URS) with appropriate people	100	10	10	0
	PC2. check your understanding of the Software Requirements Specification (SRS) with appropriate people		10	10	0
	PC3. check your understanding of High Level Design (HLD) with appropriate people		10	10	0
	PC4. design basic programming structures to implement functionality in line with requirements defined in BRS/URS, SRS and HLD		30	0	30
	PC5. review your designs with appropriate people		5	5	0
	PC6. analyze inputs from appropriate people to identify, resolve and record design defects and inform future designs		15	0	15
	PC7. document your designs using standard templates and tools		10	0	10
	PC8. comply with your organization's policies, procedures and guidelines when contributing to the design of software products and applications		10	0	10
	Total		100	35	65
2. SSC/N0502 (Develop software code to specification)	PC1. check your understanding of the Business Requirements Specification (BRS), Software Requirements Specification (SRS), High Level Design (HLD) and Low Level Design (LLD) with appropriate people	100	5	5	0
	PC2. access reusable components, code generation tools and unit testing tools from your organization's knowledge base		10	0	10

Model Curriculum for **User Interface (UI) Developer**

Assessable Outcomes	Assessment criteria for the outcome	Total Mark	Out of	Theory	Skills Practical	
	PC3. convert technical specifications into code to meet the requirements, leveraging reusable components, where available		10	0	10	
	PC4. create appropriate unit test cases (UTCs)		10	0	10	
	PC5. review codes and UTCs with appropriate people		5	5	0	
	PC6. execute UTCs and document results		10	0	10	
	PC7. rework the code and UTCs to fix identified defects		10	0	10	
	PC8. analyze inputs from appropriate people to inform future designs		5	5	0	
	PC9. record corrective actions for identified defects to inform future designs		10	0	10	
	PC10. submit tested code for approval by appropriate people		5	5	0	
	PC11. update your organization’s knowledge base with your experiences of the code developed		10	0	10	
	PC12. comply with your organization’s policies, procedures and guidelines when developing software code to specification		10	0	10	
			Total	100	20	80
	3. SSC/N0503 (Develop media content and graphic designs for software products and Applications)		PC1. check your understanding of the Business Requirements Specification (BRS), Software Requirements Specification (SRS), High Level Design (HLD) and Low Level Design (LLD) with appropriate people	100	10	10
PC2. access reusable components, media and graphical packages and tools from your organization’s knowledge base		10	0		10	
PC3. convert requirements into media content and graphic designs, leveraging reusable components where available		25	0		25	
	PC4. review media content and graphic designs with appropriate people and analyze their feedback		10	10	0	
	PC5. record any defects and corrective actions taken to inform future work		10	0	10	
	PC6. rework media content and graphic designs, incorporating feedback		10	0	10	
	PC7. submit media content and graphic designs for approval by appropriate people		5	5	0	
	PC8. update your organization’s knowledge base with your experiences of the media content and graphic designs developed		10	0	10	
	PC9. comply with your organization’s policies, procedures and guidelines when developing media content and graphic designs for software products and applications		10	0	10	
	Total	100	25	75		
4.SSC/N9001 (Manage your work to meet requirements)	PC1. establish and agree your work requirements with appropriate people	100	6.25	0	6.25	
	PC2. keep your immediate work area clean and tidy		12.5	6.25	6.25	
	PC3. utilize your time effectively		12.5	6.25	6.25	
	PC4. use resources correctly and efficiently		18.75	6.25	12.5	

Assessable Outcomes	Assessment criteria for the outcome	Total Mark	Out of	Theory	Skills Practical
	PC5. treat confidential information correctly		6.25	0	6.25
	PC6. work in line with your organization's policies and procedures		12.5	0	12.5
	PC7. work within the limits of your job role		6.25	0	6.25
	PC8. obtain guidance from appropriate people , where necessary		6.25	0	6.25
	PC9. ensure your work meets the agreed requirements		18.75	6.25	12.5
		Total	100	25	75
5.SSC/N9002 (Work with colleagues)	PC1. communicate with colleagues clearly, concisely and accurately	100	20	0	20
	PC2. work with colleagues to integrate your work effectively with theirs		10	0	10
	PC3. pass on essential information to colleagues in line with organizational requirements		10	10	0
	PC4. work in ways that show respect for colleagues		20	0	20
	PC5. carry out commitments you have made to colleagues		10	0	10
	PC6. let colleagues know in good time if you cannot carry out your commitments, explaining the reasons		10	10	0
	PC7. identify any problems you have working with colleagues and take the initiative to solve these problems		10	0	10
	PC8. follow the organization's policies and procedures for working with colleagues		10	0	10
		Total	100	20	80
6.SSC/N9003 (Maintain a healthy, safe and secure working environment)	PC1. comply with your organization's current health, safety and security policies and procedures	100	20	10	10
	PC2. report any identified breaches in health, safety, and security policies and procedures to the designated person		10	0	10
	PC3. identify and correct any hazards that you can deal with safely, competently and within the limits of your authority		20	10	10
	PC4. report any hazards that you are not competent to deal with to the relevant person in line with organizational procedures and warn other people who may be affected		10	0	10
	PC5. follow your organization's emergency procedures promptly, calmly, and efficiently		20	10	10
	PC6. identify and recommend opportunities for improving health, safety, and security to the designated person		10	0	10
	PC7. complete any health and safety records legibly and accurately		10	0	10
		Total	100	30	70
	PC1. establish and agree with appropriate people the data/information you need to provide, the formats	100	12.5	12.5	0

Assessable Outcomes	Assessment criteria for the outcome	Total Mark	Out of	Theory	Skills Practical
7.SSC/N9004 (Provide data/information in standard formats)	in which you need to provide it, and when you need to provide it				
	PC2. obtain the data/information from reliable sources		12.5	0	12.5
	PC3. check that the data/information is accurate, complete and up-to-date		12.5	6.25	6.25
	PC4. obtain advice or guidance from appropriate people where there are problems with the data/information		6.25	0	6.25
	PC5. carry out rule-based analysis of the data/information, if required		25	0	25
	PC6. insert the data/information into the agreed formats		12.5	0	12.5
	PC7. check the accuracy of your work, involving colleagues where required		6.25	0	6.25
	PC8. report any unresolved anomalies in the data/information to appropriate people		6.25	6.25	0
	PC9. provide complete, accurate and up-to-date data/information to the appropriate people in the required formats on time		6.25	0	6.25
		Total	100	25	75
8.SSC/N9005 (Develop your knowledge, skills and competence)	PC1. obtain advice and guidance from appropriate people to develop your knowledge, skills and competence	100	10	0	10
	PC2. identify accurately the knowledge and skills you need for your job role		10	0	10
	PC3. identify accurately your current level of knowledge, skills and competence and any learning and development needs		20	10	10
	PC4. agree with appropriate people a plan of learning and development activities to address your learning needs		10	0	10
	PC5. undertake learning and development activities in line with your plan		20	10	10
	PC6. apply your new knowledge and skills in the workplace, under supervision		10	0	10
	PC7. obtain feedback from appropriate people on your knowledge and skills and how effectively you apply them		10	0	10
	PC8. review your knowledge, skills and competence regularly and take appropriate action		10	0	10
		Total	100	20	80



Annexure2: Trainer Prerequisites for Job role: User Interface (UI) Developer mapped to Qualification Pack: SSC/Q0502

Sr. No.	Area	Details
1	Job Description	To deliver accredited training service, mapping to the curriculum detailed above, in accordance with the Qualification Pack SSC/Q0502.
2	Personal Attributes	<p>Aptitude to conduct training, and pre/ post work to ensure competent, employable candidates at the end of the training. Strong communication skills, interpersonal skills, ability to work as part of a team; a passion for quality and for developing others; well-organised and focused, eager to learn and keep oneself updated with the latest in the mentioned field.</p> <p>The individual should be result oriented. The individual should also be able to demonstrate skills for communication, creative and logical thinking.</p>
3	Minimum Educational Qualifications	Bachelor's Degree in Science/Technology/Computers or any graduate course
4a	Domain Certification	<p>Minimum accepted score in SSC Assessment is 90% per NOS being taught in SSC/Q0502.</p> <p>Additional certification in computers/technology/ animation/graphics</p>
4b	Platform Certification	<p>Recommended that the Trainer is certified for the Job Role: "Trainer" mapped to the Qualification Pack: "SSC/Q1402".</p> <p>Minimum accepted score is 70% per NOS.</p>
5	Experience	<p>Field experience: Minimum 2 years' experience in the same domain</p> <p>Training experience: 1 year preferred</p>



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JSS COLLEGE OF ARTS, COMMERCE & SCIENCE

(An Autonomous College of University of Mysore)

Re-accredited by NAAC with 'A' grade

OOTY ROAD, MYSORE-570 025, KARNATAKA



ESTD-1964

SYLLABUS

B. Voc. (Food Processing & Engineering)

2018 - 2019

**DEPARTMENT OF FOOD PROCESSING
& ENGINEERING**

Scheme of Instruction For B. Voc. (Food Processing & Engineering)

General Education Component

(L-Lecture; T-Tutorial; P-Practical/Practice) (1 Credit = 15 Hrs)

Semesters	Paper No.	Title	L:T:P	Theory Hours	Tutorial	Practical Hours	Total Hours	Total Credits
					Hours			
Sem I	FPA 020	Communication Language Kannada	3:0:0	45	0	0	45	3
	FPA 510	Biochemistry-I	2:0:1	30	0	15	45	3
	FPA 520	Microbiology-I	2:0:1	30	0	15	45	3
	FPA 540	Computer Fundamentals & DOS	2:0:1	30	0	15	45	3
Sem II	FPB540	Communication Language English	3:0:0	45	0	0	45	3
	FPB 510	Biochemistry-II	2:0:1	30	0	15	45	3
	FPB 520	Microbiology-II	2:0:1	30	0	15	45	3
	FPB 550	Computer C Programming	2:0:1	30	0	15	45	3
Sem III	FPC 550	Indian Constitution	3:0:0	45	0	0	45	3
	FPD 580	Bio statistics	2:0:1	30	0	15	45	3
	FPC 510	Biochemistry-III	2:0:1	30	0	15	45	3
	FPC 520	Microbiology-III	2:0:1	30	0	15	45	3
Sem IV	FPC 570	ICT	2:0:1	30	0	15	45	3
	FPD 510	Biochemistry-IV	2:0:1	30	0	15	45	3
	FPD 520	Microbiology-IV	2:0:1	30	0	15	45	3
	FPD 560	Environmental Studies	3:0:0	45	0	0	45	3
Sem V	FPE 510	Biochemistry-V	4:0:2	60	0	30	90	6
	FPE 520	Microbiology-V	4:0:2	60	0	30	90	6
Sem VI	FPF 510	Biochemistry-V	4:0:2	60	0	30	90	6
	FPF 520	Microbiology-V	4:0:2	60	0	30	90	6
								12

**DEPARTMENT OF BIOCHEMISTRY
SCHEME OF INSTRUCTION**

Scheme of Instruction For B.Voc- Food Processing and Engineering 2018-19									
General Education Component									
NSQF/ NVE QF Level	Vocational Qualification	Semester	Title	L:T:P	Theory Hours	Tutorial Hours	Practical Hours	Total Hours	Credits
Level- IV	Diploma	Semester- I	Biomolecule	2:0:1	30	0	15	45	3
Level- V		Semester-II	Enzymology and Bioenergetics	2:0:1	30	0	15	45	3
Level- VI	Advanced Diploma	Semester-III	Metabolism	2:0:1	30	0	15	45	3
		Semester-IV	Biochemical Techniques	2:0:1	30	0	15	45	3
Level- VII	Degree	Semester-V	Food and Nutrition	4:0:2	60	0	30	90	6
		Semester-VI	Applied Biochemistry	4:0:2	60	0	30	90	6

Proforma of assessment For B.Voc- Food Processing and Engineering 2018-19
General Education Component-Biochemistry

NSQF/ NVE QF Level	Vocational Qualificati on	Semester	Title	Theory				Practical				Credits
				Exam		C-1	C-2	Exam	C-1	C-2		
				Code	Marks	Marks	Marks	Code	Marks	Marks	Marks	
Level- IV	Diploma	Semester- I	Biomolecule	FPA510	70	10	10	FPA530	70	05	05	3
Level- V		Semester-II	Enzymology and Bioenergetics	FPB510	70	10	10	FPB530	70	05	05	3
Level- VI	Advanced Diploma	Semester-III	Metabolism	FPC510	70	10	10	FPC530	70	05	05	3
		Semester-IV	Biochemical Techniques	FPD510	70	10	10	FPD530	70	05	05	3
Level- VII	Degree	Semester-V	Food & Nutrition	FPE510	70	10	10	FPE530	70	05	05	6
		Semester-VI	Applied Biochemistry	FPF510	70	10	10	FPF530	70	05	05	6

SCHEME OF VALUATION FOR PRACTICAL EXAMINATION

- ✓ A candidate appearing for the first time should submit a duly signed and certified practical record
- ✓ Each candidate has to perform one experiment in the specified duration of three hours for **FORTY FIVE marks**
- ✓ Practical record has to be valued for **TEN marks** by examiners at the time of examination
- ✓ Viva-voce for **FIFTEEN marks** in practical is awarded by continuous assessment in the lab

Sl no	Component	Marks
1	Write up of the experiment	15
2	Conducting experiment	25
3	Result	05
4	Viva-voce	15
5	Practical record	10
TOTAL		70

FPA510

SEMESTER I

Paper –I BIOMOLECULES

(30 hours -2 Credits)

Course Outcome:

After completion of the course a student is able to

CO1: Explain the structure and properties of carbohydrates, lipids, proteins, and nucleic acids in living organisms

CO2: Describe the structure and functions of RNA and DNA

CO3: Understand the role of vitamins in our body

CO4: Understand the behaviour of proteins in solutions

SECTION-I : Amino Acids & Proteins:

09 hrs

1. Introduction to Bio-chemistry.
2. Definition, classification and biological functions of Amino acids and proteins.
3. **Amino Acids:** Common structural features. Stereoisomerism and RS system of designating optical isomers. Classification based on the nature of “R” groups. Amino acids present in proteins and non-protein amino acids. Specialized role of amino acids. Physical and Chemical properties of amino acids. Titration of amino acids.
4. **Proteins:** Levels of protein structure. Forces stabilising structure and shape of proteins. Native proteins and their conformations. Behavior of proteins in solutions. Salting in & salting out of proteins. Denaturation of proteins.

SECTION-II: Carbohydrates:

08 hrs

1. Definition, classification and biological functions of carbohydrates.
2. Monosaccharides-Fischer and Haworth structures of monosaccharides,. Derivatives of monosaccharides (Sugar acids, deoxysugars, amino sugars and other derivatives of biological importance). Oligosaccharides (structure of maltose, lactose, sucrose, cellobiose, and trehalose).
3. Homo-and hetero-polysaccharides (structures of starch, inulin, glycogen, cellulose, chitin). Polysaccharides of bacterial cell wall.

SECTION-III: Lipids

08 hrs

1. Definition and classification of lipids, fatty acids (saturated and unsaturated). Essential fatty acids. m. p., b. p. and their relation to molecular size. Fats as source of energy. Waxes.
2. Structures and functions of lipids : Triacylglycerols, phospholipids : lecithins (PhosphotidylCholines),

cephalins (Phosphatidylethanolamines), Phosphatidylserines, phosphatidyl inositol, sphingomyelins, plasmalogens), cerebrosides, gangliosides.

3. Lipoproteins—Composition, classification and biological functions. Liposomes.
4. Prostaglandins, Leukotrienes, Thromboxanes and their importance.
5. Sources and biochemical functions of water & fat soluble vitamins.

SECTION-IV : Nucleic Acid

05 hrs

1. Nucleic Acids: Structure and properties of purine and pyrimidine bases. Nucleosides and nucleotides. Biologically important nucleotides. Double helical model of DNA. Denaturation of DNA. Physical and chemical properties of nucleic acids.

FPA530

PRACTICALS

(15 Hrs 1 Credits)

1. Qualitative tests for : (a) Carbohydrates. (b) Amino acids and proteins (c) Cholesterol and lipids
3 Hrs X 2
2. Estimation of amino acid by formal titration **3 Hrs**
3. Estimation of ascorbic acid by dye method. **3 Hrs**
4. Estimation of reducing sugars by DNS method **3 Hrs**

FPB510

Semester II

Paper II

(30 hours – 2 credits)

ENZYMOLGY AND BIOENERGETICS

Course Outcome:

After completion of the course a student is able to;

CO1: Describe the role and functions of enzymes.

CO2: Understand the role of cofactor and coenzymes in enzyme catalysed reactions.

CO3: Describe the applications of enzymes in diagnostics.

CO4: Discuss the commercial importance of enzymes.

SECTION-I : Enzymology

08 hrs

Introduction to enzymes - General characteristics of enzymes. Prosthetic group. Holoenzymes, apoenzyme and cofactors. Coenzymes and their biochemical functions, assay of enzyme activity, units of enzyme activity. Active sites(s) of enzymes. IUB system of nomenclature and classification of enzymes. Enzymes as catalysts. Theories of enzymes catalysis: Acidbase catalysis, covalent catalysis.

SECTION-II: Enzyme Purification :

08 hrs

Need for purification. Preliminary fractionation procedures and precipitation techniques, Chromatography methods : Gel filtration, adsorption–, ion exchange– and affinity chromatography. Types of support materials. Selection of appropriate conditions and elution procedures. Criteria of enzyme purity.

SECTION-III : Enzyme Kinetics

09 hrs

Factors affecting velocity of enzyme catalysed reactions: Enzyme concentration, pH and temperature. Michaelis –Menten equation. Determination of K_m and its significance. Enzyme inhibition. Various types of enzyme inhibitions. Enzyme inhibitors and their importance. Allosteric enzymes and enzyme regulation. Isoenzymes and their clinical significance.

SECTION-IV: Bioenergetics

05 hrs

Biological systems and concept of free energy, Endergonic processes and role of ATP & other high energy compounds. Biological oxidations. Redox potential. Commercial importance of enzymes

FPB530

PRACTICALS (15 Hrs – 1 Credit)

1. Assay of salivary amylase enzyme activity. **3 Hrs**
2. Effect of pH on enzyme activity. **3 Hrs**
3. Effect of temperature on enzyme activity **3 Hrs**
4. Effect of substrate concentration on enzyme activity and determination of K_m and V_{max} **3 Hrs**
5. Effect of activators (NaCl) on salivary enzyme activity **3 Hrs**

METABOLISM**Course Outcome:**

After completion of the course a student is able to;

CO1: Gain the knowledge on cellular metabolism and their regulations

CO2: Able to describe the linked pathways of metabolism

CO3: Demonstrate an understanding of the metabolic pathways - the energy-yielding and energy-requiring reactions in life.

CO4: Understand the role of enzymes in metabolic reactions.

SECTION-I**Carbohydrate Metabolism****08****hrs**

Digestion, Absorption and transport of Carbohydrates, Metabolic Pathways, Glycolysis, Pentose Phosphate Pathway, Glucuronate and glyoxylate pathway, TCA cycle, Glycogenolysis& Glycogenesis, Gluconeogenesis. Diseases associated with carbohydrate metabolism.

SECTION-II**Lipid Metabolism****08****hrs**

Digestion, absorption & Transport of Lipids, β -Oxidation of fatty acids. α and ω oxidation of fatty acids Degradation of triglycerides and phospholipids. Formation and utilization of ketone bodies. Biosynthesis of saturated fatty acids, Biosynthesis of Cholesterol. Diseases associated with Lipid metabolism.

SECTION-III**08 hrs****Protein Metabolism**

Digestion, absorption & transport of Proteins, General Reactions of Amino Acids : Deamination, transamination and decarboxylation. Urea cycle and its significance. Ketogenic and glucogenic amino acids. Biosynthesis of amino acids (Phenyl alanine and Glutamic acid) and their degradation. Diseases associated with Proteins metabolism.

SECTION-IV**06 hrs****Nucleic acid Metabolism**

Degradation of purines and pyrimidines. Biosynthesis of purines, pyrimidines and nucleotides. Catabolism of Heme & Formation of Bile pigments. Diseases associated with Nucleic acid metabolism.

- | | | |
|----|---|--------------|
| 1. | Estimation of protein by Biuret method. | 3 Hrs |
| 2. | Estimation of Ca ⁺ in serum. | 3 Hrs |
| 3. | Estimation of iron in drumstics | 3 Hrs |
| 4. | Estimation of creatinine in serum. | 3 Hrs |

5. Estimation of uric acid in urine

3 Hrs

BIOCHEMICAL TECHNIQUES**Course Outcome:**

After completion of the course a student is able to;

CO1: Use selected analytical techniques for the separation of biomolecules.

CO2: Differentiate certain functionalities of biomolecules by using spectroscopic techniques.

CO3: Understand the intersection of life and information sciences, using different sequencing and mapping like SDS-PAGE, TLC, GLC and Chromatography.

CO4: Explain the dangers and safety precautions associated with x-rays and identify the various isotopes used in radiography.

SECTION-I**08 hrs****Spectroscopic Techniques :**

Beer-Lambert's Law. Light absorption and its transmittance. Determination and application of extinction coefficient. Principle and Applications of following spectroscopic techniques - U.V.-Visible, infra-red, Fluorescent emission and NMR spectroscopy.

SECTION-II**08 hrs****Electrophoretic Techniques :**

Principles and applications of the following electrophoresis techniques. Paper electrophoresis, PAGE, SDS- PAGE.

Centrifugation Techniques :

Principle of differential and density gradient centrifugation. Ultra centrifuge – construction and applications

SECTION-III**08 hrs****Chromatographic Techniques:**

Principles of Adsorption and Partition chromatography. Techniques of ascending, descending, circular paper chromatography.

Thin Layer Chromatography-Technique and advantages over paper chromatography

Column chromatography – Principle and applications of Gel Filtration chromatography, ion – exchange chromatography.

SECTION-IV**06 hrs**

Radio Isotopic Techniques :

Properties of radioactive emissions. Units of radioactivity. Isotopes and their applications in biological studies - ^3H , ^{14}C , ^{131}I , ^{60}CO , and ^{32}P . Techniques used to measure radioactivity- GM counter. Biological hazards of radiation and safety measures in handling radioisotopes.

FPD530**PRACTICAL****(15 Hrs = 1 Credits)**

- | | |
|--|--------------|
| 1. Identification of amino acids by circular paper chromatography. | 3 Hrs |
| 2. Identification of amino acids by ascending Paper chromatography | 3 Hrs |
| 3. Separation of phospholipids by thin layer chromatography. | 3 Hrs |
| 4. Separation of leaf pigments by column chromatography. | 3 Hrs |
| 5. Separation of proteins by PAGE. | 3 Hrs |

FPE510

Semester V

PaperV

(60 hours-4 Credits)

FOOD AND NUTRITION

Course Outcome:

After completion of the course a student is able to;

CO1: Explain the theoretical and practical uses on micro and macronutrients.

CO2: Describe the role of electrolytes in nutritional biochemistry and their functions in metabolism.

CO3: Explain the role of nutrients in the optimal functioning of key biochemical pathways in the body.

CO4: Discuss the biological roles and deficiency disorders of proteins.

SECTION-I

15 Hrs

1. Introduction:

Concept of Nutrition, Calorific value of foods and its determination (Bomb calorimeter), different components of energy expenditure, measurement of energy expenditure by direct and indirect calorimetric method (principles only) Energy expenditure at rest and work, respiratory quotient, Basal Metabolic Rate (BMR), determination of BMR by indirect calorimetric method, factors affecting BMR. Specific dynamic action of foods.

Proximate analysis of food samples:

Moisture, fiber, ash, proteins, carbohydrates, fats and their importance

Carbohydrates:

Dietary sources, dietary fibres and protein sparing action, glycemic index and its importance

Fats

Dietary sources, Visible and invisible fats, trans fats and its effects in fried foods

Water Metabolism

Distribution in the body, factors maintaining water balance and factors influencing water balance.

SECTION-II

15 Hrs

1. Proteins:

Dietary sources, nutritional classification, Nutritional value of proteins – PER, Digestive coefficient, NPU and Biological value of proteins (BV). Essential amino acids. Nitrogen balance, mutual supplementation of proteins, Malnutrition – Kwashiorkor and marasmus (causes, clinical signs with symptoms & treatment

Vitamins

Classification, example with structure, dietary sources, daily requirement, biological roles and deficiency disorders with symptoms– Thiamine, Riboflavin, Niacin, pantothenic acid, Pyridoxine, Biotin, Folic acid,.

SECTION – III

15 Hrs

Structures, dietary sources, daily requirement, biological roles and deficiency disorders with symptoms- Vitamin B12,C,A,D,E & K. Hypervitaminosis.

Minerals:

Dietary sources, physiological functions, deficiency disorders, absorption, balance and excretion of Macro nutrients- Ca, P, Na, K, Cl and Micronutrients – Fe, Zn, Cu, I & Mg

SECTION IV

15Hrs

Digestion:

Gastrointestinal tract secretions - Composition and functions of Saliva, gastric, bile, pancreatic and intestinal Juices. Appetite, gastrointestinal tract hormones.

Digestion, absorption and transport of carbohydrates, proteins and fats

Antinutritional factors: Sources and harmful effects of anti vitamins (e.g.: avidin, dicumarol), Protease inhibitors, oxalates and fitates. Natural toxicants, (e.g.: Lathyrus sativa). Food adulterants- structure and harmful effects of - Butter yellow, lead chromate and malachite green.

FPE530

PRACTICAL

(30 Hrs = 2 Credits)

- | | |
|--|----------------|
| 1. Estimation of haemoglobin in blood. | 3 Hrs |
| 2. Identification of Sugars in fruit juice using paper chromatography. | 3 Hrs |
| 3. Determination of nature of inhibition of alkaline phosphatase by cysteine. | 3 Hrs |
| 4. Determination of proteins by dye binding assay. | 3 Hrs |
| 5. Proximate analysis of food samples- Moisture, fibre, protein fat and carbohydrate (by difference) (3 experiments) | 3 HrsX3 |
| 6. Detection of adulterants in food. | 3 Hrs |
| 7. Estimation of Calcium in ragi. | 3 Hrs |
| 8. Estimation of Vitamin – C in lemon or gooseberries by DPPH method. | 3 Hrs |

PPF510

Semester VI

Paper– VI

(60 hours -4 Credits)

APPLIED BIOCHEMISTRY

Course Outcome:

After completion of the course a student is able to;

CO1: Explain concepts such as gene structure, function, and its regulation.

CO2: Discuss the molecular events and enzymes involved in DNA replication.

CO3: Understand the functions of immune system including organs, cells and receptors.

CO4: Discuss the elementary aspects of the molecular biology of cancer

SECTION-I

10 Hrs

DNA Organization: Structure of chromatin – Histones and Nucleosomes. Active and inactive chromatin. Compaction of Chromatin. Chromosomes, Structure of Genome in eukaryotes. Rearrangements in Genetic Material. Integration of Chromosomes with viruses. Transposition, Experimental proofs for DNA as genetic material.

SECTION-II

20 Hrs

DNA Replication : Semiconservative replication-proof. Molecular events and enzymes involved in DNA replication. DNA repair mechanisms. Mutations. RNA Synthesis : Initiation, elongation and termination during RNA synthesis. Transcription signals. Processing of RNA. Introns and Exons. Nucleases. Genetic Code and Protein Biosynthesis : Characteristics of Genetic code, Deciphering of Genetic Code. Initiation, elongation and termination of protein chains. Post translational modifications in proteins. Inhibitors of protein biosynthesis.

SECTION-III

15 Hrs

Definition of immune system and antigens. Cells involved in immune response. T-cell and B-cells, Immunoglobulins, chemical structure of the Antibody molecule. Haptens and carrier molecules, cell mediated immune response. Complement system, activation and its role in defense. Brief discussion of various immunological techniques; Precipitation reactions in gels Haemagglutination, Immuno-fluorescence, radio-immunoassay (RIA), enzyme linkedimmunoabsorbent assay (ELISA) and immunoblotting.

SECTION-IV

15 Hrs

Membranes : Structure and functions of biological membranes, various models of membrane structure. Transport of solutes across membranes, Sodium pump. Elementary aspects of the Molecular Biology of cancer and introduction to stem cells. Molecular basis of the Origin and Evolution of Life.

FPF530

PRACTICALS

(30 Hrs = 2 Credits)

- | | |
|---|----------------|
| 1. Estimation of DNA by diphenylamine method. | 3 Hrs |
| 2. Effect of temperature on the Viscosity of DNA using Oswald's viscometer. | 3 Hrs |
| 3. Assays of SGPT and SGOT in serum. | 3 Hrs X |
| 4. Extraction of RNA from yeast and its estimation by Orcinol method. | 3 HrsX2 |
| 5. Determination of total protein and A/G ratio in serum. | 3 Hrs |
| 6. Estimation of serum phospholipids. | 3 Hrs |
| 7. Immobilization of enzymes/ cells by entrapment in alginate gel. | 3 Hrs |
| 8. Demonstration of ELISA. | 3 Hrs |

**Pattern of Question Paper for Boc Voc
Semester I to VI
Paper I to VI**

Time : 2 Hrs 30 Mins

Max Marks 70

1. Answer all the questions in one sentence or a word

10 X 1

= 10

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----
- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

5. Answer any four of the following questions

4 X 5 =

20

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

6. Answer any four questions of the following

4 X 10 = 40

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

(Note- 10 Marks may be divided in to 6+4 or 5+5)

I SEMESTER PRACTICAL EXAMINATION

BIOMOLECULES

PRACTICAL I

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

PART

15 Marks

The candidate has to write principle and procedure with tabular column for one experiment in the first 15 min. Then the examiner has to put his/her signature and value the paper at the end of the examination

1. Estimation of amino acid by formal titration
2. Estimation of ascorbic acid by dye method.
3. Estimation of reducing sugars by DNS method

PART B

Qualitative Analysis of Biomolecules

30 Marks

Any one of the following substances may be given for identification

1 Carbohydrate – Glucose, Fructose, Galactose, Lactose, Maltose, Sucrose and Starch.

2 Amino acids – Arginine, Tryptophan, Tyrosine, Cystein, Histidine, phenyl alanine

3 Proteins- Albumin and casein.

PART C

15 Marks

Viva

SCHEME OF VALUATION

(ASSESSMENT OF EXPERIMENTAL RESULTS)

CLASS RECORDS :

i) For conducting and recording 5 experiments = 07 marks.

4 experiments = 06 marks

Less than 4 experiments = 04 marks

ii) For accuracy and neatness = 03 marks.

PART A

Principle- 7 Marks

Procedure- 8 Marks

PART B

Qualitative Analysis Of Biomolecules

- | | | |
|---|--|------------------|
| 1 | Identification of the class of biomolecule | 4 Marks |
| | If carbohydrate is given | |
| | i) Reducing test(any two test) | 7 Marks |
| | ii) Classification test | 6 Marks |
| | iii) Distinguishing test (aldose or ketose) | 6 Marks |
| | iv) Preparation of osazone and identification | 7 Marks |
| | If aminoacid or protein is given, following tests may be conducted | |
| | i) Precipitation test | |
| | ii) Xanthoproteic test | |
| | iii) Millons test | |
| | iv) Sakaguchi test | |
| | v) Lead acetate/Sodium nitroprusside test | 5 x 6 = 30 Marks |

PART C

Viva- Five questions

5 x 3 = 15 Marks

II SEMESTER PRACTICAL EXAMINATION
ENZYMOLOGY AND BIOENERGETICS
PRACTICAL II
SCHEME OF EXAMINATION

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE:- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

PART A

15 Marks

The candidate has to write procedure with tabular column for one experiment in the first 15 min. Then the examiner has to put his/her signature and value the paper at the end of the examination

1. Assay of salivary amylase enzyme activity.
2. Effect of pH on enzyme activity.
3. Effect of temperature on enzyme activity
4. Effect of substrate concentration on enzyme activity and determination of K_m and V_{max}
5. Effect of activators (NaCl) on salivary enzyme activity

PART B

30 Marks

Any one of the following experiment may be set

- 1 Determination of specific activity of Salivary amylase by DNS.
- 2 Determination of pH optimum of Salivary amylase.
- 3 Determination of K_m and V_{max} of Salivary amylase.
- 4 Determination of initial velocity (time Kinetics) of Salivary amylase.

Note-

1. Specific activity:

- i) Standard solution of reducing sugar must be supplied by the examiner.

- ii) Concentration of protein in enzyme solution must be given to the students.
- 2. Optimum pH: buffer of 5 different values from 5 to 9 may be given (buffer solutions is to be supplied named as A,B,C,D,E and asked to found out the buffer in which activity is maximum)
- 3. Optimum time: 10' , 20' , 30' , 40' and 50'range may be considered for the time kinetics.
- 4. Determination of K_m and V_{max} : substrates of different concentration range such as 2, 5, 10, 15, 20 and 30 μMole maybe given(Standard graph of the substrate must be given).

PART C

15 Marks

Viva

SCHEME OF VALUATION
(ASSESSMENT OF EXPERIMENTAL RESULTS)

CLASS RECORDS :

- i) For conducting and recording 5 experiments = 07 marks.
4 experiments = 06 marks
Less than 4 experiments = 04 marks
- ii) For accuracy and neatness = 03 marks.

PART B

Distribution of marks for enzyme assay

- | | |
|---------------------------------|----------|
| 1. Tabular column | 5 Marks |
| 2. For conduction of experiment | 12 Marks |
| 3. Graph | 5 Marks |
| 4. Calculation | 4 Marks |
| 5. Result | 4 Marks |

PART C

Viva- Five questions

5 x 3 = 15 Marks

III SEMESTER PRACTICAL EXAMINATION

METABOLISM

PRACTICAL III

SCHEME OF EXAMINATION

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE:- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

PART A

15 Marks

The candidate has to write procedure with tabular column for one experiment in the first 15 min. Then the examiner has to put his/her signature and value the paper at the end of the examination

1. Estimation of protein by Biuret method.
2. Estimation of Ca⁺ in serum.
3. Estimation of iron in drumstics
4. Estimation of creatinine in serum
5. Estimation of uric acid in urine

PART B

30

Marks

Any one of the following experiment may be set

1. Estimation of protein by Biuret method.
2. Estimation of Ca⁺ in serum.
3. Estimation of iron in drumstics
4. Estimation of creatinine in serum
5. Estimation of uric acid in urine

PART C

15

Marks

Viva

SCHEME OF VALUATION

(ASSESSMENT OF EXPERIMENTAL RESULTS)

CLASS RECORDS :

- i) For conducting and recording 5 experiments = 07 marks.
4 experiments = 06 marks
Less than 4 experiments = 04 marks
- ii) For accuracy and neatness = 03 marks.

PART A

Principle- 7 Marks

Procedure- 8 Marks

PART B

Assessment of experimental results for colorimetric estimation

Preparation standard and working solution - 5 Marks

Distribution of marks for assay

- 1. Tabular column 5 Marks
- 2. For conduction of experiment 5 Marks
- 3. Graph 5 Marks
- 4. Result
 - Up to 10% error 10 Marks
 - Up to 15% error 8 Marks
 - Up to 20% error 6 Marks
 - Any other value 4 Marks

Assessment of experimental results for Calcium estimation

Preparation of standard solution and calculation of the normality - 6 Marks

Discrepancy	Standardization	Estimation
0.1 ml	10 Marks	10 Marks
0.2 ml	8 Marks	8 Marks
0.3 ml	6 Marks	6 Marks
Any other value	4 Marks	4 Marks
Calculation	2 Marks	2 Marks

PART C

Five questions

5x 3 = 15 Marks

**IV SEMESTER PRACTICAL EXAMINATION
BIOCHEMICAL TECHNIQUES**

**PRACTICAL IV
SCHEME OF EXAMINATION**

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE:- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

PART A

15 Marks

The candidate has to write procedure with tabular column for one experiment in the first 15 min. Then the examiner has to put his/her signature and value the paper at the end of the examination

1. Identification of amino acids by circular paper chromatography.
2. Identification of amino acids by ascending Paper chromatography
3. Separation of phospholipids by thin layer chromatography.
4. Separation of leaf pigments by column chromatography
5. Separation of proteins by PAGE

PART B

30

Marks

Any one of the following experiment may be set

1. Identification of amino acids by circular paper chromatography.
2. Identification of amino acids by ascending Paper chromatography
3. Separation of phospholipids by thin layer chromatography.
4. Separation of leaf pigments by column chromatography

PART C

15

Marks

Viva

SCHEME OF VALUATION
(ASSESSMENT OF EXPERIMENTAL RESULTS)

CLASS RECORDS :

- i) For conducting and recording 5 experiments = 07 marks.
4 experiments = 06 marks
Less than 4 experiments = 04 marks
- ii) For accuracy and neatness = 03 marks.

PART A

Principle- 7 Marks

Procedure- 8 Marks

PART B

- 1. Principle and Procedure writing.....12 marks
- 2. For development of Chromatogram..... 12 marks
- 3. For correct identification.....6marks

PART C

Viva- Five questions

5x 3 = 15 Marks

V SEMESTER PRACTICAL EXAMINATION

FOOD AND NUTRITION

PRACTICAL V

SCHEME OF EXAMINATION

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE:- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

PART A

15 Marks

The candidate has to write procedure with tabular column for one experiment in the first 15 min. Then the examiner has to put his/her signature and value the paper at the end of the examination

1. Identification of Sugars in fruit juice using paper chromatography.
2. Proximate analysis of food samples- Moisture, fibre, protein fat and carbohydrate (by difference) (3 experiments)
3. Detection of adulterants in food.

PART B

30

Marks

Any one of the following experiment may be set

1. Estimation of haemoglobin in blood.
2. Determination of nature of inhibition of alkaline phosphatase by cysteine.
3. Determination of proteins by dye binding assay.
4. Estimation of Calcium in ragi.
5. Estimation of Vitamin – C in lemon or gooseberries by DPPH method

PART C

15 Marks

Viva

SCHEME OF VALUATION

(ASSESSMENT OF EXPERIMENTAL RESULTS)

CLASS RECORDS :

- i) For conducting and recording 5 experiments = 07 marks.
4 experiments = 06 marks
Less than 4 experiments = 04 marks
- ii) For accuracy and neatness = 03 marks.

PART A

Principle- 7 Marks

Procedure- 8 Marks

PART B

Assessment of experimental results for colorimetric estimation

Preparation standard and working solution - 5 Marks

Distribution of marks for assay

- 1. Tabular column 5 Marks
- 2. For conduction of experiment 5 Marks
- 3. Graph 5 Marks
- 4. Result
 - Up to 10% error 10 Marks
 - Up to 15% error 8 Marks
 - Up to 20% error 6 Marks
 - Any other value 4 Marks

Assessment of experimental results for Calcium and vitamin C estimation

Preparation of standard solution and calculation of the normality - 6 Marks

Discrepancy	Standardization	Estimation
0.1 ml	10 Marks	10 Marks
0.2 ml	8 Marks	8 Marks
0.3 ml	6 Marks	6 Marks
Any other value	4 Marks	4 Marks
Calculation	2 Marks	2 Marks

PART C

Viva- Five questions

5 x 3 = 15 Marks

VI SEMESTER PRACTICAL EXAMINATION

**APPLIED BIOCHEMISTRY
PRACTICAL VI**

SCHEME OF EXAMINATION

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE:- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

PART A

15 Marks

The candidate has to write procedure with tabular column for one experiment in the first 15 min. Then the examiner has to put his/her signature and value the paper at the end of the examination

1. Effect of temperature on the Viscosity of DNA using Oswald's viscometer.
2. Extraction of RNA from yeast.
3. Immobilization of enzymes/ cells by entrapment in alginate gel
4. Demonstration of ELISA

PART B

30

Marks

Any one of the following experiment may be set

1. Estimation of DNA by diphenylamine method.
2. Assays of SGPT and SGOT in serum.
3. Estimation RNA by Orcinol method.
4. Determination of total protein and A/G ratio in serum.
5. Estimation of serum phospholipids.
- 6.

PART C

15

Marks

Viva

SCHEME OF VALUATION

(ASSESSMENT OF EXPERIMENTAL RESULTS)

CLASS RECORDS :

- i) For conducting and recording 5 experiments = 07 marks.
4 experiments = 06 marks
Less than 4 experiments = 04 marks
- ii) For accuracy and neatness = 03 marks.

PART A

Principle- 7 Marks

Procedure- 8 Marks

PART B

Assessment of experimental results for colorimetric estimation

Preparation standard and working solution - 5 Marks

Distribution of marks for assay

- | | |
|---------------------------------|----------|
| 1. Tabular column | 5 Marks |
| 2. For conduction of experiment | 5 Marks |
| 3. Graph | 5 Marks |
| 4. Result | |
| Up to 10% error | 10 Marks |
| Up to 15% error | 8 Marks |
| Up to 20% error | 6 Marks |
| Any other value | 4 Marks |

PART C

Viva- Five questions

5 x 3 = 15 Marks

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DEPARTMENT OF MICROBIOLOGY

Revised Scheme of Instruction For B.Voc- Food Processing and Engineering 2016-17									
General Education Component									
NSQF/ NVE QF Level	Vocational Qualification	Semester	Title	L:T:P	Theory Hours	Tutorial Hours	Practical Hours	Total Hours	Credits
Level- IV	Diploma	Semester- I	Fundamental s of Microbiology	2:0:1	30	0	15	45	3
Level- V		Semester- II	Microbial Physiology And Metabolism	2:0:1	30	0	15	45	3
Level- VI	Advanced Diploma	Semester- III	Food Microbiology	2:0:1	30	0	15	45	3
		Semester- IV	Immunology	2:0:1	30	0	15	45	3
Level- VII	Degree	Semester- V	Pathogenic Microbiology	4:0:2	60	0	30	90	6
		Semester- VI	Medical Microbiology	4:0:2	60	0	30	90	6

Paper code: FPA520
Credits

MICROBIOLOGY

30hours-2

I B.Voc., I Semester

TITLE: FUNDAMENTALS OF MICROBIOLOGY

Course outcome:

After successful completion of this course students are able to:

- CO1:** Gain basic knowledge about Microbiology starting from history to Microorganisms.
- CO2:** Various Culture media and their applications and also understand various physical and chemical means of sterilization.
- CO3:** Know about microbial techniques for isolation of pure cultures of bacteria.
- CO4:** To identify the bacteria based on staining and cultural characteristics.
- CO5:** Able to perform routine culture handling tasks safely and effectively.
- CO6:** The maintenance and preservation of cultures.

UNIT-I

10Hrs

1. History: Contributions of - Antony van Leeuwenhoek, Edward Jenner, Louis Pasteur, Robert Koch, Dmitry Iwanowski, Alexander Fleming (in brief).
Development and scope of Microbiology. Branches of Microbiology.
2. Microscopy: Types of Microscope, Construction and working principle of bright field microscope. Dark field, fluorescence and phase contrast microscopy(application). Electron microscopy- Types, applications and their limitations.
3. Staining technique: Types of stains, Principle of Simple, negative and differential staining techniques (gram's staining).

UNIT-II

06hrs

Sterilization technique: Definition – Sterilization, disinfection, antiseptic, antibiosis, Fungicide, Bactericide.

A. Physical methods:

- a) Heat –
 - i) Dry heat – Hot air
 - ii) Moist heat method – Autoclave and Pressure cooker
- b) Filtration–Types of filters: Membrane filter, Hepa filter (e.g., Laminar air flow)
- c) Radiation methods – UV rays, Gamma rays and Cathode rays

B. Chemical method: Use and mode of action of- Alcohols, Aldehydes, Halogens and Phenols.

UNIT-III

06hrs

Media – Types, Pure culture and Cultural characteristics: Serial dilution, pure culture by isolation – Pour plate, Spread plate, Streak plate and Micromanipulator techniques . Colony characteristics – plate cultures/solid media and broth cultures/liquid media. Maintenance and Preservation of pure cultures – Sub culturing, overlaying with mineral oil, Refrigeration

(4°C) Lyophilization and cryopreservation.

UNIT-IV

08hrs

1. Comparative account of Prokaryotic and eukaryotic cell. Systems of classification: Haeckel's three-kingdom, Whittaker's five-kingdom and Cavalier-Smith's eight kingdom classification.
2. General characteristics of bacteria, fungi, actinomycete, virus, protozoa and algae. Organization of cell wall, cell membrane, flagella capsules and formation of spores in bacteria.
3. Bacteriophages : Morphology and multiplication(T-4 phage)

PRACTICALS

15 Hrs -1 Credit

1. Staining and mounting of algae and fungi **3Hrs x1**
2. Simple, Negative and Gram's staining **3Hrs x1**
3. Preparation of culture media- Nutrient agar, PDA and NB **3Hrs x1**
4. Methods of obtaining pure cultures of microorganism-Spread plate, pour plate and streak plate and subculturing **3Hrs x1**
5. Isolation of microorganisms from soil by serial dilution technique (Bacteria and Fungi) **3Hrs x1**

Paper code: FPB520

MICROBIOLOGY

30hours-2 Credits

I B.Voc.,II Semester

TITLE: MICROBIAL GROWTH AND METABOLISM

Enable the students to have sound knowledge about:

CO1. Inculcate the knowledge regarding microbial growth, functions, physiology and metabolism.

CO2. Understand the microbial transport systems and microbial metabolism

CO3. Know the microbial growth in response to environmental factors.

CO4. Get equipped with various methods of bacterial growth measurement.

CO5. Knowledge of properties, structure, function of enzymes, enzyme kinetics and their regulation.

UNIT: I

08Hrs

1. Major nutritional type of Microorganisms.
2. Nutritional requirements of Microorganisms. Elementary nutrients: Carbon, Nitrogen, Sulphur, Oxygen and Energy sources, Vitamins and Growth factors.
3. Uptake of nutrients: Diffusion- Simple and Facilitated, Active transport (use of Proton motive force, ATP : ABC transporter), Group translocation, Iron uptake.

UNIT: II

07Hrs

1. Definition, Growth rate and generation time. The growth curve in batch culture - Phases of growth and their significance.
2. Physical and chemical factors affecting growth-Temperature, pH, Oxygen and saline (water activity) Requirements. Measurement of growth by cell number (Haemocytometer) and cell mass (Turbidometer).

UNIT:III

05Hrs

Microbial Enzymes: Definition, Nomenclature, Classification, Properties, Mode and Mechanism of enzyme action, Factors effecting enzyme action. Cofactors and Coenzymes.

UNIT:IV

10Hrs

Aerobic respiration: Definition, Sugar degradation pathways - EMP, HMP and ED pathways. Ultra structure of Mitochondrion, Formation of acetyl CoA from pyruvate, TCA cycle, Electron transport system and Oxidative phosphorylation

Anaerobic respiration: Introduction, Anerobic respiration with special reference to dissimilatory Nitrate reduction. (Denitrification; nitrate /nitrite and nitrate/ammonia respiration; fermentative nitrate reduction) Fermentation - Alcohol fermentation and Pasteur effect; Lactate fermentation (homofermentative and heterofermentative pathways).

PRACTICALS

15 hours– 01 Credit

1. Effect of temperature and pH on growth of bacteria. **3Hrs x1**
2. Acid and gas production from carbohydrates- Demonstration of fermentation of lactose **3Hrs x1**
3. Turbidimetric/spectrophotometric monitoring of growth using liquid cultures **3Hrs x1**
4. Cell counting by Haemocytometer **3Hrs x1**
5. Starch hydrolysis **3Hrs x1**

Enable the students to get sufficient knowledge about:

CO1.Food related microorganisms, their contamination, spoilage and preservation

CO2. The significance and activities of microorganisms in food

CO3. Understand the food borne intoxication and infections.

CO4. Learn about food safety and quality control.

CO5. The principles involving various methods of food preservation.

UNIT:I

10Hrs

1. Introduction to Food Microbiology: Definition, Concept and Scope. Food as a substrate for microorganisms. .
2. Contamination, preservation and spoilage in various foods viz. – fruits and vegetables, canned foods, cereals and cereal product(cereal grains, flour, bread, pasta, macroni), sugars & sugars products (Honey, Candy), Meat (Fresh meat, fish), Milk and Milk products (cheese, butter).

UNIT:II

05Hrs

1. Methods of food preservation: Physical method – high temperature, low temperature, canning. Drying – solar drying, drum drying, spray drying. Radiation.
Chemical methods – chemical preservatives – salient features of the chemical preservatives (propionates, benzoate, sorbates, nitrates and nitrites, sulphur dioxide and sulphates, sugar and salt)

UNIT:III

05Hrs

1. Food borne intoxication and infection:
Bacterial intoxication- Botulism,
Bacterial infection- Salmonellosis.
Mycotoxin – Origin, types and importance of toxins with reference to Aflatoxins.
- 2.. Food safety and quality control. –A brief account on FPO, HACCP, Food laws and Food standards (in brief)

UNIT:IV

10Hrs

1. Microorganisms of industrial importance. Biology of industrial microorganisms: Isolation, Screening, Improvement and Preservation.
2. Raw materials-Molasses, corn steep liquor, sulphite waste liquor and whey. Buffers, Precursors, Inhibitors and Antifoam agents.
3. Fermenters and fermentation process: Design, types and basic function of fermenters, Fermentation process – Surface, Submerged and Solid state fermentation. Types- Batch and Continuous fermentation.
4. Yeasts (Baker's) and its uses, Production of : Wine , Alcohol and lactic acid.

PRACTICALS

3hrsX5 practicals

- 1. Isolation and identification of fungi from spoiled fruits and vegetables** **3hrsx1**
- 2. Isolation and enumeration of bacteria from spoiled fruits and vegetables** **3hrsx1**
- 3. Production of citric acid using *Aspergillus niger*.** **3hrsx1**
- 4. Estimation of lactic acid in milk** **3hrsx1**
- 5. Preparation of alcohol using jaggery or molasses.** **3hrsx1**

Paper code: FPD520

MICROBIOLOGY
II B.Voc.,IV Semester
TITLE: IMMUNOLOGY

30hours-2 Credits

The course provides a solid foundation to understand:

- CO1.** Demonstrate and understanding the key concepts in immunology.
- CO2.** Understand the overall organization of the immune system.
- CO3.** To make them understand the salient features of antigen antibody reaction & its uses in diagnostics and various other studies.
- CO4.** Learn about immunization, preparation and its importance.

UNIT-I

10 hrs

Introduction and history of Immunology,

Innate immunity- Physical Barriers, Biochemical, Cellular, Genetic factors, Body temperature, inflammation and fever

Acquired Immunity- Active & Passive Immunity,

Immune organs and cells: Primary lymphoid organs (Thymus, Bone marrow) & Secondary Lymphoid organs (Lymph nodes, Spleen and tonsils). Mucosa Associated Lymphoid tissue (MALT).

Immune cells- Role of immune cells. Stem cells. Lymphocytes- B lymphocytes-lymphocytes and Null cells. Types of T-cells. Macrophages: Types and function of macrophages.

Immune Response: Humoral Immune Response, Cell Mediated Immune response and Mechanism of AMI and CMI

UNIT-II

05 hrs

Antigens: Nature and types.

Antibodies: Basic structure of immunoglobulin-IgG, Biological properties of immunoglobulins, Monoclonal antibodies. Function and type of antibodies.

UNIT-III

08 hrs

Antigen- Antibody reactions: Salient features of Antigen-Antibody reaction. Precipitation reaction: Immunodiffusion test, Wasserman's test, RPR Test. Agglutination reaction: Blood grouping, Widal test. Complement fixation tests, Opsonization, Immunotechniques: ELISA

UNIT-IV

07 hrs

Immunoprophylaxis: Types of vaccines- Live and Attenuated (Bacterial and Viral) and Toxoid with an example each. National Immunization program (Tabular form).

PRACTICALS

3hrsX5=15 practicals

1. Demonstration of Immune organs (through photographs).
2. Demonstration of Immune cells in the smears prepared from Immune organs. (through photographs)
3. Agglutination- Blood Grouping test
4. Precipitation: Immunodiffusion test –ODD

5. Precipitation: Immunodiffusion test –RID

Paper code: FPE520

MICROBIOLOGY

60 hours -4 Credits

III B.Voc., V Semester

TITLE: PATHOGENIC MICROBIOLOGY

The course provides a solid foundation to understand:

CO1.The human immune response towards microbes in medical microbiology

CO2. Knowledge is gained about the relationship between microorganism and human disease, pathogenicity, Laboratory diagnosis, treatment and prophylaxis.

UNIT-I

20 Hrs

Introduction to important diseases caused by Streptococcus, Pneumococcus, Neisseria, Corynebacterium, Bacillus, Clostridium, Enterobacteriaceae (Proteus, Shigella, Salmonella), Vibrio, Yersinia, Hemophilus, Mycobacterium, The operative pathogenic mechanisms, laboratory diagnosis, prevention and control of these diseases.

UNIT-II

15 Hrs

Morphology, pathogenesis, life cycle, laboratory diagnosis, prevention and control of viral diseases viz. Rabies, Polio, Small pox, Herpes, Measles, Influenza and AIDS.

UNIT-III

15 Hrs

Introduction to Human mycotic infections viz Cryptococcosis, Dermatophytosis, Blastomycosis, Opportunistic Mycosis; Candidiasis and Aspergillosis.

UNIT-IV

10 Hrs

Life cycle, pathogenic, mechanisms and control of parasitic infections viz. amoebiasis, Kala-azar, toxoplasmosis, ascariasis, filarasis, hook worm infections.

PRACTICALS

3 hrsX10 practicals

1- 4. Identification of both gram positive and gram negative microorganisms on the basis of :

(i) Morphology.

(ii) Bio-chemical characteristics.

(iii) Serological reactions.

5. Stainings – Alberts, ZNCF.

6-7. Demonstration of pathogens (Viruses, fungi, parasites) through photographs/ permanent mounted slides.

8-9. Demonstration of cysts/ovas of protozoa/Helminths through photographs.

10. Isolation of dermatophytes from human skin.

Paper code: FPF520

MICROBIOLOGY
III B.Voc., VI Semester
TITLE: MEDICAL MICROBIOLOGY

60 hours -4 Credits

Course outcome

On successful completion of this course the student will gain knowledge about:

- CO1.** Health clinics such as examination, collection of clinical samples and diagnosis.
- CO2.** Beneficial and harmful effect of normal flora
- CO3.** Host pathogen interaction
- CO4.** Determining level of antimicrobial activity and Mechanism of action of antimicrobial agents

UNIT-I

15 Hrs

Brief introduction to terminology of Infectious diseases, Frequency of disease- morbidity and mortality rate. Characteristics of infectious disease.
Infections – Classification, sources, mode of transmission and types of infections.
Disease cycle –Sources, reservoirs, carriers and transmission of pathogen. Emerging (HIV/AIDS, Avian influenza) and reemerging (Tuberculosis, Malaria) Infectious diseases, Global travel & Health considerations.

UNIT-II

15 Hrs

Normal flora of human body –Resident flora and transient flora. Beneficial and harmful effect of Normal Flora. Distribution and occurrence of Normal Flora of Skin, Eye, Respiratory Tract, Mouth, Intestinal Tract & Genitourinary Tract. Germfree and Gnotobiotic Life.

UNIT-III

10 Hrs

Host pathogen interaction –Factors predisposing to microbial pathogenicity- Virulence, Exaltation and attenuation. Determinants of virulence-transmissibility, adhesion, invasiveness, toxigenicity-exotoxins and endotoxins. Avoidance of host defence mechanism.

UNIT-IV

20 Hrs

Development of chemotherapy, General characteristics of antimicrobial drugs, Determining level of antimicrobial activity, Mechanism of action of antimicrobial agents, factors influencing the effectiveness of antimicrobial drugs, Antibacterial drugs viz Sulfonamides, Streptomycin, Quinolones, Penicillins, Cephalosporins, Tetracyclines, Erythromycin, Chloramphenicol, Drug Resistance, Antifungal and Antiviral drugs.

PRACTICALS : 10X3Hrs

30 Hrs -2 Credits

1. Sterilization – Introduction to autoclave, hot air oven, filter sterilization.
2. Microbial flora off mouth-teeth crevices
3. Estimation of urine bacteria by calibrated loop method-Direct streak method.
4. Normal flora of human skin
5. Antibiotic sensitivity test
6. Streptomycin resistant mutant strain isolation by gradient plate technique

7. Identification of enteric pathogens using triple sugar iron agar medium
8. Determination of susceptibility to dental caries by Snyder test
9. Evaluation of antiseptics by filter paper disk method
10. Study of antimicrobial drugs as per theory syllabus.

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PATTERN OF QUESTION PAPER FOR B.VOC
SUBJECT: MICROBIOLOGY
(THEORY:I SEMESTER TO VI SEMESTER)

Time: 3hours

Max marks: 70

I. Define/Explain any ten in one/two sentences: 3X10=30
(Twelve questions to be given and ten to be answered)

- 1
- 2
- 3
- 4
- 5
- 6
- 7
- 8
- 9
- 10
- 11
- 12

II Answer any FOUR of the following: 4X4=16
(Six questions to be given and four to be answered)-short answer type

- 13
- 14
- 15
- 16
- 17
- 18

III (Three essay type questions- with all internal choices) 8X3=24

- 19
- 20
- 21

Test+ Assessment (C-1+C-2) = 30 (15+15)

SCHEME OF PRACTICAL EXAMINATION
I B.Voc., I SEMESTER: PRACTICAL- I

Time: 3hours

Max marks: 70

- I. Write critical notes on **A, B** and **C**
(Stains, Media, Pure culture plates) as per syllabus
- II. Stain the given material D by.....method. Write the principle, procedure and leave the preparation for evaluation.
(Simple staining/Gram-staining/Wet mounting of Algae and fungi/Negative staining)
(Preparation-10marks; Principle and Procedure-10 marks) 20
- III Demonstrate/ Perform the experiment E giving the principle and procedure.
Record the result. 25
(Demonstration- 10 marks; principle-5mark; procedure-5marks; result-05)
(Serial dilution/ Pour plate/Spread plate/Streak plate/subculturing)
- IV. Record. 10
-
-

SCHEME OF PRACTICAL EXAMINATION
I B.Voc.,II SEMESTER: PRACTICAL- II
Microbial physiology and Metabolism

Time: 3hours

Max marks: 70

- I. Write critical notes on **A, B** and **C**
(Haemocytometer, Effect of temperature and pH, Acid and gas production from carbohydrates, Turbidimetry/spectrophotometry, Starch hydrolysis.)
- II. Demonstrate/ Perform the experiment **A** giving the principle and procedure.
Record the result. 20
(Cell counting by Haemocytometer)
(Preparation-10marks; Principle and Procedure-10 marks)
- III Demonstrate/ Perform the experiment **A** giving the principle and procedure.
Record the result. 25
(Demonstration- 10 marks; principle-5mark; procedure-5marks; result-05)
(Acid and gas production from carbohydrates, Starch hydrolysis ,Effect of temperature and pH)
- IV. Record. 10

SCHEME OF PRACTICAL EXAMINATION
II B.Voc.,III Semester
FOOD MICROBIOLOGY

Time: 3hours.

Max.marks:70

- I. Demonstrate / Perform the experiment **A**, giving principle and procedure. Record and interpret the result.
(Demonstration -10marks; principle-5marks;procedure-3marks;results and interpretation-2marks).
(Isolation of microorganisms from spoiled vegetables/spoiled fruits). 25
- II. Conduct the test for **B**. Write the principle and procedure. Record and interpret the results.
(Demonstration -10 marks; principle-5marks; procedure-3marks; results and interpretation-2marks). (Estimation of lactic acid in milk,Estimation of citric acid) 20
- III. Write critical notes on **C, D** and **E**. (Identification -1mark; critical comments-1marks).
(Citric acid production, Estimation of lactic acid in milk,alcohol from jaggarey) 5X3=15
- IV Record 10

.....

SCHEME OF PRACTICAL EXAMINATION
II B.Voc.,IV Semester
IMMUNOLOGY

Time:3hours

Max.marks:70

- I. Demonstrate / Perform the experiment **A**, giving principle and procedure. Record and interpret the result.
(Demonstration -10marks; principle-5marks;procedure-5marks;results and interpretation-5marks).
(Determination of blood group and Rh factor /Demonstration of RID ,ODD). 25
- II. Demonstrate the experiment **B**. write the principle and procedure. Record and interpret the results.
(Demonstration -10marks; principle-5marks;procedure-3marks;results and interpretation-2marks).
(Blood grouping, ODD, RID) 20
- III. Write critical notes on **C, D**, and **E**. (Identification -1mark; critical comments-1marks)
(Immune organs and immune cells) as per syllabus 5x3=15
- IV Record 10

SCHEME OF PRACTICAL EXAMINATION
III B.Voc.,V Semester
PATHOGENIC MICROBIOLOGY

Time:3hours

Max.marks:70

- I. Demonstrate / Perform the experiment **A**, giving principle and procedure.Record and interpret the result.
(Demonstration -10marks; principle-5marks; procedure-5marks;results and interpretation-5marks).
(Serological test, Biochemical test). 25
- II. Stain the given material D by.....method. Write the principle, procedure and leave the preparation for evaluation.
(Morphology- Simple staining/Gram-staining/Negative staining)
(Preparation-10marks; Principle and Procedure-10 marks) 20
- III. Write critical notes on **C, D, and E**. (Identification -1mark; critical comments-1marks)
(pathogens ,cysts,ovas of protozoa/helminths) as per syllabus 5x3=15
- IV. Record 10

.....

SCHEME OF PRACTICAL EXAMINATION
III B.Voc.,VI Semester
PATHOGENIC MICROBIOLOGY

Time:3hours

Max.marks:70

- I. Demonstrate / Perform the experiment **A**, giving principle and procedure.Record and interpret the result.
(Demonstration -10marks; principle-5marks;procedure-5marks;results and interpretation-5marks).
(Normal flora of skin, Antibiotic sensitivity test, Microbial flora off mouth-teeth crevices, Estimation of urine bacteria by calibrated loop method-Direct streak method, Evaluation of antiseptics by filter paper disk method, Determination of susceptibility to dental caries by synder test, TSIA test). 25
- II. Demonstrate the experiment **B**. write the principle and procedure. Record and interpret the results.
(Demonstration -10marks; principle-5marks;procedure-3marks;results and interpretation-2marks).
(Streptomycin resistant mutant strain isolation by gradient plate technique, Antibiotic sensitivity test,estimation of urine bacteria by calibrated loop method, Normal flora of skin, Antibiotic sensitivity test, Microbial flora off mouth-teeth crevices,TSIA test). 20
- III. Write critical notes on **C, D, and E**. (Identification -1mark; critical comments-1marks)
(Results of Streptomycin resistant mutant strain isolation by gradient plate technique,Antibiotic sensitivity test,estimation of urine bacteria by calibrated loop method ,Normal flora of skin,mouth-teeth autoclave, hot air oven, filter sterilization, Antibiotics as per syllabus,TSIA test,Instruments) 5x3=15
- IV .Report 10

Approved list of Paper setters

Sl No.	Name	College address
1.	Dr..M .Seema	Chairperson, Dept. of Microbiology JSS College, Ooty road, Mysore
2.	Dr..K.Sumana	Assistant prof. Dept. of Microbiology JSS College, Ooty road, Mysore
3	Dr.S.Mahadevamurthy	Associate Prof & HOD Dept. of Microbiology Yuvaraja's college Mysore.
4	Dr.Syeda Kauser Fathima	Associate Prof. of Microbiology Maharani's Science College for women JLB road Mysore.
5	Dr. H.S. Jayanth.	Asso.Prof.of Microbiology Dept. of Microbiology Yuvaraja's college Mysore.
6	Dr.Nagarathnamma	Asso. Prof. of Microbiology Government women college Mandya
7	Dr. Mashooda Begum	Asso.Prof.of Microbiology Maharani's Science College for women JLB road, Mysore.
8	Sri. M. Girish	Assistant prof. Dept. of Microbiology JSS College for Women Saraswathipuram,Mysore
9	Dr. P.K.Maheshwar	Assistant Prof. Dept. of Microbiology Dept. of Microbiology Yuvaraja's college,Mysore.
10	Smt. M.S.Shobha	Assistant Prof, Dept. of Microbiology Maharani's Science College Mysore
11	Sri. Shankaregowda	Asso.Prof.of Microbiology Government Science College Mandya
12	Sri. R.A. Manjunath	Assistant Prof. Dept. of Microbiology Saradavilas College,Mysore
13	Dr.M.P. Ragavendra	Assistant Prof. Dept. of Microbiology Maharani's Science College,Mysore
14	Dr.K.Girish	Assistant Prof.

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15	Sri. G.S. Siddegowda	Assistant Prof. Dept. of Microbiology Maharani's Science College Mysore
16	Smt. M.S. Poornima	Assistant Prof. Dept. of Microbiology Yuvaraja's College, Mysore
17	Dr.N.S.Devaki	Assistant Prof. Dept. of Molecular Biology Yuvaraja's College , Mysore
18	Syeda Farahna Parveen	Assistant Prof. Dept. of Microbiology St.Philomina's College, Mysore

19	Smt. Vanitha	Assistant Prof. Dept. of Microbiology Maharani's Science College, Mysore
20	Smt. Revanamba	Assistant Prof. Dept. of Microbiology Maharani's Science College, Mysore
21	Uzma Bathool	Assistant Prof. Dept. of Microbiology St.Philomina's College, Mysore
22	Mahadeva prasad	Assistant Prof. Dept. of Microbiology JSS College for Women Saraswathipuram, Mysore

DEPARTMENT OF COMPUTER SCIENCE

**SEMESTER I
FPA 540**

COMPUTER SCIENCE

PAPER - I

COMPUTER FUNDAMENTALS & DOS

(2 hrs theory / week)

30 hours - 2 Credits

CO1.Master the basic knowledge of applications of MS office package

CO2.Get the skill of office productivity tool

CO3.Learn the usage of internet

CO4.Skill to develop program using C language

SECTION- I

15 Hours

INTRODUCTION

Computer, Characteristic of Computer, History of Computer, Generation of Computers, Components of Computer and Applications of Computers.

Key Factors of Computers: Hardware, Software - types of Software (Application and system), forms of software (firmware, shareware, freeware), Translator - Assembler, Compiler and Interpreters. Computer Application – Business, Scientific, Entertainment and educational.

CLASSIFICATION OF COMPUTERS

Mode of operations – Analog, Digital and hybrid Computers.

Size and capabilities – Micro, Mini, Main frame and Super computer.

MEMORY UNITS

Primary memory - RAM, ROM, PROM, EPROM, EEPROM, Flash memory, cache memory.

Secondary memory – Magnetic disk (Hard disk, Floppy disk, Zip disk, Jaz disk, Super disk), Optical disk (CD, CD – R, CD – RW, DVD).

COMPUTER PERIPHERALS DEVICES AND INTERFACES

Input devices – Working principle of Keyboard and mouse, Functional capabilities of Scanner, Digital Camera, OMR, OCR, touch pad, touch screen. Output Devices – Monitor, Printer, Plotter and projector.

PROGRAMMING LANGUAGES

Machine, Assembly language and High Level Language.

INFORMATION SYSTEM

Data and Information, types of information, what is an information System, Types of Information Systems – System development life cycle.

SECTION- II

15 Hours

OPERATING SYSTEM AND THE USER INTERFACE

Operating System– Functions, services, Types-Batch, Single, Multiprogramming, and Multiprocessing.

Operating System – the user interface –running programs –managing hardware –enchaining the operating system with utility software- typical operating systems in use

COMPUTERS AND COMMUNICATION

Single user, multi-user, workstation, and client server systems. Computer networks, Types of Network LAN, WAN, Internet, Internet applications, WWW, Email, FTP, web browsers (Internet explorer, Google Chrome, Mozilla).

DISK OPERATING SYSTEM (Dos)

Introduction, History & version of DOS. DOS basics – physical structures of disk, drive, Name, FAT, File & Directory structure and Naming Rules, Booting Process, DOS System files, role of config.sys file.

DOS COMMANDS

Internal – DIR, MD, CD, RD, COPY, DEL REN, VOL, DATE, TIME, CLS, PATH, TYPE etc.

External – CHKDSK, XCOPY, PRINT, DISK COPY, DISKCOMP, DOSKEY, TREE, MOVE, LABEL, APPEND, FORMAT, SORT, FDISK, BACKUP, EDIT, MODE, ATTRIB, HELP, SYS, WILD CARD Characters etc.

Executable V/S Non executable Files in DOS.

TEXT BOOKS:

- Peter Norton's 'Introduction to Computers', Second Edition, TMH
- Computer Fundamentals – P K Sinha, BPB

REFERENCE BOOKS :

- Introduction to Computers – N Subramanian, TMH
- Understanding Computers – R Rajagopalan. TMH
- Computers Today – Donald Sanders, MGH

Practical

(1Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

**SEMESTER II
COMPUTER SCIENCE**

FPB 550

**PAPER – II
PROGRAMMING IN C
(2 hrs theory / week)**

30 hours - 2 Credits

CO1.Master the basic knowledge of applications of MS office package

CO2.Get the skill of office productivity tool

CO3.Learn the usage of internet

CO4.Skill to develop program using C language

SECTION-I

15 Hours

PROBLEM SOLVING TECHNIQUES

Problem Definition, Problem Analysis, Design of problem solutions and use of design tools, Algorithm, Flowcharts, Coding, Testing, Debugging, Program documentation.

INTRODUCTION TO C LANGUAGE

History Features and Applications of 'C', Structure of C Program.

PROGRAMMING PRELIMINARIES

Character set, definitions and declarations of identifiers, Variables, Escape Sequence Characters. Constants, Keywords, Data types with examples.

OPERATORS AND EXPRESSIONS

Various operators and expressions, Operator precedence with example programs.

INPUT-OUTPUT STATEMENTS

Various types of standard input output statements, standard mathematical functions, with example programs.

SECTION - II

15 Hours

CONTROL STRUCTURES

Decision makes and branching statements, Decisions making and looping statements, break statement, continue statement and goto statement with example programs.

ARRAYS

Definitions and need of arrays, 1-d and 2-d arrays with example programs, introduction to multidimensional arrays.

STRING HANDLING

Declarations, Initialization, reading and writing of strings, operations and string functions with example programs, array of pointers to strings.

FUNCTIONS

Definitions and need of functions. Library functions, user defined functions in detail, function declaration and prototypes call by value, call by reference and functions and arrays, recursion, storage classes with example program.

STRUCTURE

Definition of structure, Array in Structures, Structure with Array, Difference between array and structure

TEXT BOOKS:

- Programming with ANSI C by: E. Balagurusamy

REFERENCE BOOKS :

- Let us C - Yashwanth kanetkar
- Computer concepts and C programming by - P. B. Kotur
- The Complete Reference C by Herbert Schildt

Practical

(1Hour per week X 15 Weeks = 15 Hours)

Experiments are based on topics mention in the Paper designed by concerned Faculty

Scheme of Examination

Sl. No	Semester	Paper	Marks for theory	Marks for internal assessment		Total
				C1 = 15	30	
1	I	COMPUTER FUNDAMENTALS & DOS	70	C1 = 15	30	100
				C2 = 15		
2	II	PROGRAMMING IN C	70	C1 = 15	30	100
				C2 = 15		

Distribution of Internal assessment

- 1) CLASS TEST (C1) -15 MARKS. } = 30 marks
2) CLASS TEST (C2) -15 MARKS. }

Scheme of Practical Examination

Sl. No	Semester	Paper	Marks for Practical	Marks for Record	Marks for Viva	Total
1	I	COMPUTER FUNDAMENTALS & DOS	60	05	05	70
2	II	PROGRAMMING IN C	60	05	05	70

Pattern of Question Paper for B.Voc

**Semester I
COMPUTER SCIENCE**

Time : 2 Hrs 30 Mins

Max Marks 70

1. Answer all the questions in one sentence or a word

5 X 1 = 5

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

2. Tick the correct answer

5 X 1 = 5

- f. -----
- g. -----
- h. -----
- i. -----
- j. -----

3. State whether the following statements are true or false

5 X 1 = 5

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

4. Fill in the blanks with suitable answers

5 X 1 = 5

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

5. Answer any ten of the following questions

10 X 3 = 30

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----
- f. -----
- g. -----
- h. -----
- i. -----
- j. -----
- k. -----

l. -----

6. Answer any four questions of the following

4 X 5 = 20

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

**Pattern of Question Paper for B.Voc
Semester II
COMPUTER SCIENCE**

Time : 2 Hrs 30 Mins

Max Marks 701. Answer

all the questions in one sentence or a word

10 X 1 = 10

- f. -----
- g. -----
- h. -----
- i. -----
- j. -----
- k. -----
- l. -----
- m. -----
- n. -----
- o. -----

5. Answer any four of the following questions

4 X 5 = 20

- f. -----
- g. -----
- h. -----
- i. -----
- j. -----

6. Answer any four questions of the following

4 X 10 = 40

- f. -----
- g. -----
- h. -----
- i. -----
- j. -----
- k. (Note- 10 Marks may be divided in to 6+4 or 5+5

FPA 020

À»vÀå WÀIPÀ

12 UÀAmÉUÀ¼ÀÄ

1. PÁªÀå

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| 4. ªÀÄªª¨ÉÉ eÁvÀPÀ | - fJ,ïJ,ï |
| 5. CqÀÄUÉ ªÀÄÉÉAiÀÄ °ÀÄqÀÄV | - ªÉÉzÉÅ» |
| 6. ÅÅÀß dÅÀUÀ¼ÀÄ | -¹zÀª°AUÀAiÀÄå |

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12 UÀAmÉUÀ¼ÀÄ

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| 1. ¤qÀAvÀgÀ ¤ÅJUÁV ¤AvÀÉèÄ NI | -ÅÁUÉÅ±ï°ÉUÀqÉ |
| 2. ªÁiÁ»w vÀAvÀæªÁÖÅ : MAzÀÄ, ÅÆÜ® ÅÉÆÄI | - f.JÆï. ÅgÀ¹AªªÅÆwð |
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| 4. ¨ÉÅqÀzÀ CwyUÀ¼ÀÄ | - J JÆï ªÀÄÆwðgÀAiÀÄgÀÄ |

¨sÁµÁ WÀIPÀ

3. DqÀ½vÀ ªÀÄvÀÄÜ ªÁtÂdÀå PÅÈÀßqÀ

21 UÀAmÉUÀ¼ÀÄ

1. DqÀ½vÀ ¨sÁµÉAiÀiÁV PÅÈÀßqÀ, ÅégÀÆªª ªÀÄvÀÄÜ ©PÀèt
2. ÅPÁðj ¤AvÀæ, ÅégÀÆªª - CçüPÀÈvÀ eÁÖªÅÅ ªÀÄvÀÄÜ, ÅÄvÉÆÜÄÉ
3. ««zsÀ jÅwAiÀÄ CfðUÀ¼ÀÄ - gÅeÉ, ªÉÅvÀÅ, §rÜ, ¤zÉÆÅÅÅßw EvÅgÉ
4. ªÁtÂdå PÅÈÀßqÀzÀ, ÅégÀÆªª ªÀÄvÀÄÜ ©PÀèt
5. ªÁtÂdå ¤AvÀæzÀ ««zsÀ CAUÀUÀ¼ÀÄ
ªÁtÂdå ¤AvÀæUÀ¼ÀÄ- «ZÁgÀuÀ ¤AvÀæ, DzÉÅ±À ¤AvÀæ, GzÀjªAvÀæ, ªÁ, ÅÆ° ¤AvÀæ, ¤ÁjzÀAiÀÄ ¤AvÀæ, ¤gÀªª±ÀðÅ ¤AvÀæ, ªªª°ÁgÉÆÄfÓªªÅÅ ¤AvÀæ, ¤ÁjªAvÀæ
6. UÀtPÀzª°è PÅÈÀßqÀ - PÅÈÀßqÀzÀ CAVÀeÁð® vÀtUÀ¼ÀÄ- ÅÄr, §gÀª- ¤Áj¨sÁ¶PÀ ¤zÀUÀ¼ÀÄ
7. ¨sÁµÁAvÀgÀ : ÅégÀÆªª ªÀÄvÀÄÜ ©PÀèt- PÅÈÀßqÀçAzÀ EAVè¶UÉ- EAVè¶ªAzÀ PÅÈÀßqÀPÉì

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| 1. ÅªªUÀæPªª | - f J,ï ªªgÀÄzÀæªª |
| 2. ©AzÀÄ ©AçUÉ | - ªÉÉzÉÅ» |
| 3. ¤AQèPª² | - PÀªªÉAªÀ |
| 4. ÅªªUÀæPªª | - ¨ÉÅzÉæ |
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| 6. ªªªªªPª PÅÈÀßqÀ | - JZï J,ï PÉ |
| 7. DqÀ½vÀPÅÈÀßqÀ | - qÀ C±ÉÆÄPïPÀªªªiÁgï gÀAeÉgÀ |

8. PÀbÉĀj PÉĒjṛ - PÄÄªÉA¶ÄÄ PÄŁÄßqÄ CzšÄâAiÄÄŁÄ ,ÄA,ÉÜ
9. ĒĒRŁÄPÄĒĒ - J,ĭ ¶ÄæªÄèzÄgÄªĭ
10. DqÄ½vÄĒsÁµĒ PÉªÄÄ «ªÄgÄUÄ¼ÄÄ - ¶ÄæzsÁŁÄ UÄÄgÄÄzÄvÄÜ

B.Voc Programme

**Language kannada
I Semester**

¶Äæ±Éß¶ÄwæPÉ ,ÄégÄĒ¶Ä

- 3.00 UÄAmÉUÄ¼ÄÄ 70 CAPÄUÄ¼ÄÄ
«ĒsÁUÄ-1
1. ĒsªªÄxÄð §gÉ-Āj : 1óó*5=5
C) CxªªÄ
D)
2. ,ÄAzÄĒsÄð ,Ä»vÄ «ªĀj¹ 1óó*5=5
C) CxªªÄ
D)
3. PÉ¼ÄVŁÄ ¶Äæ±ÉßUÉ GvÄÜj¹: 1*10=10
C) CxªªÄ
D)
5. PÉ¼ÄVŁÄ ¶Äæ±ÉßUÉ GvÄÜj¹: 1óó*10=10
C) CxªªÄ
D)
5. PÉ¼ÄVŁÄ ¶Äæ±ÉßUÉ GvÄÜj¹: 1*10=10
C) CxªªÄ
D)
6. LzÄPÉĭ n¶ÄätÄ §gÉ-Āj: 3*5=15
C) CxªªÄ
D)
7. PÉ¼ÄVŁÄ ¶Äæ±ÉßUÉ GvÄÜj¹: 1*10=10
C) CxªªÄ
D)
9. PÄŁÄßqÄPÉĭ ĒsÁµÄAvÄj¹: 1*5=5
10. EAVèĭUÉ ĒsÁµÄAvÄj¹: 1*5=5
- «ĒsÁUÄ-2
- «ĒsÁUÄ-3

FPB540

**B.Voc- II Semester
Communicative English**

No. of Credits : 3

No. of Instruction hrs : 3 / Week(45 hrs)

- AIMS:** 1) To familiarize students to basic English
2) To enable them to develop listening & speaking skills

OBJECTIVES: Students should be able to ____

- 1) Write English without grammatical errors
- 2) Speak English Language effectively and accurately
- 3) Listen and understand public announcements and news on TV & Radio

Module – 1 Grammar

	Marks	Hrs/ Week
1. Subject and Verb Agreement	5	6
2. Voice	5	5
3. Articles	5	3
4. Speech	5	6
5. Question tag	5	5
6. Framing of Questions	5	Q=05

Module – 2 Writing Skills

1. Letter Writing Letter of Application/Letter of Grievances/Resume Preparation	10	4
2. Comprehension	10	3
3. Essay Writing	10	3

Module – 3 Speaking Skills

1. Greeting		
2. Requesting		
3. Enquiring		
4. Explaining	10	Q=05
5. Reporting		
6. Permission		
7. Thanking		

70	45
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PAPER CODE :FPB540

**B.Voc II SEMESTER
COMMUNICATIVE ENGLISH**

MODEL QUESTION PAPER

TIME: 3 Hrs

MAX MARKS: 70

Instructions to students: All sections are compulsory.

SECTION A

I Fill in the blanks with correct form of verbs:

5 X 1 = 05

- a) Slow and Steady _____ the race.(to win)
- b) Christians _____ to church on Sundays. (to go)
- c) They have _____ the work. (to complete)
- d) He _____ absent yesterday. (to be)
- e) We _____ learning grammar now. (to be)

II Change the voice of the following sentences:

5 X 1 = 05

- a) He is writing a poem.
- b) We have won the match.
- c) The poet took the second road.
- d) A song will be sung by her.
- e) Close the door.

III Fill in the blanks with suitable article:

5 X 1 = 05

- a) My father is _____ MLA.
- b) _____ earth moves round the sun.
- c) The poet rejected _____ first road.
- d) The Nile is _____ long river.
- e) Brutus was _____ honest man.

IV Change the speech of the following:

5 X 1 = 05

- a) The King said to people," I am happy to be here".
- b) Mother said to the child," Don't go near water".
- c) Raju asked Rosie who she was.
- d) Velan said to Venu," Why are you crying?"
- e) The lady told the writer that he was very humorous.

V Add question tag to the following:

5 X 1 = 05

- a) He is a good singer.
- b) The poet observed the daffodils.
- c) Behrman saves the life of Johnsy.
- d) English is not a difficult language.
- e) India has won the second test.

VI Frame questions so as to get the underlined words as answers:

5 X 1 = 05

- a) Shakespeare wrote 'Macbeth'
- b) Rama killed Ravana.
- c) He goes to college by car.
- d) The College starts at 10.30 am

e) Lear's daughters were selfish.

SECTION B

P.T.O

VII Letter Writing:

10 X 1 = 10

- a) **Write a letter of application to the post of a Computer Programmer at SkillTec Ltd. Mysuru**
OR
b) **Prepare your resume.**

VIII Read the following essay carefully and answer the questions set on it:

5 X 2 =10

I was wrong to think that city life is altogether unhealthy. Perhaps it was so at one time, now a days with proper roads, pavements and drainage systems, sickness is kept at bay. Cities are planned in such a way as to provide open space with parks and playgrounds for the benefit of the dwellers.

Even when sickness does strike, there are doctors and hospitals near at hand. This is not the case in the country where people frequently suffer and sometimes die for want of medical facilities.

- 1) What does the writer think of the city life?
- 2) What causes sickness?
- 3) Where do you find doctors and hospitals in plenty?
- 4) How does the passage distinguish between city and country life?
- 5) Give a suitable title to the passage.

IX Write an essay on one of the following:

10 X 1 = 10

- a) Use and abuse of social media.
- b) Role of students in nation building
- c) Afforestation?

X Answer the following:

5 X 2 = 10

1. Write an imaginary conversation between a customer and Bank Manager regarding opening an account.
2. Write a dialogue between two friends who meet after a long time.

**B.Voc II SEMESTER
COMMUNICATIVE ENGLISH
QUESTION PAPER SCHEME**

TIME: 3 Hrs

MAX MARKS: 70

Instructions to Students: All sections are compulsory.

SECTION A

- I Fill in the blanks with correct form of verbs 5 X 1 = 05
a)
b)
c)
d)
e)
- II Change the voice of the following sentences 5 X 1 = 05
a)
b)
c)
d)
e)
- III Fill in the blanks with suitable article 5 X 1 = 05
a)
b)
c)
d)
e)
- IV Change the speech of the following 5 X 1 = 05
a)
b)
c)
d)
e)
- V Add question tag to the following 5 X 1 = 05
a)
b)
c)
d)
e)
- VI Frame questions so as to get the underlined words as answers 5 X 1 = 05
a)
b)
c)
d)
e)

SECTION B

- VII 10 X 1 = 10
1. Write a letter of application
OR
2. Prepare a Resume.
- VIII Read the following essay carefully and answer the questions. 10 X 1 = 10

- a)
- b)
- c)
- d)
- e)

IX Write an essay on one of the following:

10 X 1 = 10

- a)
- b)
- c)

X Answer the following :(Dialogue writing)

5 X 2 = 10

- 1)
- 2)

UNIT I **08 hrs**

- a) Preamble of the Indian Constitution
- b) Salient features of Indian Constitution

UNIT II **10 hrs**

- a) Fundamental Rights
- b) Fundamental Duties
- c) Directive principles of State Policy

UNIT III **14 hrs**

- a) President – Election Method, Powers and Functions
- b) The Role of the Prime Minister
- c) The Parliament – Structure, Power and Functions(Lok Sabha and Rajya Sabha)
- d) Supreme Court – Organization and Jurisdiction

UNIT IV **13 hrs**

- a) The Role of Governor in the Administration of State
- b) Powers and Functions of the Chief Minister
- c) Composition , Powers and Functions of both the Houses of State Legislature
- d) High Court – Organization and Jurisdiction

TEXT BOOKS

- An introduction to the Constitution of India by M V Pylee
- Introduction to the Constitution of India by D D Basu
- Understanding the Constitution of India by Dr. H M Rajashekar
- Indian Constitution by Sommanna, Brahamananda, H B Mallikarjuna swamy,
- Indian Constitution by H T Ramakrishana, Rajiv

Sl. no.	Information Communication Technology	Hrs
1	The humanitarian supply chain – Definition, system of organizations, people, technology, activities, information and resources involved in moving a product or service from supplier to customer	5
2	Technology framework – Front-end services, Middleware services and Infrastructure services: Supporting the food assistance supply chain; Mapping technologies; Web portals	5
3	Mobile technologies - Combining hand-held and wireless communications technologies	5
4	Beneficiary identification - Challenges in food assistance to ensure that assistance goes to the right beneficiaries	5
5	ICT in emergencies – Requirement inputs of Food assistance interventions during emergencies	5
6	Linking the humanitarian supply chain - Ways in which WFP uses technology and technological techniques to fulfill its role as the provider of food assistance in development and emergencies	5

Sl. no.	Practical	Hrs
1.	Identification of software related to Food Processing and Engineering	3
2.	Practicing the use of software	6
3.	Requirement development for Food Processing Software	6

INFORMATION AND COMMUNICATION TECHNOLOGY

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

I. Write short notes for the following(any 5): (5x2=10)

- 1. -----
- 2. -----
- 3. -----
- 4. -----
- 5. -----
- 6. -----

PART-B

II. Answer any 4 of the following: (4x5=20)

- 1. -----
- 2. -----
- 3. -----
- 4. -----
- 5. -----

PART –C

III. Answer any 4 of the following: (4x10=40)

- IV.**
- 1. -----
 - 2. -----
 - 3. -----
 - 4. -----
 - 5. -----

INFORMATION AND COMMUNICATION TECHNOLOGY

PRACTICAL

SCHEME OF EXAMINATION

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

1. Identification of software related to Food Processing and Engineering.
2. Practicing the use of software.
3. Requirement development for Food Processing Software.

Environmental Studies (One-Semester Compulsory Core Module for B.Voc Programmes)

Unit 1: Environment and natural systems

4 hrs

- Introduction to Environment and Environmental Studies
- Definition and Components of Environment, Relationship between the different components of Environment
- Man and Environment relationship
- Impact of technology on Environment, Environmental Degradation
- Multidisciplinary nature of the Environment studies
- Its scope and importance in the present day Education System

UNIT 2: Ecology and Ecosystems:

7 hrs

- Introduction: Ecology- Objectives and Classification
- Concept of an ecosystem- structure and functions of ecosystem
- Components of ecosystem- Producers, Consumers, Decomposers
- Bio-Geo- Chemical Cycles- Hydrologic Cycle, Carbon cycle, Energy Flow in Ecosystem, Food Chains, Food webs ,Ecological Pyramids
- Major Ecosystems: Forest Ecosystem, Grassland Ecosystem, Desert Ecosystem, Aquatic Ecosystem, Estuarine Ecosystem.

Unit 3: Natural Resources

7 hrs

Renewable and Nonrenewable resources, exploitation and conservation,

- Water resources: Surface and Ground water sources, Indian and Global scenario. Land as a resource, land use change and land degradation
- Forest resources: Definition and Classification of Forests Ecological and Economic importance and benefits of forest, Indian scenario, Deforestation: causes and effects, case studies remedial measures
- Food resources: Sources of food, Global and Indian food demand scenario, Limits of food production, Environmental effects of Agriculture
- Energy resources: Renewable and non renewable energy sources, use of alternate energy sources, growing energy needs, case studies
- Mineral resources: Definition and Classification of minerals, mining issues Case studies.
- Role of individual in conservation of natural resources.

Unit 4 : Biodiversity and its Conservation

6 hrs

- Biodiversity : Definition, Levels of biological diversity : genetic, species and ecosystem diversity
- Biogeographic zones of India
- Hot spots of biodiversity
- Ecosystem and biodiversity services: Ecological, economic, social, ethical, aesthetic and Informational values
- Biodiversity patterns
- India as a mega-biodiversity nation

- Threats to biodiversity: Habitat loss, poaching of wildlife, man-wildlife conflicts, biological invasions.
- Endangered and endemic species of India
- Conservation of biodiversity: In-situ and Ex-situ conservation of biodiversity.

UNIT 5: Environmental pollution:

9 hrs

- Types of Environmental Pollution:
- Water Pollution: Introduction – Water Quality Standards, Sources of Water Pollution: Industrial Agricultural, Municipal; Classification of water pollutants, Effects of water pollutants, Eutrophication.
- b) Marine pollution: Causes, effects and control.
- c) Air Pollution: Composition of air, Structure of atmosphere, Ambient Air Quality Standards, Classification of air pollutants, Sources of common air pollutants like PM, SO₂, NO_x, Natural & Anthropogenic Sources, Effects of common air pollutants
- d) Soil Pollution: causes, effects and control.
- e) Noise Pollution: Introduction, Sound and Noise, Noise measurements, Causes and Effects
- f) Thermal Pollution: Causes, effects and control.
- g) Nuclear hazards and human health risks.
- Solid waste management: Control measures of urban and industrial waste.
- Role of individual in the prevention of pollution, Pollution case studies.

UNIT 6: Sustainable development and Environmental issues and Policies.

7 hrs

- Sustainable development: Meaning, changes in resource utilization, urbanization.
- Water conservation: watershed management and Rain water harvesting.
- Environmental issues: Climate change, global warming, acid rain, ozone layer depletion.
- Disaster management: floods, drought, earthquake, cyclones and landslides.
- Wasteland reclamation.
- Environment Protection Act: Air, Water, Wildlife (Prevention and Control of Pollution)
- Forest Conservation Act.
- Issues involved in enforcement of environmental legislation.
- Environment: rights and duties.

Unit 7 : Human Population and the Environment

5 hrs

- Population growth, Explosion, demographic variation among nations.
- Family welfare Program.
- Environment, human health and welfare; infectious and lifestyle diseases in contemporary world.
- Value Education: Environmental ethics.
- HIV/AIDS
- Women and Child welfare.
- Role of information technology in Environment and human health

Unit 8: Field visit

5 hrs

- Field work Visit to an area to document environmental assets :river/ forest/ grassland/ hill/ mountain
- Visit to a local polluted site-Urban/Rural/Industrial/Agricultural.
- Visit to the solid waste treatment plant and water treatment plant.
- Video: The one degree • (Equal to 5 lectures)

REFERENCE BOOKS:

1. Textbook of Environmental Studies for Undergraduate Courses by Erach Bharucha Second edition, 2013 Publisher: Universities Press (India) Private Ltd, Hyderabad.
2. Basics of Environmental Studies by Prof Dr N S Varandani, 2013 Publisher: LAP -Lambert Academic Publishing, Germany
3. Environmental Studies by Anindita Basak, 2009 Publisher: Drling Kindersley(India)Pvt. Ltd Pearson
4. Textbook of Environmental Studies by Deeksha Dave & S S Kateva , Cengage Publishers.
5. Environmental Sciences by Daniel B Botkin & Edward A Keller Publisher: John Wiley & Sons.
6. Environmental Studies by R. Rajagopalan, Oxford University Press
7. Environmental Studies by Benny Joseph, TMH publishers
8. Environmental Studies by Dr. Suresh K Dhameja, 2007 Published by : S K Kataria & Sons New Delhi
9. Basics of Environmental Studies by U K Khare, 2011 Published by Tata McGraw Hill.
10. Environmental Studies by N.Arumugam & V.Kumaresan, saras publication.

Scheme of Examination

Sl. No	Semester	Paper	Marks for theory	Marks for internal assessment	Total
1	Even sem	Environmental studies	70	30	100

ENVIRONMENTAL STUDIES (ONE-SEMESTER COMPULSORY CORE MODULE FOR B.VOC PROGRAMMES)

Scheme of examination

Question paper pattern

Sl.no	Types of questions	Marks	No. of questions	Total marks
1	Short notes	2	5	10
2	Medium type	5	4	20
3	Long answers	10	4	40
			Total	70 marks

Distribution of Internal assessment

- 3) ASSIGNMENT/SEMINAR-15 MARKS.**
- 4) CLASS TEST-15 MARKS.**

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

V. Write short notes for the following(any 5): (5x2=10)

- 7. -----
- 8. -----
- 9. -----
- 10. -----
- 11. -----
- 12. -----

PART-B

VI. Answer any 4 of the following: (4x5=20)

- 6. -----
- 7. -----
- 8. -----
- 9. -----
- 10. -----

PART –C

VII. Answer any 4 of the following: (4x10=40)

- 6. -----
- 7. -----
- 8. -----
- 9. -----
- 10. -----

Sl. no.	Biostatistics	Hrs
1.	Statistical concepts: Data structure, sampling methods, collection, classification and tabulation of data, graphical and diagrammatic representation, histogram, frequency polygon, frequency curve, bar graph, pie chart etc.	4
2.	Measure of Central Frequency: Mean, median, mode.	2
3.	Measure of dispersion of data: Range, semi-interquartile range, mean deviation, standard deviation, standard error, coefficient of variation, confidence limits.	5
4.	Types of distribution of data: Normal, Binomial, Poisson.	7
5.	Z-test, t-test, ANOVA, multiple comparisons, LSD and DMRT, Chi-square test.	4
6.	Regression estimate, correlation coefficient.	4
7.	Experimental designs, data transformation.	4

Sl. no.	Practical	Hrs
1.	Analytical Problems / calculations	15

MODEL QUESTION PAPER

Semester-IV

CODE NO: FPD 580

BIOSTATISTICS

TIME: 3 hrs

Max marks: 70

Instructions: Draw neat and labeled diagram wherever necessary.

PART-A

VIII. Write short notes for the following(any 5): (5x2=10)

- 13. -----
- 14. -----
- 15. -----
- 16. -----
- 17. -----
- 18. -----

PART-B

IX. Answer any 4 of the following: (4x5=20)

- 11. -----
- 12. -----
- 13. -----
- 14. -----
- 15. -----

PART –C

X. Answer any 4 of the following: (4x10=40)

- XI.**
- 11. -----
 - 12. -----
 - 13. -----
 - 14. -----
 - 15. -----

IV SEMESTER PRACTICAL EXAMINATION

**BIostatISTICS
PRACTICAL
SCHEME OF EXAMINATION**

DURATION: 3 Hours

Maximum Marks: 70

Practical proper: 60

Record marks: 10

NOTE :- Candidates are required to submit the records duly signed by the teacher-in charge and certified by the Head of the Department

4. Analytical Problems / calculations.

**JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(AUTONOMOUS)
Ooty road, Mysuru - 570025.**



DEPARTMENT OF BOTANY

**Schematic Syllabus under Choice Based Credit
System (CBCS) & Continuous Assessment Grading
Pattern (CAGP) as per UGC template**

w.e.f.

2019-2020

**JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE (AUTONOMOUS)
OOTY ROAD, MYSURU-25**

DEPARTMENT OF BOTANY

**PROCEEDINGS OF THE MEETING OF BOARD OF STUDIES FOR THE PROGRAMMES
B. Sc. IN BOTANY, BIOCHEMISTRY, MICROBIOLOGY (B.Sc. 07) AND CHEMISTRY,
BOTANY, ZOOLOGY, (B.Sc. 08) HELD ON 14 JUNE 2019 AT 11.00 AM IN THE CHAMBER
OF THE CHAIRMAN, DEPARTMENT OF BOTANY, JSS COLLEGE, OOTY ROAD,
MYSURU-25**

MEMBERS PRESENT	SIGNATURE
Dr. S Prathibha Associate Professor, Dept. of Botany, JSS College, Ooty Road Mysuru Chairman	
Dr. Sowmya Associate Professor, Department of Botany Yuvaraja's college, Mysuru Member (VC Nominee)	
Dr. Syed Fasihuddin Associate professor of Botany Govt. Science College Bengaluru- 560001 Member (AC Nominee)	
Dr. V. N. Muralidhar Associate professor of Botany Govt. first grade College Sira- 572137 Tumkur District Member (AC Nominee)	

At the outset, the Chairman, BOS in Botany, welcomed the members to the meeting of BOS and briefed about the agenda to be discussed. The following agenda were placed by the Chairman which were discussed and resolved as follows:

Agenda 1: To frame/ revise, discuss and approve the Scheme/ Syllabus under Choice Based Credit System for the programmes: B.Sc. in Botany, Biochemistry, Microbiology and Chemistry, Botany, Zoology from the academic year 2019-20 onwards.

The Chairman appraised the members about the introduction of Choice Based Credit System to the above said programmes with the course matrix in 2017-18. Accordingly, a draft revised/ modified Scheme/ Syllabus was presented and placed before the Board for their opinion and approval.

Resolution: The BOS went through the Scheme/ Syllabus and discussed in length about various aspects of the same. After incorporation of the changes suggested by the members of BOS, the Syllabus was approved. Details of changes made with respect to the introduction of revised/ modified Scheme/ Syllabus in the existing courses is shown in Annexure-I.

Agenda 2: To prepare the Panel of Examiners for the examinations for the year 2019-20.

The Chairman presented the proposed Panel of Examiners to I to VI Semester examinations of 2019-20.

Resolution: After incorporating of certain changes suggested by the members, the Panel of Examiners was approved.

Agenda 3: Approval of Reference Books

The Chairman presented the proposed list of Reference Books to the Members.

Resolution: After incorporating of certain changes suggested by the members, the list of Reference Books was approved.

Agenda 4: Any other matter with the permission of the Chairman

--NIL--

Finally the meeting was concluded with the Chairman thanking the Members for their active participation in the deliberations of the meeting.

Chairman

Programme Outcomes for BSc. in Chemistry, Botany, Zoology:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1.** Demonstrate the ability to justify and explain their thinking and/or approach, both written and oral. Demonstrate the ability to present clear, logical and succinct arguments, including prose and mathematical language. Write and speak using professional norms, and demonstrate an ability to collaborate effectively.
- PO2.** Develop state-of-the-art laboratory skills and professional communication skills.
- PO3.** Apply the scientific method to design, execute, and analyze an experiment and also to explain their scientific procedures as well as their experimental observations.
- PO4.** Appreciate the central role of chemistry in our society and use this as a basis for ethical behaviour in issues facing chemists including an understanding of safe handling of chemicals, environmental issues and key issues facing our society in energy, health and medicine.
- PO5.** Explain why chemistry is an integral activity for addressing social, economic, and environmental problems.
- PO6.** Identify the taxonomic position of plants using principles and methods of nomenclature and classification in Botany.
- PO7.** Understand the impact of the plant diversity in societal and environmental contexts, and demonstrate the knowledge of, and need for sustainable development.
- PO8.** Use interdisciplinary approaches with quantitative skills to work on biological problems.
- PO9.** Identify the major groups of organisms with an emphasis on animals and be able to classify them within a phylogenetic framework.
- PO10.** Compare and contrast the characteristics of animals that differentiate them from other forms of life.
- PO11.** Give specific examples of the physiological adaptations, development, reproduction and behaviour of different forms of life.

Programme specific Outcomes for BSc. in Chemistry, Botany and Zoology

After completing the graduation in the Bachelor of Science the students are able to:

- PS01:** Communicate effectively the fundamentals and applications of chemical and Biological sciences

PS02: Possess deeper understanding of Natural laws, accuracy and validity of both theoretical and practical knowledge

PS03: Explicate ecological interconnectedness of life, by tracing energy and nutrient flows through the environment

PS04: Analyse the avenues and remedies for burning environmental issues

PS05: Pursue, enhance and appreciate conservation practices for sustainable use of plants and development

PS06: Interact with the social activities with ethical approach due to collaborative field visits, botanical tours and academic trips.

Programme Outcome for Bachelor of Science in Botany, Biochemistry & Microbiology

After completing the graduation in the Bachelor of Science the students are able to:

PO1. Identify the taxonomic position of plants using principles and methods of nomenclature and classification in Botany

PO2. Understand the impact of the plant diversity in societal and environmental context

PO3. Demonstrate the knowledge of, and need for sustainable development

PO4. Use interdisciplinary approaches with quantitative skills to work on biological problems

PO5. Demonstrate the ability to justify and explain their thinking and/or approach

PO6. Develop state-of-the-art laboratory and professional communication skills

PO7. Apply the scientific method to design, execute, and analyze an experiment

PO8. Explain scientific procedures and their experimental observations

PO9. Demonstrate an understanding of fundamental biochemical principles, structure and function

PO10. Work as a laboratory technician, biochemists or medical scientist

PO11. Explain the processes used by microorganisms for the growth

PO12. Explain the theoretical basis of the tools, technologies and methods of microbiology

Programme Specific Outcome

Bachelor of Science in Botany, Biochemistry & Microbiology

After completing the graduation in the Bachelor of Science the students are able to;

PSO1: Demonstrate applications of biochemical and biological sciences

PSO2: Inculcating proficiency in all experimental techniques and methods of analysis

PSO3: Acquire, articulate, retain and demonstrate laboratory safety skills

PSO4: Communicate scientific information effectively, relating to microbes and their role in ecosystem and health

PSO5: Gain proper procedures and regulations in handling and disposal of chemicals

PSO6: Understand biochemical and molecular processes that occur in and between the cells

I B.Sc., I Semester DSC-I
Biodiversity of Microbes and Archegoniate
Theory (Credits: 4)

Lectures: 60 Hours
(4 hours/week)

Course outcome:

After completion of the course the student is able to:

- CO1 Understand the characteristics of viruses
- CO2 Learn the classification and characteristics of bacteria
- CO3 Identify the classification and characteristics of archegoniate
- CO4 Identify the characteristics of algae
- CO5 Understand the classification and characteristics of fungi.

Unit 1- Microbial diversity:

A. Virus

(4 Lectures)

General structure, DNA virus (T₄-phage); Lytic and lysogenic cycle, RNA virus (TMV); Economic importance.

B. Bacteria

(5 Lectures)

Definition, Classification (Based on Shape, Arrangement and flagellation) and Economic importance; ultra structure, Reproduction – vegetative (fission, Budding) asexual (Endospore) Sexual (Genetic recombination-Conjugation, Transformation and Transduction).

C. Algae

(14 Lectures)

General characteristics; Classification, Reproduction and Economic importance of algae.

Type study- *Nostoc*, *Spirogyra*, *Sargassum*.

D. Fungi

(16 Lectures)

1. General characteristics, classification, nutrition, reproduction and economic importance.

Type Study - *Rhizopus*, *Saccharomyces* (Yeast), *Penicillium*,

2. Lichens: Distribution, classification and Economic importance, structure and reproduction.

Unit 2- Archegoniate:

A. Bryophytes

(7 Lectures)

General characteristics, Classification and Economic importance. Type Study- *Marchantia* and *Polytrichum*

B. Pteridophytes

(7 Lectures)

General characteristics and classification, Stellar evolution. Type Study - *Selaginella* and *Equisetum*.

C. Gymnosperms

(7 Lectures)

General characteristics, classification and Economic importance. Type Study- *Cycas* and *Pinus*.

DMA2318/ DMA2317

**I B.Sc., I Semester DSC-I
Biodiversity of Microbes and Archegoniate
Practical (Credits: 2)**

**Lectures: 60 Hours
(4 hours/week)**

1. Study of TMV and of T₄- Phage through Microphotographs
2. Study of Bacteria.
3. Study of *Nostoc* (Specimen and permanent slides)
4. Study of *Spirogyra* (Specimen and permanent slides)
5. Study of *Sargassum* (Specimen and permanent slides)
6. Study of *Rhizopus*
7. Study of *Yeast*
8. Study of *Penicillium*
9. Study of Lichens
10. Study of *Marchantia*
11. Study of *Polytrichum*
12. Study of *Selaginella*
13. Study of *Equisetum*
14. Study of *Cycas*
15. Study of *Pinus*

Suggested Readings

1. Kumar, H.D. (1999). Introductory Phycology. Affiliated East-West. Press Pvt. Ltd. Delhi. 2nd edition.
2. Tortora, G.J., Funke, B.R., Case, C.L. (2010). Microbiology: An Introduction, Pearson Benjamin Cummings, U.S.A. 10th edition.
3. Sethi, I.K. and Walia, S.K. (2011). Text book of Fungi & Their Allies, MacMillan Publishers Pvt. Ltd., Delhi.
4. Alexopoulos, C.J., Mims, C.W., Blackwell, M. (1996). Introductory Mycology, John Wiley and Sons (Asia), Singapore. 4th edition.
5. Raven, P.H., Johnson, G.B., Losos, J.B., Singer, S.R., (2005). Biology. Tata McGraw Hill, Delhi, India.
6. Vashishta, P.C., Sinha, A.K., Kumar, A., (2010). Pteridophyta, S. Chand. Delhi, India.
7. Bhatnagar, S.P. and Moitra, A. (1996). Gymnosperms. New Age International (P) Ltd Publishers, New Delhi, India.
8. Parihar, N.S. (1991). An introduction to Embryophyta. Vol. I. Bryophyta. Central Book Depot, Allahabad.

I B.Sc., I Semester DSC-I
Scheme of theory question paper
Biodiversity of Microbes and Archegoniate

Time: 3.00 Hours

Max. Marks: 70

Blue print:

Units	Hours Allotted	No. of questions from each category			Total Marks 70
		2 marks (5/8)=10	5marks (4/6)=20	10marks (4/6)=40	
Unit I :Microbial Diversity					
A&B.Virus and Bacteria	09	2X1=2	5X1=5	10X1=10	17
C. Algae	14	2X2=4	5X2=10	10X1=10	24
D. Fungi	16	2X1=2	5X3=15	10X1=10	27
Unit II: Archegoniate					
A. Bryophytes	07	2X1=2	-	10X1=10	12
B. Pteridophytes	07	2X1=2	-	10X1=10	12
C. Gymnosperms	07	2X2=4	-	10X1=10	14
Total	60	8X2=16	4X6=30	6X6=60	106

I B.Sc., I Semester DSC-I
Biodiversity of Microbes and Archeogoniate
Scheme of practical question paper

Time: 4 Hours**Max. Marks: 35(25+05+05)****I. Identify the specimens 'A' and 'B' with reasons and labeled sketches****3x2=06 marks**

(A-Algae and B-Microphotographs of virus/Bacteria/fungi)

Identification – 1 mark
Reasons with labelled sketch – 2 marks

**II. Prepare a stained temporary slide of 'C'. Sketch, label and Identify with reasons.
Leave the preparation for evaluation.****4x01= marks**(C-*Nostoc/Rhizopus/Saccharomyces* (yeast) / *Penicillium*)

Identification – 1 marks
Preparation/staining and mounting – 2 marks
Reasons with labelled sketch – 1 marks

III. Write critical notes on 'D', 'E' and 'F'**3x3=09 marks**

(D-Algae/Fungi, E-Lichens/Bryophytes, F- Pteridophytes /Gymnosperms)

Identification – 1 mark
Reasons with lablled sketch – 2 marks

**IV. Identify the Microslides 'G', 'H', and 'I' with reasons and labeled
Sketches****02x03=06 marks**

(G-Algae/Fungi, H-Lichens/Bryophytes, I- Pteridophytes /Gymnosperms)

Identification – 1 mark
Reasons with lablled sketch – 1 marks

V. Practical record**05marks****VI. Viva- Voce****05marks**

Note: Each student should submit the **duly valued and certified practical record** at the time of practical examination.

DMA2318/DMA2317

**I B.Sc., I Semester DSC-I
Biodiversity of Microbes and Archegoniate
Practical Question paper**

Time: 4 Hours

Max. Marks: 35 (25+05+05)

I. Identify the specimens 'A' and 'B' with reasons and labeled sketches 3x2=06 marks

II. Prepare a stained temporary slide of 'C'. Sketch, label and Identify with Reasons leave the preparation for evaluation. 4x1=04 marks

III. Write critical notes on 'D', 'E' and 'F' 5x3=15 marks

IV. Identify the Microslides 'G', 'H' and 'I' and with reasons and labeled Sketches 3x3=09 marks

V. Practical record 05 marks

VI. Viva- Voce 05marks

I B.Sc., II Semester DSC-II
Plant Ecology, Morphology and Angiosperm Taxonomy
Theory (Credits: 4)

Lectures: 60 Hours
(4 hours/week)

Course outcome:

After completion of the course the student is able to:

- CO1 Specify the characteristics of ecosystem
- CO2 Learn the classification and characteristics of Plant communities
- CO3 Understand in details with examples plant morphology
- CO4 Understand in depth Herbarium

Unit 1- Plant Ecology:

A. Introduction to Ecology and Ecological factors: (6 Lectures)

Introduction to ecology, Climatic factors- Light, temperature and water. Edaphic factors- soil formation, types and profile. Shelford law of tolerance.

B. Ecosystem (6 Lectures)

Structure and components of an ecosystem, study of pond and forest ecosystem, energy flow and trophic levels; Food chains, food webs, Ecological pyramids. Biogeochemical cycles- Carbon, Nitrogen and Phosphorous.

C. Plant communities (6 Lectures)

Morphological Adaptations of hydrophytes and xerophytes. Plant Succession, Hydrosere and Xerosere.

Unit 2- Leaf and Floral Morphology: (15 Lectures)

- A. Leaf- Structure, types and phyllotaxy.
- B. Types of Inflorescence
- C. Flower- structure of a typical flower (*Tribulus terrestris* / *Muntingia calabura*), Variation in floral morphology and floral organs in detail.
- D. Types of Inflorescence
- E. Types of fruits

Unit-3: Taxonomy:

A. Introduction to plant taxonomy (8 Lectures)

- 1. Taxonomic hierarchy
- 2. Types of classification (artificial, natural and phylogenetic)
- 3. Systems of classification- Bentham and Hooker, Engler and Prantl
- 4. Plant Nomenclature-Binomial system and ICBN principles.

B. Herbarium technique:

(5 Lectures)

1. Herbarium- Techniques and importance
2. Botanical gardens

C. Angiosperm families:

(14 Lectures)

Study of the following families according to Bentham and Hooker's system of classification. -
-Malvaceae, Leguminosae (Papilionaceae, Caesalpiniaceae and Mimosaceae), Apocynaceae
Asteraceae and Arecaceae.

I B.Sc., II Semester DSC-II
Plant Ecology, Morphology and Angiosperm Taxonomy
Practical (credits: 2)

Lectures: 60 Hours
(4 hours/week)

1. Study of Ecological instruments used to measure microclimatic variables: Soil thermometer, Maximum and Minimum Thermometer, Anemometer, Psychrometer/Hygrometer, Rain gauge.
2. Study of morphological adaptations of the following
 - a. Hydrophytes Eg: *Hydrilla. Pistia and Eichhornia*
 - b. Xerophytes Eg: *Opuntia, Euphorbia Tirucalli, Nerium and Casuarina*
3. Study of biotic interactions of the following:
 - a. Stem parasite Eg: *Cuscuta.*
 - b. Root parasite Eg: *Striga.*
 - c. Epiphytes, Eg: *Vanda*
 - d. Predatory plants (Insectivorous plants) Eg: *Nepenthes.*
4. Study of root modifications
5. Study of stem modifications
6. Study of leaf-structure, types, phyllotaxy and modifications.
7. Parts of a typical flower (*Tribulus terrestris / Muntingia calabura*)
8. Floral organs in detail with their variations.
9. Types of inflorescence
10. Types of fruits
11. Study of families Malvaceae, Apocynaceae
12. Study of families Leguminosae (Papilionaceae, Caesalpinaceae and Mimosaceae)
13. Study of families and Asteraceae
14. Study of Solanaceae and Arecaceae
15. Mounting of a properly dried and pressed specimen of any wild plant with herbarium label

Suggested Readings:

1. Kormondy, E.J. (1996). Concepts of Ecology. Prentice Hall, U.S.A. 4th edition.
2. Sharma, P.D. (2010) Ecology and Environment. Rastogi Publications, Meerut, India. 8th edition.
3. Simpson, M.G. (2006). Plant Systematics. Elsevier Academic Press, San Diego, CA, U.S.A.
4. Singh, G. (2012). Plant Systematics: Theory and Practice. Oxford & IBH Pvt. Ltd., New Delhi.

I B.Sc., II Semester DSC-II
Plant Ecology, Morphology and Angiosperm Taxonomy
Scheme of theory question paper

Time: 3.00 Hours

Max. Marks: 70

Units	Hours Allotted	No. of questions from each category			Total Marks 70
		2 marks (5/8)=10	5marks (4/6)=20	10marks (4/6)=40	
Unit 1: A. Plant Ecology	06	-	-	10x1=10	10
B. Ecosystem	06	-	5x2=10	-	10
C. Plant communities	06	2x1=2	-	10x1=10	12
Unit II: Leaf and Floral Morphology, Fruits	15	2x3=6	5x2=10	10x1=10	26
Unit III: Taxonomy					
A. Introduction to plant taxonomy	08	2x2=4	-	10x1=10	14
B. Herbarium technique	05	2x1=2	5x1=5	-	07
C. Angiosperm families	14	2x1=2	5x1=5	10x2=20	27
Total	60	8x2=16	5x6=30	10x6=60	106

DMA23108/ DMA23107

I B.Sc., II Semester DSC-II
Plant Ecology, Morphology and Angiosperm Taxonomy
Scheme of practical question paper

Time: 4 Hours

Max. Marks: 35 (25+05+05)

I. Write critical notes on 'A' 'B' and 'C' with reasons and labeled sketches 3x3=9 marks

(A-Ecological instruments, B-Hydrophytes/xerophytes/parasites/epiphytes,
C-Underground root & stem modifications/Leaf phyllotaxy /leaf types/)

Identification – 1 mark

Labeled sketch with reasons – 4marks

II. Assign the plant 'D' to its respective family giving reasons. 4x1=4 marks

(D- Malvaceae/Apocynaceae/Asteraceae/Araceae)

Family name – 1 mark

Salient features – 3 marks

III. Describe the plant 'E' in technical terms. 4x1=4 marks

(Papilionaceae /Caesalpinaceae)

Family name – 1 mark

Technical terms – 3 marks

IV. Draw the floral diagram and write the floral formula of the give plant 'F' 4x1=4 marks

(Malvaceae, Solanaceae, Apocynaceae)

Floral formula -1mark

Floral diagram-3marks

V. Identify the specimen 'G' and 'H' 2x2=4marks

(J-Inflorescence, H- Fruits)

Identification – 1 mark

Reasons – 1mark

VI. Practical record 05marks

VII. Viva- Voce 05marks

Note: each student should submit the **duly valued and certified practical record** at the time of practical examination.

**I B.Sc., II Semester DSC-II
Plant Ecology, Morphology and Angiosperm Taxonomy
Practical Question Paper**

Time: 4 Hours

Max. Marks: 35 (25+05+05)

- I. Write critical notes on 'A' 'B' and 'C' with reasons and labeled sketches** **3x3=9 marks**
- II. Assign the plants 'D' to its respective family giving reasons.** **4x1=4marks**
- III. Describe the plant 'E' in technical terms.** **4x1=4 marks**
- IV. Draw the floral diagram and write the floral formula of the give plant 'F'** **4x1=4marks**
- V. Identify the specimen 'G' and 'H'** **2x2=4marks**
- VII. Practical record** **05marks**
- VIII. Viva-Voce** **05marks**

**Plant Anatomy and Embryology of Embryology
Theory (Credits: 4)**

**Lectures: 60 Hours
(4 hours/week)**

Course outcome:

After completion of the course the student is able to:

- CO1 Understand the details of histology
- CO2 Understand the details of anatomy
- CO3 Understand the characteristics of secondary growth
- CO4 Learn the details of embryology

Unit 1: Histology and Anatomy (18 Lectures)

Meristem- structure, classification, based on origin, position and function. Study of Simple and complex tissues.

Internal Structure of dicot and monocot root stem and leaf.

Unit 2: Secondary Growth (4 Lectures)

Process of secondary growth in dicot stem.

Unit 3: Adaptive and protective systems (7 Lectures)

Anatomical adaptations in xerophytes (Nerium & Causarina) and Hydrophytes (Hydrilla & Eichhornia). Epidermis, cuticle, Stomata & Trichome. (Brief Account)

Unit 4: Embryology of Angiosperm (8 Lectures)

T.S of mature anther, Microsporogenesis, types of tetrads and Male gametophyte, Megasporogenesis- types of ovules, L.S of Anatropous ovule, Embryosac development- Monosporic (Polygonum), Bisporic (Allium), Tetrasporic (Frittilaria) structure of Mature Embryo sac.

Unit 5: Pollination and fertilization (11 Lectures)

Definition, types, contrivances for self and cross pollinations

Process of Double Fertilization, Post Fertilization changes

Unit 6: Embryo and endosperm (6 Lectures)

Structure and development of Dicot (Capsella) and Monocot embryo (Maize). Endosperm- nucellar, cellular, helobial and ruminant.

Unit 7: Experimental Embryology (6 Lectures)

Brief account of apomixis (recurrent and non-recurrent), apospory, polyembryony, parthenocarpy.

1. Study of Meristems through permanent slides and photographs- apical, intercalary and lateral
2. Study of simple Tissues (parenchyma, collenchyma and sclerenchyma) through Permanent slides, photographs.
3. Study of complex Tissue (Permanent slides and photographs).
4. Study of Anatomical characteristics of Root: Monocot- *Zea mays* and Dicot- *Helianthus*.
5. Study of Anatomical characteristics of Stem: Monocot- *Zea mays*; Dicot-*Helianthus*.
6. Study of Anatomical characteristics of Monocot and Dicot Leaf.
7. Adaptive anatomy: Xerophyte (*Nerium* & *Causaurina*); Hydrophyte (*Hydrilla* & *Eichhornia*).
8. Structure of anther (young and mature) and mounting of Pollen grains.
9. Calculation of percentage of germinated pollen in a given medium (Hanging drop method).

10. Types of ovules: Anatropous, Orthotropous, Circinotropous, Amphitropous/ Campylotropous.
11. Female gametophyte: Polygonum (monosporic) type of Embryo sac Development.
12. Pollination types and seed dispersal mechanisms. (Photographs and specimens).
13. Dissection/ mounting of embryo/endosperm from seeds.

Suggested Readings

1. Bhojwani, S.S. & Bhatnagar, S.P. (2011). Embryology of Angiosperms. Vikas Publication House Pvt. Ltd. New Delhi. 5th edition.
2. Mauseth, J.D. (1988). Plant Anatomy. The Benjamin/Cummings Publisher, USA.

DMC23008/DMC23007

II B.Sc., III Semester DSC-III
Plant Anatomy and Embryology of Angiosperms
Scheme of theory question paper

Time: 3.00 Hours

Max. Marks: 70

Blue print:

Units	Hours Allotted	No. of questions from each category			Total Marks 70
		2 marks (5/8)=10	5marks (4/6)=20	10marks (4/6)=40	
Unit 1: Histology and Anatomy	18	2X1=2	5X2=10	10X2=20	32
Unit 2: Secondary Growth	04	2X1=2	5X1=5	-	07
Unit 3: Adaptive and protective systems	07	2X1=2	5X2=10	-	12
Unit 4: Embryology	08	2X1=2	-	10X1=10	12
Unit 5: Pollination and fertilization	11	2X1=2	-	10X2=10	22
Unit 6: Embryo and endosperm	06	2X1=2	-	10X1=10	12
Unit 7: Experimental Embryology	06	2X2=4	5X1=5	-	09
Total	60	8X2=16	5X6=30	10X6=60	106

II B.Sc., III Semester DSC-III
Plant Anatomy and Embryology of Angiosperms
Scheme of practical question paper

Time: 4 Hours**Max. Marks: 35 (25+05+05)**

I. Write critical notes on 'A' 'B' and 'C' with reasons and labeled sketches **2x3=6 marks**
(A-Meristem- Apical, intercalary and lateral, B-Simple tissues,
C-Complex tissues)

Identification – 1 mark

Labeled sketch with reasons – 1marks

II. Prepare a stained temporary slide of 'D. Sketch, label and identify with reasons.

Leave the preparation for evaluation**5x1=5 marks**

(D-Dicot stem/ Monocot stem Anatomy)

Identification – 1 mark

Mounting and Preparation -2 marks

Labeled sketch with reasons – 2 marks

III. Identify the microslides/ photographs 'E' 'F' & 'G' sketch, label with reasons

3x3=09 marks

(E-root/ leaf, F-Xerophytes/ Hydrophytes, G-T.S of anther/ types of ovule)

Identification – 1 mark

Labeled sketch with reasons – 2marks

IV. 'H'- Dissect Embryo/Endosperm, sketch label with reasons/ Mounting of Pollen grains / calculate the percentage of germinated pollen

5x1=5 marks

Identification – 1 mark

Labeled sketch with reasons – 4marks

V. Practical record**05 marks****VI. Viva-Voce****05 marks**

Note: Each student should submit the **duly valued and certified practical record** at the time of practical examination.

DMC23108/DMC23107

**II B.Sc., III Semester DSC-III
Plant Anatomy and Embryology of Angiosperms
Practical question paper**

Time: 4 Hours

Max. Marks: 35 (25+05+05)

I. Write critical notes on 'A' 'B' and 'C' with reasons and labeled sketches 2x3=6 marks

II. Prepare a stained temporary slide of 'D'. Sketch, label and identify with reasons. Leave the preparation for evaluation 05marks

III. Identify the microslides/ photographs 'E' 'F' & 'G', sketch, label with reasons 3x3=9 marks

IV. 'H'- Dissect Embryo/Endosperm, sketch label with reasons/ Mounting of Pollen grains / calculate the percentage of germinated pollen.

5x1=5 marks

V. Practical record

5 marks

VI. Viva-Voce

5 marks

DMD23008/DMD23007

**II B.Sc., IV Semester DSC-IV
Plant Physiology and Metabolism
Theory (Credits: 4)**

**Lectures: 60 Hours
(4 hours/week)**

Course outcome:

After completion of the course the student is able to:

- CO1 Learn in depth translocation in phloem
- CO2 Specify the classification and characteristics of enzyme
- CO3 Understand the details of photosynthesis
- CO4 Identify the characteristics of plant response to light and temperature

UNIT 1 - Plant – Water Relations: (20 Lectures)

1. **Fundamental concepts:** Importance of water to plants, Diffusion, Imbibition, Osmosis, Endosmosis and Exosmosis, Plasmolysis, Osmotic Pressure, Water potential and its components.
2. **Short Distance Transport:** Absorption of water – Active and passive absorption, Absorption of minerals – Donnan's equilibrium (Passive absorption), Carrier-ion concept (Active absorption)
3. **Long distance Transport:** Ascent of sap; Root pressure theory (Vital theory), TCT Theory (Physical theory), Soil plant atmospheric continuum (SPAC)
4. **Transpiration:** Types, Mechanism of stomatal movement; Starch-sugar interconversion theory, Potassium ion pump theory, Significance of transpiration, Antitranspirants, Guttation.
5. **Mineral nutrition:** Macro and Micro nutrients; Role of Nitrogen, Phosphorous, Potassium, Sulphur, Manganese and Zinc, Hydroponics.
6. **Translocation of solutes:** Path of translocation, Munch's mass flow hypothesis with merits and demerits.

UNIT 2 – Enzymes: (4 Lectures)

Properties, Classification and Mode of action (Lock & Key theory, Induced fit theory)

UNIT 3 – Bioenergetics: (18 Lectures)

1. **Photosynthesis :** Introduction, Photosynthetic apparatus, Mechanism – Light and Dark reactions (C_3 pathway/Calvin Cycle), C_4 pathway, Significance of Photosynthesis.
2. **Respiration :** Introduction,,Types, Ultrastructure of Mitochondrion, Mechanism of Aerobic respiration – Glycolysis, Krebs' cycle and Terminal Oxidation of reduced coenzymes , Anaerobic respiration – alcoholic & lactic acid fermentation, Significance .

UNIT 4 -Nitrogen Metabolism: (6 Lectures)

Nitrogen fixation (Symbiotic and Non Symbiotic), Nitrate reduction, Aminoacids & their synthesis (Transamination & Reductive amination)

UNIT 5 - Plant Growth and Movements: (12 Lectures)

1. Growth: Definition, Phases of growth and Growth curve
2. Growth regulators Chemical nature, application of Auxins, Gibberellins, Cytokinins, Abcissic acid (ABA) & Ethylene.
3. Photoperiodism and Vernalisation : A brief account.
4. Plant movements : Tropisms & their types (Phototropism, Thigmotropism, Hydrotropism & Geotropism)

I. Minor Experiments:

1. a) Root pressure experiment
b) Ganong's Potometer experiment
2. c) Ganong's light screen experiment
d) Mohl's half leaf experiment
3. f) Kuhne's experiment to demonstrate fermentation.
g) Phototropism
4. h) Geotropism
i) Arc Auxanometer experiment
5. j) Bolting.
k) Effect of auxins on rooting.

II. Major Experiments:

1. Determination of Osmotic Potential by Plasmolytic method using *Rhoeo discolor* (Epidermal peel)
2. Experiment to demonstrate the Relationship between Absorption and Transpiration.
3. Experiment to demonstrate the Suction force due to transpiration.
4. Separation of photosynthetic pigments using paper Chromatography.
5. Evolution of Oxygen during photosynthesis
6. Calculation of stomatal index and stomatal frequency.

III. Biochemical tests:

Qualitative biochemical tests for Carbohydrates, fats and protein.

Suggested Readings

1. Taiz, L., Zeiger, E., (2010). Plant Physiology. Sinauer Associates Inc., U.S.A. 5th Edition.
2. Hopkins, W.G., Huner, N.P., (2009). Introduction to Plant Physiology. John Wiley & Sons, U.S.A. 4th Edition.
3. Bajracharya, D., (1999). Experiments in Plant Physiology- A Laboratory Manual. Narosa Publishing House, New Delhi.

**II B.Sc., IV Semester DSC-IV
Plant Physiology and Metabolism
Scheme of theory question paper**

Time: 3.00 Hours

Max. Marks: 70

Blue print:

Units	Hours Allotted	No. of questions from each category			Total Marks 70
		2 marks (5/8)=10	5marks (4/6)=20	10marks (4/6)=40	
Unit 1: Plant-water relations	20	2X3=6	5X2=10	10X2=20	36
Unit 2: Enzymes	4	2X1=2	5X1=5	-	7
Unit 3: Bioenergetics	18	2X2=4	5X1=5	10X2=10	29
Unit 4: Nitrogen metabolism	6		5X1=5	10X1=10	15
Unit 5: Plant Growth and Movements:	12	2X2=4	5X1=5	10X1=10	19
Total	60	8X2=16	5X6=30	10X6=60	106

DMD23108/DMD23107

II B.Sc., IV Semester DSC-IV
Plant Physiology and Metabolism
Scheme of practical question paper

Time: 4 Hours

Max. Marks: 35 (25+05+05)

I. Perform the major experiment 'A' write the principle, Requirements, Procedure and record the result with inference and leave the setup for evaluation 9x1 = 9 marks

(Determination of osmotic potential of plant cell sap by plasmolytic method. Study of plasmolysis and deplasmolysis on Rhoeo leaf

OR

Calculation of stomatal index and stomatal frequency of a mesophyte and a xerophytes

OR

Experiment to demonstrate the Relationship between Absorption and Transpiration.

OR

Separation of photosynthetic pigments by paper chromatography

OR

Evolution of Oxygen during photosynthesis

OR

Experiment to demonstrate the Suction force due to transpiration

Principle	-2 marks	Procedure	-3 marks
Requirements	-1 mark	Result and inference	-1 marks
Setting	-2 marks		

II. Comment on 'B' & 'C' (Minor experiments) 4x2 =8 marks

(Root pressure experiment/ Ganong's Potometer experiment / Ganong's light screen experiment/ Mohl's half leaf experiment/ Kuhne's experiment to demonstrate fermentation/ Phototropism/ Arc Auxanometer experiment/ Bolting/ Effect of auxins on rooting)

Identification -1 marks

Critical notes -2 marks

Labeled sketch -1 marks

III. Perform the biochemical test of the given sample 'D' & 'E'

Procedure -3marks Result-1 marks 4x2=8 marks

IV. Practical record 05 marks

V. Viva-Voce 05 marks

DMD23108/DMD23107

**II B.Sc., IV Semester DSC-IV
Plant Physiology and Metabolism
Practical Question Paper**

Time: 4 Hours

Max. Marks: 35 (25+05+05)

I. Perform the major experiment 'A' write the principle, Requirements, Procedure and record the result with inference and leave the setup for evaluation **9x1 =9 marks**

II. Comment on 'B' & 'C' (Minor experiments) **4x2=8 marks**

II. Perform the biochemical test of the given sample 'D' & 'E' **4x2=8 marks**

IV. Practical record **05 marks**

V. Viva-Voce **05 marks**

DME23008/DME23007

**III B.Sc., V Semester DSC-V
Cell and Molecular Biology
Theory (4 credits)**

**Lectures: 60 Hours
(4 hours/week)**

Course outcome:

After completion of the course the student is able to:

- CO1 Understand in depth microscopy
- CO2 Learn the details of cell
- CO3 Specify the details of DNA
- CO4 Learn the details of gene regulation

Unit 1- Microscopy:

(5 Lectures)

Principles of microscopy; Light Microscopy; Phase contrast microscopy; Fluorescence microscopy; Electron microscopy (EM)- Scanning EM and Scanning Transmission EM (STEM).

Unit 2- Cell:

(4 Lectures)

Cell Theory; Ultra structure of Prokaryotic and eukaryotic cells;

Unit 3- Cell Wall and Cell Membrane:

(4 Lectures)

Cell wall- Structure; Cell Membrane- Fluid mosaic model and functions

Unit 4- Cell Organelles:

(13 Lectures)

Ultrastructure and functions of Nucleus, Mitochondrion, Chloroplast, Endoplasmic reticulum, Golgi bodies, Lysosomes, Ribosomes, Peroxisomes and Glyoxisomes

Morphology of chromosomes in general, Ultrastructure of Chromosome (Nucleosome concept),

Karyotype and Ideogram

Unit 5-Cell Division:

(6 Lectures)

Cell cycle, Mitosis and Meiosis and their significance

Unit 6- Nucleic acids:

(12 Lectures)

A. DNA: Chemistry, Structure and Replication in Eukaryotes (semi- conservative method) DNA as a genetic material Griffith's and Avery's transformation experiments, Hershey-Chase bacteriophage experiment.

B.RNA: Chemistry, Structure, Types (mRNA, tRNA, rRNA) and structure. Frankel Conrat's experiment

Unit 7-Gene concept:

(12 Lectures)

A. Cistron, Recon, Muton- Prokaryotic and Eukaryotic gene structure, Split gene concept.

B. Genetic code- features, Wobble concept. Protein synthesis: Transcription, Splicing and Translation.

Central dogma of molecular Biology.

Unit 8-Regulation of gene expression:

(4 Lectures)

Lac operon and Tryp operon concepts.

DME23108/DME23107

**III B.Sc., V Semester DSC-V
Cell and Molecular Biology
Practical (2 credits)**

**Lectures: 60 Hours
(4 hours/week)**

1. Preparation of fixatives and stains: FAA, Carnoy's fixative, safranin, acetocarmine and aceto-orcein.
2. Study of viruses, prokaryotic cell (bacteria) and eukaryotic cell with the help of light and electron micrographs.
3. Study of cell organelles through photographs.
4. To study the structure of plant cell through temporary mounts- Onion peeling and tomato pulp
5. Study of Mitosis (temporary mounts and permanent slides).
6. Study of Meiosis (temporary mounts and permanent slides).
7. Study the structure of nuclear pore complex by photograph (from Gerald Karp)
8. Structure of DNA and RNA (mRNA, rRNA, tRNA).
9. Study DNA packaging through photographs- solenoid model.
10. Lac operon and Tryp operon concepts.

Suggested Readings

1. Karp, G. 2010. Cell and Molecular Biology: Concepts and Experiments. 6th Edition. John Wiley & Sons. Inc.
2. De Robertis, E.D.P. and De Robertis, E.M.F. 2006. Cell and Molecular Biology. 8th edition. Lippincott Williams and Wilkins, Philadelphia.
3. Cooper, G.M. and Hausman, R.E. 2009. The Cell: A Molecular Approach. 5th edition. ASM Press & Sunderland, Washington, D.C.; Sinauer Associates, MA. 4. Becker, W.M., Kleinsmith, L.J., Hardin. J. and Bertoni, G. P. 2009. The World of the Cell. 7th edition. Pearson Benjamin Cummings Publishing, San Francisco.

**III B.Sc., V Semester DSC-V
Cell and Molecular Biology
Scheme of theory question paper**

Time: 3.00 Hours

Max. Marks: 70

Blue print:

Units	Hours Alloted	No. of questions from each category			Total Marks 70
		2 marks (5/8)=10	5marks (4/6)=20	10marks (4/6)=40	
Unit 1: Microscopy	5	-	-	10X1=10	10
Unit 2: Cell	4	2X1=2	5X1=5	-	07
Unit 3: Cell Membrane and Cell Wall	4	2X1=2	5X1=5	-	07
Unit 4: Cell Organelles	13	2X1=2	-	10X2=20	22
Unit 5: Cell Division	06	2X1=2	-	10X1=10	12
Unit 6: Nucleic acid	12	2X1=2	5X2=10	10X1=10	22
Unit 7: Gene concept	12	2X2=4	5X1=5	10X1=10	19
Unit 8: Regulation of gene expression	04	2X1=2	5X1=5	-	07
Total	60	8X2=16	5X6=30	10X6=60	106

DME23108/DME23107

III B.Sc., V Semester DSE-I
Cell and Molecular Biology
Scheme of practical question paper

Time: 4 Hours

Max. Marks: 35 (25+05+05)

I. Prepare a temporary squash of given material 'A'. Sketch, label and identify with reasons. Leave the preparation for evaluation. 07 marks

(Onion root tip/ flower bud)

Preparation - 4marks

Identification - 1mark

Sketch and label - 1mark

Reasons - 1mark

II. Identify the cytological slide/Photograph 'B' with labeled diagram and reasons. 04 marks

(Mitosis/ Meiosis)

Identification - 1mark

Sketch and label - 1mark

Reasons - 2marks

III. Comment on 'C' and 'D' (charts/photographs) 3 X 2= 06marks

C- Cell organelle (Identification - 1mark, labeled Sketch and Reason - 2marks)

D- Fixative/ stain (Identification - 1mark, labeled Sketch and Reason - 2marks,)

IV. Prepare a temporary mount of a plant cell 'E' 04 marks

(Onion peeling/ tomato pulp)

Preparation - 2 marks

Sketch and label of eukaryotic cell - 3marks

V. Write critical notes on 'F and G 2 X 2 = 04 marks

F-DNA/ DNA packaging/RNA types G- Nuclear pore complex/ Lac operon/Tryp operon

VI. Practical record 05marks

VII. Viva-voce 05marks

DME23108/DME23107

**III B.Sc., V Semester DSE-I
Cell and Molecular Biology
practical question paper**

Time: 4 Hours

Max. Marks: 35 (25+05+05)

- I. Prepare a temporary squash of given material 'A'. Sketch, label and identify with reasons. Leave the preparation for evaluation. 07 marks**
- II. Identify the cytological slide/photograph 'B' with labeled diagram and reasons. 04 marks**
- III. Comment on 'C' and 'D' (charts/photographs) 3 X 2= 06 marks**
- IV. Prepare a temporary mount of a plant cell 'E' 04 marks**
- V. Write critical notes on F and G 2 X 2 = 04 marks**
- VI. Practical record 05 marks**
- VII. Viva-voce 05marks**

DME23208/DME23207

III B.Sc., V Semester DSE-II
Economic Botany and Biotechnology
Theory (4 credits)

Lectures: 60 Hours
(4 hours/week)

Course outcome:

After completion of the course the student is able to:

- CO1 Understand in details with application, if applicable, economic botany
- CO2 Specify the details of plant tissue culture
- CO3 Understand in details with examples recombinant DNA technology

Unit 1: Cereals and Millets (4 Lectures)

Rice, Wheat, Maize, Ragi (Botanical name, family, part used, morphology and uses)

Unit 2: Legumes (6 Lectures)

General account with special reference to Pigeon pea, Green gram, Black gram, Bengal gram (Botanical name, family, part used, morphology and uses)

Unit 3: Spices and condiments (8 Lectures)

General account with special reference to clove, black pepper, cinnamom, cardamom, garlic, onion, chilli and coriander (Botanical name, family, part used, morphology and uses)

Unit 4: Beverages (4 Lectures)

Tea and coffee (Botanical name, family, part used, morphology and uses)

Unit 5: Oils and Fats (4 Lectures)

General description with special reference to groundnut, sunflower, mustard (Botanical name, family, part used, morphology and uses)

Unit 6: Fibre Yielding Plants (4 Lectures)

General description with special reference to Cotton, Jute, kapok and sunn hemp (Botanical name, family, part used, morphology and uses)

Unit 7: Introduction to biotechnology (2 lecture)

Unit 8: Plant tissue culture (10 Lectures)

Micropropagation ; haploid production through androgenesis and gynogenesis; brief account of embryo & endosperm culture with their applications

Unit 9: Recombinant DNA Techniques (18 Lectures)

Blotting techniques: Northern, Southern and Western Blotting, DNA Fingerprinting; Molecular DNA markers i.e. RAPD, RFLP, SNPs; DNA sequencing, PCR and Reverse Transcriptase-PCR. Hybridoma and monoclonal antibodies, ELISA and Immunodetection.

DME23308/DME23307

**III B.Sc., V Semester DSE-II
Economic Botany and Biotechnology
Practical (2 credits)**

**Lectures: 60 Hours
(4 hours/week)**

1. Study of Cereals and Millets
2. Study of Legumes
3. Study of Spices and condiments
4. Study of Beverages
5. Study of Oils and Fats
6. Study of Fibre Yielding Plants
7. Familiarization with basic equipments in tissue culture.
8. Study through photographs: Anther culture, somatic embryogenesis,
9. Study through photographs: endosperm and embryo culture; micropropagation.
10. Study of molecular techniques: PCR, Blotting techniques and PAGE.

Suggested Readings

1. Kochhar, S.L. (2011). Economic Botany in the Tropics, MacMillan Publishers India Ltd., New Delhi. 4th edition.
2. Bhojwani, S.S. and Razdan, M.K., (1996). Plant Tissue Culture: Theory and Practice. Elsevier Science Amsterdam. The Netherlands.
3. Glick, B.R., Pasternak, J.J. (2003). Molecular Biotechnology- Principles and Applications of recombinant DNA. ASM Press, Washington.

III B.Sc., V Semester DSE-II
Economic Botany and Biotechnology
Scheme of Theory Question Paper

Time: 3.00 Hours

Max. Marks: 70

Blue print:

Units	Hours allotted	No. of questions from each category			Total Marks 70
		2 marks (5/8)=10	5marks (4/6)=20	10marks (4/6)=40	
Unit 1: Cereals and Millets	4		5X1=5	-	07
Unit 2: Legumes	6	2X1=2	-	10X1=10	12
Unit 3: Spices and condiments	8	-	5X1=5	10X1=10	15
Unit 4: Beverages	4	-	5X1=5	-	05
Unit 5: Oils and Fats	4	2X1=2	5X1=5	-	07
Unit 6: Fibre Yielding Plants	4	2X1=2	-	10X1=10	12
Unit 7: Introduction to biotechnology	2	2X1=2	-	-	02
Unit 8: Plant tissue culture	10	2X1=2	5X1=5	10X1=10	17
Unit 9: Recombinant DNA Techniques	18	2X2=4	5X1=5	10X2=20	29
Total		8X2=16	5X6=30	10X6=60	106

Floriculture**Theory (2 credits)****Lectures: 30 Hours
(2 hours/week)****Course outcome:**

After completion of the course the student is able to:

- CO1 Specify the classification and characteristics of gardening
- CO2 Understand in depth nursery management
- CO3 Identify in details with examples ornamental plants

Unit 1- Establishment and management of Nurseries:**(7 Lectures)**

Definition, importance of nurseries, classification of nurseries, and management of nurseries

1. **Basic requirements for Nurseries:** Agro-climatic conditions, Topography, Selection of site Selection of soil, Seed bed preparation, Water supply and irrigation.

Parts of nursery- a) Building structures, b) Propagating structures- raising of seedlings

2. **Management of nursery:** Irrigation, Nutrition, Weed control, Plant protection, Uprooting, packing and transplantation

Unit 2-Gardening and Landscaping**(8 Lectures)****A. Features of a garden:**

1. Introduction, living elements, hedges, edges, trees, flower beds, lawn, Shrubbery, climbers and creepers, paths, Steps, arches, pergola, rockery, Water garden, sunken garden, carpet beds, topiary, trophy, ,non living elements.
2. Gardening: Introduction, Formal style, Informal style, planning a garden, creating a garden, establishment of the garden.
3. Some Famous gardens of India

B. Landscaping:

1. Home garden, Public garden. Educational institution, commercial complexes and companies.

2. Importance, Scope of floriculture and landscape gardening.

Unit 3-Garden Plants:

(5 Lectures)

Introduction, Annuals, Biennials, Perennials, Shrubs, Trees, Climbers(Divine Vines) Succulents, Cacti,Ferns, Gymnosperms, Palms, Orchids, Bulbous Ornamentals.

Unit 4- Commercial Floriculture:

(8 Lectures)

Introduction, Importance of Floriculture from social, Economic, Health and Aesthetic point of view. Marketing and floristry in Indian scenario. Future and scope of Floriculture in India- Employment opportunities. Packaging of cut flowers; Flower arrangements; Methods to prolong vase life; Cultivation of Important cut flowers (Rose, Chrysanthemum, Gerbera, Gladiolous, Marigold, Orchids).

Unit 5- Diseases and pests of ornamental plants:

(2 Lectures)

References

1. Randhawa, G.S. and Mukhopadhyay, A. 1986. Floriculture in India Allied Publishers

Floriculture**Scheme of Theory Question Paper****Blue print:****Max. Marks: 50**

Units	Hours allotted	No. of questions from each category			Total Marks 50
		2 marks (5/8)=10	5marks (4/6)=20	10marks (2/4)=20	
Unit 1: Establishment and management of Nurseries.	7	2X2=4	5X1=5	10X1=10	19
Unit 2: Gardening and Landscaping:	8	2X2=4	5X2=10	10X1=10	24
Unit 3: Garden Plants:	5	2X2=4	5X2=10	-	14
Unit 4: Commercial Floriculture	8	2X1=2	-	10X2=20	22
Unit 5: Diseases and Pests of Ornamental Plants.	2	2X1=2	5X1=5	-	07
Total	30	8X2=16	5X6=30	10X4=40	86

Course outcome

After completion of the course the student is able to:

CO1. Understand the details of Ethnobotany

CO2. Learn the characteristics of traditional medicinal plants

Unit 1- Ethnobotany:**(6 Lectures)**

Introduction, concept, scope and objectives; Ethnobotany as an interdisciplinary science

The relevance of ethnobotany in the present context; Major and minor ethnic groups or

Tribals of India, and their life styles. Plants used by the tribals: a) Food plants b) intoxicants and beverages c) Resins and oils and miscellaneous uses.

Unit 2- Methodology of Ethnobotanical studies:**(6 Lectures)**

a) Field work b) Herbarium c) Ancient Literature d) Archaeological findings e) temples and sacred places.

Unit 3- Role of Ethnobotany in modern Medicine:**(10 Lectures)**

Medico-ethnobotanical sources in India; Significance of the following plants in ethno botanical practices (along with their habitat and morphology) a) *Azadiractha indica* b) *Ocimum sanctum*

c) *Vitex negundo* d) *Gloriosa superba* e) *Tribulus terrestris* f) *Pongamia pinnata*

g) *Cassia auriculata* h) *Indigofera tinctoria*

Role of ethnobotany in modern medicine with special example *Rauvolfia serpentina*, *Trichopus zeylanicus*, *Artemisia*, *Withania*.

Role of ethnic groups in conservation of plant genetic resources Endangered taxa and forest management (participatory forest management).

Unit 4- Ethnobotany and legal aspects:**(8 Lectures)**

Ethnobotany as a tool to protect interests of ethnic groups. Sharing of wealth concept with

few examples from India. Biopiracy, Intellectual Property Rights and Traditional Knowledge.

References

- 1) S.K. Jain, Manual of Ethnobotany, Scientific Publishers, Jodhpur, 1995.
- 2) S.K. Jain (ed.) Glimpses of Indian. Ethnobotny, Oxford and I B H, New Delhi – 1981
- 3) Lone et al., Palaeoethnobotany
- 4) S.K. Jain (ed.) 1989. Methods and approaches in ethnobotany. Society of ethnobotanists, Lucknow, India.
- 5) S.K. Jain, 1990. Contributions of Indian ethnobotny. Scientific publishers, Jodhpur.
- 6) Colton C.M. 1997. Ethnobotany – Principles and applications. John Wiley and sons – Chichester
- 7) Rama Ro, N and A.N. Henry (1996). The Ethnobotany of Eastern Ghats in Andhra Pradesh, India. Botanical Survey of India. Howrah.8) Rajiv K. Sinha – Ethnobotany The Renaissance of Traditional Herbal Medicine – INA –SHREE Publishers, Jaipur-19969)

DME23408 /DME23407

III B.Sc., V Semester SEC-II
Ethnobotany
Scheme of Theory Question Paper

Blue print:

Max. Marks: 50

Units	Hours allotted	No. of questions from each category			Total Marks 50
		2 marks (5/8)=10	5marks (4/6)=20	10marks (2/4)=20	
Unit 1: Ethnobotany	6	2X2=4	5X1=5	10X1=10	19
Unit 2: methodology of Ethanobotanical studies	6	2X3=6	5X2=10	-	16
Unit 3: role of ethonobotany in modern medicine	10	2X2=4	5X1=5	10X2=20	29
Unit 4: Ethnobotany and legal aspects	8	2X1=2	5X2=10	10X1=10	22
Total	30	8X2=16	5X6=30	10X4=40	86

DMF23008/ DMF23007

III B.Sc., VI Semester DSE-III

**Genetics and Plant Breeding
Theory (4 credits)**

**Lectures: 60 Hours
(4 hours/week)**

Course outcome:

After completion of the course the student is able to:

- CO1 Specify the details of heredity
- CO3 Write down the classification and characteristics of mutations
- CO4 Learn the details of plant breeding
- CO2 Identify in details with examples linkage

Unit 1- Heredity:

(24 Lectures)

1. Brief life history of Mendel
2. Terminologies
3. Laws of Inheritance
4. Modified Mendelian Ratios: incomplete dominance; complementary factors; supplementary factors, Duplicate factors, Epistasis.
6. Pedigree Analysis
7. Cytoplasmic Inheritance: leaf variegation in *Mirabilis jalapa*, Male sterility.
8. Chromosome theory of Inheritance.
9. Quantitative inheritance-Concept, mechanism, examples. Monogenic vs polygenic Inheritance.

Unit 2- Sex-determination and Sex-linked Inheritance:

(6 Lectures)

Sex – determination in *Melandrium album* by XX-XY method, Bridges Genic balance theory, Sex-linked Inheritance

Unit 3- Linkage and Crossing over:

(8 Lectures)

Linkage: complete & incomplete linkage, coupling & repulsion, recombination frequency, linkage in Maize, two point test cross, linkage maps, Coincidence and interference. Crossing over: concept and significance.

Unit 4-Mutations and Chromosomal Aberrations:

(6 Lectures)

Types of mutations, effects of physical & chemical mutagens. Numerical chromosomal changes:

Euploidy, Polyploidy and Aneuploidy; Structural chromosomal changes: Deletions, Duplications, Inversions & Translocations.

Unit 5- Plant breeding and Methods of crop improvement: (12 lectures)

Introduction, objectives and Methods: Plant introduction, selection, Hybridization- Emasculation and bagging. Mutation breeding, polyploidy breeding, genetic or molecular plant breeding. Methods of propagation– Procedure, advantages and limitations

Unit 6- Inbreeding depression and heterosis: (4 lectures)

Inbreeding depression and heterosis; Applications. Germplasm maintenance, pollen banks and quarantine measures.

DMF23108/ DMF23107

**III B.Sc., VI Semester DSE-III
Genetics and Plant Breeding
Practical (2 credits)**

**Lectures: 60 Hours
(4 hours/week)**

1. Mendel's laws through seed ratios. Laboratory exercises in probability and chisquare.
2. Chromosome mapping using point test cross data.
3. Pedigree analysis for dominant and recessive autosomal and sex linked traits.
4. Incomplete dominance and gene interaction through seed ratios (9:7, 9:6:1, 13:3, 15:1, 12:3:1, 9:3:4).
5. Study of aneuploidy: Down's, Klinefelter's and Turner's syndromes through photographs.
6. Photographs/Permanent Slides showing Translocation Ring, Laggards and Inversion Bridge.
7. Hybridization techniques - Emasculation, Bagging (For demonstration only).
8. Induction of polyploidy conditions in plants (For demonstration only).

Suggested Readings

1. Gardner EJ, Simmons MJ, Snustad DP (2008). Principles of Genetics. 8th Ed. WileyIndia.
2. Snustad, D.P. and Simmons, M.J. (2010). Principles of Genetics, John Wiley & Sons Inc., India. 5th edition.
3. Klug WS, Cummings MR, Spencer, C, Palladino, M (2011). Concepts of Genetics, 10th Ed., Benjamin Cummings
4. Griffiths, A.J.F., Wessler, S.R., Carroll, S.B., Doebley, J. (2010). Introduction to Genetic Analysis. W. H. Freeman and Co., U.S.A. 10th edition.
5. Pierce BA (2011) Genetics: A Conceptual Approach, 4th Ed., Macmillan Higher Education Learning
6. Singh, B.D. (2005). Plant Breeding: Principles and Methods. Kalyani Publishers. 7th edition.
7. Chaudhari, H.K. (1984). Elementary Principles of Plant Breeding. Oxford – IBH. 2nd edition.
8. Acquaah, G. (2007). Principles of Plant Genetics & Breeding. Blackwell Publishing.

**III B.Sc., VI Semester DSE-III
Genetics and Plant Breeding
Scheme of Practical Question Paper**

Time: 4 Hours

Max. Marks: 35 (25+05+05)

I. Perform the experiment 'A'. 6 marks

(Emasculation and bagging)

Preparation -2 marks

Sketch and label - 2marks

Reasons - 2marks

II. Conduct experiment 'B'. 4 marks

(Induction of polyploidy)

Principle -1 marks

Requirements -1 marks

Procedure -1 marks

Result and inference -1 marks

III. Problems on Chromosome mapping using point test cross data 'C'. 4 marks

IV. Comment on the given specimen 'D'. 03 marks

(Pedigree analysis)

Identification - 1mark

Reasons - 2marks

V. Problems on gene interaction 'E' 04 marks

VI. Identify the given photographs 'F' & 'G' 2x2=4 marks

(F- Aneuploidy & G- Translocation)

Identification - 1mark

Reasons - 1marks

VI. Practical record 05 marks

VII. Viva-Voce 05 marks

**III B.Sc., VI Semester DSE-III
Genetics and Plant Breeding
Scheme of Theory Question Paper**

Time: 3.00 Hours

Max. Marks: 70

Blue print:

Units	Hours Allotted	No. of questions from each category			Total Marks 70
		2 marks (5/8)=10	5marks (4/6)=20	10marks (4/6)40	
Unit 1: Heredity	24	2X3=6	5X1=5	10X3=30	41
Unit 2: Sex-determination and Sex-linked Inheritance	4	2X1=2	5X1=5	-	7
Unit 3: Linkage and Crossing over	8	2X1=2	-	10X1=10	12
Unit 4: Mutations and Chromosomal Aberrations	4	2X1=2	5X1=5	-	7
Unit 5: Plant Breeding	4	2X1=2	5X1=5	-	07
Unit 6: Methods of crop improvement	8	-	5X1=5	10X1=10	15
Unit 7: Inbreeding depression and heterosis	4	2X1=2	5X1=5	-	02
Unit 8: Crop improvement and breeding	4	-	-	10X1=10	10
Total	60	8X2=16	5X6=30	10X6=60	106

Course outcome:

After completion of the course the student is able to:

- CO2 Learn the details of Spectrophotometry
- CO3 Write down the details of chromatography
- CO1 Specify the details of cell fractioning
- CO4 Identify in details with application, if applicable, biostatistics

Unit 1: Imaging and related techniques**(15 Lectures)**

Principles of microscopy; Light microscopy; Fluorescence microscopy; Confocal microscopy; Use of fluorochromes: (a) Flow cytometry (FACS); (b) Applications of fluorescence microscopy: Chromosome banding, FISH, chromosome painting; Transmission and Scanning electron microscopy – sample preparation for electron microscopy, cryofixation, negative staining, shadow casting, freeze fracture, freeze etching.

Unit 2: Cell fractionation**(8 Lectures)**

Centrifugation: Differential and density gradient centrifugation, sucrose density gradient, CsCl₂ gradient, analytical centrifugation, ultracentrifugation, marker enzymes.

Unit 3: Radioisotopes**(4 Lectures)**

Use in biological research, auto-radiography, pulse chase experiment.

Unit 4: Spectrophotometry**(4 Lectures)**

Principle and its application in biological research.

Unit 5: Chromatography**(8 Lectures)**

Principle; Paper chromatography; Column chromatography, TLC, GLC, HPLC, Ionexchange chromatography; Molecular sieve chromatography; Affinity chromatography.

Unit 6: Characterization of proteins and nucleic acids**(6 Lectures)**

Mass spectrometry; X-ray diffraction; X-ray crystallography; Characterization of proteins and nucleic acids; Electrophoresis: AGE, PAGE, SDS-PAGE

Unit 7: Biostatistics**(15 Lectures)**

Statistics, data, population, samples, parameters; Representation of Data: Tabular, Graphical; Measures of central tendency: Arithmetic mean, mode, median; Measures of dispersion: Range, mean deviation, variation, standard deviation; Chi-square test for goodness of fit.

1. Study of Blotting techniques: Southern, Northern and Western, DNA fingerprinting, DNA sequencing, PCR through photographs.
2. Demonstration of ELISA.
3. To separate nitrogenous bases by paper chromatography.
4. To separate sugars by thin layer chromatography.
5. Isolation of chloroplasts by differential centrifugation.
6. To separate chloroplast pigments by column chromatography.
7. To estimate protein concentration through Lowry's methods.
8. To separate proteins using PAGE.
9. To separate DNA (marker) using AGE.
10. Study of different microscopic techniques using photographs/micrographs (freeze fracture, freeze etching, negative staining, positive staining, fluorescence and FISH).
11. Preparation of permanent slides (double staining).

