# JSS COLLEGE OF ARTS, COMMERCE AND SCIENCE (AUTONOMOUS) OOTY ROAD, MYSORE – 25.



# **DEPARTMENT OF BIOCHEMISTRY**

SCHEMATIC SYLLABUS UNDER CHOICE BASED CREDIT SYSTEM (CBCS)

# CORE: FUNDAMENTALS OF CHEMISTRY AND MOLECULES OF LIFE (THEORY)

Unit:1	<b>THE FOUNDATIONS OF BIOCHEMISTRY:</b> Cellular and chemical foundations of life	2 Hrs
Unit: 2	<b>WATER:</b> Unique properties, weak interactions in aqueous systems, ionization of water, buffers, water as a reactant and fitness of the aqueous environment.	4 Hrs
Unit: 3	<b>CONCENTRATION UNITS:</b> Mole, Mole fraction, Molality, Molarity and Normality (problems to be worked out).	1 Hr
Unit: 4	<b>STEREOCHEMISTRY:</b> Stereoisomerism, types.Geometrical Isomerism- Cis Trans & E/Z nomenclature. Optical Isomerism-asymmetric carbon atom, Optical activity, plane polarized light, Chirality, Specfic molecular rotation, Projection formula- Fischer & Newman projection formulae, optical isomerism in Glyceraldehyde, Lactic acid, Tartaric acid. Nomenclature of enantiomers - D and L system, Recimisation and resolution (biochemical method)	4 Hrs
Unit: 5	<b>REACTION MECHANISM:</b> Concept of inductive, mesomeric and resonance.effect.Concept of the reaction intermediates-carbanions, carbocations, free radicals, carbenes. Nucleophiles and Electrophiles.	3 Hrs
Unit: 6	<b>CARBOHYDRATES &amp; GLYCOBIOLOGY:</b> Monosaccharides - structure of aldoses and ketoses, ring structure of sugars, conformations of sugars, mutarotation, anomers, epimers and enantiomers, structure of biologically important sugar derivatives, oxidation of sugars. Formation of disaccharides, reducing and nonreducing disaccharides. Polysaccharides — homo- and heteropolysaccharides, structural and storage polysaccharides. Structure and role of proteoglycans, glycoproteins and glycolipids (gangliosides and lipopolysaccharides). Carbohydrates as informational molecules, working with carbohydrates	12 Hrs
Unit:7	LIPIDS: Building blocks of lipids - fatty acids, glycerol, ceramide. Storage lipids - triacyl glycerol and waxes. Structural lipids in membranes- glycerophospholipids, galactolipids and sulpholipids, sphingolipids and sterols, structure, distribution and role of membrane lipids. Saponification value, Iodine value, Peroxide value, Acid value and their significance. Ecosinoids: Definition, types, Protaglandins: Definition, biological functions, structure of	12 Hrs

	DOEL O EQ. 1 DOEL O EQ.	
	PGE1 & E2 and PGF1&F2.  Threbovens and lowketrianes. Piological roles	
	Throboxans and leukotrienes- Biological roles. Plant steroids. Lipids as signals, cofactors and pigments	
	Train steroids. Espids as signals, coractors and pigments	
Unit	AMINO ACIDS & PROTEINS: Structure and classification, physical, chemical and optical properties of amino acids  Peptide bond - Formation and structure. Structure and biological importance of Glutathione, Oxytosine, Vasopressine, Endorphene. Synthetic peptides – importance of polyglutamic acid, polylysine.  Classification of proteins based on structure with examples. Primary structure of proteins – Determination of amino acid composition, determination of N- and C- terminal amino acids, sequencing by Edman's degradation method. Secondary structure: - a Helix, b- sheet, b barrel and b turn. Tertiary structure – Fibrousproteins (collagen) and Globular proteins (Myoglobin). Quaternary Structure – hemoglobin, Denaturation and renaturation of proteins by Anfinsen's experiment.	8 Hrs
Unit :	9 <b>NUCLEIC ACIDS:</b> Nucleotides - structure and properties. Nucleic acid structure – Watson-Crick model of DNA. Structure of major species of RNA - mRNA, tRNA and rRNA. Nucleic acid chemistry - UV absorption, effect of acid and alkali on DNA. Other functions of nucleotides - source of energy, component of coenzymes, second messengers.	8 Hrs
Unit	: 10 VITAMINS: Structure and active forms of water soluble and fat soluble vitamins, deficiency diseases and symptoms,	6 Hrs
	hypervitaminosis	
1 5	PRACTICALS	
2 I 3 I	PRACTICALS Safety measures in laboratories. Preparation of normal and molar solutions. Preparation of buffers.	
2 I 3 I 4 I	PRACTICALS Safety measures in laboratories. Preparation of normal and molar solutions. Preparation of buffers. Determination of pKa of acetic acid and glycine.	
2 II 3 II 4 II 5 (	PRACTICALS Safety measures in laboratories. Preparation of normal and molar solutions. Preparation of buffers. Determination of pKa of acetic acid and glycine. Qualitative tests for carbohydrates.	
2 II 3 II 4 II 5 (6	PRACTICALS Safety measures in laboratories. Preparation of normal and molar solutions. Preparation of buffers. Determination of pKa of acetic acid and glycine. Qualitative tests for carbohydrates. Qualitative tests for lipids.	
2 II 3 II 4 II 5 (6 6 (7	PRACTICALS Safety measures in laboratories. Preparation of normal and molar solutions. Preparation of buffers. Determination of pKa of acetic acid and glycine. Qualitative tests for carbohydrates. Qualitative tests for lipids. Qualitative tests for amino acids, proteins.	
2 II 3 II 4 II 5 0 6 0 7 0 8 0	PRACTICALS  Safety measures in laboratories.  Preparation of normal and molar solutions.  Preparation of buffers.  Determination of pKa of acetic acid and glycine.  Qualitative tests for carbohydrates.  Qualitative tests for lipids.  Qualitative tests for amino acids, proteins.  Qualitative tests for nucleic acids.	
2 II 3 II 4 II 5 (6 6 (7 8 (9) S	PRACTICALS  Safety measures in laboratories.  Preparation of normal and molar solutions.  Preparation of buffers.  Determination of pKa of acetic acid and glycine.  Qualitative tests for carbohydrates.  Qualitative tests for lipids.  Qualitative tests for amino acids, proteins.  Qualitative tests for nucleic acids.  Separation of amino acids/ sugars/ bases by thin layer chromatography.	
2 II 3 II 5 0 6 0 7 0 8 0 9 S 10 II 11 II	PRACTICALS Safety measures in laboratories. Preparation of normal and molar solutions. Preparation of buffers. Determination of pKa of acetic acid and glycine. Qualitative tests for carbohydrates. Qualitative tests for lipids. Qualitative tests for amino acids, proteins. Qualitative tests for nucleic acids. Separation of amino acids/ sugars/ bases by thin layer chromatography. Estimation of vitamin C by iodometric method Preparation of ZnSO <sub>4</sub> standard solution, standardization of aproximatly N/EDTA solution and estimation of total hardness of water using Erichrome	
2 II 3 II 4 II 5 0 6 0 7 0 8 0 9 S 10 II 11 II	PRACTICALS  Safety measures in laboratories.  Preparation of normal and molar solutions.  Preparation of buffers.  Determination of pKa of acetic acid and glycine.  Qualitative tests for carbohydrates.  Qualitative tests for lipids.  Qualitative tests for amino acids, proteins.  Qualitative tests for nucleic acids.  Separation of amino acids/ sugars/ bases by thin layer chromatography.  Estimation of vitamin C by iodometric method  Preparation of ZnSO <sub>4</sub> standard solution, standardization of aproximatly N/	black –