Suggested Readings

1. Plummer, D.T. (1996). An Introduction to Practical Biochemistry. Tata McGrawHill Publishing Co. Ltd. New Delhi. 3rd edition.
2. Ruzin, S.E. (1999). Plant Microtechnique and Microscopy, Oxford University Press, New York. U.S.A.
3. Ausubel, F., Brent, R., Kingston, R. E., Moore, D.D., Seidman, J.G., Smith, J.A., Struhl, K. (1995). Short Protocols in Molecular Biology. John Wiley & Sons. 3rd edition.
4. Zar, J.H. (2012). Biostatistical Analysis. Pearson Publication. U.S.A. 4th edition

**III B.Sc., VI Semester DSE-III
Genetics and Plant Breeding
Scheme of Theory Question Paper**

Time: 3.00 Hours

Max. Marks: 70

Blue print:

Units	Hours Allotted	No. of questions from each category			Total Marks 70
		2 marks (5/8)=10	5marks (4/6)=20	10marks (4/6)40	
Unit 1: Imaging and related techniques	15	2X2=4	5X2=10	10X1=10	24
Unit 2: Cell fractionation	8	2X1=2	5X1=5	10X1=10	17
Unit 3: Radioisotopes	4	2X2=4	-	-	4
Unit 4: Spectrophotometry	4	2X1=2	-	10X1=10	12
Unit 5: Chromatography	8	-	5X1=5	10X1=10	15
Unit 6: Characterization of proteins and nucleic acids	6	2X1=2	5X1=5	-	7
Unit 7: Biostatistics	15	2X1=2	5X1=5	10X2=20	27
Total	60	8X2=16	5X6=30	10X6=60	106

Suggested Readings:

Author	Title of the Book	Publisher
VIRUSES AND BACTERIA R.C.Dubey and D.K. Maheshwari	A textbook of Microbiology	S. Chand & company, Ramnagar N.Delhi-110005.
P.D. Sharma	Microbiology	Rastogi Publications; Shivaji roa Meerat; 250002; India
P. D. Sharma	Microbiology and Plant patholog	Rastogi Publications; Shivaji roa Meerat; 250002; India
H. C. Dube	Text book of fungi, Bacteria & Virus	Vani Educational books , Vikas house 20/4, Industrial area, Sahidabad, 201010, Ghaziabad, UP.
Power & Daginawala	General Microbiology. Vol. I	Himalaya Publishing house, Bombay
Power & Daginawala	General Microbiology. Vol. II	Himalaya Publishing house, Bombay
Pelzar Michael.J Prescott, Lansing and Others	Text Book of Microbiology Microbiology Text Book of Microbiology	Orient and Longman, New Delhi
Ananthanarayana .R . Jayaram Panicker a) salle. A. J.	Functional Principles of Bacteriology Applied Microbiology.	Tata Mc graw Hill Himalaya Publishing house, Bombay
Vinita Kale and Kishore Bhusari	Food Microbiology	ELBS Publisher , New Delhi
Frazier William. C. Cruckishank	Text book of Medical Microbiology	Prentice Hall of India N.Delhi Vardaman Publishers , Bangalor
Rangaswamy.G. Sundar Rajan	Diseases of crop plants in India. College Microbiology	Vol. III & Vol. IV. Tata McGraw Hill Publishing company.
William. C. Frazier and Dennis C. West hoff. 3 rd Edn	Food Microbiology	R. Chand & company, Publishers N.Delhi.
ALGAE K.N. Bhatia	A Treatise on Algae	Pradeep Pub., Jalandhar. Mc graw Hill , New york.
Chopra. G.L G. M. Smith	A Text book of Algae Cryptogamic Botany Vol. I	Thomas, Nelson and Sons Rastogi Publications
Prescott, G.W Kumar, M.A and Kashyap	The Algae to Review Recent advances in physiology	Cambridge University Press

<p>A.K. Fritsch. F. E. Chapman V.J & Chapman D.J. Singh, Pande, Jain. B. P. Pandey Darley. M. W. FUNGI Smith. G. M. Alexopoulos. C. J. and Mims. C. W. Chopra G. L. and Verma. Mundkur, B. B. Rangaswamy, G. Sharma. P. D. Vashista, R.R</p>	<p>Structure and Reproduction of Algae Vol. I & Vol. II The Algae 2nd Edn. A text book of Botany Simplified course in Botany Algal Biology Cryptogamic Botany Vol. I Introduction to Mycology Text book of Fungi Fungi & Plant diseases Diseases of India 3rd Edition The fungi Fungi Bryophyta</p>	<p>Mac Milan, Publishing New York Rastogi Publications; Shivaji road Meerat; 250002; India S. Chand & company, Ltd. Ramnagar N. Delhi-110005. Black well Publishers. Mc Grawhill, New York. Wiley Eastern Ltd. New Delhi. Pradeep publications, Jalandar Mac Milan & Co Calcutta Prentice Hall of India New Delhi Rastogi Publications S. Chand and Company, New Delhi. S. Chand and Company, New Delhi. S. Chand and Company, New Delhi. Central book depot, Allahabad. Mc Grawhill, New York Pradeep Publications, Jalandar.</p>
<p>BRYOPHYTA Pandey. B.P. Vashista. B. P. Parihar. N.S. G. M. Smith G. L. Chopra</p>	<p>Bryophyta Bryophyta Bryophyta Cryptogamic Botany vol. I Class Book and Pteridophytes Bryophytes and Pteridophytes</p>	<p>MC Graw Hill, New York.</p>
<p>Chauhan D.K.S ANATOMY Eames A.J. and Mac Daniels, L. H Katherien Esau Pandey. B. P Singh. V., Pandey, P.C and Jain, D.K. Tayal M. S. Ganguli Das L Datta Venkateshvaralu EMBRYOLOGY OF ANGIOSPERMS & TAXANOMY</p>	<p>Introduction to Plant Anatomy Anatomy of seed plants Introduction to Plant Anatomy Anatomy of seed plants Plant anatomy College Botany Vol. I Cytology and Anatomy The Embryology of Angiosperms</p>	<p>Wiley Eastern, New Delhi. S. Chand and Company. Rastogi publications, Meerat. Rastogi publications, Meerat. Vikas publishing HOUSE, New Delhi. Rastogi publications, Shivaji Road, Meerat, 250002.</p>

Bhojwani. S. S. & Bhatnagar, S. P. Singh, Pandey, Jain	The Embryology of Angiosperms	MC Graw Hill publishing Company, New Delhi.
Maheshwari , P	The Embryology of Angiosperms	Ind. Sci. Acad. Bull. No.41, New Delhi.
Johri, B.M.	Comparative Embryology of Angiosperms Morphology of Angiosperms Plant cell and Tissue culture.	MC Graw Hill, New York. Narosa publishing House New Delhi.
Eames A. J. Reinert . J and Yeoman M.M	Plant Anatomy Taxonomy of Vascular plants A Text book of systematic Botany	Harper C Row, New York.
Vashishta George H.M. Lawarance. R.N. sutaria A. C. Dutta	Botany for Degree Students. Morphology of plants and Fungi	Mc Graw Hill, New York.
PTERIDOPHYTA Bold , H.C., Alexopoulos, C.J & Delevoryas, T. Eames, Arthur, J.	Morphology of vascular plants (lower groups). The Biology and Morphology of Pteridophytes.	Central book depot. Allahabad. Vikas publishing House, New Delhi. Vani educational books, New Delhi.
Parihar, N.S. 1977	Text book of Botany, Vol. II An introduction to Pteridophyta. The Morphology of Pteridophyte	Hutchinson university library, London. S. Chand and Co., New Delhi.
Pandey, S.N.& Others Rashid,A.1986 Sporne,K.R.1970	Pteridophyta An Introduction to Gymnosperm	Asia publishing house, New Delhi. K. Nath and Co. Prasaranga, University of Mysore Mysore.
Vashista,P.C. 1987 GYMNOSPERMS Datta, S.C. Pandey, B.P. Ramaswamy, S.N. 1984	Gymnosperms. Anavrutha beeja sasyagalu (Gymnosperms) Text book of Botany Vol. II. The Morphology of Gymnosperms.	Ratna Prakashana Mandir, Agra Hutchinson university library, London. Shashidhar Malaviya Prakashan. S.Chand & Co. New Delhi. Wiley, New York.
Saxena and Sarabhai 1993 Sporne, K.R.1969	An Introduction to Gymnosperm Gymnosperms.	Wiley, New York.
Trivedi, B.S.& Singh, D.K Vashista, B.R. Andrews, H.N. 1961 Biswas, C. & Johri, B.M. 1997	Studies in palaeobotany. The Gymnosperms. Out line of Biochemistry	Narosa, New Delhi. Wiley-Estern, New Delhi.
PLANT PHYSIOLOGY		Centar book Depot, Allahabad.

<p>Conn, E.E. and Stumpf, P.K. 1976 Datta, S.C. Delvin, R.M. 1969 Delvin, R.M. & Barker, A.V. 1971 Jain, V.K. 1990 Kumar, H.D. & Singh, H.N. 1975, 1993 Krishnamurthy, H.N.</p> <p>Lehninger, A.L. 1978 Noggle, G.R. and Fritz George, J. 1977. Rao, K.N. Sudhakar Rao and Bharatan, S. 1987 Rabinowitch, E. & Govindjee. 1970 Salisbury, E.E. & Ross, C.W. 1986</p> <p>ECOLOGY & ENVIRONMENTAL BIOLOGY</p> <p>Aarne Vesilid, P & Jeffrey Pierce, J. 1983 Benton Allen, H & Warner, W.E Colinvaux Paul, A. 1973 Dash, M.C.</p> <p>Dara, S.S. 1993</p> <p>Kormondy Edward, J. 198 Kochhar, P.L. 1990 Kotpal, R.L. 7 Bali, N.P. 1987 Kumar, H.D. 1990 Lloyd, J.R. 1980 Mason, C.E. 1981 Misra, K.C. 1989 Odum, E.P. 1971 Odum, E.P. 1983</p>	<p>Plant physiology Plant physiology Photosynthesis</p> <p>Fundamentals of Plant physiology Plant Metabolism I Edn. & II Edn.</p> <p>Physiology of plant Growth and Development. Biochemistry Introductory Plant physiology</p> <p>The function of plant.</p> <p>Photosynthesis</p> <p>Plant physiology</p> <p>Environmental Pollution and Control</p> <p>Field Biology and Ecology Introduction to Ecology Fundamentals of Ecology</p> <p>A Text book of Environmental Chemistry and Pollution Control Concept of Ecology Plant Ecology Concept of Ecology</p> <p>Concept of Ecology Man and the ecosystem. Biology of fresh water Pollution Manual of plant Ecology Fundamentals of Ecology Basic Ecology</p> <p>Air Pollution and Control</p>	<p>Affiliated East West, New Delhi. Affiliated East West, New Delhi.</p> <p>S.Chand & Co. New Delhi. East West Press Pvt. Ltd. New Delhi. Atma Ram & Sons, New Delhi.</p> <p>Prentice Hall of India Pvt. Ltd. S.Vishwanatha, Pvt. Ltd. Wiley Eastern, New Delhi.</p> <p>First Indian Edn. CBZ Publishers and Distributers, New Delhi.</p> <p>Ann Arbor Science, Michigan. McGraw Hill. John Wiley and Sons, New York Tata-McGraw Hill publishing Co New Delhi. S.Chand & Co, New Delhi.</p> <p>Prentice Hall of India New Delhi Ratna Prakashan mandir, Agra. Vishal Publications, Jalandar.</p> <p>Vikas, New Delhi. Macmillan Education Ltd. London Longman Inc., New York. Oxford and IBH, New Delhi. Saunders, W.B. Philadelphia Wiley, New York.</p> <p>Divyajyoti prakashan Jodhpur. Rastogi Publications, Meerut I Edn. Rastogi</p>
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Pratap Mowli, P & Venkat Subbaya, N. 1989. Sharma, P.D. Sharma, P.D. Trivedi, R.N. 1993	Ecology and Environment Environmental Biology Text book of Environmental Sciences Plant Ecology Principles of Ecology	Publications, Meerut. Anmol Publications, New Delhi. Vishal Publications Jalandhar. S.Chand & Co., New Delhi. Macmillan, New York.
Vashista, P.C. 1989 Verma, P.S. and Agarwal, V.K. 1992 Whittaker, R.H. 1975	Communities and Ecosystems II Edn.	Wiley Eastern Ltd.
CYTOLOGY, GENETICS AND EVOLUTION	Genetics Genetics-Analysis and Principles	Addison Wesley Longman, California.
Ahluwalia Kavita, B. 1985 Booker, R.J 1999	The Chromosomes Modern Genetics II Edn.	Oxford and IBH, New Delhi Benjamin Cummings.
Archana Sharma, 1990 Ayala, F.J. and Klug, Jr. 1984	Gene and Genetics Cell Biology	Tata McGraw Hill, New Delhi W.B. Saunders and Co. Philadelphia
Cherayil, J.D 1974 De Robertis, E.D.P. Solez, F.A & Nowinski, W.W. 1966	Evolution Genetics and Origin of species	Surjeet publications, New Delhi
Dobzhansky, T., Ayala, J., Stebbins Dobzhansky, T. 1951	Cell biology Principles of Genetics	Oxford and IBH publishing Co., New Delhi Harper and Row publishers, John Wiley and Sons,
Dowben Robert, M 1971 Gardner, E.J & Snustad, D. 1984 & 1990	Genetics The Science of Genetics	Rastogi publications, Meerut. Prentice Hall of India. New York.
Gupta, P.K 1987 Hexter, W and Yost Henry T. 1977	Genes and Evolution Evolution Genetics and Evolution	Macmillan, India, New Delhi George Allen & Unwin, London.
Jha, A.P. 1993. Huxley, J. 1974	Cell structure and function	15 th Edn. Rattan Prakashan Mandir, Agar. Amerind Publishing Co. New Delhi
Kochhar, P.L. 1994	Evolution and Genetics A Text book of Genetics & Evolution	Holt, Rinehart and Winston, New York.
Loewy Ariel, G. & Philip Siekevitz. 1974 Marril David, J. 1962 Nair, P.G.K. Prabhakar Achar, K.	Genetics – the community of life Essentials of Cytology	Konark Publishers Pvt. Ltd. A. 149, Main Vikar Marg, New Delhi Brooks-Cole, California.

<p>Fair banks, D.J. and Anderson, W.R. 1999 Pawar, C.B.1983</p>	<p>Evolution The Science of Evolution</p>	<p>Himalayan publishing house, Bombay.</p>
<p>Savage, J.M. 1969 Stansfields, W.D.1977</p>	<p>Principles of Genetics</p>	<p>Oxford and IBH, New Delhi</p>
<p>Sinnot, E.W., Dunn, L.C., & Dobzhansky, T 1958</p>	<p>Principles of Genetics</p>	<p>Calif polytechnic state university and Macmillan, New York.</p>
<p>Snustad, D.P., Simmons, M.J. & Jenkins, J.R. 1997</p>	<p>Cytology and Cytogenetics</p>	<p>McGraw Hill, New York</p>
<p>Swanson Carl, P. 1963 Swanson Carl, P & Webster Peter, L.</p>	<p>The Cell</p>	<p>John Wiley, New York.</p>
<p>Strickberger Monroe, W. 1968</p>	<p>Genetics</p>	<p>Macmillan & Co., Ltd. London.</p>
<p>Strichberger Monroe, W. 1996</p>	<p>Evolution</p>	<p>Prentice Hall of India Pvt. Ltd., New Delhi</p>
<p>Winchester, A.M. 1966 PLANT BREEDING, ECONOMIC BOTANY AND TAXONOMY</p>	<p>Genetics</p>	<p>Macmillan Company, New York</p>
<p>Allard, R.W. 1960</p>	<p>Principles of Plant Breeding</p>	<p>John & Bartlett Sandburry.</p>
<p>Bailey,L.H. 1966</p>	<p>Manual of cultivated plants</p>	<p>Oxford & IBH, New Delhi</p>
<p>Chandrasekharn, S.N., Parthasarathy,S.V.1973</p>	<p>Cytogenetics and Plant Breeding</p>	<p>Macmillan & Co., New York.</p>
<p>Chaudhari, H.K. 1980</p>	<p>Elementary principles of Plant Breeding</p>	<p>Varadachary and Co., Madras.</p>
<p>Hartman, h.T. & Kester, D.E. 1976</p>	<p>Plant Breeding</p>	<p>Oxford and IBH publishing Co., New Delhi.</p>
<p>Hill, Albert, F. 1983</p>	<p>Economic Botany</p>	<p>Principles and practices, Prentice Hall of India Pvt. Ltd., New Delhi.</p>
<p>Jain S.C.</p>	<p>Medicinal plants</p>	<p>.Tata-McGraw Hill publishing Co New Delhi.</p>
<p>Hutchison, J. 1973</p>	<p>The families of Flowering plants</p>	<p>Oxford University Press, London</p>
<p>Lawrence, George, H.M. 1964.</p>	<p>Taxonomy of Vascular plants</p>	<p>Oxford and IBH publishing Co., New Delhi</p>
<p>Naik, V.N. 1984</p>	<p>Taxonomy of Angiosperms</p>	<p>Tata-McGraw Hill publishing Co New Delhi.</p>
<p>Johri, B.M.& Bhatnagar,S.P.</p>	<p>Taxonomy of Angiosperms</p>	<p>Narosa publishing House New Delhi.</p>
<p>Pullaiah, T.1998</p>	<p>Taxonomy of Angiosperms</p>	<p>Regency Publications, New Delhi</p>
<p>Radford,A.E., Dickison,W.C., Massey, J & Bell, C.R 1974</p>	<p>Vascular Plant Systematics</p>	<p>Harper & Row, New York.</p>
	<p>Breeding Asian Field Crops</p>	
	<p>Flora of Shimoga District</p>	
	<p>Flora of Bangalore District.</p>	

Poehlman, J.M. & Dhirendranath, B. Ramaswamy, S.N., Radhakrishna Rao, M & Govindappa.D.A.2001 Ramaswamy, S.V.& Razi, B.A. 1973 Rendle, A.B. 1979	Classification of Flowering Monocotyledons Vol.I. (Indian Reprint Edition) Classification of Flowering plant Dicotyledons Vol. II. (Indian Reprint Edition) A text book of Economic Botany.	Mysore. Prasaranga, University of Mysore Mysore. Vikas Publishing house, New Delhi. Vikas Publishing house, New Delhi.
Rendle, A.B. 1979	Flora of Karnataka, Vol. I & Vol.II	Tata-McGraw Hill publishing Co New Delhi.
Samba Murthy,A.V.S.S.& Subramanyam. N.S. 1973	Flora of Hassan district (Karnataka, India) Text book of Botany, Vol. III.	Oxford and IBH publishing Co., New Delhi
Saldhana, Cecil, J. 1984	Plant Taxonomy	Amerind Publishing Co. Pvt.Ltd. New Delhi.
Saldhana, Cecil, J., &Nicolson Dan, H. 1976	Flora of India series 2: Flora of Karnataka.	Ratan Prakashan Mandir.Agra.
Saxena and Sarabhai. 1993	Taxonomy of Angiosperms Introduction toPrinciples of plant taxonomy	Tata-McGraw Hill publishing Co Ltd.4/12, Asif ali road, New Delhi.
O.P. Sharma	Dictionary of Economic plants in India	Botanical Survey of India & Dep of Environment, New Delhi.
Sharma, B.D., Singh, N.P. Raghavan, R.S. & Miss. Deshpande,U.R. 1984	Taxonomy of Angiosperms	Rastogi Publications. Kalyani Publications,New Delhi.
Singh, V. Sivarajan, V.V. 1984	A Text book of Practical Botany Vol.I & II	ICAR,New Delhi.
Umarao Singh,, Wadhwan A.M. & Johri, B.M. 1983	Practical Botany Vol.I & II	R.Chand & Co., New Delhi
Vashishta, P.C. 1976	College Botany Vol. I, II, III &IV	Rastogi Publications, Shivaji roa Meerut.
GENERAL Ashok Bendre and Ashok Kumar Dr. H.M. Srivastava	Medicinal Plants Vol. 1- 5 Global Biodiversity Assesment	Pradeep publications opp. Sitta Mandhir, Jalandhar. Subha's Publications, Bangalore.
Sundararajan, S. Kottakkal Arya Vaidya sala's	Biodiversity and Ecosystem functions Biodiversity and Ecosystem function, Scope. Biodiversity: Implications for global food security	Cambridge University press, U.K
BOOKS ON BIODIVERSITY Heywood, H & Watson,	Endemic plants of the Indian	Springer-verlag, Berlin. John wiley, Chichester.

R.J. 1995 Schulze, E.D. & Mooney, H. (eds.) 1992 Mooney, H.A. et al. (eds.) 1996 Swamynathan, M.S. & Jana, S. 1992 Ahmedullh, M. & Nayar, M.P. 1987 Jain, S.K. & Sastry, A.R.K. 1980 Jain, S.K. & Sastry, A.R.K. (eds) 1984 Puri, S.K.	region Vol. I. Threatened Plants of India- A State-of-the-Art report Indian plant red data book Vol.I Biodiversity Database projects in India.	Macmillan India Ltd. Madras. BSI Calcutta BSI Calcutta BSI Calcutta Indira Gandhi Conservation Monitory Centre, New Delhi.
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Annexure-I:

Revision/ modification made in the existing Syllabus for 2019-2020 batch onwards:

Existing	Proposed	Justification	Approved
Biodiversity of Microbes and Archegoniate (DMA23008/07): Unit 1: Microbial diversity: A. Virus- replication D. Fungi- <i>Puccinia</i> Unit 2: Archegoniate: A. Pteridophytes- <i>Marsilia</i> , Practicals:	Biodiversity of Microbes and Archegoniate (DMA23008/07): Deleted Fungi yeast is added	Virus- replication is a repetition. <i>Puccinia</i> Better to study in higher level.	Approved
Plant Ecology Morphology and Angiosperm Taxonomy (DMB23008/07):	Plant Ecology Morphology and Angiosperm Taxonomy (DMB23008/07):	To Represent Monocot family	

Apiaceae	Apiaceae deleted Areaceae is added		Approved
Plant Anatomy Embyology (DMC23008/07): Structure of pollen grains	Plant Anatomy and Embyology of Angiosperm (DMC23008/07): Brief Account of Epidermis, cuticle, Stomata & Trichome. Added Deleted Types of Tetrad, Male gametophyte & Embryosac development- Monosporic, Bisporic, Tetrasporic added Mechanism & Adaptation of Pollination deleted. Seed structure is deleted	Blown up syllabi For detailed embryological studies.	Approved
Practicals:			
Plant Physiology and Metabolism. (DMD23008/07): Root pressure theory	Plant Physiology and Metabolism (DMD23008/07): Transpiration pul theory Protoplasmic streaming theory is added Photosynthetic apparatus is added Respiration- Definition, types and ultra structure of mitochondrion added.	Highly essential	Approved

<p>Glyoxilate, oxidative pentose phosphate pathway</p> <p>Biological nitrogen fixation</p> <p>Plants response to light and temperature</p>	<p>Deleted.</p> <p>Enzyme- Classification added</p> <p>Nitrogen fixation(Symbiotic & non symbiotic)</p> <p>Tropic movements</p>	<p>Appropriate</p> <p>To be more accurate</p> <p>Appropriate</p>	
<p>Practicals:</p>			
<p>DSE-1: Cell and Molecular Biology (DME23008/07)</p> <p>Techniques in Biology: Sample Preparation for light microscopy; Sample Preparation for electron microscopy; X-ray diffraction analysis.</p> <p>Cell Membrane and Cell Wall : Carbohydrates in the membrane; Faces of the membranes;</p> <p>Genetic material: DNA replication (Prokaryotes and eukaryotes): bidirectional replication, semi-conservative method</p> <p>Regulation of gene expression:</p> <p>Practicals: Measure the cell size (either length or breadth/diameter) by micrometry</p>	<p>DSE-1: Cell and Molecular Biology</p> <p>deleted</p> <p>deleted</p> <p>DNA replication in prokaryotes-deleted</p> <p>Split gene concept – introduced</p> <p>Deleted</p>	<p>Repetition in microbiology</p> <p>Better to study in higher classes</p> <p>Better to study in higher classes</p> <p>Relevant</p> <p>Repetition in microbiology</p>	<p>Approved</p>
<p>DSE-2: Genetics and Plant Breeding (DMF23008/07)</p> <p>Heredity: lethal genes</p> <p>Pleiotropism, co- dominance</p> <p>9:7; 9:4:3; 13:3; 12:3:1</p>	<p>Deleted</p>	<p>Repetition in Zoology</p> <p>Proper terminologies have been used instead of ratios</p>	<p>Approved</p>

Practicals:	Complementary factors; supplementary factors, Duplicate factors, Epistasis Genetic problems should be included	For better elucidation of theoretical concepts	
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Chairman

**JSS COLLEGE OF ARTS, COMMERCE AND
SCIENCE
(Autonomous)
B N ROAD, MYSURU- 570 025**

**DEPARTMENT OF CHEMISTRY
Syllabus**

CHOICE BASED CREDIT SYSTEM

For B.Sc programmes

**Physics, Chemistry, Mathematics
Chemistry, Zoology, Biotechnology
Chemistry, Botany, Zoology**

2019-20

BSc UG Syllabus PROGRAMME PCM
PROGRAMME CODE: BSc-01,Credit pattern

L:Lecture, T:Tutorial, P:Practicals) PCM01

Semester	Course Type	Course code	Course Title	L:T:P	Total Credit
I	DSC1	DMA24001	ATOMIC STRUCTURE & ORGANIC CHEMISTRY	4:0:0	4
		DMA24101	Practical 1: ATOMIC STRUCTURE & ORGANIC CHEMISTRY	0:0:2	2
II	DSC 2	DMB24001	CHEMICAL ENERGETICS & ORGANIC CHEMISTRY	4:0:0	4
		DMB24101	Practical 2: CHEMICAL ENERGETICS & ORGANIC CHEMISTRY	0:0:2	2
III	DSC 3	DMC24001	SOLUTIONS & ORGANIC CHEMISTRY	4:0:0	4
		DMC24101	Practical 3: SOLUTIONS & ORGANIC CHEMISTRY	0:0:2	2
IV	DSC 4	DMD24001	CO-ORDINATION CHEMISTRY & PHYSICAL CHEMISTRY	4:0:0	4
		DMD24101	Practical 4: CO-ORDINATION CHEMISTRY & PHYSICAL CHEMISTRY	0:0:2	2
V	DSE 1A	DME24001	INORGANIC MATERIALS OF INDUSTRIAL IMPORTANCE	4:0:0	4
	DSE1A	DME24101	Practical 5: INORGANIC MATERIALS OF INDUSTRIAL IMPORTANCE	0:0:2	2
	DSE1B	DME24201	INDUSTRIAL CHEMICALS AND ENVIRONMENT	4:0:0	4
	DSE1B	DME24301	Practical 6: INDUSTRIAL CHEMICALS AND ENVIRONMENT	0:0:2	2
VI	DSE2A	DMF24001	ORGANOMETALLICS, BIOINORGANIC CHEMISTRY, POLY NUCLEAR HYDROCARBONS AND UV,IR SPECTROSCOPY	4:0:0	4
	DSE2A	DMF24101	Practical 7: ORGANOMETALLICS, BIOINORGANIC CHEMISTRY, POLY NUCLEAR HYDROCARBONS AND UV,IR SPECTROSCOPY	0:0:2	2
	DSE2B	DMF24201	QUANTUM CHEMISTRY, SPECTOSCOPY AND PHOTOCHEMISTRY	4:0:0	4
	DSE2B	DMF24301	Practical 8: QUANTUM CHEMISTRY, SPECTOSCOPY AND PHOTOCHEMISTRY	0:0:2	2
	SEC1A	DMF24401	FUEL CHEMISTRY	2:0:0	2
	SEC1B	DMF24501	BASIC ANALYTICAL CHEMISTRY	2:0:0	2

Scheme for Examination and Assessment

Course type	IA C1		IA C2		SEE C3		Exam duration	
	Th	Pr	Th	Pr	Th	Pr	Th	Pr
DSC	15	7.5	15	7.5	70	35	3h	3h
DSE	15	7.5	15	7.5	70	35	3h	3h
SEC	7.5	-	7.5	-	35	-	2h	-

SCHEME OF VALUATION FOR PRACTICAL EXAMINATION

- A candidate appearing for the exam should submit a duly signed and certified practical record
- Each candidate has to perform one experiment in the specified duration of three hours
- Practical record has to be valued by the examiners at the time of examination
- IA for FIFTEEN marks in practical is awarded in the lab

I. EVALUATION OF EXPERIMENTS:

Sl. no	Component	Marks
1	Conducting experiment	25
2	Viva-voce	05
3	Practical record	05
TOTAL		35

BSc UG Syllabus PROGRAMME CZBt
PROGRAMME CODE: BSc-05,Credit pattern

L:Lecture, T:Tutorial, P:Practicals)CZBt05

Semester	Course Type	Course code	Course Title	L:T:P	Total Credit
I	DSC1	DMA24005	ATOMIC STRUCTURE & ORGANIC CHEMISTRY	4:0:0	4
		DMA24105	Practical 1: ATOMIC STRUCTURE & ORGANIC CHEMISTRY	0:0:2	2
II	DSC 2	DMB24005	CHEMICAL ENERGETICS & ORGANIC CHEMISTRY	4:0:0	4
		DMB24105	Practical 2: CHEMICAL ENERGETICS & ORGANIC CHEMISTRY	0:0:2	2
III	DSC 3	DMC24005	SOLUTIONS & ORGANIC CHEMISTRY	4:0:0	4
		DMC24105	Practical 3: SOLUTIONS & ORGANIC CHEMISTRY	0:0:2	2
IV	DSC 4	DMD24005	CO-ORDINATION CHEMISTRY & PHYSICAL CHEMISTRY	4:0:0	4
		DMD24105	Practical 4: CO-ORDINATION CHEMISTRY & PHYSICAL CHEMISTRY	0:0:2	2
V	DSE 1A	DME24005	INORGANIC MATERIALS OF INDUSTRIAL IMPORTANCE	4:0:0	4
	DSE1A	DME24105	Practical 5: INORGANIC MATERIALS OF INDUSTRIAL IMPORTANCE	0:0:2	2
	DSE1B	DME24205	INDUSTRIAL CHEMICALS AND ENVIRONMENT	4:0:0	4
	DSE1B	DME24305	Practical 6: INDUSTRIAL CHEMICALS AND ENVIRONMENT	0:0:2	2
VI	DSE2A	DMF24005	ORGANOMETALLICS, BIOINORGANIC CHEMISTRY, POLY NUCLEAR HYDROCARBONS AND UV,IR SPECTROSCOPY	4:0:0	4
	DSE2A	DMF24105	Practical 7: ORGANOMETALLICS, BIOINORGANIC CHEMISTRY, POLY NUCLEAR HYDROCARBONS AND UV,IR SPECTROSCOPY	0:0:2	2
	DSE2B	DMF24205	QUANTUM CHEMISTRY, SPECTOSCOPY AND PHOTOCHEMISTRY	4:0:0	4
	DSE2B	DMF24305	Practical 8: QUANTUM CHEMISTRY, SPECTOSCOPY AND PHOTOCHEMISTRY	0:0:2	2
	SEC1A	DMF24405	FUEL CHEMISTRY	2:0:0	2
	SEC1B	DMF24505	BASIC ANALYTICAL CHEMISTRY	2:0:0	2

Scheme for Examination and Assessment

Course type	IA C1		IA C2		SEE C3		Exam duration	
	Th	Pr	Th	Pr	Th	Pr	Th	Pr
DSC	15	7.5	15	7.5	70	35	3h	3h
DSE	15	7.5	15	7.5	70	35	3h	3h
SEC	7.5	-	7.5	-	35	-	2h	-

SCHEME OF VALUATION FOR PRACTICAL EXAMINATION

- A candidate appearing for the exam should submit a duly signed and certified practical record
- Each candidate has to perform one experiment in the specified duration of three hours
- Practical record has to be valued by the examiners at the time of examination
- IA for ***FIFTEEN*** marks in practical is awarded in the lab

II. EVALUATION OF EXPERIMENTS:

Sl. no	Component	Marks
1	Conducting experiment	25
2	Viva-voce	05
3	Practical record	05
TOTAL		35

PROGRAMME CODE: BSc-08,Credit pattern**L:Lecture, T:Tutorial, P:Practicals)CBZ08**

Semester	Course Type	Course code	Course Title	L:T:P	Total Credit
I	DSC1	DMA24008	ATOMIC STRUCTURE & ORGANIC CHEMISTRY	4:0:0	4
		DMA24108	Practical 1: ATOMIC STRUCTURE & ORGANIC CHEMISTRY	0:0:2	2
II	DSC 2	DMB24008	CHEMICAL ENERGETICS & ORGANIC CHEMISTRY	4:0:0	4
		DMB24108	Practical 2: CHEMICAL ENERGETICS & ORGANIC CHEMISTRY	0:0:2	2
III	DSC 3	DMC24008	SOLUTIONS & ORGANIC CHEMISTRY	4:0:0	4
		DMC24108	Practical 3: SOLUTIONS & ORGANIC CHEMISTRY	0:0:2	2
IV	DSC 4	DMD24008	CO-ORDINATION CHEMISTRY & PHYSICAL CHEMISTRY	4:0:0	4
		DMD24108	Practical 4: CO-ORDINATION CHEMISTRY & PHYSICAL CHEMISTRY	0:0:2	2
V	DSE 1A	DME24008	INORGANIC MATERIALS OF INDUSTRIAL IMPORTANCE	4:0:0	4
	DSE1A	DME24108	Practical 5: INORGANIC MATERIALS OF INDUSTRIAL IMPORTANCE	0:0:2	2
	DSE1B	DME24208	INDUSTRIAL CHEMICALS AND ENVIRONMENT	4:0:0	4
	DSE1B	DME24308	Practical 6: INDUSTRIAL CHEMICALS AND ENVIRONMENT	0:0:2	2
VI	DSE2A	DMF24008	ORGANOMETALLICS, BIOINORGANIC CHEMISTRY, POLY NUCLEAR HYDROCARBONS AND UV,IR SPECTROSCOPY	4:0:0	4
	DSE2A	DMF24108	Practical 7: ORGANOMETALLICS, BIOINORGANIC CHEMISTRY, POLY NUCLEAR HYDROCARBONS AND UV,IR SPECTROSCOPY	0:0:2	2
	DSE2B	DMF24208	QUANTUM CHEMISTRY, SPECTOSCOPY AND PHOTOCHEMISTRY	4:0:0	4
	DSE2B	DMF24308	Practical 8: QUANTUM CHEMISTRY, SPECTOSCOPY AND PHOTOCHEMISTRY	0:0:2	2
	SEC1A	DMF24408	FUEL CHEMISTRY	2:0:0	2
	SEC1B	DMF24508	BASIC ANALYTICAL CHEMISTRY	2:0:0	2

Scheme for Examination and Assessment

Course type	IA	IA	SEE	Exam duration
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	C1		C2		C3		Th	Pr
	Th	Pr	Th	Pr	Th	Pr		
DSC	15	7.5	15	7.5	70	35	3h	3h
DSE	15	7.5	15	7.5	70	35	3h	3h
SEC	7.5	-	7.5	-	35	-	2h	-

SCHEME OF VALUATION FOR PRACTICAL EXAMINATION

- A candidate appearing for the exam should submit a duly signed and certified practical record
- Each candidate has to perform one experiment in the specified duration of three hours
- Practical record has to be valued by the examiners at the time of examination
- IA for ***FIFTEEN*** marks in practical is awarded in the lab

III. EVALUATION OF EXPERIMENTS:

Sl. no	Component	Marks
1	Conducting experiment	25
2	Viva-voce	05
3	Practical record	05
TOTAL		35

DEPARTMENT OF CHEMISTRY

Programme Outcome for Bachelor of Science in Physics, Chemistry and Mathematics:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate proficiency in Mathematics and the Mathematical concepts needed for a proper understanding of Physics.
- PO2. Demonstrate the ability to justify and explain their thinking and/or approach.
- PO3. Demonstrate the ability to think, express and present in a clear, logical and succinct arguments.
- PO4. Develop state-of-the-art laboratory skills and professional communication skills.
- PO5. Apply the scientific method to design, execute, and analyze an experiment and also to explain their scientific procedures as well as their experimental observations.
- PO6. Appreciate the role of chemistry in the society
- PO7. Use this as a basis for ethical behaviour in issues facing chemists /drugs.
- PO8. Understand chemistry as an integral part for addressing social, economic, and environmental problems.
- PO9. Develop and understand the value of Mathematical proof and demonstrate proficiency in writing and understanding proofs.
- PO10. Investigate and apply mathematical problems and solutions in aspects of science and technology.
- PO11. Gain experience investigating the real world problems
- PO12. Apply mathematical ideas and models to those problems.

Program Specific Outcome:

Bachelor of Science in Physics, Chemistry and Mathematics

After completing the graduation in Physics, Chemistry and Mathematics the students are able to:

PSO1. Find career opportunities and develop competence to write competitive examinations.

PSO2. Develop proficiency in the analysis of complex physical problems and the use of mathematical or other appropriate techniques to solve them.

PSO3. Apply appropriate techniques for the qualitative and quantitative analysis of chemicals in laboratories and in industries.

PSO4. Create a hypothesis and appreciate how it relates to broader theories.

PSO5. Demonstrate skills in the use of Computers.

Programme Outcome for Bachelor of Science in Chemistry, Zoology and Biotechnology:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate the ability to justify, explain, and/or approach the concept both in written and oral forms
- PO2. Demonstrate the ability to present clear, logical and succinct arguments
- PO3. Develop state-of-the-art laboratory skills and professional communication skills.
- PO4. Apply the scientific method to design, execute, and analyze an experiment.
- PO5. Appreciate the central role of chemistry in the society and use this as a basis for ethical behaviour in issues facing chemists/drugs.
- PO6. Understand Chemistry as an integral part for addressing social, economic, and environmental problems.
- PO7. Identify the major groups of organisms with an emphasis on animals and plants.
- PO8. Compare and contrast the characteristics of animals that differentiate themselves from other living and non-living creatures.
- PO9. Give specific examples of physiological adaptations.
- PO10. Design and develop solution to Biotechnology problems keeping in mind the safety measures for environment and society.
- PO11. Support Biotechnology research activity with strong technical background knowledge.

**Programme Specific Outcome for Bachelor of Science in Chemistry,
Zoology and Biotechnology:**

After completing the graduation in Chemistry, Zoology and Biotechnology the students are able to:

- PSO1. Find jobs at all level of chemical, pharmaceutical, food products and life oriented material Industries
- PSO2. Apply appropriate techniques for the qualitative and quantitative analysis of chemicals in laboratories and in industries.
- PSO3. Recognize the relationship between different structures and functions at different levels.
- PSO4. Characterize the biological, chemical and physical features of environments that Animals inhabit.
- PSO5. Demonstrate effectively the applications of biochemical and biological sciences.
- PSO6. Know and apply appropriate tools and techniques in biotechnological manipulation
- PSO7. Understand his or her responsibilities in biotechnological practices.

Programme Outcome for Bachelor of Science in Chemistry, Botany, Zoology:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate the ability to justify, explain, and/or approach the concept both in written and oral forms
- PO2. Demonstrate the ability to present clear, logical and succinct arguments
- PO3. Develop state-of-the-art laboratory skills and professional communication skills.
- PO4. Apply the scientific method to design, execute, and analyze an experiment.
- PO5. Appreciate the central role of chemistry in the society and use this as a basis for ethical behaviour in issues facing chemists/drugs.
- PO6. Identify the taxonomic position of plants using required principles and methods.
- PO7. Understand the impact of the plant diversity in societal and environmental context, and demonstrate the knowledge of, and need for sustainable development.
- PO8. Use interdisciplinary approaches with quantitative skills to work on biological problems.
- PO9. Understand Chemistry as an integral part for addressing social, economic, and environmental problems.
- PO10. Identify the major groups of organisms with an emphasis on animals and plants.

After completing the graduation in the Bachelor of Science the students are able to:

PSO1. Find jobs at all level of chemical, pharmaceutical, food products, life oriented material industries, etc.

PSO2. Apply appropriate techniques for the qualitative and quantitative analysis of chemicals in laboratories and in industries.

PSO3. Explicate ecological interconnectedness of life

PSO4: Analyze the avenues and remedies for burning environmental issues

PSO5. Recognize the relationship between different structures and functions at different levels.

PSO6. Characterize the biological, chemical and physical features of environments that Animals inhabit.

DMA24001/ DMA24005/ DMA24008

SEMESTER I

ATOMIC STRUCTURE AND ORGANIC CHEMISTRY

Theories: 60 Lectures

Course outcome:

After completion of the course the student is able to:

- CO1: Learn the basics of atomic structure and periodicity functions, structures and properties of chemical compounds.
- CO2: Acquire knowledge on aromaticity and aliphatic hydrocarbons
- CO3: Learn the basics of stereochemistry
- CO4: Understand the methods of analysis related to volumetric estimations.

CHEMISTRY-DSC 1:

Section A: Inorganic Chemistry-1 (30 Periods)

Atomic Structure: Review of: Bohr's theory and its limitations, dual behaviour of matter and radiation, de Broglie's relation, Heisenberg Uncertainty principle. Hydrogen atom spectra. Need of a new approach to Atomic structure.

What is Quantum mechanics? Time independent Schrodinger equation and meaning of various terms in it Significance of ψ and ψ^2 , Schrödinger equation for hydrogen atom. Radial and angular parts of the hydrogen wave functions (atomic orbitals) and their variations for $1s$, $2s$, $2p$, $3s$, $3p$ and $3d$ orbitals (Only graphical representation) Radial and angular nodes and their significance. Radial distribution functions and the concept of the most probable distance with special reference to $1s$ and $2s$ atomic orbitals. Significance of quantum numbers, orbital angular momentum and quantum numbers ml and ms . Shapes of s , p and d atomic orbitals, nodal planes. Discovery of spin, spin quantum number (s) and magnetic spin quantum number (ms). Rules for filling electrons in various orbitals, Electronic configurations of the atoms Stability of half-filled and completely filled orbitals, concept of exchange energy. Relative energies of atomic orbitals, Anomalous electronic configurations.

Periodic Table and Periodicity:

Classification of elements into s, p, d, and f-blocks, cause of periodicity. Detailed discussion of the following periodic properties of elements with examples

1). Atomic radius: Covalent, ionic, Vander Waal's and crystal radii. Additive nature of covalent radii., Determination of ionic radii by Lande's method. Variation of covalent radii in a group and in a period- explanation for the observed trends. Comparison of the size of the atoms with the corresponding anions and cations, Variation of ionic radii in isoelectronic ions.

2). Ionization enthalpy: Successive ionization enthalpy, factors affecting ionization enthalpy, applications of ionization enthalpy. Variation in a group and in a period- explanation for the observed trends.

3). Electron gain enthalpy: Successive electron gain enthalpy variation of electron gain enthalpy in period and in a group- explanation for the observed trends.

4). Electronegativity: Variation of electronegativity in a group and in a period- explanation for the observed trends. Factors determining electro negativity (charge on

the atom and hybridization). Pauling and Mullikan methods (problems to be worked out). Allred-Rochow scale of electronegativity. Applications of electronegativity.

(14 Lectures)

Chemical Bonding and Molecular Structure

Ionic Bonding: General characteristics of ionic bonding. Energy considerations in ionic bonding, lattice energy and solvation energy and their importance in the context of stability and solubility of ionic compounds. Statement of Born-Landé equation for calculation of lattice energy, Born-Haber cycle and its applications, polarizing power and polarizability. Fajan's rules, ionic character in covalent compounds, bond moment, dipole moment and percentage ionic character.

Covalent bonding: VB Approach: Shapes of some inorganic molecules and ions on the basis of VSEPR and hybridization with suitable examples of linear, trigonal planar, square planar, tetrahedral, trigonal bipyramidal and octahedral arrangements.

Concept of resonance and resonating structures of simple inorganic compounds

MO Approach: Rules for the LCAO method, bonding and antibonding MOs and their characteristics for *s-s*, *s-p* and *p-p* combinations of atomic orbitals, nonbonding combination of orbitals, MO treatment of homonuclear diatomic molecules of 1st and 2nd periods (including idea of *s-p* mixing) and heteronuclear diatomic molecules such as CO, NO and NO⁺. Comparison of VB and MO approaches. Molecular orbital structures and bond orders of homo and hetero atomic molecules like H₂, He₂, He₂⁺, N₂, O₂, HF, and CO, Prediction of magnetic properties of these species

Coordinate bond: Explanation by taking NH₃-BF₃ molecule as example.

Hydrogen bonding: Definition, inter and intra molecular Hydrogen bonding by taking HF, H₂O, and nitrophenols as examples. Anomalous properties like physical state, boiling point and solubility. Structure of ice Theories (or nature) of hydrogen bond (electrostatic approach, VBT and MOT treatments)

(16 Lectures)

Section B: Organic Chemistry-1 (30 Periods)

Fundamentals of Organic Chemistry

Physical Effects, Electronic Displacements: Inductive Effect, Electromeric Effect, Resonance and Hyperconjugation. Cleavage of Bonds: Homolysis and Heterolysis.

Basic Concepts of Organic reaction mechanism:-

Homolytic and heterolytic cleavages, electrophiles and nucleophiles (their nature with examples). Meaning of the terms with their illustration to show the formation of Carbocations, Carbanions, Free radicals. Stability and structure of primary, secondary and tertiary carbocations, carbanions, free radicals. Strength of organic acids and bases: Comparative study with emphasis on factors affecting K_a OR pK values. (acetic acid, propionic acid, Butanoic acid) & (Methylamine, ethylamine and aniline)

(8 Lectures)

Stereochemistry: Types of stereoisomers.

Optical isomerism; Definition, Elements of symmetry (Plane, centre and alternate axis) Chirality, Optical activity. Optical isomerism in lactic acid, tartaric acid and

biphenyls. Racemisation, Resolution, methods of resolution (Chemical and biochemical) Walden Inversion, Asymmetric synthesis (Partial and Absolute) Diastereomers, R/S Nomenclature- CIP rules (upto 2 carbon atoms) Geometrical isomerism- Definition with examples, Geometrical isomerism in aldoximes and ketoximes. Determination of configuration- Beckmann rearrangement.). E / Z Nomenclature (for upto two C=C systems). Conformations with respect to ethane, butane and cyclohexane. Interconversion of Wedge Formula, Newmann, Sawhorse and Fischer representations.

(10 Lectures)

Aliphatic Hydrocarbons

Functional group approach for the following reactions (preparations & reactions) to be studied in context to their structure.

Alkanes: Preparation: Catalytic hydrogenation of alkenes, Wurtz reaction, Kolbe's synthesis, from Grignard reagent. Reactions: Free radical Substitution: Halogenation.

Alkenes: Preparation: Elimination reactions: Dehydration of alcohols and dehydrohalogenation of alkyl halides (Saytzeff's rule); Reactions: cis-addition (alk. KMnO_4) and trans-addition (bromine), Addition of HX (Markownikoff's and anti-Markownikoff's addition), Hydration, Ozonolysis, oxymecuration-demercuration, Hydroboration-oxidation.

Dienes: Types, relative stabilities of dienes, 1,3 Butadiene, 1,2 and 1,4-addition reactions with H_2 and HBr, Diel's Alder reaction with an example

Alkynes: Preparation: Acetylene from CaC_2 and conversion into higher alkynes; by dehalogenation of tetra halides and dehydrohalogenation of vicinal-dihalides.

Reactions: formation of metal acetylides, Acidity of alkynes, addition of bromine, HCN, Acetic acid, water, oxidation with KMnO_4 and ozonolysis

(12 Lectures)

Aromatic hydrocarbons; Aromaticity: Benzenoids and Hückel's rule

Preparation (Case benzene): from phenol, by decarboxylation of Carboxylic acids, from acetylene. **Reactions:** Electrophilic substitution: Mechanisms of nitration, halogenation and sulphonation. Friedel-Craft's reaction (alkylation and acylation) Orientation- Orienting influence of o-p and m- directing groups

(8 Lectures)

Reference:

- Lee, J.D. Concise Inorganic Chemistry ELBS, 1991.
- Cotton, F.A., Wilkinson, G. & Gaus, P.L. Basic Inorganic Chemistry, 3rd ed., Wiley.
- Douglas, B.E., McDaniel, D.H. & Alexander, J.J. Concepts and Models in Inorganic Chemistry, John Wiley & Sons.
- Huheey, J.E., Keiter, E.A., Keiter, R.L. & Medhi, O.K. Inorganic Chemistry: Principles of Structure and Reactivity, Pearson Education India, 2006.
- Graham Solomon, T.W., Fryhle, C.B. & Snyder, S.A. Organic Chemistry, John Wiley & Sons (2014).
- McMurry, J.E. Fundamentals of Organic Chemistry, 7th Ed. Cengage Learning India Edition, 2013.
- Sykes, P. A Guidebook to Mechanism in Organic Chemistry, Orient Longman, New Delhi (1988).
- Eliel, E.L. Stereochemistry of Carbon Compounds, Tata McGraw Hill education, 2000.

- Finar, I.L. Organic Chemistry (Vol. I & II), E.L.B.S.
- Morrison, R.T. & Boyd, R.N. Organic Chemistry, Pearson, 2010.
- Bahl, A. & Bahl, B.S. Advanced Organic Chemistry, S. Chand, 2010
- R D Madan, Inorganic Chemistry

DMB24001/ DMB24005/ DMB24008

SEMESTER-II

CHEMICAL ENERGETICS AND ORGANIC CHEMISTRY

Theory: 60 Lectures

Course outcome:

After completion of the course the student is able to:

- CO1: Understand the concept of thermodynamics.
- CO2: Learn the concept of ionic equilibria.
- CO3: Understand the mechanisms involved in functional Organic Chemistry.
- CO4: Study the applications of electrochemistry.

CHEMISTRY-DSC 2B:

Section A: Physical Chemistry-1

(30 Lectures)

Chemical energetics

Laws of thermodynamics, Zeroth law of thermodynamics- statement

First law thermodynamics – statements, mathematical expressions internal energy and its significance.

Enthalpy; Calculation of bond energy, bond dissociation energy and resonance energy from thermochemical data. Variation of enthalpy of a reaction with temperature – Kirchoff's equation

Second law of thermodynamics – spontaneous, non-spontaneous and equilibrium processes, different ways of stating second law – (Clausius, spontaneity, entropy), heat engine, Carnot cycle and its efficiency (derivation). Concept of entropy and its significance in terms of randomness and probability

Free energy – Helmholtz and Gibbs free energy and their relationship, Gibbs – Helmholtz's equation at constant pressure and volume (derivations). Thermodynamic criteria of equilibrium and spontaneity, variation of free energy with temperature and pressure. Clausius – Clapeyron's equation (to be derived). Applications of integrated form of Clausius – Clapeyron equation and its applications. Van't Hoff's reaction isotherm and isochore equations (derivation), Statement of third law of thermodynamics. (Numerical problems)

Statistical Thermodynamics: Introduction, types of statistics. Importance of each statistics. Expression for Bose-Einstein's Statistics (equation to be given) **(12 Lectures)**

Chemical Equilibrium:

Characteristics of chemical equilibrium, Law of mass action, equilibrium constant, Free energy change in a chemical reaction. Thermodynamic derivation of the law of chemical equilibrium. Distinction between G and G° , Le Chatelier's principle. effect of change in concentration and temperature. Application of law of mass action for the formation of ammonia, dissociation of phosphorous pentachloride, Relationships between K_p , K_c and K_x for reactions involving ideal gases.

(6 Lectures)

Ionic Equilibria:

Strong, moderate and weak electrolytes, degree of ionization, factors affecting degree of ionization, Arrhenius theory of electrolytic dissociation, merits and demerits, Kohlrausch's law of independent migration of ions and applications. Transport number by moving

boundary method, ionization constant and ionic product of water. Ionization of weak acids and bases, pH scale, common ion effect. Salt hydrolysis-calculation of hydrolysis constant, degree of hydrolysis and effect of temperature on degree of hydrolysis, pH for different salts. Buffer solutions. Solubility and solubility product of sparingly soluble salts – applications of solubility product principle. Activity and activity co-efficients, definition and their relation, ionic strength and its calculation. Debye Huckel theory of strong electrolytes (relaxation time effect, electrophoretic effect and viscous effect). Debye-Huckel Onsager equation (no derivation), Debye-Huckel limiting equation for activity co-efficients (no derivation). Role of solvents in altering the strengths of acids and bases. **(12 Lectures)**

Section B: Organic Chemistry-2 (30 Lectures)

Functional group approach for the following reactions (preparations & reactions) to be studied in context to their structure.

Alkyl and Aryl Halides

Alkyl Halides: Nomenclature-Preparation: from alkenes and alcohols. Reactions- Types of Nucleophilic Substitution (SN^1 , SN^2 and SN^i) reactions. Mechanisms, Energy profile diagram for SN^1 and SN^2 reactions. Reactions: Elimination reactions- E_1 and E_2 , Mechanisms and hydrolysis

Aryl Halides: Preparation: (Chloro, bromo and iodo-benzene case): from phenol, Sandmeyer & Gattermann reactions. Reactions (Chlorobenzene): Aromatic nucleophilic substitution (replacement by $-OH$ group) and effect of nitro substituent. Benzyne Mechanism: KNH_2/NH_3 (or $NaNH_2/NH_3$).

Reactivity and Relative strength of C-Halogen bond in alkyl, allyl, benzyl, vinyl and aryl halides.

(8 Lectures)

Alcohols, Phenols and Ethers

Alcohols: Preparation: Preparation of 1° , 2° and 3° alcohols: using Grignard reagent, Ester hydrolysis, Reduction of aldehydes, ketones, carboxylic acid and esters.

Reactions: With sodium, HX (Lucas test), esterification, oxidation (alk. $KMnO_4$, acidic dichromate,). Oppeneauer oxidation, Interconversions among primary, secondary and tertiary alcohols.

Diols: Oxidation of diols. Pinacol-Pinacolone rearrangement.

Trihydric alcohols- Glycerol-Synthesis from propene, Reactions of glycerol with HI, oxalic acid, HNO_3 and dehydrating agent (P_2O_5 or H_2SO_4).

Phenols: (Phenol case) Preparation: Cumene hydroperoxide method, from diazonium salts. Reactions: Electrophilic substitution: Nitration, halogenation and sulphonation. Reimer-Tiemann Reaction, Gattermann-Koch Reaction, Schotten – Baumann Reaction. Acidity of phenols. Effect of substitution on acidity of phenols.

Ethers: Nomenclature, Williamson ether synthesis, reactions of ethers-Cleavage. Ziesel's method of estimation. Epoxides:- Synthesis, Acid and Base catalyzed opening of epoxides. Crown ethers: Introduction and applications

Aldehydes and ketones (aliphatic and aromatic): (Formaldehyde, acetaldehyde, acetone and benzaldehyde) Preparation: from acid chlorides and from alcohols.

Reactions – Reaction with HCN, ROH, $NaHSO_3$, NH_2-G derivatives. Iodoform test. Mechanisms of Aldol Condensation, Cannizzaro's reaction, perkins reaction and , Benzoin condensation.

Illustrations of Clemensen reduction, Wolff Kishner reduction, Meerwein-Ponndorf-Verley reduction, Gattermann-Koch reaction and Baeyer-Villiger oxidation.

(14 Lectures)

Carboxylic acids and their derivatives; Preparation: From Nitriles and by Arndt-Eistert reaction. Acidic and Alkaline hydrolysis of esters. Reactions: Hell – Vohlard–Zelinsky Reaction. Acidity of carboxylic acids. Resonance structure of carboxylate ion and its stability. Effect of substitution on acidity of carboxylic acids

Carboxylic acid derivatives (aliphatic): Preparation of Acid chlorides, Anhydrides, Esters and Amides from acids.

Hydroxy acids: - Synthesis of Lactic, Tartaric and citric acids. Effect of heat on α, β, γ -hydroxyl acids **(8 Lectures)**

Reference:

- Graham Solomon, T.W., Fryhle, C.B. & Snyder, S.A. Organic Chemistry, John Wiley & Sons (2014).
- McMurry, J.E. Fundamentals of Organic Chemistry, 7th Ed. Cengage Learning India Edition, 2013.
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- Castellan, G.W. Physical Chemistry 4th Ed. Narosa (2004).
- Kotz, J.C., Treichel, P.M. & Townsend, J.R. General Chemistry Cengage Learning India Pvt. Ltd., New Delhi (2009).
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DMC24001/ DMC24005/ DMC24008

**SEMESTER-III
SOLUTIONS AND ORGANIC CHEMISTRY**

Course outcome:

After completion of the course the student is able to:

CO1: Understand the concepts of electrochemistry.

CO2: Study organometallic compounds.

CO3: Learn the synthesis and reactions of amino acids, carbohydrates, alkaloids, vitamins, hormones and terpenes.

CO4: Understand the qualitative organic analysis of organic compounds and enthalpy reactions.

CHEMISTRY-DSC 2C:**Theory: 60 Lectures****Section A: Physical Chemistry-2 (30 Lectures)****Solutions**

Concentrations-different ways of expression, solutions of gases in gases, Henry's law, Thermodynamics of ideal solutions: Ideal solutions and Raoult's law, deviations from Raoult's law – non-ideal solutions. Vapour pressure-composition and temperature-composition curves of ideal and non-ideal solutions. Distillation of solutions. Lever rule. Azeotropes. Chemical potential of ideal and non ideal solution: Gibbs Duhem-Morgules equation. Entropy change of mixing for an ideal solution

Liquid mixtures:

Classification of binary mixtures into partially miscible, completely miscible and immiscible pairs of liquids. Principle of steam distillation Raoult's law, Critical solution temperature, partially miscible liquids: phenol water system, triethyl-amine water system and nicotine water system, Binary mixtures of completely miscible liquids, vapour pressure – composition diagrams and vapour pressure – temperature diagram. Classification into types- obeying Raoult's law (type I), positive deviation (type II), negative deviation (type III) from Raoult's law. Principles of fractional distillation, fractional distillation type I, type II and type III liquid mixtures, azeotropic mixtures. Binary mixtures of completely miscible liquids, principles of steam distillation – applications. (to be briefed)

Colligative properties

Introduction: vapour pressure, variation of vapour pressure with temperature (explanation with graph). Definition of boiling point and freezing point. Effect of dissolution of solute, vapour pressure of the solvent, lowering of vapour pressure, Raoult's law – relation between relative lowering of vapour pressure and molar mass. Determination of molar mass of solute by dynamic method, problems.

Elevation of boiling point: – definition and its relation to lowering of vapour pressure and molar mass (to be derived). Ebullioscopic constant of the solvent and its relation to the boiling point (only equation). Determination of molar mass of the solute by Walker-Lumsden's method,

Depression in freezing point:– definition. Relation to lowering of vapour pressure and molar mass (to be derived). Cryoscopic constant, its relation to the melting point (only equation). Determination of molar mass of non-volatile solute by Beckmann's method. Abnormal molecular weights – causes - vant Hoff's factor, evaluation of degree of dissociation and association. Problems pertaining to all the colligative properties

(12 Lectures)

Phase Equilibrium

Phases, components and degrees of freedom of a system, criteria of phase equilibrium. Gibbs Phase Rule and its thermodynamic derivation. Phase diagrams of one-component systems (water and sulphur) and two component systems involving eutectics, congruent and incongruent melting points (lead-silver, $\text{FeCl}_3\text{-H}_2\text{O}$ and Na-K only).

(5 Lectures)

Conductance and Electrochemistry

Introduction, conductance (specific, equivalent and molar conductance – definition and SI unit), conductance cell and cell constant, determination of equivalent conductance by meter-bridge method, variation of Λ and k with dilution, ionic mobility, ionic conductance and their relationship. Kohlrausch's law and its significance, transport number – definition and explanation, anomalous transport number, relationship between ionic conductance and transport number (to be derived). Determination of transport number by Hittorff's and moving boundary method (transport number of H^+ using CdCl_2 as supporting electrolyte) (Numerical problems to be worked out).

Application of conductance measurement:

- a) Solubility and solubility product of sparingly soluble salt.
- b) Ionic product of water
- c) degree of ionization of weak electrolyte
 - I. conductometric titration (strong acid Vs strong base, weak acid Vs strong base, strong acid Vs weak base, weak acid and weak base with examples)
 - II. hydrolysis constant (taking aniline hydrochloride as an example)

Electromotive force:

Electrolytic and electro chemical cells,.Single electrode potential, sign of electrode potential (reduction potential to be adopted) convention of representing a cell, electrode reaction of a daniel cell. EMF and standard EMF of a cell, cell reaction, reversible and irreversible cells. Nernst equation (to be derived) and calculation of electrode potential, primary reference electrode – standard hydrogen electrode, secondary reference electrode – calomel and Ag - AgCl electrode – construction and working, electro-chemical series, equilibrium constant and free energy of a cell reaction, and its derivation, concentration cells with and without transference, EMF of concentration cells, liquid junction potential and salt bridge. Numerical problems on Nernst equation and EMF calculation. Fuel cells – working of H_2O_2 fuel cell and its importance.

Application of EMF measurements:

- a) Determination of pH of a solution using quinhydrone electrode and glass electrode using dip type calomel electrode – principles and procedure.
- b) Potentiometric titration – principle, location of end points in neutralization reactions (NaOH Vs HCl), Oxidation – reduction reactions ($\text{K}_2\text{Cr}_2\text{O}_7$ Vs FAS), precipitation reaction (KCl Vs AgNO_3) and complex reactions (ZnSO_4 Vs $\text{K}_3[\text{Fe}(\text{CN})_6]$)

(13 Lectures)

Section B: Organic Chemistry-2 (30 Lectures)

Organometallic compounds:-

Definition with example. Organo magnesium compounds (Grignard reagents) Formation ethyl magnesium bromide and its synthetic applications (synthesis of alcohols, acids, aldehydes, ketones and carboxylic acids)

Organo zinc compounds:- Preparation of diethyl zinc and its applications

Organolithium Compounds:- Preparation and synthetic applications of LDA

Amines and Diazonium Salts

Amines:- Definition, classification with example. Synthesis by Gabriel phthalimide method, reduction of amides. Separation of amine mixture by Hinsberg's method. Distinction tests for 1°, 2°, 3° amines (acetylation and Hoffmann's exhaustive methylation). Action of nitrous acid on different amines (Both aliphatic and aromatic 1°, 2°, 3° amines), basicity of amines, effect of substituent on basicity of aliphatic and aromatic amines. Hoffmann-Martius rearrangement.

Diazonium Compounds: Preparation, mechanism of preparation and synthetic applications of benzene diazonium chloride. Conversion to phenol, halobenzene, phenyl hydrazine and coupling reaction.

(10 Lectures)

Amino Acids, Peptides and Proteins:

Preparation of Amino Acids: Strecker synthesis using Gabriel's Phthalimidesynthesis. Zwitterion, Isoelectric point and Electrophoresis.

Reactions of Amino acids: ester of -COOH group, acetylation of NH₂ group, complexation with Cu²⁺ ions,

Elementary account of Primary, Secondary, Structure of proteins. Peptides (Amides) Synthesis of simple peptides (upto dipeptides) by N-protection (t-butyl oxycarbonyl and Carbobenzyloxy carbonyl) (4 Lectures)

Carbohydrates: Definition and importance, classification based on composition with examples-reducing and non-reducing sugars. Monosaccharides:- Glucose- reactions of glucose (with H₂N-OH, HCN, C₆H₅NHNH₂, Br₂ water, Conc. HNO₃, reductions with HI/red P, Methanol (dry HCl), acetic anhydride and reduction reactions. Mutarotation. Structural elucidation of glucose and fructose :- open chain structure, ring structure- Fisher and Haworth structure. Determination of ring size by methylation method. Fischer and Haworth structures of fructose, galactose and mannose Interconversions reactions- 1) Ascending (Killiani's synthesis) 2) Descending (Wohl's degradation) 3) Aldose to Ketose 4) Ketose to Aldose 5) Epimerization

Disaccharides:- structural elucidation of sucrose, structural formulae of maltose and lactose (Haworth structure). Polysaccharides:- Partial structural formulae of starch and Cellulose. (8 Lectures)

Alkaloids:- definition, classification based on heterocyclic rings-isolation, synthesis and structural elucidation of nicotine. Structure of Morphine, Atropine, Cocaine & physiological importance of alkaloids.

Vitamins:- Definition, classification, structural elucidation and synthesis of vit-A, Synthesis of vit-C, Sources & importance of Vitamin-B, calciferol, E, D & K

Hormones:- definition, classification, synthesis and functions of adrenaline and thyroxine.

Terpenes:- definition, isoprene rule, Classification, isolation (Solvent extraction and Steam distillation), structural elucidation of citral and its synthesis, structural formulae of α-terpeneol, Camphor and menthol. (8 Lectures)

Reference:

- Barrow, G.M. Physical Chemistry Tata McGraw-Hill (2007).
- Castellan, G.W. Physical Chemistry 4th Ed. Narosa (2004).
- Kotz, J.C., Treichel, P.M. & Townsend, J.R. General Chemistry, Cengage Learning India Pvt. Ltd.: New Delhi (2009).
- Mahan, B.H. University Chemistry, 3rd Ed. Narosa (1998).
- Petrucci, R.H. General Chemistry, 5th Ed., Macmillan Publishing Co.: New York (1985).
- Morrison, R. T. & Boyd, R. N. Organic Chemistry, Dorling Kindersley (India) Pvt. Ltd. (Pearson Education).
- Finar, I. L. Organic Chemistry (Volume 1), Dorling Kindersley (India) Pvt. Ltd. (Pearson Education).
- Finar, I. L. Organic Chemistry (Volume 2), Dorling Kindersley (India) Pvt. Ltd. (Pearson Education).
- Nelson, D. L. & Cox, M. M. Lehninger's Principles of Biochemistry 7thEd., W. H. Freeman.
- Berg, J.M., Tymoczko, J.L. & Stryer, L. Biochemistry, W.H. Freeman, 2002.

- Puri & Sharma, A Textbook of Chemistry

SEMESTER-IV

COORDINATION CHEMISTRY AND PHYSICAL CHEMISTRY

Theory: 60 Lectures

Course outcome:

After completion of the course the student is able to:

CO1: Know about co-ordination chemistry.

CO2: Understand kinetic theory of gases, properties of liquids and crystallography.

CO3: Acquire knowledge on the qualitative analysis of mixtures.

CHEMISTRY-DSC 2D:

Chemistry of transition elements:

Position in the periodic table, electronic configuration, general characteristics- atomic and ionic radii, ionization energy, variable oxidation states, (Latimer diagrams) spectral properties, redox potentials, colour and magnetic properties, catalytic activity, complex formation and interstitial compounds formation (3d, 4d and 5d series).

Chemistry of inner transition elements: Lanthanides: Electronic configuration and position in the periodic table, oxidation states, spectral properties, colour and magnetic properties, complex formation and ionic radii, lanthanide contraction – cause & its consequences and solvent extraction method.

General survey of actinides – comparison with lanthanides, transuranic elements. Action of ion exchange resins – cation exchange and anion exchange resins, exchange of inorganic ions, ion exchange capacity, separation of lanthanides by ion-exchange method. Comparison of d and f block elements.

(12 Lectures)

Coordination Chemistry

Ligands, classification of ligands and chelation, nomenclature of co-ordination compounds, physical methods in the study of complexes – change in conductance, colour and pH. Stability of complexes – stability constant, a brief outline of thermodynamic stability of metal complexes, factors affecting the stability of complexes. Polynuclear complexes, inner metallic complexes. Sidwick-EAN rule

Isomerism in co-ordination complexes: Stereo-isomerism – Geometrical and optical isomerism exhibited by co-ordination compounds of co-ordination number 4 and 6.

Metal-ligand bonding in transition metal complexes:

Valence bond theory: Salient features, formation of octahedral complexes on the basis of VBT, outer and inner orbital octahedral complexes- $[\text{Fe}(\text{CN})_6]^{4-}$, $[\text{Fe}(\text{CN})_6]^{3-}$, $[\text{Co}(\text{CN})_6]^{3-}$, $[\text{CoF}_6]^{3-}$ $[\text{Cr}(\text{H}_2\text{O})_6]^{3+}$ and $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$. Formation of tetrahedral and square planar complexes on the basis of VBT – $[\text{Ni}(\text{CN})_4]^{2-}$, $[\text{Cu}(\text{NH}_3)]^{2+}$, $[\text{Zn}(\text{NH}_3)_4]^{2+}$ and $[\text{Ni}(\text{CO})_4]$, limitations of VBT.

Crystal field theory: Important features of crystal field theory, crystal field splitting of d-orbitals in tetrahedral, octahedral and square planar complexes, crystal field stabilization energy (CFSE), factors affecting the magnitude of Δ_o , (nature of ligand, oxidation state of the metal ion, size of the orbitals, geometry of the complex), high spin (HS) and low spin (LS) complexes Spectrochemical series, magnetic properties of metal complexes based on crystal field theory- $[\text{Co}(\text{NH}_3)_6]^{3+}$, $[\text{CoF}_6]^{3-}$, $[\text{Fe}(\text{CN})_6]^{4-}$,

[Fe(CN)₆]³⁻ and [Ni(CN)₄]²⁻. Magnetic susceptibility, measurement of magnetic moment by Gouy's method. Tetragonal distortion of octahedral geometry. Jahn-Teller distortion. Applications of complex formation in biological systems

(18

Lectures)

Section B: Physical Chemistry-3

(30 Lectures)

Kinetic Theory of Gases

Postulates of Kinetic Theory of Gases and derivation of the kinetic gas equation. Deviation of real gases from ideal behaviour, compressibility factor, causes of deviation. van der Waals equation of state for real gases. Boyle temperature (derivation not required). Critical phenomena, critical constants and their calculation from van der Waals equation. Andrews isotherms of CO₂.

Maxwell Boltzmann distribution laws of molecular velocities and molecular energies (graphic representation – derivation not required) and their importance.

Temperature dependence of these distributions. Most probable, average and root mean square velocities (no derivation). Collision cross section, collision number, collision frequency, collision diameter and mean free path of molecules. Viscosity of gases and effect of temperature and pressure on coefficient of viscosity (qualitative treatment only)

(8 Lectures)

Liquids

Properties of liquids

Viscosity : Definition of coefficient of viscosity, factors affecting viscosity – temperature, size, mass, shape of molecules, intermolecular forces, determination of viscosity of liquids by Ostwald's method..

Surface tension : Definition, effect of temperature and solute on surface tension. Determination of surface tension of liquids using stalagmometer. Effect of temperature on surface tension and coefficient of viscosity of a liquid (qualitative treatment only).

Parachor : Definition, sudegen equation, calculation of parachor and its application w.r.t. the elucidation of structures- of benzene, quinine and isocyanide ion (Numerical problems).

Polarization- induced, orientation and molar polarization – definitions, Clausius-Mossotti equation (no derivation) and its application. (8 Lectures)

Solids :

Introduction, laws of crystallography - law of constancy of interfacial angles, law of rational indices- Weiss and Miller indices. Unit cell, Space lattices and lattice planes, seven crystal systems, lattice planes in cubic crystals (Simple cubic, body centered cubic and face centered cubic).

Elements of symmetry – plane, axis and centre, elements of symmetry in cubic system, types of lattices, Bragg's equation and its derivation. X-ray diffraction and determination of crystal structure of rock salt by rotating crystal method. Application of X-ray studies – distance between lattice planes, density of crystals, determination of Avogadro number. (Numerical problems) Defects in crystals.

Liquid crystals:

Mesomorphic state – definition, classification of liquid crystals smectic and nematic with examples, molecular arrangement in the two types and uses. Nano materials – definition, properties and application **(8 Lectures)**

Chemical Kinetics:

The concept of reaction rates. Effect of temperature, pressure, catalyst and other factors on reaction rates. Order and molecularity of a reaction. Derivation of integrated rate equations for zero, first and second order reactions (both for equal and unequal concentrations of reactants). Half-life of a reaction. General methods for determination of order of a reaction (Differential method, integration, half-life period and isolation methods) Concept of activation energy and its calculation from Arrhenius equation.

Theories of Reaction Rates: Collision theory based on hard sphere model and Activated Complex theory of bimolecular reactions. Comparison of the two theories (qualitative treatment only). **(6 Lectures)**

Reference:

- Barrow, G.M. Physical Chemistry Tata McGraw-Hill (2007).
- Castellan, G.W. Physical Chemistry 4th Ed. Narosa (2004).
- Kotz, J.C., Treichel, P.M. & Townsend, J.R. General Chemistry Cengage Learning India Pvt. Ltd., New Delhi (2009).
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- Cotton, F.A. & Wilkinson, G. Basic Inorganic Chemistry, Wiley.
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- Wulfsberg, G. Inorganic Chemistry, Viva Books Pvt. Ltd.
- Rodgers, G.E. Inorganic & Solid State Chemistry, Cengage Learning India Ltd., 2008.
- R D Madan, Textbook of Chemistry
- Madan, Malik Tuli, Comprehensive Chemistry
- Satyaprakash, Text book of Chemistry

INORGANIC MATERIALS OF INDUSTRIAL IMPORTANCE

Theory: 60 Lectures

Course outcome:

After completion of the course the student is able to:

- CO1: Understand the synthesis and applications of glass and ceramics, vitamins, hormones, soaps and detergents; and higher aspects of spectroscopy.
- CO2: Understand the types and manufacture of different fertilizers.
- CO3: Understand the different methods of prevention of corrosion.

CHEMISTRY-DSE:

Silicate Industries

Glass: Rawmaterials, Glassy state and its properties, classification (silicate and non-silicate glasses). Manufacture and processing of glass. Composition and properties of the following types of glasses: Soda lime glass, lead glass, armoured glass, safety glass, borosilicate glass, fluorosilicate, coloured glass, photosensitive glass.

Ceramics: Important clays and feldspar, ceramic, their types and manufacture. Hightechnology ceramics and their applications, superconducting and semiconducting oxides, fullerenes carbon nanotubes and carbon fibre.

Cement: Classification of cement, ingredients and their role, Manufacture of cement and the setting process, quick setting cements.

Abrasives: Definition, classification with examples – hardness, manufacture and applications of carborundum, alundum and tungsten carbide.

Refractories: Definition, properties, classification with examples. Different steps involved in the manufacture of refractories. Applications of refractories.

Inorganic polymers:

Definition – examples, general properties, comparison with organic polymers, glass transition temperature Silicones: Definition, nomenclature, preparation (linear, cross-linked and cyclic). Factors affecting the nature of silicon polymers, properties (chemical and thermal stabilities(chemical properties) uses of silicon polymers, silicon fluids/oils – uses, silicon elastomers / rubbers, silicon resins (preparation and uses)

Phosphazenes: Definition, types, structures, preparation, properties and uses. Crystalline polymetaphosphates – Maddrell's and Kuroll's salts – properties and uses. Nature of bonding in phosphazenes.

Fluorocarbons: Definition, examples, preparation, properties and uses of Freon-12, Freon-22, PTFE and poly per fluorovinyl chloride.

Non-aqueous solvents:

Liquid ammonia- Reasons for the solvent properties, typical reactions- solubility of alkali metals; acid-base, precipitation, ammonolysis, Ionization of weak acids, advantages and disadvantages.

Liquid SO₂- Reasons for the solvent properties, typical reactions-acid-base, solvolysis, precipitation, amphoteric and redox reactions **(25 Lectures)**

Fertilizers:

Different types of fertilizers. Manufacture of the following fertilizers: Urea, ammonium nitrate, calcium ammonium nitrate, ammonium phosphates; polyphosphate, superphosphate, compound and mixed fertilizers, potassium chloride, potassium sulphate. **(5 Lectures)**

Surface Coatings:

Objectives of coatings surfaces, preliminary treatment of surface, classification of surface coatings. Paints and pigments-formulation, composition and related properties. Oil paint, Vehicle, modified oils, Pigments, toners and lakes pigments, Fillers, Thinners, Enamels, emulsifying agents. Special paints (Heat retardant, Fire retardant, Eco-friendly paint, Plastic paint), Dyes, Wax polishing, Water and Oil paints, additives, Metallic coatings (electrolytic and electroless), metal spraying and anodizing.

(10 Lectures)

Batteries:

Primary and secondary batteries, battery components and their role, Characteristics of Battery. Working of following batteries: Pb acid, Li-Battery, Solid state electrolyte battery. Fuel cells, Solar cell and polymer cell.

(4 Lectures)

Alloys:

Classification of alloys, ferrous and non-ferrous alloys, Specific properties of elements in alloys. Properties and applications of steel. Manufacture of Steel (removal of silicon decarbonization, demanganization, desulphurization dephosphorisation) and surface treatment (argon treatment, heat treatment, nitriding, carburizing). Composition and properties of different types of steels. Production of Ferro alloys: Ferro chrome and Ferro manganese.

(8 Lectures)

Catalysis:

General principles and properties of catalysts, homogenous catalysis (catalytic steps and examples) and heterogenous catalysis (catalytic steps and examples). Theories of catalysis, Auto catalyst. Industrial applications, Deactivation or regeneration of catalysts. Phase transfer catalysts, application of zeolites as catalysts.

(4 Lectures)

Chemical explosives:

Origin of explosive properties in organic compounds, Classification with examples, preparation and explosive properties of lead azide, PETN, cyclonite (RDX). Introduction to rocket propellants.

(4 Lectures)

Reference:

- E. Stocchi: Industrial Chemistry, Vol-I, Ellis Horwood Ltd. UK.
- R. M. Felder, R. W. Rousseau: Elementary Principles of Chemical Processes, Wiley Publishers, New Delhi.
- W. D. Kingery, H. K. Bowen, D. R. Uhlmann: Introduction to Ceramics, Wiley Publishers, New Delhi.
- J. A. Kent: Riegel's Handbook of Industrial Chemistry, CBS Publishers, New Delhi.
- P. C. Jain & M. Jain: Engineering Chemistry, Dhanpat Rai & Sons, Delhi.
- R. Gopalan, D. Venkappayya, S. Nagarajan: Engineering Chemistry, Vikas Publications, New Delhi.
- B. K. Sharma: Engineering Chemistry, Goel Publishing House, Meeru

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DSE!B

SEMESTER V

INDUSTRIAL CHEMICALS AND ENVIRONMENT

Theory: 60 Lectures

Course outcome:

After completion of the course the student is able to:

CO1: Understand the environmental issues

CO2: Know about the metallurgy of metals

CO3: Understand the hazards and handling certain gases and chemicals

Industrial Gases and Inorganic Chemicals

Industrial Gases: Large scale production, uses, storage and hazards in handling of the following gases: oxygen, nitrogen, argon, neon, helium, hydrogen, acetylene, carbon monoxide, chlorine, fluorine, sulphur dioxide and phosgene.

Inorganic Chemicals: Manufacture, application, analysis and hazards in handling the following chemicals: hydrochloric acid, nitric acid, sulphuric acid, caustic soda,

common salt, borax, bleaching powder, sodium thiosulphate, hydrogen peroxide, potash alum, chrome alum, potassium dichromate and potassium permanganate.

(10 Lectures)

Industrial Metallurgy

Preparation of metals (ferrous and nonferrous) and ultrapure metals for semiconductor technology

(4 Lectures)

Environment and its segments

Ecosystems. Biogeochemical cycles of carbon, nitrogen and sulphur.

Air Pollution: Major regions of atmosphere. Chemical and photochemical reactions in atmosphere. Air pollutants: types, sources, particle size and chemical nature; Photochemical smog: its constituents and photochemistry. Environmental effects of ozone, Major sources of air pollution

Pollution by SO₂, CO₂, CO, NO_x, H₂S and other foul smelling gases. Methods of estimation of CO, NO_x, SO_x and control procedures

Effects of air pollution on living organisms and vegetation. Greenhouse effect and Global warming, Ozone depletion by oxides of nitrogen, chlorofluorocarbons and Halogens, removal of sulphur from coal. Control of particulates

Water Pollution: Hydrological cycle, water resources, aquatic ecosystems, Sources and nature of water pollutants, Techniques for measuring water pollution, Impacts of water pollution on hydrological and ecosystems

Water purification methods: Effluent treatment plants (primary, secondary and tertiary treatment). Industrial effluents from the following industries and their treatment: electroplating, textile, tannery, dairy, petroleum and petrochemicals, agro, fertilizer, etc. Sludge disposal.

Industrial waste management, incineration of waste. Water treatment and purification (reverse osmosis, electro dialysis, ion exchange). Water quality parameters for waste water, industrial water and domestic water

(30 Lectures)

Energy & Environment

Sources of energy: Coal, petrol and natural gas. Nuclear Fusion / Fission, Solar energy, Hydrogen, geothermal, Tidal and Hydel, etc.

Nuclear Pollution: Disposal of nuclear waste, nuclear disaster and its management.

(10 Lectures)

Biocatalysis

Introduction to biocatalysis: Importance in “Green Chemistry” and Chemical Industry.

(6 Lectures)

Reference:

- E. Stocchi: *Industrial Chemistry*, Vol-I, Ellis Horwood Ltd. UK.
- R.M. Felder, R.W. Rousseau: *Elementary Principles of Chemical Processes*, Wiley Publishers, New Delhi.
- J. A. Kent: *Riegel’s Handbook of Industrial Chemistry*, CBS Publishers, New Delhi.
- S. S. Dara: *A Textbook of Engineering Chemistry*, S. Chand & Company Ltd. New Delhi.
- K. De, *Environmental Chemistry*: New Age International Pvt., Ltd, New Delhi.
- S. M. Khopkar, *Environmental Pollution Analysis*: Wiley Eastern Ltd, New Delhi.
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**ORGANOMETALLICS, BIOINORGANIC CHEMISTRY,
POLYNUCLEAR HYDROCARBONS AND UV, IR SPECTROSCOPY**

Theory: 60 Lectures

Course outcome:

After completion of the course the student is able to:

- CO1: Understand the techniques involved in metallurgy.
 CO2: Understand the role of ions in different biological systems.
 CO3: Understand the applications of spectroscopy.

DSE-1A:

Section A: Inorganic Chemistry-4

(30 Lectures)

Metallurgy: Terms and principles involved in metallurgy, Ellingham's diagram, Types of metallurgy: Pyro metallurgy- extraction of Nickel by sulphide ore- general metallurgy followed by Mond's process (purification, Manganese from oxides ores- Reduction by the Aluminothermite process- refining by electrolytic process.

Hydro metallurgy: Extraction of Gold from native ore by cyanide process, and refining by quaternary process.

Electro metallurgy: Extraction of Lithium by fusion method Followed by electrolysis of lithium chloride.

Powder metallurgy: Importance, metal powder production & applications. Production of Tungsten powder. Principles of Electroplating. **(10 Lectures)**

Organometallic Compounds

Definition and Classification with appropriate examples based on nature of metal-carbon bond (ionic, s, p and multicentre bonds). Structures of methyl lithium, Zeiss salt and ferrocene. Preparation, properties, bonding and applications of alkyls and aryls of Li, Al, Hg, Sn and Ti.

Nature of M-CO bonding in carbonyls. Preparation, properties and structures of mononuclear and binuclear metal carbonyls- Ni(CO)₄, Cr(CO)₆, Fe(CO)₅, Mn₂(CO)₁₀, Co₂(CO)₈. Applications of EAN rule to mononuclear metal carbonyls.

Behaviour of carbon monoxide. Synergic effects (VB approach)- (MO diagram of CO can be referred to for synergic effect to IR frequencies). **(10 Lectures)**

Chemistry of nonmetals:

Boron: Boron hydrides – Diborane – preparation, properties, uses and structure
Carbon: Fullerenes- Production, structure of C₆₀ and C₇₀. Diamond and Graphite- Properties and structure.

Silicon: Structure of silica. Silicates- types of silicates with examples.

Nitrogen: Preparation (any two methods), properties, uses, structure of hydrazine, hydroxyl amine and hydrazoic acid.

Sulphur: Preparation, properties, structures and applications of thionyl chloride, sulphuryl chloride and SF₆.

Halogens: Preparation, properties and structure of bleaching powder.

Pseudo halogens: preparation, properties and structure of cyanogens, thiocyanogen, tellurocyanogen and oxocyanogen. (any one method of preparation and any three properties to be discussed). **(5 Lectures)**

Bio-Inorganic Chemistry

A brief introduction to bio-inorganic chemistry. Essential and trace elements in biological process. Role of metal ions present in biological systems with special reference to Na⁺, K⁺ and Ca²⁺, Mg²⁺ ions: Na/K pump; Role of Mg²⁺ ions in energy production and chlorophyll. Role of Ca²⁺ in blood clotting, Enzymatic role of Iron in Hemoglobin and myoglobin, Magnesium in Chlorophyll, Cobalt in Vitamin B12.

Stabilization of protein structures and structural role (bones).

Biological functions and toxicity of Cr, Mn, Co, Ni, I, Hg, Mo, and Se. **(5 Lectures)**

Section B: Organic Chemistry-4

(30 Lectures)

Polynuclear and heteronuclear aromatic compounds:

Polynuclear Hydrocarbons: Resonance structures of Naphthalene, anthracene and Phenanthracene.

Structural elucidation of naphthalene. Reactions of naphthalene- oxidation, reduction and electrophilic substitution reactions

Heterocyclic Compounds: Definition, classification with examples, synthesis of Furan, thiophene, pyrrole, pyridine, indole (Fischer method), quinoline (Skrap's synthesis), isoquinoline, pyrimidine (one method each). Aromaticity and basicity of pyrrole and pyridine. Electrophilic substitution reactions of pyrrole and pyridine.

Uric acid- Structure, Synthesis. Conversion of uric to purine and caffeine

Dyes: Colour and Constitution, Witt's theory, Classification of dyes based on structures with examples, synthesis of Methyl orange, Bismark brown, indigo and malachite green, structural elucidation of alizarin and its synthesis.

Drugs: Chemotherapy and chemotherapeutic agents, definition of drugs, types of drugs, antipyretics, analgesics, anaesthetics, sedatives, narcotics, antiseptics, antibacterials, antibiotics, antimalarials and sulpha drugs with examples. Synthesis of paracetamol, sulphanilamide, sulphaguanidine **(13 lectures)**

Active methylene compounds: Definition, Ethyl acetoacetate and diethyl malonate preparation, Mechanism of Claisen condensation, keto-enol tautomerism and its evidence. Synthetic applications of EAA and DEM:- Synthesis of mono carboxylic acids, dicarboxylic acids-succinic acid, adipic acid, antipyrine, Barbituric acid, acetyl acetone, Crotonic acid and Cinnamic acid. **(4 lectures)**

Application of Spectroscopy to Simple Organic Molecules

Application of visible, ultraviolet and Infrared spectroscopy in organic molecules. Electromagnetic radiations, electronic transitions, λ_{max} , chromophore, auxochrome, bathochromic and hypsochromic shifts. Woodward rules for calculating λ_{max} of conjugated dienes

IR-Spectroscopy: Introduction, functional group region and finger print region stretching frequency, Graphical representation of IR spectra of benzoic acid and methyl benzoate. Absorption frequencies of Simple functional groups

NMR Spectroscopy: Basic principles of proton magnetic resonance , nuclear magnetic spin quantum number I, influence of the magnetic field on the spin of nuclei, magnetic resonance-chemical shift (δ value), use of TMS as reference, nuclear shielding effects, equivalent and non-equivalent protons, spin-spin splitting.

NMR spectra of Simple organic molecules (like ethyl alcohol, ethane, propane, benzene, toluene, acetone, and methyl chloride) to be discussed.. **(10 Lectures)**

Oils, Fats and Waxes:-

Definition, composition, analysis of oils and fats- determination of acid value, saponification value, iodine value, R-M value and their importance. **Waxes:-** Definition, composition, types and applications. **Soaps and Detergents:-** Definition, types, manufacture of soaps by hot process, cleansing action of soaps. Merits and demerits of soaps and detergents **(3 Lectures)**

Reference:

- James E. Huheey, Ellen Keiter & Richard Keiter: Inorganic Chemistry: Principles of Structure and Reactivity, Pearson Publication.
- G.L. Miessler & Donald A. Tarr: Inorganic Chemistry, Pearson Publication.
- J.D. Lee: A New Concise Inorganic Chemistry, E.L.B.S.
- F.A. Cotton & G. Wilkinson: Basic Inorganic Chemistry, John Wiley & Sons.
- I.L. Finar: Organic Chemistry (Vol. I & II), E.L.B.S.
- John R. Dyer: Applications of Absorption Spectroscopy of Organic Compounds, Prentice Hall.
- R.M. Silverstein, G.C. Bassler & T.C. Morrill: Spectroscopic Identification of Organic Compounds, John Wiley & Sons.
- R.T. Morrison & R.N. Boyd: Organic Chemistry, Prentice Hall.
- Peter Sykes: A Guide Book to Mechanism in Organic Chemistry, Orient Longman.
- Arun Bahl and B. S. Bahl: Advanced Organic Chemistry, S. Chand
- Spectroscopy – William Kemp
- Spectroscopy – Y.R. Sharma

DMF24201/ DMF24205/ DMF24208

DSE1B

SEMESTER VI

**QUANTUM CHEMISTRY, SPECTROSCOPY &
PHOTOCHEMISTRY**

Theory: 60 Lectures

COURSE OUTCOME:

After completion of the course the student is able to:

CO1: Understand the concepts in quantum chemistry

CO2: Know about photochemistry

CO3: Know the concept of spectroscopy with applications

Quantum Chemistry

Postulates of quantum mechanics, quantum mechanical operators, Schrödinger equation and its application to free particle and “particle-in-a-box” (rigorous treatment), quantization of energy levels, zero-point energy and Heisenberg Uncertainty principle; wavefunctions, probability distribution functions, nodal properties, Extension to two and three dimensional boxes, separation of variables, degeneracy.

Qualitative treatment of simple harmonic oscillator model of vibrational motion: Setting up of Schrödinger equation and discussion of solution and wavefunctions. Vibrational energy of diatomic molecules and zero-point energy.

Angular momentum: Commutation rules, quantization of square of total angular momentum and z-component.

Rigid rotator model of rotation of diatomic molecule. Schrödinger equation, transformation to spherical polar coordinates. Separation of variables. Spherical harmonics. Discussion of solution.

Qualitative treatment of hydrogen atom and hydrogen-like ions: setting up of Schrödinger equation in spherical polar coordinates, radial part, quantization of energy (only final energy expression). Average and most probable distances of electron from nucleus.

Setting up of Schrödinger equation for many-electron atoms (He, Li). Need for approximation methods. Statement of variation theorem and application to simple systems (particle-in-a-box, harmonic oscillator, hydrogen atom).

Chemical bonding: Covalent bonding, valence bond and molecular orbital approaches, LCAO-MO treatment of H_2 . Bonding and antibonding orbitals. Qualitative extension to H_2 . Comparison of LCAO-MO and VB treatments of H_2 (only wavefunctions, detailed solution not required) and their limitations. Refinements of the two approaches (Configuration Interaction for MO, ionic terms in VB). Qualitative description of LCAO-MO treatment of homonuclear and heteronuclear diatomic molecules (HF, LiH). Localised and non-localised molecular orbitals treatment of triatomic (BeH_2 , H_2O) molecules. Qualitative MO theory and its application to AH_2 type molecules. **(24 Lectures)**

Molecular Spectroscopy:

Interaction of electromagnetic radiation with molecules and various types of spectra; BornOppenheimer approximation.

Rotation spectroscopy: Selection rules, intensities of spectral lines, determination of bond lengths of diatomic and linear triatomic molecules, isotopic substitution.

Vibrational spectroscopy: Classical equation of vibration, computation of force constant, amplitude of diatomic molecular vibrations, anharmonicity, Morse potential, dissociation energies, fundamental frequencies, overtones, hot bands, degrees of freedom for polyatomic molecules, modes of vibration, concept of group frequencies. Vibration-rotation spectroscopy: diatomic vibrating rotator, P, Q, R branches.

Raman spectroscopy: Qualitative treatment of Rotational Raman effect; Effect of nuclear spin, Vibrational Raman spectra, Stokes and anti-Stokes lines; their intensity difference, rule of mutual exclusion.

Electronic spectroscopy: Franck-Condon principle, electronic transitions, singlet and triplet states, fluorescence and phosphorescence, dissociation and predissociation, calculation of electronic transitions of polyenes using free electron model.

Nuclear Magnetic Resonance (NMR) spectroscopy: Principles of NMR spectroscopy, Larmor precession, chemical shift and low resolution spectra, different scales, spin-spin coupling and high resolution spectra, interpretation of PMR spectra of organic molecules.

Electron Spin Resonance (ESR) spectroscopy: Its principle, hyperfine structure, ESR of simple radicals. **(24 Lectures)**

Photochemistry

Characteristics of electromagnetic radiation, Lambert-Beer's law and its limitations, physical significance of absorption coefficients. Laws, of photochemistry, quantum yield, actinometry, examples of low and high quantum yields, photochemical equilibrium and the differential rate of photochemical reactions, photosensitised reactions, quenching. Role of photochemical reactions in biochemical processes, photostationary states, chemiluminescence.

(12

Lectures)

Reference:

- Banwell, C. N. & McCash, E. M. Fundamentals of Molecular Spectroscopy 4th Ed. Tata McGraw-Hill: New Delhi (2006).
- Chandra, A. K. Introductory Quantum Chemistry Tata McGraw-Hill (2001).
- House, J. E. Fundamentals of Quantum Chemistry 2nd Ed. Elsevier: USA (2004).
- Lowe, J. P. & Peterson, K. Quantum Chemistry, Academic Press (2005).
- Kakkar, R. Atomic & Molecular Spectroscopy: Concepts & Applications, Cambridge University Press (2015).

DMF24401/ DMF24405/ DMF24408

VI SEMESTER

SEC1A

FUEL CHEMISTRY

SKILL ENHANCEMENT COURSE-SEC

30 Lectures

Course outcome:

After completion of the course a student is able to:

CO1: Understand soil sample for calcium and magnesium content.

CO2: Understand water parameters.

CO3: Identify food adulterants.

CO4: Understand chromatography.

Review of energy sources (renewable and non-renewable). Classification of fuels and their calorific value

Coal: Uses of coal (fuel and nonfuel) in various industries, its composition, carbonization of coal. Coal gas, producer gas and water gas—composition and uses. Fractionation of coal tar, uses of coal tar based chemicals, requisites of a good metallurgical coke, Coal gasification (Hydro gasification and Catalytic gasification), Coal liquefaction and Solvent Refining.

Petroleum and Petrochemical Industry: Composition of crude petroleum, Refining and different types of petroleum products and their applications.

Fractional Distillation (Principle and process), Cracking (Thermal and catalytic cracking), Reforming Petroleum and non-petroleum fuels (LPG, CNG, LNG, bio-gas, fuels derived from biomass), fuel from waste, synthetic fuels (gaseous and liquids), clean fuels Petrochemicals: Vinyl acetate, Propylene oxide, Isoprene, Butadiene, Toluene and its derivatives Xylene.

Lubricants: Classification of lubricants, lubricating oils (conducting and non-conducting) Solid and semisolid lubricants, synthetic lubricants.

Properties of lubricants (viscosity index, cloud point, pour point) and their determination.

Reference :

- Stocchi, E. Industrial Chemistry, Vol-I, Ellis Horwood Ltd. UK (1990).
- Jain, P.C. & Jain, M. Engineering Chemistry Dhanpat Rai & Sons, Delhi.

- Sharma, B.K. & Gaur, H. Industrial Chemistry, Goel Publishing House, Meerut (1996).

DMF24501/ DMF24505/ DMF24508

SEC1B

BASIC ANALYTICAL CHEMISTRY

30 Lectures

Course outcome:

After completion of the course a student is able to:

- CO1: Analyze food products
- CO2: Analyze soil sample
- CO3: Determine the contaminants in water
- CO4: Study about chromatography

Introduction: Introduction to Analytical Chemistry and its interdisciplinary nature. Concept of sampling. Importance of accuracy, precision and sources of error in analytical measurements. Presentation of experimental data and results, from the point of view of significant figures.

Analysis of soil: Composition of soil, Concept of pH and pH measurement, Complexometric titrations, Chelation, Chelating agents, use of indicators

- a. Determination of pH of soil samples.
- b. Estimation of Calcium and Magnesium ions as Calcium carbonate by complexometric titration.

Analysis of water: Definition of pure water, sources responsible for contaminating water, water sampling methods, water purification methods.

- a. Determination of pH, acidity and alkalinity of a water sample.
- b. Determination of dissolved oxygen (DO) of a water sample.

Analysis of food products: Nutritional value of foods, idea about food processing and food preservations and adulteration.

- a. Identification of adulterants in some common food items like coffee powder, asafoetida, chilli powder, turmeric powder, coriander powder and pulses, etc.
- b. Analysis of preservatives and colouring matter.

Chromatography: Definition, general introduction on principles of chromatography, paper chromatography, TLC etc.

- a. Paper chromatographic separation of mixture of metal ion (Fe^{3+} and Al^{3+}).

- b. To compare paint samples by TLC method. **Ion-exchange:** Column, ion-exchange chromatography etc.
Determination of ion exchange capacity of anion / cation exchange resin (using batch procedure if use of column is not feasible).

Analysis of cosmetics: Major and minor constituents and their function

- a. Analysis of deodorants and antiperspirants, Al, Zn, boric acid, chloride, sulphate.
b. Determination of constituents of talcum powder: Magnesium oxide, Calcium oxide, Zinc oxide and Calcium carbonate by complexometric titration.

Suggested Applications (Any one):

- a. To study the use of phenolphthalein in trap cases.
b. To analyze arson accelerants.
c. To carry out analysis of gasoline.

Suggested Instrumental demonstrations:

- a. Estimation of macro nutrients: Potassium, Calcium, Magnesium in soil samples by flame photometry.

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- b. Spectrophotometric determination of Iron in Vitamin / Dietary Tablets.
c. Spectrophotometric Identification and Determination of Caffeine and Benzoic Acid in Soft Drink.

Reference:

- Willard, H.H., Merritt, L.L., Dean, J. & Settoe, F.A. *Instrumental Methods of Analysis*. 7th Ed. Wadsworth Publishing Co. Ltd., Belmont, California, USA, 1988.
- Skoog, D.A. Holler F.J. & Nieman, T.A. *Principles of Instrumental Analysis*, Cengage Learning India Ed.
- Skoog, D.A.; West, D.M. & Holler, F.J. *Fundamentals of Analytical Chemistry 6th Ed.*, Saunders College Publishing, Fort Worth (1992).
- Harris, D. C. *Quantitative Chemical Analysis*, W. H. Freeman.
- Dean, J. A. *Analytical Chemistry Notebook*, McGraw Hill.
- Day, R. A. & Underwood, A. L. *Quantitative Analysis*, Prentice Hall of India.
- Freifelder, D. *Physical Biochemistry 2nd Ed.*, W.H. Freeman and Co., N.Y. USA (1982).
- Cooper, T.G. *The Tools of Biochemistry*, John Wiley and Sons, N.Y. USA. 16 (1977).
- Vogel, A. I. *Vogel's Qualitative Inorganic Analysis 7th Ed.*, Prentice Hall.
- Vogel, A. I. *Vogel's Quantitative Chemical Analysis 6th Ed.*, Prentice Hall.
- Robinson, J.W. *Undergraduate Instrumental Analysis 5th Ed.*, Marcel Dekker, Inc., New York (1995)

Chemistry Syllabus for practicals for B.Sc Course

Note: Students should be trained to use both chemical and electronic balances (three digit) DMA24101/ DMA24105/ DMA24108

I Semester : Practical 1

3 hours per week

1. Calibration of : (i) Pipette (ii) Burette (iii) Volumetric flask
2. Preparation of 2N solutions of H_2SO_4 , HCl, HNO_3 , CH_3COOH and NH_3
3. Preparation of standard sodium carbonate solution and standardization of hydrochloric acid solution (methyl orange indicator). Estimation of sodium hydroxide present in the solution using phenolphthalein indicator.
4. Preparation of standard oxalic acid solution and standardization of sodium hydroxide solution. Estimation of sulphuric acid present in the solution.
5. Preparation of standard potassium biphthalate solution and standardization of sodium hydroxide solution. Estimation of oxalic acid present in the solution.
6. Preparation of standard oxalic acid solution and standardization of potassium permanganate solution. Estimation of ferrous ammonium sulphate present in the solution.
7. Preparation of standard oxalic acid solution and standardization of potassium permanganate solution. Estimation of hydrogen peroxide present in the solution.
8. Estimation of sulphuric acid and oxalic acid in a mixture using standard sodium hydroxide solution and standard potassium permanganate solution.
9. Determination of the percentage of available chlorine in the given sample of bleaching powder.
10. Estimation of ferrous and ferric iron in a given mixture using standard potassium dichromate solution.
11. Preparation of standard Zinc Sulphate solution and standardization of EDTA. Estimation of total hardness of water.
12. Estimation of ammonium chloride using standard sodium hydroxide and standard hydrochloric acid solutions (back titration)

II Semester : Practical 2

3 hours per week

DMB24101/ DMB24105/ DMB24108

Part 1 : Qualitative analysis of mono functional organic compounds through functional group analysis. Determination of physical constant. Preparation of suitable derivative of the following class.

1. Acids
2. Alcohols
3. Aldehydes
4. Amides
5. Amines
6. Halogenated hydrocarbons
7. Hydrocarbons
8. Ketones
9. Nitro compounds
10. Phenols

III Semester : Practical 3

3 hours per week

DMC24101/ DMC24105/ DMC24108

Systematic semi micro qualitative analysis of a mixture of two simple salts (with no interfering radicals).

The constituent ions in the mixture to be restricted to the following

Anions : HCO_3^- , CO_3^{2-} , SO_3^- , Cl^- , Br^- , NO_3^- , BO_3^{3-} , SO_4^{2-} , PO_4^{3-}

Cations : Pb^{2+} , Bi^{3+} , Cd^{2+} , Al^{3+} , Fe^{3+} , Fe^{2+} , Mn^{2+} , Zn^{2+} , Ba^{2+} , Sr^{2+} , Ca^{2+} , Mg^{2+} , K^+ , Na^+ , NH_4^+

Note :

1. Mixture requiring elimination of phosphate and borate should not be given.
2. Combination of anions in second group shall be avoided.
3. Salts that yields double decomposition shall be avoided. (BaSO_4).
4. The combination of two cations in the mixture should belong to different groups.
However combinations like Mg^{2+} and NH_4^+ and Na^+ and NH_4^+ can be given.

IV Semester : Practical 4

3 hours per week

DMD24101/ DMD24105/ DMD24108

Part 1 :

1. Determination of the density using specific gravity bottle and viscosity of a liquid using Ostwald's viscometer
2. Determination of the density using specific gravity bottle and surface tension of a liquid using stalagmometer.
3. Determination of molecular mass of a non-volatile solute by walker-Lumsden method.
4. Determination of rate constant of the decomposition of hydrogen peroxide catalysed by FeCl_3 .
5. Determination of transition temperature of the salt hydrates .
6. Determination of rate constant of saponification of ethylacetate titrimetrically.
7. Determination of percentage composition of sodium chloride solution by determining the miscibility temperature of phenol-water system.
8. Determination of the mass present in the given solution of a strong acid using strong base by thermometric titration method
9. Determination of molecular weight of a polymer material by viscosity measurements (cellulose acetate/ Methyl acrylate)
10. Study of kinetics of reaction between $\text{K}_2\text{S}_2\text{O}_8$ and KI, second order, determination of rate constant.

V Semester: Practical 5A

3 hours per week

DME24101/ DME24105/ DME24108

Part 1: Gravimetric estimation

1. Gravimetric estimation of Barium as Barium sulphate.
2. Gravimetric estimation of Iron as Iron(III) oxide
3. Gravimetric estimation of copper as copper thiocyanate
4. Gravimetric estimation of Nickel as nickel dimethyl glyoximate
5. Gravimetric estimation of magnesium as magnesium hydroxyl quinolate

Part 2 : Volumetric estimations

1. Estimation of iron in the given sample of haematite by dichromate method.
2. Estimation of percentage of calcium in limestone by oxalate method
3. Estimation of manganese in the given sample of pyrolusite.
4. Estimation of magnesium in the given sample of dolomite by EDTA method.

- Determination of % purity of copper in the given sample of copper wire.

V Semester: Practical 5B

3 hours per week

DME24301/ DME24305/ DME24308

- Determination of dissolved oxygen in water.
- Determination of Chemical Oxygen Demand (COD)
- Determination of Biological Oxygen Demand (BOD)
- Percentage of available chlorine in bleaching powder.
- Measurement of chloride, sulphate and salinity of water samples by simple titration method (AgNO_3 and potassium chromate).
- Estimation of total alkalinity of water samples (CO_3^{2-} , HCO_3^-) using double titration method.
- Measurement of dissolved CO_2 .
- Study of some of the common bio-indicators of pollution.
- Estimation of SPM in air samples.
- Preparation of borax/ boric acid.

VI Semester : Practical 6 A

3 hours per week

DMF24101/ DMF24105/ DMF24108

Part 1: Physical chemistry experiments (instrumental)

- Determination of Equivalent conductance of the given electrolyte(both strong and weak electrolyte)
- Determination of percentage composition of benzene and carbon tetra chloride by using Abbe's refractrometer.
- Determination of concentration of an acid/ base by conductometric method.
- Potentiometric titration of Ferrous ammonium sulphate and $\text{K}_2\text{Cr}_2\text{O}_7$.
- Determination of PK_a of weak acid by potentiometric method
- PH titration of strong acid and strong base.
- Calorimetric estimation of Fe^{3+} ion using Ammonium thiocyanate
- Calorimetric estimation of Cu^{2+} ion using NH_4OH
- Calorimetric estimation of Aspirin using FeCl_3

Part 2 : Isolation of organic compounds from natural products

- Isolation of hesperidins from orange peels
- Isolation of caffeine from Tea leaves
- Isolation of Nicotine from Tobacco leaves.

DMF24301/ DMF24305/ DMF24308**DSE LAB**

UV/Visible spectroscopy

1. Study the 200-500 nm absorbance spectra of KMnO_4 and $\text{K}_2\text{Cr}_2\text{O}_7$ (in 0.1 M H_2SO_4) and determine the λ_{max} values. Calculate the energies of the two transitions in different units (J molecule^{-1} , kJ mol^{-1} , cm^{-1} , eV).
2. Study the pH-dependence of the UV-Vis spectrum (200-500 nm) of $\text{K}_2\text{Cr}_2\text{O}_7$.
3. Record the 200-350 nm UV spectra of the given compounds (acetone, acetaldehyde, 2-propanol, acetic acid) in water. Comment on the effect of structure on the UV spectra of organic compounds.
 - i. Colourimetry
4. Verify Lambert-Beer's law and determine the concentration of $\text{CuSO}_4/\text{KMnO}_4/\text{K}_2\text{Cr}_2\text{O}_7$ in a solution of unknown concentration
5. Determine the concentrations of KMnO_4 and $\text{K}_2\text{Cr}_2\text{O}_7$ in a mixture.
6. Study the kinetics of iodination of propanone in acidic medium
7. Determine the amount of iron present in a sample using 1,10-phenanthroline.
8. Determine the dissociation constant of an indicator (phenolphthalein).
9. Study the kinetics of interaction of crystal violet/ phenolphthalein with sodium hydroxide.
10. Analyse the given vibration-rotation spectrum of $\text{HCl}(\text{g})$

Reference:

- A.I. Vogel: Qualitative Inorganic Analysis, Prentice Hall, 7th Edn.
- A.I. Vogel: Quantitative Chemical Analysis, Prentice Hall, 6th Edn.
- Vogel, A.I., Tatchell, A.R., Furnis, B.S., Hannaford, A.J. & Smith, P.W.G., *Textbook of Practical Organic Chemistry*, Prentice-Hall, 5th edition, 1996.

Mann, F.G. & Saunders, B.C. *Practical Organic Chemistry* Orient-Longman, 1960

**Pattern of question paper for DSC/DSE
I to VI semesters**

Time : 3.00hrs

Max marks: 70

PART-A

I. Answer the following questions

1x10=10

- a)
- b)
- c)
- d)
- e)
- f)
- g)
- h)
- i)
- j)

PART-B

II Answer any three questions

(Questions carrying 4,4,2- 3,3,4- 6,4 and 5,5 marks may be given)

3x10=30

- 2)
- 3)
- 4)
- 5)

PART-C

III Answer any three questions

(Questions carrying 4,4,2- 3,3,4- 6,4 and 5,5 marks may be given)

3x10=30

- 6)
- 7)
- 8)
- 9)

Pattern of question paper for SEC

Time : 2 hrs

Max marks: 35

PART-A

I. Answer the following questions

1x5=5

- a)
- b)
- c)
- d)
- e)

PART-B

II Answer any three questions

(Questions carrying 4,4,2- 3,3,4- 6,4 and 5,5 marks may be given)

3x10=30

- 2)
- 3)
- 4)
- 5)

Panel of Examiners

1. Smt J S Vidya
Chairperson, JSS College,
Ooty Road, Mysore
2. Dr. B K Kendagannaswamy
JSS College, Ooty Road,
Mysore
3. D S Prabhakar
JSS College, Ooty Road,
Mysore
4. Shwetha P
JSS College, Ooty Road,
Mysore
5. Shambhavi S
JSS College, Ooty Road,
Mysore

EXTERNAL MEMBERS

1. Dr. GirijaNagendraswamy . Maharani's Science College for Women, Mysuru
2. Dr. Jayarooma .Maharani.s Science College for Women, Mysuru
3. DrKempegowda,. Maharani's Science College for Women, Mysuru
4. Dr. Jamuna rani. Maharani's Science College for Women Mysuru
5. Prof B Manjunath . Maharani's Science College for Women ,Mysuru
6. Dr Lakshmi Hebbar,.Maharani's Science College for Women ,Mysuru
7. DrMadhusudana Reddy , Maharani's Science College for Women, Mysuru
8. Dr. K KPadmanabha,.Maharani's Science College for Women,. Mysuru
9. Prof. Tara, Maharani's Science College for Women, Mysuru
10. Dr. Chennu, Maharani's Science College for Women ,Mysuru
11. Smt. Kavitha ,Maharani's Science College for Women ,Mysuru
12. Smt. Radhika ,Maharani's Science College for Women, Mysuru
13. Dr. Ramesh baba Maharani's Science College for Women, Mysuru
14. Dr. T S Yamuna ,Maharani's Science College for Women,Mysuru

15. Smt. Ayesha, Kuvempu Nagar first Grade College., Mysuru
16. Dr. Mousinabegam Maharani's Science College for Women Mysuru
17. Dr. K Ajay Kumar, Yuvaraja's College, Mysuru
18. Dr. Chandrashekar, Yuvaraja's College, Mysuru
19. Dr. DevaRaj, Yuvaraja's College, Mysuru
20. Dr. Sudha B S ,Yuvaraja college ,Mysuru
21. Dr. Shaukatharakhanum ,Yuvaraja'sCollege,Mysuru
22. Dr. Ahamabidi, Yuvaraja's College ,Mysuru
23. Jayadevappa ,Yuvaraja'sCollege,Mysuru
24. Dr. Ashaiyangar, Yuvaraja's College ,Mysuru
25. Dr. Govindaraju M ,Sharadavilas College, Mysouru
26. Prof. Prakash K, Govt. College for boys(Autonomous) ,Mandya
27. Dr. ShivaprakashVijayanagar first Grade college Mysuru
28. Dr. AlphonsusDsouza, St. PhilominasCollege,Mysuru
29. Smt. Agnes Dsouza St. Philominas College Mysuru
30. Prof. Britto Dominik,rayan,,Sri D DUrs College, Hunsur
31. Smt. Pushpa ,JSS CW, ,Sarswathipuram, Mysuru
32. SmtLokeswari ,JSS CW Sarswathipuram, Mysuru
33. Rajeshwari P R, JSS CW Sarswathipuram, Mysuru
34. SmtRakshitha, JSS college ,Nanjangud
35. Sri Siddaraju, JSS College, Chamarajanagar
36. Dr. Laval Devi, Govt first grade College, K R Nagar.
37. Archana P Mendora, St. Philominas College,Mysore
38. Bindu Noronha, St. Philominas College,Mysore
39. Sudeep P, YCM, Mysore
40. Vageesh, YCM, Mysore
41. Dileep P, YCM, Mysore
42. Prakash , YCM, Mysore
43. Pallavi H M, YCM, Mysore
44. Sumana, YCM, Mysore
45. Dr. Nandeesh, YCM, Mysore
46. Guruprasad, YCM, Mysore
47. Jagadeesh, YCM, Mysore
48. Mahima, YCM, Mysore
49. Rekha G R, YCM, Mysore
50. Kemparaje Gowda, YCM, Mysore
51. Gangadhar, YCM, Mysore
52. Vivek, YCM, Mysore
53. Sindhushree, YCM, Mysore
54. Mamatha M, MSCW, Mysore
55. Noor Fatima, MSCW, Mysore
56. Bindushree, MSCW, Mysore
57. Mamatha , JSS College, Chamarajnagar
58. Sowmya, JSS College, Chamarajnagar

JSS COLLEGE OF ARTS, COMMERCE & SCIENCE

(AUTONOMOUS)

B.N. ROAD, MYSURU



DEPARTMENT OF ECONOMICS

**Revised Syllabus for Undergraduate (UG)
CBCS Scheme - 2019-2020**

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

(AUTONOMOUS)

B.N. ROAD, MYSURU-25**DEPARTMENT OF ECONOMICS****Course BA: Syllabus****Restructured and Revised Syllabus and Scheme for BA Economics under
CBCS - 2019-2020**

Sem	Course Code	Course	Teaching Hours / week	L T P	Credit	IA Marks	Theory Exam	Total Marks
Discipline Specific Core(DSC)								
I	Core-I ELA21021/22/ /26/23/24	Principles of Micro Economics-I	5+1	5-1-0	6	30	70	100
II	Core - II ELB21021/22/ 26/23/24	Principles of Micro Economics-II	5+1	5-1-0	6	30	70	100
III	Core - III ELC21021/22/ 26/23/24	Principles of Macro Economics-I	5+1	5-1-0	6	30	70	100
IV	Core - IV ELD21021/22 /26/23/24	Principles of Macro Economics-II	5+1	5-1-0	6	30	70	100
Discipline Specific Elective (DSE) – Any one								
V	DSE – 1A ELE21021/22/ 26/23/24	Economics of Development	5+1	5-1-0	6	30	70	100
V	DSE – 1B ELE21221/22/ 26/23/24	Money and Banking	5+1	5-1-0	6	30	70	100
V	DSE – 1C ELE21421/22/ 26/23/24	Environmental Economics	5+1	5-1-0	6	30	70	100
Discipline Specific Elective (DSE) – Any one								
VI	DSE - 1A ELF21021/22/ 26/23/24	Indian Economy	5+1	5-1-0	6	30	70	100
VI	DSE - 1B ELF21221/22/ 26/23/24	Economic History of India-1857-1947	5+1	5-1-0	6	30	70	100
VI	DSE – 1C ELF21421/22/ 26/23/24	Public Finance	5+1	5-1-0	6	30	70	100

Note:

- Credits Per Course/Paper per week in all the SIX Semesters =6 Credits [L:5 + T:1]**
[Of which, it is 5 Credits for Lecture and 1 Credit for Tutorials]
- Work Load Per Course/Paper per week in all the SIX Semesters = 07 Hours[L:5 + T:2]**
[Of which, it is 5 Hours for Lecture and 2 Hours for Tutorials]

Generic Elective

Sem	Paper Code	Course	Teaching Hours / week	L T P	Credit	I.A Marks	Theory Exam	Total Marks	Remarks
V	GE-I ELE21621 /22/26/23/ 24	Introduction to Economics	2	2-0-0	2	15	35	50	To benefit those students interested to study <i>Basic Economics</i> and <i>Indian Economy</i> irrespective of his / her programme
VI	GE-II ELF21622 /22/26/23/ 24	Indian Economy	2	2-0-0	2	15	35	50	

Scheme of assessment for B.A Economics under CBCS-2019-2020

Year	SEM	COURSE CODE	TITLE OF THE PAPER	CONTINUOUS ASSESSMENT				MAX.MARKS			DURATI ON OF EXAM	
				C1		C2		C3	TH	IA		TH
				TEST	AS/ GD	TEST	SE/ PD					
I	I	Core-1 ELA21021/22/26/ 23/24	Principles of Micro Economics-I	10	5	10	5	70	70	30	3h	
	II	Core - II ELB21021/22/26/ 23/24	Principles of Micro Economics-II	10	5	10	5	70	70	30	3h	
II	III	Core - III ELC21021/22/26/ 23/24	Principles of Macro Economics-I	10	5	10	5	70	70	30	3h	
	IV	Core - IV ELD21021/22/26/ 23/24	Principles of Macro Economics-II	10	5	10	5	70	70	30	3h	
III	V	DSE - 1A ELE21021/22/ 26/23/24	Economics of Development	10	5	10	5	70	70	30	3h	
		DSE - 1B ELE21221/22/ 26/23/24	Money and Banking	10	5	10	5	70	70	30	3h	
		DSE – 1C ELE21421/22/ 26/23/24	Environmental Economics	10	5	10	5	70	70	30	3h	
		GE-I ELE21621/22/ 26/23/24	Introduction to Economics	7.5	-	7.5	-	35	35	15	2h	
	VI	DSE - 1A ELF21021/22/ 26/23/24	Indian Economy	10	5	10	5	70	70	30	3h	
		DSE - 1B ELF21221/22/ 26/23/24	Economic History of India- 1857-1947	10	5	10	5	70	70	30	3h	
		DSE – 1C ELF21421/22/ 26/23/24	Public Finance	10	5	10	5	70	70	30	3h	
		GE-II ELF21622/22/ 26/23/24	Indian Economy	7.5	-	7.5	-	35	35	15	2h	

Core and Discipline Specific Elective(DSE)
Courses in Economics

Semester I	Semester II
Core Economics I: Principles of Microeconomics-I	Core Economics II: Principles of Microeconomics-II
Semester III	Semester IV
Core Economics III: Principles of Macroeconomics-I	Core Economics IV: Principles of Macroeconomics-II
Semester V	Semester VI
Discipline Specific Elective I One of the following: i. DSE-1A : Economics of Development ii. DSE-1B : Money and Banking iii. DSE-1C :Environmental Economics	Discipline Specific Elective II One of the following: iv. DSE-1A: Indian Economy v. DSE -1B: Economic History of India 1857-1947 vi. DSE- 1C: Public Finance

Generic Electives

Semester	Course Title	Remarks
V	GE-I Introduction to Economics	Benefits those students interested to study <i>Basic Economics</i> and <i>Indian Economy</i> irrespective of his / her programme
VI	GE-II Indian Economy	

TEMPLATE

Subject: ECONOMICS

Sem	Core			DSE			GE		
	No. of Courses	Credits	Total Hours	No. of Courses	Credits	Total Hours	No. of Courses	Credits	Total Hours
I	1	6	75+15	-	-	-	-	-	-
II	1	6	75+15	-	-	-	-	-	-
III	1	6	75+15	-	-	-	-	-	-
IV	1	6	75+15	-	-	-	-	-	-
V	-	-	-	1	6	75+15	1	2	30
VI	-	-	-	1	6	75+15	1	2	30
Total	4	24	360	2	12	180	2	4	60

Programme Outcome

After completing the graduation in B A(HEG) the students are able to:

- PO1. Critically recognizes the social, political, economic and cultural aspects of History.
- PO2. Demonstrate thinking skills by analyzing, synthesizing, and evaluating them in relation to their cultural and historical context.
- PO3. Correctly extracts evidence from primary sources by analyzing and evaluating them in relation to their cultural and historical context.
- PO4. Develop an informed familiarity with multiple cultures.
- PO5. Demonstrate critical reading, writing and thinking skills.
- PO6. Explain, graph, and analyze key economics models.
- PO7. Understand current events and evaluate specific policy proposals.
- PO8. To address problem that do not have clear economic solutions.
- PO9. Develop critical and quantitative thinking skills.
- PO10. Communicate effectively in written, oral and graphical form about specific issues.
- PO11. Apply economic analysis to everyday problems in real world situations.
- PO12. Understand and appreciate relationship between man and Environment
- PO13. Read, interpret, and generate maps and other geographic representations
- PO14. To extract, analyze, and present information from a spatial perspective
- PO15. Understand physical-geographic processes, global distribution of landforms and ecosystems
- PO16. The role of physical environment on human population
- PO17. Develop the ethical aptitudes and dispositions necessary to acquire and hold leadership positions in industry, government, and professional organizations.

Programme Specific Outcomes

On Completion of BA (HEG) students will:

- PSO1. This Programme exerts its influence on life and destiny of Human beings.
- PSO2. It is a stepping stone for one's success in competitive examinations.
- PSO3. Understand the background of our religion, customs and Institutions and so on.
- PSO4. Understand the present Social, political, religious and economic conditions of the people.
- PSO5. Understand theoretical and practical aspects of Economics and Geography

- PSO6. Evaluate Economic behavior inconsonance with Geographical factors
- PSO7. Suggest the policy makers about desirable changes to be made in Micro and Macro Economic issues based on geographical factors
- PSO8. Gain ability to understand the economic problems in Geographical indicators
- PSO9. Able to offer palatable solutions for economic and geographical challenges
- PSO10. Attain Proficiency to analyze the economic decision of Government and non-Govt. entities that correlate with Geographical factors
- PSO11. Gain requisite knowledge to evaluate land use pattern and demographical profile
- PSO12. Apply GIS for understanding Market situation, Transport problem change in Weather Condition, Cropping Pattern, and Natural Calamities and so on

Programme Outcome

After completing the graduation in B A(HEP) the students are able to:

- PO1. Critically recognizes the social, political, economic and cultural aspects of History.
- PO2. Demonstrate thinking skills by analyzing, synthesizing, and evaluating them in relation to their cultural and historical context.
- PO3. Correctly extracts evidence from primary sources by analyzing and evaluating them in relation to their cultural and historical context.
- PO4. Develop an informed familiarity with multiple cultures.
- PO5. Demonstrate critical reading, writing and thinking skills.
- PO6. Explain, graph, and analyze key economics models.
- PO7. Understand current events and evaluate specific policy proposals.
- PO8. To address problem that do not have clear economic solutions.
- PO9. Develop critical and quantitative thinking skills.
- PO10. Apply economic analysis to everyday problems in real world situations.
- PO11. To extract, analyze, and present information from a spatial perspective.
- PO12. Spread the messages of equality, nationality, social harmony and other human values.
- PO13. Comprehend the basic structures and processes of government systems and/or theoretical underpinnings.
- PO14. Analyze political problems, arguments, information, and/or theories.
- PO15. Apply methods appropriate for accumulating and interpreting data applicable to the discipline of Political Science.

Programme Specific Outcomes

On Completion of BA (HEP) students will:

PSO1. Critically recognize the social, political, economic and cultural aspects of History.

PSO2. Demonstrate thinking skills by analyzing, synthesizing, and evaluating historical information from multiple sources.

PSO3. Correctly extract evidence from primary sources by analyzing and evaluating them in relation to their cultural and historical context.

PSO4. Develop an informed familiarity with multiple cultures.

PSO5. Demonstrate critical reading, writing, and thinking skills.

PSO6. Explain, graph, and analyze key economics models.

PSO7. Understand current events and evaluate specific policy proposals.

PSO8. To address problem that do not have clear economic solutions.

PSO9. Develop critical and quantitative thinking skills.

PSO10. Communicate effectively in written, oral and graphical form about specific issues.

PSO11. Apply economic analysis to everyday problems in real world situations.

PSO12. Develop the ethical aptitudes and dispositions necessary to acquire and hold leadership positions in industry, government, and professional organizations.

Programme Outcome

After completing the graduation in B A(KEP) the students are able to:

PO1. Develop human values and a sense of social service

PO2. Become a responsible and dutiful citizen.

PO3. Able to enhance critical temper and creative ability

PO4. Apply economic analysis to everyday problems in real world situations.

PO5. Explain, graph, and analyze key economics models.

PO6. Understand current events and evaluate specific policy proposals.

PO7. To address problem that do not have clear economic solutions.

PO8. Develop critical and quantitative thinking skills.

PO9. Apply economic analysis to everyday problems in real world situations.

PO10. To extract, analyze, and present information from a spatial perspective.

PO11. Spread the messages of equality, nationality, social harmony and other human values.

PO12. Comprehend the basic structures and processes of government systems and/or theoretical underpinnings.

PO13. Analyze political problems, arguments, information, and/or theories.

PO14. Apply methods appropriate for accumulating and interpreting data applicable to the discipline of Political Science.

Programme Specific Outcomes

On Completion of BA (KEP) students will:

PSO 1: Know more specific terminologies along with its etymology.

PSO2: Know the changes in the differences in formation of society and their culture

PSO3: Help to understand different races, Society, and culture.

PSO4. Understand theoretical and practical aspects of Economics and Geography

PSO5. Evaluate Economic behavior inconsonance with Geographical factors

PSO6. Suggest the policy makers about desirable changes to be made in Micro and Macro Economic issues based on geographical factors

PSO7. Act as a stepping stone for one's success in competitive examinations

PSO8. Create appropriate and efficient Historians, Political Leaders, administrators and State's man

PSO9. Educate about patriotism toleration and National Integration.

Programme Outcome

After completing the graduation in B A(KEG) the students are able to:

PO1. Develop human values and a sense of social service

PO2. Become a responsible and dutiful citizen.

PO3. Able to enhance critical temper and creative ability

PO4. Apply economic analysis to everyday problems in real world situations.

PO5. Explain, graph, and analyze key economics models.

PO6. Understand current events and evaluate specific policy proposals.

PO7. To address problem that do not have clear economic solutions.

PO8. Develop critical and quantitative thinking skills.

PO9. Apply economic analysis to everyday problems in real world situations.

PO12. Understand and appreciate relationship between man and Environment

PO13. Read, interpret, and generate maps and other geographic representations

PO14. To extract, analyze, and present information from a spatial perspective

PO15. Understand physical-geographic processes, global distribution of landforms and ecosystems

PO16. The role of physical environment on human population.

Programme Specific Outcomes

On Completion of BA (KEG) students will:

PSO 1: Know more specific terminologies along with its etymology.

PSO2: Know the changes in the differences in formation of society and their culture

PSO3: Help to understand different races, Society, and culture.

PSO4. Understand theoretical and practical aspects of Economics and Geography.

PSO5. Evaluate Economic behavior inconsonance with Geographical factors.

PSO6. Suggest the policy makers about desirable changes to be made in Micro and Macro Economic issues based on geographical factors.

PSO7. Act as a stepping stone for one's success in competitive examinations.

PSO8. Gain ability to understand the economic problems in Geographical indicators.

PSO9. Able to offer palatable solutions for economic and geographical challenges.

PSO10. Attain Proficiency to analyze the economic decision of Government and non-Govt. entities that correlate with Geographical factors.

PSO11. Gain requisite knowledge to evaluate land use pattern and demographical profile.

PSO12. Apply GIS for understanding Market situation, Transport problem change in Weather Condition, Cropping Pattern, and Natural Calamities and so on.

Programme Outcome

After completing the graduation in B A(HEE) the students are able to:

PO1. Critically recognize the social, political, economic and cultural aspects of History.

PO2. Demonstrate thinking skills by analyzing, synthesizing, and evaluating historical information from multiple sources.

PO3. Correctly extract evidence from primary sources by analyzing and evaluating them in relation to their cultural and historical context.

PO4. Develop an informed familiarity with multiple cultures.

PO5. Demonstrate critical reading, writing, and thinking skills.

PO6. Explain, graph, and analyze key economics models

PO7. Understand current events and evaluate specific policy proposals

PO8. To address problem that do not have clear economic solutions

PO9. Develop critical and quantitative thinking skills

PO10. Communicate effectively in written, oral and graphical form about specific issues

PO11. Apply economic analysis to everyday problems in real world situations

PO12. Write focused, organized, well-developed, and text-based essays using effective paragraphs.

PO13. Support a clear thesis statement, and demonstrate competence in Standard English Language and usage.

PO14. Write articles, novels, stories to spread the messages of equality, nationality, social harmony and other human values.

PO15. Acquire the ability to engage in independent and life-long learning in a broader context about socio-technological and demographic changes.

PO16. Develop and carry out research projects, and locate, evaluate, organize, and incorporate information effectively.

Programme Specific Outcomes

On Completion of BA (HEE) students will:

PSO1. Demonstrate thinking skills by analyzing, synthesizing, and evaluating historical information from multiple sources.

PSO2. Correctly extract evidence from primary sources by analyzing and evaluating them in relation to their cultural and historical context.

PSO3. Develop an informed familiarity with multiple cultures.

PSO4. Demonstrate critical reading, writing, and thinking skills.

PSO5. Explain, graph, and analyze key economics models.

PSO6. Understand current events and evaluate specific policy proposals.

PSO7. To address problem that do not have clear economic solutions.

PSO8. Develop critical and quantitative thinking skills.

PSO9. Communicate effectively in written, oral and graphical form about specific issues.

PSO10. Apply economic analysis to everyday problems in real world situations

PSO11. Write and edit clear, graceful, grammatically correct prose.

PSO12. Analyse numerical data and utilize database for multi-layered storytelling.

PSO13. Demonstrate preparation for an entry-level position in the profession through a portfolio exhibiting their work.

I – Semester
DSC I: Principles of Microeconomics-I

Course outcomes

On completion of the Course, students will:

- CO1. Understand in details with examples Concepts of Micro and Macro Economics.
- CO2. Deliberate in depth Law of Demand.
- CO3. Understand in depth laws of utility.
- CO4. Learn in details with examples meaning and properties of indifference curve.
- CO5. Deliberate in depth cost and revenue concepts.
- CO6. Understand the details of meaning and types of markets.
- CO7. Identify in details with examples perfect competitive market.
- CO8. Specify the details of concepts of Marginal cost and Marginal revenue.

Course Description:

This course exposes the student to the basic principles and application of Microeconomic Theory.

I. Introduction:

12

Economics, Meaning, Scope and Need for the study. Concepts of Micro and Macro Economics: Scope-Importance and Limitations of Micro Economics. Scarcity and choice. Basic Problems of an Economy-Opportunity Cost-Production Possibility Curve.

II. Consumer Theory:

14

- a) Marshallian Utility Theory – Concepts of Utility-Law of Diminishing Marginal Utility-Diamond Water Paradox-Theory of Consumer Surplus-Law of Equi-Marginal Utility.
- b) Indifference Curve Analysis-Meaning, Assumptions, Schedule & Constructions of Indifference Curve, Indifference Map & Properties. Budget Constraint-Consumer Equilibrium-Price, Income and substitution Effects.

III. Demand and Supply:

24

- a) Demand-Meaning, Law of Demand-Individual and Market Demand-Exceptions to Law of Demand-Determinants of Demand-Shifts in Demand versus movement along with a Demand Curve.
- b) Elasticity-Price Elasticity of Demand-Types of Price Elasticity of Demand-Determinants of Price Elasticity –Concepts of Income Elasticity and Cross Elasticity. Calculating Elasticity
- c) Supply-Meaning, Law of Supply- Determinants of Supply-equilibrium of Demand and Supply.
- d) Application of Demand and Supply-Price rationing-Price Floors-Consumer surplus and

Producer surplus.

IV. Production Analysis:

15

- a) Production-Meaning- Production Function-TP-AP-MP-Law of Variable Proportions-Economies and Diseconomies of Scale.
- b) Cost and Revenue: Concepts of Cost-short run and Long run cost curves. Concepts of Revenue- short run and Long run revenue curves. Isoquants-IsoCost-Meaning-Properties-Mathematical Calculation of TR,AR and MR.
- c) Equilibrium of Firm and Industry- Short run and Long run.

V. Market Structure and Perfect Competition:

10

- a) Market –Meaning and classification. Role of Time Elements in Market.
- b) Perfect Competition-Features-Price and output determination in short run and Long run under perfect Competition.

Reference:

Case, Karl E and Ray C. Fair- Principles of Economics, Pearson Education, Inc..8th edition, 2007.

Samuelson P.A-Economics(18th Edition, McGraw hill)

Mukarjee Sampath- Modern Economic Theory(New Age International)

H L Ahuja-Modern Economic Theory(S.Chand & Company)

Mithani. D.M -Modern Economic Analysis(Himalayan Publication publication)

McConnel Compbell & Stanley Brue-Micro Economics(16th Edition, McGraw hill)

Zen-Micro Economic Analysis(McGraw Hill)

KPM Sundaram and M.C.Vaish -Principles of Economics(Himalayan Publication publication)

S.Sankaran-Principles of Economics(Himalayan Publication publication)

M.L.Jhingan-Micro Economic Theory

K.K.Dewett – Modern Economic Theory

II – Semester

DSC II: -Principles of Microeconomics-II

Course outcomes

On completion of the Course, students will:

- CO1. Learn in depth types of Imperfect Competition.
- CO2. Deliberate the characteristics of Price Discrimination.
- CO3. Identify the classification and characteristics of Kinked Demand Curve.
- CO4. Identify in details with examples Merits and Public Goods.
- CO5. Write down the classification and characteristics of General Equilibrium of Exchange and Production.
- CO6. Understand in details with examples General Equilibrium of Exchange and Production.
- CO7. Write down in details with examples Marginal Productivity Theory.

Course Description:

This is a sequel to Principles of Microeconomics-I that highlights market functioning Structure.

- I. Imperfect Competition and Monopoly: 15**
Concept and Types of Imperfect Competition-Monopoly-Monopolistic-Oligopoly-Duopoly. Monopoly-meaning-Types and features of Monopoly-price and output determination in short run and Long run. Concept of Dumping and Price Discrimination-Meaning-Types-Remedies for Monopoly. Comparison of Perfect Competition and Monopoly.
- II. Monopolistic and Oligopoly Competition: 16**
 - a) Monopolistic competition- Meaning-Features-price and output determination in short run and Long run- Product Differentiation.
 - b) Oligopoly-Meaning-Classification-Features-Price and output determination in Oligopoly- Concepts of Kinked Demand Curve-Price Leadership and Cartels. Concept of Duopoly.
- III. General Equilibrium and Externalities: 17**
 - a) General Equilibrium of Exchange and Consumption-General Equilibrium of Exchange and Production.
 - b) Externalities-Meaning-Types-Marginal Cost Pricing-Public Goods-Private Goods-Merit Goods-Market Failure.
- IV Income Distribution and Factor Pricing: 13**
Inputs-Demand for inputs-Marginal Productivity Theory-Modern Theories of Distribution-Ricardian Theory of Rent-Modern Theory of Wages-Liquidity Theory of Interest-Innovation Theory of Profit.

- a) Trade-Meaning-Internal and International Trade-Theories of International Trade-Absolute Advantage and Comparative Cost Advantage.
- b) Free Trade and Protectionist Policy of Trade. Meaning, Relative Merits and Demerits.
- c) Concepts of Terms of Trade-Trade Barriers-Tariffs and Quotas.

Reference:

Case, Karl E and Ray C. Fair Principles of Economics, Pearson Education, Inc. 8th edition, 2007.

Samuelson P.A-Economics(18th Edition McGraw hill)

Mukarjee Sampath- Modern Economic Theory(New Age International)

H L Ahuja-Modern Economic Theory(S. Chand & Company)

Mithani. D.M -Modern Economic Analysis(Himalayan publication)

McConnel Compbell & Stanley Brue-Micro Economics(16th Edition, McGraw hill)

K. K. Dwett -Modern Economic Theory(S.Chand & Company)

S.Sankaran-Principles of Economics(Himalayan publication)

M.L.Jhingan-Micro Economic Theory

M.L.Seth-Microeconomics(Lakshminarayan Agarwal)

G.Mankiw-Microeconomics(New Age International)

Mithani.D.M - International Economics(Himalaya Publishing House, Mumbai)

Jhingan.M.L - International Economics(Virnda Publication)

Sharma.A.K - International Economics(Anmol Publication)

III – Semester
DSC III - Principles of Macroeconomics-I

Course outcomes

On completion of the Course, students will:

- CO1. Identify in details with examples Key variables of Macro Economics.
- CO2. Understand in details with examples Concepts of National Income.
- CO3. Identify in depth Marginal Efficiency of Capital.
- CO4. Specify the details of Concept of Multiplier.
- CO5. Identify the characteristics of Keynesian Macro Economics.
- CO6. Deliberate in depth Liquidity Theory of money.
- CO7. Identify the characteristics of Demand for Money.
- CO8. Identify in details with application, if applicable, Concepts of Micro and Macro Economics.

Course Description:

This course introduces the students to the basic concepts in Macroeconomics, definition, measurement and variables like GDP, consumption, savings, investment and Balance of Payments.

- I. Introduction: 15**
Macro Economics-Meaning-Definition-Scope-Importance and Limitation.
Key Variables of Macro Economics-Income-Output-Expenditure. Concepts of Stock and Flow. Equilibrium-General and Partial Equilibrium.
- II. National Income Accounting: 16**
a) National Income- Definition- Concepts of National Income-GNP-NNP-GDP-NDP-Per-capita Income-Disposable Income-NI at Factor Cost-NI at Market Cost-Nominal and Real Income. Green Accounting.
b) Measurement of National Income: Methods of measuring National Income-Importance-Difficulties.
- III. Determination of Income and Employment: 17**
a) Classical Theory of Employment: Say's Law of Market- Assumptions and Criticism.
b) Keynesian Theory of Income and Employment: Concept of Effective Demand-Consumption Function APC and MPC-APS and MPS. Investment Function-Induced Investment and Autonomous Investment-Rate of interest and MEC(Marginal Efficiency of Capital)
Determination of Equilibrium-ASF and ADF – Concept of Multiplier.
- IV. Business Cycle:**
Meaning, Types, Phases of Business Cycles- Causes of Business Cycles-Theories of Business Cycle: Sunspot Theory-Psychological Theory. Control of Business Cycle.

V. Macro Economic Policies of the Government:

Monetary Policy-Meaning-Objectives and Instruments-Quantitative and Qualitative methods
Fiscal Policy- Meaning-Objectives and Instruments-Tax and Expenditure.Budgetary
Measures.

Reference:

Case, Karl E and Ray C Fair, Principles of Economics, Pearson Education, Inc.8th edition, 2007.

Sikdar, Shoumyen, Principles of Macroeconomics, 2nd edition, Oxford University Press, India

Samuelson P.A-Economics(18th Edition, McGraw hill)

Mukarjee Sampath- Modern Economic Theory(New Age International)

H L. Ahuja-Modern Economic Theory(S. Chand & Company)

Mithani. D.M -Modern Economic Analysis(Himalayan Publication)

McConnel Compbell & Stanley Brue-Micro Economics(16th Edition, McGraw hill)

K. K. Dwett-Modern Economic Theory(S. Chand & Company)

S. Sankaran-Principles of Economics(Himalayan Publication)

M. L. Jhingan-Macro Economic Theory

M. L. Seth-Macroeconomics(Lakshminarayan Agarwal)

G. Mankiw-Macroeconomics(New Age International)

IV – Semester

DSC IV: -Principles of Macroeconomics-II

Course outcomes

On completion of the Course, students will:

- CO1. Specify in details with examples IS-LM Analysis.
- CO2. Learn in depth Supply side Economics.
- CO3. Identify the details of Rational Expectation.
- CO4. Identify in details with examples Concept if Inflation.
- CO5. Learn the classification and characteristics of Balance of Trade.
- CO6. Specify the classification and characteristics of Disequilibrium in Balance of Payment.
- CO7. Understand the details of Devaluation and its Effects.
- CO8. Identify the classification and characteristics of Exchange rate.

Course Description:

This is a sequel to Principles of Macroeconomics that analyses various theories of determination of National Income in detail. It also introduces students to concept of Inflation, its relationship with unemployment and some basic concepts in an open economy.

- | | |
|--|-----------|
| I. Money in Modern Economy & IS-LM Analysis | 17 |
| <ul style="list-style-type: none"> a) Money –Meaning-Definition and Functions-Role in Modern Economy. Quantitative Theory Money [Fisher’s Equation] and Restatement Theory of Money. b) IS-LM Analysis: Goods Market and Money and their Equilibrium-Derivation of IS-LM Functions-IS-LM Model- AS and AD Curve- Critique of IS-LM Model. | |
| II. Modern Macro Economics: | 13 |
| <ul style="list-style-type: none"> a) Supply Side Economics-Basic propositions of supply side Economics-Taxation-Labour supply-Incentives to save and investment –The Tax wedge-Tax Revenue and Laffer’s curve. b) Rational Expectations Analysis-Introduction to New Keynesian Theory-Aggregate Supply Function and Aggregate Demand Function. The New Classical Rational Expectations-Model –Policy implication[Lucas Model] | |
| III Inflation and Unemployment: | 18 |
| <p>Concept of Inflation-Deflation-Stagflation-Regression-Meaning-Definitions-Types-Determinants-Causes-Effects- Inflationary Gap- Controlling of Inflation-Relationship between Inflation and Unemployment-Phillip’s Curve.</p> | |
| IV. Balance of Payments: | 15 |

Balance of Payment-Balance of Trade and Balance of Payments-Composition of BOP-
Current Account-Capital account-Official Holdings.

Disequilibrium in BOP- Types-Causes-measures to Correct Disequilibrium in Balance of
Payments.

V. Foreign Exchange Rate:

12

Exchange Rate-Meaning and Types-Flexible-Fixed-Managed. Determination of Exchange
Rate-Devaluation and its Effects.

Reference:

Case, Karl E and Ray C. Fair, Principles of Economics, Pearson Education, Inc..8th edition,
2007.

Sikdar, Shoumyen, Principles of Macroeconomics,2nd edition, Oxford University Press, India.

Samuelson P.A-Economics(18th Edition, McGraw hill)

Mukarjee Sampath- Modern Economic Theory(New Age International)

H L. Ahuja-Modern Economic Theory(S. Chand & Company)

Mithani. D. M -Modern Economic Analysis(Himalayan Publication)

Mc Connel Campbell & Stanley Brue-Micro Economics(16th Edition, McGraw hill)

K. K. Dewett-Modern Economic Theory(S. Chand & Company)

S. Sankaran-Principles of Economics(Himalayan Publication)

M. L. Jhingan -Macro Economic Theory

M. L. Seth-Macroeconomics(Lakshminarayan Agarwal)

G. Mankiw-Macroeconomics((New Age International)

V Semester
DSE-1A: Economics of Development

Course outcomes

On completion of the Course, students will:

- CO1. Learn in depth Understand the concept of Economic development and factors affect Development.
- CO2. Deliberate in details with examples Differentiate Economic development and growth.
- CO3. Identify the characteristics of Demographic Trends.
- CO4. Specify in depth Harrod -Domar Growth Model.
- CO5. Understand the classification and characteristics of Endogenous Growth theory.
- CO6. Identify the details of Poverty Eradication Measures.
- CO7. Deliberate in depth Amartya Sen and Bhagavathi Debate.

Course Description:

This course reviews major trends in aggregate economic indicators of economic development, factors in economic development and theories of economic development and growth

- | | | |
|-------------|--|-----------|
| I | Economic Development: | 18 |
| | Economic Development –Meaning-definition- Concept of Sustainable Development and Inclusive Development-Economic Growth and Development-Factors affecting Development-Indicators of Economic Development-National Income, Per capita Income, Basic needs approach- PQLI-HDI-GEM-MDPI-Happiness Index. | |
| II | Factors in Economic Development: | 12 |
| | Capital Formation: Meaning and Importance-Capital Output Ratio- Physical Capital – Technology-Human Capital-Demographic Transition-Institutional factors. | |
| III. | General Theories of Economic Growth | 15 |
| | Adam Smith, Karl Marx and Schumpeter’s Theories of Economic Development. Harrod-Domor Growth Model-Rastow’s Growth Theory. | |
| IV. | Partial Theories of Economic Growth and Development: | 12 |
| | Arthur Lewis- Labour Surplus Model. Big push Theory-Theories of Dualistic Development-Balanced v/s Unbalanced Theories- Endogenous Growth Theory. | |
| V. | Economic Growth and Distributive Justice: | 18 |
| | Economic Growth and Social Justice: Poverty- Meaning, Types, Causes and Measures. Poverty Eradication Measures-Unemployment- Meaning, Types, Causes and Measures. Measures to reduce Unemployment. Occupational Structure-Organized and Unorganized Sector (with reference to India) Amartya Sen- Bhagavathi Debate. | |

Reference:

Michael P Todaro and Stephen Smith. Economic Development, Pearson, 11th edition (2011).

Uma Kapila, Indian Economy since Independence, Academic Foundation, 19th edition (2009).

United Nations Development Programme, Human Development Report 2010.palgrave Macmillan (2010).

Government of India, Economic Survey (latest).

Government of India, Five Year Plan (latest).

Government of India. Finance Commission Report (latest).

Dutt Ruddar & Sundaram .K. P. M -Indian Economy

Misra S. K & V. K. Puri-Indian Economy(Himalaya Publishing House)

Agarwal A. N -Indian Economy(Vishwa Publications)

P. K. Dhar -Indian Economy(New Age International)

V Semester
DSE-1B: Money and Banking

Course outcomes

On completion of the Course, students will:

- CO1. Understand the characteristics of Demand for Money.
- CO2. Specify the classification and characteristics of Capital Market.
- CO3. Deliberate the details of Theories of Interest.
- CO4. Learn the classification and characteristics of Functions of Commercial bank.
- CO5. Understand the characteristics of Indian Banking System.
- CO6. Learn the characteristics of Evaluation and Functions of Central bank.
- CO7. Deliberate the details of Monetary Policy.

Course Description:

This course exposes students to the theory and functioning of the monetary and financial sectors of the economy. It highlights the organization, structure and role of financial markets and institutions. It also discusses interest rates, monetary management and instruments of monetary control, financial and banking sector reforms and monetary system with special reference to India.

- I. Money :** **12**
Concept, Meaning and Definitions, Classifications of Money Functions and its Classification- Importance of Money. Demand for Money and determinants. Supply of money and determinants. Theories of Value of Money- Fisher's and Quantity Theory of Money.
- II. Money Market and Capital Market:** **15**
Money Market-Meaning-Structure and Instruments-Characteristics of a Good Money Market. Capital Market- Meaning-Structure and Instruments. Difference between Money Market and Capital Market.
- III. Interest Rates:** **14**
Rate of Interest-Meaning-Structure-Functions-short term and Long Term Interest. Theories of Interest rate Determination-Liquidity Theory of Interest- Don Pankin Theory-Interest rate in India.
- IV Commercial Banks & NBFCs:** **16**
 - a) Commercial Banks-Functions-Role-Balance Sheet-*Credit Creation*.
 - b) Indian Banking System-NBFCs: Concept-Structure-Functions-Banking sector reforms, Changing role and structure.
- v. Central Banking and Monetary Policy:** **18**
 - a) Central Bank-Evolution and Functions of Central Bank.

b) RBI and its role in the Development of Banking System in India- RBI and Monetary Policy. Credit control Instruments.

Reference:

F. S. Mishkin and S. G. Eakins, Financial Markets and Institutions, Pearson Education, 6th edition, 2009

F. J. Fabozzi, F. Modigliani, F. J. Jones, M. G. Ferri, Foundations of Financial markets and Institutions, Pearson Education, 3rd edition, 2009

L. M. Bhole and J. Mahukud, Financial Institutions and Markets, Tata McGraw Hill, 5th edition, 2011

M. Y. Khan, Indian Financial System, Tata McGraw Hill, 7th edition, 2011.

RBI Bulletin, Various latest issues of Annual Reports, Reports on Currency and Finance and Reports of the Working Group, IMF Staff Papers.

L. V. Chandler-Money & Banking (S. Chand & Company)

D. M. Mithani- Money & Banking and Financial System(Himalaya Publishing house)

R. R. Paul-Monetary Economics(Kalyani Publishers)

B. Gupta-Monetary Economics(S. Chand & Company)

V Semester

DSE-1C: Environmental Economics

Course outcomes

On completion of the Course, students will:

- CO1. Deliberate in details with examples Pareto Optimality.
- CO2. Understand the details of Market failure and Externalities.
- CO3. Understand in details with examples Implementation of Environmental Policy
- CO4. Identify in details with examples Economics of climate change
- CO5. Specify the details of Cost- benefit analysis of Environmental policies.
- CO6. Deliberate the characteristics of Sustainable Development.
- CO7. Identify the details of Environmental Valuation Methods and Application.
- CO8. Specify in depth Perspectives from Indian experience.

Course Description:

This course introduces students to concepts, methods and policy options in managing the environment using tools of economic analysis. Since several environmental problems are caused by economic activity, for instance, carbon emissions, over-harvesting of renewable resources and air and water pollution as a by- product of industrial activity. This course examines different approaches to adjusting behavior through economic institution such as markets and incentives.

- I. Introduction: 15**
Key environmental issues and problems, economic way of thinking about these problem, basic concepts from economics; Pareto optimality and market failure in the presence of externalities; property rights and other approaches.
- II. Economics of Natural Resources: 14**
Economics of Natural Resources: Types-Importance: Traditional-Natural Resources-Renewable and Non-Renewable Resources: Property Resources; Economics of Climate Change-Meaning-Causes and Effects.
- III. The Design and Implementation of Environmental Policy: 16**
Overview, Pigouvian taxes and effluent fees, tradable permits, implementation of environmental policies in India and international experience; trans-boundary environmental problems:
- IV. Environmental Valuation Methods and Applications: 14**
Valuation of Environmental goods and services-theory and practice; measurement methods; Direct and Indirect Application: Travel Cost Method-Cost-benefit analysis: Contingent Method.
- V. Sustainable Development: 14**
Concepts: measurement; Agriculture Sustainable Development; Industrial Sustainable Development: Rural Sustainable Development; Perspectives from Indian experience.

Reference:

Roger Perman, Yue Ma, Michael Common, David Maddison and James McGilvray, "Natural Resource and Environmental Economics" Pearson Education/Addison.

Charles Kolstad," Intermediate Environmental Economics" Oxford University press 2nd edition,2010.

Robert N. Stavins (ed.),"Economics of the Environment: Selected Readings", W. W. Norton 6th edition,2012.

V Semester
GE-I Introduction to Economics

Course outcomes

On completion of the Course, students will:

- CO1. Analyze Micro Economic issues and Scarcity, Choice & Production Possibility Curve.
- CO2. Understand Law of demand, determinants of demand.
- CO3. Gain knowledge about Law of Supply, determinants of Supply and Equilibrium of demand & supply.
- CO4. Have insight into Production function: Law of variable proportion.
- CO5. Clarify the meaning & Types of various Cost Concepts and Revenue Concepts.
- CO6. Understand to various classification of Market.

- I. Introduction: 08**
Economics-Meaning, Definitions-Need for the study: Micro and Macro Economics-Meaning-Importance and Limitations: Basic Problems of an Economy: Problem of Choice: Production Possibility Curve.
- II. Theories of Demand and Supply: 07**
Demand: Meaning-Law of Demand: Exception of the Law of Demand: Factors determining Demand, Supply-Meaning-Law of Supply, Factors determining Supply.
- III. Production, Cost and Revenue Analysis: 08**
Production Function-returns to scale-Law of Variable Proportions; Cost Concepts-Short run and Long run Cost Curves; Revenue Concept-TR,AR and MR.
- IV. Market Structure: 07**
Market-Meaning and Types of Market: Perfect Competition Market-Meaning-Features, Monopoly and Oligopoly-Features.

Reference:

D.M.Mithani-Modern Economic Analysis-Himalayan Publishing House.
H.L.Ahuja-Advanced Economic Theory
J.K.Mithra-Principles of Economics
K.K.Dewett-Modern Economic Theory
G.Mankiw-Principles of Economics
Mukhaerjee Sampat-Modern EconomTheory

VI Semester
DSE-1A : Indian Economy

Course outcomes

On completion of the Course, students will:

- CO1. Understand the characteristics of Indian Agricultural policies.
- CO2. Identify the classification and characteristics of Regional variation.
- CO3. Write down the classification and characteristics of New Industrial Policy.
- CO4. Specify in depth Public and Private Sector.
- CO5. Identify in depth Monetary Policy.
- CO6. Understand in depth FDI and WTO.
- CO7. Learn in details with examples Public Debt.
- CO8. Identify the details of Effects of Parallel Economy.

Course Description:

This course examines sector-specific trends in key indicators and their implications in the post-Independence period.

- I. Indian Agriculture: Policies and Performance** **14**
Introduction to Indian Economy-Role of Agriculture in India-Agriculture development-Agriculture-Production and Productivity-Agriculture Credit-Agriculture Labour-Agriculture Pricing and Marketing-Land Reforms-Regional Variation. National Agricultural Policy.
- II. Indian Industries: Policies and Performance** **15**
Role of Industries in India-Importance and Problems of Large Scale and MSMEs in India. Private Sector and Public Sector Industries-Disinvestment Policy. New Industrial Policy. Development of IT Industries.
- III. Development of Service Sector In India** **16**
a) Growth of Service Sector in India: Trends and Issues. Health and Education .
b)Role of commercial banks in Indian economy; Reserve Bank of India and Monetary policy; Reforms in the Banking sector(Narasimmam committee only)
- IV. India's Foreign Trade: Trends and Policies** **13**
Composition, directions, Trends in India's Foreign Trade. FDI and MNCs. WTO and India's Foreign Trade. Balance of Trade, Balance of Payments. Trade Liberalization. EXIM Policy
- V. Indian Public Finance** **17**
Revenue and expenditure of central and state governments; India's public debt; deficit financing; financial relations between central and state; parallel economy-meaning, causes, extent and consequences; measures to control it. Concept of Demonetisation.GST-Current Year Union Budget

Reference:

UmaKapila Indian Economy Since Independence, Academic Foundation,19th edition(2009)

Government of India, Economic Survey (latest).

Government of India, Five Year Plan (latest).

Michael P Todaro and Stephen Smith. Economic Development, Pearson, 11th edition (2011).

United Nations Development Programme, Human Development Report 2010.palgrave Macmillan (2010).

RBI-Handbook of Statistics on Indian Economy

Government of India. Finance Commission Report (latest).

Dutt Ruddar & Sundaram.K.P.M-Indian Economy

Misra S.K & V.K.Puri-Indian Economy(Himalaya Publishing House)

Agarwal A.N-Indian Economy(Wishva Publications)

P.K.Dhar-Indian Economy(New Age International)

VI Semester
DSE-1B: Economic History of India 1857-1947

Course outcomes

On completion of the Course, students will:

- CO1. Learn in depth Indian Economy in the pre -British period.
- CO2. Specify in depth Estimation of National Income in India.
- CO3. Identify the characteristics of Agriculture markets and Institutions- credit.
- CO4. Understand in details with examples Evaluation of Entrepreneurial and Industrial Structure.
- CO5. Learn the details of Government and Fiscal policy.

Course Description:

This course analyses key aspects of Indian economic development during second half of British rule and investigates the place of Indian economy in the wider colonial context. This course links directly to the course on India's development after independent in 1947.

- I. Introduction: Colonial India: Background and Introduction 15**
Overview of colonial Economy: Meaning of Colonialism: British Rule and Exploitation of India-The British Rule and India's Underdevelopment-State Policies and Economic Underdevelopment: Colonial Exploitation: forms and Consequences: Colonialism and Modernization: Nature of Indian Economy.
- II. Macro Trends: 16**
National Income: Estimates in India-Trends, Limitations. Population: Size and Growth Rate: Demographic Transition: Causes and Measurement of Rapid Growth of Population Occupational Structure: Occupational distribution of Labour Force in India: Historical Experience of Structural Change.
- III. Agriculture: 17**
Agrarian structure: Productivity and Efficiency: land relations; Cropping Patten-Irrigation: Agricultural markets and institutions-Credit. Commerce and technology; trends in performance and productivity; famines.
- IV. Railway and Industry: 13**
Railways; the de-industrialization debate; evolution of entrepreneurial and industrial structure; Nature of industrialization in the interwar period; constraints to industrial breakthrough; labor relations.
- v. Economy and State in the Imperial Context: 14**
The imperial priorities and the Indian economy; drain of wealth; international trade, Capital,

flows and the colonial economy- changes and continuities; government and fiscal policy.

Reference:

Lakshmi Subramaniam, History of India-1707-1857, Orient Blackswan (2010)

Tirthankar Roy, The Economic History of India 1857-1947, Oxford University (2011)

J.Krishnamurthy, Occupational Structure, The Cambridge

Irfan Habib, Indian Economy 1858-1914,(2006)

VI Semester
DSE-1C: Public Finance

Course outcomes

On completion of the Course, students will:

- CO1. Understand the classification and characteristics of Public Economics and Public Finance.
- CO2. Understand in details with examples Test of maximum social Advantage.
- CO3. Understand in depth Tax and non Tax Revenue.
- CO4. Identify the characteristics of Central and State financial Relations.
- CO5. Deliberate the classification and characteristics of Central and State financial Relations.

Course Description:

This course gives a non-technical overview of government finances with special reference to India. It will look into the efficiency and equity aspects of Taxation of the centre, states and the local governments and the issues of fiscal federalism and decentralization in India. The course will be useful for students aiming towards careers in the government sector, policy analysis, business and journalism.

I. Introduction :

13

Meaning, Nature and Scope of Public Finance, differences between Public Economics and Public Finance. Importance of the study Public Economics-Role of Public Finance in Developing Countries.

II. Principles of Public Economics:

15

Principles of maximum social advantage, Principle of allocation of resources; Test of maximum social advantage; Pareto's welfare Theory; Social welfare functions of modern governments; Distinction between private and public goods; Market imperfection; Externalities.

III. Public Revenue, Expenditure and Debt:

17

- a) Public Revenue-Tax and Non-Tax Revenue-Cannon of Taxation.
- b) Public Expenditure-Principles of Public Expenditure and Wagner's Law of Increasing state Activities.
- c) Public Debt-Meaning-Causes-Burden and Redemption.

IV. Issues In Indian Public Finance:

16

- a) Working of monetary and fiscal policies
- b) Current Issues of Indian Tax System.
- c) Central and State financial Relations.
- d) State and Local Finance.

Classification of Budgets: Programme performance budget; Budgetary deficits-Revenue and fiscal deficits; Zero based budgeting; Fiscal policy-meaning, objectives and Instruments

Reference:

Musgrave, R. A. and P. B. Musgrave, Public Finance in Theory and Practice, Mc-Graw Hill, 1989
Mahesh Purohit, "Value Added Tax: Experience of India and other Countries" Gayatri Publications, 2007

Kaushik Basu, and A. Maertens (ed). The Oxford Companion to Economics in India, Oxford University Press, 2007

M. M. Sury - Government Budgeting in India, Commonwealth publishers, 1990.

Shankar Acharya, 'Thirty Years of tax reform' in India, Economic and Political Weekly, May 2005.

Report of the 13th Finance Commission - Government of India

Economic survey - Government of India (latest)

State Finances: A Study of Budgets, Reserve Bank of India (latest).

H. L. Bhatia-Public Finance (Sulthan Chand & Sons)

S. K. Singh-Public Finance in Theory & Practice (Himalayan Publications)

K. P. M. Sundaram & K. K. Andley -Public Finance (Sulthan Chand & Sons)

B. P. Tyagi- Public Economics (Himalayan Publications)

VI Semester
GE-II Indian Economy

Course outcomes

On completion of the Course, students will:

- CO1. Identify in details with examples Human Resource.
- CO2. Learn the characteristics of Population policy.
- CO3. Identify the details of Role of Agriculture.
- CO4. Understand the details of Rural Development.
- CO5. Understand in depth Disinvestment.
- CO6. Deliberate the classification and characteristics of Foreign Trade.

- I. Introduction :** **10**
Features of Indian Economy; Concepts of National Income -GNP-NNP-GDP-NDP- Per capita Income-Disposable Income- Green GDP; of Population Explosion: India's Population Policy; Poverty: Definition; Poverty and Inequality; Poverty Alleviation Schemes; Unemployment: Definition and Types: Employment Guarantee Schemes(MGNREGA).
- II. Agriculture Sector:** **08**
Role of Agriculture; Cropping Pattern-Land Reforms-Green Revolution- Food Security; Sources of Agricultural Credit-Co-Operatives Banks, Regional Rural Banks, NABARD, Commercial Banks; Agricultural Marketing-Problems and remedies; Agriculture and WTO.
- III. Industry and Tertiary Sector:** **12**
Role of Industries; Industrial Policy since 1991; MSMEs-Role and Problems; Large Scale Industries; Disinvestment in Public Sector undertakings: Emergence of IT Industry.
Role of Commercial Banks in Indian Economy; Reserve Bank of India: Functions and Credit Control Methods: Reforms in the Banking Sector (Narasimham Committee only): Foreign Trade- Trends, Composition, directions of Trade-India's Balance of Payments; Foreign Exchange Rate.

Reference:

Dutt Ruddar & Sundaram K.P.M-Indian Economy(S. Chand & Co, New Delhi)
Kapila Uma, Indian Economy since Independence, Academic Foundations, New Delhi
Misra S.K & V.K Puri - Indian Economy (Himalaya Publishing House)
S. Sankaran - Indian Economy(Margham Publications)
Mohan Rakesh (Edition), Facts of the Indian Economy, OUP, New Delhi
Vaidyanathan A. India's Economic Reforms and Development, Oxford University Press New Delhi
Agarwal A.N- Indian Economy (Vishva Publications)
M.C.Vaish-Indian Economy (New Age International)
P. K. Dhar- Indian Economy (New Age International)
Economic Survey (Recent) – Govt. of India and Karnataka
Data Sources: Hand Book of statistics of Indian Economy
Census Report & NSSO Report.

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(AUTONOMOUS)
B.N. ROAD, MYSORE-25

DEPARTMENT OF ECONOMICS
MODEL QUESTION PAPER FOR DSC & DSE
B.A.ECONOMICS
I, II, III, IV, V&VI Semester
(CBCS Scheme)

Time: 3 Hours

Max. Marks – 70

PART-A

I. Answer any Five of the following questions.

5x2=10

1)

- a) -----
- b) -----
- c) -----
- d) -----
- e) -----
- f) -----
- g) -----

PART-B

II. Answer any four of the following questions.

4x5=20

- 2. -----
- 3. -----
- 4. -----
- 5. -----
- 6. -----
- 7. -----

PART-C

III. Answer any four of the following questions.

4x10=40

- 8. -----
- 9. -----
- 10. -----
- 11. -----
- 12. -----

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

(AUTONOMOUS)

B.N. ROAD, MYSORE-25

DEPARTMENT OF ECONOMICS

MODEL QUESTION PAPER FOR GENERIC ELECTIVE

B.A.ECONOMICS

V&VI Semester

(CBCS Scheme)

Time: 2 Hours

Max. Marks – 35

PART-A

I. Answer any Five of the following questions.

5x2=10

1. -----
2. -----
3. -----
4. -----
5. -----
6. -----
7. -----

PART-B

II. Answer any Three of the following questions.

5x3=15

8. -----
9. -----
10. -----
11. -----
12. -----

PART-C

III. Answer any One of the following questions.

10x1=10

13. -----
14. -----
15. -----



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

(Autonomous)

B N ROAD, MYSURU- 570 025

DEPARTMENT OF ELECTRONICS

Syllabus

CHOICE BASED CREDIT SYSTEM

For B.Sc programme

Physics, Mathematics, Electronics

2019-20

PROGRAMME CODE: BSc-04**Credit Pattern for Courses**

Sem	Course Type	Course code	Course Title	L:T:P	Total Credit
I	DSC1: Theory	DMA26004	Network Analysis and Analog Electronics	4:0:0	6
	DSC1: Practical-1	DMA26104	Network Analysis and Analog Electronics	0:0:2	
II	DSC2: Theory	DMB26004	Linear and Digital Integrated Circuits	4:0:0	6
	DSC2: Practical- II	DMB26104	Linear and Digital Integrated Circuits	0:0:2	
III	DSC3: Theory	DMC26004	Communication Electronics	4:0:0	6
	DSC3: Practical- III	DMC26104	Communication Electronics	0:0:2	
IV	DSC4: Theory	DMD26004	Microprocessor and microcontrollers	4:0:0	6
	DSC4: Practical-IV	DMD26104	Microprocessor and microcontrollers	0:0:2	
V	DSE 1A : Theory	DME26004	Digital Signal Processing	4:0:0	5
	DSE1A: Practical-V	DME26104	Practical 5: Digital Signal Processing	0:0:1	
	DSE1B: Theory	DME26204	Electronic Instrumentation	4:0:0	5
	DSE1B: Practical-VI	DME26304	Practical 6: Electronic Instrumentation	0:0:1	
	SEC1A: Theory	DME26404	Electrical circuits and network skills	2:0:0	2
	SEC1B: Theory	DME26604	Computer networks	2:0:0	2
VI	DSE2A: Theory	DMF26004	VERILOG & VHDL	4:0:0	6
	DSE2A: Practical-VII	DMF26104	VERILOG & VHDL	0:0:2	
	DSE2B: Theory	DMF26204	Photonic Devices and Power Electronics	4:0:0	6
	DSE2B: Practical-VIII	DMF26304	Photonic Devices and Power Electronics	0:0:2	

L: Lecture, T: Tutorial, P: Practicals

Scheme for Examination and Assessment

Course type	IA C1		IA C2		SEE C3		Exam duration	
	Th	Pr	Th	Pr	Th	Pr	Th	Pr
DSC	15	7.5	15	7.5	70	35	3h	3h
DSE	15	7.5	15	7.5	70	35	3h	3h
SEC	7.5	-	7.5	-	50	-	2h	

SCHEME OF VALUATION FOR PRACTICAL EXAMINATION

- A candidate appearing for the first time should submit a duly signed and certified practical record
- Each candidate has to perform one experiment in the specified duration of two hours for **THIRY FIVE marks**
- Practical record has to be valued for **FIVE marks** by examiners at the time of examination
- IA for **FIFTEEN marks** in practical is awarded by two tests in the lab

I. ANALOG & DIGITAL EXPERIMENTS:

Sl no	Component	Marks
1	Write up of the experiment	10
2	Conducting experiment	10
3	Result	05
4	Viva-voce	05
5	Practical record	05
TOTAL		35

II. FOR PROGRAMMING EXPERIMENTS:

Sl no	Component	Marks
1	Program Writing	10
2	Entering /Coding	10
3	Execution	05
4	Practical record	05
5	Viva	05
TOTAL		35

Programme Outcome

Bachelor of Science in Physics, Mathematics, Electronics

After completing the graduation in the Bachelor of Science the students are able to:

PO1: Demonstrate proficiency in Mathematics and the Mathematical concepts needed for a proper understanding of Physics.

PO2: Demonstrate the ability to justify and explain their thinking and/or approach.

PO3: Develop state-of-the-art laboratory and professional communication skills.

PO4: Apply the scientific method to design, execute, and analyze an experiment.

PO5: Explain scientific procedures and their experimental observations.

PO6: Understand the value of Mathematical proof.

PO7: Demonstrate proficiency in writing and understanding proofs.

PO8: Apply mathematical problems and solutions in aspects of science and technology.

PO9: Gain experience to investigate the real world problems.

PO10: Apply mathematical ideas and models to problems.

PO11: Apply appropriate troubleshooting techniques to electronic circuits / systems and perform test procedures.

PO12: Assist, Assemble, modify and test electronic circuits in accordance with job requirements.

PO13: Communicate effectively in technical and non-technical environments.

Programme Specific Outcome

Bachelor of Science in Physics, Mathematics, Electronics

After completing the graduation in the Bachelor of Science the students are able to:

PSO1: Find career opportunities.

PSO2: Develop competence to write competitive examinations.

PSO3: Develop proficiency in the analysis of complex physical problems.

PSO4: Use mathematical or other appropriate techniques to solve complex physical problems.

PSO5: Create a hypothesis and appreciate how it relates to broader theories.

PSO6: Demonstrate skills in the use of Computers for control, data acquisition, and data analysis in experimental investigations.

PSO7: Apply knowledge of Physics, Mathematics and Electronics fundamentals to the solve problems in Electronic circuits & communication systems.

PSO8: Apply appropriate troubleshooting techniques to Electronic circuits / systems and perform test procedures.

DSC1: DMA26004

SEMESTER I

NETWORK ANALYSIS AND ANALOG ELECTRONICS

Credits: Theory – 04

Practical – 02

Theory: 60 Lectures

COURSE OUTCOME:

After completion of the course the student is able to

CO1: Deliberate Network theorems with examples

CO2: Specify the classification and characteristics of semiconductor diodes and transistors

CO3: Deliberate in detail the application of semiconductor diodes and transistors

CO4: Understand the characteristics of FET & UJT.

Unit-1

Circuit Analysis:

Concept of Voltage and Current Sources. Kirchhoff's Current Law, Kirchhoff's Voltage Law. Mesh Analysis. Node Analysis. Star - Delta networks – Star to Delta and Delta to Star Conversions.

Network Theorems - Superposition Theorem. Thevenin's Theorem. Norton's Theorem. Reciprocity Theorem. Maximum Power Transfer Theorem (Statement and explanation only).

Two Port Networks: h, y and z parameters and their conversion. (15 Lectures)

Unit-2

Junction Diode and its applications:

PN junction diode (Ideal and practical) construction of diode, formation of depletion layer in diode, V-I characteristics. Static and dynamic resistance,

Zener diode, Reverse saturation current, Zener and avalanche breakdown. Qualitative idea of Schottky diode and Varactor diode

Rectifiers- Half wave rectifier, Full wave rectifiers (Centre tapped and Bridge) - Circuit diagrams, working and waveforms, Expressions for ripple factor and efficiency.

Filter- Shunt capacitor filter, working with FWR.

Regulation- Line and load regulation, Zener diode as voltage regulator and explanation for load and line regulation. (15 Lectures)

Unit-3

Bipolar Junction Transistor:

Introduction to transistors- construction, types and operation of transistors, Characteristics of transistor in CE and CB configurations, Regions of operation (active, cut off and saturation), Current gains α and β . Relations between α and β . DC load line and Q point.

Transistor biasing:

Need for biasing, DC load line and Q point, Thermal runaway, Stabilization - stability and stability factor, Expression for stability factor S. (only S_{Ico} derivation), Fixed Bias and Voltage Divider Bias. (Relevant Numericals)

Amplifiers:

Definition and classification of amplifiers, single stage CE amplifier- construction, working and frequency response

Transistor as a two port network, h-parameter equivalent circuit. Small signal analysis of single stage CE amplifier. Input and Output impedance, Current and Voltage gains.

Power amplifiers -Class A (Qualitative analysis), Class B (derivation for expression for efficiency) and class C Amplifiers (Qualitative analysis) (15 Lectures)

Unit-4

Application of transistors:

Cascaded Amplifiers:

Multistage amplifiers, Two stage RC Coupled Amplifier and its Frequency Response.

Feedback in Amplifiers:

Concept of feedback, negative and positive feedback (expression for gain), advantages of negative feedback (Relevant Numericals).

Sinusoidal Oscillators:

Barkhausen criterion for sustained oscillations. Phase shift and Colpitt's oscillator - derivation of expression for Frequency and Condition of oscillation.

Unipolar Devices:

JFET - Construction, working and I-V characteristics (output and transfer), expression for Pinch off voltage (no derivation).-Relevant Numericals

UJT - Construction, working, equivalent circuit and I-V characteristics (Relevant Numericals)

(15 Lectures)

Reference Books:

- Electric Circuits, S. A. Nasar, Schaum's outline series, Tata McGraw Hill (2004).
- Electrical Circuits, M. Nahvi & J. Edminister, Schaum's Outline Series, Tata McGraw-Hill (2005).
- Electrical Circuits, K.A. Smith and R.E. Alley, 2014, Cambridge University Press.
- Network, Lines and Fields, J.D. Ryder, Prentice Hall of India.
- Electronic Devices and Circuits, David A. Bell, 5th Edition 2015, Oxford University Press.
- Electronic Circuits: Discrete and Integrated, D.L. Schilling and C. Belove, Tata McGraw Hill.
- Electrical Circuit Analysis, Mahadevan and Chitra, PHI Learning.
- Microelectronic circuits, A.S. Sedra, K.C. Smith, A.N. Chandorkar, 2014, 6th Edn., Oxford University Press.
- J. Millman and C. C. Halkias, Integrated Electronics, Tata McGraw Hill (2001).
- J. J. Cathey, 2000 Solved Problems in Electronics, Schaum's outline Series, Tata McGraw Hill (1991).

**PRACTICAL 1
NETWORK ANALYSIS AND ANALOG ELECTRONICS**

1. To familiarize with basic electronic components (R, C, L, diodes, transistors), Digital Multimeter, Function Generator and Oscilloscope.
2. Measurement of Amplitude, Frequency & Phase difference using Oscilloscope.
3. Verification of Thevenin's theorem and (b) Norton's theorem.
4. Verification of (a) Superposition Theorem and (b) Reciprocity Theorem
5. Verification of the Maximum Power Transfer Theorem.
6. Study of the I-V Characteristics of (a) p-n junction Diode and (b) Zener diode.
7. Study of (a) Half wave rectifier and (b) Full wave rectifier (FWR) with C filter and without filter
8. Zener diode as voltage regulator (at the output of Full Wave Rectifier)
9. Study of the I-V Characteristics of UJT
10. Study of the output and transfer I-V characteristics of common source JFET.
11. Study of Fixed Bias and Voltage divider bias configuration for CE transistor.
12. Study of Single Stage CE amplifier.
13. Study of the RC Phase Shift Oscillator.
14. Study the Colpitt's oscillator.

(Note: Minimum of Eight experiments to be conducted)

DSC 2: DMB26004

**SEMESTER II
LINEAR AND DIGITAL INTEGRATED CIRCUITS**

Credits: Theory – 04

Practical – 02

Theory: 60 Lectures

COURSE OUTCOME:

After completion of the course the student is able to

CO1: Understand the characteristics and applications of operational amplifiers

CO2: Design different signal conditioning circuits like filters, A/D and D/A converters

CO3: Understand the fundamentals of converting from one number system to another

CO4: Interpret logic functions, combinational and sequential digital circuits

Unit-1

Operational Amplifiers (Black box approach):

Differential amplifier, Block diagram of Op-amp, Characteristics of an Ideal and Practical Operational Amplifier (IC 741), Open and closed loop configurations, Frequency Response. CMRR & Slew Rate.

Applications of Op-Amps:

Inverting and non-inverting amplifiers and concept of Virtual Ground, (1) Summing, Averaging and Scaling amplifiers (2) Difference Amplifier (3) Differentiator (4) Integrator (5) Active low pass and high pass Butterworth filter (1st order only). (15 Lectures)

Unit-2

Applications of Op-amp contd. & Timer (IC 555):

Wein bridge oscillator, Comparator and Zero-crossing detector

Introduction to 555 Timer - block diagram, 555 as a monostable multivibrator and applications (Frequency divider and Pulse stretcher) 555 as an astable multivibrator and applications - square wave generator (Relevant Numerical)

D-A and A-D Conversion:

D – A conversion- 4 bit binary weighted and R-2R D-A converters,. Accuracy and Resolution.

A-D conversion - characteristics, successive approximation ADC. (Mention of relevant ICs for all). (15 Lectures)

Unit-3

Number System and Codes:

Decimal, Binary, Octal and Hexadecimal number systems –conversion from one system to another. Representation of signed and unsigned binary numbers, Binary arithmetic- addition, Subtraction by 1's and 2's complement method, multiplication & division. BCD code (BCD Addition).

Logic Gates and Boolean algebra:

Discussion and Truth Tables of OR, AND, NOT, NOR, NAND, XOR, XNOR Gates. NAND & NOR as Universal gates. Basic postulates of Boolean algebra. Principle of duality. De Morgan's theorems. Simplification of Boolean expressions, Boolean expression for logic circuits and vice versa.

Combinational Logic Analysis and Design:

Standard representation of logic functions (SOP and POS), Minimization Techniques - Karnaugh map (minimization up to 4 variables for SOP only). (15 Lectures)

Unit-4

Combinational and sequential circuits:

Arithmetic Circuits:

Binary Addition and subtraction - Half and Full Adder. Half and Full Subtractor, 4-bit binary Adder/Subtractor.

Data processing circuits:

Multiplexers(4 X 1) , De-multiplexers (1 X 4) ,

Decoders- 3 to 8 lines, BCD to decimal

Encoders - 8 to 3 line, Decimal to BCD encoders

Sequential Circuits:

RS flip flop, clocked RS and D flip flops. JK flip flop. Preset and Clear, Race around condition.

Master-slave JK Flip-Flop.

Shift registers:

Block diagram of Serial-in-Serial-out, Serial-in-Parallel-out, Parallel-in-Serial-out and Parallel-in-Parallel-out Shift Registers (only up to 4 bits) Qualitative Analysis of Serial-in-Serial-out, Serial-in-Parallel-out.

Counters:

Asynchronous counters- Ripple counter, Decade counter. Up/Down Counter, Mod-3

(15 Lectures)

Reference Books:

- OP-Amps and Linear Integrated Circuit, R. A. Gayakwad, 4th edition, 2000, Prentice Hall.
- Operational Amplifiers and Linear ICs, David A. Bell, 3rd Edition, 2011, Oxford University Press.
- Digital Principles and Applications, A.P. Malvino, D.P. Leach and Saha, 7th Ed., 2011, Tata McGraw.
- Fundamentals of Digital Circuits, Anand Kumar, 2nd Edn, 2009, PHI Learning Pvt. Ltd.
- Digital Circuits and systems, Venugopal, 2011, Tata McGraw Hill.
- Digital Systems: Principles & Applications, R.J. Tocci, N.S. Widmer, 2001, PHI Learning.
- Thomas L. Floyd, Digital Fundamentals, Pearson Education Asia (1994).
- R. L. Tokheim, Digital Principles, Schaum's Outline Series, Tata McGraw- Hill (1994).

PRACTICAL 2
LINEAR AND DIGITAL INTEGRATED CIRCUITS LAB

Section – A: Op-Amp. Circuits (Hardware)

1. To design an Inverting & Non-inverting amplifier using Op-amp (741,351) for DC voltage of given gain
2. To design an Inverting amplifier using Op-amp (741) & to study its frequency response(for AC voltage)
3. To design non-inverting amplifier using Op-amp (741) & to study frequency response(for AC voltage)
4. To Add / Subtract two dc voltages using Op-amp in inverting mode.
5. To study the zero-crossing detector and comparator.
6. To investigate the use of an op-amp as an Integrator.
7. To investigate the use of an op-amp as a Differentiator.
8. To study a Wien bridge oscillator using an op-amp.
9. To design an Astable Multivibrator of given specification using IC 555 Timer.
10. To design a Butterworth Low Pass active Filter (1st order) & to study Frequency Response.
11. To design a Butterworth High Pass active Filter (1st order) & to study Frequency Response.
12. R – 2R digital to analog converter (DAC).

Section-B: Digital circuits (Hardware)

1. To simplify the given Boolean expression and to construct logic circuit using basic gates
2. To study NAND & NOR gates as Universal gates
3. Half Adder and Full Adder.
4. Half Subtractor and Full Subtractor.
5. 4 bit binary Adder-Subtractor using Full adder IC.
6. Seven segment decoder.
7. Study of Encoder
8. To build Flip-Flop (RS, Clocked RS, D-type and JK) circuits using NAND gates.
9. To build JK Master-slave flip-flop using Flip-Flop ICs.
10. To build a Counter using D-type/JK Flip-Flop ICs and study timing diagram.

11. To make a Shift Register (serial-in and serial-out) using D-type/JK Flip-Flop ICs.

Section-C: SPICE/MULTISIM simulations for electronic circuits and devices

1. To verify the Thevenin's and Norton Theorems.
2. Design and analyze the series and parallel LCR circuits
3. Design the inverting and non-inverting amplifier using an Op-Amp of given gain
4. Design and Verification of op-amp as integrator and differentiator
5. Design the 1st order active low pass and high pass filters of given cut-off frequency
6. Design a Wein Bridge oscillator of given frequency.
7. Design clocked SR and JK Flip-Flop`s using NAND Gates
8. Design 4-bit asynchronous counter using Flip-Flop ICs
9. Design the CE amplifier of a given gain and its frequency response.

(Minimum of 05 experiments from section A, B and C to be done)

DSC3: DMC26004

**SEMESTER III
COMMUNICATION ELECTRONICS**

Credits: Theory – 04,

Practical – 02

Theory: 60 Lectures

COURSE OUTCOME:

After completion of the course the student is able to

CO1: Understand the classification and characteristics of analog communication systems

CO2: Identify the classification and characteristics of pulse modulation systems

CO3: Specify the classification and characteristics of digital communication systems

CO4: Specify the classification and characteristics of satellite communication systems

CO5: Identify the classification and characteristics of mobile communication systems

Unit-1

Electronic communication:

Introduction to communication – means and modes. Block diagram of an electronic communication system. Brief idea of frequency allocation for radio communication system in India (TRAI). Electromagnetic communication spectrum, band designations and usage. Channels and base-band signals. Concept of Noise – External Noise – Atmospheric noise, Extraterrestrial noise, Industrial noise. Internal noise – Thermal agitation Noise, Shot Noise, Transit-time Noise, Miscellaneous Noise. Signal-to-Noise (S/N) ratio.

Analog Modulation – Amplitude modulation:

Modulation-types of modulation Need for modulation, Amplitude Modulation, modulation index – expression for modulation index, Analysis of AM wave, Frequency spectrum of AM wave, Power Relation, Current Calculation, Modulation by several sine waves. Generation of AM (Emitter Modulation),

Single Side Band Generation- Balanced Modulator and suppression of sidebands using filter method.

Amplitude Demodulation (diode detector),

(15 Lectures)

Unit-2

Analog Modulation contd:

Frequency Modulation (FM) modulation index, Analysis of FM Wave and frequency spectrum
Phase Modulation (PM), equivalence between FM and AM, Generation of FM using VCO,
FM detector (slope detector),

Receivers:

Characteristics of a Radio Receiver, Block diagram of AM & FM super heterodyne receiver

Analog Pulse Modulation:

Channel capacity, Sampling theorem, Basic Principles-PAM, PWM, PPM, modulation and
detection technique for PAM only, Multiplexing. (15 Lectures)

Unit-3

Digital Pulse Modulation:

Need for digital transmission, Pulse Code Modulation, Digital Carrier Modulation Techniques,
Sampling, Quantization and Encoding. Concept of Amplitude Shift Keying (ASK), Frequency
Shift Keying (FSK), Phase Shift Keying (PSK), and Binary Phase Shift Keying (BPSK).

Satellite Communication

Introduction, need for satellite communication, orbital Parameters, Kepler's Laws, Types of
Orbits, Geosynchronous satellite orbits, geostationary satellite, advantages of geostationary
satellites. Antenna Look Angles, Satellite visibility, transponders (C - Band), ground station,
simplified block diagram of earth station (15 Lectures)

Unit-4

Mobile Telephony system

Basic concept of mobile communication, frequency bands used in mobile communication,
Frequency reuse, Interference- Co-channel interference, Adjacent channel Interference, Cell
splitting, Sectoring, Segmentation and Dualization, Roaming and Handoffs, SIM number, IMEI
number, need for data encryption, Architecture (block diagram) of mobile communication
network. Idea of GSM, CDMA, TDMA and FDMA technologies, GPS navigation system
(qualitative idea only) (15 Lectures)

Reference Books:

- Electronic Communications, D. Roddy and J. Coolen, Pearson Education India.
- Advanced Electronics Communication Systems- Tomasi, 6th edition, Prentice Hall.
- Modern Digital and Analog Communication Systems, B.P. Lathi, 4th Edition, 2011, Oxford University Press.
- Electronic Communication systems, G. Kennedy, 3rd Edn., 1999, Tata McGraw Hill.
- Principles of Electronic communication systems – Frenzel, 3rd edition, McGraw Hill.
- Communication Systems, S. Haykin, 2006, Wiley India.
- Electronic Communication system, Blake, Cengage, 5th edition.
- Wireless communications, Andrea Goldsmith, 2015, Cambridge University Press.

PRACTICAL 3
COMMUNICATION ELECTRONICS

1. To study an Amplitude Modulator using Transistor.
2. To study envelope detector for demodulation of AM signal.
3. To study FM – Generator.
4. To study AM Transmitter and Receiver
5. To study FM Transmitter and Receiver
6. To study Time Division Multiplexing (TDM)
7. To study Pulse Amplitude Modulation (PAM).
8. To study Pulse Width Modulation (PWM).
9. To study Pulse Position Modulation (PPM).
10. To study ASK modulators.
11. To study PSK modulators.
12. To study FSK modulators.
13. IF amplifier.
14. RF amplifier.

(Minimum of Eight experiments to be conducted)

MICROPROCESSOR AND MICROCONTROLLER LAB

Credits: Theory – 04

Practical – 02 Theories: 60 Lectures

COURSE OUTCOME:

After completion of the course the student is able to

CO1: Understand the architecture of 8085 microprocessors

CO2: Write down the instruction set and simple programs of 8085 microprocessors.

CO3: Understand the architecture and instruction set of 8051 microcontrollers

CO4: Specify the characteristics of embedded system

.Unit-1

8085 Microprocessor:

Block diagram of Digital Computer, Microcomputer Organization, Bus organization of 8085, Memory – types of memory , Input /Output

Architecture:

Main features of 8085. Block diagram. Pin-out diagram of 8085. Control and Status Signals, Power supply and Clock Frequency. Architecture –ALU, Flags & Registers and Timing and Control unit.

8085 Programming:

Instruction format-Word sizes, Opcode format Instruction classification, Data transfer, Timing diagram of MOV and MVI, Arithmetic, logical and branch instructions, Simple Programs
(15 Lectures)

Unit-2

8051 microcontroller:

Introduction and block diagram of 8051 microcontroller, Overview of 8051 family, comparison between Microprocessor and Microcontroller. Pin diagram of 8051, I/O port pins description and their functions. Architecture of 8051 – Block diagram, Oscillator and clock, Program Counter and Data Pointer, A & B CPU registers, flags and PSW, Internal memory, Internal RAM, Internal ROM, Stack and Stack pointer, Special function registers and their addresses. Timers & Interrupts.
(15 Lectures)

Unit-3

8051 Programming:

8051 Addressing modes – Immediate, Register, Direct and Indirect addressing modes. Indexed addressing mode- MOVC & MOVX instructions.

Arithmetic instructions – Incrementing and Decrementing, Addition, Subtraction, Multiplication and Division, Decimal Arithmetic.

Logic Instructions – Byte level and Bit level logical instructions, Rotate and Swap instructions (15 Lectures)

Unit-4

JUMP, LOOP & CALL instructions - Loop and Jump instructions, Call instructions.

8051 interfacing with 8255 – Programming 8255, mode selection of 8255, DAC interfacing

Introduction to embedded system:

Embedded systems and general purpose computer systems. Architecture of embedded system. Classifications, applications of embedded systems. (15 Lectures)

Reference Books:

- Microprocessor Architecture Programming & applications with 8085, 2002, R.S.Goankar, Prentice Hall.
- Embedded Systems: Architecture, Programming & Design, Raj Kamal, 2008, Tata McGraw Hill.
- The 8051 Microcontroller and Embedded Systems Using Assembly and C, M.A.Mazidi, J.G. Mazidi, and R.D. McKinlay, 2nd Ed., 2007, Pearson Education India.
- Microprocessor and Microcontrollers, N. Senthil Kumar, 2010, Oxford University Press.
- 8051 microcontrollers, Satish Shah, 2010, Oxford University Press.
- Embedded Systems: Design & applications, S.F. Barrett, 2008, Pearson Education India.
- Introduction to embedded system, K.V. Shibu, 1st edition, 2009, McGraw Hill.
- Embedded Microcomputer systems: Real time interfacing, J.W. Valvano 2011, Cengage Learning

PRACTICAL IV
MICROPROCESSOR AND MICROCONTROLLER LAB

Section-A: Programs using 8085 Microprocessor

1. Addition and subtraction of numbers using direct addressing mode.
2. Addition and subtraction of numbers using indirect addressing mode.
3. Multiplication by repeated addition.
4. Division by repeated subtraction.
5. Finding Largest among a group of numbers.
6. Finding Smallest among a group of numbers.
7. Arranging in an ascending order of a group of numbers.
8. Arranging in descending order of a group of numbers
9. Other programs (e.g. Parity Check, etc.).

Section-B: Experiments using 8051 microcontroller:

1. Binary addition , subtraction , multiplication and division
2. 8 bits multiplication and division.
3. Fibonacci series.
4. Average of a number.
5. Square and Square root of a number.
6. Palindrome.
7. BCD to Binary conversion.
8. Finding the smallest and largest numbers from the given N binary numbers.
9. To find that the given numbers is prime or not.
10. To find the factorial of a number.
11. Write a program to make the two numbers equal by increasing the smallest number and decreasing the largest number.
12. Use one of the four ports of 8051 for O/P interfaced to eight LED's. Simulate binary counter (8 bit) on LED's. 5. Program to glow the first four LEDs then next four using TIMER application.
13. Program to rotate the contents of the accumulator first right and then left.
14. Program to run a countdown from 9-0 in the seven segment LED display.

15. To interface seven segment LED display with 8051 microcontroller and display 'HELP' in the seven segment LED display.
16. To toggle '1234' as '1324' in the seven segment LED display.
17. To generate different types of wave forms using DAC interface
18. Application of embedded systems: Temperature measurement & display on LCD

Note: At least 04 experiments from section A and B to be done

DSE 1A: DME26004

SEMESTER V DIGITAL SIGNAL PROCESSING

Credits: Theory – 04

Practical – 02

Theory: 60 Lectures

COURSE OUTCOME:

After completion of the course the student is able to

CO1: Identify the details of discrete/digital signals and systems

CO2: Understand the classification and characteristics of frequency domain analysis of discrete time signals.

CO3: Specify with examples DSP filters

Unit-1

Discrete-Time Signals and Systems:

Classification of Signals: Continuous-time and Discrete-Time signals, Analog and Digital signals, Periodic and Aperiodic Signals, Energy and Power Signals, Even and Odd Signals.

Operations on Dependent and Independent Variables. Continuous-Time System and Discrete-Time Systems: Unit Step Response, Impulse Response and Ramp Response. Exponential signals, exponentially damped sinusoidal signals, pulse signals

Basic Discrete Time signals – Step function, Impulse function, Ramp function, Exponential function, Sinusoidal signals, Exponentially damped sinusoidal signals, Pulse signals

System Properties: Linearity, Causality; Time-Invariance, Memory, Stability and Invertibility.

Linear Time Invariant Systems- Convolution Sum. Convolution Integral, Representation for LTI System – two systems in Parallel, Two systems in Cascade. (15 Lectures)

UNIT 2:

The Z-Transform:

Introduction, Definition of Z – transform, Transform and ROC of Finite Duration Sequences – Right Sided Sequence, Left Sided Sequence, Double Sided Sequences.

Z – Transform and ROC of Infinite Duration Sequences- Positive Time Exponential Sequence, Negative Time Exponential Sequence, Double sided Exponential Sequence, ROC & Stability, Properties of ROC . Problems.

Properties of Z-Transform-Linearity, Time Shift or Translation, Multiplication by an Exponential, Multiplication by a Ramp, Convolution (Time-domain), Time-reversal, Convolution (Z-domain), Initial-value theorem, Final value theorem. Transforms of some useful Sequences.

Inverse Z- transform-Partial fraction expansion method, Inverse Z- transform by long division method.

UNIT 3:

Discrete-Time Fourier Transform:

DFT - Introduction. Definition of DFT and IDFT- Lemma, Inverse DFT, Periodicity of $X(k)$ and $x(n)$.

Matrix relation for computing DFT, Matrix relation for Computing IDFT, Using DFT to find IDFT, Concept circular shift and circular symmetry, Properties of DFT – Linearity, Circular time shift, Circular frequency shift.

Fast Fourier Transform (FFT) – Decimation-in-time FFT, Further reduction: Cooley-Turkey Algorithm, In-place computations, Decimation in frequency FFT. (15 Lectures)

UNIT 4:

Filter Concepts: Linear-Phase Filter, Simple FIR Digital Filters, Simple IIR Digital Filters, All pass Filters

Design of IIR Filters:

Introduction, Analog Filter Specifications, Classification of Analog Filters, Butterworth Filters, Frequency transformations / Spectral Transformations, Design of Low pass Butterworth Filters, Digital filters, Backward difference method, Bilinear Transformations, Analog Design using Digital filters, Advantages and Disadvantages of IIR Filters. (15 Lectures)

Reference Books:

- Digital Signal Processing, Tarun Kumar Rawat, 2015, Oxford University Press, India.
- Digital Signal Processing, S. K. Mitra, McGraw Hill, India.
- Principles of Signal Processing and Linear Systems, B.P. Lathi, 2009, 1st Edn. Oxford University Press.
- Fundamentals of Digital Signal processing using MATLAB, R.J. Schilling and S.L.Harris, 2005, Cengage Learning.
- Fundamentals of signals and systems, P.D. Cha and J.I. Molinder, 2007, Cambridge University Press.
- Digital Signal Processing Principles Algorithm & Applications, J.G. Proakis and D.G. Manolakis, 2007, 4th Edn., Prentice Hall.

**PRACTICAL V
DIGITAL SIGNAL PROCESSING**

1. Verification of properties of a system : linear and convolution
2. Finding DFT of a given sequence - direct method
3. Finding IDFT of a given sequence - direct method
4. Linear convolution using DFT
5. Circular convolution using DFT
6. Solution of simple difference equations
7. Verification of sampling theorem
8. Determination of impulse response of a given system
9. Determination of Frequency response of system to any arbitrary input
10. Design of simple IIR filters – Butterworth
11. DFT using FFT
12. IDFT using FFT

(Note: Minimum of Eight experiments to be done).

ELECTRONIC INSTRUMENTATION

Credits: Theory – 04

Practical – 02

Theory: 60 Lectures

COURSE OUTCOME:

After completion of the course a student is able to

CO1: Specify in detail, basic electronic measurement instruments.

CO2: Write down the classification and working of impedance measuring instruments.

CO3: Specify the details of power supply.

CO4: Understand the working and applications of transducers

Unit 1:

Measurements

Accuracy and precision. Significant figures. Error and uncertainty analysis. Shielding and grounding. Electromagnetic Interference.

Basic Measurement Instruments:

DC measurement-ammeter, voltmeter, ohm meter, AC measurement, Digital voltmeter systems (integrating and non-integrating). Digital Multimeter; Block diagram principle of measurement of I, V, C. Accuracy and resolution of measurement.

Measurement of Impedance-

A.C. bridges, Measurement of Self Inductance (Anderson's bridge), Measurement of Capacitance (De Sauty's bridge), Measurement of frequency (Wien's bridge).

(15 Lectures)

Unit 2:

Power supply

Block Diagram of a Power Supply, Qualitative idea of C and L Filters. IC Regulators (78XX and 79XX), Line and load regulation, Short circuit protection. Idea of switched mode power supply (SMPS) and uninterrupted power supply (UPS).

Oscilloscope:

Block Diagram, CRT, Vertical Deflection, Horizontal Deflection. Screens for CRT, Oscilloscope probes, measurement of voltage, frequency and phase by Oscilloscope. Digital Storage Oscilloscopes. LCD display for instruments.

(15 Lectures)

Unit 3:

Lock-in-amplifier:

Basic Principles of phase locked loop (PLL), Phase detector (XOR & edge triggered), Voltage Controlled Oscillator (Basics, varactor), lock and capture. Basic idea of PLL IC (565 or 4046). Lock-in-amplifier, Idea of techniques for sum and averaging of signals.

Signal Generators:

Function generator, Pulse Generator, (Qualitative only).

Virtual Instrumentation:

Introduction, Interfacing techniques (RS 232, GPIB, USB), Idea about Audrino microcontroller and interfacing software like lab View). (15 Lectures)

Unit 4:

Transducers:

Classification of transducers, Basic requirement/characteristics of transducers, Active and Passive transducers, Resistive (Potentiometer- Theory, temperature compensation & applications), Capacitive (variable air gap type), Inductive (LVDT) & piezoelectric transducers. Measurement of temperature (RTD, semiconductor IC sensors), Light transducers (photo resistors & photovoltaic cells). (15 Lectures)

Reference Books:

- W.D. Cooper and A. D. Helfrick, Electronic Instrumentation and Measurement Techniques, Prentice Hall (2005).
- E.O. Doebelin, Measurement Systems: Application and Design, McGraw Hill Book - fifth Edition (2003).
- David A. Bell, Electronic Devices and Circuits, Oxford University Press (2015).
- Alan S. Morris, "Measurement and Instrumentation Principles", Elsevier (Butterworth Heinmann-2008).
- S. Rangan, G. R. Sarma and V. S. Mani, Instrumentation Devices and Systems, Tata Mcgraw Hill (1998).
- Introduction to measurements and instrumentation, 4th Edn., Ghosh, PHI Learning.

**PRACTICAL VI
ELECTRONIC INSTRUMENTATION LAB**

1. Measurement of resistance by Wheatstone bridge and measurement of bridge sensitivity.
2. Measurement of Capacitance by De Sauty's bridge.
3. To determine the Characteristics of resistance transducer - Strain Gauge (Measurement of Strain using half and full bridge).
4. To determine the Characteristics of LVDT.
5. To determine the Characteristics of Thermistors and RTD.
6. Measurement of temperature by Thermocouples.
7. Design a regulated power supply of given rating (5 V or 9V).
8. To design and study the Sample and Hold Circuit.
9. To plot the frequency response of a microphone.

(Note: Minimum of Eight experiments to be done).

**SEMESTER VI
VERILOG & VHDL**

Credits: Theory – 04,
Practical – 02

Theory: 60 Lectures

COURSE OUTCOME:

After completion of the course the student is able to

CO1: Identify the details of Digital logic design flow.

CO2: Learn the characteristics and model the digital circuits using VHDL behavioural modelling

CO3: Deliberate in detail the dataflow and structural modelling in VHDL

CO4: Describe digital circuits utilizing various constructs of Verilog

Unit-1

Digital logic design flow

Review of combinational circuits. Combinational building blocks: multiplexers, demultiplexer, decoders, encoders and adder circuits. Review of sequential circuit elements: flip-flop, latch and register. Finite state machines: Mealy and Moore. Other sequential circuits: shift registers and counter (15 lectures)

Unit-2

Verilog HDL:

Introduction to HDL. Verilog primitive operators and structural Verilog. Behavioural Verilog. Design verification. Modelling of combinational and sequential circuits (including FSM and FSMD) with Verilog Design examples in Verilog. (15 lectures)

Unit-3

VHDL: Basic Language elements

Identifiers, Data Objects, Data types, Operators

Behavioral Modelling

Entity Declaration, Architecture Body, Process statement, Variable assignment statement, Signal Assignment, Wait statement, If statement, Case statement, Null statement, Loop statement, Exit

statement, Next statement, Assertion statement, Report statement, More on Signal Assignment statement, other sequential statements. (15 Lectures)

Unit-4

Dataflow Modelling

Concurrent Signal Assignment, Concurrent versus Sequential Signal Assignment, Delta Delay Revisited, Multiple Drivers, Conditional Signal Assignment Statement, Selected Signal Assignment Statement, The UNAFFECTED value, Block Statement, Concurrent Assertion Statement, Value of signal.

Structural Modeling

An example, Component Declaration, Component Instantiation, Other examples. Resolving Signal Values (15 lectures)

Reference Books:

1. LizyKurien and Charles Roth. *Principles of Digital Systems Design and VHDL*. Cengage Publishing. ISBN-13: 978-8131505748.
2. Palnitkar, Samir, *Verilog HDL*. Pearson Education; Second edition (2003).
3. Ming-Bo Lin. *Digital System Designs and Practices: Using Verilog HDL and FPGAs*. Wiley India Pvt Ltd. ISBN-13: 978-8126536948.
4. Zainalabedin Navabi. *Verilog Digital System Design*. TMH; 2nd edition. ISBN-13: 978-0070252219.
5. Wayne Wolf. *FPGA Based System Design*. Pearson Education. S. K. Mitra, Digital Signal processing, McGraw Hill, 1998.
6. VLSI design, Debaprasad Das, 2nd Edition, 2015, Oxford University Press.
7. D.J. Laja and S. Sapatnekar, Designing Digital Computer Systems with

**PRACTICAL VII
VERILOG AND VHDL LAB**

Experiments using Verilog

1. Write code to realize basic and derived logic gates.
2. Half adder, Full Adder using basic and derived gates.
3. Half Subtractor and Full Subtractor using basic and derived gates.
4. Design and simulation of a 4 bit Adder.
5. Multiplexer (4x1) and Demultiplexer using logic gates.
6. Decoder and Encoder using logic gates.
7. Clocked D, JK and T Flip flops (with Reset inputs).
8. 3-bit Ripple counter

Experiments using VHDL

1. Behavioral modeling and simulation of basic gates
2. Structural modeling and simulation of simple Boolean expression
3. Modeling and simulation of adders and subtractors
4. Modeling and simulation of magnitude comparators
5. Modeling and simulation of Flip-flops
6. Modeling and simulation of Shift registers
7. Modeling and simulation of Counters
8. Modeling and simulation of encoders and decoders
9. Modeling and simulation of multiplexers

Note: At least 04 experiments - each from section A and B to be done.

DSE2B: DMF26204

SEMESTER VI
PHOTONIC DEVICES AND POWER ELECTRONICS

Credits: Theory – 04
Practical – 02

Theory: 60 Lectures

COURSE OUTCOME:

After completion of the course the student is able to

CO1: Deliberate the Principles and operations of Photonic devices.

CO2: Deliberate the Principle of operation and characteristics of optical fibers

CO3: Understand the mode of signal prorogation in optical fibres

CO4: Deliberate the characteristics and application of power devices

UNIT 1:

Photonic Devices

Classification of photonic devices. Interaction of radiation and matter, Radiative transition and optical absorption. Light Emitting Diodes- Construction, materials and operation. Semiconductor Laser- Condition for amplification, laser cavity, heterostructure and quantum well devices. Charge carrier and photon confinement, line shape function. Threshold current. Laser diode. Photodetectors: Photoconductor. Photodiodes (p-i-n, avalanche) and Photo transistors, quantum efficiency and responsivity. Photomultiplier tube. (15 Lectures)

UNIT 2:

Solar Cell

Construction, working and characteristics LCD Displays: Types of liquid crystals, Principle of Liquid Crystal Displays, applications, advantages over LED displays.

Introduction to Fiber Optics:

Evolution of fiber optic system- Element of an Optical Fiber Transmission link- Ray Optics- Optical Fiber Modes and Configurations –Mode theory of Circular Wave guides- Overview of Modes-Key Modal concepts- Linearly Polarized Modes -Single Mode Fibers-Graded Index fiber structure. (15 Lectures)

UNIT 3:

POWER ELECTRONICS

Power Devices:

Need for semiconductor power devices, Power MOSFET (Qualitative). Introduction to family of thyristors. Silicon Controlled Rectifier (SCR)- structure, I-V characteristics, Turn-On and Turn-Off characteristics, ratings, Gate-triggering circuits. Diac and Triac- Basic structure, working and V-I characteristics. Application of Diac as a triggering device for Triac. (15 Lectures)

UNIT 4:

Insulated Gate Bipolar Transistors (IGBT):

Basic structure, I-V Characteristics, switching characteristics, device limitations and safe operating area (SOA).

Applications of SCR:

Phase controlled rectification, AC voltage control using SCR and Triac as a switch. Power Invertors- Need for commutating circuits and their various types, dc link invertors, Parallel capacitor commutated invertors, Series Invertor, limitations and its improved versions, bridge invertors. (15 Lectures)

Reference Books:

1. J. Wilson & J.F.B. Hawkes, Optoelectronics: An Introduction, Prentice Hall India (1996).
2. S.O. Kasap, Optoelectronics & Photonics, Pearson Education (2009).
3. AK Ghatak & K Thyagarajan, Introduction to fiber optics, Cambridge Univ. Press (1998).
4. Power Electronics, P.C. Sen, Tata McGraw Hill.
5. Power Electronics, M.D. Singh & K.B. Khanchandani, Tata McGraw Hill.
6. Power Electronics Circuits, Devices & Applications, 3rd Edn., M.H. Rashid, Pearson Education.
7. Optoelectronic Devices and Systems, Gupta, 2nd edn., PHI learning.
8. Electronic Devices and Circuits, David A. Bell, 2015, Oxford University Press.

PRACTICAL VIII
PHOTONIC DEVICES AND POWER ELECTRONICS

1. To determine wavelength of sodium light using Michelson's Interferometer.
2. Diffraction experiments using a laser.
3. Study of Electro-optic Effect.
4. To determine characteristics of (a) LEDs, (b) Photo voltaic cell and (c) Photo diode.
5. To study the Characteristics of LDR and Photodiode with (i) Variable Illumination intensity, and (ii) Linear Displacement of source.
6. To measure the numerical aperture of an optical fiber.
7. Output and transfer characteristics of a power MOSFET.
8. Study of I-V characteristics of SCR.
9. SCR as a half wave and full wave rectifiers with R and RL loads.
10. AC voltage controller using TRIAC with UJT triggering.
11. Study of I-V characteristics of DIAC
12. Study of I-V characteristics of TRIAC.

SEC1A: DME26404

ELECTRICAL CIRCUITS AND NETWORK SKILLS

Credits: 02

Theory: 30 Lectures

Course Outcome:

After completion of the course the student acquires skill to

CO1: Design and trouble shoot the electrical circuits and networks

CO2: Carry-out simple domestic wiring.

UNIT 1:

Basic Electricity Principles:

Discussion of Voltage (AC & DC), Current(AC & DC), Resistance, and Power. Ohm's law. Series, parallel, Series and Parallel combinations of R, L and C. Response of inductors and capacitors with DC or AC sources. Impedance in of Inductor and Capacitor

Electrical Circuits:

DC Circuits - Basic electric circuit elements and their combination in DC circuits. Rules to analyze DC sourced electrical circuits. Current and voltage drop across the DC circuit elements.

AC Circuits - . Simple numericals on network theorms. Real, imaginary and complex power components of AC source. Power factor. Saving energy and money.

Electrical Drawing and Symbols:

Drawing symbols. Blueprints. Reading Schematics. Ladder diagrams. Electrical Schematics. Power circuits. Control circuits. Reading of circuit schematics. Tracking the connections of elements and identify current flow and voltage drop. (15 Lectures)

UNIT 2:

Generators and Transformers:

DC Power sources. AC/DC generators. Basic principle of operation, constructional features. Transformers – Principle of working, Construction and Operation of transformers.

Electric Motors:

Single-phase, three-phase & DC motors - Construction and Working. Speed & power of ac motor. Interfacing DC or AC sources/ Motors to control heaters

Solid-State Devices: Diodes, types of diodes –symbol and applications, Rectifiers - PN junction diode as rectifier (Half wave and Full wave rectifier) construction and working

Electrical Protection:

Relays - Relay as protection device, Fuses and disconnect switches. Circuit breakers. Overload devices. Ground-fault protection. Grounding and isolating. Phase reversal. Surge protection.

Electrical Wiring:

Different types of conductors and cables. Basics of wiring-Star and delta connection. Voltage drop and losses across cables and conductors. Instruments to measure current, voltage, power in DC and AC circuits. Insulation. Solid and stranded cable. Conduit. Cable trays. Splices: wirenuts, crimps, terminal blocks, and solder. Preparation of extension board. (15 Lectures)

Reference Books:

1. Electrical Circuits, K.A. Smith and R.E. Alley, 2014, Cambridge University Press.
2. A text book in Electrical Technology - B L Theraja - S Chand & Co.
3. A text book of Electrical Technology - A K Theraja.
4. Performance and design of AC machines - M G Say ELBS Edn.

<https://drive.google.com/file/d/1qzea3KMY6yomll0MIzZpsGnMQe1zCWtp/view?usp=sharing>

SEC1B: DME26604

COMPUTER NETWORKS

Credits: 02

Theory: 30 Lectures

COURSE OUTCOME:

After completion of the course the student acquires skill to

CO1: Understand the concepts of network devices

CO2: Understand the terminology and concepts of the OSI model

Unit 1:

Data communication, Components & Basic Concepts

Line configuration- point-to-point, multipoint, Topology – Mesh, Star, Tree, Bus, Ring, and Hybrid Topologies Transmission modes – Simplex, Half Duplex, Full Duplex. Categories of networks – LAN, MAN, WAN, Internet

Transmission Media

Guided media – Twisted pair cable, Co-axial cable, Optical fiber

Multiplexing:

Many to one/one to many, types of multiplexing, Frequency division multiplexing, time division multiplexing, multiplexing applications

Error detection

Types of error, multiple bit error, Burst error, Detection – redundancy, Checksum Error

correction – Single bit error correction, Hamming code (15 Lectures)

Unit 2

The OSI Model

Model – layered Architecture, Functions of layers- physical layer, Data link layer, Network layer, Transport layer, Session layer, Presentation Layer, Application layer

Networking and internetworking devices

Repeaters, Bridges- types of Bridges, Routers- Routing concepts, Gate ways

World Wide Web:

Uniform Resource Locator (URL), Browser Architect (15 Lectures)

Text Book:

Introduction to Data Communications & Networking by- BEHROUZ FOROUZAN

Reference Book:

Computer Networks by – ANDREW S TANENBAUM

Pattern of theory Question Paper for DSC/DSE

From the academic year 2019-2020 onwards

Time: 3 hours

Max. Marks: 70

Credits: 4

Part - A

- I. Answer all questions. 1 x 10 = 10**
- Ten questions to be set from the four units of the syllabus.
 - Minimum of two questions to be set from each unit.
 - Questions can be simple problems also.

Part - B

- II. Answer any Four questions. 4 x 5 = 20**
- **Six** questions to be set from four units of the syllabus.
 - Minimum of one question to be set from each unit,
 - This section can have questions / problems.

Part - C

- III. Answer any Four questions 4 x 10 = 40**
- **Six** questions to be set from four units of the syllabus.
 - Minimum of one question to be set from each unit.
 - Questions can have subdivisions.

Model Question Paper for DSC/DSE

Time: 3 hours

Max. Marks: 70

Part -A

1. Answer all questions.

1 x 10 = 10m

- (a)
- (b)
- (c)
- (d)
- (e)
- (f)
- (g)
- (h)
- (i)
- (j)

Part - B

Answer any Four questions.

4 x 5 = 20

- 2.**
- 3.**
- 4.**
- 5.**
- 6.**
- 7.**

Part - C

Answer any Four questions

4 x 10 = 40

- 8.**
- 9.**
- 10.**
- 11.**
- 12.**
- 13.**

Pattern of theory Question Paper for SEC

Time: 2 hours

Max. Marks: 35

Credits: 2

Part –A

I. Answer all questions. 1 x 05 =05m

- Five questions to be set from the two units of the syllabus.
- Minimum of two questions to be set from each unit.
- Questions can be simple problems also.

Part –B

II. Answer any Three questions 3x 10 = 30m

- Five questions to be set from the two units of the syllabus.
- Minimum Two questions/ Problems to be set from each unit.

Model Question Paper for SEC

From the academic year 2019-2020 onwards

Time: 2 hours

Max. Marks: 35

1. Answer all questions.

5 x 1 = 05m

- (a)
- (b)
- (c)
- (d)
- (e)

Part - B

Answer any Three questions.

3 x 10 = 30m

- 2.**
- 3.**
- 4.**
- 5.**
- 6.**

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DEPARTMENT OF JOURNALISM
CBCS SYLLABUS

Revised Syllabus for Undergraduate (UG) Journalism
with CBCS scheme Effective from 2019-20

Bachelor of Arts in (JPE) JOURNALISM, POL-SCIENCE AND ENGLISH

1

Programme Outcomes for Bachelor of Arts in Journalism, Pol-Science and English

After completing the graduation in the B A (JPE) the students are able to:

- P01. Acquire a functional knowledge of the underlying principles and recent emerging trends of the media industry.
- P02. Create and design emerging audio media products
- P03. Conceptualize, create, design, and strategies, high-quality media content for various digital platforms.
- PO4. Appreciate and demonstrate the ability to produce reliable outcome.
- P05. Demonstrate critical reading, writing, and thinking skills.
- P06. Locate, evaluate, organize and incorporate information effectively
- P07. Develop and carry out research projects
- P08. Demonstrate competence in Standard English Language and usage in documentation

Programme Specific Outcomes

Journalism, Political science and English PPE)

On Completion of BA (JPE) students are able to

- PSO1: Gather, analyze and create journalism on contemporary issues for print, broadcast and other digital media.
- PSO2: Produce work incorporating diverse groups and perspectives on local, national and international issues.
- PSO3: Demonstrate an awareness of journalism as an ethical practice
- PSO4: Write and edit clear, graceful, grammatically correct prose
- PSO5: Analyze numerical data and utilize databases for multi-layered storytelling
- PSO6: Demonstrate preparation for an entry-level position in the profession through a portfolio exhibiting their work.
- PSO7: Report day to day activities effectively in written and oral pattern
- PSO8: Draft reports and other written matter effectively

BA PROGRAMME IN JOURNALISM

2019-20

JPE BA 25

Semester	Course	Core	Course CODE	Credits L:T:P	Total Credits	Theory marks	IA Marks	Total marks
First	Introduction to Communication and Journalism	DS C	ELA2702 5	5:1:0	06	70	30	100
Second	Introduction to Broadcasting Media	DS C	ELA2702 5	5:1:0	06	70	30	100
Third	Media Industry and Management	DS C	ELA2702 5	5:1:0	06	70	30	100
Four	Media Law and ethics	DS C	ELA2702 5	5:1:0	06	70	30	100
Fifth	1. Reporting and Editing Techniques	DS C	ELA2702 5	5:1:0	07	70	30	100
	2. Indian Applied Journalism	DS C	ELA2707 2	5:1:0	06	70	30	100
	3. Introduction to TV and Radio	DS C	ELA2707 4	5:1:0	06	70	30	100
Sixth	1. Practices of Advertising & PR	DS C	ELA2702 5	5:1:0	07	70	30	100
	2. Cinema Journalism	DS C	ELA2707 2	5:1:0	06	70	30	100
	3. Introduction to New Media	DS C	ELA2707 4	5:1:0	06	70	30	100
GENERIC ELECTIVE PAPERS								
Fifth Semester	1. Media Gender and Human Rights 2. Documentary Production 3. Documentary production	GE	ELE276 ELE277 ELE278	2:0:0	02	35	15	50
Sixth Semester	1. Basis of Journalism 2. Development Journalism 3. Documentary Production	GE	ELE276 ELE277 ELE278	2:0:0	02	35	15	50
Total	Six+ One GE =Seven Papers			35+4	39			700

Semester: 1

Course Code – ELA27025

Introduction to Communication and Journalism

Course Out comes:

On completion of the Course, students are able to:

- CO: 1 Become an -Event anchors
- CO: 2 Become Freelance journalist
- CO: 3 Stringers for several media houses
- CO: 4 Assistant for yellow journalist
- CO: S to set up the commercial studio CO:
- 6 Become armature photo journalists

Unit: 1 Origin, Meaning, Definition, Nature and Scope of Communication- Functions of Communication and Types of Communication- principles of Communication

Unit: 2 Processes of Communication- Models of Communication- Aristotle, Shonnon and Weaver, Wilber Schram andBerlo Models

Unit: 3 Origin, meaning, definitions of journalism- functions of journalism- press theories- branches of journalism- Yellow, Penny Press, Electronic, online, citizen and community journalism

Unit: 4 Role of Press in Democracy- Ethics of journalism- Principles of journalism —press and Society

Students practical:

1. Contemporary debates on issues relating to media
2. Maintain records on LIE, Articles, features and interviews

References:

1. Mass Communication by Koval J. Kumar
2. Professional Journalism by M.V.Kamath
3. Theory and Practices of Journalism by 13.1\T. Abuja
4. Theories of Mass Communication by Daniel Learner
5. Communication and Development by Everet M. Rogers

Semester:2

ELB27025

Introduction to Broadcast Media

Course Out comes:

On completion of the Course, students are able to:

CO: 1 Become an -Event anchors

CO: 2 Become an audio and video editors

CO 3 Copy writers/editors for broadcasting media

CO: 4 Assistant for electronic journalists CO: 5 to set up the studio/ lab

CO: 6 Become armature journalists

Unit I - Basics of Sound - Concepts of sound- sound culture- Types of sound- Sound Design- Sound recording techniques- types of microphones- Characteristics of Radio.

Unit 2 - Basics of Visual- meaning of image- types of images Characteristics of TV.

Unit 3 - Writing and Editing Radio News- Preparing Radio News Bulletins- Radio News Room- Nature of Recording and editing sound.

Unit 4 - Writing and Editing TV News- Basics of a Camera- types of Lens & accessories-ENG & Electronic field Production (EFP)- Types of Shots- formats of TV news- preparing Television News Bulletins- Basics of Editing for TV- Basic Soft-wares and Techniques.

Practical & Documentaries-

- 1 . Preparing News Bulletins for ten minutes
2. Documentary- 'The future of Television News.'

References:

1. Television Production Handbook by Zettl Herbert
2. The Television Reader by Robert Allen and Annette Hill
3. Broadcasting in India by P.O Chattejee.
4. The Radio Handbook by Carrot Fleming

Semester:3
Code: ELB27025

Media Industry and Management

Course Out comes:

On completion of the Course, students are able to:

CO: 1 Become an owner of the media house

CO: 2 Become an Administrator or CEO; 3

Become as TRP agent

CO: 4 Assistant for media managing Directors

CO: 5 to set up the newspaper industry

Unit: 1 Concept of Management- Origin and growth of Media Management -
Fundamentals of management - Management School of Thought

Unit: 2 Media Industry & Challenges • Media industry as manufacturers- news and I!
content management.- Market Forces- performance evaluation TRP, BARC
and HITS - Changing Ownership patterns

Unit 3. Structure of media organizations — Role and responsibilities - Workflow &
Shift Patterns- Guidelines- Qualities and Functions of media managers.

Unit 4. Media Economics- Strategic Management and Marketing - Capital inflow,-
Budgeting- Financial management-personnel Management- Strategic
Management,-Market forces

Reference

1. Indian Media Business by Vinita Kohli Khandeka,
2. Political Economy of Communications in India by Pradip Ninan Thomas,
3. Strategic management in media by Lucy Kung
4. Media Management in the age of Giant by Dennis F. Herrick
5. Media industries-History, Theory and Method , by Jennifer Holt and Alisa Perron,, (Edited)

Semester: 4
Code:ELD27025
Media Ethics and Law

Course Out comes:On completion of the Course, students are able to

- CO: I develop live telecasting skills
- CO: 2 provide online journalism services
- CO: 3 establish own publication
- CO: 4 develop social activities skills

Unit: 1-Press commissions- PCI- media ethics- Freedom of expression-Article 19(1)A and 19 (1)2, Case study of Romesh Thaper

Unit: 2- Ethics for live telecasting and sting operations-ethical issues in social media-media violence and laws

Unit: 3-Broadcasting Bill-NBA guidelines-Media and culture-pornography and women prohibition act-1986

Unit: 4- Official Secrecy act-law of defamation-copy right act-RTI-2005, IT act, Intellectual property rights-cyber laws-contempt of court

BOOK REFERENCE-

Mass Media laws and regulations- Venkat Iyer
Freedom of the Press- M.K. Joseph
Law of Press in India-D.D.Basu
Press and the law-A.W.Grover
Legal limits to journalism-A.G.Noorani

Semester: 5

Code: ELE 27025

Paper: 1 Reporting and Editing Techniques

Course Out comes:

On completion of the Course, students are able to

CO: 1 Develop skills for news writing

CO: 2 Prepare news copy

CO: 3 sub editing the news materials

CO: 4 translate the news

CO: 5 specialize as fashion reporter.

Unit: 1- Role of Reporter, functions and qualities of reporter. News agencies- types of reporting- press conferences

Unit: 2- Types of leads, interviews, article writing, types of features

Unit:3-Organisational pattern of a news paper-set up of a news room, editorial department, editing-principles of editing, style sheet, jargons, importance of picture, selection of picture, headlines-functions and types o headlines. Design of editorial page, letters to the editors, contents of editorial page.

Unit: 4- Trends in sectional news-supplements-columns and columnists

Unit: 5- Paid news-agenda setting-gate keepers-pressures in news room.

Practical:

1. Internal Assessment
2. Exercise of Record books
3. Discussions on contemporary issues.

References:

1. Baskette and Scissors- The Art of Editing
2. S.N. Chaturvedi- Dynamics of Journalism an Art of Editing
3. Bruce Mille and Dougals Andrson- News Writing and Reporting for Today's Media
4. F.W. Hodson- Modern News paper Practices
5. Fedler Fred- Reporting for Print Media

Semester: 5
Code: ELE 27025

Paper: 2 Introductions to TV and Radio

Course Out comes:

On completion of the Course, students are able to

CO: 1 become news reporters for electronic media

CO: 2 become audio/video editor

CO: 3 develop cinema shooting and script writing skills

CO: 4 become radio/video jockey.

Unit:1- Origin and growth of radio-characteristics of radio, structure of AIR, FM and community stations-broadcast format-stages of radio productions

Unit:2- Writing for radio-broadcast formats-use of voice-sound effects-KR-studio-recording and mixing-sound editing

Unit:3- Origin and growth of TV-characteristics of tv media, types of tv programmes-scripts for TV-studio set up

Unit:4- Types of camera-uses of different lenses-uses of archives-editing skills-mixing and dubbing

References:

1. Mass Communicat on-Keval J. Kumar
2. Radio and TV journalism- K.M.Srivastava
3. Broadcast journalism-itas basic principles-S.C. Bhatt
4. Audio-Visual journalism-B.N.Ahuja
5. TV media and Radio Voice-Dr.Nagendra

Semester: 5
Code: ELE 27025
Paper: 3 Indian Applied Journalism

Course Out comes:

On completion of the Course, students are able to

- CO: 1 become news reporters
- CO: 2 become circulation manager
- CO: 3 become script writer
- CO: 4 become stringer

Unit: 1 Evolution of printing- history of Indian Printing press- William Bolts- Bengal Gazette- J.A. Hickey

Unit: 2 Birth of Indian journalism- James Silk Buckingham- Raja Ram Mohan Roy- Balagangadhar Tilak

Unit: 3 Freedom Movement and Press- Gandhi, Nehru, Annie Besant and Ambedkar as pioneers of Indian journalism

Unit: 4 Brief history of English Publications- the Times of India- the Hindu- the Indian Express- Hindustan Times.

Students Presentations:

1. Cartoons on Indian press during the freedom movement- Indo-Pak divide and emergency rule.
2. Growth of regional press- Samyukta Karnataka- Mohare Hanumatha Roy- D.V. Gundappa- Patil Puttappa- Hermon Mongling-Mangalore Samachar- Amrutha Bazara Patrike- Free Press Journal

References:

1. History of Indian Journalism by Nadig Krishnamurthy
2. Journalism in India by Parthasarthi Rangaswamy
3. Social History of Media from Gutenberg to the Internet by Briggs A and Burke
4. History of Press laws and Communication by B.N. Abuja
5. Indian journalism by K.M.Srivatsava

Semester: 6

Code: ELF 27025

Paper: I. Practice of Advertising and Public Relations

Course Out comes:

On completion of the Course, students are able to

CO: I set up advertising agency

CO: 2 prepare the advertising copy for print

CO: 3 become Script writer-marketing research

CO: 4 become PRO and event campaigner

Unit: 1- Introduction to Advertising, meaning, history and functions of Advertising-Theories, models-Apex bodies Like-AA-AI and ASCI

Unit: 2- Types of Advertising and new trends, ethics and tools for advertising, types of media for advertising, Advertising department and Ad agency structure.

Unit: 3- Introduction to PR, Definition, growth of PR, functions, principles and tools of PR, structure of PR department

Unit: 4- Types of PR, corporate communication, apex bodies like-IPRA, PRSI, PSPF and their codes.

References:

1. Advertising made simple- Frank Jefkins
2. Advertising Theory and Practices- Chunawalla •
3. Public Relations Techniques- Jefkins Frank Butterworth
4. Hand book of Public Relations- Heath Robert
5. Effective Public Relations- Cutlip S.M. and Center A.H.
6. Public Relations in India- Kaul El/ and Noya Prakash

Semester: 6

Code: Code: ELF 27025

Paper: 2 Introduction to new Media

Course Out comes:

On completion of the Course, students arc able to

CO: 1 become as Video Journalist

CO: 2 become cinema story writer

CO: 3 become Audio and video editor

CO: 4 become sound mixer and film maker.

CO: 5 become digital content editor

Unit: 1- Definition of new media, Basics- information society and new media- virtual culture and digital journalism-online communities

Unit:2- cyber security-privacy, new media ethics- web writing-linear and non-writing- visual and content. design-website planning and visual design

Unit:3- Original and growth of blogging, creating a blog, video blog, photo blog, and micro blog, video, podcast

Unit:4- User general content and web 2.0-digital art scripting and screening-cyber security

References: 1 Understanding digital culture- Vincent Miller

2. What is new media?-Lev -Manovich

3. Understanding new media- Siapera Eugenia

4. Who Controls the Internet- Goldsmith Jack and Tim W

5. New media and new technologies- Lister Covey and Griddings Grant

Semester: 6

Code: Code: ELF 27025

Paper: 3 Cinema Journalism

Course Out comes:

On completion of the Course, students are able to

CO: 1 work as cinema journalist

CO: 2 work as camera crew

CO: 3 work in cinema industry as assistant

CO: 4 find the job in TV media

CO: 5 work as script writer in cinema

Unit I Language of Cinema - visual Language- Shot, Scene- Montage Language of Cinema
types of Sound- use of Colors- Differences between story, plot and screenplay

Unit II Film Form - German Expressionism - Italian Neorealism- French New Wave- Classical
Hollywood Cinema

Unit III Third Cinema - Non Fiction Cinema - Feminist Film Theory - Film Authorship.

Unit IV Hindi Cinema 1950s Cinema and the Nation- The Indian New-Wave -Globalization
and Indian Cinema- The multiplex Era.

Recommended Screenings:

1. Rear Window by Alfred Hitchcock (Language of Cinema)
2. Battle ship Potempkin by Sergei Eisenstein (Language of Cinema)
3. Man with a Movie Camera by DzigaVertov
4. Germany Year Zero by Roberto Rosselini (Italian Neo Realism)
5. Double Indemnity by Billy Wilder (German Expressionism and Film Noir)
6. . Aakrosh by Govind Nihalani (Indian New wave)

References:

1. "The Ontology of the Photographic Image" by Andre Bazin,
2. From his book What is Cinema by Berkeley,
3. "A Dialectic Approach to Film .Form" by Sergei Eisenstein
4. Film Form: Essays in Film Theory (Edited and Translated) by Jay Leyda)
5. "Classical Hollywood Cinema: Narrational Principles and Procedures" by David Bordwell.

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DEPARTMENT OF JOURNALISM
MODEL QUESTION PAPER
B,AJOURNALISM
I, II, III, IV, V&VI Semester
(CBCS Scheme)

Time: 3 Hours

Max. Marks — 70

Q: I- Write any five questions. Each question carries two marks. 2X5=10

- 1
- 2
- 3
- 4
- 6
- 7

Q: TT - Write any four questions. Each question carries five marks. 5)(4=20

- 1
- 2
- 3
- 4
- 5
- 6

Q: III - Write any four questions. Each question carries ten marks. 10X4=40

Generic Elective Paper
Semester: 5
Code: DLF27215:
Paper: I- Development Journalism

Course Out comes:

On completion of the Course, students are able to

- CO: 1 set up the commercial studio
- CO: 2 become an armature photo journalist
- CO: 3 become visual composer and editor
- CO: 4 set up community radio
- CO: 5 provide cable services

Unit:1- Concept, meaning of development, models of development, evolution of development journalism-communication for social change

Unit:2- Development news and features, writing headline and captions, innovative leads, research and planning- print and broad cast development journalism

Unit:3- Visual and documentation-folk media, community radio for development, niche newspapers and magazines, Development communication agencies and web sites

Unit:4- Mass media as a tool for development, role of development agencies, and NGO, using new media technologies for development-devcom.-SITE, Krishi Darshan, KHeda, cyber media and e-governance

References:

1. 1. Manual of Development Journalism- Alan Chalkley
2. Development Communication and media Debate- Mridala Niel-ion
3. Television and Social change in rural India-Kirk Johnson
4. Everybody loves a good drought- P. Sainath•

Generic Elective Paper

Semester: 5

Code: ELE27025

Paper: 2 - Media Gender and Human Rights

Course Out comes:

On completion of the Course, students are able to

CO: 1 become as Social activist

CO: 2 appear for competitive examinations

CO: 3 know the media impacts on the communities

CO: 4 gain knowledge on media culture

Unit: 1-Media and the Social world- media impact on society- democratic policy and mass media- media and cultural change- grass roots media

Unit:2- Gender conceptual framework- feminist theory- media and gender- media and masculinity-media power- media and social differences- genres- romances- television-soap opera- sports

Unit:3- Media and Human Rights- Theoretical perspectives- universal declaration of human rights-case studies.

References:

1. Street John- Mass media, politics and democracy
2. Balnaves Mark, StephanieDonald and Bria Shoe Smith- Media Theories and Approaches: A global Perspective
3. The media Reader: Continuity, and transformation
- 4, Mc Quill Denis- Mass Communication theory
- 5, Berger Arthur Asia- Media and Society: A critical Perspective
6. Nicholas Joe and Price John- Advanced Studies in Media
7. Rajagopal Arvind- The Indian Public Sphere: Readings in Media History
8. Bannerjee Menon and Priyameds- Human Rights Gender and Environmer

Generic Elective Paper
Semester: 5
Code: ELE2705
Paper: 3 Photo Journalism

Course Out comes:

On completion of the Course, students are able to

CO: 1 set up own studio

CO: 2 work as camera crew in cinema/TV industry

CO: 3 work as photographer in print and electronic media

CO: 4 become a freelance photographer

Unit: 1-History, definition.-birth of camera, modernization of Photography, uses in media, digital photography.

Unit:2- Equipments of photography camera- lenses- tripods- monopods-camera bags- digital storage.

Unit:3- Lighting-types of lighting-Recommended equipments for outdoor — introduction to indoor lighting and photographing

Unit:4- Types of photography- news photography- sports photography- the basics of photojournalism- importance of context in photo journalism-

1

Unit:5- Editing photo- editing softwares-correcting imperfect images-

References:

1. The photography Book by Editors of Phaidon press.
2. All about photography by Ashok Dilwali
3. Practical Photography by O.P. Sharma
4. The photographers Guide to Light by Freeman John Collins and Brown
5. Lonely Planet's Best Ever Photography Tips by Richard I' Anson

Generic Elective Paper
Semester: 6
Code:ELF27025
Paper: 1 Basics of Journalism

Course Out comes:

On completion of the Course, students are able to

CO: 1 set up own publication

CO: 2 work as stringer in local media industry

CO: 3 work as freelancer

CO: 4 become a reporter

Unit 1 —News meaning, definition, nature- Hard news and Soft news -parts of a news

Unit 2 structure of news- SW's and IH-inverted pyramid- sources of news - Principles of clear writing

Unit 3 — Branches of journalism- Yellow- Penny press- tabloid press- electronic - online and Citizen Journalism

Unit 4 - Role of Media in a Democracy- Responsibility to Society -Ethics of journalism

References

1. News writing and reporting for today's media- Bruce D. hide and Douglas A. Anderson. 2. News writer's Handbook- M.L. Stein,Susan Paterno Sc R. Christopher Burnett.
3. An Introduction to Journalism- Blackwell
- 4 Mass Media in a Changing World -George Rodrnann
5. An introduction to Journalism- Carole Flemming and Emma Hemmingway.

Generic Elective Paper
Semester: 6
Code:ELF27025
Paper: 2 Documentary Productions

Course Out comes:

On completion of the Course, students are able to

CO: 1 Become as documentary artist

CO: 2 work as assistant in cinema industry

CO: 3 work as short movie maker

CO: 4 become as visual editor

Unit 1: Understanding the Documentary- Six Modes of Documentary-1.Representation, 2.Participatory, 3.Expository, d. Observational, 5.Performative, 6.Reflexive, -Defining the Subject/Social Actor/Participant Voice in the Documentary- Different Posturing of the Narration, Participant, Filmmaker, & Audience Camcorder.

Unit 2: Pre-Production -Researching the Documentary- Library, Archives, location, life stories, ethnography- Writing concept- telling a story- Writing a Treatment-Proposal and Budgeting- Structure and scripting the documentary- Issues of Funding and Issues of Primary and Secondary Audience

Unit 3: Production Documentary- Sound Documentary- Cinematography —Technologies and Techniques- Shooting Schedule - Call list Production Team- Meetings-Checklist -Crowd Funding

Unit 4: Post Production- Grammar of editing Transitions -Sound Effects -Visual Effects-Aspect Ratio-Language, Duration and marketing of DVD and issues of piracy.

Suggested Practical Exercise: Making a short documentary for 5-10 minutes

Suggested Screenings:

1. Michael Moore: Roger and Me
2. Nanook of the North by Robert J Flaherty
3. Bombay Our City by Anand Patwardhan
4. Black Audio Collective City of Photos by Nishtha Jain

References:

1. "Documentary" in Geoffrey by Charles Musser
2. "The Truth about Non Fiction" and "Towards a Poetics of Documentary" by Michael Renov
3. How to Write a Documentary Double Take by PSBT DOX magazine by Trisha Das
4. Introduction to Documentary by Nichols, Bill

Generic Elective Paper
Semester: 6
Code: ELF2705
Paper: 3 Global Media and Politics

Course Out comes:

On completion of the Course, students are able to

- CO: 1 Become as documentary artist
- CO: 2 work as assistant in cinema industry
- CO: 3 work as short movie maker
- CO: 4 become as visual editor

Unit I: Media and international communication- Radio and international communication- a brief overview Propaganda in the inter-war years-Nazi Propaganda

Unit II: Media and super power - Media during the Cold War- Radio Liberty, Voice of American Communication — NWICO- McBride Commission and UNESCO - Third World concerns.

Unit III: Global Conflict and World Wars- Media Coverage post 1990- Rise of Al Jazeera- CNN's satellite transmission- embedded Journalism

Unit IV: Media and Cultural Globalization- Cultural Imperialism-Cultural politics-media hegemony and Global cultures.

Unit V: Media and the Global market- technological developments- Media conglomerates- Ted Turner-Rupert Murdoch -Zee TV as a Pan Indian Channel- Hollywood Entertainment.

Internal Assessment: The internal assessment will be based on assignments, group discussions.

References:

1. International Communication: Continuity and Change by Daya Kishan Thussu.
2. War, Media and Propaganda-.A Global Perspective by Yahya R. Karnatipour and Nancy Snow.
3. Communication and Society, Today and Tomorrow "Many Voices One World" Unesco Publication, Rowman and Littlefield publishers, 2004.
4. Journalism after 9/11 by Barbie Zelizer and Stuart
5. War and the media: Reporting conflict 24x7 by DayaKishanThussu

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(AUTONOMOUS)
B.N. ROAD, MYSORE-25

DEPARTMENT OF JOURNALISM
MODEL QUESTION PAPER FOR GENERIC ELECTIVE PAPERS
V&VI Semesters
(CBCS Scheme)

Max. Marks — 35

I - Write any three questions. Each question carries five marks. 5X3 15

- 1.
- 2.
- 3.
- 4.
- 5.

Q: III - Write any two questions. Each question carries ten marks. 10X2-20

- 1.
- 2.
- 3.
- 4.
- 5.

Syllabus for Practical

20 Marks for test

20 marks for practical Records

10 marks for viva

Practical Paper — Total Marks — 20 4 Questions of 5 Marks each = 20

Any One Paper

Semester: 5

Code: DLE 27015

Paper: -I Reporting and Editing Techniques

Practical contents:

1. Types of News writing-05
2. Two copies editing
3. Creation of .Headline=10
4. Structuring dummy editorial page
5. Two editorial writing
6. Ten captions writing

Semester: 5

Code: DLE27015

Paper: 2 Introductions to TV and Radio

Semester: .5

Code: DLE

Paper: 3 Indian Applied Journalism

Any One Paper

Semester: 6

Code: DLF 27015

Paper: 1 Practice of Advertising and Public Relations

Practical contents:

1. Press release
2. Preparing posters
3. Preparing brochures
4. Radio scripting writing
5. TV script writing
6. Planning and designing PR campaign

Semester: 6

Code: DLF27215

Paper: 2 Introduction to new Media

Semester: 6

Code: D1_127415

Paper: 3 Cinema Journalism

Guidelines

1. Colleges must show practical and tutorial classes in the timetable.
2. In practical !tutorial class make batches of 20 students each.
3. In papers having credits for practical's, End Semester Examination in Praciicals like Theory is mandatory.
4. BoE has to set papers for practical examination based on practical syllabus.
5. Establishment of computer lab and AV Studio with Journalism related software in the college are mandatory.

BoS Chair Person

Dr. Sathish K. Itagi

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE (AUTONOMOUS)
OOTY ROAD, MYSURU-25

DEPARTMENT OF JOURNAISM

PROCEEDINGS OF THE MEETING OF BOARD OF STUDIES FOR THE PROGRAMMES BA IN JOURNALISM, POLITICAL SCIENCE AND ENGLISH ((.1PE-BATS) HELD ON 21st JUNE 2019 AT 12.30.PM IN THE DEPARTMENT OF JOURNALISM , JSS COLLEGE, OOTY ROAD, MYSURU-25

MEMBERS	PRESENT	Asso	Media
Dr. Sathish K. Itagi		ciate	Studies
Dept. of Journalism, JSS College, Ooty Road		Prof	SSIT,
Chairman		essor	
Dr. C.K. Puttaswamy		,	Maraluru
HoD, DoS in Communication and Media		Depa	
Studies		rtme	,
University of Mysore, Manasagaugutri,		nt of	Tumakur
Mysore		Com	
Member (VC Nominee)		muni	u
Dr. Nagendra D.		catio	
		n and	Member

At the outset, the Chairman, BOS in Journalism, welcomed the members to the meeting of BOS and briefed about the agenda to be discussed. The following agenda was placed by the Chairman which was discussed and resolved as follows:

Agenda 1: To frame/ revise, discuss and approve the Scheme/ Syllabus under Choice Based Credit System for the programmes: BA in JOURNALISM, POLITICAL SCIENCE AND ENGLISH ((JPE-BA15) from the academic year 2019-20 onwards.

The Chairman appraised the members about the introduction of Choice Based Credit System to the above said programmes with the course matrix in 2017-18. Accordingly, a draft revised/ modified _Scheme/ Syllabus was presented and placed before the Board for their opinion and approval.

Resolution: The BOS went through the Scheme/ Syllabus and discussed in length about.. various aspects of the same. After incorporation of the changes suggested by the members of BOS, the Syllabus was approved. Details of changes made with respect to the introduction of revised/ modified Scheme/ Syllabus in the existing courses is shown in Annexure-I.

Agenda 2: To prepare the Panel of Examiners for the examinations for the year 2019-20.

The Chairman presented the proposed Panel of Examiners to I to VI Semester examinations of 2019-20.

Resolution: After incorporating of certain changes suggested by the members, the Panel of Examiners was approved.

Agenda 3: To frame the Generic Elective Paper for V and VI non Journalism semester students

The Chairman presented the proposed to frame the CBCS syllabus

Resolution: After incorporating of certain changes suggested by the members, the syllabus was framed and approved.

Finally the meeting was concluded with the Chairman thanking the Members for their active participation in the deliberations of the meeting.

Annexure-I:

Revision/ modification made in the existing Generic Elective Syllabus for 2019-20 batches onwards:

Existing	Proposed	Justification	Approved
Title of the paper: Media Gender and Human Rights(Course Code): DLE2765	Bot.4.c.A	\-)vr'otA oy	PcIT(.7eet



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(Autonomous)
B N ROAD, MYSURU- 570 025

DEPARTMENT OF ENGLISH

Syllabus under
CHOICE BASED CREDIT SYSTEM (CBCS)

OPTIONAL ENGLISH

2019 - 20 onwards
(Approved by BOS)

B.A Programme syllabus –

Proposed Scheme of study under CBCS scheme to be implemented from

2019- 20

BA (Optional English) DSE

Paper No.	Sem	Course code	Course Title	Teaching Hours Per Week	LTP	Credit	I.A Marks	Theory	Total Marks
1	I	ELA22024 ELA22025	Poetry, Drama & Essays	5+1	5-1-0	6	30	70	100
2	II	ELB22024 ELB22025	Poetry, Fiction & Essays	5+1	5-1-0	6	30	70	100
3	III	ELC22024 ELC22025	Poetry, Drama & Fiction	5+1	5-1-0	6	30	70	100
4	IV	ELD22024 ELD22025	Poetry, Fiction & Prose	5+1	5-1-0	6	30	70	100
*DSE- Discipline Specific Elective – One of the following									
5	V	ELE22024 ELE22025	Indian Writing in English	5+1	5-1-0	6	30	70	100
6	V	ELE22224 ELE22225	Modern Literature	5+1	5-1-0	6	30	70	100
*DSE- Discipline Specific Elective – One of the following									
7	VI	ELF22024 ELF22024	American Literature	5+1	5-1-0	6	30	70	100
8	VI	ELF22224 ELF22424	English Writing in Third World Countries	5+1	5-1-0	6	30	70	100
GE									
9	V	ELE22424 ELE22425	Drama	2	2-0-0	2	15	35	50
10	VI	ELF22424 ELF22425	Drama	2	2-0-0	2	15	35	50

**Scheme of Assessment for B.A (History English, Economics) and (English, Political
Science, Journalism) under CBCS
2019-20**

Year	Sem	COURSE CODE	COURSE TYPE	TITLE OF THE PAPER	CONTINUOUS ASSESSMENT		MAX. MARKS			Duration of Exam
					TH	TH	C3	TH	IA	TH
I	I		DSE - A	Poetry, Drama and Essays	15	15	70	70	30	3h
	II		DSE- B	Poetry, Fiction and essays	15	15	70	70	30	3h
II	III		DSE- C	Poetry, Drama and Fiction	15	15	70	70	30	3h
	IV		DSE- D	Poetry, Fiction and Prose	15	15	70	70	30	3h
	V		DSE V A	Indian Writing in English	15	15	70	70	30	3h
	V		DSE V B	Modern Literature	15	15	70	70	30	3h
	VI		DSE VI C	American Literature	15	15	70	70	30	3h
	VI		DSE VI D	English Writing in Third World Countries	15	15	70	70	30	3h
	V		GE 1	Drama	15	15	35	35	15	2h
	VI		GE 2	Drama	15	15	35	35	15	2h

TEMPLATE SYLLABUS IN ENGLISH

Programme: B.A

Sem	Core			DSE			GE		
	No. of Courses	Credits	Total Hours	No. of Courses	Credits	Total Hours	No. of Courses	Credits	Total Hours
I	1	6	75+15	-	-	-	-	-	
II	1	6	75+15	-	-	-	-	-	
III	1	6	75+15	-	-	-	-	-	
IV	1	6	75+15	-	-	-	-	-	
V	-	-	-	1	6	75+15	1	2	24
VI	-	-	-	1	6	75+15	1	2	24

DSE - English (Credit: 06each)

Sl. No.	Sem	Course Code	Title of the paper	Total Credits	
				Th	Tu
1	I		DSE -A Poetry, Drama and Essays	5	1
2	II		DSE - B Poetry, Fiction and essays	5	1
3	III		DSE - C Poetry, Drama and Fiction	5	1
4	IV		DSE -D Poetry, Fiction and Prose	5	1

Discipline Specific Elective Papers (Credit: 06 each)

Sl. No.	Sem	Course Code	Title of the paper	Total Credits	
				Th	Tu
1	V		Indian Writing in English	5	1
			Modern Literature	5	1
2	VI		American Literature	5	1
			English Writing in Third World Countries	5	1

Generic Elective (Credit: 02 each)

1	V		Drama	2	0
2	VI		Drama	2	0

Programme: Bachelor of Arts in History Economics and English

Programme Outcomes:

After completing the Bachelor of Arts in History, Economics and English students will be able to

- PO1. Critically recognize the social, political, economic and cultural aspects of History.
- PO2. Develop an informed familiarity with multiple cultures.
- PO3. Correctly extract evidence from primary sources of analysing and evaluating them in relation to their cultural and historical context.
- PO4. Demonstrate critical thinking skills by analysing and evaluating historical information from multiple sources
- PO5. Support a clear thesis statement and demonstrate competence in Standard English Language and usage.
- PO6. Spread the messages of equality, nationality, social harmony and other human values.
- PO7. Demonstrate critical reading, writing, and thinking skills.
- PO8. Write focused, organized, well-developed, and text-based essays using effective paragraphs.
- PO9. Develop and carry out research projects, and locate, evaluate, organise, and incorporate information effectively.
- PO10. Acquire the ability to engage in independent and life-long learning in a broader context about socio-technological and demographic changes.

Program Specific Outcome:

Program: Bachelor of Arts in History Economics and English

After completing the Bachelor of Arts in History, Economics and English students will be able to

- PSO1. Understand background of our religion, customs, institutions, administration and so on.
- PSO2. Understand the present existing social, political, religious and economic conditions of the people
- PSO3. Identify cultural and literary synthesis.
- PSO4. Imbibe and inculcate patriotism
- PSO5. Major knowledge of English as Literature.
- PSO5. Basic knowledge of English Grammar.
- PSO6. Basic knowledge of English as Language
- PSO7. Critical study of English Literary studies.
- PSO8. Understand Cultural diaspora
- PSO9. Opens window for world market

Programme: Bachelor of Arts in Journalism Political Science and English

Programme Outcomes:

After completing the Bachelor of Arts in Journalism, Political Science and English students will be able to

PO1. Write focused, organized, well-developed, and text-based essays using effective paragraphs.

PO2. Support a clear thesis statement and demonstrate competence in Standard English Language and usage.

PO3. Develop and carry out research projects, and locate, evaluate, organize, and incorporate information effectively.

PO4. Write articles, novels, stories to spread the messages of equality, nationality, social harmony and other human values.

PO5. Acquire the ability to engage in independent and life-long learning in the broadest context socio-technological changes.

PO6. Demonstrate critical reading, writing, and thinking skills.

PO7. Report the day to day activities effectively either written or oral.

PO8. Develop competence and prepare for competitive examinations.

PO9. Enhance their presentation skills in mass media (Print/electronic)

Program Specific Outcome:

Program: Bachelor of Arts in Journalism Political Science and English

After completing the Bachelor of Arts in Journalism, Political Science and English students will be able to

PSO1. Opens window for world market

PSO2. Write and edit clear, graceful, grammatically correct prose

PSO3. Demonstrate an awareness of Journalism as an ethical practise

PSO4. Basic knowledge of English as Language.

PSO5. Major knowledge of English as Literature.

PSO6. Basic knowledge of English Grammar.

PSO7. Language helps to prepare important documentations, preserve documents/prepare reports of important events.

PSO8. Present themselves as good journalists with good communicative skills

Course code- ELA 22024/25

**SEMESTER I
DSE-I
Poetry, Drama and Essays**

Course Outcome:

After completion of the course the student is able to

- CO1. Know the history of English literature in the chronological order
- CO2. Enjoy the literary forms such as novel, poem, play, and essay.
- CO3. Critically understand the literature
- CO4. Emotionally develop his mind.
- CO5. Understand the culture in that particular period of time
- CO6. Enhance narrative capacity and be rational and decisive in his approach to life.

UNIT 1: Poetry

- | | |
|------------------------|---|
| 1. Geoffrey Chaucer | Knight |
| 2. Ben Jonson | To Celia |
| 3. William Shakespeare | Shall I Compare Thee
When to the Session |
| 4. John Milton | On His Blindness
How Soon Hath Time |
| 5. Sir Philip Sidney | The Nightingale |

UNIT II: Drama

William Shakespeare The Tempest

UNIT III

Essays

Chaucer to Milton	History of English Literature (Hudson)
Study of Drama	An Introduction to English Literature (Hudson)

Literary Terms (Self-study)

Epic, Ballad, Ode, Elegy, Mock Epic, Lyric, Dramatic Monologue, Soliloquy, Metaphor, Simile, Irony, Paradox.

Ref: A Glossary of Literary Terms M.H.Abrams

Golden Treasury- poetry

Narain Series – Poetry

Sonnets of Shakespeare

Geoffery Chaucer – Canterbury Tales

Narain Series – Drama

Complete works of Hudson: An outline History of English Literature

**2019-20- CBCS
Pattern of Question Paper
Optional English
I BA- I Semester
Poetry, Drama & Essays**

Max. Hrs-3

Max. Marks-100

Section A

Poetry

I. Annotate Four (Two from poetry and two from drama)of the following:

4x5=20

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.

Section B

Poetry

II. Answer Two of the following:

2x7=14

- 1.
- 2.
- 3.
- 4.

Section C

Drama

III. Answer Two of the following:

2x8=16

- 1.
- 2.
- 3.
- 4.

Section D

Essays

IV. Answer two of the following questions:

2x10=20

- 1.
- 2.
- 3.
- 4.

V. Internal Assessment

30

Course code: ELB 22024/25

SEMESTER II
DSE - II
Poetry, Fiction & Essays

Course Outcome:

After completion of the course the student is able to

- CO1. Understand the language, culture and pattern of writing of the 18th C writers.
- CO2. Enjoy the literary forms such as novel, poem, and essay.
- CO3. Critically analyse the literature
- CO4. Understand the relation between literature and real life.
- CO5. Connect, compare and contrast the life of fantasy and fact.
- CO6. Distinguish the human qualities

UNIT 1: Poetry

- | | |
|-------------------|--|
| 1. John Donne | Go and Catch a Falling Star
The Good Morrow
Death Be Not Proud |
| 2. Andrew Marvel | To His Coy Mistress
The Definition of Love |
| 3. George Herbert | The Gifts of God
The Collar |
| 4. Robert Herrick | To Blossoms
To Dianeme |
| 5. Henry Vaughan | The Retreat |

UNIT II: Fiction

Thomas Hardy : Far From The Madding Crowd

UNIT III: Essays

Francis Bacon	:	Of Travel, Of Studies
Joseph Addison	:	Ladies Head Dress
Oliver Goldsmith	:	Beau Tibbs
T.S.Eliot	:	Metaphysical Poets
History of English Literature	:	John Milton to Renaissance

Ref., Books:

Metaphysical Poets – Narain Series—Poetry

Fiction- Victorian Age

20th C reviews

Essays- Penguin Classics

A Book of English essays

2019-20- CBCS
Pattern of Question Paper
Optional English
I BA- II Semester
Poetry, Fiction &Essays

Max. Hrs-3

Max. Marks-100

Section A

Poetry

- I. Annotate Four of the following: 4x5=20**
- 1.
 - 2.
 - 3.
 - 4.
 - 5.
 - 6.

Section B

Poetry

- II. Answer Two of the following: 2x7=14**
- 1.
 - 2.
 - 3.
 - 4.

Section C

Fiction

- III. Answer Two of the following: 2x8=16**
- 1.
 - 2.
 - 3.
 - 4.

Section D

Essays

- IV. Answer two of the following questions: 2x10=20**
- 1.
 - 2.
 - 3.
 - 4.

V. Internal Assessment

30

2019-20- CBCS
Pattern of Question Paper
Optional English
II BA- III Semester
Poetry, Drama & Short Story

Max. Hrs-3

Max. Marks-100

Section A

Poetry

- I. Annotate Four (Two from poetry and two from drama) of the following:**
1. **4x5=20**
 - 2.
 - 3.
 - 4.
 - 5.
 - 6.

Section B

Poetry

- II. Answer Two of the following:** **2x7=14**
- 1.
 - 2.
 - 3.
 - 4.

Section C

Drama

- III. Answer Two of the following:** **2x8=16**
- 1.
 - 2.
 - 3.
 - 4.

Section D

Short Story

- IV. Answer two of the following questions:** **2x10=20**
- 1.
 - 2.
 - 3.
 - 4.

V. Internal Assessment

30

Course code: ELD22024/25

SEMESTER IV

**DSE -IV
Poetry, Fiction & Prose**

Course Outcome:

After completion of the course the student is able to

- CO1. Understand the culture and tradition prevailed in 20th C
- CO2. Enhance the narrative capacity and be rational and decisive in his approach to life
- CO3. Re-relate historical events in a more apprehensive language.
- CO4. Relation between literature and real life.
- CO5. Learn and lead a life filled with humanitarian concern.

UNIT1: Poetry

- | | |
|-----------------------|-----------------------------|
| 1. William Wordsworth | Resolution and Independence |
| 2. John Keats | Ode on a Grecian Urn |
| 3. P.B. Shelley | To A Skylark |
| 4. S. T. Coleridge | Frost at Midnight |
| 5. Robert Browning | My Last Duchess |
| 6. Lord Tennyson | Ulysses |

Fiction

Emily Bronte	Wuthering Heights
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Prose

William Wordsworth	Preface to the Lyrical Ballads
History of English Literature	Romantics and Victorians

Literary Terms (Self-study)

Comedy, Tragedy, Picturesque Novel, Gothic Novel, Science Fiction, Allegory, Parable.

*Ref., Books: Poetry: The Golden Treasury, Narain Series, Romantic Period
The Oxford History of English Literature, The Romantic Imagination by C.M.Bowra
The Mirror and The Lamp: Romantic Theory and The Critical Condition by
M.H.Abrams*

The Romantics by S.Prickett

Fiction: Jane Austen: Feminism and Fiction by M. Kirkham

20th C reviews

Prose: Tradition and Experiment in Wordsworth's Lyrical Ballads by M.Jacobus

William Wordsworth Biography by M. Moorman

The Romantic Age

*General Reference: R.J.Rees, M.H.Abrams, I A Richards' The Study of DramAn
Introduction of The Study of English Literature- W.H.Hudson*

**2019-20- CBCS
Pattern of Question Paper
Optional English
II BA- IV Semester
Poetry, Fiction & Prose**

Max. Hrs-3

Max. Marks-100

Section A

Poetry

- I. Annotate Four of the following: 4x5=20**
- 1.
 - 2.
 - 3.
 - 4.
 - 5.
 - 6.

Section B

Poetry

- II. Answer Two of the following: 2x7=14**
- 1.
 - 2.
 - 3.
 - 4.

Section C

Fiction

- III. Answer Two of the following: 2x8=16**
- 1.
 - 2.
 - 3.
 - 4.

Section D

Prose

- IV. Answer two of the following questions: 2x10=20**
- 1.
 - 2.
 - 3.
 - 4.

V. Internal Assessment

30

Course code: ELE22024/25

**BA V Semester
DSE –V A
Indian Writing in English**

Course Outcome:

After completion of the course the student is able to

CO1.Understand rich culture and heritage of India in a better way.

CO2.Develop the feeling of oneness.

CO3.Bridge the gap that divide the country with regional languages.

CO4.Relation between literature and real life.

CO5. Develop translation skills.

Unit-I-Poetry

Basavanna	Vachanas
	Rich Build Temple for Shiva
Akkamahadevi	They Say Shiva Loves Music
	I Am The One With
	O Mother, I Am Burnt
Kamala Das	An Introduction
Arun Kolatkar	The Old Woman
Toru Dutt	Sita
Sarojini Naidu	Indian Weavers
Mamang Dai	Remembrance

Unit-II-Drama

Mahesh Dattani	The Final Solution
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Unit-III- Fiction

Aravind Adiga	The White Tiger
---------------	-----------------

Ref. Books:

Poetry :Narain Series

Critical Study of Mahesh Dattani-Surjeet Publication

20th C Indian Writing in English- S.Iyengar

Fiction: Untouchable – Narain Series

The White Tiger

Social criticism in AravindaAdiga's The White Tiger

Research Gate www.researchgate.nethttps://indianbookcritics.in

A Survey of Virashaiva Religion and Literature- Prof. Chandrashekaraiah

Indian Writing in English- K.R.Srinivas Iyengar

Indian Drama in English edited by Kaustav Chakravarthy

2019-20- CBCS
Pattern of Question Paper
Optional English
III BA- V Semester
DSE V A
Poetry, Drama & Fiction

Max. Hrs-3

Max. Marks-100

Section A

I. Annotate Four (Two from poetry and two from drama) of the following:
4x5=20

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.

Section B
Poetry

I. Answer Two of the following: **2x7=14**

- 1.
- 2.
- 3.
- 4.

Section C
Drama

II. Answer Two of the following: **2x8=16**

- 1.
- 2.
- 3.
- 4.

Section D
Fiction

III. Answer two of the following questions: **2x10=20**

- 1.
- 2.
- 3.
- 4.

V. Internal Assessment

30

Course code: ELE22224/25

**SEMESTER V
Modern Literature
DSE – V B**

Course Outcome:

After completion of the course the student is able to

- CO1. Have better understanding of life.
- CO2. Develop analytical and critical quality.
- CO3. Be creative in his day to day life and face the problems
- CO4. Relation between literature and real life.
- CO5. Compare and contrast the historical and modern works

UNIT 1: Poetry

- | | |
|------------------|--|
| 1. G. M. Hopkins | Pied Beauty
Thou Art Indeed Just, Lord
Starlight Night |
| 2. W. B. Yeats | Easter 1916
A Prayer for My Daughter |
| 3. T. S. Eliot | Geration
The Hippopotamus |
| 4. W. H. Auden | Unknown Citizen
Muse De Beaux Arts |

Prose

- | | |
|-------------|---|
| M.K. Gandhi | My Experiments with Truth
(Autobiography) Part I |
|-------------|---|

Criticism

- | | |
|---------------|---|
| T.S. Eliot | Tradition and Individual Talent |
| D.H. Lawrence | Why the Novel Matters? |
| Chinua Achebe | Novelist as a Teacher |
| Mathew Arnold | The Functions of Criticism at the
present time |

Self Study

- | | |
|-------------------------------|----------------|
| History of English Literature | The Modern age |
|-------------------------------|----------------|

Ref. Books:

Poetry: Forces in Modern British Literature by W.Y.Tindall

Narain Series

20th C Literature Penguin Classics

Criticism

An Outline of English Literature- Pat Rogers

Narain Series

20th C views

My Experiment with Truth – Mahadev Desai

An Introduction to the History of English Literature- W.H.Hudson

The Life and Death of Mahatma Gandhi- Robert Payne

The American Theatre- Allan S Downer

2019-20- CBCS
Pattern of Question Paper
Optional English
III BA- V Semester
DSE V B – Modern Literature

Max. Hrs- 3 Hrs

Max Marks - 100

Section A
Poetry

- I. Annotate Four of the following:** **4x5=20**
- 1.
 - 2.
 - 3.
 - 4.
 - 5.
 - 6.

Section B
Poetry

- II. Answer Two of the following:** **2x7=14**
- 1.
 - 2.
 - 3.
 - 4.

Section C
Prose

- III. Answer Two of the following:** **2x8=16**
- 1.
 - 2.
 - 3.
 - 4.

Section D
Criticism

- III. Answer two of the following questions:** **2x10=20**
- 1.
 - 2.
 - 3.
 - 4.

V. Internal Assessment

30

Course code: ELF22024/25

**Semester VI
American Literature
DSE – V1 C**

After completion of the course the student is able to

CO1. Introduce literature produced by countries other than England

CO2. Read the great writers of this country

CO3. Bring in a wide variety of writers and thinkers into study pattern.

CO4. Learn the socio cultural growth

Unit –I-Poetry

- | | |
|--------------------|--|
| 1. Walt Whitman | Out of the Cradle Endlessly Rocking
When Lilacs in the Dooryard Bloomed |
| 2. Robert Frost | Birches
Mending Wall |
| 3. Langston Hughes | I too Sing America
Life is Fine |
| 4. Wallace Stevens | Sunday Morning
Anecdote of a Jar |

Unit-II-Prose

The Old Man and the Sea Ernest Hemmingway

Unit-III-Criticism

- | | |
|--------------------|----------------------|
| 1. Emerson | The Poet |
| 2. Robert Frost | Figure A Poem Makes |
| 3. Edgar Allan Poe | The Poetic Principle |

Ref.,:

A Critical Evaluation – Unique Publishers

20th C Reviews

English Critical essays – 20th C – Phyllis M Jones

History of American Literature

Narain Series

The Pelican Guide to English Literature – Vol.7

A History of English Literature by Emily Legouis

**2019-20- CBCS
Pattern of Question Paper
Optional English
III BA- VI Semester
DSE – VI C
American Literature**

Max. Hrs- 3 Hrs

Max Marks - 100

Section A

Poetry

- I. Annotate Four of the following: 4x5=20**
- 1.
 - 2.
 - 3.
 - 4.
 - 5.
 - 6.

Section B

Poetry

- II. Answer Two of the following: 2x7=14**
- 1.
 - 2.
 - 3.
 - 4.

Section C

Prose

- III. Answer Two of the following: 2x8=16**
- 1.
 - 2.
 - 3.
 - 4.

Section D

Criticism

- IV. Answer two of the following questions: 2x10=20**
- 1.
 - 2.
 - 3.
 - 4.

V. Internal Assessment

30

Course code: ELF22224/225

**BA VI Semester
English Writing in Third World Countries
DSE – V1 D**

Course Outcome:

After completion of the course the student is able to

- CO1. Understand the problems the of third world countries
- CO2. Know the rift between colonised and coloniser
- CO3. Understand the spirit of independence and limitations of freedom.
- CO4. Get the knowledge of pre and post independent socio-political and economic aspects of India.
- CO5. Develop critical and rational thinking.
- CO6. Understand the diversity and similarity in socio-cultural aspects of the third world countries.

Unit- I- Poetry

Rabindranath Tagore	Where the Mind is Without Fear
Peter Porter	Your Attention Please
Gabriel Okara	You Laughed, Laughed, and Laughed
Margaret Atwood	Journey to the Interior
Derek Walcott	A Far Cry from Africa
Kishwar Naheed	I Am Not That Woman

Unit-II-Novel

The Alchemist	Paulo Coelho
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Unit-III- Fiction

The Lion and the Jewel	Wole Soyinka
------------------------	--------------

Ref.,:

Narain series

Online reference

2019-20- CBCS
Pattern of Question Paper
Optional English
III BA- VI Semester
DSE - VI D English Writing in Third World Countries

Max. Hrs- 3 Hrs

Max Marks - 100

Section A
Poetry

I. Answer three of the following:

3x10=20

- 1.
- 2.
- 3.
- 4.

Section B
Fiction

II. Answer Two of the following:

2x10=20

- 1.
- 2.
- 3.
- 4.

Section C
Fiction

III. Answer all the questions:

2x10=20

- 1.
- 2.
- 3.
- 4.

IV. Internal Assessment

30

Course Code: ELE22424/425

**BA V Semester
GE 1
Drama**

Marks: 50

2Credits/Week

**Unit- I
Drama**

A Christmas Carol

Charles Dickens

Course Code: ELE22424/425

Pattern of Question paper

V Semester

GE 1

Drama

Max. Hrs – 2 Hrs

Max. marks – 50

Unit I

I. Annotate Three of the following:

3x5=15

- 1.
- 2.
- 3.
- 4.
- 5.

Unit II

II. Answer Four of the following:

4x5=20

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.

Course Code: ELF22424/425

**BA VI Semester
GE 2
Drama**

Marks: 50

2 Credits/Week

**Unit- I
Drama**

Candida

Bernard Shaw

Course Code: ELF22424/425

Pattern of Question paper

V Semester

GE 2

Drama

Max. Hrs – 2 Hrs

Max. marks – 50

Unit I

III. Annotate Three of the following:

3x5=15

- 1.
- 2.
- 3.
- 4.
- 5.

Unit II

IV. Answer Four of the following:

4x5=20

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.



**JSS COLLEGE OF ARTS, COMMERCE AND
SCIENCE**

(Autonomous)

B N ROAD, MYSURU- 570 025

DEPARTMENT OF ENGLISH

Syllabus

CHOICE BASED CREDIT SYSTEM

AECC

LANGUAGE ENGLISH

2019 - 20 onwards

(Approved by BOS)

BA, BSc and BCA

Templates - BA, BSc, & BCA

Paper No	Sem	Course Code	Course	Teaching Hours Per Week	L.T.P	Credit	I.A Marks	Theory Exam	Total Marks
1.	I	ELA 02021 ELA02022 ELA02023 ELA02024 ELA02025 ELA02026 DMA02001 DMA 02002 DMA 02003 DMA 02004 DMA 02005 DMA 02006 DMA 02007 DMA 02008 ECA02001	Poetry, Prose & Grammar	2+2	2-1-0	3	30	70	100
2.	II	ELB 02021 ELB02022 ELB02023 ELB02024 ELB02025 ELB02026 DMB02001 DMB 02002 DMB 02003 DMB 02004 DMB 02005 DMB 02006 DMB 02007 DMB 02008 ECB02001	Poetry, One Act Plays & Grammar	2+2	2-1-0	3	30	70	100
3.	III	ELC 02021 ELC02022 ELC02023 ELC02024 ELC02025 ELC02026 DMC02001 DMC 02002 DMC 02003 DMC 02004 DMC 02005 DMC 02006 DMC 02007 DMC 02008	Novel & Grammar	2+2	2-1-0	3	30	70	100

		ECC02001							
4.	IV	ELD 02021 ELD02022 ELD02023 ELD02024 ELD02025 ELD02026 DMD02001 DMD 02002 DMD 02003 DMD 02004 DMD 02005 DMD 02006 DMD 02007 DMD 02008 ECD02001	Drama & Grammar	2+2	2-1-0	3	30	70	100

Course type BA, BSc, & BCA

	Semester	Course Code	Course Title	Course Type	No. of Credits			
					L	T	P	Total
1.	I	ELA 02021 ELA02022 ELA02023 ELA02024 ELA02025 ELA02026 DMA02001 DMA 02002 DMA 02003 DMA 02004 DMA 02005 DMA 02006 DMA 02007 DMA 02008 ECA02001	Poetry, Prose & Grammar	AECC	2	1	-	03
2.	II	ELB 02021 ELB02022 ELB02023 ELB02024 ELB02025 ELB02026 DMB02001 DMB 02002 DMB 02003 DMB 02004	Poetry, One Act Plays & Grammar	AECC	2	1	-	03

		DMB 02005 DMB 02006 DMB 02007 DMB 02008 ECB02001						
3.	III	ELC 02021 ELC02022 ELC02023 ELC02024 ELC02025 ELC02026 DMC02001 DMC 02002 DMC 02003 DMC 02004 DMC 02005 DMC 02006 DMC 02007 DMC 02008 ECC02001	Novel & Grammar	AECC	2	1	-	03
4.	IV	ELD 02021 ELD02022 ELD02023 ELD02024 ELD02025 ELD02026 DMD02001 DMD 02002 DMD 02003 DMD 02004 DMD 02005 DMD 02006 DMD 02007 DMD 02008 ECD02001	Drama & Grammar	AECC	2	1	-	03

Internal Assessment BA, BSc, & BCA

Max.Marks-100

Course type	C1		C2		C3		Total
	Marks	Duration (Hr)	Marks	Duration (Hr)	Marks	Duration (Hr)	
Language	15	1	15	1	70	3	100
Communication	15	1	15	1	70	3	100

Scheme of assessment: BA BSc, & BCA

Year	Sem	Course Code	TITLE OF THE PAPER	PERCENTAGE				
				C1	C2	C3	IA	Duration
				Theory	Theory	Theory		
I	I	ELA 02021 ELA02022 ELA02023 ELA02024 ELA02025 ELA02026 DMA02001 DMA 02002 DMA 02003 DMA 02004 DMA 02005 DMA 02006 DMA 02007 DMA 02008 ECA02001	Poetry, Prose & Grammar	15	15	70	30	3h
	II	ELB 02021 ELB02022 ELB02023 ELB02024 ELB02025 ELB02026 DMB02001 DMB 02002 DMB 02003 DMB 02004 DMB 02005 DMB 02006 DMB 02007 DMB 02008 ECB02001	Poetry, One Act Plays & Grammar	15	15	70	30	3h
II	III	ELC 02021 ELC02022 ELC02023 ELC02024 ELC02025 ELC02026 DMC02001 DMC 02002 DMC 02003 DMC 02004 DMC 02005 DMC 02006 DMC 02007 DMC 02008	Novel & Grammar	15	15	70	30	3h

		ECC02001						
	IV	ELD 02021 ELD02022 ELD02023 ELD02024 ELD02025 ELD02026 DMD02001 DMD 02002 DMD 02003 DMD 02004 DMD 02005 DMD 02006 DMD 02007 DMD 02008 ECD02001	Drama & Grammar	15	15	70	30	3h

Course code: ELA02021, 022, 023, 024, 025, 026

DMA02001, 002, 003, 004, 005, 006, 007, 008

ECA02001

**BA/BSc/BCA
I Semester
English Language AECC
Poetry, Prose & Grammar**

Unit I: Poetry

- | | |
|---|-----------------------|
| 1. An Attribute of God | - William Shakespeare |
| 2. Death, The Leveller | - James Shirley |
| 3. Home they Brought her Warrior dead | - Alfred Tennyson |
| 4. Stopping by Woods On A Snowy Evening | - Robert Frost |
| 5. Teach Him if You Can | - Abraham Lincoln |
| 6. Ajamil and the Tigers | - Arun Kolatkar |

Unit II: Prose

- | | |
|------------------------------|---------------------|
| 1. Water: The Elixir of Life | - C V Raman |
| 2. A Day's Wait | - Ernest Hemmingway |
| 3. An Astrologer's Day | - R K Narayan |
| 4. The Clay Mother-in-Law | - A K Ramanujam |
| 5. The Eyes are not here | - Ruskin Bond |

Unit III: Grammar

1. Parts of Speech
2. Punctuation
3. Letter Writing – Letter of Application/ grievance letter/ Personal letter
4. Comprehension

Tutorial Activities:

Newspaper Reading, Report writing, Story-telling, Recitation, Review of current literature activities, dialogue development, elocution, extempore speech, conversation writing.

**Pattern of Question Paper - 2019-20- CBCS
BA/BSC/BCA I Sem
Poetry, Prose & Grammar**

Max. Hrs-3

Max. Marks-100

**Section A
Poetry**

I. Annotate Three of the following:

3x4=12

- 1.
- 2.
- 3.
- 4.
- 5.

Section B

II. Answer Eight of the following questions in a word or a sentence:

8x1=8

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.
- 9.
- 10

**Section C
Poetry and Prose**

III. Answer any Two of the following questions: (poetry)

2x7=14

- 1.
- 2.
- 3.
- 4.
- 5.

IV. Answer any Two of the following questions: (prose)

2x8=16

- 1.
- 2.
- 3.
- 4.

**Section D
Grammar**

V. Language Component

1. Parts of Speech
2. Punctuation
3. Letter Writing
4. Comprehension

**5
5
5
5**

V. Internal Assessment

30

Course code: ELB02021, 022, 023, 024, 025, 026

DMB02001, 002, 003, 004, 005, 006, 007, 008

ECB02001

**BA/BSc/BCA
II Semester
English Language AECC
Poetry, One Act Plays & Grammar**

Unit I Poetry:

- | | |
|-------------------------------|----------------------|
| 1. The Human Seasons | - John Keats |
| 2. On Shakespeare | - John Milton |
| 3. The Second Coming | - W B Yeats |
| 4. Characters of a Happy Life | - Sir Henry Wotton |
| 5. Fidelity | - William Wordsworth |
| 6. The Crutches | - Bertolt Brecht |

Unit II One Act Plays:

- | | |
|-------------------------|------------------|
| 1. The Pie and the Tart | - Hug Chesterman |
| 2. The Man Upstairs | - Hugh Beresford |
| 3. The Never-Never Nest | - Cedric Mount |

Unit III Grammar:

1. Subject & Verb Agreement
2. Antonyms
3. Synonyms
4. Articles

Tutorial Activities:

Newspaper Reading, Report writing, Story-telling, Recitation, Review of current literature activities, dialogue development, elocution, extempore speech, conversation writing

Pattern of Question Paper
BA/BSc/BCA II Sem
Poetry, One Act Play & Grammar

Max. Hrs-3

Max. Marks-100

Section-A

I. Annotate Three of the following: (poetry)

3x4=12

- 1.
- 2.
- 3.
- 4.
- 5.

Section B

II. Answer Eight of the following questions in a word or a sentence:

8x1=8

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.
- 9.
- 10.

Section C

Poetry and One act play

III. Answer any Two of the following questions:(poetry)

2x7=14

- 1.
- 2.
- 3.
- 4.
- 5.

IV. Answer any Two of the following questions: (prose)

2x8=16

- 1.
- 2.
- 3.
- 4.

Section D
Grammar

IV. Language Component

1. Subject & Verb Agreement
2. Antonyms
3. Synonyms
4. Articles

5
5
5
5

V. Internal Assessment

30

Course code: ELC02021, 022, 023, 024, 025, 026

DMC02001, 002, 003, 004, 005, 006, 007, 008

ECC02001

**BA/BSc/BCA
III Semester
English Language (AECC)
Fiction & Grammar**

Unit I: Fiction

The Guide - R K Narayan

Unit II: Grammar

1. Question Tag
2. Framing of questions
3. One-word substitutes
4. Words often confused

Tutorial Activities:

Newspaper Reading, Report writing, Story-telling, Recitation, Review of current literature activities, dialogue development, elocution, extempore speech, conversation writing

Pattern of Question Paper

BA/BSC/BCA

III Sem

Fiction & Grammar

Max. Hrs-3

Max. Marks-100

Section A

I. Answer three of the following:

3x10=30

- 1.
- 2.
- 3.
- 4.
- 5.

Section B

II. Write short notes for four of the following:

4x5=20

- 1
- 2
- 3
- 4
- 5

Section C

Grammar

III. Language Component

- | | |
|-------------------------|---|
| 1. Question Tag | 5 |
| 2. Framing of questions | 5 |
| 3. One-word substitutes | 5 |
| 4. Words often confused | 5 |

V. Internal Assessment

30

Course code: ELD02021,022,023,024,025,026

DMD02001,002,003,004,005,006,007,008

ECD02001

**BA/BSc/BCA
IV Semester
English Language (AECC)
Drama & Grammar**

Unit I: Drama

Othello

William Shakespeare

Unit II: Grammar

1. Active and passive voice
2. Linkers
3. Precise Writing
4. Essay Writing

Tutorial Activities:

Newspaper Reading, Report writing, Story-telling, Recitation, Review of current literature activities, dialogue development, elocution, extempore speech, conversation writing

Pattern of Question Paper
BA/BSC/BCA
IV Sem
Drama & Grammar

Max. Hrs-3

Max. Marks-100

Section A

I. Annotate Four of the following:

4x5=20

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.

Section B

II. Answer three of the following:

3x10=30

- 1.
- 2.
- 3.
- 4.

Section C
Grammar

III. Language Component

- | | |
|-----------------------------|----------|
| 1. Active and passive voice | 5 |
| 2. Linkers | 5 |
| 3. Precise Writing | 5 |
| 4. Essay Writing | 5 |

. Internal Assessment **30**

**JSS COLLEGE OF ARTS, COMMERCE AND
SCIENCE
(Autonomous)
B N ROAD, MYSURU- 570 025**

**DEPARTMENT OF ENGLISH
Syllabus
CHOICE BASED CREDIT SYSTEM**

**AECC
LANGUAGE ENGLISH**

**2019 - 20 onwards
(Approved by BOS)
B Com and BBA**

Templates - B. Com and BBA

Paper No	Sem	Paper Code	Course	Teaching Hours Per Week	L.T.P	Credit	IA Marks	Theory Exam	Total Marks
1.	I	ENA02001 CDA02001	Poetry, Prose & Grammar	2+2	2-1-0	3	30	70	100
2.	II	ENB02001 CDB02001	Poetry, One Act Plays & Grammar	2+2	2-1-0	3	30	70	100
3.	III	ENC02001 CDC02001	Novel & Grammar	2+2	2-1-0	3	30	70	100
4.	IV	END02001 CDD02001	Drama & Commercial correspondence	2+2	2-1-0	3	30	70	100

	Semester	Course Code	Course Title	Course Type	No. of Credits			
					L	T	P	Total
1.	I	ENA02001 CDA02001	Poetry, Prose & Grammar	AECC	2	1	-	03
2.	II	ENB02001 CDB02001	Poetry, One Act Plays & Grammar	AECC	2	1	-	03
3.	III	ENC02001 CDC02001	Novel & Grammar	AECC	2	1	-	03
4.	IV	END02001 CDD02001	Drama & Commercial correspondence	AECC	2	1	-	03

Internal Assessment

Max.Marks-100

Course type	C1		C2		C3		Total
	Marks	Duration (Hr)	Marks	Duration (Hr)	Marks	Duration (Hr)	
Language	15	1	15	1	70	3	100
Communication	15	1	15	1	70	3	100

Scheme of assessment: B. Com and BBA

Year	Sem	Course Code	TITLE OF THE PAPER	PERCENTAGE				
				C1	C2	C3	IA	Duration
				Theory	Theory	Theory		
I	I	ENA02001 CDA02001	Poetry, Prose & Grammar	15	15	70	30	3h
	II	ENB02001 CDB02001	Poetry, One Act Plays & Grammar	15	15	70	30	3h
II	III	ENC02001 CDC02001	Novel & Grammar	15	15	70	30	3h
	IV	END02001 CDD02001	Drama & Commercial correspondence	15	15	70	30	3h

Course code: ENA02001
CDA02001

BCom/BBA
I Semester
English Language (AECC)
Poetry, Prose and Grammar

Unit I: Poetry

- | | |
|-----------------------------|-----------------------|
| 1. An Attribute of God | - William Shakespeare |
| 2. The Village Schoolmaster | - Oliver Goldsmith |
| 3. Daffodils | - William Wordsworth |
| 4. Ozymandias | - P B Shelley |
| 5. The Human Seasons | - John Keats |
| 6. The Road Not Taken | - Robert Frost |

Unit II: Prose

- | | |
|---------------------------|-----------------------|
| 1. The Golden Touch | - Nathaniel Hawthorne |
| 2. The Last leaf | - O. Henry |
| 3. Half A Rupee Worth | - R K Narayan |
| 4. The Clay Mother-in-law | - A K Ramanujan |
| 5. The Eyes are not here | - Ruskin Bond |

Unit III: Grammar

1. Parts of Speech
2. Punctuation
3. Letter Writing – Letter of Application/ grievance letter/ Personal letter
4. Comprehension

Tutorial Activities:

Newspaper Reading, Report writing, Storytelling, Recitation, Review of current literature, dialogue development, elocution, extempore-speech, conversation writing.

Pattern of Question Paper
BCom and BBA
I Sem
Poetry, Prose & Grammar

Max. Hrs-3

Max. Marks-100

Section A
Poetry

I. Annotate Three of the following:

3x4=12

- 1.
- 2.
- 3.
- 4.
- 5.

Section B

II. Answer Eight of the following questions in a word or a sentence:

8x1=8

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.
- 9.
- 10

Section C
Poetry and Prose

III. Answer any Two of the following questions: (poetry)

2x7=14

- 2.
- 3.
- 4.
- 5.

IV. Answer any Two of the following questions: (prose)

2x8=16

- 1.
- 2.
- 3.
- 4.

Section D
Grammar

1. Parts of Speech
2. Punctuation
3. Letter Writing
4. Comprehension

5
5
5
5

V. Internal Assessment

30

**Course code: ENB02001
CDB02001**

**BCom/BBA
II Semester
English Language (AECC)
Poetry, One Act Plays & Grammar**

Unit I: Poetry

- | | |
|-----------------------------|-----------------------|
| 1. On Shakespeare | - John Milton |
| 2. Fidelity | - William Wordsworth |
| 3. La Belle Dame Sans Merci | - John Keats |
| 4. Teach Him If You Can | - Abraham Lincoln |
| 5. Go Not to the Temple | - Rabindranath Tagore |
| 6. Five Ways to Kill a Man | - Edwin Brock |

Unit II: One Act Plays

- | | |
|-------------------------|-------------------------------|
| 1. Progress | - St. John Ervin |
| 2. The Blue Carbuncle | - Michael and Mollie Hardwick |
| 3. The Never-Never Nest | - Cedric Mount |

Unit III: Grammar

1. Subject & Verb Agreement
2. Antonyms
3. Synonyms
4. Articles

Tutorial Activities:

Newspaper Reading, Report writing, Storytelling, Recitation, Review of current literature, dialogue development, elocution, extempore-speech, conversation writing.

Pattern of Question Paper
II Sem
BCom and BBA
Poetry, One Act Play & Grammar

Max. Hrs-3

Max. Marks-100

Section A

I. Annotate Three of the following: (poetry)

3x4=12

- 1.
- 2.
- 3.
- 4.
- 5.

Section B

II. Answer Eight of the following questions in a word or a sentence:

8x1=8

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.
- 9.
- 10

Section C

Poetry and One act play

III. Answer any Two of the following questions: (poetry)

2x7=14

- 1.
- 2.
- 3.
- 4.
- 5.

IV. Answer any Two of the following questions:(play)

2x8=16

- 1.
- 2.
- 3.
- 4.

Section D
Grammar

IV. 1. Subject & Verb Agreement

5

2. Antonyms

5

3. Synonyms

5

4. Articles

5

V. Internal Assessment

30

**Course code ENC02001
CDC02001**

**BCom/BBA
III Semester
English Language (AECC)
Fiction & Grammar**

Max. Marks-70

3 Credits/Week

**Unit I
Fiction**

The Financial Expert

R K Narayan

**Unit II
Grammar**

1. Question Tag
2. Framing of questions
3. One word substitutes
4. Words often confused

Tutorial Activities:

Newspaper Reading, Report writing, Story-telling, Recitation, Review of current literature activities, dialogue development, elocution, extempore speech, conversation writing

Pattern of Question Paper

BCom and BBA

III Sem

Fiction & Grammar

Max. Hrs-3

Max. Marks-100

Section A

I. Answer three of the following:

3x10=30

- 1.
- 2.
- 3.
- 4.
- 5.

Section B

II. Write short notes for four of the following:

4x5=20

- 1
- 2
- 3
- 4
- 5

Section C

Grammar

III. Language Component

- | | |
|-------------------------|---|
| 1. Question Tag | 5 |
| 2. Framing of questions | 5 |
| 3. One-word substitutes | 5 |
| 4. Words often confused | 5 |

V. Internal Assessment

30

**Course code END02001
CDD02001**

**BCom/BBA
IV Semester
English Language (AECC)
Drama & Commercial Correspondence**

Max. Marks-70

3 Credits/Week

Unit I

Drama

A Christmas Carols

Charles Dickens

Unit II

Commercial correspondence

1. Quotation and Enquire Letters
2. Orders and Complaint letters
3. Circulars
4. Report Writing

Tutorial Activities:

Newspaper Reading, Report writing, Story-telling, Recitation, Review of current literature activities, dialogue development, elocution, extreme speech, conversation writin

Pattern of Question Paper
BCom/BBA
IV Semester
English Language (AECC)
Drama & Commercial Correspondence

Section-A
Drama

I. Annotate any three of the following: **3x5=15**

- 1.
- 2.
- 3.
- 4.
- 5.

II. Answer two of the following: **2x10=20**

- 1.
- 2.
- 3.
- 4.

Section-B
Commercial Correspondence

III. Draft the following letters. **4x5=20**

1. Quotation Letters
2. Enquiry Letter
3. Placing an order
4. Complaint Letter

IV. Circular Letter. **1x7=7**

V. Report Writing. **1x8=8**

VI. Internal Assessment **30**

**JSS COLLEGE OF ARTS, COMMERCE AND
SCIENCE**

(Autonomous)

B N ROAD, MYSURU- 570 025

DEPARTMENT OF ENGLISH

Syllabus

CHOICE BASED CREDIT SYSTEM

AECC

LANGUAG ENGLISH

Special English

2019 - 20 onwards

(Approved by BOS)

BA/B.Sc/BCA/B Com/ BBA

Course Code: ELA/DMA/ECA/ENA 030

**SPECIAL ENGLISH
SEMESTER-I
Prose, Short Stories & Language Component**

Unit-I

Prose: **Will Durant Reader**

30marks

1. The Condition of Civilization
2. What is beauty?
3. The origin of Science
4. Is progress real?
5. I fall in love

Unit-II

Short stories:

20marks

1. Witches' loaves- O. Henry
2. Gift of Magi- O. Henry
3. The Parting- Chuman Nahal
4. The luncheon- Somerset Maugham

Unit-III

Language Component

20marks

1. Antonyms
2. Synonyms
3. Preposition
4. Words often confused

SPECIAL ENGLISH
Pattern of Question Paper
BBA/B. Voc
SEMESTER-I
Prose, Short stories & Language Component

Marks-70

Max.Hrs-3

Unit-I Prose

I. Answer three questions: 3x10=30

- 1.
- 2.
- 3.
- 4.
- 5.

Unit-II Short stories

II. Answer two question: 2x10=20

- 1.
- 2.
- 3.
- 4.

Unit-III Language Component

- | | |
|-------------------------|---|
| 1. Antonyms | 5 |
| 2. Synonyms | 5 |
| 3. Preposition | 5 |
| 4. Words often Confused | 5 |

Course Code: ELB/DMB/ECB/ENB 030

SPECIAL ENGLISH

Semester II

Prose and Language Component

Unit I:

Chapter 1 to 3 of S Radhakrishnan 'The Foundation of Civilization Ideas and Ideals

2x10 =20

Unit II

Chapter 4 to 6 of S Radhakrishnan 'The Foundation of Civilization Ideas and Ideals

2x10 =20

Unit III

Chapter 7 to 9 of S Radhakrishnan 'The Foundation of Civilization Ideas and Ideals

2x10 =20

Unit IV

Language Component:

1 Articles	5 marks
2 Punctuation	5 marks
3 Framing Questions	5 marks
4 Question tag	5 marks

Course Code: ELC/DMC/ECC/ENC 030

SPECIAL ENGLISH

Semester III

Novel and Language Component

THINGS FALL APART

CHINUA ACHEBE

Language Component

Question paper pattern:

Unit I

Two questions out of four on Characters

2x10 =20

Unit II

Two questions out of four on Plot

2x10 =20

Unit III

Two questions out of four on Theme

2x10 =20

Unit IV

Language Component

1 Comprehension

10 marks

2 Expansion of an idea

5 marks

3 Idioms and Phrases

5 marks

Course Code: ELD/DMD/ECD/END 030

SPECIAL ENGLISH

Semester IV

Drama and Language Component

Drama: THE MERCHANT OF VENICE

WILLIAM SHAKESPEARE

Question paper pattern:

Unit I

Annotations

4x5 =20

Unit II

Three questions out of four on characterisation

3x10 =30

Unit III

Two questions out of four on Theme

2x10 =20

Unit IV

Language Component

1 Paraphrase

10 marks

2 Correction of sentences

5 marks

3 Linkers

5 marks

**JSS COLLEGE OF ARTS, COMMERCE AND
SCIENCE
(Autonomous)
B N ROAD, MYSURU- 570 025**

**DEPARTMENT OF ENGLISH
Syllabus
CHOICE BASED CREDIT SYSTEM**

**AECC
LANGUAGE ENGLISH
B. VOC (II Semester)**

**2019 - 20 onwards
(Approved by BOS)**

Course code FPB/SDB/AFP 540

**B.VOC
II SEMESTER
COMMUNICATIVE SKILLS**

AIMS:

- 1 To familiarise students to basic English
- 2 To enable them to develop listening and speaking skills

OBJECTIVES: Students should be able to -

Unit I: Grammar

- | | |
|------------------------------|---------|
| 1 Subject and Verb agreement | 5 marks |
| 2 Voice | 5 marks |
| 3 Speech | 5 marks |
| 4 Question tag | 5 marks |
| 5 Framing of Question | 5 marks |

Unit II: Writing Skills

- | | |
|--|----------|
| 1 Letter Writing
(Letter of Application/ Grievances/Resume Preparation) | 10 marks |
| 2 Comprehension | 10 marks |
| 3 Essay Writing | 10 marks |

Unit III: Speaking Skills

- | | |
|--------------|----------|
| 1 Greeting | 10 marks |
| 2 Requesting | |
| 3 Enquiring | |
| 4 Explaining | |
| 5 Reporting | |
| 6 Permission | |
| 7 Thanking | |

ಕನ್ನಡ ಐಚ್ಛಿಕ
ಚತುರ್ಮಾಸ ಯೋಜನೆ

ಪಠ್ಯಕ್ರಮ ಮತ್ತು ಪರೀಕ್ಷಾ ಯೋಜನೆ 2019 ರಿಂದ 2022 ರವರೆಗೆ

ಡಿಎಸ್‌ಸಿ - 04 ಪತ್ರಿಕೆಗಳು : (ಮೊದಲನೆಯ, ಎರಡನೆಯ, ಮೂರನೆಯ ಮತ್ತು ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸಗಳು)

BA(DSC) KEG

ಸೆಮಿಸ್ಟರ್	ಪತ್ರಿಕೆ ಸಂಖ್ಯೆ	ಪತ್ರಿಕೆ ಶೀರ್ಷಿಕೆ					ಪರೀಕ್ಷಾ ಅವಧಿ (ಗಂಟೆಗಳು)	ಅಂಕಗಳು	ಆಂತರಿಕ ಮೌಲ್ಯಮಾಪನ ಅಂಕಗಳು	ಒಟ್ಟು ಅಂಕಗಳು
			ಕೋರ್ಸ್ ಕೋಡ್	ಕ್ರೆಡಿಟ್ ವಿನ್ಯಾಸ LTP	ಒಟ್ಟು ಕ್ರೆಡಿಟ್	ಬೋಧನಾ ಗಂಟೆಗಳು ಸೆಮಿಸ್ಟರ್‌ಗೆ				
ಡಿ ಎಸ್ ಸಿ - 1 ಮೊದಲನೆಯ ಚತುರ್ಮಾಸ										
1	1	ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯ ಪೂರಕ ಚಟುವಟಿಕೆ	ELA25023	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
ಡಿ ಎಸ್ ಸಿ - 2 ಎರಡನೆಯ ಚತುರ್ಮಾಸ										
2	2	ಕನ್ನಡ ಭಾಷಾ ಚರಿತ್ರೆ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELB25023	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
ಡಿ ಎಸ್ ಸಿ - 3 ಮೂರನೆಯ ಚತುರ್ಮಾಸ										
3	3	ಭಾರತೀಯ ಕಾವ್ಯಮೀಮಾಂಸೆ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELC25023	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100
ಡಿ ಎಸ್ ಸಿ - 4 ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸ										
4	4	ಛಂದಸ್ಸು ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELD25023	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 15)	100

(ಐದು ಮತ್ತು ಆರನೆಯ ಚತುರ್ಮಾಸಗಳು)

BA (DSE) KEG

ಐದನೆಯ ಚತುರ್ಮಾಸ

ಡಿ ಎಸ್ ಇ - 5.1 ಐದನೆಯ ಚತುರ್ಮಾಸ										
5	1	ಸಾಹಿತ್ಯ ವಿಮರ್ಶೆ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯ ಪೂರಕ ಚಟುವಟಿಕೆ	ELE250	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
ಡಿ ಎಸ್ ಇ - 5.2 ಐದನೆಯ ಚತುರ್ಮಾಸ										
5	2	ವಿಶೇಷ ಕವಿ ಕಾವ್ಯ ಅಧ್ಯಯನ ಹೊಸಗನ್ನಡ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELE252	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
ಡಿ ಎಸ್ ಇ - 5.3 ಐದನೆಯ ಚತುರ್ಮಾಸ										
5	3	ಕನ್ನಡ ನಾಟಕ ಸಾಹಿತ್ಯ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELE254	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
ಡಿ ಎಸ್ ಇ - 5.4 ಐದನೆಯ ಚತುರ್ಮಾಸ										
5	4	ಮಹಿಳಾ ಸಾಹಿತ್ಯ ಅಧ್ಯಯನ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELE256	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
ಡಿ ಎಸ್ ಇ - 5.5 ಐದನೆಯ ಚತುರ್ಮಾಸ										
5	5	ಕನ್ನಡ ಜನಪದ ಸಾಹಿತ್ಯ ಅಧ್ಯಯನ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELE258	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100

ಆರನೆಯ ಚತುರ್ಮಾಸ

ಡಿ ಎಸ್ ಇ - 6.1 ಆರನೆಯ ಚತುರ್ಮಾಸ										
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6	1	ಹಳೆಗನ್ನಡ ವ್ಯಾಕರಣ ಮತ್ತು ಪ್ರಯೋಗ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯ ಪೂರಕ ಚಟುವಟಿಕೆ	ELF250	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2 -15)	100
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ಡಿ ಎಸ್ ಇ - 6.2 ಅರನೆಯ ಚತುರ್ಮಾಸ

6	2	ವಿಶೇಷ ಕವಿ ಕಾವ್ಯ ಅಧ್ಯಯನ (ಹಳೆಗನ್ನಡ) ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELF252	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
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ಡಿ ಎಸ್ ಇ - 6.3 ಅರನೆಯ ಚತುರ್ಮಾಸ

6	3	ಕನ್ನಡ ಸಾಹಿತ್ಯ ಮತ್ತು ಸಿನಿಮಾ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELF254	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
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ಡಿ ಎಸ್ ಇ - 6.4 ಅರನೆಯ ಚತುರ್ಮಾಸ

6	4	ಪ್ರಕಾರ ಕೇಂದ್ರಿತ ಅಧ್ಯಯನ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELF256	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
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ಡಿ ಎಸ್ ಇ - 6.5 ಅರನೆಯ ಚತುರ್ಮಾಸ

6	5	ಕನ್ನಡ ಸಾಹಿತ್ಯ ಮತ್ತು ಸಮೂಹ ಮಾಧ್ಯಮ ಪೂರಕ ಪಠ್ಯ ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು	ELF258	03 02 01 5:1:0	06	48 32 32 80+32=112	03	40 30 - 70	30 (ಸಿ -1 - 15 ಮತ್ತು ಸಿ - 2- 15)	100
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ಭಾಷೆ ಕನ್ನಡ

ಸಿಬಿಸಿಎಸ್ ಪದ್ಧತಿಯ ಪಠ್ಯವಿನ್ಯಾಸ

2019-20ರಿಂದ ಆರಂಭವಾಗುವ ಶೈಕ್ಷಣಿಕ ವರ್ಷಕ್ಕೆ ಸಿಬಿಸಿಎಸ್ ಪದ್ಧತಿಯ ಪಾಠಕ್ರಮ ಮತ್ತು ಪರೀಕ್ಷಾ ಯೋಜನೆ

ಮುಕ್ತ ಐಚ್ಛಿಕ ಪತ್ರಿಕೆ

ಜಿ.ಇ - Generic Elective(GE) HEG

ಸೆ ಮಿ ಸ್ಟಂ ಕೆ	ಪತ್ರಿಕೆ	ಕೋರ್ಸ್ ಕೋಡ್	ಪತ್ರಿಕೆ ಶೀರ್ಷಿಕೆ	ಕ್ರೆಡಿಟ್ ವಿನ್ಯಾಸ	ಒಟ್ಟು ಕ್ರೆಡಿಟ್	ಬೋಧನಾ ಗಂಟಿಗಳ ಸಂಖ್ಯೆ (ಸೆಮಿಸ್ಟರ್‌ಗ ಲಿ)	ಪರೀಕ್ಷಾ ಅವಧಿ (ಗಂಟಿಗಳು)	ಅಂಕಗ ಳು	ಆಂತರಿಕ ಮೌಲ್ಯಮಾಪನ ಅಂಕಗಳು	ಒಟ್ಟು ಅಂಕ ಗಳು
				LTP						
ಐದನೆಯ ಚತುರ್ಮಾಸ ಕನ್ನಡ ಸಾಹಿತ್ಯ ಪರಿಚಯ										
5	1	ELE259	ಘಟಕ 1 : ಪ್ರಾಚೀನ ಕನ್ನಡ ಸಾಹಿತ್ಯ ಪರಿಚಯ	01	02	16	02	20	15 (ಸಿ -1 - 10 ಮತ್ತು ಸಿ - 2- 10)	50
			ಘಟಕ 2 : ಆಧುನಿಕ ಕನ್ನಡ ಸಾಹಿತ್ಯ ಪರಿಚಯ	01		16		15		
				2:0:0		32		35		
ಆರನೆಯ ಚತುರ್ಮಾಸ ಕನ್ನಡ ಕಲಿಕೆ - ಗಳಿಕೆ										
6	1		ಘಟಕ 1 : ಕನ್ನಡ ಸಾಹಿತ್ಯ ಕಲಿಕೆ	01	02	16	02	20	15 (ಸಿ -1 - 10 ಮತ್ತು ಸಿ - 2 -10)	50
		ಘಟಕ 2 : ಕನ್ನಡ ಜ್ಞಾನಗಳಿಕೆ	01	16		15				
				2:0:0		32		35		

ಪ್ರಶ್ನೆ ಪತ್ರಿಕೆ ಮಾದರಿ

ಮೊದಲನೆಯ, ಎರಡನೆಯ, ಮೂರನೆಯ ಮತ್ತು ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸಗಳ (ಸೆಮಿಸ್ಟರ್)

ಸಿಬಿಸಿಎಸ್ - ಪಠ್ಯಕ್ರಮದ ಭಾಷಾ ಪತ್ರಿಕೆಗಳ ಪ್ರಶ್ನೆ ಪತ್ರಿಕೆ ಮಾದರಿ

ಕಲಾ, ವಾಣಿಜ್ಯ ಮತ್ತು ವಿಜ್ಞಾನ - ಮೂರು ನಿಕಾಯಗಳಿಗೂ ಅನ್ವಯವಾಗುವಂತೆ

ಸಮಯ: 3 ಗಂಟೆಗಳು

ಗರಿಷ್ಠ

ಅಂಕಗಳು: 70

ಪದ್ಯ / ನಾಟಕ / ಕಾವ್ಯ ಭಾಗ -1 / ಕಾವ್ಯ ಭಾಗ -2

ಅಂಕಗಳು: 40

೧. ಅ) ಒಂದು ಪದ್ಯ ಭಾಗದ ಭಾವವನ್ನು ಸಂದರ್ಭ ಸಹಿತ ವಿವರಿಸಿ

1 X 5 = 5

(ಎರಡು ಪದ್ಯ ಭಾಗಗಳನ್ನು ಕೇಳುವುದು)

ಆ) ಒಂದು ವಾಕ್ಯದಲ್ಲಿ ಉತ್ತರಿಸಿ (ಐದು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

5 X 1 = 5

2. ನಾಲ್ಕು ವಾಕ್ಯಗಳನ್ನು ಸಂದರ್ಭ ಸಹಿತ ವಿವರಿಸಿ (ಆರು ವಾಕ್ಯಗಳನ್ನು ಕೇಳುವುದು)

2. 1/2 X

4= 10

3. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

1 X 10 =

10

1. ಎರಡು ಪ್ರಶ್ನೆಗಳಿಗೆ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

2 X 5 = 10

ಕಥೆ / ಕಾದಂಬರಿ / ಗದ್ಯವೈವಿಧ್ಯ

30 ಅಂಕಗಳು

2. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

1 X 10 =

10

3. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

1 X 10 =

10

4. ಎರಡನ್ನು ಕುರಿತು ಸಂಕ್ಷಿಪ್ತವಾಗಿ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

2 X 5 = 10

ಜೆ ಎಸ್ ಎಸ್ ಕಲಾ, ವಿಜ್ಞಾನ ಮತ್ತು ವಾಣಿಜ್ಯ ಕಾಲೇಜು, (ಸ್ವಾಯತ್ತ) ಬಿ. ಎನ್. ರಸ್ತೆ, ಮೈಸೂರು

ಪದವಿ ತರಗತಿಗಳ ಪಠ್ಯಕ್ರಮ

ಆಯ್ಕೆ ಆಧಾರಿತ ಗುಣಾಂಕ ಪದ್ಧತಿ (ಸಿ.ಬಿ.ಸಿ.ಎಸ್)

ಪದವಿ ಕನ್ನಡ ಐಚ್ಛಿಕ ಪತ್ರಿಕೆಗಳ ಪಠ್ಯಕ್ರಮ

ಭಾಷೆ ಕನ್ನಡ (ಸಿ.ಬಿ.ಸಿ.ಎಸ್)

ಈ ಪಠ್ಯಕ್ರಮವು 2019-20ನೇ ಶೈಕ್ಷಣಿಕ ಸಾಲಿನಿಂದ ಮೊದಲುಗೊಂಡು

ಮುಂದಿನ 2020-21, 201-22ರವರೆಗೆ ಜಾರಿಯಲ್ಲಿರುತ್ತದೆ.