13	Determination of density the given protein solution or liquid using specific gravity
	bottle and viscosity by using ostwald's viscometer.
14	Conductometric titration of Amino acid against astrong base.

- **1.** Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H.Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-4292-3414-8.
- **2.** Textbook of Biochemistry with Clinical Correlations (2011) 7th ed., Devlin, T.M., JohnWiley & Sons, Inc. (New York), ISBN:978-0-470-28173-4.

# SEMESTER - II

# **CORE: PHYSIOLOGY (THEORY)**

Unit: 1	HOMEOSTASIS & THE ORGANIZATION OF BODY FLUID	8 Hr
	<b>COMPARTMENTS:</b> Intracellular, extracellular and interstitial fluid.	
	Homeostasis, control system and their components. Plasma as an	
	extracellular fluid, RBC, molecular mechanism of blood	
	coagulation, role of vitamin K in coagulation, anticoagulant and	
	fibrinolytic systems.	
TI 14 0	Anemias, polycythemia, haemophilia and thrombosis	0 TT .
Unit: 2	<b>CARDIOVASCULAR PHYSIOLOGY:</b> Anatomy of heart. Physiology	8 Hr
	of the cardiac muscle, Pressure, flow and resistance. Ventricular	
	volumes and the ECG, control of cardiac function and output. The	
	arterial system, venous system, the microcirculation and	
	mechanics of capillary fluid exchange. Control of blood flow to	
	the tissues. Portal circulations. Hypertension, congestive heart	
TI '4 2	disease, atherosclerosis and myocardial infarction.	
Unit: 3	<b>RESPIRATION:</b> Mechanism of respiration. Principles of gas	6Hr
	exchange and transport. Regulation of respiration. Pulmonary	
	oedema and regulation of pleural fluid. Hypoxia, hypercapnea,	
TT 1. 4	pulmonary distress, emphesema, ARDS.	<b>7</b> 11
Unit: 4	<b>RENAL PHYSIOLOGY:</b> Anatomy of the kidney and the nephron.	7Hr
	Regulation of renal blood flow. Cell biology of the Bowmans'	
	capsule. Physiology of glomerular filtration and GFR. Tubular	
	processing of the glomerular filtrate. Blood buffer systems, renal	
	and pulmonary control of blood pH, renal clearance. Assessment	
	of kidney function. Acidosis and alkalosis. Glomerular nephritis,	
	renal failure, dialysis and diuretics.	CTT
Unit: 5	GASTROINTESTINAL & HEPATIC PHYSIOLOGY: Propulsion and	6Hr
	motility of food and digested material digestion and absorption of	
	macro and micronutrients. Peptic ulcer, Sprue, celiac disease,	
	IBD, regurgitation, diarrhoea and constipation. Anatomy of the	
	hepatic lobule and blood flow into the liver. Formation and	
	secretion of bile. enterohepatic cycle, reticuloendothelial system,	
	metabolic importance of liver. Liver function tests. Jaundice, liver	
	cirrhosis and fatty liver.	
<b>T</b> T •4 6	MUSCULOSKETETAL SYSTEM: Bone structure and formation.	4Hr
Unit :6	Physiology of muscle contraction in striated and non-striated	4Hr
TT •4 =	muscle.	
Unit: 7	<b>REPRODUCTIVE PHYSIOLOGY:</b> Sex determination and	7Hr
	differentiation. Development of female and male genital tracts.	
	Spermatogenesis, capacitation and transport of sperm, blood testis	
	barrier. Ovarian function and its control. Uterine changes,	
	fertilization and implantation. Placenta as a feto- maternal unit,	
	gestation and parturition.	
Unit:8	<b>NEUROCHEMISTRY &amp; NEUROPHYSIOLOGY:</b> Central Nervous	6Hr