ಐಚ್ಛಿಕ ಕನ್ನಡ (ಸಿ.ಬಿ.ಸಿ.ಎಸ್)

- ಅ. ಡಿ.ಎಸ್.ಸಿ - 1 ಮೊದಲನೆಯ ಚತುರ್ಮಾಸ
ಡಿ.ಎಸ್.ಸಿ - 2 ಎರಡನೆಯ ಚತುರ್ಮಾಸ
ಡಿ.ಎಸ್.ಸಿ - 3 ಮೂರನೆಯ ಚತುರ್ಮಾಸ
ಡಿ.ಎಸ್.ಸಿ - 4 ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸ
- ಆ. ಡಿ.ಎಸ್.ಇ - 5 ಐದನೆಯ ಚತುರ್ಮಾಸ (ಐದು ಐಚ್ಛಿಕ ಮಾದರಿ ಪತ್ರಿಕೆಗಳು)
ಡಿ.ಎಸ್.ಇ - 6 ಆರನೆಯ ಚತುರ್ಮಾಸ (ಐದು ಐಚ್ಛಿಕ ಮಾದರಿ ಪತ್ರಿಕೆಗಳು)

ಮೊದಲನೆಯ ಚತುರ್ಮಾಸ

ಐಚ್ಛಿಕ ಕನ್ನಡ (DSC)

ಡಿ.ಎಸ್.ಸಿ - ಪತ್ರಿಕೆ -1

Course Code:ELA 25023 ELA25026

ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ ಮತ್ತು ಪೂರಕ ಪಠ್ಯ

ಘಟಕ 1 : ಹಳಗನ್ನಡ

- ಅ) ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆಯ ಸ್ವರೂಪ, ಅಧ್ಯಯನ ಉದ್ದೇಶ ಹಾಗೂ ಮಹತ್ವ
ಕನ್ನಡ ಸಾಹಿತ್ಯದ ಪ್ರಾಚೀನತೆ: ಹಲ್ಮಿಡಿ ಶಾಸನ, ಕವಿರಾಜಮಾರ್ಗ, ವಡ್ಡಾರಾಧನೆ
- ಆ) ಚಂಪೂ ಸಾಹಿತ್ಯದ ಸ್ವರೂಪ, ಲಕ್ಷಣಗಳು
- ಇ) ಪಂಪ, ರನ್ನ, ನಾಗವರ್ಮ, ದುರ್ಗಸಿಂಹ, ನಾಗಚಂದ್ರ, ಜನ್ನ, ಆಂಡಯ್ಯ

ಘಟಕ 2 - ನಡುಗನ್ನಡ

- ಅ) ವಚನ ಮತ್ತು ದೇಸಿ ಸಾಹಿತ್ಯ : ಜೇಡರ ದಾಸಿಮಯ್ಯ, ಅಲ್ಲಮ ಪ್ರಭು, ಬಸವಣ್ಣ, ಹರಿಹರ
ರಾಘವಾಂಕ, ಚಾಮರಸ, ರತ್ನಾಕರವರ್ಣಿ, ಷಡಕ್ಷರಿ, ಸರ್ವಜ್ಞ
- ಆ) ಭಾಗವತ ಮತ್ತು ದಾಸಸಾಹಿತ್ಯ: ಕುಮಾರವ್ಯಾಸ, ಲಕ್ಷ್ಮೀಶ, ಕನಕದಾಸ,
ಪುರಂದರದಾಸ, ಸಿಂಗರಾಯ
ಜನಪದ ಕಾವ್ಯ : ಮಂಟೇಸ್ವಾಮಿ, ಮಾದಪ್ಪ, ಜುಂಜಪ್ಪ

ಘಟಕ : 3 ಹೊಸಗನ್ನಡ

- ಅ) ಅರಣೋದಯ: ಕೆಂಪುನಾರಾಯಣ, ದೇವಚಂದ್ರ, ಮುದ್ದಣ, ಶಿಶುನಾಳ ಶರೀಫ,
ಹೊಸಗನ್ನಡ ಸಾಹಿತ್ಯದ ಮುಖ್ಯಘಟ್ಟಗಳು, ನವೋದಯ, ಪ್ರಗತಿಶೀಲ, ನವ್ಯ, ದಲಿತ ಬಂಡಾಯ ಮತ್ತು
ಸ್ತ್ರೀವಾದಿ
ಪಂಥಗಳ ಸಂಕ್ಷಿಪ್ತ ಪರಿಚಯ
- ಆ) ಬಿ.ಎಂ.ಶ್ರೀ, ದ.ರಾ. ಬೇಂದ್ರೆ, ಕುವೆಂಪು, ಶಿವರಾಮ ಕಾರಂತ, ಬಸವರಾಜ ಕಟ್ಟೀಮನಿ, ಗೋಪಾಲಕೃಷ್ಣ
ಅಡಿಗ,
ಪಿ. ಲಂಕೇಶ್, ಕೆ. ಎಸ್, ನಿಸಾರ್ ಅಹಮದ್, ಸಿದ್ದಲಿಂಗಯ್ಯ, ದೇವನೂರ ಮಹಾದೇವ - ಇವರ
ಪರಿಚಯ
(ಪರೀಕ್ಷೆಯಲ್ಲಿ ಈ ಸಾಹಿತಿಗಳ ಸಾಧನೆಯ ಬಗ್ಗೆ ಮಾತ್ರವೇ ಪ್ರಶ್ನೆ ಕೇಳಬೇಕು)
- ಇ) ಮಹಿಳಾ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ: ಅಕ್ಕಮಹಾದೇವಿ, ಸಂಚಿಯ ಹೊನ್ನಮ್ಮ, ಕೊಡಗಿನ ಗೌರಮ್ಮ., ತ್ರಿವೇಣಿ,
ವೈದೇಹಿ,
ವಿಜಯಾದಬ್ಬೆ, ವೀಣಾ ಶಾಂತೇಶ್ವರ, ಸಾರಾ ಅಬೂಬಕ್ಕರ್, ಇವರ ಪರಿಚಯ
(ಪರೀಕ್ಷೆಯಲ್ಲಿ ಈ ಸಾಹಿತಿಗಳ ಸಾಧನೆಯ ಬಗ್ಗೆ ಮಾತ್ರವೇ ಪ್ರಶ್ನೆ ಕೇಳಬೇಕು)

ಪೂರಕ ಪಠ್ಯ : ಸಾಹಿತ್ಯ ಸಂಗಾತಿ

ಘಟಕ : 4 ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

1. ಹಲ್ಮಿಡಿ ಶಾಸನ - (ಭಾವಾರ್ಥ ರೂಪ)
ಕವಿರಾಜಮಾರ್ಗ (_____ವಿವೇಕಿಗಳ್ ನಾಡವರ್ಗಲ್)
ವಡ್ಡಾರಾಧನೆ (ಸುಕುಮಾರ ಸ್ವಾಮಿ ಕಥೆಯ ಆಯ್ದುಭಾಗ)
2. ಪಂಪ (ಚಾಗದ ಭೋಗದಕ್ಕರದ ಗೇಯದ)
ರನ್ನ (ಎಂತು ಮರೆವಂ ದುರ್ಯೋಧನಂ)
ಜನ್ನ (ಒಲವಾದೊಡೆ ರೂಪಿನ ಕೋಟಲೆಯೇವುದೊ)
3. ವಚನಗಳಲ್ಲಿ ಭಕ್ತಿ ಕಾಯಕ - ದಾಸೋಹ, ಸಮಾಜ ಪರಿಕಲ್ಪನೆ ಸಂಬಂಧಿ 10 ವಚನಗಳು
ಹರಿಹರ (ಕುಂಬಾರ ಗುಂಡಯ್ಯನ ರಗಳೆಯ ಆಯ್ದು ಭಾಗ)
ರಾಘವಾಂಕ (__ನಾಲ್ಕುಧಮವೊಂದಧಿಕವೆ)
4. ಕುಮಾರವ್ಯಾಸ (ಬಡವರ ಬಿನ್ನಪವನಿನ್ನಾರು ಕೇಳುವರು)
ಪುರಂದರದಾಸ (ಎಲ್ಲಾ ಲೋಳಲೊಟ್ಟೆ)
ಕನಕದಾಸ (ಕುಲ ಕುಲವೆಂದು)

ಘಟಕ : 5 ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

1. ದ.ರಾ. ಬೇಂದ್ರೆ (ನೀ ಹೀಂಗ ನೋಡಬ್ಯಾಡ ನನ್ನ)
ಗೋಪಾಲಕೃಷ್ಣ ಅಡಿಗ (ಮೋಹನ ಮುರಲಿ)
ಸಿದ್ದಲಿಂಗಯ್ಯ (ನನ್ನ ಜನಗಳು)
2. ವಿಜಯಾದಿತ್ಯ (ಕುಂತಿಯ ಬೆಳಗು)
ಸವಿತಾ ನಾಗಭೂಷಣ (ಜಾತ್ರೆಯಲ್ಲಿ ಶಿವ)
3. ಚದುರಂಗ (ಶವದ ಮನೆ - ಕತೆಯ ಆಯ್ದುಭಾಗ)
ದೇವನೂರ ಮಹದೇವ (ಮಾರಿಕೊಂಡವರು - ಕತೆಯ ಆಯ್ದುಭಾಗ)
4. ಕುವೆಂಪು (ಕಾನೂರು ಹೆಗ್ಗಡತಿ - ಕಾದಂಬರಿಯ ಆಯ್ದುಭಾಗ)
ಸಾರಾ ಅಬೂಬ್ಕರ್ - ಚಂದ್ರಗಿರಿಯ ತೀರದಲ್ಲಿ - ಕಾದಂಬರಿಯ ಆಯ್ದು ಭಾಗ)

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಸಾಹಿತ್ಯ ಚರಿತ್ರೆಯಲ್ಲಿ ಅಭ್ಯಸಿಸುವ ಕವಿಗಳ ಮೂಲ ಕೃತಿಗಳನ್ನು ವಿದ್ಯಾರ್ಥಿಗಳಿಗೆ ತೋರಿಸುವುದು.
- ಕಾವ್ಯಗಳ ಆಯ್ದುಭಾಗಗಳನ್ನು ವಾಚನ ಮತ್ತು ವ್ಯಾಖ್ಯಾನ ಮಾಡಿಸುವುದು.
- ಕವಿಗಳು ಮತ್ತು ಅವರ ಕೃತಿಗಳನ್ನು ಕುರಿತು ಪೂರಕ ಮಾಹಿತಿಯನ್ನು ಸಂಗ್ರಹಿಸುವಂತೆ ಸೂಚಿಸುವುದು.
- ಕವಿಗಳ ಕಾಲಕ್ಕೆ ಸಂಬಂಧಿಸಿದ ಐತಿಹಾಸಿಕ ಮಾಹಿತಿಗಳನ್ನು ಅನ್ವಯಿಸಿ ಶಿಸ್ತುಗಳಿಂದ (ಇತಿಹಾಸಮ, ಶಾಸನ ಇತ್ಯಾದಿ..) ಸಂಗ್ರಹಿಸುವಂತೆ ಮಾರ್ಗದರ್ಶನ ಮಾಡುವುದು.
- ತಮ್ಮ ಪರಿಸರದ ಕವಿಗಳನ್ನು ಕುರಿತು ಲೇಖನ ರಚಿಸಲು ಮಾರ್ಗದರ್ಶನ ಮಾಡುವುದು.
- ಕವಿಗಳ ಜೀವನ ಮತ್ತು ಸಾಧನೆ ಕುರಿತಂತೆ ವಿಚಾರ ಮಂಡನೆ, ಗುಂಪು ಚರ್ಚೆಗಳನ್ನು ಏರ್ಪಡಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳಿಗೆ ನನ್ನ ಮೆಚ್ಚಿನ ಕವಿ ಕೃತಿ ವಿಷಯ ಕುರಿತು ಅಭಿಪ್ರಾಯ ಮಂಡನೆಗೆ ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.
- ಕವಿಮನೆಗಳಿಗೆ ಭೇಟಿಗೆ ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳು ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳಿಗೆ ಸಂಬಂಧಿಸಿದ ದಾಖಲೆಗಳನ್ನು ಸಂಗ್ರಹಿಸುವಂತೆ ಸೂಚಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ರಂ. ಶ್ರೀ. ಮುಗಳಿ
2. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ ಸಂಪುಟಗಳು : ಕನ್ನಡ ಅಧ್ಯಯನ ಸಂಸ್ಥೆ ಪ್ರಕಟಣೆ
3. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಸಮಗ್ರ ಚರಿತ್ರೆ ಸಂಪುಟಗಳು : ಬೆಂಗಳೂರು ವಿಶ್ವವಿದ್ಯಾಲಯ
4. ಸಾಮಾನ್ಯನಿಗೆ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ ಸಂಪುಟಗಳು : ಬೆಂಗಳೂರು ವಿಶ್ವವಿದ್ಯಾಲಯ
5. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ಕೆ. ವೆಂಕಟರಾಮಪ್ಪ
6. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ತ.ಸು. ಶಾಮರಾಯ
7. ಹೊಸಗನ್ನಡ ಕವಿತೆಯ ಮೇಲೆ ಇಂಗ್ಲಿಷ್ ಕಾವ್ಯದ ಪ್ರಭಾವ : ಎಸ್. ಅನಂತನಾರಾಯಣ
8. ಸಂಕ್ಷಿಪ್ತ ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ಎಂ. ಮರಿಯಪ್ಪ ಭಟ್ಟ
9. ಯುಗಧರ್ಮ ಹಾಗೂ ಸಾಹಿತ್ಯ ದರ್ಶನ : ಕೀರ್ತಿನಾಥ ಕುರ್ತುಕೋಟೆ
10. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಸಂಗಾತಿ : ಕೀರ್ತಿನಾಥ ಕುರ್ತುಕೋಟೆ
11. ಹೊಸಗನ್ನಡ ಅರುಣೋದಯ : ಶ್ರೀನಿವಾಸ ಹಾವನೂರ
12. ಸಾಲುದೀಪಗಳು : ಕರ್ನಾಟಕ ಸಾಹಿತ್ಯ ಅಕಾಡೆಮಿ
13. ಹೊಸಗನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ಎಲ್ . ಎಸ್. ಶೇಷಗಿರಿರಾವ್
14. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ ಚಾರಿತ್ರಿಕ ಬೆಳವಣಿಗೆ : ಸಿ. ವೀರಣ್ಣ
15. ಮಹಿಳಾ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ಎಚ್. ಎಸ್, ಶ್ರೀಮತಿ
16. ಕನ್ನಡ ಸಾಹಿತ್ಯಮೀಮಾಂಸೆ : ರಹಮತ್ ತರೀಕೆರೆ
17. ವಚನ ಚಿಂತನ ಮಾಲೆ ಸಂಪುಟಗಳು : ಬರಗೂರು ರಾಮಚಂದ್ರಪ್ಪ

18. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಕೋಶ : ರಾಜಪ್ಪ ದಳವಾಯಿ
19. ಕನ್ನಡ ರಾಷ್ಟ್ರೀಯತೆ : ಬಂಜೆಗೆರೆ ಜಯಪ್ರಕಾಶ
20. ಕರ್ನಾಟಕ ಸಮಗ್ರ ತತ್ವಪದಗಳ ಸಂಪುಟಗಳು : ಪ್ರ. ಸಂ: ಕಾ. ತ. ಚಿಕ್ಕಣ್ಣ

ಎರಡನೆಯ ಚತುರ್ಮಾಸ

ಐಚ್ಛಿಕ ಕನ್ನಡ(DSC)

Course Code:ELA 25023 ELB25026

ಡಿ.ಎಸ್.ಸಿ - ಪತ್ರಿಕೆ -2

ಕನ್ನಡ ಭಾಷಾ ಚರಿತ್ರೆ ಮತ್ತು ಪೂರಕ ಪಠ್ಯ

ಘಟಕ 1 :

- ಅ) ಕನ್ನಡ ಭಾಷೆ ಸ್ವರೂಪ, ಲಕ್ಷಣ ಮತ್ತು ವ್ಯಾಖ್ಯಾನಗಳು
- ಆ) ಧ್ವನ್ಯಂಗಗಳ ಪರಿಚಯ, ಧ್ವನಿ, ಧ್ವನಿಮಾ, ಆಕೃತಿಮಾ, ಪದ, ವಾಕ್ಯ - ಇವುಗಳ ಸಂಕ್ಷಿಪ್ತ ವಿವರಣೆ
- ಇ) ಭಾಷಾ ಪರಿವಾರದ ಕಲ್ಪನೆ, ಭಾರತ ದೇಶದಲ್ಲಿರುವ ಪ್ರಮುಖ ಭಾಷಾಪರಿವಾರಗಳು, ದ್ರಾವಿಡ ವರ್ಗದ ಪರಿಚಯ ಮತ್ತು ಲಕ್ಷಣ, ಸಂಸ್ಕೃತ ಮತ್ತು ಕನ್ನಡ ಭಾಷೆಯ ಸಂಬಂಧ

ಘಟಕ 2 - ನಡುಗನ್ನಡ

- ಅ) ಕನ್ನಡ ಪದದ ನಿಷ್ಪತ್ತಿ, ಕನ್ನಡ ಭಾಷೆಯ ವಿವಿಧ ಹಂತಗಳು; ಪೂರ್ವದ ಹಳಗನ್ನಡ, ನಡುಗನ್ನಡ ಮತ್ತು ಹೊಸಗನ್ನಡ, ವಿದ್ವಾಂಸರ ಅಭಿಪ್ರಾಯಗಳು
- ಆ) ಭಾಷೆಯ ವಿವಿಧ ಅಂಗಗಳು: ಕನ್ನಡ ವರ್ಣಮಾಲೆ, ನಾಮಪದ, ಲಿಂಗ, ಮಹತ್, ಅಹಮತ್, ಪುಲ್ಲಿಂಗ, ಸ್ತ್ರೀಲಿಂಗ, ನಪುಂಸಕಲಿಂಗ, ವಚನ: ಏಕವಚನ, ಬಹುವಚನ, ವಿಭಕ್ತಿ, ಏಳು ವಿಭಕ್ತಿಗಳ ಪ್ರತ್ಯಯಗಳು, ಸರ್ವನಾಮದ ಬಗೆಗಳು, ಸಸಮ್ಮುಖ - ಅಸಮ್ಮುಖ (ಅಭಿವ್ಯಾಪಕ - ವ್ಯಾವರ್ತಕ)
- ಇ) ಪುರುಷತ್ರಯಗಳು: ಉತ್ತಮ, ಮಧ್ಯಮ ಪ್ರಥಮ, ಕ್ರಿಯಾಪದ, ಕಾಲತ್ರಯಗಳು, (ಭೂತ ವರ್ತಮಾನ ಭವಿಷ್ಯತ್) ನಿಷೇಧಾರ್ಥಕ, ವಿಧ್ಯರ್ಥಕ, ಪ್ರೇರಣಾರ್ಥಕ, ಸಂಭಾವನಾರ್ಥಕ, ಕರ್ಮಣಿ ಪ್ರಯೋಗ, ಗುಣವಾಚಕ, ಸಂಖ್ಯಾವಾಚಕ ಮತ್ತು ಅವ್ಯಯಗಳ ಸ್ಥೂಲ ಸಮೀಕ್ಷೆ

ಘಟಕ : 3

- ಅ) ಪ್ರಾದೇಶಿಕ ಉಪಭಾಷೆಗಳು: ಉಪಭಾಷೆಗಳು ಉಂಟಾಗಲು ಕಾರಣಗಳು, ಮೈಸೂರು ಕನ್ನಡ, ಮಂಗಳೂರು (ಕರಾವಳಿ) ಕನ್ನಡ, ಧಾರವಾಡ ಕನ್ನಡ; ವ್ಯಕ್ತಿ ಭಾಷೆ, ಶಿಷ್ಟಭಾಷೆ, ಆಡುಭಾಷೆ ಮತ್ತು ಲಿಖಿತ ಭಾಷೆ ಇವುಗಳ ಸ್ಥೂಲ ಪರಿಚಯ
- ಆ) ವಿಶಿಷ್ಟ ಪದಗಳ ನಿಷ್ಪತ್ತಿ

ಪೂರಕ ಪಠ್ಯ : ಸಾಹಿತ್ಯ ಸಂಗಾತಿ

ಘಟಕ : 4 ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

1. ಬಾದಾಮಿ ಶಾಸನ
2. ವಡ್ಡಾರಾಧನೆ : ಬ್ರಾಜಿಷ್ಣು
3. ತ್ರಿಷಷ್ಟಿ ಲಕ್ಷಣ ಮಹಾಪುರಾಣ : ಚಾವುಂಡರಾಯ

4. ಜಗನ್ನಾಥ ವಿಜಯ : ರುದ್ರಭಟ್ಟ
5. ಪಂಚತಂತ್ರ : ದುರ್ಗಸಿಂಹ
6. ಧರ್ಮಾಮೃತ : ನಯಸೇನ

ಘಟಕ : 5 ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ಕೆ ಭಾಗಗಳು)

1. ಮುದ್ರಾಮಂಜೂಷ : ಕೆಂಪುನಾರಾಯಣ
2. ರಾಮಾಶ್ವಮೇಧ : ಮುದ್ದಣ
3. ವಾಣಿಮಾಯಿ : ವೈದೇಹಿ
4. ರುಚಿ : ಸುನಂದ ಬೆಳಗಾವಕರ್
5. ಅಮಾಸ : ದೇವನೂರ ಮಹದೇವ
6. ಕರುಣ ಗಂಗೆ : ಪ್ರಭುಶಂಕರ

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಪ್ರಾಚೀನ ಕನ್ನಡ ಭಾಷೆಯಲ್ಲಿನ ಪ್ರಮುಖ ಭಾಷಾ ವೈವಿಧ್ಯತೆಗಳನ್ನು ಗುರುತಿಸಲು ಸೂಚಿಸುವುದು.
- ಸ್ಥಳೀಯ ಭಾಷೆಯ ಸಾಮಾಜಿಕ ಭೇದಗಳನ್ನು ಗುರುತಿಸಲು ಸೂಚಿಸುವುದು
- ಭಾಷೆಯಲ್ಲಿನ ಪ್ರಾದೇಶಿಕ ಅಂಶಗಳನ್ನು ಗುರುತಿಸಿ ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು.
- ವ್ಯಾಕರಣದ ಸರಿಯಾದವನ್ನು ತಿಳಿಸುವ ಚಟುವಟಿಕೆಗಳನ್ನು ನಡೆಸುವುದು.
- ವ್ಯಾಕರಣದ ಸರಿಯಾದವನ್ನು ತಿಳಿಸುವ ಚಟುವಟಿಕೆಗಳನ್ನು ನಡೆಸುವುದು.
- ದೇಶ್ಯ, ಅನ್ಯದೇಶ್ಯ, ಅನ್ಯಭಾಷೆಯ ಪದಗಳನ್ನು ಗುರುತಿಸಿ ಸಂಗ್ರಹಿಸಲು ಪ್ರೋತ್ಸಾಹಿಸುವುದು.
- ಭಾಷೆಯಲ್ಲಿನ ನುಡಿಗಟ್ಟು ಗಾಡೆ, ಒಗಟು, ದ್ವಿರೂಪಿಗಳು, ತತ್ಸಮ, ತದ್ಭವ, ಸಮಾಸ ಪದಗಳನ್ನು ಗುರುತಿಸುವುದು.
- ಭಾಷಾ ವೈಶಿಷ್ಟ್ಯಗಳನ್ನು ಅರಿಯುವುದು ಮತ್ತು ಸಂಗ್ರಹಿಸುವುದು. ವಿಶಿಷ್ಟ ಪದಗಳನ್ನು ಗುರುತಿಸುವುದು, ಈ ಬಗ್ಗೆ ಅಭಿಪ್ರಾಯ ಮಂಡನೆಗೆ ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಕನ್ನಡ ಭಾಷಾಶಾಸ್ತ್ರ : ಆರ್. ವೈ. ಧಾರವಾಡಕರ
2. ಹಿನ್ನರಿ ಆಫ್ ದಿ ಕನ್ನಡ ಲ್ಯಾಂಗ್ವೇಜ್ : ಆರ್. ನರಸಿಂಹಾಚಾರ್
3. ಭಾಷಾವಿಜ್ಞಾನದ ಮೂಲ ತತ್ವಗಳು : ಎಂ. ಚಿದಾನಂದ ಮೂರ್ತಿ
4. ಸಂಕ್ಷಿಪ್ತ ಕನ್ನಡ ಭಾಷೆಯ ಚರಿತ್ರೆ : ಡಿ. ಎನ್. ಶಂಕರ ಭಟ್
5. ಕನ್ನಡ ಭಾಷೆಯ ಸಂಕ್ಷಿಪ್ತ ಚರಿತ್ರೆ : ಎಂ. ಎಚ್. ಕೃಷ್ಣಯ್ಯ

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| 6. ಸಾಮಾನ್ಯ ಭಾಷಾವಿಜ್ಞಾನ | ; ಕೆ. ಕೆಂಪೇಗೌಡ |
| 7. ಉಪಭಾಷಾ ವಿಜ್ಞಾನ | : ಶಾಲಿನಿ ರಘುನಾಥ ಶೆಟ್ಟಿ |
| 8. ಇಂದಿನ ಕನ್ನಡ ರಚನೆ ಮತ್ತು ಬಳಕೆ | : ಎಚ್. ಎನ್. ಶ್ರೀಧರ್ |
| 9. ಕನ್ನಡ ಜಗತ್ತು ಅರ್ಥ ಶತಮಾನ | : ಕೆ. ವಿ. ನಾರಾಯಣ |
| 10. ಕನ್ನಡ ಮಧ್ಯಮ ವ್ಯಾಕರಣ | : ತೀ.ನಂ.ಶ್ರೀ |
| 11. ಕನ್ನಡ ಭಾಷಾಲೋಕ | : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ |
| 12. ಕನ್ನಡ ವ್ಯಾಕರಣ ಮತ್ತು ಭಾಷಾರಚನೆ | : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ |
| 13. ಹೊಸಗನ್ನಡ ವ್ಯಾಕರಣ ಸಂವೇದನೆಗಳು | : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ |

ಮೂರನೆಯ ಚತುರ್ಮಾಸ

ಐಚ್ಛಿಕ ಕನ್ನಡ(DSC)

Course Code:ELC 25023 ELC25026

ಡಿ.ಎಸ್.ಸಿ - ಪತ್ರಿಕೆ -3

ಭಾರತೀಯ ಕಾವ್ಯಮೀಮಾಂಸೆ ಮತ್ತು ಪೂರಕ ಪಠ್ಯ

ಘಟಕ 1 :

ಅ) ಭಾರತೀಯ ಕಾವ್ಯಮೀಮಾಂಸೆಯ ಉಗಮ ಮತ್ತು ವಿಕಾಸ: ಭರತ, ಭಾಮಹ, ದಂಡಿ, ವಾಮನ, ಆನಂದವರ್ಧನ, ಕುಂತಕ, ಕ್ಷೇಮೇಂದ್ರ, ಈ ಪ್ರಮುಖ ಅಲಂಕಾರಿಕರು ಮತ್ತು ಅವರ ಕೃತಿಗಳ ಸಂಕ್ಷಿಪ್ತ ಪರಿಚಯ.

ಆ) ಕಾವ್ಯದ ಆಕರ (ಪ್ರತಿಭೆ) ಮತ್ತು ಪರಿಕರ (ವ್ಯುತ್ಪತ್ತಿ)ಗಳು, ಕಾವ್ಯ, ಲಕ್ಷಣ, ಕಾವ್ಯ ಪ್ರಯೋಜನಗಳು, ಕವಿ ಮತ್ತು ಸಹೃದಯ

ಘಟಕ 2 - ಭಾರತೀಯ ಕಾವ್ಯಮೀಮಾಂಸೆಯ ಪ್ರಮುಖ ಪ್ರಸ್ಥಾನಗಳು

ಅ) ಅಲಂಕಾರ ಪ್ರಸ್ಥಾನ : ಭಾಮಹನನ್ನು ಮುಖ್ಯವಾಗಿಟ್ಟುಕೊಂಡು ಅಲಂಕಾರ ಪ್ರಸ್ಥಾನದ ಸಾಮಾನ್ಯ ಸಮೀಕ್ಷೆ

ಶಬ್ದಾಲಂಕಾರಗಳು : ಅನುಪ್ರಾಸ, ಯಮಕ

ಅರ್ಥಾಲಂಕಾರಗಳು : 1. ಉಪಮೆ 2. ದೀಪಕ 3. ರೂಪಕ 4. ದೃಷ್ಟಾಂತ

5. ಉತ್ಪ್ರೇಕ್ಷೆ 6. ಅರ್ಥಾಂತರನ್ಯಾಸ 7. ಶ್ಲೇಷ

8. ಅತಿಶಯೋಕ್ತಿ 9. ಸ್ವಭಾವೋಕ್ತಿ

ಆ) ರೀತಿ (ಶೈಲಿ) ಪ್ರಸ್ಥಾನ : ವಾಮನನ್ನು ಮುಖ್ಯವಾಗಿಟ್ಟುಕೊಂಡು ರೀತಿ ಪ್ರಸ್ಥಾನದ ಸಾಮಾನ್ಯ ಸಮೀಕ್ಷೆ

ಘಟಕ : 3

ಅ) ಧ್ವನಿಪ್ರಸ್ಥಾನ : ಶಬ್ದ, ಅರ್ಥ, ವಾಚ್ಯಾರ್ಥ - ಅಭಿದಾವೃತ್ತಿ, ಲಕ್ಷಾರ್ಥ - ಲಕ್ಷಣಾವೃತ್ತಿ,

ವ್ಯಂಗ್ಯಾರ್ಥ - ವ್ಯಂಜನಾವೃತ್ತಿ; ಆನಂದವರ್ಧನನನ್ನು ಮುಖ್ಯವಾಗಿಟ್ಟುಕೊಂಡು ಧ್ವನಿಯ ಸಾಮಾನ್ಯ ಸ್ವರೂಪದ ಪರಿಚಯ, ಧ್ವನಿ ಪ್ರಭೇದಗಳು: ವಸ್ತುಧ್ವನಿ, ಅಲಂಕಾರ ಧ್ವನಿ, ರಸಧ್ವನಿ

ಆ) ರಸಪ್ರಸ್ಥಾನ : ಭಾವ, ಸ್ಥಾಯೀಭಾವ, ವಿಭಾವ, ಅನುಭಾವ, ಸಂಚಾರಿ ಭಾವಗಳು. ಅನುಕಾರ್ಯ,

ಅನುಕರ್ತ ಮತ್ತು ಸಾಮಾಜಿಕರು, ಭರತನ ರಸಸೂತ್ರ, ಭಟ್ಟಲೋಲ್ಲಟ, (ಉತ್ಪತ್ತಿವಾದ) ಶ್ರೀಶಂಕುಕ (ಅನುಮಿತಿವಾದ) ಭಟ್ಟನಾಯಕ (ಭುಕ್ತಿವಾದ) ಮತ್ತು ಅಭಿನವಗುಪ್ತ (ವ್ಯಕ್ತಿವಾದ) ಕಾವ್ಯದ ಆತ್ಮವಾಗಿ ರಸ, ರಸ ಪ್ರಭೇದಗಳು

ಇ) ವಕ್ರೋಕ್ತಿ ಮತ್ತು ಔಚಿತ್ಯ ಪ್ರಸ್ಥಾನಗಳ ಸಾಮಾನ್ಯ ಪರಿಚಯ

ಪೂರಕ ಪಠ್ಯ : ಮೀಮಾಂಸೆಯ ಸಂಗಾತಿ

ಘಟಕ : 4 ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

ಅ) ಅಲಂಕಾರ ಕಾವ್ಯ 1. ಕಮಲದಳ ನಯನ : ಲಕ್ಷ್ಮೀಶನ ಜೈಮಿನಿಭಾರತ

2. ಕಣ್ಣೋಟಿ ಶಿವನ ಕೈಯಲಗು : ಜನಪದ

3. ಬೆಳಗು : ಬೇಂದ್ರೆ ಹಾಗೂ ಇನ್ನಿತರ ಕವಿಗಳ ಆಯ್ದು ಕವಿತೆಗಳು

ಆ) ಧ್ವನಿಕಾವ್ಯ

1. ಸೂಳ್ನಡೆಯಲಪ್ಪುದು ಕಾಣಾ ಮಹಾಜಿರಂಗದೊಳ್: ಪಂಪಭಾರತ

2. ಸುತ್ತಿ ಸುತ್ತಿ ಬರುತಿಯಾ ಯಾತಕ್ಕಾಗಿ : ತವರ ಬಣ್ಣ : ಜನಪದ

3. ಕಲ್ಪಿ : ಕುವೆಂಪು ಹಾಗೂ ಇನ್ನಿತರ ಕವಿಗಳ ಆಯ್ದು ಕವಿತೆಗಳು

ಘಟಕ : 5 ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

ಅ) ರಸ ಕಾವ್ಯ: 1, ನೀರೊಳಗಿದುರ್ಬಿ ಬೆಮರ್ತನ್ ಉರಗಪತಾಕನ್ : ರನ್ನನ ಗದಾಯುದ್ಧ

2. ಬಡವರಿಗೆ ಸಾವ ಕೊಡಬೇಡ - ಜನಪದ

3. ಯದುಗಿರಿಯ ಮೌನವಿಕಾಸ : ಪುತಿನ ಮತ್ತು ಇನ್ನಿತರ ಕವಿಗಳ ಆಯ್ದ ಕವಿತೆಗಳು

ಅ) ಔಚಿತ್ಯ ಕಾವ್ಯ : 1. ದಶಾಸ್ಕನೀಕ್ಷಿಸಿದಂ : ನಾಗಚಂದ್ರನ ರಾಮಚಂದ್ರ ಚರಿತ ಪುರಾಣಂ

2. ನರಸರ ಜುಂಜುಹೊಳೆಯಲಿ ಹಾರುವನೇ ಜಾಹ್ನವೀಧರ : ಕುಮಾರವ್ಯಾಸ

3. ಸಾವಿರ ಕೊಳ್ಳಿ ಒಲೆಯಾಗ : ಜನಪದ

4. ಅಮ್ಮ ಆಚಾರ ಮತ್ತು ನಾನು ಕೆ. ಎಸ್. ನಿಸಾರ್ ಹಾಗೂ ಇತರ ಕವಿಗಳ ಆಯ್ದ

ಕವಿತೆಗಳು

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಕೆಲವು ಅಲಂಕಾರಗಳನ್ನು ಬಳಸಿ ಪದ್ಯ ರಚನೆ ಮಾಡಿಸುವುದು.
- ಭಾರತೀಯ ಅಲಂಕಾರಿಕರ ಜೊತೆಗೆ ಕನ್ನಡ ಮೀಮಾಂಸೆಯ ಅಲಂಕಾರಿಕರನ್ನು ಗುರುತಿಸಲು ಸೂಚಿಸುವುದು
- ಕೆಲವು ಪ್ರಮುಖ ಕವಿತೆಗಳನ್ನಿಟ್ಟುಕೊಂಡು ಧ್ವನಿ, ರಸ, ರೀತಿಗಳನ್ನು ಗುರುತಿಸುವುದು
- ಕಾವ್ಯದ ಚರ್ಚೆಗೆ ಅವಕಾಶ ಕಲ್ಪಿಸಿಕೊಡುವುದು
- ಕಾವ್ಯವನ್ನು ರಸಕ್ಕೆ ಅನುಗುಣವಾಗಿ ಓದಿಸುವುದು
- ಕಾವ್ಯಮೀಮಾಂಸೆಯ ಪರಿಕಲ್ಪನೆಗಳ ಹಿನ್ನೆಲೆಯಲ್ಲಿ ವಿದ್ಯಾರ್ಥಿಗಳಲ್ಲಿ ಕವಿತೆ, ಗದ್ಯ, ಕಥೆ, ಪ್ರಬಂಧ ಇತ್ಯಾದಿ ಸಾಹಿತ್ಯ ಪ್ರಕಾರಗಳ ರಚನೆಗೆ ಪ್ರೇರೇಪಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

- | | |
|--------------------------------|---|
| 1. ಭಾರತೀಯ ಕಾವ್ಯಮೀಮಾಂಸೆ | : ತೀ.ನಂ.ಶ್ರೀ |
| 2. ಕನ್ನಡ ಕೈಪಿಡಿ (ಸಂ 1 ಮತ್ತು 2 | : ಮೈಸೂರು ವಿ.ವಿ |
| 3. ಅಲಂಕಾರ ಶಾಸ್ತ್ರ | : ಬಿ.ಕೆ ಶಿವರಾಮಯ್ಯ |
| 4. ತೌಲನಿಕ ಕಾವ್ಯಮೀಮಾಂಸೆ | : ಎಚ್. ತಿಪ್ಪೇರುದ್ರಸ್ವಾಮಿ |
| 5. ಕಾವ್ಯಾರ್ಥ ಪದಕೋಶ | : ಜಿ.ಎಸ್. ಶಿವರುದ್ರಪ್ಪ ಮತ್ತು ಕೆ. ವಿ., ನಾರಾಯಣ |
| 6. ಕಾವ್ಯಾರ್ಥ ಚಿಂತನ | : ಜಿ. ಎಸ್. ಶಿವರುದ್ರಪ್ಪ |
| 7. ಭಾರತೀಯ ಕಾವ್ಯಶಾಸ್ತ್ರ ಪರಿಭಾಷೆ | : ಮಲ್ಲೇಪುರಂ ಜಿ. ವೆಂಕಟೇಶ |
| 8. ಕಾವ್ಯಪರಿಕರಗಳು | : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ |

ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸ
ವಿಚ್ಛಿಕ ಕನ್ನಡ
ಡಿ.ಎಸ್.ಸಿ - ಪತ್ರಿಕೆ -4(DSC)
Course Code: ELD 25023 ELD25026

ಛಂದಸ್ಸು ಮತ್ತು ಪೂರಕ ಪಠ್ಯ

ಘಟಕ 1 :

- ಅ) ಛಂದಸ್ಸು - ಸಾಮಾನ್ಯ ಸ್ವರೂಪ, ಅಧ್ಯಯನ ಅಗತ್ಯ, ಗುರು, ಲಘು ಮಾತ್ರಗಳ ಸ್ವರೂಪ,
ಆ) ಪ್ರಾಸದ ಸ್ವರೂಪ, ವಿಧಗಳು, ಆದಿ ಪ್ರಾಸ (ಸಿಂಹ, ಗಜ, ವೃಷಭ, ಶರಭ, ಅಜ, ಹಯ)
ಅಂತ್ಯಪ್ರಾಸ, ಯತಿ - ಇವುಗಳ ಪರಿಚಯ
ಇ) ವರ್ಣವೃತ್ತಗಳು - ಖ್ಯಾತ ಕರ್ಣಾಟಕಗಳು : ಚಂಪಕ, ಉತ್ಪಲ, ಮತ್ತೇಭ, ಶಾರ್ದೂಲ, ಸ್ತಗ್ಧರಾ,
ಮಹಾಸ್ತಗ್ಧರಾ ವೃತ್ತಗಳು

ಘಟಕ 2 :

- ಅ) ಮಾತ್ರಾವೃತ್ತಗಳು : ಕಂದ, ರಗಳೆ, ಉತ್ಸಾಹ, ಮಂದಾನಿಲ, ಲಲಿತ ರಗಳೆಗಳು, ತ್ರಿಪದಿ ಮತ್ತು
ಷಟ್ಪದಿಯ ಪರಿಚಯದೊಡನೆ ಶರ, ಕುಸುಮ, ಭೋಗ, ಭಾಮಿನಿ, ಪರಿವರ್ಧಿನಿ, ವಾರ್ಧಕ ಮತ್ತು
ಉದ್ಧಂಡ ಈ ಮಾತ್ರಾ ಷಟ್ಪದಿಗಳ ಪರಿಚಯ ಇವುಗಳ ಲಕ್ಷಣ ಮತ್ತು ಸಂಕ್ಷಿಪ್ತ ಇತಿಹಾಸ

ಅ) ಅಂಶವೃತ್ತಗಳು ; ಗಣಗಳು, ಬ್ರಹ್ಮ, ವಿಷ್ಣು, ರುದ್ರ, ಆಯಾ ಗಣಗಳಿಗೆ ಪ್ರಸ್ತಾರ ಹಾಕುವ ವಿಧಾನ,
ತ್ರಿಪದಿ, ಪಿರಿಯಕ್ಕರ, ಷಟ್ಪದಿ ಮತ್ತು ಸಾಂಗತ್ಯ - ಇವುಗಳ ಲಕ್ಷಣ ಮತ್ತು ಸಂಕ್ಷಿಪ್ತ ಇತಿಹಾಸ

ಘಟಕ : 3

ಅ) ಹೊಸಗನ್ನಡ ಛಂದಸ್ಸು : ಗಣಗಳು ಮತ್ತು ವಿಭಿನ್ನ ಲಯಗಳೂ - ಪುತ, ಕಂಪಿತ,
ಮಾನ, ಗಣಪರಿವೃತ್ತಿ, ಅನಾಗತ, ಮುಡಿ, ಪದ್ಯಗಣ, ವಿಷಮಗಣ ಇವುಗಳ ಸಂಕ್ಷಿಪ್ತ ಪರಿಚಯ.
ಇ) ಹೊಸಗನ್ನಡ ಛಂದಸ್ಸಿನ ಸ್ವರೂಪಗಳು : ಸುನೀತ, ಸರಳ ರಗಳೆ, ಪ್ರಗಾಥ, ಮುಕ್ತಛಂದಸ್ಸು

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ಛಂದೋ ಸಂಗಾತಿ

ಘಟಕ : 4 ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

ಕಾವ್ಯಭಾಗಗಳು:

1. ಖ್ಯಾತ ಕರ್ಣಾಟಕಗಳು (ಕಾವ್ಯದಿಂದ ಆಯ್ದುಪದ್ಯಗಳು)
2. ಕಂದ, ರಗಳೆ ಮತ್ತು ಷಟ್ಪದಿ (ಕಾವ್ಯದಿಂದ ಆಯ್ದುಪದ್ಯಗಳು)
3. ತ್ರಿಪದಿ ಮತ್ತು ಸಾಂಗತ್ಯ (ಕಾವ್ಯದಿಂದ ಆಯ್ದುಪದ್ಯಗಳು)
4. ಹೊಸಗನ್ನಡ ಛಂದಸ್ಸು (ಕಾವ್ಯದಿಂದ ಆಯ್ದುಪದ್ಯಗಳು)

ಘಟಕ : 5 ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

ಲೇಖನಗಳು

1. ಡಿ.ಎಸ್. ಕರ್ಕಿಯವರ ಕನ್ನಡ ಛಂದೋವಿಕಾಸ ಕೃತಿಯಿಂದ ಆಯ್ದು ಭಾಗ
2. ತೀ.ನಂ.ಶ್ರೀ ಅವರ ಸಮಾಲೋಕನ ಕೃತಿಯಿಂದ ಆಯ್ದುಭಾಗ
3. ಟಿ.ವಿ.ವೆಂಕಟಾಚಲ ಶಾಸ್ತ್ರಿ ಅವರ ಕನ್ನಡ ಛಂದಃಸ್ವರೂಪ ಕೃತಿಯಿಂದ ಆಯ್ದುಭಾಗ
4. ಸಾ.ಶಿ ಮರುಳಯ್ಯನವರ ವರಸೆಗಳು ಕೃತಿಯಿಂದ ಆಯ್ದುಭಾಗ

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಹಳಗನ್ನಡ ಪದ್ಯಗಳಿಗೆ ಪ್ರಸ್ತಾರ ಹಾಕಿ ಛಂದೋಪ್ರಕಾರವನ್ನು ಗುರುತಿಸುವುದು.
- ಪದ್ಯಗಳನ್ನು ಬಿಡಿಸಿಬಿಡಿಸುವುದು
- ತಾಳ ಲಯ ಗತಿಗಳನ್ನು ಅನುಸರಿಸಿ ಕವಿತೆಗಳನ್ನು ರಾಗಬದ್ಧವಾಗಿ ಹಾಡುವಂತೆ ಪ್ರೇರೇಪಿಸುವುದು.
- ಹಳಗನ್ನಡ ಛಂದಸ್ಸಿಗೆ ಅನ್ವಯ ಪದ್ಯ ರಚನೆ ಮಾಡುವುದು
- ಹೊಸಗನ್ನಡ ಛಂದಸ್ಸನ್ನು ಅನ್ವಯಿಸಿ ಪದ್ಯ ರಚನೆ

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಕನ್ನಡ ಛಂದೋವಿಕಾಸ : ಡಿ. ಎಸ್. ಕರ್ಕಿ
2. ಸಮಾಲೋಕನ : ತೀ.ನಂ.ಶ್ರೀ

- | | |
|---|----------------------------|
| 3. ಕನ್ನಡ ಛಂದಸ್ಸಿನ ಚರಿತ್ರೆ ಭಾಗ 1 ಮತ್ತು 2 | : ಸಂ ಸಿ.ಪಿ.ಕೆ |
| 4. ಕನ್ನಡ ಛಂದಃಸ್ವರೂಪ | : ಟಿ. ವಿ. ವೆಂಕಟಾಚಲಶಾಸ್ತ್ರಿ |
| 5. ಛಂದೋಮಿತ್ರ | : ಅ.ರಾ.ಮಿತ್ರ |
| 6. ಕನ್ನಡ ಛಂದೋದರ್ಪಣ | : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ |
| 7. ಕನ್ನಡ ಛಂದಃಸ್ತಂಪದ | : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ |
| 8. ಕನ್ನಡ ಛಂದಸ್ಸಿನ ಬೆಳವಣಿಗೆ ಮತ್ತು ಇತಿಹಾಸ | : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ |

ಐದನೆಯ ಚತುರ್ಮಾಸ
(ಆಯ್ಕೆ ಮಾದರಿ ಶಿಸ್ತು ಡಿ.ಎಸ್.ಇ ಅಧ್ಯಯನ)
ಐಚ್ಛಿಕ ಕನ್ನಡ
ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -5.1(DSE)
Course code : ELE25023 ELE25026

ಸಾಹಿತ್ಯ ವಿಮರ್ಶೆ ಮತ್ತು ಪೂರಕ ಪಠ್ಯ

ಘಟಕ 1

- ಅ) ಸಾಹಿತ್ಯ ವಿಮರ್ಶೆಯ ಸ್ವರೂಪ ಮತ್ತು ಲಕ್ಷಣಗಳು
- ಆ) ಸಾಹಿತ್ಯ ವಿಮರ್ಶೆ ಬೆಳೆದು ಬಂದ ದಾರಿ, ಪ್ಲೇಟೋ, ಅರಿಸ್ಟಾಟಲ್, ಹೊರೇಸ್, ಲಾಂಜೈನಸ್, ಕೋಲರಿಜ್,
- ಐ.ಎ. ರಿಚರ್ಡ್ಸ್, ಟಿ. ಎನ್. ಇಲಿಯಟ್ ಇವರ ಪ್ರಮುಖ ಸಿದ್ಧಾಂತಗಳನ್ನು ಪರಿಚಯಿಸುವುದು.

ಘಟಕ 2 - ಆಧುನಿಕ ವಿಮರ್ಶಾ ಪ್ರಸ್ಥಾನಗಳ ಸ್ಥೂಲ ಪರಿಚಯ

ಕರ್ತೃನಿಷ್ಠ ವಿಮರ್ಶೆ, ಕೃತಿನಿಷ್ಠ, ಮನೋವೈಜ್ಞಾನಿಕ, ಚಾರಿತ್ರಿಕ, ಮಾರ್ಕ್ಸವಾದಿ ಮತ್ತು ಸ್ತ್ರೀ ನಿಷ್ಠ ವಿಮರ್ಶೆ

ಘಟಕ : 3

ಕನ್ನಡ ಸಾಹಿತ್ಯ ವಿಮರ್ಶೆ

- ಅ) ನವೋದಯ ವಿಮರ್ಶೆ ಸ್ವರೂಪ, ಲಕ್ಷಣ, ನಿಲುವುಗಳು - ಪ್ರಮುಖ ವಿಮರ್ಶಕರನ್ನು ಅನುಲಕ್ಷಿಸಿ
- ಆ) ನವ್ಯ ವಿಮರ್ಶೆ - ಪ್ರಭಾವ, ಪ್ರೇರಣೆ, ದೃಷ್ಟಿಕೋನ ಪ್ರಮುಖ ವಿಮರ್ಶಕರನ್ನು ಅನುಲಕ್ಷಿಸಿ
- ಇ) ದಲಿತ, ಬಂಡಾಯ ವಿಮರ್ಶೆ - ಸೈದ್ಧಾಂತಿಕ ನಿಲುವುಗಳು ಪ್ರಮುಖ ವಿಮರ್ಶಕರನ್ನು ಅನುಲಕ್ಷಿಸಿ

- ಈ) ಸ್ತ್ರೀವಾದಿ ವಿಮರ್ಶೆ - ಪ್ರಭಾವ, ಪ್ರೇರಣೆ, ಆಶಯಗಳು ಪ್ರಮುಖ ವಿಮರ್ಶಕರನ್ನು ಅನುಲಕ್ಷಿಸಿ
 ಉ) ಸಾಂಸ್ಕೃತಿಕ ವಿಮರ್ಶೆ ಭಿನ್ನ ನೆಲೆಗಳು ಪ್ರಮುಖ ವಿಮರ್ಶಕರನ್ನು ಅನುಲಕ್ಷಿಸಿ

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ವಿಮರ್ಶಾ ಸಂಗಾತಿ

ಅ) ವಿಮರ್ಶಾ ಮಾದರಿ ಅಧ್ಯಯನ

1. ಸರೋವರದ ಸಿರಿಗನ್ನಡಿಯಲ್ಲಿ - ಕುವೆಂಪು
2. ಪ್ರಜ್ಞೆ ಮತ್ತು ಪರಿಸರ ಒಂದು ಟಿಪ್ಪಣಿ - ಯು.ಆರ್. ಅನಂತಮೂರ್ತಿ
3. ಸಾಹಿತ್ಯಾಭಿವ್ಯಕ್ತಿಯಲ್ಲಿ ಸಾಮಾನ್ಯ ಮನುಷ್ಯ - ಸಿದ್ದಲಿಂಗಯ್ಯ
4. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಒಂದು ಸ್ತ್ರೀಪರದೃಷ್ಟಿ : ಬಿ.ಎನ್. ಸುಮಿತ್ರಾಬಾಯಿ

ಘಟಕ : 5

ಆ) ವಿಮರ್ಶೆಯ ಅಭ್ಯಾಸಕ್ಕಾಗಿ ಆಯ್ದ ಪಠ್ಯ ಭಾಗಗಳು

1. ಪದ್ಯಭಾಗಗಳು

ಗೊಬ್ಬರ	: ಕುವೆಂಪು
ಅವ್ವ	: ಪಿ. ಲಂಕೇಶ್
ಮಾಂಸದಂಗಡಿಯ ನವಿಲು	: ಎನ್.ಕೆ. ಹನುಮಂತಯ್ಯ
ದೇವಿ ಮಹಾತ್ಮೆ	: ಲಲಿತ ಸಿದ್ದಬಸವಯ್ಯ

2. ಕತೆಗಳು

ಸೆರೆ	: ಯಶವಂತ ಚಿತ್ತಾಲ
ದೇವರ ದಾರಿ	: ಮೊಗ್ಗಿ ಗಣೇಶ

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ವಿಮರ್ಶೆಯ ನೆಲೆಗಳನ್ನು ಗಮನಿಸುವುದು ಮತ್ತು ಪ್ರಾಯೋಗಿಕ ವಿಮರ್ಶೆ ಮಾಡುವುದು
- ಪೂರಕ ವಿಮರ್ಶೆಯ ಲೇಖನಗಳ ಸಂಗ್ರಹ
- ವಿಮರ್ಶಕರನ್ನು ಸಂದರ್ಶಿಸುವುದು, ಸಂವಾದ ನಡೆಸುವುದು
- ವಿದ್ಯಾರ್ಥಿಗಳು ಸ್ವತಃ ವಿಮರ್ಶೆ ಬರೆಯುವಂತೆ ಪ್ರೋತ್ಸಾಹಿಸುವುದು.
- ಕನ್ನಡದ ಪ್ರಮುಖ ವಿಮರ್ಶಕರನ್ನು, ಪಾಶ್ಚಾತ್ಯ ವಿಮರ್ಶಕರನ್ನು ಗುರುತಿಸುವುದು ಮತ್ತು ವಿವರಗಳನ್ನು ಸಂಗ್ರಹಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳಿಂದ ವಿಮರ್ಶೆ ಕುರಿತು ಪ್ರಬಂಧಗಳನ್ನು ಸಿದ್ಧಪಡಿಸುವುದು ಅದನ್ನು ಮಂಡಿಸಲು ಉತ್ತೇಜಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

- | | |
|--|--------------------------|
| 1. ತೌಲನಿಕ ಕಾವ್ಯಮೀಮಾಂಸೆ | : ಎಚ್. ತಿಪ್ಪೇರುದ್ರಸ್ವಾಮಿ |
| 2. ಪಾಶ್ಚಾತ್ಯ ಕಾವ್ಯಮೀಮಾಂಸೆ | : ಇನಾಂದಾರ್. ವಿ.ಎಂ |
| 3. ವಿಮರ್ಶೆಯ ಪರಿಭಾಷೆ | : ಓ.ಎಲ್. ನಾಗಭೂಷಣಸ್ವಾಮಿ |
| 4. ಪ್ರಿನ್ಸಿಪಲ್ಸ್ ಆಫ್ ಲಿಟರರಿ ಕ್ರಿಟಿಸಿಜಂ | : ಐ. ಎ. ರಿಚರ್ಡ್ಸ್ |
| 5. ದಿ ಫಂಕ್ಷನ್ ಆಫ್ ಕ್ರಿಟಿಸಿಜಂ | : ಟಿ. ಎಸ್. ಇಲಿಯಟ್ |
| 6. ಅರಿಸ್ಟಾಟಲನ ಕಾವ್ಯಮೀಮಾಂಸೆ | : ಎನ್. ಬಾಲಸುಬ್ರಹ್ಮಣ್ಯ |
| 7. ಪಾಶ್ಚಾತ್ಯ ವಿಮರ್ಶೆಯ ಪಕ್ಷಿನೋಟ | : ಸಂ. ಜಿ. ಶಂಕರಯ್ಯ |
| 8. ಸಾಂಸ್ಕೃತಿಕ ಅಧ್ಯಯನ | : ರಹಮತ್ ತರಿಕೆರೆ |
| 9. ಸಾಹಿತ್ಯ ವಿಮರ್ಶೆ | : ಸಿ.ಎನ್. ರಾಮಚಂದ್ರನ್ |
| 10. ವಚನ ವಿನ್ಯಾಸ | : ಗಿರಡ್ಡಿ ಗೋವಿಂದರಾಜ್ |

ವಿದನೆಯ ಚತುರ್ಮಾಸ

ಐಚ್ಛಿಕ ಕನ್ನಡ

ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -5.2(DSE)

Course code : ELE25223 ELE25226

ವಿಶೇಷ ಕವಿ ಕಾವ್ಯ ಅಧ್ಯಯನ ಮತ್ತು ಪೂರಕ ಪಠ್ಯ

ವಿಶೇಷ ಕವಿ : ಕುವೆಂಪು

ಘಟಕ 1 : ಕವಿಕಾವ್ಯ ಹಿನ್ನೆಲೆ : ಸ್ಥೂಲ ಪರಿಚಯ

- ಅ) ಕಾಲ, ಸಂದರ್ಭ ಮತ್ತು ಪ್ರೇರಣೆ ಪ್ರಭಾವಗಳು
- ಆ) ಕವಿಯ ಜೀವನ ಮತ್ತು ಸಾಹಿತ್ಯಕ ಸಾಧನೆಗಳು
- ಇ) ಸಾಹಿತ್ಯ ರಚನೆಯ ಸ್ವರೂಪ ಮತ್ತು ಲಕ್ಷಣಗಳು
- ಈ) ಸಮಕಾಲೀನ ಸ್ಪಂದನ ಮತ್ತು ಚಾರಿತ್ರಿಕ ಮಹತ್ವ

ಘಟಕ 2 - ಪದ್ಯ ನಾಟಕ (ಆಯ್ದು ಭಾಗಗಳು)

- ಅ) ಪ್ರಜಾಪ್ರಭುತ್ವ ತತ್ವ : ಅತ್ತಲಾ ದೈತ್ಯ ಸಭೆಯೋಳ್ (ಮಹಾಕಾವ್ಯ)
- ಆ) ಸ್ತ್ರೀ ಸಂವೇದನೆ : ನಾಗಿ (ಕಥನ ಕವನ)
- ಇ) ಜೀವಪರಿಸರದ ಪ್ರಜ್ಞೆ : ಭಾವಗೀತೆಗಳು
- ಈ) ಸರ್ವೋದಯ : ಜಲಗಾರ (ನಾಟಕ)

ಘಟಕ : 3 - ಗದ್ಯ ಕಥನ (ಆಯ್ದು ಭಾಗಗಳು)

- ಅ) ತಳ ಲೋಕದ ನಾಯಕತ್ವ : ಮಲೆಗಳಲ್ಲಿ ಮದುಮಗಳು (ಕಾದಂಬರಿ)
- ಆ) ಅನ್ನದಾತನ ಸಮಸ್ಯೆ : ಧನ್ವಂತರಿ ಚಿಕಿತ್ಸೆ (ಕಥೆ)
- ಇ) ವೈಚಾರಿಕತೆ : ವಿಚಾರಕ್ರಾಂತಿಗೆ ಆಹ್ವಾನ (ಚಿಂತನೆ)
- ಈ) ನಾಡು ನುಡಿ ವಿಚಾರ : ಬಾರಿಸು ಕನ್ನಡ ಡಿಂಡಿಮ (ಲೇಖನ)

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ಕುವೆಂಪು ಸಾಹಿತ್ಯ ಸಂಗಾತಿ

ಆತ್ಮಕಥನಾತ್ಮಕ ಬರಹಗಳು: ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

- ಅ. ಅಣ್ಣನ ನೆನಪು : ಕೆ. ಪಿ. ಪೂರ್ಣಚಂದ್ರ ತೇಜಸ್ವಿ
ಆ. ಮಗಳು ಕಂಡ ಕುವೆಂಪು : ತಾರಿಣಿ ಚಿದಾನಂದ
ಇ. ಕುವೆಂಪುಗೆ ಪುಟ್ಟ ಕನ್ನಡಿ : ಕೆ.ವಿ. ಸುಬ್ಬಣ್ಣ
ಈ. ಕುವೆಂಪು ಅವರ ಕೊನೆಯ ದಿನಗಳು : ದೇಜಗೌ

ಘಟಕ : 5

ಸಾಹಿತ್ಯ ಅವಲೋಕನ ಬರಹಗಳು : ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

- ಅ) ಕುವೆಂಪು ಕೃತಿ ಪ್ರವೇಶ : ಸುಜನಾ
ಆ) ಕುವೆಂಪು ಕಾವ್ಯಮೀಮಾಂಸೆ : ಜಿ. ಎಸ್. ಶಿವರುದ್ರಪ್ಪ
ಇ) ಕುವೆಂಪು ನಾಟಕಗಳು : ನರಹಳ್ಳಿ ಬಾಲಸುಬ್ರಹ್ಮಣ್ಯ
ಈ) ಕುವೆಂಪು ಕಾದಂದರಿಗಳು : ಬಿ.ಎನ್. ಸುಮಿತ್ರಾಬಾಯಿ

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಕುವೆಂಪು ಅವರಿಗೆ ಸಂಬಂಧಿಸಿದ ಮಾಹಿತಿಗಳನ್ನು ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು.
- ಕುವೆಂಪು ಅವರ ಕೃತಿಗಳನ್ನು ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು.
- ಕುವೆಂಪು ಅವರ ಭಾಷಣಗಳ ಆಡಿಯೋಗಳನ್ನು ಕೇಳಿಸುವುದು
- ಕುವೆಂಪು ಅವರಿಗೆ ಸಂಬಂಧಿಸಿದ ವಿಡಿಯೋಗಳನ್ನು ಪ್ರದರ್ಶಿಸುವುದು.
- ಕುವೆಂಪು ಗೀತೆಗಳಿಗೆ ರಾಗಸಂಯೋಜನೆ ಮಾಡಿಸಿ ಹಾಡಿಸುವುದು.
- ಕುವೆಂಪು ನಾಟಕಗಳನ್ನು ಅಭಿನಯಿಸಲು ಪ್ರೇರೇಪಿಸುವುದು.
- ಕುವೆಂಪು ಅವರ ಹುಟ್ಟೂರಿಗೆ ಪ್ರವಾಸ ಕೈಗೊಳ್ಳುವುದು
- ಕುವೆಂಪು ಅವರ ವಿಶ್ವಮಾನವ ಸಂದೇಶವನ್ನು ಪ್ರಸಾರ ಮಾಡಲು ಸಲಹೆ ನೀಡುವುದು.
- ಇತರ ಕವಿಗಳು ಕಂಡ ಕುವೆಂಪು ಚಿತ್ರಣಗಳನ್ನು ಸಂಗ್ರಹಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಕುವೆಂಪು ಪುನರಾವಲೋಕನ : ಡಾ. ಜಿ. ಎಸ್. ಶಿವರುದ್ರಪ್ಪ
2. ಹೀಗಿದ್ದರು ಕುವೆಂಪು : ಡಾ. ಪ್ರಭುಶಂಕರ
3. ಪರಂಪರೆ ಮತ್ತು ಕುವೆಂಪು : ಸುಜನಾ
4. ಕುವೆಂಪು ದರ್ಶನ : ದೇಜಗೌ
5. ತರಗತಿಯಲ್ಲಿ ಕುವೆಂಪು : ವ್ಯಷಭೇಂದ್ರ ಸ್ವಾಮಿ
6. ಒಲವು - ನಿಲವು : ಹರಿಹರ ಪ್ರಿಯ
7. ಕುವೆಂಪು ಸಮಗ್ರ ಸಾಹಿತ್ಯ ಸಂಪುಟಗಳು : ಸಂ : ಕೆ. ಸಿ. ಶಿವಾರೆಡ್ಡಿ
8. ಕುವೆಂಪು ಕೃತಿ ವಿಮರ್ಶೆ : ಸಂ. ಅರವಿಂದ ಮಾಲಗತ್ತಿ
9. ಅನಿಕೇತನ : ಸಂ. ಸಾಶಿ. ಮರುಳಯ್ಯ
10. ಕುವೆಂಪು ಒಂದು ಪುನರನ್ವೇಷಣೆ : ಸಂ. ಡಿ.ಎಸ್. ನಾಗಭೂಷಣ

11. ಶ್ರೀ ಕುವೆಂಪು ; ಬೆಂಗಳೂರು ವಿಶ್ವವಿದ್ಯಾಲಯ
 12. ಮಲೆಗಳಲ್ಲಿ ಮದುಮಗಳು ಸಾಂಸ್ಕೃತಿಕ ಮುಖಾಮುಖಿ - ಸಂ: ರಹಮತ್ ತರಿಕೆರೆ
 13. ಕುವೆಂಪು ಕಾದಂಬರಿಗಳ ಸಾಂಸ್ಕೃತಿಕ ಅಧ್ಯಯನ : ಸಂ : ಗುರುಪಾದ ಮರಿಗುದ್ದಿ
 ಕನ್ನಡದ ಮುಖ್ಯ ವಿಮರ್ಶಕರ ಎಲ್ಲ ಪರಿಕಲ್ಪನಾತ್ಮಕ ಅಧ್ಯಯನದ ಲೇಖನಗಳು

ವಿದನೆಯ ಚತುರ್ಮಾಸ

ವಿಚ್ಛಿಕ ಕನ್ನಡ

ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -5.3(DSE)

Course code : ELE25423 ELE25426

ಕನ್ನಡ ನಾಟಕ ಸಾಹಿತ್ಯ ಮತ್ತು ಪೂರಕ ಪಠ್ಯ

ಘಟಕ 1 : ನಾಟಕದ ಪರಿಕಲ್ಪನೆ ಮತ್ತು ಪರಂಪರೆ

ಅ) ಸ್ವರೂಪ. ಲಕ್ಷಣ

ಆ) ನಾಟಕ ಸಾಹಿತ್ಯ ಬೆಳೆದು ಬಂದ ಬಗೆ

- ಸಂಸ್ಕೃತ ನಾಟಕಗಳ ಸಂಕ್ಷಿಪ್ತ ಚರಿತ್ರೆ
- ಗ್ರೀಕ್ ಮತ್ತು ಪಾಶ್ಚಾತ್ಯ ನಾಟಕಗಳ ಸಂಕ್ಷಿಪ್ತ ಚರಿತ್ರೆ
- ಜನಪದ ರಂಗಭೂಮಿಯ ಸಂಕ್ಷಿಪ್ತ ಚರಿತ್ರೆ

ಘಟಕ 2 - ನಾಟಕ ಸಾಹಿತ್ಯ ಪರಿಚಯ

ಅ) ಸಂಸ್ಕೃತ ನಾಟಕಗಳು : ಅಶ್ವಘೋಷ, ಕಾಳಿದಾಸ, ಭಟ್ಟನಾರಾಯಣ

ಆ) ಪಾಶ್ಚಾತ್ಯ ನಾಟಕಗಳು : ಸೋಪೋಕ್ಲಿಸ್, ಈಸ್ಟಿಲಸ್, ಯೂರಿಪಿಡೀಸ್, ಷೇಕ್ಸ್ಪಿಯರ್

ಇ) ಜನಪದ ರಂಗಭೂಮಿ : ಪಗರಣ, ಬಯಲಾಟ ಯಕ್ಷಗಾನ

ಘಟಕ : 3 - ಕನ್ನಡ ನಾಟಕ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ

ಅ) ಸಂಸ್ಕೃತದ ಪ್ರಭಾವ : ಬಸವಪ್ಪ ಶಾಸ್ತ್ರಿ (ಅಭಿಜ್ಞಾನ ಶಾಕುಂತಲ)

ಆ) ಪಾಶ್ಚಾತ್ಯ ಪ್ರಭಾವ : ಬಿ.ಎಂ.ಶ್ರೀ (ಅಶ್ವತ್ಥಾಮನ್)

ಇ) ಜನಪದದ ಪ್ರಭಾವ : ಗಿರೀಶ್ ಕಾರ್ನಾಡ್ (ನಾಗಮಂಡಲ)

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ನಾಟಕ ಸಂಗಾತಿ

ಮಾದರಿ ಅಧ್ಯಯನ ಆಯ್ದ ಭಾಗಗಳು

ಅ) ಶ್ರೀರಂಗರ ಶೋಕಚಕ್ರ

ಆ) ಚಂದ್ರಶೇಖರ ಕಂಬಾರ : ಸಂಗ್ರಾಹಾಳ್ಯ

ಇ) ಎಚ್. ಎಸ್. ವೆಂಕಟೇಶ್ ಮೂರ್ತಿ : ಉರಿಯ ಉಯ್ಯಾಲೆ

ಘಟಕ : 5

ಮಾದರಿ ಅಧ್ಯಯನ (ಇತ್ತೀಚಿನ ಪ್ರವೃತ್ತಿಗಳು)

ಏಕಾಂಕ, ಬೀದಿನಾಟಕ, ರೇಡಿಯೋ (ಬಾನುಲಿ) ನಾಟಕ,

ಕಿರು ಪ್ರಹಸನ ಇತ್ಯಾದಿಗಳ ಪರಿಚಯ

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 :

ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ನಾಟಕದ ಅಭಿನಯ ಪೂರ್ವಕ ಓದು (ಸಾಮೂಹಿಕವಾಗಿ ಭಿನ್ನ ಪಾತ್ರಗಳನ್ನು ನಿರ್ವಹಿಸುವಿಕೆ).
- ನಾಟಕದ ಪರಿಕರ ಸಿದ್ಧಪಡಿಸುವುದು (ಬೆಳಕು, ರಂಗವಿನ್ಯಾಸ ಇತ್ಯಾದಿ)
- ನಾಟಕ ರಂಗಪ್ರದರ್ಶನ
- ಕಿರುನಾಟಕ ರಚನೆಗೆ ಉತ್ತೇಜನ
- ಏಕಪಾತ್ರಾಭಿನಯಕ್ಕೆ ಪ್ರೋತ್ಸಾಹ

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ನಾಟ್ಯಶಾಸ್ತ್ರ : ಭರತ
2. ದಶರೂಪಕ : ಧನಂಜಯ
3. ಅಭಿನಯ ದರ್ಪಣ : ನಂದಿಕೇಶ್ವರ
4. ಭಾರತೀಯ ಕಾವ್ಯ ಮೀಮಾಂಸೆ : ತೀ.ನಂ. ಶ್ರೀಕಂಠಯ್ಯ
5. ಅರಿಸ್ವಾಟಲ್‌ನ ಕಾವ್ಯಮೀಮಾಂಸೆ : ಎನ್. ಸುಬ್ರಹ್ಮಣ್ಯ
6. ರಿಕ್ತ ರಂಗಭೂಮಿ : ಬರ್ಲೋಲ್ಟ್ ಜೈಕ್ಸ್
7. ನಟನೆಯ ಪಾಠಗಳು : ಪ್ರಸನ್ನ
8. ಕನ್ನಡ ರಂಗಭೂಮಿ : ಮರುಳು ಸಿದ್ಧಪ್ಪ.ಕೆ
9. ರಂಗಪ್ರಯೋಗ : ಕೆ.ವಿ.ಅಕ್ಷರ
10. ಗ್ರೀಕ್ ರಂಗಭೂಮಿ ಮತ್ತು ನಾಟಕಗಳು : ಎಲ್. ಎಸ್. ಶೇಷಗಿರಿರಾವ್
11. ಶಿಕ್ಷಣದಲ್ಲಿ ನಾಟಕ ಮತ್ತು ಕಲೆ : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ

ವಿದನೆಯ ಚತುರ್ಮಾಸ

ವಿಚ್ಛಿಕ ಕನ್ನಡ

ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -5.4(DSE)

Course code : ELE25623/ELE25626

ಮಹಿಳಾ ಸಾಹಿತ್ಯದ ಅಧ್ಯಯನ ಮತ್ತು ಪೂರಕ ಪಠ್ಯ

ಘಟಕ 1 : ಅ) ಮಹಿಳಾ ಸಾಹಿತ್ಯ :

ಸ್ವರೂಪ. ಉಗಮ, ವಿಕಾಸ ಮತ್ತು ಅಧ್ಯಯನದ ಅಗತ್ಯತೆ

ಘಟಕ 2 - ಮಹಿಳಾ ಸಾಹಿತ್ಯ ಅಧ್ಯಯನ ವಿವಿಧ ಹಂತಗಳು

ಅ) ಜನಪದ

ಆ) ವಚನ ಮತ್ತು ದಾಸಸಾಹಿತ್ಯ

ಇ) ಒಡೆಯರ ಕಾಲದಸಾಹಿತ್ಯ

ಘಟಕ : 3- ಇಪ್ಪತ್ತನೆಯ ಶತಮಾನದ ಮಹಿಳಾ ಸಾಹಿತ್ಯ

ಅ) ಸ್ವಾತಂತ್ರ್ಯ ಪೂರ್ವಸಾಹಿತ್ಯ

ಆ) ಜನಪ್ರಿಯ ಮಹಿಳಾಸಾಹಿತ್ಯ

ಇ) ಸ್ವಾತಂತ್ರ್ಯೋತ್ತರ ಮಹಿಳಾಸಾಹಿತ್ಯ

ಈ) ಸ್ತ್ರೀವಾದ - ಪ್ರಭಾವ, ಪ್ರೇರಣೆಗಳು

ಉ) ಬಂಡಾಯ, ದಲಿತ, ಮುಸ್ಲಿಂ ಸಂವೇದನೆಗಳು

ಊ) ಇತ್ತೀಚಿನ ಪ್ರವೃತ್ತಿಗಳು

ಋ) ಸ್ತ್ರೀ ಅನನ್ಯತೆಯ ಹುಡುಕಾಟ

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ಮಹಿಳಾ ಸಾಹಿತ್ಯ ಸಂಗಾತಿ

ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

ಆಯ್ದು ಕವಿತೆಗಳು

1. ಬೆಳೆಗೆರೆ ಜಾನಕಮ್ಮ
2. ಸ. ಉಷಾ
3. ಎಸ್. ಮಾಲತಿ
4. ಎಂ.ಆರ್. ಕಮಲ
5. ಸಬೀಹಾ ಭೂಮಿಗೌಡ
6. ಕವಿತಾ ರೈ

ಘಟಕ : 5

ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

ಆಯ್ದು ಕತೆಗಳು:

1. . ವೀಣಾ ಶಾಂತೇಶ್ವರ
2. ಬಾನುಮುಷ್ತಾಕ್

ಆಯ್ದು ಗದ್ಯಭಾಗಗಳು

1. . ವೈದೇಹಿ
2. ಭುವನೇಶ್ವರಿ ಹೆಗಡೆ

ಆತ್ಮಕಥನ ಆಯ್ದು ಭಾಗಗಳು

1. ಇಳಾವಿಜಯ
2. ಜಯಶ್ರೀ

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಮಹಿಳಾ ಅಭಿವೃದ್ಧಿಯ ವಿಶೇಷ ಗುಣಗಳನ್ನು (ಭಾಷೆ, ನುಡಿಗಟ್ಟು, ಶೈಲಿ ಇತ್ಯಾದಿ) ಗುರುತಿಸಲು ಸೂಚಿಸುವುದು.
- ಸಾಧಕಿಯರ ಮತ್ತು ಲೇಖಕಿಯರ ಬದುಕು, ಸಾಧನೆ ಕುರಿತು ವಿಚಾರ ಗೋಷ್ಠಿಗಳನ್ನು ನಡೆಸುವುದು.
- ಮಹಿಳಾ ಕೇಂದ್ರಿತ ವಿಷಯಗಳ ಬಗ್ಗೆ ಕವಿತೆ, ಕತೆ ಬರೆಯಲು ಉತ್ತೇಜಿಸುವುದು
- ಮಹಿಳಾ ಸಾಹಿತ್ಯ ಮತ್ತು ಮುಖ್ಯ ಸಾಹಿತ್ಯವನ್ನು ತೌಲನಿಕವಾಗಿ ಅಧ್ಯಯನ ಮಾಡಲು ಉತ್ತೇಜಿಸುವುದು.
- ವಿದ್ಯಾರ್ಥಿಗಳಿಗೆ ಅವರ ಪರಿಸರದಲ್ಲಿ ವಿಶೇಷವಾಗಿ ಗುರುತಿಸಿಕೊಂಡಿರುವ ಮಹಿಳೆಯರ/ ಮಹಿಳಾ ಸಾಧಕಿಯರ ಬಗ್ಗೆ ಪ್ರಬಂಧ ಬರೆಯಲು ಸೂಚಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಕನ್ನಡ ಮಹಿಳಾ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ಎಚ್. ಎಸ್. ಶ್ರೀಮತಿ
2. ಆಧುನಿಕ ಮಹಿಳಾ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ಸಂ. ಎಂ. ಉಷಾ
3. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಮತ್ತು ಸ್ತ್ರೀವಾದಿ ಚಿಂತನೆ : ವಿವಿಧ ಲೇಖಕಿಯರು
4. ಚಹರೆ : ಎಚ್. ಎಸ್. ಶ್ರೀಮತಿ
5. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಮತ್ತು ಮಹಿಳೆ : ಸಬಿಹಾ ಭೂಮಿಗೌಡ
6. ಸ್ತ್ರೀ ವಾದ : ಬಿ.ಎನ್. ಸುಮಿತ್ರಾಬಾಯಿ

ವಿದನೆಯ ಚತುರ್ಮಾಸ

ಐಚ್ಛಿಕ ಕನ್ನಡ

ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -5.5(DSE)

Course code : ELE25823/ELE25826

ಕನ್ನಡ ಜನಪದ ಸಾಹಿತ್ಯ ಅಧ್ಯಯನ

ಘಟಕ 1 :

- ಅ) ಕನ್ನಡ ಜನಪದ ಸಾಹಿತ್ಯ ಅಧ್ಯಯನ ಸ್ವರೂಪ
- ಆ) ಕನ್ನಡ ಸಾಹಿತ್ಯ ಮತ್ತು ಜನಪದದ ಅಂತರ್ ಸಂಬಂಧಗಳು

ಘಟಕ 2 -

- ಅ) ಜನಪದ ಕಾವ್ಯ : ಗೀತೆಗಳು, ಕಥನಕವನ, ಲಾವಣಿ, ಮಹಾಕಾವ್ಯ - ಇವುಗಳ ಸ್ವರೂಪ, ಲಕ್ಷಣ (ಮಾದರಿ ಪರಿಚಯದೊಂದಿಗೆ)

ಘಟಕ : 3

- ಅ) ಜನಪದ ಗದ್ಯ : ಗಾದೆ, ಒಗಟು, ಐತಿಹ್ಯ, ನುಡಿಗಟ್ಟು - ಇವುಗಳ ಸ್ವರೂಪ, ಲಕ್ಷಣ (ಮಾದರಿ ಪರಿಚಯದೊಂದಿಗೆ)
- ಆ) ಜನಪದ ಕಥೆ: ಸ್ವರೂಪ, ಲಕ್ಷಣಗಳು, ವರ್ಗೀಕರಣ, ಜನಪದ ಕಥೆಗಳ ವಿಧಗಳು - ರಮ್ಯ ಕಥೆಗಳು, ವಿನೋದ ಕಥೆಗಳು, ನೀತಿ ಕತೆಗಳು

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ಜನಪದ ಸಾಹಿತ್ಯ ಸಂಗಾತಿ

ಮಾದರಿ ಅಧ್ಯಯನ ಆಯ್ದು ಭಾಗಗಳು

- ಅ) ಜನಪದ ಕಾವ್ಯ - ಆಯ್ದು ಜನಪದ ಗೀತೆ, ಲಾವಣಿ, ಕಥನ ಕವನಗಳ ಅಧ್ಯಯನ
- ಆ) ಜನಪದ ಗದ್ಯ - ಗಾದೆ, ಒಗಟು, ಐತಿಹ್ಯ, ನುಡಿಗಟ್ಟುಗಳ ಅಧ್ಯಯನ
- ಇ) ಜನಪದ ಮಹಾಕಾವ್ಯ - ಆಯ್ದು ಮಹಾಕಾವ್ಯಗಳ ಅಧ್ಯಯನ

ಘಟಕ : 5

ಪೂರಕ ಲೇಖನಗಳು

- ಅ) ಜನವಾಣಿ ಬೇರು ಕವಿವಾಣಿ ಹೂವು - ಬಿ.ಎಂ.ಶ್ರೀ
- ಆ) ಜನಪದ ಸಾಹಿತ್ಯ ಅದರ ಸ್ವರೂಪ - ಹಾಮಾ ನಾಯಕ
- ಇ) ಜನಪದ ಮತ್ತು ಆಧುನಿಕ ಸಾಹಿತ್ಯ - ಕೆ.ರಂ. ನಾಗರಾಜ
- ಈ) ಕೆಲವು ಊರ ಹೆಸರುಗಳು ಮತ್ತು ಅವುಗಳ ಹಿನ್ನೆಲೆ - ಹ.ಕ. ರಾಜೇಗೌಡ

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7

ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಕನ್ನಡದಲ್ಲಿ ಪ್ರಕಟಗೊಂಡಿರುವ ಜನಪದ ಸಾಹಿತ್ಯಕ್ಕೆ ಸಂಬಂಧಿಸಿದ ಪುಸ್ತಕಗಳನ್ನು ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು.

- ಜನಪದ ಗೀತೆಗಳನ್ನು ಅವುಗಳ ಮೂಲಮಟ್ಟಗಳಲ್ಲಿ ಹಾಡಲು ಉತ್ತೇಜಿಸುವುದು.
- ಒಗಟುಗಳನ್ನು ಹೇಳಿ ಅವುಗಳೊಳಗಿನ ನಿಗೂಢ ಅರ್ಥವನ್ನು ಗ್ರಹಿಸಿ ತಿಳಿಸುವ ಚಟುವಟಿಕೆಯನ್ನು ನಡೆಸುವುದು.
- ಜನಪದ ಗಾದೆಗಳ ಅರ್ಥವನ್ನು ವಿಸ್ತರಿಸಲು ಸೂಚಿಸುವುದು.
- ತಮ್ಮ ಪರಿಸರದ ಕಲಾವಿದರನ್ನು ಸಂದರ್ಶಿಸಿ ಸಂಬಂಧಿಸಿದ ಮಾಹಿತಿಗಳನ್ನು ದಾಖಲಿಸಲು ತಿಳಿಸುವುದು.
- ಜನಪದ ಕರಕುಶಲ ಕಲಾಕೃತಿಗಳನ್ನು ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು.
- ಒಂದು ನಿರ್ದಿಷ್ಟ ವಿಷಯವನ್ನು ಸೂಚಿಸಿ ಅದನ್ನು ಕುರಿತು ಗ್ರಾಮೀಣ ಭಾಷೆಯಲ್ಲಿ ಮಾತನಾಡಲು ತಿಳಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

- | | |
|---|--------------------|
| 1. ಜಾನಪದ ಅಧ್ಯಯನ | : ದೇ.ಜ.ಗೌ |
| 2. ಜಾನಪದ ಸ್ವರೂಪ | : ಹಾ.ಮಾ. ನಾಯಕ |
| 3. ಜಾನಪದ ಪುರಾಣಗಳು | : ರಾಗೌ |
| 4. ಜನಪದ ಪ್ರವೇಶಿಕೆ | : ಸಿ.ಪಿ.ಕೆ |
| 5. ಆದಿಮ ಜನಪದ | : ಮೊಗಳ್ಳಿ ಗಣೇಶ |
| 6. ಗ್ರಾಮದೇವತೆಗಳು | : ಸಿದ್ದಲಿಂಗಯ್ಯ |
| 7. ಜಾನಪದ ಅಧ್ಯಯನ ಸಂಕ್ಷಿಪ್ತ ಇತಿಹಾಸ | : ನಂ. ತಪಸ್ವೀಕುಮಾರ್ |
| 8. ದಕ್ಷಿಣ ಕರ್ನಾಟಕದ ಜನಪದ ಕಾವ್ಯ ಪ್ರಕಾರಗಳು | : ಜಿ.ಶಂ.ಪರಮಶಿವಯ್ಯ |

ಆರನೆಯ ಚತುರ್ಮಾಸ
ವಿಚ್ಛೇದ ಕನ್ನಡ
ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -6.1(DSE)
Course code : ELF25026 ELF25023

ಪಠ್ಯ (ಅ) ಹಳಗನ್ನಡ ವ್ಯಾಕರಣ - ಪಠ್ಯ ಮತ್ತು ಪ್ರಯೋಗ

ಘಟಕ 1 :

- ಅ) ಕೇಶಿರಾಜನ ಕಾಲ, ದೇಶ, ವ್ಯಕ್ತಿತ್ವ, ಕೃತಿ ಪರಿಚಯ, ವ್ಯಾಕರಣ ಅಧ್ಯಯನದ ಅವಶ್ಯಕತೆ
ಆ) ವರ್ಣಮಾಲೆ : ಅಕ್ಷರ, ವರ್ಣ, ಶ್ರವಣ, ಚಾಕ್ಷುಷ, ಸ್ವರಗಳು, ವ್ಯಂಜನಗಳು, ವರ್ಗೀಯ ವ್ಯಂಜನಗಳು, ಯೋಗವಾಹಗಳು, ದೇಶೀಯ ಅಕ್ಷರಗಳು, ಮಹಾಪ್ರಾಣಗಳು, ಉಳ, ಕುಳ, ಕ್ಷಳ ವಿಚಾರ, ಪಂಚವಿಧ ಶುದ್ಧಗುಣಗಳು, ವರ್ಣಮಾಲೆಯ ವರ್ಣಗಳ ಸಂಖ್ಯಾನಿರ್ಣಯ
ಸೂತ್ರಗಳು: 1, 2 , 3, 6, 7, 8, 10, 12, 14, 18, 22, 23, 24, 25, 26, 27, 41, 42, 43,-
20 ಸೂತ್ರಗಳು

ಘಟಕ 2

- ಅ) ಸಂಧಿಗಳು : ಸಂಧಿವಿಚಾರ, ಲೋಪ, ಆಗಮ, ಆದೇಶ, ದ್ವಿತ್ವ ಸಂಧಿಗಳು, ಪ್ರಕೃತಿ ಭಾವ, ಶ್ರುತಿಕಷ್ಟ,

ಶ್ರುತಿ ಸಹ್ಯ, ಸಂಧಿವಿಕಲ್ಪ

ಸೂತ್ರಗಳು: 53, 55, 56, 57, 58, 59, 60, 62, 63, 65, 66, 73 - 13 ಸೂತ್ರಗಳು

- ಇ) **ನಾಮಪ್ರಕರಣ :** ನಾಮಪದ - ರೂಢ, ಅನ್ವರ್ಥ, ಅಂಕಿತ, ಸರ್ವನಾಮ ಭೇದಗಳು, ಸಮಸಂಸ್ಕೃತ ಲಿಂಗಮೊಂಬತ್ತು ತೆರಿಗೆ, ವಚನಗಳ ವಿಚಾರ, ವಚನ ಪಲ್ಲಟ, ವಿಭಕ್ತಿ, - ಸಪ್ತವಿಧ ವಿಭಕ್ತಿಗಳು ಮತ್ತು ಅವುಗಳ ಬಳಕೆ, ವಿಭಕ್ತಿ ಪಲ್ಲಟ, ಪಕಾರಕ್ಕೆ ಹಕಾರ

ಸೂತ್ರಗಳು: 75, 76, 77, 81, 85, 86, 87, 88, 89, 100, 103, 106 - 18 ಸೂತ್ರಗಳು

ಘಟಕ : 3

- ಅ) **ಸಮಾಸ: ಸಮಾಸ ಮತ್ತು ಅದರ ಭೇದಗಳು:** ತತ್ಪುರುಷ, ದ್ವಿಗು, ಕರ್ಮಧಾರಾಯ, ಬಹುವ್ರೀಹಿ, ದ್ವಂದ್ವ, ಅವ್ಯಯೀಭಾವ (ಅಂಶೀ) ಕ್ರಿಯಾ, ಗಮಕ ಮತ್ತು ಅರಿಸಮಾಸ

ಸೂತ್ರಗಳು: 107, 109, 110, 111, 112, 114 - 06 ಸೂತ್ರಗಳು

- ಈ) ಅಖ್ಯಾತ - ಧಾತು, ಅಖ್ಯಾತ ಪ್ರತ್ಯಯಗಳು, ದ - ದಪ - ವ ಎಂಬ ಕಾಲಸೂಚಕ ಪ್ರತ್ಯಯಗಳು, ಪ್ರತಿಷೇಧ, ಪುರುಷಪ್ರತ್ಯಯಗಳಲ್ಲಿ ಕ್ರಿಯಾಪದದ ರಚನೆ, ಪ್ರತ್ಯಯಗಳು, ಕಾಲಪ್ರತ್ಯಯಗಳಲ್ಲಿ ಕ್ರಿಯಾಪದಗಳ ರಚನೆ, ಸತಿಸಪ್ತಮಿ

ಸೂತ್ರಗಳು: 115, 116, 127, - 04 ಸೂತ್ರಗಳು

- ಉ) ಧಾತು : ತದ್ವಿಂತ ಲಕ್ಷಣಗಳು, ಅಪಭ್ರಂಶ, ಅವ್ಯಯಗಳ ಸಾಮಾನ್ಯ ವಿಚಾರ, ಕನ್ನಡದ ಅಸಾಧಾರಣ ಲಕ್ಷಣಗಳ ವಿವೇಚನೆ

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ವ್ಯಾಕರಣ ಸಂಗಾತಿ

ಮಾದರಿ ಅಧ್ಯಯನ

ಟಿ.ವಿ. ವೆಂಕಟಾಚಲ ಶಾಸ್ತ್ರಿಯವರು ಸಂಪಾದಿಸಿರುವ ಶಬ್ದಮಣಿದರ್ಪಣಂ ಕೃತಿಯಲ್ಲಿನ ಆಯ್ದು 16 ಲಕ್ಷ್ಯ ಪದ್ಯಗಳು

ಘಟಕ : 5

ಮಾದರಿ ಅಧ್ಯಯನ

ವ್ಯಾಕರಣ ಸಂಗತಿಗಳನ್ನು ಕುರಿತ ಆಯ್ದು 4 ಭಾಗಗಳು

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಸೂತ್ರಗಳನ್ನು ಕ್ರಮಬದ್ಧವಾಗಿ ಓದಿಸುವುದು.
- ಸೂತ್ರದ ವಿಚಾರಗಳಿಗೆ ಪ್ರಯೋಗಗಳನ್ನು ಅನ್ವಯಿಸಲು ಸೂಚಿಸುವುದು.
- ಕಾವ್ಯಗಳ ಭಾಗಗಳನ್ನು ನೀಡಿ ಅಲ್ಲಿರುವ ವ್ಯಾಕರಣಾಂಶಗಳನ್ನು ಗುರುತಿಸಲು ತಿಳಿಸುವುದು.
- ಸಂಧಿ ಮತ್ತು ಸಮಾಸ ಪದಗಳ ನಡುವಣ ವ್ಯತ್ಯಾಸಗಳನ್ನು ಪಟ್ಟಿ ಮಾಡಿಸುವುದು.
- ಕನ್ನಡ ವ್ಯಾಕರಣಕಾರರ ಬದುಕು - ಬರಹ ಕುರಿತ ವಿಚಾರ ಗೋಷ್ಠಿ ಏರ್ಪಡಿಸುವುದು.
- ಪದಬಂಧಗಳನ್ನು ರಚಿಸಲು ಮತ್ತು ಬಿಡಿಸಲು ಸೂಚಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಕೇಶೀರಾಜನ ಶಬ್ದಮಣಿ ದರ್ಪಣ : ಡಿ.ಎಲ್ ನರಸಿಂಹಾಚಾರ್
2. ಕೇಶೀರಾಜನ ಶಬ್ದಮಣಿದರ್ಪಣ : ಟಿ.ವಿ. ವೆಂಕಟಾಚಲ ಶಾಸ್ತ್ರಿ
3. ಶಬ್ದಮಣಿದರ್ಪಣ : ಎಲ್ . ಬಸವರಾಜು
4. ಶಬ್ದಮಣಿದರ್ಪಣ : ಜೆ. ಎಸ್. ಕುಳ್ಳಿ
5. ಅಲೋಕ : ಎಚ್. ಎಸ್. ಬಿಳಿಗಿರಿ
6. ಶಬ್ದಮಣಿ ದರ್ಪಣದ 101 ಸೂತ್ರಗಳು : ಎ.ಎಸ್. ಜಯರಾಂ ಮತ್ತು ಕೆ ಮಹದೇವಪ್ಪ
7. ದರ್ಪಣ ದೀಪಿಕೆ : ಅಬ್ದುಲ್ ಬಷೀರ್
8. ಹಳಗನ್ನಡ ತೌಲನಿಕ ಅಧ್ಯಯನ : ಕೆ. ಕುಶಾಲಪ್ಪ ಗೌಡ

ಆರನೆಯ ಚತುರ್ಮಾಸ

ಐಚ್ಛಿಕ ಕನ್ನಡ

ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -6.2(DSE)

Course code :ELF25226 ELF25223

ವಿಶೇಷ ಕವಿ ಕಾವ್ಯ ಅಧ್ಯಯನ ಮತ್ತು ಪೂರಕ ಪಠ್ಯ

ವಿಶೇಷ ಕವಿ : ಪಂಪ

ಘಟಕ 1 : ಪಂಪನ ಕಾಲ, ದೇಶ, ಚಾರಿತ್ರಿಕತೆ ಮತ್ತು ಯುಗಧರ್ಮ

ಘಟಕ 2 - ಪಂಪನ ಕಾವ್ಯಗಳು

- ಅ) ಆದಿ ಪುರಾಣ - ಆಕರ, ವಸ್ತು, ಸ್ವರೂಪ, ಪಾತ್ರ ಚಿತ್ರಣ, ಸನ್ನಿವೇಶ ಸೃಷ್ಟಿ
- ಆ) ವಿಕ್ರಮಾರ್ಜುನ ವಿಜಯ - ಆಕರ, ವಸ್ತು, ಸ್ವರೂಪ, ಪಾತ್ರ ಚಿತ್ರಣ, ಸನ್ನಿವೇಶ ಸೃಷ್ಟಿ

ಘಟಕ : 3 -

- ಅ. ಆದಿ ಪುರಾಣ - ಆಯ್ದು ಭಾಗಗಳ ಅಧ್ಯಯನ
- ಆ) ವಿಕ್ರಮಾರ್ಜುನ ವಿಜಯ - ಆಯ್ದು ಭಾಗಗಳ ಅಧ್ಯಯನ

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ಪಂಪ ಸಾಹಿತ್ಯ ಸಂಗಾತಿ

ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

- ಅ. ನಾಡೋಜ ಪಂಪ : ಮುಳಿಯ ತಿಮ್ಮಪ್ಪಯ್ಯ
- ಆ. ಪಂಪ : ತೀ.ನಂ.ಶ್ರೀ
- ಇ. ಪಂಪ ಕೆಲವು ಮುಖಗಳು : ಸಿ.ಪಿ.ಕೆ
- ಈ. ಮತ್ತೆ ಮತ್ತೆ ಪಂಪ : ಜಿ. ಎಚ್. ನಾಯಕ

ಘಟಕ : 5

ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

ಪಂಪನ ಕಾವ್ಯ ಕುರಿತ ವಿಮರ್ಶಾ ಲೇಖನಗಳು

- ಅ) ಪಂಪನ ಎರಡು ಪದ್ಯಗಳು : ಟಿ. ಎಸ್. ವೆಂಕಣ್ಣಯ್ಯ
- ಆ) ಪಂಪನಲ್ಲಿ ಭವ್ಯತೆ : ಕುವೆಂಪು
- ಇ) ಪಂಪ ಭಾರತದ ವಸ್ತು ವಿನ್ಯಾಸ : ಸುಜನಾ
- ಈ) ಪಂಪ ಒಂದು ಮೌಲ್ಯ ವಿವೇಚನೆ : ಕೆ. ಪಿ. ಪೂರ್ಣಚಂದ್ರ ತೇಜಸ್ವಿ

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಪಂಪನಿಗೆ ಸಂಬಂಧಿಸಿದ ಶಾಸನದ ಮಾಹಿತಿಗಳನ್ನು ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು.
- ಪಂಪನ ಕೃತಿಗಳಿಗೆ ಸಂಬಂಧಿಸಿದ ವಿಮರ್ಶೆಗಳನ್ನು ಸಂಗ್ರಹಿಸಲು ಸೂಚಿಸುವುದು.
- ಪಂಪನ ಕಾವ್ಯಗಳ ಗಮಕವಾಚನವನ್ನು ಕೇಳಿಸುವುದು
- ಪಂಪನ ಕಾವ್ಯಗಳಿಗೆ ಪ್ರಸ್ತಾರ ಹಾಕಲು ಸೂಚಿಸುವುದು.
- ಪಂಪನ ಬನವಾಸಿಯ ಕುರಿತು ಮಾಹಿತಿ ಸಂಗ್ರಹಿಸುವುದು.

- ಪಂಪನ ಕಾವ್ಯಗಳ ಧ್ವನಿ, ರಸ, ಅಲಂಕಾರಗಳನ್ನು ಗುರುತಿಸಲು ತಿಳಿಸುವುದು.
- ಪಂಪನ ಕಾವ್ಯಗಳ ಭಾಷಿಕ ಅಧ್ಯಯನವನ್ನು ಕೈಗೊಳ್ಳಲು ಸೂಚಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಪಂಪಭಾರತ ದೀಪಿಕೆ : ಡಿ.ಎಲ್. ಎನ್.
2. ಸರಳ ಪಂಪ ಭಾರತ : ಎಲ್. ಬಸವರಾಜ್
3. ಸರಳ ಆದಿಪುರಾಣ : ಎಲ್. ಬಸವರಾಜ್
4. ಪಂಪನ ಸಮಸ್ತ ಭಾರತ ಕಥಾಮೃತ : ಎಲ್.ಬಸವರಾಜ್
5. ಆದಿಪುರಾಣ ಕಥಾಮೃತ : ಎಲ್.ಬಸವರಾಜ್
6. ಪಂಪ ಮಹಾಕವಿ : ವಿ.ಸೀ
7. ಆದಿಪುರಾಣ ದೀಪಿಕೆ : ತ.ಸು. ಶಾಮರಾಯ
8. ಪಂಪ ಒಂದು ಅಧ್ಯಯನ : ಜಿ. ಎಸ್. ಶಿವರುದ್ರಪ್ಪ
9. ಪಂಪಭಾರತ ಸಾಂಸ್ಕೃತಿಕ ಅಧ್ಯಯನ : ಶಾಂತಿನಾಥ ದಿಬ್ಬದ
10. ಪಂಪ ಭಾರತ (ನಾಟಕ) : ಕೆ. ವೈ. ನಾರಾಯಣ ಸ್ವಾಮಿ
11. ರೂಪಮಂ ಧರಿಯಿಸಿ ನಿಂದುದಲ್ಲು : ಜಯಪ್ರಕಾಶ್ ಶೆಟ್ಟಿ. ಎಚ್.
12. ವಚನ ಪಂಪ ಭಾರತ : ಸಿದ್ದಲಿಂಗಯ್ಯ
13. ಆದಿಪುರಾಣ : ಜಿ. ಬ್ರಹ್ಮಪ್ಪ
14. ಆದಿಪುರಾಣ ಸಂಗ್ರಹ : ಎಲ್. ಗುಂಡಪ್ಪ

ಆರನೆಯ ಚತುರ್ಮಾಸ

ಐಚ್ಛಿಕ ಕನ್ನಡ

ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -6.3(DSE)

Course code :ELF25426 ELF25423

ಕನ್ನಡ ಸಾಹಿತ್ಯ ಮತ್ತು ಕನ್ನಡ ಸಿನಿಮಾ

ಘಟಕ 1 : ಸಿನಿಮಾ ಪರಿಕಲ್ಪನೆ ಮತ್ತು ಇತಿಹಾಸ

- ಅ. ಸಿನಿಮಾದ ಸ್ವರೂಪ, ಲಕ್ಷಣ
- ಆ. ಕನ್ನಡ ಸಿನಿಮಾ ಬೆಳೆದು ಬಂದ ಬಗೆ

ಘಟಕ 2 - ಸಾಹಿತ್ಯ ಮತ್ತು ಸಿನಿಮಾ ಅಂತರ್ ಸಂಬಂಧ

ಅ. ಜನಪದ ಸಾಹಿತ್ಯ ಮತ್ತು ಸಿನಿಮಾ

ಆ. ಪೌರಾಣಿಕ - ಚಾರಿತ್ರಿಕ ಸಾಹಿತ್ಯ ಮತ್ತು ಸಿನಿಮಾ

ಘಟಕ : 3 - ಸಿನಿಮಾ ಮಾಧ್ಯಮದ ವಸ್ತುಗಳಾಗಿ ಸಾಹಿತ್ಯ ಪ್ರಕಾರಗಳು

ಅ) ಕಥೆ ಕಾದಂಬರಿ ಆಧಾರಿತ ಸಿನಿಮಾಗಳು : ಭೂತಯ್ಯನ ಮಗ ಅಯ್ಯು, ನಾಗರಹಾವು,

ಕಿರಗೂರಿನ ಗಯ್ಯಾಳಿಗಳು

ಆ) ನಾಟಕ, ಕವಿತೆ, ಸಿನಿಮಾಗಳು - ನಾಗಮಂಡಲ, ಉದ್ಭವ, ಸಂಗ್ರಾಬಾಳ್ಯ, ಮೈಸೂರು ಮಲ್ಲಿಗೆ, ಕೆರೆಗೆ
ಹಾರ

ಇ) ಜೀವನ ಚರಿತ್ರೆ ಆಧಾರಿತ ಸಿನಿಮಾಗಳು : ಶಿಶುನಾಳ ಷರೀಫ, ಜಗಜ್ಯೋತಿ ಬಸವೇಶ್ವರ, ಭಕ್ತ
ಕನಕದಾಸ

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ಸಿನಿಮಾ ಸಂಗಾತಿ

ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದ ಭಾಗಗಳು)

ಅ) ಸಂಭಾಷಣೆ ರಚನೆ

ಆ) ಚಿತ್ರಕಥೆ ರಚನೆ

ಇ) ಗೀತೆ ರಚನೆ

ಘಟಕ : 5

ಕನ್ನಡದ ಸಿನಿಮಾಗಳು : ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದ ಭಾಗಗಳು)

ಅ. ಭಾಷೆ : ಜನಪದ, ಪೌರಾಣಿಕ - ಬಾಲನಾಗಮ್ಮ ಬಬ್ರುವಾಹನ

ಆ. ವಸ್ತು : ಸಾಮಾಜಿಕ, ಕಲಾತ್ಮಕ - ಫಣಿಯಮ್ಮ, ತಬರನ ಕಥೆ

ಇ. ಶೈಲಿ : ನೇರ ನಿರೂಪಣೆ, ಹಿನ್ನೋಟ ತಂತ್ರ, (ಆಯ್ದ ಸಿನಿಮಾ ಭಾಗಗಳ ಮಾದರಿ ಅಧ್ಯಯನ)

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

➤ ಪಠ್ಯಕ್ಕೆ ಪೂರಕವಾದ ಸಿನಿಮಾ ಪ್ರದರ್ಶನವನ್ನು ಏರ್ಪಡಿಸುವುದು,

- ಸಿನಿಮಾ ರಸಗ್ರಹಣ ಶಿಬಿರಗಳನ್ನು ಏರ್ಪಡಿಸುವುದು.
- ಸಿನಿಮಾಧ ವಸ್ತು, ಆಶಯ, ಪಾತ್ರ ಚಿತ್ರಣಗಳನ್ನು ಕುರಿತ ಗುಂಪು ಚರ್ಚೆಗಳನ್ನು ಏರ್ಪಡಿಸುವುದು.
- ಆಯ್ದು ಕಥೆಗಳನ್ನು ಸೂಚಿಸಿ ಅವುಗಳಿಗೆ ಚಿತ್ರಕಥೆ ಮತ್ತು ಸಂಭಾಷಣೆಗಳನ್ನು ಬರೆಯುವುದು.
- ಸಿನಿಮಾ ಪುರವಣಿಗಳಲ್ಲಿ ಬರುವ ಚಲನ ಚಿತ್ರ ವಿಮರ್ಶೆಗಳನ್ನು ಸಂಗ್ರಹಿಸುವುದು.
- ಗೀತ ರಚನೆ ಮಾಡುವಂತೆ ಪ್ರೇರೇಪಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

- | | |
|---|--------------------------|
| 1. ಕನ್ನಡ ಸಿನಿಮಾ | : ಗಂಗಾಧರ್ ಮೊದಲಿಯಾರ್ |
| 2. ಅಮೆರಿಕಾ ಅಮೆರಿಕಾ ಚಿತ್ರಕಥೆ | : ನಾಗತಿಹಳ್ಳಿ ಚಂದ್ರಶೇಖರ್ |
| 3. ಕಥೆ , ಚಿತ್ರಕಥೆ, ಸಂಭಾಷಣೆ | : ಜೋಗಿ |
| 4. ಸಿನಿಮಾ ಸಂಕಥನ | : ಆರ್. ಸಿ. ಗೋವಿಂದರಾಜು, |
| 5. ಸಿನಿಮಾಯಾನ | : ಪುಟ್ಟಸ್ವಾಮಿ.ಕೆ |
| 6. ಪುಟ್ಟಣ್ಣ ಕಣಗಾಲ್: ಸಿನಿಮಾಗಳು | : ಎಸ್. ಪ್ರಸಾದ್ ಸ್ವಾಮಿ |
| 7. ಸಾಹಿತ್ಯ ಮತ್ತು ಚಲನಚಿತ್ರ | : ಕೆ. ವಿ. ಸುಬ್ಬಣ್ಣ |
| 8. ಸಾಯುವ ಮುನ್ನ ನೋಡಲೇಬೇಕಾದ 101 ಕನ್ನಡ ಸಿನಿಮಾಗಳು | : ರವೀಂದ್ರ |
| 9. ಕನ್ನಡ ಚಿತ್ರರಂಗ ಒಂದು ವಿವೇಚನೆ | : ಎಸ್. ಶ್ರೀಧರ ಮೂರ್ತಿ |
| 10. ಡಾ. ರಾಜಕುಮಾರ್ ಸಮಗ್ರ ಚರಿತ್ರೆ | : ದೊಡ್ಡಹುಲ್ಲೂರು ರುಕ್ಮಿಣಿ |

ಆರನೆಯ ಚತುರ್ಮಾಸ

ಐಚ್ಛಿಕ ಕನ್ನಡ

ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -6.4(DSE)

Course code : ELF25623/ELF25626

ಪ್ರಕಾರ ಕೇಂದ್ರಿತ ಅಧ್ಯಯನ : ಕನ್ನಡ ಸಾಹಿತ್ಯ ಪ್ರಕಾರ - ಸಣ್ಣ ಕಥೆ

ಘಟಕ 1 : ಕಥೆಯ ಸ್ವರೂಪ. ಮತ್ತು ಲಕ್ಷಣ

- ಅ. ಸಣ್ಣಕಥೆ ಬೆಳೆದು ಬಂದ ದಾರಿ
1. ಸಂಸ್ಕೃತ ಕಥೆಗಳ ಹಿನ್ನೆಲೆ
 2. ಪಾಶ್ಚಾತ್ಯ ಪ್ರಭಾವ ಪ್ರೇರಣೆ
 3. ಜನಪದ ಪರಂಪರೆ

ಘಟಕ 2 – ಕನ್ನಡ ಸಣ್ಣಕಥೆ ಸಾಹಿತ್ಯ ಪ್ರಕಾರದ ಪರಿಚಯ

- ಅ) ಹಳಗನ್ನಡ, ನಡುಗನ್ನಡ ಕತೆಗಳು : ವಡ್ಡಾರಾಧನೆ, ಧರ್ಮಾಮೃತ, ಪಂಚತಂತ್ರ, ರಾಜರತ್ನಂ ಕಥೆ
ಆ) ಸಾಮಾಜಿಕ ಕಥೆಗಳು : ನನ್ನ ಚಿಕ್ಕತಾಯಿ, ಧನಿಯರ ಸತ್ಯನಾರಾಯಣ

ಘಟಕ : 3 – ಕನ್ನಡ ಸಣ್ಣ ಕತೆಗಳ ಬೆಳವಣಿಗೆಯ ಇತಿಹಾಸ

- ಅ) ಪಂಜೆ ಮಂಗೇಶರಾಯರಿಂದ ಇಂದಿನ ವರೆಗೂ ಆಯ್ದು ಕತೆಗಾರರ ಪರಿಚಯ
ಆ) ಕನ್ನಡ ಸಣ್ಣ ಕತೆಗಳ ಭಾಷೆ, ವಿನ್ಯಾಸ, ತಂತ್ರಗಳ ಅಧ್ಯಯನ (ಮಾದರಿ ಕಥೆಗಳೊಂದಿಗೆ)

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ಕಥಾ ಸಾಹಿತ್ಯ ಸಂಗಾತಿ

ಕನ್ನಡ ಸಣ್ಣ ಕತೆಗಳು : ಮಾದರಿ ಅಧ್ಯಯನ ಆಯ್ದು ಭಾಗಗಳು

- ಅ. ಸೇಡಿಯಾಪು ಕೃಷ್ಣಭಟ್ಟ : ಚೆನ್ನೆಮಣೆ
ಆ. ಆನಂದ : ನಾ ಕೊಂದ ಹುಡುಗಿ
ಇ. ಶಾಂತಿನಾಥ ದೇಸಾಯಿ : ಕ್ಷಿತಿಜ
ಈ. ನೇಮಿಚಂದ್ರ : ಆಯ್ಕೆ ನಮ್ಮ ಕೈಲಿದೆ
ಉ. ಅಮರೇಶ್ ನುಗಡೋಣಿ : ಸವಾರಿ
ಊ. ರಾಘವೇಂದ್ರ ಪಾಟೀಲ : ದೇಸಗತಿ

ಘಟಕ : 5

ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

ಇತ್ತೀಚಿನ ಪ್ರವೃತ್ತಿಗಳ ಅಧ್ಯಯನ

- ಅ. ಒಂದು ಪುಟದ ಕಥೆ
ಆ. ಮಿನಿ ಕಥೆ
ಇ. ಹನಿ ಕಥೆ
ಈ. ವಾಟ್ಸ್‌ಆಪ್ ಮತ್ತು ಫೇಸ್‌ಬುಕ್ ಕಥೆಗಳು

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಕಥೆಗಳನ್ನು ಓದಿಸುವುದು.
- ಕಥೆಗಳನ್ನು ರಚಿಸುವಂತೆ ಪ್ರೇರೇಪಿಸುವುದು.
- ತಮ್ಮ ಪರಿಸರದ ಕತೆಗಾರರೊಂದಿಗೆ ಸಂವಾದ ಮತ್ತು ಸಂದರ್ಶನಗಳನ್ನು ಏರ್ಪಡಿಸುವುದು.
- ಕಥೆಗಳಲ್ಲಿರುವ ಚಿಂತನಾತ್ಮಕ ಅಂಶಗಳನ್ನು ಚರ್ಚಿಸುವುದು.

- ಸಾಹಿತ್ಯ ನಿಯತಕಾಲಿಕಗಳಲ್ಲಿ ಪ್ರಕಟವಾಗುವ ಇತ್ತೀಚಿನ ಕಥೆಗಳ ಮಾದರಿಗಳನ್ನು ಸಂಗ್ರಹಿಸುವುದು.
- ಕಥಾರಚನ ಕಮ್ಮಟಗಳನ್ನು ಏರ್ಪಡಿಸುವುದು.
- ಸಣ್ಣಕತೆಗಳನ್ನಾಧರಿಸಿದ ಸಿನಿಮಾಗಳ ಪ್ರದರ್ಶನವನ್ನು ಏರ್ಪಡಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಶತಮಾನದ ಸಣ್ಣಕತೆಗಳು : ಕರ್ನಾಟಕ ಸಾಹಿತ್ಯ ಅಕಾಡೆಮಿ
2. ಹೊಸಗನ್ನಡದ ಅರುಣೋದಯ : ಶ್ರೀನಿವಾಸ ಹಾವನೂರ
3. ಹೊಸಗನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ಎಲ್. ಎಸ್. ಶೇಷಗಿರಿರಾವ್
4. ಯುಗಧರ್ಮ ಮತ್ತು ಕನ್ನಡ ಸಾಹಿತ್ಯ : ಕೀರ್ತಿನಾಥ ಕುರ್ತುಕೋಟೆ
5. ಕಥನ ಸಾಹಿತ್ಯ : ಜಿ. ಎಸ್. ಅಮೂರ
6. ಕನ್ನಡದ ಸಣ್ಣಕತೆಗಳು : ಜಿ. ಎಚ್, ನಾಯಕ
7. ಜೀವನ ಪ್ರೀತಿ : ಸಂ. ಕಾಳೇಗೌಡ ನಾಗವಾರ
8. ಸ್ವಾತಂತ್ರ್ಯೋತ್ತರ ಕಾಲದ ಕನ್ನಡ ಸಾಹಿತ್ಯ : ಉದಯಭಾನು ಕಲಾಸಂಘ
9. ಕನ್ನಡ ಕಥನ ಸಾಹಿತ್ಯದಲ್ಲಿ ಸಾಂಸ್ಕೃತಿಕ ಸಂಘರ್ಷ : ಎನ್. ಕೆ. ಕೋದಂಡ ರಾಮ
10. ಮಾಸ್ತಿಯರ ಕಥೆಗಳು : ಎ.ಎನ್. ಮೂರ್ತಿರಾವ್

ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆ, ಪರೀಕ್ಷೆ ಮತ್ತು ಮೌಲ್ಯಮಾಪನ ಪ್ರಕ್ರಿಯೆ

ಕಲಾ, ವಾಣಿಜ್ಯ ಮತ್ತು ವಿಜ್ಞಾನ ನಿಕಾಯಗಳಿಗೆ ಅನ್ವಯವಾಗುವಂತೆ ಆಯ್ಕೆ ಆಧಾರಿತ ಗುಣಾಂಕ ಪದ್ಧತಿ (ಸಿಬಿಸಿಎಸ್) ಕನ್ನಡ ಭಾಷೆ ಮತ್ತು ಐಚ್ಛಿಕ ಪತ್ರಿಕೆಗಳ ಪಠ್ಯಕ್ರಮಕ್ಕೆ ಸಂಬಂಧಿಸಿದಂತೆ ಕನ್ನಡ ವಿಭಾಗದ ಮುಖ್ಯಸ್ಥರು ಮತ್ತು ಅಧ್ಯಾಪಕರಿಗೆ ಸೂಚನೆಗಳು.

- ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳನ್ನು ಕಡ್ಡಾಯವಾಗಿ ನಡೆಸುವುದು
- ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳ ದಾಖಲೆಗಳನ್ನು ಅಧ್ಯಾಪಕರು ವಿದ್ಯಾರ್ಥಿಗಳಿಂದ ಸಂಗ್ರಹಿಸಿ ಸಂರಕ್ಷಿಸುವುದು.
- ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳಲ್ಲಿ ಭಾಗವಹಿಸುವ ವಿದ್ಯಾರ್ಥಿ ವಿವರಗಳನ್ನು (ಚಟುವಟಿಕೆಗಳ ಸ್ವರೂಪ, ವಿದ್ಯಾರ್ಥಿ ಹೆಸರು, ತರಗತಿ, ದಿನಾಂಕ) ಟಿಪ್ಪಣಿ ಪುಸ್ತಕದಲ್ಲಿ ದಾಖಲಿಸುವುದು.
- ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳಿಗೆ ಸಹಾಯಕವಾಗುವಂತೆ ವಿಭಾಗದ ಗ್ರಂಥಾಲಯವನ್ನು ರೂಪಿಸಿಕೊಳ್ಳಬೇಕು
- ಸಿ-1 : 15 ಅಂಕಗಳು ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್) ಆರಂಭವಾದ 8 ವಾರಗಳ ಒಳಗೆ ಪ್ರತಿ ಘಟಕದಲ್ಲೂ ಶೇ. 50ರಷ್ಟು ಪಠ್ಯಬೋಧನೆ ಮಾಡಿರಬೇಕು. ಬೋಧಿಸಿರುವ ಪಠ್ಯ ವಿಷಯಗಳ ಮೇಲೆ ಸಿ-1 ಮೌಲ್ಯ ಮಾಪನ ಪ್ರಕ್ರಿಯೆಯನ್ನು ನಿಯೋಜಿತ ಕಾರ್ಯ ಅಥವಾ ವಿಚಾರ ಮಂಡನೆ ಮೂಲಕ ನಡೆಸುವುದು.

- ಸಿ -2 : ಅಂಕಗಳು, ಚತುರ್ಮಾಸ (ಸೆಮಿಸ್ಟರ್) ಆರಂಭವಾದ 15 ವಾರಗಳ ಒಳಗೆ ಪ್ರತಿಘಟಕದಲ್ಲೂ ಶೇ. 100ರಷ್ಟು ಪಠ್ಯ ಬೋಧನೆ ಮಾಡಿರಬೇಕು. ಬೋಧಿಸಿರುವ ಪಠ್ಯವಿಷಯಗಳ ಮೇಲೆ ಸಿ-2 ಮೌಲ್ಯಮಾಪನ ಪ್ರಕ್ರಿಯೆಯನ್ನು ಕಿರು ಪರೀಕ್ಷೆ ಮೂಲಕ ನಡೆಸುವುದು.
- ವಿಶ್ವವಿದ್ಯಾಲಯದ ನಿಯಮಗಳನ್ವಯ ಸಿ-1 ಮತ್ತು ಸಿ-2 ಮೌಲ್ಯಮಾಪನ ಪ್ರಕ್ರಿಯೆಯ ದಾಖಲೆಗಳನ್ನು ಸಂರಕ್ಷಿಸಿಡುವುದು.
- ಸಿ-3 : 70 ಅಂಕಗಳು ಮುಖ್ಯ ಪರೀಕ್ಷೆ; ಮೂರು ಗಂಟೆಗಳ ಅವಧಿ; ಪರೀಕ್ಷೆಯು ಕಾಲೇಜು ಮಟ್ಟದಲ್ಲಿ ನಡೆಯುತ್ತದೆ.
- ಸಿ-1, ಸಿ-2, ಸಿ-3 : ಇವುಗಳ ಬಗ್ಗೆ ಹೆಚ್ಚಿನ ಮಾಹಿತಿಗೆ ಕನ್ನಡ ವಿಭಾಗವನ್ನು ಸಂಪರ್ಕಿಸುವುದು.

ಆರನೆಯ ಚತುರ್ಮಾಸ

ಐಚ್ಛಿಕ ಕನ್ನಡ

ಡಿ.ಎಸ್.ಇ - ಪತ್ರಿಕೆ -6.5(DSE)

Course code : ELF25823/ELF25826

ಕನ್ನಡ ಭಾಷೆ ಮತ್ತು ಸಮೂಹ ಮಾಧ್ಯಮ

ಘಟಕ - 1 :

ಭಾಷೆ ಮತ್ತು ಸಮೂಹ ಮಾಧ್ಯಮ : ಅಂತರ್ ಸಂಬಂಧ

- ಅ) ಸಮೂಹ ಮಾಧ್ಯಮದ ಸ್ವರೂಪ ಮತ್ತು ಲಕ್ಷಣಗಳು
- ಆ) ಮಾಧ್ಯಮಗಳ ಬೆಳವಣಿಗೆಯ ಇತಿಹಾಸ ಮತ್ತು ಮಹತ್ವ
- ಇ) ವಿವಿಧ ಪ್ರಕಾರದ ಸಮೂಹ ಮಾಧ್ಯಮಗಳು

ಘಟಕ - 2

ಕನ್ನಡ ಪತ್ರಿಕಾ ಮಾಧ್ಯಮ : ಸ್ವರೂಪ - ಲಕ್ಷಣಗಳು

ಅ. ಪತ್ರಿಕಾ ಭಾಷೆ ಮತ್ತು ಬರಹಗಳ ವೈಶಿಷ್ಟ್ಯ: ವರದಿ, ಸುದ್ದಿ ಲೇಖನ, ನುಡಿಚಿತ್ರ, ಶೀರ್ಷಿಕೆ ಬರಹ, ಅಂಕಣ ಬರಹ

ವ್ಯಂಗ್ಯ ಚಿತ್ರ, ರೇಖಾಚಿತ್ರ, ಓದುಗರ ಪತ್ರಗಳು

ಆ. ಸಾಹಿತ್ಯ ಪುರವಣಿಗಳ ಸ್ವರೂಪ ಮತ್ತು ಮಹತ್ವ, ಸೃಜನಶೀಲ ಬರಹಗಳು, ಪುಸ್ತಕ ಅವಲೋಕನ, ಸಂದರ್ಶನ, ವ್ಯಕ್ತಿಚಿತ್ರ

ಇ. ಸಾಹಿತ್ಯ ಪತ್ರಿಕೆಗಳ ಸ್ವರೂಪ ಮತ್ತು ಪ್ರಕಾರಗಳು : ಹವ್ಯಾಸಿ ಪತ್ರಕರ್ತರು, ಪುಸ್ತಕ ಪ್ರಕಾಶನ

ಘಟಕ : 3

ಕನ್ನಡ ಭಾಷೆ : ಶ್ರವ್ಯ ಮತ್ತು ದೃಶ್ಯ ಮಾಧ್ಯಮ

- ಅ) ಬಾನುಲಿ, (ಆಕಾಶವಾಣಿ) ಭಾಷೆಯ ಸ್ವರೂಪ : ಸಂಭಾಷಣೆ, ನಿರೂಪಣೆ, ಉದ್ಯೋಷಣೆಯಲ್ಲಿ ಉಚ್ಚಾರಣೆ ಮತ್ತು ಧ್ವನಿವಿನ್ಯಾಸದ ಮಹತ್ವ
- ಆ) ಬಾನುಲಿ ಬರಹದ ಸ್ವರೂಪ : ವಾರ್ತಾವಾಚನ, ನಾಟಕ, ಚಿಂತನೆ, ಕಥಾವಾಚನ, ಗಮಕ ವ್ಯಾಖ್ಯಾನ
- ಇ) ದೂರದರ್ಶನದಲ್ಲಿ ಭಾಷೆ ಮತ್ತು ಬರಹ : ವಾರ್ತೆ, ಸಂದರ್ಶನ, ನಿರೂಪಣೆಯಲ್ಲಿ ಉಚ್ಚಾರಣೆಯ ಸ್ವರೂಪ, ಶೀರ್ಷಿಕೆ, ಜಾಹಿರಾತು, ಚಲಿಸುವ ಬರಹ (ಸ್ಯೋಲಿಂಗ್) ಸಂಪಾದನಾ ಸಾಹಿತ್ಯದಲ್ಲಿ ಬರಹದ ಸ್ಥಾನ ಮತ್ತು ಮಹತ್ವ

ಘಟಕ : 4

ಪೂರಕ ಪಠ್ಯ : ಮಾಧ್ಯಮ ಸಂಗಾತಿ

ಕನ್ನಡ ಭಾಷೆ ಮತ್ತು ನವಮಾಧ್ಯಮ : ಮಾದರಿ ಅಧ್ಯಯನ ಆಯ್ದು ಭಾಗಗಳು

- ಅ) ಅಂತರ್ಜಾಲದಲ್ಲಿ ಕನ್ನಡ ಬಳಕೆಯ ಸ್ವರೂಪ
- ಆ) ಭಾಷೆ ಮತ್ತು ಭಾಷಾಂತರ ತಂತ್ರಾಂಶಗಳು ನುಡಿ, ಬರಹ, ಕುವೆಂಪು, ಜಸ್ಟ್ ಕನ್ನಡ, ಯುನಿಕೋಡ್, ವಿಕಿಪೀಡಿಯಾ

ಘಟಕ : 5

ಕನ್ನಡ ಸಮೂಹ ಮಾಧ್ಯಮ : ಮಾದರಿ ಅಧ್ಯಯನ (ಆಯ್ದು ಭಾಗಗಳು)

- ಅ) ಪತ್ರಿಕೆಯ ಚತುರಂಗ : ಡಿ.ವಿ.ಜಿ
ಆ) ಆಕಾಶವಾಣಿ : ವಸಂತಕುಮಾರ್ ಪೆರ್ಲ
ಇ) ದೂರದರ್ಶನ : ಡಿ.ಬಿ. ನಾಯಕ
ಈ) ಮಾಧ್ಯಮ - ವರ್ತಮಾನದ ಬಿಕ್ಕಟ್ಟುಗಳು : ಪದ್ಮರಾಜ ದಂಡಾವತಿ

ಘಟಕ - 6 ಮತ್ತು ಘಟಕ 7 : ಪಠ್ಯಾಧಾರಿತ ಚಟುವಟಿಕೆಗಳು (ಟ್ಯುಟೋರಿಯಲ್)

- ಕನ್ನಡ ಪತ್ರಿಕೆಗಳನ್ನು ಓದಿಸುವುದು
- ಕನ್ನಡ ಪತ್ರಿಕೆಗಳನ್ನು ಸಂಗ್ರಹಿಸುವಂತೆ ಸಲಹೆ ನೀಡುವುದು
- ಕನ್ನಡ ಪತ್ರಿಕೆಗಳಿಗೆ ಪ್ರಾಯೋಗಿಕವಾಗಿ ಲೇಖನ ಬರೆಯುವುದು
- ಪ್ರಾಯೋಗಿಕ ಪತ್ರಿಕೆಗಳನ್ನು ರೂಪಿಸಲು ಮಾರ್ಗದರ್ಶನ ನೀಡುವುದು.
- ಕನ್ನಡದ ಸಾಹಿತ್ಯ ಮತ್ತು ಪತ್ರಿಕೆಗಳ ವಿಚಾರಸಂಕೀರ್ಣವನ್ನು ಏರ್ಪಡಿಸುವುದು
- ಪತ್ರಿಕೆಗಳ ಲೇಖನಗಳಿಗೆ ಸಂಬಂಧಪಟ್ಟಂತೆ ಅಭಿಪ್ರಾಯ ಮಂಡನೆಗೆ ಅವಕಾಶ ಕಲ್ಪಿಸುವುದು.

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ವೃತ್ತ ಪತ್ರಿಕೆ : ಡಿ.ವಿ.ಜಿ
2. ಭಾರತೀಯ ಪತ್ರಿಕೋದ್ಯಮದ ಇತಿಹಾಸ : ನಾಡಿಗ ಕೃಷ್ಣಮೂರ್ತಿ
3. ಇದೂ ಪತ್ರಿಕೋದ್ಯಮ : ನರೇಂದ್ರ ರೈ ದೇರ್ಲ
4. ಬಾನುಲಿ ಬರಹ : ಡಿ. ಕೆ. ರವಿ
5. ಸಮೂಹ ಸಂಪರ್ಕ ಮಾಧ್ಯಮಗಳು : ಬಿ.ಎಸ್. ಚಂದ್ರಶೇಖರ್
6. ಜಾಹೀರಾತು : ಕೆ. ವಿ. ನಾಗರಾಜು
7. ವಾಚಕರ ವಾಣಿ : ಸಾಹಿತ್ಯಾನಂದ
8. ಪತ್ರಿಕೋದ್ಯಮ ಕೈಪಿಡಿ . ವಿ. ಆನಂದ ಮೂರ್ತಿ
9. ಆಕಾಶವಾಣಿ ಸಾಹಿತ್ಯ : ಎನ್. ಎಸ್. ವಿಜಯ ಹರನ್
10. ಬರವಣಿಗೆ ಒಂದು ಕಲೆ : ನಿರಂಜನ ವಾನಳ್ಳಿ
11. ಮಾಧ್ಯಮ ಚಿಂತನ : ಲೋಹಿತ್. ಜಿ
12. ಪತ್ರಿಕೋದ್ಯಮದ ಪರಿಚಯ : ಕೆ. ಜಿ. ಜೋಸೆಫ್
13. ಸೃಜನಶೀಲತೆ ಮತ್ತು ಸಮೂಹ ಮಾಧ್ಯಮಗಳು : ಡಾ. ಎನ್. ಎಂ. ಗಿರಿಜಾಪತಿ

ಪ್ರಶ್ನೆ ಪತ್ರಿಕೆ ಮಾದರಿ

ಮೊದಲನೆಯ, ಎರಡನೆಯ, ಮೂರನೆಯ ಮತ್ತು ನಾಲ್ಕನೆಯ ಚತುರ್ಮಾಸಗಳ (ಸೆಮಿಸ್ಟರ್) - ಸಿಬಿಸಿಎಸ್

ಪಠ್ಯಕ್ರಮದ ಕನ್ನಡ ಐಚ್ಛಿಕ ಪತ್ರಿಕೆಗಳ ಪ್ರಶ್ನೆ ಪತ್ರಿಕೆ ಮಾದರಿ

ಡಿ.ಎಸ್.ಸಿ ಮತ್ತು ಡಿ.ಎಸ್.ಇ ಪತ್ರಿಕೆಗಳಿಗೆ ಅನ್ವಯವಾಗುವಂತೆ

ಸಮಯ: 3 ಗಂಟೆಗಳು

ಗರಿಷ್ಠ

ಅಂಕಗಳು: 70

ಒಂದು ಎರಡು ಮೂರು ಘಟಕಗಳಿಗೆ

ಅಂಕಗಳು: 40

1. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು) 1 X 10 =
10
2. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು) 1 X 10 =
10
3. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು) 1 X 10 =
10
4. ಎರಡು ಪ್ರಶ್ನೆಗಳಿಗೆ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು) 2 X 5 = 10

ಪೂರಕ ಪಠ್ಯ

ಅಂಕಗಳು:

30

5. ಅ) ಒಂದು ಪದ್ಯ/ಗದ್ಯ ಭಾಗದ ವಿಶೇಷತೆಗಳನ್ನು ಗುರುತಿಸಿ. 1 X 5 = 5
(ಆಯ್ದು ಎರಡು ಭಾಗಗಳನ್ನು ಕೇಳುವುದು)
ಆ) ಒಂದು ವಾಕ್ಯದಲ್ಲಿ ಉತ್ತರಿಸಿ (ಐದು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು) 5 X 1 = 5
6. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು) 1 X 10 =
10
7. ಎರಡು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು) 2 X 5 = 10

ಐದನೆಯ ಚತುರ್ಮಾಸ
ಕನ್ನಡ ಸಾಹಿತ್ಯ ಪರಿಚಯ(GE)
Course Code :BA- ELE26021/ELE26022/ELE26023/ELE26024/ELE26025/ELE26026
ಪತ್ರಿಕೆ - 1 ಕನ್ನಡ ಸಾಹಿತ್ಯ ಪರಿಚಯ

ಘಟಕ - 1 ಪ್ರಾಚೀನ ಕನ್ನಡ ಸಾಹಿತ್ಯ ಪರಿಚಯ

- ಅ) ಕನ್ನಡ ಸಾಹಿತ್ಯದ ಪ್ರಾಚೀನತೆ; ಕವಿರಾಜಮಾರ್ಗ, ವಡ್ಡಾರಾಧನೆ, ಚಂಪೂ ಸಾಹಿತ್ಯ
- ಆ) ದಾಸಿಮಯ್ಯ, ಅಲ್ಲಮಪ್ರಭು, ಬಸವಣ್ಣ, ಅಕ್ಕಮಹಾದೇವಿ, ಹರಿಹರ, ರಾಘವಾಂಕ, ಚಾಮರಸ, ರತ್ನಾಕರವರ್ಣಿ, ಸರ್ವಜ್ಞ
ಭಾಗವತ ಮತ್ತು ದಾಸಸಾಹಿತ್ಯ; ಕುಮಾರವ್ಯಾಸ, ಲಕ್ಷ್ಮೀಶ, ಪುರಂದರದಾಸ, ಕನಕದಾಸ

ಘಟಕ 2 - ಆಧುನಿಕ ಕನ್ನಡ ಸಾಹಿತ್ಯ ಪರಿಚಯ

- ಅ) ಹೊಸಗನ್ನಡ ಅರುಣೋದಯ: ಕೆಂಪುನಾರಾಯಣ, ಮುದ್ದಣ
ಹೊಸಗನ್ನಡ ಸಾಹಿತ್ಯದ ಮುಖ್ಯಘಟ್ಟಗಳು: ನವೋದಯ, ಪ್ರಗತಿಶೀಲ, ನವ್ಯ ಮತ್ತು
ದಲಿತ ಬಂಡಾಯ ಪಂಥಗಳ ಸಂಕ್ಷಿಪ್ತ ಪರಿಚಯ
- ಆ)ಬಿ,ಎಂ,ಶ್ರೀ., ದ,ರಾ, ಬೇಂದ್ರೆ, ಕುವೆಂಪು, ಶಿವರಾಮ ಕಾರಂತ, ಗೋಪಾಲಕೃಷ್ಣ ಅಡಿಗ,
ತಿರುಮಲಾಂಬ, ತ್ರಿವೇಣಿ, ಸಾರಾ ಅಬೂಬ್ಕರ್, ಇವರ ಸಂಕ್ಷಿಪ್ತ ಪರಿಚಯ

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ರಂ. ಶ್ರೀ. ಮುಗಳಿ
2. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ತ.ಸು. ಶಾಮರಾಯ
3. ಹೊಸಗನ್ನಡ ಸಾಹಿತ್ಯ ಚರಿತ್ರೆ : ಎಲ್. ಎಸ್. ಶೇಷಗಿರಿರಾವ್,
4. ಸಾಲುದೀಪಗಳು : ಕರ್ನಾಟಕ ಸಾಹಿತ್ಯ ಅಕಾಡೆಮಿ
5. ಕನ್ನಡ ಸಾಹಿತ್ಯ ಕೋಶ : ರಾಜಪ್ಪ ದಳವಾಯಿ

ಅರನೆಯ ಚತುರ್ಮಾಸ

ಕನ್ನಡ ಸಾಹಿತ್ಯ ಪರಿಚಯ(GE)

Course Code :BA- ELF25821/ELF25822/ELF25823/ELF25824/ELF25825/ELF25826

ಪತ್ರಿಕೆ - 1 ಕನ್ನಡ ಕಲಿಕೆ - ಗಳಿಕೆ

ಘಟಕ - 1 ಕನ್ನಡ ಸಾಹಿತ್ಯ ಕಲಿಕೆ

ಅ) ಕತೆಗಳ ಪರಿಚಯ : ರಂಗನ ಮದುವೆ - ಶ್ರೀನಿವಾಸ, ಶವದಮನೆ - ಚದುರಂಗ

ಆ) ಕವಿತೆಗಳ ಪರಿಚಯ: ಕಾಣಿಕೆ - ಬಿ,ಎಂ.ಶ್ರೀ, ನಾನು ಬಡವಿ - ಬೇಂದ್ರೆ,

ಕನ್ನಡ ಪದಗೋಳ - ಜಿ.ಪಿ. ರಾಜರತ್ನಂ, ನಿತ್ಯೋತ್ಸವ - ಕೆ.ಎಸ್. ನಿಸಾರ್ ಅಹಮದ್

ಅ) ನಾಟಕಗಳ ಪರಿಚಯ : ಜಲಗಾರ - ಕುವೆಂಪು, ಸಂಕ್ರಾಂತಿ - ಪಿ. ಲಂಕೇಶ್ (ಆಯ್ದ ಭಾಗಗಳು)

ಘಟಕ 2 - ಕನ್ನಡ ಜ್ಞಾನ ಗಳಿಕೆ

ಅ) ಪದಜ್ಞಾನ - ಮೂಲ ಪದಗಳು, ತತ್ಸಮ - ತದ್ಭವ, ಗಾದೆ, ಒಗಟು, ನುಡಿಗಟ್ಟುಗಳು

ಆ) ವ್ಯಾಕರಣಜ್ಞಾನ : ನಾಮ. ಲಿಂಗ, ವಚನ. ವಿಭಕ್ತಿ, ಸಂಧಿ, ಸಮಾಸ

ಇ) ವಾಕ್ಯಜ್ಞಾನ : ವಾಕ್ಯ ಮತ್ತು ವಾಕ್ಯದ ವಿಧಗಳು, ಪ್ರಬಂಧ ರಚನೆ, ಸಂಕ್ಷೇಪ ಲೇಖನ ಲೇಖನ ಚಿಹ್ನೆಗಳು, ಭಾಷಾಂತರ ಕೌಶಲ

ಪ್ರಶ್ನೆ ಪತ್ರಿಕೆ ಮಾದರಿ

ಮುಕ್ತ ಐಚ್ಛಿಕ ಪತ್ರಿಕೆ

ಜಿ.ಇ - Generic Elective

ಸಮಯ: 2 ಗಂಟೆಗಳು

ಗರಿಷ್ಠ

ಅಂಕಗಳು: 35

ಘಟಕ -1

(ಅಂಕಗಳು:

20)

1. ಒಂದು ಪ್ರಶ್ನೆಗೆ ಉತ್ತರಿಸಿ (ಎರಡು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

1 X 10 =

10

2. ಎರಡು ಪ್ರಶ್ನೆಗಳಿಗೆ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

2 X 5 = 10

ಘಟಕ - 2

(ಅಂಕಗಳು:

15)

3. ಎರಡು ಪ್ರಶ್ನೆಗಳಿಗೆ ಉತ್ತರಿಸಿ (ನಾಲ್ಕು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

2 X 5 = 10

4. ಒಂದೊಂದು ವಾಕ್ಯದಲ್ಲಿ ಉತ್ತರಿಸಿ (ಐದು ಪ್ರಶ್ನೆಗಳನ್ನು ಕೇಳುವುದು)

5 X 1 = 5

**JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(AUTONOMOUS)
B.N. ROAD, MYSURU-570025**



DEPARTMENT OF POLITICAL SCIENCE

**UG Syllabus
CHOICE BASED CREDIT SYSTEM
For B.A.Programmes**

- History,Economics,Political science(HEP)
- Kannada, Economics,Political science(KEP)
- Journalisam,Political science,English(JPE)

2019-20

**JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(AUTONOMOUS)
B.N. ROAD, MYSURU-25
DEPARTMENT OF **POLITICAL SCIENCE**
(CBCS Scheme of Syllabus 2019-20)**

**CODE:BA 22: HISTORY, ECONOMICS AND POLITICAL SCIENCE
CODE:BA 26: KANNADA, ECONOMICS AND POLITICAL SCIENCE
CODE:BA 25: JOURNALISAM, POLITICAL SCIENCE,ENGLISH**

Sem	CORE COURSE	Course	Teaching Hours Per week	L T P	Credit	I.A Marks	Theory Exam	Total Marks
I	Core-I ELA26022/25/26	Political Theory	5+1	5:1:0	6	30	70	100
II	Core-II ELB26022/25/26	Indian Government and Politics	5+1	5:1:0	6	30	70	100
III	Core - III ELC26022/25/26	Comparative Government and Politics	5+1	5:1:0	6	30	70	100
IV	Core – IV ELD26022/25/26	Introduction to International Relations	5+1	5:1:0	6	30	70	100
V	DSE – 1A ELE26022/25/26	Themes in comparative political Theory	5+1	5:1:0	6	30	70	100
V	DSE – 1B ELE26222/25/26	Legislative Support	5+1	5:1:0	6	30	70	100
V	GE-1 ELE26242/25/26	Interdisciplinary Reading Gandhi	2	2:0:0	2	15	35	50
VI	DSE – 1A ELF26022/25/26	Modern governments (UK, USA ,Swiss)	5+1	5:1:0	6	30	70	100
VI	DSE – 1B ELF26222/25/26	Public Administration concepts and theories	5+1	5:1:0	6	30	70	100
VI	GE-2 ELF26242/25/26	Interdisciplinary Reading Gandhi	2	2:0:0	2	15	35	50
I,II BA, B.Com BSc, BCA, BBA, B. Voc	AECC DMA/ CDA/ ECA/ ELB/ ENB13001	Indian constitution	4	3:1:0	4	30	70	100

Scheme of assessment

Year	SEM	COURSE CODE	TITLE OF THE PAPER	CONTINUOUS ASSESSMENT				MAX. MARKS			DURATION OF EXAM
				C1		C2		C3	TH	IA	TH
				TH	AS	TH	SE				
I	I	ELA26022	Political Theory	10	5	10	5	70	70	30	3h
	II	ELB26022	Indian Government and Politics	10	5	10	5	70	70	30	3h
II	III	ELC26022	Comparative Government and Politics	10	5	10	5	70	70	30	3h
	IV	ELD26022	Introduction to International Relations	10	5	10	5	70	70	30	3h
III	V	ELE26022	Theme in comparative political theory	10	5	10	5	70	70	30	3h
		ELE26222	Legislative Support	10	5	10	5	70	70	30	
	VI	ELF26022	Modern governments (UK, USA ,Swiss)	10	5	10	5	70	70	30	3h
		ELF26222	Administration and public policy concept & theories	10	5	10	5	70	70	30	
	V	ELE26422	Interdisciplinary Reading Gandhi	7.5	-	7.5	-	35	35	15	2h
	VI	ELF26422	Interdisciplinary Reading Gandhi	7.5	-	7.5	-	35	35	15	2h
BA, B.Com, BSc, BCA, BBA, B. Voc	I.II.V	DMA13001 CDA13001 ECA13001 ELB13001 ENB13001	Indian Constitution	15	-	15	-	70	70	30	3h

TEMPLATE

Subject: Political Science

Sem	Core			DSE			GE		Total Hours
	No. of Courses	Credits	Total Hours	No. of Courses	Credits	Total Hours	No. of Courses	Credits	
I	1	6	75+15	-	-	-	-	-	90
II	1	6	75+15	-	-	-	-	-	90
III	1	6	75+15	-	-	-	-	-	90
IV	1	6	75+15	-	-	-	-	-	90
V	-	-	-	1	6	75+15	1	2	30
VI	-	-	-	1	6	75+15	1	2	30

Programme Outcome

After completing the graduation in B A (HEP) the students are able to:

- PO1. Critically recognize the social, political, economic and cultural aspects of History.
- PO2. Demonstrate thinking skills by analysing, synthesizing, and evaluating historical information from multiple sources.
- PO3. Extract evidence from primary sources by analysing and evaluating them in relation to their cultural and historical context.
- PO4. Develop an informed familiarity with multiple cultures.
- PO5. Apply economic analysis to everyday problems in real world situations
- PO6. Develop critical and quantitative thinking skills
- PO7. Explain, graph, and analyze key economic models
- PO8. Spread the messages of equality, nationality, social harmony and other human values.
- PO9. Comprehend the basic structures and processes of government systems and/or theoretical underpinnings.
- PO10. Analyse political problems, arguments, information, and/or theories.
- PO11. Apply methods appropriate for accumulating and interpreting data applicable to the discipline of political science.
- PO12. Acquire the ability to engage in independent and life-long learning in the broadest context socio-technological changes

Programme Specific Outcomes

On Completion of BA (HEP) students will:

- PSO1. Exerts its Influence on life and destiny of Human beings
- PSO2. Act as a stepping stone for one's success in competitive examinations
- PSO3. Understand theoretical and practical aspects of Economics and Geography
- PSO4. Evaluate Economic behaviour inconsonance with Geographical factors
- PSO5. Suggest the policy makers about desirable changes to be made in Micro and Macro Economic issues based on geographical factors
- PSO6. Gain ability to understand the economic problems in Geographical indicators
- PSO7. Create appropriate and efficient Historians, Political Leaders, administrators and State's man
- PSO8. Educate about patriotism toleration and National Integration

Programme Outcome

After completing the graduation in B A (KEP) the students are able to:

- PO1. Develop human values and a sense of social service
- PO2. Become a responsible and dutiful citizen.
- PO3. Able to enhance critical temper and creative ability
- PO4. Apply economic analysis to everyday problems in real world situations
- PO5. Develop critical and quantitative thinking skills
- PO6. Explain, graph, and analyze key economic models
- PO7. Spread the messages of equality, nationality, social harmony and other human values.
- PO8. Comprehend the basic structures and processes of government systems and/or theoretical underpinnings.
- PO9. Analyse political problems, arguments, information, and/or theories.
- PO10. Apply methods appropriate for accumulating and interpreting data applicable to the discipline of political science.
- PO11. Acquire the ability to engage in independent and life-long learning in the broadest context socio-technological changes

Programme Specific Outcomes

On Completion of BA (KEP) students will:

- PSO 1: Know more specific terminologies along with its etymology
- PSO2: Know the changes in the differences in formation of society and their culture
- PSO3: Help to understand different races, Society, and culture.
- PSO4. Understand theoretical and practical aspects of Economics and Geography
- PSO5. Evaluate Economic behaviour inconsonance with Geographical factors
- PSO6. Suggest the policy makers about desirable changes to be made in Micro and Macro Economic issues based on geographical factors
- PSO7. Act as a stepping stone for one's success in competitive examinations
- PSO8. Create appropriate and efficient Historians, Political Leaders, administrators and State's man
- PSO9. Educate about patriotism toleration and National Integration

Programme Outcome

After completing the graduation in B A (JPE) the students are able to:

- PO1. Acquire a functional knowledge of the underlying principles and recent emerging trends of the media industry
- PO2. Create and design emerging audio/video media products
- PO3. Conceptualise, create and design strategies for high-quality media content for various digital platforms
- PO4. Appreciate and demonstrate the ability to produce reliable data outcome
- PO5. Comprehend the basic structures and processes of government systems and/or theoretical underpinnings.
- PO6. Analyse political problems, arguments, information, and/or theories.
- PO7. Apply methods appropriate for accumulating and interpreting data applicable to the discipline of political science.
- PO8. Demonstrate critical reading, writing and thinking skills
- PO9. Locate, evaluate, organise and incorporate information effectively
- PO10. Develop and carry out research projects

Programme Specific Outcomes

On Completion of BA (JPE) students will:

- PSO1. Gather, analyse and create journalism on contemporary issues for print, broadcast and other digital media
- PSO2. Produce work incorporating diverse groups and perspectives on local, national and international issues
- PSO3 demonstrate an awareness of journalism as an ethical practice
- PSO4. Exerts it's Influence on life and destiny of Human beings
- PSO5. Act as a stepping stone for one's success in competitive examinations
- PSO6. Create appropriate and efficient Historians, Political Leaders, administrators and State's man
- PSO7. Educate about patriotism toleration and National Integration
- PSO8. Write and edit clear, graceful, grammatically correct prose
- PSO9 .Analyse numerical data and utilise database for multi-layered storytelling
- PSO10. Demonstrate preparation for an entry-level position in the profession through a portfolio exhibiting their work

Course outcome
I – Semester

Political Science I: Political Theory

Course Outcome:

On completion of the Course, students will:

- CO1. Learn in depth meaning and nature of political theory
- CO2. Deliberate in details with examples differences between politics and political theory
- CO3. Understand the characteristics of elements of state
- CO4. Specify the details of civil society
- CO5. Understand the classification and characteristics of rights
- CO6. Specify the classification and characteristics of democracy

I. Introduction to Political Theory and Politics

- a. Political science – Meaning, Nature, scope and Importance
- b. Political Theory – Meaning, Nature and Importance
- c. Differences between Politics and Political Theory
- d. Approaches the study of methods in political science Historical, Behavioural, comparative

II. State – Citizenship – Civil Society

- a. State – Meaning and Elements
- b. Theories- Historical, Social Contract theory, Marxist theory
- c. Citizenship – Concepts, Kinds, Methods of accruing and Loosing Citizenship
- d. Civil Society – Meaning Role and Importance

III. Basic Concepts – I

- a. Liberty – Meaning, Types, Importance and Limitations
- b. Equality – Meaning, Types, Importance, Limitations
- c. Justice - Meaning, Types, Importance, Limitations

IV. Basic Concepts – II

- a. Rights - Meaning, Types, Importance, Limitations
- b. Duties - Meaning, Types, Importance, Limitations

V. Democracy

- a. Democracy – Meaning, Nature, Principles & Forms
- b. Democracy Developments-economic ,social cultural, political
- c. Essentials & Hindrance

References:

1. Bharagava, R and Acharya-political theory
2. Acharya, a-Equality
3. R.G. Gettle-Principles of political science
4. B.K. Ghokhale-A Study of political theory
5. Appadori-Substance of politics
6. s.p varma-modern political theory
7. A.C. kapur-principles of political theory
8. V.D mahajan-political theory
9. R.C. Agrawala- political theory
10. PROF.N. Hallappa-political science

ELB26022/25/26

II Semester

Political Science II: Indian Government and Politics

Course Outcome:

On completion of the Course, students will:

- CO1 Identify the characteristics of Indian politics
- CO2 Understand the characteristics of Indian constitution
- CO3 Understand in details with application, if applicable, federalism
- CO4 Identify the classification and characteristics of power structure in India
- CO5 Identify the details of party system in India

Unit I

- a. Indian Politics: Nature of the State in India
- b. Approaches to Indian Politics:
 - (1) Liberal (2) Marxist and Gandhian
- c. Federalism in India – Federal features, unitary features, challenges to Indian Federalism

Unit II

- a. Basic features of Indian Constitution
- b. Debates on Fundamental Rights
- c. Directive Principles of State Policy,

Unit III

- a. Parliament
- b. Prime Minister & Council Ministers
- c. Supreme Court of India

Unit IV

- a. Power Structure in India
- b. Caste, Class, Patriarchy
- c. Religion and Politics, Secularism and Communalism
- d. Social movements, Peasants, Environmental, Women's Movement

Unit V

- a. Features of Political Party System in India
- b. Organization and Functions – INC, BJP
- c. Role of Regional Parties

Reference:

1. Indian Government and Politics -Abbas, H, Kumar R & Alam
2. Contemporary India - Chandhoke N
3. Indian Government and Politics - Chakravarthy B
4. Indian Politics -Singh
- 5.indian constitution -faida
6. Indian Government and Politics -Dr.S.N.dubey
- 7.Indian political system - R.C.Agarwal
8. indian constitution -Vidya bhushan and Vishnu bhagavan

III Semester
Political Science III: Comparative Government & Politics

Course Outcome:

On completion of the Course, students will:

- CO1 Specify the details of comparative governments
- CO2 Understand the details of classification of political systems
- CO3 Understand the classification and characteristics of electoral systems
- CO4 Learn the classification and characteristics of contemporary debates on state
- CO5 Understand in details with application, if applicable, contemporary debates

Unit I. Comparative Politics

- a. Nature, Scope, Methods
- b. Authoritarian and Democratic Regimes

II. Classifications of Political Systems

- a. Parliamentary and Presidential : UK & USA
- b. Federal & Unitary : India and China

III. Electoral Systems

- a. Universal adult franchise-Meaning, merits and demerits
- b. Election commission-composition, powers and functions
- c. Electoral Systems

IV. Party Systems

- a. Single Party System
- b. Bi-party System
- c. Multi-party System

V. Issues in comparative politics

- a. Political Development – Meaning and characteristics
- b. political socialization
- c. political modernization

Reference:

1. Baraj & Pennington m – comparative politics
2. Caramani – Comparative Politics
3. Hague Rand Harrop. M – The Political executive in comparative government and politics
4. Evans, Jocelyn A J Government and Politics

ELD26022/25/26

IV Semester
Political Science IV: Introductions to International Relations

Course Outcome:

On completion of the Course, students will:

- CO1 Identify the classification and characteristics of approaches of international relations
- CO2 Specify the classification and characteristics of cold war
- CO3 Specify in depth India and her neighbours relationship
- CO4 Learn the details of relations of India with great powers
- CO5 Write down the characteristics of India's foreign policy

I. Introduction:

- a. Meaning Scope & Importance of INR
- b. Approaches – Realism, Neo-Realism, Neo Liberalism
- c. Structural Approaches

II. National power

- a. Meaning and importance of national power
- b. Elements of National power
- c. Limitations of national power

III. Instruments of Foreign policy

- a. War-meaning, causes, effects and remedies
- b. Diplomacy-meaning, kinds and functions privileges and immunities of diplomats
- c. Propaganda-meaning and techniques
- d. Pacific settlement of in disputes

IV. Balance of power-collective security-disarmament

- a. Balance of power-Meaning Importance and techniques
- b. collective security-Meaning and Importance
- c. disarmament - Meaning and Importance

V. India & Her neighbours

- a. India & Pakistan
- b. India & China

Reference:

1. R.J. and Tervis R – Concepts and Contemporary Issues
2. Smith M – Perspective on World Politics
3. Basu, Rumki – International Politics

ELE26022/25/26

V Semester
Discipline Specific Elective (DSE1A)
Themes on Comparative Political Theory

Course Outcome:

On completion of the Course, students will:

- CO1 Understand in details with application, if applicable, democracy and governance
- CO2 Understand in details with application, if applicable, Indian political thought
- CO3 Specify in depth Indian political thought
- CO4 Identify the classification and characteristics of western political thought
- CO5 Understand in details with examples western political thought
- CO6 Understand in depth local government
- CO7 Learn the details of regulatory institutions
- CO8 Identify the classification and characteristics of lobbying institutions

Unit I

- a. Distinctive features of Indian political thought
- b. Kautilya on State
- c. Tilak and Gandhi on Swaraj

Unit II

- a. Ambedkar and Lohia on Social Justice
- b. Nehru and Jayaprakash Narayan on Democracy
- c. John Locks on Rights

Unit III

- a. Distinctive features of western political thought
- b. Aristotle on citizenship
- c. plato –ideal state

Unit IV

- a. J.J. Rousseous on Inequality
- b. J.S. Mill on liberty and democracy
- c. H.J.Laski- Rights,Liberty,Equality

Reference:

- Dallmayr F – Comparative Political Theory
- Sparks, Oh and Isaacs s – Political Theory its in context
- Zelliot E – Political Thought in Modern India

ELE26222/25/26

V- SEMESTER
Discipline Specific Elective (DSE1B)
Legislative support

Course Outcome:

On completion of the Course, students will:

- CO1 Understand in depth legislative support
- CO2 Write down the details of legislative process
- CO3 Write down the details of legislative committees
- CO4 Learn in details with examples legislative committees
- CO5 Identify in details with application, if applicable, budget process

Unit I a. Parliament, composition Power and functions,
 b.State legislature - composition Power and functions,
 .c.Rural and urban local self government from Zilla parishads / municipal
 corporation to panchayat

Unit II a. law making process in parliament
 b.Role of the Standing Committee in reviewing a Bill, Legislative
 Consultations, amendments to a Bill, the framing of Rules and Regulations.

Unit III a. meaning ,nature and scope financial administration
 b. Importance of financial administration
 c.Budget- preparation
 Enactment
 execution

Unit IV. a. Importance of Accounting Auditing
 b.The comptroller and auditor general of India-powers and functions
 c. public accounts committee, estimate committee

Essential Readings:

- Avasthi and maheshwari public administration
 - Madhavan, M R & N Wahi financing of election campaigns
 - Vanka, S primer on MPLADS centre for policy research
 - Karla, H Public engagement with the legislative process
 - Agerwal o.P and T V Somanathan, “Public Policy Making in India: Issues and Remedies,” Feb, 2005
 - Debroy, bibek, “Why we Need Law Reform”.

ELF26022/25/26

VI SEMESTER(DSE1A)
MODERN GOVERNMENTS (U.K U,S.A.SWISS)

Course Outcome:

On completion of the Course, students will:

CO1 Understanding the world politics

CO2 Enlightening the world governmental system

CO3 Develop comparative study on governmental systems

Unit I UNITED KINGDOM

- a. Evolution, features of UK Constitution and role of conventions
- b. Parliament- Composition powers and functions
- c. Crown-prime minister- powers and functions

Unit II

- a. Judiciary in UK- Composition and jurisdiction
- b. Role of political parties in UK

Unit III : UNITED STATES OF AMERICA

- a. Features of USA Constitution
- b. Congress - Composition powers and functions
- c. President-Election, Role powers and functions

Unit IV:

- a. Judiciary in USA- Composition and jurisdiction
- b. political parties in USA

Unit IV: SWISS

- a. Features of Swiss Constitution
- b. Federal Assembly- Composition powers
- c. Federal Council- Composition powers
- d. Federal Tribunal- Composition powers
- e. Devices of Direct Democracy in Swiss

Reference:

- A.C Kapoor - Select constitution
- H.t.Ramakrishna - Adunika sarakaragalu
- U.Gurumurthy - Adunika sarakaragalu
- Vidyabhushan and- World Constitution
- Vishnu Bagavan

ELF26222/25/26

VI Semester
Skill Enhancement Course (DSE1B):
Public Administration Concepts & Theories

Course Outcome:

On completion of the Course, students will:

- CO1 Learn in depth Administration and Public Policy
- CO2 Specify the details of administrative theories
- CO3 Learn the classification and characteristics of administrative theories
- CO4 Deliberate the details of public policy
- CO5 Deliberate in details with examples public policy in India
- CO6 Identify the characteristics of public policy in India

Unit I Public Administration

- a. Meaning, Scope and Significance
- b. Public & Private Administration
- c. Approaches to Public Administration

Unit II Administrative Theories

- a. Classical Theory & Human Relations Theory
- b. Scientific Theory
- c. Rational decision making theory

Unit III. Management

- a. Meaning, importance and principles
- b. Features and functions of management
- c. Chief executive –functions and is role

Unit IV. contemporary issues in administration

- a. Meaning and modes of corruption
- b. Good governance
- c. E- governance

Reference:

- Awasthi A and Maheshwari - Public Administration
- BasuRumki, Public Administration - Concepts and Theories
- Henry N - Public Administration and Public Affairs
- Mouzelis N P - The Ideal type of Bureaucracy
- Mooney - The Principles of Organization
- Esman M T - Politics & Development Administration

ELE26242/25/26

Generic Elective: Interdisciplinary Reading Gandhi

Course Outcome:

On completion of the Course, students will:

- CO1 Specify the details of reading Gandhi
- CO2 Deliberate in depth Gandhi and hind swaraj
- CO3 Learn the details of Gandhi's views on nationalism
- CO4 Understand the details of Gandhi's views on communal unity
- CO5 Understand in details with examples Gandhi's views on women

Unit I:Introduction

- a. Biography (a brief)
- b. Commentaries on hind swaraj
- c. Truth –non-violence
- d. Satyagraha
- e. Modernity

Unit II:

- a. State and religion
- b. Nationalism
- c. Communal Unity
- d. Women's Question
- e. Untouchability
- f. Crapt of ramarajya
- g. Economical idias

Reference:

- A.J. Parel(1997)- Introduction, M.K Gandhi , Hinda swaraj and other writings
- B Parekh, -Gandhi(1997)
- D.hardiman, -Gandhi in his time and ours(2003)

DMA13001/CDA13001/ECA13001/

ELB13001/ENB13001/ FPC/SDC550

Indian Constitution
Ability enhancement compulsory course (AECC)
(BA, B.Com, BSc, BCA, BBA, B. Voc)

Course Outcome:

On completion of the Course, students will:

- CO1 Deliberate in details of preamble of Indian constitution
- CO2 Deliberate the details with examples fundamental rights
- CO3 Understand the classification and characteristics of union government of India
- CO4 Understand the details of supreme court of India
- CO5 specify the details of state government of India
- CO6 specify in details with application, if applicable, right to information act

Unit-I

- a. Meaning and Importance of Constitution
- b. Preamble of the Indian Constitution
- c. Salient features of Indian Constitution

Unit II

- a. Fundamental Rights
- b. Fundamental Duties
- c. Directive Principles of State Policy

Unit III

- a. President – Election methods, Powers and Functions
- b. The Role of the Prime Minister
- d. The Parliament – Structure, Power and Functions (Lok Sabha and Rajya Sabha)

Unit IV

- a. The Role of Governor in the Administration of State
- b. Powers and Functions of the Chief Minister
- c. Composition, Powers and Functions of both the Houses of State Legislature

Unit V

- a. Supreme Court – Organization and Jurisdiction
- b. High Court – Organization and Jurisdiction
- c. Public Interest Litigation
- d. Right to Information Act

Reference:

- M V Pylee - An Introduction to the Constitution of India
- D D Basu - An Introduction to the Constitution of India
- H B Mallikarjuna Swamy - Constitution of India
- J C Johri - Indian Government and Politics
- Faida - India's Constitution
- S R Maheshwari - Indian Government
- Dr.S.N Dubey - Indian Government and Politics
- R.C Agarwal - Indian Political System
- Vidya Bhushan and Vishnu Bhagavwan - Indian Constitution
- qÁ. °ÉZi. JªÀiï gÁd±ÉÃRgĩ - ``sÁgÀvÀzÀ ,ÁPÁðgÀ ªÀvÀÄÛ gÁdQÃAiÄÄ
- qÁ.PÉ.eÉ. ,ÄÄgÉÃ±i - ``sÁgÀvÀzÀ ,ÁPÁðgÀ ªÀvÀÄÛ gÁdQÃAiÄÄ

Department of Political Science
Model Question Paper for BA Student under CBCS

Time: 3 Hours

Marks: 70

PART – A

Note: Answer any four questions each question carries Ten marks 4 x 10 = 40

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____

PART – B

Note: Answer any Two questions each question carries Fifteen Marks 2 x 15 = 30

8. _____
9. _____
10. _____

Department of Political Science
Model Question for Genric Elective Paper under CBCS

Time: 2Hours

Marks: 35

PART – A

Note: Answer any Three questions each question carries five marks 3x 5 = 15

1. _____
2. _____
3. _____
4. _____
5. _____

PART – B

Note: Answer any two questions each question carries Ten Marks 2 x 10 = 20

6. _____
7. _____
8. _____
9. _____



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE

(Autonomous)

B N ROAD, MYSURU- 570 025

DEPARTMENT OF ZOOLOGY

Syllabus

CHOICE BASED CREDIT SYSTEM

For B.Sc programmes

Chemistry, Botany, Zoology

Chemistry, Zoology, Biotechnology

2019-20

Year	Sem	Course	Title of the Paper	Paper code	Credits	Maximum Marks						Exam Duration		
						L:T:P	Th	Pr	THEORY		PRACTICAL		Th	Pr
									C1	C2	C1	C2		
I BSc	I	DSC	Theory-Animal Diversity	DMA30005	4:0:0	70	-	15	15					
			Practical- Animal Diversity	DMA30105	0:0:2	-	35	-	-	7.5	7.5	3H	3H	
	II	DSC	Theory -Comparative Anatomy and Developmental Biology of Vertebrates	DMB30005	4:0:2	70	-	15	15	-	-	3H	3H	
			Practical- Comparative Anatomy and Developmental Biology of Vertebrates	DMB30105	0:0:2	-	35			7.5	7.5			
II BSc	III	DSC	Theory-Physiology and Biochemistry	DMC30005	4:0:0	70	-	15	15	-	-			
			Practical- Physiology and Biochemistry	DMC30105	0:0:2	-	35			7.5	7.5	3H	3H	
	IV	DSC	Theory-Genetics and Evolutionary Biology	DMD30005	4:0:0	70	-	15	15	-	-			
			Practical- Genetics and Evolutionary Biology	DMD30105	0:0:2	-	35			7.5	7.5	3H	3H	
III BSc	V	DSE (ELECTIVE 1)	Theory-Applied Zoology	DME30005	4:0:0	70	-	15	15	-	-			
			Practical- Applied Zoology	DME30105	0:0:2	-	35			7.5	7.5	3H	3H	
		DSE (ELECTIVE 2)	Theory-Insect, Vector and Diseases	DME30205	4:0:1	70	-	15	15	-	-			
			Practical- Insect, Vector and Diseases	DME30305	0:0:2	-	35			7.5	7.5	3H	3H	
	VI	DSE (ELECTIVE 1)	Theory-Aquatic Biology	DMF30005	4:0:0	70	-	15	15	-	-			
			Practical- Aquatic Biology	DMF30105	0:0:2	-	35			7.5	7.5	3H	3H	
		DSE (ELECTIVE 2)	Theory-Immunology	DMF30205	4:0:1	70	-	15	15	-	-			
			Practical- Immunology	DMF30305	0:0:2	-	35			7.5	7.5	3H	3H	
VI	SEC-1	Theory-Medical Diagnostics	DMF30405	2:0:0			15	15	-	-	2H	-		
	SEC-2	Theory-Aquarium Fish keeping	DMF30605	2:0:0	35	-	15	15	-	-				

Scheme of study for B.Sc. Zoology under CBCS scheme 2019-20

B.Sc. Degree syllabus – Programme- CZBt

Year	Sem	Course	Title of the Paper	Paper code	Credits	Maximum Marks						Exam Duration		
						L:T:P	Th	Pr	THEORY		PRACTICAL		Th	Pr
									C1	C2	C1	C2		
I BSc	I	DSC	Theory-Animal Diversity	DMA30008	4:0:0	70	-	15	15					
			Practical- Animal Diversity	DMA30108	0:0:2	-	35	-	-	7.5	7.5	3H	3H	
	II	DSC	Theory -Comparative Anatomy and Developmental Biology of Vertebrates	DMB30008	4:0:2	70	-	15	15	-	-	3H	3H	
			Practical- Comparative Anatomy and Developmental Biology of Vertebrates	DMB30108	0:0:2	-	35			7.5	7.5			
II BSc	III	DSC	Theory-Physiology and Biochemistry	DMC30008	4:0:0	70	-	15	15	-	-			
			Practical- Physiology and Biochemistry	DMC30108	0:0:2	-	35			7.5	7.5	3H	3H	
	IV	DSC	Theory-Genetics and Evolutionary Biology	DMD30008	4:0:0	70	-	15	15	-	-			
			Practical- Genetics and Evolutionary Biology	DMD30108	0:0:2	-	35			7.5	7.5	3H	3H	
III BSc	V	DSE (ELECTIVE 1)	Theory-Applied Zoology	DME30008	4:0:0	70	-	15	15	-	-			
			Practical- Applied Zoology	DME30108	0:0:2	-	35			7.5	7.5	3H	3H	
		DSE (ELECTIVE 2)	Theory-Insect, Vector and Diseases	DME30208	4:0:1	70	-	15	15	-	-			
			Practical- Insect, Vector and Diseases	DME30308	0:0:2	-	35			7.5	7.5	3H	3H	
	VI	DSE (ELECTIVE 1)	Theory-Aquatic Biology	DMF30008	4:0:0	70	-	15	15	-	-			
			Practical- Aquatic Biology	DMF30108	0:0:2	-	35			7.5	7.5	3H	3H	
		DSE (ELECTIVE 2)	Theory-Immunology	DMF30208	4:0:1	70	-	15	15	-	-			
			Practical- Immunology	DMF30308	0:0:2	-	35			7.5	7.5	3H	3H	
VI	SEC-1	Theory-Medical Diagnostics	DMF30408	2:0:0	35	-	15	15	-	-	2H	-		
	SEC-2	Theory-Aquarium Fish keeping	DMF30608	2:0:0	35	-	15	15	-	-				

Scheme of study for B.Sc. Zoology under CBCS scheme 2019-20
B.Sc. Degree syllabus – Programme- CBZ

B.Sc., UG SYLLABUS- PROGRAMME - CZBT

Scheme of study for B.Sc. Zoology under CBCS scheme from 2019-20

YEAR	SEMESTER	COURSE CODE	Title of the paper & Credits	NO. OF CREDITS		LECTURE/ PRACTICAL/ HOUR/WEEK		TOTAL TEACHING HOURS	
				TH	P	TH	P	TH	P
I	I	DMA30005	Animal Diversity	4	2	4	4	60	60
	II	DMA30105	Comparative Anatomy and Developmental Biology of Vertebrates	4	2	4	4	60	60
II	III	DMB30005	Physiology and Biochemistry	4	2	4	4	60	60
	IV	DMB30105	Genetics and Evolutionary Biology	4	2	4	4	60	60
III	V	DMC30005	Applied Zoology	4	2	4	4	60	60
		DMC30105	Insect, Vector and Diseases	4	2	4	4	60	60
	VI	DMD30005	Aquatic Biology	4	2	4	4	60	60
		DMD30105	Immunology	4	2	4	4	60	60
III	VI SEC	DME30005	Medical Diagnostics	2	-	2	-	30	-
		DME30105	Aquarium Fish keeping	2	-	2	-	30	-

B.Sc., UG SYLLABUS- PROGRAMME - CBZ

Scheme of study for B.Sc. Zoology under CBCS scheme from 2019-20

YEAR	SEMESTER	COURSE CODE	Title of the paper & Credits	NO. OF CREDITS		LECTURE/ PRACTICAL/ HOUR/WEEK		TOTAL TEACHING HOURS	
				TH	P	TH	P	TH	P
I	I	DMA30008	Animal Diversity	4	2	4	4	60	60
	II	DMA30108	Comparative Anatomy and Developmental Biology of Vertebrates	4	2	4	4	60	60
II	III	DMB30008	Physiology and Biochemistry	4	2	4	4	60	60
	IV	DMB30108	Genetics and Evolutionary Biology	4	2	4	4	60	60
		DMC30008	Applied Zoology	4	2	4	4	60	60

III	V	DMC30108	Insect, Vector and Diseases	4	2	4	4	60	60
	VI	DMD30008	Aquatic Biology	4	2	4	4	60	60
		DMD30108	Immunology	4	2	4	4	60	60
III	VI SEC	DME30008	Medical Diagonistics	2	-	2	-	30	-
		DME30108	Aquarium Fish keeping	2	-	2	-	30	-

Programme Outcome for Bachelor of Science in Chemistry, Zoology and Biotechnology:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1.** Demonstrate the ability to justify, explain, and/or approach the concept both in written and oral forms
- PO2.** Demonstrate the ability to present clear, logical and succinct arguments
- PO3.** Develop state-of-the-art laboratory skills and professional communication skills.
- PO4.** Apply the scientific method to design, execute, and analyze an experiment.
- PO5.** Appreciate the central role of chemistry in the society and use this as a basis for ethical behaviour in issues facing chemists/drugs.
- PO6.** Understand Chemistry as an integral part for addressing social, economic, and environmental problems.
- PO7.** Identify the major groups of organisms with an emphasis on animals and plants.
- PO8.** Compare and contrast the characteristics of animals that differentiate themselves from other living and non-living creatures.
- PO9.** Give specific examples of physiological adaptations.
- PO10.** Design and develop solution to Biotechnology problems keeping in mind the safety measures for environment and society.
- PO11.** Support Biotechnology research activity with strong technical background knowledge.

Programme Specific Outcome

After completing the graduation in Chemistry, Zoology and Biotechnology the students are able to:

PSO1. Find jobs at all level of chemical, pharmaceutical, food products and life oriented material Industries

PSO2. Apply appropriate techniques for the qualitative and quantitative analysis of chemicals in laboratories and in industries.

PSO3. Recognize the relationship between different structures and functions at different levels.

PSO4. Characterize the biological, chemical and physical features of environments that Animals inhabit.

PSO5. Demonstrate effectively the applications of biochemical and biological sciences.

PSO6. Know and apply appropriate tools and techniques in biotechnological manipulation

PSO7. Understand his or her responsibilities in biotechnological practices.

Programme Outcome for Bachelor of Science in Chemistry, Botany, Zoology:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1.** Demonstrate the ability to justify, explain, and/or approach the concept
- PO2.** Demonstrate the ability to present clear, logical and succinct arguments
- PO3.** Develop state-of-the-art laboratory skills and professional communication skills.
- PO4.** Apply the scientific method to design, execute, and analyze an experiment.
- PO5.** Appreciate the role and use of chemistry for ethical issues facing chemists/drugs.
- PO6.** Identify the taxonomic position of plants using required principles and methods.
- PO7.** Understand the impact of the plant diversity in societal and environmental context.
- PO8.** Use interdisciplinary approaches with quantitative skills to work on biological problems.
- PO9.** Understand Chemistry as an integral part for addressing social, economic, and environmental problems.
- PO10.** Identify the major groups of organisms with an emphasis on animals and plants.

Programme Specific Outcome

Bachelor of Science in Chemistry, Botany and Zoology

After completing the graduation in the Bachelor of Science the students are able to:

PSO1. Find jobs at all level of chemical, pharmaceutical, food products, life oriented material industries

PSO2. Apply appropriate techniques for the qualitative and quantitative analysis of chemicals in laboratories and in industries.

PSO3. Explicate ecological interconnectedness of life

PSO4. Analyze the avenues and remedies for burning environmental issues

PSO5. Recognize the relationship between different structures and functions at different levels.

PSO6. Characterize the biological, chemical and physical features of environment of Animals inhabits.

DMA30005/ DMA30008

**I SEMESTER
CORE COURSE I
ANIMAL DIVERSITY**

Theory

Credits -4(60hrs)

Course outcome

After completion of the course the student is able to:

CO1. Identify in depth non chordata

CO2. Write down the characteristics of non chordata

CO3. Identify in details with examples non chordata

CO5. Learn the classification and characteristics of non chordata

CO6. Write down the details of non chordate

Unit I

15 hrs

INTRODUCTION TO BIODIVERSITY AND NON CHORDATA

Subdivisions of Zoology, Biodiversity and its importance. Principles of animal classification. Binomial and Trinomial nomenclature. Animal kingdom classification based on symmetry, coelom and metamerism. Definition of species.

Protozoa

General characters and classification up to classes with examples

Locomotory organelles in Protozoa: Pseudopodia, Flagella, cilia and myonemes

Locomotion in Protozoa: Amoeboid Movement and Sol- Gel Theory, Euglenoid

movement and Flagellar movement, Ciliary movement and Paddle- Stroke Theory Brief account of locomotion and nutrition in protozoa.

Porifera

General characters and classification upto classes with examples. Canal systems in porifera (Ascon, Sycon, Leucon and Rhagon types), Larval forms of sponges. Economic importance of Porifera.

UnitII

15 hrs

Cnidaria

General characters and classification upto classes with examples. Structure and life cycle of *Obelia* / Metagenesis. Polymorphism in *Physalia*. Structure of a corallite. Types of coral reefs. Importance of corals.

Platyhelminthes

General Character and classification upto classes with examples. External features, Parasitic adaptations, pathogenecity and preventive measures of *Taenia solium* & *Schistoma hematobium*

Aschelminthes

General characters and classification upto classes; Structure,transmission, pathogenicity and preventive measures of *Ascaris* and *Wuchereria*.

UnitIII

15hrs

Annelida

General characters and classification upto classes with examples.

Leech: External morphology, parasitic adaptations, digestive system and reproductive system. Metamerism in annelids. Differentiate between Neries and Heteroneries.

Arthropoda

General characters and classification upto classes with examples

Metamorphosis in insects – (Ametabola, Hemimetabola, Paurometabola, Hypermetabola, Holometabola), hormonal control of metamorphosis. Social life in bees and termites.

Economic importance of insects.

UnitIV

15hrs

Mollusca

General characters and classification upto classes with examples; Torsion in gastropods, Foot in Mollusca- Amphineura, Scaphopoda, Pelicypoda, Gastropoda, Cephalopoda.

Fresh water mussel: External morphology, Digestive system.

Economic importance of Mollusca

Echinodermata

General characters and classification upto classes with examples.

Starfish: External features, Digestive system & water vascular system in starfish.

Pedicellaria and Aristotle's Lantern of Sea urchin

Larval forms and phylogenetic significance.

Reference:

1. Barnes, R.D. 1974, Invertebrate Zoology – III edition by W.B. Saunders Co., Philadelphia.
2. Barrington, E.J.W., 1976, Invertebrate structure and function by Thomas Delson & Sons Ltd., London.
3. Chandler A.C. An Introduction to Parasitology, John Willy & Co.,
4. Earthworm Cinderella of Organic Farming-Radha D. Kale
5. Economic Zoology. - Shukla and Upadhyia.
6. Ekambarnatha Ayyar – Invertebrate – Volumes
7. Hyman, L.H. 1940. The Invertebrates Vol. 1, 2.
8. Jordan and Verma P.S. Invertebrate Zoology
9. Kotpal R.L. Invertebrates.
10. Parker. J.J. Haswell, W.A. 1961 Text Book of Zoology vol. 1, 2nd edition. By Macmillan Co., London.
11. [Sandhu](#) G.S, [Harshvardhan Bhaskar](#), Text book on Invertebrate Zoology, Campus Books International, 2002, 2005 – [Vol](#) 1 and 2
12. Simpson G.C. Principles of Taxonomy

DMA30105/ DMA30108

PRACTICAL –I ANIMAL DIVERSITY

1. **A-Study of Microscope.**
B-PROTOZOA-Amoeba, Entamoeba, Polystomella. Euglena, Paramecium, Balantidium, Vorticella.
2. **PORIFERA**-Sycon, Hyalonema, Gemmule, Monaxon spicules.
3. **COELETERATA**-Obelia, Physalia, Aurelia, Ephyra larva, Metridium.
4. **HELMINTHES**-Taenia solium, Planaria, Fasciola, Ascaris male and female.
5. **ANNELIDA**- Pheretima , Nereis, Leech.
Onychophora- Peripatus
6. **ARTHROPODA**- Palaemon, Palamnaeus, Scolopendra, Spirostreptus.
7. **MOLLUSCA**- Chiton, Unio, Sepia, Octopus.
8. **ECHINODERMATA**- Asteropecten, Ophiothrix, Holothuria, Antedon, Pedicellariae of Seaurchin.
9. Study of mouth parts of insects: Cockroach, Mosquito, House fly, and Honey bee (permanent slides)
10. Preparation of protozoan culture by students and observation of protozoan culture
11. Mounting of setae OR foraminiferans
12. Preparation of animal album containing photographs, cut outs, with appropriate write up the above mentioned taxa. Different taxa/topics may be given to different sets of students.**(Invertebrates-05)**
13. Study of animal dissection through digi frog software.
14. Repetition

Note: Charts, Photographs and models can be shown.

SCHEME OF EXAMINATION

Duration: 3 hrs.

Max. Marks: 35 Marks

- | | |
|--|----------|
| 1. Identify and comment on A to C (3 X 3) | 09 MARKS |
| 2. Identify and comment on D to F (3 X 3) | 09 MARKS |
| 3. Mounting | 03 MARKS |
| 4. Identify and write neat labeled diagram the given mouth parts | 04 MARKS |
| 5. Submission of Animal album | 05 MARKS |
| 6. Class Records | 05 MARKS |

SCHEME OF VALUATION

1. Identification & Classification-01,Diagram-01,Salient features-01
2. Identification & Classification-01,Diagram- 01,Salient features -01
3. Procedure-01, Slide preparation – 02
4. Identification-01, labeled Diagram – 03

DMB30005/ DMB30008

**II SEMESTER
CORE COURSE II
COMPARATIVE ANATOMY AND DEVELOPMENTAL BIOLOGY OF
VERTEBRATES**

Theory

Credits -4(60 hrs)

Course outcome

After completion of the course the student is able to:

- CO1.** Write down the characteristics of chordata
- CO2.** Identify in details with examples chordata
- CO3.** Deliberate the details of comparative anatomy
- CO4.** Identify in details with application of developmental biology
- CO5.** Learn in depth developmental biology
- CO6.** Deliberate in details with examples developmental biology

Unit I

15 hrs

CHORDATA

Protochordates

General features and Phylogeny of Protochordata.

Agnatha

General features of Agnatha and classification of cyclostomes upto classes.

Pisces

General features and Classification upto orders with examples; Migration of fishes- Anadromous and Catadromous. Osmoregulation in Fishes, Differentiation of osteichthyes and chondrichthyes.

Amphibia

General features and Classification upto orders with examples
Parental care in Amphibia- Ichthyophis, Alytes, Pipa and Gastrotheca.

Reptiles

General features and Classification upto orders with examples.
Poisonous and non-poisonous snakes, biting mechanism in snakes, Types and importance of snake venom, first aid for snake bite.

Unit II

15 hrs

Aves

General features and Classification upto orders with examples. Flightless birds and their distribution, major types of beaks. Kinds of migration in birds

Mammals

General features Classification up to orders with examples. Origin of mammals, Distribution of prototheria and metatheria with example

Dentition in mammals. Structure of teeth and their types. Adaptive radiation in mammals.

Comparative anatomy: Derivatives of integuments w.r.t glands and digital tips, brief account of Gills, lungs, air sacs and swim bladder. Types of receptors.

Circulatory system: Comparative account of Heart (Shark, Frog, Lizard, Pigeon and Man) and aortic arches (General plan of Shark, Frog, Lizard, Pigeon and Man)

Nervous system: Brain of Shark, Frog, Lizard, Pigeon and Man.

Evolution of Kidneys in vertebrates: Pronephros, mesonephros and metanephros.

Unit III

15 hrs

Early Embryonic Development

Gametogenesis: Spermatogenesis and oogenesis w.r.t. mammals, vitellogenesis in birds,

Fertilization: external (amphibians), internal (mammals)

monospermy and polyspermy, types of egg and egg membrane

Cleavage: Planes of cleavage: meridional, vertical, equatorial and latitudinal cleavages.

Types of cleavages: holoblastic, radial, spiral, meroblastic and superficial cleavage with examples.

Early development of frog ; structure of mature egg, cleavage, blastula, fate map, gastrula and Neurulation in frog embryo.

Unit IV

15 hrs

Late Embryonic Development

Structure of mature spermatozoon, Graafian follicle, Human menstrual cycle, Ovulation, fertilization, morula, blastocyst, implantation and placentation.

Placenta: Histological and morphological classification of mammalian placenta with examples.

Foetal membranes: Development, structure and functions of amnion, chorion, yolk sac and allantoises.

Modern trends in human reproduction : Invitro fertilization, cloning, sperm and egg banks, sexually transmitted diseases(AIDS, syphilis and gonorrhoea).

Reference:

1. Balinsky, B.I. (2008). An introduction to Embryology, International Thomson Computer
2. Carlson, Bruce M (1996). Patten's Foundations of Embryology, McGraw Hill, Inc
3. Gilbert, S. F. (2006). Developmental Biology, VIII Edition, Sinauer Associates, Inc.,
4. Hildebrand, M and Gaslow G.E. Analysis of Vertebrate Structure, John Wiley and Sons
5. Kent, G.C. and Carr R.K. (2000). Comparative Anatomy of the Vertebrates. IX Edition. The McGraw-Hill Companies.
6. Kardong, K.V. (2005) Vertebrates' Comparative Anatomy, Function and
7. Evolution. IV Edition. McGraw-Hill Higher Education
8. Walter, H.E. and Sayles, L.P; Biology of Vertebrates, Khosla Publishing House. Publishers, Sunderland, Massachusetts, USA. Press.
9. Young J.Z. 1950 - Life of Vertebrates – Oxford University Press, London

DMB30105/ DMB30108

PRACTICAL –II COMPARATIVE ANATOMY AND DEVELOPMENTAL BIOLOGY OF VERTEBRATES

1. **PROTOCHORDATA**- Balanoglossus, Herdmania, Myxine, Ammocoetes larva.
2. **PISCES**- Pristis, Torpedo, Labeo, Exocoetus.
3. **AMPHIBIA**- Salamandra, Bufo, Hyla.
4. **REPTILES**- Chamaeleon, Draco, Vipera, Naja, Chelone, Bungarus, Enhydrina.
5. **AVES AND MAMMALS**- Koel, Kite, Duck, Bat, Loris, Guinea pig.
6. Skull of Frog, Pigeon, Rabbit.
7. Vertebrae of Frog (Atlas, Pro, Amphi and Acoelous), Pigeon (Heterocoelous Vertebra, Synsacrum) Rabbit (Atlas, Axis, Thoracic vertebrae).
8. Pectoral and Pelvic girdles of Frog, Pigeon and Rabbit. Fore and hind limb skeletons of Frog, pigeon and Rabbit.
9. Frog - cleavage stages, Blastula, gastrula, neurula stage (whole mount)
10. Chick embryo stages-18hrs, 24hrs, 36 hrs, 48hrs (whole mount and sections)
11. Study of the different types of placenta- Sheep, placenta.
Study of development of hen's egg – Window technique.
12. Identification of various family planning devices.
13. Study of skeletal system through software,
14. Preparation of animal album containing photographs, cut outs, with appropriate write up the above mentioned taxa. Different taxa/topics may be given to different sets of students.
(Vertebrates-05)
15. Repetition,

SCHEME OF EXAMINATION

Duration: 3 hrs.

Max. Marks: 35 Marks

1. Identify and comment on A to C (3 X 3)	09 MARKS
2. Identify and comment on D to F (3 X 3)	09 MARKS
3. Window technique	03 MARKS
4. Identify and comment on the given specimen/device	04 MARKS
5. Submission of Animal album	05 MARKS
6. Class Records	05 MARKS

SCHEME OF VALUATION

1. Identification & Classification-01, Diagram – 01, Salient features – 01
2. Identification & Classification-01, Diagram – 01, Salient features -01
3. Procedure-01, Diagram-01, significance – 01
4. Identification-01, Diagram – 01, Comment-02

DMC30005/ DMC30008

**III SEMESTER
CORE COURSE III
PHYSIOLOGY AND BIOCHEMISTRY**

Theory

Credits -4(60 hrs)

Course outcome

After completion of the course the student is able to:

- CO1.** Specify in depth physiology
- CO2.** Learn in details with examples physiology
- CO3.** Learn the classification and characteristics of biochemistry
- CO4.** Learn the characteristics of biochemistry

Unit I

15 hrs

Nerve and muscle: Structure of a neuron, types of neuron, axonic conduction and synaptic transmission of nerve impulse, neurotransmitters. Ultra-structure of skeletal muscle, sliding filament theory. Chemical composition of muscle.

Digestion: Physiology of digestion in the alimentary canal; Absorption of carbohydrates, proteins, lipids, mechanical process in digestion

Unit II

15 hrs

Respiration : Pulmonary ventilation, respiratory volumes and capacities, Transport of Oxygen and carbon dioxide in blood: Respiratory pigments (Haemoglobin, Haemocyanin, Haemoerythrin and chlorocruorin)

Excretion: Ammonotelism, Ureotelism, Uricotelism with examples .Structure of nephron, Mechanism and physiology of Urine formation, Counter-current Mechanism

Cardiovascular system: Composition of blood, Structure of Human Heart, Origin and conduction of the cardiac impulse, Cardiac cycle

Unit III

15 hrs

Reproduction and Endocrine Glands: Physiology of male reproduction: hormonal control of spermatogenesis; Physiology of female reproduction: hormonal control of menstrual cycle
Gross morphology and function of pituitary, thyroid, pancreas, adrenal, testis and ovary.

Carbohydrate Metabolism : Glycolysis, Krebs Cycle ,Gluconeogenesis; Path way, regulation and significance Glycogen metabolism: Glycogenolysis, glycogenesis, Regulation of blood sugar level (Homeostasis), Diabetes mellitus

Unit IV

15hrs

Protein metabolism: Transamination, Deamination and Urea Cycle.

Lipid Metabolism: Biosynthesis and β oxidation of palmitic acid.

Vitamins: Definition and classification, (fat soluble and water soluble vitamins) physiological role and disorders.

Enzymes: Introduction, Mechanism of action- lock and key hypothesis, induced fit hypothesis
Enzyme kinetics- Effect on enzyme concentration, substrate concentration

Reference:

1. Berg, J.M., Tymoczko, J.L. and Stryer, L. (2006) Biochemistry. VI Edition. W.H Freeman and CO.
2. Guyton, A.C. and Hall, J.E (2011) Text book of medical physiology. XII Edition, Harcourt Asia Pvt. Ltd/W.B. Saunders Company.
3. Harper's Illustrated Biochemistry. XXVIII Edition. Lange Medical Book/McGraw Hill.
4. Jayaraman J. 1981. Laboratory Manual in Biochemistry Wiley Eastern Ltd.
5. Lehninger A.L.: Nelson D. Land Co., M.M. 2nd edition 1993. Principles of Biochemistry, CBS Publishers, New Delhi.
6. Murray, R.K., Granner, D.K., Mayes, P.A and Rodwell, V.W. (2009).
7. Tortora, G.J. and Derrickson, B.H. (2009). Principles of Anatomy and Physiology, XII Edition, John Wiley & Sons, Inc.
8. Nelson, D.L. Cox, m.m. Lehninger, A.L (2009). Principles of Biochemistry. IV Edition. W.H Freeman and Co.
9. Widmaier, E.P., Raff H. and Strang, K.T. (2008) Vander's Human physiology XI Edition. McGraw Hill/

DMC30105/ DMC30108

PRACTICAL –III

PHYSIOLOGY AND BIOCHEMISTRY

PRACTICAL

(CREDITS 2)

1. Qualitative test for carbohydrate -Molisch's test, Iodine test, Fehling's test and Picric acid test.
2. Qualitative test for proteins- Biuret test, Ninhydrin test, Millon's test and Xanthoproteic test.
3. Qualitative test for lipids-Acrolein Test, Leiberman-Burchard test, Salkowasky test.
4. Identification of vitamins presents in the food materials with their physiological role and their disorders.
5. Salivary amylase activity on starch.
6. Dehydrogenase activity in milk.
7. Preparation of hematin crystals.
8. Blood typing –ABO type and Rh factors.
9. Detection of nitrogenous wastes – Ammonia (Nessler's test), Urea (Urease test), Uric acid (Folin's U test).
10. Study of permanent mammalian histological sections of pituitary, thyroid, parathyroid.
11. Study of permanent mammalian histological sections of Pancreas, adrenal gland, testis and ovary.
12. Study of permanent mammalian slides of liver, lung and kidney.
13. Study of permanent slides of spinal cord, bone marrow and T.S of cartilage.
14. **Field Study:** Visit to a Pathology lab / Hospital to know about different techniques of biochemical analysis and submission of the report.
15. Repetition.

SCHEME OF PRACTICAL EXAMINATION

Duration: 3 hrs.

Max Marks: 35

- | | |
|---|---------|
| 1. Biochemistry experiments by lots (1 to 3) | 09Marks |
| 2. Physiology experiments by lots (5 to 9) | 09Marks |
| 3. Identify and comment the slide A (10 to 13) | 03Marks |
| 4. Identify the vitamin present in the food material and comment on their physiological role and disorders. | 04Marks |
| 5. Report | 05Marks |
| 6. Practical records | 05Marks |

SCHEME OF VALUATION

1. Procedure - 03, Experiment- 04, Result - 02
2. Procedure - 03, Experiment- 04, Result - 02,
3. Identification -01, Diagram –01, Comment– 01
4. Identification -01, Sources-01, Physiological role –01, Deficiency disease – 02

034. Identification - 01 mark, Comment - 02 marks,

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03

DMD30005/ DMD30008

**IV SEMESTER
CORE COURSE IV
GENETICS AND EVOLUTIONARY BIOLOGY**

Theory

Credits -4(60 hrs)

After completion of the course the student is able to:

CO1. Learn in depth genetics

CO2. Identify the detail of evolution

CO3. Understand in depth evolution

CO4. Identify in details with application, if applicable, genetics

Unit I

15 hrs

Introduction to Genetics: History of genetics mono and dihybrid crosses, Mendel's law.

Mendelian Genetics and its Extension: Principles of Inheritance, Chromosome theory of inheritance, Incomplete dominance and codominance.

Multiple alleles, lethal alleles, Supplementary gene, Complementary gene, Epistasis, Pleiotropy. Extra-chromosomal inheritance-Maternal inheritance-shell coiling in *Limnaea*. Cytoplasmic inheritance-Kappa partical in *Paramecium*.

Sex Determination: Chromosomal basis of sex determination in animals. Non-disjunction (primary and secondary). Genic balance theory in *Drosophila*. Klinefelter's and Turner's syndromes. Environmental and hormonal effects on determination of sex (*Bonellia* and free martin).

Unit II

15 hrs

Linkage, Crossing Over and Chromosomal Mapping

Linkage and crossing over: Linkage in *Drosophila*, linkage in man; Significance of crossing over. Genetic maps of chromosomes. Construction of chromosome map.

Inborn errors of metabolism: phenylketonuria, alkaptonuria and albinism.

Sex linked inheritance: *Drosophila*- Red and white color inheritance

Man- Haemophilia and colour blindness, Y-linked genes.

Mutations: Chromosomal Mutations: Deletion, Duplication, Inversion, Translocation, Aneuploidy and Polyploidy; Gene mutations: Induced mutations, Spontaneous mutations,

Reverse mutations, Suppressor mutations.

Disorders due to mutant gene in man: sickle cell anaemia, thalassemia, brachydactyly.

Unit III

15 hrs

Introduction to Evolutionary Theories

Concept of organic evolution. Lamarckism, Darwinism, Neo-Darwinism-Industrial melanism (Natural selection).

Evidences of Evolution:

Indirect evidences: comparative morphology & anatomy, connecting links. Analogy, Homology, vestigial organs, Physiology and biochemistry – enzyme, hormones, serology, and amino acid sequence analysis, excretory product analysis, cytology, genetics and embryology.

Direct evidences from palaeontology: Nature, types of fossils. Geological time scale mentioning dominant groups in each period.

Unit IV

15 hrs

ADAPTATIONS

Adaptive radiation : examples, micro and macro evolution – examples.

Adaptations: Aquatic adaptations: Primary (Shark) and secondary (Turtle).

Volant adaptations: Active (Insect, Bird and Bat) and passive flights (Exocoetes, Rhacophorus and Draco).

Arboreal adaptations: Chameleon, Loris.

Desert adaptations: Phrynosoma, Camel. Coloration and mimicry.

Trends in Human evolution: Australopithecus, Paranthropus, Kenyanthropus and *Homo erectus*, *Homo sapiens*.

Ethology: Definition, innate – taxes, reflexes, instincts and motivation, learned behaviour – habituation, imprinting, conditioned reflexes and insight.

Reference:

1. Barton.N.H. Briggs, D.E.G., Eisen, J.A., Goldstein, D.B.and N.H. (2007).
2. Campbell, N.A.and Reece J.B. (2011) Biology.IX Edition, Pearson, Benjamin, Cumming.
3. Douglas,J.Futuyma(1997)Evolutionary Biology.Sinauer Associates Evolution. Cold Spring, Harbour Laboratory Press.
4. Gardner, E.J.Simmons, M.J., Snustad, D.P (2008).Principles of Genetics.
5. Hall,B.K. and Hallgrimsson, B.(2008) Evolution.IV Edition,Jones and Bartlett
6. Klug,W.S.,Cumming,M.R., Spencer,C.A.(2012).Concepts of Genetics Organic Evolution by Veer Bala Rastogi
7. Principles of Genetics by Robert.H.Tamarin, Tata McGraw-Hill pub. Publishers
8. Ridiey, M. (2004) Evolution.III Edition. Blackwell Publishing
9. Russell, P.J. (2009) Genetics-A Molecular Approach.III Edition. Benjamin
10. Snustad, D.P., Simmons, M.J (2009). Principles of Genetics Edition. Stebbins G.L. (1970) The process of organic evolution, Prentice Hall Publ.

11. Strickberger, M.W, (1995) Genetics, Mac Millan Co.VIII Edition. Wiley India X Edition. Benjamin Cummings.

DMD30105/ DMD30108

PRACTICAL –IV

GENETICS AND EVOLUTIONARY BIOLOGY

PRACTICAL

(CREDITS 2)

1. Genetic problems- Monohybrid and Dihybrid cross.
2. Complementary genes: Flower color in sweet pea.
Supplementary genes: Comb patterns in Fowls.
3. Epistatic genes: Plumage colors in Fowls
Multiple genes: Skin color in man.
4. Multiple alleles: ABO blood group in human.
Sex linked inheritance: *Drosophila* (red and white eye colour) and human (Haemophilia and Colourblindness).
5. Study of *Drosophila* and Grasshopper Karyotypes.
6. General morphology of *Drosophila melanogaster* (male and female).
7. Mounting of sex comb and wing of *Drosophila*.
8. Identification of mutants of white eye, bar eye, sepia eye, vestigial wing and ebony body of *Drosophila*.
9. Preparation of media and maintenance and breeding of *Drosophila* (Demonstration).
10. Study of homology – vertebrate forelimbs and analogy-wing of bird & butterfly, serial homology in crustacia from suitable specimens/ pictures.
11. Study of aquatic adaptations: Shark & Turtle.
Study of arboreal adaptations: Chamaeleon, Loris.
12. Study of Volant adaptations : Exocoetus Fish, Bat, Pigeon, Draco.
13. Submission of report on- a. Evolution of Horse

- b. Evolution of Camel
 - c. Evolution of Elephant
 - d. Evolution of Man
14. Experiment on genetic drift-Sampling error and population size.
15. Repetition.

SCHEME OF PRACTICAL EXAMINATION

DURATION: 3 HOURS

MAX.MARKS:35

- | | |
|--|----------|
| 1. Genetics problem any 2 | 10marks |
| 2. Identify with reasons A (Wild male & Female <i>Drosophila melanogaster</i>)
B (Drosophila mutant) | 04 marks |
| 3. Construct, Identify and comment on given Karyotypes | 05 marks |
| 4. Mounting of wing/ sex comb of Drosophila. | 03marks |
| 5. Identify and comment on given specimen | 03 marks |
| 6. Submission of report. | 05 marks |
| 7. Class record. | 05 marks |

SCHEME OF VALUATION

1. Genetic Problems
- 2 .Identification-01, Reasons- 01
3. Identification -01, Construct -02, Comment-02
4. Procedure-01, Mounting-02
5. Identification -01, Comment-02

DME30005/ DME30008

**V SEMESTER
DISCIPLINE SPECIFIC ELECTIVE COURSE
APPLIED ZOOLOGY**

Theory

Credits -4(60 hrs)

Course outcome

After completion of the course the student is able to:

- CO1.** Understand the detail of communicable diseases
- CO2.** Understand the classification and characteristics of medical zoology
- CO3.** Learn the characteristics of applied zoology
- CO4.** Understand in detail with examples applied zoology
- CO5.** Know about the importance of insects in forensic science and medicine.

Unit I

15 hrs

Introduction to Host-parasite Relationship: Host, Definitive host, Intermediate host, Parasitism, Symbiosis, Commensalism, Reservoir, Zoonosis

Epidemiology of Diseases: Transmission, Prevention and control of diseases: Tuberculosis, typhoid

Rickettsiae and Spirochaetes: Brief account of *Rickettsia prowazekii*, *Borrelia recurrentis* and *Treponema pallidum*

Insect Vector Biology: Introduction to vector biology, economic importance and control of fleas, lice, bugs, mosquitoes, flies and parasitoids. Vector-parasite interaction; host-pathogen interaction

Unit II

15 hrs

Parasitic Protozoa: Structure, Life history, pathogenicity and preventive measures of *Entamoeba histolytica*, *Plasmodium vivax*, *Trypanosoma gambiense* and *Leishmania donovani*

Parasitic Helminthes: Structure, Life history and pathogenicity of *Ancylostoma duodenale*, *Tape worm* and *Liver fluke*

Medical Zoology: Infectious/ communicable diseases: small pox, hepatitis, AIDS, influenza, SARS, Ebola, tuberculosis, plague, cholera, amoebiasis, malaria, dengue, chikungunya, trypanosomiasis, and elephantiasis. (Pathogen, Vector, transmission, symptoms and prevention).

Unit III

15 hrs

Insects of Economic Importance: Biology, Control and damage caused by *Helicoverpa armigera* (cottonbal worm), *Pyrilla perpusilla* (sugar cane leaf hopper), *Papilio demoleus* (lime butterfly) *Callosobruchus chinensis*(beetle), and *Tribolium castaneum*(red flour beetle) Lac insect.

Sericulture: Morphology and life cycle of *Bombyx mori*, Types of silks and distribution.

Apiculture: Types of honey bees, products, byproducts and their usage. Pollination and insects with examples.

Unit IV

15 hrs

Animal Husbandry: Preservation and artificial insemination in cattle; Induction of early puberty and synchronization of estrus in cattle.

Poultry Farming: Principles of poultry breeding, Management of breeding stock and broilers, processing and preservation of eggs.

Fish Technology: Genetic improvements in aquaculture industry; Induced breeding and transportation of fish seed.

Reference:

1. Arora, D. R and Arora, B. (2001). Medical Parasitology. II Edition. CBS Publications
2. Atwal, A.S. (1986). Agricultural Pests of India and South East Asia, Kalyani CBCS Undergraduate Program in Zoology
4. Dennis, H. (2009). Agricultural Entomology. Timber Press (OR).
5. Dunham R.A. (2004). Aquaculture and Fisheries Biotechnology Genetic Approaches.CABI publications, U.K.
6. Hafez, E. S. E. (1962). Reproduction in Farm Animals. Lea & Fabiger Publisher
7. Kumar and Corton. Pathological Basis of Diseases.
8. Park, K. (2007). Preventive and Social Medicine. XVI Edition. B.B Publishers.
9. Pedigo, L.P. (2002). Entomology and Pest Management, Prentice Hall. Publishers.

DME30105/ DME30108

**PRACTICAL –V
APPLIED ZOOLOGY**

PRACTICAL

(CREDITS 2)

1. Study of *Plasmodium vivax*, *Entamoeba histolytica*, *Trypanosoma gambiense*, and *Wuchereria bancrofti* and their life stages through permanent slides/photomicrographs or specimens.
2. Study of arthropod vectors associated with human diseases: Pediculus, Culex, Anopheles, Aedes and Xenopsylla.
3. Study of insect damage to different plant parts/stored grains through damaged products/photographs.
4. Identifying feature and economic importance of *Helicoverpa (Heliothis) armigera*, *Papilio demoleus*
5. Demonstration of Plastination techniques by using some dead insects.
6. Maintenance of freshwater aquarium.
7. Collection and mounting of Ants.
8. Animal associations: - Mutualism – Termites and Trichonympha.
9. Commensalism – Echenies and shark.
Protocooperation – Hermit crab and Sea anemone.
10. Predation – Snake and Frog. Parasitism – Head louse, Bed bug, Mosquito, Ticks, Mites.
11. Identification of mulberry and non mulberry silkworms. Identification of different larvae of silk worm- Using specimens / pictures
12. Identification of food fishes of Karnataka (Tilapia, Notopterus, Catla catla, Clarius).
13. Field visits to a Vermiculture / Sericulture / fisheries / apiculture / poultry / dairy farm-
submission of any 1 Report.
14. Repetition.

SCHEME OF PRACTICAL EXAMINATION

Duration 3 Hours

Max Marks: 35

- | | |
|--|----------|
| 1. Identify and comment on A to C (From 1,2,3 & 4) | 09 Marks |
| 2. Identify and comment upon animal association D &E | 06 Marks |
| 3. Identify mulberry or non mulberry silkworms | 03Marks |
| 4. Identify food fishes of Karnataka | 04 Marks |
| 5. Identify different larvae of silk worm- | 03 Marks |
| 6. Report submission | 05 Marks |
| 7. Class record. | 05 Marks |

SCHEME OF VALUATION

1. Identification & Classification-01,Diagram – 01,Comment & role–01
2. Identification – 01,Diagram – 01, Comment-01
3. Identification –01, Diagram – 01, Comment- 01
4. Identification – 01, Diagram – 01, Comment- 01
5. Identification – 01, Diagram – 01, Comment- 02

DME30205/ DME30208

**V SEMESTER ZOOLOGY
DISCIPLINE SPECIFIC ELECTIVE COURSE
INSECT, VECTORS AND DISEASES**

Theory

Credits -4(60 hrs)

Course outcome

After completion of the course the student is able to:

CO1. Learn in detail of insect, vector diseases

CO2. Identify in detail with examples insect, vector diseases

CO3. Deliberate the detail of insect, vector diseases

CO4. Learn in depth Insect, vectors and diseases

Unit I

15 hrs

Introduction to Insects

General Features of Insects, Morphological features, Head – Eyes, Types of antennae, Mouth parts w.r.t. feeding habits

Concept of Vectors : Brief introduction of Carrier and Vectors (mechanical and biological vector), Reservoirs, Host-vector relationship, Vectorial capacity, Adaptations as vectors, Host Specificity

Insects as Vectors: Classification of insects up to orders, detailed features of orders with insects as vectors – Diptera, Siphonaptera, Siphunculata, Hemiptera

Unit II

15 hrs

Dipteran as Disease Vectors

Dipterans as important insect vectors – Mosquitoes, Sand fly, Houseflies;

Study of mosquito-borne diseases – Malaria, Dengue, Chikungunya, Viral encephalitis, Filariasis; Control of mosquitoes

Study of sand fly-borne diseases – Visceral Leishmaniasis, Cutaneous

Leishmaniasis, Phlebotomus fever; Control of Sand fly

Study of house fly as important mechanical vector, Myiasis, Control of house fly

Unit III

15 hrs

Siphonaptera as Disease Vectors

Fleas as important insect vectors; Host-specificity, Study of Flea-borne diseases –Plague, Typhus fever; Control of fleas

Siphunculata as Disease Vectors

Human louse (Head, Body and Pubic louse) as important insect vectors; Study of

louse-borne diseases –Typhus fever, Relapsing fever, Trench fever, Vagabond's disease, Phthiriasis; Control of human louse

Unit VI

15 hrs

Hemiptera as Disease Vectors

Bugs as insect vectors; Blood-sucking bugs; Chagas disease, Bed bugs as mechanical vectors, Control and prevention measures

Reference:

1. Imms, A.D. (1977). *A General Text Book of Entomology*. Chapman & Hall, UK
2. Chapman, R.F. (1998). *The Insects: Structure and Function*. IV Edition, Cambridge University Press, UK
3. Pedigo L.P. (2002). *Entomology and Pest Management*. Prentice Hall Publication
4. Mathews, G. (2011). *Integrated Vector Management: Controlling Vectors of Malaria and Other Insect Vector Borne Diseases*. Wiley-Blackwell

DME30305/ DME30308

V SEMESTER ZOOLOGY

PRACTICAL –V

INSECT VECTORS AND DISEASES

PRACTICAL

(CREDITS 2)

1. Study of different kinds of mouth parts of insects
2. Study of following insect vectors through permanent slides/ photographs:*Pediculus humanus capitis*
3. Study of following insect vectors through permanent slides/ photographs:
4. *Aedes, Culex, Anopheles*
5. Study of following insect vectors through permanent slides/ photographs:*Pediculus humanus*
6. Study of following insect vectors through permanent slides/ photographs:*corporis, Phithirus pubis, Xenopsylla cheopis*
7. Study of following insect vectors through permanent slides/ photographs: *Cimex lectularius,*
8. Study of following insect vectors through permanent slides/ photographs:*Phlebotomus argentipes, Musca domestica,*
9. Study of different diseases transmitted by above insect vectors
10. Study of different diseases transmitted by above insect vectors
11. Repetation.
12. **Submission of a project report on any one of the insect vectors and disease transmitted**

DMF30005/ DMF30008

**VI SEMESTER ZOOLOGY
DISCIPLINE SPECIFIC ELECTIVE COURSE
AQUATIC BIOLOGY**

Theory

Credits -4(60 hrs)

Course outcome

After completion of the course the student is able to:

- CO1.** Learn in detail with examples aquatic biology
- CO2.** Write down in detail with examples aquatic biology
- CO3.** Identify the classification and characteristics of aquatic biology
- CO4.** Identify in depth aquatic biology

UNIT I

15 hrs

Aquatic Biomes

Ecosystems: Concept, types and structure of ecosystem

Hydrosphere: (Water) Physical and chemical properties.

Brief introduction of the aquatic biomes: Freshwater ecosystem (lakes, wetlands, Streams and rivers), estuaries, intertidal zones, oceanic pelagic zone, marine Benthic zone and coral reefs

UNIT II

15 hrs

Fresh water biology

Physico–chemical Characteristics: Light, Temperature, Thermal stratification, Dissolved Solids, Carbonate, Bicarbonates, Phosphates and Nitrates, Turbidity; dissolved gases (Oxygen, Carbon dioxide).

Pond: The pond as an ecosystem – Origin and classification, flora and fauna of pond ecosystem. Freshwater adaptations.

Lakes: Origin and classification, Lake as an Ecosystem

UNIT III

15 hrs

Marine Biology

Salinity and density of Sea water, Continental shelf, Adaptations of deep sea organisms, Coral reefs, Sea weeds.

Aquaculture in India: An overview – nutritional value and food security - Site selection and preparation of culture ponds

Fish culture: carps, marine fishes and ornamental fishes.

Prawn culture: Freshwater prawns and marine shrimps. Fattening of crabs. Crayfish and Lobster

Pearl culture: Types of pearls, composition, Biology of pearl oyster, pearl formation, culture of pearls .

UNIT IV

15hrs

Management of Aquatic Resources

Water pollution: Definition, sources Types and classification of pollutants. Effects of Water Pollution, Oxygen sag curves and Eutrophication of drinking water: Collection, purification and distribution. Waste water treatment: Primary, secondary and tertiary treatment

Causes of pollution: Agricultural, Industrial, Sewage, Thermal and Oil spills, Management and conservation (legislations), Sewage treatment, Water quality assessment- BOD and COD.

Reference:

1. B. Das, (2012). Parasitology, Books & Allied Pvt. Ltd-II
2. S. Bhattacharya (2011). Environmental toxicology, Books & Allied Pvt. Ltd
3. S. Bhattacharya (2011). Environmental toxicology, Books & Allied Pvt. Ltd
4. T. K. Saha. (2012). Ecology and Environmental Biology. Books & Allied Pvt.

DMF30105/ DMF30108

**PRACTICAL VI
AQUATIC BIOLOGY**

PRACTICAL

(Credits 2)

- 1&2.** Study the important macrophytes, phytoplanktons and zooplanktons Present in a lake ecosystem/ photographs.
- 3** Estimation of dissolved oxygen in different water samples.
- 4** Estimation of dissolved carbon dioxide in different water samples.
- 5** Estimation of chlorides in different water samples.
- 6** Estimation of hardness in different water samples.
- 7** Measurement of pH, using pH-meter, pH paper.
- 8** Study of pond ecosystem.
- 9** Study of aquarium ecosystem.
- 10** Morphometric measurement of locally available freshwater fish and marine water fish.
- 11** Identification of fish(any six).
- 12&13.** Fish by products.
- 14** . Project Report on a visit to a Sewage treatment plant/Marine bioreserve/ Fisheries Institutes.
- 15.** Repetition.

SCHEME OF PRACTICAL EXAMINATION

Duration 3 Hours

Max Marks: 35 Marks

1. Estimation of any 2 parameters – by lots.	10Marks
2. Identify the given specimen (A&B.)	06 Marks
3. Identify and comment on phyto or zooplanktons.	04Marks
7. Fish by products.	05 Marks
8. Report.	05 Marks
9. Record.	05 Marks

SCHEME OF VALUATION

- 1.Procedure - 04, Experiment- 04, Result - 02
- 2.Identification -01, Diagram –01,Comment- 01
3. Identification -01,Diagram –01,Comment- 02
4. Identification -01,Comment- 04

DMF30205/ DMF30208

**VI SEMESTER ZOOLOGY
DISCIPLINE SPECIFIC ELECTIVE COURSE
IMMUNOLOGY**

Theory

Credits -4(60hrs)

Course outcome

After completion of the course the student is able to:

CO1. Deliberate in details with application, if applicable immunology.

CO2. Learn in details with examples immunology.

CO3. Identify in details with examples immunology.

CO4. Identify the characteristics of immunology.

CO5. Deliberate in depth immunology.

Unit I: Overview of the Immune System

15 Hrs

Introduction to basic concepts in immunology, components of immune system, principles of innate and adaptive immune system.

Cells and Organs of the Immune System

Haematopoiesis, Cells of immune system and organs (primary and secondary lymphoid organs) of the immune system.

Antigens: Basic properties of antigens, B and T cell epitopes, haptens and adjuvants.

Unit II: Antibodies

15 Hrs

Structure, classes and function of antibodies, monoclonal antibodies, antigen antibody interactions as tools for research and diagnosis

Vaccines: General introduction to vaccines, Various types of vaccines

CBCS Undergraduate Program in Zoology

Unit III: Working of the immune system

15 Hrs

Structure and functions of MHC, exogenous and endogenous pathways of antigen presentation and processing, Basic properties and functions of cytokines, Complement system: Components and pathways.

Unit IV: Immune system in health and disease

15 Hrs

Gell and Coombs' classification and brief description of various types of hypersensitivities, Introduction to concepts of autoimmunity and immunodeficiency

Reference:

1. Kindt, T. J., Goldsby, R.A., Osborne, B. A. and Kuby, J (2006). *Immunology*, VI Edition. W.H. Freeman and Company.
- 2, David, M., Jonathan, B., David, R. B. and Ivan R. (2006). *Immunology*, VII Edition, Mosby, Elsevier Publication.
3. Abbas, K. Abul and Lechtman H. Andrew (2003.) *Cellular and Molecular Immunology*. V Edition. Saunders Publication.

PRACTICAL VI

IMMUNOLOGY

PRACTICAL

(CREDITS 2)

1. Demonstration of lymphoid organs
2. Histological study of spleen, thymus and lymph nodes through slides/ photographs
3. Preparation of stained blood film to study various types of blood cells.
4. Ouchterlony's double immuno-diffusion method.
5. ABO blood group determination.
6. ABO blood group typing.
7. Cell counting and viability test from splenocytes of farm bred animals/cell lines.
8. Demonstration of ELISA.
9. Demonstration Immuno-electrophoresis.

10. Demonstration of Blotting technique.

11. Demonstration of PCR.

12. Repitation

DMF30405/DMF30408

**SKILL ENHANCEMENT COURSE
MEDICAL DIAGNOSTICS**

Theory

Credits -2(30 hrs)

Course outcome

After completion of the course the student is able to:

CO1. Write down the characteristics of medical diagnostics

CO2. Write down in depth medical diagnostics

CO3. Specify the details of non infectious diseases

Unit I

10 hrs

Introduction to Medical Diagnostics and its Importance

Diagnostics Methods Used for Analysis of Blood

Blood composition, Preparation of blood smear and Differential Leucocyte Count (D.L.C) using Leishman's stain, Platelet count using haemocytometer, Erythrocyte Sedimentary Rate (E.S.R), Packed Cell Volume (P.C.V.).

Unit II

10 hrs

Non-infectious Diseases

Causes, types, symptoms, complications, diagnosis and prevention of Diabetes (Type I and Type II), Hypertension (Primary and secondary), Testing of blood glucose using Glucometer/Kit.

Infectious Diseases

Causes, types, symptoms, diagnosis and prevention of Tuberculosis and Hepatitis.

Unit III:

10 hrs

Diagnostic Methods Used for Urine Analysis

Urine Analysis: Physical characteristics; Abnormal constituents.

Tumours

Types (Benign/Malignant), Detection and metastasis; Medical imaging: X-Ray of Bone fracture, PET, MRI and CT Scan (using photographs).

Reference:

1. Asis. Das. (2012). Medical Physiology, Books & Allied Pvt. Ltd
2. Cheesbrough M., A Laboratory Manual for Rural Tropical Hospitals, A Basis for Training Courses.

3. Godkar P.B. and Godkar D.P. Textbook of Medical Laboratory Technology, II Edition, Bhalani Publishing House.
4. Guyton A.C. and Hall J.E. Textbook of Medical Physiology, Saunders
5. Park, K. (2007), and Social Medicine, B.B. Publishers.
6. Prakash, G. (2012), Lab Manual on Blood Analysis and Medical Diagnostics, S.Chand and Co.

DMF30605/DMF30608

SKILL ENHANCEMENT COURSE AQUARIUM FISH KEEPING

Theory

Credits -2(30 hrs)

Unit1: Introduction to Aquarium Fish Keeping

10 hrs

The potential scope of Aquarium Fish Industry as a Cottage Industry, Exotic and Endemic species of Aquarium Fishes

Unit 2: Biology of Aquarium Fishes

10 hrs

Common characters and sexual dimorphism of Fresh water and Marine Aquarium fishes such as Guppy, Molly, Sword tail, Gold fish, Angel fish, Blue morph, Anemone fish and Butterfly fish

Unit 3: Food and feeding of Aquarium fishes

10 hrs

Use of live fish feed organisms. Preparation and composition of formulated fish feeds

Fish Transportation

Live fish transport - Fish handling, packing and forwarding techniques.

Maintenance of Aquarium

General Aquarium maintenance – budget for setting up an Aquarium Fish Farm as a Cottage Industry

Reference:

1. E.P. Odum (Revised edition) Fundamental of Ecology, W. B. Saunders Co.,
2. P.D. Sharma (1991) Ecology and Environment – Rastogi Puplium
3. H.R. Singh (1989) An Introduction Animal Ecology and Envirmental Biology, Shobanlal Magin chand & Co.,
4. S.Anderson (1991) Managing on Wildlife, Resources, Prentive Hall, Englewood Chit, New Jermamy.
5. Nair, S.M. (1992) Endangered animals of India and their conservation M.B.J Delhi.
6. H.D. Kumar – Modern concepts of Ecology, Vikas Publishing House.
7. Elton, Plant ecology.

8. WWF manuals.
 9. Prater S H, The book of Indian Animals. 1998.
- *****

**Pattern of Question Paper
Semester I to VI**

Time : 3 Hrs

Max Marks: 70

I. Answer all the questions

5 X 1 = 5

- 1 -----
- 2-----
- 3-----
- 4-----
- 5-----

II. Answer any five questions

5 X 3 = 15

- 6-----
- 7-----
- 8-----
- 9-----
- 10-----
- 11-----

III. Answer any four questions

4 X 5 = 20

- 12 -----
- 13-----
- 14-----
- 15-----
- 16-----

IV. Answer any three questions

3 X 10 = 30

- 17-----
- 18-----
- 19-----
- 20-----

(Note- 10 Marks may be divided in to 6+4 or 5+5)

B.Sc, Examinations (SEC)

Zoology Question Paper

Duration: 02 Hours

Maximum Marks: 35

I. Answer all questions

1X 3 = 03

- 1.
- 2.
- 3.

II. Write short note on any Four of the following

3X 4 = 12

- 4.
- 5.
- 6.
- 7.
- 8.

III. Answer any Two of the following

5X 2 = 10

- 9.
- 10.
- 11.

IV. Write an essay on any one of the following

10X1 = 10

- 12.
- 13.



**JSS COLLEGE OF ARTS COMMERCE
& SCIENCE**

(Autonomous)

B.N Road, Mysuru – 25

**DEPARTMENT OF COMMERCE
AND MANAGEMENT**

Syllabus

CHOICE BASED CREDIT SYSTEM

BBA Programme

2017-18

Scheme of Study for BBA under CBCS Scheme implemented from 2017-18

YEAR	SEM	COURSE CODE	TITLE OF THE PAPER	NO. OF CREDITS			LECTURE/PRACTICAL HOURS /WEEK			TOTAL TEACHING HOURS			
				L (Hrs)	T (Hrs)	PRA (Hrs)	L (Hrs)	T (Hrs)	PRA (Hrs)	L (Hrs)	T (Hrs)	PRA (Hrs)	
I	I	BDA21011	ENTREPRENEURSHIP DEVELOPMENT	3	1	0	3	1	0	48	16	0	
		BDA22011	FINANCIAL ACCOUNTING AND ANALYSIS	4	1	1	4	1	2	64	16	32	
		BDA23011	BUSINESS DECISIONS	5	1	0	5	1	0	80	16	0	
	II	BDB23011	ORGANISATIONAL BEHAVIOUR	3	1	0	3	1	0	48	16	0	
		BDB21011	STATISTICS FOR BUSINESS	5	1	0	5	1	0	80	16	0	
		BDB22011	COST AND MANAGEMENT ACCOUNTING	5	1	0	5	1	0	80	16	0	
II	III	BDC21011	INCOME TAX	4	1	1	4	1	2	64	16	32	
		BDC22011	CORPORATE FINANCE	5	1	0	5	1	0	80	16	0	
	IV	BDD22001	FINANCIAL MARKETS AND INSTITUTIONS	5	1	0	5	1	0	80	16	0	
		BDD23001	QUANTITATIVE TECHNIQUES	5	1	0	5	1	0	80	16	0	
		BDD21001	COMPUTER APPLICATIONS IN BUSINESS	2	0	2	2	0	4	32	0	64	
III	V	BDE21001	INVESTMENT ANALYSIS AND PORTFOLIE MANAGEMENT	5	1	0	5	1	0	80	16	0	
		BDE22001	GST AND CUSTOMS DUTY	5	1	0	5	1	0	80	16	0	
			ELECTIVE -1 (ANY ONE FROM THE LIST DSE GROUP)	LIST GIVEN BELOW									
			ELECTIVE -2 (ANY ONE FROM THE LIST DSE GROUP)	LIST GIVEN BELOW									
	VI	BDF21001	CORPORATE RESTRUCTURING	5	1	0	5	1	0	80	16	0	
		BDF22001	INTERNATIONAL FINANCIAL INSTITUTIONS	5	1	0	5	1	0	80	16	0	
			ELECTIVE -3 (ANY ONE FROM THE LIST DSE GROUP)	LIST GIVEN BELOW									
			ELECTIVE - 4 (ANY ONE FROM THE LIST DSE GROUP)	LIST GIVEN BELOW									

LIST OF COURSES IN DSE GROUP

YEAR	SEM	COURSE CODE	TITLE OF THE PAPER	NO. OF CREDITS			LECTURE/PRACTICAL HOURS /WEEK			TOTAL TEACHING HOURS		
				L (Hrs)	T (Hrs)	PRA (Hrs)	L (Hrs)	T (Hrs)	PRA (Hrs)	L (Hrs)	T (Hrs)	PRA (Hrs)
III	V	BDE23001	INVESTMENT BANKING AND FINANCIAL SERVICES	5	1	0	5	1	0	80	16	0
		BDE24001	CORPORATE ANALYSIS AND VALUATION	5	1	0	5	1	0	80	16	0
		BDE25001	RESEARCH METHODOLOGY	5	1	0	5	1	0	80	16	0
		BDE26001	INTERNATIONAL TRADE BLOCKS & MULTILATERAL AGENCIES	5	1	0	5	1	0	80	16	0
		BDE27001	HUMAN RESOURCE MANAGEMENT	5	1	0	5	1	0	80	16	0

VI	BDF23001	STRATEGIC CORPORATE FINANCE	5	1	0	5	1	0	80	16	0
	BDF24001	MANAGEMENT OF FINANCIAL INSTITUTIONS	5	1	0	5	1	0	80	16	0
	BDF25001	FINANCIAL DERIVATIVES	5	1	0	5	1	0	80	16	0
	BDF26001	RESEARCH SOFTWARE PACKAGE	2	0	2	2	0	4	32	0	64
	BDF27001	COMPUTERISED ACCOUNTING SYSTEM	4	0	2	4	0	4	64	0	64
	BDF28001	PROJECT WORK	1	1	4	1	1	8	16	16	128

Proposed Scheme of Assessment for BBA under CBCS Scheme to be implemented from 2017-18

Year	SEM	COURSE CODE	TITLE OF THE PAPER	CONTINUOUS ASSESSMENT					MAX MARKS			PERCENTAGE			DURATION OF EXAM		
				C1		C2		C3	TH	PR	IA	TH	PR	IA	TH	PR	
				T	SDR	T	VI VA										
I	I	BDA21011	ENTREPRENEURSHIP DEVELOPMENT	10	05	10	05	70	70	-	30	70	-	30	3h	-	
		BDA22011	FINANCIAL ACCOUNTING AND ANALYSIS	10	05	10	05	70	70	70	30	50	20	30	3h	2h	
		BDA23011	BUSINESS DECISIONS	10	05	10	05	70	70	-	30	70	-	30	3h	-	
	II	BDB23011	ORGANISATIONAL BEHAVIOUR	10	05	10	05	70	70	-	30	70	-	30	3h	-	
		BDB21011	STATISTICS FOR BUSINESS	10	05	10	05	70	70	-	30	70	-	30	3h	-	
II	III	BDC21011	INCOME TAX	10	05	10	05	70	70	70	30	50	20	30	3h	2h	
		BDC22011	CORPORATE FINANCE	10	05	10	05	70	70	-	30	70	-	30	3h	-	
	IV	BDD22001	FINANCIAL MARKETS AND INSTITUTIONS	10	05	10	05	70	70	-	30	70	-	30	3h	-	
		BDD23001	QUANTITATIVE TECHNIQUES	10	05	10	05	70	70	-	30	70	-	30	3h	-	
		BDD21001	COMPUTER APPLICATIONS IN BUSINESS	10	05	10	05	70	50	70	30	35	35	30	3h	2h	
III	V	BDE21001	INVESTMENT ANALYSIS AND PORTFOLIO MANAGEMENT	10	05	10	05	70	70	-	30	70	-	30	3h	-	
		BDE22001	GST AND CUSTOMS DUTY	10	05	10	05	70	70	-	30	70	-	30	3h	-	
			ELECTIVE -1 (ANY ONE FROM THE LIST DSE GROUP)	LIST GIVEN BELOW													
			ELECTIVE -2 (ANY ONE FROM THE LIST DSE GROUP)	LIST GIVEN BELOW													
	VI	BDF21001	CORPORATE RESTRUCTURING	10	05	10	05	70	70	-	30	70	-	30	3h	-	
		BDF22001	INTERNATIONAL FINANCIAL INSTITUTIONS	10	05	10	05	70	70	-	30	70	-	30	3h	-	
			ELECTIVE -3 (ANY ONE FROM THE LIST DSE GROUP)	LIST GIVEN BELOW													
			ELECTIVE -4 (ANY ONE FROM THE LIST DSE GROUP)	LIST GIVEN BELOW													

LIST OF COURSES IN DSE GROUP

Year	SEM	COURSE CODE	TITLE OF THE PAPER	CONTINUOUS ASSESSMENT					MAX MARKS			PERCENTAGE			DURATION OF EXAM	
				C1		C2		C3	TH	PR	IA	TH	PR	IA	TH	PR
				T	SDR	T	VI VA									
III	V	BDE23001	INVESTMENT BANKING AND FINANCIAL SERVICES	10	05	10	05	70	70	-	30	70	-	30	3h	-
		BDE24001	CORPORATE ANALYSIS AND VALUATION	10	05	10	05	70	70	-	30	70	-	30	3h	-
		BDE25001	RESEARCH METHODOLOGY	10	05	10	05	70	70	-	30	70	-	30	3h	-
		BDE26001	INTERNATIONAL TRADE BLOCKS & MULTILATERAL AGENCIES	10	05	10	05	70	70	-	30	70	-	30	3h	-
		BDE27001	HUMAN RESOURCE MANAGEMENT	10	05	10	05	70	70	-	30	70	-	30	3h	-
	VI	BDF23001	STRATEGIC CORPORATE FINANCE	10	05	10	05	70	70	-	30	70	-	30	3h	-
		BDF24001	MANAGEMENT OF FINANCIAL INSTITUTIONS	10	05	10	05	70	70	-	30	70	-	30	3h	-
		BDF25001	FINANCIAL DERIVATIVES	10	05	10	05	70	70	-	30	70	-	30	3h	-
		BDF26001	RESEARCH SOFTWARE PACKAGE	10	05	10	05	70	50	70	30	35	35	30	2h	2h
		BDF27001	COMPUTERISED ACCOUNTING SYSTEM	10	05	10	05	70	70	70	30	50	20	30	3h	3h
		BDF28001	PROJECT WORK	-	-	-	-	-	-	70	30	-	70	30	-	-

SCHEME OF VALUATION FOR PRACTICAL EXAMINATION

- A candidate appearing for the first time should submit a duly signed and certified practical record
- Practical record has to be valued for **FIVE marks** by examiners at the time of examination
- **FIVE marks** for conducting viva-voce.
- Computer Lab exams are conducted for 70 marks for each candidates

EVALUATION OF PROJECT WORK:

SI No	Component	Marks
1	Proposal of Project Work	15
2	PPT Presentation	15
TOTAL		30

Sl No	Component	Marks
1	Evaluation of Project by Internal and External Examinors	50
2	Viva Voce	20
	TOTAL	70

DEPARTMENT OF COMMERCE

Discipline Specific Course

Sl. No	Sem	Course Code	Title of the paper	Total credits		
				L	T	P
1	I	BDA22011	FINANCIAL ACCOUNTING AND ANALYSIS	4	1	1
2		BDA23011	BUSINESS DECISIONS	5	1	0
3	II	BDB21011	STATISTICS FOR BUSINESS	5	1	0
4		BDB22011	COST AND MANAGEMENT ACCOUNTING	5	1	0
5	III	BDC21011	INCOME TAX	4	1	1
6		BDC22011	CORPORATE FINANCE	5	1	0
7	IV	BDC22001	FINANCIAL MARKETS AND INSTITUTIONS	5	1	0
8		BDC23001	QUANTITATIVE TECHNIQUES	5	1	0
9	V	BDE21001	INVESTMENT ANALYSIS AND PORTFOLIO MANAGEMENT	5	1	0
10		BDE22001	GST AND CUSTOMS DUTY	5	1	0
11	VI	BDF21001	CORPORATE RESTRUCTURING	5	1	0
12		BDF22001	INTERNATIONAL FINANCIAL INSTITUTIONS	5	1	0

Generic Elective

Sl. No	Sem	Course Code	Title of the paper	Total credits		
				L	T	P
1	I	BDA21011	ENTREPRENEURSHIP DEVELOPMENT	3	1	0
2	II	BDB23011	ORGANISATIONAL BEHAVIOUR	3	1	0

Skill Enhancement Course

Sl. No	Sem	Course Code	Title of the paper	Total credits		
				L	T	P
1	IV	BDD21011	COMPUTER APPLICATION IN BUSINESS	2	0	2

Discipline Specific Elective

Sl. No	Sem	Course Code	Title of the paper	Total credits		
				L	T	P
1	V	BDE23001	INVESTMENT BANKING AND FINANCIAL	5	1	0

			SERVICES			
2		BDE24001	CORPORATE ANALYSIS AND VALUATION	5	1	0
3		BDE25001	RESEARCH METHODOLOGY	5	1	0
4		BDE26001	INTERNATIONAL TRADE BLOCKS & MULTILATERAL AGENCIES	5	1	0
5		BDE27001	HUMAN RESOURCE MANAGEMENT	5	1	0
6	VI	BDF23001	STRATEGIC CORPORATE FINANCE	5	1	0
7		BDF24001	MANAGEMENT OF FINANCIAL INSTITUTIONS	5	1	0
8		BDF25001	FINANCIAL DERIVATIVES	5	1	0
9		BDF26001	RESEARCH SOFTWARE PACKAGE	2	0	2
10		BDF27001	COMPUTERISED ACCOUNTING SYSTEM	4	0	2
11		BDF28001	PROJECT WORK	1	1	4

Programme code: BBA11

Programme Outcome:

This program could provide well trained dynamic personnel and professionals for

PO1: Industries and Multinational companies

PO2: Banking Sectors and Insurance Companies

PO3: Financing and Leasing Companies

PO4: Transport Agencies and Warehousing

PO5: Stock Markets and Foreign Trade,

This program could provide well trained professionals to practice and work as

PO6: Chartered accountants, advocates, cost accountants and company secretaries

PO7: Financial Analysts, Tax consultants, Tax Practitioners and Investment consultants

PO8: Financial and management accountants

PO9: Marketing Manager, Store manager, Purchase Manager and Sales Manager

PO10: Human Resources Manager, Counsellor

PO11: Retail Manager, Middle men and Customer relation manager

PO12: Decision Maker

PO13: Stock broker

PO14: Official receiver and Liquidator,

PO15: Market researcher, supply chain manger and Franchisee

PO16: Administrator of the different types of Business and Non-business organizations

Programme Specific Outcomes

The students at the end of the BBA programme can become a

PSO1: Business Entrepreneur

PSO2: Business Administrator

PSO3: Financial, Cost and Management Accountant

PSO4: Business Researcher

PSO5: Bank Manager

PSO6: Personal Secretary

PSO7: Project Manager

PSO8: Legal adviser

PSO9: Stock Broker

BDA22011

SEMESTER I

FINANCIAL ACCOUNTING AND ANALYSIS – DSC 1

(Credits: Lecture – 04, Tutorial – 01, Practical – 01)

Lectures: 64 Hrs

Tutorials: 16 Hrs

Practical: 32 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn the characteristics of financial accountant and able to work as financial accountant

CO2: Learn the classification and characteristics and apply the principles of accounting in preparation of financial statements

CO3: Understand in details with application and generate financial statement using accounting software

CO4: Understand in details with examples become a financial analyst

CO5: Identify the characteristics of financial analyst and able to become financial analyst

Unit 1: Introduction to Financial Accounting:

Accounting as an Information System, Importance and Scope, Limitations; Users of accounting information, Concepts, Principles and Conventions. Generally Accepted Accounting Principles; The Accounting Equation; Nature of Accounts, Types of books (Primary and Secondary) and Rules of Debit and Credit; Recording Transactions in Journal; Preparation of Ledger Accounts; Opening and Closing Entries; Preparation of Trial Balance.

Unit 2: Preparation of Financial Statements:

Trading Account, Profit & Loss Account and Balance Sheet, Adjustment Entries, Understanding contents of financial statements of a joint stock company as per the Companies Act 2013; Understanding the contents of annual report of a company.

Unit 3: Indian Accounting Standards (Ind-AS):

Concept, benefits, procedure for issuing Ind-AS in India, salient features of Ind-AS issued by ICAI; International Financial Reporting Standards(IFRS): Main features, uses and objectives of IFRS, IFRS issued by IASB and concept of harmonization and convergence, obstacle in harmonization and convergence, suggestions for increased convergence and harmonization.

Unit 4: Computerised Accounting Systems:26 Practical Lab

Computerised Accounting Systems: Computerized Accounts by using any popular accounting software: Creating a Company; Configure and Features settings; Creating Accounting Ledgers and Groups; Creating Stock Items and Groups; Vouchers Entry; Generating Reports – Cash Book, Ledger Accounts, Trial Balance, Profit and Loss Account, Balance Sheet, Funds Flow Statement, Cash Flow Statement. Selecting and shutting a Company; Backup and Restore data of a Company.

Unit 5: Financial Statement Analysis:

Objective of financial statement analysis, sources of information; Techniques of financial statement analysis: Horizontal analysis, Vertical analysis and Ratio Analysis; Financial Ratios: Meaning and Usefulness of Financial Ratios. Analysis of ratios from the perspective

of Stakeholders like Investors, Lenders, and Short-term Creditors. Liquidity Ratios, Solvency Ratios, Profitability Ratios, and Turnover Ratios; Limitation of ratios.

Reference:

1. Monga, J.R., Financial Accounting: Concepts and Applications, Mayur Paperbacks
2. Tulsian, P.C., Financial Accounting, Pearson
3. Maheshwari, S.N. & Maheshwari, S.K. , Financial Accounting for B. Com., CA, CS, & ICWA (Foundation) Courses, Vikas Publishing House Pvt. Ltd.
4. Ghosh, T.P., Financial Accounting for Managers, Taxmann Allied Services (P) Ltd.
5. Balwani, Nitin, Accounting and Finance for Managers

6. Gupta, Ambrish: Financial Accounting for Management
7. Bhattacharyya, Asish K., Financial Accounting for Business Managers
8. Jain, S.P. & Narang, K.L., Advanced Accountancy.
8. Charles T. Horngren, Gart L. Sundem, John A. Elliot and Donna R. Philbrick, Introduction to Financial Accounting, Pearson.

Note: Latest edition of the text books should be used.

BDA23011

SEMESTER I
BUSINESS DECISIONS – DSC 2

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Identify in detail the cost and income effect on business

CO2: Learn in depth the decision making areas and able to make appropriate business decisions

CO3: Understand in details with examples Analyse the behaviour of consumer

CO4: Learn in detail the characteristics of forecast and able to forecast the demand for products

CO5: Understand and analyse business problems and able to solve them

Unit 1: Market Dynamics:

Individual demand, market demand, individual supply, market supply, market equilibrium; Elasticity of demand and supply: Price elasticity of demand, income elasticity of demand, cross price elasticity of demand, elasticity of supply; Theory of consumer behaviour: cardinal utility theory, ordinal utility theory(indifference curves, budget line, consumer choice, price effect, substitution effect, income effect for normal, inferior and Giffen goods), revealed preference theory.

Unit 2: Producer and optimal production choice:

Optimizing behaviour in short run (geometry of product curves, law of diminishing margin productivity, three stages of production), optimizing behaviour in long run (isoquants, iso-cost line, optimal combination of resources) Costs and scale: traditional theory of cost (short run and long run, geometry of cost curves, envelope curves), modern theory of cost (short run and long run), economies of scale, economies of scope.

Unit 3: Theory of firm and market organization:

Perfect competition (basic features, short run equilibrium of firm/industry, long run equilibrium of firm/industry, effect of changes in demand, cost and imposition of taxes) ; monopoly (basic features, short run equilibrium, long run equilibrium, effect of changes in demand, cost and imposition of taxes, comparison with perfect competition, welfare cost of monopoly), price discrimination, multi plant monopoly; monopolistic competition (basic features, demand and cost, short run equilibrium, long run equilibrium, excess capacity) ; oligopoly (Cournot's model, kinked demand curve model, dominant price leadership model, prisoner's dilemma

Unit 4: Factor Market:

Demand for a factor by a firm under marginal productivity theory (perfect competition in the product market, monopoly in the product market), market demand for a factor, supply of labour, market supply of labour, factor market equilibrium.

Reference:

1. Dominick Salvatore (2009). Principles of Microeconomics(5th ed.) Oxford University Press
2. Lipsey and Chrystal. (2008). Economics.(11th ed.) Oxford University Press
3. Koutosyannis (1979). Modern Micro Economics. Palgrave Macmillan
4. Pindyck, Rubinfeld and Mehta. (2009). Micro Economics. (7th ed.). Pearson.

Note: Latest edition of the text books should be used.

BDA21011

SEMESTER I

ENTREPRENEURSHIP DEVELOPMNET - GE 1

(Credits: Lecture – 03, Tutorial – 01, Practical – 0)

Lectures: 48 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn in depth qualities of an entrepreneur and able to become an entrepreneur

CO2: Write down the details of financial schemes offered by banks and government agencies and able to access them easily

CO3: Learn the details of mobilisation of resources

CO4: Learn in depth the characteristics of customer and able to identify the customer

Unit 1: Entrepreneurial Development Perspective:

Concept of entrepreneurship development and their dynamics. Importance of skill, knowledge and motivation in ED. Entrepreneurial Competition, generation of business, ideas and final selection of an activity. Market survey report and business plan preparation. Pooling of resources, forms enterprise ownership and their details.

Unit 2: Enterprise Management

Logistics and launching formalities, probable pitfalls, managing money, men, machinery, material and marketing. Support organization, entrepreneurial growth, following the law of the land and social obligation. Managing organisation for innovation and creativity. Importance of leadership, business ethics and business skills on good team building

Unit 3: Running A Family Business :

Concept, structure and kinds of family firms. Understanding its reputation and brand. Enhancing the knowledge and skill. Managing family and shareholders relationship. Managing leadership succession and understanding the group dynamics, encouraging family women into business. Identifying the changed customer needs and encouraging growth and change in the family business.

Unit 4: Social Entrepreneurship:

Introduction, Role and Characteristics of Social Entrepreneurs, Starting of a Non-profits Organization innovatively through local resources in a social context, sustainability, Business Strategies and Scaling up.

Unit 5: Role Of Government And Financial Institutions

Role of Central and State Government in promoting entrepreneurship. Types of schemes, loans, incentives, grants and subsidies. Different types of financial institutions, role of commercial banks, types of loans for MSMEs schemes, appraisal, sanctions, repayment.

Reference:

1. Entrepreneurship – Tata MC Graw hill
2. Entrepreneurship and Small Business – New Jersey: Palgrave
3. Creativity and Entrepreneurship – Jhon Kao
4. Corporate Creativity Tata MC Graw hill
5. Innovative Entrepreneurship – Practice and Principles – Drucker P.F.

Note: Latest edition of the text books should be used.

BDB23011

SEMESTER II
ORGANISATIONAL BEHAVIOUR - GE 2

(Credits: Lecture – 03, Tutorial – 01, Practical – 0)

Lectures: 48 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Understand in detail behaviour of employees and able to manage them efficiently

CO2: Identify in details employees performance and able to motivate for effective performance

CO3: Learn in depth and analyse the behaviour of employees

CO4: Understand in details key positions in an organisation and able to occupy them

CO5: Learn in details with examples frame policies and strategies in organisation

Unit 1: Introduction to Organisational Behaviour:

Organisation- Concept, features and types. Organisational Behaviour – concept, meaning, scope, characteristics and role. Evolution, challenges and opportunities for O.B.

Unit 2: Personality:

Personality – Meaning, characteristics and determinants. Theories – Psychoanalytical Theory. Erikson stages, CheisArgyeis’s immateriality – Maturity Theory, Traits Theory and Self theory. Personality traits.

Unit 3: Perception, Attitude and Learning:

Perception – concept, nature, process and factors influencing perceptual set.

Attitudes – Meaning, definition, nature, components and sources.

Learning – concept, nature, theories of learning, principles and determinants of learning

Unit 4: Groups and Teams:

Groups – Definitions, types, reasons for group formation. Groups Dynamics – Definition and features. Teams – Meaning , Groups v/s Teams, features, importance and types.

Reference:

1. Koontz & Heinz Wehrich: Essential of management McGraw Hill (1999)
2. Kaul, Vijay kumar, Management- Text & Cases, Vikas Publishing, New Delhi, 2015
3. Stoner & Wankel: Management
4. Stephen P. Robbins and Mary Coulter: Management, Pearson
5. Y.K. Bhushan: Fundamentals of Business Organisation & Management X Edition
6. Richard L. Daft, Principles Of Management, Cengage Learning, India
7. Robbins Stephen P. : Organisational Behaviour, Pearson Education, 12th Edition

Note: Latest edition of the text books should be used.

BDB21011

SEMESTER II
STATISTICS FOR BUSINESS - DSC 3

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn in details with examples Measures of Central tendency

CO2: Understand the classification and characteristics of Measures of dispersion

CO3: Learn in detail the correlation and determine the relation between two variables

CO4: Understand in depth regression and able to find unknown variable value based on known variable value

CO5: Learn the characteristics and components of time series analysis and able fit a trend line

Unit 1: Measures of Central Value:

Meaning, Need for measuring central value. Characteristics of an ideal measure of central value. Types of averages - mean, median, mode, harmonic mean and geometric mean. Merits, Limitations and Suitability of averages. Relationship between averages. Measures of Dispersion: Meaning and Significance. Absolute and Relative measures of dispersion - Range, Quartile Deviation, Mean Deviation, Standard Deviation, Moments, Skewness.

Unit 2: Correlation Analysis:

Meaning and significance. Types of correlation. Methods of studying simple correlation - Karl Pearson's coefficient of correlation, Spearman's Rank correlation coefficient.

Regression Analysis: Meaning and significance, Regression vs. Correlation. Linear Regression, Regression lines (X on Y and Y on X) and Standard error of estimate.

Unit 3: Analysis of Time Series and Probability

Analysis of Time Series: Meaning and significance. Utility, Components of time series, Models (Additive and Multiplicative), Measurement of trend: Method of least squares, parabolic trend and logarithmic trend.

Probability: Meaning and need. Theorems of addition and multiplication. Conditional probability. Bayes' theorem, Random Variable- discrete and continuous. Probability Distribution: Meaning, characteristics (Expectation and variance) of Binomial, Poisson, and Normal distribution. s

Unit 4: Introduction to testing of Hypothesis:

Concept; Level of Significance; Process of testing; Test of hypothesis concerning Mean; Test of hypothesis concerning Proportion. Z -test, t – test for single mean and difference of means and ANNOVA – one way and two way.

Reference:

1. S.P. Gupta (S.P.) : Statistical Methods, Sultan Chand & Sons, 34th Edition

2. Richard Levin & David Rubin : Statistics for management, Prentice Hall
3. M. Wilson – Business Statistics
4. Sancheti&Kapoor – Business Statistics
5. C.B. Gupta - Business Statistics
- 6.Tulsian - Business Statistics

Note: Latest edition of the text books should be used.

BDB22011

SEMESTER II

COST AND MANAGEMENT ACCOUNTING - DSC 4

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO 1: Understand in depth budgetary control system and able to prepare different types of budgets

CO 2: Understand in detail the techniques of costing and able to become cost accountant

CO 3: Learn in detail the classification of cost and able to prepare cost sheet, tender and quotation

CO 4: Learn in depth variance analysis and able to determine Material, Labour and overhead variance

CO5: Learn in detail marginal costing techniques and able to determine BEP, desired profit

Unit 1: Cost concepts:

Meaning, Scope, Objectives, and Importance of Cost Accounting, Cost, Costing, Cost Control, and Cost Reduction. Elements of Cost, Components of total Cost, Cost sheet.

Classification of costs: Fixed, Variable, Semi-variable, and Step costs; Product, and Period costs; Direct, and Indirect costs; Relevant, and Irrelevant costs; Shut-down, and Sunk costs; Controllable, and Uncontrollable costs; Avoidable, and Unavoidable costs; Imputed / Hypothetical costs; Out-of-pocket costs; Opportunity costs; Expired, and Unexpired costs; Conversion cost. Cost Ascertainment: Cost Unit, Cost Center, Profit Center, Cost Allocation and Cost apportionment; Cost Reduction and Cost Control.

Unit 2: Cost-Volume-Profit Analysis:

Contribution, PV Ratio, Breakeven-point, Margin of safety, cost break-even-point, cash break-even-point, Composite break-even-point, Key Factor, Break-even Analysis. Relevant Costs and Decision Making: Pricing, Product Profitability, Make or Buy, Exploring new markets, Shut down or continue.

Process costing: Process losses and wastage, Abnormal effectives.

Unit 3: Budgets and Budgetary Control:

Meaning, Types of Budgets (sales, production, purchase raw material consumption, cash budget). Steps in Budgetary Control, Fixed and Flexible Budgeting, Responsibility Accounting.

Unit 4: Standard Costing and Variance Analysis:

Material, Labour & Overhead variances.

Reference:

1. Maheshwari, S.N., and Mittal, S.N. Cost Accounting: Theory and Problems, Shree Mahavir Book Depot (Publishers), Delhi.

2. M.N.Arora, Management Accounting, Theory, Problems and Solutions, Himalaya Publishing House
3. Horngren, C.T., Foster, G, and Datar, S.M., Cost Accounting: A Managerial Emphasis, Prentice Hall of India Pvt. Ltd., New Delhi.
4. Henke, E.O., and Spoede, C.W., Cost Accounting: Managerial Use of Accounting Data,PWS-KENT Publishing Company, Boston.

Note: Latest edition of the text books should be used.

BDC21011

SEMESTER III
INCOME TAX - DSC 5

(Credits: Lecture – 04, Tutorial – 01, Practical – 01)

Lectures: 64 Hrs

Tutorials: 16 Hrs

Practical: 32 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn in depth the procedure of online filing and able to file online income tax returns

CO2: Understand in depth Income Tax Act of 1961 and can become tax consultant practitioner

CO3: Deliberate in detail with examples and appear before IT tribunal on behalf of clients

CO4: Learn in detail different sections Under IT Act to reduce tax liability

CO5: Identify the different heads of income and able to compute tax liability

Unit 1: Basic concepts:

Income, agricultural income, person, assessee, assessment year, previous year, gross total income, total income, maximum marginal rate of tax. Residential status of persons and its effect on tax incidence .Exempted income under section 10(in relation to individuals).

Unit 2: Computation of income under the heads:

Salaries, Income from house property

Unit 3: Computation of income under the heads:

Profits and gains of business & profession, Capital gain, Income from other sources.

Unit 4: Computation Total Income and Tax Liability

Deductions under Chapter VI-A- Computation of total income and tax liability of individuals. Preparation of return of income manually and through software. Provision & Procedures of Compulsory online filing of returns for specified assesses.

Reference:

1. Dr.VinodkSinghania and Dr.MonicaSinghania; Students guide to income tax, Taxman Publications.
2. GirishAhuja and Ravi Gupta; Systematic Approach to Income Tax: Bharat Law House.
3. Mahesh Chandra, D.C Shukla; Income Tax Law and Practice: Pragati Publications.
4. S.P Goyal; Direct tax planning: SahityaBhawan
5. Finance Act for relevant Assessment Year
6. CBDT Circulars
7. Latest court judgements

Note: Latest edition of the text books should be used.

BDC22011

SEMESTER III
CORPORATE FINANCE - DSC 6

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn in details financial management and become financial adviser

CO2: Understand and analyse the interrelationship between finance and corporate

CO3: Learn in depth working capital management and able to estimate the working capital requirements

CO4: Understand in detail the various approaches to capital structure theories and able to design capital structure

CO5: Understand in depth long term investment decisions

Unit 1: Nature of Financial Management:

Finance and related disciplines; Scope of Financial Management; Profit Maximization, Wealth Maximization - Traditional and Modern Approach; Functions of finance – Finance Decision, Investment Decision, Dividend Decision; Objectives of Financial Management; Organisation of finance function; Concept of Time Value of Money: present value, future value, and annuity.

Unit 2: Long-term investment decisions:

Capital Budgeting - Principles and Techniques; Nature and meaning of capital budgeting; Estimation of relevant cash flows and terminal value; Evaluation techniques - Accounting Rate of Return, Net Present Value, Internal Rate of Return, Net Terminal Value, Profitability Index Method. Concept and Measurement of Cost of Capital: Explicit and Implicit costs; Measurement of cost of capital; Cost of debt; Cost of perpetual debt; Cost of Equity Share; Cost of Preference Share; Cost of Retained Earning; Computation of over-all cost of capital based on Historical and Market weights.

Unit 3: Capital Structure:

Approaches to Capital Structure Theories - Net Income approach, Net Operating Income approach, Modigliani-Miller (MM) approach, Dividend Policy Decision - Dividend and Capital; Irrelevance of dividends: General, MM hypothesis.

Leverage Analysis: Operating and Financial Leverage; EBIT -EPS analysis; Combined leverage.

Unit 4: Working Capital Management:

Management of Cash - Preparation of Cash Budgets (Receipts and Payment Method only); Cash management technique (Lock box, concentration banking), Receivables Management – Objectives; Credit Policy, Cash Discount, Debtors Outstanding and Ageing Analysis; Costs - Collection Cost, Capital Cost, Default Cost, Delinquency Cost, Inventory Management (Briefly) - ABC Analysis; Minimum Level; Maximum Level; Reorder Level; Safety Stock; EOQ (Basic Model), Determination of Working Capital.

Reference:

1. M.Y. Khan & P.K. Jain , Financial Management, Tata McGraw Hill Pubilshng Co. Ltd.
2. Rustogi , Financial Management
3. I.M. Pandey , Financial Management
4. L.J. Gitman& C.J. Zutter, Managerial Finance.R.A. Brealey, S.C. Myers, F. Allen& P. Mohanty, Principles of Corporate Finance.
5. J.V. Horne & J.M. Wachowicz, Fundamentals of Financial Management.

Note: Latest edition of the text books should be used.

BDD22001

SEMESTER IV

FINANCIAL MARKETS AND INSTITUTIONS - DSC 7

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can;

CO1: Understand in details transaction in stock exchange and able to become inter-mediator

CO2: Learn in depth the classification and characteristics of financial markets and institutions and able to set up his own institution

CO3: Understand in detail characteristics of financial market and able to participate in financial market

CO4: Identify the government policies for investment and able apply them

Unit 1: Indian Financial System:

Indian Financial System – Meaning - Overview – needs – objectives, major reforms – issues in financial reforms –future agenda –role of RBI- Banking Companies Regulation Act 1949, services, Retail banking – Payment Banks. Corporate universal Banking system – CBS – RTG’S – MBFC – Internet banking.

Unit 2: Financial Markets in India:

Role and Importance of Financial Markets, Types of Financial Markets: Money Market; Capital Market; Factors affecting Financial Markets, Linkages Between Economy and Financial Markets, Integration of Indian Financial Markets with Global Financial Markets, Primary & secondary market, Currency Market, Debt Market- role and functions of these markets.

Unit 3: Primary Market:

Primary Market for Corporate Securities in India: Issue of Corporate Securities: Public Issue through Prospectus, Green shoe option, Offer for sale, Private Placement, Rights Issue, On-Line IPO, Book Building of Shares, Disinvestment of PSU, Employees Stock Options, Preferential Issue of Shares, Venture Capital, Private Equity Firms, Performance of Primary Market in India, Corporate Listings : Listing and Delisting of Corporate Stocks.

Unit 4: Secondary Market in India:

Introduction to Stock Markets, Regional and Modern Stock Exchanges, International Stock Exchanges, Raising of funds in International Markets: ADRs and GDRs, FCCB and Euro Issues; Indian Stock Indices and their construction, maintenance, adjustment for corporate actions (rights, bonus and stock split;) on index with numerical, free float v/s full float methodology, Classification of Securities to be included in the Index, Bulls and Bears in Stock Markets, Factors influencing the movement of stock markets, indicators of maturity of stock markets, Major Instruments traded in stock markets: Equity Shares, Debentures, Myths attached to Investing in Stock Markets.

Unit 5: Trading of Securities:

Trading of securities on a stock exchange; Selection of broker, capital and margin requirements of a broker, MTM and VAR Margins, kinds of brokers, opening of an account to trade in securities, DEMAT System, placing an order for purchase/sale of shares, margin trading and margin adjustment, contract note and settlement of contracts, Algorithmic trading, Demutualization of Securities, NSE and BSE: Sensex & Nifty -Settlement mechanism at BSE & NSE..

Unit 6: Money & Debt Markets:

Money Markets & Debt Markets in India: Money Market - Meaning, role and participants in money markets, Segments of money markets, Call Money Markets, Repos and reverse Repo concepts, Treasury Bill Markets, Market for Commercial Paper, Commercial Bills and Certificate of Deposit. Role of STCI and DFHI in money market, Secondary market for government securities. Over subscription and devolvement of Government Securities, Government securities issued by State Governments, Municipal Bonds.

Reference:

1. Saunders , Anthony & Cornett , Marcia Millon (2007). Financial Markets and Institutions (3rd ed.). Tata McGraw Hill
2. Khan, M Y. (2010). Financial Services (5th ed.). McGraw Hill Higher Education
3. Shahani, Rakesh(2011). Financial Markets in India : A Research Initiative. Anamica Publications
4. Goel, Sandeep. (2012). Financial services.PHI.
5. Gurusamy,S. (2010). Financial Services.TMH.

Note: Latest edition of the text books should be used

BDD23001

SEMESTER IV
QUANTITATIVE TECHNIQUES - DSC 8

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

After the completion this course the students are able to:

CO1: Understand in depth about OR techniques and solve the business problems

CO2: Learn in detail to convert the business problem into mathematical for optimum utilization of scarce resources

CO3: Understand the details of elementary transportation and able to solve business problems

CO4: Learn in depth about Network analysis and able to construct network diagram

CO5: Learn the details of Decision Theory

Unit 1: Linear Programming:

Formulation of L.P. Problems, Graphical Solutions, Two Variables, introduction of slack variables and additional variables – Duality concept, formulation of LPP Model using slack and additional variables.

Unit 2: Elementary Transportation:

Formulation of Transport Problem, Solution by N.W. Corner Rule, Least Cost method, Vogel's Approximation Method (VAM), Modified Distribution Method. (Special cases: Multiple Solutions, Maximization case, unbalanced case, prohibited routes) Elementary Assignment: Hungarian Method, (Special cases: Multiple Solutions, Maximization case, unbalanced case, Restrictions on assignment.)

Unit 3: Network Analysis:

Construction of the Network diagram, Critical Path- float and slack analysis (Total float, free float, independent float), PERT, Project Time Crashing.

Unit 4: Decision Theory:

Pay off Table, Opportunity Loss Table, Expected Monetary Value, Expected Opportunity Loss, Expected Value of Perfect Information and Sample Information Markov Chains: Predicting Future Market Shares, Equilibrium Conditions (Questions based on Markov analysis) Limiting probabilities, Chapman Kolmogrov equation.

Introduction to Game Theory: Pay off Matrix- Two person Zero-Sum game, pure strategy, Saddle point; Dominance Rule, Mixed strategy, Reduction of $m \times n$ game and solution of 2×2 , $2 \times s$, and $r \times 2$ cases by Graphical and Algebraic methods; Introduction to Simulation: Monte Carlo Simulation

Reference:

1. N. D. Vohra, Quantitative Management, Tata McGraw Hill
2. P. K. Gupta, Man Mohan, KantiSwarup, Operations Research, Sultan Chand
3. V. K. Kapoor, Operations Research, Sultan Chand & Sons
4. J. K. Sharma, Operations Research Theory & Applications, Macmillan India Limited.

Note: Latest edition of the text books should be used.

BDD21001

SEMESTER IV

COMPUTER APPLICATIONS IN BUSINESS – SEC - 2

(Credits: Lecture – 02, Tutorial – 0, Practical – 2)

Lectures: 32 Hrs

Practical: 64 Hrs

Course Outcome:

On successful completion of this course the students are able to:

CO1: Learn in depth and create business documents by using MS Word and Excel

CO2: Learn in details with application of PowerPoint and present of business information through PPTs

CO3: Write down the characteristics of BPOs and KPOs and able to work in BPOs and KPOs

CO4: Learn the classification and characteristics of chart and graphs using computers

CO5: Identify the characteristics of projects and develop projects using computer

Unit 1: Word Processing

3 Lectures, Practical Lab 6

Introduction to word Processing, Word processing concepts, Use of Templates, Working with word document: Creating and Editing Text, Find and replace text, Formatting, spell check, Autocorrect, Autotext; Bullets and numbering, Tabs, Paragraph Formatting, Indent, Page Formatting, Header and footer, Tables: Inserting/Drawing, filling and formatting a table; OLE concepts (Object Linking and Embedding) Inserting Pictures and Video; Mail Merge: including linking with Database; Printing documents

Creating Business Documents using the above facilities

Unit 2: Presentation Graphics

3 Lectures, Practical Lab 6

Presentation-Basic concepts, Creating/Editing slides, Formatting slides, Inserting drawings, Charts, Tables, Images, Symbols. Embedding media and animation. Preparing and presenting a slide show..

Creating Business Presentations using above facilities

Unit 3: Spreadsheets and its Business Applications

10 Lectures, Practical Lab 20

Spreadsheet concepts, Managing worksheets; Formatting, Entering data, Editing, and Printing a worksheet; Handling operators in formula, Project involving multiple spreadsheets, Organizing Charts and graphs

Generally used Spreadsheet functions: Mathematical, Statistical, Financial, Logical, Date and Time, Lookup and reference, Database, and Text functions.

Unit 4: Creating Business Spreadsheet

10 Lectures, Practical Lab 20

Creating spreadsheet in the area of: Loan and Lease statement; Ratio Analysis; Payroll Processing and statements; Capital Budgeting; Depreciation Accounting; Graphical representation of data; Frequency distribution and its statistical parameters; Correlation and Regression.

Note:

1. The General Purpose Software referred in this course will be notified by the University Departments every three years. If the specific features, referred in the detailed course

above, is not available in that software, to that extent it will be deemed to have been modified.

2. Teaching arrangement need to be made in the computer Lab

Reference:

The suggested readings and guidelines shall be notified by the university department at least once in three years based on the selected software

BDE21001

SEMESTER V

INVESTMENT ANALYSIS & PORTFOLIO MANAGEMENT – DSC 9

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn in depth various financial models to analyse risk and return

CO2: Understand in depth share valuation

CO3: Identify and use different models for portfolio analysis

CO4: Learn in depth capital asset pricing model

Unit 1: Basics of risk and return:

Concept of returns, application of standard deviation, coefficient of variation, beta, alpha. Bonds : present value of a bond, yield to maturity, yield to call, yield to put, systematic risk, price risk, interest rate risk, default risk. Yield curve and theories regarding shape of yield curve. Unsystematic risk and non-risk factors that influence yields. Duration and modified duration, immunization of a bond portfolio. Fundamental analysis: EIC framework; Economic analysis: Leading lagging & coincident macro-economic indicators, Expected direction of movement of stock prices with macroeconomic variables in the Indian context; Industry analysis: stages of life cycle, Porter's five forces model, SWOT analysis, financial analysis of an industry; Company analysis.

Unit 2: Share valuation:

Dividend discount models- no growth, constant growth, two stage growth model, multiple stages; Relative valuation models using P/E ratio, book value to market value. Technical analysis: meaning, assumptions, difference between technical and fundamental analysis; Price indicators- Dow theory, advances and declines, new highs and lows- circuit filters. Volume indicators- Dow Theory, small investor volumes. Other indicators- futures, institutional activity, Trends: resistance, support, consolidation, momentum- Charts: line chart, bar chart, candle chart, point & figure chart. Patterns: head & shoulders, triangle, rectangle, flag, cup & saucer, double topped, double bottomed, Indicators: moving averages. Efficient market hypothesis; Concept of efficiency: Random walk, Three forms of EMH and implications for investment decisions. (No numerical in EMH and technical analysis)

Unit 3: Portfolio analysis:

Portfolio risk and return, Markowitz portfolio model: risk and return for 2 and 3 asset portfolios, concept of efficient frontier & optimum portfolio. Market Model: concept of beta systematic and unsystematic risk. Investor risk and return preferences: Indifference curves and the efficient frontier, and anticipated inflation. Asset allocation: Asset allocation pyramid, investor life cycle approach, Portfolio management services: Passive – Index funds, systematic investment plans. Active – market timing, style investing.

Unit 4: Capital Asset Pricing Model (CAPM):

Efficient frontier with a combination of risky and risk free assets. Assumptions of single period classical CAPM model. Characteristic line, Capital Market Line, Security market Line. Expected return, required return, overvalued and undervalued assets. Mutual Funds :Introduction, calculation of Net Asset Value (NAV) of a Fund, classification of

mutual fund schemes by structure and objective, advantages and disadvantages of investing through mutual funds. Performance Evaluation using Sharpe's Treynor's and Jensen's measures and Fama's Decomposition.

Reference:

1. Fischer, D.E. & Jordan, R.J. : Security Analysis & Portfolio Management ; Pearson Education.
2. Prasanna Chandra : Investment Analysis and Portfolio Management ; Tata Mcgraw Hill Education Private Limited

Note: Latest edition of the text books should be used

BDE22001

SEMESTER V

GST AND CUSTOMS DUTY – DSC 10

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students are able to:

CO1: Learn in depth the provisions of GST to handle TDS and POS online and off line more efficiently

CO2: Learn the details of computation of GST

CO3: Learn in detail the steps to be followed to determine the assessable value and customs duty

CO4: Understand the details of technology of GST and flow of return filing under GST

CO5: Learn in details and gain knowledge to practice as GST Consultant

Unit 1: Introduction to GST- Indirect tax Structure in India, Constitutional requirements, GST Council, Issues in Indirect Tax, Rationale for Transition to GST. GST- Meaning, Definition of GST - Structure of GST -Types of GST- Features of GST, Benefits of GST. Difficulties in implementation of GST.

GST Compliance requirement definitions of Dealer, Manufacturer and Trader. Registration under GST-Persons liable for registration, compulsory registration, Procedure for Registration, Rejection of application for registration, cancellation of Registration,

Unit 2: a) GST Definitions - Aggregate Turnover, Agriculturist, Business, Credit note and Debit note, Exempt Supplies, Input, Input service, Input Service Distributor, Intra-state supply of Goods, Job work, Invoice. Composition Levy, Mixed Supply, outward supply, Person, Turnover in State

b) Levy and Collection of Tax: Introduction, Supply- meaning and scope of supply, treatment of mixed and composite supply, Liability of tax payable person, Rate and value of tax, transactions without considerations, list of transactions for supply of goods and services and list of transactions for non-supply of goods and services, Reverse charge mechanism,.

Unit 3: Time of supply and Value of taxable supply

Time of Supply Introduction, time of supply-forward charge, reverse charge, residuary, special charges Time of supply of service- forward charge, reverse charge, Vouchers, Residuary, Special charges. Problems on determination of time of supply.

Value of taxable Supply-conditions, inclusions, Consideration not wholly in money, Supply between two related persons, Supply through agent, cost based value, Residual valuation, specific supplies, Service of pure agent. Problems on determination of value of supply.

Unit 4 : Input tax credit and Returns

Input tax credit- Meaning, conditions for taking credit, ineligible input tax credit, availability of credit in special circumstances, Input tax credit and change in constitution of registered person, Taking input tax credit in respect of inputs and capital goods sent for job work, Manner of Distribution of Credit by Input Service Distributor (ISD)

Returns-Furnishing details of outward supplies and inward supplies, a brief introduction to GST forms-1 to 8, Steps for filing forms, Levy of late fee.

Unit 5: Customs Act 1962

Meaning-Notified Goods-Specified goods-Prohibition of Importation and Exportation under section 11-Types of Customs duty. Computation of Assessable Value and Customs duty.

Suggested Readings:

1. GST and Customs Duty - Singhania

BDE23001

SEMESTER V
INVESTMENT BANKING & FINANCIAL SERVICES- DSE

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students are able to:

CO1: Learn the characteristics of Investment banking in India

CO2: Learn the details of Public Issue

CO3: Understand the details of types of leasing

CO4: Understand in detail the feature of Hire Purchase

CO5: Understand the classification and characteristics of Insurance

CO6: Learn in detail the history and evolution of venture Capital

CO7: Learn the details of Securitization

Unit 1: Introduction:

An Overview of Indian Financial System, Investment Banking in India, Recent Developments and Challenges ahead, Institutional structure and Functions of Investment /Merchant Banking; SEBI guidelines for Merchant Bankers, Registration, obligations and responsibilities of Lead Managers, Regulations regarding Continuance of association of lead manager with an issue

Unit 2: Issue Management:

Public Issue: classification of companies, eligibility, issue pricing, promoter's contribution, minimum public offer, prospectus, allotment, preferential allotment, private placement, Book Building process, designing and pricing, Green Shoe Option; Right Issue: promoter's contribution, minimum subscription, advertisements, contents of offer document, Bought out Deals, Post issue work & obligations, Investor protection, Broker, subbroker and underwriters

Unit 3: Leasing and Hire Purchase :

Concepts of leasing, types of leasing – financial & operating lease, direct lease and sales & lease back, advantages and limitations of leasing, Lease rental determination; Finance lease evaluation problems (only Lessee's angle), Hire Purchase interest & Installment, difference between Hire Purchase & Leasing, Choice criteria between Leasing and Hire Purchase mathematics of HP, Factoring, forfaiting and its arrangement, Housing Finance : Meaning and rise of housing finance in India, Fixing the amount of loan, repricing of a loan, floating vs. fixed rate, Practical problems on housing finance.

Unit 4: Venture Capital, Insurance, Credit ratings and Securitization:

Concept, history and evolution of VC, the venture investment process, various steps in venture financing, incubation financing.

Insurance: Concept, classification, principles of insurance, IRDA and different regulatory norms, operation of General Insurance, Health Insurance, Life Insurance.

Credit Ratings: Introduction, types of credit rating, advantages and disadvantages of credit ratings, Credit rating agencies and their methodology, International credit rating practices.

Securitization: Concept, securitization as a funding mechanism, Traditional and nontraditional mortgages, Graduated-payment mortgages (GPMs), Pledged-Account Mortgages(PAMs), Centralized Mortgage obligations (CMOs), Securitization of non mortgage assets, Securitization in India

Reference:

1. M.Y.Khan,-Financial Services‘ – Tata McGraw –Hill, 3 rd Edition, 2005.
2. Machiraju - Indian Financial System –- Vikas Publishing House, 2 nd Edition, 2002.
3. J.C.Verma - A Manual of Merchant Banking –, Bharath Publishing House, New Delhi,2001.
4. K.Sriram - Hand Book of Leasing, Hire Purchase & Factoring‘, ICFAI, Hyderabad, 1992.

Note: Latest edition of the text books should be used

BDE24001

**SEMESTER V
CORPORATE ANALYSIS AND VALUATION- DSE**

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students are able to:

CO1: Learn the details of Analysis of Corporate Financial Statements

CO2: Understand the details of various methods of valuation

CO3: Learn in details with application of Discounted Cash flow valuation method

CO4: Learn the classification of estimating discount rates

CO5: Understand in depth Relative valuation

CO6: Identify the details of Impact of inflation on valuation

Unit 1: Analysis of Corporate Financial Statements:

Income statements and Balance sheets through ratio analysis and analyzing the Chairman's statement, Directors' report, management discussion & analysis, report on corporate governance, auditor's report to evaluate the financial soundness of the company. Understanding financial statements of manufacturing and service organisations. Common size analysis and relevant ratios.

Unit 2: Introduction to Valuation:

Value and price, Balance sheet-based methods, Income statement-based methods, Multiples, Goodwill-based methods. Cash flow discounting-based methods. Deciding the appropriate cash flow for discounting. The free cash flow to the firm, free cash flow to equity. Forecasting Cash flows: simple model for forecasting income and cash flows. Earnings, Tax effect, Reinvestment needs, dividend.

Unit 3: Discounted Cash flow Valuation:

Valuation of a company with no growth, constant growth, variable growth and infinite life. Estimating discount rates-cost of equity, cost of debt, tax shield, weighted average cost of capital. The estimation of equity risk premium, calculation of beta, instability of beta, adjusted beta, levered and unlevered beta, bottoms up beta.

Unit 4: Relative Valuation:

standard multiples, comparable companies, potential pitfalls; estimating multiples using regression. Valuation of brands and intellectual capital. Interest rates and company valuation. Impact of inflation on valuation. Reconciling relative and discounted cash flow valuation. Case studies in valuation.

Reference:

1. Foster, George Financial Statement Analysis, 2nd ed., Pearson Education Pvt Ltd
2. Damodaran, A. (2008). Damodaran on Valuation, Security Analysis for investment and Corporate Finance (2nd ed.). Wiley India Pvt. Ltd.
3. Chandra, P. (2011). Corporate Valuation and Value Creation, (1st ed). TMH
4. Pablo Fernandez, Valuation and Common Sense, free download from ssrn.

Note: Latest edition of the text books should be used

BDE25001

SEMESTER V

RESEARCH METHODOLOGY-DSE

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Understand in details with application of research methodology and become a good researcher

CO2: Learn the characteristics and classification of research Design

CO3: Understand in details with examples Sources of Data Collection

CO4: Learn the characteristics of sampling techniques

CO5: Understand in details with application of methods of analysis

CO6: Understand the details of research report

Unit 1: Business Research:

Meaning - Nature and Scope of Business Research – Role of Business Research in decision making. Applications of Business Research; The Research process – Steps in the research process; the research proposal; Problem Formulation: Management decision problem vs. Business Research problem. Research Design: Exploratory, Descriptive & Causal.

Unit 2: Primary Data Collection:

Survey v/s Observations. Random sample collection methods. Comparison of self-administered, telephone, mail, emails techniques. Qualitative Research
Tools: Depth Interviews focus groups and projective techniques.

Unit 3: Measurement & Scaling:

Primary scales of Measurement-Nominal, Ordinal, and Interval & Ratio. Scaling techniques paired comparison, rank order, constant sum, semantic differential, itemized ratings, Likert Scale; Questionnaire-form & design. Sampling: Sampling techniques, determination of sample size using statistical techniques, Cronbach's Alpha test for reliability(using software).

Unit 4: Data and the Methods of Analysis:

Analysis of Variance (ANOVA) One-Way & Two-Way, Chi square test (goodness of Fit). Multivariate Data Analysis: Factor Analysis (Principal Component Analysis), Discriminant Analysis. Above statistical test also to be explained using statistical software package. Report writing: Contents of a Research Report, Plagiarism in Business Research – Meaning and effects - Plagiarism detection software

Reference:

1. Zikmund, W.G., Babin, B.J., Carr, J.C. & Griffin, M. (2013). Business Research Methods (9th ed.). Cengage Learning.
2. Johnson, R.A. & Wichern, D.W. (1997) Business Statistics-Decision Making with Data (1st ed.). John Wiley & Sons.
3. Cooper, D.R. & Schindler, P.S. (2008) Business Research Methods (10th ed.). McGraw Hill Education.

- 4..Chawla, D, &Sondhi,N. (2011) Research Methodology Concepts and Cases (1st ed.).
Vikas Publishing House
5. Malhotra, N & Dash. S (2010) Marketing Research An Applied Orientation (6th ed.).
Pearson, Prentice Hall of India.

Note: Latest edition of the text books should be use

BDE26001

SEMESTER V

**INTERNATIONAL TRADE BLOCKS AND
MULTILATERAL AGENCIES- DSE**

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO 1: Learn the details of Economic theory on international trade

CO2: Understand the characteristics of domestic, foreign, global environment and analyse their impact on international business decision

CO3: Understand the details of regional integration and trade blocks

CO4: Understand the characteristics and types of international investment

CO5: Identify the impact of reform on competitiveness of the Indian firms

CO6: Understand in depth economic institutions

Unit 1: Review of Economic Theory on International Trade:

Basis for international trade; gains from trade; distributional issues, policy instruments and their impact, political economy. Importance, nature and scope of international relation, modes of entry into international business, internationalization process and managerial implications; Domestic, foreign and global environments and their impact on international business decision; Growing concern for green trades.

Unit 2: International economic & trading environment:

Regional integration and trade blocks, regionalism v/s. multilateralism, European union, integration of developing countries -BRICS, ASEAN, SAARC, SAFTA, NAFTA, G-20; World trade in goods and services -Major trends and developments; World trade and protectionism - Tariff and non-tariff barriers ; Counter trade, UNCTAD, WTO, GATT, GATS, TRIM, TRIPS; India's role in facilitating trade relations under BRICS, SAARC, SAFTA, ASEAN and to WTO.

Unit 3: International investment:

Types and significance of foreign investments, factors affecting international investment, growth and dispersion of FDI, Cross border mergers and acquisition, foreign investment in India-Impact of reforms on competitiveness of the Indian Firms, EURO/ADR issues, ECBs; current economic crises in US/Europe/Asia and its impact on economic growth in India.

Unit 4: Economic institutions:

International Monetary Funds (IMF), World Bank (IBRD, IDA, IFC), Asian Development Bank, BRICS Development Bank, Bilateral funding arrangements with special reference to Japan International Cooperation Agencies (JICA), agencies of USA; Case studies on Bilateral financing arrangements of Indian projects like Delhi Metro, Dedicated Freight corridor, Nuclear Power Plant etc.

Reference:

1. Hill, W. L. Charles and Jain, A.K. (2008). International Business (6th ed.). India: McGrawHill.
2. Fernando, A.C. (2011) .Corporate Governance: Principles, Policies and Practices. India: Pearson Education.
3. Roger, Bennet (1999). International Business, Financial Times. London: Pitman Publishing.
4. Sharan, Vyuptakesh (2003). International Business (2nd ed.). India: Pearson Education.
5. Krueger, Anne O. (2002). Economic Policy Reforms and the Indian Economy.OUP.

Note: Latest edition of the text books should be used

BDE27001

SEMESTER V

HUMAN RESOURCE MANAGEMENT- DSE

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students are able to:

CO1: Understand in depth the techniques to manage Human Resources at work place and able to occupy a position of H R Manager

CO2: Deliberate the methods and techniques of training to train the Human Resources and create assets for the organisation

CO3: Understand in depth the methods of recruitment

CO4: Understand the basis and purpose of promotion

CO5: Learn the details of transfer

Unit 1: Human Resource Management

Meaning, Definition – objectives, Importance , Functions and process. HR policies, HR Manager- Duties and responsibilities – Recent trends in HRM

Unit 2: Human Resource Planning, Recruitment and Selection

Meaning – Importance of Human Resource Planning – Benefits of Human Resource Planning. Recruitment – Meaning – methods of recruitment. Selection – meaning – steps in selection process. Interview – meaning and types. Induction – Meaning, objectives and purposes.

Unit 3: Training and Development

Training – Need for training, benefits of training, identification of training needs and methods of training. Training v/s development, methods of development. Career Development.

Unit 4: Performance Appraisal and Compensation

Performance Appraisal – Meaning and Definition, objectives, methods, uses and limitation. Compensation – meaning and objectives. Fringe benefits.

Unit 5: Promotion and Transfers

Meaning and Definition of Promotion – purpose of promotion, basis of promotion, meaning of transfer, reasons for transfer, types of transfer, promotion v/s transfer.

Reference:

1. Gary Dessler. *A Framework for Human Resource Management*. Pearson Education.
2. DeCenzo, D.A. and S.P. Robbins, *Personnel/Human Resource Management*, Pearson Education.
3. Bohlendar and Snell, *Principles of Human Resource Management*, Cengage Learning
4. Ivancevich, John M. *Human Resource Management*. McGraw Hill.
5. Wreather and Davis. *Human Resource Management*. Pearson Education.
6. Robert L. Mathis and John H. Jackson. *Human Resource Management*. Cengage Learning.
7. TN Chhabra, *Human Resource Management*, DhanpatRai& Co., Delhi

8. Biswajeet Pattanayak, *Human Resource Management*, PHI Learning
9. Neeru Kapoor, *Human Resource Management*, Taxmann Publication

Note: Latest edition of text books may be used.

BDF21001

SEMESTER VI

CORPORATE RESTRUCTURING – DSC 11

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Understand in detail the problems faced in joint ventures

CO2: Learn the types and structure of joint venture

CO3: Understand in detail the theories of Mergers and Acquisitions

CO4: Learn the details of legal frame work for mergers and acquisitions

CO5: Understand the methods and factors affecting Valuation

CO6: Understand the details of Post merger evaluation and its impact on shareholders' wealth

Unit 1: Joint Ventures:

Concept & Meaning of Joint Ventures, Need & Types of Joint Ventures, Structures & Problems faced in Joint Ventures, Joint Ventures and Strategic Alliance. Some relevant case study of successful and failed joined ventures.

Unit 2: Mergers and Acquisitions:

Introduction to mergers, types of mergers, theories of mergers & acquisitions; Cross-border mergers and acquisitions, issues and challenges in cross border M&A. Analysis of Post-Merger Performance. Demerger, types of demerger, reverse merger, buyback of shares, leverage buy-out strategy, merger strategy - growth, synergy, operating synergy, financial synergy, diversification. Takeover and its types, takeover strategy, takeover bids, legal framework for mergers and acquisitions, leverages and buyouts; Hostile tender offers and various anti-takeover strategies.

Unit 3: Deal Valuation and Evaluation:

Factors affecting valuation basics, methods of valuation, cash flow approaches, Economic Value Added (EVA), sensitivity analysis, valuation under takeover regulation, valuation for slump sale, cost-benefit analysis and swap ratio determination.

Unit 4: Post-Merger Evaluation:

Financial Evaluation of Mergers & Acquisitions, Impact on shareholders' Wealth; Methods of payment and financing options in mergers & acquisitions, financing decision, Merger, Acquisition and Competition law 2002, SEBI (Securities & Exchange Board of India) Takeover Code 2011 and criteria for negotiating friendly takeover.

Reference:

1. Weston, Fred; Chung, Kwang S. &Siu, Jon A.: Takeovers, Restructuring and Corporate Governance, (2nd ed.). Pearson Education
2. Gupta, Manju (2010): Contemporary Issues in Mergers and Acquisitions. Himalaya Publishing
3. Sundarsanam (2006); Creating Value from Mergers and Acquisitions, (1st ed.) Pearson Education
4. Ramanujan. S. (1999); Mergers: The New Dimensions for Corporate Restructuring, McGraw Hill

5. Narayankar, Ravi, (2013): Merger and Acquisitions Corporate Restructuring, Strategy and Practices, (2nded.). International Book House Pvt. Ltd.

Note: Latest edition of the text books should be used

BDF22001

SEMESTER VI

INTERNATIONAL FINANCIAL INSTITUTIONS – DSC 12

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn the details of IFIS and their implications on business

CO2: Understand the characteristics of world bank

CO3: Learn the History and Organisational Structure of world bank

CO4: Understand in detail the functioning of IDA and IFC

CO5: Understand the steps taken by IMF to solve problems of international liquidity

CO6: Understand the Dispute Settlement facilities of international Centre for investment Disputes

Unit 1: International financial institutions

Meaning - objectives- need-functions. International liquidity: meaning, problems of international liquidity-IFIs and their implications on business.

Unit 2: International monetary fund (IMF):

Origin- objectives functions- membership- organization and structure- financial resources- role played by IMF in global business- steps taken by IMF to solve problems of international liquidity.

Unit 3: World Bank:

Origin-History –membership-organisational structure- capital resources- funding objectives-functions of World Bank in international business-loan provision: projects loans, sectoral loans and structural adjustment loans.

Unit 4: IDA and IFC:

IDA: Origin- history-objectives- membership -organizational structure-IDA assistance – Donor contributions.

IFC: Origin-history-objectives - organizational structure-funding mechanisms-Financial resources- Assistance to member countries.

Unit 5:UNCTAD and MIGA:

UNCTAD : Origin-history-functions-organisational structure-activities.

MIGA: Origin-history-functions-organisational structure-activities.

Unit 6:ADB and International Centre for Investment Disputes(ICID):

ADB: Origin- history-objectives- membership -organizational structure-Functions.

International centre for investment Disputes (ICID): Origin- history- membership - organizational structure-activities-Disputes settlement facilities.

Reference:

1. Principles & Practice of Banking – B.S. Raman
2. International Banking operations – Indian institute of Banking and finance
3. Corporate finance for Non finance Security – Prasanna Chandra

Note: Latest edition of the text books should be used

BDF23001

**SEMESTER VI
STRATEGIC CORPORATE FINANCE- DSE**

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can

CO1: Learn in detail the risk management choices

CO2: Understand the details of value enhancement tools and techniques

CO3: Learn the different types of financial strategy for shareholders wealth maximization of strategic corporate finance

CO4: Understand the significance of strategy in financial decision

CO5: Learn the details of Financial Distress and restructuring

CO6: Learn in depth capital structure and factors affecting the capital structure

Unit 1: Introduction to strategic corporate finance:

Strategy v/s Planning, significance of strategy in financial decisions, Different types of financial strategy for Shareholders Wealth Maximization, overall corporate value addition and Economic Value Addition. Strategic Cost Management: Traditional costing Vs Strategic Costing, Relevant costs v/s Irrelevant costs, Different types of strategic costing and their relevance- Target Costing, Activity based Costing, Life Cycle Costing, Quality Costing, Zero Based Budgeting, Strategic cost reduction techniques and value chain analysis. Valuing Real assets in the presence of risk: tracking portfolios and Real Asset valuation, Different Approaches of Valuing Real Assets, Capital Budgeting and Strategic policy.

Unit 2: Fundraising:

Identification of different sources of development capital, determination of capital structure and factors affecting the capital structure, cost of capital and cost saving strategy, production of a business plan and financial forecasts to enable potential funders to assess the proposition. Alternative sources of financing – alternative sources of financing, Different approach to infrastructure projects financing- Public Private Partnership (PPP) and its relevance. Managing credit ratings. Dividend v/s share repurchase policy, problem of too much cash. The issues of stock liquidity and illiquidity.

Financial Distress and restructuring: Meaning of Bankruptcy, Factors leading to bankruptcy, symptoms and predictions of bankruptcy, reorganization of distressed firms, liquidation of firms. Company disposals: retirement sale or the sale of a non-core subsidiary, planned exit, forceful retirement and other disposals. Exit strategy- most appropriate exit route, valuation, timing of sale and tax planning opportunities, identification of potential purchasers, approaching the potential purchaser, negotiate with potential acquirers and selection of a preferred purchaser. Real options: Financial and real options compared, various types of real options, application of Real options, Drawbacks of Real options

Unit 3: Company Valuation:

An overview of valuation, valuation principles and practices more, the impact of “what if” scenarios, the key financial and commercial factors affecting the business. Value enhancement tools & techniques, the link between valuation and corporate finance Management Buy-outs: Establishing feasibility of the buy-out, Negotiating the main terms of

the transaction with the vendor including price and structure, Developing the business plan and financial forecasts in conjunction with the buy-out team for submission to potential funders, negotiations with potential funders so that the most appropriate funding offers are selected. Management Buy-ins: Management Buy-in/Buy-outs (“BIMBOs”), Vendorinitiatedbuy-outs/buy-ins. Due Diligence: financial due diligence for both purchasers and financial institutions.

Unit 4: Strategic risk management:

Strategic risk management, the substitutability of capital structure and risk management choices, such as process control efforts, financial, physical, and operational hedging, value-based management.

Reference:

- 1.AswathDamodaran: Corporate finance theory and practice; John willey& sons, Inc.
- 2.AswathDamodaran: Strategic Risk Taking:A Framework for Risk Management ; Prentice Hall
- 3.IMPandey: Financial Management; Vikas Publishing House
- 4.Strategic Financial Management: Prasanna Chandra; McGraw Hill Education (India) Private Limited

Note: Latest edition of text books may be used.

BDF24001

**SEMESTER VI
MANAGEMENT OF FINANCIAL INSTITUTIONS- DSE**

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn the details of financial institution and its kind

CO2: Understand and analyse the financial statement of Financial institutions

CO3: Identify the financial strength and weakness of Financial institutions

CO4 :Understand the details of capital adequacy norms

CO5: Learn the classification of Institutional risk and the theories of interest rate

CO6: Understand the causes of liquidity risk

Unit 1 Financial Intermediation;

Kinds of Intermediation; Financial Institution and its kinds; An overview of the Indian financial system; Regulation of Banks, NBFCs & FIs; Products offered by Banks and FIs. CRR & SLR management; Capital Adequacy: Capital adequacy norms; Basel agreement-II&III; effect of capital requirements on bank operating policies

Unit 2 Statement of Financial Sector:

Flow of Funds Accounts – Sector wise and Instrument wise. Statements of Financial Institution: Analyzing Bank's Financial Statement: The balance sheet; income statement; Cash Flow Statement; profitability, liquidity and solvency analysis; Performance Analysis of banks: CAMELS Risk system; KPIs; Data Envelopment Analysis. Asset Liability Management: RBI guidelines on asset liability management

Unit 3 Institutional Risk Management:

Interest Rate Risk; Market Risk; Credit Risk; Liquidity Risk; Operational Risk. Determination of Interest Rate. Theories of Interest Rates: Classical Theory; Loan able Funds Theory; Liquidity Preference Theory; Term Structure of Interest Rates. Interest Rate Risk Management: Measurement of Interest Rate Risk; Duration and its kinds; Convexity. Managing Interest Rate Risk: Reprising Gap Model, Maturity Matching Model, Duration Gap Model, Cash Flow Matching Model; Convexity Adjustments.

Unit 4: Credit & Liquidity Risk Management:

Types of Assets, NPA & its types, Management of NPA, Measurement of Credit Risk – Qualitative and Quantitative models. Modelling Credit Risk; Term Structure of Credit Risk; Managing Credit Risk: Credit Analysis and kinds of Loans; Pricing of Loans. Liquidity Risk Management: Measurement of Liquidity Risk; Measures of Liquidity Exposure; Causes of Liquidity risk: Asset-Side and Liability-Side; Managing Liquidity Risk :Purchased Liquidity management and Stored Liquidity management; Liquidity Planning; Deposit Insurance; Discount Window

Reference:

1. Saunders & Cornett – “Financial Institutions Management – A risk management approach”
Tata McGraw Hill
2. Resti & Sironi – “Risk management and shareholders’ value in banking” John Wiley
3. Rose & Hudgins – “Bank management and financial services”
4. IIBF-“Bank Financial Management ”
5. Paul & Suresh – “Management of Banking and Financial Services ”
6. Subramanyam – “Investment Banking ”
7. Madhuvij -- “Management of financial institutions”
8. <http://nptel.iitm.ac.in/courses/110106040/>

Note: Latest edition of text books may be used.

BDF25001

SEMESTER VI
FINANCIAL DERIVATIVES- DSE

(Credits: Lecture – 05, Tutorial – 01, Practical – 0)

Lectures: 80 Hrs

Tutorials: 16 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn the classification and characteristics of financial derivatives in India

CO2: Identify and study the relationship between derivatives and other variables

CO3: Learn the characteristics of options and factors affecting option prices

CO4 : Understand the various types of SWAPS

CO5: Understand in depth forward and future contracts and play an important role in dealing with these contract

Unit 1: Introduction:

History of derivatives - origin of derivatives in India - the classification of derivatives – the important features of derivatives – Early delivery, extension and cancellation of forward contracts – financial derivatives market in India.

Unit 2: Forward and Future Contracts:

Meaning of forwards and futures – difference between forwards and futures – Clearing house mechanism - Margin requirements: Initial margin, maintenance margin and margin call – Making to Market – pricing of futures (Problems in preparation of Margin Accounts and Pricing of futures) – Valuation of futures contracts.

Unit 3: Options:

Meaning – Types of Option contracts – factors affecting option prices – Distinction between futures and options. Prices – Upper bounds and lower bounds, early exercise, put-call parity – Time value and Intrinsic value of options – Concept of options In the money, At the money and Out of the money – Computation of Gross pay off and Net pay off from options contracts – Graphical representation of Pay off from options contracts.

Unit 4: Understanding of SWAPS:

Meaning and types – Interest rates swap – Currency Swaps; Credit Derivatives: Credit ratings, Credit Default Swaps.

Reference:

1. Ruey S. Tsay (2005). Analysis of Financial Time Series (2nd ed.). John Wiley.
2. John C. Hull. Options, Futures and Other Derivatives (7th ed.). Pearson Education.
3. Jurgen Franke, Wolfgang Hardle and Christian Hafner. Introduction to Statistics of Financial Markets.
4. R. Madhumathi, M. Ranganatham. Derivatives and risk management (1st ed.)
Redhead, K. Financial Derivatives- An introduction to futures, forwards, options, swaps
Prentice Hall of India
5. Kotreshwar – Derivative Market

Note: Latest edition of text books may be used

BDF26001

**SEMESTER VI
RESEARCH SOFTWARE PACKAGE- DSE**

(Credits: Lecture – 02, Tutorial – 0, Practical – 02)

Lectures: 32 Hrs

Practical: 64 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Understand the details of EViews

CO2: Learn the details of importing data to EViews

CO3: Understand the details of estimating a Multiple Regression Equation by Ordinary Least Squares using software

CO4: Identify the details of printing output and graph

CO5 : Understand the details of estimating Logit and Probit equation

A Brief Introduction to EViews (Econometric Views)

A. Importing Data to EViews

B. Executing Simple Procedures

i. Showing Data

ii. Graphing Data

iii. Descriptive Statistic of data

iv. Estimating a Multiple Regression Equation by Ordinary Least Squares

v. Testing of Homoscedasticity

vi. Testing of Error Term

vii. Testing of Multicollinearity

viii. Testing of stationarity

ix. Estimating a Logit and Probit Equation

x. “Views” of the Regression Equation

C. Printing Output and Graph

Readings:

http://www.eviews.com/illustrated/EViews_Illustrated.pdf

Note: Latest edition of text books may be used.

BDF27001

SEMESTER VI
COMPUTERISED ACCOUNTING SYSTEM- DSE

(Credits: Lecture – 04, Tutorial – 00, Practical – 02)

Lectures: 64 Hrs

Tutorials: 64 Hrs

Course Outcome:

On successful completion of this course the students can:

CO1: Learn the details of Processing and preparing ledgers, trial balance using SQL query and report modules

CO2: Understand the details of designing accounting support system

CO3: Understand the details of auditing in computerized accounting system using Generic Software

CO4: Learn in detail the creation of sample Data base using view, query and report modules

Unit-1: Computerized Accounting: Using Generic Software (12 Lectures, 12 Practical Lab)

Taxation: TDS, VAT and Service Tax Auditing in Computerized Accounting system: Statutory Audit, Voucher verification, Verification of related party transaction, CAAT: Various Tools

Unit-2: Introduction to DBMS(10 Lectures, 10 Practical Lab)

DBMS - concepts-DBMS Modules - Table, Form, View, Query and Report. Familiarising with SQL. Creating a sample data base and using view, query and report modules.

Unit -3 Designing Computerised Accounting System (14 Lectures, 14 Practical Lab)

Requirement Analysis, Designing Main Tables. Creating Voucher entry form, Populating the data base, Processing and preparing ledgers, Trial Balance using SQL query and report modules,

Unit-4 : Designing Accounting Support System (16 Lectures, 16 Practical Lab)

Designing Bills Payable and Bills Receivable Systems for Accounting using Form, Query and Report Modules. Designing Payroll System for Accounting using Form, Query and Report Modules.

Note:

1. The General Purpose Software referred in this course will be notified by the University Departments every three years. If the specific features, referred in the detailed course above, is not available in that software, to that extent it will be deemed to have been modified.

2. Teaching arrangements need to be made in the computer Lab

Reference:

The suggested readings and guidelines shall be notified by the university department at least once in three years based on the selected software.

Note: Latest edition of text books may be used.

BDF28001

SEMESTER VI
Project Work- DSE

(Credits: Lecture – 01, Tutorial – 01, Practical – 04)

Lectures: 16 Hrs

Tutorials: 16 Hrs

Practical: 128 Hrs

Course Outcome:

On successful completion of the project work the students are able to:

CO1: Understand in depth to fill the gap between theory and practical through internship

CO2: Understand in detail with examples the procedure and able to write a report on the problems of human resources, Production, Marketing purchase, financial, consumer's preferences and behaviour, pricing, competition etc of the organisations

CO3: Specify the details in depth and able to communicate effectively

CO4: Learn in detail and able to absorb as an employee by the employer

CO5: Specify and analyse the components of project report and prepare the report effectively

C1 – Proposal of Project Work - 15 Marks

C2 – Progress of Project Work - 15 Marks

Viva - 20 Marks

Valuation of Report – 50 Marks

JSS COLEGE OF ARTS COMMERCE & SCIENCE
(Autonomous)
BN Road, Mysuru – 25
Department of Commerce and Management

Model Question Paper
(For all courses except QT, computer Applications in Business, Computerized Accounting System, Research Software Package)

BBA Programme

Time 3 hrs

Maximum marks: 70

Allocation of marks and Model Question Paper Commerce

- I. The question paper carries 70 marks.
- II. It is divided into 3 Parts, Part A, Part B and Part C
- III. Part A carries 30 marks. 2x15=30
- IV. Part B, carries 20 marks..... 2x10=20
- V. Part C, carries 20 marks..... 4x5=20

BBA

Question Paper Pattern- 2017-18 (CBCS)

(For all courses except QT, computer Applications in Business, Computerized Accounting System, Research Software Package)

PART-A

Answer the following. Each question carries 15 marks. 2X15=30

1.

OR

2.

3.

OR

4.

PART-B

**Answer the following. Each question carries 10 marks.
2X10=20**

5.

OR

6.

7.

OR

8.

PART-C

Answer any four of the following. Each question carries 5 marks.4X5=20

9.

10.

11.

.....

12.

13.

BBA

**Question Paper Pattern
4.4 Quantitative Techniques**

Time: 3hrs

Max. Marks: 70

Part-A

Answer the following. Each question carries two marks 10X2=20

- | | |
|------------|---------|
| 1. a. | f. |
| b. | g. |
| c. | h. |
| d. | i. |
| e. | j. |

Part-B

Answer any four of the following. Each question carries five marks 4X5=20

- 2.
- 3.
- 4.
- 5.
- 6.

Part-C

Answer any three of the following. Each question carries ten marks 3X10=30

- 7.
- 8.
- 9.
- 10.

BBA

Question Paper Pattern

4.5 Computer Application in Business / DSE – Research Software Package

Time: 2 hrs

Max. Marks: 50

Part-A

Answer the following. Each question carries two marks.10X2=20

- 1. a.
- b.
- c.
- d.
- e.
- f.
- g.
- h.
- i.
- j.

Part-B

Answer any two of the following. Each question carries five marks.

2X5=10

- 2.....
- 3.
- 4.....

Part-C

Answer any two of the following. Each question carries ten marks

2X10=20

- 5.
- 6.....
- 7.

BBA
Question Paper Pattern
DSE – Computerized Accounting System

Time: 3 hrs

Max. Marks:70

Part-A

Answer the following. Each question carries two marks.10X2=20

1. a.
- b.
- c.
- d.
- e.
- f.
- g.
- h.
- i.
- j.

Part-B

Answer any four of the following. Each question carries five marks.
4X5=20

2.
3.
4.
5.
6.

Part-C

Answer any three of the following. Each question carries ten marks
3X10=30

7.
8.
9.
10.



**JSS COLLEGE OF ARTS, COMMERCE AND
SCIENCE**

(Autonomous)

B N ROAD, MYSURU- 570 025

**DEPARTMENT OF COMPUTER
SCIENCE**

Syllabus

CHOICE BASED CREDIT SYSTEM

For BCA PROGRAMME

Bachelor of Computer Applications

2017-18

CBCS Syllabus - BCA for 2017-2018 onwards

		Course	Title	Hours / Week		Credits	Maximum Marks			Exam Duration	Total
				L	T/ P		L:T:P	IA			
I Year	I Semester	DCA 21001	Computer Fundamentals & MIS	4	4	4:0:2	15	15	70	3 Hours	100
		DCA 23001	Computer System Organization and Architecture	4	4	4:0:2	15	15	70	3 Hours	100
		DCA 25001	Object Oriented Programming In C++	4	4	4:0:2	15	15	70	3 Hours	100
	II Semester	DCB 21001	Data Structures and File Processing	4	4	4:0:2	15	15	70	3 Hours	100
		DCB 23001	Operating Systems with Linux	4	4	4:0:2	15	15	70	3 Hours	100
		DCB 25001	Programming in JAVA	4	4	4:0:2	15	15	70	3 Hours	100
II Year	III Semester	DCC 21001	Discrete Mathematics & Logic Computation	5	1	5:1:0	15	15	70	3 Hours	100
		DCC 23001	Database Management Systems	4	4	4:0:2	15	15	70	3 Hours	100
		DCC 25001	Web Technologies	4	4	4:0:2	15	15	70	3 Hours	100
	IV Semester	DCD 21001	Numerical Analysis And Statistics	4	4	4:0:2	15	15	70	3 Hours	100
		DCD 23001	J2EE	4	4	4:0:2	15	15	70	3 Hours	100
		DCD 25001	Software Engineering & Software Testing	4	4	4:0:2	15	15	70	3 Hours	100
		DCD 31001	Mathematics	3	1	3:1:0	15	15	70	3 Hours	100
		DCD 33001	Business Mathematics	3	1	3:1:0	15	15	70	3 Hours	100
		DCD 35001	Accountancy & Financial Management	3	1	3:1:0	15	15	70	3 Hours	100
		DCD 37001	Entrepreneurship Development	3	1	3:0:1	15	15	70	3 Hours	100

	Course	Title	Hours / Week		Credits	Maximum Marks			Exam Duration	Total	
			L	T/ P		IA		Exam			
					L:T:P	C1	C2		Marks		
III Year	V Semester	DCE 21001	Data Communication and Computer Networks	4	3	4:0:1	15	15	70	3 Hours	100
		DCE 23001	Computer Graphics	4	3	4:0:1	15	15	70	3 Hours	100
		DCE 25001	Multimedia Systems and Applications	4	3	4:0:1	15	15	70	3 Hours	100
		DCE 31001	ASP. Net	4	3	4:0:1	15	15	70	3 Hours	100
		DCE 33001	Visual Programming	4	3	4:0:1	15	15	70	3 Hours	100
		DCE 35001	Artificial Intelligence and Expert Systems	4	3	4:0:1	15	15	70	3 Hours	100
		DCE 41001	Cloud Computing	4	3	4:0:1	15	15	70	3 Hours	100
		DCE 43001	Enterprise Resource Planning	4	3	4:0:1	15	15	70	3 Hours	100
		DCE 45001	Data Mining & Warehousing	4	3	4:0:1	15	15	70	3 Hours	100
		DCE 51001	Object Oriented Modelling & Design With UML	3	3	3:0:1	15	15	70	3 Hours	100
		DCE 53001	Analysis and Design of Algorithms	3	3	3:0:1	15	15	70	3 Hours	100
		DCE 55001	E-Commerce Technologies	3	3	3:0:1	15	15	70	3 Hours	100
		VI Semester	DCF	31001	Operation Research	4	3	4:0:1	15	15	70
33001	System Software & Compiler design			4	3	4:0:1	15	15	70	3 Hours	100
35001	Digital Image Processing			4	3	4:0:1	15	15	70	3 Hours	100
40001	Project			0	12	0:0:6	15	15	70	3 Hours	100
51001	PHP Programming			1	2	1:0:1	15	15	70	3 Hours	100
53001	Computer Simulations			1	2	1:0:1	15	15	70	3 Hours	100
55001	Information Security & Cyber Law			1	2	3:0:1	15	15	70	3 Hours	100

GENERAL RECOMMENDATIONS:

- Teacher who handles theory is completely responsible to prepare the lab exercise well in advance which has to be placed before the department meeting for approval.
- Related practical programs' skeleton has to be discussed in the theory class.
- Lab instructors

Assessment Maximum marks - 100

Course Type	C1		C2		C3 Exam Marks		Assigned Marks (Percentage)			Total
	Theory	Lab	Theory	Lab	Theory	Lab	Theory	Lab	IA	
DSC	10	05	10	05	70	70	50	20	30	100
DSE	10	05	10	05	70	70	50	20	30	100
DSE (non practical)	15	-	15	-	70	-	70	--	30	100
SEC	15	-	15	-	50	-	70	--	30	100

Note:

1. C1 will be conducted for 20 Marks (Theory) with one hour duration - 10 Marks (Lab) with continuous assessment and it will be reduced to assigned marks.
2. C2 will be conducted for 20 Marks (Theory) with one hour duration - 10 Marks (Lab) with continuous assessment and it will be reduced to assigned marks.
3. C3 will be conducted for 70 Marks (Theory) with three hours duration - 70 Marks (Lab) with 3 hours duration and to be reduced to assigned marks.
4. For non-practical course C3 will be conducted for 70 Marks (Theory) with three hours duration.
5. In case of SEC, C1 and C2 will be conducted for 15 Marks each with one hour duration and C3 will be conducted for 50 Marks with 2 hours duration.

Programme Outcomes

After completing the graduation in the Bachelor of Computer Applications the students are able to:

- PO1. Get expected skills to be placed in IT sector and self-employment.
- PO2. To develop abilities for data analysis and interpretation using ICT.
- PO3. Acquire comprehensive knowledge with equal emphasis on theory and practice.
- PO4. Analyze and apply latest technologies to solve problems in the areas of computer applications.
- PO5. Develop the basic programming skills to enable students to build Utility tools.
- PO6. Get the foundation knowledge for higher studies in the field of Computer Application.
- PO7. Analyze and synthesis computing systems through quantitative and qualitative techniques
- PO8. Develop practical skills to provide solutions to industry, society and business.
- PO9. Work effectively both as an individual and a team leader on multidisciplinary projects.
- PO10. Improves communication skills so that they can effectively present technical information in oral and written reports
- PO11. To integrate ethics and values in designing computer application.

Programme Specific Outcomes

After completing the graduation in the Bachelor of Computer Applications the students are able to:

- PSO1. Knowledge of contemporary and emerging issues in computer science
- PSO2. Ability to identify, critically analyse, formulate and develop computer application
- PSO3. Learn techniques, skills and modern hardware and software tools necessary for innovative software solutions
- PSO4. Devise and conduct experiments, interpret data and provide well informed conclusions.
- PSO5. Information about computer, technology, organization and management.
- PSO6. Know various computer applications and latest development in IT and communication system.
- PSO7. Act as software programmer, system and Database administrator, web designer, faculty for computer science and computer applications.
- PSO8. Design and conduct experiments, analyze and interpret data.

DCA21001

I SEMESTER**DSC 1A: Computer Fundamentals & MIS****Credit (L: T: P = 4: 0: 2)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Deliberate the details of computer system
- CO2. Learn the classification and characteristics of computer system
- CO3. Understand in details with examples software
- CO4. Identify the characteristics of devices
- CO5. Learn the classification and characteristics of software
- CO6. Understand the classification and characteristics of Memory units
- CO7. Learn the classification and characteristics of CPU
- CO8. Identify the characteristics of Computer Components
- CO9. Understand the classification and characteristics of Computer Technologies
- CO10. Learn the details of Computer Application in Education and research
- CO11. Identify in details with examples MIS
- CO12. Specify in depth MIS

Unit - 1**15 Hours**

Introduction: Introduction to computer system, uses, types.

Human Computer Interface: Types of software, Operating system as user interface, utility programs

Devices: Input and output devices (with connections and practical demo), keyboard, mouse, joystick, scanner, OCR, OMR, bar code reader, web camera, monitor, printer, plotter

Unit - 2**15 Hours**

Memory: Primary, secondary, auxiliary memory, RAM, ROM, cache memory, hard disks, optical disks

Computer Organization and Architecture: C.P.U., registers, system bus, main memory unit, cache memory, Inside a computer, SMPS, Motherboard, Ports and Interfaces, expansion cards, ribbon cables, memory chips, processors.

Unit - 3**15 Hours**

Overview of Emerging Technologies: Bluetooth, cloud computing, big data, data mining, mobile computing and embedded systems.

Use of Computers in Education and Research: Data analysis, Heterogeneous storage, e-Library, Google Scholar, Domain specific packages such as SPSS, Mathematical etc.

Unit - 4**15 Hours**

MANAGEMENT INFORMATION SYSTEM: Introduction to data and information, Types of Information, Types of information System. Impact of MIS, Role and Importance, Managers and Activities in IS, Types of Computers Used by Organizations in Setting up MIS, Hardware support for MIS.

Reference Books:

1. A. Goel, Computer Fundamentals, Pearson Education, 2010.
2. P. Aksoy, L. DeNardis, Introduction to Information Technology, Cengage Learning, 2006
3. P. K.Sinha, P. Sinha, Fundamentals of Computers, BPB Publishers, 2007

Computer Fundamentals & MIS Lab**LIST OF EXPERIMENTS**

1. Study and Identification of standard desktop personal computer
2. Understanding of Motherboard and its interfacing components
3. Install and configure computer drivers and system components.
4. Disk formatting, partitioning and Disk operating system commands
5. Install, upgrade and configure Windows operating systems.
6. Remote desktop connections and file sharing.
7. Identify, Install and manage network connections Configuring IP address and Domain name system
8. Install, upgrade and configure Linux operating systems.
9. Installation Antivirus and configure the antivirus.
10. Installation of printer and scanner software.
11. Disassembly and Reassembly of hardware.
12. Trouble shooting and Managing Systems

DCA23001**I SEMESTER****DSC 2A: Computer System Organization and Architecture****Credit (L: T: P = 4: 0: 2)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Deliberate in details with examples Boolean algebra and logic circuits
- CO2. Learn the details of Data Representation and Computer Arithmetic
- CO3. Learn in depth Computer Organization and Design
- CO4. Learn the details of architecture of CPU
- CO5. Deliberate the classification and characteristics of Basic Computer Programming Concepts
- CO6. Write down in depth Basic Computer Programming Concepts
- CO7. Learn the classification and characteristics of Input -Output organization

Unit - 1**15 Hours**

Introduction: Logic gates, Boolean algebra, combinational circuits, circuit simplification, flip-flops and sequential circuits, decoders, multiplexors, registers, counters and memory units.

Data Representation and basic Computer Arithmetic: Number systems, complements.

Unit - 2**15 Hours**

Fixed and floating point representation, character representation, addition, subtraction, magnitude comparison.

Basic Computer Organization and Design: Computer registers, bus system, instruction set

Unit - 3**15 Hours**

Timing and control, instruction cycle, memory reference, input-output and interrupt.

Central Processing Unit: Register organization, arithmetic and logical micro-operations, stack organization, micro programmed control.

Unit - 4**15 Hours**

Programming the Basic Computer: Instruction formats, addressing modes, instruction codes, Machine language, and Assembly language, input output programming.

Input-output Organization: Peripheral devices, I/O interface, Modes of data transfer, direct memory access.

Reference Books:

1. M. Mano, Computer System Architecture, Pearson Education 1992.
2. A. J. Dos Reis, Assembly Language and Computer Architecture using C++ and JAVA, Course Technology, 2004
3. W. Stallings, Computer Organization and Architecture Designing for Performance, 8th Edition, Prentice Hall of India, 2009
4. Digital Design, M.M. Mano, Pearson Education Asia, 1979

Computer System Organization and Architecture Lab

1. Verification of Basic gates(AND,OR,NOT)
2. Verification of Universal gates(NAND,NOR,EX-OR)
3. Verification of NAND gate as a Universal gate
4. Verification of NOR gate as a Universal gate
5. Verification of DeMorgan's theorem
6. Verification of Half adder & Full Adder
7. Verification of Half subtractor& Full Subtractor
8. Verification of Half adder & Half subtractor using NAND gate
9. Conversion of Binary to Gray &gray to Binary Code
10. Simplification of Boolean Expressions
11. Simplification of Boolean Expressions using K-Map
12. Flip-Flops: SR FF (clock, without clock)
13. JK FF
14. Toggle FF
15. Delay FF
16. Multiplexer
17. De-multiplexer
18. Simulate the machine for the following memory-reference instructions with $I= 0$ and address part = 082. The instruction to be stored at address 022 in RAM. Initialize the memory word at address 082 with the operand B8F2 and AC with A937. Determine the contents of AC, DR, PC, AR and IR in hexadecimal after the execution.
 - a. ADD f. BSA
 - b. AND g. ISZ
 - c. LDA
 - d. STA
 - e. BUN
19. Simulate the machine for the memory-reference instructions referred in above question with $I= 1$ and address part = 082. The instruction to be stored at address 026 in RAM. Initialize the memory word at address 082 with the value 298. Initialize the memory word at address 298 with operand B8F2 and AC with A937. Determine the contents of AC, DR, PC, AR and IR in hexadecimal after the execution.
20. Modify the machine created in Practical 1 according to the following instruction format:

Instruction format

0	2	3	4	15
Opcode	I	address		

- a. The instruction format contains a 3-bit opcode, a 1-bit addressing mode and a 12-bit address. There are only two addressing modes, I = 0 (direct addressing) and I = 1(indirect addressing).
- b. Create a new register I of 1 bit.

C. Create two new microinstructions as follows:

- i. Check the opcode of instruction to determine type of instruction (Memory Reference/Register Reference/Input-Output) and then jump accordingly.
- ii. Check the I bit to determine the addressing mode and then jump accordingly.

DCA25001**I SEMESTER****DSC 3A: Object Oriented Programming in C++****Credit (L: T: P = 4: 0: 2)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Deliberate the classification and characteristics of Basic of Problem Solving Techniques
- CO2. Understand the details of Basic of Problem Solving Techniques
- CO3. Learn in depth Basic concepts of OOPs and C++ Programming Language
- CO4. Deliberate the details of Program Development using OOPs Concepts
- CO5. Specify in details with application and Use of Polymorphism Concepts
- CO6. Specify the details of implements of Inheritance Using C++

Unit - 1**15 Hours**

Problem solving aspects: Introduction, Problem definition, Problem analysis, Design of problem solution, Algorithm, Flowchart, Coding, Debugging, Types of errors in programming, Program Documentation and Program maintenance.

Techniques of Problem Solving: Flowcharting, decision table, algorithms, structured programming concepts, Programming methodologies viz- top-down and bottom-up programming

Introduction to C++: Concepts of Object-oriented programming, benefits of OOP, Structure of C++ program & Applications of OOP.

Fundamentals: Tokens, Keywords, Identifiers and constants, Basic Data Types, User-defined data types, Derived data Types, Symbolic constants, Declaration of variables.

Operators in C++: Scope resolution operator, Memory management operators, Manipulators, Type cast operator, Expressions and their types.

Unit - 2**15 Hours**

Control structures & Functions: The main function, Function prototyping, Call by Reference, Return by Reference, Inline functions, Function overloading, Friend and Virtual functions.

Object Oriented Concepts: Abstraction, encapsulation, objects, classes, methods, constructors, inheritance, polymorphism, static and dynamic binding, overloading.

Classes and Objects: Specifying a Class, Defining member functions, Making an Outside function Inline, Nesting of member functions, Private member functions, Arrays within a Class, Static data members, Static member functions, Arrays of Objects.

Unit - 3**15 Hours**

Program Development: Object oriented analysis, design, unit testing & debugging, system testing & integration, maintenance.

Constructors and Destructors: Constructors, Parameterized constructors, copy constructor, Dynamic constructor and Destructor.

Unit - 4**15 Hours**

Operator overloading and Type Conversions: Defining operator overloading, Overloading unary operators, Overloading Binary operators, Rules for overloading operators, Type conversions.

Inheritance- introduction, defining derived classes, single inheritance, making a private member inheritable, multilevel inheritance, hierarchical inheritance and hybrid inheritance.

Polymorphism – introduction, pointers, pointers to objects, this pointer, pointers to derived classes, virtual functions, pure virtual functions.

Reference Books:

1. P. K. Sinha&PritiSinha, “Computer Fundamentals”, BPB Publications, 2007.
2. Dr. Anita Goel, Computer Fundamentals, Pearson Education, 2010.
3. Object Oriented Programming with C++ , M.T. Somashekara, D.S. Guru, H.S.
4. Nagendraswamy, K.S. Manjunatha, PHI Learning, New Delhi, 2012
5. Object Oriented Programming with C++ by E. Balagurusamy
6. Richard Johnson, An Introduction to Object-Oriented Application Development, Thomson Learning, 2006
7. B. Stroustrup, the C++ Programming Language, Addison Wesley, 2004.

Object Oriented Programming in C++ Lab**PART-A**

1. PROGRAM TO FIND THE ROOTS OF A QUADRATIC EQUATION
2. PROGRAM TO FIND WHETHER GIVEN NUMBER IS EVEN OR NOT
3. PROGRAM TO FIND LARGEST OF 3 NOS USING NESTED IF
4. PROGRAM TO DISPLAY RAINBOW COLOURS USING SWITCH STATEMENT
5. PROGRAM TO CALCULATE SIMPLE & COMPOUND INTEREST
6. PROGRAM TO FIND MINIMA & MAXIMA IN 1-DIMENSIONAL ARRAY
7. PROGRAM TO GENERATE FIBONACCI SERIES OF A GIVEN NO
8. PROGRAM TO FIND FACTORIAL OF A GIVEN NO
9. PROGRAM TO SEARCH AN ELEMENT IN 1-DIMENSIONAL ARRAY
10. PROGRAM TO DISPLAY N NATURAL NUMBERS & THEIR SUM

PART-B

1. PROGRAM TO SWAP TWO NOS USING CALL BY REFERENCE
2. PROGRAM TO ILLUSTRATE INLINE FUNCTION
3. PROGRAM TO ILLUSTRATE FRIEND FUNCTION
4. PROGRAM TO ILLUSTRATE OPERATOR OVERLOADING
5. PROG PROGRAM TO ILLUSTRATE INLINE FUNCTION
6. PROGRAM TO ILLUSTRATE MULTIPLE INHERITANCE
7. PROGRAM TO CREATE A CLASS CALLED EMPLOYEE. ACCEPT PROGRAM TO ILLUSTRATE INLINE FUNCTION USING ARRAY OF OBJECTS.
8. PROGRAM TO CREATE A CLASS CALLED STUDENT & TO ACCEPT & DISPLAY NECESSARY DETAILS OF A STUDENT USING NESTED CLASS.
9. PROGRAM TO CREATE A CLASS CALLED BANK & ACCEPT CUSTOMER DATA.
10. PROGRAM TO ILLUSTRAT FUNCTION OVER LOADING.

DCB21001

II SEMESTER

DSC 1B: Data Structures and File Processing

Credit (L: T: P = 4: 0: 2)

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Learn the classification, characteristics and understanding of Data structures
- CO2. Specify the details of Searching Techniques
- CO3. Deliberate in details with examples Basic Concepts of Memory Management Techniques
- CO4. Understand in depth File System Operations
- CO5. Specify the characteristics of File Organization Methods
- CO6. Deliberate in details with examples of Storage Devices

Unit – 1**15 Hours**

Basic Data Structures: Introduction, Abstract data structures- stacks, queues, linked lists and binary trees. **Sets:** Dictionary implementation, use of priority queues, hashing, binary trees, balanced trees, sets with merge-find operations.

Unit - 2**15 Hours**

Searching: Internal and external searching, use of hashing and balancing techniques.

Memory Management: Garbage collection algorithms for equal sized blocks, storage allocation for objects with mixed size, buddy systems.

Unit - 3**15 Hours**

Physical Devices: Characteristics of storage devices such as disks and tapes, I/O buffering.

Basic File System Operations: Create, open, close, extend, delete, read-block, write-block, protection mechanisms.

Unit - 4**15 Hours**

File Organizations: Sequential, indexed sequential, direct, inverted, multi-list, directory systems, Indexing using B-tree, B+ tree and their variants, hashing – hash function, collision handling methods, extendible hashing.

Reference Books:

1. M.T. Goodrich, R. Tamassia and D. Mount, Data Structures and Algorithms in C++, John Wiley and Sons, Inc., 2004.
2. T.H. Cormen, C.E. Leiserson, R.L. Rivest and C. Stein, Introduction to Algorithms, 2nd Ed., Prentice-Hall of India, 2006.
3. Robert L. Kruse and A.J. Ryba, Data Structures and Program Design in C++, Prentice Hall, Inc., NJ, 1998.
4. B. Stroustrup, The C++ Programming Language, Addison Wesley, 2004
5. D.E. Knuth, Fundamental Algorithms (Vol. I), Addison Wesley, 1997

Data Structures and File Processing Lab**Part - A**

1. Program to find lower triangular and upper triangular matrices for the given matrix.
2. Write an interactive program to insert an element at the given position and delete an element at the specified position in the given array.
3. Program to search an element identifies the number of occurrences with locations in linear array.
4. Program to sort the given M x N matrix row-wise and column-wise using bubble sorting technique.
5. Write an interactive program to search an element in the given linear array using linear and binary searching technique.
6. Write a program to merge two sorted arrays.

Part – B

7. Write an interactive program to implement the following operations on stack using arrays
 - a. PUSH
 - b. POP
8. Program to implement Tower of Hanoi problem.
9. Write an interactive program to perform insertion and deletion operations in Linear Queue using arrays.
10. Write an interactive program to perform insertion and deletion operations in Circular Queue using arrays.
11. Write an interactive program to insert a node in a linked list at the front, delete a node from the rear and display.
12. Write an interactive program to implement pre order, post order and in order traversal of a binary tree using linked list.

Note: Lecturer May Change the Programs without deviating Theory Paper
DCB23001

II SEMESTER

DSC 2B: Operating Systems with Linux

Credit (L: T: P = 4: 0: 2)

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Learn in details with examples system software
- CO2. Learn the details of Operating System organization
- CO3. Understand the classification and characteristics of Process Management and Scheduling mechanisms
- CO4. Understand in depth Memory Management and allocation strategies
- CO5. Learn in details with examples basic concepts of shell scripting
- CO6. Understand in depth basic Linux environment

Unit - 1

15 Hours

Introduction: System Software, Resource Abstraction, OS strategies. Types of operating systems - Multiprogramming, Batch, Time Sharing, Single user and Multiuser, Process Control & Real Time Systems.

Operating System Organization: Factors in operating system design, basic OS functions, implementation consideration; process modes, methods of requesting system services – system calls and system programs.

Unit - 2

15 Hours

Process Management: System view of the process and resources, initiating the OS, process address space, process abstraction, resource abstraction, process hierarchy, Thread model

Unit - 3

15 Hours

Scheduling: Scheduling Mechanisms, Strategy selection, non-pre-emptive and pre-emptive strategies.

Memory Management: Mapping addresses space to memory space.