Unit:	system. Peripheral Nervous system. Blood brain barrier and CSF. Membrane potentials. Synaptic transmission. Neurotransmitters. Sensory receptors and neural pathways. Somatic sensation, EEG, sleep, coma, learning and memory.  PLANT RESPIRATION: Overview of glycolysis, Alternative reactions of glycolysis, Regulation of plant glycolysis, Translocation of metabolites across mitochondrial membrane, TCA cycle, Alternative NAD(P)H oxidative pathways; Cyanide	8Hr		
	resistant respiration.			
	PRACTICALS			
1 He	Hematology.			
2 a. I	a. RBC and WBC counting			
3 b. l	b. Differential leucocyte count.			
	c. Clotting time.			
5 Est	imation of haemoglobin.			
	paration of plasma proteins.			
<del></del>	termination of total iron binding capacity.			
	Pulmonary function tests, spirometry and measurement of blood pressure.			
	paration of isoenzymes of LDH by electrophoresis.			
<del>                                     </del>	stology of connective tissue, liver and/ brain permanent slides.			
<del>                                     </del>	se studies (Renal clearance, GFR, ECG).			
<del>                                     </del>	imation of serum phospholipids			
	monstration of ELISA			
<b>14</b> Ide	entification of PMN leucocytes from human peripheral blood sample by sta	aining		

- **1.** Vander's Human Physiology (2008) 11th ed., Widmaier, E.P., Raff, H. and Strang, K.T., McGraw Hill International Publications (New York), ISBN: 978-0-07-128366-3.
- 2. Harper's Biochemistry (2012) 29th ed., Murray, R.K., Granner, D.K., Mayes and
- P.A., Rodwell, V.W., Lange Medical Books/McGraw Hill. ISBN:978-0-07-176-576-3.
- **3.** Textbook of Medical Physiology (2011) 10th ed., Guyton, A.C. and Hall, J.E., Reed Elseviers India Pvt. Ltd. (New Delhi). ISBN: 978-1-4160-4574-8.
- **4.** Fundamental of Anatomy and Physiology (2009), 8th ed., Martini, F.H. and Nath, J.L., Pearson Publications (San Francisco), ISBN: 10:0-321-53910-9 / ISBN: 13: 978-0321-53910-6

# SEMESTER - III

**CORE: METABOLISM (THEORY)** 

Unit:1	<b>INTRODUCTION TO METABOLISM:</b> Anabolism and catabolism, compartmentalization of metabolic pathways.	2 Hr
Unit: 2	METABOLISM OF CARBOHYDRATES: Glycolysis -	10Hr
CIMU 1 Z	reactions, regulation and energetics. Entry of other	
	carbohydrates (Fructose, Galactose and Mannose) into	
	glycolytic pathway.	
	Fates of pyruvate – Under airobic and anaerobic conditions	
	(conversion of pyruvate to lactate, alcohol and acetyl coA	
Unit: 3	<b>CITRIC ACID CYCLE:</b> Reactions, regulation and energetics.	10Hr
	Amphibolic and integrating roles of TCA cycle.	
	Anaplerosis. Pentose Phosphate pathway and its	
	significance. cori cycle - its significance . Rapoport and	
	luebering cycle, Gluconeogenesis pathway & their	
	significance.	0.611
Unit: 4	Uronic acid pathway – Reactions and its significance.	06Hr
	glyoxalate pathway – Reactions and its significance.	
	Glycogen metabolism – glycogenolysis and glycogen	
TT:4. <i>E</i>	synthesis and its regulation	OTT
Unit: 5	<b>METABOLISM OF LIPID:</b> Oxidation of fatty acid - $\alpha$ , $\beta$ and	8Hr
	$\omega$ - oxidation of saturated fatty acids, Energetics of $\beta$ -	
	oxidation.	
	Biosynthesis of even number saturated fatty acids, ketone	
IInit. 6	bodies- formation and its significance.  Bio synthesis of triglycerides, glycolipids, phospholipids	8Hr
Unit: 6	and spingolipids	0111
	Cholesterol - Outline of biosynthesis and its degradation	
Unit:7	METABOLISM OF AMINO ACIDS: General reaction of	8Hr
Cint .7	amino acid degradation – Transamination, deamination and	OIII
	decarboxylation (oxidative and non oxidative). Ketogenic	
	and glucogenic amino acids. Urea cycle and its significance.	
	Biosynthesis of amino acids (Phenyl alanine and Glutamic	
	acid) and their degradation.	
Unit: 8	NUCLEIC ACID METABOLISM: Degradation of Nucleic	8Hr
	acids by DNase I & II, pancreatic RNase and	
	phosphodiesterases. Biosynthesis and degradartion of purine	
	and pyrimidine nucleotides, salwage pathways, regulation of	
	purine and pyrimidine synthesis.	
	PRACTICALS	
1 Glucose	by DNS method.	
2 Amino a	acid by Ninhydrine method.	
3 Protein	by Lowry's method.	

4	Urea by DAMO or nitroprusside method.
5	Creatinine by Jaffe's method.
6	Phosphorous by Fiske and Subbarow's method.
7	Iron by Wong's method.
8	Ketoacids by DNPH method
9	Glycogen by anthrone method
10	Cholesterol by Zak's method
11	Isolation of lecithin, identification by TLC, and its estimation.
12	Isolation of cholesterol from egg yolk and its estimation.
13	Assay of serum transaminases – SGOT and SGPT.

- **1.** Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H.Freeman and Company (New York), ISBN:13:978-1-4641-0962-1 / ISBN:10:1-4641-0962-1.
- **2.** Textbook of Biochemistry with Clinical Correlations (2011) 7th ed., Devlin, T.M., JohnWiley & Sons, Inc. (New York), ISBN: 978-0-470-28173-4 / BRV ISBN: 978-0-470-60152-5.