Unit - 4

15 Hours

Memory allocation strategies, fixed partition, variable partition, paging, virtual memory

Shell introduction and Shell Scripting: What is shell and various type of shell, Various editors present in Linux, Different modes of operation in vi editor, What is shell script, Writing and executing the shell script, Shell variable (user

defined and system variables), System calls, Using system calls, Pipes and Filters, Decision making in Shell Scripts (If else, switch), Loops in shell, Functions, Utility programs (cut, paste, join, tr , unique utilities), Pattern matching utility (grep)

Reference Books:

1. A Silberschatz, P.B. Galvin, G. Gagne, Operating Systems Concepts, 8th Edition, John Wiley Publications 2008.
2. A.S. Tanenbaum, Modern Operating Systems, 3rd Edition, Pearson Education 2007.
3. G. Nutt, Operating Systems: A Modern Perspective, 2nd Edition Pearson Education 1997.
4. W. Stallings, Operating Systems, Internals & Design Principles, 5th Edition, Prentice Hall of India. 2008.
5. M. Milenkovic, Operating Systems- Concepts and design, Tata McGraw Hill 1992.

Operating Systems with Linux Lab

Software Lab based on Operating Systems

Note: Following exercises can be performed using Linux or UNIX

1. Usage of following commands: ls, pwd, tty, cat, who, who am I, rm, mkdir, rmdir, touch, cd.
2. Usage of following commands: cal, cat (append), cat (concatenate), mv, cp, man, date.
3. Usage of following commands: chmod, grep, tput (clear, highlight), bc.
4. Write a shell script to check if the number entered at the command line is prime or not.
5. Write a shell script to modify “cal” command to display calendars of the specified months.
6. Write a shell script to modify “cal” command to display calendars of the specified range of months.
7. Write a shell script to accept a login name. If not a valid login name display message – “Entered login name is invalid”.
8. Write a shell script to display date in the mm/dd/yy format.
9. Write a shell script to display on the screen sorted output of “who” command along with the total number of users.
10. Write a shell script to display the multiplication table any number,
11. Write a shell script to compare two files and if found equal asks the user to delete the duplicate file.
12. Write a shell script to find the sum of digits of a given number.
13. Write a shell script to merge the contents of three files, sort the contents and then display them page by page.
14. Write a shell script to find the LCD (least common divisor) of two numbers.
15. Write a shell script to perform the tasks of basic calculator.
16. Write a shell script to find the power of a given number.
17. Write a shell script to find the factorial of a given number.
18. Write a shell script to check whether the number is Armstrong or not.
19. Write a shell script to check whether the file have all the permissions or not.
20. Program to show the pyramid of special character “*”.

DCB25001

II SEMESTER**DSC 3B: Programming in JAVA****Credit (L: T: P = 4: 0: 2)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Deliberate in depth java programming fundamental
- CO2. Specify in details with examples Basic java oops Concepts
- CO3. Understand in depth java Interface and packages
- CO4. Deliberate the details of Exception handling in java
- CO5. Identify the classification and characteristics of File handling in java
- CO6. Learn the details of File handling in java
- CO7. Learn the characteristics of Applet Programming

Unit - 1**15 Hours**

Introduction to Java: Features of Java, JDK Environment, Object Oriented Programming Concept Overview of Programming, Paradigm, Classes, Abstraction, Encapsulation, Inheritance, Polymorphism, Difference between C++ and JAVA

Java Programming Fundamental: Structure of java program, Data types, Variables, Operators, Keywords, Naming Convention, Decision Making (if, switch), Looping (for, while), Type Casting

Unit - 2**15 Hours**

Classes and Objects: Creating Classes and objects, Memory allocation for objects, Constructor, Implementation of Inheritance, Implementation of Polymorphism, Method Overloading, Method Overriding, Nested and Inner classes

Arrays and Strings: Arrays, Creating an array, Types of Arrays, String class Methods, String Buffer methods.

Unit - 3**15 Hours**

Abstract Class, Interface and Packages: Modifiers and Access Control, Abstract classes and methods, Interfaces, Packages Concept, Creating user defined packages

Exception Handling: Exception types, Using try catch and multiple catch, Nested try, throw throws and finally, Creating User defined Exceptions.

Unit - 4

15 Hours

File Handling: Byte Stream, Character Stream, File IO Basics, File Operations, Creating file, Reading file, Writing File.

Applet Programming: Introduction, Types Applet, Applet Life cycle, Creating Applet, Applet tag.

Reference Books:

1. Ivan Bayross, Web Enabled Commercial Application Development Using Html, Dhtml, javascript, Perl Cgi , BPB Publications, 2009.
2. Cay Horstmann, BIG Java, Wiley Publication , 3rd Edition., 2009
3. Herbert Schildt , Java 7, The Complete Reference, , 8th Edition, 2009.
4. E Balagurusamy , Programming with JAVA, TMH, 2007

Programming in JAVA Lab

Software Lab based on Java

1. WAP to find the largest of n natural numbers.
2. WAP to find whether a given number is prime or not.
3. Write a menu driven program for following:
 - a. Display a Fibonacci series
 - b. Compute Factorial of a number
 - c. WAP to check whether a given number is odd or even.
 - d. WAP to check whether a given string is palindrome or not.
4. WAP to print the sum and product of digits of an Integer and reverse the Integer.
5. Write a program to create an array of 10 integers. Accept values from the user in that array. Input another number from the user and find out how many numbers are equal to the number passed, how many are greater and how many are less than the number passed.
6. Write a program that will prompt the user for a list of 5 prices. Compute the average of the prices and find out all the prices that are higher than the calculated average.

7. Write a program in java to input N numbers in an array and print out the Armstrong numbers from the set.
8. Write java program for the following matrix operations:
 - a. Addition of two matrices
 - b. Summation of two matrices
 - c. Transpose of a matrix
 - d. Input the elements of matrices from user.
9. Write a java program that computes the area of a circle, rectangle and a Cylinder using function overloading.
10. Write a Java for the implementation of multiple inheritance using interfaces to calculate the area of a rectangle and triangle.
11. Write a java program to create a frame window in an Applet. Display your name, address and qualification in the frame window.
12. Write a java program to draw a line between two coordinates in a window.
13. Write a java program to display the following graphics in an applet window.
 - a. Rectangles
 - b. Circles
 - c. Ellipses
 - d. Arcs
 - e. Polygons
14. Write a program that reads two integer numbers for the variables a and b. If any other character except number (0-9) is entered then the error is caught by NumberFormatException object. After that ex.getMessage () prints the information about the error occurring causes.
15. Write a program for the following string operations:
 - a. Compare two strings
 - b. Concatenate two strings
 - c. Compute length of a string
16. Create a class called Fraction that can be used to represent the ratio of two integers. Include appropriate constructors and methods. If the denominator becomes zero, throw and handle an exception.

DCC21001

III SEMESTER

DSC 1C: Discrete Mathematics & Logic Computation

Credit (L: T: P = 5: 1: 0)

Course Outcome:

After successful completion of the course, the student is able to

CO 1. Identify the classification and characteristics of Set theory

CO 2. Specify in details with examples graph theory

CO 3. Deliberate the classification and characteristics of relation

CO 4. Write down the details of relation

CO 5. Learn the classification and characteristics of function

CO 6. Write down in details with examples function

CO 7. Identify the characteristics of mathematical logic

CO 8. Understand in depth mathematical logic

CO 9. Identify the classification and characteristics of Boolean algebra

CO 10. Identify the details of Boolean algebra

Unit - 1**15 Hours**

Basics of Set Theory: Notation, Inclusion and Equality of Sets, The Power set, Operations on sets, Venn diagram, Set identities, Ordered pairs and Cartesian Products.

Graph Theory: Basic Definitions, Paths and Connectedness, Matrix Representation of Graphs, Trees.

Unit - 2**15 Hours**

Relations and ordering – Properties of binary relations in a Set, Relation Matrix and the Graph of a Relation, Equivalence Relations, Compatibility Relations, Composition of Binary Relation

Unit – 3**15 Hours**

Functions: Definition and Introduction, Composition of Functions, Inverse Functions.

Unit - 4**15 Hours**

Mathematical Logic: Statements and Notation, Connectives, Negation, Conjunction, Disjunction, Statement Formulas and Truth Tables, Conditional

and Bi-conditional, Tautologies, Equivalence of Formulas, Tautological Implications.

Unit - 5**15 Hours**

Boolean Algebra & Formal: Boolean algebra - Application of Boolean Algebra to switching theory. Languages - Recognition and generation - Phase structure grammars and languages – Finite state Machine - Recognition in regular languages.

Reference Books:

1. Discrete Mathematical Structures with Applications to Computer Science by J.P. Tremblay, R Manohar 3rd Edition – Tata McGraw Hill.
2. Discrete mathematical structures by B. Kolman, R.C. Busby and S. Ross, 3rd edition.
3. Introduction to discrete mathematics by Liu, C.L., McGraw Hill, 2nd edition, 1985.
4. Discrete mathematics by S.A. Witala, McGraw Hill, 1987.

DCC23001**III SEMESTER****DSC 2C: Database Management Systems****Credit (L: T: P = 4: 0: 2)****Course Outcome**

After successful completion of the course, the student is able to

CO 1. Understand the characteristics of DBMS with examples

CO 2. Deliberate the details of types of database languages with examples

CO 3. Learn the details of ER- Diagrams and Relationship

CO 4. Understand in depth Basic concepts of Relational Model

CO 5. Learn in details with examples MYSQL Commands

CO 6. Learn in details with examples in PL-SQL

Unit - 1**15 Hours**

Introduction to Database Management Systems: Definition of Data, Information, DBMS, Data base system application, Purpose of database systems, Characteristics of DB – Self describing nature, Insulation between programs, data and data Abstraction (data Independence), support of multiple views of the data, sharing of data and multiples transaction processing, Storage management, Database language – DDL, DML,DCL. File processing system v/s DBMS, Data models, Levels of Abstraction in a DBMS, Three Schema architecture, Characteristics of database approach,, data models, DBMS architecture and data independence.

Unit - 2**15 Hours**

Entity Relationship and Enhanced ER Modelling: Entity types, Entity Sets, Attributes, and Keys, Relationships, Relationship Types, Roles, and Structural Constraints, Weak Entity Types, ER Diagrams, Naming Conventions,SQL99: Schema Definition, constraints, and object modelling

Unit - 3**15 Hours**

Relational Data Model: Basic concepts, Relational Constraints and Relational Database Schemas, Update Operations and Dealing with Constraint Violations, Basic Relational Algebra Operations.

Database design: ER and EER to relational mapping, functional dependencies, normal forms-first normal form, second normal forms. Third normal form BCNF

Unit - 4**15 Hours**

MYSQL (SQL/PL-SQL): SQL VS. SQL * PLUS: SQL COMMANDS AND DATA TYPES, OPERATORS AND EXPRESSIONS, INTRODUCTION TO SQL * PLUS.

Managing Tables and Data:

- CREATING AND ALTERING TABLES (INCLUDING CONSTRAINTS)
- DATA MANIPULATION COMMAND LIKE INSERT, UPDATE, DELETE
- SELECT STATEMENT WITH WHERE, GROUP BY AND HAVING, ORDER BY, DISTINCT, SPECIAL OPERATOR LIKE IN, ANY, ALL BETWEEN, EXISTS, LIKE
- JOIN, BUILT IN FUNCTIONS OTHER DATABASE OBJECTS
- VIEW • SYNONYMS, INDEX TRANSACTION CONTROL STATEMENTS
- COMMIT, ROLLBACK, SAVEPOINT INTRODUCTION TO PL/SQL
- SQL V/S PL/SQL • PL/SQL BLOCK STRUCTURE
- LANGUAGE CONSTRUCT OF PL/SQL (VARIABLES, BASIC AND COMPOSITE DATA TYPE, CONDITIONS LOOPING ETC.)
- % TYPE AND % ROWTYPE
- USING CURSOR (IMPLICIT, EXPLICIT)

Reference Books:

1. R. Elmasri, S.B. Navathe, Fundamentals of Database Systems 6th Edition, Pearson Education, 2010.
2. R. Ramakrishnan, J. Gehrke, Database Management Systems 3rd Edition, McGraw-Hill, 2002.
3. A. Silberschatz, H.F. Korth, S. Sudarshan, Database System Concepts 6th Edition, McGraw Hill, 2010.
4. R. Elmasri, S.B. Navathe Database Systems Models, Languages, Design and application Programming, 6th Edition, Pearson Education, 2013.

Database Management Systems Lab

Software Lab based on Database Management Systems

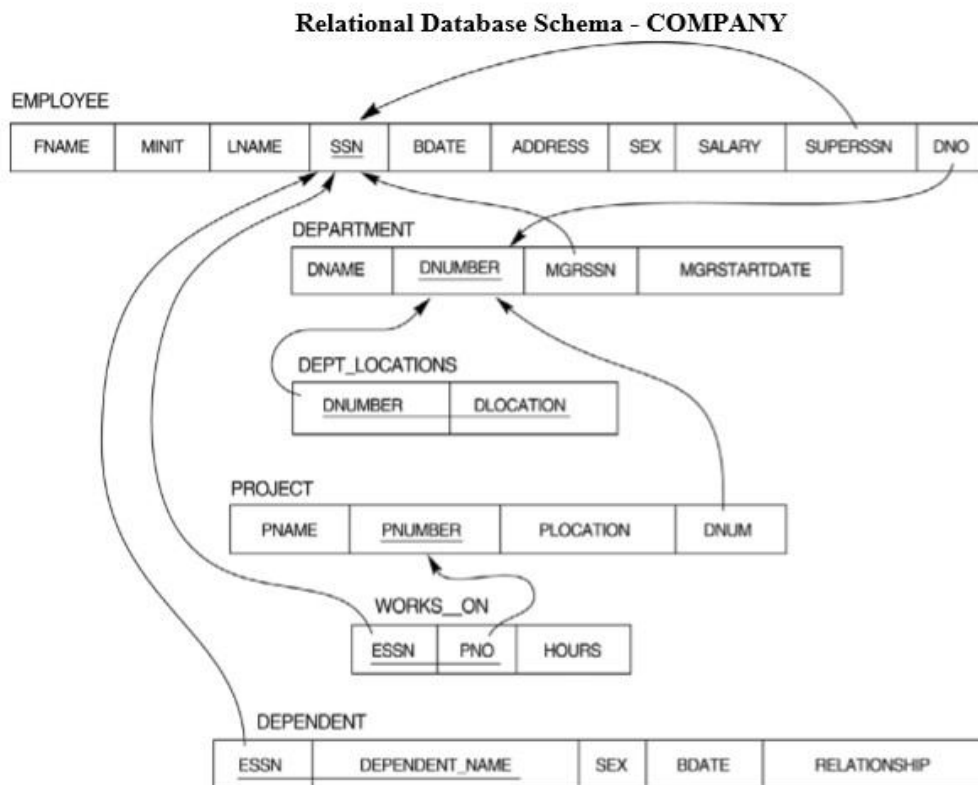
The following concepts must be introduced to the students: **Note:** MS Access / MySQL may be used.

DDL Commands

- Create table, alter table, drop table

DML Commands

- Select, update, delete and insert statements
- Condition specification using Boolean and comparison operators (and, or, not, =, <>, >, <, >=, <=)
- Arithmetic operators and aggregate functions (Count, sum, avg, Min, Max)
- Multiple table queries (join on different and same tables) • Nested select statements
- Set manipulation using (any, in, contains, all, not in, not contains, exists, not exists, union, intersect, minus, etc.)
- Categorization using group by.....having
- Arranging using order by



1. Create tables with relevant foreign key constraints

2. Populate the tables with data
3. Perform the following queries on the database:
 - a. Display all the details of all employees working in the company.
 - b. Display ssn, lname, fname, address of employees who work in department no 7.
 - c. Retrieve the birthdate and address of the employee whose name is 'Franklin T. Wong'
4. Retrieve the name and salary of every employee
5. Retrieve all distinct salary values
6. Retrieve all employee names whose address is in 'Bellaire'
7. Retrieve all employees who were born during the 1950s
8. Retrieve all employees in department 5 whose salary is between 50,000 and 60,000(inclusive)
9. Retrieve the names of all employees who do not have supervisors
10. Retrieve SSN and department name for all employees
11. Retrieve the name and address of all employees who work for the 'Research' department
12. For every project located in 'Stafford', list the project number, the controlling department number, and the department manager's last name, address, and birth date.
13. For each employee, retrieve the employee's name, and the name of his or her immediate supervisor.
14. Retrieve all combinations of Employee Name and Department Name
15. Make a list of all project numbers for projects that involve an employee whose last name is 'Narayan' either as a worker or as a manager of the department that controls the project.
16. Increase the salary of all employees working on the 'ProductX' project by 15%. Retrieve employee name and increased salary of these employees.
17. Retrieve a list of employees and the project name each works in, ordered by the employee's department, and within each department ordered alphabetically by employee first name.
18. Select the names of employees whose salary does not match with salary of any employee in department 10.
19. Retrieve the name of each employee who has a dependent with the same first name and same sex as the employee.
20. Retrieve the employee numbers of all employees who work on project located in Bellaire, Houston, or Stafford.
21. Find the sum of the salaries of all employees, the maximum salary, the minimum salary, and the average salary. Display with proper headings.

22. Find the sum of the salaries and number of employees of all employees of the 'Marketing' department, as well as the maximum salary, the minimum salary, and the average salary in this department
23. Select the names of employees whose salary is greater than the average salary of all employees in department 10.
24. For each department, retrieve the department number, the number of employees in the department, and their average salary.
25. For each project, retrieve the project number, the project name, and the number of employees who work on that project.
26. Change the location and controlling department number for all projects having more than 5 employees to 'Bellaire' and 6 respectively.
27. For each department having more than 10 employees, retrieve the department no, no of employees drawing more than 40,000 as salary.
28. Insert a record in Project table which violates referential integrity constraint with respect to Department number. Now remove the violation by making necessary insertion in the Department table.
29. Delete all dependents of employee whose ssn is '123456789'.
30. Delete an employee from Employee table with ssn = '12345' (make sure that this employee has some dependents, is working on some project, is a manager of some department and is supervising some employees). Check and display the cascading effect on Dependent and Works on table. In Department table MGRSSN should be set to default value and in Employee table SUPERSSN should be set to NULL
31. Perform a query using alter command to drop/add field and a constraint in Employee table.

DCC25001

III SEMESTER

DSC 3C: Web Technologies

Credit (L: T: P = 4: 0: 2)

Course Outcome:

After successful completion of the course, the student is able to

- CO 1. Learn the details of HTML tags
- CO 2. Understand the details of Basic CSS and implements
- CO 3. Understand the details of Basic Concepts of Java Scripts
- CO 4. Write down in details with application and Usage of Java scripts
- CO 5. Understand in details with examples Document object Model
- CO 6. Deliberate in depth Basic of XML

Unit - 1**15 Hours**

Introduction to Web Design: Introduction to HyperTextMarkupLanguage (HTML), header, footer, formatting tags, graphical elements, inserting images, lists, hyperlinks, tables. **Frames**-introduction, frameset. Forms- attributes of forms. Creating web pages

Unit - 2**15 Hours**

Cascading Style Sheets: Introduction, Understanding the Basic CSS syntax, Types of style sheets, multiple sheets, Background properties, Text properties, Font properties, Border properties, Margin properties Padding list & table properties. DIV, SPAN

Unit - 3**15 Hours**

JavaScript: Introduction, Java script in HTML, Java script statement, comments, Expressions, Data types, operators, Conditional statements, Loop statements, functions, Popup boxes ,Array & Boolean Objects Math & Date Objects String & Number Objects, events and event handling & form document object.

Unit - 4**15 Hours**

DOM and XML: Basics of DOM, DOM methods, functions Forms collection, table collections Inner HTML.

XML Introduction; Syntax: Document structure; Document Type definitions; Namespaces; XML schemas; Displaying raw XML documents; Displaying XML documents with CSS; XSLT style sheets.

Reference Books:

1. M.Deitel, P.J.Deitel, A.B.Goldberg: Internet & World Wide Web How to program, 3rd Edition, Pearson Education / PHI, 2004.
2. Chris Bates: Web Programming Building Internet Applications, 3rd Edition, Wiley India, 2006.
3. 3. XueBai et al: The Web Warrior Guide to Web Programming, Thomson, 2003.

Web Technologies Lab

1. Program for formatting tags.
2. Creating a Webpage having Hyperlink.
3. Creating Types of Lists (Ordered, UnOrdered, Definition).
4. Creating a Nested List.
5. Creating a Time Table.
6. Creating a HTML document having vertical frames.
7. Creating Student Application Form.
8. Program to insert audio & video files
9. Creating Internal & External Style Sheets.
10. Program to Margin & Padding.
11. Program to create a Greeting card
12. Program to Image Transparency
13. Program to generate Fibonacci series in Javascript.
14. Program to display Rainbow Colours in Javascript.
15. Program to create Pop-Up Boxes.
16. Program to generate multiplication table.
17. Program to find even and odd numbers.
18. Program to add 2 numbers.
19. Program to find factorial of a numbers.
20. Program to generate 2 different patterns.
21. Program to change background color after 5 sec of page load.
22. Display reverse of a given number.
23. Program to generate random numbers.
24. Program to find the sum of individual numbers.
25. Program to display Book information in XML.

DCD21001

IV SEMESTER
DSC 1D: Numerical Analysis and Statistics
Credit (L: T: P = 4: 0: 2)

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Understand the details of Computer Numerical data and arithmetic
- CO2. Understand the classification and characteristics of Iterative Methods in numerical analysis
- CO3. Deliberate in details with examples Matrices and linear system of Equations
- CO4. Specify in details with examples Interpolation
- CO5. Understand in depth Numerical integration and differentiation
- CO6. Learn the details of Importance and limitations of statistics

Unit - 1 **15 Hours**

Computer Arithmetic: Fixed and Floating point representation, Normalization of numbers. Errors in numbers.

Iterative methods: Bisection method, Regula-Falsi method, Secant method, Newton-Raphson method

Unit - 2 **15 Hours**

Matrices and Linear System of Equations: LU decomposition method, Gauss elimination, Gauss seidal and Gauss Jordan for solving system of equations

Interpolation: Polynomial interpolation, Newton-Gregory forward and backward interpolation, Newton's divided differences interpolation formulae.

Unit - 3 **15 Hours**

Numerical Integration: Trapezoidal rule, Simpson's 1/3rd and 3/8th rule,

Numerical Differentiation: Euler's, modified Euler's and Runge-Kutta (RK) 2nd order and 4th order.

Unit - 4 **15 Hours**

Statistics: Definition, Importance, Functions and Limitations of statistics.

Graphic presentation: Frequency distribution, Histogram, Frequency polygon, frequency curve and O gives Measures of central tendency: (Mean, Median, Mode) Dispersion, Correlation, Regression.

Reference Books:

1. K.E. Atkinson, W. Han, Elementary Numerical Analysis, 3rd Ed., Wiley, 2003.
2. C. Xavier, S.S. Iyengar, Introduction to Parallel Algorithms, Wiley-Interscience, 1998.
3. A. Kharab, R.B. Guenther, An Introduction to Numerical Methods: A MATLAB Approach, 1st Ed. Chapman and Hall/CRC, 2001.
4. B. Bradie, A Friendly Introduction to Numerical Analysis, Pearson Education, 2007.
5. S.R. Otto and J.P. Denier, An Introduction to Programming and Numerical Methods in MATLAB, Springer, 2005.
6. M.K. Jain, S.R.K. Iyengar and R.K. Jain, Numerical Methods for Scientific and Engineering Computation, 7th Ed., New Age International Publishers, 2007.
7. Computer oriented numerical methods by V Rajaraman
8. Statistics Theory and Practice by R S N Pillai, Bagavathi
9. Practical statistics by S P Gupta

Numerical Analysis and Statistics Lab**Software lab based on numerical techniques and statistics**

1. BRUTE FORCE METHOD
2. BISECTION METHOD
3. REGULA –FALSI METHOD
4. NEWTON RAPHSON
5. SECANT
6. LU DECOMPOSITION
7. GAUSS ELIMINATION
8. GAUSS JORDAN
9. GAUSS SEIDAL
10. EULER'S
11. MODIFIED EULER'S
12. RUNGE KUTTA II ORDER
13. RUNGE KUTTA IV ORDER
14. TRAPEZOIDAL
15. SIMPSONS 1/3 RD
16. SIMPSONS 3/8 TH
17. FINDING THE MEAN, MEDIAN AND MODE OF A SET OF DATA
18. FINDING THE RANGE OF A SET OF DATA
19. FINDING THE STANDARD DEVIATION OF A SET OF DATA
20. NEWTONS FORWARD AND BACKWARD INTERPOLATION
21. NEWTONS DIVIDED DIFFERENCE

DCD23001**IV SEMESTER****DSC 2D: J2EE****Credit (L: T: P = 4: 0: 2)****Course Outcome:**

After successful completion of the course, the student is able to

CO 1. Learn the details of Basic elements of J2EE

CO 2. Deliberate the details of Concepts of Multi-Tier Architectures

CO 3. Understand the characteristics of Enterprise Application Strategy

CO 4. Write down in depth Basic Concepts of JDBC

CO 5. Identify in details with examples implementation of SQL Commands Using JDBC objects

CO 6. Learn in details with examples Basic Concepts of Servlet

CO 7. Learn in details with examples Basic Concepts of JSP

Unit - 1**15 Hours**

Interdiction: The ABC of Programming Languages, taking programming languages up a notch, the beginning of java, java byte-code, the advantages of Java, J2EE and J2SE.

J2EE Multi-Tier Architecture: Distributive systems, the Tier, J2EE Multi-Tier Architecture, Client Tier Implementation, Web Tier Implementation, Enterprise JavaBeans Tier Implementation, Enterprise Information Systems Tier Implementation, Challenges.

J2EE Nest Practices: Enterprise Application Strategy, The enterprise application, clients, Sessions Management, Web Tier and Java Server pages, Enterprise Java Beans Tier, The Myth of using inheritance, Maintainable classes, Performance Enhancements, The power of Interfaces, The power of threads, The power of Notification.

Unit - 2**15 Hours**

J2EE Database Concepts: Data, Database, Database Schema, the Art of Indexing.

JDBC Objects: The concept of JDBC, JDBC Driver Types, JDBC packages, A Brief overview of the JDBC Process, Database Connection, Associating the JDBC / ODBC Bridge with the database, Statement Objects, Result Set, Transaction Processing, Metadata.

Unit - 3**15 Hours**

JDBC and Embedded SQL: Model Programs: Model A Program, Model B Program, Tables: Creating a Table, Dropping a Table, Indexing: Creating an Index, Dropping an Index, Inserting Data into tables: Inserting a Row, Inserting the systems date into a column, Inserting the system Time into a column, Inserting a Timestamp into a column, Selecting Data from a Table: Selecting all data from a Table, Requesting one column, Requesting Multiple column, Requesting rows, Requesting rows and columns, AND, OR, and NOT clauses, Joining multiple compound expressions, equal & not equal operators, Less than & greater than operators, Less than equal to & greater than equal to, BETWEEN, LIKE, IS NULL Operator, DISTINCT Modifier, IN modifier,

Unit - 4**15 Hours**

Metadata: Number of columns in result set, Data type of a column, Name of a column, Column Size, updating Tables: Updating a row and column, updating multiple rows, Deleting Data from a table: Deleting a Row from a table, Joining tables, Calculating Data, Grouping and ordering data, sub queries, view.

Java Servlets: Java servlets and common gateway interface programming: Benefits of using a Java servlet, A simple Java servlet, Anatomy of a java servlet: Development Descriptor, Reading Data from a client, Reading HTTP request Headers, Sending Data to a Client & writing the HTTP response Header, Working with cookies, Tracking Sessions, Quick reference guide.

JAVA Server Pages: JSP installation, JSP Tags: Variables & objects, Methods, Control Statements, Loops, Tomcat, Request string: Parsing other information, User sessions, cookies, Session objects, Quick reference guide.

Reference Books:

1. The complete reference J2EE seventh edition - Java 2 Enterprise edition overview
2. J2EE: The complete Reference - McGraw-Hill Education

J2EE Lab

J2EE LAB CYCLE

1. Program to Create Jdbc Connection
2. Application to access the database using the Java Database Connectivity (JDBC).
3. Perform a Database Query and View Results.
4. Write a program to demonstrate Basic Servlet.
5. Write a program to Display request header information.
6. Write a program to design web application
7. Write a program to display cookie value, cookie age and cookie path.
8. Write a program in JSP file to set and then display the cookie.
9. Write a program for Java script validation.
10. Write a JAVA Servlet Program to implement a dynamic HTML using Servlet (user name and password should be accepted using HTML and displayed using a Servlet).
11. Write a JAVA Servlet Program to Download a file and display it on the screen (A link has to be provided in HTML, when the link is clicked corresponding file has to be displayed on Screen)
12. Write a JAVA Servlet Program to implement RequestDispatcher object (use include() and forward() methods)
13. Write a JAVA Servlet Program to implement and demonstrate get() and Postmethods(Using HTTP Servlet Class).
14. Write a JAVA Servlet Program to implement sendRedirect() method(using HTTP Servlet Class).
15. Write a JAVA Servlet Program to implement sessions (Using HTTP Session Interface).
16. Write a JAVA JSP Program to print 10 even and 10 odd number.
17. Write a JAVA JSP Program to implement verification of a particular user login and display a welcome page.
18. Write a JAVA JSP Program to get student information through a HTML and create a JAVA Bean Class, populate Bean and display the same information through another JSP.
19. Write a JAVA JSP Program which uses <jsp:plugin> tag to run a applet
20. Write a JAVA JSP Program which implements nested tags and also uses Tag Support Class.

DCD25001**IV SEMESTER****DSC 3D: Software Engineering and Software Testing****Credit (L: T: P = 4: 0: 2)****Course Outcome:**

After successful completion of the course, the student is able to

- CO 1. Understand in details with examples Concepts of Software process
- CO 2. Specify the details of Software requirements and analysis
- CO 3. Learn in depth Design concepts and principles of software engineering
- CO 4. Understand in depth software Configuration Management and Project Management
- CO 5. Learn in details with examples Software Testings
- CO 6. Specify in depth trends in software engineering

Unit - 1**15 Hours**

Software Process: Introduction, S/W Engineering Paradigm, life cycle models (water fall, incremental, spiral, evolutionary, prototyping, object oriented) , System engineering, computer based system, verification, validation, life cycle process, development process, system engineering hierarchy.

Software requirements: Functional and non-functional, user, system, requirement engineering process, feasibility studies, requirements, elicitation, validation and management, software prototyping, prototyping in the software process, rapid prototyping techniques, user interface prototyping, S/W document.

Unit - 2**15 Hours**

Software Analysis: Analysis and modelling, data, functional and behavioural models, structured analysis and data dictionary.

Design Concepts and Principles: Design process and concepts, modular design, design heuristic, design model and document, Architectural design, software architecture, data design, architectural design, transform and transaction mapping, user interface design, user interface design principles. Real time systems, Real time software design, system design, real time executives, data acquisition system, monitoring and control system.

Unit - 3**15 Hours**

Software Configuration Management: The SCM process, Version control, Change control, Configuration audit, SCM standards.

Software Project Management: Measures and measurements, S/W complexity and science measure, size measure, data and logic structure measure, information flow measure. Estimations for Software Projects, Empirical Estimation Models, Project Scheduling.

Unit - 4**15 Hours**

Testing: Taxonomy of software testing, levels, test activities, types of s/w test, black box testing, testing boundary conditions, structural testing, test coverage criteria based on data flow, mechanisms, regression testing, testing in the large. S/W testing strategies, strategic approach and issues, unit testing, integration testing, validation testing, system testing and debugging.

Trends in Software Engineering: Reverse Engineering and Re-engineering – wrappers – Case Study of CASE tools.

Reference Books:

1. Roger S.Pressman, Software engineering- A practitioner's Approach, McGraw-Hill
2. Ian Sommerville, Software engineering, Pearson education Asia, 6th edition, 2000.
3. PankajJalote- An Integrated Approach to Software Engineering, Springer Verlag, 1997.
4. James F Peters and WitoldPedryez, "Software Engineering – An Engineering Approach", John Wiley and Sons, New Delhi, 2000.
5. Ali Behforooz and Frederick J Hudson, "Software Engineering Fundamentals", Oxford University Press, New Delhi, 1996.
6. Pfleeger, "Software Engineering", Pearson Education India, New Delhi, 1999.
7. Carlo Ghezzi, Mehdi Jazayari and Dino Mandrioli, "Fundamentals of Software Engineering", Prentice Hall of India, New Delhi, 1991.

Software Engineering and Software Testing Lab

Lab based on Software Engineering

1. Practical Title
 - Problem Statement,
 - Process Model
2. Requirement Analysis
 - Creating a Data Flow
 - Data Dictionary,
 - Use Cases
3. Project Management
 - Computing FP
 - Effort
 - Schedule, Risk Table, Timeline chart
4. Design Engineering
 - Architectural Design
 - Data Design, Component Level Design
5. Testing
 - Basis Path Testing

Sample Projects like

- DTC Route Information: Online information about the bus routes and their frequency and fares
- Car Pooling: To maintain a web based intranet application that enables the corporate
- Employees within an organization to avail the facility of carpooling effectively.
- Patient Appointment and Prescription Management System
- Organized Retail Shopping Management Software
- Parking Allocation System
- Wholesale Management System

DCD31001**IV SEMESTER****SEC 1A: Elective: Mathematics****Credit (L: T: P = 3: 1: 0)****Course Outcome:**

After successful completion of the course, the student is able to

- CO 1. Deliberate in details with examples partial fraction
- CO 2. Learn the characteristics of Theory of equation
- CO 3. Understand in details with examples trigonometry
- CO 4. Learn in details with examples complex number
- CO 5. Understand the classification and characteristics of Analytic geometry
- CO 6. Deliberate in details with examples straight lines
- CO 7. Specify in details with examples pair of lines
- CO 8. Specify the classification and characteristics of conics
- CO 9. Understand the classification and characteristics of differentiation
- CO 10. Identify in details with examples integration

Unit - 1**15 Hours**

PARTIAL FRACTIONS: Proper & improper fractions-all four types.

LOGARITHMS: All problems, except common logarithms.

MATHEMATICAL INDUCTION: Simple problems on all types.

THEORY OF EQUATIONS:

- i) Solutions of cubic, bi quadratic equations when complex and irrational roots are given
- ii) Solutions of cubic, bi quadratic equations when roots are in AP, GP and HP.
- iii) Solutions of cubic, bi quadratic equations using synthetic division.
- iv) Operations on complex numbers.

BINOMIAL THEOREM: NO PROOF.

- i) Expansion - problems thereon.
- ii) Finding middle terms.
- iii) Finding constant terms or terms independent of x.

TRIGONOMETRY:

- i) Definition of radian (no proof for constant angle)
- ii) Problems on conversion of radians to degree and vice versa
- iii) Problems on $s = r\theta$, $s = \frac{1}{2} r^2\theta$ (no proofs)

TRIGONOMETRIC FUNCTIONS AND IDENTITIES: Simple problems

GRAPHS OF TRIGONOMETRIC FUNCTIONS: for sine, cos and tan functions.

Allied angles: Problems thereon

Unit - 2**15 Hours****COMPLEX NUMBERS:**

- i) Finding modulus and amplitude of complex numbers
- ii) Solving problems using Domoivre's Theorem.

ANALYTICAL GEOMETRY

- i) Problems on distance formula - Proving parallelogram, square, rhombus, equilateral triangle, Co linearity.
- ii) Problems on section formula - internal division, external division, midpoint formula, centroid of a triangle.
- iii) Problems on area of a triangle.

Unit - 3**15 Hours****STRAIGHT LINES:**

- i) By finding slopes - show that lines are parallel and perpendicular.
- ii) Finding slopes - when two points are given
- iii) Equation of straight lines - passing through given point, parallel and perpendicular to given line.
- iv) Problems on intercept form, slope form, normal form, two point form.
- v) Problems on angle between two lines.
- vi) Concurrency of three lines and point of concurrency.

PAIR OF LINES

- ii) Angle between two lines $ax^2 + 2hxy + by^2 = 0$. $ax^2 + 2hxy + by^2 + 2gx + 2fy + C = 0$.
- iii) Point of Intersection.
- iv) Condition for an equation to represent pair of lines.

CIRCLES:

- i) Finding center and radius.
- ii) Finding equation of a circle passing through three points, when different conditions are given, passing through x and y- axis.

CONICS: PARABOLA -

Finding vertex, focus, tangent, normal, length of latus rectum, eccentricity. (No proofs)

Unit - 4**15 Hours****LIMITS AND CONTINUITY:**

- i) Simple direct problems on limits of the form $\left(\frac{x^n - a^n}{x - a}\right), \frac{\sin \theta}{\theta}, \frac{\tan \theta}{\theta}$ (no determinate forms).
- ii) Simple problems on continuity.

DIFFERENTIATION:

- i) Problems on sum, product, quotient, chain rule (No parametric, logarithmic functions)

DIFFERENTIAL EQUATIONS:

Solving problems by variable separable form.

INTEGRATION:

- i) By substitution
 ii) By parts
 iii) By partial fractions
 iv) Problems of types

$$\int \frac{dx}{a^2 + x^2}, \int \frac{dx}{a^2 - x^2}, \int \frac{dx}{x^2 - a^2}, \int \frac{dx}{\sqrt{a^2 - x^2}}, \int \frac{dx}{\sqrt{a^2 + x^2}}, \int \frac{dx}{\sqrt{x^2 - a^2}}, \int \frac{dx}{x\sqrt{x^2 - a^2}}, \int \frac{dx}{x\sqrt{x^2 + a^2}}$$

$$\int \sqrt{a^2 - x^2} dx, \int \sqrt{a^2 + x^2} dx, \int \sqrt{x^2 - a^2} dx.$$

$$\int \frac{dx}{ax^2 + bx + c}, \int \frac{dx}{\sqrt{ax^2 + bx + c}}$$

APPLICATION OF INTEGRATION:

Simple problems on area

- i) Find the area of circle, ellipse, parabola & the ordinate $x=a$ by integration
 ii) Find the area bounded by the parabola $y^2 = 4ax$, x - axis and $x = 1$, $x = y$.
 Find the area bounded by $y = \sin x$, x - axis & $x = 0$, $y = z$.

Reference Books:

1. Theory and Problems in Mathematics – I by BOSCO Publications 2004.
2. Theory and Problems in Mathematics – II by BOSCO Publications 2005.
3. Engineering Mathematics, Volumes I–IV by S Chandrasekhar.

DCD33501**IV SEMESTER****SEC 1B: Elective: Business Mathematics****Credit (L: T: P = 3: 1: 0)****Course Outcome:**

After successful completion of the course, the student is able to

CO 1. Specify the characteristics of matrices and determinants

CO 2. Understand the details of straight lines

CO 3. Write down in details with examples matrices and determinants

CO 4. Deliberate the characteristics of algebra

CO 5. Learn the classification and characteristics of permutation and combination

CO 6. Deliberate in details with examples mathematical induction

CO 7. Deliberate the characteristics of sequence and series

CO 8. Understand the classification and characteristics of progression

CO 9. Learn in details with examples compound interest

CO 10. Deliberate the classification and characteristics of coordinate geometry

Unit - 1**15 Hours**

Matrices and Determinants: Order - Types of matrices - Addition and subtraction of matrices and Multiplication of a matrix by a scalar - Product of matrices. Evaluation of determinants of order two and three - Properties of determinants (Statements only) - Singular and non singular matrices - Product of two determinants.

Unit - 2**15 Hours**

Algebra: Partial fractions - Linear non repeated and repeated factors - Quadratic non repeated types. Permutations - Applications - Permutation of repeated objects - Circular permutation.

Combinations - Applications - Mathematical induction - Summation of series using $\sum n$, $\sum n^2$ and $\sum n^3$. Binomial theorem for a positive integral index - Binomial coefficients.

Unit - 3**15 Hours**

Sequences and series: Harmonic progression - Means of two positive real numbers - Relation between A.M., G.M., and H.M. - Sequences in general - Specifying a sequence by a rule and by a recursive relation - Compound interest - Nominal rate and effective rate - Annuities - immediate and due.

Unit - 4**15 Hours**

Ordinate Geometry: Rectangular castes ion Co–ordinates in a Plane, Equations of straight lines and the concept of gradient with its practical applications in real life business problems.

Reference Books:

1. Mathematics for Economics and Business by R.S. Bhardwaj .
2. Business Mathematics by PadmalochanHazarika.
3. Business Mathematics by D.C. Sancheti and V.K. Kapoor.
4. Mathematical Economics by Dowling, T. Edword.
5. Mathematical Analysis for Economics by Allen, RGD.

DCD37001**IV SEMESTER****SEC 1C: Elective: Accountancy & Financial Management****Credit (L: T: P = 3: 1: 0)****Course Outcome:**

After successful completion of the course, the student is able to

- CO 1. Deliberate the details of Basic Accounting Concepts
- CO 2. Specify in depth Examples of Accounting
- CO 3. Deliberate the details of Meaning and Scope of Financial Management
- CO 4. Learn in depth Functions of Financial Manager
- CO 5. Learn the characteristics of Ratio Analysis
- CO 6. Understand the detail Concepts of Costing
- CO 7. Learn in details with examples Budgetary Control

Unit - 1**15 Hours****ACCOUNTING:**

1. Introduction: Principles, concepts and conventions, double entry system of Accounting, ledger keeping.
2. Subsidiary books with special reference to simple cash book and three columns cash book.
3. Trial balance and final accounts of sole trader: Preparation trial balance, adjusting entries, including revenue for bad debts, revenue for discount on debtors and creditors, preparation of final accounts.

Unit - 2**15 Hours****FINANCIAL MANAGEMENT:**

4. Final accounts of joint stock companies.
5. Introduction: Meaning and scope of financial management, functions of the financial manager.
6. Ratio analysis: Meaning of ratio, advantages, limitations, types of ratios and their usefulness, liquidity and ratios, profitability ratios, efficiency ratios, solvency ratios, problems including preparation of balance sheet.

Unit - 3**15 Hours****COSTING & BUDGETARY CONTROL:**

7. Funds flow statement: Meaning and concepts of funds, preparation of fund flow statement.
8. Unit costing: Preparation of cost sheet and tender price statement.

9. Marginal costing: Concepts, Marginal cost equations, P/V ratio, B.E.P., Margin of safety, Sales to earn a desired profit, Problems on the above.

Unit - 4**15 Hours**

10. Budgetary Control: Meaning and definition, preparation of flexible budget and cash budget.
11. Standard costing: Meaning of standard cost and standard costing, analysis of variances – material and labour variances only.

Reference Books:

1. Accountancy Vol. 1 by B.S. Raman.
2. Accountancy Vol. 2 by B.S. Raman.
3. Management Accounting by R.K. Sharma and Gupta.
4. Financial Management by I.M. Pandey.

DCD39001**IV SEMESTER****SEC 1D: Elective: Entrepreneurship Development****Credit (L: T: P = 3: 1: 0)****Course Outcome**

After successful completion of the course, the student is able to

- CO1. Specify the characteristics of Entrepreneurship
- CO2. Deliberate the details of identification of opportunities
- CO3. Understand in depth Feasibility and financial management of the Project
- CO4. Write down the details of Project Report
- CO5. Learn the characteristics of SWOT Techniques
- CO6. Learn the details of Enterprise Rules and regulations

Unit - 1**15 Hours**

Need scope and characteristics of entrepreneurships special schemes for a technical entrepreneur STED.

Identification of opportunities: Exposure to Demand based, Resource based, Service based important substitutes and expert promotion industries. Market surveys techniques.

Need scope and approaches for the project formulation.

Criteria for the principles of products selection and development.

Unit - 2**15 Hours**

Structure of the project report, Choice of technology, plant and equipment.

Institutions financing procedure and financial incentives. financial ratios and their significance. Books of accounts, financial statements and funds flow analysis.

Resource management, man machine and material. The critical path method (CPM) and project evolution, review techniques (PERT).

Planning tools for establishing SSI

- a. CREATIVITY AND INNOVATION b. PROBLEM SOLVING APPROACH

- c. STRENGTH AND WEAKNESS OPPORTUNITY AND THREAT (SWOT) techniques.

Unit - 3**15 Hours**

Techno economics feasibility of the project. Plan layout and process planning for the product. Quality control / quality assurances and testing of the products. Costing and pricing.

Management of self and understanding human behavior. Sickness in small-scale industries and their remedial measures.

Unit - 4**15 Hours**

Coping and uncertainties, stress management, and positive reinforcement

- a. Licensing registration, b. Municipal by laws and insurance coverage
Important provision of factory acts sales of goods act partnership act.
- a. DILUTION CONTROL,
- b. SOCIAL RESPONSIBILITY AND BUSINESS ETHIQUES
Income tax, sales tax and excise rule.

Reference Books:

1. Entrepreneurship Development – Kanaka
2. Entrepreneurship Development – VasanthDesa

DCE21001**V SEMESTER****DSE 1A: Elective: Data Communication and Computer Networks****Credit (L: T: P = 4: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Learn in depth Elements of Data Communications and network Systems
- CO2. Learn in depth Transmission Media
- CO3. Understanding the various classifications and characteristics of Signals
- CO4. Understand in details with examples Network Models
- CO5. Learn in depth Error Detection and Corrections Algorithms
- CO6. Deliberate in details with examples Switching Concepts
- CO7. Deliberate the classification and characteristics of networking and internetworking Devices

Unit - 1**15 Hours**

Data Communications: Components, Data Representation, Dataflow

Networks: Network criteria, Physical Structures, Topology (Mesh, Star, Tree, Bus, Ring, Hybrid)

Categories of Network: LAN, MAN, WAN

History of Network: Protocols and Standards: Protocols (Standards organization),

Addressing: Physical, Logical, Port, Specific.

Unit - 2**15 Hours**

Transmission Media: Guided Media – Twisted pair cable, co-axial cable, optical fiber, Unguided Media – Radio waves, microwaves, Infrared.

Signals: Analog and Digital Data, Analog and Digital Signals, periodic and non periodic signals. Analog Signals – Sine wave, Peak Amplitude, Period and frequency, Phase, wave length, composite signals. Digital Signals – Band width, Bit length, Bit rate, base band transmission, Digital v/s Analog. Transmission Impairment, Data rate limits (Noisy and noiseless channel)

Unit - 3**15 Hours**

Network Models: Layered tasks, OSI model (peer – to – peer), Layered Architecture. Functions of Layers (OSI), TCP / IP Protocol suite

Multiplexing: FDM (MUX and DEMUX process, Application of FDM), WDM, TDM (Interleaving, synchronizing, bit padding)

Switching Concept: Working principle of circuit switching and packet switching. Circuit switched networks, three phases efficiency, delay. Data grams network, routing table, delay efficiency, virtual.

Error Detection and Correction: Types of Errors, Redundancy, Error detection virus Error Correction.

Error Detection: Parity check, Cyclic Redundancy Check (CRC), Check Sum. Error Correction - Retransmission, Forward Error Correction, Burst error Correction.

Unit - 4

15 Hours

Networking And Internetworking Devices: Connecting Devices - Hubs, Repeaters, Amplifiers, Bridges – LAN bridges, transparent bridges, Source-route bridges, Routers, Gateways, 2 layer and 3 layers switches.

Routing Concepts: Types, Shortest path, flooding.

Wireless Lan's: Blue tooth - Architecture, Blue tooth layers.

Network Layer : IPV4, IPV6 addresses

Transport Layer: UDP – user datagram, operations, Application. TCP - Services, TCP segment, SCTP - Services, packet format.

Application Layer: - SMTP, SNMP, HTTP, FTP

Reference Books:

1. Data Communication and Networking – Forouzan
2. Computer Network – Tanenbaurn – 3rd Editions
3. Computer Network – Larry L. Peterson & Bruce S. Davie

Data Communication and Computer Networks Lab

1. Program for Identifying well known Ports
2. Program for Data Retrieval from Remote Database.
3. Program for Simulating SMTP Client.
4. Program for Simulating Telnet Client
5. Program for Simple file transfer between two systems, (without using Protocols)
6. Program for implementing HTTP.
7. Program for Downloading Image files.
8. Simulate Checksum Algorithm.
9. Simulate Stop & Wait Protocol.
10. Simulate Go-Back-N Protocol.
11. Simulate Selective Repeat Protocol.
12. Take an example subnet of hosts. Obtain broadcast tree for it.
13. Network address with automatic subnet address generation:

DCE23001**V SEMESTER****DSE 1B: Elective: Computer Graphics****Credit (L: T: P = 4: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Learn the classification and characteristics of Elements of Graphics Systems
- CO2. Learn in depth Graphics Algorithms
- CO3. Deliberate the classification and characteristics of 2D Graphics
- CO4. Understand the characteristics of 3D Graphics
- CO5. Deliberate the details of Transformation and Viewing Techniques
- CO6. Learn the details of Illumination and Colour Models

Unit - 1**15 Hours**

INTRODUCTION: Survey of computer graphics, Overview of graphics systems – Video display devices, Raster scan systems, Random scan systems, Graphics monitors and Workstations, Input devices, Hard copy Devices, Graphics Software; Output primitives – points and lines, line drawing algorithms, loading the frame buffer, line function; circle and ellipse generating algorithms; Pixel addressing and object geometry, filled area primitives.

Unit - 2**15 Hours**

TWO DIMENSIONAL GRAPHICS: Two dimensional geometric transformations – Matrix representations and homogeneous coordinates, composite transformations; Two dimensional viewing – viewing pipeline, viewing coordinate reference frame; widow-to-viewport coordinate transformation, Two dimensional viewing functions; clipping operations – point, line, and polygon clipping algorithms.

Unit - 3**15 Hours**

THREE DIMENSIONAL GRAPHICS: Three dimensional concepts; Three dimensional object representations – Polygon surfaces Polygon tables- Plane equations – Polygon meshes; Curved Lines and surfaces, Quadratic surfaces; Blobby objects; Spline representations – Bezier curves and surfaces -B-Spline curves and surfaces.

TRANSFORMATION AND VIEWING: Three dimensional geometric and modeling transformations – Translation, Rotation, Scaling, composite transformations; three dimensional viewing – viewing pipeline, viewing coordinates, Projections, Clipping; Visible surface detection methods.

Unit - 4**15 Hours**

ILLUMINATION AND COLOUR MODELS: Light sources – basic illumination models – halftone patterns and dithering techniques; Properties of light – Standard primaries and chromaticity diagram; Intuitive colour concepts – RGB colour model – YIQ colour model – CMY colour model – HSV colour model – HLS colour model; Colour selection.

Reference Books:

1. Computer Graphics C Version by Donald Hearn & M. Pauline Baker Pearson Education, New Delhi, 2004
2. Procedural Elements for Computer Graphics by David F. Rogers, Tata McGraw Hill Book Company, New Delhi, 2003
3. Computer Graphics: Principles & Practice in C by J. D. Foley, S. K Feiner, A Van Dam F. H John, Pearson Education, 2004
4. Computer Graphics using Open GL by Francis S Hill Jr Pearson Education, 2004.

Computer Graphics Lab

1. Implementation of Bresenham's Algorithm – Line, Circle, Ellipse.
2. Implementation of Line, Circle and ellipse Attributes
3. Two Dimensional transformations - Translation, Rotation, Scaling, Reflection, Shear.
4. Composite 2D Transformations
5. Cohen Sutherland 2D line clipping and Windowing
6. Sutherland – Hodgeman Polygon clipping Algorithm
7. Three dimensional transformations - Translation, Rotation, Scaling
8. Composite 3D transformations
9. Drawing three dimensional objects and Scenes
10. Generating Fractal images

DCE25001**V SEMESTER****DSE 1C: Elective: Multimedia Systems and Applications****Credit (L: T: P = 4: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Understand the details of Components of Multimedia with applications
- CO2. Identify in details with examples Text, Images, Sound and Videos
- CO3. Learn in depth Animation Techniques
- CO4. Understand the details of Multimedia in internet
- CO5. Deliberate the characteristics of Making Multimedia
- CO6. Deliberate in depth Multimedia Making Tools

Unit - 1**15 Hours**

Multimedia: Introduction to multimedia, components, uses of multimedia, multimedia applications, virtual reality.

Text: Fonts & Faces, Using Text in Multimedia, Font Editing & Design Tools, Hypermedia & Hypertext.

Images: Still Images – bitmaps, vector drawing, 3D drawing & rendering, natural light & colors, computerized colors, color palettes, image file formats.

Unit - 2**15 Hours**

Sound: Digital Audio, MIDI Audio, MIDI vs Digital Audio, Audio File Formats.

Video: How video works, analog video, digital video, video file formats, video shooting and editing.

Unit - 3**15 Hours**

Animation: Principle of animations, animation techniques, animation file formats.

Internet and Multimedia: www and HTML, multimedia on the web – web servers, web browsers, web page makers and site builders.

Unit - 4**15 Hours**

Making Multimedia: Stages of a multimedia project, Requirements to make good multimedia, Multimedia Hardware - Macintosh and Windows production

Platforms, Hardware peripherals - Connections, Memory and storage devices, Multimedia software and Authoring tools.

Reference Books:

1. Tay Vaughan, “Multimedia: Making it work”, TMH, Eighth edition.
2. Ralf Steinmetz and KlaraNaharstedt, “Multimedia: Computing, Communications Applications”, Pearson.
3. Keyes, “Multimedia Handbook”, TMH.
4. K. Andleigh and K. Thakkar, “Multimedia System Design”, PHI.

Multimedia Systems and Applications Lab

Practical exercises based on concepts listed in theory using Presentation tools in office automation tool / GIMP / Blender / Audacity / Animation Tools / Image Editors / Video Editors.

Implement the followings using Blender -

1. Create an animation using the tools panel and the properties panel to draw the following – Line, pe , oval, circle, rectangle , square, pencil , brush , lasso tool
2. Create an animation using text tool to set the font , size , color etc.
3. Create an animation using Free transform tool that should use followings-
 - Move Objects
 - Skew Objects
 - Stretch Objects
 - Rotate Objects
 - Stretch Objects while maintaining proportion
 - Rotate Objects after relocating the center dot
4. Create an animation using layers having following features- Insert layer, Delete layer, guide layer, Mask layer.
5. Modify the document (changing background color etc.)using the following tools
 - Eraser tool
 - Hand tool
 - Ink bottle tool
 - Zoom tool
 - Paint Bucket tool
 - Eyedropper tool
6. Create an animation for bus car race in which both starts from the same point and car wins the race.
7. Create an animation in which text Hello gets converted into GoodBye (using motion/shape tweening).
8. Create an animation having five images having fade-in fade-out effect.
9. Create an scene to show the sunrise (using multiple layers and motion tweening)
10. Create an animation to show the ripple effect.
11. Create an animation (using Shape tweening and shape hints) for transforming one shape into another.
12. Create an animation for bouncing ball (you may use motion guide layer).

OR

Project :

Design a minimum 10 page interactive website using Joomla or WordPress.

DCE31001**V SEMESTER
DSE 2A: Elective: ASP.Net
Credit (L: T: P = 4: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO 1. Learn the details of ASP.NET Framework
- CO 2. Learn the details of ASP.NET working Environment
- CO 3. Deliberate in details with examples Standard Control Of Asp.NET
- CO 4. Understand the details of Developing Simple Websites Using Asp.net Controls
- CO 5. Deliberate in depth Developing Simple Web Application Using Asp.net Controls
- CO 6. Learn the details of Database Access Controls
- CO 7. Identify in details with examples Database Access Controls

Unit - 1**15 Hours**

Overview of the ASP.NET: Introduction of different Web Technology, What is Asp.Net, How Asp.Net Works, Use of visual studio, Different Languages used in ASP.Net. Summary.

Framework: Common Language Runtime (CLR), .NET Framework Class Library, Summary

Unit - 2**15 Hours**

Setting up and Installing ASP.NET: Installing Internet Information Server, Installation of Asp.Net, virtual directory, Application Setting in IIS, Summary.

Unit - 3**15 Hours**

Asp.Net Standard Controls, Displaying information, Label Controls, Literal Controls, Bulleted List, Accepting User Input, Textbox controls, Radio Button and Radio Button List Controls, Checkbox and Checkbox List Controls, Button controls, Link Button Control, Image Button Control, Using Hyperlink Control, Dropdown List, List Box, Displaying Images, Image Control, Image Map Control, Using Panel Control, Using Hyperlink Control, Asp.Net, Page & State Management, Overview of events in page, Summary.

Unit - 4**15 Hours**

Designing Websites with master pages, creating master pages, Creating default contents, nesting master pages, registering master pages in web configuration, Summary.

ASP.Net Theme: ASP.NET Website Theme, Named Skin and Default Skin In ASP.NET Theme, Style Sheet Theme and Theme Attributes Of A Page Directive

Using the Rich Controls: Accepting File Uploads, Saving files to file system, Calendar Control, Displaying advertisements, Displaying Different Page view, Displaying a Tabbed Page View, Wizard Control, Summary.

Reference Books:

1. Mathew Mac Donald, ASP . Net The Complete Reference, McGraw –Hill, 2002.

ASP . Net Lab

LAB MANUAL:

1. Write a Program to generate the factorial operation.
2. Write a Program to perform Money Conversion.
3. Write a Program to generate the Quadratic Equation.
4. Write a Program to generate the Login control.
5. Write a Program to perform Asp.Net state.
6. Write a Program to perform validation operation.
7. Write a Program to perform Tree view operation.
8. Write a Program to display the phone no of an author using database.
9. Write a Program to insert the data in to database using Execute-Non Query.
10. Write a Program to bind data using template in data list.
11. Write a Program to bind data using Hyperlink column in data grid.

DCE35001**V SEMESTER****DSE 2B: Elective: Visual Programming****Credit (L: T: P = 4: 0: 1)**

{**Note:** Use any open source alternative such as Tkinter with Python /Sharp Develop/GAMBAS/OPENXAVA with JAVA}

Course Outcome:

After successful completion of the course, the student is able to

CO 1. Learn in details with examples Basic concept Of GUI Environment

CO 2. Deliberate the details of GUI Controls

CO 3. Learn in details with examples Data types and Operations in Visual Programming

CO 4. Learn in details with examples Control statements in Visual Programming

CO 5. Write down in details with examples Modular Programming

CO 6. Learn the details of Forms Handling in Visual Programming

CO 7. Understand in depth Database Connectivity in Visual Programming

Unit - 1**15 Hours**

GUI Environment: Introduction to graphical user interface (GUI), programming language (procedural, object oriented, event driven), the GUI environment, compiling, debugging, and running the programs.

Controls: Introduction to controls textboxes, frames, check boxes, option buttons, images, setting borders and styles, the shape control, the line control, working with multiple controls and their properties, designing the user interface, keyboard access, tab controls, default & cancel property, coding for controls.

Operations: Data types, constants, named & intrinsic, declaring variables, scope of variables, val function, arithmetic operations, formatting data.

Unit - 2**15 Hours**

Decision Making: If statement, comparing strings, compound conditions (and, or, not), nested if statements, case structure, using if statements with option buttons & check boxes, displaying message in message box, testing whether input is valid or not.

Modular programming: Menus, sub-procedures and sub-functions defining / creating and modifying a menu, using common dialog box, creating a new sub-procedure, passing variables to procedures, passing argument by value or by reference, writing a function/ procedure.

Unit - 3**15 Hours**

Forms Handling: Multiple forms creating, adding, removing forms in project, hide, show method, load, unload statement, me keyword, referring to objects on a different forms

Iteration Handling: Do/loops, for/next loops, using msgbox function, using string function

Arrays and Grouped Data Control: Arrays - 1-dimension arrays, initializing an array using for each, user-defined data types, accessing information with user-defined data types, using list boxes with array, two dimensional arrays.

Unit - 4**15 Hours**

lists, loops and printing list boxes & combo boxes, filling the list using property window / add item method, clear method, list box properties, removing an item from a list, list box/ combo box operations.

Database Connectivity: Database connectivity of forms with back end tool like mysql, populating the data in text boxes, list boxes etc. searching of data in database using forms. Updating/ editing of data based on a criterion.

Reference Books:

1. Reference: Programming in Visual Basic 6.0 by Julia Case Bradley, Anita C. Millispangh (Tata Mcgraw Hill Edition 2000 (Fourteenth Reprint 2004))

Visual Programming Lab

1. Print a table of numbers from 5 to 15 and their squares and Cubes.
2. Print the largest of three numbers.
3. Find the factorial of a number n.
4. Enter a list of positive numbers terminated by zero. Find the sum and average of these numbers.
5. A person deposits Rs. 1000 in a fixed account yielding 5% interest. Complete the amount in the account at the end of each year for n years.
6. Read n numbers. Count the number of negative numbers, positive numbers and zeros in the list.
7. Read n numbers. Count the number of negative numbers, positive numbers and zeroes in the list. use arrays.
8. Read a single dimension array. Find the sum and average of these numbers.
9. Read a two dimension array. Find the sum of two 2D Array.
10. Create a database Employee and Make a form to allow data entry to Employee Form with the following command buttons:

DCE37001**V SEMESTER****DSE 3C: Elective: Artificial Intelligence and Expert Systems****Credit (L: T: P = 4: 0: 1)****Course Outcome**

After successful completion of the course, the student is able to

- CO1. Deliberate in details with examples Artificial intelligence system
- CO2. Learn the characteristics of Concepts of Representation of knowledge
- CO3. Understand in details with examples Concepts of Representation of knowledge
- CO4. Understand the details of knowledge inference methods
- CO5. Understand in details with examples Concepts of Machine Learning Techniques
- CO6. Learn the details of Expert System

Unit - 1**15 Hours**

INTRODUCTION TO AI AND PRODUCTION SYSTEMS: Introduction to AI-Problem formulation, Problem Definition -Production systems, Control strategies, Search strategies. Problem characteristics, Production system characteristics -Specialized production system- Problem solving methods - Problem graphs, Matching, Indexing and Heuristic functions -Hill Climbing-Depth first and Breadth first, Constraints satisfaction - Related algorithms, Measure of performance and analysis of search algorithms.

REPRESENTATION OF KNOWLEDGE: Game playing - Knowledge representation, Knowledge representation using Predicate logic,

Unit - 2**15 Hours**

Introduction to predicate calculus, Resolution, Use of predicate calculus, Knowledge representation using other logic-Structured representation of knowledge.

KNOWLEDGE INFERENCE: Knowledge representation -Production based system, Frame based system. Inference - Backward chaining, Forward chaining, Rule value approach, Fuzzy reasoning - Certainty factors, Bayesian Theory-Bayesian Network-Dempster -Shafer theory.

Unit - 3**15 Hours**

PLANNING AND MACHINE LEARNING: Basic plan generation systems - Strips -Advanced plan generation systems - Kstrips -Strategic explanations - Why, Why not and how explanations. Learning- Machine learning, adaptive Learning.

Unit - 4**15 Hours**

EXPERT SYSTEMS: Expert systems - Architecture of expert systems, Roles of expert systems - Knowledge Acquisition –Meta knowledge, Heuristics. Typical expert systems - MYCIN, DART, XOON, Expert systems shells.

Reference Books:

1. Kevin Night, Elaine Rich, Nair B., “Artificial Intelligence (SIE)”, McGraw Hill- 2008. (Unit-1,2,4,5)
2. Dan W. Patterson, “Introduction to AI and ES”, Pearson Education, 2007. (Unit-III)
3. Peter Jackson, “Introduction to Expert Systems”, 3rd Edition, Pearson Education, 2007.
4. Stuart Russel, Peter Norvig “AI – A Modern Approach”, 2nd Edition, Pearson Education 2007.

Artificial Intelligence and Expert Systems Lab

1. Implement Breadth First Search (for 8 puzzle problem or Water Jug problem or any AI search problem)
2. Implement Depth First Search (for 8 puzzle problem or Water Jug problem or any AI search problem)
3. Implement Best First Search (for 8 puzzle problem or Water Jug problem or any AI search problem)
4. Implement Single Player Game (Using Heuristic Function)
5. Implement Two Player Game (Using Heuristic Function)
6. Implement A* Algorithm
7. Implement Propositional calculus related problem
8. Implement First order propositional calculus related problem
9. Implement Certainty Factor problem
10. Implement Syntax Checking of English sentences-English Grammar
11. Develop an Expert system for Medical diagnosis.
12. Develop any Rule based system for an application of your choice.

DCE41001**V SEMESTER****DSE 3A: Cloud Computing****Credit (L: T: P = 4: 0: 1)****Course Outcome**

After successful completion of the course, the student is able to

CO1. Learn in depth Fundamentals of Cloud Computing

CO2. Understand the details of Cloud Services and File System

CO3. Learn in depth Concept of Collaborating with Cloud

CO4. Understand the details of Virtualization in cloud

CO5. Learn the classification and characteristics of Security challenges in Cloud Computing

CO6. Specify the classification and characteristics of Security challenges in Cloud Computing

CO7. Understand the details of Security challenges in Cloud Computing

CO8. Understand the Common standards of Cloud Computing

CO9. Deliberate in details with examples Various Application of Cloud Computing

Unit - 1**15 Hours**

Cloud Introduction: Cloud Computing Fundamentals: Cloud Computing definition, Types of cloud, Cloud services: Benefits and challenges of cloud computing, Evolution of Cloud Computing , usage scenarios and Applications , Business models around Cloud– Major Players in Cloud Computing - Issues in Cloud - Eucalyptus - Nimbus - Open Nebula, CloudSim.

Cloud Services and File System: Types of Cloud services: Software as a Service - Platform as a Service – Infrastructure as a Service - Database as a Service - Monitoring as a Service –Communication as services.

Unit - 2**15 Hours**

Service providers- Google App Engine, Amazon EC2, Microsoft Azure, Sales force.

Collaborating With Cloud: Collaborating on Calendars, Schedules and Task Management – Collaborating on Event Management, Contact Management, Project Management –Collaborating on Word Processing , Databases – Storing and Sharing Files- Collaborating via Web-Based Communication Tools – Evaluating Web Mail Services – Collaborating via Social Networks – Collaborating via Blogs and Wikis.

Unit - 3**15 Hours**

Virtualization For Cloud: Need for Virtualization – Pros and cons of Virtualization – Types of Virtualization – System Vm, Process VM, Virtual Machine monitor – Virtual machine properties - Interpretation and binary translation, HLL VM - Hypervisors – Xen, KVM , VMWare, Virtual Box, Hyper-V.

Unit - 4**15 Hours**

Security, Standards, And Applications: Security in Clouds: Cloud security challenges – Software as a Service Security, Common Standards: The Open Cloud Consortium –The Distributed management Task Force – Standards for application Developers – Standards for Messaging – Standards for Security, End user access to cloud computing, Mobile Internet devices and the cloud.

Reference Books:

1. Bloor R., Kanfman M., Halper F. Judith Hurwitz “Cloud Computing ” Wiley India Edition,2010
2. John Rittinghouse& James Ransome, “Cloud Computing Implementation Management and Strategy”, CRC Press, 2010
3. Antohy T Velte ,Cloud Computing : “A Practical Approach”, McGraw Hill,2009
4. Michael Miller, Cloud Computing: “Web-Based Applications That Change the Way You Work and Collaborate Online”, Que Publishing, August 2008.
5. James E Smith, Ravi Nair, “Virtual Machines”, Morgan Kaufmann Publishers, 2006.

Online Reading/Supporting Material

1. Haley Beard, “Cloud Computing Best Practices for Managing and Measuring Processes for On-demand Computing”, Applications and Data Centers in the Cloud with SLAs, Emereo Pty Limited, July 2008
2. Webpages.iust.ac.ir/hsalimi/.../89.../Cloud%20Common%20standards.pptop ennebula.org,
3. www.cloudbus.org/cloudsim/, <http://www.eucalyptus.com/>
4. hadoop.apache.org
5. http://hadoop.apache.org/docs/stable/hdfs_design.html
6. http://static.googleusercontent.com/external_content/untrusted_dlcp/research.google.com/en//archive/mapreduce-osdi04.pdf

Cloud Computing Lab

Software Lab based on Cloud Computing:

1. Create virtual machines that access different programs on same platform.
2. Create virtual machines that access different programs on different platforms.
3. Exploring Google cloud for the following
 - a) Storage
 - b) Sharing of data
 - c) Manage your calendar, to-do lists,
 - d) A document editing tool
4. Exploring Open source cloud (Any two)

DCE43001**V SEMESTER****DSE 3B: Elective: Enterprise Resource Planning****Credit (L: T: P = 4: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

CO1. Deliberate the details of ERP

CO2. Learn in depth Models of ERP

CO3. Write down in depth Business Process Mapping for ERP

CO4. Understand in details with applications of ERP and Related Technologies

CO5. Deliberate the details of ERP Modules

CO6. Specify in details with examples SAP

Unit - 1**15 Hours**

Introduction to ERP, Evolution of ERP, What is ERP? Reasons for the growth of ERP, Scenario and Justification of ERP in India, Evaluation of ERP, Various Modules of ERP, Advantage of ERP and Disadvantage of ERP.

Unit - 2**15 Hours**

An overview of Enterprise, Integrated Management Information, Business Modeling, ERP for Small Business, ERP for make to order companies, Business Process Mapping for ERP Module Design, Hardware Environment and its Selection for ERP Implementation.

Unit - 3**15 Hours**

ERP and Related Technologies, Business Process Reengineering (BPR), Management Information System (MIS), Executive Information System (EIS), Decision support System (DSS), Supply Chain Management (SCM) (With Example)

Unit - 4**15 Hours**

ERP Modules, Introduction to Finance, Plant Maintenance, Quality Management, Materials Management, ERP Market, Introduction, SAP AG, Baan Company, Oracle Corporation, People Soft, JD Edwards World Solutions Company, System Software Associates.

Reference Books:

1. C.S. V Murthy Enterprise Resource Planning
2. R.G. Saha – Enterprise Resource Planning - HPH
3. Alexis Leon, Leon Publishers: Enterprise Resource Planning

4. Ravi Anupindi, Sunil Chopra, Pearson Education”. “Managing Business Process Flows
5. Altekar, PHI. Enterprise Resource Planning
6. Srivatsava, I.K. International Publishers, Enterprise Resource Planning
7. P. Diwan Vinod Kumar Garg and N.K. Venkitakrishnan, PHI. Enterprise Resource Planning
8. Introduction to SAP, an Overview of SD: MM, PP, FI/CO Modules of SAP. 10. Zaveri Jyotindra Enterprise Resource Planning

Enterprise Resource Planning Lab

Students should be Prepare ERP Solution Report for his / her Case Study under the supervision of Teacher/ Lecturer

DCE45001**V SEMESTER****DSE 3C: Elective: Data Mining and Data Warehousing****Credit (L: T: P = 4: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Understand the characteristics of Data Warehousing
- CO2. Understand the details of Data Warehousing Architecture
- CO3. Deliberate in depth Data Mining
- CO4. Learn in details with examples Association Rule Mining
- CO5. Specify the details of Classification and Prediction Techniques
- CO6. Learn in depth Clustering Methods
- CO7. Write down in depth Application of Data Mining

Unit - 1**15 Hours**

Data Warehousing: Introduction- Definition and description need for data ware housing, need `for strategic information, failures of past decision support systems, OLTP vs DWH-DWH requirements-trends in DWH-Application of DWH.

Data Warehousing Architecture: Reference architecture- Components of reference architecture - Data warehouse building blocks, implementation, physical design process and DWH deployment process.

Unit - 2**15 Hours**

A Multidimensional Data, Model Data Warehouse Architecture.

Data Mining: Data mining tasks-Data mining vs KDD- Issues in data mining, Data Mining metrics, Data mining architecture - Data cleaning- Data transformation- Data reduction - Data mining primitives.

Unit - 3**15 Hours**

Association Rule Mining: Introduction - Mining single dimensional Boolean association rules from transactional databases - Mining multi-dimensional association rules.

Classification and Prediction: Classification Techniques - Issues regarding classification and prediction - decision tree - Bayesian classification –Classifier accuracy.

Unit - 4**15 Hours**

Clustering: Clustering Methods - Outlier analysis.

Applications and Other Data Mining Methods: Distributed and parallel Data Mining Algorithms, Text mining- Web mining.

Reference Books:

1. Jiawei Han and MichelineKamber, ” Data Mining Concepts and Techniques”, Morgan Kaufmann Publishers, USA, 2006.
2. Berson,”DataWarehousing, Data Mining and OLAP”, Tata McGraw Hill Ltd, New Delhi, 2004.
3. Pang-Ning Tan, Michael Steinbach, Vipin Kumar, Introduction to Data Mining, , Pearson Education.
4. Arun K Pujari,”Data mining techniques”, Oxford University Press, London, 2003.
5. Dunham M H,”Data mining: Introductory and Advanced Topics”. Pearson Education, New Delhi, 2003.
6. MehmedKantardzic,” Data Mining Concepts, Methods and Algorithms”, John Wiley and Sons, USA, 2003.
7. Soman K. P., DiwakarShyam, Ajay V., Insight into Data mining: Theory and Practice, PHI 2006

Data Mining and Data Warehousing Lab

Software Lab based on Data Mining:

Practical List: Practical are to be done using Weka, and a report prepared as per the format*.

The operations are to be performed on built-in dummy data sets of weka and / or the downloadable datasets mentioned in references below. Also wherever applicable, the parameter values are to be varied (upto 3 distinct values). The 'Visualize' tab is to be explored with each operation.

1. Preprocessing: Apply the following filters –
 - a. weka>filter>supervised>attributed>
AddClassification ,AttributeSelection, Discretize , NominalToBinary
 - b. weka>filter>supervised>instance:
StratifiedRemoveFolds, Resample
 - c. weka>filter>unsupervised>attribute>
Add, AddExpression, AddNoise ,Center , Discretize ,
MathExpression ,MergeTwoValues , NominalToBinary ,
NominalToString, Normalize
NumericToBinary ,NumericToNominal , NumericTransform ,
PrincipalComponent,
RandomSubset , Remove , RemoveType , ReplaceMissingValues ,
Standardize
 - d. weka>filter>unsupervised>instance>
Normalize , Randomize , Standardize, RemoveFrequentValues,
RemoveWithValues , Resample , SubsetByExpression
2. Explore the 'select attribute' as follows
weka>attributeSelection> , FilteredSubsetEval , WrapperSubsetEval
3. Association mining
weka>associations> , Apriori, FPGrowth
4. Classification**
weka>classifiers>bayes> , NaïveBayes , weka>classifiers>lazy> : IB1 ,
IBkweka>classifiers>trees , SimpleCart , RandomTree , ID3
5. Clustering**
weka>clusters> , SimpleKMeans , FarthestFirst algorithm, DBSCAN,
hierarchicalClusterer

DCE51001**V SEMESTER****SEC 2A: Elective: Object Oriented Modelling & Design with UML****Credit (L: T: P = 3: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Deliberate in details with examples Object Oriented Development
- CO2. Write down the details of OO Modeling Concepts
- CO3. Learn the details of OO process Overview
- CO4. Identify in depth Design of System Using OO Model
- CO5. Specify the details of Steps for Implementation of OO Modeling
- CO6. Learn the details of Design a System Using UML Tool

Unit - 1**15 Hours**

INTRODUCTION: What is object orientation? What is OO development? OO themes, Evidence for usefulness of OO development, OO Modeling history. Introduction to UML, Importance of modeling, Principles of modeling, Object oriented modeling, Overview of UML, Conceptual model of the UML, Architecture, Software development life cycle.

MODELING CONCEPTS: Modeling as a design technique- Modeling, Abstraction, the three models. Class modeling -object and class concepts, link and association concepts, Generalization and inheritance, a sample model, navigation of class models, Advanced class modeling, advanced object and class concepts, association ends, n-ray associations, aggregation, abstract classes, multiple inheritance, metadata, reification, constraints, derived data, packages.

Unit - 2**15 Hours**

State modeling events, states, transitions and conditions, state diagrams, state diagram behavior. Interaction modeling- use case models, sequence models, activity models. Advanced Interaction Modeling- use case relationships, procedural sequence models, special constructs for activity models

PROCESS OVERVIEW- development stages, development life cycle, Domain analysis- overview of analysis, domain class model, domain state model, domain interaction model, iterating the analysis.

Unit - 3**15 Hours**

SYSTEM DESIGN-overview of system design, estimating performance, making a reuse plan, breaking a system into subsystems, identifying

concurrency, allocation of subsystems, management of data storage, handling global resources, choosing a software control strategy, handling boundary conditions.

CLASS DESIGN- Overview of class design, Bridging the gap, realizing use cases, designing algorithms, Recursing downward, refactoring, Design Optimization Reification of Behavior, adjustment of inheritance, organizing a class design.

IMPLEMENTATION MODELING-Overview of Implementation, fine-tuning classes Fine-tuning generalizations, Realizing associations testing.

Reference Books:

1. Object –oriented modeling and design- Michael R Blaha and James R Rumbaugh
2. Object Technology- David A.Taylor
3. Designing Flexible Object Oriented systems with UML – Charles Ritcher
4. Object Oriented Analysis & Design, Sat/.inger. Jackson, BurdThomson
5. Object Oriented Modeling and Design - James Rumbaugh
6. Teach Yourself UML in 24 Hours - Joseph Schmuilers

Object Oriented Modelling& Design with UML Lab

Term Work / Assignment

Each candidate will submit an approximately 10-page written report on a case study or mini project. Students have to do OO analysis & design for the project problem, and develop use case model, analysis model and design model for it, using UML.

Practical assignment

Nine assignments, one on each of the diagrams learnt in UML.

DCE53001**V SEMESTER****SEC 2B: Elective: Analysis and Design of Algorithms****(Credit L: T: P = 3: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Learn the details of Types of notion of Algorithm
- CO2. Learn in details with examples Algorithm Design Techniques
- CO3. Deliberate in depth Sorting Techniques
- CO4. Deliberate in depth of Searching Techniques
- CO5. Identify in details with examples Analysis of Graph Algorithms
- CO6. Learn the details of Dynamic Programming Methods

Unit - 1**15 Hours**

Introduction: Notion of Algorithm, Review of Asymptotic Notations, Mathematical Analysis of Non-Recursive and Recursive Algorithms Correctness of Algorithm

Algorithm Design Techniques: Iterative techniques Divide and conquer greedy algorithms.

Sorting Techniques: Selection sort, bubble sort, insertion sort, more sorting techniques-quick sort, merge sort. Radix sort,

Unit - 2**15 Hours**

Searching Techniques: Linear and Binary search, Complexity Analysis.

Graphs: Analysis of Graph algorithms -Depth-First Search Breadth-First Search and its applications, minimum Spanning Trees and Shortest Paths - PRIM 'S, KRUSKAL, Dijkstra's algorithm. Branching-Hamiltonian Circuit problem.

Unit - 3**15 Hours**

DYNAMIC PROGRAMMING: The General Method, Warshall's Algorithm, Floyd's Algorithm for the All-Pairs Shortest Paths, Single-Source Shortest Paths: The Travelling Salesperson problem.

Reference Books:

1. Analysis & design of Algorithm-Padma Reddy
2. A.V. Levitin, Introduction to the Design and Analysis of Algorithms, Pearson Education, 2006.
3. J. Kleinberg and E. Tardos, Algorithms Design, Pearson Education, 2006.

Analysis and Design of Algorithms Lab

1. Implement Insertion Sort.
2. Implement Merge Sort.
3. Implement recursive algorithm
4. Implement Randomized Quick sort.
5. Implement Radix Sort.
6. Implement Searching Techniques (linear & Binary)
7. Implement selection sort
8. Implement Bubble sort
9. Implement Prim's Algorithm
10. Implement Dijkstra's Algorithm
11. Implement Krushkal's Algorithm
12. Implement Travelling Salesperson problem
13. Implement Floyd's Algorithm
14. Implement Depth First Search
15. Implement Binary Search tree.

DCE55001**V SEMESTER****SEC 2C: Elective: E-Commerce Technologies****Credit (L: T: P = 3: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

CO1. Understand the details of E-Commerce

CO2. Learn the details of Basic Concepts Of Internet and WWW

CO3. Identify in depth Internet Security Methods

CO4. Learn in details with examples Concepts of Electronic Data Exchange and applications

CO5. Learn in details with examples Planning For E-Commerce

CO6. Understand in depth Features of Internet Marketing

Unit - 1**15 Hours**

An introduction to Electronic commerce: What is E-Commerce (Introduction And Definition), Main activities E-Commerce, Goals of E-Commerce, Technical Components of E-Commerce, Functions of E-Commerce, Advantages and disadvantages of E-Commerce, Scope of E-Commerce, Electronic Commerce Applications, 9 Electronic Commerce and Electronic Business(C2C)(C2G,G2G, B2G, B2P, B2A, P2P, B2A, C2A, B2B, B2C)

The Internet and WWW: Evolution of Internet, Domain Names and Internet Organization (.edu, .com, .mil, .gov, .net etc.) , Types of Network, Internet Service Provider, World Wide Web, Internet & Extranet, Role of Internet in B2B Application, building own website, Cost, Time, Reach, Registering a Domain Name, Web promotion, Target email, Baner, Exchange, Shopping Bots.

Unit - 2**15 Hours**

Internet Security: Secure Transaction, Computer Monitoring, Privacy on Internet, Corporate Email privacy, Computer Crime(Laws , Types of Crimes), Threats, Attack on Computer System, Software Packages for privacy, Hacking, Computer Virus(How it spreads, Virus problem, virus protection, Encryption and Decryption, Secret key Cryptography, DES, Public Key Encryption, RSA, Authorization and Authentication, Firewall, Digital Signature(How it Works)

Electronic Data Exchange: Introduction, Concepts of EDI and Limitation, Applications of EDI, Disadvantages of EDI, EDI model, Electronic Payment System: Introduction, Types of Electronic Payment System, Payment Types,

Value Exchange System, Credit Card System, Electronic Fund Transfer, Paperless bill, Modern Payment Cash, Electronic Cash.

Unit - 3**15 Hours**

Planning for Electronic Commerce: Planning Electronic Commerce initiates, linking objectives to business strategies, measuring cost objectives, Comparing benefits to Costs, Strategies for developing electronic commerce web sites.

Internet Marketing: The PROS and CONS of online shopping, The cons of online shopping, Justify an Internet business, Internet marketing techniques, The E-cycle of Internet marketing, Personalization e-commerce.

Reference Books:

1. G.S.V.Murthy, E-Commerce Concepts, Models, Strategies- :- Himalaya Publishing House, 2011.
2. Kamlesh K Bajaj and DebjaniNag , E- Commerce , 2005.
3. Gray P. Schneider , Electronic commerce, International Student Edition, 2011,
4. HENRY CHAN, RAYMOND LEE, THARAM DILLON, ELIZABETH CHANG E COMMERCE, FUNDAMENTALS AND APPLICATIONS, Wiely Student Edition, 2011

E-Commerce Technologies Lab

Software Lab based on E-Commerce Technologies

E-commerce concepts are to be implemented in developing a website using a combination of following technologies:

1. Hyper Text Markup Language (HTML)
2. Cascading Style Sheets (CSS)
3. JavaScript
4. ASP
5. PHP
6. XML
7. Joomla

DCF31001

VI SEMESTER

DSE 4A: Elective: Operation Research**Credit (L: T: P = 4: 0: 1)****Course Outcome**

After successful completion of the course, the student is able to

- CO1. Write down the details of Origin and Development of Operation Research
- CO2. Understand the characteristics of Linear Programming Problems and Methods
- CO3. Deliberate in depth Transportation Problems
- CO4. Deliberate in depth Assignment Problem
- CO5. Identify in details with examples Network Analysis
- CO6. Learn in depth Application of Operation Research

Unit - 1**15 Hours**

Linear Programming Problems: Origin and development of operations research, formulation of Linear Programming problem, Graphical solution. Theory of simplex method, Use of artificial variables and their solution.

Unit - 2**15 Hours**

Transportation Problem: Mathematical formulation of transportation problem, Initial basic Feasible solution, North West corner rule, Matrix minima method, Vogel's approximation method, MODI method to find optimal solution.

Unit - 3**15 Hours**

Assignment Problem: Mathematical formulation of an Assignment problem, Assignment algorithm, Hungarian Method to solve Assignment Problem.

Unit - 4**15 Hours**

Network Analysis: Basic components of Network, Rules for drawing Network diagram Time calculation in Networks. Critical Path Method and PROJECT Evaluation and Review Techniques. Algorithm and flow chart for CPM and PERT.

Reference Books:

1. Taha, "Operations Research", 7th edition, Pearson Education, 2007.
2. Billey E. Gillett, "Introduction to Operations Research", Himalaya Publishing House, Delhi, 1979.
3. Hamady A. Taha "Operations Research", Collin Mac Millan, 1982

Operation Research Lab

Lab based on Operation Research

1. LPP
2. Simplex
3. Dual Simplex
4. Big – M
5. Vogel's
6. Maxima and Minima
7. North West corner
8. Sequencing Problems
9. Modi Method
10. Hugarian Method
11. Assignment Problem

DCF33001**VI SEMESTER****DSE 4B: Elective: System Software & Compiler design****Credit (L: T: P = 4: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Understand the classification and characteristics of language Processors
- CO2. Learn the details of Concepts Of Macros
- CO3. Learn the details of Linkers and Loaders
- CO4. Understand in details with examples of Compiler Concepts
- CO5. Understand the details of Parsing Methods
- CO6. Write down the details of Code optimization

Unit - 1**15 Hours**

ASSEMBLERS & MACROS: Overview of Language processors – Assemblers: Design of two pass assemblers - single pass assemblers

MACRO: Macro definition- macro call – macro expansion- nested macro advanced

Macro facilities- Design of Microprocessor.

LINKERS & LOADERS: Relocation and linking concepts – Design of linker – self relocating programs – linking in MS-DOS – overlays

DYNAMIC LINKING: Loaders – Absolute loaders- relocating loaders

Unit - 2**15 Hours**

COMPILERS: GRAMMARS & AUTOMATA - Languages – Grammars – Types of grammars – Context free grammar – regular expression - Recognizing of patterns - finite automation (deterministic & non deterministic) Conversion of NDFFA to DFA - Conversion of regular expression of NDFFA – Thompson's construction- minimization of NDFFA - Derivation - parse tree - ambiguity – Lexical analysis- handles - token specification - design of lexical analysis (LEX) - Automatic generation of lexical analyzer - input buffering – A language for specifying lexical analyzers - implementation of lexical analyzer

SYNTAX ANALYSIS – PARSING: Definition - role of parsers - top down parsing - bottom-up parsing - Left recursion - left factoring - Handle pruning , Shift reduce parsing - operator precedence parsing – FIRST- FOLLOW- LEADING- TRAILING

Unit - 3**15 Hours**

Predictive parsing – recursive descent parsing. LR parsing – LR (0) items - SLR parsing – Canonical LR – LALR parsing - generation of LALR - Ambiguous grammars - error recovery

SYNTAX DIRECTED TRANSLATION & CODE OPTIMIZATION:

Intermediate Languages - prefix - postfix - Quadruple - triple - indirect triples – syntax tree- Evaluation of expression - three-address code- Synthesized attributes – Inherited attributes – Conversion of Assignment statements- Boolean expressions –Backpatching - Declaration - CASE statements

Unit - 4**15 Hours**

CODE OPTIMIZATION: Local optimization- Loop Optimization techniques – DAG –Dominators- Flow graphs – Storage allocations- Peephole optimization – Issues in Code Generation.

Reference Books:

1. Alfred V Aho , Jeffery D Ullman , Ravi Sethi, " Compilers , Principles techniques and tools ", Pearson Education 2011
2. Dhamdhare D.M., “Systems Programming”, Tata McGraw Hill Education Pvt. Ltd., 2011.
3. Srimanta Pal, “Systems Programming”, Oxford University Press, 2011.
4. Raghavan V., “Principles of Compiler Design”, Tata McGraw Hill Education Pvt. Ltd., 2010.
5. David Galles, “Modern Compiler Design”, Pearson Education, Reprint 2012.
6. DasaradhRamaiah. K., “Introduction to Automata and Compiler Design”, PHI, 2011.

System Software & Compiler design Lab

1. Implementation of a text editor
2. Implementation of an Assembler
3. Implementation of Macro processor
4. Converting a regular expression to NFA
5. Conversion of an NFA to DFA
6. Computation of FIRST and FOLLOW sets
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7. Computation of Leading and Trailing Sets
8. Construction of Predictive Parsing Table
9. Implementation of Shift Reduce Parsing
10. Computation of LR(0) items
11. Construction of DAG
12. Intermediate code generation

DCF35001**VI SEMESTER****DSE 4C: Elective: Digital Image Processing****Credit (L: T: P = 4: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

CO1. Learn in details with examples Element of Digital Image Processing Systems

CO2. Specify in depth Image Enhancement Techniques

CO3. Write down in depth Image Restoration Methods

CO4. Learn the details of Image Segmentation Algorithms

CO5. Identify in details with examples Data Image Compression

CO6. Learn in depth - Various Application of Image Processing

Unit - 1**15 Hours**

DIGITAL IMAGE FUNDAMENTALS Elements of digital image processing systems, Videocon and Digital Camera working principles, Elements of visual perception, brightness, contrast, hue, saturation, mach band effect, Color image fundamentals - RGB, HSI models, Image sampling, Quantization, dither, Two-dimensional mathematical preliminaries, 2D transforms - DFT, DCT, KLT, SVD.

IMAGE ENHANCEMENT: Histogram equalization and specification techniques, Noise distributions, Spatial averaging,

Unit - 2**15 Hours**

Directional Smoothing, Median, Geometric mean, Harmonic mean, Contra harmonic mean filters, Homomorphism filtering, Color image enhancement.

IMAGE RESTORATION: Image Restoration - degradation model, Unconstrained restoration - Lagrange multiplier and Constrained restoration, Inverse filtering-removal of blur caused by uniform linear motion, Wiener filtering,

Unit - 3**15 Hours**

Geometric transformations-spatial transformations.

IMAGE SEGMENTATION: Edge detection, Edge linking via Hough transform – Thresholding - Region based segmentation – Region growing – Region splitting and Merging – Segmentation by morphological watersheds – basic concepts – Dam construction – Watershed segmentation algorithm.

Unit - 4**15 Hours**

IMAGE COMPRESSION: Need for data compression, Huffman, Run Length Encoding, Shift codes, Arithmetic coding, Vector Quantization, Transform coding, JPEG standard, MPEG.

Reference Books:

1. Rafael C. Gonzalez, Richard E. Woods, , Digital Image Processing', Pearson, Second Edition, 2004.
2. Anil K. Jain, , Fundamentals of Digital Image Processing', Pearson 2002.

Digital Image Processing Lab

Lab 1: Write a program for image enhancement

Lab2: Write a program for image compression

Lab3: Write a program for color image processing

Lab4: Write a program for image segmentation

Lab 5: Write a program for image morphology

Lab 6: Image Restoration

Lab 7: Edge detection

Lab 8: Blurring 8 bit color versus monochrome

DCF40001**VI SEMESTER****DSE 5: Elective: Dissertation / Project**
Credit (L: T: P = 0: 0: 6) 12 Hours/Week**Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Identify in details with examples Problem identification
- CO2. Write down in depth System Analysis
- CO3. Understand and Develop SRS for selected System Problem
- CO4. Understand and Develop System Design for selected System Problem
- CO5. Learn in details and Develop a Code and Test the System
- CO6. Understand the details of Presentation and Demo of Project Work

- This option is to be offered only in 6th Semester.
- The students will be allowed to work on any project based on the concepts studied in
Core/elective or skill based elective courses.
- The group size should be maximum of TWO (02) students.
- Each group will be assigned a teacher as a supervisor who will handle both their theory as Well as lab classes.
- A maximum of Four (04) projects would be assigned to one teacher.

DCF51001**VI SEMESTER****SEC 3A: PHP Programming****Credit (L: T: P = 1: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Learn in depth Elements of PHP
- CO2. Learn in depth Interaction Methods Between HTML and PHP
- CO3. Understand in depth PHP function
- CO4. Understand in depth String Manipulation
- CO5. Learn the characteristics of Regular Expression
- CO6. Learn the details of Developing PHP Web Application

Unit - 1**15 Hours**

Introduction Basic PHP Development Control Structure: Introduction to www, History, Understanding client/server roles Apache, PHP, MySQL, XAMPP Installation PHP Basic syntax, PHP data Types, PHP Variables PHP Constants, PHP Expressions, PHP Operators Control Structures & Loop

Working With the File System Working With Regular Expressions, Opening a File, Reading from a File, Writing to a File, File Locking, Uploading Files via an HTML Form, Getting File Information, Directory Functions, Getting a Directory Listing, The basic regular expressions, Matching patterns, Finding matches, Replace patterns

WORKING WITH FORMS: PHP Form handling, PHP GET/POST, PHP Form Validation, Accessing user input, Combine HTML and PHP code using hidden fields, Redirecting the user, File upload

CLASSES AND OBJECTS: Object oriented concepts, Define a class, attributes, Object, Object properties, methods, constructors and destructors, Class constants, Static method, Class inheritance, Abstract classes, Final keyword, Implementing Interface, Object serialization

Reference Books:

1. [Complete Beginner's Guide to PHP: Programming & Web Development](#) by Cedric Palmer (22 February 2014)
2. [PHP and MySQL Web Development](#) by Laura Thomson and Luke Welling
3. [PHP Reference: Beginner to Intermediate PHP5](#) by Mario Lurig
4. [PHP 4: A Beginner's Guide](#) by William Mccarty

PHP Programming Lab

Students should do Mini Project Using PHP under the supervision of Teacher/ Lecturer

DCF53001**V SEMESTER****SEC 3B: Elective: Computer Simulations****Credit (L: T: P = 1: 0: 1)****Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Deliberate the details of Concepts of Simulation and Methods
- CO2. Understand the details of Random Numbers/variates
- CO3. Learn the details of Design of Simulation Experiments
- CO4. Understand the characteristics of Discrete System simulation languages
- CO5. Learn in details with Case studies using Simulation Languages
- CO6. Identify the details of applications of simulation

Unit - 1**15 Hours**

INTRODUCTION: Concept of simulation – simulation as a decision making tool-Monte Carlo simulation.

RANDOM NUMBERS/VARIATES: Pseudo random numbers – methods of generating random variates – random variates for uniform, normal, binominal, poisson, exponential distributions.

DESIGN OF SIMULATION EXPERIMENTS: Problem formulation – data collection and reduction – logic developments – initial conditions – run length, tabular method of simulation – development of models using higher level languages for systems like queuing, production, inventory and maintenance – output analysis and interpretation, validation.

DISCRETE SYSTEM SIMULATION LANGUAGES: Need for simulation language – Comparison of simulation languages: SIMSCRIPT, GASP, SIMULA, GPSS, PROMODEL, etc...

CASE STUDIES USING SIMULATION LANGUAGES: Development of simulation models using the simulation language studies for systems for systems like, queuing systems, production systems, inventory systems, maintenance and replacement systems, investment analysis and network.

Reference Books:

1. Jerry Banks and John S.Carson, Barry L Nelson, David M.Nicol, P.Shahabudeen “Discrete event system simulation” Pearson, 2007.
2. Thomas J.Schriber, “Simulation using GPSS”, John Wiley, 2002.
3. Law A.M. and Kelton W.D “Simulation Modeling and Analysis, McGraw Hill, 2003

Computer Simulations Lab

**Design lab Cycle based on Computer Simulations (Theory) Paper by
Concerned Teacher / Lecturer**

DCF55001

V SEMESTER

SEC 3C: Elective: Information Security and Cyber Laws
Credit (L: T: P = 1: 0: 1)**Course Outcome:**

After successful completion of the course, the student is able to

- CO1. Learn in depth Computer Network Threats
- CO2. Understand the details of Digital Crimes
- CO3. Understand the details of Risk Analysis
- CO4. Understand in details with examples Concepts of Cryptography
- CO5. Learn in details with examples Application of Cryptography
- CO6. Learn in details with examples safety tools and Cyber Laws

Unit - 1**15 Hours**

Introduction: Computer network as a threat, hardware vulnerability, software vulnerability, importance of data security.

Digital Crime: Overview of digital crime, criminology of computer crime.

Information Gathering Techniques: Tools of the attacker, information and cyber warfare, scanning and spoofing, password cracking, malicious software, session hijacking

Risk Analysis and Threat: Risk analysis, process, key principles of conventional computer security, security policies, authentication, data protection, access control, and internal vs. external threat, security assurance, passwords, authentication, and access control, computer forensics and incident response

Introduction to Cryptography and Applications: Important terms, Threat, Flaw, Vulnerability, Exploit, Attack, Ciphers, Codes, Substitution Cipher (Caesar), Transposition Cipher (Rail-Fence), Public key cryptography (Definitions only), Private key cryptography (Definition and Example), Cyber forensics, Steganography

Safety Tools and Issues: Firewalls, logging and intrusion detection systems, Windows and windows XP / NT security, Unix/Linux security, ethics of hacking and cracking

Cyber laws to be covered as per IT 2008

Reference Books:

1. M. Merkow, J. Breithaupt, Information Security Principles and Practices, Pearson Education.
2. G.R.F. Snyder, T. Pardoe, Network Security, Cengage Learning, 2010
3. A. Basta, W.Halton, Computer Security: Concepts, Issues and Implementation, Cengage Learning India, 2008
4. Anderson, Ross. Security engineering. John Wiley & Sons, 2008. (Freely available online)

Information Security and Cyber Laws Lab

1. Demonstrate the use of Network tools: ping, ipconfig, ifconfig, tracert, arp, netstat, whois
2. Use of Password cracking tools: John the Ripper, Ophcrack. Verify the strength of passwords using these tools.
3. Perform encryption and decryption of Caesar cipher. Write a script for performing these operations.
4. Perform encryption and decryption of a Rail fence cipher. Write a script for performing these operations.
5. Use nmap / zenmap to analyse a remote machine.
6. Use Burp proxy to capture and modify the message.
7. Demonstrate sending of a protected word document.
8. Demonstrate sending of a digitally signed document.
9. Demonstrate sending of a protected worksheet.
10. Demonstrate use of steganography tools.
11. Demonstrate use of gpg utility for signing and encrypting purposes.

Question Paper Pattern
Theory (3 or 4 Credits)

Time: 3 Hours

Max. Marks: 70

Part – A

I. Answer any Eleven Question out of given Twelve Questions. 11 X 2 = 22

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.
- 9.
- 10.
- 11.
- 12.

Note: Three Questions each from the Units 1, 2, 3, and 4

Part - B

II. Answer any Two Sub Questions from each main Question.

- | | |
|--------|------------|
| 13. A) | 2 X 6 = 12 |
| B) | |
| C) | |
| 14. A) | 2 X 6 = 12 |
| B) | |
| C) | |
| 15. A) | 2 X 6 = 12 |
| B) | |
| C) | |
| 16. A) | 2 X 6 = 12 |
| B) | |
| C) | |

Note: Each Main questions contains 3 sub questions carries 06 Marks (may have internal Split-ups) and from Units 1, 2, 3 and 4 respectively

Question Paper Pattern**Theory (1 or 2 Credits)**

Time: 2 Hours

Max. Marks: 50

Part – A

I. Answer all Questions.**05 X 02 = 10**

- 1.
- 2.
- 3.
- 4.
- 5.

Part - B

II. Answer any Four Questions out of given Five Questions. 4 X 10 = 40

- 1.
- 2.
- 3.
- 4.
- 5.

Note: Each Main questions contains 3 sub questions carries 06 Marks (may have internal Split-ups) and from Units 1, 2, 3 and 4 respectively

Question Paper Pattern**Practical / Project (1 or 2 Credits)**

Time: 3 Hours

Max. Marks: 70

Practicals

1. Two Experiments/ Programs 20 Marks Each 2 x 20 = 40 Marks
 - a. Write-ups 10 Marks
 - b. Conducting & Results 10 Marks (Any One for 2 Hours of Practical's)
2. Viva – Voice 10 Marks
3. Record 10 Marks

Project

1. Project Presentation 10% Marks
2. Project Demo 20% Marks
3. Viva – Voice 20% Marks
4. Record / Report 20% Marks
5. IA (C1 & C2) 30% Marks



JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(An autonomous College of University of Mysuru)
Re-accredited by NAAC with 'A' grade
Ooty road, Mysuru-570 025, Karnataka

DEPARTMENT OF BIOCHEMISTRY

SYLLABUS

CHOICE BASED CREDIT SYSTEM

FOR
B.Sc. PROGRAMME

Biochemistry, Microbiology & Biotechnology

Botany, Biochemistry & Microbiology

2019-2020

BSc- B.Sc UG- syllabus - Programme: Biochemistry, Microbiology & Biotechnology

Scheme for examination and assessment for B.Sc. Biochemistry under CBCS scheme 2019-20

Year	Sem	Course Type	Title of the Course	Course Code	Credits	Maximum Marks in the Exam/ Assessment				Exam Duration		
						SEE		IA		Th	Pr	
						L:T:P	Th	Pr	Th			Pr
I BSc	I	DSC-1 Theory	Fundamentals Of Chemistry And Molecules Of Life	DMA 21006	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSC-1 Practicals	Fundamentals Of Chemistry And Molecules Of Life	DMA 21106	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
	II	DSC-2 Theory	Physiology	DMB 21006	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSC-2 Practicals	Physiology	DMB 21106	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
II BSc	III	DSC-3 Theory	Enzymology & Bioenergetics	DMC 21006	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSC-3 Practicals	Enzymology & Bioenergetics	DMC 21106	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
	IV	DSC-4 Theory	Metabolism	DMD 21006	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSC-4 Practicals	Metabolism	DMD 21106	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
III BSc	V		No. Of Course 1									
		DSE-1 Theory	Nutritional Biochemistry	DME 21006	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSE-1 Practicals	Nutritional Biochemistry	DME 21106	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
		DSE-2 Theory	Molecular Basis Of Infectious Diseases	DME 21206	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSE-2 Practicals	Molecular Basis Of Infectious Diseases	DME 21306	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
		SEC 1-Theory	Tools And Techniques In Biochemistry	DME 21406	2:0:0	35	-	15 (15+15)	-	2h	-	
	SEC 2-Theory	Clinical Biochemistry	DME 21606	2:0:0	35	-	15 (15+15)	-	2h	-		
	VI			No. Of Course 1								
		DSE-3 Theory	Advanced Cell Biology And Endocrinology	DMF 21006	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSE-3 Practicals	Advanced Cell Biology And Endocrinology	DMF 21106	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
		DSE-4 Theory	Plant Biochemistry	DMF 21206	4:0:0	70	-	30 (15+15)	-	3h	-	
	DSE-4 Practicals	Plant Biochemistry	DMF 21306	0:0:2	-	35	-	15 (7.5+7.5)	-	3h		

Practical weightage

Component	Marks
Knowledge of relevant processes	30%
Skills and operations involved	50%
Result/ products including calculations and reporting	20%

B.Sc UG syllabus - Programme: Botany, Biochemistry & Microbiology

Scheme for examination and assessment for B.Sc. Biochemistry under CBCS scheme
2019-20

Year	Sem	Course Type	Title of the Course	Course Code	Credits	Maximum Marks in the Exam/ Assessment				Exam Duration		
						SEE		IA		Th	Pr	
						L:T:P	Th	Pr	Th			Pr
I BSc	I	DSC-1 Theory	Fundamentals Of Chemistry And Molecules Of Life	DMA 21007	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSC-1 Practicals	Fundamentals Of Chemistry And Molecules Of Life	DMA 21107	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
	II	DSC-2 Theory	Physiology	DMB 21007	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSC-2 Practicals	Physiology	DMB 21107	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
II BSc	III	DSC-3 Theory	Enzymology & Bioenergetics	DMC 21007	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSC-3 Practicals	Enzymology & Bioenergetics	DMC 21107	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
	IV	DSC-4 Theory	Metabolism	DMD 21007	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSC-4 Practicals	Metabolism	DMD 21107	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
III BSc	V		No. Of Course 1									
		DSE-1 Theory	Nutritional Biochemistry	DME 21007	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSE-1 Practicals	Nutritional Biochemistry	DME 21107	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
		DSE-2 Theory	Molecular Basis Of Infectious Diseases	DME 21207	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSE-2 Practicals	Molecular Basis Of Infectious Diseases	DME 21307	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
			No. Of Course 1									
		SEC 1- Theory	Tools And Techniques In Biochemistry	DME 21407	2:0:0	35	-	15 (15+15)	-	2h	-	
		SEC 2- Theory	Clinical Biochemistry	DME 21607	2:0:0	35	-	15 (15+15)	-	2h	-	
	VI		No. Of Course 1									
		DSE-3 Theory	Advanced Cell Biology And Endocrinology	DMF 21007	4:0:0	70	-	30 (15+15)	-	3h	-	
		DSE-3 Practicals	Advanced Cell Biology And Endocrinology	DMF 21107	0:0:2	-	35	-	15 (7.5+7.5)	-	3h	
		DSE-4 Theory	Plant Biochemistry	DMF 21207	4:0:0	70	-	30 (15+15)	-	3h	-	
	DSE-4 Practicals	Plant Biochemistry	DMF 21307	0:0:2	-	35	-	15 (7.5+7.5)	-	3h		

Practical weightage

Component	Marks
Knowledge of relevant processes	30%
Skills and operations involved	50%
Result/ products including calculations and reporting	20%

SCHEME OF VALUATION FOR PRACTICAL EXAMINATION

- A candidate appearing for the first time should submit a duly signed and certified practical record
- Each candidate has to perform one experiment in the specified duration of three hours for **FIFTEEN marks**
- Practical record has to be valued for **FIVE marks** by examiners at the time of examination
- Viva-voce for **FIVE marks** in practical is awarded by continuous assessment in the lab

Sl no	Component	Marks
1	Write up of the experiment	05
2	Conducting experiment	15
3	Result	05
4	Viva-voce	05
5	Practical record	05
TOTAL		35

Programme Outcome for Bachelor of Science in Biochemistry, Microbiology & Biotechnology

After completing the graduation in the Bachelor of Science the students are able to:

- PO1.** Demonstrate the ability to justify and explain their thinking and/or approach
- PO2.** Develop state-of-the-art laboratory and professional communication skills
- PO3.** Apply the scientific method to design, execute, and analyze an experiment
- PO4.** Explain scientific procedures and their experimental observations
- PO5.** Demonstrate an understanding of fundamental biochemical principles, structure and function
- PO6.** Work as a laboratory technician, biochemists or medical scientist
- PO7.** Explain the processes used by microorganisms for the growth
- PO8.** Explain the theoretical basis of tools, technologies and methods of microbiology
- PO9.** Design and develop solution to Biotechnology problems
- PO10.** Applying appropriate tools keeping in mind safety factor for environment & society
- PO11.** Create, select, and apply appropriate techniques, resources, and modern tools
- PO12.** Support biotechnology research activity with strong technical background

Programme Specific Outcome

Bachelor of Science in Biochemistry, Microbiology & Biotechnology

After completing the graduation in the Bachelor of Science the students are able to;

- PSO 1:** Gain and understand biochemical and molecular processes
- PSO2:** Communicate scientific information effectively, relating to microbes and their role in ecosystem and health
- PSO3:** Acquire, articulate, retain and demonstrate laboratory safety skills
- PSO4:** Demonstrate applications of biochemical and biological sciences
- PSO5:** Apply appropriate tools and techniques in biotechnological manipulation
- PSO6:** Understand the responsibilities of biotechnological practices

Programme Outcome for Bachelor of Science in Botany, Biochemistry & Microbiology

After completing the graduation in the Bachelor of Science the students are able to:

- PO1.** Identify the taxonomic position of plants using principles and methods of nomenclature and classification in Botany
- PO2.** Understand the impact of the plant diversity in societal and environmental context
- PO3.** Demonstrate the knowledge of, and need for sustainable development
- PO4.** Use interdisciplinary approaches with quantitative skills to work on biological problems
- PO5.** Demonstrate the ability to justify and explain their thinking and/or approach
- PO6.** Develop state-of-the-art laboratory and professional communication skills
- PO7.** Apply the scientific method to design, execute, and analyze an experiment
- PO8.** Explain scientific procedures and their experimental observations
- PO9.** Demonstrate an understanding of fundamental biochemical principles, structure and function
- PO10.** Work as a laboratory technician, biochemists or medical scientist
- PO11.** Explain the processes used by microorganisms for the growth
- PO12.** Explain the theoretical basis of the tools, technologies and methods of microbiology

Programme Specific Outcome

Bachelor of Science in Botany, Biochemistry & Microbiology

After completing the graduation in the Bachelor of Science the students are able to;

- PSO 1:** Demonstrate applications of biochemical and biological sciences
- PSO2:** Inculcating proficiency in all experimental techniques and methods of analysis
- PSO3:** Acquire, articulate, retain and demonstrate laboratory safety skills
- PSO4:** Communicate scientific information effectively, relating to microbes and their role in ecosystem and health
- PSO5:** Gain proper procedures and regulations in handling and disposal of chemicals
- PSO6:** Understand biochemical and molecular processes that occur in and between the cells

CORE: FUNDAMENTALS OF CHEMISTRY AND MOLECULES OF LIFE

(Credits: Theory – 04, Practical – 02)

Theory: 60 Hrs

Course Outcome:

After completion of the course the student is able to

- CO1:** Understand in detail with examples stereo-chemistry
- CO2:** Specify the characteristics of carbohydrates & glycobiology
- CO3:** Learn the characteristics of proteins
- CO4:** Understand the classification and characteristics of vitamins

- | | |
|---|---------------|
| Unit : 1 THE FOUNDATIONS OF BIOCHEMISTRY : Cellular and chemical foundations of life | 2 Hrs |
| Unit : 2 WATER: Unique properties, weak interactions in aqueous systems, ionization of water, buffers, water as a reactant and fitness of the aqueous environment. | 4 Hrs |
| Unit : 3 CONCENTRATION UNITS: Mole, Mole fraction, Molality, Molarity and Normality (problems to be worked out). | 1 Hrs |
| Unit : 4 STEREOCHEMISTRY: Stereoisomerism, types. Geometrical Isomerism- Cis Trans & E/Z nomenclature. Optical Isomerism- asymmetric carbon atom, Optical activity, plane polarized light, Chirality, Specific molecular rotation, Projection formula- Fischer & Newman projection formulae, optical isomerism in Glyceraldehyde, Lactic acid, Tartaric acid. Nomenclature of enantiomers - D and L system, Racemisation and resolution (biochemical method) | 6 Hrs |
| Unit: 5 REACTION MECHANISM: Concept of inductive, mesomeric and resonance effect. Concept of the reaction intermediates- carbanions, carbocations, free radicals, carbenes. Nucleophiles and Electrophiles. | 3 Hrs |
| Unit : 6 CARBOHYDRATES & GLYCOBIOLOGY: Monosaccharides - structure of aldoses and ketoses, ring structure of sugars, conformations of sugars, mutarotation, anomers, epimers and enantiomers, oxidation of sugars. Formation of disaccharides, reducing and nonreducing disaccharides. Polysaccharides – homo- and heteropolysaccharides, structural and storage polysaccharides. Structure and role of proteoglycans, glycoproteins and glycolipids (gangliosides and lipopolysaccharides). | 12 Hrs |
| Unit : 7 LIPIDS: Building blocks of lipids - fatty acids, glycerol, ceramide. Storage lipids - triacyl glycerol and waxes. Structural lipids in membranes- glycerophospholipids, galactolipids and sulpholipids, sphingolipids and sterols, structure, distribution and role of membrane lipids. Saponification value, Iodine value, Peroxide value, Acid value and their significance. Eicosanoids: Definition, types, Prostaglandins: Definition, biological functions, structure of PGE1 & E2 and PGF1&F2. | 12 Hrs |

Thromboxans and leukotrienes- Biological roles.

- Unit : 8 AMINO ACIDS & PROTEINS:** Structure and classification, physical, chemical and optical properties of amino acids **9 Hrs**
Peptide bond - Formation and structure. Structure and biological importance of Glutathione, Oxytosine, Vasopressine, Endorphine. Synthetic peptides – importance of polyglutamic acid, polylysine.
Classification of proteins based on structure with examples. Primary structure of proteins – Determination of amino acid composition, determination of N- and C- terminal amino acids, sequencing by Edman's degradation method. Secondary structure: - a Helix, b- sheet, b barrel and b turn. Tertiary structure – Fibrous proteins (collagen) and Globular proteins (Myoglobin). Quaternary Structure – hemoglobin, Denaturation and renaturation of proteins by Anfinsen's experiment.
- Unit : 9 NUCLEIC ACIDS:** Nucleotides - structure and properties. Nucleic acid structure – Watson-Crick model of DNA. Structure of major species of RNA - mRNA, tRNA and rRNA. Nucleic acid chemistry - UV absorption, effect of acid and alkali on DNA. **6 Hrs**
- Unit: 10 VITAMINS:** Structure and active forms of water soluble and fat soluble vitamins. **5 Hrs**

DMA 21106/ DMA 21107

PRACTICALS

- 1 Safety measures in laboratories.
- 2 Preparation of normal and molar solutions.
- 3 Preparation of buffers.
- 4 Determination of pKa of acetic acid and glycine.
- 5 Qualitative tests for carbohydrates.
- 6 Qualitative tests for lipids.
- 7 Qualitative tests for amino acids, proteins.
- 8 Qualitative tests for nucleic acids.
- 9 Estimation of amino acid by formal titration
- 10 Determination of Acid value of an oil or fat
- 11 Preparation of ZnSO₄ standard solution, standardization of approximately N/10 EDTA solution and estimation of total hardness of water using Erichrome black –T indicator.
- 12 Preparation of standard sodium carbonate solution. Standardization of approximately N/10 NaOH and estimation of HCl in the given solution.
- 13 Determination of density the given protein solution or liquid using specific gravity bottle and viscosity by using ostwald's viscometer.
- 14 Conductometric titration of Amino acid against a strong base.

Reference:

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-4292-3414-8.
2. Textbook of Biochemistry with Clinical Correlations (2011) 7th ed., Devlin, T.M., John Wiley & Sons, Inc. (New York), ISBN:978-0-470-28173-4.

SEMESTER II

CORE: PHYSIOLOGY

(Credits: Theory – 04, Practical – 02)

Theory: 60 Hrs

Course Outcome:

After completion of the course the student is able to;

CO1: Understand in depth cardiovascular physiology

CO2: Specify the characteristics of renal physiology

CO3: Deliberate the detail of musculoskeletal system

CO4: Learn the detail of reproductive physiology

- Unit : 1 HOMEOSTASIS & THE ORGANIZATION OF BODY FLUID 8 Hrs**
COMPARTMENTS: Intracellular, extracellular and interstitial fluid. Homeostasis, control system and their components. Plasma as an extracellular fluid, RBC, molecular mechanism of blood coagulation, role of vitamin K in coagulation, anticoagulant and fibrinolytic systems. Anemias, polycythemia, haemophilia and thrombosis
- Unit: 2 CARDIOVASCULAR PHYSIOLOGY: 8 Hrs**
Anatomy of heart. Physiology of the cardiac muscle. Ventricular volumes and the ECG, control of cardiac function and output. The arterial system, venous system and the microcirculation. Portal circulations. Hypertension, congestive heart disease, atherosclerosis and myocardial infarction.
- Unit: 3 RESPIRATION: 6 Hrs**
Mechanism of respiration. Principles of gas exchange and transport. Regulation of respiration. Pulmonary oedema and regulation of pleural fluid. Hypoxia, emphysema.
- Unit: 4 RENAL PHYSIOLOGY: 7 Hrs**
Anatomy of the kidney and the nephron. Regulation of renal blood flow. Cell biology of the Bowmans' capsule. Physiology of glomerular filtration and GFR. Tubular reabsorption and secretion. Blood buffer systems, renal and pulmonary control of blood pH. Assessment of kidney function. Acidosis and alkalosis. Glomerular nephritis, renal failure and dialysis.
- Unit: 5 GASTROINTESTINAL & HEPATIC PHYSIOLOGY: 7 Hrs**
Propulsion and motility of food and digested material. digestion and absorption of macro and micronutrients. Peptic ulcer, Sprue, celiac disease, IBD, regurgitation, diarrhoea and constipation. Anatomy of the hepatic lobule and blood flow into the liver. Formation and secretion of bile. enterohepatic cycle, metabolic importance of liver. Liver function tests. Jaundice.
- Unit :6 MUSCULOSKETETAL SYSTEM: 4 Hrs**
Bone structure and formation. Physiology of muscle contraction in striated and non-striated muscle.

- Unit: 7 REPRODUCTIVE PHYSIOLOGY:** Sex determination and 7 Hrs
differentiation. Development of female and male genital tracts.
Spermatogenesis, capacitation and transport of sperm, blood testis barrier.
Ovarian function and its control. Uterine changes, fertilization and
implantation. Placenta as a feto- maternal unit, gestation and parturition.
- Unit :8 NEUROCHEMISTRY & NEUROPHYSIOLOGY:** Central Nervous 6 Hrs
system. Peripheral Nervous system. Blood brain barrier and CSF.
Membrane potentials. Synaptic transmission. Neurotransmitters.
- Unit :9 PLANT RESPIRATION:** Overview of glycolysis, Alternative reactions 7 Hrs
of glycolysis, Regulation of plant glycolysis, Translocation of metabolites
across mitochondrial membrane, TCA cycle, Alternative NAD(P)H
oxidative pathways; Cyanide resistant respiration.

DMB21106 / DMB21107

PRACTICALS

- 1 Hematology
- 2 RBC counting
- 3 WBC counting
- 4 Clotting time
- 5 Determination of ABO blood group
- 6 Estimation of serum phospholipids
- 7 Qualitative analysis of abnormal constituents of urine – Glucose, albumin, bile pigments, bile salts & ketone bodies
- 8 Determination of titrable acidity of urine
- 9 Pulmonary function tests, spirometry and measurement of blood pressure.
- 10 Separation of isoenzymes of LDH by electrophoresis.
- 11 Histology of connective tissue- liver, kidney, lungs and stomach permanent slides.
- 12 Case studies (Renal clearance, GFR, ECG).
- 13 Demonstration of ELISA
- 14 Identification of PMN leucocytes from human peripheral blood sample by staining

Note: Minimum of ten experiments to be done.

Reference:

1. Vander's Human Physiology (2008) 11th ed., Widmaier, E.P., Raff, H. and Strang, K.T., McGraw Hill International Publications (New York), ISBN: 978-0-07-128366-3.
2. Harper's Biochemistry (2012) 29th ed., Murray, R.K., Granner, D.K., Mayes and P.A., Rodwell, V.W., Lange Medical Books/McGraw Hill. ISBN:978-0-07-176-576-3.
3. Textbook of Medical Physiology (2011) 10th ed., Guyton, A.C. and Hall, J.E., Reed Elseviers India Pvt. Ltd. (New Delhi). ISBN: 978-1-4160-4574-8.
4. Fundamental of Anatomy and Physiology (2009), 8th ed., Martini, F.H. and Nath, J.L., Pearson Publications (San Francisco), ISBN: 10:0-321-53910-9 / ISBN: 13: 978-0321-539106

SEMESTER III

CORE: ENZYMOLOGY & BIOENERGETICS

(Credits: Theory – 04, Practical – 02)

Theory: 60 Hrs

Course Outcome:

After completion of the course the student is able to;

CO1: Learn the characteristics of enzyme kinetics

CO2: Learn in depth enzyme inhibitions

CO3: Specify in detail with examples enzyme activity

CO4: Understand the classification and characteristics of bioenergetics

- Unit : 1 INTRODUCTION TO ENZYMES:** Nature of enzymes – protein and non-protein (ribozyme). Cofactor and prosthetic group, apoenzyme, holoenzyme. IUBMB classification of enzymes. **02 Hrs**
- Unit : 2 FEATURES OF ENZYME CATALYSIS:** Factors affecting the rate of chemical reactions, activation energy and transition state theory. Catalytic power and specificity of enzymes (concept of active site), Fischer's lock and key hypothesis, Koshland's induced fit hypothesis. **06 Hrs**
- Unit : 3 ENZYME KINETICS:** Relationship between initial velocity and substrate concentration, Michaelis-Menten equation, Lineweaver-Burk plot, Eadie-Hofstee and Hanes plot. K_m and V_{max} , K_{cat} and turnover number. Effect of pH, temperature and metal ions on the activity of enzyme. **08 Hrs**
- Unit : 4 ENZYME INHIBITION:** Reversible inhibition (competitive, uncompetitive, non-competitive). Mechanism based inhibitors – antibiotics as inhibitors **05 Hrs**
- Unit : 5 MECHANISM OF ACTION OF ENZYMES:** General features – proximity and orientation, strain and distortion, acid base and covalent catalysis (chymotrypsin, lysozyme). **05 Hrs**
- Unit : 6 REGULATION OF ENZYME ACTIVITY:** Control of activities of single enzymes (end product inhibition) and metabolic pathways, feedback inhibition (aspartate transcarbamoylase). Occurrence and isolation, phylogenetic distribution and properties (pyruvate dehydrogenase, fatty acyl synthase) Isoenzymes – properties and physiological significance (lactate dehydrogenase). **06 Hrs**
- Unit : 7 INVOLVEMENT OF COENZYMES IN ENZYME CATALYSED REACTIONS:** TPP, FAD, NAD, Pyridoxal Phosphate, Biotin, Coenzyme A, Tetrahydrofolate, Lipoic Acid. **06 Hrs**
- Unit: 8 APPLICATIONS OF ENZYMES:** Application of enzymes in diagnostics (SGPT, SGOT, creatine kinase, alkaline and acid **06 Hrs**

phosphatases), enzyme immunoassay (HRPO), enzyme therapy (Streptokinase). Immobilized enzymes.

Unit : 9 INTRODUCTION TO BIOENERGETICS: Laws of thermodynamics, equilibrium constant, coupled reactions, energy charge, ATP cycle, phosphorylation potential, phosphoryl group transfers. Chemical basis of high standard energy of hydrolysis of ATP, other phosphorylated compounds and thioesters. Redox reactions, standard redox potentials and Nernst equation. Universal electron carriers. **08 Hrs**

Unit: 10 OXIDATIVE PHOSPHORYLATION: Mitochondria. Electron transport chain – its organization and function. Inhibitors of ETC and uncouplers. Peter Mitchell's chemiosmotic hypothesis. Proton motive force. Fo F1ATP synthase, structure and mechanism of ATP synthesis. Metabolite transporters in mitochondria. Regulation of oxidative phosphorylation. ROS production and antioxidant mechanisms. Thermogenesis. Alternative respiratory pathways in plants. **08 Hrs**

DMC21106/ DMC21107

PRACTICALS

- 1 Determination of activity of Salivary amylase by DNS
- 2 Determination of specific activity of Salivary amylase by DNS
- 3 Determination of optimum temperature and energy of activation of Salivary amylase.
- 4 Determination of pH optimum of Salivary amylase.
- 5 Determination of Km and Vmax of Salivary amylase
- 6 Determination of initial velocity (time Kinetics) of Salivary amylase.
- 7 Effect of activators (NaCl) on Salivary amylase activity
- 8 Assay of enzyme activity of acid phosphatase.
- 9 Assay of specific activity of acid phosphatase.
- 10 Separation of photosynthetic pigments by column chromatography
- 11 Partial purification of acid phosphatase from germinating mung bean.
- 12 Enzyme inhibition – calculation of Ki for competitive inhibition.
- 13 Estimation of SGOT
- 14 Estimation of SGPT

Note: Minimum of ten experiments to be done.

Reference:

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H.Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1/ ISBN:10:1-4292-3414-8.
2. Biochemistry (2011) 4th ed., Donald, V. and Judith G.V., John Wiley & Sons Asia Pvt.Ltd. (New Jersey), ISBN:978-1180-25024.
3. Fundamentals of Enzymology (1999) 3rd ed., Nicholas C.P. and Lewis S., Oxford University Press Inc. (New York), ISBN:0 19 850229 X.

SEMESTER IV

CORE: METABOLISM

(Credits: Theory – 04, Practical – 02)

Theory: 60 Hrs

Course Outcome:

After completion of the course the student is able to;

CO1: Specify the detail of metabolism of lipids

CO2: Understand the detail of metabolism of carbohydrates

CO3: Deliberate the characteristics of metabolism of proteins

CO4: Understand the detail of metabolism of nucleic acids

- Unit : 1 INTRODUCTION TO METABOLISM:** Anabolism and catabolism, compartmentalization of metabolic pathways. **02 Hrs**
- Unit : 2 METABOLISM OF CARBOHYDRATES:** Glycolysis - reactions, regulation and energetics. Entry of other carbohydrates (Fructose, Galactose and Mannose) into glycolytic pathway. Fates of pyruvate – Under aerobic and anaerobic conditions (conversion of pyruvate to lactate, alcohol and acetyl coA). **10 Hrs**
- Unit: 3 CITRIC ACID CYCLE:** Reactions, regulation and energetics. Amphibolic and integrating roles of TCA cycle. Anaplerosis. Pentose Phosphate pathway and its significance. Cori cycle - its significance. Rapoport and luebering cycle, Gluconeogenesis pathway & their significance. **10 Hrs**
- Unit: 4 Uronic acid pathway** –Reactions and its significance. **06 Hrs**
Glyoxalate pathway – Reactions and its significance.
Glycogen metabolism – glycogenolysis and glycogen synthesis and its regulation
- Unit: 5 METABOLISM OF LIPIDS 1:** Oxidation of fatty acid - α , β and ω - oxidation of saturated fatty acids, Energetics of β - oxidation. Biosynthesis of even number saturated fatty acids, ketone bodies-formation and its significance. **08 Hrs**
- Unit: 6 METABOLISM OF LIPIDS 2:** Bio synthesis of triglycerides, glycolipids, phospholipids and spingolipids. Cholesterol - Outline of biosynthesis and its degradation **08 Hrs**
- Unit :7 METABOLISM OF AMINO ACIDS:** General reaction of amino acid degradation – Transamination, deamination and decarboxylation (oxidative and non oxidative). Ketogenic and glucogenic amino acids. Urea cycle and its significance. Biosynthesis of amino acids (Phenyl alanine and Glutamic acid) and their degradation. **08 Hrs**

Unit: 8 NUCLEIC ACID METABOLISM: Degradation of Nucleic acids by **08 Hrs**
DNase I & II, pancreatic RNase and phosphodiesterases. Biosynthesis and
degradation of purine and pyrimidine nucleotides, salvage pathways,
regulation of purine and pyrimidine synthesis.

DMD21106/ DMD21107

PRACTICALS

- 1 Glucose by DNS method.
- 2 Amino acid by Ninhydrine method.
- 3 Protein by Lowry's method.
- 4 Urea by DAMO or nitroprusside method.
- 5 Creatinine by Jaffe's method.
- 6 Inorganic Phosphate by Fiske and Subbarow's method.
- 7 Iron by Wong's method.
- 8 Ketoacids by DNPH method
- 9 Glycogen by anthrone method
- 10 Cholesterol by Zak's method
- 11 Isolation of phospholipids by egg yolk
- 12 Isolation of cholesterol from egg yolk and its estimation.
- 13 Separation of plasma proteins
- 14 Protein by Biuret method

Note: Minimum of ten experiments to be done.

Reference:

1. Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H. Freeman and Company (New York), ISBN:13:978-1-4641-0962-1 / ISBN:10:1-4641-0962-1.
2. Textbook of Biochemistry with Clinical Correlations (2011) 7th ed., Devlin, T.M., John Wiley & Sons, Inc. (New York), ISBN: 978-0-470-28173-4 / BRV ISBN: 978-0-470-60152-5.

DSE -1: NUTRITIONAL BIOCHEMISTRY**(Credits: Theory – 04, Practical – 01)****Theory: 60 Hrs****Course Outcome:****After completion of the course the student is able to;****CO1:** Understand the characteristics of energy metabolism**CO2:** Specify the characteristics of dietary carbohydrates**CO3:** Identify in detail with examples dietary lipid & health**CO4:** Understand the characteristics of minerals

- Unit : 1 INTRODUCTION TO NUTRITION & ENERGY METABOLISM 09 Hrs**
 Defining Nutrition, role of nutrients. Unit of energy, Biological oxidation of foodstuff. Measurement of energy content of food, Physiological energy value of foods, SDA.
 Measurement of energy expenditure. Direct and Indirect Calorimetry, factors affecting thermogenesis, energy utilization by cells, energy output – Basal and Resting metabolism, physical activity, factors affecting energy input - hunger, appetite, energy balance Energy expenditure in man. Estimating energy requirements, BMR factors Recommended Nutrient Intakes (RNI) and Recommended Dietary Allowances for different age groups (Adult man, women and children).
- Unit: 2 DIETARY CARBOHYDRATES & HEALTH: 07 Hrs**
 Review functions of carbohydrates. Digestion, absorption, utilization and storage, hormonal regulation of blood glucose. Dietary requirements and source of carbohydrates, blood glucose level.
- Unit :3 DIETARY LIPID & HEALTH: 08 Hrs**
 Review of classification, sources, functions, digestion, absorption, utilization and storage.
 Essential Fatty Acids; Functions of EFA, RDA, – excess and deficiency of EFA. Lipotropic factors, role of saturated fat, cholesterol, lipoprotein and triglycerides. Importance of the following: a) Omega – fatty acids. Omega 3/ omega 6 ratio b) Phospholipids c) Cholesterol in the body d) Mono, Polyunsaturated and Saturated Fatty Acids.
- Unit: 4 DIETARY PROTEINS & HEALTH: 08 Hrs**
 Review of functions of proteins in the body, Digestion and absorption. Essential and Nonessential amino acids. Amino acid Supplementation. Effects of deficiency. Food source and Recommended Dietary Allowances for different age group. Amino acid pool. NPU, Biological Value, Nitrogen balance. PEM and Kwashiorkor.
- Unit :5 FAT & WATER SOLUBLE VITAMINS: 10 Hrs**
 Vitamin A, C, E, K and D Dietary sources, RDA, Absorption, Distribution, Metabolism and excretion (ADME), Deficiency. Role of Vitamin A as an antioxidant, in Visual cycle, dermatology and immunity. Role of Vitamin K in Gamma carboxylation. Role of Vitamin E as an antioxidant. Role of Vitamin D and its effect on

bone physiology.

Hypervitaminosis. Vitamin C role as cofactor in amino acid modifications. Niacin- Metabolic interrelation between tryptophan, Niacin and NAD/ NADP. Vitamin B6-Dietary source, RDA, conversion to Pyridoxal Phosphate. Role in metabolism, Biochemical basis for deficiency symptoms. Vitamin B12 and folate; Dietary source, RDA, absorption, metabolic role Biochemical basis for deficiency symptoms.

Unit: 6 MINERALS: Calcium, Phosphorus and Iron - Distribution in the body digestion, Absorption, Utilization, Transport, Excretion, Deficiency, Sources, RDA. **09 Hrs**

Iodine, Fluoride, Mg, Cu, Zn, Se, Manganese - Absorption, Utilization, Transport, Excretion, Deficiency, Sources, RDA.

Unit: 7 ASSESSMENT OF NUTRITIONAL STATUS: Anthropometric measurements; Z scores, BMI, skinfold, circumference ratios. Biochemical assessment; Basal metabolic panel, Comprehensive metabolic panel, CBC, Urine Analysis, Assessment of Anemia, ROS assessment. **05 Hrs**

Unit: 8 FOOD & DRUG INTERACTIONS & NUTRICEUTICALS: Nutrient interactions affecting ADME of drugs, Alcohol and nutrient deficiency, Antidepressants, psychoactive drugs and nutrient interactions. **04 Hrs**

DME21106/ DME21107 PRACTICALS

- 1 Bioassay for vitamin B12/B1.
- 2 Extraction of oil from oil seeds by soxlet.
- 3 Separation of sugars by circular paper chromatography
- 4 Anthropometric identifications for Kwashiorkor, Marasmus and Obesity.
- 5 Extraction of caffeine from tea leaves
- 6 Vitamin A/E estimation in serum.
- 7 Determination of moisture content of food sample
- 8 Detection of adulterants in food.
- 9 Estimation of Calcium in ragi.
- 10 Estimation of Vitamin – C in lemon or gooseberries by DPPH method
- 11 Estimation of Lactose in milk by Benedict's method
- 12 Estimation of iron in drumsticks
- 13 Determination of iodine value of an oil or fat
- 14 Determination of saponification value of an oil or fat

Note: Minimum of eight experiments to be done.

Reference:

1. Textbook of Biochemistry with Clinical Correlations (2011) Devlin, T.M. John Wiley & Sons, Inc. (New York), ISBN: 978-0-4710-28173-4.
2. Nutrition for health, fitness and sport (2013) Williams.M.H,Anderson,D.E, Rawson,E.S. McGraw Hill international edition. ISBN-978-0-07-131816-7.

3. Krause's Food and Nutrition Care process (2012); Mahan, L.K. Stang, S.E., Raymond, J. Elsevier's Publications. ISBN- 978-1-4377-2233-8.
4. The vitamins, Fundamental aspects in Nutrition and Health (2008); G.F. Coombs Jr. Elsevier's Publications. ISBN-13- 978-0-12- 183493-7.
5. Principles of Nutritional Assessment (2005) Rosalind Gibson. Oxford University Press.

DSE -2: MOLECULAR BASIS OF INFECTIOUS DISEASES**(Credits: Theory – 04, Practical – 01)****Theory: 60 Hrs****Course Outcome:**

After completion of the course the student is able to;

CO1: Deliberate in depth infectious agents**CO2:** Learn the detail of bacterial diseases**CO3:** Understand the detail of viral diseases**CO4:** Learn the characteristics of fungal diseases

- Unit : 1 CLASSIFICATION OF INFECTIOUS AGENTS:** Bacteria, Viruses, protozoa and fungi. Past and present emerging and re-emerging infectious diseases and pathogens. Source, reservoir and transmission of pathogens, Antigenic shift and antigenic drift. Host parasite relationship, types of infections associated with parasitic organisms. Overview of viral and bacterial pathogenesis. Infection and evasion **12 Hrs**
- Unit: 2 OVERVIEW OF DISEASES CAUSED BY BACTERIA:** Detailed study of tuberculosis: History, causative agent, molecular basis of host specificity, infection and pathogenicity, Diagnostics, Therapeutics, inhibitors and vaccines. Drug resistance and implications on public health. Other bacterial diseases including Typhoid, Diphtheria, Pertussis, Tetanus, Typhoid and Pneumonia. **18 Hrs**
- Unit: 3 OVERVIEW OF DISEASES CAUSED BY VIRUSES:** Detailed study of AIDS, history, causative agent, pathogenesis, Diagnostics, Drugs and inhibitors. Other viral diseases including hepatitis, influenza, rabies, chikungunya and polio. **12 Hrs**
- Unit: 4 OVERVIEW OF DISEASES CAUSED BY PARASITES:** Detailed study of Malaria, history, causative agents, Vectors, life cycle, Host parasite interactions, Diagnostics, Drugs and Inhibitors, Resistance, Vaccine development. Other diseases including leishmaniasis, amoebiasis. **08 Hrs**
- Unit :5 OVERVIEW OF DISEASES CAUSED BY OTHER ORGANISMS:** **10 Hrs**
Fungal diseases, General characteristics. Medical importance of major groups, pathogenesis, treatment.

- 1 WIDAL test
- 2 Gram staining
- 3 Acid fast staining
- 4 PCR based diagnosis
- 5 Preparation of media.

- 6 Isolation of bacteria from soil
- 7 Isolation of bacteria from water
- 8 Structural staining- (cellwall and endospore of bacteria).
- 9 Simple staining and Negative staining
- 10 Effect of temperature on growth of microorganisms
- 11 Effect of pH on growth of microorganisms
- 12 Demonstration of Bacterial Conjugation
- 13 Demonstration of bacterial transformation and transduction
- 14 Photographic demonstration of infectious diseases

Note: Minimum of eight experiments to be done.

Reference:

1. Prescott, Harley, Klein's Microbiology (2008) 7th Ed., Willey, J.M., Sherwood, L.M., Woolverton, C.J. Mc Graw Hill International Edition (New York) ISBN: 978-007-126727.
2. Mandell, Douglas and Bennett.S, Principles and practices of Infectious diseases, 7th edition, Volume, 2. Churchill Livingstone Elsevier.
3. Sherris Medical Microbiology: An Introduction to Infectious Diseases by Kenneth J.Ryan, C. George Ray, Publisher: McGraw-Hill.
4. Medical Microbiology by Patrick R. Murray, Ken S. Rosenthal, Michael A. Pfaller, Elsevier Health Sciences

SEMESTER V

SEC-1: TOOLS AND TECHNIQUES IN BIOCHEMISTRY

(Credits: Theory – 02)

Theory: 30 Hrs

Course Outcome:

After completion of the course the student is able to;

CO1: Understand in depth chromatography

CO2: Learn in depth electrophoresis technique

CO3: Deliberate the characteristics of centrifugation

CO4: Understand in detail with examples spectrophotometry

- Unit : 1 BIOCHEMICAL REAGENTS & SOLUTIONS:** Safety practices in the laboratory. Preparation and storage of solutions. Concepts of solution concentration and storing solutions. Quantitative transfer of liquids. Concept of a buffer, Henderson-Hasselbach equation, working of a pH meter. **06 Hrs**
- Exercise**
Preparation of a buffer of given pH and molarity.
- Unit : 2** Chromatography- Definition, types, Principles of Adsorption and Partition chromatography. Techniques of circular, 2D chromatography, Thin Layer Chromatography- and its advantages
Column chromatography – Principle and applications of Gel Filtration chromatography, HPLC **07 Hrs**
- Unit : 3** Electrophoresis: Principle and applications of electrophoresis technique- PAGE, SDS - PAGE **03 Hrs**
- Unit : 4** Centrifugation: Principle of differential and density gradient centrifugation. Ultra centrifuge – construction and applications **03 Hrs**
- Unit: 5 SPECTROPHOTOMETRIC TECHNIQUES:** Principle and instrumentation of UV-visible and fluorescence spectroscopy. **05 Hrs**
- Exercises**
Determination of the absorption maxima and molar extinction coefficient (of a relevant organic molecule).
Measurement of fluorescence spectrum.
Determination of concentration of a protein solution by Lowry/BCA method.
- Unit :6 Introduction and importance of virtual labs in biochemistry** **06 Hrs**

Reference:

1. Physical Biochemistry: Principles and Applications (2010) 2nd ed., Sheehan, D., Wiley Blackwell (West Sussex), ISBN:978-0-470-85602-4 / ISBN:978-0-470-85603-1.
2. Physical Biochemistry: Applications to Biochemistry and Molecular Biology (1982) 2nd ed., Freifelder, D., W.H. Freeman and Company, ISBN:0-7167-1315-2/ISBN:0-7167-1444-2.
3. An Introduction to Practical Biochemistry (1998) 3rd ed., Plummer D. T., Tata McGraw Hill Education Pvt. Ltd. (New Delhi), ISBN:13: 978-0-07-099487-4 / ISBN:10: 0-07-099487-0.

SEMESTER V

SEC-2: CLINICAL BIOCHEMISTRY

(Credits: Theory – 02)

Theory: 30 Hrs

Course Outcome:

After completion of the course the student is able to;

CO1: Specify the characteristics of clinical laboratory

CO2: Identify in depth blood glucose

CO3: Deliberate the detail of lipid profile

CO4: Learn in detail with examples cardiovascular diseases

- Unit : 1 INTRODUCTION:** Organization of clinical laboratory, Introduction to instrumentation and automation in clinical biochemistry laboratories safety regulations and first aid. General comments on specimen collection, types of specimen for biochemical analysis. Precision, accuracy, quality control, precautions and limitations. **04 Hrs**
- Exercises**
Collection of blood and storage.
Separation and storage of serum.
- Unit :2 EVALUATION OF BIOCHEMICAL CHANGES IN DISEASES:** Basic hepatic, renal and cardiovascular physiology. Biochemical symptoms associated with disease and their evaluation. Diagnostic biochemical profile. **04 Hrs**
- Unit: 3 ASSESSMENT OF GLUCOSE METABOLISM IN BLOOD:** Clinical significance of variations in blood glucose. Diabetes mellitus. **04 Hrs**
- Exercises**
Estimation of blood glucose by glucose oxidase peroxidase method.
- Unit :4 LIPID PROFILE:** Composition and functions of lipoproteins. Clinical significance of elevated lipoprotein. **04 Hrs**
- Exercises**
Estimation of triglycerides.
- Unit :5 LIVER FUNCTION TESTS** **04 Hrs**
- Exercises**
Estimation of bilirubin (direct and indirect).
- Unit: 6 RENAL FUNCTION TESTS & URINE ANALYSIS:** Use of urine strip / dipstick method for urine analysis. **06 Hrs**
- Exercises**
Quantitative determination of serum creatinine and urea.
- Unit: 7 TESTS FOR CARDIOVASCULAR DISEASES:** Involvement of enzymes in diagnostics of heart disease including aspartate **04 Hrs**

transaminase, isoenzymes of creatine kinase and lactate dehydrogenase and troponin.

Exercises

Estimation of creatine kinase MB.

Reference:

1. Medical Laboratory Technology - a Procedure Manual for Routine Diagnostic Tests Vol.I (2010), Mukherjee, K.L., Tata Mc Graw–Hill Publishing Company Limited (New Delhi). ISBN: 9780070076594 / ISBN: 9780070076631.
2. Medical Laboratory Technology - a Procedure Manual for Routine Diagnostic Tests Vol. II (2010), Mukherjee, K.L., Tata Mc Graw – Hill Publishing Company Ltd. (New Delhi), ISBN: 9780070076648.
3. Medical Biochemistry (2005) 2nd ed., Baynes, J.W. and Dominiczak, M.H., Elsevier Mosby Ltd. (Philadelphia), ISBN:0-7234-3341-0.

SEMESTER VI

DSE-1: ADVANCED CELL BIOLOGY AND ENDOCRINOLOGY

(Credits: Theory – 04, Practical – 01)

Theory: 60 Hrs

Course Outcome:

After completion of the course the student is able to;

CO1: Learn the detail of cell membrane

CO2: Understand in detail with examples cell-cell interactions

CO3: Identify in depth endocrine organs

CO4: Deliberate the detail of immunohistochemistry

- | | | |
|-----------------|--|---------------|
| Unit : 1 | PLASMA MEMBRANE & NUCLEAR TRANSPORT: Properties and Composition of Cell Membrane; Structure of Nuclear Envelope; Nuclear Pore Complex; Transport Across Nuclear Envelope; Regulation of Nuclear Protein Import and Export. | 08 Hrs |
| Unit: 2 | CELL-CELL INTERACTION: Cell-Cell Interactions and Cell-Matrix Interactions; Components of Extracellular Matrix: Collagen and Non-Collagen Components; Tight Junctions; Gap Junctions; Desmosomes; Hemidesmosomes; Focal Adhesions And Plasmodesmata; Cell Wall; Role Of Cell Interaction In Development. | 10 Hrs |
| Unit :3 | CELL CYCLE & PROGRAMMED CELL DEATH: Overview of The Cell Cycle; Eukaryotic Cell Cycle; Events Of Mitotic Phase; Cytokinesis; Events Of Meiosis And Fertilization; Regulation Of Cell Division And Cell Growth; Apoptosis And Necrosis, Stem Cells And Maintenance of Adult Tissues, Hematopoiesis, Embryonic Stem Cells and Therapeutic Cloning. | 12 Hrs |
| Unit :4 | CANCER BIOLOGY: Development and causes Of Cancer; Genetic Basis of Cancer; Oncogenes, Tumor Viruses; Molecular Approach to Cancer Treatment. | 10 Hrs |
| Unit: 5 | ADVANCED METHODS IN CELL BIOLOGY: Ultracentrifugation, Fluorescence Microscopy- FACS, Confocal Microscopy, Electron Microscopy, Plant and Animal Cell Culture, Immunohistochemistry. | 10 Hrs |
| Unit: 6 | ENDOCRINE SYSTEM: Endocrine organs, hormones-classification, Hierarchy, interplay, dynamic balance and regulation of their secretions. Functions of the hormones of Hypothalamus, Pituitary, Adrenal, Thyroid, pancreas and Gonads. Concept of receptors- Membrane and cytosolic. Mechanism of hormone action - Steroid hormone and Peptide hormone (second messengers hypothesis. Eg: cAMP, DAG, IP3). | 10 Hrs |

PRACTICALS

- 1** Isolation of WBC using fical reagent method
- 2** Study of cell viability /death assay by use of trypan blue and MTT assay.
- 3** Study of apoptosis through analysis of DNA fragmentation patterns in mitochondria.
- 4** Identification and study of cancerous cells using permanent slides and photomicrographs.
- 5** Antigen-Antibody reaction
- 6** Cell counting methods: counting using Haemocytometer.
- 7** Calibration of ocular micrometer and Measurement of average cell size using Stage micrometer
- 8** Separation of cell organelles by differential centrifugation and assay of marker enzymes. (2 Practicals)
- 9** Photographic demonstration of different types of cells.
- 10** Isolation of chloroplast by differential centrifugation & its identification.
- 11** Buccal smear- barr bodies
- 12** Identification of normal & abnormal karyotype using permanant slides
- 13** Estimation of iodine by titrometric method from commercial salt
- 14** Industrial /Institution visit

Note: Minimum of eight experiments to be done.

Reference:

- 1.** Cooper, G.M. and Hausman, R.E. 2009 The Cell: A Molecular Approach. 5th edition ASM Press & Sunderland, Washington, D.C.; Sinauer Associates, MA.
- 2.** Karp, G. 2010 Cell and Molecular Biology: Concepts and Experiments. 6 edition. JohnWiley & Sons. Inc.
- 3.** Alberts, B., Johnson,A., Lewis, J., and Enlarge, M. 2008 Molecular Biology of the Cell.5th ed., Garland Science (Princeton).
- 4.** Lodish, H., Berk, A., Zipursky, S.L., Matsudaira, P., Baltimore, D. and Darnell. J. 2012. Molecular Cell Biology. 7th ed., W.H. Freeman & Company (New York).

SEMESTER VI

DSE-2: PLANT BIOCHEMISTRY

(Credits: Theory – 04, Practical – 01)

Theory: 60 Hrs

Course Outcome:

After completion of the course the student is able to;

CO1: Specify the characteristics of plant cell structure

CO2: Deliberate in detail with examples photosynthesis

CO3: Understand the detail of nitrogen metabolism

CO4: Learn in detail with examples secondary metabolites

Unit : 1	INTRODUCTION TO PLANT CELL STRUCTURE: Plasma membrane, Vacuole and tonoplast membrane, cell wall, plastids and peroxisomes.	5Hr
Unit :2	PHOTOSYNTHESIS & CARBON ASSIMILATION: Structure of PSI and PSII complexes, Light reaction, Cyclic and non cyclic photophosphorylation, Calvin cycle and regulation; C4 cycle and Crassulacean acid metabolism (CAM), Photorespiration.	14Hr
Unit :3	NITROGEN METABOLISM: Biological Nitrogen fixation by free living and in symbiotic association, structure and function of enzyme Nitrogenase. Nitrate assimilation: Nitrate and Nitrite reductase. Primary and secondary ammonia assimilation in plants; ammonia assimilation by Glutamine synthetase-glutamine oxoglutarate amino transferase (GS-GOGAT) pathway. Seed storage proteins in legumes and cereals.	14Hr
Unit :4	REGULATION OF PLANT GROWTH: Introduction to plant hormones and their effect on plant growth and development, Regulation of plant morphogenetic processes by light.	7Hr
Unit :5	SECONDARY METABOLITES: Representatives alkaloid group and their amino acid precursors, function of alkaloids, Examples of major phenolic groups; simple phenylpropanoids, Coumarins, Benzoic acid derivatives, flavonoids, tannins and lignin, biological role of plant phenolics, Classification of terpenoids and representative examples from each class, biological functions of terpenoids.	12Hr
Unit :6	PLANT TISSUE CULTURE: Cell and tissue culture techniques, types of cultures: organ and explants culture, callus culture, cell suspension culture and protoplast culture. Plant regeneration pathways: organogenesis and somatic embryogenesis. Applications of cell and tissue culture and somoclonal variation.	8Hr

PRACTICALS

- 1 Induction of hydrolytic enzymes β -galactosidase during germination
- 2 Extraction and assay of Urease from Jack bean
- 3 Estimation of carotene in fruits and vegetables
- 4 Estimation of ascorbic acid in fruits and vegetables
- 5 Estimation of phenols in fruits and vegetables
- 6 Estimation of tannins in fruits and vegetables
- 7 Culture of plant plants (explants)
- 8 Extraction of genomic DNA from coconut endosperm, quantification & determination of its purity by Spectrophotometric method.(2 Practicals)
- 9 Extraction of RNA from spinach leaves
- 10 Estimation of DNA by DPA method
- 10 Estimation of RNA by orcinol method
- 11 Extraction of Starch from potato
- 12 Extraction Casein from milk
- 13 Extraction of polyphenols
- 14 Industrial /Institution visit

Note: Minimum of eight experiments to be done.

Reference:

1. Plant Biochemistry (2008), Caroline Bowsher, Martin steer, Alyson Tobin, Garland science ISBN 978-0-8153-4121-5.
2. Biochemistry and molecular Biology of plant-Buchanan. (2005) 1 edition. Publisher: IK International. ISBN-10: 8188237116, ISBN-13: 978-8188237111.
3. Plant Biochemistry by P.M Dey and J.B. Harborne (Editors) (1997) Publisher: Academic Press ISBN-10:0122146743, ISBN-13:978-01221467.

Pattern of Question Paper
Semester I to VI
Paper I to V and VII (DSC and DSE)

Time : 3 Hrs

Max Marks: 70

1. Answer all the questions

5 X 1 = 5

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

2. Answer any five of the following questions

5 X 3 = 15

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----
- f. -----
- g. -----

3. Answer any four of the following questions

4 X 5 = 20

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----
- f. -----

4. Answer any three questions of the following

3 X 10 = 30

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

(Note- 10 Marks may be divided in to 6+4 or 5+5)

Pattern of Question Paper
Semester V
Paper VI (SEC)

Time : 2 Hrs

Max Marks: 35

1. Answer all the questions

5 X 1 = 5

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

2. Answer any five of the following questions

5 X 3 = 15

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----
- f. -----
- g. -----

3. Answer any three of the following questions

3 X 5 = 15

- a. -----
- b. -----
- c. -----
- d. -----
- e. -----

**JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE
(AUTONOMOUS)
OOTY ROAD, MYSORE – 25.**



DEPARTMENT OF BIOTECHNOLOGY

SCHEMATIC SYLLABUS UNDER CHOICE BASED CREDIT SYSTEM (CBCS)

For B.Sc., programmes

Chemistry, Zoology and Biotechnology

Biochemistry, Microbiology and Biotechnology

2019-20

B.Sc., UG SYLLABUS- PROGRAMME – CZBt

Scheme of study for B.Sc. Biotechnology under CBCS scheme from 2019-20

YEAR	SEMESTER	CORE COURSE	COURSE CODE	TITLE OF THE PAPER	NO. OF CREDITS	LECTURE/ PRACTICAL/ HOUR/WEEK	TOTAL TEACHING HOURS
I BSc	I	DSC -I:Theory	DMA22005	Cell biology & genetics	4	4	60
		DSC -I:Pract	DMA22105	Cell biology & genetics	2	4	60
	II	DSC-II:Theory	DMB22005	Biomolecules and Bio-analytical techniques	2	4	60
		DSC-II: Pract	DMB22105	Biomolecules and Bio-analytical techniques	2	4	60
II BSc	III	DSC-III:Theory	DMC22005	Molecular biology and Genetic engineering	2	4	60
		DSC-III: Pract	DMC22105	Molecular biology and Genetic engineering	2	4	60
	IV	DSC-IV:Theory	DMD22005	Plant Tissue culture and Animal Cell culture	2	4	60
		DSC-IV: Pract	DMD22105	Plant Tissue culture and Animal Cell culture	2	4	60
III BSc	V	DSE Elective 1: Theory	DME22005	Elective 1: Immunology and Medical Biotechnology	2	4	60
		Elective 1: Pract	DME22105	Elective 1: Immunology and Medical Biotechnology	2	4	60
		Elective 2: Theory	DME22205	Elective 2: Microbial technology and Agricultural Biotechnology	2	4	60
		Elective 2: Pract	DME22305	Elective 2: Microbial technology and Agricultural Biotechnology	2	4	60
		SEC 1	DME22405	SEC 1:Microbial Techniques	2	2	30
		SEC 2	DME22505	SEC 2:Enzymology	2	2	30
	VI	DSE Elective 1: Theory	DMF22005	Elective 1: Environmental Biotechnology and Biostatistics	2	4	60
		Elective 1: Pract	DMF22105	Elective 1: Environmental Biotechnology and Biostatistics	2	4	60
		Elective 2: Theory	DMF22205	Elective 2: Bioinformatics and Bioprocess technology	2	4	60
		Elective 2: Pract	DMF22305	Elective 2: Bioinformatics and Bioprocess technology	2	4	60

B.Sc., UG SYLLABUS- PROGRAMME – BMBt

Scheme of study for B.Sc. Biotechnology under CBCS scheme from 2019-20

YEAR	SEMESTER	CORE COURSE	COURSE CODE	TITLE OF THE PAPER	NO. OF CREDITS	LECTURE/ PRACTICAL/ HOUR/WEEK	TOTAL TEACHING HOURS
I BSc	I	DSC -I:Theory	DMA22006	Cell biology & genetics	4	4	60
		DSC -I:Pract	DMA22106	Cell biology & genetics	2	4	60
	II	DSC-II:Theory	DMB22006	Biomolecules and Bio-analytical techniques	2	4	60
		DSC-II: Pract	DMB22106	Biomolecules and Bio-analytical techniques	2	4	60
II BSc	III	DSC-III:Theory	DMC22006	Molecular biology and Genetic engineering	2	4	60
		DSC-III: Pract	DMC22106	Molecular biology and Genetic engineering	2	4	60
	IV	DSC-IV:Theory	DMD22006	Plant Tissue culture and Animal Cell culture	2	4	60
		DSC-IV: Pract	DMD22106	Plant Tissue culture and Animal Cell culture	2	4	60
III BSc	V	DSE Elective 1: Theory	DME22006	Elective 1: Immunology and Medical Biotechnology	2	4	60
		Elective 1: Pract	DME22106	Elective 1: Immunology and Medical Biotechnology	2	4	60
		Elective 2: Theory	DME22206	Elective 2: Microbial technology and Agricultural Biotechnology	2	4	60
		Elective 2: Pract	DME22306	Elective 2: Microbial technology and Agricultural Biotechnology	2	4	60
		SEC 1: Theory	DME22406	SEC 1:Microbial Techniques	2	2	30
		SEC 2: Theory	DME22506	SEC 2:Enzymology	2	2	30
	VI	DSE Elective 1: Theory	DMF22006	Elective 1: Environmental Biotechnology and Biostatistics	2	4	60
		Elective 1: Pract	DMF22106	Elective 1: Environmental Biotechnology and Biostatistics	2	4	60
		Elective 2: Theory	DMF22206	Elective 2: Bioinformatics and Bioprocess technology	2	4	60
		Elective 2: Pract	DMF22306	Elective 2: Bioinformatics and Bioprocess technology	2	4	60

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE, OOTY ROAD, MYSORE
Scheme of Examination Programme – B.Sc., CZBt ; Programme code –BSC05

Year	Semester	Core course	Course code	Title of the paper	credits	Maximum Marks in exam/Assessment				Exam Duration			
						L:T:P	IA(Theory)			Total	Th	Pr	
							C-1	C-2	C-3				
I B.Sc	I	DSC-I :Theory	DMA22005	Cell Biology and Genetics	4: 0: 0	15	15	70	100	3h	3h		
		DSC-I: Pract	DMA22105	Cell Biology and Genetics	0: 0: 2	7.5	7.5	35	50				
	II	DSC-II:Theory	DMB22005	Biomolecules and Bio-analytical techniques	4:0:0	15	15	70	100	3h	3h		
		DSC-II: Pract	DMB22105	Biomolecules and Bio-analytical techniques	0:0:2	7.5	7.5	35	50				
II B.Sc	III	DSC-III:Theory	DMC22005	Molecular biology and Genetic engineering	4:0:0	15	15	70	100	3h	3h		
		DSC-III: Pract	DMC22105	Molecular biology and Genetic engineering	0:0:2	7.5	7.5	35	50				
	IV	DSC-IV: Theory	DMD22005	Plant Tissue culture and Animal Cell culture	4:0:0	15	15	70	100	3h	3h		
		DSC-IV: Pract	DMD22105	Plant Tissue culture and Animal Cell culture	0:0:2	7.5	7.5	35	50				
III B.Sc.	V	DSE: Elective 1:Theory	DME22005	Elective 1: Immunology and Medical Biotechnology	4:0:0	15	15	70	100	3h	3h		
		Elective 1: Pract	DME22105	Elective 1: Immunology and Medical Biotechnology	0:0:2	7.5	7.5	35	50				
		Elective 2: Theory	DME22205	Elective 2: Microbial technology and Agricultural Biotechnology	4:0:0	15	15	70	100				
		Elective 2: Pract	DME22305	Elective 2: Microbial technology and Agricultural Biotechnology	0:0:2	7.5	7.5	35	50				
		SEC 1: Theory	DME22405	SEC 1:Microbial Techniques	2:0:0	7.5	7.5	35	50			2h	-
		SEC 2: Theory	DME22505	SEC 2:Enzymology	2:0:0	7.5	7.5	35	50			2h	-
	VI	DSE: Elective 1:Theory	DMF22005	Elective 1: Environmental Biotechnology and Biostatistics	4:0:0	15	15	70	100	3h	3h		
		Elective 1: Pract	DMF22105	Elective 1: Environmental Biotechnology and Biostatistics	4:0:0	15	15	70	100				
		Elective 2: Theory	DMF22205	Elective 2: Bioinformatics and Bioprocess technology	4:0:0	15	15	70	100				
		Elective 2: Pract	DMF22305	Elective 2: Bioinformatics and Bioprocess technology	4:0:0	15	15	70	100				

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE, OOTY ROAD, MYSORE
Scheme of Examination Programme – B.Sc., BMBt ; Programme code –BSC06

Year	Semester	Core course	Course code	Title of the paper	credits	Maximum Marks in exam/Assessment				Exam Duration			
						L:T:P	IA(Theory)			Total	Th	Pr	
							C-1	C-2	C-3				
I B.Sc	I	DSC-I :Theory	DMA22006	Cell Biology and Genetics	4: 0: 0	15	15	70	100	3h	3h		
		DSC-I: Pract	DMA22106	Cell Biology and Genetics	0: 0: 2	7.5	7.5	35	50				
	II	DSC-II:Theory	DMB22006	Biomolecules and Bio-analytical techniques	4:0:0	15	15	70	100	3h	3h		
		DSC-II: Pract	DMB22106	Biomolecules and Bio-analytical techniques	0:0:2	7.5	7.5	35	50				
II B.Sc	III	DSC-III:Theory	DMC22006	Molecular biology and Genetic engineering	4:0:0	15	15	70	100	3h	3h		
		DSC-III: Pract	DMC22106	Molecular biology and Genetic engineering	0:0:2	7.5	7.5	35	50				
	IV	DSC-IV: Theory	DMD22006	Plant Tissue culture and Animal Cell culture	4:0:0	15	15	70	100	3h	3h		
		DSC-IV: Pract	DMD22106	Plant Tissue culture and Animal Cell culture	0:0:2	7.5	7.5	35	50				
III B.Sc.	V	DSE: Elective 1:Theory	DME22006	Elective 1: Immunology and Medical Biotechnology	4:0:0	15	15	70	100	3h	3h		
		Elective 1: Pract	DME22106	Elective 1: Immunology and Medical Biotechnology	0:0:2	7.5	7.5	35	50				
		Elective 2: Theory	DME22206	Elective 2: Microbial technology and Agricultural Biotechnology	4:0:0	15	15	70	100				
		Elective 2: Pract	DME22306	Elective 2: Microbial technology and Agricultural Biotechnology	0:0:2	7.5	7.5	35	50				
		SEC 1: Theory	DME22406	SEC 1:Microbial Techniques	2:0:0	7.5	7.5	35	50			2h	-
		SEC 2: Theory	DME22506	SEC 2:Enzymology	2:0:0	7.5	7.5	35	50			2h	-
	VI	DSE: Elective 1:Theory	DMF22006	Elective 1: Environmental Biotechnology and Biostatistics	4:0:0	15	15	70	100	3h	3h		
		Elective 1: Pract	DMF22106	Elective 1: Environmental Biotechnology and Biostatistics	4:0:0	15	15	70	100				
		Elective 2: Theory	DMF22206	Elective 2: Bioinformatics and Bioprocess technology	4:0:0	15	15	70	100				
		Elective 2: Pract	DMF22306	Elective 2: Bioinformatics and Bioprocess technology	4:0:0	15	15	70	100				

Programme Outcomes for Bachelor of Science in Chemistry, Zoology and Biotechnology:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate the ability to justify, explain, and/or approach the concept both in written and oral forms
- PO2. Demonstrate the ability to present clear, logical and succinct arguments
- PO3. Develop state-of-the-art laboratory skills and professional communication skills.
- PO4. Apply the scientific method to design, execute, and analyze an experiment.
- PO5. Appreciate the central role of chemistry in the society and use this as a basis for ethical behaviour in issues facing chemists/drugs.
- PO6. Understand Chemistry as an integral part for addressing social, economic, and environmental problems.
- PO7. Identify the major groups of organisms with an emphasis on animals and plants.
- PO8. Compare and contrast the characteristics of animals that differentiate themselves from other living and non-living creatures.
- PO9. Give specific examples of physiological adaptations.
- PO10. Design and develop solution to Biotechnology problems keeping in mind the safety measures for environment and society.
- PO11. Support Biotechnology research activity with strong technical background knowledge.

Programme Outcomes for Bachelor of Science in Biochemistry, Microbiology and Biotechnology:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate the ability to justify and explain their thinking approach, both written and oral.
- PO2. Develop state-of-the-art laboratory skills and professional communication skills.
- PO3. Apply the scientific method to design, execute, and analyze an experiment, to explain their scientific procedures and experimental observations.
- PO4. Demonstrate an understanding of fundamental biochemical principles, structure and biological function and metabolic pathways.
- PO5. Work as a laboratory technician, biochemists or medical scientist.
- PO6. Describe/ explain the processes used by microorganisms for their replication, survival, and interaction with their environment and host populations.
- PO7. Explain the theoretical basis of the tools, technologies and methods common to microbiology.
- PO8. Design and develop solution to Biotechnology problems by applying appropriate tools while keeping in mind safety factor for environment & society.
- PO9. Create, select, and apply appropriate techniques, resources, and modern tools with an understanding of the limitations.
- PO10. Support biotechnology research activity with strong technical background knowledge.

Programme Specific Outcomes

Bachelor of Science in Chemistry, Zoology and Biotechnology:

After completing the graduation in the Bachelor of Science the students are able to:

PSO1. Find jobs at all level of chemical, pharmaceutical, food products and life oriented material Industries

PSO2. Apply appropriate techniques for the qualitative and quantitative analysis of chemicals in laboratories and in industries.

PSO3. Recognize the relationship between different structures and functions at different levels.

PSO4. Characterize the biological, chemical and physical features of environments that Animals inhabit.

PSO5. Demonstrate effectively the applications of biochemical and biological sciences.

PSO6. Know and apply appropriate tools and techniques in biotechnological manipulation

PSO7. Understand his or her responsibilities in biotechnological practices.

Programme Specific Outcomes

Bachelor of Science in Biochemistry, Microbiology and Biotechnology:

After completing the graduation in the Bachelor of Science the students are able to;

PSO 1: Gain and understand biochemical and molecular processes that occur in and between cells to expand understanding of biology

PSO2: Be knowledgeable in proper procedures and regulations in handling and disposal of chemicals.

PSO3: Communicate scientific information effectively, especially relating to microbes and their role in ecosystem and health related issues.

PSO4: Acquire, articulate, retain and demonstrate laboratory safety skills applicable to microbiological research or clinical methods.

PSO5: Demonstrate effectively the applications of biochemical and biological sciences

PSO6: Decide and apply appropriate tools and techniques in biotechnological manipulation.

POS7: Justify societal, health, safety and legal issues and understand his or her responsibilities in biotechnological practices.

**SEMESTER I
CELL BIOLOGY AND GENETICS**

(4 CREDITS)

Course Outcomes:

After completing the course students are able to:

CO1.Develop an understanding of the structure and functions of organelles.

CO2. Understand the structure of chromosomes, types, cell differentiation and features of cancer cells.

CO3. Gain comprehensive understanding of the chemical basis of heredity and methods.

CO4.Understand effect of mutation, mechanism and Chromosomal Aberrations.

CELL BIOLOGY

NO. HOURS

UNIT I

15

Cell: Introduction and Historical perspective, the cell theory, ultra structure of plant and animal cell.

Cell organelles: Structure and functions of – cell wall, plasma membrane, membrane protein, cytoplasm, mitochondria, chloroplast, Golgi complex, endoplasmic reticulum, ribosome, lysosomes, peroxisomes, nucleus.

Cell division: Cell cycle and phases, regulation of cell cycle and significances of cell cycle. Mitosis and meiosis, interphase nucleus, achromatic apparatus, synaptosomal complex and significances of mitosis and meiosis.

UNIT II

15

Eukaryotic chromosomes: Types, chromatin structure, nucleosomes, and higher order chromatin organization.

Special chromosomes – Polytene and B chromosome, lamp brush chromosome.

Cell interaction and motility: Cell motility flagellar and ciliary motion. Structure and function of muscle cells, muscle contraction, nerve cell structure and function.

Stem cells, differentiation of stem cells (eg: Haematopoietic stem cells) and their application. Special cells : Blood cells, identification, structure and different types of blood cells and cancer cells.

GENETICS

UNIT III

15

Introduction: Historical developments in the field of genetics. Organisms suitable for genetic experimentation and their genetic significance. Mendelian Theory: Law of dominance, Mendel's experimental on monohybrid and di-hybrid crosses, Law of segregation & Principle of independent assortment. Verification of segregates by test and back crosses. Deviation to Mendelian inheritance of genes (13:3 ratio), incomplete dominance (Flower colour in sweet peas), co dominance (Blood groups in human beings), epistasis (Dominant & recessive epistasis). Sex-linked inheritance (colour blindness), linkage, crossing over and cytoplasmic inheritance (Plastid inheritance in *Mirabilis*).

UNIT IV

15

Mutation: Natural and induced mutations, chemical, physical and biological mutagens with an example each.

Structure and characteristics of bacterial and eukaryotic chromosome, chromosome morphology, concept of euchromatin and heterochromatin.

Chromosomal aberrations: Deletion, duplication, inversion and translocation. Chromosomal disorders in human beings, abnormalities – Aneuploidy and Euploidy.

PRACTICALS

(2 CREDITS)

HOURS : 4 HOURS /WEEK

1. Cell counting methods: using Haemocytometer.
2. Measurements with the help of light microscope.
 - a. Calibration of ocular micrometer
 - b. Measurement of biological materials (cells/spores etc.).
 - c. Demonstration-Separation of cell organelles by differential centrifugation
3. Study of Mitosis -onion root tips.
4. Study of Meiosis –onion flowers buds/rheo flowers
5. Demonstration of plasmolysis and deplasmolysis
6. Isolation of chloroplast from leaves
7. Study of at least five simple mutants of Drosophila-Photographic demonstration
8. Preparation of polytene chromosome from salivary glands of Drosophila
9. Genetic Problems; Monohybrid, Di hybrid and interactions of Genes
10. Special Chromosomes; Lampbrush and Polytene chromosomes
11. Comment (Types of chromosome (slide/picture), chromosomal disorders in humans- Humans -Down's Turner's and Klinefelter's Syndrome

REFERENCES

1. Karp, G. 2010. Cell and Molecular Biology: Concepts and Experiments. 6th Edition. John Wiley & Sons. Inc.
2. De Robertis, E.D.P. and De Robertis, E.M.F. 2006. Cell and Molecular Biology. 8th edition. Lippincott Williams and Wilkins, Philadelphia.
3. Cooper, G.M. and Hausman, R.E. 2009. The Cell: A Molecular Approach. 5th edition. ASM Press & Sunderland, Washington, D.C.; Sinauer Associates, MA.
4. Gardner, E.J., Simmons, M.J., Snustad, D.P. (2006). Principles of Genetics. VIII Edition John Wiley & Sons.
5. Snustad, D.P., Simmons, M.J. (2009). Principles of Genetics. V Edition. John Wiley and Sons Inc.
6. Klug, W.S., Cummings, M.R., Spencer, C.A. (2009). Concepts of Genetics. IX Edition. Benjamin Cummings.

SEMESTER II
BIOMOLECULES & BIO-ANALYTICAL TECHNIQUES **(4 CREDITS)**

Course Outcomes:

After completing the course students are able to:

- CO1.** Understand the properties, mechanisms and biological importance of Bio-molecules .
- CO2.** Comprehend the mechanism of enzyme action, factors affecting it and its applications.
- CO3.** Understand and able to relate the principles underlying various instruments in the field of Biology.
- CO4.** Compare and contrast the role of bio -molecules and enzymes.

BIOMOLECULES

NO. HOURS

UNIT I:

15

Carbohydrates: Structure (Fischer and Haworth structure), function and properties of Monosaccharide's (Glucose, Fructose), disaccharides (Sucrose, Maltose and Lactose) and Heteropolysaccharide's- hyaluronic acid and heparin. Reducing and Non reducing Sugars, Stereochemistry- Epimers, Enantiomers, Anomers and Isomers.
Proteins: Amino acids- General structure , essential and non essential amino acids, , classification based on polarity , zwitter ionic structure, pka value. D and L amino acids, optical activity. Peptide bond, primary, secondary, tertiary and quaternary structural organization of proteins. Globular and fibrous proteins with special reference to structure of haemoglobin and collagen.

UNIT II:

15

Lipids: Classification of lipids with examples. Simple and compound lipids, unsaturated and saturated fatty acids, physical and chemical properties of fats and oils. Structure and biological importance of phospholipids and cholesterol.
Nucleic acids: Structure of bases, nucleosides, nucleotides and secondary structure of DNA and different forms of DNA. Types and functions of RNA, cloverleaf structure of tRNA.

UNIT III:

15

General characteristics of enzymes, nomenclature and classification of enzymes. Mechanism of enzyme action: active site, enzyme substrate complex formation-lock and key and induced fit theory. Concept of co-enzymes and cofactors with an example. Factors influencing enzyme activity: pH, temperature, substrate concentration, metal ion, inhibitors (allosteric) and activators, energy of activation. Isozymes, multienzyme complex and multifunctional enzymes with an example to each

BIO-ANALYTICAL TECHNIQUES

UNIT IV:

15

Bio-analytical Techniques: Lambert-Beer Law, working principles of UV-Visible spectrophotometry and colorimetry.
Centrifugation: Basic principle of centrifugation, ultracentrifuge and its application.
Chromatography: Principles of chromatography, Types- Partition chromatography- paper and thin layer chromatography & Adsorption chromatography - column chromatography, ion exchange & molecular sieve (principle & application).
Isotopes: Their importance in biological studies, measure of radioactivity, GM counters

PRACTICALS

(2 CREDITS)

HOURS : 4 HOURS /WEEK

1. Qualitative analysis of Carbohydrates.
2. Qualitative analysis of Lipids.
3. Estimation of reducing sugar by DNS method.
4. Estimation of Protein by Biuret method.
5. Estimation of amino acid by ninhydrin method /formal titration
6. Determination of activity and specific activity of enzyme-Salivary amylase.
7. Effect of pH on enzyme activity
8. Effect of temperature on enzyme activity.
9. Effect of metal ions on enzyme activity.
10. Preparation of buffer solution.
11. Identification of amino acids by circular paper chromatography.

REFERENCES

1. Nelson, D.L., Cox, M.M. 2004 Lehninger Principles of Biochemistry, 4 th edition, W.H. Freeman and Company, New York, USA.
2. Biochemistry, Lubert Stryer, 6th Edition, WH Freeman, 2006.
3. Harper's illustrated Biochemistry by Robert K. Murray, David A Bender, Kathleen M. Botham, Peter J. Kennelly, Victor W. Rodwell, P. Anthony Weil. 28th Edition, McGrawHill, 2009.
4. Biochemistry, Donald Voet and Judith Voet, 2nd Edition, Publisher: John Wiley and Sons, 1995.
5. Biochemistry by Mary K. Campbell & Shawn O. Farrell, 5th Edition, Cengage Learning, 2005.
6. Fundamentals of Enzymology Nicholas Price and Lewis Stevens Oxford University Press 1999
7. Fundamentals of Enzyme Kinetics Athel Cornish-Bowden Portland Press 2004
8. Practical Enzymology Hans Bisswanger Wiley-VCH 2004

**SEMESTER III
MOLECULAR BIOLOGY & GENETIC ENGINEERING**

(4 CREDITS)

Course Outcomes:

After completing the course students are able to:

CO1. Display a broad understanding of core molecular Biology.

CO2. Discuss and differentiate the process of Transcription and Translation

CO3 Explain key concepts of genome organization and manipulation.

CO4. Demonstrate working knowledge in a defined skill set of molecular biology and biotechnology protocols.

MOLECULAR BIOLOGY

NO. HOURS

UNIT I

15

DNA as genetic material: Experiments of Griffith and Hershey & Chase. Central Dogma of Molecular biology.

Concept of gene: Definition, generalized structure of Prokaryotes and Eukaryotes.

DNA Replication: Modes of DNA replication- Semiconservative, conservative and dispersive method.

Replication of DNA in prokaryotes and eukaryotes. Components of replication –lagging strand leading strand Okazaki fragment, role of SSBP, gyrase, helicase, RNA polymerase, DNA polymerase. Inhibitors of replication- role of actinomycin, acridin, novobiocin, novobiocin, amphidicolin and N-ethyl maleimide.

Genetic code: Major features of genetic code, outline of Deciphering of genetic code and Wobble hypothesis.

UNIT II

15

Transcription and RNA processing : RNA structure and types of RNA, Transcription in prokaryotes: Prokaryotic RNA polymerase, role of sigma factor, promoter, Initiation, elongation and termination of RNA chains. Inhibitors of Transcription- rifampicin, actinomycin, alpha amanitin and platinum antitumor drugs. Transcription in eukaryotes: Eukaryotic RNA polymerases, transcription factors, promoters, enhancers, mechanism of transcription initiation, promoter clearance and elongation RNA splicing and processing: processing of pre-mRNA: 5' cap formation, polyadenylation, splicing, rRNA and tRNA splicing.

Translation: Activation of amino acids, ribosome (composition & components), formation of initiation complex. Initiation, elongation and termination, inhibitors of protein synthesis.

GENETIC ENGINEERING

UNIT III

15

Enzymes in Genetic engineering and its importance-Restriction endonucleases-types of restriction enzymes, ligases, alkaline phosphatases, polynucleotide kinase, terminal deoxynucleotidyltransferase, S1 nuclease, Klenow fragment, taq DNA polymerases, ribonuclease, reverse transcriptase

Gene cloning vectors: Types of vectors –Cloning vector and expression vector . Plasmids (pBR322, pUC 19) and cosmids (pLFR5, pJB8). Importance of plasmids as cloning vectors, stability of plasmids, different forms of plasmid, concepts of YAC and BAC.

UNIT IV

15

Recombinant DNA technology: Isolation of gene, construction and preparation of complementary DNA. Probes- types, preparation and hybridization, genomic library.

Genetic engineering techniques: Gel electrophoresis, southern and northern blotting techniques, PCR and

its types, Sanger's, Maxam & Gilbert method of DNA sequencing.

Applications of Genetic Engineering: Therapeutic products produced by genetic engineering-blood proteins, human hormones. Genetic engineering in plants: Use of *Agrobacterium tumefaciens* and *A. rhizogenes*, Ti plasmids, Direct DNA transfer to plants.

PRACTICALS

(2 CREDITS)

HOURS : 4 HOURS /WEEK

1. Preparation of stock solution for molecular biology experiments.
2. Colorimetric estimation of DNA.
3. Colorimetric estimation of RNA.
4. Demonstration of T_m value of DNA.
5. Extraction of DNA from plant and microbial source.
6. Quantification of DNA by spectrophotometry.
7. Determination of purity of DNA.
8. Agarose gel electrophoresis of DNA.
9. Southern blotting (demonstration).
10. Isolation of plasmid DNA.

REFERENCES

1. Russell, P.J. 2009 Genetics – A Molecular Approach. 3rd edition. Benjamin Co. 7. Sambrook & Russell. Molecular Cloning: A laboratory manual. (3rd edition) 8. Slater, A., Scott, N.W. & Fowler, M.R. 2008 Plant Biotechnology: The Genetic Manipulation of Plants, Oxford University Press.
2. Brown, T.A. (1998). Molecular biology Labfax II: Gene analysis. II Edition. Academic Press, California, USA.
3. Griffiths, A.J.F., J.H. Miller, Suzuki, D.T., Lewontin, R.C. and Gelbart, W.M. (2009). An introduction to genetic analysis. IX Edition. Freeman & Co., N.Y., USA.
4. Watson, J.D., Myers, R.M., Caudy, A. and Witkowski, J.K. (2007). Recombinant DNA genes and genomes- A short course. III Edition. Freeman and Co., N.Y., USA.
5. Brown, T. A. Gene cloning and DNA analysis: An Introduction. Blackwell Publication.

DMD22005/ DMD22006

SEMESTER IV

PLANT TISSUE & ANIMAL CELL CULTURE

(4 CREDITS)

Course Outcomes:

After completing the course students are able to:

CO1.Develop concept of plant tissue and animal cell culture techniques and their application in biotechnology.

CO2. Comprehend the knowledge of transgenic plants in industrial and agricultural applications.

CO3.Establish and maintain various cell lines used in tissue culture.

PLANT CELL CULTURE

NO. HOURS

UNIT I

15

Plant tissue culture introduction: History and development, Importance of plant tissue culture. Laboratory organization and culture techniques: general requirements and aseptic conditions. Media preparation, culture media, sterilization, and pre-treatment to explants. Principles of tissue culture: Callus culture- Definition of callus, initiation, maintenance, sub culture and organogenesis .Factors affecting organogenesis . organ culture- culture protocols and importance of root and meristem culture. Micropropagation in plants: stages of micropropagation, methods, advantages, applications.

UNIT- II

15

Somaclonal variation for disease resistance and agronomic traits. Somatic embryogenesis: Embryoid and embryogenesis. Protocol and importance of somatic embryogenesis, Synthetic seeds and its applications, germplasm conservation and preservation. Suspension culture: Batch and continuous cell suspension culture. Importance of suspension culture in production of secondary metabolites. Protoplast culture and fusion: Definition of protoplast, isolation principle, culture protocol, action of enzymes, regeneration of plants, protoplast fusion, somatic cell hybridization and its application.

ANIMAL TISSUE CULTURE

UNIT - III

15

Introduction: History, developments and importance of animal cell culture. Characteristics of animal cell growth, Advantages and disadvantages of tissue culture methods and laboratory facilities. Culture procedure and culture media: Preparation and sterilization of glasswares and media. Culture media containing naturally occurring ingredients, blood plasma, blood serum, serum-free media, tissue extracts, complex natural media, chemically defined media, and basal salt solution –HBSS.

UNIT – IV

15

Primary culture, cell lines and cloning: Preparation of primary culture –mechanical and enzymatic method. Primary and established cell lines, somatic cell fusion. Tissue cultures- cover slip method, watch glass method and use of agar. Whole embryo culture. (e.g. Chick embryo). Hybridoma technology: Production of monoclonal antibodies.

Animal propagation –Artificial insemination, superovulation, embryo transfer, in-vitro fertilization, embryo splitting.

PRACTICALS

(2 CREDITS)

HOURS : 4 HOURS /WEEK

1. Media preparation and sterilization techniques.
2. Callus cultures: choice of explants, preparation of explants, callus induction, subculture and maintenance.
3. Regeneration of plants from growth hormones.
4. Meristem culture for pathogen free plants.
5. Preparation synthetic seed
6. Suspension culture – initiation of suspension culture from callus.
7. Plant protoplast Isolation.
- 8 . Cell viability test by tryphan blue method.
9. Preparation of HSS and glasswares of cell culture experiments
- 10.Isolation of PMN leucocytes from human peripheral blood sample and staining and identification.(lishman stain).
11. Demonstration of dissegration of cells by mechanical and enzymatic methods.
12. Photographic Demonstration of Animal Cell culture Lab equipments

REFERENCES

1. Hopkins, W.G. and Huner, P.A. 2008 Introduction to Plant Physiology. John Wiley and Sons.
2. Mauseth, J.D. 1988 Plant Anatomy. The Benjammin/Cummings Publisher, USA.
3. Bhojwani, S.S. and Razdan 2004 Plant Tissue Culture and Practice.
4. Butler, M. (2004). Animal cell culture and technology: The basics. II Edition. Bios scientific publishers. 3. Glick, B.R. and Pasternak, J.J. (2009). Molecular biotechnology- Principles and applications of recombinant DNA. IV Edition. ASM press, Washington, USA.
5. Reinert, J. and Bajaj, Y.P.S. 1997 Applied and Fundamental Aspects of Plant Cell, Tissue and Organ Culture. Narosa Publishing House.

SEMESTER V
DSE: IMMUNOLOGY AND MEDICAL BIOTECHNOLOGY (4 CREDITS)

Course Outcomes:

After completing the course students are able to:

CO1. Understand the role of different types of Cells in immune system .

CO2. Discuss the principles and applications of immunological techniques.

CO3. Understand to diagnose diseases.

CO4. Comprehend the knowledge of therapeutic applications of enzyme and hormone.

IMMUNOLOGY

NO. HOURS

UNIT I

15

Historical account and chronological events of Edward Jenner and Louis Pasteur.

Antigens: Definition, chemical nature, haptens, epitopes, antigenicity, blood group antigens.

Antibodies: Definition, types, structure of IgG.

Types of immunity – Innate- mechanism of innate immunity. Adaptive immunity – active and passive and adoptive immunity.

Cells and organs involved in immune system – T- cells, B-cells, antigen presentation and macrophages, their role in antigen recognition, clonal selection, and immunological memory. Immunological aspects of viral (HIV), bacterial and parasitic infection (one example each)

UNIT II

15

Immune disorders: Hypersensitivity, auto immune disorders- organ specific and systemic specific Grave's diseases, Hashimoto's disease , systemic lupus erythematosus.

Immuno techniques: Precipitation reaction, immuno diffusion-ODD and RID, RIA, Hemagglutination, ELISA, immunofluorescent, Western blotting. Major Histocompatibility complexes – class I & class II MHC antigens, antigen processing. Vaccines & Vaccination – adjuvants, cytokines, DNA vaccines, recombinant vaccines, bacterial vaccines, viral vaccines, vaccines to other infectious agents, passive & active immunization.

MEDICAL BIOTECHNOLOGY

UNIT III

15

Vaccine production: Introduction, new developments, types of vaccines – Inactivate, Attenuated and Recombinant Vaccines-Peptide and DNA, production of vaccines using genetically engineered microorganisms (Ex:HBV).

Enzymes in diagnosis: Enzymes used to detect and quantify blood glucose, cholesterol, triglycerides, urea and uric acid. Immobilized enzymes as diagnostic tools.

Nucleic acid analysis: Features of DNA probes and its applications in diagnosis, identification of *Mycobacterium tuberculosis* in clinical samples using PCR.

Enzymes in therapy: List of enzymes and their therapeutic applications.(Ex: DNase, debriding enzymes, asparaginase, superoxide dismutase, rennin, streptokinase, urokinase ,lipase and adenosine deaminase)

UNIT IV

15

Hormone therapy: List of hormones and their therapeutic applications (Ex: Insulin, human growth hormone, erythropoietin, calcitonin, lipocortin, somatotropin and alpha melanocyte stimulating hormone), production of humulin by recombinant DNA technology.

Therapeutic proteins: Cytokines as therapeutic proteins, production of interferon by recombinant DNA technology.

Human gene therapy: Definition, differences between somatic and germ line gene therapy, embryo, ex vivo, in vivo and antisense gene therapy -one example each, principle and applications.

Transgenic plants for production of biopharmaceutical (tobacco, tomatoes, and potatoes)

PRACTICALS

(2 CREDITS)

HOURS : 4 HOURS /WEEK

1 Determination of blood group

a) ABO blood grouping

b) Rh blood grouping.

2 Immuno diffusion :

a) ODD

b) RID.

3 Separation of serum from blood

4 Demonstration of ELISA

5 Demonstration of Western blotting

6 MIC assay

7 Isolation of antibiotic resistant strains using gradient plate method

8 Estimation of urea by BAMO method

9 Qualitative analysis of normal and abnormal constituents of urine

10 Photographic demonstration of transgenic animals and plants for production of biopharmaceutical

REFERENCES

1. Abbas AK, Lichtman AH, Pillai S. (2007). Cellular and Molecular Immunology. 6 th edition Saunders Publication, Philadelphia.

2. Delves P, Martin S, Burton D, Roitt IM. (2006). Roitt's Essential Immunology. 11th edition Wiley-Blackwell Scientific Publication, Oxford.

3. Goldsby RA, Kindt TJ, Osborne BA. (2007). Kuby's Immunology. 6th edition W.H. Freeman and Company, New York.

4. Murphy K, Travers P, Walport M. (2008). Janeway's Immunobiology. 7th edition Garland Science Publishers, New York.

5. Peakman M, and Vergani D. (2009). Basic and Clinical Immunology. 2nd edition Churchill Livingstone Publishers, Edinberg.

6. Richard C and Geiffrey S. (2009). Immunology. 6th edition. Wiley Blackwell Publication.

DME22205/DME22206

SEMESTER V

DSE: MICROBIALTECHNOLOGY & AGRICULTURAL BIOTECHNOLOGY (4 CREDITS)

Course Outcomes:

After completing the course students are able to:

CO1.Develop skills associated with screening of Industrially Important Strains.

CO2.Understand principles underlying design of Fermentor, Fermentation Process and downstream processing

CO3.Discuss the various aspects for the improvement of crop plants.

CO4. Understand the application of r-DNA technology to enhance the production of crop plant .

MICROBIALTECHNOLOGY

NO. HOURS

UNIT I

15

Introduction to biotechnological importance of microorganisms.

Metabolic pathway involved in microbial products (EMP,PPP and ED), primary and secondary metabolites, enzymes and microbial biomass.

Microbial production: Use of microbes in production of vitamins (vit-C), enzymes (Amylase), organic acids (citricacid),amino acids (glutamicacid),polysaccharides (xanthan),growth regulators (auxins), colorants (phycocyanin) ,flavors (diactyl), antibiotics (penicillin).

UNIT II

15

Kinetics of microbial growth and product formation: Phase of cell growth in batch cultures and continuous culture. Growth associated and non-growth associated product formation kinetics, substrate and product inhibition on cell growth and product formation.

Bioreactors- Types and functions. Purification & characterization of proteins, Upstream and downstream processing, solids and liquid handling. Disruption of microbial cells- centrifugation, filtration of fermentation broth, ultra centrifugation, liquid extraction, ion-exchange recovery of biological products.

Immobilization of cell- Introduction and methods of microbial cell immobilization.

AGRICULTURAL BIOTECHNOLOGY

UNIT III

15

Introduction: Biotechnology for crop improvement, future prospects of biotechnology for agriculture.

Biological nitrogen fixation: Nitrogen fixing microorganisms, role of nitrogenase, genetics of nitrogen fixing microorganisms, regulation of nif gene expression and mechanism of nitrogen fixation.

Bio fertilizers and phyto-stimulations: Mechanism of growth promotion by microbial inoculants- microbial production and application methods of microbial inoculants- *Rhizobium*, *azospirillum*, *azotobacter*, *mycorhizae*.

UNIT IV

15

Genetic engineering of crop plant: Gene transfer technique for desirable traits in crop plants. Agro bacterium mediated gene transfer, Direct gene transfer methods to protoplast. Few examples of transgenic plants, plants obtained through gene transfer techniques –BT cotton, herbicide tolerant soybean, virus resistance (papaya ring spot).

Microbial pesticides: Fungicides and herbicides. Bacterial, fungal and viral bio agents- *Bacillus*

Thurengensis (BT) and *BeaveriaBassiana*. Mechanism of control of plant disease-hypo virulence, competition antibiotics, induced resistance, mycoparasitism.

PRACTICALS

(2 CREDITS)

HOURS : 4 HOURS /WEEK

- 1 .Identification of important microorganisms relevant to biotechnology: E.coli, sacchromycescervisiae, spirulina.
- 2 .Demonstration of commercial products-single cell proteins microbial flavours.
- 3 .Entrapment of yeast for enzyme action & estimation of invertase activity
4. Preparation of wine.
5. Estimation of percentage of alcohol by Specific gravity method .6 .Seed inoculation with rhizobium culture and observation for root nodulation.
7. Preparation of bio control formulations.
8. Biofertilizers formulation.
9. Isolation and identification of *Rhizobium*.
10. Isolation and identification of *azospirillum*. Isoalation and Identification of *azotobacter*. Study of morphology of *mycorhizae*.
11. Photographic demonstration of BT cotton, herbicide tolerant soybean, virus resistance (papaya ring spot).
12. Demonstration of steps involved in large scale production of biofertilizers.

REFERENCES

1. Casida LE. (1991). Industrial Microbiology. 1st edition. Wiley Eastern Limited.
2. Crueger W and Crueger A. (2000). Biotechnology: A textbook of Industrial Microbiology. 2nd edition. Panima Publishing Co. New Delhi.
3. Mauseth, J.D. 1988 Plant Anatomy. The Benjammin/Cummings Publisher, USA.
4. Bhojwani, S.S. and Razdan 2004 Plant Tissue Culture and Practice.
5. Butler, M. (2004). Animal cell culture and technology: The basics. II Edition. Bios scientific publishers. 3. Glick, B.R. and Pasternak, J.J. (2009). Molecular biotechnology- Principles and applications of recombinant DNA. IV Edition. ASM press, Washington, USA.
6. Agricultural Biotechnology, S.S. Purohit

DMF22005/ DMF22006

SEMESTER VI

DSE: ENVIRONMENTAL BIOTECHNOLOGY AND BIOSTATISTICS

(4 CREDITS)

Course Outcomes:

After completing the course students are able to:

CO1. Gain an understanding of the causes, types and control methods for Environmental Pollution.

CO2. Differentiate the application of different life forms in Environmental Remediation.

CO3. Apply Statistical Tools for Analysis of Biological Data.

ENVIRONMENTAL BIOTECHNOLOGY

NO. HOURS

UNIT I

15

Introduction: Major issues in environment pollution. Role of Biotechnology to solve the problems.

Biotechnological methods of pollution detection: General bioassay, cell biological methods, immunoassay, DNA based methods, use of biosensor.

Biotechnological methods in pollution abatement: reduction of CO₂ emission, Waste water treatment – conventional waste treatment, Use of Algae, Eutrophication, Use of Cell Immobilization.

UNIT II

15

Biotechnology and biodegradation: Degradation of Xenobiotic compounds-organic (chlorinated hydrocarbons, substituted simple aromatic compounds, polyaromatic hydrocarbons, pesticides and surfactants.

Biohydrometallurgy and Biomining: Bioleaching, biosorption, oil degradation and creation of super bugs. Treatment of Industrial wastes: Pulp, Dye, leather and solid waste management. Genetically engineered microbes for waste treatment.

Ecofriendly bioproducts: Biomass resources, biogas, and alcohol as a fuel, biological hydrogen generation and biodegradable plastics.

BIOSTATISTICS

UNIT III

15

Introduction, Basic concepts- population, data, sample and variable. Types of data-primary and secondary, methods of data collection- direct personal interview, indirect oral interview, through correspondence, questionnaire and census. Classification of data- qualitative, quantitative and simple classification. Sampling methods- random and non-random. Tabulation of data- structure of a table, simple and complex table.

UNIT IV

15

Graphical and diagrammatic representation of data- histogram, bar graph and pie diagram. Frequency of distribution- without class intervals, with class intervals and cumulative frequency distribution. Measures of central tendency- mean, median and mode. Measure of dispersion- range, mean deviation, co-efficient of deviation and standard deviation.

PRACTICALS

(2 CREDITS)

HOURS : 4 HOURS /WEEK

- 1 & 2. Analysis of sewage water for BOD & COD.
- 3 Estimation of Hydrogen sulphides in the sewage water.
 - b. Estimation of chloride in sewage water sample.
 - c. Estimation of residual chloride in sewage water sample.
 - d. Estimation of carbon dioxide in sewage water sample.
4. Identification of microbial flora in the given water sample.
- 5 . Estimation of percentage of alcohol by specific gravity bottle method
- 6 a. Photographic demonstration of septic tank, sand filters, Imhoff's tank and biosensors.
 - b. Photographic demonstration of creation of superbug.
 - c. Photographic demonstration of genetically modified microbes.
 - d. Photographic demonstration of genetically modified plants.
 - e. Photographic demonstration of genetically modified animals.

Biostatistics problems

- 7 Problems on graphical and diagrammatic representation of data (histogram, bar graph and pie chart)
- 8 Calculation of mean, median, mode, standard deviation

REFERENCES

1. Environmental Science, S.C. Santra
2. Environmental Biotechnology, Pradipta Kumar Mohapatra
3. Environmental Biotechnology – Concepts and Applications, Hans-Joachim Jordening and Jeseff Winter
4. Le CT (2003) Introductory biostatistics. 1st edition, John Wiley, USA
5. Glaser AN (2001) High Yield™ Biostatistics. Lippincott Williams and Wilkins, USA
6. Edmondson A and Druce D (1996) Advanced Biology Statistics, Oxford University Press.
7. Danial W (2004) Biostatistics : A foundation for Analysis in Health Sciences, John Wiley and Sons Inc.

DMF22205/ DMF22206

SEMESTER VI

DSE: BIOINFORMATICS AND BIOPROCESS TECHNOLOGY

(4 CREDITS)

Course Outcomes:

After completing the course students are able to:

CO1. Understand the basic concepts and tools used in Bioinformatics.

CO2. Comprehend the knowledge of Genomics and Proteomics.

CO3. Develop an understanding of the various aspects of Bioprocess Technology.

CO4. Enhance the skills associated in Fermentation Process.

BIOINFORMATICS

NO. HOURS

UNIT I

15

Bioinformatics and the Internet: Introduction, Internet basics, connecting to the internet electronic mail, File transfer protocol, The World Web.

Database- DNA, protein, genomic mapping database, sequence alignment software-pair wise & multiple alignments, gene families

UNIT II

15

Information retrieval from databases: Databases similarity searching, FASTA, BLAST SEARCH, Clustal W, Clustal X, DIALIGN2, Multalign Navigating the NCBI web site.

Genomics and Proteomics: Types of genomes, bacterial genome sequence project.

Human genome project, Micro array technologies-types and applications.

BIOPROCESS TECHNOLOGY

UNIT-III

15

Introduction to bioprocess technology. Range of bioprocess technology and its chronological development. Basic principle components of fermentation technology. Types of microbial culture and its growth kinetics- Batch, Fed batch and Continuous culture.

UNIT IV

15

Design of bioprocess vessels- Significance of Impeller, Baffles, Sparger; Types of culture/production vessels- Airlift; Cyclone Column; Packed Tower and their application in production processes. Principles of upstream processing – Media preparation, Inoculation, development and sterilization. Introduction to oxygen requirement in bioprocess; mass transfer coefficient; factors affecting KLa. Bioprocess measurement and control system with special reference to computer aided process control.

PRACTICALS

(2 CREDITS)

HOURS : 4 HOURS /WEEK

1. Sequence information resource
2. Understanding and use of various web resources: EMBL, Genbank, Entrez, Unigene, Protein information resource (PIR)
3. Understanding and using: PDB, Swissprot, TREMBL
4. Using various BLAST and interpretation of results.
5. Retrieval of information from nucleotide databases.
6. Sequence alignment using BLAST.
7. Multiple sequence alignment using Clustal W.
8. Bacterial growth curve.
9. Production and analysis of ethanol.
10. Production and analysis of amylase.
11. Production and analysis of lactic acid.
12. Isolation of industrially important microorganism from natural resource.

REFERENCES

1. Stanbury PF, Whitaker A and Hall SJ. (2006). Principles of Fermentation Technology. 2nd edition, Elsevier Science Ltd.
2. Salisbury, Whitaker and Hall. Principles of fermentation Technology,
4. Waste Water Engineering, Metcalf and Eddy, Tata McGraw hill
5. Wong, K.C. (2016). Computational biology and bioinformatics: gene regulation, CRC press/ Taylor & Francis Group.
6. Joyce, A. P.; Zhang, C.; Bradley, P.; Havranek, J. J. (2015). “Structure –based modeling of protein : DNAspecificity”. Briefings in Functional Genomics.

DME22405/ DME22406

**SEC
MICROBIAL TECHNIQUES**

(2 CREDITS)

Course Outcomes:

After completing the course students are able to:

CO1. Understand and demonstrate basic sterilization techniques.

CO2. Analyze the anatomy of prokaryotic cell and structural detail of eukaryotic cell.

CO3. Acquire the knowledge of Culture media and their applications.

CO4. Assess the growth measurement and pattern of microorganism.

MICROBIAL TECHNIQUES

NO. HOURS

UNIT I

07

General introduction. Concept of Prokaryotes and Eukaryotes. General account on Structure, Classification & Reproduction of Bacteria, Fungi & Viruses.

UNIT II

08

Microbial Techniques: Sterilization: Principles and applications of
a. Physical Methods: Autoclave, Hot air oven, Laminar airflow, Seitz filter, Sintered glass Filter, membrane filter.
b. Chemical Methods: Alcohol, Aldehydes, Phenols, Halogens and Gaseous agents.
c. Radiation Methods: UV rays and Gamma rays.

UNIT III

08

Microscopy: working principle and applications of Light microscopy, phase contrast microscopy and electron microscopy.
Staining-Types, Simple and differential (Gram's and acid fast)

UNIT IV

07

Microbial nutrition and growth: nutritional classes of microorganisms, culture media, pure culture, microbial growth pattern and methods of growth measurements, method of maintenance and preservation of cultures.

REFERENCES

- 1 Prescott L.M. Harley J.P and Klein D.A (Microbiology 5th Edition)
2. Pelzar Jr, M.J. Chan, E.C.S. and Krieg N.R (Microbiology)
3. Salle. A.J Fundamental Principles of Bacteriology .
4. Caldmell, D.R. Microbial Physiology and metabolism

DME22505/ DME22506:

**SEC
ENZYMOLOGY**

(2 CREDITS)

Course Outcomes:

After completing the course students are able to:

CO1. Understand relationship between the structure and function of enzymes.

CO2. Interpret mechanisms of enzymatic action kinetics and molecular interactions and specifies importance of enzymes.

CO3. Characterise the inhibitor affects on enzyme kinetics.

CO4. Enumerate methods in immobilization of enzymes.

ENZYMOLOGY	NO. HOURS
UNIT – I	7
Enzyme classification . Enzyme substrate complex: concept of E-S complex, binding sites, active site, specificity, factors affecting initial rate, E, S, temp. &pH.	
UNIT – II	8
Kinetics of enzyme activity, Michaelis-Menten equation, Different plots for the determination of Km and Vmax and their physiological significance. Enzyme inhibition types of inhibition, Mechanism of enzyme action: General mechanistic principle.	
UNIT – III	7
Allosteric enzymes with special reference to phosphofructokinase. Kinetics of allosteric enzymes. Isoenzymes– multiple forms of enzymes with special reference to lactate dehydrogenase. Multienzyme complexes. Ribozymes. Multifunctional enzyme- eg Fatty Acid synthase.	
UNIT – IV	8
Enzyme Technology: Methods for large scale production of enzymes. Immobilized enzyme and their comparison with soluble enzymes, Methods for immobilization of enzymes. Immobilized enzyme reactors. Application of Immobilized and soluble enzyme in health and industry.	

REFERENCES

- 1.Nelson, D.L., Cox, M.M. 2004 Lehninger Principles of Biochemistry, 4 th edition, W.H. Freeman and Company, New York, USA.
2. Biochemistry, LubertStryer, 6th Edition, WH Freeman, 2006.
3. Harper’s illustrated Biochemistry by Robert K. Murray, David A Bender, Kathleen M.Botham, Peter J. Kennelly, Victor W. Rodwell, P. Anthony Weil. 28th Edition, McGrawHill, 2009.
4. Biochemistry, Donald Voet and Judith Voet, 2nd Edition, Publisher: John Wiley andSons, 1995.
5. Biochemistry by Mary K.Campbell& Shawn O.Farrell, 5th Edition, Cenage Learning,2005.
6. Fundamentals of Enzymology Nicholas Price and Lewis Stevens Oxford University Press 1999

**Pattern of Question Paper
Semester I to VI
Paper I to VI (DSC)**

Time : 3 Hrs

Max Marks: 70

I. Answer all the questions

5 X 1 = 5

- 1 -----
- 2-----
- 3-----
- 4-----
- 5-----

II. Answer any five questions

5 X 3 = 15

- 6-----
- 7-----
- 8-----
- 9-----
- 10-----
- 11-----

III. Answer any four questions

4 X 5 = 20

- 12 -----
- 13-----
- 14-----
- 15-----
- 16-----

IV. Answer any three questions

3 X 10 = 30

- 17-----
- 18-----
- 19-----
- 20-----

(Note- 10 Marks may be divided in to 6+4 or 5+5)

**Pattern of Question Paper
Semester V
(SEC)**

Time : 2 Hrs

Max Marks: 35

I. Answer all the questions

5 X 1 = 5

- 1 -----
- 2-----
- 3-----
- 4-----
- 5-----

II. Answer any five questions

5 X 3 = 15

- 6-----
- 7-----
- 8-----
- 9-----
- 10-----
- 11-----

III. Answer any four questions

4 X 5 = 20

- 12 -----
- 13-----
- 14-----
- 15-----
- 16-----

IV. Answer any one question

1 X 10 = 10

- 17-----
- 18-----

(Note- 10 Marks may be divided in to 6+4 or 5+5)

Scheme for Practical Examination

I B.Sc., I Semester

CELL BIOLOGY & GENETICS

Duration : 3 hours

Max. Marks :35

Practical paper :30 and Record:05

1. Determine the total no. of cells in the given sample by haemocytometer **18**
or
Measure the length of the given material by micrometry
or
Identify any three mitotic stages from root tip squash preparation.
2. Identify and comment on the spotter A, B and C. **3x4=12**

I B.Sc., II Semester

BIOMOLECULES & BIO-ANALYTICAL TECHNIQUES

Duration : 3 hours

Max. Marks :35

Practical paper :30 and Record:05

1. Estimate the amount of glucose in the given sample by DNS method. **18**
or
Estimate the amount of protein in the given sample by biuret method.
or
Determine the optimum pH for the given enzyme.
or
Identify the given biomolecule by performing qualitative analysis.
2. Identify and comment on the spotter A, B and C. **3x4=12**

II B.Sc., III Semester

MOLECULAR BIOLOGY & GENETIC ENGINEERING

Duration : 3 hours

Max. Marks :35

Practical paper :30 and Record:05

1. Estimate the amount of DNA in the given sample by DPA method. **18**
or
Estimate the amount of RNA in the given sample by orcinol method.
or
Isolate DNA from the given material .
2. Identify and comment on the spotter A, B and C. **3x4=12**

II B.Sc., IV Semester

PLANT TISSUE & ANIMAL CELL CULTURE

Duration : 3 hours

Max. Marks :35

Practical paper :30 and Record:05

1. Determine the total no. of living and dead cells in the given sample by tryphen blue exclusion method. **18**
or
Demonstrate the production of synthetic seeds. Write the principle and procedure for the same.
or
Demonstrate the initiation of suspension culture from callus. Write the principle and procedure for the same. Or
Demonstrate the surface sterilization of explants. Write the principle and procedure for the same.
- 2 . Identify and comment on the spotter A, B and C. **3x4=12**

III B.Sc., V Semester

IMMUNOLOGY & MEDICAL BIOTECHNOLOGY

Duration : 3 hours

Max. Marks :35

Practical paper :30 and Record:05

1. Determine the blood group of the given sample. **18**
or
Determine the antigen antibody reaction by conducting ODD.
or
Demonstrate MIC assay. Write the principle and procedure for the same.
2. Identify and comment on the spotter A, B and C. **3x4=12**

III B.Sc., VI Semester

MICROBIALTECHNOLOGY & AGRICULTURAL BIOTECHNOLOGY

Duration : 3 hours

Max. Marks :35

Practical paper :30 and Record:05

1. Estimate the invertase activity from the given immobilised cells. **18**
or
Estimate the percentage of alcohol in the given sample by specific gravity bottle method.
2. Identify and comment on the spotter A, B and C. **3x4=12**

III B.Sc., VI Semester

ENVIRONMENTAL BIOTECHNOLOGY & BIostatISTICS

Duration : 3 hours

Max. Marks :35

Practical paper :30 and Record:05

1. Determine the BOD of the given water sample. **10**

or

Estimate the amount of hydrogen sulphide present in the given water sample.

Or

Estimate the amount of carbon di oxide and chloride present in the given water sample.

Or

Estimate the percentage of alcohol in the given sample by specific gravity bottle method.

2. Identify and comment on the spotter A, B and C. **3x4=12**

3. Statistics problem

III B.Sc., VI Semester

BIOINFORMATICS AND BIOPROCESS TECHNOLOGY

Duration : 3 hours

Max. Marks :35

Practical paper :30 and Record:05

1. Estimate the amylase activity. **18**

Or

Identify and write critical note on the given microorganism.

2. Identify and comment on the spotter A, B and C. **3x4=12**

JSS MAHAVIDYAPEETHA



ESTD-1964

JSS COLLEGE OF ARTS, COMMERCE & SCIENCE

(An Autonomous College of University of Mysore)

Re-accredited by NAAC with 'A' grade

OOTY ROAD, MYSORE-570 025, KARNATAKA

Syllabus

B. Voc in Animation and Multimedia

2018-19

COMPUTER SCIENCE DEPARTMENT (UG)

**B. Voc in Animation and Multimedia
Certificate – Level 4- Character Designer
(National Competency Certificate)**

**Semester 1
Vocational Content**

MES / N 0501 - Understanding the script

Section – A

Pre-Production Techniques: Ideas, Themes, Concepts, Story Development. Script- Format, Storyboard. Short Film, Documentary, Feature Film. Script Terms, Understanding Genre, Different Types of Genres and Sub Genres, Read-through and Identifying

Character types. Understanding Narration Style, Time, Place and society condition status of the Character. Plot Interpretation. Three act Structure.

Introduction on how to make drawings for animation, Shapes and forms, About 2d and drawings, Clay modeling, Layout design according to a storyboard. Human anatomy, skeleton structure, animal anatomy and mythical creatures. Studying Human and animal movement through sequential drawings, gesture drawing, Caricaturing – fundamentals, Exaggeration, Attitude, Silhouettes, Boundary breaking exercises and warm ups, gesture drawing, Line drawing and quick sketches, Drawing from observation, memory and imagination.

Section – B

1. Story Reading (Read-through)
2. Group Discussion
3. Observation of Character development.
4. Acting (Read-through)

MES / N 0502 - Ensuring consistency across all scenes

Section – A

Introduction to Pencil Drawing- Introduction to basic drawing techniques including proportions and Line, Using Pencils for shading, People and Animals, Proportion in portrait drawing, Highlighting and shadows, Drawings with the help of basic shapes, understanding of the relationship of bones and muscles at rest and in movement to able to translate and maintaining correct volumes.

Introduction to Perspectives: Perspective in 1 point, Perspective in 2 point, Perspective in 3 point, Eye levels, Vanishing point, Significance of Perspective in animation and composition, Humans and animal forms in perspective, Perspective-blocks, boxes and shapes, Scale diagrams in perspective, Different viewpoints, Importance of eye level.

Section – B

1. Drawing Basic shapes
2. Shapes in to forms
3. One-point Perspective - Table
4. Two-points Perspective - Exterior
5. Three-points perspective – Interior
6. Drawing simple characters using simple shapes
7. Drawing cartoons
8. Drawing male characters (6&7 Heads)
9. Drawing Female Characters (4&5 Heads)
10. Drawing an animal Character
11. Drawing a Hybrid Character (Fantasy)

MES / N 0503 - Design Characters

Section – A

Introduction on how to make drawings for animation, Shapes and forms, About 2d and drawings, Clay modeling, Layout design according to a storyboard. Human anatomy, skeleton structure, animal anatomy and mythical creatures.

Studying Human and animal movement through sequential drawings, gesture drawing, Caricaturing – fundamentals, Exaggeration, Attitude, Silhouettes, Boundary breaking exercises and warm ups, gesture drawing, Line drawing and quick sketches, Drawing from observation, memory and imagination.

Intro to Raster: Foreground & background, Changing Foreground and Background colors, using the large color selection Boxes and small color swathes, using the Eyedropper tool to sample Image color, changing the Foreground Color While using a Painting Tool. Using Brushes, Selecting the Brush Shape, drawing a vertical and Horizontal Straight lines with any brush, Drawing connecting Straight Lines (at any angle) with any brush, creating a New Brush, Saving Brushes, Loading Brushes, creating a Custom Brushes, Using the Painting Modes, Fade, Airbrush Options, Pencil Options.

Intro to Vector: Understanding paths, views, selection tools, fills and strokes, setting up preferences and color settings, Creating basic geometric shapes with the Shape tools, Using a grid and smart guides to aid symmetrical drawing, Using the Bezier Pen, Direct Selection tool, and Convert tool efficiently, Applying and editing color gradients to filled regions, Creating and using swatches, tints, gradients, and patterns on filled regions.

3D MODELING: Introduction to various 3D modeling Techniques: - Organic Modeling, Mechanical & Technical Modeling. Using Templates for Modeling. Polygon and Patch. Concept of edit meshes and edit poly. Low poly modeling. Modifiers and compound objects. How to manage vertex, faces and polygon selections.

Introduction to, curves, create a curve from a surface edge, isoparm, or curve-on-surface and Bezier curve editing curves, Surface tools, Trimming, Stitching. Sculpting NURBS surfaces, NURBS Modeling Reference, import vector data as curves or generate polygonal surfaces.

Section – B(1)

1. Perspective art.
2. Traditional designs.
3. 2D Character Designing
4. Photo Collage
5. Matt Painting
6. Digital painting
7. 2d Animation basics
8. Gif Animation.

Section – B(2)

1. Modeling an Apple.
2. Modeling a simple Robot Character.
3. Modeling a Human Face.
4. Modeling Human Body.
5. Modeling Prop for Character (Costume).
6. Modeling an animal character.
7. Modeling a Fantasy Character.

MES / N 0513 - Manage and store assets

Section – A

Understanding Management, SOP, research appropriate tools, techniques, technologies and procedures for effective asset storage, Saving the work with appropriate names and/or naming conventions so that it can be identified easily. Storing the work in an appropriate place using appropriate storage techniques to ensure it is protected from damage.

Making backup copies at appropriate time intervals of any digital files. Routinely archive any work and store it securely in a second location, if required. Identify and retrieve previous work from storage, as required.

Section – B

1. Arranging Materials in order
2. Writing Logs
3. Group discussion on Assets Allocation.
4. Assets handling and Proper storage Techniques
5. Data storage techniques.

MES / N 0104 - Maintain workplace health and safety

Section – A

Understand and comply with the organization's current health, safety and security policies and procedures

Understand the safe working practices pertaining to own occupation. Understand the government norms and policies relating to health and safety including emergency procedures for illness, accidents, fires or others which may involve evacuation of the premises. Participate in organization health and safety knowledge sessions and drills

Section – B

1. Identify the people responsible for health and safety in the workplace, including those to contact in case of an emergency.
2. Identify security signals e.g. fire alarms and places such as staircases, fire warden stations, first aid and medical rooms.
3. Identify aspects of your workplace that could cause potential risk to own and others health and safety.
4. Ensure own personal health and safety, and that of others in the workplace through precautionary measures.
5. Identify and recommend opportunities for improving health, safety, and security to the designated person.
6. Report any hazards outside the individual's authority to the relevant person in line with organizational procedures and warn other people who may be affected.
7. Follow organization's emergency procedures for accidents, fires or any other natural calamity in case of a hazard.
8. Identify and correct risks like illness, accidents, fires or any other natural calamity safely and within the limits of individual's authority.

Semester 1 General Content

L4 – Sem 1 - GC - Communication Language Kannada

ಸಾಹಿತ್ಯ ಘಟಕ

12 ಗಂಟೆಗಳು

1. ಕಾವ್ಯ
1. ಯಾವ ಕಾಲದ ಶಾಸ್ತ್ರವೇನು ಹೇಳಿದರೇನು? - ಕುವೆಂಪು
2. ಪುಟ್ಟ ವಿಧವೆ - ಬೇಂದ್ರೆ
3. ರಂಗೋಲಿ - ಪುತಿನ
4. ಮುಂಬೈ ಜಾತಕ - ಜಿಎಸ್‌ಎಸ್
5. ಅಡುಗೆ ಮನೆಯ ಹುಡುಗಿ - ವೈದೇಹಿ
6. ನನ್ನ ಜನಗಳು - ಸಿದ್ದಲಿಂಗಯ್ಯ

2. ಗದ್ಯ

12 ಗಂಟೆಗಳು

1. ನಿರಂತರ ನೀರಿಗಾಗಿ ನಿಂತಲ್ಲೇ ಓಟ - ನಾಗೇಶ್ ಹೆಗಡೆ
2. ಮಾಹಿತಿ ತಂತ್ರಜ್ಞಾನ : ಒಂದು ಸ್ಥೂಲ ನೋಟ - ಜಿ.ಎನ್. ನರಸಿಂಹಮೂರ್ತಿ
3. ಮೂಢನಂಬಿಕೆಗಳ ಬೀಡಿನಲ್ಲಿ - ಭುವನೇಶ್ವರಿ ಹೆಗಡೆ
4. ಬೇಡದ ಅತಿಥಿಗಳು - ಎ ಎನ್ ಮೂರ್ತಿರಾಯರು

ಭಾಷಾ ಘಟಕ

3. ಆಡಳಿತ ಮತ್ತು ವಾಣಿಜ್ಯ ಕನ್ನಡ

21 ಗಂಟೆಗಳು

1. ಆಡಳಿತ ಭಾಷೆಯಾಗಿ ಕನ್ನಡ ಸ್ವರೂಪ ಮತ್ತು ಲಕ್ಷಣ
2. ಸರ್ಕಾರಿ ಪತ್ರ ಸ್ವರೂಪ ಅಧಿಕೃತ ಜ್ಞಾಪನ ಮತ್ತು ಸುತ್ತೋಲೆ
3. ವಿವಿಧ ರೀತಿಯ ಅರ್ಜಿಗಳು - ರಜೆ, ವೇತನ, ಬಡ್ಡಿ, ಪದೋನ್ನತಿ ಇತರೆ
4. ವಾಣಿಜ್ಯ ಕನ್ನಡದ ಸ್ವರೂಪ ಮತ್ತು ಲಕ್ಷಣ
5. ವಾಣಿಜ್ಯ ಪತ್ರದ ವಿವಿಧ ಅಂಗಗಳು ವಾಣಿಜ್ಯ ಪತ್ರಗಳು- ವಿಚಾರಣಾ ಪತ್ರ, ಆದೇಶ ಪತ್ರ, ಉದರಿಪತ್ರ, ವಸೂಲಿ ಪತ್ರ, ಪರಿಚಯ ಪತ್ರ, ಪರಾಮರ್ಶನ ಪತ್ರ, ವ್ಯವಹಾರೋದ್ದೇಶನ ಪತ್ರ, ಪರಿಪತ್ರ
6. ಗಣಕದಲ್ಲಿ ಕನ್ನಡ ಕನ್ನಡದ ಅಂತರ್ಜಾಲ ತಾಣಗಳು- ನುಡಿ, ಬರಹ- ಪಾರಿಭಾಷಿಕ ಪದಗಳು
7. ಭಾಷಾಂತರ : ಸ್ವರೂಪ ಮತ್ತು ಲಕ್ಷಣ- ಕನ್ನಡದಿಂದ ಇಂಗ್ಲಿಷಿಗೆ- ಇಂಗ್ಲಿಷಿನಿಂದ ಕನ್ನಡಕ್ಕೆ

ಪರಾಮರ್ಶನ ಗ್ರಂಥಗಳು

1. ಸಮಗ್ರಕಾವ್ಯ - ಜಿ ಎಸ್ ಶಿವರುದ್ರಪ್ಪ
2. ಬಿಂದು ಬಿಂದಿಗೆ - ವೈದೇಹಿ
3. ಪಕ್ಷಿಕಾಶಿ - ಕುವೆಂಪು
4. ಸಮಗ್ರಕಾವ್ಯ - ಬೇಂದ್ರೆ
5. ಶತ್ರುವಿಲ್ಲದ ಸಮರ - ನಾಗೇಶ್‌ಹೆಗ್ಡೆ
6. ವ್ಯಾವಹಾರಿಕ ಕನ್ನಡ - ಎಚ್ ಎಸ್ ಕೆ
7. ಆಡಳಿತಕನ್ನಡ - ಡಾ ಅಶೋಕ್‌ಕುಮಾರ್ ರಂಜೇರ ಮತ್ತು ಇತರರು
8. ಕಛೇರಿ ಕೈಪಿಡಿ - ಕುವೆಂಪು ಕನ್ನಡ ಅಧ್ಯಯನ ಸಂಸ್ಥೆ
9. ಲೇಖನಕಲೆ - ಎಸ್ ಪ್ರಹ್ಲಾದರಾವ್
10. ಆಡಳಿತಭಾಷೆ ಕೆಲವು ವಿವರಗಳು - ಪ್ರಧಾನ ಗುರುದತ್ತ

L4 – Sem 1 - GC - Mathematics:

Unit 1: Algebra

Quadratic and cubic equations – Progressions – Vectors-Scalar product- vector product – vector triple product – standard results and simple problems.

Unit 2: Analytical Geometry

Cartesian coordinate system – Distance formula – section formula – Equation of lines. simple problems – circles – parabola-ellipse-hyperbola-simple problems.

Unit 3: Trigonometry and Calculus

Trigonometric ratios- inverse trigonometric functions – simple problems .

Limits – Rules of differentiation – simple problems.

L4 – Sem 1 - GC – Physics:

PHYSICAL WORLD

Physics: Scope and excitement of physics - Physics, technology and society - Mention of fundamental forces in nature - Nature of physical laws.

UNITS AND MEASUREMENTS

Unit of measurement - System of units - SI units - Fundamental and derived units - Length, mass and time measurements - Accuracy and precision of measuring instruments, Errors in measurement. Significant figures, Numerical problems.

SCALAR AND VECTORS

Scalars and Vectors – Position and displacement vectors - Equality of vectors - Multiplication of a vector by real number, Addition and subtraction of two vectors, Triangle method and parallelogram method.

List of Experiments:

- 1) To measure diameter of a small spherical body using Vernier Calipers.
- 2) To measure diameter of a small cylindrical body using Vernier Calipers.
- 3) To measure internal diameter and depth of a given beaker using Vernier Calipers and hence find its volume.
- 4) To measure internal diameter and depth of a given calorimeter using Vernier Calipers and hence find its volume.
- 5) To measure diameter of a given wire using screw gauge.
- 6) To measure thickness of a given sheet using screw gauge.
- 7) To measure volume of an irregular lamina using screw gauge.
- 8) To determine the masses of two different objects using a beam balance.

L4 – Sem 1 – GC - Computer science:

L4-Sem-1-GC: COMPUTER SCIENCE – I

Computer Fundamentals

Unit 1:

Knowing computer: What is Computer, Basic Applications of Computer; Components of Computer System, Central Processing Unit (CPU), VDU, Keyboard and Mouse, Other input/output Devices, Computer Memory, Concepts of Hardware and Software; Concept of Computing, Data and Information; Applications of IECT; Connecting keyboard, mouse, monitor and printer to CPU and checking power supply. Operating Computer using GUI Based Operating System: What is an Operating System; Basics of Popular Operating Systems; The User Interface, Using Mouse; Using right Button of the Mouse and Moving Icons on the screen, Use of Common Icons, Status Bar, Using Menu and Menu-selection, Running an Application, Viewing of File, Folders and Directories, Creating and Renaming of files and folders, Opening and closing of different Windows; Using help; Creating Short cuts, Basics of O.S Setup; Common utilities.

Unit 2:

Programming Development life cycle

Understanding Word Processing: Word Processing Basics; Opening and Closing of documents; Text creation and Manipulation; Formatting of text; Table handling; Spell check, language setting and thesaurus; Printing of word document.

Using Spread Sheet: Basics of Spreadsheet; Manipulation of cells; Formulas and Functions; Editing of Spread Sheet, printing of Spread Sheet.

Introduction to Internet, WWW and Web Browsers: Basic of Computer networks; LAN, WAN; Concept of Internet; Applications of Internet; connecting to internet; What is ISP; Knowing the Internet; Basics of internet connectivity related troubleshooting, World Wide Web; Web Browsing software's, Search Engines; Understanding URL; Domain name; IP Address; Using e-governance website.

Communications and collaboration: Basics of electronic mail; Getting an email account; Sending and receiving emails; Accessing sent emails; Using Emails; Document collaboration; Instant Messaging; Netiquettes.

Small Presentation Making: Basics of presentation software; Creating Presentation; Preparation and Presentation of Slides; Slide Show; Taking printouts of presentation / handouts.

Reference Books:

Computer today by Donald H. Sanders - McGraw-Hill, 1983

Semester 2
Certificate – Level 5- Script Researcher
(National Competency Certificate)

Vocational Content

MES / N 3001 - Understand and detail the concept

Section – A

What Is a Short Story? Finding a Story to Write, A Short Story's Basic Ingredients, Sitting Down to Write.

Choosing a Protagonist, choosing a Point of View, Bringing Your Characters to Life, Tip Sheet: Three-Dimensional Characters, Character's Bio Chart, Giving Your Characters a Voice, Tip Sheet: Dialogue

How Conflict Works in a Short Story, The Protagonist's Predicament, Bad Guys, Hurricanes, and Fatal Flaws, Conflict Equals Suspense.

Scriptwriting terminology: Action, Angle, BG close up, Exterior, fade in fade out, Pan, Parenthetical, POV, Scene Heading, Slug line, SFX, Sotto voce, track with, Voice over, Character name, Cast List, Dialogue, Script Length, Action Description, Extension, Dual-Column Dialogue, Act numbers, Scene Numbers, short lines, dialogue paragraphs, Character Arc.

Section - B

1. Exercises: Generating Ideas.
2. Exercises: Creating Characters.
3. Exercises: Finding Story Conflict.
4. Exercises: Constructing a Plot.
5. Exercises: Making a Setting Vivid.
6. Exercises: Discovering and Developing Your Voice.
7. Public domain material, Protecting the idea, Proof of ownership, Confidentiality agreement, Piracy, Animation Industry in India, Case Study.

MES / N 3002 - Undertake research for scripts

Section - A

Creating Your Own Show, Writing a TV Proposal, Drama and Comedy, Documentary and Reality, Formatting and Presentation, One hour Drama, Procedural, Serialized, Half-hour Comedy or Sitcom, Single-Camera Format, Multi-Camera Format, Limited Series or Miniseries.

Scriptwriting for both short film: Dialogue, Parenthetical, Extension, Shot Transition, Dual-Column Dialogue, Act numbers, Scene Numbers, Cast List, short lines, dialogue paragraphs, Script Styles & Types, Script Length Scene Heading Action Description , Character Name, Dialogue, Parenthetical, Extension, Transition, Shots.

Section - B

1. Do a Research.
2. Understanding the impact on the society.
3. Finding the common interest.
4. Creating the effective narration format.

MES / N 3005 - Assist in drafting scripts

Section – A

Scriptwriting for feature film: Dialogue split by Action, Emphasis in Action, Abbreviations, Short Lines, Dialogue Paragraphs, Montage & Series of Shots, Supers - Titles, Signs, etc, Phone Calls and Intercuts Two people talking at the same time.

Introduction to storyboard: Introduction, Multimedia Storyboarding Tools, The Advantages of Storyboarding, Interactive Storyboarding, Using Interactive Storyboarding to Speed-up the Content-writing Phase, Using Interactive Storyboarding to Speed-up Report Document Production.

Steps of storyboard writing: Storyboard table, Topic, Timeline, Sketches, color scheme, text attributes, Audio, camera angle, Interaction of buttons and text.

Section - B

Understanding Camera terms: Extreme Lon Shot, Long Shot, Mid Shot, Close-up, Extreme Close-up, Over the shoulder. Panning, Tilt, Dolly shot, Crane Shot, Moving Shot.

MES / N 0104 - Maintain workplace health and safety

Section – A

Understand and comply with the organization's current health, safety and security policies and procedures

Understand the safe working practices pertaining to own occupation. Understand the government norms and policies relating to health and safety including emergency procedures for illness, accidents, fires or others which may involve evacuation of the premises. Participate in organization health and safety knowledge sessions and drills

Section – B

1. Identify the people responsible for health and safety in the workplace, including those to contact in case of an emergency.

2. Identify security signals e.g. fire alarms and places such as staircases, fire warden stations, first aid and medical rooms.

3. Identify aspects of your workplace that could cause potential risk to own and others health and safety.

4. Ensure own personal health and safety, and that of others in the workplace through precautionary measures.

5. Identify and recommend opportunities for improving health, safety, and security to the designated person.

6. Report any hazards outside the individual's authority to the relevant person in line with organizational procedures and warn other people who may be affected.

7. Follow organization's emergency procedures for accidents, fires or any other natural calamity in case of a hazard.

8. Identify and correct risks like illness, accidents, fires or any other natural calamity safely and within the limits of individual's authority.

Semester 2 General Content

L5 – Sem 2 - GC - Communication English:

AIMS: 1) To familiarize students with basic English
2) To enable them to develop listening & speaking skills

OBJECTIVES: Students should be able to ____

- 1) Write English without grammatical errors
- 2) Speak English Language Effectively and Accurately
- 3) Listen and understand public announcements and news on TV & Radio

Module – 1 Grammar

1. Subject and Verb Agreement
2. Voice
3. Articles
4. Speech
5. Question tag
6. Framing of Questions

Module – 2 Writing Skills

1. Letter Writing
Letter of Application/Letter of Grievances/Resume Preparation
2. Comprehension
3. Essay Writing

Module – 3 Speaking Skills

1. Greeting
2. Requesting
3. Enquiring
4. Explaining
5. Reporting
6. Permission
7. Thanking

L5 – Sem 2 - GC - Mathematics:

Section – A

Unit 1: Algebra

Sets –Relation - functions- Mathematical logic and Graph theory, Matrices and determinants –invertible matrices (only 2×2 matrices) - Characteristic equations and Eigen values (only 2×2 matrices).

Unit 2: Quadric Surfaces

Sphere - Ellipse –Cone-Ellipsoid-Paraboloid-Hyperboloid of one and two sheets – Intersection of Quadric surface and plane – Tetrahedron.

Unit 3: Integration and differential equations

Indefinite and definite integrals –simple problems – Differential equations of first order – separation of variables.

L5 – Sem 2 - GC - Physics:

Section – A

THERMODYNAMICS:

Definition of temperature, heat, change of state, green-house effect, first law of thermodynamics, thermal equilibrium, classification of thermodynamic system, zeroth law of thermodynamics, internal energy, work, isothermal process, adiabatic process, isochoric, Isobaric Process, Second law of thermodynamics, reversible and irreversible process.

MOTION IN A STRAIGHT LINE:

Rest, motion, Position and frame of reference - Definitions of path length and displacement - Definitions of average speed and average velocity, instantaneous speed and instantaneous velocity & uniform and non-uniform motion – Uniformly accelerated motion.

Section – B

List of Experiments:

- 1) Using a simple pendulum, plot L-T and L-T² graphs, hence find the effective length of second's pendulum using appropriate graph.
- 2) To find the weight of a given body using parallelogram law of vectors.
- 3) To determine the coefficient of viscosity of a given liquid by measuring the terminal velocity of a spherical body by stoke's method.
- 4) To determine Surface tension of Water by capillary rise method.
- 5) Interfacial between Water and Kerosene.
- 6) Verification of Lami's theorem.
- 7) Verification of Gaussian distribution and calculation of standard deviation in a Monte Carlo experiment.
- 8) Determination of the moment of inertia of an irregular body using Torsional pendulum.

L5 – Sem 2 - GC - Computer Science:

Section - A

1. Concepts of object oriented programming: object class. Encapsulation data hiding Inheritance. Polymorphism. Analysis and design of system using Object Oriented Approach.
2. Structure of a C++Program: Include files, Declaration of an object, Main function, I/O streams.
3. Classes: Class Declaration: Data Members, Member Functions, Private and Public members. Data hiding and encapsulation, arrays within a class. Class function Defection, scope resolution operator, Private and Public member function, Nesting of member functions.
4. Objects: Creating Objects, accessing class data members, Accessing member functions. Arrays of Objects, Objects as function arguments: Pass by value by Reference, Pointers, to Objects.

Section - B

5. Constructors and Destructors: Constructors: Declaration and Definition, Default Constructors, Parameterized Constructors, Copy Constructors. Destructors: Definition and use.

Section - C

6. Function Overloading: Function Overloading: Declaration and definition.
7. Inheritance – Extending Classes: Concept of inheritance, base, derived class, defining derived, classes, visibility modes, private, public, protected; sinle inheritance: private and protected members by member functions of a derived class, multilevel inheritance, nesting of classes.

Reference Books:

Object Oriented Programming with C++ By Balagurusamy- McGraw-Hill,
Object Oriented Programming And C++ By R. Rajaram
PRACTICAL: Practical's based OBJECT ORIENTED PROGRAMING (USING C++)
Objective

Semester 3
Certificate – Level 6 – Animation Director(A)
(National Competency Certificate)

Vocational Content

MES / N 1304 - Communicate requirements to the team

Section - A

Introduction to Communication, Verbal Communication, Face to Face communication, Tone of Voice, Body Language, Physical Communication.

Introduction to Listening Skills, Self-Awareness, Active Listening, Listening in Difficult Situations, Shades of Meaning. Formal Communication, and Informal communication.

Effective Communication: Voice with Emotions, Speaking what you thinking, Having a group discussions, Making Teams and Dividing responsibility.

Section – B

1. Writing Letter

- a. Writing a letter for Long leave
- b. Writing a letter for Pramotion

2. Improvising the Content

- a. Adding an emotional touch to your letter.
- b. Correcting the meaning.

3. Group Discussion and Games for Communication.

- a. Wordless Acting
- b. Famous Pair
- c. Listen, Interpret, Draw
- d. Role-playing and Conflict Resolution
- e. y-o-u-r-n-a-m-e

4. Communication Through Activities.

- a. Listening with Intention
- b. Record/Watch yourself talk
- c. DPR on Given Content.

MES / N 1306 - Direct the animation process

Section – A

Basic Principles in animation: Squash and stretch, Anticipation, Staging, straight ahead and pose to pose, follow through and overlapping action, Slow in and slow out, Arcs, Secondary action, Timing, Exaggeration, Solid drawing, Appeal, Mass and weight, Character acting, Volume, Line of action, Path of action, Walk cycles- animal and human.

Introduction to Sculpting software interface: Mesh Manipulation, Using different, applying strokes, Projection mapping, Dynamic topology, Adding detail to poly mesh, Using symmetry tool, Exporting the mesh.

Introduction to basic material types & procedurals: Study of concepts:- opacity, smoothness, specularity and color. Drawing 2D art templates, Creating complex effects like water fire and smoke. Unwrapping the map for various 3D characters.

Introduction to the mapping and advanced texturing techniques. Shadow maps, Raytraced shadows & radiosity. Concept of lighting system and shadows. Introduction to 3 point, 2 point and dramatic lighting. Creating photo real environments and textures. Applying on to a 3D objects. Understanding how to produce final output, rendering the scene, rendering the effects, network rendering.

Introduction to advance lighting effects: Mental ray rendering and Toon shade rendering. Creating various outputs as per the end user requirements and maintaining the resolution.

Introduction to Character setup: Riggers role, Criteria for a good rig, Joints and skeletons, Creating skeleton hierarchy, Constraints, Forward(FK) and Inverse kinematics(IK), FK, IK joint structures, Animation controllers, Blend shapes, Clusters, Biped Rig- Analyzing reference, Anatomy of human body, Bone placements, Setting up Torso, Biped Arms, Fingers, Legs/Foot controls, Skinning, Facial Rig- Anatomy of a face, The Facial Action Coding System(FACS), Mouth shapes, Phonemes, Animation controllers for Face, Character GUI.

Introduction to Fur, Procedural textures: Inclination, roll and polar, Fur volume and Noise, Painting fur attributes, modifying fur direction, Modifying color of a fur descriptions, Creating nCloth collision, Constraints, Hair System: Artisan, Hair system components, Modify curve tools, Paintfx with hair, Hair collision, Hair system caching, Hairstyles.

Muscle systems Introduction to 3D Muscle system: Components of Muscle, Capsules, Spline based muscle system, Stretch based muscle systems, Muscle Objects and skins, Assigning weights to Muscle skin, Muscle Builder, Muscle Parameters, Custom muscle systems, Wrinkles, fold, jiggle, Slide, Collide, Muscle control.

Section – B(1)

1. Exercises: Ball Bounce (Heavy & Light).
2. Exercises: pendulum.
3. Exercises: Walk Cycle.
4. Exercises: Run.
5. Exercises: Vanilla Walk.
6. Exercises: Animal Walk Cycle.

Section – B(2)

1. Exercises: Human Nose sculpt.
2. Exercises: Human Eyes Sculpt.
3. Exercises: Human Mouth Sculpt.
4. Exercises: Human Ears Sculpt.
5. Exercises: Human Character planning.
6. Exercises: Human Character body sculpting.
7. Exercises: Human Character Face sculpting.

Section – B(3)

1. Exercises: Shading networks.
2. Exercises: Tire Texturing.
3. Exercises: Rusty Automotive Texturing.
4. Exercises: Human Skin Texturing.
5. Exercises: Animal Skin Texturing.
6. Exercises: 2 Point lighting.
7. Exercises: 3 Point lighting.
8. Exercises: Interior lighting (Lamps, Bulbs).

Section – B(4)

1. Exercises: Rigging with Deformers.
2. Exercises: Rigging simple vehicle.
3. Exercises: Rigging cartoon character-Legs.
4. Exercises: Rigging cartoon character- Arms.
5. Exercises: Rigging cartoon character- Face rigging.
6. Exercises: Face expressions using Blend shapes.

Section – B(5)

1. Exercises: Creating fur.
2. Exercises: Creating Hair.
3. Exercises: Simulating Sparks.
4. Exercises: Jiggle animation.
5. Exercises: Adding Muscle System to a human Arm.
6. Exercises: Muscle Weight Paint.

Semester 3 General Content

L6-Sem 3 – GC – Physics:

LAWS OF MOTION

Newton's first law of motion: Concept of inertia and force – Concept of momentum - Newton's second law of motion - Impulse, impulsive force and examples - Newton's third law of motion: Identification of action and reaction pairs with examples in everyday life.

Collisions: Elastic and inelastic collisions

SYSTEMS OF PARTICLES AND ROTATIONAL MOTION

Definitions of a rigid body, translator motion and rotatory motion - Centre of mass of a two-particle system - Mention of expression for position coordinates of centre of mass of (a) n particle system (b) a rigid body and (c) a uniform thin rod.

Section – B

List of Experiments:

- 1) To study the relation between force of limiting friction and normal reaction and to find the coefficient of friction between surface of a moving block and horizontal surface.
- 2) To find the downward force, along an inclined plane, acting on a roller due to gravity and study its relationship with the angle of Inclination by plotting graph between force and $\sin\theta$
- 3) Determination of the Moment of inertia of rectangular lamina by using torsional pendulum.
- 4) Determination of the Moment of inertia of circular lamina by using torsional pendulum.
- 5) Determination of the acceleration due to gravity and the radius of gyration by using Bar pendulum(Graphical method)
- 6) Determination of the Moment of inertia of irregular body by using torsional pendulum.
- 7) To find the force constant and effective mass of a helical spring by plotting $T^2 - m$ graph using the method of oscillation.
- 8) To study the relationship between the temperature of a hot body and time by plotting a cooling curve.

L6-Sem 3 – GC – Electronics:

Unit 1: Introduction to Electronics

AC and DC sources.

Ohm's law – statement and limitations, numerical as applicable. Definition of power and unit. Kirchhoff's laws- statement and explanation, numerical.

Passive components

Resistors – types, fixed resistors – definition and example. Colour code. Variable resistors – definition and example. Combination of resistors in series and parallel (no derivation) numerical.

Capacitors- Types – fixed Capacitors – definition and example, Variable Capacitors – definition and example. Equivalent of capacitors in series and parallel combination (no derivation) numerical.

Inductors – types – fixed inductor - definition and example. Equivalent of inductors in series and parallel combination (no derivation) numerical.

(15 hrs)

Unit 2: Semiconductor devices

Classification of solids, Introduction to semiconductors – Intrinsic and extrinsic Semiconductors, PN Junction diodes - characteristics.

Mention of Zener diode, tunnel diode, photo diode and LED and their applications.

(15 hrs)

L6-Sem 3 – GC – Computer Science:

Section – A

Unit 1: Python: Introduction, features, First Step with Python, Interactive Mode, Script Mode, Variables and Types, Mutable and Immutable Variables, Keywords, Operators and Operands, Expression and Statements, Input and Output, Comments.

Functions: Introduction, Module, Built in Function, Composition, User Defined Functions, Parameters and Arguments, Scope of Variables, ways of defining Functions.

Conditional and Looping Construct: Control Flow Structure, Looping Constructs.

Strings: introduction, Creating and initializing strings, Strings are immutable, Traversing a string, Strings Operations & string Slicing, String methods & built in functions, Regular expressions and Pattern matching

Section - B

Unit 2:

Lists: Introduction, creating a list, accessing an element of list, traversing a List, Appending in the list, Updating array elements, Deleting Elements, Other functions & methods, List as arguments, Matrix implementation using list, Functions with list, Function call.

Dictionaries: What is dictionary? Key-value pair, Creation, initializing and accessing the elements in a Dictionary, Traversing a dictionary, Creating, initializing values during run time (Dynamic allocation), Appending values to the dictionary, Merging dictionaries: An update, Removing an item from dictionary, Dictionary functions and methods.

Tuples: What is a Tuple? Tuple Creation, Add new element to Tuple, Tuple Assignment, Tuple Slices, Tuple Functions

Reference Books:

Beginning Programming with Python for Dummies by stef Maruch & Aahz Maruch.

L6-Sem 3 – GC – Indian Constitution:

UNIT I

Preamble of the Indian Constitution

- a) Salient features of Indian Constitution

UNIT II

- a) Fundamental Rights
- b) Fundamental Duties
- c) Directive principles of State Policy

UNIT III

- a) President – Election Method, Powers and Functions
- b) The Role of the Prime Minister
- c) The Parliament – Structure, Power and Functions(Lok Sabha and Rajya Sabha)
- d) Supreme Court – Organization and Jurisdiction

UNIT IV

- a) The Role of Governor in the Administration of State
- b) Powers and Functions of the Chief Minister
- c) Composition, Powers and Functions of both the Houses of State Legislature
- d) High Court – Organization and Jurisdiction

Semester 4
Certificate – Level 6 – Animation Director (B)
(National Competency Certificate)

Vocational Content

MES / N 1307 - Direct the post-production process

Advanced Dynamics and pyrotechnics: Introduction to Dynamics, and Dynamic solver, Particles, Emitters, Fields: Air, Drag, Gravity, Newton, Turbulence, Vortex, Volume, Particle collisions, Particle cache, Goals, Soft bodies, Springs, Rigid bodies, Constraints, Effect: Fire, Smoke, Fireworks, Lightning, Shatter, Curve flow, Surface flow, Rendering particles and effects, Maya Paint Effects, baking simulations, Render types. UNIT – 2 Fluid Effects Introduction to Fluids, Fluid field interaction, Fluid attributes, , Creating a non-dynamic 3d fluid effects, Creating dynamic 3D effect, Creating fire and smoke using Fluid dynamics, creating an ocean.

Introduction to nParticles and Nucleus solver, Nucleus node, Nucleus forces, Nucleus plane, Nucleus attributes, nParticles interaction, nConstraints, nCloth: simulations, nCloth dynamics properties, Working with nConstraints, Tearing cloth, Dynamic Property maps, Simulating cloth on moving character, nParticle caching, nConstraints, Creating Smoke simulations in nParticles, Creating liquid simulations in nParticles, Introduction to nHair, Creating Basic hair style, Creating a dynamic curve simulations.

Introduction to Video Compositing techniques and Colour Keying advanced 2D compositing and Ink paint techniques. Creating color models as per the model sheets. Creating color pallets as required paint and ink fields. Understand the dope sheets / X- sheets in production level. Arranging and adjusting the layers as per X- sheet. Advanced panning of camera and background, multiple cameras for showing depth in-between background, over lay and character layers. Introduction to compositing special effects into a scene using 3d graphics and 3d special effects in 2d layers. Concepts for Broadcast animation for logos, channel IDs and montages. Multi-Layer Compositing, Special Effects, Superimposition and Titling. Exporting various file format outputs as per the end user requirements.

Introduction to Audio Editing sound design in animated films, including recording, gathering audio, editing, mixing, and audio design principles, high-end Pro Tools editing, including use of plug-ins to premix a soundtrack, sound is a catalyst for the imagination, master/slave relationship of picture/sound, sound and memory, psychological implications of sound. Psycho-acoustics and the physics of sound. voice-over, musical composition, psychological role of Sound and aesthetic

3D Compositing and Color Grading: Compositing in Z axis, Creating camera and lights in compositing program, Creating shadows and reflections using layers, Blending videos, Roto Paint, using particles and advanced effects, 3D camera tracking and polishing, Integrating 3 D elements. Using Color correction tools, curve tool, masking and Isolating elements, Using adjustment layer for color grading, Noir theme, color and its meaning and impact on audience.

Advanced Video Editing and Visual Story Narration: post-production strategies, narrative structure, editing rhythms and cutting for performance logging and organization exercises; cutting a short scene provided by the instructor; analysis of editing in a feature film; and active participation in class discussions of student.

Section – B(1)

1. Exercises: 3d Disintegration/Shattering effects.
2. Exercises: Smoke Simulation and Rendering.
3. Exercises: Creating Ocean and waves.
4. Exercises: Creating Fire Blast Simulation-Configuring Container and emitter properties.
5. Exercises: Creating Fire Blast Simulation-Creating Debris.
6. Exercises: Creating Rain and fog.

Section – B(2)

1. Exercises: Video stabilization.
2. Exercises: Chroma Keying & Matte.
3. Exercises: 2D Camera Tracking.
4. Exercises: 3D Camera Tracking.
5. Exercises: 3D Camera Tracking and Alignment.
6. Exercises: Motion Graphics-Planning and Asset preparation
7. Exercises: Motion graphics- Final Animation
8. Exercises: Rotoscopy-Biped
9. Exercises: Rotoscopy-Quadraped.

Section – B(3)

1. Exercises: Adding BG Music.
2. Exercises: Mixing two tracks.
3. Exercises: Recording audio.
4. Exercises: Noise reduction Process.
5. Exercises: Adding Effects.
6. Exercises: Creating Physiological Effects
7. Exercises: Foley Lab.

Section – B(4)

1. Exercises: 3D camera tracking and exporting to 3D Program.
2. Exercises: adjusting Track Points.
3. Exercises: Set extension.
4. Exercises: Match Moving.
5. Exercises: matching Lighting

Section – B(5)

1. Creating Titles using text and images.
2. Storytelling with images.
3. Editing using video effects.
4. 180-degree dialogue shot editing.
5. Editing a Montage (Future, Past).
6. Shooting and editing a documentary- Shoot/Acquiring a footage
7. Shooting and Editing a documentary-Rough cuts
8. Shooting and Editing a documentary-Syncing Audio and Video
9. Shooting and Editing a documentary-Final cut and Export to media format.

Semester 4 General Content

L6 – Sem 4 – GC – Physics:

Section – A

OPTICS:

Reflection, Refraction, Spherical mirror, Concave mirror, refractive index, Relation between focal length and radius of curvature. Laws of Refraction, Snell's law, Dispersion, wave front, interference, diffraction.

CURRENT ELECTRICITY:

Resistance, Capacitance and Inductance, Ohm's law, Current and Voltage, Conductors, Insulators, Semiconductors and Super conductors and their applications, Alternating current and Direct current, Transformers and their constructions, rectification.

Section – B

List of Experiments:

- 1) To find the focal length of a convex mirror, using a convex lens.
- 2) To find the refractive index of glass.
- 3) To find the refractive index of water.
- 4) To find the focal length of a concave mirror.
- 5) Verification of ohm's law.
- 6) To determine effective resistance of two resistances connected in series.
- 7) To determine effective resistance of two resistances connected in parallel.
- 8) To determine resistance per unit length of the wire.

L6 – Sem 4 – GC – Physics:

Unit 1: Number Systems, Boolean algebra and Logic gates

Number Systems: Introduction to decimal, binary Inter – Conversion. Addition, Subtraction, Multiplication and Division in binary Number system.

1's and 2's Complement method in binary number system. Subtraction using 1's and 2's compliment

Boolean Algebra: Laws of Boolean Algebra, Logic gates - AND gate, OR gate, NOT gate, NAND gate and NOR gate Logic symbol and truth table

De – Morgan's theorem, simplification of Boolean expressions. Logic circuit for Boolean expressions and vice versa.

Universal gates - NAND gate and NOR gate.

Unit 2: Logic circuit and Memories

Combinational logic circuit: Half adder, Full adder, half subtractor, Full subtractor.

Sequential logic circuit: Flip – flop: RS, D, JK, and T flip - flop.

Shift registers: SIPO, SISO, PISO and PIPO.

Digital computer: Block diagram of digital computer and function of each block.

Semiconductor Memories: Idea of different types of semiconductor memories (RAM, ROM, PROM, EPROM and EEPROM).

L6 – Sem 4 – GC – Computer Science:

Section – A

Unit 1:

Classes in Python: Introduction, Namespaces, Scope Rules, LEGB Rule, Defining Classes, Importance of self, Class Attributes v/s Instance Attributes, Instances attributes, Adding methods dynamically, Accessing Attributes and methods, Accessing Methods, Built in class attributes, Private Members - Limited Support, Data Hiding, Static methods, Destroying Objects (Garbage Collection)
Inheritance: Introduction, types of Inheritance, Method-I- By using super() function, Method-II- By using name of the super class, Multiple Inheritance, Overriding Methods, Abstract Methods,

Unit 2:

Linear List Manipulation- Data Structures, Implementation of List in memory, Sequential Memory Allocation, List Operations, Traversal in a List, Insertion of an element in a sorted list, Searching Techniques, Sorting a list.

Stacks and Queues in List: Stack , Push operation, Pop operation, Traversal in a stack, Expression- Conversion of an infix expression to postfix expression, Evaluation of Postfix Expression

Queue- Queue operations,

Data File Handling: Introduction, File access modes:

Exception Handling & Generator Functions: Generator Functions, Advantages of using generator

Reference Books:

Beginning Programming with Python for Dummies by stef Maruch & Aahz Maruch.

Semester 5

Certificate – Level 7 – Live Action Director(A)

Vocational Content

MES / N 1301 (Evaluate production concepts and ideas)

SECTION A

Unit-I: Film as a Storytelling Device- The history of Storytelling - Plays vs. novels vs. film - What is a “story”? - The “idea” vs. “story” vs. “screenplay”. Structure of Screen play

Unit -II: The three act screenplay - The scene - Plot points. Script writing formats - Master Scene script format - Split page format – Fiction Script writing - Dramatic structure - Nonfiction forms and formats-Characterization

Unit- III: Narrative, tone, contrast, coincidence, Surprise, Suspense, tension techniques. Nature of protagonist, antagonist. Genres of screenplays. Principles of Characterisation.

Unit-IV Director’s preparation. Director’s role in film. Introduction to scene study and staging. Beats, fulcrum, dramatic blocks. Scene analysis. Art direction - location - floor management - out- doors and indoors.

Unit-V: Planning - pre-production- Concept, Story development - Casting, Locations, Financing. Production –Shooting, Direction and Cinematography. Distribution and Exhibition.

References:

- Writing the Script by Wells Root ,Jan 15, 1980
- Secrets of Film Writing by Tom Lazarus Jun 2, 2001
- Arora: Encyclopedia of Indian Cinema

SECTION B

Exercises

- 1. Story Development (Conceptualization)**
- 2. Script Improvisation (Shooting Script)**
- 3. Casting and Team building**
- 4. Location Scouting**
- 5. Working out the finance**
- 6. Presentation of full production planning(by PPT)**

MES / N 1302 (Sourcing financiers)

SECTION A

UNIT 1: Film business: distribution and exhibition of films - the basic mechanism of distribution and its relationship to various exhibition platforms. How the entrepreneurship of production relates to the distribution and exhibition.

UNIT 2: Publicity platforms: the who's/when/how's of publicity and a brief exposition of the various platforms available now.

Certification: a brief history and the current structure of central board of film certification. The procedure to go about film certification in India.

UNIT 3: Organizational structure of indian film industry: an overview of the different bodies that regulate and monitor the various constituents directly involved in making and marketing of films in india.

Budgeting for different types of films: relationship between budget, cost and time frame.

SECTION B

Producing practical:

location study: creative/logistics, script breakdown, mix-n-match
Budget top sheet, scheduling & call sheet preparation.

MES / N 1303 (Conceptualise the creative vision)

SECTION A

Unit I: Vision and visualisation - The social conditions and effects of visual objects. Visual culture. Critical visual methodology- Metaphor and metonymy – codes – semiology of the television medium

Unit II: An introduction to compositional interpretation - Colour - Spatial organization – Light - Expressive content- the Oedipus complex – id, ego, super ego – symbols – defence mechanisms – Dreams – Aggression and guilt – psychoanalytic analysis of the media

Unit III: Cultural analysis of visual – semiotics, denotation, connotation and iconography.

Unit IV: Analysis of film and television – six levels of analysis - psychoanalytic understanding of visual images.

Unit V: Selling magic – Breaking the advertising code – commercials and anxiety – Laconian gaze: other ways of seeing -Laura Mulvey and visual pleasure.

SECTION B

Practical Assignment: Actuality - field trip and writing observation report, and write a short film script on its basis. Produce photos of trip.

References:

- Media Analysis Techniques, Arthur Asa Berger, SAGE Publications, New Delhi, 1976
- Visual Methodologies, Gillian Rose, SAGE Publications, New Delhi, 2001
- Mythologies, Palatin. Barthes, R. London, 1973
- Techniques of interpretation, Dallas J.R.Ewing. SAGE Publication, New Delhi.

MES / N 1304 (Communicate requirements to the team)

SECTION A

Unit I: Introduction to Communication: Purpose of Communication; Process of Communication; Importance of Communication in Business; Differences between Technical and General Communication; Barriers to Communication; Measures to Overcome the Barriers to Communication.

Unit 2: Types of Communication: Types of Communication; Verbal Communication- Importance of verbal communication- Advantages of verbal communication- Advantages of written communication; Significance of Non-verbal Communication
Listening Skills: Listening Process; Classification of Listening; Purpose of Listening; Common Barriers to the Listening Process; Measures to Improve Listening; Listening as an Important Skill in Work Place.

Unit 3: Language for Communication: Language and Communication; General Principles of Writing; Improving Writing Skills, Essentials of good style, Expressions and words to be avoided; Grammar and Usage

Unit 4: Communication in Organizations: Internal Communication; Stake Holders in Internal Communication; Channels of Internal Communication; External Communication; Stake Holders in External Communication; Channels of External Communication.

Communication Network: Scope and Types of Communication Network; Formal and Informal Communication Network; Upward Communication; Downward Communication; Horizontal Communication; Diagonal Communication.

SECTION B:

Practical Assignment:

1. **Reading Skills:** Reading Skill; Purpose of Reading; Types of Reading; Techniques for Effective Reading.

2. **Team Co-ordination:** Create small teams of students and assign them in different production role and ask them to prepare a PPT of there production planning.

Semester 5 **General Content**

L7 – Sem 5 – GC – Advertising-Concepts and Principles:

Section – A

Unit I: Advertising - Definition, Nature, scope, Origin and Growth. Roles of advertising:

Social, Communication, Marketing and Economic.

Unit II: Functions of advertising. - Advertising in marketing mix - Types of advertising - merits and demerits - advertising and consumers - buying systems - target plans.

Unit III: Advertising objectives - Advertising campaign. Conceptualization of Advertising for TV, Radio and Print. Copy, Slogans and Writings.

Media planning - developing media objectives - media budget - selection of media - implementing media plans - pre-testing and launch.

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Unit IV: PR Definition, Elements of PR – functions of PR – Need of PR – Growth of PR in India, - Publicity, propaganda and public opinion Corporate Communication with the public – internal and external - Community Relations.

Unit V: Advertising agencies, Luminaries of Advertising, Advertising Ethics. Corporate Social Responsibilities and Advertising case studies.

SECTION B:

Practical Exercise: Each Student Should Choose a product and have to Create

1. Print Ad
2. Flier
3. Poster
4. Handouts
5. CD Cover
6. Ad video

Reference:

Advertising Basics Bove||Michael Newman Wiley, Creative Leaps(Reference), John Wiley & Sons (Asia), 2003

Innovative Promotions That Work, Lisa l.Cyr, Rock Port Publishers, 2006.

Mass Media, Anmol Publications Pvt ltd., J.L Kumar, New Delhi.2006

The Public Relations, 2nd Edition, Alison Theaker, Routledge, USA, 2004

Public Relations: Theory and Practice, Jane Johnston, Clara Zawawi, Allen & Unwin, 2009

L7 – Sem 5 – GC – Photography and Cinematography:

Section – A

Unit I: History of Photography, Painting and Photography, Types of Cameras – parts and functions of camera

Unit II: Lens and types of lenses for photography - Short, medium and long focal lengths. Exposure – Focusing, aperture, shutter speed, depth of field. Points of View, Composition and perspectives.

Unit III: Kinds of light- indoor and outdoor - Electronic flash and artificial lights. Capturing Colours. Light meters. Accessories and equipments for photography. Aesthetics and Technology.

Unit IV: Photo journalism – Basics of News Photography, specialization of Photography - Nature – Wild life – everyday life - People and places - Sports - Advertising - object photography.

Unit V: Ethical issues in Photography – Codes of Ethics for photographers. Tragedy image, digital improvements, privacy, moral rights of subjects etc., – Basic software of Photo Editing tools.

SECTION B:

Practical Assignments:

1. Intro to DSLR
2. Portrait Photography
3. Landscape Photography
4. Rule of third
5. Shot Composition
6. Camera moments

Reference:

- John hedgecoe's, New Introductory Photographic Course, Mitchell Beazley, 1990
- , 35mm Photographers Handbook, Pan Macmillan, Julian Clader, John Garrelt 1990
- Richard New man, How to take great Photographs at Night, Collins Brown,2003
- Photo-Journalism, Terry Hope Rotovision SA, 2001
- Lighting for Portrait Photography, Steve Bavister, Rotovision SA, 2001
- Roger hicks &Francis schultz, Darkroom basics and beyond, Patterson, 2000

L7 – Sem 5 – GC – Introduction to television broadcasting:

Section – A

Unit I: Origin and Development of Electronic Journalism – Broadcasting System UK, US and other developed and Third World Nations – Comparison.

Unit II: Indian Broadcasting Policy – Major recommendations of committees and working groups – Prasar Bharathi Bill – Question of autonomy.

Unit III: Planning and Production of TV, Radio Programmes – General and Special audience Programmes – Principles of writing for radio – Language – Different formats of radio programmes – Features, Debate, Talks, Drama, Interview etc.,

Unit IV: Planning and Production of TV Programmes – TV crews – Writing for TV –

Visualization – Anatomy of News Cast & News Gathering – Various Programmes – General and Special Audience Programmes - Various formats of TV Programmes – ETV – News, Soaps, Epics, Sports, Talk Shows etc., - TV Commercials – Sponsors – TV News Agencies – Private Channels.

Semester 6

Certificate – Level 7 – Live Action Director(B)

Vocational Content

L7 - Sem 6 – SC - MES / N 1305 (Direct the production process)

SECTION A

Unit I: Direction : Role of a contemporary director, screen grammar, spatial connections, temporal connections, mise-en-scene.

Unit II: Principles of film, narrative form, non-narrative films, dividing a feature film into parts and Genres.

Unit III: Planning, pre-production- Concept / Story development, Scripting / Screen play writing, Budgeting, Casting – Locations - Financing.

Unit IV: Basics of mise-en-scene. Space and time, narrative aspect of mise-en-scene. Cinematographer – set properties - composition, duration of the shot, scene and sequences.

Unit V: Production – Shooting – Shots and scenes. Direction & Cinematography. Post production- Editing, Editing Equipments Sound recording, Dubbing, Special effects, Graphics & final editing.

Unit V: Architecture of the film's style and plot patterning, Codified formal system of narration devices. Ideological Constraints.

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SECTION B:

Practical Assignment: Students should Submit a Short Film or a Documentary of 5 Minutes in team or individually.

References

- The Cinema of India(1896-2000), Thoraval, Yves(2000)
- The Subject of Cinema, Roberge, Gaston
- Films for an ecology of Mind, Roberge, Gaston (1977):
- Encyclopedia of indian Cinema, Arora.

L7 - Sem 6 – SC - MES / N 1307 (Direct the post-production process)

SECTION A

Unit I: Direction : Role of a contemporary director, screen grammar, spatial connections, temporal connections, mise-en-scene.

UNIT II : Bird's-eye view of the modern editing process and modern editing terminology

Principles of Video Editing: Various principles of Editing like Contrast, Parallelism, Symbolism, Simultaneity Continuity, Making an edit invisible, Motivation for every edit, Delivering a message, Bearing audio in mind, editing is creating, Control of Overuse technique or Visual effects.

Editing setup:-efficient editing habit, factory-standard systems to support the editing process

Non- Linear Editing (NLE) Concept.

The Three- Point Edit: Understanding Three-Point Editing, Overview of the Three-Point Editing Process, Different Ways to Do Three-Point Editing.

Working in the Timeline / Transitions / Key framing / Applying Filters / NLE

Compositing / ColorCorrection & Color Grading / Titling / Final Review &Project.

UNIT III: Stabilizing a Shot , Controlling shaky video, Cropping the borders efficiently, Analysing and Tracking a Point in the footage, The Tracker Panel & Motion Tracker Options, Corner Pin Tracking, Exporting Track data, Introduction to Masks, Animating Masks, Working with Mask Interpolation, Basics of Rotoscopy, Analysing the shot for Rotoscopy, Using Masks for Position Key frames, Creating a Simple Rotoscopic Animation.

UNIT IV: Introduction to Particle Playground, Filters, Plugins, Understanding the Gravity, Mass, Vortex, Turbulance, Color, Fields etc.

Animating the Emitter, Key frame animation, Path Animation, Basic expressions, BASIC COMPOSITING, Applying Layer Blending Modes, Creating a Track Matte, Keying & Key light, Compound Effects: Gradient Wipe Displacement Map, Wave World & Caustics, Pre- composing & Nesting.

SECTION B:

Practical Assignment :

1. Arrange the previously shot Rushes in Sequence(Rough cut 1)
2. Dubbing the audio (Recording the audio)
3. Aligning the audio with the Rushes (Rough Cut 2)
4. Compositing the Shots
5. Match moving
6. Finalizing the Edit and Adding Title Card and Credits
7. Rendering the Edit (Exporting)

Reference Books:

1. Eve Light Honthaner, "The Complete film Production Handbook, Volume 1", Focal Press Publications, 2001, 3rd Illustrated Edition.
2. Adele Droblas and Seth Greenbeg, "Adobe Premiere Pro 2 Bible (W/ Cd)", Wiley – India Publications, 2007 Edition.
3. J. J. Marshall and Zed Saeed, "After Effects 5 Bible", John Wiley and Sons Publication, 2002.

L7 - Sem 6 – SC - MES / N 1 308 (Manage the production's marketing and release)**SECTION A**

Unit I: Concept of Brand: Introduction, Role of brand, Advantages of Branding, Core Values of Brand. Brand Equity: Concept, Brand elements. Principles of Branding, Communication Mix Strategy. Brand Media: Communication Channels and techniques, Advertising, Brand Perception.

UNIT II: Role of media, selection of media for advertising, formulation of message, art of copywriting. Branding and packaging strategies: Brand equity, image and personality, packaging decisions, perceptual mapping of customers, control aspects of advertising, advertising budget.

UNIT III: Strategic marketing- GAP analysis, Porter's five force model, Ansoff's matrix, SWOT analysis, customer relation management.

SECTION B:

Practical Assignment:

Attend a Film Launch event and write a brief Note on it with Photographic reference.

MES / N 0104 - Maintain workplace health and safety

Section – A

Understand and comply with the organization's current health, safety and security policies and procedures

Understand the safe working practices pertaining to own occupation. Understand the government norms and policies relating to health and safety including emergency procedures for illness, accidents, fires or others which may involve evacuation of the premises. Participate in organization health and safety knowledge sessions and drills

Section – B

1. Identify the people responsible for health and safety in the workplace, including those to contact in case of an emergency.
2. Identify security signals e.g. fire alarms and places such as staircases, fire warden stations, first aid and medical rooms.
3. Identify aspects of your workplace that could cause potential risk to own and others health and safety.

4. Ensure own personal health and safety, and that of others in the workplace through precautionary measures.
5. Identify and recommend opportunities for improving health, safety, and security to the designated person.
6. Report any hazards outside the individual's authority to the relevant person in line with organisational procedures and warn other people who may be affected.
7. Follow organisation's emergency procedures for accidents, fires or any other natural calamity in case of a hazard.
8. Identify and correct risks like illness, accidents, fires or any other natural calamity safely and within the limits of individual's authority.

Semester 6 General Content

L7 – Sem 6 – GC – Film Aesthetics and appreciation:

Section – A

Unit I: Introduction - Genres of films – Film History - language, style, grammar, syntax. Film perception: levels of understanding – Film as a medium: Development of cinema as a medium of art and communication.

Unit II: Understanding basic elements of Film: Aesthetics, Acting, Costume, Music Cinematography, Mise-en scene, Sound, Editing, Visual Effects and roles of artists. Film and Meaning.

Unit-III: Classic Film theory and semiotics – Sociology of films - film language - film and psycho – analysis. Post Modernism and structuralism and deconstruction. Post Colonial theory. Impressionism, expressionism, and surrealism - Subjectivity, causality and time and various schools of thoughts.

Unit-IV: Concepts of films – identities and issues - narrative form - non-narrative form- Film style: French New Wave, Soviet Montage Cinema, Asian Cinema.

Unit- V: Indian Classic Cinema. Indian Luminaries of Film and their films. Trend setters. Tamil Cinema and trends - Impact in politics, culture and economy. Film and post modernism - Film and cultural identity: Criticism and Film Review Writing.

Section – B

Practical Assignment: Students must watch select a genre and watch at least 10 movies of that genre and rank them from 1 to 10 with a brief note.

References:

- Indian Film, Eric Baranenn & Krishnaswamy OVP, 1980 2nd Edition
- How films are made, Khwaja Ahemad Abbas, National Book Trust, 1977
- Film as an art and appreciation, Maric Setton, NCERT, New Delhi
- Cinematography Censorship rules, Govt. of India Press, Nasik, 1969

L7 – Sem 6 – GC – Introduction to modern media:

Section – A

Unit I: Communication – Technology – Genesis – Application of Scientific Advancements – Electronic Revolution – Elementary Idea and about Electronic Devices – Valves – Transmitters, Integrated Circuits etc., Amplifiers – Oscillators – Modulators – Antenna etc.,

Unit II: Basic Block Diagram of Telecommunication Systems – Telegraphy – Telephone Exchange – Fax etc., - Telecommunication Networks – Policies.

Unit III: Principles of Radio Communication – Working of Radio Transmitters and Receivers
Elements of Radio Transmitters – Receivers – FM / AM Radio Systems – Narrowcasting.

Unit IV: Printing Technology – Type Faces – Fonts – Point System – Method of Composing Machine, Photo, Computers – VDT – DTP – Printing Methods – Traditional and Modern Methods of Electronic Scanners – Laser Printers – Facsimile.

Unit V: Computers – Digital – Computer Networks – E-Mail – Teleconferencing – Internet
Optical Fiber – Interactive Video – CD ROM – Transponder Data Flows – New Communication Technologies and Social Implications, Social Media, Blogging, Micro Blogging. Etc.,

SECTION B:

Practical Assignment:

Each student should create a Blog and a YouTube Channel.

Reference Books

1. Asok Mitra, Information Imbalance in Asia – AMIC, 1978.
2. Binod C. Agarwal, SITE INSAT.
3. Everett M. Rogers, Communication Technology, The New Media in Society.
4. Everett M. Rogers & Others, India's Information Revolution, Sage Publication, New Delhi, 1986.

L7 – Sem 6 – GC – Introduction to modern media:

Section – A

Unit I: Nature and Principles of Constitution of India (a) Fundamental Rights (b) Rights to Information (c) Freedom of Expression (d) and Freedom of Press in various political setup (i) in Capitalistic (ii) in socialistic Countries (iii) India before and after Independence.

Unit II: Constitutional restrictions on Media – Privileges of Media Personnel – The Indian Penal Code, Indian Evidence Act – Libel – Slander – Defamation – Contempt of Court. The Press Registration of Books Act 1867 – Copy Right Act – Periodical Changes – Post and Telegraphy Act – Official Secrets Act of 1923.

Unit III: Press Council – Working Journalists Act – MRPTC (Monopolies and Restrictive Trade Practice Commission) – Industrial Dispute Act – Incident Representation of Women Act of 1989. Rights and Abilities of the Editor, Printer and Publisher – Editorial autonomy and Independence – Government Information Services and their controls – PIB.

Unit IV: Code of Ethics in Different media – Morality of the Press and Cinema as Private Sectors - Radio and TV as States owned – Public utility undertakings. Evolution of code of ethics for media – Historical Perspectives & details.

Unit V: Freedom and Social Responsibility of Media - Tabloid Journalism – Investigative Journalism. Cable Media, its regulation and laws governing cable media –Internet as a media and the governing of rules of internet.

Reference Books

1. Adhikari Gautam, Press Council, Press Institute of India, New Delhi.
2. Arun Bhattacharjee, The Indian Press, Profession to Industry, Vikas Publication, New Delhi, 1972.
3. Chatterjee P.C., Broadcasting in India, Sage Publication, New Delhi, 1988.
4. Christians K. Rozeth Media Ethics, Cases and Moral Reasoning, Longmans, New York/ London, 1987.
5. Clement J. Jones, Mass Media, Code of Ethics and Councils.
6. Durga Das Basu Laws of the Press in India, Joy Print Pack Pvt. Ltd, New Delhi, 1986.

ENE21001

SEMESTER V

DISCIPLINE SPECIFIC COURSE - 13

5.1 ENTREPRENEURSHIP DEVELOPMENT

L: T: P-3:1:0

Course Outcome:

On successful completion of this course the students can:

CO1: Understand in depth the required characteristics and procedure to become an young entrepreneur

CO2: Specify in details with application, if applicable, easily access different financial scheme to small business

CO3: Identify in detail with examples to easily different financial schemes offered by Banks and Government Agencies

C04: Understand in depth and identify the social responsibility of an entrepreneur towards different sectors

C05: Learn in depth the Self employment opportunities

C06: Learn in detail the ethics in business

C07: Identify the content of project report

C08: Understand in depth the procedure of setting up of new business

Unit 1: Introduction

Meaning, definition of Entrepreneur, Enterprise, Entrepreneurship, Characteristics of successful entrepreneur, Functions, Role of entrepreneur in economic development, women entrepreneur, Rural entrepreneur, Agricultural entrepreneur-meaning and challenges.

Unit 2: Entrepreneurship Development Program (EDP)

Meaning, objective, importance, institutions doing EDP in India, DIC, CEDOCK, SSI, NSIC, EDII, AWAKE, KVIC, RUDSET, Industrial estate-Meaning and importance.

Unit 3: Financing of Small Business in India

Institutional and non institutional assistance SFCs, banks, SIDBI, NBFC-meaning and schemes; venture capital, bills discounting, factoring, state and central government subsidies and incentives for SSI (existing) - recent industrial policy(2011), PM MUDRA YOJANA- meaning, objectives, procedures for obtaining loan under MUDRA.

Unit 4: Setting up of new business, forms for small business

Small proprietorship, partnership, private company, cooperative society-meaning and nature, project formulation, project report-meaning, importance, general format of project report, project appraisal, financial, technical, marketing, social feasibility study, obtaining license, clearance certificate, registration procedure.

Unit 5: Business Ethics

Meaning, ethics in business, importance, various social responsibility of an entrepreneur towards customers, suppliers, government and society, self-employment-recent trends in the areas of self employment-event management-meaning and areas of business in event management (party organizing , catering, wedding plan and corporate event plan) tourism-meaning, tourism products, E-marketing as self employment opportunity.

Reference:

1. Entrepreneurship And Small Business Management- C B Gupta And S S Khanka
2. Entrepreneurship Development – C B Gupta And Srinivasan
3. Entrepreneurship development development –Shankaraiah
4. Entrepreneurship development-S S Khanka

ENE22001

SEMESTER V
DISCIPLINE SPECIFIC COURSE -14
5.2 IFRS (IND - AS)

L:T:P-3:1:0

Course Outcome:

On successful completion of this course the students can:

CO1: Deliberate the characteristics of IFRS

CO2: Understand in depth frame work for the preparation and presentation of financial statement

CO3: Learn in details with examples Accounting for assets and liabilities

CO4: Learn in details with examples IND AS on business combination

CO5: Understand the details of IND AS in relation to accounting for Revenue and Expenses

Unit 1: IFRS

Meaning, Need for IFRS, IASB, IASB Current Structure, IFRS Due Process, Benefits and limitations of IFRS; Introduction to Indian Accounting Standards (IND-ASs - Converged IFRSs), Accounting Regulations in New Companies Act – 2013, IFRSs issued by IASB (1-17), List of IND-ASs.

Unit 2: Framework for the Preparation and Presentation of Financial Statements

Meaning of Framework, Purpose and status - Scope - Application - Uses and their information and needs – objectives - Underlying assumptions – Qualitative characteristics – Elements of Financial statements – Recognition and Measurement of the elements of financial statement Balance sheet, Statement of changes in Equity, Statement of Profit and Loss consolidated financial statements, other comprehensive income as per schedule III of the Companies Act, 2013 – **Simple problems on each statement.**

Unit 3: Accounting for Assets and Liabilities

Recognition and Measurement criteria for Investment Property (IND-AS 40) – Borrowing Cost (IND-AS 23) – Provisions, Contingent Assets and Contingent Liabilities (IND-AS 37) - Share-based Payment (IND-AS 102) - . Recognition and Measurement for Property, Plant and Equipment (IND-AS 16) – Intangible Assets (IND-AS 38) - Inventories (IND-AS 2) - Leases (IND-AS 17) – Impairment of Assets (IND-AS 36) - **Theory and Simple Problems only**

Unit 4: Accounting for Revenue and Expenses

Income Taxes (IND-AS 12) – Employee Benefits (IND-AS 19) -Construction contracts (IND-AS 11) - Revenue (IND-AS 18) - Revenue from Contracts with Customers (IND-AS 115).

Unit 5: IND-AS on Business Combination

IND-AS on Business Combination, Consolidation and Disclosure-Consolidated Financial Statements (IND-AS 110) – Joint Arrangements (IND-AS 111) – Business Combinations (IND-AS 103) - Related Party Disclosures (IND-AS 24) – Operating Segments (IND-AS 108) – First-time adoption of International Financial Reporting Standards (IND-AS101)– Financial Instruments: Disclosures (IND-AS 107) – Disclosure of interests in Other Entities (IND-AS 112) –Earning Per Share (IND-AS 33) – Interim Financial Reporting (IND-AS34)– Insurance Contracts (IND-AS 104)– **Theory and Simple Problems only.**

References:

1. A Quick Guide to Indian Accounting Standards (Ind-AS) by Chethan N. Patel and BhupendraMantri, Taxmann Publication (P.) Ltd.

2. Students' Guide to Ind ASs – Converged IFRSs by Dr. D.S. Rawat, Taxmann Publication (P.) Ltd.
3. IFRS for India, Dr.A.L. Saini, Snow white publications
4. Roadmap to IFRS and Indian Accounting Standards by CA ShibaramaTripathy
5. IFRS concepts and applications by Kamal Garg, Bharath law house private limited
6. IFRS: A quick reference guide by Robert J Kirk, Elsevier Ltd.

ENF21001

SEMESTER VI

DISCIPLINE SPECIFIC COURSE - 15

6.1 PRINCIPLES AND PRACTICE OF AUDITING

L:T:P-3:1:0

Course Outcome:

On successful completion of this course the students can:

CO1: Learn in depth to practice as an Auditor

CO2: Learn the characteristics of errors and frauds and minimize them in maintenance of books of accounts

CO3: Identify in detail the importance of Internal Control and Internal Check

C04: Identify the details of audit planning

C05: Learn in depth verification and valuation of Assets and Liabilities

C06: Deliberate in details with examples audit of different types of organizations

Unit 1: Introduction

Meaning and definition of auditing- Nature and importance of auditing objectives of auditing-advantages, different types of audit, qualities of an auditor- audit report-auditing in a computerized environment.

Unit 2: Audit planning and control

Factors affecting audit planning - audit programme advantages-audit note book-appointment of a company auditor- qualifications, disqualifications-rights and duties of a company auditor.

Unit 3: Internal check and internal control

Meaning and objective-internal check for various transactions-limitations of internal control. Vouching-meaning and importance – vouching of cash transactions.

Unit 4: verification and valuation of assets and liabilities

Meaning- problems in valuation of assets, verification and valuation of assets and liabilities- goodwill, Stock in trade, Investments, Patents, Copy rights and trademarks, plant and machinery- capital, creditors, debentures, outstanding expenses, contingent liabilities.

Unit 5: Audit of different types of organizations

Audit of sole trader, audit of partnership firms, audit of hotels, audit of educational institutions, audit of trust, audit of co-operative societies.

References:

- | | |
|------------------------|--------------------|
| 1. Auditing | T R Sharma |
| 2. Practical auditing | B N Tandon |
| 3. Practical auditing- | Spicer and Spegler |
| 4. Auditing | Jagadeesh prakash |

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SEMESTER VI
DISCIPLINE SPECIFIC COURSE -16
6.2: BUSINESS LAWS

L:T:P-3:1:0

Course Outcome:

On successful completion of this course the students can:

CO1: Understand in details various laws related to business and able to work as legal adviser of business enterprises

CO2: Understand the characteristics of legal environment and practice business ethics

C03: Learn in depth and apply the basic legal knowledge to business enterprises

C04: Understand the characteristics of different intellectual properties and protect them

C05: Identify and appointed as member of various commerce and legal boards / committee

C06: Specify the details of Information technologies Act

C07: Learn the provisions of Special Contract

Unit 1: Introduction

Concept of law, Sources Of Law- Mercantile Law; Agreement, Contract-Definition, Essentials Of a Contract, Legal Rules As To Valid Offer And Acceptance; Termination Of An Offer.

Unit 2: Contractual Capacity

Minor's Agreement, Consideration-Definition, Essentials and Exceptions. Free Consent-Coercion, Undue Influence, Fraud, Misrepresentation, Mistake, Definition and Features only.

Unit 3: Special Contract

Contingent contract, quasi contracts, Wagering Agreement, Discharge of a Contract, Remedies for Breach of Contract.

Unit 4: Intellectual Property Act

Definition and Registration Procedure for Patent, Copy Right, Trademarks.

Unit 5: Information Technology Act 2000

Definition of Information-Digital Signature, Legal Recognition of Electronic Records, License to issue Digital Signature Certificate And Acceptance Of Digital Signature.

References:

1. Mercantile Law - N.D.Kapoor P C Tulsian and Bharat
2. Business Law - Tulsian
3. Mercantile Law - P P S Gogna .

ENC24001 / END24001

GROUP A1
SKILL ENHANCEMENT COURSE-1
COMPUTERIZED ACCOUNTING SYSTEM

L: T: P- 2:0:2

Course Outcome:

On successful completion of this course the students can:
CO1: Learn in depth the characteristics of computerized accounting system

C02: Identify the reason for differences between cash book and bank statement

C03: Learn in details with examples computerised accounting software

C04: Specify in details with examples periodic reports

C05: Learn the classification and characteristics of data entry system

C06: Prepare the quotation, purchase order etc., using computer software

Unit 1: a) – Computerised Accounting System

Meaning of computerised Accounting Traditional / Manual System of Accounting; process of Computerised Accounting; features of Computerised Accounting; difference between manual accounting system and computerised accounting. Process/Steps in Computerised Financial Accounting; Advantages of the Computerised Financial Accounting System; Various Components of a Computerised Financial Accounting System

b) - Voucher Types of Voucher; Bank Payment Voucher; Bank Receipt Voucher; Cash Payment Voucher ; Cash Receipt Voucher ; General Voucher.

Unit 2: - Bank Reconciliation Statement

Common transactions with bank; Cashbook and bank statement ; Reasons for difference between cash book and bank statement ; Bank reconciliation statement

Unit 3: Computerized Accounting

Scope of Computerized Accounting; Computerized Accounting Software; Creating a Corporation in Computerized Accounting Software. Chart of Accounts ; Creating Chart of Accounts ; Update Chart of accounts ; Charts of accounts Beginning Balances ; Organizational codes ; Editing charts of accounts;

Unit 4: Periodic Reports

General Journal ; Ledger; Trial Balance ; Income Statement ; Balance Sheet Inventory report; Account Receivable report; Account Payable report

Unit 5: Data Entry System

Voucher and Entry of vouchers ; Sales Entries ; Purchase Entries; Cash Receipt Entries ; Cash Payment Entries ; Bank transaction Entries Maintain the Records of Sales &

Customers ; Quotations to customers ; Sales orders ; Sales Invoices ; Sales receipts ; Record of Customers ; Sales report ; Customer Report.

Unit 6: Maintain the Records of Purchases and Vendors/ Suppliers

Preparing the Quotations; Purchase orders; Payment to vendors; Record of vendors; Utility Bills Payable; Purchase Report; Vendors Report.

References:

- 1.Computer Accounting (Accounting & Finance) Paperback – Michael Fardon
- 2.Computerized Accounting Spiral-bound – 2008- Arens and Ward (Author)

ENC25001 / END25001

GROUP A1

SKILL ENHANCEMENT COURSE-2

E-FILLING OF RETURNS

L:T:P-2:0:2

Course Outcome:

On successful completion of this course the students can:

C01: Learn in depth the types of e-filing and e-verification of returns

C02: Specify in details with examples E-filing of returns

C03: Understand the details of steps involved in e- filing of different forms of IT Return

C04: Learn the details of E-filing of returns under GST

C05: Learn the details of returns to be filled by composition tax payer

Unit 1: E-filing of Returns

Meaning- objectives- advantages, features of E-filing; Types of e- filing, e-filing Process flow, Types of e-Verification of Returns, How to e-file, who should file e-return;

Unit 2: ITR-1 and ITR2- ITR-3- ITR-4 introduction

Format- Heads of income covered- contents- filling the return- steps involved in filing of ITR-1, ITR-2, ITR-3, ITR-4

Unit 3: ITR-5- ITR-6; ITR-7- introduction

Format- Heads of income covered- contents- filling the return- steps involved in filing of ITR-5, ITR-6, ITR 7.

Unit 4: E-filing of returns under GST

GSTR-1- GSTR-1A, GSTR-2, GSTR 2A; GSTR-3, GSTR 3A, GSTR-3B, GSTR 9, Assesses required to file, contents to be filled, Documents required, steps in filing the above returns;

Unit 5: Returns to be filed by composition tax payers

GSTR-4A, GSTR-4, GSTR-9; Returns to be submitted by an input service distributor- GSTR 6, GSTR-6A; Returns to be filed by Tax Deductor-GSTR-7, GSTR-7A;Contents- steps to be followed in filing the above returns

References:

1.<https://incometaxindiaefiling.gov.in/eFiling/Portal>

2.<https://www.bankbazaar.com/tax/step-by-step-guide-efile-income-tax-return-online.html>

3.<https://www.icicibank.com/knowledge-base/tax/steps-for-e-filing>

4.<https://taxguru.in/income-tax/download-free-ebook-the-process-efiling-income-tax-returns.html>

5.Students Guide To e>Returns- vinod K singhania- taxmann

6.<https://www.profitbooks.net/gst-returns/>

ENC26001 / END26001

GROUP A1

SKILL ENHANCEMENT COURSE -3

PRINCIPLES AND PRACTICE OF GENERAL INSURANCE

L: T: P-3:1:0

Course Outcome:

On successful completion of this course the students can:

C01: Identify the classification and characteristics of General insurance

C02: Learn the details of fire insurance policies, insurance coverage and consequential loss

C03: Deliberate in details with examples underwriting and settlement of claims

C04: Understand in depth the details of claim procedure and Settlement limitation

C05: Learn in detail with examples non life miscellaneous insurance

Unit 1: Introduction

Meaning of General Insurance – The Evolution and Growth of General Insurance – Types of General Insurance – Fundamentals of General Insurance –Recent innovations. Organization and Management of General Insurance Companies – Regulatory Framework for General Insurance in India.

Unit 2: Fire Insurance

Standard policies – Fire Insurance coverage – Consequential loss (fire) Insurance policies – Declaration policies, Marine Insurance: Marine Cargo policies – Hull policies – Institute cargo clauses – Institute hull clauses – Open policies – Accumulation of risk per location - Motor Insurance: Types of policies – Third party Insurance – Comprehensive coverage – Conditions and Exclusions – premium.

Unit 3: Non life miscellaneous insurances

Personal Accident Insurance, Health Insurance and Mediclaim policies, Liability Insurance, Burglary Insurance other Miscellaneous Insurances, Rural Insurance covers, Engineering Insurance and its Consequential loss covers, Aviation hull and Aviation liability.

Unit 4: Underwriting and Settlement of Claims

Proposal forms, Cover notes, Certificates of Insurance, Endorsements, Moral and Physical Hazards, Statistics Spreading of Risks, Premium Rating, Premium Loading.

Unit 5: Settlement of Claims

Claim procedure, TPAs: Claim forms, Investigation / Assessment, Essential Claim Documents, Settlement Limitation, Arbitration, Loss Minimization and Salvage.

References:

1. Insurance Institute of India – IC 34 – General Insurance
2. Insurance Institute of India – IC 45- General Insurance Underwriting
3. Module I, Principles and Practice of General Insurance, The Institute of Chartered Accountants of India: New Delhi.
4. H Narayanan, Indian Insurance: A Profile, Jaico Publishing House: Mumbai.
5. K.C. Mishra and G.E. Thomas, General Insurance - Principles and Practice, Cengage Learning: New Delhi

ENC27001 / END27001

GROUP A1

SKILL ENHANCEMENT COURSE - 4

LOGISTICS AND SUPPLY CHAIN MANAGEMENT

L: T: P-3:1:0

Course Outcome:

On successful completion of this course the students can:

C01: Understand in depth supply chain management

C02: Write down the details of designing the supply chain network

C03: Learn in details with examples designing and planning transportation network

C04: Learn the details of Information technology in supply chain

C05: Identify the dimensions of logistics

C06: Understand in depth the details of demand management and customer care

Unit 1: Introduction to Supply Chain Management

Supply chain – objectives, importance, decision phases, process view, competitive and supply chain strategies, achieving strategic fit, supply chain drivers, obstacles, framework, facilities, inventory, transportation, information, sourcing, pricing.

Unit 2: Designing the Supply Chain Network

Designing the distribution network, role of distribution, factors influencing distribution, design options – e-business and its impact, distribution networks in practice, network design in the supply chain, role of network, factors affecting the network design decisions, modelling for supply chain.

Unit 3: Designing and Planning Transportation Networks

Role of transportation, modes and their performance, transportation, infrastructure and policies, design options and their trade-offs, tailored transportation. SOURCING AND PRICING: Sourcing, In-house or Outsource – 3rd and 4th PLs – supplier scoring and assessment, selection, design collaboration, procurement process, sourcing planning and analysis. Pricing and revenue management for multiple customers, perishable products, seasonal demand, bulk and spot contracts.

Unit 4: Information Technology in the Supply Chain - IT Framework

Customer relationship management, internal supply chain management – supplier relationship management, transaction management, future of IT. Coordination in a supply chain: Lack of supply chain coordination and the Bullwhip effect, obstacle to coordination, managerial levers, building partnerships and trust, continuous replenishment and vendor-managed inventories, collaborative planning, forecasting and replenishment.

Unit 5: Dimensions of Logistics

Introduction: A macro and micro dimension, logistics interfaces with other areas, approach to analysing logistics systems, logistics and systems analysis, techniques of logistics system analysis – factors affecting the cost and importance of logistics.

Unit 6: Demand Management and Customer Service

Outbound to customer logistics systems, Demand Management, Traditional Forecasting, CPFRRP, customer service, expected cost of stock-outs, channels of distribution.

References:

1. Sunil Chopra and Peter Meindl, Supply Chain Management – Strategy, Planning and Operation, Pearson.
2. Coyle, Bardi, Longley, The management of Business Logistics – A supply Chain Perspective, Thomson Press.
3. Supply Chain Management by Janat Shah Pearson Publication.
4. Donald J Bowersox, Dand J Closs, M Bixby Coluper, Supply Chain Logistics Management, TMH, Second Edition.

ENE23001 / ENF23001

GROUP A2

SKILL ENHANCEMENT COURSE-5

CORPORATE TAX PLANNING

L: T: P-3:1:0

Course Outcome:

On successful completion of this course the students can:

C01: Learn in depth and specify the tax saving strategies for decision making

C02: Understand in depth the provisions and laws of tax and able to become tax consultant

C03: Identify the tax provisions and deductions and able to become tax planner with reference to business restructuring

C04: Write down the details and identify special provisions in respect of Free Trade Zone, Infrastructure Development and Backward areas

C05: Learn in detail the provision of Income Tax Act relating to amalgamation

C06: Understand the classification and characteristics of taxation relief

C07: Deliberate the provisions of IT Act in relation to managerial decisions

Unit 1: Introduction:

Corporation tax, Tax Planning, Tax Evasion, Tax Avoidance, Tax Management, Dividend Tax, Domestic Company, Foreign Company.

Unit 2: a) Tax Planning for new Business: Location and Nature of Business, Forms of Business Organization

b) **Tax Planning and Financial Management Decisions:** Tax Planning relating to Capital Structure Decision, Dividend Policy, Inter-Corporate Dividends and Bonus Shares

Unit 3: Tax Planning and Managerial Decisions

Tax planning in respect of Own or Lease, sale of assets used for scientific research, make or buy decisions, repair, replace, renewal or renovation of an asset, shut-down or continue decisions.

Unit 4: Special Tax Provisions

Tax provisions in respect of Free Trade Zone, Tax provisions in respect of Infrastructure Development, Tax provisions in respect of Backward Areas, Tax provisions in respect of Tax Incentives to Exporters.

Unit 5: Amalgamation

Meaning of amalgamation under the Income-tax Act, Transactions not treated as amalgamation, Actual cost and written down value when assets are transferred in a scheme of amalgamation, When a capital asset (other than a block of assets) is transferred, When a block of asset is transferred, Assets in amalgamation not treated as transfer, Transfer of capital assets to amalgamated Indian company.

Unit 6: Tax Payment

Tax deduction at source, Tax collection at source, and Advance payment of tax, Relief for double taxation [Secs. 90, 90A and 91], ADT agreements [Sec. 90], Modes of granting relief under ADT agreements , Unilateral relief [Sec. 91], Double taxation relief in case of specified associations [Sec. 90A]

References:

1. Direct Tax Laws and Practices- Vibnod K. Singhanian
2. Direct Taxes – H C Meharothra
3. Corporate tax planning & business tax procedures- Vinod K songhanian

ENE24001 / ENF24001

GROUP A2

SKILL ENHANCEMENT COURSE-6

COMPANY LAW AND SECRETARIAL PRACTICE

L: T: P-3:1:0

Course Outcome:

On successful completion of this course the students can:

C01: Understand in depth features of Companies Act -2013 and types of companies

C02: Learn in details the promotion of a Joint Stock Company and conversion

C03: Write down the details of conducting the Board of Directors and Subcommittee meetings

C04: Understand the procedure for alteration of Articles of Association and Memorandum of Association

C05: Learn in detail the appointment of company secretary

C06: Learn the preparation of minutes of various meetings

Unit 1: Companies Act

Introduction- companies Act 2013- features of companies Act - 2013, Types of companies- Public companies, Pvt company, statutory corporation, One person company, Dormant company, Associate company, Small company, Limited Liability Partnership- Application of Company Law to banking/insurance sector- Registrar of companies- functions, Ministry of Corporate affairs-functions; SEBI-functions of SEBI.

Unit 2: Secretary

Definition, Who can be company secretary, Appointment, General Legal position, Duties of a Company Secretary, Rights of Company Secretary, Liabilities of Company Secretary, Qualification for Appointment as secretary, Dismissal of the Secretary, Secretary in the Whole time practice, Secretarial Compliance certificate, Specimen form

Unit 3: Company Formation and Conversion

Company Formation and Conversion Choice of the form of the business entity, Conversion/reconversion of one form of business entity into another, Procedure for incorporation of private/public companies, Companies limited by guarantee and unlimited companies and their conversion/re-conversion registration., Obtaining certificate of commencement of business, obtaining certificate of re-registration, Commencement of new business and certification,

Unit 4: Procedure for alteration of various clauses of memorandum

Procedures for alteration of articles, Effect of alteration, specimen forms: Procedure for issue of Shares – Public Issue, Rights Issue and Bonus Shares, Issue of Shares at Par/Premium/Discount; Issue of Shares on Preferential /Private Placement Basis – Allotment, Calls on Shares and Issue of Certificates – Issue of Sweat Equity Shares,

Employees Stock Option Scheme (ESOPs), Employees Stock Purchase Scheme (ESPS), Shares with Differential Voting Rights

Unit 5: Meetings

Collective Decision Making Forums, Authority, Accountability, Delegation and Responsibility ; Board Meetings , Convening and Management of Meetings of Board and Committees; Preparation of Notices and Agenda Papers, General Meetings, Convening and Management of Statutory Meeting, Annual and Extra-Ordinary General Meetings, Voting

through Electronic Means; Conducting a Poll and Adjournment of a Meeting; Post-Meeting Formalities, Preparation of Minutes and Dissemination of Information and Decisions

References:

1. S. Srikanth , Shanti Rekha Rajagopal ,Revathy Blakrishnan, Corporate Laws and Secretarial Practice, Jain Book
2. M C Kuchhal, Secretarial Practice, Vikas Publishing House, New Delhi.
3. Sangeet Kedia, Advanced Company Law And Practice, Pooja Law Publishing Company,

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GROUP A2

SKILL ENHANCEMENT COURSE-7

QUANTITATIVE DECISION TOOLS

L: T: P-3:1:0

Course Outcome:

On successful completion of this course the students can:

C01: Understand in depth the characteristics and features of statistics

C02: Learn the details of methods and sources of collection of data

C03: Understand in details with examples Measures of central tendency

C04: Understand in detail the characteristics and application of correlation and regression

C05: Understand in details the classification and application of correlation

C06: Deliberate the characteristics of Index numbers

C07: Learn the characteristics of Testing of hypothesis with application

C08: Learn the creation of variables and entering data using SPSS

Unit 1: Introduction

Meaning and definition of statistics, functions, advantages, limitations- collection of data - methods of collecting primary data, and sources of secondary data - classification and tabulation-SPSS -Introduction, uses, creation of variables and entering data.

Unit 2: Measure of central tendency

Meaning and Definition of Averages-Arithmetic mean, Median, Mode [grouping method], standard deviation, Calculation of mean, median and Standard deviation using SPSS.

Unit 3: Correlation

Meaning and Definition, types of correlation, Methods of calculating correlation coefficient [Karl pear sons and Spearman's correlation] calculation of correlation using SPSS; Regression-Meaning and Definition, Distinction between correlation and Regression, Regression equations and estimations Calculation of Regression using SPSS.

Unit 4: Index Numbers

Definition – Types – Methods of Construction and Problems in the Construction – Cost of living index numbers.

Note-Besides solving problems chapter-wise in class rooms, teach using SPSS in computer lab.

Unit 5: Testing of Hypothesis

Meaning of hypothesis, types of hypotheses, test of Significance Procedure of testing hypothesis – Z test, one sample t test, Chi square test.

References:

1. Fundamentals of Statistics: D. N. Elhance, Veena Elhance and B. M. Aggarwal
2. Statistical Methods: S. P Gupta.
3. Fundamentals of Statistics: S.C Gupta
4. Practical Statistics: R S N Pillai and Bhagavathi
5. Statistics (Theory, Methods and Application): D.C. Sancheti and V.K. Kapoor
6. Statistics for Management: Richard I. Levin and David S. Rubin
7. Statistics: Dr. B.H. Suresh, Dr. G.H. Mahadevaswamy, Nithya Publications, Mysore.

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GROUP A2

SKILL ENHANCEMENT COURSE-8

BUSINESS RESEARCH METHODS

Course Outcome:

On successful completion of this course the students can:

C01: Learn in depth different methods of research, methodology, data collection, analysis and interpretation of data to become a good business researcher

C02: Understand and able to report about various issues of different organisations through research report

C03: Understand the details of types of Business Research and Research design

C04: Identify and contribute to the discipline of commerce and management through the research

C05: Deliberate the details of Data analysis

C06: Identify the details of Research Report

C07: Specify in details with examples sampling and hypothesis testing

Unit 1: Introduction to Business Research

Meaning, types, criteria of good research, scientific approach to research in physical and management science, limitations of applying scientific methods in business research problems, ethical issues in business research, research process, problem formulation, preparation of business research plan/proposal.

Unit 2: Business Research Design

Types of business research, Exploratory, Descriptive, and Causal research, Exploratory research: Meaning, suitability, collection, hypothesis, formulation, Descriptive research: Meaning, types of descriptive studies, data collection methods, Causal research: Meaning, various types of experimental designs, types of errors affecting research design.

Unit 3: Data Collection

Primary and Secondary data – Sources – advantages/disadvantages, Data collection Methods – Observations, Survey, Interview and Questionnaire design, Qualitative Techniques of data collection. Measurement and Scaling Techniques: Nominal Scale, Ordinal Scale, Interval Scale, Rating Scale, Criteria for good measurement, attitude measurement.

Unit 4: Sampling and Hypothesis Testing

Sampling: Meaning, Steps in Sampling process, Types of Sampling – Probability and non probability Sampling Techniques, Errors in sampling. Hypothesis: Meaning, Types, characteristics, sources, Formulation of Hypothesis, Errors in hypothesis testing.

Unit 5: Data Analysis

Unit 6: Research Report

Types, advantages, disadvantages, Components of research reports, format, chapterisation, language, referencing.

References:

1. Marketing Research – Naresh K Malhotra – Pearson Education.
2. Business Research Methods- S.N.Murthy/U.Bhojanna- Excel Books.
- 3 .Business Research Methods–Donald R. Cooper & Pamela S Schindler, TMH.
4. Marketing research: Text and cases- Rajendra Nargundkar – TMH.
5. Business Research Methods – Alan Bryman & Emma Bell, Oxford.
6. Research Methodology – C R Kothari- Vishwa Prakashan.
7. Business Research Methods – William G Zikmund- Thomson,
8. Methodology of Research in Social Sciences – O R Krishnaswami, M Ranganatham, HPH.

L:T:P- 1:0:3

C1 - Proposal of Project Work - 15 Marks

C2 - Progress of Project Work - 15 Marks

Viva - 20 Marks

Valuation of Report – 50 Marks

Course Outcome:

On successful completion of the project work the students are able to:

CO1: Understand in depth the gap between theory and practical through internship

CO2: Understand in detail with examples the procedure and able to write a report on the various issues of an organization

CO3: Convince and communicate effectively

CO4: Learn in detail and able to absorb as an employee by the employer

CO5: Specify and analyze the components of project report and prepare the report effectively

CO6: develop leadership qualities

DISCIPLINE SPECIFIC ELECTIVE**CONSUMER AFFAIRS****L: T: P-4:1:0****Course Outcome:**

On successful completion of the project work the students are able to:

CO1: Learn conceptual frame work of consumer and consumer market

CO2: Understand in depth the characteristics of consumer protection law in India

CO3: Deliberate the details of role played by the advisory bodies at different level

CO4: Identify the grievance redressal mechanism

CO5: Specify the details of role played by industry regulator in consumer protection

CO6: Understand in depth the consumer movement in India

Unit 1: Conceptual Framework

Consumer and Markets: Concept of Consumer, Nature of markets: Liberalization and Globalization of markets with special reference to Indian Consumer Markets, E-Commerce with reference to Indian Market, Concept of Price in Retail and Wholesale, Maximum Retail Price (MRP), Fair Price, GST, labelling and packaging along with relevant laws, Legal Metrology. Experiencing and Voicing Dissatisfaction: Consumer buying process, Consumer Satisfaction/dissatisfaction-Grievances-complaint, Consumer Complaining Behaviour: Alternatives available to Dissatisfied Consumers; Complaint Handling Process: ISO 10000 suite

Unit 2: The Consumer Protection Law in India

Objectives and Basic Concepts: Consumer rights and UN Guidelines on consumer protection, Consumer goods, defect in goods, spurious goods and services, service, deficiency in service, unfair trade practice, restrictive trade practice. Organizational set-up under the Consumer Protection Act

Unit 3: Advisory Bodies

Consumer Protection Councils at the Central, State and District Levels; Adjudicatory Bodies: District Forums, State Commissions, National Commission: Their Composition, Powers, and Jurisdiction (Pecuniary and Territorial), Role of Supreme Court under the CPA with important case law.

Unit 4: Grievance Redressal Mechanism

Grievance Redressal Mechanism under the Indian Consumer Protection Law, Who can file a complaint? Grounds of filing a complaint; Limitation period; Procedure for filing and hearing of a complaint; Disposal of cases, Relief/Remedy available; Temporary Injunction, Enforcement of order, Appeal, frivolous and vexatious complaints; Offences and penalties. Leading Cases decided under Consumer Protection law by Supreme Court/National

Commission: Medical Negligence; Banking; Insurance; Housing & Real Estate; Electricity and Telecom Services; Education; Defective Products; Unfair Trade Practices.

Unit -5: Role of Industry Regulators in Consumer Protection.

Banking: RBI and Banking Ombudsman. ii. Insurance: IRDA and Insurance Ombudsman iii. Telecommunication: TRAI iv. Food Products: FSSAI v. Electricity Supply: Electricity Regulatory Commission vi. Real Estate Regulatory Authority

Unit 6: Contemporary Issues in Consumer Affairs

Consumer Movement in India: Evolution of Consumer Movement in India, Formation of consumer organizations and their role in consumer protection, Misleading Advertisements and sustainable consumption, National Consumer Helpline, Comparative Product testing, Sustainable consumption and energy ratings.

Quality and Standardization: Voluntary and Mandatory standards; Role of BIS, Indian Standards Mark (ISI), Ag-mark, Hallmarking, Licensing and Surveillance; Role of International Standards: ISO an Overview

Note: Unit 2, 3 and 4 refer to the Consumer Protection Act, 1986. Any change in law would be added appropriately after the new law is notified

References:

1. Khanna, Sri Ram, Savita Hanspal, Sheetal Kapoor, and H.K. Awasthi. (2007) Consumer Affairs, Universities Press.
2. Choudhary, Ram Naresh Prasad (2005). Consumer Protection Law Provisions and Procedure, Deep and Deep Publications Pvt Ltd.
3. G. Ganesan and M. Sumathy. (2012). Globalisation and Consumerism: Issues and Challenges, Regal Publications
4. Suresh Misra and Sapna Chadah (2012). Consumer Protection in India: Issues and Concerns, IIPA, New Delhi
5. Rajyalaxmi Rao (2012), Consumer is King, Universal Law Publishing Company
6. Girimaji, Pushpa (2002). Consumer Right for Everyone Penguin Books.
7. E-books :- www.consumereducation.in
8. Empowering Consumers e-book,
9. e-book, www.consumeraffairs.nic.in

SEMESTER V**DISCIPLINE SPECIFIC ELECTIVE****INTERNATIONAL BUSINESS****L: T: P-4:1:0****Course Outcome:**

On successful completion of this course the students can:

CO1: Learn in detail about import and export and able to become an importer and exporter

CO2: Specify in detail the application of foreign trade policies and analyse how international factors affect domestic concern

CO3: Learn in depth and analyse legal issues related to international business

CO4: Identify and analyse various social culture and responsibility awareness on global issues

CO5: Understand in detail and identify sources of trade finance and forms of payment

CO6: Identify the function of WTO and regional integrations

CO7: Understand in depth characteristics of foreign exchange market

CO8: Identify the details of facilities provided by the International monetary institutions

Unit 1: Introduction

Globalisation – Meaning and implications - Globalisation of markets and production – The emerging global economy - Drivers of Globalisation. Modes and entry strategies of international business – arguments for and against – trends in international trade. Differences between domestic and international business.

Unit 2: International Business Environment

Cultural aspects, values and norms, social structure, religious and ethical systems, language, education, implications of cultural differences on business. International business environment, Political and legal factors, political systems, legal systems, International business environment, Economic factors, the determinants of economic development. Tariffs, subsidies, local content requirements, administrative policies, anti dumping policies, political and economic arguments for intervention ,Development of the world trading system.

Unit 3: WTO and Regional Integrations

GATT, the Uruguay round of negotiations. WTO, genesis and functions, the future of WTO. Regional Integrations, Trading Blocks, nature and levels of integration, arguments for and against regional integration, Trading blocks, European Union, ASEAN, APEC, NAFTA, SAARC.

Unit 4: Multinational Corporations

Organisation, design and structures, head quarters and subsidiary relations in multinational corporations.

Unit5: Foreign Exchange Market

Functions, nature of foreign exchange market, the trading mechanism, exchange rate determination, balance of trade, stability of exchange rate, currency convertibility

Unit 6: International Monetary System

Funding facilities and strategies of IMF and World Bank, Expatriation and Repatriation, Ethical dimensions in International Business.

References:

1. Charles W L Hill. And Arun Kumar Jain. International Business: competing in the global market place, Mc Graw-Hill.
2. John D. Daniels Lee H Radebaugh, International Business: Environments and Operations Addison Wesley.
3. Justin Paul – International Business – Prentice Hall of India.
4. Oded Shenkar Yadong Luo: International Business – John Wiley and Co.
5. Wild J. John, Wild L. Keneth and Han C. Y. Jerry, International Business: An integrated approach, Prentice Hall
6. Alan M. Rugman and Richard M. Hodgetts – International Business by Pearson Education.

GROUP B
SEMESTER V
DISCIPLINE SPECIFIC ELECTIVE
GOODS AND SERVICES TAX-I

L: T: P-4:1:0

Course Outcome:

On successful completion of this course the students can:

- CO1: Understand the technology and flow of return filing under GST
- CO2: Learn in details and gain knowledge to practice as GST Consultant
- CO3: Learn in details provisions of GST to handle TDS and POS online and off line more efficiently
- CO4: Understand in depth tax provisions to make managerial decisions effectively in various tax related matters
- CO5: Understand the provisions of integrated goods and service Tax Act, 2017
- CO6: Learn in depth the provisions relating to place of supply of goods imported into, or exported from India
- CO7: Identify the details of provisions in relation to Time of supply of goods

Unit 1: Introduction to GST

Indirect tax Structure in India, Issues in Indirect Tax, Rationale for Transition to GST; GST-Meaning, Definition of GST, Types of GST, Features of GST, Benefits of GST, Problems on Computation of GST.

Unit 2: Definitions

Actionable claim, Address of Delivery, Aggregate Turnover, Agriculturist, Associated enterprises, Business, GST Council, Credit note and Debit note, Deemed exports, Draw-back, Electronic Credit ledger, Exempt Supplies, Input, Input service, Input Service Distributor, Input Tax, Input tax Credit, Intra-state supply of Goods, Job work, Reverse Charge, Invoice, Composition Levy, Mixed Supply, outward supply, Person, Turnover in State

Unit 3: Levy and Collection of Tax

Scope of supply; Levy of GST; Liability of tax payable person, Rate and value of tax, meaning and conditions of supply, list of transactions without consideration list of neither a supply of goods, nor supply of services; meaning and treatment of mixed supply: meaning and treatment of composite supply: reverse charge mechanism: Composition levy.

Unit 4 : The Integrated Goods and Service Tax Act,2017

Short title, extent and commencement; Definitions; Central tax, Customs frontier of India, Export of Goods and Services, Import of goods and services; Location of Recipient of service, Location of supplier of service; Appointment of officers; Levy and Collection; Power to grant exemption from tax; Inter-State supply; Intra-State supply; Supplies in territorial waters

Unit 5: Place of supply of goods other than supply of goods imported into, or exported from India

Place of supply of goods imported into, or exported from India; Place of supply of services where location of supplier and recipient¹ is in India; Place of supply of services where location of supplier or location of recipient is outside India; Special provision for payment of tax by a supplier of online information and database access or retrieval services

Unit 6: Time of supply

Introduction, time of supply-forward charge, reverse charge, residuary, special charges-Time of supply of service- forward charge, reverse charge, Vouchers, Residuary, Special charges. Problems on determination of time of supply.

References:

1. Taxmann publications
2. Compendium on Goods and service tax-Dr. Manju S 3.www.cbec.gov.in/
4. www.ICSI.edu.in
5. www.icaai.org.
6. Students Guide To GST & Customs Law Vinod K Singhania

GROUP B
SEMESTER V
DISCIPLINE SPECIFIC ELECTIVE
FINANCIAL MANAGEMENT-I

L: T: P-4:1:0

Course Outcome:

On successful completion of this course the students can:

CO1: Identify the details of various sources of finance

CO2: Learn the characteristics of different methods of time value of money and its application to investment decision

CO3: Learn the classification and characteristics of cost of capital

CO4: Identify the characteristics of capital structure and factors affecting the capital structure

CO5: Learn the details of Capital Budgeting

CO6: Understand the classification and characteristics of Dividend theories

Unit 1: Introduction to financial management

Meaning – scope – goals of financial management – sources of finance.

Unit 2: Time value of Money

Present value and future value concepts-present value of annuity, application of present and future value to investment decisions, preparation of amortization table.

Unit 3: Cost of capital

Meaning- importance of cost of capital in financial decisions, determination of specific costs-cost of debt- cost of preference share capital-cost of equity, cost of retained earnings-weighted average cost of capital-Leverages- meaning- types problems on Leverages.

Unit 4: Capital Structure

Meaning-optimum capital structure-features of appropriate capital structure-factors influencing capital structure-theories of capital structure-NI approach- NOI approach, MM approach, traditional approach, determination of optimal debt-equity mix.

Unit 5: Capital budgeting

Meaning-features, role of capital budgeting, techniques of capital budgeting-payback period, average rate of return, net present value, profitability index, internal rate of return, discounted pay back method. Application of excel in capital budgeting techniques.

Unit 6: Dividend decisions

Meaning, Dividend decisions, Dividend policies - objectives of dividend policy- determinants of dividend policy-dividend relevance- Walter's model, Gordon model- Dividend irrelevance-MM hypothesis.

References:

1. Financial Management I.M.Pandey.
2. Financial Management Ravi Kishore
3. Financial Management Dr.V.R.Palanivelu
4. Financial Management Kulkarni
5. Financial Management Tulsian P C
6. Financial Management Khan and Jain

GROUP B
SEMESTER V
DISCIPLINE SPECIFIC ELECTIVE
ADVANCED COST AND MANAGEMENT ACCOUNTING-I

L: T: P-4:1:0

Course Outcome:

On successful completion of this course the students can:

- CO1: Learn in depth various Costing methods
- CO2: Understand the details of contract costing and process costing
- CO3: Identify reasons for reconciliation of cost and financial accounts
- CO4: Learn in depth the details of Activity based costing
- CO5: prepare the operating cost sheet

Unit 1: Introduction to Costing Methods

Meaning, Importance and Categories, Cost accounting Standards- Generally Accepted Cost Accounting Principles (GACAP)- Purpose, Objective and Applicability.

Unit 2: Contract costing

Introduction- Contract account, Profit on incomplete contracts, work in progress, Contractee's Accounts, Escalation clause.

Unit 3: Process costing: Introduction, Distinction between Job costing, and process costing, process losses, inter-process profits, Joint products and by-products- Meaning, features, differences, problems on process accounts including joint and by products.

Unit 4: Operating Costing

Introduction, transport costing, standing charges, operating/running charges, and preparations of operating cost sheet.

Unit 5: Reconciliation of cost and financial accounts

Meaning, need for reconciliation, reasons for disagreement, reconciliation procedure, problems on reconciliation.

Unit 6: Activity based costing (ABC)

Definition, Features, Advantages, Differences between ABC and traditional costing, Allocation of overheads; Objectives of ABC, Development of ABC, Implementation of ABC, Problems on Computation of Activity Based Costing and Traditional Costing;

References:

1. Cost Accounting: N.K. Prasad
2. Cost Accounting: Nigam & Sharma :
3. Practical Costing: Khanna, Pandey & Ahuja
- 4: Cost Accounting: M.L. Agarwal
5. Cost Accounting: Jain & Narang
6. Cost Accounting: S.P. Iyengar

ENE33001

GROUP B
SEMESTER V
DISCIPLINE SPECIFIC ELECTIVE
RETAIL MANAGEMENT

L: T: P-4:1:0**Course Outcome:**

On successful completion of this course the students can:

- CO1: Learn in depth the characteristics of retailing
- CO2: Understand in depth the details of retail consumer
- CO3: Identify and basis of retail market segmentation and strategies
- CO4: Specify the factors determining the retail location selection
- CO5: Understand in depth merchandise and managing them
- CO6: Learn in depth the details of retail operation and retail pricing

Unit 1: Introduction to Retailing

Concept of retailing, Functions of retailing, Terms and Definition, retail formats and types, Retailing Channels, Retail industry in India, Importance of retailing, Changing trends in retailing.