# **CORE:** ENZYMOLOGY & BIOENERGETICS (THEORY)

Unit: 1	<b>INTRODUCTION TO ENZYMES:</b> Nature of enzymes - protein	2 Hr
	and non-protein (ribozyme). Cofactor and prosthetic group,	
	apoenzyme, holoenzyme. IUBMB classification of enzymes.	
Unit: 2	i i	6Hr
	of chemical reactions, collision theory, activation energy and	
	transition state theory, catalysis, reaction rates and	
	thermodynamics of reaction. Catalytic power and specificity of	
	enzymes (concept of active site), Fischer's lock and key	
	hypothesis, Koshland's induced fit hypothesis.	
Unit: 3	<b>ENZYME KINETICS:</b> Relationship between initial velocity and	7Hr
Omt . 3	substrate concentration, steady state kinetics, equilibrium	/111
	constant - monosubstrate reactions. Michaelis-Menten equation,	
	Lineweaver-Burk plot, Eadie-Hofstee and Hanes plot. Km and	
	Vmax, Kcat and turnover number. Effect of pH, temperature	
T124 - A	and metal ions on the activity of enzyme.	6TT
Unit: 4	<b>ENZYME INHIBITION:</b> Reversible inhibition (competitive, propertitive, privad and substrate)	6Hr
	uncompetitive, non-competitive, mixed and substrate).	
TT 1. =	Mechanism based inhibitors - antibiotics as inhibitors	<b>7</b> 11
Unit:5	MECHANISM OF ACTION OF ENZYMES: General features -	5Hr
	proximity and orientation, strain and distortion, acid base and	
	covalent catalysis (chymotrypsin, lysozyme).	
Unit: 6	<b>REGULATION OF ENZYME ACTIVITY:</b> Control of activities of	6Hr
	single enzymes (end product inhibition) and metabolic	
	pathways, feedback inhibition (aspartate transcarbomoylase).	
	Occurrence and isolation, phylogenetic distribution and	
	properties (pyruvate dehydrogenase, fatty acyl synthase)	
	Isoenzymes - properties and physiological significance (lactate	
	dehydrogenase).	
<b>Unit: 7</b>	INVOLVEMENT OF COENZYMES IN ENZYME CATALYSED	6Hr
	<b>REACTIONS:</b> TPP, FAD, NAD, Pyridoxal Phosphate, Biotin,	
	Coenzyme A, Tetrahydrofolate, Lipoic Acid.	
Unit: 8	<b>APPLICATIONS OF ENZYMES:</b> Application of enzymes in	6`Hr
	diagnostics (SGPT, SGOT, creatine kinase, alkaline and acid	
	phosphatases), enzyme immunoassay (HRPO), enzyme therapy	
1	= = = = = = = = = = = = = = = = = = =	
	(Streptokinase). Immobilized enzymes.	
Unit:9	= = = = = = = = = = = = = = = = = = =	8Hr
Unit: 9	(Streptokinase). Immobilized enzymes.	8Hr
Unit:9	(Streptokinase). Immobilized enzymes.  INTRODUCTION TO BIOENERGETICS: Laws of	8Hr
Unit:9	(Streptokinase). Immobilized enzymes.  INTRODUCTION TO BIOENERGETICS: Laws of thermodynamics, state functions, equilibrium constant, coupled	8Hr
Unit:9	(Streptokinase). Immobilized enzymes.  INTRODUCTION TO BIOENERGETICS: Laws of thermodynamics, state functions, equilibrium constant, coupled reactions, energy charge, ATP cycle, phosphorylation potential,	8Hr
Unit:9	(Streptokinase). Immobilized enzymes.  INTRODUCTION TO BIOENERGETICS: Laws of thermodynamics, state functions, equilibrium constant, coupled reactions, energy charge, ATP cycle, phosphorylation potential, phosphoryl group transfers. Chemical basis of high standard	8Hr

		Daday magations, standard maday natartials and Namest agyetion	
		Redox reactions, standard redox potentials and Nernst equation. Universal electron carriers.	
<b>T</b> T	. 10		OTT
Un	it: 10		8Hr
transport chain - its organization and function. Inhibitors of			
ETC and uncouplers. Peter Mitchell's chemiosmotic hypothesis.			
Proton motive force. Fo F1ATP synthase, structure and			
mechanism of ATP synthesis. Metabolite transporters in			
mitochondria. Regulation of oxidative phosphorylation. ROS			
production and antioxidant mechanisms. Thermogenesis.			
Alternative respiratory pathways in plants.			
PRACTICALS			
1	Partial purification of acid phosphatase from germinating mung bean.		
2	Assay of enzyme activity and specific activity, e.g. acid phosphatase.		
3	Effect of pH on enzyme activity		
4	Determination of Km and Vmax using Lineweaver-Burk graph.		
5	Enzy	me inhibition - calculation of Ki for competitive inhibition.	
6	Continuous assay of lactate dehydrogenase.		
7	Coup	led assay of glucose-6-phosphate dehydrogenase	
8			
9			
10	Determination of Km and Vmax of Salivary amylase		
11	Deter	mination of initial velocity (time Kinetics) of Salivary amylase.	
12	Deter	mination of optimum temperature and energy of activation of Sa	ılivary
	amyla	ase.	
13	Separ	ation of photosynthetic pigments by TLC	

- **1.** Lehninger: Principles of Biochemistry (2013) 6th ed., Nelson, D.L. and Cox, M.M., W.H.Freeman and Company (New York), ISBN:13: 978-1-4641-0962-1 / ISBN:10:1-4292-3414-8.
- **2.** Biochemistry (2011) 4th ed., Donald, V. and Judith G.V., John Wiley & Sons Asia Pvt.Ltd. (New Jersey), ISBN:978-1180-25024.
- **3.** Fundamentals of Enzymology (1999) 3rd ed., Nicholas C.P. and Lewis S., Oxford University Press Inc. (New York), ISBN:0 19 850229 X.

# DSE -1: NUTRITIONAL BIOCHEMISTRY (THEORY)

Unit:1	INTRODUCTION TO NUTRITION & ENERGY METABOLISM	8Hr
	Defining Nutrition, role of nutrients. Unit of energy, Biological	
	oxidation of foodstuff. Measurement of energy content of food,	
	Physiological energy value of foods, SDA.	
	Measurement of energy expenditure. Direct and Indirect	
	Calorimetry, factors affecting thermogenesis, energy utilization by cells, energy output — Basal and Resting metabolism, physical	
	activity, factors affecting energy input - hunger, appetite, energy	
	balance Energy expenditure in man. Estimating energy	
	requirements, BMR factors Recommended Nutrient Intakes (RNI)	
	and Recommended Dietary Allowances for different age groups.	
Unit: 2	<b>DIETARY CARBOHYDRATES &amp; HEALTH:</b> Review functions of	8Hr
	carbohydrates. Digestion, absorption, utilization and storage,	
	hormonal regulation of blood glucose. Dietary requirements and	
	source of carbohydrates, Dietary fiber, role of fibre in lipid	
	metabolism, colon function, blood glucose level and GI tract	
II	functions.	OTT
Unit :3	<b>DIETARY LIPID &amp; HEALTH:</b> Review of classification, sources, functions digestion characters willigation and storage	8Hr
	functions, digestion, absorption, utilization and storage. Essential Fatty Acids; Functions of EFA, RDA, – excess and	
	deficiency of EFA. Lipotropic factors, role of saturated fat,	
	cholesterol, lipoprotein and triglycerides. Importance of the	
	following: a) Omega – fatty acids. Omega 3/ omega 6 ratio b)	
	Phospholipids c) Cholesterol in the body d) Mono,	
	Polyunsaturated and Saturated Fatty Acids. Dietary implications	
	of fats and oils, Combination ratios of n6 and n3, MUFA, PUFA	
	and SFA.	
Unit: 4	<b>DIETARY PROTEINS &amp; HEALTH:</b> Review of functions of proteins	8Hr
	in the body, Digestion and absorption. Essential and Nonessential	
	amino acids. Amino Acid Availability Antagonism, Toxicity and Imbalance, Amino acid Supplementation. Effects of deficiency.	
	Food source and Recommended Dietary Allowances for different	
	age group. Amino acid pool. NPU, Biological Value, Nitrogen	
	balance. PEM and Kwashiorkor.	
Unit :5	FAT & WATER SOLUBLE VITAMINS: Vitamin A, C, E, K and D	8Hr
	Dietary sources, RDA, Adsorption, Distribution, Metabolism and	
	excretion(ADME), Deficiency. Role of Vitamin A as an	
	antioxidant, in Visual cycle, dermatology and immunity. Role of	
	Vitamin K in Gamma carboxylation. Role of Vitamin E as an	
	antioxidant. Extra-skeletal role of Vitamin D and its effect on	
	bone physiology.  Hyporvitaminosis Vitamin C role as cofeator in amino acid	
	Hypervitaminosis. Vitamin C role as cofactor in amino acid	

		modifications. Niacin- Metabolic interrelation between tryptophan, Niacin and NAD/ NADP. Vitamin B6-Dietary source,	
		RDA, conversion to Pyridoxal Phosphate. Role in metabolism, Biochemical basis for deficiency symptoms. Vitamin B12 and folate; Dietary source, RDA, absorption, metabolic role	
		Biochemical basis for deficiency symptoms.	
Uı	nit: 6	MINERALS: Calcium, Phosphorus and Iron - Distribution in the	12Hr
		body digestion, Absorption, Utilization, Transport, Excretion, Balance, Deficiency, Toxicity, Sources, RDA. Calcium:	
		Phosphorus ratio, Role of iron in prevention of anemia. Iodine and	
		iodine cycle. Iodine, Fluoride, Mg, Cu, Zn, Se, Manganese,	
		Chromium, Molybdenum Distribution in the human body,	
		Physiology, Function, deficiency, Toxicity and Sources.	
Uı	nit: 7	ASSESSMENT OF NUTRITIONAL STATUS: Anthropometric	4Hr
		measurements; Z scores, BMI, skinfold, circumference ratios.	
		Biochemical assessment; Basal metabolic panel, Comprehensive	
		metabolic panel, CBC, Urine Analysis, Assessment of Anemia,	
		ROS assessment, GTT and glycosylated Hb, Differential diagnosis	
		of B12 and folate.	
Uı	nit: 8	FOOD & DRUG INTERACTIONS & NUTRICEUTICALS: Nutrient	4Hr
		interactions affecting ADME of drugs, Alcohol and nutrient	
		deficiency, Antidepressants, psychoactive drugs and nutrient	
		interactions, Appetite changes with drug intakes and malnutrition.	
		Food as medicine.	
		PRACTICALS	
1		ay for vitamin B12/B1.	
2		cystiene estimation.	
3		/ urine MMA estimation.	
4		ppometric identifications for Kwashiorkor, Marasmus and Obesity.	
5		nination of oxidative stress: TBARS, antioxidant enzymes in hemolys	ate.
6		in A/E estimation in serum.	• `
7		lensitometry /bone ultrasound test demonstration (visit to a nearby cli	
8	Proximate analysis of food samples- Moisture, fibre, protein fat and carbohydrate (by difference) (3 practicals)		
9	Detect	ion of adulterants in food.	
10	Estima	tion of Calcium in ragi.	
11		tion of Vitamin – C in lemon or gooseberries by DPPH method	
12		nation of Lactose in milk by Benedict's method	
13			
	liver o		