Unit 2: Understanding the Retail Consumer

Retail consumer behaviour, Factors influencing the Retail consumer, Customer decision making process, Types of decision making, Market research for understanding retail consume.

Unit 3: Retail Market Segmentation and Strategies

Market Segmentation and its benefits, kinds of markets, definition of Retail strategy, Strategy for effective market segmentation, Strategies for penetration of new markets, Growth strategies, Retail value chain.

Unit 4: Retail Location Selection

Importance of Retail locations, Types of retail locations, Factors determining the location decision, Steps involved in choosing a retail locations, Measurement of success of location.

Unit 5: Merchandise Management

Meaning of Merchandising, Factors influencing Merchandising, Functions of Merchandising Manager, Merchandise planning, Merchandise buying, Analyzing Merchandise performance.

Unit 6: Retail Operations and Retail Pricing

Store administration, Premises management, Inventory Management, Store Management, receipt Management, Customer service, Retail Pricing, Factors influencing retail prices pricing strategies, controlling costs.

References:

1. Retail Management 01 Edition S. C. Bhatia
2. Retail Management: Text and Cases (English, Paperback, Swapna Pradhan
3. Retail Management; Principles and Practices – R. Sudarshan
4. Retail Management PB (English, Paperback, Mathur U C)

ENF28001

GROUP C
SEMESTER VI
DISCIPLINE SPECIFIC ELECTIVE
INVESTMENT ANALYSIS AND PORTFOLIO MANAGEMENT

L: T: P-4:1:0**Course Outcome:**

On successful completion of this course the students can:

C01: Learn in depth the characteristics of investment

C02: Understand in depth characteristics of financial system and stock market

C03: Identify and use different models for portfolio analysis

C04: Learn in depth capital asset pricing model

C05: Identify the risk and return of investment

C06: Understand in depth portfolio selection

C07: Understand in depth the efficient market hypothesis and portfolio analysis

Unit 1: Introduction to Investment

Meaning, Financial and Economic Meaning, Characteristics, Objectives, Investment v/s Speculation, Investment v/s Gambling, Types of Investors and Investment avenues.

Unit 2: Indian Financial System and Stock Market-Investment Planning,

Introduction to Stock Market, Overview of Indian Financial System, Market Indices, Methods of Computation of Market Indices.

Unit 3: Portfolio Management

Meaning, Phases, Evolution, Role of Portfolio Management, Calculation of Risk and Return, Fundamental Analysis: Economic Analysis, Industry analysis and Company analysis and Technical Analysis: Meaning, Dow Theory, basic Principles, Trends and charts.

Unit 4: Share and Bond Valuation-Share

Concepts and present value, share valuation model, constant Growth Model, Multiple Growth Model, Discount rate, Multiplier Approach to share values and Regression Analysis. Bonds: Bond Returns, Prices, Pricing Theories, Bond Risks and Bond Duration.

Unit 5: Efficient Market Hypothesis and Portfolio Analysis

EMH: Random Walk Theory, the Efficient Market Hypothesis, Forms of Market Efficiency. Portfolio Analysis: Expected Return and Risk of Portfolio, Reduction of Portfolio Risk through Diversification, Portfolio with more than two Securities – solved examples.

Unit 6: Portfolio Selection

Feasible set of Portfolios, Selection of Optimal Portfolio, Limitations of Markowitz Model and Single Index Model and Multi Index Model. Capital Asset Pricing Model and Arbitrage Pricing Theory.

References:

1. Investment analysis and Portfolio Management Prasanna Chandra
2. Security Analysis and Portfolio Management Punithavathi pandian
3. Security Analysis and portfolio Management Ambika Prasad Dash
4. Security Analysis and Portfolio Management, M. Ranganatham, R. Madhumathi

ENF29001

GROUP C
SEMESTER VI
DISCIPLINE SPECIFIC ELECTIVE
FINANCIAL DERIVATIVES

L: T: P-4:1:0**Course Outcome:**

On successful completion of this course the students can:

C01: Learn in depth the characteristics of financial derivatives and derivatives market

C02: Deliberate the characteristics of futures and forwards

C03: Understand in details with examples options

C04: Deliberate the classification and characteristics of financial swaps

C05: Write down the details of commodity market

C06: Identify in depth credit derivatives

Unit 1: Financial Derivatives

Introduction, -meaning- Types of financial derivatives - Features of derivatives market - Factors contributing to the growth of derivatives - functions of derivative markets - traders in derivatives markets - Derivatives market in India

Unit 2: Futures and forwards

Meaning, differences-valuation of futures, Mechanics of buying & selling futures, Margins, Hedging using futures -specification of futures - Commodity futures, Index futures, interest rate futures – arbitrage opportunities.

Unit 3: Options:

Types of options, option pricing, factors affecting option pricing – call and put options on dividend and non-dividend paying stocks put-call parity - mechanics of options - stock options - options on stock index - options on futures – interest rate options. Concept of exotic option. Hedging & Trading strategies involving options, valuation of option: basic model, one step binomial model, Black and Scholes Model,

Unit 4: Financial Swaps

features and uses of swaps - Mechanics of interest rate swaps – valuation of interest rate swaps – currency swaps – valuation of currency swaps

Unit 5: Commodity derivatives

Commodity futures market-exchanges for commodity futures in India, Forward Market Commissions and regulation-commodities traded – trading and settlements – physical delivery of commodities

Unit6: Credit Derivatives

Meaning, common credit derivatives, types of credit derivatives, Credit Default swaps(CDS), Total Return swaps, Collateralized debt obligations(CDO), Indian scenario.

References:

1. Financial Derivatives- Vohra and Bagri
2. Capital Market Instruments Kotreshwar G
3. Fundamental Financial Derivatives N R Parasuraman
4. Financial Derivatives S L Guptha

ENF30001

GROUP C
SEMESTER VI
DISCIPLINE SPECIFIC ELECTIVE
GOODS AND SERVICES TAX AND CUSTOM DUTY-II

L: T: P-4:1:0**Course Outcome:**

On successful completion of this course the students can:

CO1: Understand the technology and flow of return filing under GST

CO2: Learn in details and gain knowledge to practice as GST Consultant

CO3: Learn in details provisions of GST in relation to value of taxable supply and input tax credit

CO4: Understand in depth tax provisions to make managerial decisions effectively in various tax related matters

CO5: Learn in detail the procedure to be followed to assess the value and determine customs duty

CO6: Understand the procedure of registration under GST

CO7: Prepare tax invoice, credit and debit notes

Unit 1: Value of taxable supply

Conditions, inclusions, Consideration not wholly in money, Supply between two related persons, Supply through agent, cost based value, Residual valuation, specific supplies, Service of pure agent. Problems on determination of value of supply.

Unit 2: Input tax credit

Meaning, conditions for taking credit, ineligible input tax credit, availability of credit in special circumstances, Input tax credit and change in constitution of registered person, Taking input tax credit in respect of inputs and capital goods sent for job work, Manner of Distribution of Credit by Input Service Distributor (ISD)

Unit 3: Tax Invoice, Credit and Debit Notes

Tax invoice; Prohibition of un authorised collection of tax; Amount of tax to be indicated in tax invoice and other documents ; Credit and debit notes.

Unit 4: Registration under GST

Persons liable for registration, compulsory registration, Procedure for Registration, Rejection of application for registration, cancellation of Registration

Unit 5: Returns

Brief introduction to various GSTRS-procedure for filing various returns.

Unit 6: Customs Act 1962

Meaning- Notified goods –specified goods- Prohibition of importation and exportation under sec 11- types of customs duty- Basic customs duty, Education Cess, Anti dumping duty, Safeguard Duty, IGST, GST Compensation Cess- Computation of Assessable value and applicable duties. Exports – Meaning- zero rated supply.

References:

1. Taxmann publications
2. Compendium on Goods and service tax-Dr. Manju S
3. www.cbec.gov.in/
4. Systematic Approach GST- Dr. Ravi. Gupta, Dr. Girish. Ahuja

ENF31001

GROUP C

SEMESTER VI

DISCIPLINE SPECIFIC ELECTIVE

FINANCIAL MANAGEMENT –II

L: T: P-4:1:0

Course Outcome:

On successful completion of this course the students can:

CO1: Deliberate the details of working capital management

CO2: Learn in depth the details of cash management

CO3: Understand the details of working capital financing

CO4: Deliberate in details with examples Venture capital financing

CO5: Learn in depth the details of shareholders value creation

CO6: Deliberate in depth International financial management

Unit 1: Working Capital Management

Meaning, Features, types of working capital, factors influencing working capital, level of current assets, operating cycle and cash cycle, current assets financing policy

Unit 2: Cash Management

Cash budget cash collection and disbursement, options for investment of surplus funds, credit management- credit policy variables-credit evaluation. Inventory management- need for inventories; order quantity-EOQ model- monitoring and control of inventories-ABC- JIT techniques.

Unit 3: Working Capital Financing

Leasing-types of leases, Rationale for leasing, operating leases, leasing as a financing decision; hire purchase financing- Hire purchase financing v/s lease financing, instalment sale, evaluation of Hire purchase financing

Unit 4: Venture Capital Financing

Meaning, features, development of venture capital in India, stages in venture financing- the business plan- essentials of a business plan, the process of venture capital financing- Methods of venture financing; Disinvestment mechanisms

Unit 5: Share Holder Value Creation

Financial goals and strategy, shareholder value creation- market value added, Market to book value, Economic value added(EVA)- Balanced scorecard- the learning and growth perspective, significance of balanced score card , implementation of score card.

Unit 6: International Financial Management

Foreign exchange market, foreign exchange rates- spot exchange rates, bid-ask rate, forward exchange rates- foreign exchange risk- transaction exposure, economic exposure, translation exposure, hedging of foreign exchange risk- foreign currency option, money market operations- financing international operations.

References:

1. Financial Management I.M.Pandey.
2. Financial Management Ravi Kishore
3. Financial Management Dr.V.R.Palanivelu
4. Financial Management Kulkarni

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GROUP C
SEMESTER VI
DISCIPLINE SPECIFIC ELECTIVE
ADVANCED COST AND MANAGEMENT ACCOUNTING-II

L: T: P-4:1:0

Course Outcome:

On successful completion of this course the students can:

C01: Understand the details of management accounting

C02: Learn in depth the details of financial statement analysis techniques

C03: Analyze the inflow and outflow of cash and able to prepare cashflow statement

C04: Understand the characteristics of different types of ratios

C05: Learn in depth budget and budgetary control and prepare various budget for different activities of the business

C06: Learn in depth marginal costing and able to effective decision regarding make or buy, accepting foreign order, continuance or discontinuance of manufacturing a products

Unit 1: Introduction

Meaning and Definition of Management Accounting, Scope and Objectives of Management Accounting-Differences between Management Accounting and Financial Accounting –Management accounting and Cost accounting-Limitations of Management Accounting.

Unit 2: Analysis of Financial Statements

Common Size Statements, Comparative Statement, Trend analysis.

Unit 3: Ratio Analysis

Meaning and Objectives-Types of rations - (A) Profitability Ratios-GP ratio - NP Ratio- Operating ratio- Operating profit ration-Return on capital employed ratio- EPS, (B) Turnover Ratios-working capital turnover ratio- Stock Turnover ratio-Fixed assets turnover ratio-Debtors turnover Ratio-Creditors turnover Ratio, (C) Financial ratios – Current Ratio-liquidity ratio-Debt-equity ratio-Proprietary Ratio-Capital gearing Ration-Advantagesand Limitations of Ratios- Construction of Balance sheet using ratios.

Unit 4: Cost Flow Statement

Meaning, Definition, Uses and Limitations-Differences between funds flow statement and cash flow statement-Preparation of Cash flow statement (AS-7): Direct method and Indirect Method.

Unit 5: Marginal Costing

Definition - Basic concepts – Assumptions - Marginal Cost statement – Contribution-Break Even Analysis-P/V Ratio-Margin of Safety - Decision areas - Make or Buy and Pricing.

Unit 6: Budget and Budgeting Control

Definition - Basic Concepts - Budget Manual - Key factor - Classification of Budgets - Problems on cash budget, sales budget, Flexible Budget, Cash Planning and Motives for holding cash.

References:

1. Cost Accounting: N.K. Prasad
2. Cost Accounting: Nigam & Sharma :
3. Practical Costing: Khanna, Pandey & Ahuja
4. Cost Accounting: M.L. Agarwal
5. Cost Accounting: Jain & Narang
6. Cost Accounting: S.P. Iyengar

ENF33001

GROUP C

VI SEMESTER

DISCIPLINE SPECIFIC ELECTIVE

ORGANISATIONAL BEHAVIOUR

L: T: P-4:1:0

Course Outcome:

On successful completion of this course the students can:

C01: Understand in detail behaviour of employees and able to manage them efficiently

C02: Identify in details employees performance and able to motivate for effective performance

C03: Learn in depth and analyse the behaviour of employees

C04: Understand in details key positions in an organisation and able to occupy them

C05: Learn in details with examples frame policies and strategies in organisation

Unit 1: Introduction

Meaning, definition, historical development, fundamental principles of OB, contributing disciplines, approaches, challenges and opportunities.

Unit 2: Foundations of Individual Behaviour

Individual behaviour: Foundations of individual behaviour. Ability: Intellectual abilities, Physical ability, the role of disabilities. Personality: Meaning, formation, determinants, traits of personality, personality attributes influencing OB. Attitude: Formation, components of attitudes, relation between attitude and behaviour.

Unit 3: Perception and Emotions- Perception: Process of perception, factors influencing perception, link between perception and individual decision making. Emotions: Affect, mood and emotion and their significance, basic emotions, emotional intelligence, self-awareness, self-management, social awareness, relationship management.

Unit 4: Motivation and Leadership

Motivation- meaning, theories of motivation-needs theory, two factor theory, Theory X and Y, application of motivational theories. Leadership: Meaning, styles of leadership, leadership theories, trait theory, behavioural theories, managerial grid, situational theories-Fiedler's model, SLT, transactional and transformation leadership.

Unit 5: Group Behaviour

Definition, types, formation of groups, building effective teams. Conflict: Meaning, nature, types, process of conflict, conflict resolution. Power and politics: Basis of power, effectiveness of power tactics.

Unit 6: Emerging Challenges

Emerging challenges, managing diversity, globalisation, technology transformation, e - business, promoting ethical behaviour.

References:

- 1.Organisational behaviour, Stephen P Robbins, Timothy A. Judge, Neharika Vohra, 14th Edition, Pearson
2. Organization Behaviour – Ashwathappa, Himalaya Publication House

3. Organisational Behaviour: A modern approach - Arun Kumar and Meenakshi, Vikas Publishing House
4. Organisational Behaviour - Fred Luthans, 12/e, McGraw Hill International
- 5 Management and Organisational Behaviour - Laurie J Mullins, Pearson education
6. Fundamentals of Organisational Behaviour - Slocum/Hillriegel, Cengage Learning

B.COM

Question Paper Pattern

(For all courses except Quantitative Techniques, Computerised Accounting and E-filing of returns)

PART-A

Answer the following. Each question carries 15 marks.

2X15=30

1.

OR

2.

3.

OR

4.

PART-B

Answer the following. Each question carries 10 marks.

2X10=20

5.

OR

1.

7.

OR

8.

PART-C

Answer any four of the following. Each question carries 5 marks.

4X5=20

9.

10.

11.

12.

13.

14.

B.COM

**Question Paper Pattern
4.5 Quantitative Techniques**

Time: 3 hrs

Max. Marks: 70

Part-A

Answer the following. Each question carries two marks

10X2=20

- | | |
|------------|---------|
| 1. a. | f. |
| b. | g. |
| c. | h. |
| d. | i. |
| e. | j. |

Part-B

Answer any four of the following. Each question carries five marks 4X5=20

- 2.
- 3.
- 4.
- 5.
- 6.

Part-C

Answer any three of the following. Each question carries ten marks

3X10=30

- 7.
- 8.
- 9.
- 10.

B.COM

Question Paper Pattern

**Computerized Accounting System (SEC-1) / E-filing of returns
(SEC-2)**

Part-A

Answer the following. Each question carries two marks.

10X2=20

- 1. a.
- b.
- c.
- d.
- e.
- f.
- g.
- h.
- i.
- j.

Part-B

Answer any four of the following. Each question carries five marks.

4X5=20

- 2.
- 3.
- 4.
- 5.
- 6.

Part-C

Answer any three of the following. Each question carries ten marks

3X10=30

- 5.
- 6.
- 7.
- 8.



**JSS COLLEGE OF ARTS, COMMERCE AND
SCIENCE**

(Autonomous)

B N ROAD, MYSURU- 570 025

**DEPARTMENT OF COMPUTER
SCIENCE**

Syllabus

CHOICE BASED CREDIT SYSTEM

For B.Sc PROGRAMME

Physics, Mathematics, Computer Science

2017-18

PROGRAMME: BSc PMCs, PROGRAMME CODE: BSc-02 (2017-18)

Year	Sem	Corse Code	Title	Hours / Week		Credits	Maximum Marks			Exam Duration	Total Marks
				L	T/ P		IA		Exam		
				L:T:P	C1	C2					
I Year	I	CMA 25002	DSC 3A - Computer Science – I								
			Problem Solving Using Computer	4	4	4:0:2	15	15	70	3 Hours	100
	II	CMB 25002	DSC 3B - Computer Science – II								
			Computer System Organization and Architecture	4	4	4:0:2	15	15	70	3 Hours	100
II Year	III	CMC 25002	DSC 3C - Computer Science – III								
			Data Structures and File Processing	4	4	4:0:2	15	15	70	3 Hours	100
	IV	CMD 25002	DSC 3D - Computer Science – IV								
			Operating Systems	4	4	4:0:2	15	15	70	3 Hours	100
III Year	V	DSE 3 - Computer Science – V (Choose Any One)									
		CME 25002	A - Database Management Systems	4	0	4:0:1	15	15	70	3 Hours	100
		CME2 5202	B - Computer Networks	4	0	4:0:1	15	15	70	3 Hours	100
		CME 25402	C - Software Engineering	4	0	4:0:1	15	15	70	3 Hours	100
		SEC 3# - Computer Application(Choose Any One)									
		CME 38002	A - Office Automation	0	2	0:0:1	15	15	70	2 Hours	100
		CME 38202	B - XML Programming	0	2	0:0:1	15	15	70	2 Hours	100
		CME 38402	C - R Programming	0	2	0:0:1	15	15	70	2 Hours	100

introducing this paper is only after Approvals of AC / GB

UG-Computer Science - CBCS Scheme

Year	Sem	Corse Code	Title	Hours / Week		Credits	Maximum Marks			Exam Duration	Total Marks
				L	T/ P		IA		Exam		
				L:T:P	C1	C2					
III Year	VI		DSE 6 - Computer Science –VI(Choose Any One)								
		CMF 25002	A - Internet Technology	4	0	4:0:1	15	15	70	3 Hours	100
		CMF 25202	B - Mobile Applications	4	0	4:0:1	15	15	70	3 Hours	100
		CMF 25402	C - Cloud Computing	4	0	4:0:1	15	15	70	3 Hours	100
			SEC 4 - Computer Science –VII(Choose Any One)								
		CMF 38002	A - Android Programming	1	2	1:0:1	15	15	70	2 Hours	100
		CMF 38202	B - PHP Programming	1	2	1:0:1	15	15	70	2 Hours	100
		CMF 38602	C - System Administration & Maintenance	1	2	1:0:1	15	15	70	2 Hours	100

Assessment Maximum marks - 100

Course type	C1		C2		C3 Exam Marks		Assigned Marks (Percentage)			Total
	Theory	Lab	Theory	Lab	Theory	Lab	Theory	Lab	IA	
DSC	10	05	10	05	70	70	50	20	30	100
DSE	10	05	10	05	70	70	50	20	30	100
DSE (non practical)	15	-	15	-	70	-	70	--	30	100
SEC	15	-	15	-	50	-	70	--	30	100

Note:

1. C1 will be conducted for 20 Marks (Theory) with onehour duration - 10 Marks (Lab) with continuous assessment and it will be reduced to assigned marks.
2. C2 will be conducted for 20 Marks (Theory) with onehour duration - 10 Marks (Lab) with continuous assessment and it will be reduced to assigned marks.
3. C3 will be conducted for 70 Marks (Theory) with three hours duration - 70 Marks (Lab) with 3 hours duration and to be reduced to assigned marks.
4. For non-practical course C3 will be conducted for 70 Marks (Theory) with three hours duration.
5. In case of SEC, C1 and C2 will be conducted for 15 Marks each with one hour duration and C3 will be conducted for 50 Marks with 2 hours duration.

Programme Outcome for Bachelor of Science in Physics, Mathematics, Computer Science:

After completing the graduation in the Bachelor of Science the students are able to:

- PO1. Demonstrate proficiency in Mathematics and the Mathematical concepts needed for a proper understanding of Physics.
- PO2. Demonstrate the ability to justify and explain their thinking and/or approach.
- PO3. Develop state-of-the-art laboratory and professional communication skills.
- PO4. Apply the scientific method to design, execute, and analyze an experiment
- PO5. Explain scientific procedures and experimental observations.
- PO6. Understand the value of Mathematical proof
- PO7. Demonstrate proficiency in writing and understanding proofs.
- PO8. Apply mathematical problems and solutions in aspects of science and technology.
- PO9. Gain experience to investigate the real world problems
- PO10. Apply mathematical ideas and models to those problems.
- PO11. Apply Mathematical principles for computing and logical design.
- PO12. Design, implement, and evaluate a computational system to meet desired needs within realistic constraints.
- PO13. Use the System principles in the design and development of software for systems of varying complexity.

Programme Specific Outcome

Bachelor of Science in Physics, Mathematics, Computer Science

After completing the graduation in the Bachelor of Science the students are able to:

PSO1. Find career opportunities

PSO2. Develop competence to write competitive examinations.

PSO3. Develop proficiency in the analysis of complex physical problems

PSO4. Use of mathematical or other appropriate techniques to solve problems

PSO5. Create a hypothesis and appreciate how it relates to broader theories.

PSO6. Demonstrate skills in the use of Computers

PSO7. Join as Entry level Technical job role for an IT Industry

PSO8. Build small database ERP software/ web applications.

SEMESTER I

Course code: CMA25002

DSC 3A: Computer Science-I

Problem Solving Using Computer

Credits: Theory – 04, Practical – 02

Theories: 60 Lectures

COURSE OUTCOME:

After successful completion of the course, the student is able to

- CO1. Deliberate in depth Computer Fundamentals
- CO2. Understand in details with features Computer Organization
- CO3. Learn the details of Problem Solving Techniques
- CO4. Deliberate the characteristics of Programming Languages
- CO5. Understand in details with examples - Python Programming Languages
- CO6. Specify in depth OOPs, Event Driven and GUI features in Python

Unit - 1

(15Lectures)

Computer Fundamentals: Introduction to Computers: Characteristics of Computers, Uses of computers, Types and generations of Computers.

Basic Computer Organization: Units of a computer, CPU, ALU, memory hierarchy, registers, I/O devices.

Techniques of Problem Solving: Concept of problem solving, Problem definition, Program design, Debugging, Types of errors in programming, Documentation, Flowcharting, decision table, algorithms, Structured programming concepts, Programming methodologies viz. top-down and bottom-up programming.

Unit - 2

(15 Lectures)

Overview of Programming: Structure of a Python Program, Elements of Python

Introduction to Python: Python Interpreter, Using Python as calculator, Python shell, Indentation. Atoms, Identifiers and keywords, Literals, Strings, Operators (Arithmetic operator, Relational operator, Logical or Boolean operator, Assignment, Operator, Ternary operator, Bit wise operator, Increment or

Decrement operator).Input and Output Statements, Control statements (Looping-while Loop, for Loop, Loop Control)

Unit - 3 **(15 Lectures)**

Creating Python Programs: Conditional Statement- if...else, Difference between break, continue and pass.

Structures: Numbers, Strings, Lists, Tuples, Dictionary, Date & Time, Modules, Defining Functions, Exit function, default arguments.

Unit - 4 **(15 Lectures)**

Introduction to Advanced Python: Objects and Classes, Inheritance, Regular Expressions, Event Driven Programming, GUI Programming.

Reference:

1. P. K. Sinha&PritiSinha , “Computer Fundamentals”, BPB Publications, 2007.
2. Dr. Anita Goel, Computer Fundamentals, Pearson Education, 2010.
3. T. Budd, Exploring Python, TMH, 1st Ed, 2011
4. Python Tutorial/Documentation www.python.org 2010
5. Allen Downey, Jeffrey Elkner, Chris Meyers , How to think like a computer scientist : learning with Python , Freely available online.2012

Problem Solving Using ComputerLab

Software Lab using Python

Section: A (Simple programs)

1. Write a menu driven program to convert the given temperature from Fahrenheit to Celsius and vice versa depending upon user’s choice.
2. WAP to calculate total marks, percentage and grade of a student. Marks obtained in each of the three subjects are to be input by the user. Assign grades according to the following criteria
 - Grade A: Percentage ≥ 80
 - Grade B: Percentage ≥ 70 and < 80
 - Grade C: Percentage ≥ 60 and < 70
 - Grade D: Percentage ≥ 40 and < 60
 - Grade E: Percentage < 40

3. Write a menu-driven program, using user-defined functions to find the area of rectangle, square, circle and triangle by accepting suitable input parameters from user.
4. WAP to display the first n terms of Fibonacci series.
5. WAP to find factorial of the given number
6. WAP to find sum of the following series for n terms: $1 - 2/2! + 3/3! - \dots - n/n!$
7. WAP to calculate the sum and product of two compatible matrices.

Section: B (Visual Python)

All the programs should be written using user defined functions, wherever possible.

1. Write a menu-driven program to create mathematical 3D objects
 - I. Curve
 - II. Sphere
 - III. Cone
 - IV. Arrow
 - V. Ring
 - VI. Cylinder.
2. WAP to read n integers and display them as a histogram.
3. WAP to display sine, cosine, polynomial and exponential curves.
4. WAP to plot a graph of people with pulse rate p vs. height h. The values of p and h are to be entered by the user.
5. WAP to calculate the mass m in a chemical reaction. The mass m (in gms) disintegrates according to the formula $m=60/(t+2)$, where t is the time in hours. Sketch a graph for t vs. m, where $t \geq 0$.
6. A population of 1000 bacteria is introduced into a nutrient medium. The population p grows as follows:
$$P(t) = (15000(1+t))/(15 + e)$$
Where the time t is measured in hours. WAP to determine the size of the population at given time t and plot a graph for P vs t for the specified time interval.
7. Input initial velocity and acceleration, and plot the following graphs depicting equations of motion:
 - I. Velocity wrt time ($v=u+at$)
 - II. Distance wrt time ($s=u*t+0.5*a*t*t$)
 - III. Distance wrt velocity ($s=(v^2-u^2)/2*a$)

SEMESTER II

Course code: CMB25002

DSC 3B: Computer Science-II

Computer System Organization and Architecture

Credits: Theory – 04, Practical – 02

Theories: 60 Lectures

Course Outcome:

After successful completion of the course, the student is able to

CO1. Deliberate in details with examples Boolean algebra and logic circuits

CO2. Learn the details of Data Representation and Computer Arithmetic

CO3. Learn in depth Computer Organization and Design

CO4. Learn the details of architecture of CPU

CO5. Deliberate the classification and characteristics of Basic Computer Programming Concepts

CO6. Write down in depth Basic Computer Programming Concepts

CO7. Learn the classification and characteristics of Input -Output organization

Unit - 1

(15 Lectures)

Introduction: Logic gates, Boolean algebra, combinational circuits, circuit simplification, flip-flops and sequential circuits, decoders, multiplexors, registers, counters and memory units.

Data Representation and basic Computer Arithmetic: Number systems, complements.

Unit - 2

(15 Lectures)

Fixed and floating point representation, character representation, addition, subtraction, magnitude comparison.

Basic Computer Organization and Design: Computer registers, bus system, instruction set.

Unit - 3

(15 Lectures)

timing and control, instruction cycle, memory reference, input-output and interrupt.

Central Processing Unit: Register organization, arithmetic and logical micro-operations, stack organization, micro programmed control.

Unit - 4

(15 Lectures)

Programming the Basic Computer: Instruction formats, addressing modes, instruction codes, machine language, assembly language, input output programming.

Input-output Organization: Peripheral devices, I/O interface, Modes of data transfer, direct memory access.

Reference:

1. M. Mano, Computer System Architecture, Pearson Education 1992.
2. A. J. Dos Reis, Assembly Language and Computer Architecture using C:: and JAVA, Course Technology, 2004
3. W. Stallings, Computer Organization and Architecture Designing for Performance, 8th Edition, Prentice Hall of India ,2009
4. Digital Design, M.M. Mano, Pearson Education Asia, 1979

Computer System Organization and Architecture Lab

1. Verification of Basic gates(AND,OR,NOT)
2. Verification of Universal gates(NAND,NOR,EX-OR)
3. Verification of NAND gate as a Universal gate
4. Verification of NOR gate as a Universal gate
5. Verification of Demorgan's theorem
6. Verification of Half adder & Full Adder
7. Verification of Half subtractor & Full Subtractor
8. Verification of Half adder & Half subtractor using NAND gate
9. Conversion of Binary to Gray & gray to Binary Code
10. Simplification of Boolean Expressions
11. Simplification of Boolean Expressions using K-Map
12. Flip-Flops:SR FF(clock, without clock)
13. JK FF
14. Toggle FF
15. Delay FF
16. Multiplexer
17. De-multiplexer
18. Simulate the machine for the following memory-reference instructions with I= 0 and address part = 082. The instruction to be stored at address 022 in RAM. Initialize the memory word at address 082 with the operand B8F2 and AC with A937. Determine the contents of AC, DR, PC, AR and IR in hexadecimal after the execution.
 - a. ADD f. BSA
 - b. AND g. ISZ
 - c. LDA
 - d. STA
 - e. BUN
19. Simulate the machine for the memory-reference instructions referred in above question with I= 1 and address part = 082. The instruction to be stored at address 026 in RAM. Initialize the memory word at address 082 with the value 298. Initialize the memory word at address 298 with operand B8F2 and AC with A937. Determine the contents of AC, DR, PC, AR and IR in hexadecimal after the execution.
20. Modify the machine created in Practical 1 according to the following instruction format:

Instruction format

0	2	3	4	15
Opcode	I		Address	

- a. The instruction format contains a 3-bit opcode, a 1-bit addressing mode and a 12-bit Address. There are only two addressing modes, I = 0 (direct addressing) and I = 1(indirect addressing).
- b. Create a new register I of 1 bit.

C. Create two new microinstructions as follows:

- i. Check the opcode of instruction to determine type of instruction (Memory Reference/Register Reference/Input-Output) and then jump accordingly.
- ii. Check the I bit to determine the addressing mode and then jump accordingly.

SEMESTER III

Course code: CMC25002

DSC 3C:Computer Science-III

Data Structures and File Processing

Credits: Theory – 04, Practical – 02

Theories: 60 Lectures

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Learn the classification, characteristics and understanding of Data structures
- CO2. Specify the details of Searching Techniques
- CO3. Deliberate in details with examples Basic Concepts of Memory Management Techniques
- CO4. Understand in depth File System Operations
- CO5. Specify the characteristics of File Organization Methods
- CO6. Deliberate in details with examples of Storage Devices

Unit - 1

(15 Lectures)

Basic Data Structures:Introduction, Abstract data structures- stacks, queues, linked lists and binary trees. **Sets:**Dictionary implementation, use of priority queues, hashing, binary trees, balanced trees, setswith merge-find operations.

Unit - 2

(15 Lectures)

Searching: Internal and external searching, use of hashing and balancing techniques.

Memory Management: Garbage collection algorithms for equal sized blocks, storage allocationfor objects with mixed size, buddy systems.

Unit - 3

(15 Lectures)

Physical Devices: Characteristics of storage devices such as disks and tapes, I/O buffering.

Basic File System Operations: Create, open, close, extend, delete, read-block, write-block, protection mechanisms.

Unit - 4

(15 Lectures)

File Organizations: Sequential, indexed sequential, direct, inverted, multi-list, directory systems, Indexing using B-tree, B+ tree and their variants, hashing – hash function, collision handling methods, extendible hashing.

Reference:

1. M.T. Goodrich, R. Tamassia and D. Mount, Data Structures and Algorithms in C++, John Wiley and Sons, Inc., 2004.
2. T.H. Cormen, C.E. Leiserson, R.L. Rivest and C. Stein, Introduction to Algorithms, 2nd Ed., Prentice-Hall of India, 2006.
3. Robert L. Kruse and A.J. Ryba, Data Structures and Program Design in C++, Prentice Hall, Inc., NJ, 1998.
4. B. Stroustrup, The C++ Programming Language, Addison Wesley, 2004
5. D.E. Knuth, Fundamental Algorithms (Vol. I), Addison Wesley, 1997

Data Structures and File Processing Lab

Part - A

1. Program to find lower triangular and upper triangular matrices for the given matrix.
2. Write an interactive program to insert an element at the given position and delete an element at the specified position in the given array.
3. Program to search an element identify the number of occurrences with locations in linear array.
4. Program to sort the given $M \times N$ matrix row-wise and column-wise using bubble sorting technique.
5. Write an interactive program to search an element in the given linear array using linear and binary searching technique.
6. Write a program to Merge two sorted arrays.

Part – B

7. Write an interactive program to implement the following operations on stack using arrays
 - a.PUSH
 - b.POP
8. Program to implement Tower of Hanoi problem.
9. Write an interactive program to perform insertion and deletion operations in Linear Queue using arrays.
10. Write an interactive program to perform insertion and deletion operations in Circular Queue using arrays.
11. Write an interactive program to insert a node in a linked list at the front, delete a node from the rear and display.
12. Write an interactive program to implement preorder, post order and in order traversal of a binary tree using linked list.

SEMESTER IV

Course code: CMD25002

DSC 3D - Computer Science – IV

Operating Systems

Credits: Theory – 04, Practical – 02

Theories: 60 Lectures

Course Outcome:

After successful completion of the course, the student is able to

CO1. Learn in details with examples system software

CO2. Learn the details of Operating System organization

CO3. Understand the classification and characteristics of Process Management and Scheduling mechanisms

CO4. Understand in depth Memory Management and allocation strategies

CO5. Learn in details with examples basic concepts of shell scripting

CO6. Understand in depth basic Linux environment

Unit - 1

(15 Lectures)

Introduction: System Software, Resource Abstraction, OS strategies. Types of operating systems - Multiprogramming, Batch, Time Sharing, Single user and Multiuser, Process Control & Real Time Systems

Operating System Organization: Factors in operating system design, basic OS functions, implementation consideration; process modes, methods of requesting system services – system calls and system programs.

Unit - 2

(15 Lectures)

Process Management: System view of the process and resources, initiating the OS, process address space, process abstraction, resource abstraction, process hierarchy, Thread model

Unit - 3

(15 Lectures)

Scheduling: Scheduling Mechanisms, Strategy selection, non-pre-emptive and pre-emptive strategies.

Memory Management: Mapping address space to memory space.

Unit - 4

(15 Lectures)

Memory allocation strategies, fixed partition, variable partition, paging, virtual memory

Shell introduction and Shell Scripting: What is shell and various type of shell, Various editors present in Linux, Different modes of operation in vi editor, What is shell script, Writing and executing the shell script, Shell variable (user defined and system variables), System calls, Using system calls, Pipes and Filters, Decision making in Shell Scripts (If else, switch), Loops in shell, Functions, Utility programs (cut, paste, join, tr, uniq utilities), Pattern matching utility (grep)

Reference:

1. A Silberschatz, P.B. Galvin, G. Gagne, Operating Systems Concepts, 8th Edition, John Wiley Publications 2008.
2. A.S. Tanenbaum, Modern Operating Systems, 3rd Edition, Pearson Education 2007.
3. G. Nutt, Operating Systems: A Modern Perspective, 2nd Edition Pearson Education 1997.
4. W. Stallings, Operating Systems, Internals & Design Principles , 5th Edition, Prentice Hall of India. 2008.
5. M. Milenkovic, Operating Systems- Concepts and design, Tata McGraw Hill 1992.

Operating Systems Lab

Software Lab based on Operating Systems

Note: Following exercises can be performed using Linux or Unix

1. Usage of following commands: ls, pwd, tty, cat, who, who am I, rm, mkdir, rmdir, touch, cd.
2. Usage of following commands: cal, cat(append), cat(concatenate), mv, cp, man, date.
3. Usage of following commands: chmod, grep, tput (clear, highlight), bc.
4. Write a shell script to check if the number entered at the command line is prime or not.
5. Write a shell script to modify “cal” command to display calendars of the specified months.

6. Write a shell script to modify “cal” command to display calendars of the specified range of months.
7. Write a shell script to accept a login name. If not a valid login name display message – “Entered login name is invalid”.
8. Write a shell script to display date in the mm/dd/yy format.
9. Write a shell script to display on the screen sorted output of “who” command along with the total number of users .
10. Write a shell script to display the multiplication table any number,
11. Write a shell script to compare two files and if found equal asks the user to delete the duplicate file.
12. Write a shell script to find the sum of digits of a given number.
13. Write a shell script to merge the contents of three files, sort the contents and then display them page by page.
14. Write a shell script to find the LCD(least common divisor) of two numbers.
15. Write a shell script to perform the tasks of basic calculator.
16. Write a shell script to find the power of a given number.
17. Write a shell script to find the factorial of a given number.
18. Write a shell script to check whether the number is Armstrong or not.
19. Write a shell script to check whether the file have all the permissions or not.
20. Program to show the pyramid of special character “*”.

SEMESTER V

Course code: CME25002

DSE 3A: Elective: Computer Science – V

Database Management Systems

Credits: Theory – 04, Practical – 01

Theory: 60 Lectures

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Understand the characteristics of DBMS with examples
- CO2. Deliberate the details of types of database languages with examples
- CO3. Learn the details of ER- Diagrams and Relationship
- CO4. Understand in depth Basic concepts of Relational Model
- CO5. Learn in details with examples MYSQL Commands
- CO6. Learn in details with examples in PL-SQL

Unit - 1

(15 Lectures)

Introduction to Database Management Systems: Definition of Data, Information, DBMS, Data base system application, Purpose of database systems, Characteristics of DB – Self describing nature, Insulation between programs, data and data Abstraction (data Independence), support of multiple views of the data, sharing of data and multiples transaction processing, Storage management, Database language – DDL, DML,DCL.

File processing system v/s DBMS, Data models, Levels of Abstraction in a DBMS, Three Schema architecture, Characteristics of database approach,, data models,DBMS architecture and data independence.

Unit - 2

(15 Lectures)

Entity Relationship and Enhanced ER Modeling: Entity types,Entity Sets, Attributes, and Keys, Relationships, Relationship Types, Roles, and Structural Constraints,Weak Entity Types, ER Diagrams, Naming Conventions ,SQL99:Schema Definition, constraints, and object modeling

Unit - 3

(15 Lectures)

Relational Data Model: Basic concepts, Relational Constraints and Relational Database Schemas, Update Operations and Dealing with Constraint Violations, Basic Relational Algebra Operations.

Database design: ER and EER to relational mapping, functional dependencies, normal forms-first normal form, second normal forms, third normal form BCNF

Unit - 4

(15 Lectures)

MYSQL (SQL/PL-SQL) :SQL VS. SQL * PLUS: SQL COMMANDS AND DATA TYPES, OPERATORS AND EXPRESSIONS, INTRODUCTION TO SQL * PLUS.

MANAGING TABLES AND DATA:

- CREATING AND ALTERING TABLES (INCLUDING CONSTRAINTS)
- DATA MANIPULATION COMMAND LIKE INSERT, UPDATE, DELETE
- SELECT STATEMENT WITH WHERE, GROUP BY AND HAVING, ORDER BY, DISTINCT, SPECIAL OPERATOR LIKE IN, ANY, ALL BETWEEN, EXISTS, LIKE
- JOIN, BUILT IN FUNCTIONS OTHER DATABASE OBJECTS
- VIEW • SYNONYMS, INDEX TRANSACTION CONTROL STATEMENTS
- COMMIT, ROLLBACK, SAVEPOINT INTRODUCTION TO PL/SQL
- SQL V/S PL/SQL • PL/SQL BLOCK STRUCTURE
- LANGUAGE CONSTRUCT OF PL/SQL (VARIABLES, BASIC AND COMPOSITE DATA TYPE, CONDITIONS LOOPING ETC.)
- % TYPE AND % ROWTYPE
- USING CURSOR (IMPLICIT, EXPLICIT)

Reference:

1. R. Elmasri, S.B. Navathe, Fundamentals of Database Systems 6th Edition, Pearson Education, 2010.
2. R. Ramakrishanan, J. Gehrke, Database Management Systems 3rd Edition, McGraw-Hill, 2002.
3. A. Silberschatz, H.F. Korth, S. Sudarshan, Database System Concepts 6th Edition, McGraw Hill, 2010.
4. R. Elmasri, S.B. Navathe Database Systems Models, Languages, Design and application Programming, 6th Edition, Pearson Education, 2013.

Database Management Systems Lab

Software Lab based on Database Management Systems

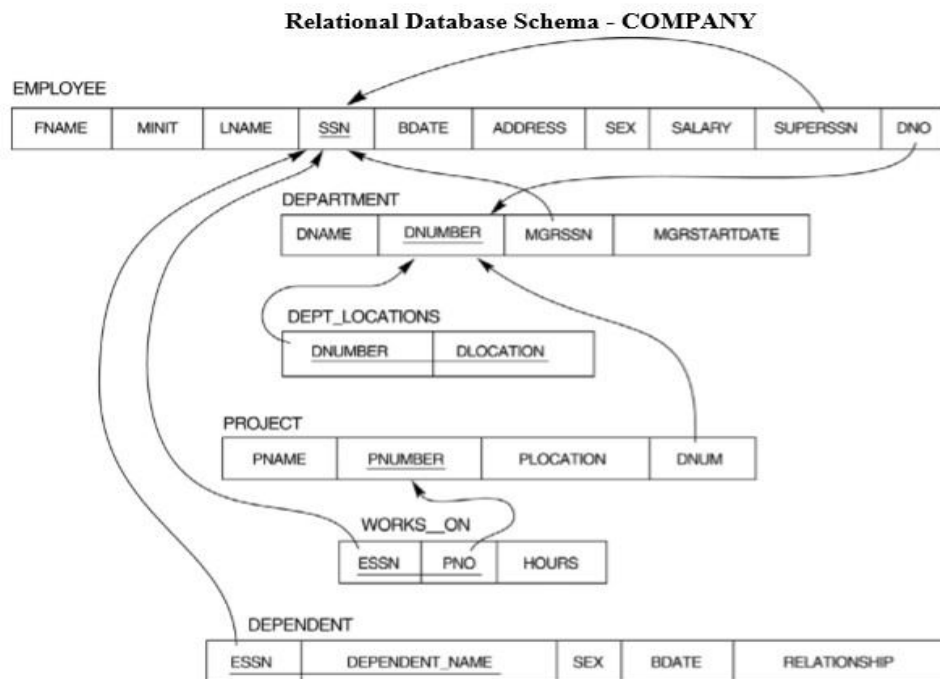
The following concepts must be introduced to the students: **Note:** MS Access/MySQL may be used.

DDL Commands

- Create table, alter table, drop table

DML Commands

- Select, update, delete, insert statements
- Condition specification using Boolean and comparison operators (and, or, not, =, <>, >, <, >=, <=)
- Arithmetic operators and aggregate functions (Count, sum, avg, Min, Max)
- Multiple table queries (join on different and same tables) • Nested select statements
- Set manipulation using (any, in, contains, all, not in, not contains, exists, not exists, union, intersect, minus, etc.)
- Categorization using group byhaving
- Arranging using order by



1. Create tables with relevant foreign key constraints
2. Populate the tables with data
3. Perform the following queries on the database :
 - a. Display all the details of all employees working in the company.
 - b. Display SSN, LNAME, FNAME, address of employees who work in department no 7.

- c. Retrieve the birthdate and address of the employee whose name is 'Franklin T. Wong'
4. Retrieve the name and salary of every employee
5. Retrieve all distinct salary values
6. Retrieve all employee names whose address is in 'Bellaire'
7. Retrieve all employees who were born during the 1950s
8. Retrieve all employees in department 5 whose salary is between 50,000 and 60,000(inclusive)
9. Retrieve the names of all employees who do not have supervisors
10. Retrieve SSN and department name for all employees
11. Retrieve the name and address of all employees who work for the 'Research' department
12. For every project located in 'Stafford', list the project number, the controlling department number, and the department manager's last name, address, and birthdate.
13. For each employee, retrieve the employee's name, and the name of his or her immediate supervisor.
14. Retrieve all combinations of Employee Name and Department Name
15. Make a list of all project numbers for projects that involve an employee whose last name is 'Narayan' either as a worker or as a manager of the department that controls the project.
16. Increase the salary of all employees working on the 'ProductX' project by 15%. Retrieve employee name and increased salary of these employees.
17. Retrieve a list of employees and the project name each works in, ordered by the employee's department, and within each department ordered alphabetically by employee first name.
18. Select the names of employees whose salary does not match with salary of any employee in department.
19. Retrieve the name of each employee who has a dependent with the same first name and same sex as the employee.
20. Retrieve the employee numbers of all employees who work on project located in Bellaire, Houston, or Stafford.
21. Find the sum of the salaries of all employees, the maximum salary, the minimum salary, and the average salary. Display with proper headings.
22. Find the sum of the salaries and number of employees of all employees of the 'Marketing' department, as well as the maximum salary, the minimum salary, and the average salary in this department
23. Select the names of employees whose salary is greater than the average salary of all employees in department 10.
24. For each department, retrieve the department number, the number of employees in the department, and their average salary.
25. For each project, retrieve the project number, the project name, and the number of employees who work on that project.
26. Change the location and controlling department number for all projects having more than 5 employees to 'Bellaire' and 6 respectively.
27. For each department having

UG-Computer Science - CBCS Scheme

more than 10 employees, retrieve the department no, no of employees drawing more than 40,000 as salary.

28. Insert a record in Project table which violates referential integrity constraint with respect to Department number. Now remove the violation by making necessary insertion in the Department table.
29. Delete all dependents of employee whose ssn is '123456789'.
30. Delete an employee from Employee table with ssn = '12345'(make sure that this employee has some dependents, is working on some project, is a manager of some department and is supervising some employees). Check and display the cascading effect on Dependent and Works on table. In Department table MGRSSN should be set to default value and in Employee table SUPERSSN should be set to NULL
31. Perform a query using alter command to drop/add field and a constraint in Employee table.

SEMESTER V

Course code: CME25202

DSE3B: Elective:Computer Science - V Computer Networks

Credits: Theory – 04, Practical – 01

Theory: 60 Lectures

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Understand the Elements of Data Communications and network Systems
- CO2. Learn in depth Transmission Media
- CO3. Understand in details with examples Network Models
- CO4. Understanding the various classifications and characteristics of Protocols
- CO5. Learn in depth Error Detection and Corrections Algorithms
- CO6. Learn in detail of Network Security

Unit - 1

(15 Lectures)

Basic concepts: Components of data communication, standards and organizations, Network Classification, Network Topologies ; network protocol; layered network architecture; overview of OSI reference model; overview of TCP/IP protocol suite.

Unit - 2

(15 Lectures)

Physical Layer: Cabling, Network Interface Card, Transmission Media Devices- Repeater, Hub, Bridge, Switch, Router, Gateway.

Data Link Layer: Framing techniques; Error Control; Flow Control Protocols; Shared media protocols - CSMA/CD and CSMA/CA.

Unit - 3

(15 Lectures)

Network Layer: Virtual Circuits and Datagram approach, IP addressing methods – Subnetting; Routing Algorithms (adaptive and non-adaptive)

Transport Layer: Transport services, Transport Layer protocol of TCP and UDP

Unit - 4

(15 Lectures)

Application Layer: Application layer protocols and services – Domain name system, HTTP, WWW, telnet, FTP, SMTP

Network Security: Common Terms, Firewalls, Virtual Private Networks.

Reference:

1. B.A. Forouzan: Data Communication and Networking, 4th Edition, Tata McGraw Hill, 2007.
2. D.E. Comer, Internetworking with TCP/IP, Vol. I, Prentice Hall of India, 1998.
3. W. Stalling, Data & Computer Communication, 8th edition, Prentice Hall of India, 2006.
4. D. Bertsekas, R. Gallager, Data Networks, 2nd edition, Prentice Hall of India, 1992.

Computer Networks Lab

Software Lab based on Computer Networks:

Implement the concepts of Computer Networks such as:

1. Simulate Checksum Algorithm.
2. Simulate CRC Algorithm
3. Simulate Stop & Wait Protocol.
4. Simulate Go-Back-N Protocol.
5. Simulate Selective Repeat Protocol.and so on....

SEMESTER V

Course code: CME25402

DSE 3C: Elective: Computer Science - V Software Engineering

Credits: Theory – 04, Practical – 01

Theory: 60 Lectures

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Understand in details with examples Concepts of Software process
- CO2. Specify the details of Software requirements and analysis
- CO3. Learn in depth Design concepts and principles of software engineering
- CO4. Understand in depth software Configuration Management and Project Management
- CO5. Learn in details with examples Software Testing
- CO6. Specify in depth trends in software engineering

Unit - 1

(15 Lectures)

Software Process: Introduction ,S/W Engineering Paradigm , life cycle models (water fall, incremental, spiral, evolutionary, prototyping, object oriented) , System engineering, computer based system, verification, validation, life cycle process, development process, system engineering hierarchy.

Software requirements: Functional and non-functional, user, system, requirement engineering process, feasibility studies, requirements, elicitation, validation and management, software prototyping, prototyping in the software process, rapid prototyping techniques, user interface prototyping, S/W document.

Unit - 2

(15 Lectures)

Analysis and modeling, data, functional and behavioral models, structured analysis and data dictionary.

Design Concepts and Principles: Design process and concepts, modular design, design heuristic, design model and document, Architectural design, software architecture, data design, architectural design, transform and transaction mapping, user interface design, user interface design principles. Real time systems, Real time

software design, system design, real time executives, data acquisition system, monitoring and control system.

Unit - 3 **(15 Lectures)**

Software Configuration Management: The SCM process, Version control, Change control, Configuration audit, SCM standards.

Software Project Management: Measures and measurements, S/W complexity and science measure, size measure, data and logic structure measure, information flow measure. Estimations for Software Projects, Empirical Estimation Models, Project Scheduling.

Unit - 4 **(15 Lectures)**

Testing: Taxonomy of software testing, levels, test activities, types of s/w test, black box testing, testing boundary conditions, structural testing, test coverage criteria based on data flow, mechanisms, regression testing, testing in the large. S/W testing strategies, strategic approach and issues, unit testing, integration testing, validation testing, system testing and debugging.

Trends in Software Engineering: Reverse Engineering and Re-engineering – wrappers – Case Study of CASE tools.

Reference:

1. Roger S.Pressman, Software engineering- A practitioner's Approach, McGraw-Hill
2. Ian Sommerville, Software engineering, Pearson education Asia, 6th edition, 2000.
3. PankajJalote- An Integrated Approach to Software Engineering, Springer Verlag, 1997.
4. James F Peters and WitoldPedrycz, "Software Engineering – An Engineering Approach", John Wiley and Sons, New Delhi, 2000.
5. Ali Behforooz and Frederick J Hudson, "Software Engineering Fundamentals", Oxford University Press, New Delhi, 1996.
6. Pfleeger, "Software Engineering", Pearson Education India, New Delhi, 1999.
Carlo Ghezzi, Mehdi Jazayari and Dino Mandrioli, "Fundamentals of Software Engineering", Prentice Hall of India, New Delhi, 1991.

Software Engineering Lab

Lab based on Software Engineering

1. Practical Title
 - Problem Statement,
 - Process Model
2. Requirement Analysis
 - Creating a Data Flow
 - Data Dictionary,
 - Use Cases
3. Project Management
 - Computing FP
 - Effort
 - Schedule, Risk Table, Timeline chart
- 1 Design Engineering
 - Architectural Design
 - Data Design, Component Level Design
5. Testing
 - Basis Path Testing

Sample Projects

- DTC Route Information: Online information about the bus routes and their frequency and fares
- Car Pooling: To maintain a web based intranet application that enables the corporate employees within an organization to avail the facility of carpooling effectively.
- Patient Appointment and Prescription Management System
- Organized Retail Shopping Management Software
- Parking Allocation System
- Wholesale Management System

SEMESTER V

Course code: CME38002

SEC 4A: Elective: Computer Application (Practical) Office Automation

Credits: Theory – 00,

Practical – 01

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Understand the details of fundamentals of Computer
- CO2. Learn in depth Hardware and Software
- CO3. Learn the details of Computer Peripherals
- CO4. Understand the details of Programming Languages
- CO5. Deliberate in details with examples office automation Tools
- CO6. Deliberate in depth Operating System and the User Interface
- CO7. Understand in details of Internet and its usages

Practical List for WORD:

1. Create a **telephone directory**.
 - The heading should be 16-point Arial Font in bold
 - The rest of the document should use 10-point font size
 - Other headings should use 10-point Courier New Font.
 - The footer should show the page number as well as the date last updated.
2. Design a time-table form for your college.
 - The first line should mention the name of the college in 16-point Arial Font and should be bold.
 - The second line should give the course name/teacher's name and the department in 14-point Arial.
 - Leave a gap of 12-points.
 - The rest of the document should use 10-point Times New Roman font.
 - The footer should contain your specifications as the designer and date of creation.
3. Create the following document: A newsletter with a headline and 2 columns in portrait orientation, including at least one image surrounded by text.

Practical List for EXCEL:

1. Create a student worksheet containing roll numbers, names and total marks. Open a document in Word and insert the excel worksheet using:-
 - i) Copy/Paste
 - ii) Embedding

UG-Computer Science - CBCS Scheme

- iii) Linking
2. The term wise marks for APS class of 20 students are stored in 3 separate sheets named term1, term2 and term3. Create 4th worksheet that contains student names and their total and average marks for the entire year. Give proper headings using headers. Make the column headings bold and italic. The 4th worksheet should contain college name as the first line. Make it bold, italic and center it.
3. Consider the following employee worksheet:-

Full Name (First Last)	Grade 1/2/3	Basic Salary	HRA	PF	Gross	Net	(VA)VehicleAllowance

HRA is calculated as follows:

Grade	HRA %(of Basic)
1	40%
2	35%
3	30%

Gross = Basic + HRA + VA

Net = Gross - PF

PF is 8% for all Grades

VA is 15000, 10000 and 7000 for Grades 1, 2 and 3.

- Find max, min and average salary of employees in respective Grade
 - Count no. of people where VA>HRA
 - Find out most frequently occurring grade.
 - Extract records where employee name starts with "A" has HRA>10000
 - Print Grade wise report of all employees with subtotals of net salary and also grand totals. Use subtotal command.
 - Extract records where Grade is 1 or 2 and salary is between 10000 and 20000 both inclusive.
4. In a meeting of a marketing department of an organization it has been decided that price of selling an item is fixed at Rs40. It was resolved to increase the sell of more of more items and getting the profit of Rs40,000/.Use Goal Seek of find out how many items you will have to sell to meet your profit figure.

5. Consider the following worksheet for APS 1st year students:-

S.No.	Name	PH	CH	BY	MT	CS	Total Marks	%	Grade
1									
2									

Grade is calculated as follows:-

If % ≥ 90 Grade A
If % ≥ 80 & < 90 Grade B
If % ≥ 70 & < 80 Grade C
If % ≥ 60 & < 70 Grade D

Otherwise students will be declared fail.

- i) Calculate Grade using if function
- ii) Sort the data according to total marks
- iii) Apply filter to display the marks of the students having more than 65% marks.
- iv) Draw a pie chart showing % marks scored in each subject by the topper of the class.
- v) Draw the doughnut chart of the data as in (iv)
- vi) Enter the S.No. of a student and find out the Grade of the student using VLOOKUP.
- vii) Extract all records where name
 - a) Begins with "A"
 - b) Contains "A"
 - c) Ends with "A"

Practical List for Power Point:

- 2 Create five Power point slides. Each slide should support different format. In these slides Explain areas of applications of IT. Make slide transition time as 10 seconds.
- 3 Create five Power Point slides to give advantages/disadvantages of computer, application of computers and logical structure of computer.
- 4 Create five Power Point slides detailing the process of internal assessment. It should be a self-running demo.

SEMESTER V

Course code: CME38202

SEC 4B: Elective: Computer Application (Practical) Elective:XML Programming

Credits: Theory – 00, Practical – 01

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Understand the details of Elements of XML Programming
- CO2. Write down in details with examples DTD
- CO3. Deliberate in details with examples XML Schemas
- CO4. Deliberate the characteristics of CSS
- CO5. Learn in details with application CSS
- CO6. Deliberate in details with examples XSL
- CO7. Learn the details of XML Security
- CO8. Learn in details with examples XML and JAVA
- CO9. Learn in details with examples XML and ASP.Net

Software Lab Based on XML:

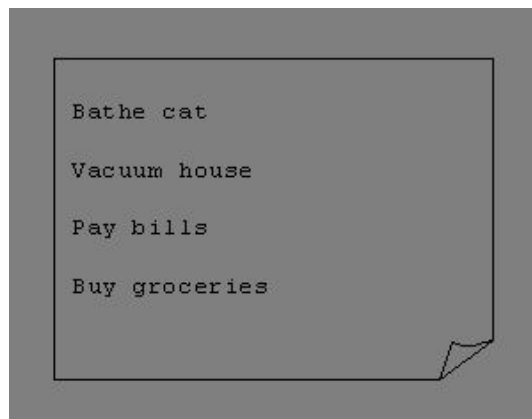
Exercise #1 – Information Structure

In this exercise, student will practice identifying the structure of an information object.

For the sample document provided below:

Label the information structures you see, including containing structures.

12. Draw a tree representation of the structure.



Exercise 2# Deconstructing an XML Document

In this exercise, student will practice identifying the explicit structure within an XML document. In a sense, this is the reverse of what you did in Exercise #1. For the sample XML markup below, create a document-like representation (or a simple drawing) for the content contained within the XML tags:

```
<book>
<coverInfo>
<title>The XML Handbook</title>
<author>Charles F. Goldfarb</author>
<author>Paul Prescod</author>
<edition>Second</edition>
<description>The definitive XML resource: applications, products, and technologies. Revised
and expanded—over 600 new pages.
</description>
</coverInfo>
</book>
```

Exercise #3 – Creating XML Markup

In this exercise, create some XML markup based on the tree representation from Exercise #1 above, and the content from the original sample document.

Exercise #4 – Well-Formedness

This exercise checks your understanding of the constraints for well-formedness. Are the following document instances well-formed? Explain any NO answers.

```
<list><title>The first list</title><item>An item</list>
<item>An item</item><item>Another item</item>
<para>Bathing a cat is a <emph>relatively</emph> easy task as long as the cat is
willing.</para>
<bibl><title>How to Bathe a Cat<author></title>Merlin Bauer<author></bibl>
```

Exercise #5-Well Formedness

This exercise is a bit more challenging than the previous example. Here is a fragment of an XML document instance. Identify all the places where it fails to match the constraints for well-formedness.

```
<PROCEDURE><TITLE>How to Bathe a Cat</TITLE>
<OVERVIEW>
```

```
This procedure tells you how to bathe a cat. <WARNING></OVERVIEW>Cats don't like to
take baths. You could get hurt doing this. Be sure to obtain all the required protective gear
before you start. </WARNING><EQUIPEMENT><ITEM>Hockey Mask <ITEM>Padded Full-
```

body Kevlar Armor</ITEM><ITEM>Tub full of warm water</ITEM><ITEM>Towels
</ITEM><ITEM>First Aid kit</ITEM><ITEM>Cat
Shampoo</ITEM><EQUIPMENT><INSTRUCTIONS><STEP> Locate the cat, who by now is
hiding under the bed.</STEP><STEP>Place the cat in the tub of water.</STEP><ITEM>Using
the First Aid kit, repair the damage to your head and arms.</STEP><STEP>Place the cat back
in the tub and hold it down.</STEP><STEP>Wash it really fast, then make an effort to dry it
with the towels.</STEP><STEP>Decide not to do this again. </STEP></INSTRUCTIONS>

Note: Cover more exercises based on XML Programming theory concepts.

SEMESTER V

Course code: CME38402

SEC 4C: Elective: Computer Application (Practical) R Programming

Credits: Theory – 00, Practical – 01

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Learn in depth Elements of R Programming
- CO2. Understand the details of R programming Packages
- CO3. Understand in details with examples of Data Management
- CO4. Deliberate the characteristics of Data Management
- CO5. Understand in depth Graphical Procedures
- CO6. Deliberate in depth Portability Distribution
- CO7. Software Lab Based on R Programming

1. Write a program that prints 'Hello World' to the screen.
2. Write a program that asks the user for a number n and prints the sum of the numbers 1 to n
3. Write a program that prints a multiplication table for numbers up to 12.
4. Write a function that returns the largest element in a list.
5. Write a function that computes the running total of a list.
6. Write a function that tests whether a string is a palindrome.
7. Implement the following sorting algorithms: Selection sort, Insertion sort, Bubble Sort
8. Implement linear search.
9. Implement binary search.
10. Implement matrices addition, subtraction and Multiplication

SEMESTER VI

Course code: CMF25002

DSE6A: Elective:Computer Science - VII Internet Technologies

Credits: Theory – 04, Practical – 01

Theory: 60 Lectures

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Learn the details of HTML tags
- CO2. Understand the details of Basic CSS and implements
- CO3. Understand the details of Basic Concepts of Java Scripts
- CO4. Learn the Core Java Programming
- CO5. Write down in details with application and Usage of JDBC
- CO6. Learn in detail of JSP Environment

Unit - 1

(15 Lectures)

Introduction to Web Design: Introduction to hypertext markup language (html) document type definition, creating web pages, graphical elements, lists, hyperlinks, tables, web forms, inserting images, frames.

Customized Features: Cascading style sheets, (css) for text formatting and other manipulations.

JavaScript: Data types, operators, functions, control structures

Unit - 2

(15 Lectures)

Java Script events and event handling.

Java: Use of Objects, Array and Array List class, Designing classes, Inheritance, Input/Output, Exception Handling.

Unit - 3

(15 Lectures)

JDBC: JDBC Fundamentals, Establishing Connectivity and working with connection interface, Working with statements, Creating and Executing SQL Statements, Working with Result Set Objects.

JSP: Introduction to JavaServer Pages, HTTP and Servlet Basics, The Problem with Servlets, The Anatomy of a JSP Page, JSP Processing, JSP Application Design with MVC

Unit - 4

(15 Lectures)

Setting Up the JSP Environment, Implicit JSP Objects, Conditional Processing, Displaying Values, Using an expression to Set an Attribute, Declaring Variables and Methods, Error Handling and Debugging, Sharing Data Between JSP Pages, Requests, and Users, Database Access.

Reference Books:

1. Web Enabled Commercial Application Development Using Html, Dhtml, javascript, Perl CgiBy Ivan Bayross, BPB Publications, 2009.
2. BIG Java Cay Horstmann, Wiley Publication , 3rd Edition., 2009
3. Java 7 ,The Complete Reference, Herbert Schildt, 8th Edition, 2009.
4. The Complete Reference J2EE, TMH, Jim Keogh, 2002.
5. Java Server Pages, Hans Bergsten, Third Edition, O'Reilly Media December 2003.

Internet TechnologiesLab

Software Lab based on Internet Technologies

JAVA Script

1. Create a student registration form. Create functions to perform the following checks:
 - a. Roll number is a 7-digit numeric value
 - b. Name should be an alphabetical value (String)
 - c. Non-empty fields like DOB
2. Implement a static password protection.
3. Write a java script
 - a. To change the colour of text using SetTimeout()
 - b. To move an image across screen using SetInterval()

JAVA Programs

1. WAP to find the largest of n natural numbers.
2. WAP to find whether a given number is prime or not.

3. WAP to print the sum and product of digits of an Integer and reverse the Integer.
4. Write a program to create an array of 10 integers. Accept values from the user in that array. Input another number from the user and find out how many numbers are equal to the number passed, how many are greater and how many are less than the number passed.
5. Write java program for the following matrix operations:
 - a. Addition of two matrices
 - b. Summation of two matrices
 - c. Transpose of a matrixInput the elements of matrices from user
6. Write a java program that computes the area of a circle, rectangle and a Cylinder using function overloading.

JDBC

1. Create a table 'Student' and 'Teacher' in 'College' database and insert two rows in this newly created table using JDBC API and do the following:
 - a. Update an already created table 'Teacher' in 'College' database by updating a teacher's name, with "Dr." appended before the name, whose name is "Rita".
 - b. Repeat the same thing for all the teachers using PreparedStatement.
 - c. Delete the student with ID=3 from 'Student' database.
 - d. Insert two students to the ResultSet returned by the query which selects all students with FirstName="Ayush". The database must also get updated along with ResultSet.
2. Create a procedure in MySQL to count the number of Rows in table 'Student'. Use Callable Statement to call this method from Java code.

JSP Practical list

1. Display the pattern:

1

1 2

1 2 3

Take 'n' in a textbox from user. Display this pattern using

- Scriptlets
- <c:forEach> loop

2. Make two files as follows:

- a. main.html: shows 2 text boxes and 3 radio buttons with values "addition", "subtraction" and "multiplication"
- b. operate.jsp: depending on what the user selects perform the corresponding function (Give two implementations: using request.getParameter() and using expression language)
3. Validate User input entered in a form. The input must include Name, DOB, Email ID, Lucky Number, Favorite food etc. (Refer Chapter 8)
4. Display Good Morning <uname>, Good Afternoon <uname> or Good Evening <uname> based on the current time of the day.
5. Create your custom library which contains two tags: <hello>, <choco>.

Usage of the tags:

- <hello name="Ajay">: Output should be Hello Ajay. It contains a mandatory attribute 'name' which can accept Dynamic value.
- <choco texture="Chewy">: Output should be FiveStar, BarOne.
- <choco texture="Crunchy">: Output should be Munch. KitKat.

That means the mandatory attribute must accept a value, and based on the attributes value, it should give output. You must use a bean ChocoBean for this purpose.

SEMESTER VI

Course code: CMF25202

DSE6B: Elective:Computer Science - VI Mobile Applications

Credits: Theory – 04, Practical – 01

Theory: 60 Lectures

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Deliberate the details of Concepts of Event Driven Programming
- CO2. Learn in details with examples issues of Mobile applications
- CO3. Specify the details of Mobile applications Development tools and Frameworks
- CO4. Deliberate in details with examples common Mobile device UI's
- CO5. Write down in depth Data persistence Remote data storage and communication
- CO6. Learn in details with examples Code signing

Unit - 1

(15 Lectures)

Event Driven Programming: UI event loop, Threading for background tasks, Outlets / actions, delegation, notification, Model View Controller (MVC) design pattern.

Mobile application issues: limited resources (memory, display, network, file system), input / output (multi-touch and gestures), sensors (camera, compass, accelerometer, GPS)

Unit - 2

(15 Lectures)

Development tools: Apple iOSToolchain: Objective-C, Xcode IDE, Interface Builder, Device simulator.

Frameworks: Objective-C and Foundation Frameworks, Cocoa Touch, UIKit, Others: Core Graphics, Core Animation, Core Location and Maps, Basic Interaction.

Unit - 3

(15 Lectures)

Common UI's for mobile devices: Navigation Controllers, Tab Bars, Table Views, Modal views, UI Layout.

Data Persistence: Maintaining state between application invocations, File system, Property Lists, SQLite, Core Data.

Unit - 4

(15 Lectures)

Remote Data-Storage and Communication: "Back End" / server side of application, RESTful programming, HTTP get, post, put, delete, database design, server side JavaScript / JSON.

Code signing: security, Keychain, Developers and App Store License Agreement (6L)

Reference:

1. Rajiv Ramnath, Roger Crawfis, and Paolo Sivilotti, Android SDK 3 for Dummies, Wiley, 2011.
2. Valentino Lee, Heather Schneider, and Robbie Schell, Mobile Applications: Architecture, Design, and Development, Prentice Hall, 2004.
3. Brian Fling, Mobile Design and Development, O'Reilly Media, 2009. Maximiliano
4. Firtman, Programming the Mobile Web, O'Reilly Media, 2010.
5. Christian Crumlish and Erin Malone, Designing Social Interfaces, O'Reilly Media, 2009.
6. James E Smith, Ravi Nair, "Virtual Machines", Morgan Kaufmann Publishers, 2006.

Mobile Applications Lab

Software Lab based on Mobile Applications:

1. Installing Android Environment
2. Create Hello World Application
3. Sample Application about Android Resources
4. Sample Application about Layouts

5. Sample Application about Intents
6. Sample Application I about user interfaces
7. Sample Application about Animations
8. Make a Project based on above labs
9. Sample Application about Android Data
10. Sample Application about SQLite I
11. Sample Application about SQLite II
12. Project Presentation

SEMESTER VI

Course code: CMF25402

DSE6C: Elective:Computer Science - VI Cloud Computing

Credits: Theory – 04, Practical – 01

Theory: 60 Lectures

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Learn in depth Fundamentals of Cloud Computing
- CO2. Understand the details of Cloud Services and File System
- CO3. Learn in depth Concept of Collaborating with Cloud
- CO4. Understand the details of Virtualization in cloud
- CO5. Learn the classification and characteristics of Security challenges in Cloud Computing
- CO6. Specify the classification and characteristics of Security challenges in Cloud Computing
- CO7. Understand the details of Security challenges in Cloud Computing
- CO8. Understand the Common standards of Cloud Computing
- CO9. Deliberate in details with examples Various Application of Cloud Computing

Unit - 1

(15 Lectures)

Cloud Introduction: Cloud Computing Fundamentals: Cloud Computing definition, Types of cloud, Cloud services: Benefits and challenges of cloud computing, Evolution of Cloud Computing, usage scenarios and Applications, Business models around Cloud– Major Players in Cloud Computing - Issues in Cloud - Eucalyptus - Nimbus - Open Nebula, CloudSim.

Cloud Services And File System: Types of Cloud services: Software as a Service - Platform as a Service – Infrastructure as a Service - Database as a Service - Monitoring as a Service – Communication as services.

Unit - 2

(15 Lectures)

Service providers- Google App Engine, Amazon EC2, Microsoft Azure, Sales force.

Collaborating With Cloud: Collaborating on Calendars, Schedules and Task Management – Collaborating on Event Management, Contact Management, Project Management – Collaborating on Word Processing , Databases – Storing and Sharing Files- Collaborating via Web-Based Communication Tools – Evaluating Web Mail Services – Collaborating via Social Networks – Collaborating via Blogs and Wikis. 185 CS-Engg&Tech-SRM-2013

Unit - 3

(15 Lectures)

Virtualization For Cloud: Need for Virtualization – Pros and cons of Virtualization – Types of Virtualization – System Vm, Process VM, Virtual Machine monitor – Virtual machine properties - Interpretation and binary translation, HLL VM - Hypervisors – Xen, KVM , VMWare, Virtual Box, Hyper-V.

Unit - 4

(15 Lectures)

Security, Standards, And Applications: Security in Clouds: Cloud security challenges – Software as a Service Security, Common Standards: The Open Cloud Consortium – The Distributed management Task Force – Standards for application Developers – Standards for Messaging – Standards for Security, End user access to cloud computing, Mobile Internet devices and the cloud.

Reference:

1. Bloor R., Kanfman M., Halper F. Judith Hurwitz “Cloud Computing ” Wiley India Edition, 2010
2. John Rittinghouse & James Ransome, “Cloud Computing Implementation Management and Strategy”, CRC Press, 2010
3. Antohy T Velte , Cloud Computing : “A Practical Approach”, McGraw Hill, 2009
4. Michael Miller, Cloud Computing: “Web-Based Applications That Change the Way You Work and Collaborate Online”, Que Publishing, August 2008.
5. James E Smith, Ravi Nair, “Virtual Machines”, Morgan Kaufmann Publishers, 2006.

Online Reading/Supporting Material

1. Haley Beard, "Cloud Computing Best Practices for Managing and Measuring Processes for On-demand Computing", Applications and Data Centers in the Cloud with SLAs, Emereo Pty Limited, July 2008
2. webpages.iust.ac.ir/hsalimi/.../89.../Cloud%20Common%20standards.pptopennebula.org,
3. www.cloudbus.org/cloudsim/, <http://www.eucalyptus.com/>
4. hadoop.apache.org
5. http://hadoop.apache.org/docs/stable/hdfs_design.html
6. http://static.googleusercontent.com/external_content/untrusted_dlcp/research.google.com/en//archive/mapreduce-osdi04.pdf

Cloud Computing Lab

Software Lab based on Cloud Computing:

1. Create virtual machines that access different programs on same platform.
2. Create virtual machines that access different programs on different platforms.
3. Exploring Google cloud for the following
 - a) Storage
 - b) Sharing of data
 - c) Manage your calendar, to-do lists,
 - d) A document editing tool
4. Exploring Open source cloud (Any two)

SEMESTER VI

Course code: CMF38002

SEC4A: Elective:Computer Science - VII Android Programming

Credits: Theory – 01, Practical – 01

Theory: 15 Lectures

Course Outcome:

After successful completion of the course, the student is able to

CO1. Learn the details of Overview of Android

CO2. Understand in details with examples Concepts of OOP's Using JAVA

CO3. Learn in depth Development Tools

CO4. Understand in depth User Interface Architecture

CO5. Learn the details of User Interface Design

CO6. Understand in depth SQLite Database Connectivity

Unit - 1

(15 Lectures)

Introduction: History of Android, Introduction to Android Operating Systems, Android Development Tools, Android Architecture.

Overview of object oriented programming using Java: OOPs Concepts: Inheritance, Polymorphism, Interfaces, Abstract class, Threads, Overloading and Overriding, Java Virtual Machine.

Development Tools: Installing and using Eclipse with ADT plug-in, Installing Virtual machine for Android sandwich/Jelly bean (Emulator), configuring the installed tools, creating a android project – Hello Word, run on emulator, Deploy it on USB-connected Android device.

User Interface Architecture: Application context, intents, Activity life cycle, multiple screen sizes.

User Interface Design: Form widgets, Text Fields, Layouts, Button control, toggle buttons, Spinners (Combo boxes), Images, Menu, and Dialog.

Database: Understanding of SQLite database, connecting with the database.

Reference:

1. Android application development for java programmers. By James C. Sheusi. Publisher: Cengage Learning, 2013.

ONLINE READING / SUPPORTING MATERIAL:

1. <http://www.developer.android.com>
2. <http://developer.android.com/about/versions/index.html>
3. <http://developer.android.com/training/basics/firstapp/index.html>
4. <http://docs.oracle.com/javase/tutorial/index.htm> (Available in the form of free downloadable ebooks also).
5. <http://developer.android.com/guide/components/activities.html>
6. <http://developer.android.com/guide/components/fundamentals.html>
7. <http://developer.android.com/guide/components/intents-filters.html>.
8. <http://developer.android.com/training/multiscreen/screensizes.html>
9. <http://developer.android.com/guide/topics/ui/controls.html>
10. <http://developer.android.com/guide/topics/ui/declaring-layout.html>
11. <http://developer.android.com/training/basics/data-storage/databases.html>

Android Programming Lab

Software Lab Based on Android Programming:

1. Create “Hello World” application. That will display “Hello World” in the middle of the screen in the emulator. Also display “Hello World” in the middle of the screen in the Android Phone.
2. Create an application with login module. (Check username and password).
3. Create spinner with strings taken from resource folder (res >> value folder) and on changing the spinner value, Image will change.
4. Create a menu with 5 options and selected option should appear in text box.
5. Create a list of all courses in your college and on selecting a particular course teacher-in-charge of that course should appear at the bottom of the screen.
6. Create an application with three option buttons, on selecting a button colour of the screen will change.
7. Create and Login application as above. On successful login, pop up the message.
8. Create an application to Create, Insert, update, Delete and retrieve operation on the database.

SEMESTER VI

Course code: CMF38202

SEC4B: Elective:Computer Science - VII PHP Programming

Credits: Theory – 01, Practical – 01

Theory: 15 Lectures

Course Outcome:

After successful completion of the course, the student is able to

- CO1. Learn in depth Elements of PHP
- CO2. Learn in depth Interaction Methods Between HTML and PHP
- CO3. Understand in depth PHP function
- CO4. Understand in depth String Manipulation
- CO5. Learn the characteristics of Regular Expression
- CO6. Learn the details of Developing PHP Web Application

Unit - 1

(15 Lectures)

Introduction to PHP: PHP introduction, inventions and versions, important tools and software requirements (like Web Server, Database, Editors etc.), PHP with other technologies, scope of PHP, Basic Syntax, PHP variables and constants, Types of data in PHP , Expressions, scopes of a variable (local, global),, PHP Operators : Arithmetic, Assignment, Relational , Logical operators, Bitwise , ternary and MOD operator.

PHP operator Precedence and associativity

Handling HTML form with PHP: Capturing Form Data, GET and POST form methods Dealing with multi value fields, Redirecting a form after submission, PHP conditional events and Loops: PHP IF Else conditional statements (Nested IF and Else) Switch case, while ,For and Do While Loop, Goto , Break ,Continue and exit

PHP Functions: Function, Need of Function , declaration and calling of a function, PHP Function with arguments, Default Arguments in Function, Function argument with call by value, call by reference, Scope of Function Global and Local

String Manipulation and Regular Expression: Creating and accessing String , Searching & Replacing String, Formatting, joining and splitting String , String Related Library functions, Use and advantage of regular expression over inbuilt function, Use of preg_match(), preg_replace(), preg_split()

functions in regular expression Array: Anatomy of an Array ,Creating index based and Associative array ,Accessing array, Looping with Index based array, with associative array using each() and foreach(),Some useful Library function

Reference:

1. Rajiv Ramnath, Roger Crawfis, and Paolo Sivilotti, Android SDK 3 for Dummies, Wiley, 2011.
2. Valentino Lee, Heather Schneider, and Robbie Schell, Mobile Applications: Architecture, Design, and Development, Prentice Hall, 2004.
3. Brian Fling, Mobile Design and Development, O'Reilly Media, 2009. Maximiliano
4. Firtman, Programming the Mobile Web, O'Reilly Media, 2010.
5. Christian Crumlish and Erin Malone, Designing Social Interfaces, O'Reilly Media, 2009.

PHP Programming Lab

Software Lab Based on PHP:

1. Create a PHP page using functions for comparing three integers and print the largest number.
2. Write a function to calculate the factorial of a number (non-negative integer). The function accept the number as an argument.
3. WAP to check whether the given number is prime or not.
4. Create a PHP page which accepts string from user. After submission that page displays the reverse of provided string.
5. Write a PHP function that checks if a string is all lower case.
6. Write a PHP script that checks whether a passed string is palindrome or not? (A palindrome is word, phrase, or sequence that reads the same backward as forward, e.g., madam or nurses run)

UG-Computer Science - CBCS Scheme

7. WAP to sort an array.
8. Write a PHP script that removes the whitespaces from a string.
Sample string : 'The quick " " brown fox'
Expected Output :Thequick""brownfox
9. Write a PHP script that finds out the sum of first n odd numbers.
10. Create a login page having user name and password. On clicking submit, a welcome message should be displayed if the user is already registered (i.e.name is present in the database) otherwise error message should be displayed.
11. Write a PHP script that checks if a string contains another string.
12. Create a simple 'birthday countdown' script, the script will count the number of days between current day and birth day.
13. Create a script to construct the following pattern, using nested for loop.

```
*  
  
* *  
  
* * *  
  
* * * *  
  
* * * * *
```

14. Write a simple PHP program to check that emails are valid.
15. WAP to print first n even numbers.
16. \$color = array('white', 'green', 'red')

Write a PHP script which will display the colors in the following way :

Output :

white, green, red,

- green
- red
- white

UG-Computer Science - CBCS Scheme

17. Using switch case and dropdown list display a “Hello” message depending on the language selected in drop down list.
18. Write a PHP program to print Fibonacci series using recursion.
19. Write a PHP script to replace the first 'the' of the following string with 'That'.

Sample : 'the quick brown fox jumps over the lazy dog.'

Expected Result : That quick brown fox jumps over the lazy dog.

SEMESTER VI

Course code: CMF38402

SEC4C: Elective:Computer Science - VII System Administration and Maintenance

Credits: Theory – 01, Practical – 01

Theory: 15 Lectures

Course Outcome:

After successful completion of the course, the student is able to

- CO1. System Administration & Maintenance
- CO2. Understand in details of Networking OS Environment
- CO3. Deliberate in depth Basic LINUX Commands
- CO4. Deliberate in details with examples Basic Windows Commands
- CO5. Learn the details of Server Configuration and Maintenance

Unit - 1

(15 Lectures)

Part I - Linux/Unix:Basics of operating system, services,Installation and configuration, maintenance. What is linux/unix Operating systems?, Kernel, API, cli, gui,Difference between linux/unix and other operating systems, Features and Architecture, Linux features, advantages, disadvantages

Part II - Windows:Windows as operating system, history, versions. PC hardware, BIOS, Devices and drivers, Kernal Configuration and building, Application installation, configuration and maintenance, Server services and Client services, Difference between WindowsXP/windows7 and windows server 2003/2008.

Reference:

1. Evi Nemeth, Garth Snyder and others,Unix and Linux System Administration Handbook, 4th Ed,Pearson Education, 2016.
2. PDaniel J. Barrett, Linux Pocket Guide: Essential Commands, 3rd Edition,Shroff/O'Reilly; Second edition 4 June 2012
3. Woody Leonhard,Windows 7 All-in-One For Dummies,2009

System Administration and Maintenance Lab

Linux:

Linux Desktop tour. Configuring desktop environment and desktop settings.

Basic Commands : Terminal, shell, Cat, ls, cd, date, cal, man, echo, pwd, Mkdir, rm, rmdir Ps, kill

Package Installation

Synaptic package manager

Windows:

Creating users – Admin and regular.

Path of their personal files. Adding and changing passwords.

Difference between workgroup and domain. Concept of roles.

user profiles – creating and roaming Concept of Active Directory. Creating active directory in windows 2003/2008.

Process and Disk management

Windows Task manager. File systems – NTFS, FAT.

Services

Control Panel

C:/program Files, C:/system C:/windows

Add /remove new hardware (like printer), Add/remove new programmes.

Network Administration

Ipconfig, Ping, tracert, route, hostname, net, netstat, whoami

Set manual IP address, check connectivity – ipv4, ipv6

Administrator Tools

Control Panel -> Administrative Tools

Computer Management, Local security Policy, Performance Monitor, Task Scheduler, Antivirus and firewall.

Misc

Start->Accessories->System tools ->All options (Remote desktop, backup/restore etc.)

LAN – sharing printer, files and folder over the network.

**Question Paper Pattern
Theory (4 Credits)**

Time: 3 Hours

Max. Marks: 70

Part – A

I. Answer any Eleven Question out of given Twelve Questions. 11 X 2 = 22

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.
- 9.
- 10.
- 11.
- 12.

Note: Three Questions each from the Units 1, 2, 3, and 4

Part - B

II. Answer any Two SubQuestions from each main Question.

- | | |
|--------|------------|
| 13. A) | 2 X 6 = 12 |
| B) | |
| C) | |
| 14. A) | 2 X 6 = 12 |
| B) | |
| C) | |
| 15. A) | 2 X 6 = 12 |
| B) | |
| C) | |
| 16. A) | 2 X 6 = 12 |
| B) | |
| C) | |

Note: Each Main questions contains 3 sub questions carries 06 Marks (may have internal Split-ups) and from Units 1, 2, 3 and 4 respectively

Question Paper Pattern

Theory (1 Credit)

Time: 2 Hours

Max. Marks: 50

Part – A

I. Answer all Questions.

05 X 02 = 10

- 1.
- 2.
- 3.
- 4.
- 5.

Part - B

II. Answer any Four Questions out of given Five Questions. 4 X 10 = 40

- 1.
- 2.
- 3.
- 4.
- 5.

Note: Each Main questions contains 3 sub questions carries 06 Marks (may have internal Split-ups) and from Units 1, 2, 3 and 4 respectively

Question Paper Pattern

Practical / Project (1 or 1.5 or 2 Credits)

Time: 3 Hours

Max. Marks: 70

Practicals

1. Two Experiments/ Programs 20 Marks Each 2 x 20 = 40 Marks
 - a. Write-ups 10 Marks
 - b. Conducting & Results 10 Marks (Any One for 2 Hours of Practical's)
2. Viva – Voice 10 Marks
3. Record 10 Marks

Project

1. Project Presentation 15 Marks
2. Project Demo 20 Marks
3. Viva – Voice 10 Marks
4. Record 25 Marks

Question Paper Pattern

Practical (1 Credit)

Time: 2 Hours

Max. Marks: 50

- a. Experiment/ Programs 30 Marks Each 30 Marks
- b. Conducting & Results 10 Marks
2. Viva – Voice 10 Marks
3. Record 10 Marks

NATIONAL EDUCATION POLICY- 2020

Universities of Karnataka State under NEP-2020 in ENVIRONMENTAL SCIENCE

AECC - ENVIRONMENTAL STUDIES SYLLABUS

Number of Theory Credits	Number of lecture hours	Number of field work hours
2	45	5

	Content of AECC – Environmental Studies	45 hours
Unit 1	Introduction to Environmental Studies	45HOURS
	Multidisciplinary nature of environmental studies Scope and importance; Concept of sustainability and sustainable development.	2
Unit 2	Ecosystems	6
	What is an ecosystem? Structure and function of ecosystem; Energy flow in an ecosystem: food chains, food webs and ecological succession. Case studies of the following ecosystems: a) Forest ecosystem, c) Desert ecosystem, Aquatic ecosystems (ponds, , lakes, rivers,)	
Unit 3	Natural Resources: Renewable	8

	and Non-Renewable	
	land-use change; Land degradation, soil erosion and desertification. Deforestation: Causes and impacts due to mining, dam building on environment, forests, biodiversity and tribal populations. Water: Use and over-exploitation of surface and ground water, floods, droughts, conflicts over water (international & inter-state). Energy resources: Renewable and non-renewable energy sources, use of alternate energy sources, growing energy needs, case studies.	
Unit 4	Biodiversity and Conservation	8
	biodiversity services: Ecological, economic, social, ethical, aesthetic and Informational value Levels of biological diversity: Genetic, species and ecosystem diversity; Biogeographic zones of India; Biodiversity patterns and global biodiversity hot spots. India as a mega-biodiversity nation; Endangered and endemic species of India. Threats to biodiversity: Habitat loss, poaching of wildlife, man-wildlife conflicts, biological invasions; Conservation of biodiversity: In-situ and Ex-situ conservation of biodiversity.	

Unit 5	Environmental Pollution	8
	Environmental pollution: types, causes, effects and controls; Air, water, soil and noise pollution, Nuclear hazards and human health risks Solid waste management, Control measures of urban and industrial waste	
Unit 6	Environmental Policies & Practices	7
	Climate change, global warming, ozone layer depletion, acid Environment Laws: Environment Protection Act; Air (Prevention & Control of Pollution) Act; Water (Prevention and control of Pollution) Act; Wildlife Protection Act; Forest Conservation Act. International agreements: Montreal and Kyoto protocols and Convention on Biological Diversity (CBD).	
Unit 7	Human Communities and the Environment	6
	Human population growth: Impacts on environment, human health and welfare. Resettlement and rehabilitation of project affected persons; case studies. Disaster management: floods, earthquake, cyclones and landslides. Environmental	

	<p>movements: Chipko, Silent valley, Bishnois of Rajasthan Environmental ethics: Role of Indian and other religions and cultures in environmental conservation</p>	
Unit 8	Field work	5

NEP SYLLABUS B.A/B.Sc Semester

1

Title of the Course: Principles of Geomorphology

Code: GEOGDSC T1.1

Number of Theory Credits	Number of lecture hours/ semester	Number of practical Credits	Number of practical hours/ semesters
4	52 or 56	2	52 or 56
Course Outcomes: <ol style="list-style-type: none"> 1. After the completion of this course, students should be able to: 2. Define the field of Geomorphology and to explain the essential principles of it. 3. To outline the mechanism of dynamic nature of the Earth's surface and interior of the Earth. 4. To illustrate and explain the forces affecting the crust of the earth and its effect on it. 5. To understand the conceptual and dynamic aspects of landform development 			
Course Objectives: This course aims to: <ol style="list-style-type: none"> 1. To define the concepts in Geomorphology and Physical Geography 2. To introduce various concept to understand cycles of the solid Earth surface 3. To understand the dynamic nature of the Earth's surface, various processes, and landforms. 4. To study the impact human on geomorphic system. 			

Content of Theory Course 1	52/56Hrs
Unit – 1 Geomorphology	13/14
Introduction to geography: physical and human geography Introduction to Geomorphology: meaning, nature, development, and scope Principles of Geomorphology Geological Time Scale Distribution of continents and oceans	
Unit – 2 Systems and Cycles of the Solid Earth	13/14
Internal structure of the earth Alfred Wegener's continental drift Theory of Isostasy: Views of Pratt and Airy Convectional current theory and concept of sea floor spreading Theory of Plate Tectonics: plate boundaries, subduction, Case Studies: Volcano, Earthquake: reporting of latest incidents Case Studies: Volcano, Earthquake: reporting of latest incidents	
Unit – 3 The Dynamics of Earth	13/14
Earth's Movements: Endogenetic and Exogenetic forces, Sudden and Diastrophic movements- Epeirogenetic and Orogenetic Movements-Process of folding and faulting Vulcanicity and earthquake Rocks: Characteristics, types, importance, and rock cycle Weathering: meaning, types and controlling factors Mass Movement: meaning, controlling factors, types-landslides, rock-falls	

Unit – 4 Evolution of Landforms	13/14
<p>Evolution of Landforms</p> <p>Landforms: meaning, types and factors controlling landforms development Slope development: concept and types</p> <p>Concept of Cycle of Erosion–W.M. Davis and W. Penck</p> <p>Agents of Denudation: river; drainage patterns, groundwater, Sea waves, Wind and Glaciers and resultant landforms.</p> <p>Application of geomorphology: in India and Karnataka (Regional planning, Urban planning and transportation, Mining, Hazard management, Agriculture and Environmental management).</p>	

References

1. Ahmed E. (1985) Geomorphology, Kalyani Publishers, New Delhi.
2. Bloom A.L. (1978) Geomorphology: A Systematic Analysis of Late Cenozoic Landforms Prentice – Hall of India, New Delhi.
3. Brunnsden D. (1985) Geomorphology in the Service of Man: The Future of Geography, Methuen, U.K.
4. Chorley, R.J., Schumm, S. A. and Sugden, D.E. 1984: Geomorphology, Methuen, London
5. Cooke, R.U. and Warren, 1973: Geomorphology in Deserts, Batsford, London
6. Dayal, P. 1996: Textbook of Geomorphology, Shukla Book Depot, Patna.
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8. Goudie Anrew et.al. (1981) Geomorphological Techniques, George Allen & Unwin, London.
9. Homes A. (1965) Principles of Physical Geology, 3rd Edition, ELBSS Edn.
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12. Nanjannavar S S: Physical Geography (Kannada Version)
13. P Mallappa, Physical Geography (Kannada Version)
14. Ranganath Principles of Physical Geography (Kannada Version)
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16. Thornberry W.D. (1969) Principles of Geomorphology 2nd Edition, Wiley Intl. Edn. & Wiley, 1984.
17. Verstappen H. (1983) Applied Geomorphology, Geomorphological Surveys for Environmental Development, Elsevier, Amsterdam

Reference Websites

1. <http://www.solarviews.com/eng/earth.htm>
2. <http://www.moorlandschool.co.uk/earth/tectonic.htm>
3. <https://www.usgs.gov/>
4. <https://www.ksndmc.org/>

Pedagogy

Formative Assessment	
Assessment Occasion/type	Weightage in Marks
Case Studies	30%
Assignment	20%
CIA	50%
Total	100%

GEOGDSC P1.1 **Geomorphology Practical**

CREDIT:02

Content of Practical Course 1: List of Experiments to be conducted

Exercise-1: Identification of Rocks and Minerals. Mineral samples: Iron ore, Bauxite ore and Manganese. Rock Samples: Granite, Basalt, Lime Stones, Sandstone, quartzite, and marble.

Exercise-2: Extraction and interpretation of Geomorphic information from Topographical maps

Exercise-3: Preparation of contour map from toposheet, Construction of Relief Profiles-serial, Super imposed, Projected & Composite.

Exercise-4: Slope Analysis - Slope Maps (Wentworth method) Slope calculation and conversion (isotan and isosin) and aspect maps & Hypsometric curve and integral

Exercise-5: Drainage Morphometry: delineation of watershed, stream ordering and Morphometric analysis: mean stream length, drainage density and drainage frequency. **Field Work:** Measurement of channel cross-sections in the field, Geomorphic map of channel bed, Study of erosional and depositional features in the field.

Case Study: students must be taken to observe local land formation and degradation and write areport on their effectiveness.

B.A. / BSc Semester 1

Title of the Course: GEOG0E T1.1 - 1. Introduction to Physical Geography

Number of Theory Credits	Number of lecture hours/ semester
3	42 - 45
Course Outcomes: <ol style="list-style-type: none">1. Students will be able to understand the fundamental concepts in Earth Sciences2. Understands basic terminology used to describe physical processes and landscapeforms.3. Describe elements of the atmosphere and the oceans	
Course Objectives: <p>This course aims to</p> <ol style="list-style-type: none">1. Study basic principles of the Earth Sciences2. Understand the landforms, atmospheric elements and structure and basics of oceanography	
Content of Theory Course 1	
Unit – 1	11
Origin, Shape and Size of the Earth, Movement of the Earth- Rotation and Revolution, Effects of the movement of Earth,Coordinates -Latitude, Longitude and Time. Structure of the Earth,	
Unit – 2	11
Rocks - types,significance, Weathering – types. Agents of Denudation - River, Glacier, Wind and Under Ground water. Volcanicity, Earthquakes and Tsunamis	
Unit – 3	11
Structure and Composition of Atmosphere,Weather and Climate. Atmospheric Temperature, Heat Budget of the atmosphere Atmospheric Pressure, Winds and Precipitation	
Unit – 4	12
Distribution of Land and Sea, Submarine Relief of the Ocean, Temperature and Salinity of Sea Water. Ocean Tides, Waves and Deposits, Ocean currents - Atlantic, Pacific and IndianOceans. Marine Resources: Biotic, mineral and energy resources	

References

1. B.S. Negi (1993) Physical Geography. S.J. Publication, Meerut
2. D.S. Lal (1998) Climatology. Chaitnya publishing house, Allahabad
3. K. Siddhartha (2001) Atmosphere, Weather and Climate. Kisalaya publication, NewDelhi
4. R.N. Tikka (2002) Physical Geography. Kedarnath Ramnath & co, Meerut.
5. Willian D. Thornbury (1997) Principle of Geomorphology. New Age International (PvtLtd.) New Delhi.

Pedagogy

Formative Assessment	
Assessment Occasion/type	Weightage in Marks
Quiz	30%
Assignment	20%
CIA	50%
Total	100%

B.A. / BSc Semester 1

Title of the Course: OE 1.1.2 Fundamentals of Remote Sensing

Code: GEOGOE T1.1.2

Number of Theory Credits	Number of lecture hours/ semester	
3	42 - 45	
Course Outcomes: <ol style="list-style-type: none">1. This course is to make understand the basic concepts of Remote Sensing and to impart necessary skills of remote sensing analysis, and image interpretation to the students. So that, students acquire employable skills in remote sensing.2. Students will learn how to handle and process the satellite images for understanding of biophysical phenomena of the earth system.		
Course Objectives: <ol style="list-style-type: none">1. To congregate the basic concepts and fundamentals of physical principles of remote sensing2. To create a firm basis for successful integration of remote sensing in any field of application.3. To study basics of digital image processing and image interpretation techniques.4. To study the applications of the remote sensing to solve the real-world problems.		
Content of Theory Course 1		52/56Hrs
Unit – 1 Introduction		11
Definition of Remote Sensing, developmental stages, Laws of Physics, electromagnetic waves, spectrum, regions, wavelength, frequencies, and applications. Types-Satellites, Sensors, Payloads, Orbits, telemetry of satellites.		
Process and types of Remote Sensing		11
Process of remote sensing, interaction of radiation with atmosphere and targets, atmospheric noises, attenuation in radiance, resolutions of remote sensing, optical remote sensing, visible region of the spectrum, thermal remote sensing, microwave remote sensing, Hyperspectral remote sensing, LiDAR, and other remote sensing Platforms.		

Unit – 3 Image Classification and Interpretation	11
Satellite products and its spectral characteristics, composite images, band ratios; Land use land cover classification schemes-Anderson and NRSC; Visual image interpretation, elements, stages of interpretation and interpretation keys. Image classification- supervised, unsupervised, and principal component analysis (PCA) and accuracy assessment.	
Unit – 4 Applications of Remote Sensing	12
Disaster Management, Meteorological Studies, Agricultural and Irrigation Studies, Forestry Studies, Hydrological Studies, Natural Resource, Oceanic and Coastal mapping, Soil resource mapping, Urban and Rural Mapping and Management.	

References

1. Image processing and GIS for remote sensing: techniques and applications; Second Edition (2016) - Liu, Jian-Guo, Mason, Philippa J
 2. Introduction to Remote Sensing and Image Interpretation (2003); Lillesand T.M.
 3. Introduction to Remote Sensing, Fifth Edition (2011); James B. Campbell, Randolph H.Wynne
 4. Introductory Digital Image Processing: A Remote Sensing Perspective, Fourth Edition(2015) - John R. Jensen
 5. Practical handbook of remote sensing, First Edition (2016) - Lavender, Andrew, Lavender, Samantha
 6. Remote Sensing and GIS, Second Edition (2011), Bhatta, B.
 7. Remote sensing and image interpretation (2015); Chipman, Jonathan W., Kiefer, Ralph W., Lillesand
 8. Remote Sensing of the Environment: An Earth Resource Perspective (Prentice HallSeries in Geographic Information Science) - Second Edition (2006), John Jensen
1. https://onlinecourses.nptel.ac.in/noc19_ce41/preview

Pedagogy

Formative Assessment	
Assessment Occasion/type	Weightage in Marks
Quiz	30%
Assignment	20%
CIA	50%
Total	100%

B.A. / BSc Semester 2

Title of the Course: Introduction to

Climatology CODE: GEOGDSC T2.1

Number of Theory Credits	Number of lecture hours/ semester	Number of practical Credits	Number of practical hours/ semesters
4	52 or 56	2	52 or 56
<p>Course Outcomes:</p> <p>After the completion of this course, students should be able to</p> <ol style="list-style-type: none"> 1. define the field of climatology and to understand the atmospheric composition and structure. 2. to outline the mechanism and process of solar radiation transfer to earth surface and to explain the temperature distribution and variation according to time and space. 3. to illustrate and explain the air pressure system, wind regulating forces and the formation of the Atmospheric Disturbance. 4. to understand and compute the air humidity as well as to explain the process of Condensation and formation of precipitation and its types. 			
<p>Course Objectives:</p> <p>This course aims to:</p> <ol style="list-style-type: none"> 1. to define the field of climatology and components of the climate system 2. to introduce various dimensions of climatology like structure and composition. 3. to understand the global atmospheric pressure, temperature, and wind system. 4. to study the concept of atmospheric moisture and its types 			
Content of Theory Course 1			52/56Hrs
Unit – 1 Composition and Structure of the Atmosphere			14
Nature and Scope of Climatology, Atmospheric Sciences; Climatology and Meteorology Origin and structure of the Atmosphere: Troposphere, Stratosphere, Mesosphere, Ionosphere, Exosphere and their characteristics. Composition of the atmosphere Weather and Climate			
Unit – 2 Atmospheric Temperature			14
Insolation: Definition, Mechanism, Solar Constant. Factors affecting the Insolation: Angle of incidence, length of the day, Sunspots, Distance between the earth and the sun, effect of the atmosphere. Heating and cooling process of the atmosphere-Radiation, Conduction, convection, and advection. Temperature: meaning and Influencing Factors on the Distribution of Temperature Distribution of the temperature: Vertical, Horizontal, and Inversion of temperature. Global Energy Budget: Incoming shortwave solar radiation, Outgoing Longwave Terrestrial radiation, Albedo. Net Radiation and Latitudinal Heat Balances.			
Unit – 3 Atmospheric Pressure and Winds			14

Atmospheric Pressure: Influencing factors on atmospheric pressure. Vertical and Horizontal Distribution of the atmospheric pressure and Pressure Belts, Pressure Gradient. Tri-cellular-Hadley, Ferrel's and Polar Cells. Winds: influencing factors, Types - planetary, seasonal, local wind Variable winds-Cyclones and anti-cyclones. Air-Masses and Fronts: Definition, Nature, Source Regions, Classification.	
Unit – 4 Atmospheric Moisture	14
Humidity: Sources, influencing factors and types-Absolute, Relative and Specific. Hydrological cycle: process of evaporation, condensation. Clouds and its types Precipitation and its forms. Climate Change: Causes and consequences, recent issues-floods, drought,	

References

1. Lal, D. S. (1998). Climatology. Allahabad: Chaitanya Publishing House.
2. P Mallappa, Physical Geography (Kannada Version)
3. Ranganath Principles of Physical Geography (Kannada Version)
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15. Trewartha, G.T. (1980): An Introduction to Climate; McGraw Hill, New York, 5th edition, (International Student Edition)

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Pedagogy

Formative Assessment	
Assessment Occasion/type	Weightage in Marks
Quiz	30%
Assignment	20%
CIA	50%
Total	100%

GEOGDSC P2.1- Climatology Practical CREDITS: 2

Content of Practical Course 1: List of Experiments to be conducted

Conduct all exercises with Goal, Procedure, devices, and findings.

Exercise 1: Understanding Structure and functions of the Indian Meteorological Department (IMD).

Exercise 2: Collection of climatic data from IMD website.

<https://mausam.imd.gov.in/bengaluru/>

Exercise 3: Plotting of downloaded climatic data using graphical methods-

Elementary Instrumental Observation:

Exercise 4: Centigrade and Fahrenheit thermometer for measuring temperature.

Exercise 5: Mercurial Barometer and Aneroid Barometer for measuring atmospheric pressure

Exercise 6: Wind Vane and cup-anemometer.

Exercise 7: Wet and Dry bulb thermometer for measuring humidity

Exercise 8: Rain gauge- Dial type for measuring rainfall Exercise 3: Rainfall Trend Analysis.

Exercise 9: Interpretation of Indian Daily Weather charts.

Exercise 10: Deriving water balance chart, Actual and potential evapotranspiration

Note: Students are expected to download weather charts of the four seasons.

B.A. / BSc Semester 2

Title of the Course: 1 Introduction to Human Geography
CODE: GEOGOE T2.1.1

Number of Theory Credits	Number of lecture hours/ semester
3	42 - 45
<p>Course Outcomes:</p> <ol style="list-style-type: none"> 1. Students will learn how human, physical, and environmental components of the world interact. 2. Students will be familiarized with economic processes such as globalization, trade and their impacts on economic, cultural and social activities. 3. The student will describe what geography and human geography are. 4. Understand population dynamics and migration. 	
<p>Course Objectives: This course aims to</p> <ol style="list-style-type: none"> 1. Understand the basic concepts of human geography 2. Study population attributes and dynamic nature of it 3. Introduce economic, cultural, and trade activities and their impact on the development of the region 	
Content of Theory Course 1	
45Hrs	
Unit – 1 Introduction to Human Geography	
11	
Nature and scope, Development Environmental Determinism and Possibilism, Neo determinism (stop and go determinism) Approaches to human geography: Exploration and Descriptive approach, regional analysis Approach, Areal Differentiation Approach, Spatial organization Approach. Modern approaches: Welfare or Humanistic Approach, Radical Approach, Behavioral Approach, Post Modernism in geography Fields and sub fields in Human geography	
Unit – 2 Geographical Analysis of Population	
11	
Distribution and Growth of Population Density of population: meaning and Types: Arithmetic Density and Physiological Density. Regional distribution of Density of Population. Population Movement: Migration, Ravenstein's Law of Migration, Factors of population Migration, Economic Push and Pull factors, Cultural Push and Pull Factors, Environmental Push and Pull Factors. Migration Types: Immigration and Emigration, Internal and International Migration	
Unit – 3 Cultural Patterns and Processes	
11	
Concept of Culture, Material and Non material culture Cultural Regions, cultural Traits and Complexes, cultural Hearths, cultural Diffusion. Languages of the World: Types, Classification and Distribution. Religions: Types and Classification. Distribution. Universalizing Religions: Christianity, Islam, Buddhism. Ethnic Religions: Hinduism, the Chinese religion, Shintoism, Judaism. The Major tribal population of the world.	

Unit – 4 Human Economic Activities, Development and Settlements	12
<p>Primary Economic Activities – Agriculture, Types: Primitive Subsistence, Intensive subsistence, Plantation Agriculture, Extensive Commercial grain cultivation, Mixed Farming, Dairy Farming</p> <p>Secondary Activities: Manufacturing, classification – based on size – Small Scale and Large scale. Based on Raw material – Argo-based, Mineral based, Chemical Based and Forest based. Industrial Regions of the world.</p> <p>Tertiary Activities: Types: Trade and commerce, Retail Trading services, Wholesale trading. Transport and communications: Factors, communication services – Telecommunication. Services: Informal and Non formal sector. Information technology and service.</p> <p>Human Settlements: Factors, Classification, Types and Patterns: Rural, Urban. Compact or Nucleated and Dispersed settlements. Rural settlement Patterns: linear, rectangular, circular, star shaped, T shaped.</p>	

References

1. Hartshorne, T. A., & Alexander, J. W. (2010). Economic Geography. New Delhi: PHI Learning.
2. Knox, P., Agnew, J., & McCarthy, L. (2008). The Geography of the World Economy. London: Hodder Arnold.
3. Lloyd, P., & Dicken, B. (1972). Location in Space: A Theoretical Approach to Economic Geography. New York: Harper and Row.
4. Siddhartha, K. (2000). Economic Geography: Theories, Process and Patterns, New Delhi: Kosalaya Publications.
5. Smith, D. M. (1971). Industrial Location: An Economic Geographical Analysis, New York: John Wiley and Sons.

Pedagogy

Formative Assessment	
Assessment Occasion/type	Weightage in Marks
Quiz	30%
Assignment	20%
CIA	50%
Total	100%

B.A. / BSc Semester 2

Title of the Course: 2. Basics of Geographic Information Systems (GIS)

CODE: GEOGOE T2.1.2

Number of Theory Credits	Number of lecture hours/ semester
3	39 or 42
<p>Course Outcomes:</p> <ol style="list-style-type: none"> 1. Students are trained to adapt the theoretical concepts in a practical way through the mathematical models of geography. 2. Students will have the hands-on training on various modes of spatial and non-spatial data collection, data storage, data analytics, data interpretation and data display through the thematic maps. 3. Students are exposed on spatial thinking to solve the geographical problems with range of proven mathematical and statistical models. 4. Students can employ in various corporate and government organisation where they deal to solve geographical problems. 	
<p>Course Objectives: This course aims to:</p> <ol style="list-style-type: none"> 1. Understand the concept and techniques of the Geographic Information Systems. 2. Define the GIS data types and structures. 3. Study geo processing and visualization concepts and techniques in GIS. 	
Content of Theory Course 1	
52/56Hrs	
Unit – 1 Introduction	
10	
Emergence of GI Science, Milestone and Developmental stages in GIS, Definition, scope, role of GIS in digital world; Components, functionalities, merits and demerits, global market, interdisciplinary domains, and its integration with GIS.	
Unit – 2 Geodesy and Spatial Mathematics	
10	
Cartesian coordinates, latitude, longitudes, formats of angular units, geographical coordinates, Datum: WGS84, vs NAD32. UTM, Aerial Distance measurement using Geographic and projected coordinates, Area, Perimeter, length by coordinates and various international measures.	
Unit – 3 GIS Data and Scale	
10	
Spatial Data and its structures; sources and types of data collection; data errors, topology of data and relationship. Large Scale vs Small Scale, generalization; precision and accuracy of data-logical consistency and non-spatial data integration	
Unit – 4 Geoprocessing and Visualization	
12	
Spatial and Non-Spatial Queries, proximity analysis, Preparation of Terrain and Surface models. Hotspot and density mapping. Types of maps, thematic maps and Its types, relief maps, flow maps and cartograms. Tabulations: Graphs and Pivottables	

References

1. An Introduction to Geographical Information Systems - Ian Heywood (2011)
2. Geographic Information Systems and Cartographic Modelling - Tomlin, C.D. (1990)
3. Geographic Information Systems and Environmental Modelling - Clarke, C., K. (2002)
4. Geographic [Information Systems](#) and Science - Paul A. Longley, et. al. (2015)
5. Geographic Information Systems: A Management Perspective - Aronoff, S. (1989)
6. GIS - Fundamentals, Applications, and Implementations - Elangovan, K. (2006)
7. Introduction to Geographical Information Systems - Chang, Kang-Tsung (2015)
8. Mathematical Modeling in Geographical Information System, Global Positioning System and Digital Cartography - Sharma, H.S. (2006)
9. Remote Sensing and GIS - Bhatta, B. (2011)
10. Spatial analysis and Location-Allocation Models - Ghosh, A. and G. Rushton (1987)

Reference Websites

1. IIRS MOOC programme: <https://isat.iirs.gov.in/mooc.php>
2. ITC Netherlands, Principles of GIS
https://webapps.itc.utwente.nl/librarywww/papers_2009/general/principlesgis.pdf
3. Geographical Information Systems: Principles, Techniques, Management and Applications https://www.geos.ed.ac.uk/~gisteac/gis_book_abridged/

Pedagogy

Formative Assessment	
Assessment Occasion/type	Weightage in Marks
Quiz	30%
Assignment	20%
CIA	50%
Total	100%

CBCS SCHEME

B.A III Semester Core Course (Paper – III)

(For Students admitted in 2017-18 and onwards)

Teaching hours: 8hrs (theory 4hrs + Practical 4hrs)

Credit 4:0:2 Total 6 credits

60 Hrs lectures

3. General Cartography

1. Evolution of cartography – Traditional and Digital	06
3. Maps – Types, Elements and Uses	09
4. Map Scale – Types and Application, Reading Distances on a Map.	10
5. Map projection- Meaning, Types, Importance and uses	15
6. Representation of Data – Symbols, Dots, Choropleth, Isopleth and Flow Diagrams, Interpretation of Thematic Maps.	20

Note: This paper is not a practical paper, and the objective is to give basic information about various tools and techniques used in making maps. Students will not be involved in any laboratory work or hands on exercises, though a few demonstrations in the laboratories by teachers are recommended.

Reading List

1. Dent B. D., 1999: *Cartography: Thematic Map Design*, (Vol. 1), McGraw Hill.
2. Gupta K. K and Tyagi V. C., 1992: *Working with Maps*, Survey of India, DST, New Delhi.
3. Mishra R. P. and Ramesh A., 1989: *Fundamentals of Cartography*, Concept Publishing.
4. Robinson A., 1953: *Elements of Cartography*, John Wiley.
5. Sharma J. P., 2010: *Prayogic Bhugol*, Rastogi Publishers.
6. Singh R. L. and Singh R. P. B., 1999: *Elements of Practical Geography*, Kalyani Publishers
7. Singh R. L., 1998: *Prayogic Bhoogol Rooprekha*, Kalyani Publications.
8. Steers J. A., 1965: *An Introduction to the Study of Map Projections*, University of London.

III SEMESTER
Practical Paper – III
MAP PROJECTION

Unit	Topic	Total teaching hours: 60
Unit I	Cylindrical Map Projections: 1. Simple cylindrical projection 2. Cylindrical Equal- area projection 3. Mercator's projection	25
Unit II	Conical Map projections 1. Simple Conical projection 2. Bonne's projection 3. Polyconic projection	10
Unit III	Zenithal map projections (Polar Case) 1. Zenithal Equal -distant. 2. Zenithal Equal – area 3. Zenithal Gnomonic 4. Zenithal Stereographic	25

Note: The above map projections should be constructed with exercises, properties and uses.

References:

1. Salar Massod. M. : Map Projections, Roa and Raghavam Co., Mysore.
2. Ranganath & Mallappa : Map Projections (kan version), Chetana Book House, Mysore.
3. Erwin Raisz : General Cartography; Mc Graw- Hill book company Inc.
4. Singh R L : Elements of Practical Geography, Student's Friends, Allahbad.
5. George P Kellaway : Methuen & Co. , Ltd., London.
6. Gopal Singh : Mapwork & Practical Geography, Surjeet Book Depot, New Dehli.

Teaching hours : 8hrs (theory 4hrs + Practical 4hrs)

Credit 4:0:2 Total 6 credits

Environmental Geography

60 Hrs lectures

1. Meaning and importance of Environmental Geography, 15
Concepts and Approaches; Ecosystem – Structure, Functions.
2. Human-Environment Relationship in Equatorial, Desert, Mountain and 15
Coastal Regions.
3. Environmental Problems and Management: Air Pollution; Biodiversity 16
Loss; Solid and Liquid Waste.
4. Environmental Programmes and Policies: Developed Countries; 14
Developing Countries, with special reference to India.

Reading List

1. Casper J.K. (2010) Changing Ecosystems: Effects of Global Warming. Infobase Pub. New York.
2. Hudson, T. (2011) Living with Earth: An Introduction to Environmental Geology, PHI Learning Private Limited, New Delhi.
3. Miller, G.T. (2007) Living in the Environment: Principles, Connections, and Solutions, Brooks/ Cole Cengage Learning, Belmont.
4. Singh, R.B. (1993) Environmental Geography, Heritage Publishers, New Delhi.
5. UNEP (2007) Global Environment Outlook: GEO4: Environment For Development, United Nations Environment Programme. University Press, Cambridge.
6. Wright R. T. and Boorse, D. F. (2010) Toward a Sustainable Future, PHI Learning Pvt Ltd, New Delhi.
7. Singh, R.B. and Hietala, R. (Eds.) (2014) Livelihood security in Northwestern Himalaya: Case studies from changing socio-economic environments in Himachal Pradesh, India. Advances in Geographical and Environmental Studies, Springer
8. Singh, Savindra 2001. *Paryavaran Bhugol*, Prayag Pustak Bhawan, Allahabad. (in Hindi)
9. L.T Nayak – Environmental Geography(in Kannada)
10. Dr.Ranganath - Environmental Studies(in Kannada)

**APPLICATIONS OF STATISTICAL
METHODS IN GEOGRAPHY**

Unit	Topic	Total teaching hours: 60
1.	a). Methods of data collection – Primary and secondary sources, census and sampling methods. b). Measures of Central Tendency: Direct and shortcut Methods- a)Arithmetic mean b) Median c) Mode (Grouping and formula)	35
2.	a). Measures of dispersion – Mean and standard deviations b). Measures of association correlation c). Time series analysis	25

References:

1. Singh R. L. : Elements of Practical Geography
2. Gopal Singh : Map Work and Practical Geography
3. Misra R. P. : Fundamentals of Cartography
4. Zamir Alvi : Statistical Geography, Methods and Applications.
5. Aslam Mahmood : Statistical Methods in Geography.
6. Ashis Sarkar : Practical Geography, Orient Longman, Kolkata.
7. Dr. C K Renukarya : Basic statistics (Kan & Eng Version)

Discipline Specific Elective Papers (2 Compulsory Papers)

B.A V Semester DSE-I (Paper – V)

(For Students admitted in 2017-18 and onwards)

Teaching hours theory 4hrs Practical 4hrs

Credit 4:0:2 Total 6 credits

1. Geography of India

- | | |
|--|----|
| 1. Physical Setting – Location, Structure and Relief, Drainage, Climate. | 14 |
| 2. Population – Size and Growth since 1901, Population Density and Distribution, Literacy, Sex Ratio. | 10 |
| 3. Settlement System – Types – Rural and urban, Patterns of Settlements | 10 |
| 4. Resource Base –Livestock (cattle & fisheries),Power (Coal,& hydroelectricity) Minerals (iron ore and bauxite). | 10 |
| 5. Economy – Agriculture (Rice, Wheat, Sugarcane, Groundnut, Cotton); Industries (Cotton Textile, Iron-Steel, Automobile), Transportation Modes (Road and Rail). | 16 |

Reading List

1. Hussain M., 1992: *Geography of India*, Tata McGraw Hill Education.
2. Mamoria C. B., 1980: *Economic and Commercial Geography of India*, Shiva Lal Agarwala.
3. Miller F. P., Vandome A. F. and McBrewster J., 2009: *Geography of India: Indo- Gangetic Plain, Thar Desert, Major Rivers of India, Climate of India, Geology of India*, Alphascript Publishing.
2. Nag P. and Sengupta S., 1992: *Geography of India*, Concept Publishing.
3. Pichamuthu C. S., 1967: *Physical Geography of India*, National Book Trust.
4. Sharma T. C. and Coutinho O., 1997: *Economic and Commercial Geography of India*, Vikas Publishing.
7. Singh Gopal, 1976: *A Geography of India*, Atma Ram.
8. Spate O. H. K. and Learmonth A. T. A., 1967: *India and Pakistan: A General and Regional Geography*, Methuen.
9. Rana, Tejbir Singh, 2015, *Diversity of India*, R.K. Books, Delhi.

V SEMESTER
Practical Paper – V
FUNDAMENTALS OF G.I.S

Unit	Topic	Total teaching hours: 60
Unit 1	a) Meaning, definitions, components and importance of GIS b) Spatial entities – Point, line and polygon Sources of spatial data- Census, Topographical Maps, Aerial Photographs and Satellite Imageries	20
Unit 2	a) Spatial Data Structure - Raster and vector data Structures Linking spatial and non spatial data b) Introduction to MapInfo software	20
Unit 3	a) Geo – referencing , Choice of map projection b) Digitization c) Attaching attribute data (Creating data base) d) Editing e) Map layout f) Thematic map	20

References:

1. Burrough P.A. : Geographical Information Systems for Land Resources
2. Maguire D. J. : Computers in Geography
3. Star J. C and J.E. : Geographic Information Systems
4. Internet : GIS. Development
5. Heywood : Introduction to GIS, 2002.
6. Mahesh : Introduction to GSI Shivalingappa Chandrashekar

Discipline Specific Elective Papers (2 Compulsory Papers)

B.A VI Semester DSE-II (Paper – VII)

(For Students admitted in 2017-18 and onwards)

Teaching hours theory 4hrs Practical 4hrs

Credit 4:0:2 Total 6 credits

4. Geography of Tourism

1. Concepts, Nature and Scope; Inter-Relationships of Tourism, Recreation and Leisure; Geographical Parameters of Tourism by Robinson. 12
2. Type of Tourism: Nature Tourism, Cultural Tourism, Medical Tourism, Pilgrimage 14
3. Recent Trends of Tourism: International and Regional; Domestic (India); Eco-Tourism, Sustainable Tourism, Meetings, Incentives, Conventions and Exhibitions (MICE), Carrying capacity of Tourism 14
4. Impact of Tourism: Economy; Environment; Society 10
5. Tourism in India: Tourism Infrastructure; Case Studies of Himalaya, Desert and Coastal and Heritage; National Tourism Policy 10

Reading List

1. Dhar, P.N. (2006) International Tourism: Emerging Challenges and Future Prospects. Kanishka, New Delhi.
2. Hall, M. and Stephen, P. (2006) Geography of Tourism and Recreation – Environment, Place and Space, Routledge, London.
3. Kamra, K. K. and Chand, M. (2007) Basics of Tourism: Theory, Operation and Practise, Kanishka Publishers, Pune.
4. Page, S. J. (2011) Tourism Management: An Introduction, Butterworth- Heinemann- USA. Chapter 2.
5. Raj, R. and Nigel, D. (2007) Morpeth Religious Tourism and Pilgrimage Festivals Management: An International perspective by, CABI, Cambridge, USA, www.cabi.org.
6. Tourism Recreation and Research Journal, Center for Tourism Research and Development, Lucknow
5. Singh Jagbir (2014) “Eco-Tourism” Published by - I.K. International Pvt. Ltd. S-25, Green Park Extension, Uphaar Cinema Market, New Delhi, India(www.ikbooks.com).

**VI SEMESTER
Practical Paper – VI
COMPUTER MAPPING AND GPS SURVEYING**

Unit	Topic	Total teaching hours: 60
1.	Introduction to Computer : Generation of Computers, Hardware and Software Components	20
2.	Computer graphics : Creating Data base in computer, creation of Line, Bar and Pie diagrams. Thematic Maps - Choro chromatic and schematic Maps	20
3.	GPS Surveying: Concepts, Segments and applications, plotting way points by using map source software .	20
4.	Tour report / Factory visit	

References:

1. Singh L.R. : Fundamentals of Practical Geography, Sharadha Pustaka Bhavan, Alahabad, 2006
2. Dr. M.A. Siddaqui : Introduction to Geographical Information System, Sharadha Pustaka Bhavan, Alahabad, 2006
3. Chang : Introduction to GIS, Tata McGraw Hill W, New Delhi.

B.A V Semester SEC –I (Paper – VI)
(For Students admitted in 2017-18 and onwards)
Teaching hours theory 4hrs
Credit 4:0:0 Total 4 credits

1. Regional Planning and Development

- | | |
|--|----|
| 1. Concept, Need and Types of Regional Planning. | 8 |
| 2. Characteristics and Delineation of Planning Region. | 10 |
| 3. Regionalization of India for Planning (Agro Ecological Zones). | 12 |
| 4. Models for Regional Planning: Growth Pole Theory; Core Periphery Model and Growth Foci Concept in Indian Context. | 15 |
| 5. Backward Regions and Regional Plans- Special Area Development Plans in India; DVC-The Success Story and the Failures; NITI Aayog. | 15 |

Reading List

1. Blij H. J. De, 1971: *Geography: Regions and Concepts*, John Wiley and Sons.
2. Claval P.I, 1998: *An Introduction to Regional Geography*, Blackwell Publishers, Oxford and Massachusetts.
3. Friedmann J. and Alonso W. (1975): *Regional Policy - Readings in Theory and Applications*, MIT Press, Massachusetts.
4. Gore C. G., 1984: *Regions in Question: Space, Development Theory and Regional Policy*, Methuen, London.
5. Gore C. G., Köhler G., Reich U-P. and Ziesemer T., 1996: *Questioning Development; Essays on the Theory, Policies and Practice of Development Intervention*, Metropolis- Verlag, Marburg.
6. Haynes J., 2008: *Development Studies*, Polity Short Introduction Series.
7. Johnson E. A. J., 1970: *The Organization of Space in Developing Countries*, MIT Press, Massachusetts.
8. Peet R., 1999: *Theories of Development*, The Guilford Press, New York.
9. UNDP 2001-04: *Human Development Report*, Oxford University Press.
10. World Bank 2001-05: *World Development Report*, Oxford University Press, New

Skill Enhancement Course (2 Compulsory Papers)

B.A VI Semester SEC- II (Paper – VIII)

(For Students admitted in 2017-18 and onwards)

Teaching hours theory 4hrs

Credit 4:0:0 Total 4 credits

4. Field Techniques and Survey based Project Report

1. Field Work in Geographical Studies – Role, Value and Ethics of Field-Work. 10
2. Defining the Field and Identifying the Case Study – Rural /Urban /Physical /Human /Environmental. 10
3. Field Techniques – Merits, Demerits and Selection of the Appropriate Technique; Observation (Participant / Non Participant). 12
4. Questionnaires (Open/ Closed / Structured / Non-Structured); Interview with Special Focus on Focused Group Discussions; Space Survey (Transects and Quadrants, Constructing a Sketch). 16
5. Designing the Field Report – Aims and Objectives, Methodology, Analysis, Interpretation and Writing the Report. 12

Practical Record

1. Each student will prepare an individual report based on primary and secondary data collected during field work.
2. The duration of the field work should not exceed 10 days.
3. The word count of the report should be about **8000 to 12,000** excluding figures, tables, photographs, maps, references and appendices.
4. One copy of the report on A 4 size paper should be submitted in soft binding.

Reading List

1. Creswell J., 1994: *Research Design: Qualitative and Quantitative Approaches* Sage Publications.
2. Dikshit, R. D. 2003. *The Art and Science of Geography: Integrated Readings*. Prentice-Hall of India, New Delhi.
3. Evans M., 1988: "Participant Observation: The Researcher as Research Tool" in *Qualitative Methods in Human Geography*, eds. J. Eyles and D. Smith, Polity.
4. Mukherjee, Neela 1993. *Participatory Rural Appraisal: Methodology and Application*. Concept Publs. Co., New Delhi.
5. Mukherjee, Neela 2002. *Participatory Learning and Action: with 100 Field Methods*. Concept Publs. Co., New Delhi
6. Robinson A., 1998: "*Thinking Straight and Writing That Way*", in *Writing Empirical Research Reports: A Basic Guide for Students of the Social and Behavioural Sciences*, eds. by F. Pryczak and R. Bruce Pryczak, Publishing: Los Angeles.
7. Special Issue on "Doing Fieldwork" *The Geographical Review* 91:1-2 (2001).
8. Stoddard R. H., 1982: *Field Techniques and Research Methods in Geography*, Kendall/Hunt.
9. Wolcott, H. 1995. *The Art of Fieldwork*. Alta Mira Press, Walnut Creek, CA.

DEPARTMENT OF GEOGRAPHY
MODEL QUESTION PAPER FOR CBCS SCHEME
B.A GEOGRAPHY
(For I,II,III,IV,V AND VI semesters)

Time : 3 Hours

Max. Marks : 70

Part-A

I. Answer any five of the following questions. 5x2=10

- 1).....
- 2).....
- 3).....
- 4).....
- 5).....
- 6).....
- 7).....

Part-B

II. Answer any Six of the following questions. 6x5=30

- 8).....
- 9).....
- 10).....
- 11).....
- 12).....
- 13).....
- 14).....
- 15).....

Part –C

III. Answer any Three of the following questions. 3x10=30

- 16).....
- 17).....
- 18).....
- 19).....
- 20).....

JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE, OOTY ROAD, MYSURU-25
(AUTONOMOUS)



DEPARTMENT OF HINDI

LANGUAGE - HINDI

SYLLABUS

CBCS

2017-2020

SCHEME OF STUDY & ASSESSMENT

**B-COM/BBA/BCA
CBCS**

Programme: Bachelor of Commerce

Programme: Bachelor of Business Administration

Programme: Bachelor of Computer Application

Semester	Course Title	Course code	No of credits L: T: P:	Assessment				Duration of Exam
				C1	C 2	C3	TO	
I	Hindi Kahani aur Vyakarna	DNA 05001	2 : 1 : 0	15	15	70	100	03
		BDA 05001						
		DCA 05001						
II	Hindi Gadya aur Vyakarna	DNB 05001	2 : 1 : 0	15	15	70	100	03
		BDB 05001						
		DCB 05001						
III	Hindi Kavya aur Anuvada Paribhashik Shabdavali	DNC 05001	2 : 1 : 0	15	15	70	100	03
		BDC05001						
		DCC 050001						
IV	Hindi Upanyas Tatha Vanijya Hindi	DND 05001	2 : 1 : 0	15	15	70	100	03
		BDD05001						
		DCD 05001						

B-COM
CBCS AE

Semester	Course Title	Course code	No of credits L: T: P	Assessment				Duration of Exam
				C1	C 2	C3	TO	
III	<u>Communicative Hindi</u>	DNC 30001	2 : 0 : 0	15	15	50	80	02

Programme Outcome for Bachelor of Commerce, Bachelor of Business Administration, Bachelor of Computer Applications

After the completion of the graduation students are:

PO1: Motivated for higher education.

PO2: Able to write resume, application and business letters

PO3: Able to write and communicate effectively.

Programme Specific Outcome for Bachelor of Commerce, Bachelor of Business Administration, Bachelor of Computer Applications

After the completion of the graduation students are:

PSO1: Able to write and communicate effectively.

PSO2: Able to avail job opportunity in translation.

DNA 05001 /BDA 05001 /DCA 05001

**I Semester
Language Hindi
Kahani aur Vyakarna**

60hrs (4hrs/week)

3 Credits

Course out come

After successful completion of this course students are able to

CO1. Deliberate in detail, short stories of 20th century

CO2. Deliberate in detail, Bade bhahe Sahab by Premchand

CO3. Understand the classification and characteristics of Akasha deep by Jayashankar Prasad

CO4. Understand with application, details of Hindi vyakaran

UNIT A & B Saat Shrest Kahaniyan

1. Bade bhai Saheb
2. Aakasha Beep
3. Chif ke Davath
4. Sabachar ka Thavej
5. Jalathe Huha Daine
6. Fas ke edar udar
7. Jahar

UNIT C Hindi Vyakaran

Portions Prescribed:

- 1) Vyakaran ka samanya parichay
- 2) Varnamala : Swaron ka vargeekaran, vyanjanon ka vargeekaran
- 3) Shabdabhed; a) Vyuthpatthi ki dristi se
b) Uthpatthi ki drishti se
c) Roopantar ki drishti se
- 4) Sandhi : Paribhasha aur bhed
- 5) Sangya : Paribhasha aur bhed
- 6) Sangya ke roopantar:
a) Ling - Paribhasha aur bhed – Ling Nirnay ke niyam
Ling parivarthan ke niyam
b) Vachan – Paribhasha aur bhed – Vachan parivarthan ke niyam
c) Karak – Paribhasha aur bhed
- 7) Sarvanaam : Paribhasha aur bhed
- 8) Visheshan – Paribhasha aur bhed

Reference

- 1) Saral Hindi Vyakaran –Pb. Prabhath Prakashan, # 4/19, Asaf Ali road, New Delhi-02
- 2) Parishkrith Hindi Vyakaran - Badarinath Kapur
- 3) Abhinav Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
- 4) Shikshaarthe Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
- 5) Sankshipta Hindi Vyakaran - Kamatha prasad Guru, Nagari Pracharini Sabha, kaashi.
- 6) Vyavaharik Hindi Vyakaran Tatha Rachana –Dr. Hardev Bahari, Lokabharathi prakashan, Allahabad -1

DNB 05001 /BDB050 01/DCB05001

II Semester
Language Hindi
Hindi Gadya aur Vyakarna

60hrs (4hrs/week)

3 Credits

Course out come

After successful completion of this course students are able to

CO1. Understand with application, details of Hindi vyakaran

CO2. Understand the details of Prose of 20th cenyury

CO3. Identify the classification and characteristics of Hindi vyakaran

UNIT A&B Sahitya sapthak (prescribed Lessons – 1,2,4,,7,8,11,13)

Judge ka peissla

Vapasi

Irsha tu na gayi meri man se

Mai Narak se Bol Raha hoon

Adyaksha Mahodaya

Girimitiyonke Desh Mei

Jab maa Ro Padi

UNIT C Hindi Vyakaran :

- 1) Kriya-Paribhasha aur bhed ,‘NE–NiyaZm Kriya ke roopantar-Kaal, Vachya, Prayog , Arth
- 2) Kriya Visheshan - Paribhasha aur bhed
- 3) Sambandha boodhak avyay – Paribhasha aur bhed
- 4) Samucchaya boodhak avyay -- Paribhasha aur bhed
- 5) Vismayadi boodhak avyay --Paribhasha aur bhed
- 6) Upasarg aur Pratyay --Paribhasha aur bhed
- 7) Samas --Paribhasha aur bhed
- 8) Pad-parichay

Reference

- 1) Saral Hindi Vyakaran –Pb. Prabhath Prakashan, # 4/19, Asaf Ali road,New Delhi-02
- 2) Parishkrith Hindi Vyakaran - Badarinath Kapur
- 3) Abhinav Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
- 4) Shikshaarathi Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
- 5) Sankshipta Hindi Vyakaran - Kamatha prasad Guru, Nagari Pracharini Sabha, kaashi.
- 6) Vyavaharik Hindi Vyakaran Tatha Rachana –Dr. Hardev Bahari, Lokabharathi prakashan, Allahabad -1

DNC05001 /BDC05001 /DCC05001

III Semester
Language Hindi
Hindi Kavya aur Anuvada Paribhashik Shabdavali

60hrs (4hrs/week)

3 Credits

Course out come

After successful completion of this course students are able to

CO1. Deliberate the classification and characteristics of medieval and modern hindi kavya

CO2:. Understand the details of Kaber by saakhe

CO3: Identify the characteristics of Hemala by ramadhare simha dinakar

CO4: Learn in depth preyatham by suryakanta threepati nirala

CO5: Understand the characteristics of Hindi Anuvada

UNIT A&B- KAVYAGANGA (Prescribed Lessons 1,2,5,8,11,12,13,14,16,19)

Kaber

Thulse Das

Bihare Lal

Maithilisharan Gupta

Surya Kanth Tripathi Nirala

sumitranandan pant

Subhadra Kumari Chauhan

Ramkumar Verma

Ramdhari Simha Dinkar

Kuvempu

UNIT C -Anuvad tatha Paribhashik Shabdavali

Portions Prescribed :

- 1) Anuvad ki Paribhashayen , Adhunik jagat mein Anuvad ka mahatva
Anuvadak kee Yogyathayen , Anuvad kala hai yaa vignan ?
- 2) Anuvad ke prakar:
- 3) Shabdanuvad
- 4) Bhavanuvad
- 5) Kavyanuvad
- 6) Saranuvad
- 7) Angreji / Kannad se Hindi me Anuvad – vice versa (passages)
- 8) Paribhashik Shabdavali

Reference

- 1) Anuvad Vignan – Bholanath Thiwari Pb. Shabdaker, Delhi -110092
- 2) Anuvad Kala Kuch Vichar –Anuvad Prakash Khemani , S.Chand & co. Delhi.

- 3) Anuvad Siddhant aur Samasyayen- R.N. Srivastava and K.K.Goswami
- 4) Alok Prakashan , Delhi Anuvad Patrika Ank, Pb. Anuvad Sahitya Parishad , New Delhi

DND05001 /BDD05001 /DCD05001

**IV Semester
Language Hindi**

Hindi Upanyas Tatha Vanijya Hindi

60hrs (4hrs/wee

3 Credits

Course out come

After successful completion of this course students are able to

- CO1:** Learn in details Novel-Gaban by Premchand
CO2: Learn the classification and characteristics of Vanijya Hindi
CO3: Learn the application of Vanijya Hindi

UNIT A&B - Gaban(sankshith)

UNIT C - Vanijya Hindi

Portions Prescribed :

- 1) a. Patra Vyavahar ka Samanya Parichay tatha Mahatva
b. Patron ke prakar
- 2) Vyavasaik patra Vyavahar :
 - a. Bank Sambandhi Patra
 - b. Beema Sambandhi Patra
 - c. Pooch-Taach Sambandhi Patra
 - d. Shikayati Sambandhi Patra
 - e. Naukari Sambandhi Patra
 - f. Pari patra
 - g. Agency Sambandhi Patra
- 3) Alekhan - Uttam Alekhan ke gun
- 4) Sampreshan - uske Prakar
- 5) Sankshiptikaran - Sankshipt Lekhan ke Pradhan gun aur udaaharan

Reference

- 1) Vyavasaik Sampreshan –Dr. Anup Chandra Bhayani Pb. Rajpal &sons, New Delhi-6
- 2) Karyalay Alekhan aur Tippan – Karnatak Mahila Hindi Seva Samithi, Chamarajpet,Bangalore.
- 3) Prashasanik Hindi – Ed. Ramdarash Mishra and amswaroop Shastry
- 4) Vanijya Patra Vyavahar,Anuvad,Nibandh tatha Sankshiptikaran – A V Narti, Gnanoday Prakashan , Dharwad.

DNC 30001

**II B-COM
III Semester
Language Hindi**

Communicative Hindi

30hrs (2hrs/week)

2 Credits

Course out come

After successful completion of this course students are able to

CO1. Learn the details of Communicative Hindi

CO2. Identify in depth Communicative Hindi

CO3. Deliberate in depth Communicative Hindi

UNIT I: HINDI GRAMMAR

- 1) Parts of speech – A general outlook
- 2) Gender -
 - a) Ling pehchanne ke niyam
 - b) Ling parivarthan ke niyam
- 3) Synonymns and Antonymns
 - a)Definition
 - b)Difference

UNIT II : GENERAL COMMUNICATION AND PRACTICE

- 1) Conversation - in market, hotel , journey etc.
- 2) Conversation on health , education etc.
- 3) Practice.

Reference

1. Saral Hindi Vyakaran –Pb. Prabhath Prakashan, # 4/19, Asaf Ali road,New Delhi-02
2. Parishkrith Hindi Vyakaran - Badarinath Kapur
3. Abhinav Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
4. Shikshaarathi Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
5. Sankshipta Hindi Vyakaran - Kamatha prasad Guru, Nagari Pracharini Sabha, kaashi.
6. Anuvad Vignan - Bholanath Thiwari , Shabdakar , Delhi -92
7. Anuvad Kala – Kuch Vichar - Anandprakash Khemani , S.Chand & Co., Delhi

Anuvad siddhant aur Samasyayen

Model Question Paper

Language HINDI I ,II , & III SEMESTERS of B.COM / BBA / BCA

Time : 3 Hrs

Max. Marks : 70

- I) Total Main questions with internal choice on the prescribed text book 10 marks each
10 x 2 = 20
- II) One Annotation question – three out of five on the prescribed text book 05 marks
each 05 x 3
= 15
- III) Two short answer questions with internal choice on Vyakaran /prayojan moolak
Hindi / Anuvad as the case may be, 10 marks each 10 x 2 = 20
- IV) 15 objective type questions on Vyakaran /prayojan moolak Hindi / Anuvad as the
case may be, 01 mark each 01 x 15 = 15

MODEL QUESTION PAPER

- Qn.1. OR 01 x 10 = 10
- Qn.2.OR 01 x 10 = 10
- Qn. 3. Annotate three out of six 05 x 03 = 15
- Qn.4.OR..... 01 x 10 = 10
- Qn.5.OR..... 01 x 10 = 10
- Qn.6. Objective type questions 01 x 15 = 15

INTERNAL ASSESSMENT - 30 MARKS

III SEMESTER - PAPER III
Communicative Hindi

MODEL QUESTION PAPER

TIME : 2HOURS

MAX. MARKS : 50

Instructions: All sections are compulsory

UNIT-I: Hindi Grammar

I . Answer The Following Questions in a sentence or a ward. 1 X 10 = 10

1).....2)..... 3).....4).....5).....

6).....7).....8).....9).....10).....

II . Answer The Following Questions (Any four) 4 X 5 = 20

1).....2).....

3).....4).....

5).....6).....

UNIT II : GENERAL COMMUNICATION AND PRACTICE 2 X 10 = 20

I . Answer The Following Questions (any two)

1..... 2.....

3..... 4.....

INTERNAL ASSESSMENT - 30 MARKS ,F-1 -15,F-2-15

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SCHEME OF STUDY & ASSESSMENT
INSTRUCTION

BA CBCS

BAEG11/HP12/KG13/HE14/JE15

Semester	Course Title	Course code	No of credits L: T: P:	Assessment				Duration of Exam
				C1	C2	C3	TO	
I	Hindi Gadya aur Vyakarna	DLA 05011	2 : 1 : 0:	15	15	70	100	03
		DLA 05012						
		DLA 05013						
		DLA 05014						
		DLA 05015						
II	Hindi Kahani aur Vyakarna	DLB 05011	2 : 1 : 0:	15	15	70	100	03
		DLB 05012						
		DLB 05013						
		DLB 05014						
		DLB 05015						
III	Hindi Nataka aur Vanjya Hindi	DLC 05011	2 : 1 : 0:	15	15	70	100	03
		DLC 05012						
		DLC 05013						
		DLC 05014						
		DLC 05015						
IV	Hindi Kavya aur Anuvada Paribhashik Shabdavali	DLD 05011	2 : 1 : 0:	15	15	70	100	03
		DLD 05012						
		DLD 05013						
		DLD 05014						
		DLD 05015						

BA
CBCS AECC

Semester	Course Title	Course code	No of credits L: T: P:	Assessment				Duration of Exam
				C1	C2	C3	TO	
III	Communicative Hindi	DLC 30001	2 : 0 : 0:	15	15	50	80	02

SCHEME OF STUDY & ASSESSMENT
BA/BSc

Programme: BA-HE/HP/EG/KG/JE

Programme: BSc-PCM / PMCS / PMCM / PME /CZBT/BMBt /BBM / CBZ

Semester	Course Title	Course code	No of credits L: T: P	Assessment				Duration of Exam
				C1	C2	C3	TO	
I	Hindi Gadya aur Vyakarna	CMA 05001	2 : 1 : 0	15	15	70	100	03
		CMA 05002						
		CMA 05003						
		CMA 05004						
		CMA 05005						
		CMA 05006						
		CMA 05007						
		CMA 05008						
II	Hindi Kahani aur Vyakarna	CMB 05001	2 : 1 : 0	15	15	70	100	03
		CMB 05002						
		CMB 05003						
		CMB 05004						
		CMB 05005						
		CMB 05006						
		CMB 05007						
		CMB 05008						
III	Hindi Nataka aur Vanjya Hindi	CMC 05001	2 : 1 : 0	15	15	70	100	03
		CMC 05002						
		CMC 05003						
		CMC 05004						
		CMC 05005						
		CMC 05006						
		CMC 05007						
		CMC 05008						
IV	Hindi Kavya aur Anuvada Paribhashik Shabdavali	CMD 05001	2 : 1 : 0	15	15	70	100	03
		CMD 05002						
		CMD 05003						
		CMD 05004						
		CMD 05005						
		CMD 05006						
		CMD 05007						
		CMD 05008						

Programme Outcome for Bachelor of Arts

After the completion of the graduation student are able to:

PO1: Understand culture and heritage

PO2: Manage business affairs

PO3: Create interest in literature

PO4: Report and edit public events effectively

PO5: Develop reading, writing, communication and reasoning skills

Programme Specific Outcome for Bachelor of Arts

After the completion of the Graduation Student are able to:

PSO1: Understand basic knowledge of Hindi grammar

PSO2: Understand the History of short stories and their relevance to society at large

PSO3: Avail job opportunity in translation

**Programme Outcome for Bachelor of Science–PCM, PMCs, PMCm, PME,
CZBt, BMBt, BBM, CBZ**

Programme Outcome for Bachelor of Science

After the completion of the graduation student are able to:

PO1: Inculcate human values

PO2: Avail job opportunities in translation

PO3: Create interest in literature

Programme Specific Outcome for Bachelor of Science

After the completion of the graduation student are able to:

PSO1: Get information about the History ancient medieval and modern Hindi literature

PSO2: Develop research aptitude

Hindi Gadya aur Vyakarna

60hrs (4hrs/week)

3 Credits

Course outcome

After successful completion of this course students are able to

CO1: Identify in details with examples prose of 20th century

CO2: Write down in depth prose of 20th century

CO3: Deliberate in depth prose of 20th century

CO4: Specify the classification and characteristics of Hindi vyakaran

CO5: Identify the characteristics of Hindi vyakaran

UNIT-A & B: Sahitya sapthak (prescribed Lessons – 1,2,4,,7,8,11,13)

Judge ka peissla

Vapasi

Irsha tu na gayi meri man se

Mai Narak se Bol Raha hoon

Adyaksha Mahodaya

Girimitiyonke Desh Mei

Jab maa Ro Padi

UNIT C Hindi Vyakaran :Portions Prescribed :

- 1) Vyakaran ka samanya parichay
- 2) Varnamala : Swaron ka vargeekaran, vyanjanon ka vargeekaran
- 3) Shabdabhed; a) Vyuthpatthi ki dristi se
b) Uthpatthi ki drishti se
c) Roopantar ki drishti se
- 4) Sandhi : Paribhasha aur bhed
- 5) Sangya : Paribhasha aur bhed
- 6) Sangya ke roopantar:
 - a) Ling - Paribhasha aur bhed – Ling Nirnay ke niyam
Ling parivarthan ke niyam
 - b) Vachan – Paribhasha aur bhed – Vachan parivarthan ke niyam
 - c) Karak – Paribhasha aur bhed
- 7) Sarvanaam : Paribhasha aur bhed
- 8) Visheshan – Paribhasha aur bhed

Reference

1. Saral Hindi Vyakaran –Pb. Prabhath Prakashan, # 4/19, Asaf Ali road, New Delhi-02

2. Parishkrith Hindi Vyakaran - Badarinath Kapur
3. Abhinav Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
4. Shikshaarathi Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
5. Sankshipta Hindi Vyakaran - Kamatha prasad Guru, Nagari Pracharini Sabha, kaashi.
6. Vyavaharik Hindi Vyakaran Tatha Rachana –Dr. Hardev Bahari, Lokabharathi prakashan, Allahabad -1

DLB 05011 /12/13/14/15

CMB 05001/2/3/4/5/6/7/8

II Semester Hindi Kahani aur Vyakarna

60hrs (4hrs/week)

3 Credits

Course out comes

After successful completion of this course students are able to

CO1:. Write down the details of short stories of 20th century

CO2: Identify in depth short stores of 20th century

CO3: Identify in details with application, if applicable, short stores of 20th century

CO4: Identify the classification and characteristics of Hindi vyakaran

CO5: Write down the characteristics of Hindi vyakaran

UNIT A & B Hindi ke Prathinidi Kahaneeya (Prescribed Lessons 1,2,3,7 ,9 10,12)

Usane kaha thank .

Puraskar

Pus ke Rath

Chif ke saathl .Jaj ka peissla

Dopahar ka Bhojan

Akeli

Prayaschita

UNIT C Hindi Vyakaran

Portions prescribed:

- 1) Kriya - Paribhasha aur bhed , 'NE' -Niyam
Kriya ke roopantar - Kaal, Vachya, Prayog , Arth
- 2) Kriya Visheshan - Paribhasha aur bhed
- 3) Sambandha boodhak avyay – Paribhasha aur bhed
- 4) Samucchaya boodhak avyay -- Paribhasha aur bhed
- 5) Vismayadi boodhak avyay --Paribhasha aur bhed
- 6) Upasarg aur Pratyay --Paribhasha aur bhed
- 7) Samas --Paribhasha aur bhed
- 8) Pad-parichay

Reference

1. Saral Hindi Vyakaran –Pb. Prabhath Prakashan, # 4/19, Asaf Ali road,New Delhi-02
- 2 .Parishkrith Hindi Vyakaran - Badarinath Kapur
3. Abhinav Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
4. Shikshaarthe Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
5. Sankshipta Hindi Vyakaran - Kamatha prasad Guru, Nagari Pracharini Sabha, kaashi.

6 .Vyavaharik Hindi Vyakaran Tatha Rachana –Dr. Hardev Bahari, Lokabharathi prakashan,
Allaha

CMC 05001/2/3/4/5/6/7/8

III Semester

Hindi Nataka aur Vanjya Hindi

60hrs (4hrs/week)

3 Credits

Course outcome

After successful completion of this course students are able to

- CO1. Understand the characteristics of Hindi Natak
- CO2. Deliberate in details with application, if applicable, Hindi Natak -deep daan by Ramkumar varma
- CO3. Deliberate the characteristics of Hindi Natak -Red ke haddi by Jagadeshachandra mathur
- CO4. Understand the details of Hindi Natak -sukhe dale by Upendranath ashka
- CO5. Write down in details with examples Hindi Natak -mai bee manav hu by Vishnu prabakar
- CO6. Identify the details of Hindi Vanijya Hindi
- CO7. Specify in depth Vanijya Hindi

UNIT A & B -EKanki Sapthak (1 to 6)

Deep daan
Reed ke hadde
Sukhi daal
mai bhe Manav hu
Ande ke chillke
Bahu ke Bida

UNIT C Vanijya Hindi

Portions Prescribed :

- 1) a. Patra Vyavahar ka Samanya Parichay tatha Mahatva
b. Patron ke prakar
- 2) Vyavasaik patra Vyavahar :
 - a. Bank Sambandhi Patra
 - b. Beema Sambandhi Patra
 - c. Pooch-Taach Sambandhi Patra
 - d. Shikayati Sambandhi Patra
 - e. Naukari Sambandhi Patra
 - f. Pari patra
 - g. Agency Sambandhi Patra
- 3) Alekhan - Uttam Alekhan ke gun

- 4) Sampreshan - uske Prakar
- 5) Sankshiptikaran - Sankshipt Lekhan ke Pradhan gun aur udaaharan

Reference Books

- 1) Vyavasaik Sampreshan –Dr. Anup Chandra Bhayani Pb. Rajpal &sons, New Delhi-6
- 2) Karyalay Alekhan aur Tippan – Karnatak Mahila Hindi Seva Samithi, Chamarajpet, Bangalore.
- 3) Prashasanik Hindi – Ed. Ramdarash Mishra and amswaroop Shastry
- 4) Vanijya Patra Vyavahar, Anuvad, Nibandh tatha Sankshiptikaran – A V Narti, Gnanoday Prakashan , Dharwad.

DLD 05011 /12/13/14/15
CMD 05001/2/3/4/5/6/7/8

IV Semester

Hindi Kavya aur Anuvada Paribhashik Shabdavali

60hrs (4hrs/week)

3 Credits

Course out comes

After successful completion of this course students are able to

- CO 1. Write down the classification and characteristics of medieval and madran Hindi Kavya
- CO2. Deliberate in details with application, if applicable, medieval - saakhi by Kaber
- CO3. Specify in details with examples Hemala by Ramadhare Simha Dinakar
- CO4. Specify in details with application, if applicable, Gurukul by Ramkumar Varma
- CO5. Specify the characteristics of Hindi Anuvada Paribhasik Shabdavali
- CO6. Learn in details with examples Hindi Anuvada Paribhasik Shabdavali

UNIT A & B KAVYAGANGA (Prescribed Lessons-1, 2,5, 8, 11, 12, 13, 14, 16,19)

Kaber
Thulse Das
Bhihare Lal
Maithilisharan Gupta
Surya Kanth Tripathi Nirala
sumitranandan pant
Subhadra Kumari Chauhan
Ramkumar Verma
Ramdhari Simha Dinkar
Kuvempu

UNIT C Anuvad tatha Paribhashik Shabdavali

Portions Prescribed :

- 1) Anuvad ki Paribhashayen , Adhunik jagat mein Anuvad ka mahatva
Anuvadak kee Yogayathayen , Anuvad kala hai yaa viganan ?
- 2) Anuvad ke prakar:
 - a) Shabdanuvad
 - b) Bhavanuvad
 - c) Kavyanuvad

- d) Saranuvad
3) Angreji / Kannad se Hindi me Anuvad – vice versa (passages)
4) Paribhashik Shabdavali

Reference Books:

- 1) Anuvad Vignan – Bholanath Thiwari Pb. Shabdaker, Delhi -110092
- 2) Anuvad Kala Kuch Vichar –Anuvad Prakash Khemani , S.Chand & co. Delhi.
- 3) Anuvad Siddhant aur Samasyayen- R.N. Srivastava and K.K.Goswami
- 4) Alok Prakashan , Delhi Anuvad Patrika Ank, Pb. Anuvad Sahitya Parishad , New Delhi

III Semester Communicative Hindi

30hrs (2hrs/week)

2 Credits

Course out comes

After successful completion of this course students are able to

CO1. Learn the details of Communicative Hindi

CO2. Identify in depth Communicative Hindi

CO3. Deliberate in depth Communicative Hindi

UNIT I: HINDI GRAMMAR

- 1) Parts of speech – A general outlook
- 2) Gender -
 - a) Ling pehchanne ke niyam
 - c) Ling parivarthan ke niyam
- 3) Synonymns and Antonymns
 - a) Definition
 - b) Difference

UNIT II : GENERAL COMMUNICATION AND PRACTICE

- 4) Conversation - in market, hotel , journey etc.
- 5) Conversation on health , education etc.
- 6) Practice.

Reference

8. Saral Hindi Vyakaran –Pb. Prabhath Prakashan, # 4/19, Asaf Ali road, New Delhi-02
9. Parishkrith Hindi Vyakaran - Badarinath Kapur
10. Abhinav Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
11. Shikshaarathi Hindi Vyakaran - Prof. N.Nagappa, Pb. Rajpal &sons, New Delhi
12. Sankshipta Hindi Vyakaran - Kamatha prasad Guru, Nagari Pracharini Sabha, kaashi.
13. Anuvad Vignan - Bholanath Thiwari , Shabdakar , Delhi -92
14. Anuvad Kala – Kuch Vichar - Anandprakash Khemani , S.Chand & Co., Delhi
Anuvad siddhant aur Samasyayen

Model Question Paper

Language HINDI I ,II , & III SEMESTERS of BA / BSC

Time : 3 Hrs

Max. Marks : 70

- II) Total Main questions with internal choice on the prescribed text book 10 marks each
10 x 2 = 20
- II) One Annotation question – three out of five on the prescribed text book 05 marks
each 05 x 3
= 15
- III) Two short answer questions with internal choice on Vyakaran /prayojan moolak
Hindi / Anuvad as the case may be, 10 marks each 10 x 2 = 20
- IV) 15 objective type questions on Vyakaran /prayojan moolak Hindi / Anuvad as the
case may be, 01 mark each 01 x 15 = 15

MODEL QUESTION PAPER

- Qn.1. OR 01 x 10 = 10
- Qn.2.OR 01 x 10 = 10
- Qn. 3. Annotate three out of six 05 x 03 = 15
- Qn.4.OR..... 01 x 10 = 10
- Qn.5.OR..... 01 x 10 = 10
- Qn.6. Objective type questions 01 x 15 = 15

INTERNAL ASSESSMENT - 30 MARKS

III SEMESTER -PAPER III
Communicative Hindi

MODEL QUESTION PAPER

TIME : 2 HOURS

MAX. MARKS : 50

Instructions: All sections are compulsory

UNIT-I:Hindi Grammar

I . Answer The Following Questions in a sentence or a ward.

1).....2)..... 3).....4).....5).....

6).....7).....8).....9).....10).....

1 X 10 = 10

II . Answer The Following Questions

(Any four)

4 X 5 = 20

1).....2).....

3).....4).....

5).....6).....

UNIT II : GENERAL COMMUNICATION AND PRACTICE

2 X 10 = 20

I . Answer The Following Questions (any two)

1..... 2.....

3..... 4.....

INTERNAL ASSESSMENT - 30 MARKS ,F-1 -15,F-2-15