- **1.** Textbook of Biochemistry with Clinical Correlations (2011) Devlin, T.M. John Wiley & Sons, Inc. (New York), ISBN: 978-0-4710-28173-4.
- **2.** Nutrition for health, fitness and sport (2013); Williams.M.H,Anderson,D.E, Rawson,E.S. McGraw Hill international edition. ISBN-978-0-07-131816-7.
- **3.** Krause's Food and Nutrition Care process.(2012); Mahan, L.K Strings, S.E, Raymond, J. Elsevier's Publications. ISBN-978-1-4377-2233-8.
- **4.** The vitamins, Fundamental aspects in Nutrition and Health (2008); G.F. Coombs Jr.Elsevier's Publications. ISBN-13- 978-0-12- 183493-7.
- 5. Principles of Nutritional Assessment (2005) Rosalind Gibson. Oxford University Press.

# DSE -2: MOLECULAR BASIS OF INFECTIOUS DISEASES

Uni	it:1	CLASSIFICATION OF INFECTIOUS AGENTS: Bacteria,	12Hr
		Viruses, protozoa and fungi. Past and present emerging and re-	
		emerging infectious diseases and pathogens. Source, reservoir	
		and transmission of pathogens, Antigenic shift and antigenic	
		drift. Host parasite relationship, types of infections associated	
		with parasitic organisms. Overview of viral and bacterial	
<b>W</b> T	• 4 • 2	pathogenesis. Infection and evasion	10TT .
Un	it: 2	OVERVIEW OF DISEASES CAUSED BY BACTERIA: Detailed	18Hr
		study of tuberculosis: History, causative agent, molecular	
		basis of host specificity, infection and pathogenicity,	
		Diagnostics, Therapeutics, inhibitors and vaccines. Drug	
		resistance and implications on public health. Other bacterial	
		diseases including Typhoid, Diphtheria, Pertussis, Tetanus, Typhoid and Pneumonia.	
IIn	it: 3	OVERVIEW OF DISEASES CAUSED BY VIRUSES: Detailed	12Hr
	It. J	study of AIDS, history, causative agent, pathogenesis,	12111
		Diagnostics, Drugs and inhibitors. Other viral diseases	
		including hepatitis, influenza, rabies, chikungunya and polio.	
Un	it: 4	OVERVIEW OF DISEASES CAUSED BY PARASITES:	8Hr
	100 1	Detailed study of Malaria, history, causative agents, Vectors,	0222
		life cycle, Host parasite interactions, Diagnostics, Drugs and	
		Inhibitors, Resistance, Vaccine development. Other diseases	
		including leishmaniasis, amoebiasis.	
Un	it :5	<b>OVERVIEW OF DISEASES CAUSED BY OTHER ORGANISMS:</b>	10Hr
		Fungal diseases, General characteristics. Medical importance	
		of major groups, pathogenesis, treatment.	
		PRACTICALS	
1	WID	OAL test	
2		n staining	
3		I fast staining	
4		based diagnosis	
5		no acid by Ninhydrine method.	
6		ein by Lowry's method.	
7	Urea by DAMO or nitroprusside method.		
8	Creatinine by Jaffe's method.		
9		sphorous by Fiske and Subbarow's method.	
10		by Wong's method.	
11		oacids by DNPH method	
12		monia by nitroprusside method	
13		cogen by anthrone method	
14	Cho	lesterol by Zak's method	

- **1.**Prescott, Harley, Klein's Microbiology (2008) 7th Ed., Willey, J.M., Sherwood, L.M., Woolverton, C.J. Mc Graw Hill International Edition (New York) ISBN: 978-007-126727.
- **2.** Mandell, Douglas and Bennett.S, Principles and practices of Infectious diseases, 7<sup>th</sup> edition, Volume, 2. Churchill Livingstone Elsevier.
- **3.** Sherris Medical Microbiology: An Introduction to Infectious Diseases by Kenneth J.Ryan, C. George Ray, Publisher: McGraw-Hill
- **4.** Medical Microbiology by Patrick R. Murray, Ken S. Rosenthal, Michael A. Pfaller, Elsevier Health Sciences

#### **SEC-1: TOOLS AND TECHNIQUES IN BIOCHEMISTRY**

TOTAL HOURS: 30 CREDITS: 2

Unit: 1 Unit: 2	BIOCHEMICAL REAGENTS & SOLUTIONS: Safety practices in the laboratory. Preparation and storage of solutions. Concepts of solution concentration and storing solutions. Quantitative transfer of liquids. Concept of a buffer, Henderson-Hasselbach equation, working of a pH meter.  Exercise  • Preparation of a buffer of given pH and molarity.  Chromatography- Definition, types, Principles of Adsorption and Partition chromatography. Techniques of circular, 2D chromatography, Thin Layer Chromatography-	7Hr 6 Hrs
Unit: 3	and its advantages  Column chromatography – Principle and applications of Gel  Filtration chromatography, HPLC and GLC  Electrophoresis: Principle and applications of electrophoresis technique- PAGE, SDS - PAGE	3 Hrs
Unit: 4	<b>Centrifugation :</b> Principle of differential and density gradient centrifugation. Ultra centrifuge – construction and applications	3 Hrs
Unit: 5	<ul> <li>SPECTROPHOTOMETRIC TECHNIQUES: Principle and instrumentation of UV-visible and fluorescence spectroscopy.</li> <li>Exercises</li> <li>Determination of the absorption maxima and molar extinction coefficient (of a relevant organic molecule).</li> <li>Measurement of fluorescence spectrum.</li> <li>Determination of concentration of a protein solution by Lowry/BCA method.</li> </ul>	5Hr
Unit :6	Introduction and importance of virtual labs in biochemistry	6Hr

- **1.** Physical Biochemistry: Principles and Applications (2010) 2nd ed., Sheehan, D., Wiley Blackwell (West Sussex), ISBN:978-0-470-85602-4 / ISBN:978-0-470-85603-
- **2.** Physical Biochemistry: Applications to Biochemistry and Molecular Biology (1982) 2<sup>nd</sup> ed., Freifelder, D., W.H. Freeman and Company (New York), ISBN:0-7167-1315-2 /ISBN:0-7167-1444-2.
- **3.** An Introduction to Practical Biochemistry (1998) 3rd ed., Plummer D. T., Tata McGraw Hill Education Pvt. Ltd. (New Delhi), ISBN:13: 978-0-07-099487-4 / ISBN:10: 0-07-099487-0.

# **SEC-2: CLINICAL BIOCHEMISTRY**

TOTAL HOURS: 30 CREDITS: 2

Unit:1	INTRODUCTION: Organization of clinical laboratory, Introduction to instrumentation and automation in clinical biochemistry laboratories safety regulations and first aid. General comments on specimen collection, types of specimen for biochemical analysis. Precision, accuracy, quality control, precautions and limitations.  Exercises  Collection of blood and storage.  Separation and storage of serum.	4Hr
Unit :2	EVALUATION OF BIOCHEMICAL CHANGES IN DISEASES:  Basic hepatic, renal and cardiovascular physiology. Biochemical symptoms associated with disease and their evaluation.	4Hr
	Diagnostic biochemical profile.	
Unit: 3	ASSESSMENT OF GLUCOSE METABOLISM IN BLOOD: Clinical significance of variations in blood glucose. Diabetes mellitus.  Exercises  • Estimation of blood glucose by glucose oxidase peroxidase	4Hr
	method.	
Unit :4	LIPID PROFILE: Composition and functions of lipoproteins. Clinical significance of elevated lipoprotein. Exercises  • Estimation of triglycerides.	4Hr
Unit :5	LIVER FUNCTION TESTS  Exercises  • Estimation of bilirubin (direct and indirect).	4Hr
Unit: 6	RENAL FUNCTION TESTS & URINE ANALYSIS: Use of urine strip / dipstick method for urine analysis.  Exercises  • Quantitative determination of serum creatinine and urea.	6Hr
Unit: 7	<b>TESTS FOR CARDIOVASCULAR DISEASES:</b> Involvement of enzymes in diagnostics of heart disease including aspartate transaminase, isoenzymes of creatine kinase and lactate dehydrogenase and troponin. <b>Exercises</b> • Estimation of creatine kinase MB.	4Hr

- **1.** Medical Laboratory Technology a Procedure Manual for Routine Diagnostic Tests Vol.I (2010), Mukherjee, K.L., Tata Mc Graw–Hill Publishing Company Limited (New Delhi). ISBN: 9780070076594 / ISBN: 9780070076631
- **2**. Medical Laboratory Technology a Procedure Manual for Routine Diagnostic Tests VoI. II (2010), Mukherjee, K.L., Tata Mc Graw Hill Publishing Company Ltd. (New Delhi), ISBN: 9780070076648.
- **3.** Medical Biochemistry (2005) 2nd ed., Baynes, J.W. and Dominiczak, M.H., Elsevier Mosby Ltd. (Philadelphia), ISBN:0-7234-3341-0.
- **4.** Experimental Biochemistry: A Student Companion (2005) Rao, B.S. and Deshpande, V., IK International Pvt. Ltd. (New Delhi), ISBN: 81-88237-41-8.

# DSE-1: ADVANCED CELL BIOLOGY AND ENDOCRINOLOGY (THEORY)

Uni	t: 1	PLASMA MEMBRANE & NUCLEAR TRANSPORT: Properties and	8Hr
		Composition of Cell Membrane; Structure of Nuclear Envelope;	
		Nuclear Pore Complex; Transport Across Nuclear Envelope;	
		Regulation of Nuclear Protein Import and Export.	
Ur	nit: 2	<b>CELL-CELL INTERACTION:</b> Cell-Cell Interactions and Cell-	10Hr
		Matrix Interactions; Components of Extracellular Matrix:	
		Collagen and Non-Collagen Components; Tight Junctions; Gap	
		Junctions; Desmosomes; Hemidesmosomes; Focal Adhesions	
		And Plasmodesmata; Cell Wall; Role Of Cell Interaction In	
		Development.	
Ur	it :3	<b>CELL CYCLE &amp; PROGRAMMED CELL DEATH:</b> Overview of The	12Hr
		Cell Cycle; Eukaryotic Cell Cycle; Events Of Mitotic Phase;	
		Cytokinesis; Events Of Meiosis And Fertilization; Regulation Of	
		Cell Division And Cell Growth; Apoptosis And Necrosis, Stem	
		Cells And Maintenance of Adult Tissues, Hematopoiesis,	
		Embryonic Stem Cells and Therapeutic Cloning.	
Ur	it :4	<b>CANCER BIOLOGY:</b> Development and causes Of Cancer;	10Hr
		Genetic Basis of Cancer; Oncogenes, Tumor Viruses; Molecular	
	•	Approach to Cancer Treatment.	1011
Ur	nit: 5	ADVANCED METHODS IN CELL BIOLOGY: Ultracentrifugation,	10Hr
		Fluorescence Microscopy- FACS, Confocal Microscopy,	
		Electron Microscopy, Plant and Animal Cell Culture,	
<b>T</b> T	nit: 6	Immunohistochemistry. <b>ENDOCRINE SYSTEM</b> : Endocrine organs, hormones-	10Hrs
UI	ш: о	$\mathcal{S}$	101118
		classification, Hierarchy, interplay, dynamic balance and regulation of their secretions. Functions of the hormones of	
		Hypothalamus, Pituitary, Adrenal, Thyroid, pancreas and	
		Gonads.Concept of receptors- Membrane and cytosolic.	
		Mechanism of hormone action - Steroid hormone and Peptide	
		hormone (second messengers hypothesis. Eg: cAMP, DAG,	
		IP3).	
		H 3).	
		PRACTICALS	
1		ion of organelles by sub-cellular fractionation.	
2		of cell viability /death assay by use of trypan blue and MTT assay.	
3	•	of apoptosis through analysis of DNA fragmentation patterns of the property of	erns in
_		hondria.	
4		fication and study of cancerous cells using permanent slid	es and
5		micrographs. ion of organelles by sub-cellular fractionation.	
	T 1 1		

6	Cell counting methods: counting using Haemocytometer.	
7	Calibration of ocular micrometer and Measurement of average cell size using	
	Stage micrometer	
8	Separation of cell organelles by differential centrifugation and assay of marker	
	enzymes. (2 Practicals)	
9	Study of different types of cells.	
10		
10	Isolation of chloroplast by differential centrifugation & its identification.	
11	Buccal smear- barr bodies	
<b>12</b>	Identification of normal & abnormal karyotype	

- **1.**Cooper, G.M. and Hausman, R.E. 2009 The Cell: A Molecular Approach. 5th edition ASM Press & Sunderland, Washington, D.C.; Sinauer Associates, MA.
- **2.** Karp, G. 2010 Cell and Molecular Biology: Concepts and Experiments. 6 edition. JohnWiley & Sons. Inc.
- **3.** Alberts, B., Johnson, A., Lewis, J., and Enlarge, M. 2008 Molecular Biology of the Cell.5th ed., Garland Science (Princeton),
- **4.** Lodish, H., Berk, A., Zipursky, S.L., Matsudaira, P., Baltimore, D. and Darnell. J. 2012. Molecular Cell Biology. 7th ed., W.H. Freeman & Company (New York),
- **5.** Becker, W.M., Kleinsmith, L.J., Hardin. J. and Bertoni, G. P. 2009 The World of the Cell. 7th edition. Pearson Benjamin Cummings Publishing, San Francisco.

# **DSE-2: PLANT BIOCHEMISTRY (THEORY)**

Unit: 1	INTRODUCTION TO PLANT CELL STRUCTURE: Plasma	5Hr
	membrane, Vacuole and tonoplast membrane, cell wall, plastids	
	and peroxisomes.	
Unit:2	PHOTOSYNTHESIS & CARBON ASSIMILATION: Structure of	14Hr
	PSI and PSII complexes, Light reaction, Cyclic and non cyclic	
	photophosphorylation, Calvin cycle and regulation; C4 cycle	
	and Crassulacean acid metabolism (CAM), Photorespiration.	
Unit:3	<b>NITROGEN METABOLISM:</b> Biological Nitrogen fixation by free	14Hr
	living and in symbiotic association, structure and function of	
	enzyme Nitrogenase. Nitrate assimilation: Nitrate and Nitrite	
	reductase. Primary and secondary ammonia assimilation in	
	plants; ammonia assimilation by Glutamine synthetase-	
	glutamine oxoglutarate amino transferase (GS-GOGAT)	
	pathway. Seed storage proteins in legumes and cereals.	
Unit :4	REGULATION OF PLANT GROWTH: Introduction to plant	7Hr
	hormones and their effect on plant growth and development,	
	Regulation of plant morphogenetic processes by light.	4 4 4 4 4
Unit :5	<b>SECONDARY METABOLITES:</b> Representatives alkaloid group	12Hr
	and their amino acid precursors, function of alkaloids,	
	Examples of major phenolic groups; simple phenylpropanoids,	
	Coumarins, Benzoic acid derivatives, flavonoids, tannins and	
	lignin, biological role of plant phenolics, Classification of	
	terpenoids and representative examples from each class,	
Timit of	biological functions of terpenoids.	8Hr
Unit :6	<b>PLANT TISSUE CULTURE:</b> Cell and tissue culture techniques, types of cultures: organ and explants culture, callus culture, cell	om
	suspension culture and protoplast culture. Plant regeneration	
	pathways: organogenesis and somatic embryogenesis.	
	Applications of cell and tissue culture and somoclonal	
	variation.	
Unit: 7	Yununon.	
	PRACTICALS	
1 Induc	tion of hydrolytic enzymes proteinases /amylases/lipase during ger	mination
	ction and assay of Urease from Jack bean	
	ation of carotene/ascorbic acid/phenols/tannins in fruits and vegeta	bles
	ation of photosynthetic pigments by TLC	
	are of plant plants (explants).	
	1 L (L).	

# **SUGGESTED BOOKS**

**1.** Plant Biochemistry (2008), Caroline Bowsher, Martin steer, Alyson Tobin, Garland science ISBN 978-0-8153-4121-5

<ol> <li>Biochemistry and molecular Biology of plant-Buchanan. (2005) 1 edition. Publisher: IK International. ISBN-10: 8188237116, ISBN-13: 978-8188237111.</li> <li>Plant Biochemistry by P.M Dey and J.B. Harborne (Editors) (1997) Publisher: Academic Press ISBN-10:0122146743, ISBN-13:978-0122146749.</li> </ol>		